



Research and Development

COMBUSTION MODIFICATION CONTROLS
FOR STATIONARY GAS TURBINE
Volume II. Utility Unit Field Test

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COMBUSTION MODIFICATION CONTROLS FOR STATIONARY GAS TURBINES
VOLUME II. UTILITY UNIT FIELD TEST

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ABSTRACT

This test report describes the methods and results of an environmental assessment test program conducted at Houston Lighting and Power's T. H. Wharton Generating Station, Unit 52. The purpose of the test program was to measure changes in emissions as a result of applying NO_x controls. Emissions of trace elements, organic materials, sulfur species, and the criteria pollutants, SO_2 , NO_x , CO, and particulate matter, were measured. Comparisons of these emissions under normal operating conditions and controlled (for NO_x) operating conditions were then made. Source operating data were also analyzed so that changes in operating parameters and efficiency could be assessed.

Unit 52 is a General Electric MS 7001C simple-cycle, single-shaft, heavy duty gas turbine rated at 70.8 MW nominal electrical output. This gas turbine may use either natural gas or distillate oil fuels. The test program was conducted using oil fuel.

Water injection was used for NO_x control. A water-to-fuel ratio of 0.42 resulted in a 58 percent reduction in NO_x from baseline levels. Changes in other emissions were within the limits of the analyses.

Operating efficiency decreased with water injection. The unit heat rate showed approximately 2 percent change in going from baseline to controlled (for NO_x) operation.

The test program concludes that using water injection for NO_x control in this unit reduced NO_x and showed little effect on other emissions. Water injection implementation did reduce operating efficiency.

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SECTION 1

INTRODUCTION

This report is part of a series of test reports resulting from the experimental testing task of the "Environmental Assessment of Stationary Source NO_x Control Technologies" Program (NO_x EA), being performed under Environmental Protection Agency (EPA) contract 68-02-2160. The NO_x EA is a 3-year program to: (1) identify the multimedia environmental impact of stationary combustion sources and combustion modification NO_x controls; and (2) identify the most cost-effective environmentally-sound NO_x controls for attaining and maintaining current and projected NO₂ air quality standards to the year 2000.

During the first year of the NO_x EA a preliminary environmental assessment (Reference 1) concluded that emissions and operating data needed to perform adequate process engineering and environmental assessment activities were severely lacking in several key areas. Most noteworthy was the virtual absence of data on noncriteria flue gas emissions and liquid and solid effluents. In response to these identified data needs, seven field test programs were initiated. Source selection was based on a source/control priority listing developed in the preliminary environmental assessment. These test programs were designed to provide information on changes in emissions and operation due to NO_x controls. The NO_x EA Field Test Program is outlined in Table 1-1.

The test program documented in this report was conducted on Unit 52 of the T. H. Wharton Generating Station of the Houston Lighting and Power Company in Houston, Texas from April 21-24, 1978. Unit 52 was selected because its design is typical of large scale simple cycle utility gas turbines equipped with water injection and because of the possibility of collaborating with the engine manufacturer in detailed process evaluation tests. Unit 52 is a General Electric Model MS 7001C simple-cycle, single-shaft, heavy duty gas turbine rated at 70.8 MW nominal electrical

TABLE 1-1. NO_x EA FIELD TEST PROGRAM

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Coal-fired Utility Boiler	Kingston #6; 180 MW tangential; twin furnace, 12 burners/ furnace, 3 elevations; cyclone, 2 ESP's for particulate control	Baseline Biased Firing (2) BOOS (2)	Continuous NO _x , SO ₂ , CO, CO ₂ , O ₂ Inlet to 1st ESP: -- SASS -- Method 5 -- Method 8 -- Gas grab (C ₁ -C ₆ HC) Outlet of 1st ESP: -- SASS -- Method 5 -- Method 8 -- Gas grab (C ₁ -C ₆ HC) Bottom ash Hopper ash (1st ESP, cyclone) Fuel Operating data	TVA	Complete, August 1977
Coal-fired Utility Boiler	Crist #7, 500 MW opposed wall fired; 24 burners, 3 elevations; ESP for particulate control	Baseline BOOS (2)	Continuous NO _x , CO CO ₂ , O ₂ ESP inlet -- SASS -- Method 5 -- Method 8 -- Gas grab (C ₁ -C ₆ HC) ESP outlet -- SASS -- Method 5 -- Method 8 -- Gas grab (C ₁ -C ₆ HC) Bottom ash ESP hopper ash Fuel Operating data Bioassay	Exxon	Complete, June 1978
Oil-fired Utility Boiler	Moss Landing #6, 740 MW opposed wall fired; 48 burners, 6 elevations	Baseline FGR FGR + OFA	Continuous NO _x , CO, CO ₂ , O ₂ Flue gas -- SASS -- Method 5 -- Method 8 -- Gas grab (C ₁ -C ₆ HC) Fuel Operating data Bioassay	New test start	Complete, September 1978

TABLE 1-1. Continued

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Coal-fired Industrial Boiler	Traveling grate spreader stoker, 38 kg/s (300,000 lb/hr); ESP for particulate control; wet scrubber for SO _x control	Baseline LEA + high OFA	Continuous NO _x , CO, CO ₂ , O ₂ Boiler exit: -- SASS -- Method 5 -- Shell-Emeryville -- Gas grab (C ₁ -C ₆ HC) ESP outlet: -- SASS -- Method 5 -- Shell-Emeryville -- Gas grab (C ₁ -C ₆ HC) Bottom ash Cyclone hopper ash Fuel Operating data	KVB	Complete, October 1977
Coal-fired Industrial Boiler	Traveling grate spreader stoker, 25 kg/s (200,000 lb/hr) ESP for particulate	Baseline LEA + High OFA	Continuous NO _x , CO, CO ₂ , O ₂ Boiler exit: -- SASS -- Method 5 -- Shell-Emeryville -- Gas grab (C ₁ -C ₆ HC) ESP Outlet: -- SASS -- Method 5 -- Shell-Emeryville -- Gas grab (C ₁ -C ₆ HC) Bottom ash ESP hopper ash Fuel Operating data Bioassay	KVB	Complete, February 1978
Oil-fired Gas Turbine	T.H. Wharton Station, 60 MW GE MS 7001 C machine	Baseline water injection to meet proposed NSPS	Continuous NO _x , CO, CO ₂ , O ₂ Exhaust gas: -- SASS -- Method 5 -- Method 8 Fuel Water Operating data	General Electric	Complete, April 1978

TABLE 1-1. Concluded

Source Category	Description	Test Points (Unit Operation)	Sampling Protocol	Test Collaborator	Status
Oil-fired Residential Heating Unit	Blue Ray low NO _x furnace, Medford, New York	Continuous Cycling	Continuous NO _x , CO, CO ₂ , O ₂ Flue gas: -- SASS -- Method 5 -- Method 8 Fuel	New test start with EPA/IERL-RTP	Complete, November 1977

EE-073

output. Through the cooperation of the Houston Lighting and Power Company and the General Electric Company, this unit was made available for testing in the NO_x EA Field Test Program.

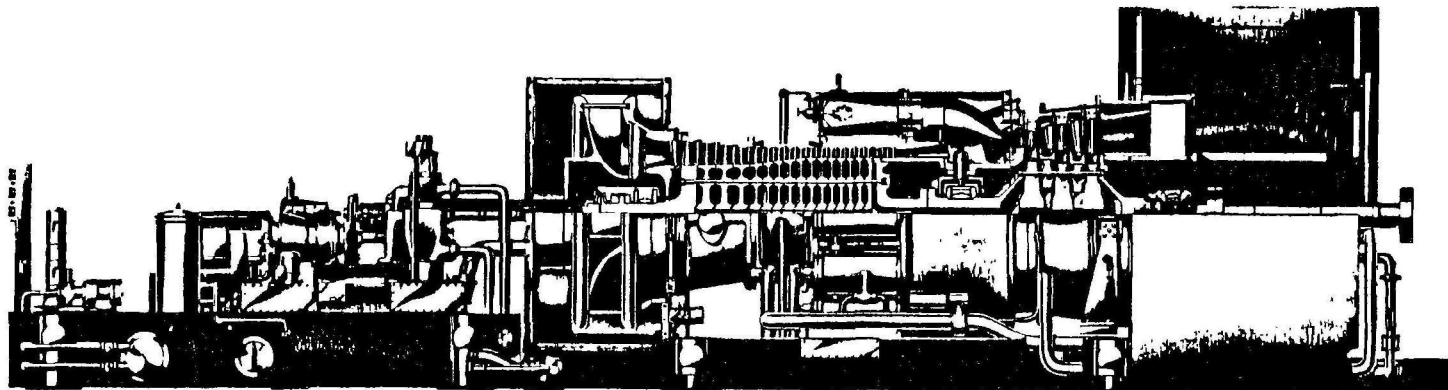
The test program at the T. H. Wharton Station consisted of a baseline (normal operation) test and a test with water injection being used for NO_x control. The test program results will be used in both process analysis and source assessment modeling, conducted as part of the Environmental Assessment and Process Engineering Task of the NO_x EA.

SECTION 2

PLANT DESCRIPTION

The field tests were conducted on Unit 52 of the T. H. Wharton Generating Station of the Houston Lighting and Power Company in Houston, Texas. Unit 52 is a General Electric Model MS 7001C (Figure 2-1) simple-cycle, single-shaft, heavy-duty stationary gas turbine rated at 70.8 MW nominal electrical output and is one of six such units at the Wharton Station. The Station also has eight GE 7001B combined cycle units, one Westinghouse 15 MW unit and two conventional steam boilers producing a total rated electrical output of 280 MW.

Unit 52 is fired with No. 2 distillate fuel oil with 0.11 percent sulfur by weight and approximately 46.054×10^3 kJ/kg (19,800 Btu/lb) heat content. Table 2-1 lists the rated operating parameters of the unit. There is no flue gas cleaning equipment on a turbine of this type due to the clean fuel used and the unit's inherent efficient combustion. Unit 52 is, however, equipped with a water injection system used to control the formation of NO_x within the combustion chambers. NO_x formation is repressed when atomized water is injected directly into the primary zone of the combustor resulting in reduced flame temperatures. The degree of NO_x control is adjusted by altering the quantity of water injected -- the more water injected the greater the degree of control. The first test on Unit 52, a baseline test, was run with no water injected. The second test was run while 2.52 l/sec (40 gpm) water was being injected. This corresponds to a water to fuel mass ratio of approximately 0.42, a ratio sufficiently high to bring NO_x emissions to within 75 ppm at 15 percent O_2 which is the level of the proposed New Source Performance Standards.



TC-7509A

Figure 2-1. Model series 7001 simple-cycle, single-shaft heavy-duty gas turbine.

TABLE 2-1. UNIT 52 RATED OPERATING PARAMETERS

Output power	70.8 MW
Overall pressure ratio	10.5
Heat rate	11.44 MJ/kWh (10,847 Btu/kWh)
Air flow	268 kg/sec (592 lb/sec)
Fuel flow	5.2 kg/sec (11.5 lb/sec)

SECTION 3

SAMPLING AND ANALYSIS METHODOLOGIES

The sampling and analysis procedures used in the test program closely follow the procedures recommended in the IERL-RTP Level 1 Environmental Assessment Procedures Manual (Reference 2). The following subsections will contain notations of where the procedures differ significantly from the standard methods. Level 1 testing, according to EPA's phased environmental assessment approach, is for screening purposes. Through chemical and biological tests potential problem areas and needs for further analysis are identified. Furthermore, Level 1 testing provides the basis for setting priorities for discharge streams, components, and classes of materials for further consideration in an overall environmental assessment. Thus, the results of the sampling and analysis procedures used in Level 1 are semiquantitative, yielding an accuracy factor of ± 2 to 3.

All analyses for trace elements, organic species, particulates and sulfur species in the Method 5/8 and SASS trains and water samples were performed in the Acurex analytical laboratory. Commercial Testing and Engineering Company analyzed the fuels and the bioassay analyses were performed by Litton Bionetics, Incorporated.

3.1 SAMPLING PROTOCOL

In order to effectively evaluate how emissions of compounds and pollutant species are affected by the use of water injection, all influent and effluent streams must be characterized during the baseline and water injection tests. The following streams on Unit 52 were sampled:

- Water feed (water injection system)
- Fuel feed
- Exhaust gas

Ambient air was not sampled. Descriptions of the specific sampling methods are given in the following paragraphs. Figure 3-1 shows the duct configuration and the location of the sampling ports.

3.1.1 Feed Streams

Water Feed

Samples of the demineralized feed water from the water injection purification system were periodically sampled throughout the five hour duration of the NO_x control test. Samples were tapped off the inlet lines preceeding the combustor section and then composited into one integrated sample for each run.

Fuel Feed

Fuel oil samples were obtained for both tests. Samples were tapped off the fuel inlet lines, collected throughout the test period and finally composited into one integrated sample for each test. Sampling of the fuel feed commenced one hour into the test run, then approximately once for each 90 minute period throughout the test.

3.1.2 Flue Gas

The flue gas was monitored on a continuous basis during both test runs for O_2 , CO_2 , NO , total NO_x , total unburned hydrocarbons and CO . The continuous monitoring was provided by General Electric (GE) personnel and equipment. Table 3-1 lists the instrumentation used by General Electric. All sample lines were of Teflon construction and heated to 450 K (350°F) to assure the integrity of all sampled species. The sample flow was filtered to remove particulate matter and then split into two streams. One stream supplied the nitrogen oxides instrument and the total hydrocarbon monitor, while the other supplied the nondispersive infrared (NDIR) instruments and the paramagnetic oxygen analyzer. The latter stream was further conditioned in a saturator and refrigerated dryer before connecting to the CO_2 , CO and O_2 analyzers. All monitors were frequently zeroed and calibrated with certified gases.

All continuous gaseous sampling was done through a single point probe located in the center of the exhaust duct approximately 1m (40 inches) upstream of the main row of sampling ports used for the SASS and Method 5/8 sampling.

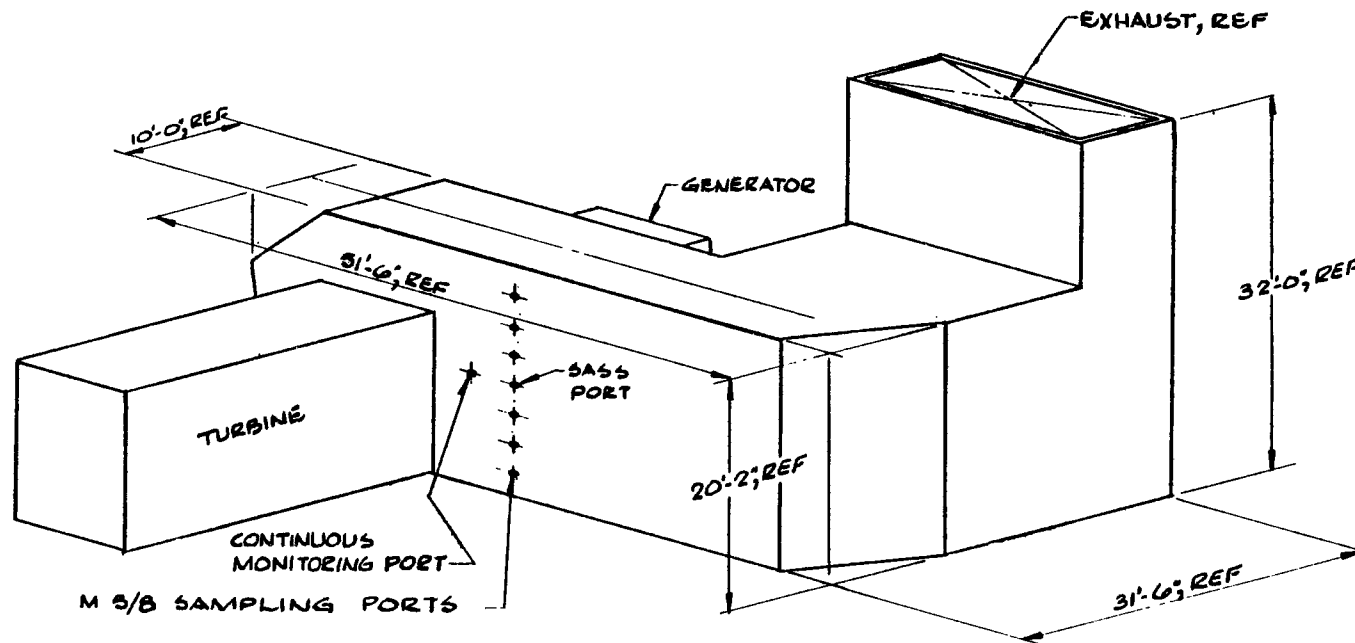


Figure 3-1. Exhaust duct configuration and sampling location.

TABLE 3-1. INSTRUMENTATION USED BY GENERAL ELECTRIC

Instrument	Technique	Measuring Range
Unburned hydrocarbons: Beckman Model 402	Flame ionization detector	5 ppm - full scale to 25% - full scale
NO and NO ₂ : Beckman 955	Chemiluminescence	0 - 10 ppm 0 - 1,000 ppm 0 - 25 ppm 0 - 2,500 ppm 0 - 100 ppm 0 - 10,000 ppm 0 - 250 ppm
O ₂ : Beckman Model F3	Paramagnetic	0 - 15% 13% - 18% 16% - 21% 0 - 25%
CO: Beckman 315B	Nondispersive infrared	0 - 50 ppm 0 - 200 ppm 0 - 500 ppm
CO ₂ : Beckman 364	Nondispersive infrared	0 - 5% 0 - 10% 0 - 15%

Particulate and Sulfur Species

Particulate and sulfur species in the flue gas were collected simultaneously with one sampling train -- a combined EPA Method 5 and Method 8 train. Such a system collects particulate samples on a filter heated to 394 K (250°F) in a conventional Method 5 arrangement. But rather than the conventional Method 5 water filled impinger train, the modified train employs a Method 8 impinger train containing isopropanol to remove SO₃ and hydrogen peroxide to remove SO₂. Particulate sulfate (SO₄) is also collected with this system. One run was completed for each test.

C₁-C₆ Hydrocarbons

Flue gas grab samples were collected in evacuated glass grab flasks. These samples were chromatographed onsite to determine C₁ to C₆ hydrocarbon compounds. A Carle Model 8500 portable gas chromatograph with a flame ionization detector was used for this analysis.

Source Assessment Sampling System

A Source Assessment Sampling System (SASS) train was used to sample the gas turbine exhaust gas. The SASS train was conventional in every way except that cyclones were not used to classify the particulate by size. A single fiberglass mat filter was used to collect the small amounts of particulate produced. A special oil cooled probe was used to maintain the sample tube temperature at 394 K (250°F). This SASS train arrangement generates the following samples:

Particulate: filter 99.99 percent efficient for particulate greater than 0.2 m.

Vapor phase: 1) XAD-2 porous polymer resin sorbent cartridge
2) Aqueous condensate
3) Hydrogen peroxide impinger
4) Ammonium persulfate-silver nitrate impingers

These samples were analyzed for trace elements and organic species to give both vapor and condensed phase composition.

A schematic of the SASS train is shown in Figure 3-2. The analysis protocol is given in Figure 3-3.

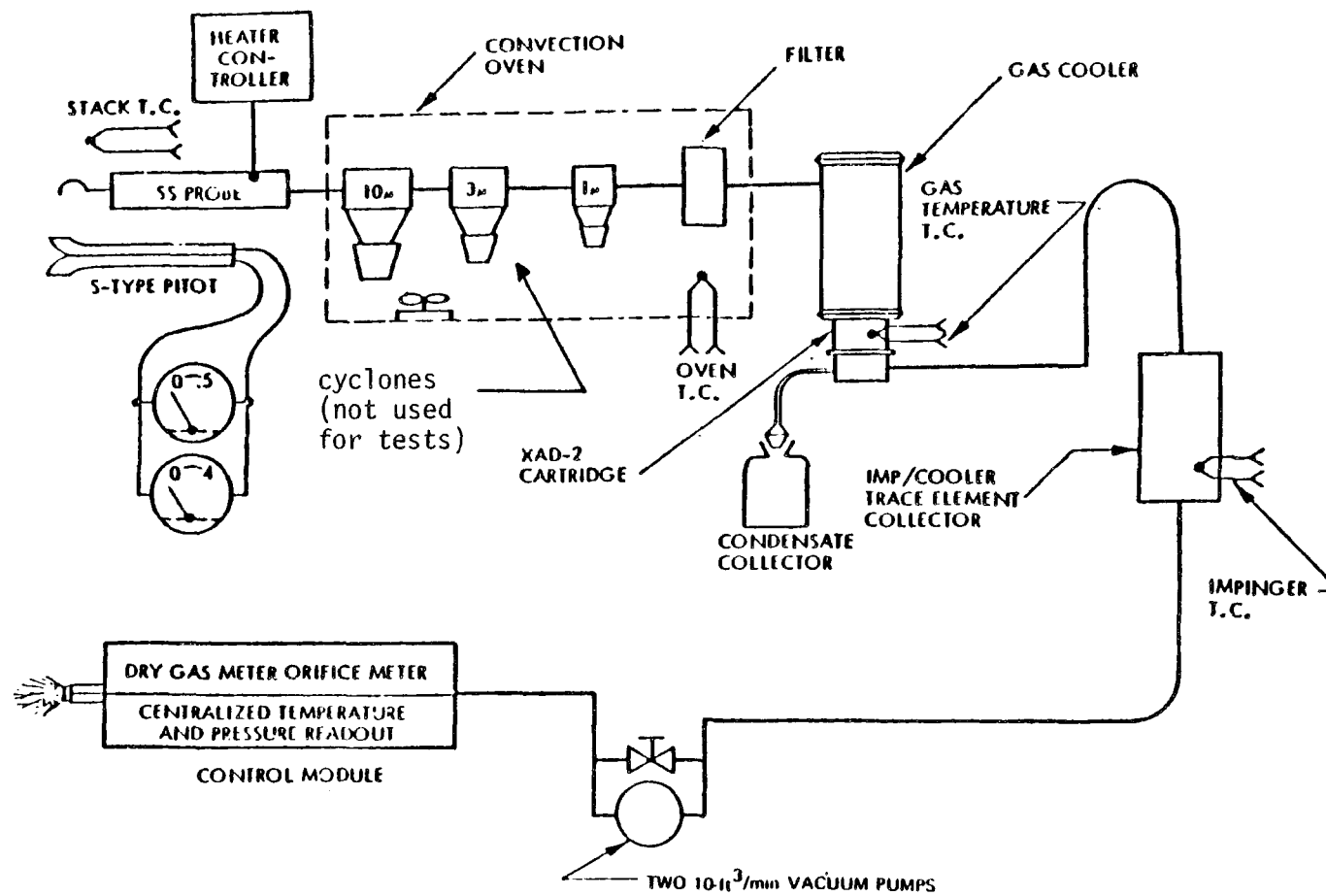


Figure 3-2. Source Assessment Sampling System (SASS) schematic.

Figure 3-3. SASS analysis protocol.

3.2 ANALYSIS PROTOCOL

Table 3-2 lists the analyses performed on the samples collected during both tests. Descriptions of these analyses are given in the following paragraphs.

In addition, machine operating data were collected by General Electric personnel. Sufficient data were taken during both tests so that airflow rates and the operating condition of the machine could be established. A detailed field test report submitted by General Electric is contained in Appendix G. In summary, GE treats operating data, information on gaseous emissions, fuel composition, machine geometry and internal flow splits, using a data analysis program which calculates machine operational characteristics. Table 3-3 illustrates the kind of information that can be obtained. This program proved an excellent means of crosschecking actual flue gas measurements as well as checking calculation methods against each other.

3.2.1 Inorganic Analysis

Trace element analyses for 23 selected trace elements were performed on the fuel, injected water, flyash, SASS XAD-2 and SASS impinger solutions. The procedure used to determine each trace element is outlined in Appendix H. Proximate and ultimate analyses were done on the fuel samples.

3.2.2 Organic Analysis

Organic analyses were performed in accordance with EPA Level 1 protocol (Reference 2). These analyses included C_1 - C_6 hydrocarbons in the flue gas, organic material condensed on the ash samples, and organic material caught in the XAD-2 sorbent trap and condensate trap.

3.2.3 Bioassay

Bioassays were performed on the SASS train XAD-2 extract sample from the water injection test. Microbial mutagenesis and cytotoxicity assays were performed by Litton Bionetics, Incorporated.

TABLE 3-2. SAMPLE ANALYSIS

Test/Analysis	Baseline no water injection	Low NO _x with water injection
Fuel		
Proximate and ultimate	X	X
Trace elements	X	X
Water		
Trace elements		X
Organic material		X
SASS train - outlet		
Trace elements	X	X
Organic material	X	X
Method 5/8 -- outlet		
Particulate	X	X
Sulfur species	X	X
Flue gas		
O ₂ , CO ₂ , NO _x , NO, CO	X	X
and total unburned hydrocarbons	X	X
C ₁ -C ₆ hydrocarbons	X	X

TABLE 3-3. GENERAL ELECTRIC CALCULATED OPERATING DATA^a

Calculation Method	Measured	Calculated
Factory test flow	Fuel flow and composition, airflow during factory test, inlet guide vane position, ambient conditions	Machine airflow, O ₂ , CO ₂ , water in exhaust, turbine inlet temperature.
Choked flow	Compressor discharge pressure and temperature, first stage nozzle area, fuel flow and composition.	Machine airflow, O ₂ , CO ₂ , H ₂ O, turbine inlet temperature.
Oxygen concentration	O ₂ , fuel flow and composition	Machine airflow, CO ₂ , H ₂ O, turbine inlet temperature.
CO ₂ concentration	CO ₂ , fuel flow and composition	Machine airflow, O ₂ , H ₂ O, turbine inlet temperature.

^aSee Appendix G

SECTION 4

TEST PROGRAM RESULTS

Data from the test program provided information on unit operation, effluent gaseous composition, particulate emissions, trace element emissions, sulfur species emissions and organic material emissions.

4.1 UNIT OPERATION

Unit 52 operated under steady-state conditions at rated continuous load and with operating parameters nominally the same for both the baseline (no water injection) and the NO_x control (with water injection) test. The load for each test was approximately 62 MW electrical generator output. Table 4-1 lists the process operating conditions and parameters during each test. The only significant difference between Test 1 and Test 2 is that Test 2 had water injection while Test 1 did not. Since it is fairly easy to duplicate engine operating conditions in a gas turbine, one can be reasonably confident in comparing emissions from tests where only the one variable, water injection rate, was changed.

The operating variable readings were recorded on an hourly basis throughout the tests. The results shown in Table 4-1 are an average of those values. The actual data sheets can be found in Appendix G.

One of the most significant penalties resulting from the use of water injection for NO_x control is the reduction in unit thermal efficiency or increased heat rate manifested as increased fuel consumption. As indicated in Table 4-1, the unit heat rate increased 2.4 percent with water injection at a water/fuel ratio equal to 0.42. This is because a portion of the fuel is required to vaporize the injected water. These effects on heat rate and fuel consumption are quite typical (Reference 3). Most users have reported heat rate penalties ranging from 2 to 5 percent, depending on the water to fuel ratio.

TABLE 4-1. UNIT 52 -- OPERATING CONDITIONS

	<u>Baseline</u>	<u>Injection</u>
Ambient barometric pressure - mm Hg (in. Hg)	755 (29.74)	756 (29.79)
Ambient temperature -- dry bulb -- K (°F)	295 (71.2)	301 (82.0)
Relative humidity	83.6	58.7
Compressor discharge pressure P_{CD} -- kpa (psia)	915 (132.7)	901 (130.7)
Compressor discharge temperature T_{CD} -- K (°F)	593 (607)	602 (624)
Speed (rpm)	3600	3600
Inlet guide vane angle (IGV degrees)	77	77
Load (MW)	61.9	61.5
Turbine exhaust temperature -- K (°F)	809 (997)	813 (1000)
Water injection rate -- liters/sec and (gpm)	0	2.52 (40)
Water/fuel ratio	0	0.42
Fuel temperature -- K (°F)	295.4 (71.7)	298 (76.4)
Fuel flow -- liters/sec and (gpm)	5.93 (94.0)	6.03 (95.6)
Atomizing air pressure -- kpa (psia)	1372 (199)	1372 (199)
Atomizing air temperature -- K (°F)	473 (392)	471 (387)
Combustion efficiency (%)	99.9	99.9
Exhaust flow -- m^3/s (10^6 SCFH)	205.4 (26.14)	200.9 (25.52)
Compressor inlet flow -- kg/s (lbm/sec)	253 (556.7)	255.7 (562.7)
Fuel/air ratio	0.0190	0.0196
Heat rate -- MJ/kWh (Btu/Kwh - based on LHV)	12.55 (11,892)	12.84 (12,173)

As noted in Section 3.2, GE personnel recorded operating data and monitored unit operation throughout the test program. In addition GE also evaluated recorded data using an in-house data analysis code. This program can be used to calculate inlet airflow and exhaust gas flow (in addition to other parameters -- see Table 3-3) using gaseous emissions data and other operating information. Four different calculational modes are possible, as outlined in Table 3-3. Calculated exhaust gas flowrates for each test, using the program, are listed in Table 4-2 for each of the calculation methods. Agreement among the methods is excellent (within one percent). Also shown in Table 4-2 are measured exhaust gas flowrates obtained by performing an EPA Method 5 velocity traverse across the exhaust duct. As indicated, measured rates are approximately 55 percent greater than calculated rates. This was not unexpected, though. The exhaust duct configuration was such that gas flow obstructions (i.e., bends) were very close to the sampling location, thus accurate velocity measurements were very difficult to obtain. In an attempt to equalize the effects of a poor sampling location, 42 sampling points were sampled. Nevertheless, measured gas flowrates were still unreasonably high due to the highly variable velocity readings.

Thus all exhaust flowrate values reported herein, including those noted in Table 4-1, are calculated values, averaged over the four possible calculational methods.

4.2 FUEL ANALYSIS

Duplicate proximate and ultimate fuel analyses were performed by General Electric and Commercial Testing and Engineering (CT&E). General Electric's results are reported in Appendix G. CT&E's analysis is reported in Appendix A. Results from both analyses were very similar and typical of distillate fuel oil. In addition, a trace element analysis of the fuel oil was performed as part of the mass balance and reported in Concentration and mass flowrate units in Appendices B-E.

4.3 EXHAUST GAS EMISSIONS

Exhaust emissions were tested for gaseous species, particulate emissions, sulfur species, trace elements and organic material emissions. Gaseous species were measured by General Electric personnel on a continuous basis throughout both tests. A combined EPA Method 5/8 train

TABLE 4-2. CALCULATED AND MEASURED EXHAUST GAS FLOWRATES -- m^3/s (10^6 SCFH)

Test No.	GE -- Calculated Values ^a				Average	Measured
	Factory Test Flow	Choked Flow	Oxygen Concentration	CO ₂ Concentration		
1	203.8 (25.9244)	205.3 (26.1206)	207.2 (26.3644)	205.8 (26.1828)	205.4 (26.1481)	316.4 (40.2632)
2	199.3 (25.3136)	201.0 (25.5280)	202.5 (25.7149)	220.5 (27.9991)	200.9 (25.5188 ^b)	311.4 (39.6247)

^aSee Appendix G for explanation of calculations

^bExhaust gas flowrate calculated by the CO₂ concentration scheme not included in average. CO₂ values believed to be affected by moisture in flue gas.

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was used to simultaneously sample particulates and sulfur species. A Source Assessment Sampling System (SASS) was used to collect samples for analysis of trace elements and organic material. This section presents the results of these analyses.

4.3.1 Gaseous Emissions

Total NO_x , NO , O_2 , CO_2 , CO and total unburned hydrocarbons (UHC) were measured at a single point in the exhaust duct. Supporting tests conducted by General Electric, reported in Appendix G, have concluded that emissions of NO_x and O_2 can be reliably and accurately measured from a single sampling point. However, species that are present only in very low concentrations, UHC for example (ppmv <2), require a traverse of the duct when sampling.

Table 4-3 presents gaseous emissions data in a form summarized from the General Electric report in Appendix G. With regard to the proposed New Source Performance Standards (NSPS) for stationary gas turbines, there are two things of importance to note from this information. First, with water injection operating at a water/fuel weight ratio of 0.42, NO_x emissions were reduced by 58 percent from the baseline levels -- from 177.5 to 74.2 ppm at 15 percent O_2 dry. This controlled level is within the NSPS proposed level of 75 ppm. The second item to note is that SO_2 emissions are substantially below the proposed NSPS level of 150 ppmv at 15 percent O_2 . The SO_2 values for Unit 52 were calculated directly from the fuel sulfur content assuming 100 percent conversion. The calculated value for SO_2 concentration, which assumes all fuel sulfur is converted to SO_2 , is reasonably close to the measured total SO_x emission concentration (within 30 percent) as determined by the Method 8 analysis.

The results of the sulfur species analysis are shown in Table 4-4. The data show that the actual emission levels of sulfur species, as well as the SO_2/SO_3 ratio, are not significantly affected by the use of water injection for NO_x control. Table 4-5 shows the results of a sulfur balance across the gas turbine. The quantity of sulfur recovered in the flue gas was approximately 70 percent of the inlet sulfur. Duplicate fuel oil sulfur analyses gave a sulfur content of approximately 0.11 percent, so inlet sulfur calculations should be correct. Consequently, the source of the inconsistency probably lies in the Method 8 sampling train and subsequent analysis.

TABLE 4-3. GASEOUS EMISSIONS RESULTS -- ppmv at 15 percent O₂ dry

	<u>Baseline</u>	<u>Water Injection</u>
NO _x	175.5	74.2
CO	5.6	8.1
CO ₂ (%)	4.1	4.3
SO ₂ ^a	19.5	20.5
UHC ^b	2.3	3.5

^aCalculated from fuel sulfur assuming
100 percent conversion to SO₂

^bppmv wet as CH₄

TABLE 4-4. SULFUR SPECIES EMISSIONS

Test	Species	Emissions			
		ppmv dry	μg/m ³	kg/min	μg/J
Baseline	SO ₂	11.7	3.12 x 10 ⁴	0.385	0.029
	SO ₃	1.1	3.48 x 10 ³	0.043	0.003
	SO ₄	1.2	4.61 x 10 ³	0.057	0.004
Water Injection	SO ₂	12.7	3.37 x 10 ⁴	0.407	0.030
	SO ₃	1.8	6.04 x 10 ³	0.073	0.005
	SO ₄ ^a	--	--	--	--

^aSample destroyed

TABLE 4-5. SULFUR BALANCE

	Baseline	Water Injection
Sulfur Input		
Fuel feedrate (kg/s)	4.85	4.96
Fuel sulfur content (% by wt.)	0.11	0.11
Total sulfur input (kg/s)	5.33×10^{-3}	5.50×10^{-3}
Sulfur Output		
SO ₂ (kg/s)	6.42×10^{-3}	6.78×10^{-3}
SO ₃ (kg/s)	0.72×10^{-3}	1.22×10^{-3}
SO ₄ (kg/s)	0.95×10^{-3}	
Total sulfur output (kg/s)	3.83×10^{-3}	3.83×10^{-3}
Sulfur recovery at outlet	72%	70%

An increase in emissions of unburned species due to lowered peak flame temperatures, is generally associated with the use of water injection for NO_x control. During the NO_x control test on Unit 52 average emissions of CO and UHC increased 54 and 52 percent respectively. While the increases seem significant, the actual emission concentrations for CO and UHC are still very low (<10 ppm) when water injection is being used.

Onsite analyses of C₁ to C₆ exhaust gas hydrocarbons were conducted for both the baseline and the water injection tests. The test results show that in the baseline test, C₁ to C₆ hydrocarbons were 6.5ppm at 15 percent O₂ wet, characterized as methane. In the water injection test, C₁ to C₆ hydrocarbon emissions were 1ppm at 15 percent O₂ wet, characterized as methane. These results are in general agreement with the total unburned hydrocarbon emissions measured by the continuous monitor.

4.3.2 Particulate Emissions

Particulate emissions for Unit 52 are shown in Table 4-6. As expected from a gas turbine burning distillate fuel oil, particulate emissions were very low, on the order of 0.0037 to 0.0042 kg/s as measured by the EPA Method 5 train. However, correlation between particulate emission rates from the EPA Method 5 train and the SASS train is poor. SASS measurements are almost a factor of 10 lower. This is not surprising however, when one considers that a SASS train is run at a single point in the exhaust duct. In a duct such as that of Unit 52, where flow patterns are irregular due to the duct configuration, particulate matter can be highly stratified. Furthermore, since particulate matter generated in a gas turbine will be very small in size, it will have a greater tendency to stratify with a strong bias to high velocity regions. Since the SASS train is required to operate at a point of average velocity, away from the high velocity regions, particulate capture is expected to be considerably lower in the SASS train than in the Method 5 train, which fully traverses the duct cross section, as this gives representative results.

TABLE 4-6. PARTICULATE EMISSIONS

Test	Particulate Emissions		
	kg/s	μg/Joule	μg/DSCM
Method 5 -- Baseline	4.2×10^{-3}	.019	572
Method 5 -- Water injection	3.7×10^{-3}	.016	509
SASS -- Baseline	0.45×10^{-3}	.002	63
SASS -- Water injection	0.97×10^{-3}	.004	137

While according to the Method 5 measurements, particulate emissions dropped with water injection, the reduction was not significant. Water injection then appears to have little effect on particulate emissions. This is supported by data presented in Reference 3.

4.3.3 Trace Element Characterization and Emissions

Fuel oil, injected water and flue gas samples were collected and analyzed for selected trace elements for the baseline and water injection tests. Grab samples were taken for the oil and water. The flue gas was sampled by using a SASS train. The detailed results of these analyses are presented in Appendices A through F.

The probe wash and the filter have been combined into one sample, as have the aqueous condensate and the first impinger. The XAD-2 cartridge was analyzed independently and the second and third impingers were combined into one sample as outlined in the Level 1 procedures manual (Reference 2).

Solid And Vapor Phase Trace Element Partitioning

The SASS train allows determining both solid phase and gas phase composition. Solid phase species are collected in the probe, cyclones, filter and interconnecting tubing, while the vapor phase species are collected in the organics module or the impinger portions of the SASS train. All SASS train components up to the filter are maintained at 394-478K (250 - 400°F). From there, the flue gas goes to the organics module, where it is cooled to approximately 293 K (68°F) and passed through a cross linked porous polymer resin (XAD-2) cartridge. From this section, two samples are generated: the condensate and the XAD-2 sorbent extract. From the organic module, the flue gas goes through an impinger train. The first impinger contains hydrogen peroxide and the second and third impingers contain silver nitrate-ammonium persulfate solutions. For trace element analysis, the organic module aqueous condensate sample is combined with the hydrogen peroxide impinger sample to form one sample for analysis. Thus three samples representing vapor phase composition are analyzed: the XAD-2, the aqueous condensate and hydrogen peroxide impinger solution, and the combined silver nitrate-ammonium persulfate impinger solution.

To determine whether a particular trace element was concentrated in the solid or vapor phase, trace element flowrates (kg/s) were compared. In order to partition the samples as to whether they were solid or vapor, the following partitioning criterion was used: trace elements were

considered to be preferentially concentrated in the vapor phase if their vapor phase concentrations were at least twice their solid phase concentrations. The partitioning results are shown in Table 4-7 for elements where sufficient data to determine partitioning were obtained.

Elemental Mass Balance

A trace element mass balance was performed across the gas turbine system using emissions flowrate data from Appendix C. Table 4-8 presents the results of the mass balance. In general and where sufficient data are available, the element mass balances are within the reliability of the Level 1 sampling and analysis procedures which are assumed to be quantitative within a factor of 2 to 3. Zinc and copper are somewhat outside of these boundaries but not significantly so. Iron, as measured at the outlet for both tests, far exceeds the amount entering the turbine as contained in the fuel oil and injected water. It is possible that the source of this excess iron is rust and scale coming loose from the internal gas turbine ductwork and being captured in the SASS train. The analysis of the fuel for iron was supported by a duplicate analysis performed by GE (Appendix G).

Effects Due to NO_x Control

It appears that the use of water injection to control NO_x emissions has an insignificant effect on trace element emissions. Outlet emissions of all trace elements analyzed remained within a factor of three when comparing the baseline and water injection emission flowrates. Also, water injection has an insignificant effect on trace element emissions with respect to solid/vapor phase partitioning. For those elements where sufficient data were available, the solid/vapor partitioning remained virtually the same.

4.3.4 Organic Analyses

Organic analyses were performed on selected samples according to the EPA Level 1 protocol (Reference 2). Any differences from the Level 1 protocol will be noted in the following discussion. The analytical laboratory data are reported in Appendix F.

As recommended by Level 1 analysis procedures the samples were first extracted with methylene chloride in a Soxhlet apparatus. A Total Chromatographable Organic (TCO) and a gravimetric (GRAV) analysis were then performed on the sample extracts. This analysis separates each

TABLE 4-7. TRACE ELEMENT PARTITIONING -- SOLID PHASE/VAPOR PHASE

	<u>Baseline</u>	<u>Water Injection</u>
Arsenic	X	V
Barium	X	V
Beryllium	X	V
Cadmium	V	V
Chromium	V	EQ
Copper	V	V
Iron	V	V
Lead	V	EQ
Manganese	X	V
Mercury	V	X
Nickel	V	S
Thallium	X	V
Vanadium	V	V
Zinc	X	S

EQ -- Material partitioned equally between vapor and solid phase

S -- Material preferentially concentrated in solid phase

V -- Material preferentially concentrated in vapor phase

X -- Insufficient data

TABLE 4-8. TRACE ELEMENT MASS BALANCE -- OUTLET (g/min)/INLET(g/min)

	<u>Baseline</u>	<u>Water Injection</u>
Boron	<1	1.2
Cadmium	.62	--
Chromium	.24	>3
Cobalt	--	<1
Copper	>5	>7
Iron	>100	>100
Lead	.26	44
Mercury	.16	1.4
Nickel	>.7	>2
Selenium	<.1	--
Vanadium	>.3	>.6
Zinc	3.5	4.6

sample extract into two separate samples having definite boiling point ranges. The TCO fraction contains species with boiling points in a range from 373 K to 573 K. Those species with boiling points above 573 K are contained in the gravimetric sample.

An infrared spectrophotometric (IR) analysis was also performed on the total sample extracts. This aided in the identification of functional organic groups within the complex sample mixture. The organic material in the sample extract was not sufficient to warrant separation by liquid chromatography with further analyses of the fractions eluted. The total sample extracts were analyzed by gas chromatography-mass spectrometry (GCMS) for specific polycyclic organic molecules and priority pollutants.

C₁ to C₆ hydrocarbon compounds were analyzed onsite by gas chromatography. The same set of organic analyses was performed on the samples from the baseline and the water injection test. A discussion of the analytical results follows.

Total Chromatographable Organics (TCO) and Gravimetric Analyses (GRAV) of Organic Extracts

Total Chromatographable Organics (TCO) and Gravimetric Analyses (GRAV) were performed on the XAD-2 resin extracts from the baseline and water injection tests. The extract samples were combined with the organic portion of the sorbent module condensate. The results from these analyses are shown in Table 4-9. Three conclusions can be drawn from these results. First, virtually all of the organics in the flue gas can be found in compounds with the boiling point ranging from 373 K to 573 K. Second, there is little effect on the distribution of compounds, with regard to the boiling point, between the baseline and water injection test. Third, the use of water injection has a very small effect on the total amount of organics in the sample extract, decreasing the quantity by approximately 6 percent.

Infrared Spectra of Total Extracts

The results of the infrared spectral analyses done on the total XAD-2 sample extracts for the baseline and water injection test are shown in Tables 4-10 and 4-11 respectively. Comparison of the wave numbers and assignments (as well as the spectra themselves) indicate that the organics were almost identical for both tests. Both spectra indicate that the

TABLE 4-9. TCO AND GRAV ANALYSES RESULTS OF THE XAD-2 EXTRACT

Test	Sample Type	Gravimetric Result (mg)	TCO Result (mg)	Total Organics in Extract (mg)	Total Organics Concentration (mg/m ³)
Baseline	XAD-2 extract	0.6	27.5	28.1	1.3 ^a
Water injection	XAD-2 extract	<0.1	26.3	26.3	1.1 ^b

^aBased on sample volume = 20.9 m³

^bBased on sample volume = 23.2 m³

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principal constituents were an ester or a carboxylic acid and an alcohol. Unsaturated compounds and/or aromatic groups were also present although the intensity of the bands suggest that they may not be part of the principal constituents.

Gas Chromatography-Mass Spectrometry (GCMS) Analysis of Total Extracts

Liquid column separation and low resolution mass spectrometry were not performed on the sample extracts because an insufficient sample volume remained after concentration. However, the sample extracts were analyzed by GCMS. Specific compounds were identified and quantified with this technique.

Tables 4-12 and 4-13 show the GCMS results for the baseline and water injection tests. Table 4-14 lists the specific compounds which were analyzed with the GCMS.

4.3.5 Bioassay

Mutagenicity and cytotoxic evaluations were performed on the XAD-2 extract sample from the water injection test. Results from the Ames salmonella/microsome plate test show the sample nonmutagenic. Results of the cytotoxicity assay indicate the extract has low toxicity to WI-38 human cells and that the viability index showed an EC50 value would be obtained at approximately 152 liters gas/ml. Complete results and supporting data are located in Appendix I.

4.3.6 Conclusions

The use of water injection for NO_x control on Unit 52 appears to have little effect on organic emissions. Neither the total organics, as reported in the TCO and GRAV analyses, nor the species and classes of organics changed significantly from the baseline test to the water injection test.

TABLE 4-10. INFRARED SPECTRA ANALYSIS RESULTS -- XAD-2 EXTRACT,
BASELINE TEST

Wave Number (cm ⁻¹)	Intensity ^a	Assignment	Comments
3400-3500	S	O-H	Broad peak
2960, 2920, 2850	S	C-H	Aliphatic
1720-1700	S	C=O	Carbonyl possibly ester
1600	W	C=C	Unsaturated, aromatic
1450	M	-CH ₃ bending	Methyl groups
1370	M		Possibly methyl
1260	S	-O- bending	Ether/ester
1070-1090	M	Unassigned	Broad peak
800	W	Unassigned	Possibly aromatic Substitution bands
710	W	Unassigned	Possibly aromatic Substitution bands

^aIntensity: S - strong, M - medium, W - weak

TABLE 4-11. INFRARED SPECTRA ANALYSIS RESULTS -- XAD-2 EXTRACT,
WATER INJECTION TEST

Wave Number (cm ⁻¹)	Intensity ^a	Assignment	Comments
3400-3500	S	O-H	Broad peak
2960, 2920, 2850	S	C-H	Aliphatic
1690-1720	S	C=O	Carbonyl broad
1600	W	C=C	Unsaturated, aromatic
1450, 1460	M	-CH ₃	Methyl band
1380	M		Possibly methyl
1260	S	-O-	Ether/ester
1070-1100	S	Unassigned	Broad peak
800	M	Unassigned	Possibly aromatic Substitution bands
710, 700	M	Unassigned	Possibly aromatic Substitution bands

^aIntensity: S - strong, M - medium, W - weak

TABLE 4-12. GCMS RESULTS -- XAD-2 EXTRACT -- BASELINE TEST

Compound	Concentration ($\mu\text{g}/\text{m}^3$) ^b
Bis(2-ethylhexyl)phthalate ^a	1.0
Other phthalates	1.0
Phenanthrene/anthracene ^a	0.5
Diphenyl ether	0.5
Phenol	1.0

^aConfirmed by comparison with standard^bBased on sample volume = 20.9 m³

TABLE 4-13. GCMS RESULTS -- XAD-2 EXTRACT -- WATER INJECTION TEST

Compound	Concentration ($\mu\text{g}/\text{m}^3$) ^b
Bis(2-ethylhexyl)phthalate ^a	1.0
Other phthalates	1.0
Phenanthrene/anthracene	1.0
Fluoranthene	0.5
Pyrene	0.5
Terphenyl	5.0
Diphenylcyclohexane (2 isomers)	10.0
Phenol	1.0
Naphthalene	1.0

^aBased on sample volume = 23.2 m³

TABLE 4-14. COMPOUNDS SCREENED FOR IN GCMS ANALYSIS OF XAD-2 EXTRACTS

Compound	Representative m/e Values	Compound	Representative m/e Values
7,12 dimethyl benz (a) anthracene	256	Methylnaphthalene	42,127
Dibenz (a,h) anthracene*	278	Biphenyl	154
Benzo (c) phenanthrene	228	Phthalic Anhydride	148,104
3-methyl cholanthrene	268	Nitronaphthalene	173,115
Benzo (a) pyrene*	252	Dibenzothiophene	184,139
Dibenzo (a,H) pyrene	302	Alkanes	57,71
Dibenzo (a,i) pyrene	302	Decalin	67,138
Dibenzo (c,g) carbozole	267	Anthracene*	178
Fluoranthene*	202	Phenanthrene*	178
Pyrene*	202		
Anthanthrene	276		
Benz (a) anthracene*	228		
Benzo (g,h,i) perylene*	276		
Benzo (e) pyrene	252		
Perylene	252		
Naphthalene	128		
Acenaphthylene	152		
Acenaphthene	154		
Fluorene	166		
Diphenyl ether	170		
Dibenzofuran	168		
Fluorenone	180,152		
Naphthoquinone	158,130		
Xanthone	196,168		
Xanthene	182,181		
Quinoline	129		
Phenol	94,65		

*Contained in standard mixture.

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SECTION 5

SUMMARY

Exhaust emissions sampling and analysis was performed on Houston Lighting and Power's Unit 52 at the T. H. Wharton Generating Station in Houston, Texas from April 21-24, 1978. Unit 52 is a General Electric Model MS 70001C simple-cycle, single-shaft, heavy-duty gas turbine rated at 70.8 MW nominal electrical output and fired with distillate oil fuel. The unit is equipped with a water injection system for controlling NO_x emissions. The purpose of the tests was to determine the effectiveness of water injection in reducing NO_x and to assess the effects that water injection has on emissions other than NO_x. In addition, operating parameters were recorded so that effects on turbine operation due to water injection could also be observed.

One baseline test (without water injection) and one water injection test, under nominally similar operating conditions, were performed. A summary of the results is presented below.

Unit Operation

The use of water injection did not appear to have any significant impact on unit operations other than an increase in heat rate of approximately 2 percent. This results from some of the fuel heat content being used to vaporize the water. A water/fuel weight ratio of 0.42 was used to reduce NO_x emissions to a level just below the proposed New Source Performance Standard (NSPS) of 75 ppm for stationary gas turbines. A higher or lower water/fuel ratio would respectively raise or lower the resulting heat rate. No other significant operational effects were observed as a result of water injection.

Emissions

Exhaust emissions were measured for changes resulting from the use of water injection for NO_x control. Gaseous, particulate, sulfur species, trace element and organic species emissions were evaluated.

Total NO_x , NO , CO , O_2 , CO_2 and total unburned hydrocarbons were measured by continuous monitoring. With water injection operating at a water/fuel weight ratio of 0.42, NO_x emissions were reduced 58 percent from the baseline levels -- from 177.5 to 74.2 ppmv NO_x at 15 percent O_2 dry. SO_2 emissions are wholly determined by the fuel sulfur content and are not affected by water injection. Changes in emissions of CO and total unburned hydrocarbons were within the limits of the analyses. Particulate emissions were very low due to the clean fuel and efficient combustion and did not change significantly with the use of water injection.

It appears that the use of water injection to control NO_x emissions has an insignificant effect on trace element emissions. Furthermore, water injection was found to have little effect on trace element emissions with respect to solid/vapor phase partitioning.

Comparisons of organic species emissions between the baseline and water injection test indicate that water injection has little effect on these emissions. Neither the total organics nor the species and classes of organics changed significantly.

Bioassay tests on the XAD-2 extract from the water injection test showed the exhaust gas to be nonmutagenic and of low toxicity as determined by the Ames Salmonella/microsome plate test and the WI-38 cytotoxicity test respectively.

REFERENCES

REFERENCES

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3. Goodwin, D. R. et al., "Standard Support and Environmental Impact Statement. Volume 1: Proposed Standards of Performance for Stationary Gas Turbines, EPA-450/2-77-017a, NTIS-PB 272 422/7BE.

APPENDIX A
FUEL ANALYSIS

PART III - Section No. 1

Table No. 1 - Proximate Analysis
Concentration in Wt. %

<u>Parameter</u>	<u>Baseline</u>	<u>Water Injection</u>
Moisture	≤ 0.01	≤ 0.01
Volatile Matter	-----	-----
Ash	0.01	≤ 0.005
Carbon	-----	-----
Sulfur	0.10	0.12
BTU (per lb.)	19849	19751
Specific Gravity at 60°F	0.833	0.831

Table No. 2 - Ultimate Analysis
Concentration in Wt. %

<u>Parameter</u>	<u>Baseline</u>	<u>Water Injection</u>
Moisture	≤ 0.01	≤ 0.01
Ash	0.01	0.005
Carbon	81.98	84.91
Hydrogen	13.16	13.25
Nitrogen	0.11	0.17
Chlorine	0.14	0.13
Sulfur	0.10	0.12
Oxygen (by diff.)	<u>4.5</u>	<u>1.42</u>
TOTAL	100.00	100.00

APPENDIX B
TRACE ELEMENT CONCENTRATIONS -- ppm

Symbols appearing in the tables:

DSCM	Dry Standard Cubic Meter
ESP	Electrostatic Precipitator
kg	Kilogram
MCG	Microgram
min	Minute
ppm	Part per million by weight
<	Less than
*	Sample not analyzed for the particular element/or ionic specie
N	Sample not analyzed
-	Concentration in the sample less than the concentration in the blank

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - FUEL

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .800	< .800
ARSENIC	< 1.00	< 1.00
BARIUM	< .700	< .700
BERYLLIUM	< .300-01	< .300-01
BISMUTH	< .400	< .400
BORON	69.0	53.0
CADMIUM	.900	< .600-01
CHROMIUM	2.90	< .100+00
COBALT	< .200-01	.500-01
COPPER	< .400	< .400
IRON	< .300-01	< .300-01
LEAD	13.0	2.20
MANGANESE	< .100+00	< .100+00
MERCURY	.700	.600
MOLYBDENUM	< 1.00	< 1.00
NICKEL	< .100-01	< .100-01
SELENIUM	3.00	< 2.00
TELLURIUM	< .800	< .800
THALLIUM	< 2.00	< 2.00
TIN	< 3.00	< 3.00
TITANIUM	< 7.00	< 7.00
URANIUM	< 1.00	< 1.00
VANADIUM	< 5.00	< 5.00
ZINC	8.70	7.10
ZIRCONIUM	< 15.0	< 17.0

**GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - WATER**

TRACE ELEMENT CATIONS	TEST CONDITION			
	BASELINE, NO WATER		WATER INJECTION	
ANTIMONY	N	.000	<	.700-02
ARSENIC	N	.000	<	.200-01
BARIUM	N	.000	<	.400-02
BERYLLIUM	N	.000	<	.200-03
BISMUTH	N	.000	<	.200-02
BORON	N	.000	<	3.00
CADMIUM	N	.000	<	.400-03
CHROMIUM	N	.000	<	.600-03
COBALT	N	.000	<	.100-03
COPPER	N	.000	<	.300-02
IRON	N	.000	<	.300-03
LEAD	N	.000		.190
MANGANESE	N	.000	<	.100-03
MERCURY	N	.000		.880-02
MOLYBDENUM	N	.000		.160
NICKEL	N	.000	<	.100-03
SELENIUM	N	.000	<	.200-01
TELLURIUM	N	.000	<	.600-02
THALLIUM	N	.000	<	.900-02
TIN	N	.000	<	.200-01
TITANIUM	N	.000	<	.600-01
URANIUM	N	.000	*	.000
VANADIUM	N	.000	<	.200-01
ZINC	N	.000		.780-01
ZIRCONIUM	N	.000	*	.000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< 10.0	< 10.0
ARSENIC	30.0	40.0
BARIUM	< 7.00	38.0
BERYLLIUM	.600	2.90
BISMUTH	< 2.00	< 2.00
BORON	< .400+04	< .300+04
CADMIUM	8.50	7.10
CHROMIUM	260.	210.
COBALT	.500	6.60
COPPER	160.	77.0
IRON	21.0	190.
LEAD	150.	880.
MANGANESE	< .100+00	.400
MERCURY	< .900	< .900
MOLYBDENUM	< 9.00	< 7.00
NICKEL	3.00	35.0
SELENIUM	< 30.0	< 20.0
TELLURIUM	< 10.0	< 8.00
THALLIUM	< 20.0	30.0
TIN	< 300.	.280+04
TITANIUM	< 60.0	< 50.0
URANIUM	* .000	* .000
VANADIUM	57.0	45.0
ZINC	.540+05	.630+05
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET
XAD-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .700	< .800
ARSENIC	< 2.00	< 2.00
BARIUM	< .600	< .600
BERYLLIUM	< .200-01	< .200-01
BISMUTH	< .300	< .400
BORON	< 370.	< 360.
CADMIUM	< .400-01	< .400-01
CHROMIUM	< .500-01	< .600-01
COBALT	< .800-02	< .800-02
COPPER	2.80	6.60
IRON	11.0	15.0
LEAD	15.0	2.40
MANGANESE	< .800-02	< .800-02
MERCURY	.440	4.30
MOLYBDENUM	< .900	< .900
NICKEL	< .800-02	< .800-02
SELENIUM	< 2.00	< 2.00
TELLURIUM	< .600	< .600
THALLIUM	< 2.00	< 2.00
TIN	< 2.00	< 2.00
TITANIUM	< 5.00	< 6.00
URANIUM	* .000	* .000
VANADIUM	4.00	10.0
ZINC	< .600	4.00
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .600-02	< .500-02
ARSENIC	< .300-01	< .300-01
BARIUM	.500-02	< .400-02
BERYLLIUM	< .200-03	< .200-03
BISMUTH	< .400-02	< .500-02
BORON	< 3.00	< 3.00
CADMIUM	.200	.370-02
CHROMIUM	.200	.630-01
COBALT	< .100-03	< .200-03
COPPER	.390	.370
IRON	.210	.210
LEAD	.300-01	.160-01
MANGANESE	< .700-04	< .100-03
MERCURY	< .700-03	< .100-02
MOLYBDENUM	.150-01	< .600-02
NICKEL	.240-02	.210-02
SELENIUM	< .100-01	< .100-01
TELLURIUM	< .600-02	< .600-02
THALLIUM	< .500-02	< .600-02
TIN	< .200-01	< .200-01
TITANIUM	< .400-01	< .500-01
URANIUM	* .000	* .000
VANADIUM	< .700-02	< .100-01
ZINC	.560	.530
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (PPM) - UNIT OUTLET
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .500-02	< .600-02
ARSENIC	< .200-01	< .300-01
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CADMIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	.500-02	< .300-02
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX C

TRACE ELEMENT FLOWRATES -- kg/min

**GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - FUEL**

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .235-03	< .236-03
ARSENIC	< .294-03	< .296-03
BARIUM	< .206-03	< .207-03
BERYLLIUM	< .883-05	< .887-05
BISMUTH	< .118-03	< .118-03
BORON	.203-01	.157-01
CADMIUM	.265-03	< .177-04
CHROMIUM	.853-03	< .296-04
COBALT	< .588-05	.148-04
COPPER	< .118-03	< .118-03
IRON	< .883-05	< .887-05
LEAD	.382-02	.650-03
MANGANESE	< .294-04	< .296-04
MERCURY	.206-03	.177-03
MOLYBDENUM	< .294-03	< .296-03
NICKEL	< .294-05	< .296-05
SELENIUM	.883-03	< .591-03
TELLURIUM	< .235-03	< .236-03
THALLIUM	< .588-03	< .591-03
TIN	< .883-03	< .887-03
TITANIUM	< .206-02	< .207-02
URANIUM	< .294-03	< .296-03
VANADIUM	< .147-02	< .148-02
ZINC	.256-02	.210-02
ZIRCONIUM	< .441-02	< .502-02

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - WATER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	N .000	< .103-05
ARSENIC	N .000	< .295-05
BARIUM	N .000	< .591-06
BERYLLIUM	N .000	< .295-07
BISMUTH	N .000	< .295-06
BORON	N .000	< .443-03
CADMIUM	N .000	< .591-07
CHROMIUM	N .000	< .886-07
COBALT	N .000	< .148-07
COPPER	N .000	< .443-06
IRON	N .000	< .443-07
LEAD	N .000	.281-04 †
MANGANESE	N .000	< .148-07
MERCURY	N .000	.130-05
MOLYBDENUM	N .000	.236-04 †
NICKEL	N .000	< .148-07
SELENIUM	N .000	< .295-05
TELLURIUM	N .000	< .886-06
THALLIUM	N .000	< .133-03
TIN	N .000	< .295-05
TITANIUM	N .000	< .886-05
URANIUM	N .000	* .000
VANADIUM	N .000	< .295-05
ZINC	N .000	.115-04
ZIRCONIUM	N .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .165-05	< .143-05
ARSENIC	.494-05	.571-05
BARIUM	< .115-05	.542-05
BERYLLIUM	.989-07	.414-06
BISMUTH	< .330-06	< .285-06
BORON	< .659-03	< .428-03
CADMIUM	.140-05	.101-05
CHROMIUM	.429-04	.300-04
COBALT	.824-07	.942-06
COPPER	.264-04	.110-04
IRON	.346-05	.271-04
LEAD	.247-04	.126-03
MANGANESE	< .165-07	.571-07
MERCURY	< .148-06	< .128-06
MOLYBDENUM	< .148-05	< .999-06
NICKEL	.494-06	.500-05
SELENIUM	< .494-05	< .285-05
TELLURIUM	< .165-05	< .114-05
THALLIUM	< .330-05	.428-05
TIN	< .494-04	.400-03
TITANIUM	< .989-05	< .714-05
URANIUM	* .000	* .000
VANADIUM	.939-05	.642-05
ZINC	.890-02	.899-02
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET
XAU-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .450-04	< .459-04
ARSENIC	< .129-03	< .115-03
BARIUM	< .386-04	< .344-04
BERYLLIUM	< .129-05	< .115-05
BISMUTH	< .193-04	< .229-04
BORON	< .238-01	< .206-01
CADMIUM	< .257-05	< .229-05
CHROMIUM	< .322-05	< .344-05
COBALT	< .515-06	< .459-06
COPPER	.180-03	.378-03
IRON	.708-03	.860-03
LEAD	.965-03	.138-03
MANGANESE	< .515-06	< .459-06
MERCURY	.283-04	.246-03
MOLYBDENUM	< .579-04	< .516-04
NICKEL	< .515-06	< .459-06
SELENIUM	< .129-03	< .115-03
TELLURIUM	< .386-04	< .344-04
THALLIUM	< .129-03	< .115-03
TIN	< .129-03	< .115-03
TITANIUM	< .322-03	< .344-03
URANIUM	* .000	* .000
VANADIUM	.257-03	.573-03
ZINC	< .386-04	.229-03
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .478-05	< .456-05
ARSENIC	< .239-04	< .274-04
BARIUM	.398-05	< .365-05
BERYLLIUM	< .159-06	< .182-06
BISMUTH	< .319-05	< .456-05
BORON	< .239-02	< .274-02
CADMIUM	.159-03	.337-05
CHROMIUM	.159-03	.575-04
COBALT	< .797-07	< .182-06
COPPER	.311-03	.337-03
IRON	.167-03	.192-03
LEAD	.239-04	.146-04
MANGANESE	< .558-07	< .912-07
MERCURY	< .558-06	< .912-06
MOLYBDENUM	.120-04	< .547-05
NICKEL	.191-05	.192-05
SELENIUM	< .797-05	< .912-05
TELLURIUM	< .478-05	< .547-05
THALLIUM	< .398-05	< .547-05
TIN	< .159-04	< .182-04
TITANIUM	< .319-04	< .456-04
URANIUM	* .000	* .000
VANADIUM	< .558-05	< .912-05
ZINC	.446-03	.483-03
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (KG/MIN) - UNIT OUTLET
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .537-05	< .494-05
ARSENIC	< .215-04	< .247-04
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CAUDIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	.537-05	< .247-05
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX D

TRACE ELEMENT FLOWRATES -- MCG/Joule

**GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - FUEL**

TRACE ELEMENT CATIONS	TEST CONDITION			
	BASELINE, NO WATER		WATER INJECTION	
ANTIMONY	<	.174-04	<	.175-04
ARSENIC	<	.218-04	<	.219-04
BARIUM	<	.152-04	<	.153-04
BERYLLIUM	<	.653-06	<	.656-06
BISMUTH	<	.871-05	<	.875-05
BORON		.150-02		.116-02
CADMIUM		.196-04	<	.131-05
CHROMIUM		.631-04	<	.219-05
COBALT	<	.435-06		.109-05
COPPER	<	.871-05	<	.875-05
IRON	<	.653-06	<	.656-06
LEAD		.283-03		.481-04
MANGANESE	<	.218-05	<	.219-05
MERCURY		.152-04		.131-04
MOLYBDENUM	<	.218-04	<	.219-04
NICKEL	<	.218-06	<	.219-06
SELENIUM		.653-04	<	.437-04
TELLURIUM	<	.174-04	<	.175-04
THALLIUM	<	.435-04	<	.437-04
TIN	<	.653-04	<	.656-04
TITANIUM	<	.152-03	<	.153-03
URANIUM	<	.218-04	<	.219-04
VANADIUM	<	.109-03	<	.109-03
ZINC		.189-03		.155-03
ZIRCONIUM	<	.327-03	<	.372-03

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - WATER

TRACE ELEMENT CATIONS	TEST CONDITION		
	BASELINE, NO WATER		WATER INJECTION
ANTIMONY	N	.000	< .765-07
ARSENIC	N	.000	< .219-06
BARIUM	N	.000	< .437-07
BERYLLIUM	N	.000	< .219-08
BISMUTH	N	.000	< .219-07
BOHON	N	.000	< .328-04
CADMIUM	N	.000	< .437-08
CHROMIUM	N	.000	< .656-08
COBALT	N	.000	< .109-08
COPPER	N	.000	< .328-07
IRON	N	.000	< .328-08
LEAD	N	.000	.208-05
MANGANESE	N	.000	< .109-08
MERCURY	N	.000	.962-07
MOLYBDENUM	N	.000	.175-05
NICKEL	N	.000	< .109-08
SELENIUM	N	.000	< .219-06
TELLURIUM	N	.000	< .656-07
THALLIUM	N	.000	< .984-07
TIN	N	.000	< .219-06
TITANIUM	N	.000	< .656-06
URANIUM	N	.000	* .000
VANADIUM	N	.000	< .219-06
ZINC	N	.000	.853-06
ZIRCONIUM	N	.000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .122-06	< .106-06
ARSENIC	.366-06	.423-06
BARIUM	< .855-07	.402-06
BERYLLIUM	.732-08	.307-07
BISMUTH	< .244-07	< .211-07
BORON	< .488-04	< .317-04
CADMIUM	.104-06	.751-07
CHROMIUM	.317-05	.222-05
COBALT	.610-08	.698-07
COPPER	.195-05	.814-06
IRON	.256-06	.201-05
LEAD	.183-05	.930-05
MANGANESE	< .122-08	.423-08
MERCURY	< .110-07	< .951-08
MOLYBDENUM	< .110-06	< .740-07
NICKEL	.366-07	.370-06
SELENIUM	< .366-06	< .211-06
TELLURIUM	< .122-06	< .846-07
THALLIUM	< .244-06	.317-06
TIN	< .366-05	.296-04
TITANIUM	< .732-06	< .529-06
URANIUM	* .000	* .000
VANADIUM	.696-06	.476-06
ZINC	.659-03	.666-03
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MC6/JOULE) - UNIT OUTLET
XAD-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .334-05	< .340-05
ARSENIC	< .953-05	< .849-05
BARIUM	< .286-05	< .255-05
BERYLLIUM	< .953-07	< .849-07
BISMUTH	< .143-05	< .170-05
BORON	< .176-02	< .153-02
CADMIUM	< .191-06	< .170-06
CHROMIUM	< .238-06	< .255-06
COBALT	< .381-07	< .340-07
COPPER	.133-04	.280-04
IRON	.524-04	.637-04
LEAD	.715-04	.102-04
MANGANESE	< .381-07	< .340-07
MERCURY	.210-05	.183-04
MOLYBDENUM	< .429-05	< .382-05
NICKEL	< .381-07	< .340-07
SELENIUM	< .953-05	< .849-05
TELLURIUM	< .286-05	< .255-05
THALLIUM	< .953-05	< .849-05
TIN	< .953-05	< .849-05
TITANIUM	< .238-04	< .255-04
URANIUM	* .000	* .000
VANADIUM	.191-04	.425-04
ZINC	< .286-05	.170-04
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET
FIRST IMPINGER

TRACE ELEMENT	TEST CONDITION	
CATIONS	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .354-06	< .338-06
ARSENIC	< .177-05	< .203-05
BARIUM	.295-06	< .270-06
BERYLLIUM	< .118-07	< .135-07
BISMUTH	< .236-06	< .338-06
BORON	< .177-03	< .203-03
CADMIUM	.118-04	.250-06
CHROMIUM	.118-04	.426-05
COBALT	< .590-08	< .135-07
COPPER	.230-04	.250-04
IRON	.124-04	.142-04
LEAD	.177-05	.108-05
MANGANESE	< .413-08	< .675-08
MERCURY	< .413-07	< .675-07
MOLYBDENUM	.885-06	< .405-06
NICKEL	.142-06	.142-06
SELENIUM	< .590-06	< .675-06
TELLURIUM	< .354-06	< .405-06
THALLIUM	< .295-06	< .405-06
TIN	< .118-05	< .135-05
TITANIUM	< .236-05	< .338-05
URANIUM	* .000	* .000
VANADIUM	< .413-06	< .675-06
ZINC	.330-04	.358-04
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/JOULE) - UNIT OUTLET
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .398-06	< .366-06
ARSENIC	< .159-05	< .183-05
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CADMIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	* .398-06	< .183-06
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX E
TRACE ELEMENT CONCENTRATION -- MCG/DSCM

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET
DUST SAMPLE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .133	< .118
ARSENIC	.400	.473
BARIUM	< .934-01	.449
BERYLLIUM	.800-02	.343-01
BISMUTH	< .267-01	< .236-01
BORON	< 53.4	< 35.4
CADMIUM	.113	.839-01
CHROMIUM	3.47	2.48
COBALT	.667-02	.780-01
COPPER	2.13	.910
IRON	.280	2.25
LEAD	2.00	10.4
MANGANESE	< .133-02	.473-02
MERCURY	< .120-01	< .106-01
MOLYBDENUM	< .120	< .827-01
NICKEL	.400-01	.414
SELENIUM	< .400	< .236
TELLURIUM	< .133	< .946-01
THALLIUM	< .267	.354
TIN	< 4.00	33.1
TITANIUM	< .800	< .591
URANIUM	* .000	* .000
VANADIUM	.760	.532
ZINC	720.	744.
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MC6/DSCM) - UNIT OUTLET
XAD-2 CARTRIDGE

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< 3.64	< 3.80
ARSENIC	< 10.4	< 9.49
BARIUM	< 3.12	< 2.85
BERYLLIUM	< .104	< .949-01
BISMUTH	< 1.56	< 1.90
BORON	< .193+04	< .171+04
CADMIUM	< .208	< .190
CHROMIUM	< .260	< .285
COBALT	< .417-01	< .380-01
COPPER	14.6	31.3
IRON	57.3	71.2
LEAD	78.1	11.4
MANGANESE	< .417-01	< .380-01
MERCURY	2.29	20.4
MOLYBDENUM	< 4.69	< 4.27
NICKEL	< .417-01	< .380-01
SELENIUM	< 10.4	< 9.49
TELLURIUM	< 3.12	< 2.85
THALLIUM	< 10.4	< 9.49
TIN	< 10.4	< 9.49
TITANIUM	< 26.0	< 28.5
URANIUM	* .000	* .000
VANADIUM	20.8	47.5
ZINC	< 3.12	19.0
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET
FIRST IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .387	< .378
ARSENIC	< 1.93	< 2.27
BARIUM	.322	< .302
BERYLLIUM	< .129-01	< .151-01
BISMUTH	< .258	< .378
BORON	< 193.	< 227.
CADMIUM	12.9	.279
CHROMIUM	12.9	4.76
COBALT	< .645-02	< .151-01
COPPER	25.1	27.9
IRON	13.5	15.9
LEAD	1.93	1.21
MANGANESE	< .451-02	< .755-02
MERCURY	< .451-01	< .755-01
MOLYBDENUM	.967	< .453
NICKEL	.155	.159
SELENIUM	< .645	< .755
TELLURIUM	< .387	< .453
THALLIUM	< .322	< .453
TIN	< 1.29	< 1.51
TITANIUM	< 2.58	< 3.78
URANIUM	* .000	* .000
VANADIUM	< .451	< .755
ZINC	36.1	40.0
ZIRCONIUM	* .000	* .000

GAS TURBINE
TRACE ELEMENT CONCENTRATION - (MCG/DSCM) - UNIT OUTLET
2ND & 3RD IMPINGER

TRACE ELEMENT CATIONS	TEST CONDITION	
	BASELINE, NO WATER	WATER INJECTION
ANTIMONY	< .435	< .409
ARSENIC	< 1.74	< 2.05
BARIUM	* .000	* .000
BERYLLIUM	* .000	* .000
BISMUTH	* .000	* .000
BORON	* .000	* .000
CADMIUM	* .000	* .000
CHROMIUM	* .000	* .000
COBALT	* .000	* .000
COPPER	* .000	* .000
IRON	* .000	* .000
LEAD	* .000	* .000
MANGANESE	* .000	* .000
MERCURY	* .435	< .205
MOLYBDENUM	* .000	* .000
NICKEL	* .000	* .000
SELENIUM	* .000	* .000
TELLURIUM	* .000	* .000
THALLIUM	* .000	* .000
TIN	* .000	* .000
TITANIUM	* .000	* .000
URANIUM	* .000	* .000
VANADIUM	* .000	* .000
ZINC	* .000	* .000
ZIRCONIUM	* .000	* .000

APPENDIX F
ORGANIC ANALYSIS RESULTS

TABLE F-1. IR ANALYSIS REPORT

Contractor Acurex
 Sample Site Houston Lighting + Power Sample Acquisition Date 4-22-78
 Type of Source Organic extracts from XAD-2 resin, SASS Train,
 Test Number _____ Sample ID Number HLP-1-X
 Sample Description Baseline
 Responsible Analyst Harriette Atkins Date Analyzed 8-21-78
 Calculations and Report Reviewed By M Beth Heffernan Report Date 9-26-78
 Instrument Perkin-Elmer M. 297 Sample Cell Type Smear on NaCl plate
 Utilized Max/Min Signal Intensity Values _____
 Observations obviously, carboxylic acid and/or ester
and alcohol are major component

Wave Number (cm ⁻¹)	Intensity	Assignment	Comments
3400-3500	S	O-H	broad peak
2960, 2920, - 2850	S	C-H	aliphatic
1720-1700	S	C=O	carbonyl possibly ester
1600	w	C=C	unsaturated aromatic
1450	m	-CH ₃ bending	methyl groups
1370	m		possibly methyl
1260	S	-O- bending	ether/ester
1070-1090	m	unassigned	broad peak
800	w	unassigned	possibly aromatic substitution bands
710	w	unassigned	

TABLE F-2. IR ANALYSIS REPORT

Contractor Acurex
 Sample Site Houston L+P unit #52 Sample Acquisition Date 4-24-78
 Type of Source XAD-2 resin, SASS train turbine outlet
 Test Number _____ Sample ID Number HLP-2-X
 Sample Description condensed organic extracts from XAD-2 resin
Water addition
 Responsible Analyst Harriette Atkins Date Analyzed 8-21-78
 Calculations and Report Reviewed By M. Beth Heffernin Report Date _____
 Instrument Perkin Elmer M. 297 Sample Cell Type Screen on NaCl plate
 Utilized Max/Min Signal Intensity Values _____
 Observations obviously carboxylic acid and ester (or ketone)
principal constituent

Wave Number (cm ⁻¹)	Intensity	Assignment	Comments
3400-3500	s	O-H	broad peak
2960 2920			
- 2850	s	C-H	aliphatic
1690-1720	s	C=O	carbonyl, broad
1600	w	C=C	unsaturation, aromatic
1450, 1460	m	-CH ₂	methyl band
1380	m		possibly methyl
1260	s	-O-	ether/ester
1070-1100	s	unassigned	broad peak
800	m	unassigned	aromatic substituents
710, 700cm	m	unassigned	

04/27/79 16:44:00 CALI: C42779 035
SAMPLE: HLP-1-X 2 MLS. CONC. EXTRACT 7/12/78
RANGE: G 1.1200 LABEL: N 20.0 QUAN: A 0.1.0 BASE: U 20. 3

11712.

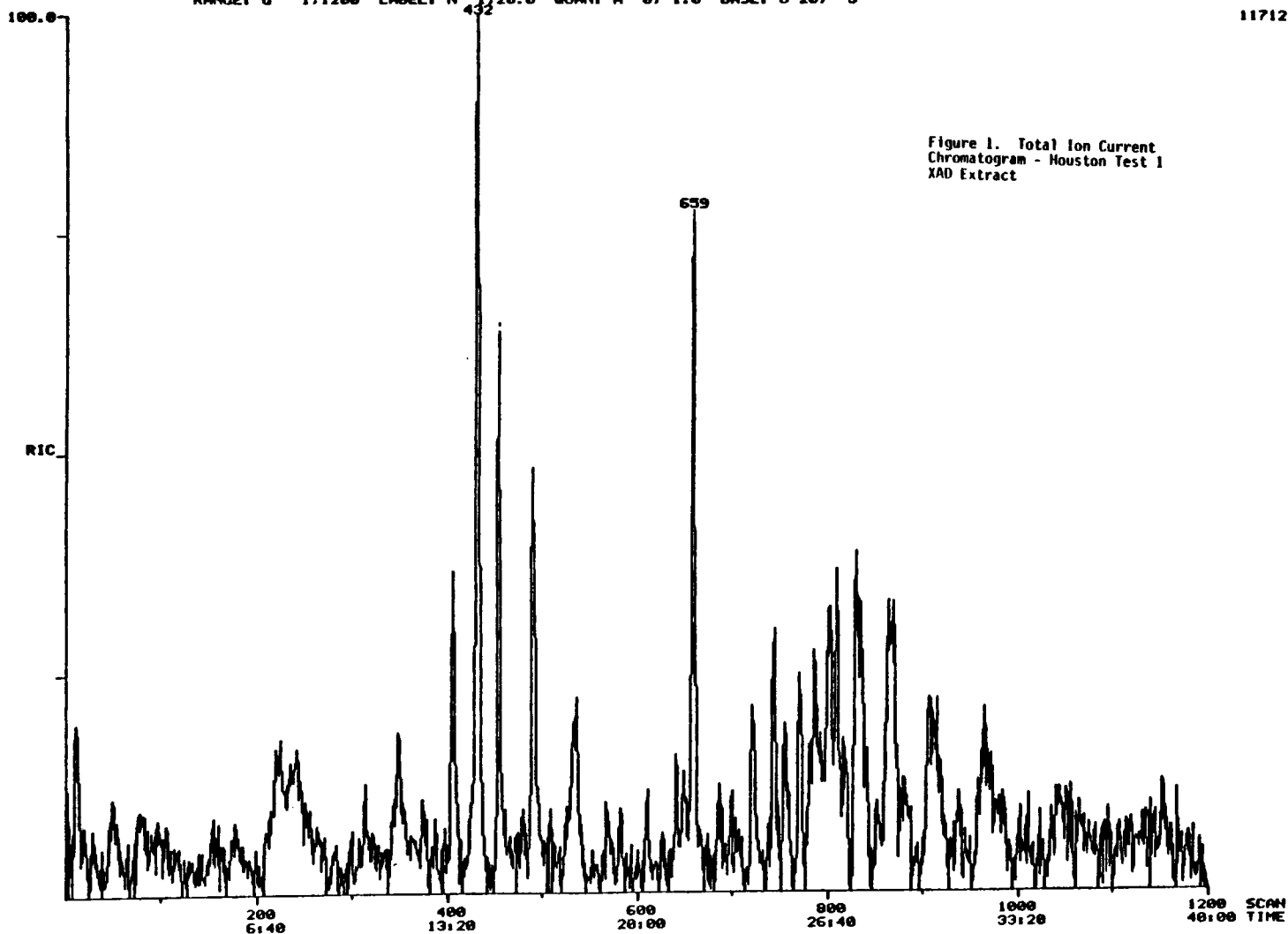


Figure F-1. Total ion current chromatogram -- Houston test 1 XAD extract.

F-6

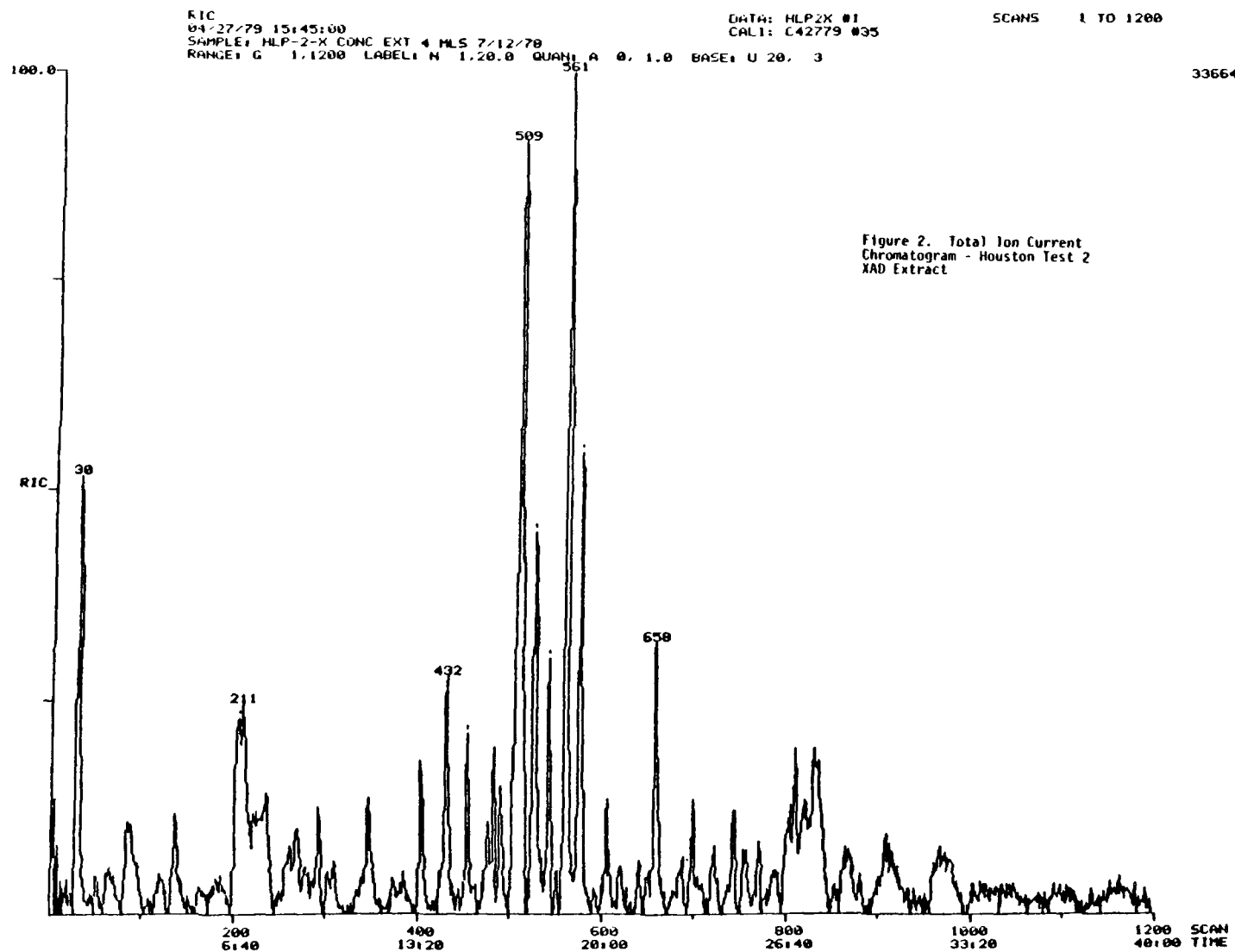


Figure F-2. Total ion current chromatogram -- Houston test 2 XAD extract.

APPENDIX G
GENERAL ELECTRIC TEST RESULTS

GENERAL ELECTRIC

GAS TURBINE
PRODUCTS DIVISION

GENERAL ELECTRIC COMPANY, ONE RIVER ROAD, SCHENECTADY, N. Y., U.S.A. 12345
Phone (518) 374-2211, Telex 145354

July 13, 1978

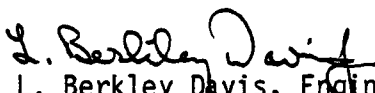
Mr. Brent Higginbotham
Acurex Aerotherm
485 Clyde Avenue
Mountain View, CA. 94042

Dear Brent:

Five copies of General Electric's report on the Acurex-GE joint test at Houston Lighting and Power are attached. These should provide sufficient data for your analysis, but don't hesitate to contact us if you have further questions.

Working with the Acurex test team was a pleasant experience. Perhaps there will be other opportunities for such tests.

Sincerely yours,


L. Berkley Davis, Engineer
Combustion Development-LGT
Bldg. 53 - Rm. 322

Attachments

LBD:rhb

cc: Nancy Fitzroy, 500-224
M. B. Hilt, 53-322

MS7001C FIELD TEST RESULTS
UNIT 52, HOUSTON LIGHTING AND POWER
APRIL 1978

During the recent field test of a MS7001C gas turbine, Unit 52, at Houston Lighting and Power, personnel from General Electric's Gas Turbine Division were responsible for measuring gaseous emissions and assessing turbine operation.

This report details the results of these tests and fulfills General Electric's reporting requirements to Acurex under contract RB68439A.

1. RESULTS

Some nineteen test points were run over a period from April 21 to 24, 1978. As indicated, these were spread over three days, with the first day

<u>DATE</u>	<u>TEST</u>
APRIL 21	PRELIMINARY TEST, VELOCITY TRAVERSE
APRIL 22	BASE LOAD, NO WATER INJECTION
APRIL 24	BASE LOAD, WATER INJECTION

devoted to a preliminary test to establish the velocity profile in the exhaust duct.

At each test point, gaseous emissions (O_2 , CO_2 , NO_x , NO, unburned hydrocarbons (UHC), and CO were measured at a single point in the exhaust duct.

Machine operating data, sufficient to establish the airflow rate and operating state of the machine, were also recorded.

The subsequent discussion in part III of this report will address the quality of results expected from single point sampling, as compared to those from a traverse.

Tables (1-2) list the data points for each of the three days. Data from a point are input to a data analysis program called FIRCAL9.

This program utilizes machine performance data, gaseous emissions, fuel composition, and machine geometry and internal flow splits to predict machine operational characteristics. For example, compressor inlet airflow is calculated using four methods; refer to Table (3) for an output sheet from FIRCAL9.

Each column of results (e.g., FT. TEST FLOW) makes use of certain of the data to calculate machine airflow, turbine inlet temperature and exhaust composition. This is illustrated in the table given below.

<u>COLUMN</u>	<u>MEASURED</u>	<u>CALCULATED</u>
FT. TEST FLOW	fuel flow and composition, airflow during factory test, inlet guide lane position, ambient conditions	machine airflow, O_2 , CO_2 , water in exhaust, turbine inlet temperature.
CHKD. FLOW	compressor discharge pressure and temperature, first stage nozzle area, fuel flow and composition.	machine airflow/turbine inlet temp., O_2 , CO_2 , H_2O
OXYGEN CONC.	O_2 , fuel flow and composition	machine airflow, CO_2 , H_2O , turbine inlet temperature
CO_2 CONC.	CO_2 , fuel flow and composition	machine airflow, turbine inlet temperature, O_2 , H_2O

Turbine temperatures are proprietary to General Electric and are not included in the results.

Results tables for each test point are interpreted as described in Table (3). The complete sets of results are given in Tables (3-17).

II. DISCUSSION

The core of the test results is contained in the test points 6-11 and 14-18. The first set is at base load with no water injection, while the second is at base load with 40 gpm water injection. The main points to note are the NO_x , $\text{NO}_x(\text{ISO})$, O_2 , and airflow rates.

Emissions of nitrogen oxides from a gas turbine are strongly affected by ambient humidity. This is taken into account using the relationship*

$$\text{NO}_x(\text{ISO}) = \text{NO}_x(\text{MEASURED}) e^{23.2(\text{H}-.0063)}$$

While measured NO_x varied considerably during the course of the dry test (pts. 6-11) in response to a cold front moving through, $\text{NO}_x(\text{ISO})$ changes about ± 1 percent. A similar statement applies to results from the water injection test (pts. 14-18).

* The constant 23.2 appearing in the exponential is derived from General Electric data. It gives numbers 2.7 percent higher than the EPA constant of 19.0.

NO_x EMISSIONS

<u>OPERATING POINT</u>	<u>MEAN NO_x(ISO) ppmv</u>	<u>MEAN O₂ (% by Vol., dry)</u>
BASE, DRY	159.1 \pm 1.0%	15.30 $\begin{matrix} +1.7\% \\ -3.5\% \end{matrix}$
BASE, 40 gpm	67.7 \pm 1.1%	15.12 \pm 1%

The variation^{*} in O₂ measurements during the dry test is somewhat larger than that in NO_x(ISO). This is apparently an outright measurement error in one point (pt. 8) that was taken during the period of severe weather.

Airflow in a gas turbine is directly affected by ambient variations and as the numbers in the tables vary by some \pm 2.5 percent with time. The mean levels (for, say, FT. TEST FLOW) are representative of the expected performance of the MS7001C axial compressor.

A comparison of the compressor inlet airflow values from each of the four calculation schemes (CHKD. FLOW, etc.) reveals excellent agreement (\pm 1.0%) between airflow from FT. TEST FLOW, CHKD. FLOW, and OXYGEN CONC. Note that pt. 8 is an exception (\pm 2.2%).

The airflow calculated from measured CO₂ (CO₂ CONC.) shows relatively poor consistency and agreement with the other values. This is particularly apparent in the water injection test, where the airflows predicted from measured CO₂ concentration are approximately ten percent higher than expected.

^{*}This variation is calculated as a deficit from 21 percent atmospheric O₂.

The large amounts of water vapor in the exhaust obviously affect the instrument.

The SO₂ values in the tables are calculated assuming one hundred percent conversion of fuel sulfur. This is consistent with General Electric's experience.

Two fuel analyses are given in Table (18). The variation in measured hydrogen is typical of that found from repeated measurements of distillate fuels. The ash content is, in both cases, higher than expected for distillates.

Table (19) gives a breakdown of the ash, with the primary constituent being an oxide of zinc. The hydrogen values in Table (19) are determined using the Galbraith method and are not as accurate as those in Table (18).

III. SUPPORTING TESTS

Subsequent to the Acurex tests, considerable effort was put into obtaining very accurate hydrocarbon measurements. In preparation for these tests, the stainless sampling probe was washed with acetone and methylene chloride, passivated with nitric acid, and washed again with methylene chloride.

In addition, an abbreviated stack traverse (18 pts.) was made.

There are two aspects of this portion of the testing of interest to Acurex: the hydrocarbon readings themselves, and the variation of NO_x and O₂ over the cross section of the exit duct.

Figures (1-3) give the results taken from a base load point with 40 gpm of water being injected. Each figure shows a definite profile across the stack, with a 3 NO_x variation of $\pm 5.9\%$ (± 3.3 ppmv). The point customarily used

for single point sampling is just above position D-4. These data indicate that the NO_x is approximately 2.2% lower than the mean value at that point. Thus, the readings reported in the previous section would be within two percent of the true mean.

Oxygen readings have a $\pm 3\sigma$ variation of $\pm 4.5\%$ (referenced to $21 - \overline{O}_2$), and verify the trends observed in NO_x. Unburned hydrocarbons are quite low (≤ 2 ppmv); they show a very wide spread (3σ is ± 79 percent).

Other points presented in the tables give supporting data. For example, points 12 and 19 are at base load, dry; they were taken just before and just after the water injection test. Note that NO_x(ISO) values are in excellent agreement with those previously obtained.

Machine performance and emissions measurements taken from Unit 52 at Houston Lighting and Power are both self-consistent and in agreement with data acquired from other machines of this class.

Close agreement between airflows from different methods of calculation give considerable confidence in the airflow levels. Any discrepancy between these values and that obtained during a stack traverse should be carefully reviewed.

Single point sampling should give adequate results for gaseous emissions such as NO_x and O₂. Measurement of species present at very low concentrations requires a traverse.

TABLE 1

GENERAL ELECTRIC CO. SCHENECTADY, N.Y.
GAS TURBINE TEST DATA

To Acurex

(12)

APPARATUS HOUSTON UNIT 52

TEST _____

CONDITIONS _____

DATA _____

POINT _____

SHEET _____

DATE 4-21-78BAR. 4-22-78 0 0

BAR. TIME _____

OBSERVER _____

DATE	4-21-78	4-21-78	4-21-78	4-21-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78	4-22-78			
TIME	17:35	18:45	19:20	19:45	15:17	17:08	18:00	20:00	21:00	22:00	23:00			
POINT	1	2	3	4	5	6	7	8	9	10	11			
P _{CD}	130.5		132.4		119.2	131.0	131.9	134.2	132.0	132.2	131.9			
T _{CD}	613.0		605	602	590	612	609	605	607	608	609			
S _{FEET}	3600		3600	3600	3600	3600	3600	3600	3600	3600	3600			
V _{CE}	7.4													
161	77°	77°	77°	77°	77°	77°	77°	77°	77°	77°	77°			
LOAD	59.3	59.0	60.0	60.0	40.0	61.0	62.0	61.8	62.0	63.0	62.0			
1-2														
3-4														
5-6														
7-8														
9-10														
11-12														
GrE	978	975	971	969		1002	999	995	996	999	999			
Grm	0		0		0	0	0	0	0	0	0			
TEMP	0		0						0					
1-2														
3-4														
5-6														
7-8														
9-10														
TEMP	72°F		66°F			73°F	72°F	73°F	70°F	72°F	72°F			
Time	1 min		1 min			1:04.5	1:04.1	2:06.5	2:08.3	2:07.5	2:07.9			
Cal	91 gm		92 gm			100	100	200	200	200	200			
THRM	196		199		180	198	198	200	200	199	198			
THRM	390		390		390	395	390	395	390	392	390			
T ₁														
T ₂														
T ₃														
TEMP														
TEMP														

TABLE 2

To Acurex

GENERAL ELECTRIC CO. SCHENECTADY, N.Y.
GAS TURBINE TEST DATA

(12)

APPARATUS HOUSTON UNIT 52

TEST _____

CONDITIONS _____

DATA _____

POINT _____

SHEET _____

DATE 4-24-78BAR. 0 PO

BAR. TIME _____

OBSERVER MATZONKE - RUSSELL

DATE	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78	4-24-78											
TIME	13:10	14:10	15:00	16:00	17:00	18:00	19:00												
POINT	12	13	14	15	16	17	18	19											
P ₀₁	129.5	130.2	130.5	130.1	129.9	130.9	132.2	131.8											
T ₀₁	619	624	627	626	628	624	616	617											
Speed	3600	3600	3600	3600	3600	3600	3600	3600											
V ₀₁																			
161	77°	77°	77°	77°	77°	77°	77°	77°											
162	60.0	60.0	61.0	61.0	61.0	61.5	63.0	62.0											
1-2																			
3-4																			
5-6																			
7-8																			
9-10																			
11-12																			
Ave	1001	1001	1001	1001	1001	1001	996	999											
61.1	0	20	40	40	40	40	40	0											
Temp																			
1-2																			
3-4																			
5-6																			
7-8																			
9-10																			
Temp	80°	80°	81°	80°	77°	74°	70°	70°											
Time	2:11.2	2:08.5	2:06.5	2:06.0	2:04.6	2:05.5	2:03.6	2:03.3											
Cal	200	200	200	200	200	200	200	200											
Flow	195	198	199	198	198	199	200	198											
Flow	395	-	385	385	390	385	390	395											
T ₁																			
T ₂																			
T ₃																			
Flow																			
Temp																			

04/22/78 12,52HOURS
CIRCLY REVISED 03/31/78
HOUSTON LIGHTING & POWER
MS7061C APRIL 1978

TABLE 3

OUTPUT	~ 65.00	HN
NOX	148.3	PMV(WET)
NOX(150)	152.0	PMV(WET)
NO	143.5	PMV(WET)
NO(150)	149.1	PMV(WET) EMISSIONS
O2	15.50	VOL(DRY)
CO2	3.97	VOL(DRY)
CO	5.0	PMV(DRY)
H2O	5.0	PMV(WET) CH4
H2O (INJECTION)	0	GMH
FUEL	10.769	PHS
WATER/FUEL	0	LWS/LQ
HHN FUEL CORR	0.004	X BY HT.

	FT TEST FLOW	CHUKU FLOW	OXYGEN CONC.	CO2 CONC.	
WEX(DRY) MSCF/Hr	26.232	26.443	26.536	26.9289	EXHAUST FLOW (10 ⁶ STD FT ³ /Hr)
WA PPS	570.3	574.8	576.9	585.3	COMPRESSOR INLET FLOW (lbm/SEC)
WAT(SO) PPS	589.6	596.2	596.8	605.1	" " " AT ISO AMBIENT (lbm/SEC)
I FTH DEG.F					
I NOZ DEG.F					
H2O X DE COMP	0.	0.	0.	0.	WATER INJECTION RATE (% OF WA)
F/A	0.6169	0.6187	0.6187	0.6184	MACHINE FUEL ADR RATIO
NOX PPH	424.2	493.0	499.9	507.0	} NOx (lbm/hr)
NOX(SO) PPH	516.5	510.7	512.6	519.9	
NOX 1% O2 LHV	173.2	173.4	174.0	176.4	ISO NOx @ 15% O2 (ppmv, dry)
NOX PPMBTU INPUT	0.65	0.66	0.66	0.67	} NOx (lbm/10 ⁶ BTU)
NOX(SO) PPMBTU	0.67	0.68	0.68	0.69	
E1 NOX PPTP FUEL	12.6	12.9	12.9	13.1	} EMISSIONS INDEX (lbm NOx / 10 ³ lbm FUEL)
E1 NOX(SO)	13.1	13.2	13.2	13.4	
ORG, NOX 100% YLD	1.5	1.5	1.5	1.5	ORGANIC NO (ppmv), at 100% yield of fuel bound nitrogen
O2 CONC AVBL DRY	15.45	15.46	15.50	15.53	} Exhaust gas composition
CO2 X VOL DRY	4.07	4.04	4.03	3.97	
CO AT 15% O2 DRY	5.4	5.4	5.5	5.5	
E1 CO PPTP FUEL	0.26	0.26	0.26	0.27	CO
SO2 PPMV DRY	19.1	19.1	19.0	18.8	calculation SO2, at 100% conversion of fuel sulfur
DRY TO WET VOL	1.049	1.049	1.049	1.048	exhaust moisture correction
PPM(EXHAUST)	28.43	28.57	28.86	28.80	mean molecular weight of exhaust gas
COHB, OPRF %					

TABLE 4

06/29/78 9.19HOURS

FIRCALY REVISED 03/30/78

HOUSTON LIGHTING & POWER

MS7001C APRIL 1976

TEST PT	3D-61MW	6	OUTPUT	61.00	M.
DATE	780422		NOX	131.5	PPMV (WET)
TIME	1657	HOURS	NOX (ISO)	158.0	PPMV (WET)
BAROMETER	29.740	INS. HG.	NO	126.1	PPMV (WET)
T (DRY)	74.4	DEG. F.	NO (ISO)	151.5	PPMV (WET)
T (WET)	69.1	DEG. F.	O ₂	15.43	VOL. (DRY)
REL. HUM.	77.0	%	CO ₂	4.04	VOL. (DRY)
ABS. HUM.	0.0142	LB/LB	CO	5.4	PPMV (DRY)
CO ₂	145.5	PPM	UNE	2.7	PPMV (WET) CH ₄
CDT	612.0	DEG. F.	H ₂ O	0.	GPA
I.G.V.	77.0	DEG. F.	FUEL	10.763	PPS
			WATER/FUEL	0.	LB/LB
COMB. EFF.	99.90	%	FM	0.008	% BY WT.

		FI. TEST	CHAL.	DAY SE.	CONC.
		FLOW	FLOW	CONC.	CONC.
WEX (DRY) MSCF/HR		25.7259	25.6775	26.2590	26.0525
WA	PPS	563.7	560.0	575.8	570.0
WA (ISO)	PPS	589.5	592.9	601.4	596.9
H ₂ O % OF COMP.		0.	0.	0.	0.
F/A		0.0191	0.0190	0.0187	0.0189
NOX	PPM	435.4	437.6	443.9	440.5
NOX (ISO)	PPM	523.0	525.8	533.3	529.3
NOX 15% O ₂ DRY		174.7	178.1	180.0	179.3
NOX PPMATJ 15% O ₂		1.57	1.57	1.57	1.56
NOX (ISO) PPMATJ		0.69	0.69	0.70	0.69
EI NOX PPTP FUEL		11.2	11.3	11.5	11.4
EI NOX (ISO)		13.5	13.6	13.8	13.7
ORG. NOX 100% YLD		3.1	3.1	3.0	3.0
O ₂ CONC % VOL. DRY		15.40	15.35	15.43	15.39
CO ₂ % VOL. DRY		4.14	4.12	4.10	4.09
CO AT 15% O ₂ DRY		5.8	5.8	5.9	5.8
EI CO PPTP FUEL		0.28	0.28	0.29	0.29
SO ₂ PPMV DRY		19.4	19.5	19.7	19.4
DRY TO WET VOL		1.062	1.062	1.061	1.062
MMW (EXHAUST)		26.74	26.74	26.74	26.74

TABLE 5

06/29/78 9.19HOURS
FIRCAL9 REVISED 03/30/78

HOUSTON LIGHTING & POWER
MS7001C APRIL 1978

TEST PT	40-627M	7	OUTPUT	62.00	MA
DATE	780422		NOX	130.3	PPMV (WET)
TIME	1802	HOURS	NOX (ISO)	157.3	PPMV (WET)
BAROMETER	29.780	INS. HG.	NO	124.4	PPMV (WET)
T (DRY)	71.2	DEG. F.	NO (ISO)	150.4	PPMV (WET)
T (WET)	68.5	DEG. F.	O2	15.44	VOL. (DRY)
REL. HUM.	87.5	%	CO2	4.04	VOL. (DRY)
ABS. HUM.	0.0144	LB5/LB	CO	5.3	PPMV (DRY)
ODP	140.5	PSIA	UHC	2.1	PPMV (WET) CH4
ODT	609.0	DEG. F.	H2O	0.	GGT
LOG. V.	71.5	DEG. F.	FUEL	10.831	PPS
COMB. EFF.	94.40	%	WATER/FUEL	0.	LB5/LB
			FEEL	0.006	W BT BT.
		FT. LBS	CH2O		CONC.
		FLOW	FLOW	CONC.	CONC.
WEX (DRY) NSCF/HK		25.9491	26.0559	26.4766	25.2149
MA	PPS	568.7	571.0	575.8	574.3
MA (ISO)	PPS	589.6	592.0	591.7	595.5
H2O % OF COMB.	0.	0.	0.	0.	0.
F/A	0.0190	0.0190	0.0187	0.0189	0.0189
NOX	PPM	432.3	437.0	443.7	439.5
NOX (ISO)	PPM	528.2	528.2	531.3	531.3
NOX 15% O2 DRY		177.2	177.2	180.5	176.9
NOX PPM AT 15% O2		0.57	0.57	0.57	0.57
NOX (ISO) PPM AT 15% O2		0.69	0.69	0.70	0.69
EI NOX WITH FUEL		11.2	11.2	11.4	11.3
EI NOX (ISO)		13.5	13.5	13.8	13.6
ORG. NOX 100% YEL		3.1	3.1	3.1	3.0
O2 CONC W/ VOL. DRY		15.33	15.35	15.44	15.39
CO2 % VOL DRY		4.13	4.11	4.05	4.04
CO AT 15% O2 DRY		5.8	5.8	5.9	5.9
EI CO PPM FUEL		0.29	0.29	0.29	0.29
SO2 PPMV DRY		19.6	19.5	19.2	19.4
DRY TO WET VOL		1.062	1.062	1.062	1.062
MMW (EXHAUST)		28.74	28.74	28.73	28.73

TABLE 6

06/29/78 9.19HOURS					
FIRCA 9 REVISED 03/30/78					
HOUSTON LIGHTING & POWER					
MS7001C APRIL 1978					
TEST PT	50-62MA	8	OUTPUT	62.00	MA
DATE	780422		NOX	138.4	PPMV(WET)
TIME	2007	HOURLY	NOX(ISO)	157.5	PPMV(WET)
BAROMETER	29.800	INS.HG.	NO	131.7	PPMV(WET)
T (DRY)	67.0	DEG.F.	NO(ISO)	149.9	PPMV(WET)
T (WET)	63.4	DEG.F.	O2	15.47	VOL. (DRY)
REL.HUM.	83.4	%	CO2	4.19	VOL. (DRY)
ABS.HUM.	0.0119	LB5/LB	CO	4.9	PPMV(DRY)
CDP	14.4	PSIA	CH4	1.5	PPMV(WET)CH4
CDT	605.0	DEG.F.	H2O	0.	GPI.
1.6.V.	77.0	DEG.F.	FUEL	11.002	PPS
COND.EFF.	99.00	%	WATER/FUEL	0.	LB5/LB
			FLOW	0.000	LB BY WT.
		IT. TEST	CHCL.	DAYGEN.	CONC.
WEX(DRY) MSCF/HK		FLOW	FLOW	CONC.	CONC.
WA	PPS	26.2866	26.5736	27.0390	25.9918
WA(ISO)	PPS	574.6	580.8	590.7	568.2
		569.9	595.9	596.1	563.0
		0.	0.	0.	0.
H2O % OF CONC.		0.0191	0.0186	0.0186	0.0194
F/A		466.3	471.2	474.1	461.2
NOX	PPM	530.5	536.4	545.4	525.1
NOX(ISO)	PPM	175.5	177.7	180.7	174.0
NOX 15% O2 DRY		0.60	0.60	0.61	0.59
NOX PPM(ISO) 15% O2		0.68	0.69	0.70	0.67
EI NOX PPM FUEL		11.8	11.9	12.1	11.6
EI NOX(ISO)		13.4	13.5	13.6	13.3
ORG. NOX 100% YEL		3.1	3.1	3.0	3.1
O2 CONC %VOL. DRY		15.30	15.36	15.47	15.25
CO2 % VOL DRY		4.14	4.10	4.13	4.19
CO AT 15% O2 DRY		5.2	5.2	5.3	5.1
EI CO PPM FUEL		0.25	0.26	0.26	0.25
SO2 PPMV DRY		19.7	19.4	19.1	19.9
DRY TO WET VOL		1.058	1.058	1.057	1.059
MMK(EXHAUST)		28.78	28.76	28.78	28.78

TABLE 7

06/29/78 9.19HOURS					
FIRCAL9 REVISID 03/30/78					
HOUSTON LIGHTING & POWER					
MST001C APRIL 1976					
TEST PT	60-62MW	9	OUTPUT	62.00	MW
DATE	780422		NOX	129.6	PPMV(WET)
TIME	2101	HOURS	NOX(ISO)	155.7	PPMV(WET)
BAROMETER	29.620	INS.HG.	NO	123.1	PPMV(WET)
T (DRY)	71.0	DEG.F.	NO(ISO)	147.9	PPMV(WET)
T (WET)	68.0	DEG.F.	O2	15.31	VOL (DRY)
REL.HUM.	86.1	%	CO2	4.11	VOL (DRY)
ABS.HUM.	0.0142	LBS/LB	CO	5.1	PPMV(DRY)
ODP	146.5	PSIA	UNO	2.1	PPMV(WET) CH4
ODT	607.0	DEG.F.	H2O	0.	GPM
I.G.V.	71.0	DEG.F.	FUEL	10.022	PPS
			WATER/FUEL	0.	LBS/LB
COMB.EFF.	94.50	%	FEI.	0.006	% BY WT.
		FT. TEST	CH. 2	OXYGEN	CO2
		FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HR		25.8254	26.0902	26.0406	26.0662
WA	PPS	563.9	571.6	566.1	571.0
WA(ISO)	PPS	589.0	595.5	589.9	594.9
H2O % OF COMB.		0.	0.	0.	0.
F/F		0.0191	0.0189	0.0191	0.0190
NOX	PPM	430.7	434.9	420.4	434.5
NOX(ISO)	PPM	517.5	522.6	517.1	522.1
NOX 15% O2 DRY		174.6	176.1	174.4	175.9
NOX PPM(T) 15% O2		0.56	0.57	0.55	0.57
NOX(ISO) PPM(T)		0.67	0.68	0.67	0.68
EI NOX PPM(T) FUEL		11.1	11.2	11.1	11.2
EI NOX(ISO)		13.4	13.4	13.3	13.4
ORG.NOX 100% YLD		3.1	3.1	3.1	3.1
O2 CONC %VOL DRY		15.32	15.37	15.31	15.30
CO2 % VOL DRY		4.15	4.11	4.14	4.11
CO AT 15% O2 DRY		5.4	5.5	5.4	5.5
EI CO PPM(T) FUEL		0.27	0.27	0.27	0.27
SO2 PPMV DRY		19.6	19.5	19.7	19.5
DRY TO WET VOL		1.062	1.062	1.062	1.062
MMW(EXHAUST)		26.74	26.74	26.74	26.74

TABLE 8

06/29/78 9.19HOURS FIRCAL 9 REVISED 04/30/78 HOUSTON LIGHTING & POWER M57001C APRIL 1976					
TEST PT	70-63MW	10	OUTPUT	63.00	MW
DATE	780422		NOX	131.6	PPMV(WET)
TIME	2200	HOURS	NOX(ISO)	158.4	PPMV(WET)
BAROMETER	29.720	INS.HG.	NO	124.5	PPMV(WET)
T (DRY)	70.0	DEG.F.	NO(ISO)	150.3	PPMV(WET)
T (WET)	68.0	DEG.F.	O2	15.36	VOL.(DRY)
REL.HUM.	90.5	%	CO2	4.04	VOL.(DRY)
ABS.HUM.	0.0144	LBS/LB	CO	5.3	PPMV(WET)
ODP	168.7	PSIA	UHC	2.9	PPMV(WET) CH4
ODI	608.0	DEG.F.	H2O	0.	GPM
L.G.V.	77.0	DEG.F.	FUEL	10.890	PDS
			WATER/FUEL	0.	LBS/LB
COMB.EFF.	94.90	%	FLH	0.008	% BY WT.
		FT. TEST	CH2D.	OXYGEN	CO2
		FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/H		25.9677	26.0787	26.2352	26.6873
WA PPS		569.1	571.5	574.8	584.5
WA(ISO) PPS		589.6	592.1	595.5	605.6
H2O % OF COMB.		0.	0.	0.	0.
F/A		0.0191	0.0191	0.0189	0.0186
NOX PPM		439.8	441.6	444.2	451.5
NOX(ISO) PPM		530.8	533.0	536.0	544.9
NOX 15% O2 DRY		177.3	178.5	179.4	182.4
NOX DRY PPM		0.57	0.57	0.58	0.58
NOX(ISO) PPM		0.69	0.69	0.69	0.71
EI NOX PPTH FUEL		11.2	11.3	11.3	11.5
EI NOX(ISO)		13.5	13.6	13.7	13.9
UPG NOX 100% YLD		3.1	3.1	3.1	3.0
O2 CONC %VOL DRY		15.31	15.33	15.36	15.40
CO2 % VOL DRY		4.15	4.13	4.11	4.06
CO AT 15% O2 DRY		5.6	5.6	5.7	5.8
EI CO PPTH FUEL		0.28	0.28	0.28	0.28
SO2 PPMV DRY		19.7	19.6	19.5	19.2
DRY TO WET VOL		1.062	1.062	1.062	1.061
MMW(EXHAUST)		28.74	28.74	28.73	28.73

TABLE 9

06/29/78 9.19HOURS
FIRCAL9 REVISED 04/30/78
HOUSTON LIGHTING & POWER
MS7001C APRIL 1978

TEST PT	8D-62MW	11	OUTPL1	62.00	MW
DATE	780422		NOX	130.7	PPMV(WET)
TIME	2255	HOURS	NOX(ISO)	155.4	PPMV(WET)
BAROMETER	29.760	INS.HG.	NO	125.1	PPMV(WET)
T (DRY)	73.8	DEG.F.	NO(ISO)	149.2	PPMV(WET)
T (WET)	68.5	DEG.F.	O2	15.40	VOL (DRY)
REL.HUM.	76.8	%	CO2	4.12	VOL (DRY)
ABS.HUM.	0.0135	LBS/LB	CO	5.0	PPMV(DRY)
ODP	146.5	PSIA	UHC	2.0	PPMV(WET),CH4
CDT	609.0	DEG.F.	H2O	0.	GPM
I.G.V.	77.0	DEG.F.	FUEL	10.856	PPS
			WATER/FUE	0.	LBS/LB
COMB.EFF.	95.90	%	FEH	0.008	% BY WT.
		FT. TEST	CH4 L.	O2 YG L.	CO2
		FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HR		25.7864	26.0476	26.3419	26.0841
WA	PPS	566.9	570.5	576.7	571.2
WA(ISO)	PPS	565.6	565.4	561.4	566.2
H2O - % OF COMP.	0.	0.	0.	0.	
F/A	0.0192	0.0190	0.0168	0.0190	
NOX	PPH	433.6	437.6	442.5	438.4
NOX(ISO)	PPH	517.0	522.0	527.6	522.7
NOX 15% O2 DRY		173.6	175.3	177.2	175.5
NOX PPTP IMP T		0.56	0.57	0.58	0.57
NOX(ISO) PPM LTO		0.67	0.68	0.69	0.68
E1 NOX PPTP FUEL		11.1	11.2	11.3	11.2
E1 NOX(ISO)		13.2	13.4	13.5	13.4
ORG.NOX 100% YLD		3.1	3.1	3.0	3.1
O2 CONC %VOL DRY		15.26	15.34	15.40	15.35
CO2 % VOL DRY		4.17	4.12	4.08	4.12
CO AT 15% O2 DRY		5.8	5.9	6.0	5.9
E1 CO PPTP FUEL		0.29	0.29	0.29	0.29
SU2	PPMV DRY	19.8	19.6	19.3	19.5
DRY TO WET VOL		1.062	1.061	1.061	1.061
MMW(EXHAUST)		28.75	28.75	28.74	28.74

TABLE 10

06/29/78 10.37HOURS

FISCALY REVISED 03/30/78

HOUSTON LIGHTING & POWER
MS7001C APRIL 1978

TEST PT	90-60MW 12	OUTPUT	60.00	MA
DATE	780424	NOX	135.8	PPMV (WET)
TIME	1306 HOURS	NOX (ISO)	158.7	PPMV (WET)
BAROMETER	29.850 INS.HG.	HC	130.1	PPMV (WET)
T (DRY)	83.0 DEG.F.	NO (ISO)	152.1	PPMV (WET)
T (WET)	70.2 DEG.F.	O2	15.46	VOL. (DRY)
REL.HUM.	53.5 %	CO2	3.94	VOL. (DRY)
ABS.HUM.	0.0130 LBS/L	CO	5.5	PPMV (WET)
ODP	144.1 PSIA	UIC	4.1	PPMV (WET) CH4
CDT	619.0 DEG.F.	H2O	0.	GF
I.G.V.	77.0 DEG.F.	FUEL	10.340	P/S
		FUELR/FUEL	0.	LBS/LB
COMB.EFF.	99.90 %	HH	0.005	% BY WT.

	FT. TEST	COND.	DAYTIME	COND.
	FLOW	FLOW	CONC.	CONC.
WEX (DRY) MSCF/HK	25.3430	25.7001	25.8098	26.5054
WA	PPH	554.5	562.2	565.5
WA (ISO)	PPH	589.0	597.7	601.7

H2O	% OF COMB.	0.	0.	0.	0.
F/A		0.0190	0.0108	0.0180	0.0182
NOX	PPH	441.9	447.5	450.7	461.4
NOX (ISO)	PPH	516.5	523.4	526.1	539.2
NOY	1% O2 DRY	157.0	161.6	157.0	160.3
NOX PPMV (WET)		3.25	3.31	3.33	3.42
NOX (ISO) PPMV		0.69	0.70	0.70	0.72
EI NOX PPMV FUEL		11.6	11.8	11.9	12.2
EI NOX (ISO)		13.6	13.1	13.7	14.2
URG. NOX 100% YEL		3.1	3.0	3.0	2.9
O2 CONC %VOL DRY		15.42	15.43	15.46	15.59
CO2	%VOL DRY	4.12	4.06	4.07	3.94
CO AT 15% O2 DRY		5.9	5.9	5.9	6.1
EI CO PPMV FUEL		0.29	0.29	0.29	0.30
SO2	PPMV DRY	19.3	19.3	19.1	18.7
DRY TO WET VOL		1.060	1.059	1.059	1.058
MMW (EXHAUST)		26.76	26.76	26.76	26.76

TABLE 11

06/29/78 10.37HOURS
 FIRCAL9 REVISED 04/30/78
 HOUSTON LIGHTING & POWER
 MS7001C APRIL 1976

TEST PT	100-60MW-20GPM	13	OUTPUT	60.00	M.
DATE	780424		NOX	89.7	PPMV(WET)
TIME	1400	HOURS	NOX(ISO)	106.6	PPMV(WET)
BAROMETER	29.830	INS.HG.	NO	84.0	PPMV(WET)
T (DRY)	84.5	DEG.F.	NO(ISO)	99.8	PPMV(WET)
T (WET)	71.5	DEG.F.	O2	15.29	VOL.(DRY)
REL.HUM.	53.6	%	CO2	3.85	VOL.(DRY)
ABS.HUM.	0.0137	LBS/LB	CO	6.1	PPMV(DRY)
ODP	144.1	PSIA	URE	2.4	PPMV(WET) CH4
CDT	624.0	DEG.F.	H2O	20.0	GA.
I.G.V.	77.0	DEG.F.	FUEL	10.768	PPS
COMB.EFF.	99.90	%	WATER/FUEL	0.26	LBS/LB
			EBL	0.008	% BY WT.
ET TEST		CFR	DAY	CO2	
FLOW		FLOW	CONC.	CONC.	
WEX(DRY) MSCF/HR	25.2109	25.6333	45.7423	27.8290	
WA	PPS	552.2	561.3	563.8	608.7
WA(ISO)	PPS	569.4	599.3	602.0	649.8
H2O % OF COM.	0.55	0.56	0.49	0.46	
F/A	0.0195	0.0192	0.0191	0.0177	
NOX PPM	252.3	258.0	255.2	322.4	
NOX(ISO) PPM	348.5	354.0	325.4	323.0	
NOX 15% O2 DRY	117.5	119.3	116.8	129.0	
NOX PPM(T) INLET	0.32	0.32	0.27	0.32	
NOX(ISO) PPM(T)	0.46	0.46	0.47	0.50	
EI NOX PPTP FUEL	7.6	7.7	7.7	8.3	
EI NOX(ISO)	9.0	9.1	9.2	9.9	
URG.NOX 100% YLT	3.1	3.1	3.1	2.8	
O2 CONC %VOL DRY	15.17	15.27	15.29	15.72	
CO2 % VOL DRY	4.25	4.15	4.10	3.85	
CO AT 15% O2 DRY	6.3	6.4	6.4	6.9	
EI CO PPTP FUEL	0.31	0.32	0.32	0.34	
SO2 PPMV DRY	20.1	19.8	19.7	18.2	
DRY TO WET VOL	1.071	1.070	1.070	1.066	
MMW(EXHAUST)	28.65	28.65	28.65	28.67	

TABLE 12

06/29/78 10.37HOURS

FIRCAL9 REVISED 04/30/78

HOUSTON LIGHTING & POWER
MS/001C APRIL 1976

TEST PT	110-61NW-40GPM	14	OUTPUT	61.00	HW
DATE	780424		NOX	57.5	PPMV (WET)
TIME	1501	HOURS	NO ₂ (ISO)	67.8	PPMV (WET)
BAROMETER	29.810	INS. HG.	NO	53.4	PPMV (WET)
T (DRY)	85.0	DEG. F.	NO (ISO)	62.4	PPMV (WET)
T (WET)	71.2	DEG. F.	O ₂	15.14	VOL. (DRY)
REL. HUM.	51.4	%	CO ₂	4.00	VOL. (DRY)
ABS. HUM.	0.0135	LBS/LB	CO	7.4	PPMV (DRY)
ODP	145.1	PSIA	UFC	4.3	PPMV (WET) CH ₄
CDT	627.0	DEG. F.	H ₂ O	40.0	GRM
J.G.V.	77.0	DEG. F.	FUEL	10.955	PPS
			WATER/FUE	0.21	LBS/LB
COMB. EFF.	99.90	%	FEI	0.008	% BY WT.
		FT. TEST	CH ₄	DAYGRI	CO ₂
		FLOW	FLOW	CONC.	CONC.
WEX (DRY) MSCF/HK	25.1664		25.5431	45.6357	27.3753
WA PPS	551.2		554.3	581.7	599.1
WA (ISO) PPS	559.6		596.2	600.8	640.7
H ₂ O % OF COMB.	1.01		0.99	0.99	0.93
F/A	0.0199		0.0196	0.0195	0.0183
NOX PPS	185.6		192.1	172.8	205.1
NOX (ISO) PPS	223.2		226.4	247.1	241.7
NOX 15% O ₂ DRY	73.5		74.6	74.4	79.7
NOX PPMBTU (WET)	0.24		0.25	0.23	0.26
NOX (ISO) PPMBTU	0.29		0.29	0.29	0.31
EI NOX PPTP FUEL	4.8		4.9	4.9	5.2
EI NOX (ISO)	5.7		5.7	5.8	6.1
URG. NOX 100% YLL	3.1		3.1	3.1	2.9
O ₂ CONC %VOL. DRY	15.03		15.12	15.14	15.51
CO ₂ % VOL DRY	4.35		4.29	4.27	4.00
CO AT 15% O ₂ DRY	7.9		8.0	8.0	8.6
EI CO PPTP FUEL	0.40		0.40	0.41	0.43
SO ₂ PPMV DRY	20.6		20.3	20.2	18.9
DRY TO WET VOL	1.080		1.074	1.079	1.075
MMW (EXHAUST)	28.56		28.56	28.58	28.59

TABLE 13

06/29/78 10.37HOURS					
FISCALY REVISED 03/30/78					
HOUSTON LIGHTING & POWER					
MS7001C APRIL 1978					
TEST PT	120-61HW-40GPM	15	OUTPUT	61.00	MA
DATE	780424		NOX	57.8	PPMV(WET)
TIME	1600	HOURS	NOX(ISO)	67.8	PPMV(WET)
BAROMETER	29.790	INS.HG.	NO	53.7	PPMV(WET)
T (DRY)	83.5	DEG.F.	NO(ISO)	63.0	PPMV(WET)
T (WET)	70.5	DEG.F.	O2	15.13	VOL.(DRY)
REL.HUM.	53.1	%	CO2	3.92	VOL.(DRY)
ABS.HUM.	0.0131	LBS/LB	CO	8.1	PPMV(DRY)
CDP	144.7	PSIA	UHC	2.4	PPMV(MULTI)CH4
CDT	626.0	DEG.F.	H2O	40.0	GPI
LEG.V.	77.0	DEG.F.	FUEL	10.481	PPS
			WATER/FUEL	0.21	LBS/LB
COMB.EFF.	99.90	%	FLUE	0.006	% BY WT.
		FT. LIST	CHCL	OXYGEN	CO2
		FLCH	FLCH	CONC.	CONC.
WEX(DRY) MSCF/HF		25.2426	25.4344	25.6521	28.0028
WA	PPS	552.5	554.9	552.0	612.4
WA(ISO)	PPS	569.6	594.6	599.4	653.2
H2O % OF COMP.		1.01	1.00	0.99	0.91
F/A		0.0199	0.0197	0.0195	0.0179
NOX	PPH	190.9	192.2	193.8	210.0
NOX(ISO)	PPH	223.4	225.5	247.3	247.0
NOX 15% O2 DRY		73.8	74.1	74.8	81.3
NOX PPTP FULL		0.25	0.25	0.22	0.27
NOX(ISO) PPTP FULL		0.29	0.29	0.29	0.32
EI NOX PPTP FULL		4.8	4.9	4.9	5.3
EI NOX(ISO)		5.7	5.7	5.8	6.2
URG.NOX 100% YLE		3.1	3.1	3.1	2.9
O2 CONC %VOL.DRY		15.03	15.00	15.13	15.62
CO2 % VOL DRY		4.35	4.32	4.23	3.92
CO AT 15% O2 DRY		8.1	8.2	8.3	9.0
EI CO PPTP FUEL		0.41	0.41	0.42	0.45
SO2 PPMV DRY		20.6	20.5	20.3	18.9
DRY TO WET VOL		1.079	1.079	1.078	1.074
MMW(EXHAUST)		26.56	26.57	26.59	26.60

TABLE 14

06/29/76 10.37HOURS
FIRCALY REVISED 04/30/78
HOUSTON LIGHTING & POWER
MST001C APRIL 1976

TEST PT	130-61ML-40GPM 16	OUTPUT	61.00	MW
DATE	780424	NOX	57.6	PPMV(WET)
TIME	1700 HOURS	NOX(ISO)	67.6	PPMV(WET)
BAROMETER	29.780 INS.HG.	NO	53.6	PPMV(WET)
I (DRY)	83.3 DEG.F.	NO(ISO)	62.9	PPMV(WET)
I (WET)	70.5 DEG.F.	O2	15.14	VOL.(DRY)
REL.HUM.	55.6 %	CO2	3.93	VOL.(DRY)
ABS.HUM.	0.0132 LBS/LB	CO	7.8	PPMV(DRY)
CDP	164.5 PSIA	UIC	4.1	PPMV(WET) CH4
CDT	628.0 DEG.F.	H2O	40.0	GPM
I.G.V.	77.0 DEG.F.	FUEL	10.944	PPS
		WATER/FUEL	0.21	LBS/LB
COMB.EFF.	99.90 %	FEI	0.008	% BY WT.

	FT. TEST	CHCL	OXYGEN	CO2
	FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HR	25.2463	25.4036	25.6080	27.8356
WA	PPH 552.9	556.2	561.0	608.8
WA(ISO)	PPH 589.6	593.2	598.3	649.3

H2O % OF COMP.	1.01	1.00	0.99	0.91
F/A	0.0196	0.0197	0.0195	0.0180
NOX	PPH 190.1	191.2	192.7	206.5
NOX(ISO)	PPH 223.3	224.6	226.3	244.9
NOX 15% O2 DRY	74.6	74.1	74.7	80.7
NOX PPM(T) 15% O2	0.25	0.25	0.25	0.27
NOX(ISO) PPM(T)	0.29	0.29	0.29	0.32
EI NOX PPTP FUEL	4.8	4.9	4.9	5.3
EI NOX(ISO)	5.7	5.7	5.7	6.2
URG.NOX 100% YLL	3.1	3.1	3.1	2.9
O2 CONC %VOL.DRY	15.13	15.09	15.14	15.61
CO2 % VOL DRY	4.33	4.31	4.27	3.93
CO AT 15% O2 DRY	8.0	7.9	8.0	8.7
EI CO PPTP FUEL	0.40	0.40	0.40	0.44
SO2 PPMV DRY	20.3	20.4	20.2	18.6
DRY TO WET VOL	1.074	1.079	1.076	1.074
MMW(EXHAUST)	28.56	28.56	28.59	28.60

TABLE 15

06/29/78 10.37HOURS

FIRCALLY REVISED 03/30/78

HOUSTON LIGHTING & POWER
MS7001C APRIL 1978

TEST PT	140-61.5MW-40GPM	17	OUTPUT	61.50	MV
DATE	780424		NOX	55.8	PPMV(WET)
TIME	1800	EDUPS	NOX(ISO)	66.4	PPMV(WET)
BAROMETER	29.760	INS.HG.	NO	51.8	PPMV(WET)
T (DRY)	81.2	DEG.F.	NO(ISO)	62.1	PPMV(WET)
T (WET)	71.0	DEG.F.	O2	15.10	VOL (DRY)
REL.HUM.	61.2	%	CO2	3.87	VOL (DRY)
ABS.HUM.	0.0141	LBS/LB	CO	8.0	PPMV(DRY)
CDP	165.6	PSI	UNO	4.1	PPMV(WET) CH4
CDT	624.0	DEG.F.	H2O	40.0	GPM
I.G.V.	77.0	DEG.F.	FULL	11.054	PPS
CONC.EPI.	99.40	%	WATER/FUEL	0.26	LBS/LB
			FIR	0.006	% BY WT.
	FT. INT	CHL	DAY 1	CO2	
	FLOW	FLOW	CONC.	CONC.	
WEX(DRY) MSCF/HK	25.3440	25.5269	25.6873	28.5513	
WA	555.5	559.5	553.3	624.9	
WA(ISO)	559.0	553.8	547.6	663.1	
H2O % OF COMP.	1.00	0.99	0.99	0.89	
F/A	0.0199	0.0195	0.0196	0.0177	
NOX	165.6	166.5	147.6	207.4	
NOX(ISO)	222.0	223.5	244.8	248.6	
NOX 15% O2 DRY	73.5	73.0	73.5	81.3	
NOX PPM TO 15% O2	0.24	0.24	0.24	0.26	
NOX(ISO) PPM TO 15% O2	0.28	0.29	0.29	0.32	
EI NOX PPM FUEL	4.7	4.7	4.7	5.2	
EI NOX(ISO)	5.6	5.6	5.7	6.2	
UNO NOX 100% YLD	3.1	3.1	3.1	2.8	
O2 CONC %VOL DRY	15.10	15.06	15.10	15.69	
CO2 3 VOL DRY	4.36	4.33	4.32	3.87	
CO AT 15% O2 DRY	8.2	8.1	8.2	9.1	
EI CO PPM FUEL	0.41	0.41	0.41	0.46	
SOX PPM DRY	20.4	20.5	20.4	18.3	
DRY TO WET VOL	1.081	1.081	1.080	1.074	
MMW(EXHAUST)	28.55	28.55	28.57	26.59	

TABLE 16

06/29/78 10.37HOURS

EIRCALY REVISED 03/30/78

HOUSTON LIGHTING & POWER
MS7001C APRIL 1978

TEST PT	15D-63HF-40GPM	18	OUTPUT	63.00	M.
DATE	780424		NOX	56.1	PPMV(WET)
TIME	1400	HOURS	NOX(ISO)	68.5	PPMV(WET)
BAROMETER	29.780	INS.HG.	NO	51.9	PPMV(WET)
T (DRY)	77.0	DEG.F.	NO(ISO)	63.4	PPMV(WET)
T (WET)	70.4	DEG.F.	O2	15.07	VOL.(DRY)
REL.HUM.	74.2	%	CO2	3.48	VOL.(DRY)
ABS.HUM.	0.0149	LBS/LB	CO	7.9	PPMV(DRY)
CDP	146.8	PSIA	UNE	2.8	PPMV(WET), CH4
CDT	616.0	DEG.F.	H2O	40.0	GPM
I.G.V.	77.0	DEG.F.	FUEL	11.244	PPS
			WATER/FUEL	0.49	LBS/LB
COMB.EFF.	99.90	%	FEH	0.008	% BY WT.
		FT. TEST	CH4	0.97	CONC.
		FLOW	CONC.	CONC.	CONC.
WEX(DRY) MSCF/HK	25.5687	25.7321	25.9413	28.2305	
W3	561.0	564.8	570.5	618.7	
W4(ISO)	589.0	593.3	599.6	650.1	
H2O % OF COMB.	0.009	0.009	0.009	0.009	
F/A	0.0200	0.0199	0.0197	0.0182	
NOX	125.1	124.2	121.0	206.6	
NOX(ISO)	229.8	231.2	233.4	252.4	
NOX 15% O2 DRY	74.9	74.3	75.0	81.1	
NOX PPTP FULL	0.24	0.24	0.24	0.26	
NOX(ISO) PPTP FULL	0.29	0.29	0.29	0.32	
EI NOX PPTP FULL	4.6	4.7	4.7	5.1	
EI NOX(ISO)	5.7	5.7	5.8	6.2	
ORG. NOX 100% YLE	3.2	3.2	3.1	2.9	
O2 CONC %VOL DRY	15.06	15.01	15.07	15.54	
CO2 % VOL DRY	4.39	4.37	4.32	3.98	
CO AT 15% O2 DRY	8.0	7.9	8.0	8.7	
EI CO PPTP FUEL	0.40	0.40	0.41	0.44	
SO2 PPMV DRY	20.5	20.7	20.5	18.9	
DRY TO WET VOL	1.063	1.062	1.082	1.077	
MMW(EXHAUST)	28.54	28.54	28.56	28.57	

TABLE 17

06/29/78 10.37HOURS

FISCALY REVISED 03/30/78

HOUSTON LIGHTING & POWER
M57001C APRIL 1978

TEST PT	160-62MW 19	OUTPUT	62.00	MW
DATE	780424	NOX	130.8	PPMV(WET)
TIME	2000 HOURS	NOX(ISO)	158.3	PPMV(WET)
BAROMETER	29.820 INS.HG.	NO	123.6	PPMV(WET)
T (DRY)	73.5 DEG.F.	NO(ISO)	149.6	PPMV(WET)
T (WET)	69.3 DEG.F.	O2	15.35	VOL.(DRY)
REL.HUM.	81.4 %	CO2	3.82	VOL.(DRY)
ABS.HUM.	0.0145 LBS/LB	CO	5.1	PPMV(WET)
ODP	146.4 PSIA	UNO	1.2	PPMV(WET) CH4
ODT	612.0 DEG.F.	H2O	0.	GPV
I.G.V.	77.0 DEG.F.	FUEL	10.832	PPS
		WATER/FUEL	0.	LBS/LB
COND.EFF.	99.90 %	FEI	0.008	% BY WT.

	FT. TEST	CH4	DAY 1	CO2
	FLOW	FLOW	CONC.	CONC.
WEX(DRY) MSCF/HK	25.8436	26.0124	26.0477	26.0818
WA PPS	568.4	570.1	570.8	614.5
WA(ISO) PPS	589.6	592.4	594.1	639.6

H2O % OF COND.	0.	0.	0.	0.
F/A	0.0191	0.0190	0.0190	0.0170
NOX PPM	432.1	437.5	438.4	471.4
NOX(ISO) PPM	526.7	530.0	530.8	570.8
NOX 15% O2 DRY	177.2	178.7	178.5	191.9
NOX PPM(T) 15% O2	0.57	0.57	0.57	0.61
NOX(ISO) PPM(T)	0.69	0.69	0.69	0.74
EI NOX PPTP FUEL	11.2	11.2	11.2	12.1
EI NOX(ISO)	13.5	13.6	13.6	15.6
ORG.NOX 100% YLP	3.1	3.1	3.1	2.8
O2 CONC %VOL.DRY	15.31	15.34	15.35	15.76
CO2 % VOL DRY	4.15	4.12	4.12	3.82
CO AT 15% O2 DRY	6.0	6.0	6.0	6.5
EI CO PPTP FULL	0.29	0.30	0.30	0.32
SO2 PPMV DRY	19.7	19.5	19.5	18.1
DRY TO WET VOL	1.063	1.062	1.062	1.060
MMW(EXHAUST)	28.73	28.73	28.73	28.73

TABLE 18
MATERIALS AND PROCESSES LABORATORY
SCHENECTADY, NEW YORK
ANALYSIS REQUEST-REPORT

RECEIVED 5/24/78 LAB. NO. 78C-1068
REPORTED 5/31/78 SPECTRUM NO. _____
REQUESTED BY R.A. Hiskam BLDG. 262 RM. 105 EXT. _____
PROJECT NO. _____
SHOP ORDER 5910-440-309-114 DEPT. _____ CC. _____

DESCRIPTION OF MATERIAL Gas Turb. Lab
Fuel Type #2 (4-26) HL&P
SH 209

Analysis Requested	Wet	X-Ray		Spectro		Probe	Organic	Thermal	Gas	Part. Study	Other
		AA	emiss	diff	qual						
	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>

Met: HHV, ash, H X-ray: S Evendale: H McBride: Spec. Grav. & Visc. @ 100°F
% water 210°F, C residue, Distillation curve,
Galbraith: C/H ratio aniline pt.

RESULTS	HHV (BTU/lb)	19,730
	ppm ash	32
	% H	0.008
	% S	0.11
	% N	13.55
	% H ₂ O	<0.02
	Sp. Gr. @ 100°F	0.820
	Sp. Gr. @ 210°F	0.782
	Viscosity @ 100°F	2.55
	Viscosity @ 210°	1.07
	Aniline Point °F	156
	Carbon Residue (10% Bottoms)	0.18
	Distillation °F	°F
	Initial Point	326 50% Distilled 482
	10% Distilled	396 60% " 502
	20% "	422 70% " 526
	30% "	440 80% " 550
	40% "	464 90% " 576

Inquiries should be directed to: Dick McIlroy/ Bob McBride Ext. 5-2113

TABLE 18 CONCLUDED

RECEIVED	5/24/78		LAB. NO.		78C-1069					
PORTED	5/31/78		SPECTRUM NO.							
REQUESTED BY	R.A. Miskarin		BLDG.	262	RM.	105 EXT.				
PROJECT NO.										
SHOP ORDER	5818-440-300-113		DEPT.			CC.				
DESCRIPTION OF MATERIAL										
Gas Turb. Lab										
Fuel Type #2 (4-22) HL&P										
SN 208										
Analysis Requested	Wet	AA	X-Ray	Spectro	Probe	Organic	Thermal	Gas	Part.	Other
	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>
Wet: HHV, ash, N X-ray: S Evendale: H McBride: Spec. Grav. & Visc. @ 100°F										
% water 210°F, C residue, Distillation curve.										
Galbrath: C/H ratio aniline pt.										
RESULTS	HHV (BTU/lb)		19,665							
	ppm ash		23							
	ZN		0.009							
	ZS		0.11							
	ZH		13.40							
	ZH ₂ O		<0.02							
	Sp. Gr. @ 100°F		0.823							
	Sp. Gr. @ 210°F		0.782							
	Viscosity @ 100°F		2.59							
	Viscosity @ 210°F		1.03							
	Aniline Point		155°F							
	Carbon Residue (10% bottoms)		0.29							
	Distillation		°F						°F	
	Initial Point		324		50% Distilled				487	
	10% Distilled		401		60% "				508	
	20% "		428		70% "				528	
	30% "		448		80% "				552	
	40% "		466		90% "				560	

Inquiries should be directed to: Dick Northrup / Bob McBride Ext. 5-2113

TABLE 19
MATERIALS AND PROCESSES LABORATORY
SCHENECTADY, NEW YORK
ANALYSIS REQUEST-REPORT

LAB. NO. 78C-1068
SPECTRUM NO. BF1
DATE 1/17/78 BLDG. 262 RM. 105 EXT.
ANALYST E.A. Hiskam DEPT. CC.
SOURCE OF MATERIAL Gas Turb. Lab
Fuel Type #2 (4-26) HL&P
SN 289
ANALYSIS REQUESTED: ☐ Wet ☐ AA ☐ X-Ray ☐ emiss ☐ diff ☒ Spectro ☐ qual ☐ quant ☐ Probe ☐ Organic ☐ Thermal ☐ Gas ☐ Part. ☒ Study ☐ Other

Galbraith: C/H ratio

RESULTS	approx. % in ash		
	Zn	major	>10
	Pb		0.3
	Sn		0.2
	Si		0.2
	Fe		0.2
	Cu		0.1
	Ca		0.04
	Mg		0.02
	V		0.01
	Mo		0.01
	Al		0.009
	N		0.008
	Mn		0.004
	Ni		0.003
	C		85.07
	H		13.23

Inquiries should be directed to: BH Kint

Ext. 5-2113

TABLE 19 CONCLUDED

RECEIVED 5/24/78 LAB. NO. 79C-1069
 REPORTED 6/7/78 SPECTRUM NO. BF1
 REQUESTED BY R.A. Miskunin BLDG. 262 RM. 105 EXT.
 PROJECT NO. DEPT. CC.
 SHOP ORDER 5818-440-300-113

DESCRIPTION OF MATERIAL Gas Turb. Lab
Fuel Type #2 (4-22) HL&P
SN 288

Analysis Requested	X-Ray				Spectro		Probe	Organic	Thermal	Gas	Part.	
	Wet	AA	emiss	diff	qual	quant					Study	Other
<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	<input checked="" type="checkbox"/>

Galbraith: C/H ratio

RESULTS	approx. % in ash
Zn	major >10
Pb	0.2
Fe	0.2
Sn	0.1
Cu	0.1
V	0.05
Si	0.04
Ca	0.03
Ni	0.02
Mn	0.01
Mo	0.006
Al	0.005
Mn	0.002
C	84.83
H	13.07

Inquiries should be directed to: BH Kelt Ext. 5-2113

DATA PRESENTED ARE
MEANS (\bar{x}) AND STANDARD
DEVIATIONS AT EACH
SAMPLE POINT, AND
THE MEAN (\bar{X}) AND
DEVIATION FOR THE
ENTIRE STACK.
ALL QUANTITIES ARE
IN PPM BY VOLUME

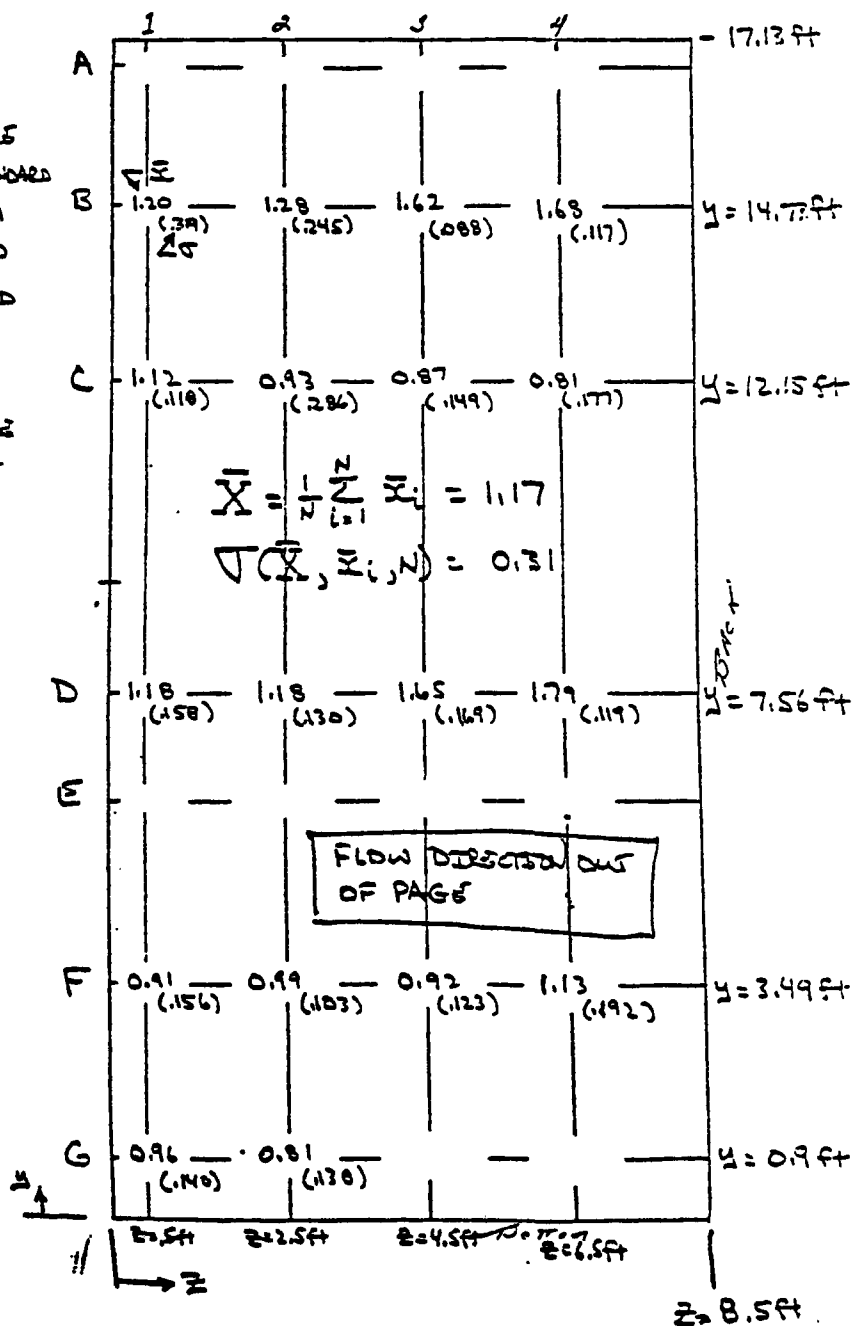


Figure 1. M57001C total hydrocarbon emissions at base load with water injection.

DATA ARE MEAN
READINGS AT EACH SAMPLE
POINT. STANDARD
DEVIATIONS FOR THESE
READINGS ARE LESS
THAN 0.01.

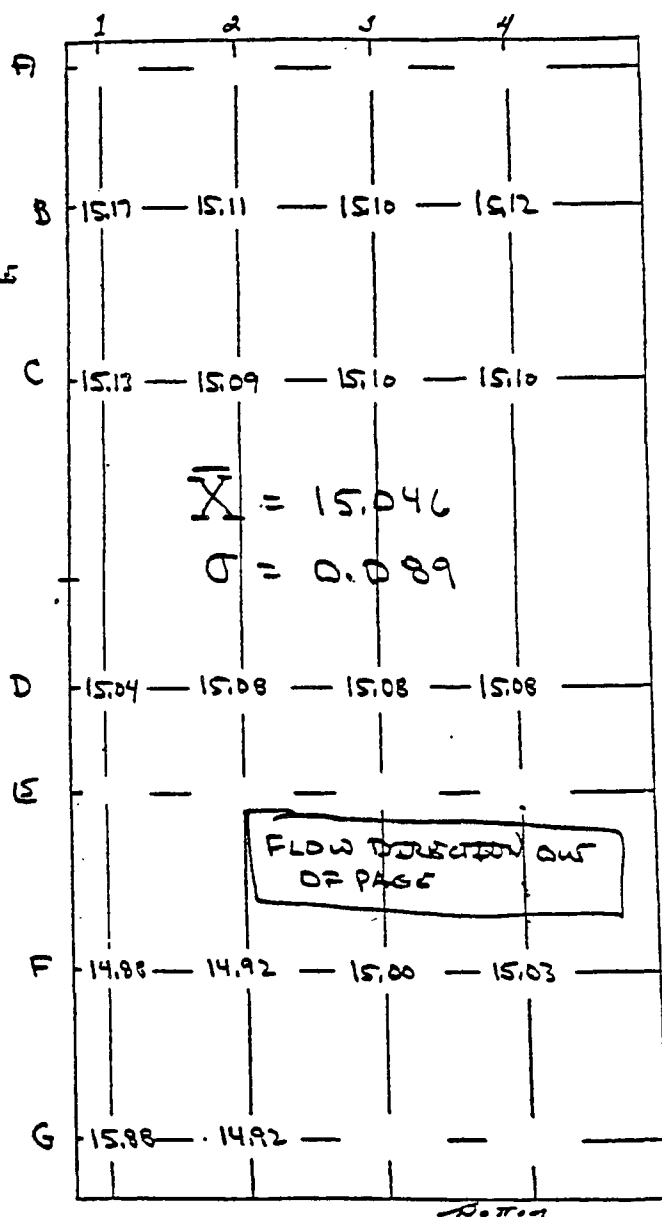


Figure 2. M57001C oxygen concentration in exhaust at base load with water injection.

FACING
AGAINST AIR
FLOW

NO₂ DATA

As P/2nd.

PPMV

THIS PORT
is 40" up Section

$$\bar{x} = 57.19$$

$$s = 1.113$$

$$\frac{s}{\bar{x}} = .0195$$

2nd 2d

40 ppm

40 PPM

	1	2	3	4
A				
B	57.35	58.29	58.45	59.10
C	56.25	57.89	58.59	57.87
D	57.82	55.47	55.36	55.62
E				56.52
F	56.86	56.80	57.67	56.62
G	56.19	56.13		

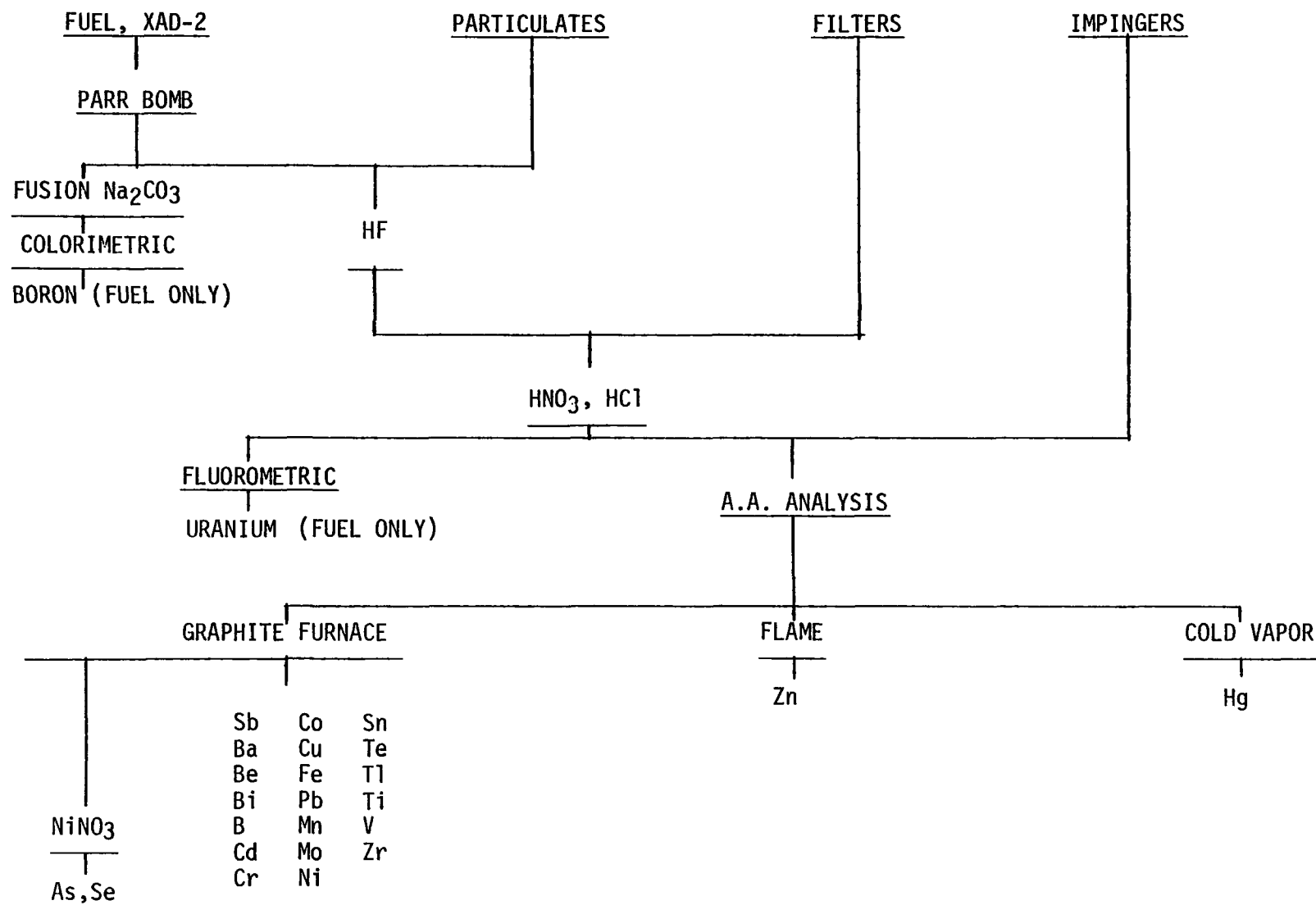
DATA

Bottom

Figure 3.

APPENDIX H
ANALYTICAL PROCEDURES

ELEMENTAL ANALYSIS FLOWSHEET



APPENDIX I
BIOASSAY RESULTS

LBI ASSAY NO. 3986

LBI SAFETY NO. 3643

MUTAGENICITY EVALUATION OF
OIL FIRED GAS TURBINE NO. 1 SASS TRAIN
XAD-2 EXTRACT (IN METHYLENE CHLORIDE)
IN THE
AMES SALMONELLA/MICROSOME
PLATE TEST
FINAL REPORT

SUBMITTED TO:

ACUREX CORPORATION
485 CLYDE AVENUE
MOUNTAIN VIEW, CA.
94042

SUBMITTED BY:

LITTON BIONETICS, INC.
5516 NICHOLSON LANE
KENSINGTON, MARYLAND 20795

LBI PROJECT NO. 20988

REPORT DATE: JUNE 1979



BIONETICS

PREFACE

This report contains a summary of the data compiled during the evaluation of the test compound. The report is organized to present the results in a concise and easily interpretable manner. The first part contains items I-IX. Items I-IV provide sponsor and compound identification information, type of assay, and the protocol reference number. All protocol references indicate a standard procedure described in the Litton Bionetics, Inc. "Screening Program for the Identification of Potential Mutagens and Carcinogens." Item V provides the initiation and completion dates for the study, and Item VI provides identification of supervisory personnel. Item VII identifies the tables and/or figures containing the data used by the study director in interpreting the test results. The interpretation itself is in Item VIII. Item IX provides the conclusion and evaluation.

The second part of the report, entitled PROTOCOL, describes the materials and procedures employed in conducting the assay. This part of the report also contains evaluation criteria used by the study director, and any appendices. The evaluation criteria are included to acquaint the sponsor with the methods used to develop and analyze the test results.

All test and control results presented in this report are supported by fully documented raw data which are permanently maintained in the files of the Department of Genetics and Cell Biology or in the archives of Litton Bionetics, Inc., 6516 Nicholson Lane, Kensington Maryland, 20795.



BIONETICS

- I. SPONSOR: Acurex Corporation
- II. MATERIAL (TEST COMPOUND): LBI ASSAY NUMBER 3986
- A. Identification: Oil Fired Gas Turbine No. 1 Sass Train
XAD-2 Extract (in Methylene Chloride)*
- B. Date Received: February 23, 1979
- C. Physical Description: Clear yellow liquid
- III. TYPE OF ASSAY: Ames Salmonella/microsome Mutagenesis Assay
- IV. PROTOCOL NUMBER: 401
- V. STUDY DATES:
- A. Initiation: May 18, 1979
- B. Completion: June 8, 1979
- VI. SUPERVISORY PERSONNEL:
- A. Study Director: D.R. Jagannath, Ph.D.
- B. Laboratory Supervisor: Sibyl Goode
- VII. RESULTS:
- The results of this assay are presented in Tables 1 and 2.
- VIII. INTERPRETATION OF RESULTS:
- The test compound was examined for mutagenic activity in a series of in vitro microbial assays employing Salmonella indicator organisms. The compound was tested directly and in the presence of liver microsomal enzyme preparations from Aroclor-induced rats.
- The compound was tested at four concentrations according to the IERL-RTP procedures Manual: Level I (1977). The compound was tested for its mutagenic activity as well as for its toxicity at 0.01 mg, 0.1 mg, 1.0 mg and 10 mg per plate.
- The toxicity results presented in Table 1 indicate that the test compound was not toxic at the doses employed in these studies both in the presence and absence of metabolic activation systems.

*See Sample Preparation and Handling.

VIII. INTERPRETATION OF RESULTS (continued):

The mutagenicity results presented in Table 2 indicate that the test compound did not induce any genetic activity in any of the test organisms employed in these assays. The results of tests conducted on the test compound in the presence of a rat liver activation system were also negative. The test with TA-1537 was repeated in nonactivation and activation assays because of high solvent values in the initial test. The repeat tests were also negative.

IX. CONCLUSIONS:

The test compound, Oil Fired Gas Turbine No. 1 Sass train XAD-2 extract (in methylene chloride) did not demonstrate genetic activity or toxicity in any of the assays conducted in this evaluation and was considered as not mutagenic under these test conditions.

Submitted by:

Study Director

D.R. Jagannathan

6.11.79

D.R. Jagannathan, Ph.D.
Section Chief
Submammalian Genetics
Department of Genetics
and Cell Biology

Reviewed by:

David J. Brusick (for)
David J. Brusick, Ph.D.
Director

6/11/79

Department of Genetics
and Cell Biology



BIONETICS

V. RESULTS

TABLE 1

A. NAME OR CODE DESIGNATION OF THE TEST COMPOUND: OIL FIRED GAS TURBINE NO1 SASS TRAIN XAD-2 EXTRACT (IN METHYLENE CHLORIDE)
 B. SOLVENT: CMSO
 C. TEST INITIATION DATES: 05/18/79 05/30/79
 D. TEST COMPLETION DATE: 06/08/79
 E. S-9 LCT#: CH024
 NOTE: CONCENTRATIONS ARE GIVEN IN MILLIGRAMS (MG) PER PLATE.

TEST	SPECIES	TISSUE	INDICATOR ORGANISMS (POPULATION/10 ⁶)							
			TA-1535		TA-1537		TA-98		TA-100	
			1	2	1	2	1	2	1	2
NONACTIVATION										
SOLVENT CONTROL	---	---	393		264	167	290		1853	
POSITIVE CONTROL **	---	---	315		159	91	290		1649	
TEST COMPOUND										
0.010000 MG	---	---	341		280	153	221		440	
0.100000 MG	---	---	362		244	194	280		391	
1.000000 MG	---	---	346		282	125	305		353	
10.000000 MG	---	---	326		267	177	354		353	
ACTIVATION										
SOLVENT CONTROL	RAT	LIVER	378		311	149	351		400	
POSITIVE CONTROL ***	RAT	LIVER	551		191	101	312		331	
TEST COMPOUND										
0.010000 MG	RAT	LIVER	437		285	153	388		389	
0.100000 MG	RAT	LIVER	380		299	158	341		348	
1.000000 MG	RAT	LIVER	383		297	186	402		350	
10.000000 MG	RAT	LIVER	196		270	168	378		346	

**

TA-1535 SODIUM AZIDE 1 UG/PLATE
 TA-1537 9-AMINOACRIDINE 50 UG/PLATE
 TA-98 2-NITROFLUORENE 10 UG/PLATE
 TA-100 SODIUM AZIDE 1 UG/PLATE
 SOLVENT 50 UG/PLATE

TA-1535 2-ANTHRAMINE 2.5 UG/PLATE
 TA-1537 2-ANTHRAMINE 2.5 UG/PLATE
 TA-98 2-ANTHRAMINE 2.5 UG/PLATE
 TA-100 2-ANTHRAMINE 2.5 UG/PLATE

V. RESULTS

TABLE 2

A. NAME OR CODE DESIGNATION OF THE TEST COMPOUND: DIL FIRED GAS TURBINE NO1 SASS TRAIN XAD-2 EXTRACT (IN METHYLENE CHLORIDE)
B. SOLVENT: DMSO
C. TEST INITIATION DATES: 05/18/79
D. TEST COMPLETION DATE: 06/08/79
E. S-9 LOT#: _____

NOTE: CONCENTRATIONS ARE GIVEN IN MILLIGRAMS (MG) PER PLATE

			INDICATOR ORGANISMS (REVERTANTS/PLATE)									
TEST		SPECIES	TISSUE	TA-1535		TA-1537		TA-98		TA-100		
				1	2	1	2	1	2	1	2	
NONACTIVATION												
SOLVENT CONTROL		---	---	20		45	11	55		124		
POSITIVE CONTROL **		---	---	1023		352	265	1304		1923		
TEST COMPOUND												
	0.010000 MG	---	---	20		41	6	46		140		
	0.100000 MG	---	---	10		34	8	45		106		
	1.000000 MG	---	---	21		25	7	42		110		
	10.000000 MG	---	---	11		25	5	68		108		
ACTIVATION												
SOLVENT CONTROL		RAT	LIVER	16		52	11	65		112		
POSITIVE CONTROL ***		RAT	LIVER	414		272	110	2926		2046		
TEST COMPOUND												
	0.010000 MG	RAT	LIVER	14		43	15	58		91		
	0.100000 MG	RAT	LIVER	7		53	13	75		100		
	1.000000 MG	RAT	LIVER	13		6	7	55		89		
	10.000000 MG	RAT	LIVER	16		42	8	63		109		

**	TA-1535	SODIUM AZIDE		1 UG/PLATE				***		TA-1535	2-ANTHRAMINE	2.5 UG/PLATE
	TA-1537	9-AMINOACRIDINE		50 UG/PLATE					TA-1537	2-ANTHRAMINE	2.5 UG/PLATE	
	TA-98	2-NITROFLUORENE		10 UG/PLATE					TA-98	2-ANTHRAMINE	2.5 UG/PLATE	
	TA-100	SODIUM AZIDE		1 UG/PLATE					TA-100	2-ANTHRAMINE	2.5 UG/PLATE	
	SOLVENT	50 UL/PLATE										

C INDICATES CONTAMINATION

SAMPLE PREPARATION AND HANDLING

The test material was received as a solution in 1.2 ml of methylene chloride and was stored at 4°C until solvent exchanged. The entire sample was exchanged into DMSO by first adding 2 ml DMSO and reducing the volume to 2 ml under a stream of nitrogen in a warm water bath (33°C). Then 0.5 ml DMSO was added and the solution evaporated again to 2 ml. This last process was repeated once more, leaving the sample in a final volume of 2.0 ml. This sample was stored at 4°C until use in the cytotoxicity assay. Since the original test sample represented 307 ft³ of exhaust gas, the solvent exchanged sample corresponded to 153.5 ft³ gas/ml or 4346.5 L gas/ml.

A solvent exchanged DMSO blank was also prepared by the above procedure, starting with 1.2 ml methylene chloride (same volume as the original test material). Since the test material did not exhibit any mutagenic or toxic effect on the indicator organism in these assays, solvent exchanged DMSO blank was not tested separately.



PROTOCOL NO. 401

AMES SALMONELLA/MICROSOME PLATE ASSAY

1. OBJECTIVE

The objective of this study is to evaluate a test material for mutagenic activity in a bacterial assay with and without a mammalian S9 activation system.

2. RATIONALE

The Salmonella typhimurium strains used at LBI are all histidine auxotrophs by virtue of mutations in the histidine operon. When these histidine-dependent cells are grown in a minimal media petri plate containing a trace of histidine, only those cells that revert to histidine independence (his⁺) are able to form colonies. The trace amount of histidine allows all the plated bacteria to undergo a few divisions; this growth is essential for mutagenesis to occur. The his⁺ revertants are easily scored as colonies against the slight background growth. The spontaneous mutation frequency of each strain is relatively constant; but when a mutagen is added to the agar, the mutation frequency is increased 2- to 100-fold. Cells which grow to form colonies on the minimal media petri plates are therefore assumed to have reverted, either spontaneously or by the action of a test substance to his⁺ genotype.

3. MATERIALS

A. Indicator Microorganisms

The Salmonella typhimurium strains used in this assay were obtained from Dr. Bruce Ames, University of California at Berkeley.¹⁻³ The following 5 strains are routinely used.

Strain Designation	Gene Affected	Additional Mutations			Mutation Type Detected
		Repair	LPS	R Factor	
TA-1535	<u>his</u> G	Δ <u>uvr</u> S	<u>rfa</u>	-	Base-pair substitution
TA-1537	<u>his</u> C	Δ <u>uvr</u> S	<u>rfa</u>	-	Frameshift
TA-98	<u>his</u> D	Δ <u>uvr</u> B	<u>rfa</u>	pKM101	Frameshift
TA-100	<u>his</u> G	Δ <u>uvr</u> B	<u>rfa</u>	pKM101	Base-pair substitution



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All the above strains have, mutation in the histidine operon, mutation (rfa⁻) that leads to defective lipopolysaccharide coat, a deletion that covers genes involved in the synthesis of vitamin biotin (bio⁻) and in the repair of ultraviolet (uv) - induced DNA damage (uvrB⁻). The rfa⁻ mutation makes the strains more permeable to many large molecules. The uvrB⁻ mutation decreases repair of some types of chemically or physically damaged DNA and thereby enhances the strain's sensitivity to some mutagenic agents. The resistant transfer factor plasmid (R factor) pKM101, in TA-98 and TA-100 is believed to cause an increase in error-prone DNA repair that leads to many more mutations for a given dose of most mutagens⁵. In addition, plasmid pKM101 confers resistance to the antibiotic ampicillin, which is a convenient marker to detect the presence of plasmid in the cells.

All indicator strains are kept at 4°C on minimal medium plates supplemented with a trace of biotin and an excess of histidine. The plates with plasmid-carrying strains contain in addition ampicillin (25 µg/ml), to ensure stable maintenance of plasmid pKM101. New stock culture plates are made every two months from the frozen master cultures or from single colony reisolates that were checked for their genotypic characteristics (his, rfa, uvrB, bio) and for the presence of plasmid. For each experiment, an inoculum from the stock culture plates is grown overnight at 37°C in nutrient broth (Oxoid CM67) and used.

5. Media

The bacterial strains were cultured in Oxoid Media #2 (nutrient Broth). The selective medium was Vogel Bonner Medium 5 with 2% glucose. The overlay agar will consist of 0.6% purified agar with 0.5 mM histidine, 0.05 mM biotin and 0.1M NaCl according to the methods of Ames et al.⁶

C. Activation System

(1) S9 Homogenate

A 9,000 x g supernatant prepared from Sprague-Dawley adult male rat liver induced by Aroclor 1254 (described by Ames et al.⁶) was purchased from Bionetics Laboratory Products, Litton Bionetics, Inc. and used in this assay.

(2) S9 Mix

Components	Concentration per Milliliter S9 Mix
NACP (sodium salt)	4 µmoles
D-glucose-6-phosphate	8 µmoles
MgCl ₂	8 µmoles
KCl	33 µmoles
Sodium phosphate buffer pH 7.4	100 µmoles
Organ homogenate from rat liver (S9 fraction)	100 µliters



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4. EXPERIMENTAL DESIGN

A. Dosage Selection

The tests are run at four concentrations according to the EPA Level I Manual. The recommended doses are 0.01, 0.1, 1.0 and 10 mg per plate. Both mutagenicity testing and toxicity testing are performed using these four doses.

B. Mutagenicity Testing

The procedure used is based on the paper published by Ames et al.⁶ and is performed as follows:

(1) Nonactivation Assay

To a Sterile 13 x 100 mm test tube placed in a 43°C water bath the following is added in order:

- (a) 2.00 ml of 0.6% agar containing 0.05 mM histidine and 0.05 mM biotin.
- (b) 0.05 ml of a solution of the test chemical to give approximate dose.
- (c) 0.1 ml - 0.2 ml of indicator organism/s.
- (d) 0.50 ml of 0.01M phosphate buffer, pH 7.4.

This mixture is swirled gently and then poured into minimal agar plates (see 3B, Media). After the top agar has set, the plates are incubated at 37°C for approximately 2 days. The number of his⁺ revertant colonies growing in the plates is counted and recorded.

(2) Activation Assay

The activation assay is run concurrently with the nonactivation assay. The only difference is the addition of 0.5 ml of S9 mix (see 3C:2, Activation System) to the tubes in place of 0.5 ml of phosphate buffer which is added in nonactivation assays. All other details are similar to the procedure for nonactivation assays.

A detailed flow diagram for the plate incorporation assay is provided in Figure 1.

C. Control Compounds

A negative control consisting of the solvent used for the test material is performed in all cases. For negative controls, step 'b' of Nonactivation Assays is replaced by 0.05 ml of the solvent. The negative controls are employed for each indicator strain and is performed in the absence and presence of S9 mix. The solvent used to prepare the stock solution of the test material is given in the Results section of this report. All dilutions of the test material made using this solvent.

Specific positive control compounds known to revert each strain are also used in the assays. The concentrations and specificities of these compounds to specific strains are given in the following table.

Assay	Chemical	Solvent	Concentration per Plate (μ g)	<u>Salmonella</u> Strains
Nonactivation	Sodium azide	Water	1	TA-1538, TA-100
	2-Nitrofluorene (NF)	Dimethyl- sulfoxide	10	
	9-aminoacridine (9AA)	Ethanol	50	TA-1537
Activation	2-anthramine (ANTH)	Dimethyl- sulfoxide	2.5	For all strains

D. Toxicity Test

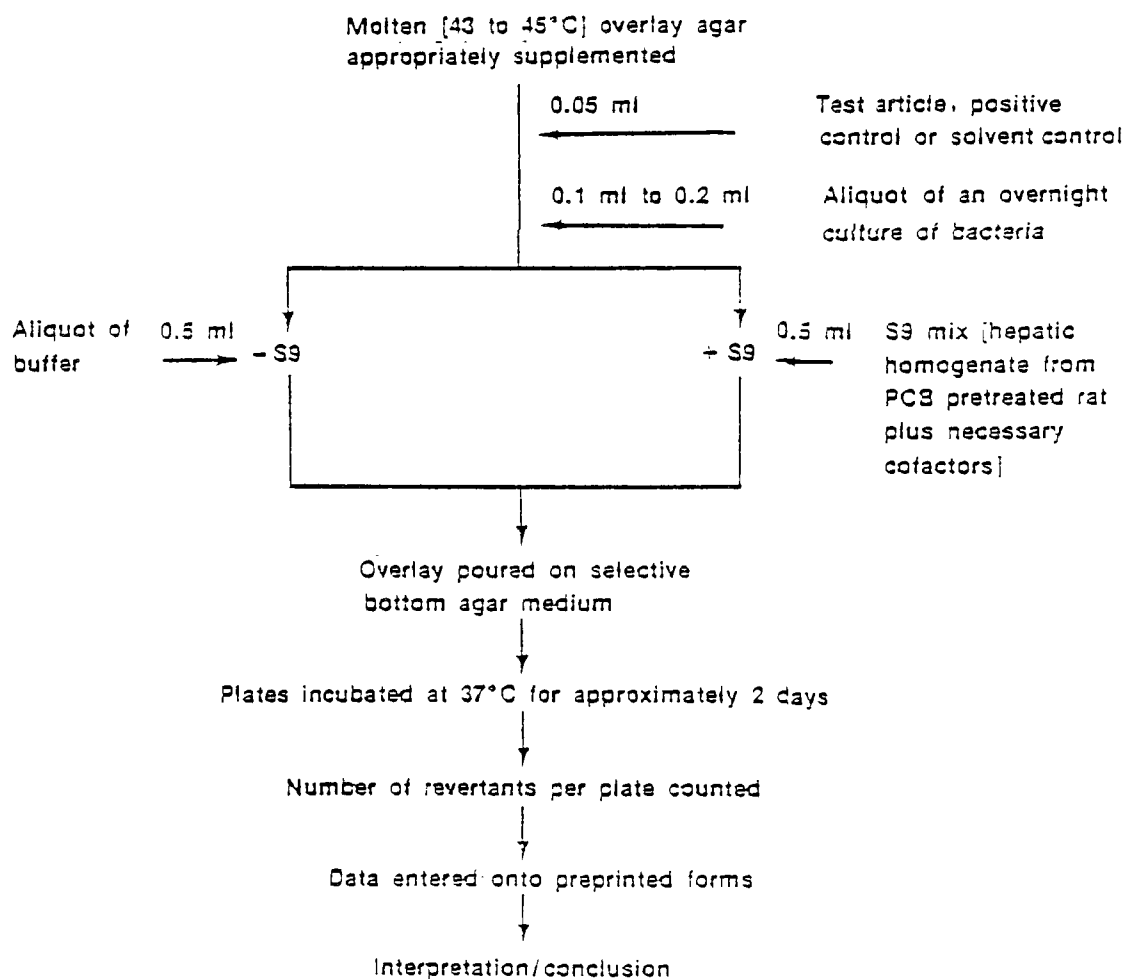
To a sterile 13 x 100 mm test tube placed in a 43°C water bath the following is added in order:

- (a) 2.0 ml of 0.6% agar containing 0.05 mM histidine and 0.05 mM biotin.
- (b) 0.05 ml of a solution of the test chemical to give approximate dose.
- (c) 0.1-0.2 ml of indicator cells (approximately 200 cells from an overnight culture appropriately dilute)
- (d) 0.50 ml of 0.01M phosphate buffer, pH 7.4 (for nonactivation assays) or S9 mix (see 3c:2) (for activation assays)

This mixture is swirled gently and then poured over the surface of nutrient agar plates. After the top agar has set, the plates are incubated at 37°C for 2 days. The number of colonies growing on the plates is counted and recorded.

FIGURE 1

REVERSE MUTATION ASSAY
[Agar Incorporation Method]



5. EVALUATION CRITERIA

Statistical methods are not currently used, and evaluation is based on the criteria included in this protocol.

Plate test data consists of direct revertant colony counts obtained from a set of selective agar plates seeded with populations of mutant cells suspended in a semisolid overlay. Because the test material and the cells are incubated in the overlay for approximately 2 days and a few cell divisions occur during the incubation period, the test is semiquantitative in nature. Although these features of the assay reduce the quantitation of results, they provide certain advantages not contained in a quantitative suspension test:

- . The small number of cell divisions permits potential mutagens to act on replicating DNA, which is often more sensitive than nonreplicating DNA.
- . The combined incubation of the test article and the cells in the overlay permits constant exposure of the indicator cells for approximately 2 days.

A. Surviving Populations

Plate test procedures do not permit exact quantitation of the number of cells surviving chemical treatment. At low concentrations of the test material, the surviving population on the treatment plates is essentially the same as that on the negative control plate. At high concentrations, the surviving population is usually reduced by some fraction. Our protocol will normally employ several doses ranging over two or three log concentrations, the highest of these doses being selected to show slight toxicity as determined by subjective criteria.

B. Dose-Response Phenomena

The demonstration of dose-related increases in mutant counts is an important criterion in establishing mutagenicity. A factor that might modify dose-response results for a mutagen would be the selection of doses that are too low (usually mutagenicity and toxicity are related). If the highest dose is far lower than a toxic concentration, no increases may be observed over the dose range selected. Conversely, if the lowest dose employed is highly cytotoxic, the test material may kill any mutants that are induced, and the test material will not appear to be mutagenic.



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C. Control Tests

Positive and negative control assays will be conducted with each experiment and will consist of direct-acting mutagens for nonactivation assays and mutagens that require metabolic biotransformation in activation assays. Negative controls will consist of the test material solvent in the overlay agar together with the other essential components. The negative control plate for each strain will give a reference point to which the test data will be compared. The positive control assay will be conducted to demonstrate that the test systems are functional with known mutagens.

D. Evaluation Criteria for Ames Assay

Because the procedures to be used to evaluate the mutagenicity of the test material are semiquantitative, the criteria to be used to determine positive effects are inherently subjective and are based primarily on a historical data base. Most data sets will be evaluated using the following criteria.

(1) Strains TA-1535, TA-1537

If the solvent control value is within the normal range, a test material that produces a positive dose response over three concentrations with the highest increase equal to three times the solvent control value will be considered to be mutagenic.

(2) Strains TA-98 and TA-100

If the solvent control value is within the normal range, a test material that produces a positive dose response over three concentrations with the highest increase equal to twice the solvent control value for TA-98 and TA-100 will be considered to be mutagenic.

(3) Pattern

Because TA-1535 and TA-100 are both derived from the same parental strain (G-46) and because TA-1538 and TA-98 are both derived from the same parental strain (J3052), to some extent there is a built-in redundancy in the microbial assay. In general, the two strains of a set respond to the same mutagen and such a pattern is sought. Generally, if a strain responds to a mutagen in nonactivation tests, it will do so in activation tests.



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(4) Reproducibility

If a test material produces a response in a single test that cannot be reproduced in additional runs, the initial positive test data lose significance.

The preceding criteria are not absolute, and other extenuating factors may enter into a final evaluation decision. However, these criteria will be applied to the majority of situations and are presented to aid those individuals not familiar with this procedure. As the data base is increased, the criteria for evaluation can be more firmly established.

E. Relation between Mutagenicity and Carcinogenicity

It must be emphasized that the Ames Salmonella/Microsome Plate Assay is not a definitive test for chemical carcinogens. It is recognized, however, that correlative and functional relations have been demonstrated between these two endpoints. The results of comparative tests on 300 chemicals by McCann et al.² show an extremely good correlation between results of microbial mutagenesis tests and in vivo rodent carcinogenesis assays.

All evaluations and interpretation of the data to be presented in the final report will be based only on the demonstration, or lack, of mutagenic activity.



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LBI ASSAY NO. 3986

LBI SAFETY NO. 3643

CYTOTOXIC EVALUATION OF
OIL FIRED GAS TURBINE NO. 1
SASS TRAIN XAD-2 EXTRACT

IN THE
WI-38 HUMAN CELL
CYTOTOXICITY ASSAY

FINAL REPORT

SUBMITTED TO:

ACUREX CORPORATION
485 CLYDE AVE.
MOUNTAIN VIEW, CA 94042

SUBMITTED BY:

LITTON BIONETICS, INC.
5516 NICHOLSON LANE
KENSINGTON, MARYLAND 20795

LBI PROJECT NO. 20993

REPORT DATE: JUNE, 1979



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PREFACE

This report contains a summary of the data compiled during the evaluation of the test compound. The report is organized to present the results in a concise and easily interpretable manner. The first part contains items I-IX. Items I-IV provide sponsor and compound identification information, type of assay, and the protocol reference number. All protocol references indicate a standard procedure described in the Litton Bionetics, Inc. "Screening Program for the Identification of Potential Mutagens and Carcinogens." Item V provides the initiation and completion dates for the study, and Item VI provides identification of supervisory personnel. Item VII identifies the tables and figures containing the data used by the study director in interpreting the test results. The interpretation itself is in Item VIII. Item IX provides the conclusion and evaluation.

The second part of the report, entitled PROTOCOL, describes the materials and procedures employed in conducting the assay. This part of the report also contains evaluation criteria used by the study director, and any appendices. The evaluation criteria are included to acquaint the sponsor with the methods used to develop and analyze the test results.

All test and control results presented in this report are supported by fully documented raw data which are permanently maintained in the files of the Department of Genetics and Cell Biology or in the archives of Litton Bionetics, Inc., 5516 Nicholson Lane, Kensington Maryland, 20795.

Copies of raw data will be supplied to the sponsor upon request.

- I. SPONSOR: ACUREX CORPORATION
- II. MATERIAL (TEST COMPOUND): LBI ASSAY NUMBER 3986
- A. Identification: Oil Fired Gas Turbine No. 1, SASS Train XAD-2 Extract
- B. Date Received: February 23, 1979
- C. Physical Description: Light yellow solution in DMSO
- III. TYPE OF ASSAY: WI-38 Human Cell Cytotoxicity Assay
- IV. PROTOCOL NUMBER: Special Protocol
- V. STUDY DATES:
- A. Initiation: May 29, 1979
- B. Completion: June 5, 1979
- VI. SUPERVISORY PERSONNEL:
- A. Study Director: Brian C. Myhr, Ph.D.
- B. Laboratory Supervisor: Robert Young
- VII. RESULTS:
- The data are presented in Table 1 on page 3 and in Figures 1 and 2 on pages 4 and 5.
- VIII. INTERPRETATION OF RESULTS:
- The methylene chloride extract of the test sample on XAD-2 resin, after solvent exchange into DMSO, appeared to remain soluble in the culture medium at the highest assayed concentration of 20 μ l/ml. Higher concentrations could not be tested because of the introduction of greater than 2% organic solvent by volume. As shown in Table 1, 1% DMSO reduced the viability index, total protein, and total ATP to about 70-80% of the untreated negative control; 2% DMSO reduced these parameters even further to about 40-65%. The corresponding concentrations of solvent exchanged DMSO were somewhat less toxic to these assay parameters, showing that residual methylene chloride does not contribute to the solvent toxicity. Because of the solvent toxicity, the effect of the test material was measured relative to the assay parameters obtained for the appropriate solvent exchanged DMSO negative control.

VIII. INTERPRETATION OF RESULTS (continued):

The most responsive assay parameter appeared to be the viability index, although the protein and ATP contents started to decrease similarly at the highest dose of 20 μ l/ml. The percent viability and ATP per 10^6 cells parameters gave no indication of any toxicity. A 50% reduction was not achieved for any assay parameter, but the curve for the viability index (Figure 1) indicated that an EC50 would occur near 35 μ l/ml. In terms of the volume of exhaust gas represented by the DMSO test solution (4346.5 L gas/ml), this EC50 corresponds to 152.1 L gas/ml. Therefore, on the basis of the viability index and expectations for the ATP and protein parameters, the test material appears to yield EC50 values in the low toxicity region (100 L/ml to 1000 L/ml).

IX. CONCLUSIONS:

The test material, SASS train XAD-2 Extract, Oil Fired Gas Turbine No. 1, is evaluated as having low toxicity to WI-38 human cells. The viability index indicated an EC50 value would be obtained near 152 L gas/ml, and the ATP and protein contents were decreasing in the same toxicity range.

Submitted by:

Study Director

Brian Myhr 6/11/79
Brian Myhr, Ph.D. date
Section Chief
Mammalian Genetics
Department of Genetics
and Cell Biology

Reviewed by:

Robert J. Brusick (for) 6/11/79
David J. Brusick, Ph.D. date
Director
Department of Genetics
and Cell Biology



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TABLE 1

WI-38 HUMAN CELL CYTOTOXICITY ASSAY

Test Date: MAY 29, 1979

Initial Cell Viability: 97.2%

LBI Assay No.: 3986

Viable WI-38 Cells Seeded/Flask: 2.0×10^5 Test Material Identity: Oil Fired Gas Turbine No. 1,
SASS Train XAD-2 Extract

Passage number: 28

Vehicle: DMSO/growth medium

TEST RESULTS

Sample	Concentration ^{††} μl/ml	Average Values per Culture Flask			ATP Per		Expressed as Percent of Negative Control (NC) ^{***}					
		Viable Cells 10 ⁶ Units	Total Cells 10 ⁶ Units	Cellular Protein μg	ATP 10 ⁸ fg	ATP Per 10 ⁶ Cells 10 ⁸ fg	Viability %	Viability	Viability Index	Protein	ATP	ATP Per 10 ⁶ Cells
NEGATIVE CONTROL ---		.602	.610	145	50.4	82.6	98.7%	100.0	100.0	100.0	100.0	100.0
1% DMSO BLANK S.E.**		.448	.449	101	43.3	96.4	99.8	101.1	74.4	69.7	85.9	116.7
1% DMSO†		.446	.446	98	38.9	87.2	100.0	101.3	74.1	67.6	77.2	105.6
2% DMSO BLANK S.E.**		.463	.481	80	33.1	68.8	96.3	97.6	76.9	55.2	65.7	83.3
2% DMSO		.397	.406	55	27.3	67.2	97.8	99.1	65.9	37.9	54.2	81.4
TEST	0.5	.467	.473	100	43.3	91.5	98.7	98.9	104.2	99.0	100.0	94.9
TEST	2.0	.453	.454	102	39.0	85.9	99.8	100.0	101.1	101.0	90.1	89.1
TEST	5.0	.420	.425	91	39.4	92.7	98.8	99.0	93.8	90.1	91.0	96.2
TEST	10.0	.411	.420	79	38.9	92.6	97.9	98.1	91.7	78.2	89.8	96.1
TEST	20.0	.325	.333	61	27.5	82.6	97.6	101.3	70.2	76.3	83.1	120.1
^{†††} change in culture medium: None observed					^{††} EC50 VALUES:							
^{**} 1.2 ml methylene chloride solvent exchanged to 2.0 ml DMSO					μl/ml							
^{***} or appropriate DMSO blank					>20.0 (35) >20.0 >20.0 >20.0							

*Determined from data plots in Figures 1 and 2

Toxicity

†Average of 2 flasks

Classification: Low toxicity

FIGURE 1

EC50 DETERMINATION FOR

PERCENT VIABILITY (○) AND VIABILITY INDEX (●)

OIL FIRED GAS TURBINE NO. 1

SASS TRAIN XAD-2 EXTRACT

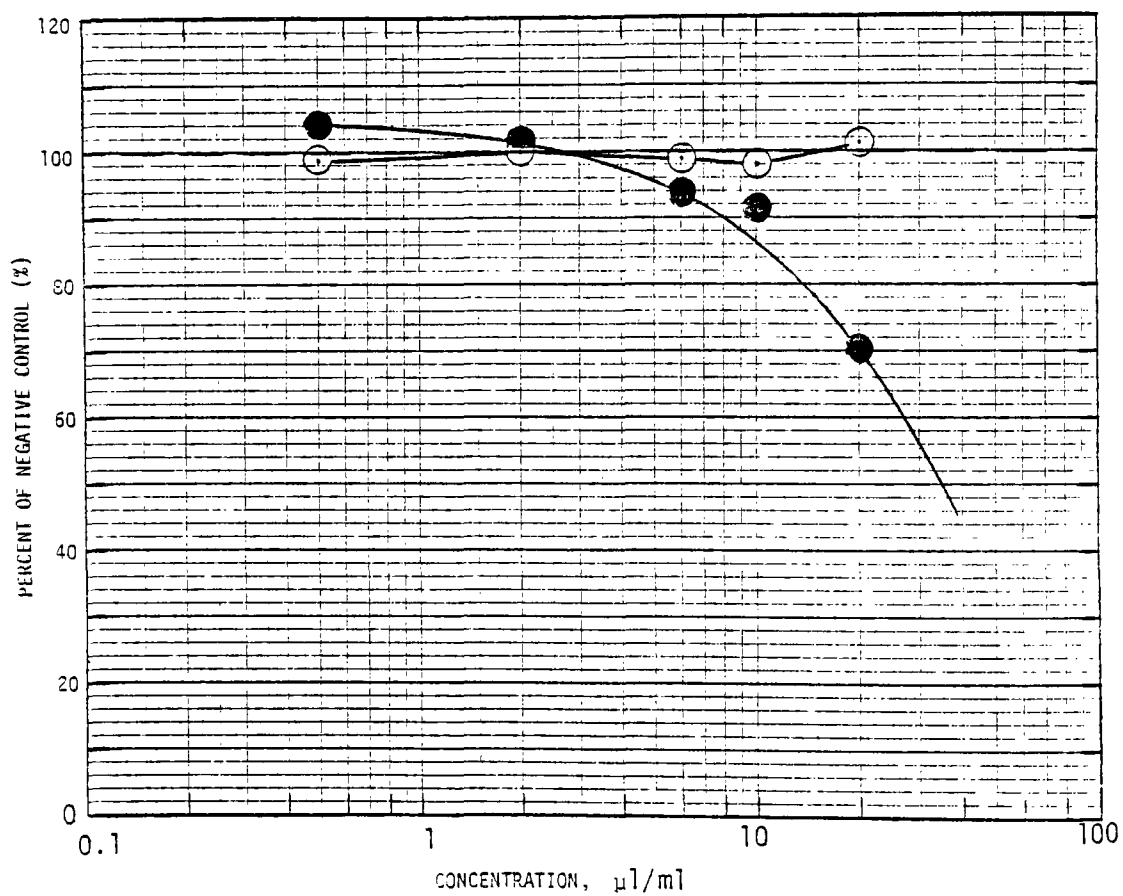
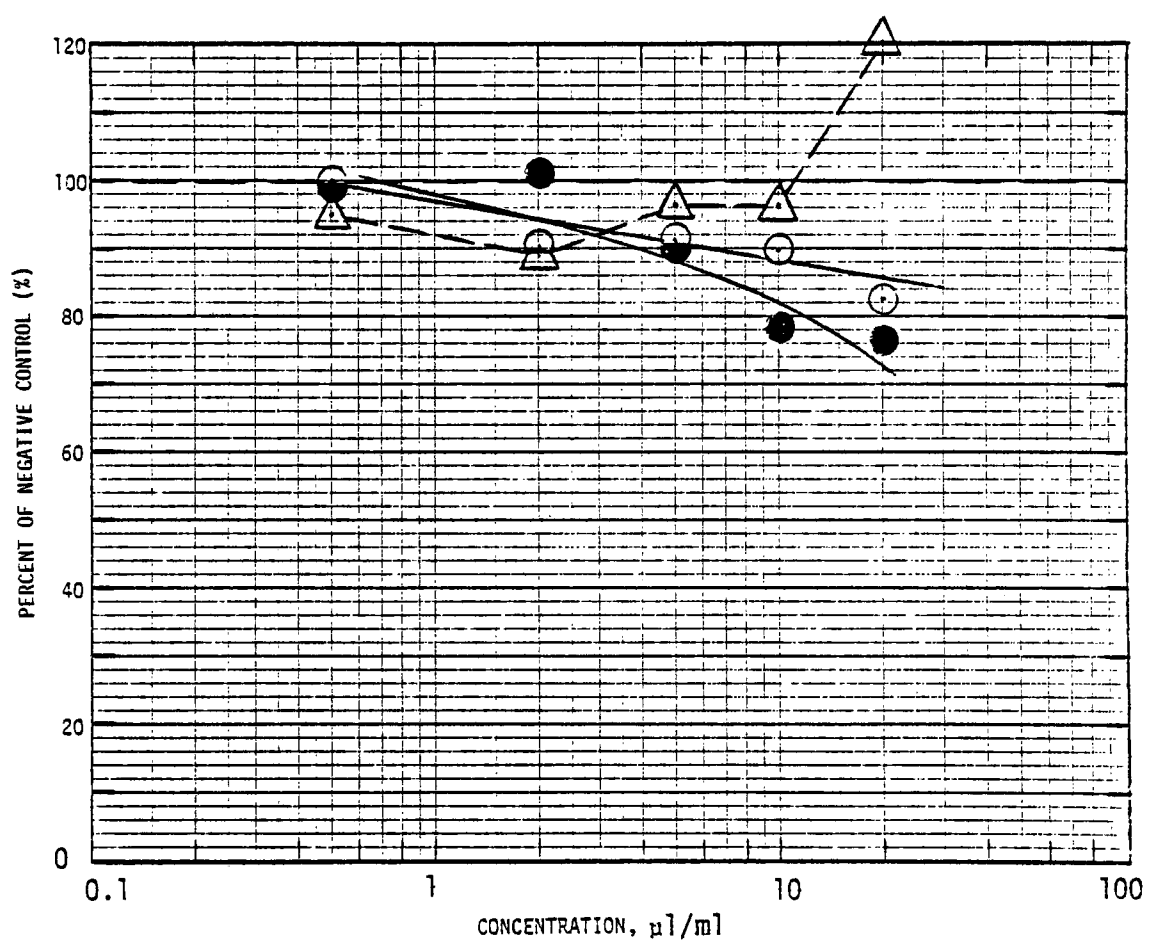


FIGURE 2

EC50 DETERMINATION FOR
PROTEIN (●), ATP (○), and ATP/10⁶ CELLS (Δ)

OIL FIRED GAS TURBINE NO. 1

SASS TRAIN XAD-2 EXTRACT



ASSAY PROTOCOL

1. OBJECTIVE

The objective of this assay is to determine the concentrations of test material that reduces by 50% the number of viable cells, the cellular protein, and the ATP content after a 20 hour exposure. These concentrations are referred to as the EC50 values for each measured parameter.

2. MATERIALS

A. Indicator Cells

The indicator cells used for this study were WI-38 human embryonic lung fibroblasts obtained from Flow Laboratories, Inc., Rockville, Maryland. The cells were supplied as confluent monolayers at passage numbers 23 or 24 in Eagle's Minimum Essential Medium. This test system is specified by the Environmental Protection Agency's Level 1 Environmental Assessment Program.¹

B. Medium and Cell Maintenance

The cells were maintained and treated in Basal Medium Eagle (BME) supplemented with 10% fetal bovine serum, 2mM L-glutamine, 100 units/ml penicillin, 100 µg/ml streptomycin, and 1.0 µg/ml amphotericin B (Fungizone). Subcultures were prepared twice weekly at a 1:2 split ratio using 0.25% trypsin. Cultures were discarded after the 35th subculture (passage).

C. Negative Controls

Five sets of negative control cultures, each in triplicate, were carried through the same experimental time period as the treated cells. One set was an untreated negative control consisting of cultures exposed only to BME culture medium. Two sets were solvent controls containing 1% and 2% of the solvent-exchanged DMSO blank, prepared as described below. In addition, two solvent control sets containing 1% and 2% pure DMSO were assayed in order to determine whether residual methylene chloride in the solvent-exchanged blank was contributing to solvent toxicity. The average viability, ATP content, and protein content of the solvent-exchanged negative controls provided the reference points for determining the effects of different concentrations of the test material on the assay parameters. The 2% solvent-exchanged control was the reference for the highest assayed concentration (20 µl/ml) and the 1% solvent-exchanged control was the reference for the remaining test concentrations.



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2. MATERIALS (Continued)

D. Sample Preparation and Handling

The test material was received as a solution in 1.2 ml of methylene chloride and was stored at 4°C until solvent exchanged. The entire sample was exchanged into DMSO by first adding 2 ml DMSO and reducing the volume to 2 ml under a stream of nitrogen in a warm water bath (33°C). Then 0.5 ml DMSO was added and the solution evaporated again to 2 ml. This last process was repeated once more, leaving the sample in a final volume of 2.0 ml. This sample was stored at 4°C until use in the cytotoxicity assay. Since the original test sample represented 307 ft³ of exhaust gas, the solvent exchanged sample corresponded to 153.5 ft³ gas/ml or 4346.5 L gas/ml.

A solvent exchanged DMSO blank was also prepared by the above procedure, starting with 1.2 ml methylene chloride (same volume as the original test material).

3. EXPERIMENTAL DESIGN

A. Dose Selection

The solvent exchanged sample was tested from 20 µl/ml to 0.5 µl/ml in five dose steps. The 20 µl/ml treatment was the maximum dose because of the introduction of 2% DMSO in the growth medium. All other concentrations were obtained by 1:100 dilutions of the test sample and dilutions thereof (using DMSO) into the growth medium to give a 1% final concentration of solvent.

B. Culture Preparation

Stock cultures were subcultured into 100-mm plastic culture dishes 24 to 72 hours prior to use. This procedure provides a population of actively growing, sub-confluent cells to initiate the assay.

The cells were then suspended in BME culture medium by treatment with 0.25% trypsin for 3-5 minutes and the cell number determined by hemocytometer. A series of 25-cm² culture flasks were seeded with 20×10^4 cells and 4 ml culture medium per flask. The cultures were incubated overnight at 37°C in a humidified atmosphere containing 5% CO₂ to allow attachment of the cells and resumption of growth.



3. EXPERIMENTAL DESIGN (Continued)

C. Treatment

The medium was aspirated from the cultures and 4 ml of BME culture medium containing the test sample was applied. Three cultures were exposed to each test concentration and solvent exchanged DMSO blank. The flasks were then placed on a rocker platform in a 37°C incubator with a humidified atmosphere containing 5% CO₂. The flasks were slowly rocked for a 20 hour exposure period. Any color changes in the culture medium caused by the test material were noted and the pH determined in additional treated flasks.

D. Cell Viability Assay

At the end of the treatment period, the medium containing unattached cells was decanted into a centrifuge tube on ice. The cell monolayer was washed with 1 ml 0.05% trypsin/versene and this wash combined with the decanted media. The attached cells were then removed with 2 ml of 0.25% trypsin at 37°C and the suspended cells combined with the decantate. The cells from each flask were thereby resuspended in 7 ml volumes for subsequent analysis.

A 1.0 ml aliquot was removed for cell count and viability determination. The aliquot was combined with 0.2 ml or 0.5 ml of 0.4% trypan blue and counted by hemocytometer about 5 to 15 minutes later. Between 60 and 154 cells were counted per flask and the number of live (colorless) and dead (blue) cells were recorded.

E. ATP Assay

ATP was immediately analyzed by extraction of a 0.1 ml cell suspension sample with 0.9 ml of 90% DMSO. After 2 minutes at room temperature, 5.0 ml cold MOPS buffer (0.01 M morpholinopropane sulfonic acid) at pH 7.4 was added and the extract was vortexed and placed on ice. Aliquots of 10 µl were injected into a cuvette containing a luciferin-luciferase reaction mixture in a DuPont Model 760 Luminescence Biometer. The Biometer was calibrated with standard ATP solutions to provide a direct read-out of the ATP content. Each test sample was assayed three times to demonstrate consistent readings.

F. Lowry Protein Assay²

A 3.0 ml aliquot of the cell suspension was taken for protein analysis by the Lowry method. The aliquot was centrifuged at 365 x g for 10 minutes, the medium decanted, and the cell pellet resuspended in 3 ml PBS. After two additional centrifugation washes with PBS, the pellet was resuspended in 1.5 ml of PBS and frozen at -20°C or analyzed immediately. A 1.0 ml aliquot was used for the Lowry assay.



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3. EXPERIMENTAL DESIGN (Continued)

At the end of the color development period, the tubes were centrifuged to remove any particulate test material prior to making absorbance readings at 750 nm. Lowry protein standard curves were constructed with bovine serum albumin for each assay.

4. REFERENCES

¹IERL-RTP Procedures Manual: Level I. EPA-600/7-77-043, April 1977.

²Lowry, O.H., Rosebrough, N.J., Farr, A.L., and Randall, R.J.: Protein Measurement with Folin Phenol Reagent. J.Bio.Chem., 193:265-275, 1951.



BIONETICS

ASSAY ACCEPTANCE CRITERIA

The assay will be considered acceptable for evaluation of the test results if the following criteria are met:

1. The passage level of the cells (number of subcultures) prior to use in the assay does not exceed 35.
2. The percent viability of the WI-38 cells used to initiate the assay is 95% or greater.
3. At least 1.5×10^5 cells are seeded per flask. The untreated negative control cultures must increase in cell number by at least 2-fold over the 20 hour treatment period.
4. A sufficient number of data points (for five test concentrations or less) are available to clearly locate the EC50 of the most sensitive test parameter within a toxicity region as defined under Evaluation Criteria.
5. The data points critical to the location of the EC50 for the most sensitive parameter are the averages of at least two treated cultures.
6. If all the test parameters yield EC50 values greater than 1000 $\mu\text{g/ml}$ or 600 $\mu\text{l/ml}$, the plotted curves for any parameter will not exceed 120% of the negative control.

ASSAY EVALUATION CRITERIA

The EC50 value represents the concentration of test material that reduces an assay parameter to 50% of the negative control value. EC50 values are determined graphically by fitting a curve by eye through the data points associated with each test parameter plotted as a function of the logarithm of the applied concentration. Each point normally represents the average of three culture flasks for each treatment. Statistical analysis is unnecessary in most cases for evaluation.

The evaluation of the test material is based upon determinations of the EC50 values for five parameters: percent viability (ratio of viable cells to total cells x 100% for each treatment), viability index (ratio of viable cells for each treatment to viable cells in the negative control x 100%), cellular protein, total ATP content, and ATP per 10^6 cells. Except for the ATP content, these parameters are specified in the EPA Procedures Manual.¹ The ATP content will generally be a more sensitive parameter than ATP per 10^6 cells because any cell loss due to treatment will increase the latter parameter. ATP released into the growth medium by disrupted cells contributes to the ATP measurement.

The toxicity of the test material is evaluated as high, moderate, low or nondetectable according to the ranges of EC50 values defined in the following table. The actual concentration of extract at the EC50 is converted to the equivalent volume of exhaust gas per milliliter of culture medium prior to the evaluation. The assay parameter yielding the lowest EC50 will classify the test material.

Toxicity*	EC50 Values
High	EC50 < 10 L gas/ml
Moderate	EC50 range of 10-100 L gas/ml
Low	EC50 range of 100-1000 L gas/ml
Nondetectable	EC50 > 1000 L gas/ml

*Formulated by Litton Bionetics, Inc., under contract to the Environmental Protection Agency, Contract No. 68-02-2681.



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TECHNICAL REPORT DATA

(Please read Instructions on the reverse before completing)

1. REPORT NO. EPA-600/7-81-122b		2.		3. RECIPIENT'S ACCESSION NO.	
4. TITLE AND SUBTITLE Combustion Modification Controls for Stationary Gas Turbine: Volume II. Utility Unit Field Test				5. REPORT DATE July 1981	
				6. PERFORMING ORGANIZATION CODE	
7. AUTHOR(S) R. Larkin and E. B. Higginbotham				8. PERFORMING ORGANIZATION REPORT NO.	
9. PERFORMING ORGANIZATION NAME AND ADDRESS Acurex/Energy and Environmental Division 485 Clyde Avenue Mountain View, California				10. PROGRAM ELEMENT NO. EHE 624A	
				11. CONTRACT/GRANT NO. 68-02-2160 and 68-02-3176, Task 12	
12. SPONSORING AGENCY NAME AND ADDRESS EPA, Office of Research and Development Industrial Environmental Research Laboratory Research Triangle Park, NC 27711				13. TYPE OF REPORT AND PERIOD COVERED Task Final; 7/78-7/79	
				14. SPONSORING AGENCY CODE EPA/600/13	
15. SUPPLEMENTARY NOTES IERL-RTP project officer is Joshua S. Bowen, Mail Drop 65, 919/541-2470.					
16. ABSTRACT The report gives methods and results of an environmental assessment test program at Houston Lighting and Power's T.H. Wharton Generating Station, Unit 52. The aim of the program was to measure emissions changes resulting from applying NOx controls. Emissions of trace elements, organic materials, sulfur species, SO2, NOx, CO, and particulate matter were measured. These emissions, under normal and controlled (for NOx) operating conditions, were compared. Source operating data were also analyzed so that changes in operating parameters and efficiency could be assessed. Unit 52 is a General Electric MS 7001C simple-cycle, single-shaft, heavy duty gas turbine, rated at 70.8 MW nominal electrical output. This gas turbine may use either natural gas or distillate oil fuels. The test program was conducted using oil fuel. Water injection was used for NOx control. A water-to-fuel ratio of 0.42 reduced NOx by 58% from baseline levels. The unit heat rate showed about 2% change in going from baseline to controlled (for NOx) operation. Test results indicate that using water injection for NOx control in this unit reduced NOx and showed little effect on other emissions. Water injection reduced operating efficiency.					
17. KEY WORDS AND DOCUMENT ANALYSIS					
a. DESCRIPTORS		b. IDENTIFIERS/OPEN ENDED TERMS		c. COSATI Field/Group	
Pollution Gas Turbines Combustion Control Utilities Nitrogen Oxides		Pollution Control Stationary Sources Combustion Modification		13B 13G 21B 07B	
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