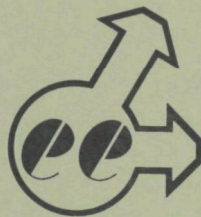


TEST NO. 72 - CI - 1
W. R. GRACE AND COMPANY
WET PROCESS PHOSPHORIC ACID
BARTOW, FLORIDA

JANUARY 4 - 5, 1972



environmental engineering, inc.

2324 S. W. 34th STREET / GAINESVILLE, FLORIDA 32601 / PHONE 904 / 372-3318

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Test Conducted By:
Environmental Engineering, Inc.
Contract # CPA - 70 - 82

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INTRODUCTION

Under the direction of the Environmental Protection Agency, Environmental Engineering, Inc., conducted emission tests at the W. R. Grace and Company wet process phosphoric acid plant located in Bartow, Florida. Three test runs were conducted on January 4 - 5, 1972. The purpose of the test was to obtain data to be used by both the Industrial Studies Branch and the Performance Standards Branch of the EPA.

Measurements for soluble and insoluble fluorides were made in the inlet ducts and outlet stack of the crossflow scrubber. The inlet leading from the attack tank and the inlet leading from the filtration system were sampled. Numerous grab samples of the process reactants and products were taken and analyzed for fluoride and P_2O_5 content. A schematic diagram of the process flow and all sampling locations is given in Figure 1.

Pertinent results of the test are listed in Tables 1 - 3; complete test results are given in Appendix A.

SUMMARY OF RESULTS

Although no major problems were encountered with process operation, some of the filter pans were not dumping dry. This condition had existed for some time before the tests were begun. Other operating conditions were typical of normal operation. There were no apparent irregularities in the actual sampling or analytical procedures.

For runs one, two, and three, respectively, the scrubber inlet from the

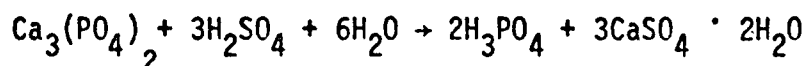
attack tank showed fluoride emissions of 0.033, 0.182, and 0.187 lb/ton P_2O_5 fed. Emissions at the inlet from the filtration system were 0.031, 0.025, and 0.024 lb/ton P_2O_5 fed, while at the scrubber outlet to the atmosphere, total fluoride emissions were 0.004, 0.008, and 0.004 lb/ton P_2O_5 fed. From this data, scrubber efficiency was calculated to be 93.3%, 96.0%, and 97.7% for runs one, two, and three, respectively.

Tables 1 - 3 give a complete summary of stack gas conditions, sample volumes collected, and fluoride emissions.

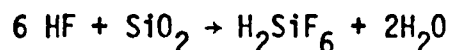
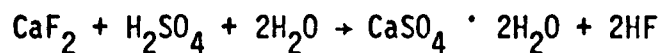
PROCESS DESCRIPTION

The first step in the basic process involves mixing phosphate rock, sulfuric acid, and water to form a reaction slurry.

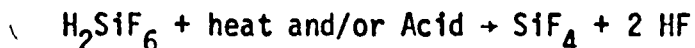
The basic reaction is the acidulation of tricalcium phosphate in the rock with sulfuric acid and water to produce phosphoric acid and calcium sulfate dihydrate (gypsum). The reaction is:



Hydrogen fluoride gas (HF) is produced by a side reaction between the fluorine in the rock and sulfuric acid. HF subsequently reacts with the silicates in the digesting slurry to form fluosilicic acid as follows:

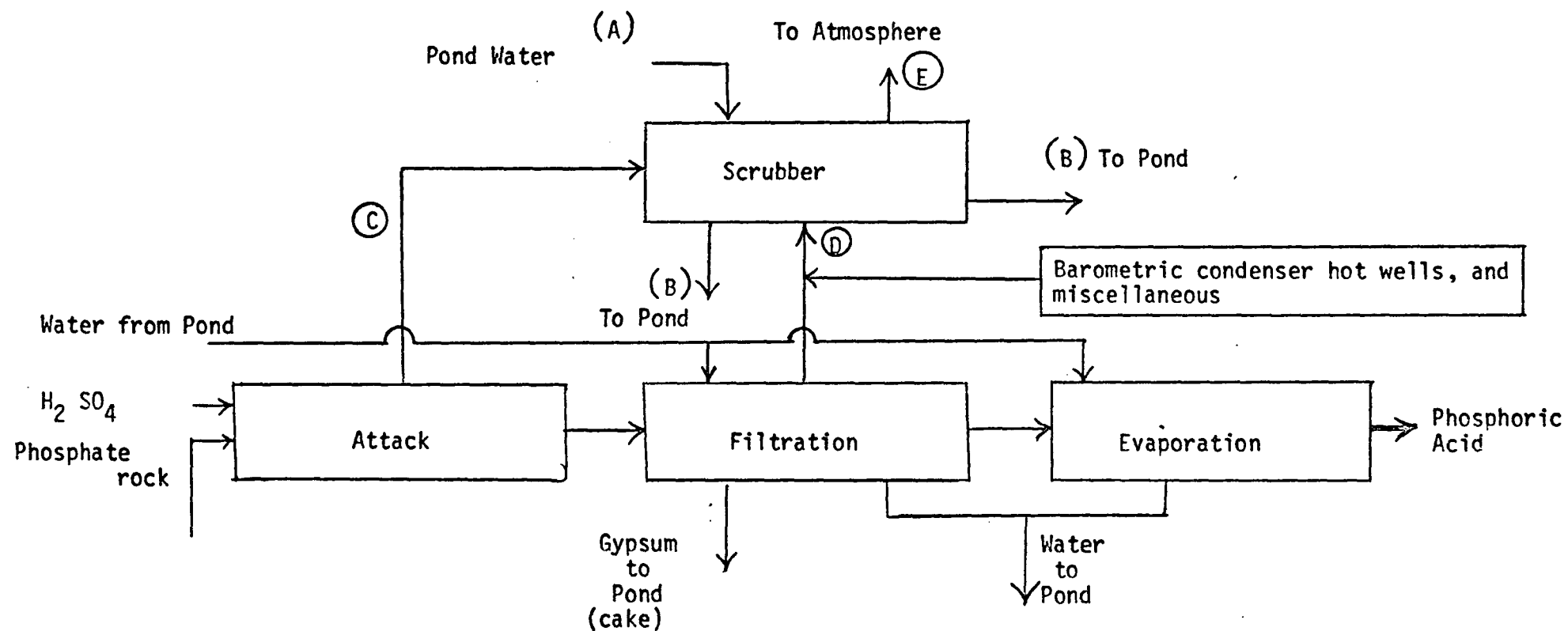


The fluosilicic acid in turn can decompose:



○ Stack Sample

() Grab Sample



Schematic Flow Diagram with Locations of Sampling Stations

W. R. GRACE

Wet Phosphoric Acid

Figure 1

TABLE 1
SUMMARY OF RESULTS
FLUORIDES

STATION C
Scrubber Inlet from Attack Tank

Run No.	1	2	3
Date	1-4-72	1-5-72	1-5-72
Barometric pressure, inches Hg	29.92	29.92	29.92
Stack pressure, inches Hg	29.92	29.92	29.92
Stack gas moisture, % volume	10.4	28.2	27.2
Average stack gas temperature, °F.	150	151	154
Stack gas flow rate @ S.T.P., SCFM	5813	4867	5010
Vol. dry gas @ S.T.P.* , SCF	78.951	97.143	83.965
Fluoride, water soluble, mg	106.95	837	603
Fluoride, total, mg	106.95	837	737
Fluoride, water soluble, gr/SCF	0.020	0.132	0.108
Fluoride, total, gr/SCF	0.020	0.132	0.132
Fluoride, water soluble, gr/CF stk. cond.	0.016	0.082	0.067
Fluoride, total, gr/CF stk. cond.	0.016	0.082	0.082
Fluoride, water soluble, lb/hour	1.01	5.51	4.62
Fluoride, total, lb/hour	1.01	5.51	5.65
Fluoride, water soluble, lb/ton P_2O_5 Fed.	0.033	0.182	0.153
Fluoride, total, lb/ton P_2O_5 Fed.	0.033	0.182	0.187
Scrubber efficiency, %			

* Dry, 70°F., 29.92 inches Hg.

TABLE 2
SUMMARY OF RESULTS
FLUORIDES

STATION D
Scrubber Inlet from Filtration System

Run No.	1	2	3
Date	1-4-72	1-5-72	1-5-72
Barometric pressure, inches Hg	30	30	30
Stack pressure, inches Hg	29.68	29.68	29.68
Stack gas moisture, % volume	6.6	6.2	6.9
Average stack gas temperature, °F.	103	103	105
Stack gas flow rate @ S.T.P., SCFM	11686	11231	11407
Vol. dry gas @ S.T.P.* , SCF	77.449	77.631	78.343
Fluoride, water soluble, mg	47.7	39.165	38.1
Fluoride, total, mg	47.7	39.755	38.1
Fluoride, water soluble, gr/SCF	0.009	0.008	0.007
Fluoride, total, gr/SCF	0.009	0.008	0.007
Fluoride, water soluble, gr/CF stk. cond.	0.008	0.007	0.006
Fluoride, total, gr/CF stk. cond.	0.008	0.007	0.006
Fluoride, water soluble, lb/hour	0.92	0.71	0.71
Fluoride, total, lb/hour	0.92	0.75	0.71
Fluoride, water soluble, lb/ton P_2O_5 Fed.	0.031	0.025	0.024
Fluoride, total, lb/ton P_2O_5 Fed.	0.031	0.025	0.024
Scrubber efficiency, %			

* Dry, 70°F., 29.92 inches Hg.

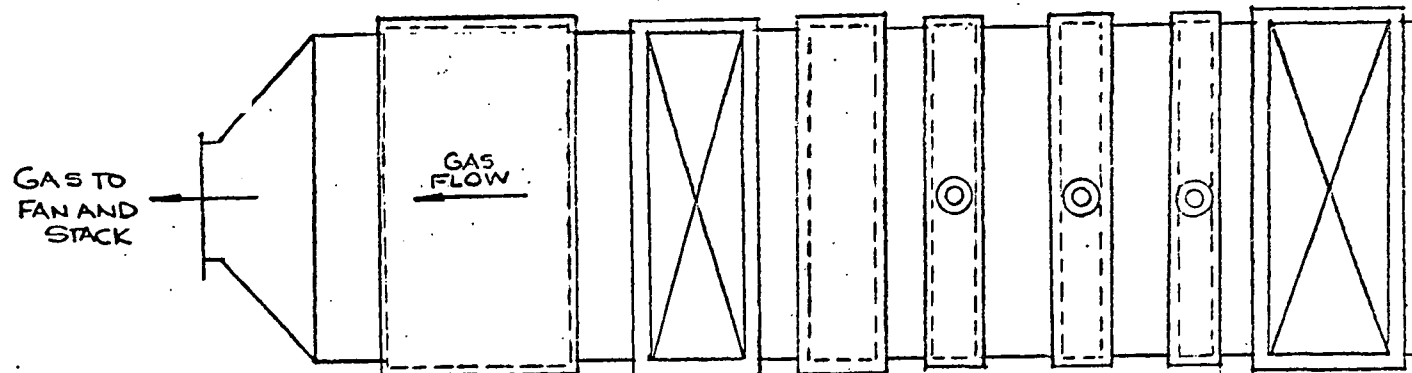
TABLE 3
SUMMARY OF RESULTS
FLUORIDES

STATION E
Scrubber Outlet

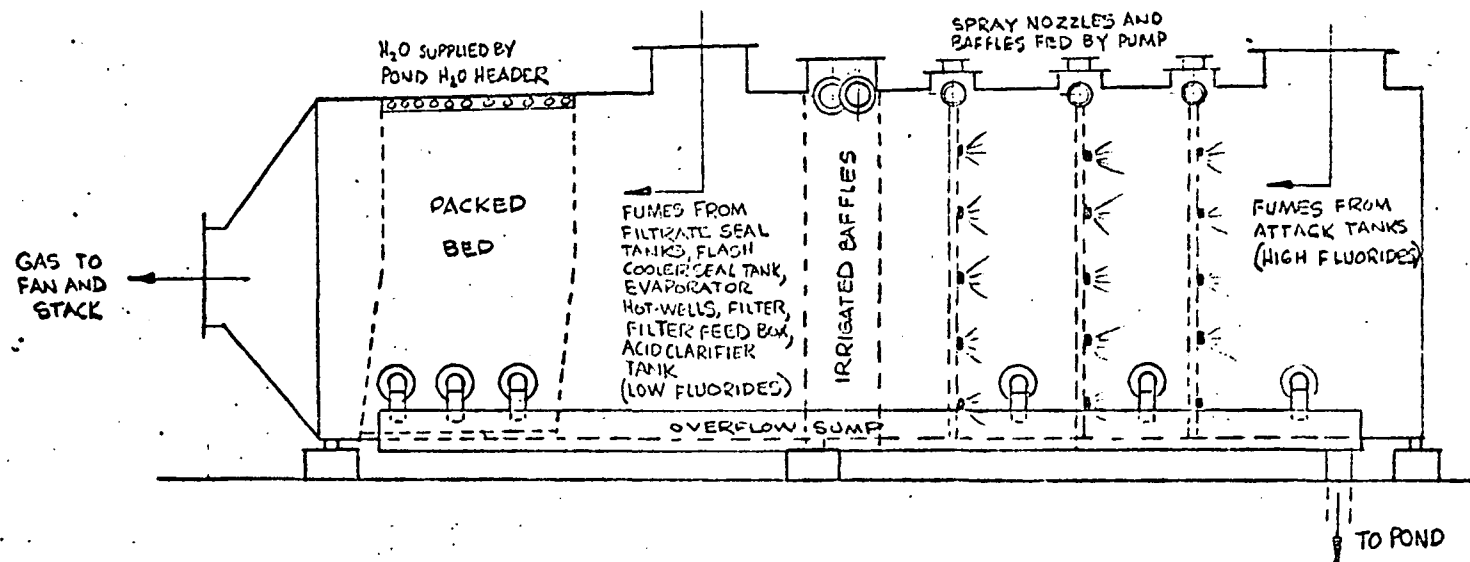
Run No.	1	2	3
Date	1-4-72	1-5-72	1-5-72
Barometric pressure, inches Hg	30	30	30
Stack pressure, inches Hg	30	30	30
Stack gas moisture, % volume	4.9	5.2	5.2
Average stack gas temperature, °F.	93	94	96
Stack gas flow rate @ S.T.P., SCFM	15981	16832	16825
Vol. dry gas @ S.T.P.* , SCF	55.057	63.498	63.282
Fluoride, water soluble, mg	3.6	7.2	3.535
Fluoride, total, mg	3.6	7.2	3.609
Fluoride, water soluble, gr/SCF	0.001	0.002	0.001
Fluoride, total, gr/SCF	0.001	0.002	0.001
Fluoride, water soluble, gr/CF stk. cond.	0.001	0.002	0.001
Fluoride, total, gr/CF stk. cond.	0.001	0.002	0.001
Fluoride, water soluble, lb/hour	0.13	0.25	0.12
Fluoride, total, lb/hour	0.13	0.25	0.12
Fluoride, water soluble, lb/ton P ₂ O ₅ Fed.	0.004	0.008	0.004
Fluoride, total, lb/ton P ₂ O ₅ Fed.	0.004	0.008	0.004
Scrubber efficiency, %	93.3	96.0	97.7

* Dry, 70°F., 29.92 inches Hg.

W.R. GRACE & COMPANY
FUME SCRUBBER - WET PROCESS PHOSPHORIC ACID



PLAN



ELEVATION

Figure 2

The reactor consists of a series of tanks with the slurry alternately overflowing and underflowing from one compartment to the next. The multi-compartment design allows temperature and agitation to vary throughout the reaction sequence as slurry recirculates through the tank arrangement. The acidulation or digestion step is a highly exothermic reaction requiring considerable apparatus for cooling. A vacuum flash cooler maintains temperature in the reactor and degasifies the recirculated slurry of dissolved air, carbon dioxide, and fluorides.

The acid slurry from the last attack compartment of the reactor is pumped to a rotating-tilting-pan filter where the phosphoric acid is filtered from the gypsum. The byproduct gypsum is repulped and pumped to a nearby pond. The product acid is pumped to a storage vessel and then to vacuum evaporators in which the acid can be concentrated from 30 to 54% P_2O_5 .

The crossflow scrubber design has a primary scrubbing section consisting of countercurrent sprays of gypsum pond water. The gases then pass through a section of irrigated baffles before flowing through the packing in the secondary scrubbing section.

PROCESS OPERATION

Run 1 was conducted from 2:15 p.m. to 4:15 p.m. on January 4, 1972, on WRG's "V" train phosphoric acid plant. This test involved three sampling points (2 inlets and one outlet). Process operation was normal except that some filter pans were dumping wet. This condition had existed for several days.

Less than optimum filtration continued throughout Run II, which was begun at 9:00 a.m. on January 5. Other process conditions were essentially normal. No. 1 and 2 evaporators were brought on stream at the beginning of the run, thereby doubling the production rate of 54% acid. With all four evaporators running, the production of fluosilicic acid was also increased.

Scrubber conditions changed slightly during Run II. The booster water pump discharge pressure decreased from 42 psig during Run I to 40 psig just prior to Run II. This was caused by reduced pond water header pressure. Pressure drop through the scrubber remained constant at 5" H₂O during Run II.

The third run got underway at 12:00 p.m. the same day. Process conditions remained basically the same as for Run II except that evaporation rates were increased slightly.

LOCATION OF SAMPLING POINTS

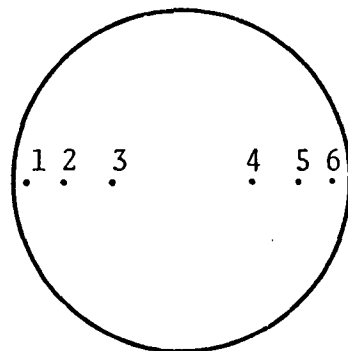
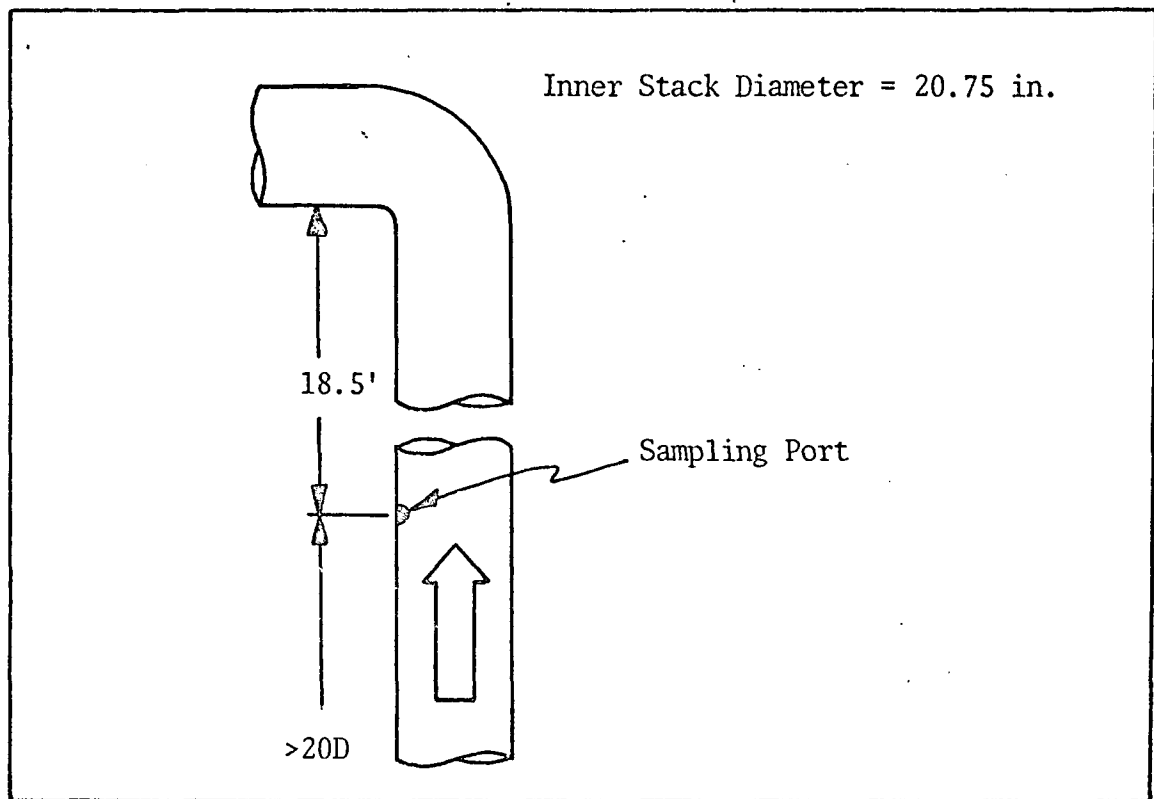
The sampling sites and number of traverse points were selected as per "Method I - Sampling and Velocity Traverses for Stationary Sources, Part 60, Subchapter C, Chapter 1, Title 40, "Federal Register, No. 247-Pt. II-1.

The above method suggests using two perpendicular diameters of traverse points per sampling station; however, on-site conditions necessitated the use of only one traverse diameter. The suggested number of traverse points per diameter was used where possible without sampling within one inch of the inner wall.

Figures 3 through 5 are schematic diagrams of the stack configuration near the sampling location, and the sampling points traversed during the emission tests.

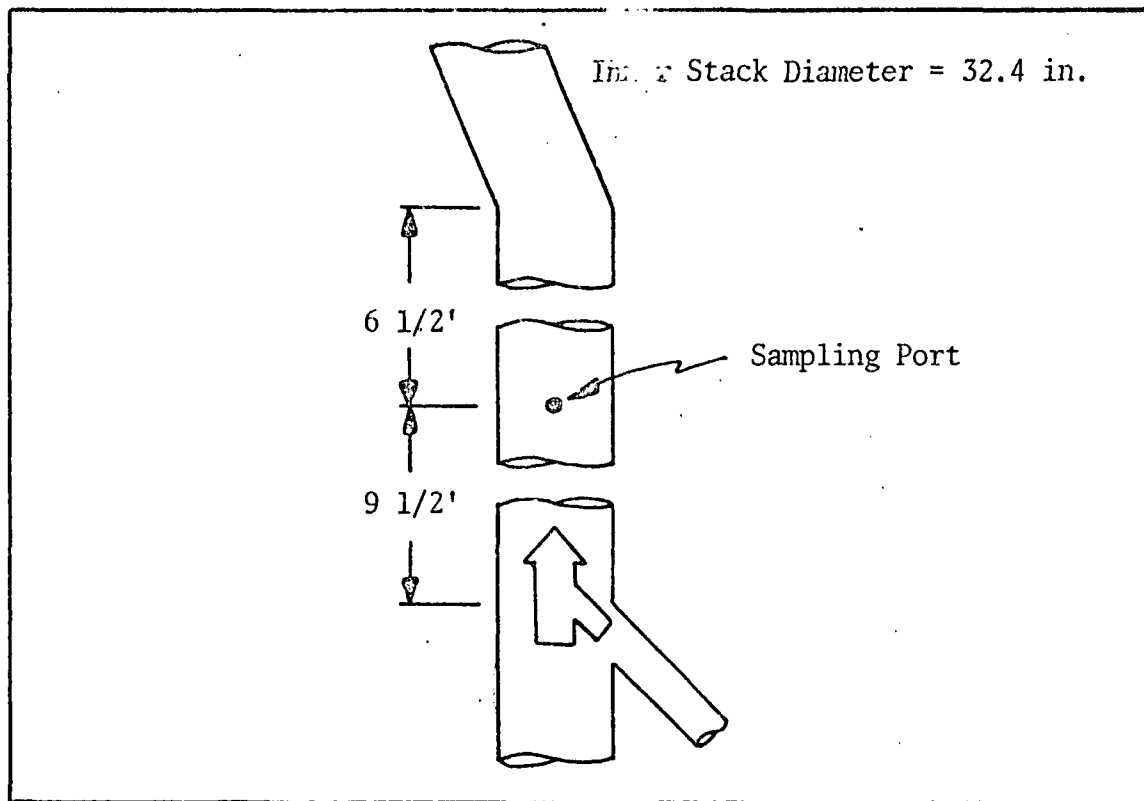
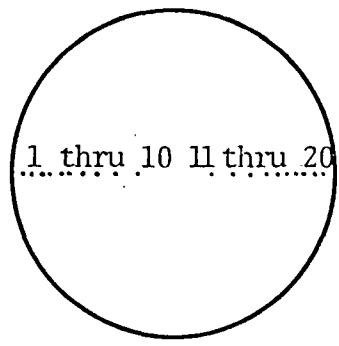
TABLE 4
DESCRIPTION OF SAMPLING POINTS

Sampling Point Description	Sampling Point Identification	Number of Traverse Points	Stack Diameter (Ft.)
Attack tank effluent	C	6	1.72
Filtration effluent	D	20	2.70
Scrubber outlet	E	6	3.9



Sampling Point	Distance From Inner Wall (in.)
1	1
2	3
3	6
4	14 5/8
5	17 3/4
6	19 3/4

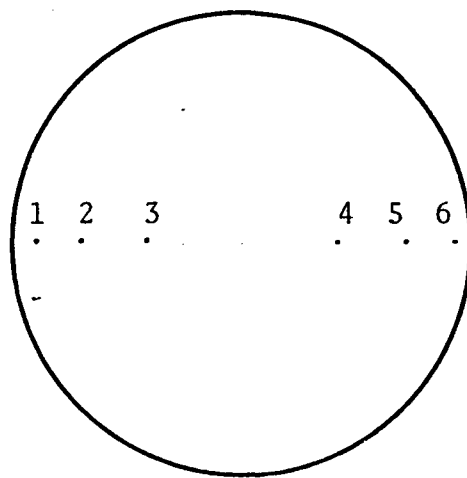
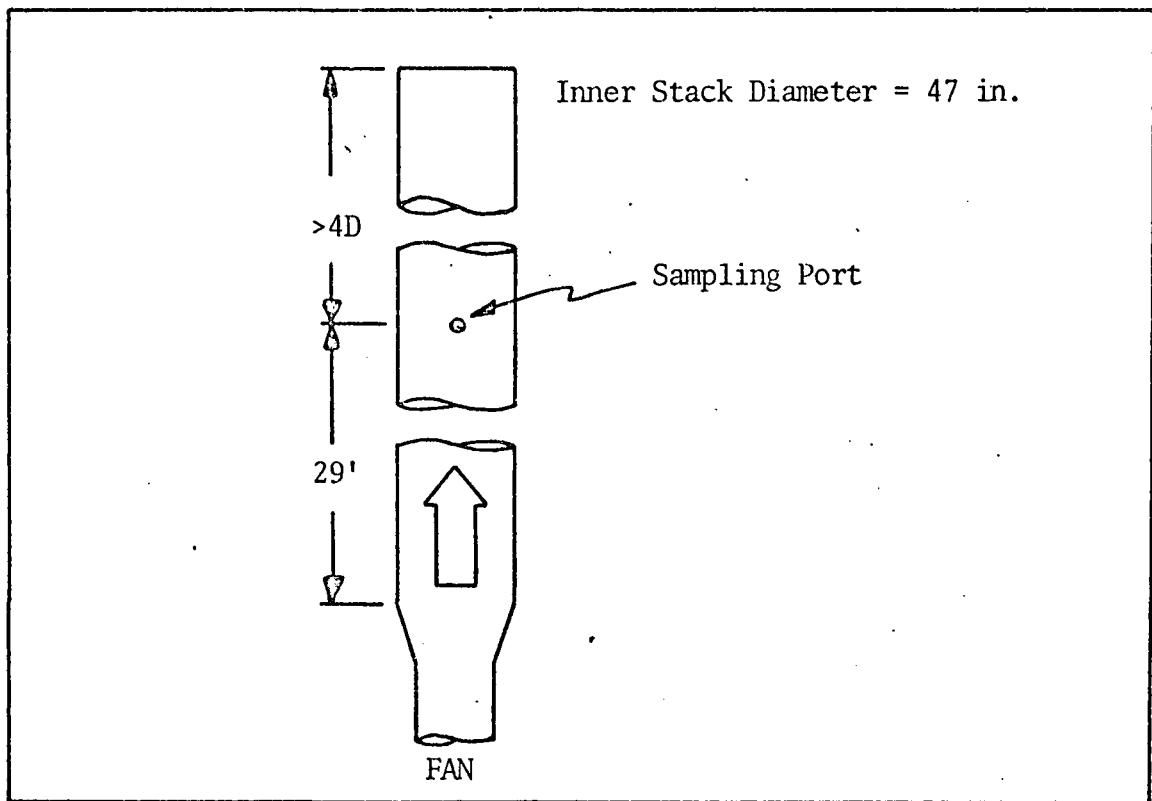
FIGURE 3
LOCATION OF PORT AND SAMPLING POINTS
STATION C, W.R. GRACE

A circular diagram representing the cross-section of the stack. It is divided into 20 equal segments, numbered 1 through 20. The text "1 thru 10 11 thru 20" is written across the center of the circle.

Sampling Point	Distance from Inner Wall (in.)
1	1 1/8
2	1 15/16
3	2 13/16
4	3 3/4
5	4 23/32
6	5 27/32
7	7 1/16
8	8 15/32
9	10 7/32
10	12 3/4
11	19 11/16
12	22 3/16
13	23 15/16
14	25 11/32
15	26 9/16
16	27 11/16
17	28 21/32
18	29 19/32
19	30 15/32
20	31 1/8

FIGURE 4
LOCATION OF PORT AND SAMPLING POINTS
STATION D, W.R. GRACE



Sampling Point	Distance From Inner Wall (in.)
1	2 1/16
2	6 7/8
3	13 7/8
4	33 1/8
5	40 3/32
6	44 15/16

FIGURE 5
LOCATION OF PORT AND SAMPLING POINTS
STATION E, W.R. GRACE

SAMPLING AND ANALYTICAL PROCEDURES

A. Preliminary Moisture Determination

The preliminary moisture content of the stack gases at each sampling site was determined by Method 4 of the Federal Register (Volume 36, Number 247, Part II, December 23, 1971).

The only significant difference between F.R. Method 4 and the method used was the configuration of the sampling train (see Figure 6). The sampling train used in these tests consisted of the first two midget impingers with 5 grams of distilled-deionized water followed by two dry midget impingers in place of a silica gel cartridge.

At sampling sites where liquid entrainment was a problem, the preliminary and final moisture contents were determined from wet and dry bulb thermometry. See Appendix B for the data used in determining the preliminary moisture content of the stack gases.

After completing the moisture run, the total impinger liquid plus water rinsings of the probe tip through the fourth impinger were placed in an 8 ounce polyethylene container. The samples were held by EPA personnel for further analyses.

Field data sheets are contained in Appendix B.

B. Preliminary Velocity Determination

Method 2 of the above mentioned Federal Register was used as a guide in determining the preliminary stack gas velocity for each source tested. The major difference was that only the maximum and minimum velocity heads

across each stack area were determined so that a proper nozzle size could be selected. During each of the three fluoride emission tests, velocity head readings were taken at points selected by using Method 1 of the Federal Register.

Stack pressure and temperature measurements were also made during the preliminary velocity determinations.

C. Sampling for Fluoride Emissions

The sampling procedure used for determining fluoride emissions was similar to Method 5 of the Federal Register. The major difference between the two methods was the configuration of the sampling train. The sampling train described in the Federal Register has a heated box containing the filter holder directly following the glass probe. The sampling train used in these tests contained no heated box and the filter holder was placed between the third and fourth impingers (between dry impinger and silica gel impinger) to prevent sample carryover. Figure 7 is a schematic diagram of the sampling train used.

After the selection of the sampling site and the minimum number of sampling points per Method 2 of the above mentioned Federal Register, three separate test runs were performed. For each run, the required stack and sampling parameters were recorded on field data sheets. They are included in Appendix B. Readings were taken at each traverse point at least every five minutes, and when significant changes in stack parameters necessitated additional adjustments to maintain an isokinetic flow rate. Nomographs were used to aid in the rapid adjustment of the sampling rate. The traverse points were selected to maintain at least one inch from the inner stack wall.

After each run, the liquid volume in the first three impingers was measured volumetrically and the silica gel was reweighed. The impinger liquid, the filter, plus the water washings of the probe and other sampling train components up to the silica gel were placed into polyethylene containers. During some runs the different sample fractions were placed in separate containers, while during others, all of the recovered sample was placed into one container.

D. Liquid and Product Grab Samples

Periodically during each test run, grab samples of the raw materials, finished product, and scrubber liquid were taken, and the temperature and pH were determined at the site. On some occasions, the samples were split with the plant personnel so that comparative analyses could be performed. Field data sheets are included in Appendix B.

E. Laboratory Analysis Procedures

Water soluble fluorides were done by a sulfuric acid distillation followed by the SPADNS-ZIRCONIUM LAKE METHOD. Water insoluble fluorides were first fused with NaOH followed by a sulfuric acid distillation then by the SPADNS-ZIRCONIUM LAKE METHOD.

P_2O_5 analysis of the stack effluent was done by EPA personnel. All other P_2O_5 analyses were done by plant personnel.

For more details of exact method used see Appendix C.

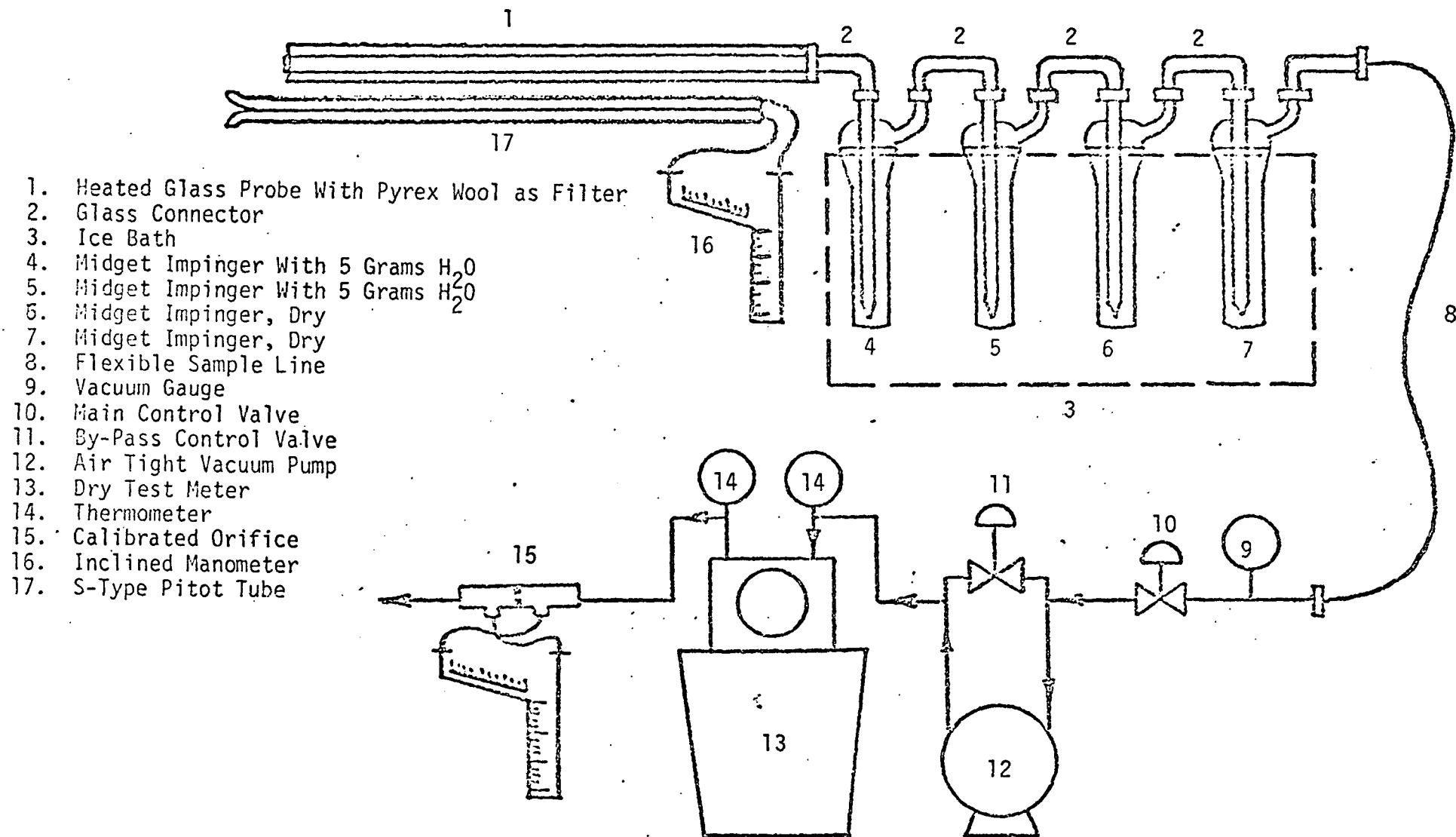


Figure 6
 MOISTURE SAMPLING TRAIN

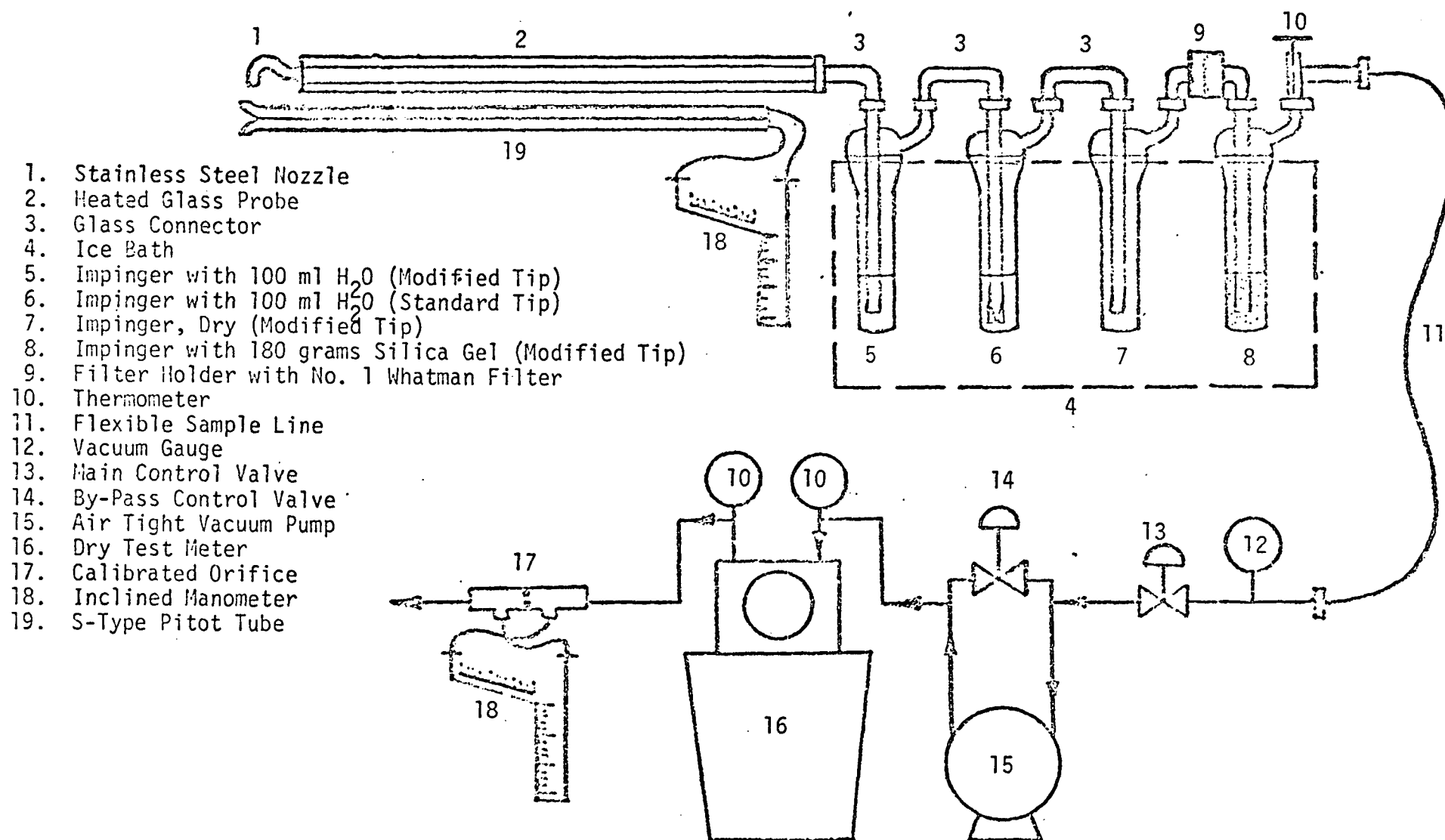


Figure 7
 FLUORIDE SAMPLING TRAIN

APPENDIX

APPENDIX A

Emission Calculations and Results

E.E.I. SOURCE SAMPLING NOMENCLATURE SHEET

- PB - Barometric pressure, inches Hg
 - PS - Stack pressure, inches Hg
 - As - Stack area, sq. ft.
 - AS' - Effective area of positive stack gas flow, sq. ft.
 - NPTS - Number of traverse points where the pitot velocity head was greater than zero
 - TS - Stack temperature, °R
 - TM - Meter temperature, °R
 - $\bar{H}$ - Average square root of velocity head, $\sqrt{\text{inches H}_2\text{O}}$
 - ΔH - Average meter orifice pressure differential, inches H₂O
 - AN - Sampling nozzle area, square feet
 - CP - S-type pitot tube correction factor
 - VM - Recorded meter volume sample, cubic feet (meter conditions)
 - VC - Condensate and silica gel increase in impringers, milliliters
 - Po - Pressure at the dry test meter orifice, $\left[\bar{P}B + \frac{\Delta H}{13.6} \right]$ inches Hg
 - STP - Standard conditions, dry, 70°F, 29.92 inches Hg
- - - - -
- VWV - Conversion of condensate in milliliters to water vapor in cubic feet (STP)
 - VSTPD - Volume sampled, cubic feet (STP)
 - VT - Total water vapor volume and dry gas volume sampled, cubic feet (STP)
 - W - Moisture fraction of stack gas
 - FDA - Dry gas fraction
 - MD - Molecular weight of stack gas, lbs/lb-mole (dry conditions)
 - MS - Molecular weight of stack gas, lbs/lb-mole (stack conditions)
 - GS - Specific gravity of stack gas, referred to air
 - EA - Excess air, %
 - $\sqrt{H \times TS}$ - Average square root of velocity head times stack temperature
 - $\bar{U}$ - Stack gas velocity, feet per minute
 - QS - Stack gas flow rate, cubic feet per minute (stack conditions)
 - QD - Stack gas flow rate, cubic feet per minute (dry conditions)
 - QSTPD - Stack gas flow rate, cubic feet per minute (STP)
 - PISO - Percent isokinetic volume sampled (method described in Federal Register)
 - TIME - Total Sample Time (minutes)

EQUATIONS FOR CALCULATING FLUORIDE EMISSIONS

$$VWV = (0.0474) \times (VC)$$

$$VSTPD = (17.71 \times (VM) \times (PB + \frac{\Delta H}{13.6}) \div TM$$

$$VT = (VWV) + (VSTPD)$$

$$W = (VWV) \div (VT)$$

$$FDA = (1.0) - (W)$$

FMOIST = Assumed moisture fraction

$$MD = (0.44 \times \% CO_2) + (0.32 \times \% O_2) + (0.28 \times \% N_2) + (0.28 \times \% CO)$$

$$MS = (MD \times FDA) + (18 \times W)$$

$$GS = (MS) \div (28.99)$$

$$EA = \left[(100) \times (\% O_2 - \frac{\% CO}{2}) \right] \div \left[(0.266 \times \% N_2) - (\% O_2 - \frac{\% CO}{2}) \right]$$

$$\underline{U} = (174) \times (CP) \times (\underline{H}) \times \sqrt{(TS \times 29.92) \div (GS \times PS)}$$

$$QS = (\underline{U}) \times (AS)$$

$$QD = (QS) \times (FDA)$$

$$QSTPD = (QD) \times (\frac{530}{29.92}) \times (\frac{PS}{TS})$$

$$PISO = (0.00267 \times VC \times TS) + (P_o \times TS \times VM \div TM) \div (\text{Time} \times \underline{U} \times PS \times AN)$$

Fluoride Emissions:

MG = Milligrams of fluoride from lab analysis

$$\text{Grains/SCF} = (0.01543) \times (MG) \div VSTPD$$

$$\text{Grains/CF, Stack Cond.} = (17.71) \times (PS) \times (FDA) \times (\text{Grains/SCF}) \div (TS)$$

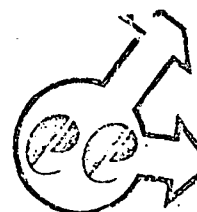
$$\text{Lbs/hour} = (\text{Grains/SCF}) \times (0.00857) \times (QSTPD)$$

P₂O₅ Fed = Tons/hour, determined from plant data.

$$\text{Lbs/ton P}_{20}_5 \text{ Fed} = (\text{lbs/hour}) \div (\text{Tons/hour P}_{20}_5 \text{ Fed})$$

FLUORIDE EMISSIONS

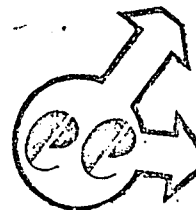
SOURCE TEST DATA



TEST NO. _____ NO. OF RUNS 3
 NAME OF FIRM W. R. Grace
 LOCATION OF PLANT Bartow, Florida
 TYPE OF PLANT Wet Phosphoric Acid
 CONTROL EQUIPMENT _____
 SAMPLING POINT LOCATION Station "C" -- Inlet to the Scrubber
 POLLUTANTS SAMPLED Fluoride

	1	2	3
1) RUN NO.			
2) DATE	1/4/72	1/5/72	1/5/72
3) TIME BEGAN	14:18	9:00	12:00
4) TIME END	15:58	11:45	14:05
5) BAROMETRIC PRESSURE, "Hg ABSOLUTE	29.92	29.92	29.92
6) METER ORIFICE PRESSURE DROP, "H ₂ O	2.02	2.29	1.63
7) VOL DRY GAS @ METER CONDITIONS, ft ³	81.085	97.605	86.296
8) AVERAGE GAS METER TEMPERATURE, °F	87	75.5	86.9
9) VOL DRY GAS @ S.T.P.*, ft ³	78.951	97.143	83.965
10) TOTAL H ₂ O COLLECTED, ml	193.5	803.8	661.9
11) VOL WATER VAPOR COLL. @ S.T.P., ft ³	9.17	38.1	31.37
12) STACK GAS MOISTURE, %VOLUME	10.4	28.2	27.2
13) ASSUMED STACK GAS MOISTURE, %VOLUME	1850	1050	3000
14) % CO ₂			
15) % O ₂			
16) % CO			
17) % N ₂			
18) % EXCESS AIR	0	0	0
19) MOLECULAR WT. OF STACK GAS, DRY	28.99	28.99	28.99
20) MOLECULAR WT. OF STACK GAS, STACK COND.	27.85	25.89	26
21) STACK GAS SP. GRAVITY, REF. TO AIR	0.96	0.89	0.9
22) AVG VEL. HEAD OF STACK GAS, "H ₂ O	0.873	0.88	0.897
23) AVERAGE STACK GAS TEMPERATURE, °F	150.4	151.8	154
24) PITOT CORRECTION FACTOR	0.83	0.83	0.83
25) STACK PRESSURE, "Hg ABSOLUTE	29.9	29.9	29.9
26) STACK GAS VEL @ STACK COND., fpm	3178.9	3328	3392.1
27) STACK AREA, ft ²	2.35	2.35	2.35
28) STACK GAS FLOW RATE @ S.T.P., scfm	5813	4867	5010
29) NET TIME OF TEST, min.	100	120	120
30) SAMPLING NOZZLE DIAMETER, in.	0.25	0.25	0.25
31) PERCENT ISOKINETIC	93.7	114.7	96.3
32) FLUORIDE - WATER SOLUBLE, MG	106.95	837	603
33) FLUORIDE - TOTAL, MG	106.95	837	737
34) FLUORIDE - WATER SOLUBLE, GR/SCF	0.0203	0.1321	0.1076
35) FLUORIDE - TOTAL, GR/SCF	0.0203	0.1321	0.1315
36) FLUORIDE - WATER SOL. GR/CF STK COND.	0.0158	0.082	0.0675
37) FLUORIDE - TOTAL GR/CF STK COND.	0.0158	0.082	0.0825
38) FLUORIDE - WATER SOLUBLE, LB/HOUR	1.0119	5.5083	4.6203
39) FLUORIDE - TOTAL, LB/HOUR	1.0119	5.5083	5.647
41) FLUORIDE - WATER SOL. LB/TON P205 FED.	0.0335	0.1824	0.153
42) FLUORIDE - TOTAL, LBS/TON P205 FED.	0.0335	0.1824	0.187

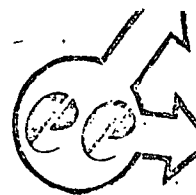
SOURCE TEST DATA



TEST NO. _____ NO. OF RUNS 3
 NAME OF FIRM W. R. Grace
 LOCATION OF PLANT Bartow, Florida
 TYPE OF PLANT Wet Phosphoric Acid
 CONTROL EQUIPMENT _____
 SAMPLING POINT LOCATION Station "D" -- Inlet to the Scrubber
 POLLUTANTS SAMPLED Fluoride

	1	2	3
1) RUN NO.			
2) DATE	1/4/72	1/5/72	1/5/72
3) TIME BEGAN	14:20	9:05	12:00
4) TIME END	16:20	11:05	14:00
5) BAROMETRIC PRESSURE, "Hg ABSOLUTE	30	30	30
6) METER ORIFICE PRESSURE DROP, "H ₂ O	1.38	1.33	1.38
7) VOL DRY GAS @ METER CONDITIONS, ft ³	79.552	78.054	80.261
8) AVERAGE GAS METER TEMPERATURE, °F	87.7	76.1	86.3
9) VOL DRY GAS @ S.T.P.*, ft ³	77.449	77.631	78.343
10) TOTAL H ₂ O COLLECTED, ml	116.3	109	121.9
11) VOL WATER VAPOR COLL. @ S.T.P., ft ³	5.51	5.17	5.78
12) STACK GAS MOISTURE, %VOLUME	6.6	6.2	6.9
13) ASSUMED STACK GAS MOISTURE, %VOLUME	9	670	670
14) % CO ₂	0.2	0.2	0.4
15) % O ₂	20.5	20.5	20.4
16) % CO	0	0	0
17) % N ₂	79.3	79.3	79.2
18) % EXCESS AIR	3452	3452	3058
19) MOLECULAR WT. OF STACK GAS, DRY	28.85	28.85	28.88
20) MOLECULAR WT. OF STACK GAS, STACK COND.	28.13	28.17	28.13
21) STACK GAS SP. GRAVITY, REF. TO AIR	0.97	0.97	0.97
22) AVG VEL. HEAD OF STACK GAS, "H ₂ O	0.665	0.637	0.652
23) AVERAGE STACK GAS TEMPERATURE, °F	103.1	103.4	105
24) PITOT CORRECTION FACTOR	0.83	0.83	0.83
25) STACK PRESSURE, "Hg ABSOLUTE	29.68	29.68	29.68
26) STACK GAS VEL @ STACK COND., fpm	2322.9	2223.7	2280.5
27) STACK AREA, ft ²	5.73	5.73	5.73
28) STACK GAS FLOW RATE @ S.T.P., scfm	11686	11231	11407
29) NET TIME OF TEST, min.	120	120	120
30) SAMPLING NOZZLE DIAMETER, in.	0.25	0.25	0.25
31) PERCENT ISOKINETIC	93.5	97.5	96.9
32) FLUORIDE - WATER SOLUBLE, MG	47.7	39.165	38.1
33) FLUORIDE - TOTAL, MG	47.7	39.755	38.1
34) FLUORIDE - WATER SOLUBLE, GR/SCF	0.0092	0.0077	0.0073
35) FLUORIDE - TOTAL, GR/SCF	0.0092	0.0078	0.0073
36) FLUORIDE - WATER SOL. GR/CF STK COND.	0.008	0.0068	0.0063
37) FLUORIDE - TOTAL GR/CF STK COND.	0.008	0.0069	0.0063
38) FLUORIDE - WATER SOLUBLE, LB/HOUR	0.9248	0.7437	0.7146
39) FLUORIDE - TOTAL, LB/HOUR	0.9248	0.7549	0.7146
41) FLUORIDE - WATER SOL. LB/TON P2O5 FED.	0.0306	0.0246	0.0237
42) FLUORIDE - TOTAL, LBS/TON P2O5 FED.	0.0306	0.025	0.0237

SOURCE TEST DATA



TEST NO. _____ NO. OF RUNS 3

NAME OF FIRM W. R. Grace

LOCATION OF PLANT Bartow, Florida

TYPE OF PLANT Wet Phosphoric Acid

CONTROL EQUIPMENT Fume Scrubber

SAMPLING POINT LOCATION Station "E" -- Outlet from the Scrubber

POLLUTANTS SAMPLED Fluoride

	1	2	3
1) RUN NO.	1	2	3
2) DATE	1/4/72	1/5/72	1/5/72
3) TIME BEGAN	14:15	9:00	12:00
4) TIME END	16:15	11:00	14:00
5) BAROMETRIC PRESSURE, "Hg ABSOLUTE	30	30	30
6) METER ORIFICE PRESSURE DROP, "H ₂ O	0.58	0.76	0.76
7) VOL DRY GAS @ METER CONDITIONS, ft ³	56.293	63.593	64.278
8) AVERAGE GAS METER TEMPERATURE, °F	84.1	73.2	80.8
9) VOL DRY GAS @ S.T.P.*, ft ³	55.057	63.498	63.282
10) TOTAL H ₂ O COLLECTED, ml	60.1	74.2	73.8
11) VOL WATER VAPOR COLL. @ S.T.P., ft ³	2.85	3.52	3.5
12) STACK GAS MOISTURE, %VOLUME	4.9	5.2	5.2
13) ASSUMED STACK GAS MOISTURE, %VOLUME	1200	500	50
14) % CO ₂			
15) % O ₂			
16) % CO			
17) % N ₂			
18) % EXCESS AIR	0	0	0
19) MOLECULAR WT. OF STACK GAS, DRY	28.99	28.99	28.99
20) MOLECULAR WT. OF STACK GAS, STACK COND.	28.45	28.41	28.41
21) STACK GAS SP. GRAVITY, REF. TO AIR	0.98	0.98	0.98
22) AVG VEL. HEAD OF STACK GAS, "H ₂ O	0.425	0.45	0.45
23) AVERAGE STACK GAS TEMPERATURE, °F	93.8	94.2	96.4
24) PITOT CORRECTION FACTOR	0.83	0.83	0.83
25) STACK PRESSURE, "Hg ABSOLUTE	30	30	30
26) STACK GAS VEL @ STACK COND., fpm	1457.8	1541.7	1547.2
27) STACK AREA, ft ²	12.05	12.05	12.05
28) STACK GAS FLOW RATE @ S.T.P., scfm	15981	16832	16825
29) NET TIME OF TEST, min.	120	120	120
30) SAMPLING NOZZLE DIAMETER, in.	0.25	0.25	0.25
31) PERCENT ISOKINETIC	101.2	110.8	110.5
32) FLUORIDE - WATER SOLUBLE, MG	3.6	7.2	3.535
33) FLUORIDE - TOTAL, MG	3.6	7.2	3.609
34) FLUORIDE - WATER SOLUBLE, GR/SCF	0.001	0.0017	0.0008
35) FLUORIDE - TOTAL, GR/SCF	0.001	0.0017	0.0009
36) FLOURIDE - WATER SOL. GR/CF STK COND.	0.0009	0.0016	0.0008
37) FLOURIDE - TOTAL GR/CF STK COND.	0.0009	0.0016	0.0008
38) FLUORIDE - WATER SOLUBLE, LB/HOUR	0.1349	0.2515	0.1221
39) FLUORIDE - TOTAL, LB/HOUR	0.1349	0.2515	0.1247
41) FLOURIDE - WATER SOL. LB/TON P205 FED.	0.0045	0.0083	0.004
42) FLUORIDE - TOTAL, LBS/TON P205 FED.	0.0045	0.0083	0.0041

*DRY, 70 °F, 29.92 in. Hg

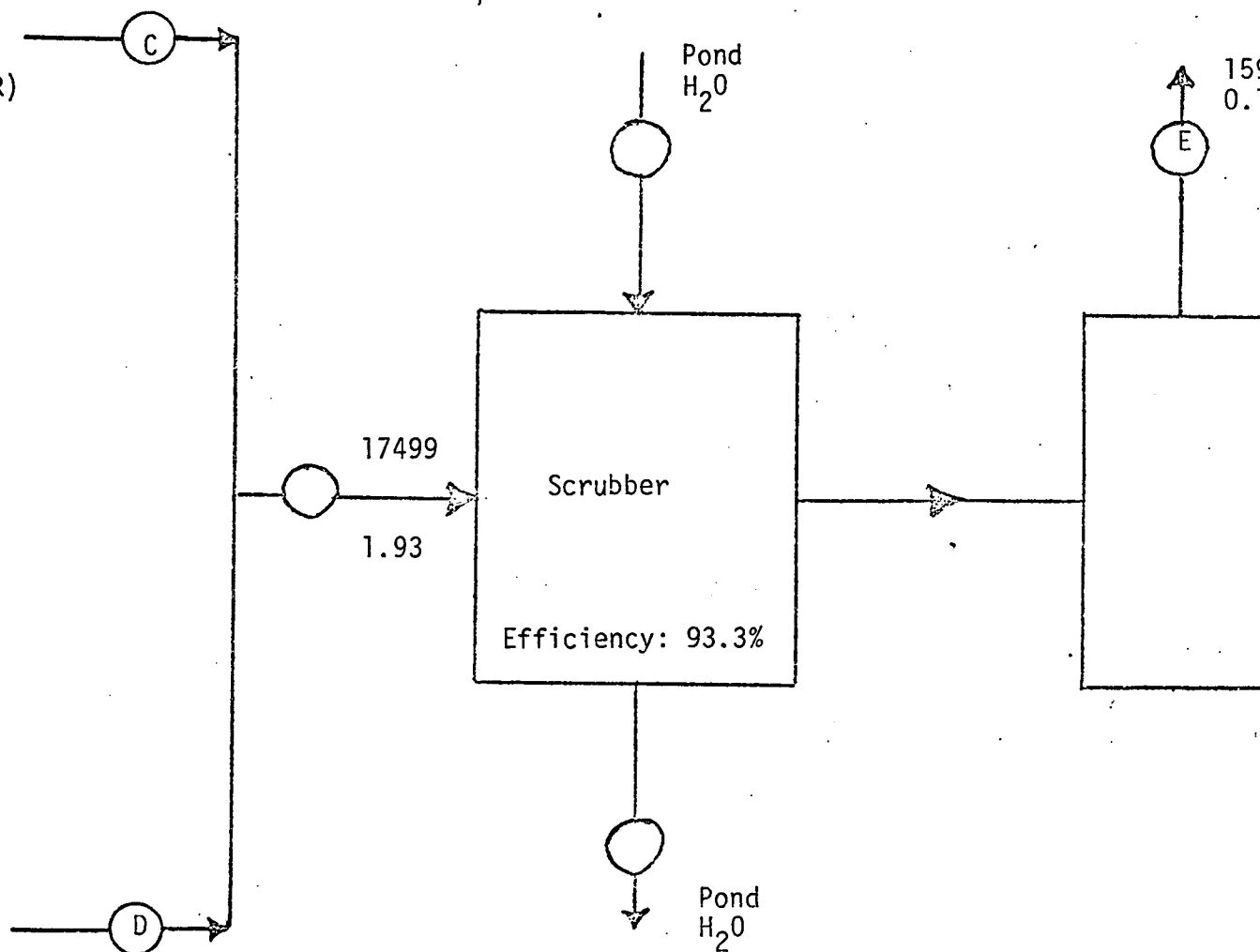
SCRUBBER EFFICIENCY

W. R. Grace
Wet Acid
Run 1

To atmosphere

Flow: 5813 (USCFM)
Total
Fluorides: 1.01 (#/HR)

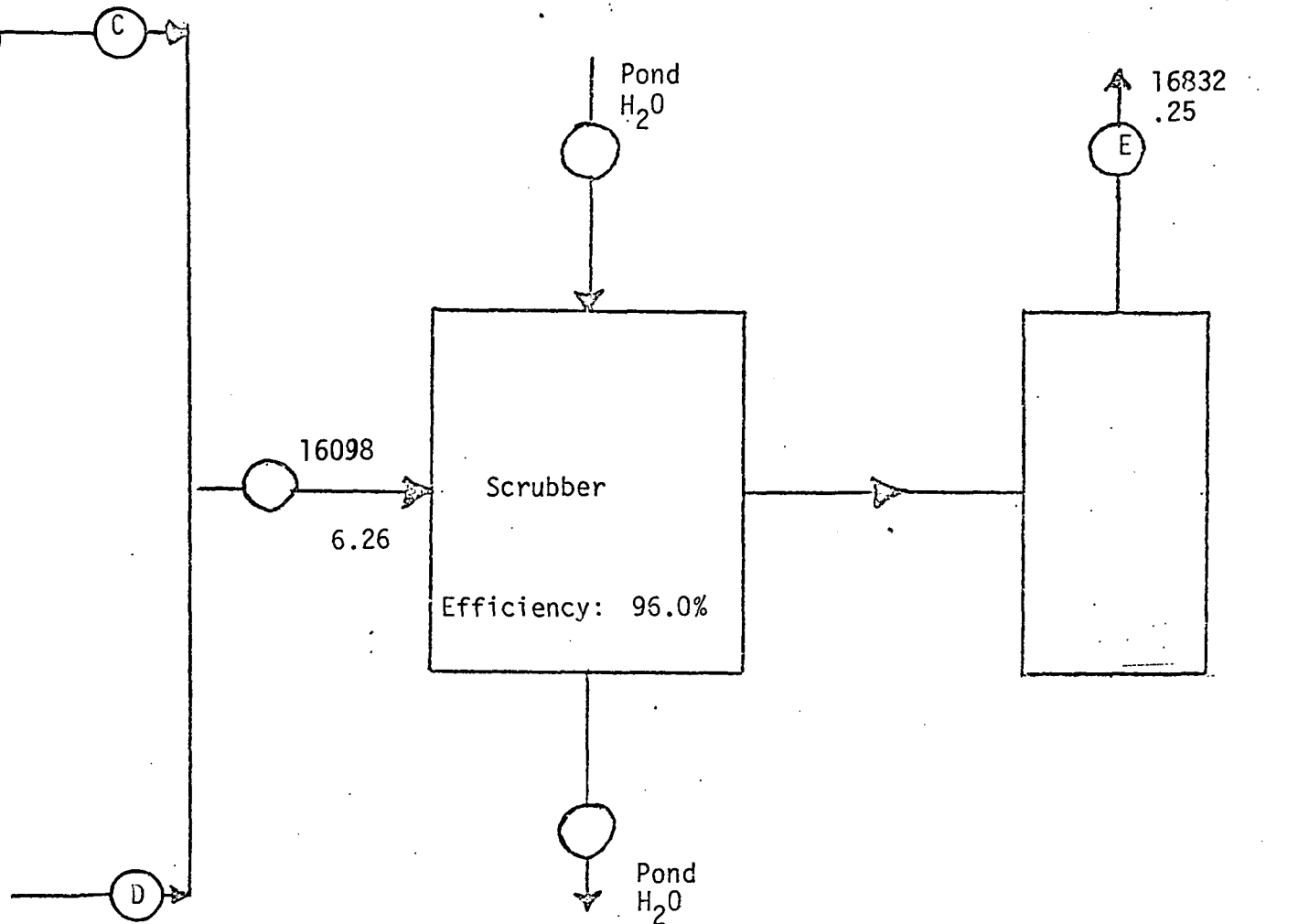
11686
0.92



W. R. Grace
Wet Acid
Run 2

Flow: 4867 (USCFM)
Total
Fluorides: 5.51 (#/HR)

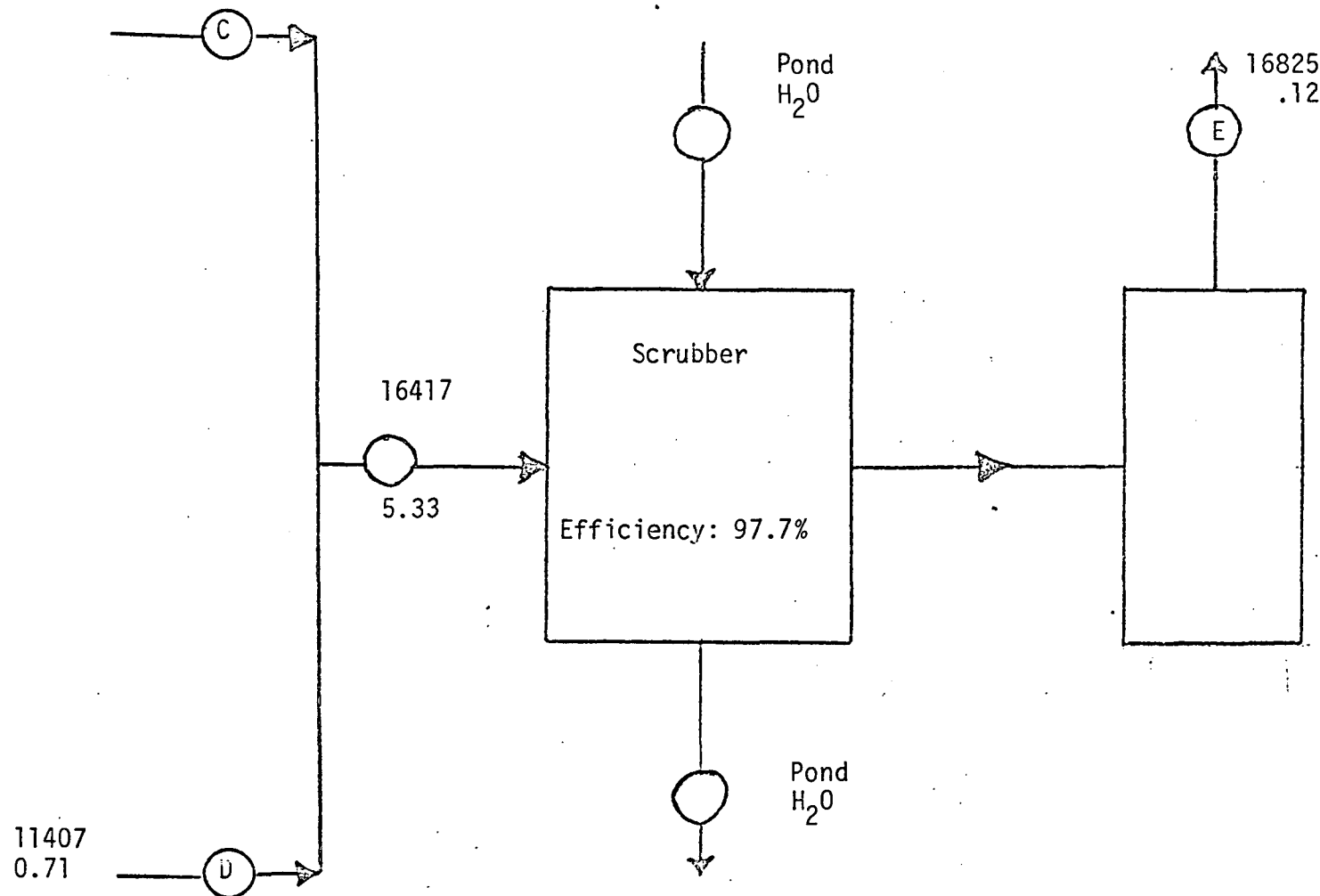
11231
0.75



Flow: 5010 (DSCFM)
Total 4.62 (#/HR)
Fluorides:

W. R. Grace
Wet Acid
Run 3

To atmosphere



APPENDIX B

Field Data

PRELIMINARY MOISTURE DETERMINATION

PRELIMINARY CHECK FOR STACK GAS
MOISTURE CONTENT AND SPECIFIC GRAVITY

Plant W. R. GRACE Stack Wet Phos. Acid - Station "C"
Date 1-4-72 Sample Time 1800 Barometric Pressure 30 "Hg

Moisture Content -- Method 1

Final Dry Test Meter Reading 213.725 Ft³
Initial Dry Test Meter Reading 212.725 Ft³
Dry Test Meter Volume Sampled 1.000 Ft³
Average Meter Temperature 71 °F
Average Meter Vacuum — "Hg
Average Meter Orifice ΔH 0.05 "H₂O
Sampling Rate 2.8 LPM
Barometric Pressure @ Meter Orifice 30 "Hg
Dry Gas Volume Sampled @ 70 °F, 29.92 "Hg 1.00 Ft³
Condensate Volume 5 ml
Water Vapor Volume @ 70 °F, 29.92 "Hg 0.24 Ft³
Moisture Fraction, H₂O 0.19 Fraction Of Dry Air, FDA 0.81

Moisture Content -- Method 2

Dry Bulb Temp. 153 °F Wet Bulb Temp. 149 °F Dew Point Temp. 147.8 °F
Vapor Pressure Of H₂O @ DP 2.18 "Hg Stack Pressure 29.9 "Hg
Moisture Fraction, H₂O 0.24 Fraction Of Dry Air, FDA 0.76

Specific Gravity

Dry Molecular Weight, $M_d = [0.44(\%CO_2)] + [0.32(\%CO_2)] + [0.28(\%N_2 + CO)] =$ _____
Molecular Weight @ Stack Conditions, $M_s = [(M_d) \times (FDA)] + [(18) \times (H_2O)] =$ _____
Specific Gravity (Referred to air), $G_s = (M_s) \div (28.99) =$ _____

PRELIMINARY CHECK FOR STACK GAS
MOISTURE CONTENT AND SPECIFIC GRAVITY

Plant W. R. GRACE Stack Wet Phos. Acid - Station "D"
Date 1-4-72 Sample Time 1500-1015 Barometric Pressure 30 "Hg

Moisture Content -- Method 1

Final Dry Test Meter Reading 908.700 Ft³
Initial Dry Test Meter Reading 906.700 Ft³
Dry Test Meter Volume Sampled 2.000 Ft³
Average Meter Temperature 71 °F
Average Meter Vacuum — "Hg
Average Meter Orifice Δ H 0.05 "H₂O
Sampling Rate 3.8 LPM
Barometric Pressure @ Meter Orifice 30 "Hg
Dry Gas Volume Sampled @ 70 °F, 29.92 "Hg 2.00 Ft³
Condensate Volume 9 ml
Water Vapor Volume @ 70 °F, 29.92 "Hg 0.43 Ft³
Moisture Fraction, H₂O 0.18 Fraction Of Dry Air, FDA 0.82

Moisture Content -- Method 2

Dry Bulb Temp. 101 °F Wet Bulb Temp. 91 °F Dew Point Temp. 88 °F
Vapor Pressure Of H₂O @ DP 1.3 "Hg Stack Pressure 29.68 "Hg
Moisture Fraction, H₂O 0.04 Fraction Of Dry Air, FDA 0.96

Specific Gravity

Dry Molecular Weight, $M_d = [0.44(\%CO_2)] + [0.32(\%O_2)] + [0.28(\%N_2 + CO)] =$ _____
Molecular Weight @ Stack Conditions, $M_s = [(M_d) \times (FDA)] + [(18) \times (H_2O)] =$ _____
Specific Gravity (Referred to air), $G_s = (M_s) \div (28.99) =$ _____

PRELIMINARY CHECK FOR STACK GAS
MOISTURE CONTENT AND SPECIFIC GRAVITY

Plant W. R. Grace Stack Wet Plus. Acid-Station "E"
Date 1-4-72 Sample Time 1000 Barometric Pressure 30 "Hg

Moisture Content -- Method 1

Final Dry Test Meter Reading 702.589 Ft³
Initial Dry Test Meter Reading 700.607 Ft³
Dry Test Meter Volume Sampled 1.972 Ft³
Average Meter Temperature 74 °F
Average Meter Vacuum — "Hg
Average Meter Orifice ΔH 0.06 "H₂O
Sampling Rate 2.8 LPM
Barometric Pressure @ Meter Orifice 30 "Hg
Dry Gas Volume Sampled @ 70 °F, 29.92 "Hg 1.96 Ft³
Condensate Volume 16 ml
Water Vapor Volume @ 70 °F, 29.92 "Hg 0.76 Ft³
Moisture Fraction, H₂O 0.28 Fraction Of Dry Air, FDA 0.72

Moisture Content -- Method 2

Dry Bulb Temp. 100 °F Wet Bulb Temp. 95 °F Dew Point Temp. 95 °F
Vapor Pressure Of H₂O @ DP 1.66 "Hg Stack Pressure 30 "Hg
Moisture Fraction, H₂O 0.06 Fraction Of Dry Air, FDA 0.94

Specific Gravity

Dry Molecular Weight, $M_d = [0.44(\%CO_2)] + [0.32(\%O_2)] + [0.28(\%N_2 + CO)] =$ _____
Molecular Weight @ Stack Conditions, $M_s = [(M_d) \times (FDA)] + [(18) \times (H_2O)] =$ _____
Specific Gravity (Referred to air), $G_s = (M_s) \div (28.99) =$ _____

FLUORIDE EMISSIONS

19
200
124

IRONMOUNTAIN ENGINEERING, INC.
Gainesville, Florida

wt. of Silica Gel 240.5 gm

SOURCE SAMPLING FIELD DATA SHEET

Plant W.R. GRACE

Sampling Location Wet Phos. Acid, STAT "C"

Date Tues Jan 4 Run No. 1

Time Start 14:18 Time End 15:58

Sampling Time/Point (100 min) (5 at 20 min)

DB 153 °F, WB 149 °F, VP @ DP 7.18 "Hg

Moisture 18.5 %, FDA .815, Gas Density Factor

Barometric Press 29.92 Hg, Stack Press 29.90 Hg

Weather Over Cast

Temp. 80 °F, W/D, W/S

Sample Box No. 4 Meter Box No. 4

Meter ΔH 1.62 Pitot Corr. Factor .83

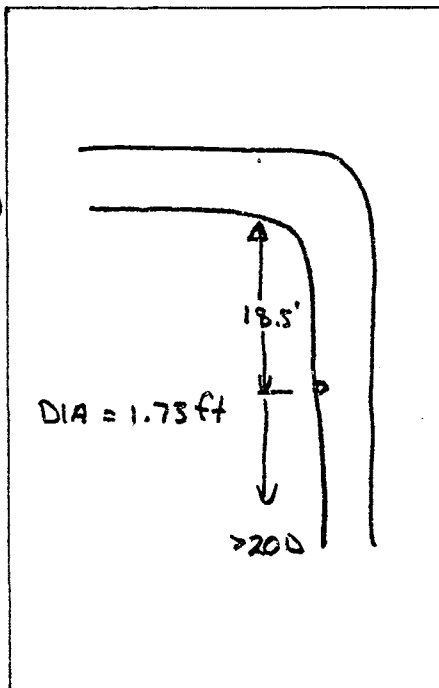
Nozzle Dia. .25 in., Probe Length 4 ft

Probe Heater Setting 35 %

Stack Dimensions: Inside Diameter 20.76 in

Inside Area 2.35 ft²

Height



Mat'l Processing Rate

Final Gas Meter Reading 295.085 ft³

Initial Gas Meter Reading 214.000 ft³

Total Condensate in Impingers 8.5 ml

Moisture in Silica Gel 9.5 gm

Silica Gel Container No. 305 Filter No. 72037

Orsat: CO₂

O₂

CO

N₂

Excess Air

Test Conducted by: A.L. WILSON
BILL DEMERY

Remarks: 1st POINT IS TOO CLOSE TO STACK WALL.

Port and Traverse Point No.	Distance from End of Port (ft.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
1	.08	Too	CLOSE TO WALL									
1		1418	214.0									
2	.25	1423	217.4	0.4	1.05	1.05	152	82	87	—	83	13
2	.25	1428	220.5	0.39	1.0	1.0	152	87	87	—	83	11
2	.25	1433	223.6	0.4	1.05	1.05	152	87	87	—	83	11

[illegible]

SOURCE SAMPLING FIELD DATA SHEET

Plant W.R. GRACE

Sampling Location Wet Phos Acid, STAT "C"

Date July 31 Run No. 2

Time Start 9:00 Time End 11:05

Sampling Time/Point 120 min (6 at 20 min)

DB °F, WB °F, VF @ DP "Hg

Moisture 10.5 %, FDA , Gas Density Factor

Barometric Press "Hg, Stack Press "Hg

Weather Over Cast

Temp. °F, W/D , W/S

Sample Box No. Meter Box No. 4

Meter ΔH 1.62 Pitot Corr. Factor .83

Nozzle Dia. .25 in., Probe Length 4 ft

Probe Heater Setting 35%

Stack Dimensions: Inside Diameter 20.76 in

Inside Area 2.35 ft²

Height ft

Handwritten calculations:
4.2
785
105
3725

40
146
= .3

Mat'l Processing Rate

Final Gas Meter Reading 392.605 ft³

Initial Gas Meter Reading 295.000 ft³

Total Condensate in Impingers 785 ml

Moisture in Silica Gel 18.8 gm

Silica Gel Container No. 305 Filter No. 72041

Orsat: CO₂

O₂

CO

N₂

Excess Air

Test Conducted by: A. L. WILSON

W. P. DEMERY

Remarks:

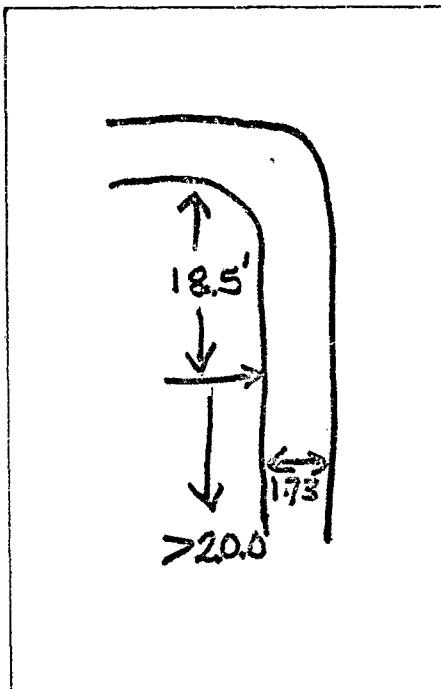
Points No. 1 & 6 were ≈ .005 ft from wall.

Port and Traverse Point No.	Distance from End of Port (in.) (ft.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
		0900	295.0									
1	.08+	0905	298.4	.4	1.18	1.18	150	73	74	—	70	6
1	.08+	0910	301.5	.4	1.18	1.18	150	73	74	—	70	6
1	.08+	0915	304.4	.4	1.18	1.18	150	73	74	—	70	6
1	.08+	0920	307.3	.4	1.18	1.18	150	73	74	—	70	6
2	.25	0925	311.0	.6	1.75	1.75	150	73	74	—	70	9
2	.25	0930	314.7	.65	1.9	1.9	150	73	74	—	70	10
2	.25	0935	318.6	.65	1.9	1.9	150	73	74	—	70	10

[illegible]

SOURCE SAMPLING FIELD DATA SHEET

Plant W.R. GRAE
Sampling Location Wet Phos. Acid, STAT "C"
Date Jan 5 Run No. 3
Time Start 1200 Time End 1405
Sampling Time/Point 120mm (6 at 20min)
DB Assumed 30% °F, WB 30% °F, VF @ DP 30% °F
Moisture 30%, FDA 30%, Gas Density Factor 30%
Barometric Press 29.91 Hg, Stack Press 29.90 Hg
Weather Over Cast
Temp. 85 °F, W/D 85 °F, W/S 85 °F
Sample Box No. 4 Meter Box No. 4
Meter ΔH 1.62 Pitot Corr. Factor .83
Nozzle Dia. .25 in., Probe Length 4 ft
Probe Heater Setting 35%
Stack Dimensions: Inside Diameter 20.76 in
Inside Area 2.35 ft²
Height 20.0 ft



Mat'l Processing Rate 478.901 ft³
Final Gas Meter Reading 478.901 ft³
Initial Gas Meter Reading 392.605 ft³
843 - 200 = 643
Total Condensate in Impingers 643 ml
Moisture in Silica Gel 18.9 gm
Silica Gel Container No 3/0 Filter No. 72044

Orsat: CO₂
O₂
CO
N₂
Excess Air

Test Conducted by: A.L. Wilson
W.P. DEMERY

Remarks: Onset CO₂ reading 721%

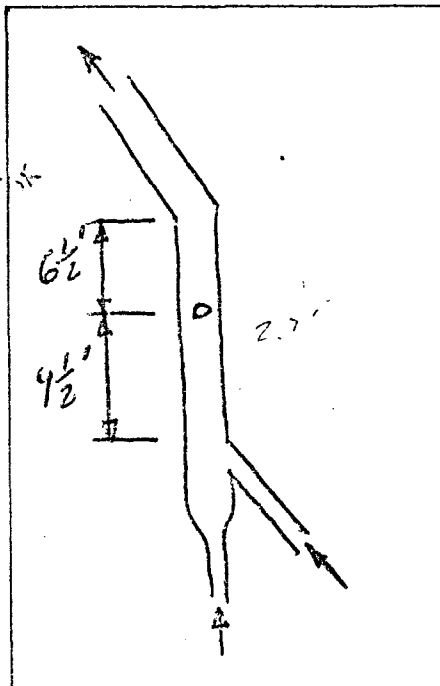
Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
		<u>1200</u>	<u>392.605</u>									
<u>1</u>	<u>.08+</u>	<u>1205</u>	<u>394.9</u>	<u>.3</u>	<u>.58</u>	<u>.58</u>	<u>154</u>	<u>85</u>	<u>85</u>	<u>—</u>	<u>90</u>	<u>4</u>
<u>1</u>	<u>.08+</u>	<u>1210</u>	<u>397.4</u>	<u>.3</u>	<u>.58</u>	<u>.58</u>	<u>154</u>	<u>82</u>	<u>85</u>	<u>—</u>	<u>90</u>	<u>4</u>
<u>1</u>	<u>.08+</u>	<u>1215</u>	<u>399.9</u>	<u>.4</u>	<u>.79</u>	<u>.79</u>	<u>154</u>	<u>81</u>	<u>86</u>	<u>—</u>	<u>95</u>	<u>5</u>
<u>1</u>	<u>.08+</u>	<u>1220</u>	<u>402.3</u>	<u>.3</u>	<u>.58</u>	<u>.58</u>	<u>154</u>	<u>81</u>	<u>84</u>	<u>—</u>	<u>95</u>	<u>4</u>
<u>2</u>	<u>.25</u>	<u>1225</u>	<u>405.0</u>	<u>.45</u>	<u>.89</u>	<u>.89</u>	<u>154</u>	<u>83</u>	<u>84</u>	<u>—</u>	<u>90</u>	
<u>2</u>	<u>.25</u>	<u>1230</u>	<u>407.9</u>	<u>.45</u>	<u>.89</u>	<u>.89</u>	<u>154</u>	<u>84</u>	<u>84</u>	<u>—</u>	<u>90</u>	
<u>2</u>	<u>.25</u>	<u>1235</u>	<u>411.0</u>	<u>.45</u>	<u>.89</u>	<u>.89</u>	<u>154</u>	<u>85</u>	<u>85</u>	<u>—</u>	<u>90</u>	

[illegible]

ENVIRONMENTAL ENGINEERING, INC.
Gainesville, Florida

SOURCE SAMPLING FIELD DATA SHEET

Plant W.R. GRACE
Sampling Location STATION D
Date 1/4/72 Run No. 1
Time Start 14:20 Time End 16:20
Sampling Time/Point 6 min @ 20 pts = 120
DB °F, WB °F, VF @ DP "Hg
Moisture 0.09, F.D. 0.91, Gas Density Factor
Barometric Press 30 "Hg, Stack Press 29.68 "Hg
Weather CLEAR
Temp. 25 °F, W/D , W/S
Sample Box No. Meter Box No. 4
Meter ΔH 1.74 Pitot Corr. Factor 0.83
Nozzle Dia. 0.25 in., Probe Length 6 ft
Probe Heater Setting
Stack Dimensions: Inside Diameter 32.4 in
Inside Area 5.72 ft²
Height Horizontal ft



Mat'l Processing Rate
Final Gas Meter Reading 988.257 ft³
Initial Gas Meter Reading 908.705 ft³
Total Condensate in Impingers 101 ml
Moisture in Silica Gel 249.0-233.7=15.3 (15.3)
Silica Gel Container No. 306 Filter No. 72-038
Orsat: CO₂ 0.2
O₂ 20.5
CO 0
N₂ 79.3
Excess Air
Test Conducted by: J. Dollar
A. Areoyo
R. Wickert

Remarks: Optimum sampling pts. = 45
Only one port available, thus 22 pts.

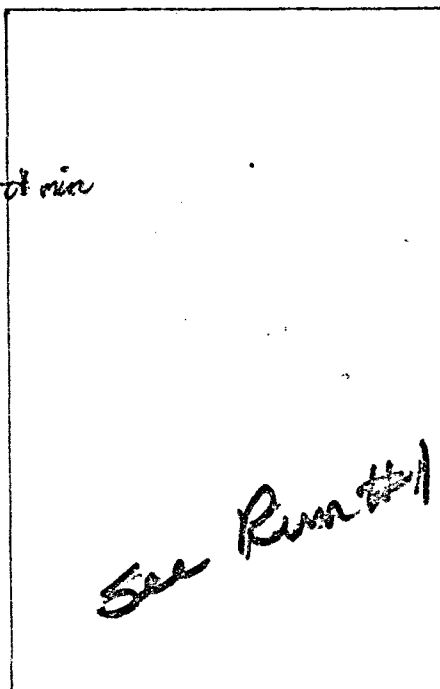
Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Temp. (°F) Stop Watch	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
1	23/64	14:20	908.705	—	—	—	—	—	—	0	—	—
2	1/8	14:26	913.100	.58	1.75	1.75	104	87	87	6	75	.5
3	1 5/16	14:32	916.050	.25	.73	.73	104	85	85	1	80	1.8
4	2 13/16	14:38	918.82	.23	.65	.65	103	86	85	3	83	1.8
5	3 3/4	14:44	921.65	.22	.63	.63	103	87	85	9	86	1.8
6	4 23/32	14:50	924.57	.25	.73	.73	103	8	1	15	83	2.1

[illegible]

Environmental Engineering, Inc.
Gainesville, Florida

SOURCE SAMPLING FIELD DATA SHEET

Plant W. R. GRACE
 Sampling Location STATION "D"
 Date 1/5/72 Run No. 2
 Time Start 0905 Time End 11:05
 Sampling Time/Point 6 min @ 20 ft 120 ft min
 DB — °F, WB — °F, VP @ DP — "Hg
 Moisture 6.7%, FDA 93.3, Gas Density Factor .97
 Barometric Press 30 "Hg, Stack Press 29.68 "Hg
 Weather Overcast
 Temp. 70 °F, W/D —, W/S —
 Sample Box No. — Meter Box No. 1
 Meter ΔH@ 1.71 Pitot Corr. Factor 0.83
 Nozzle Dia 0.25 in., Probe Length 4 ft
 Probe Heater Setting —
 Stack Dimensions: Inside Diameter 32.4 in
 Inside Area 5.72 ft²
 Height — ft



Mat'l Processing Rate —
 Final Gas Meter Reading 1066.254 ft³
 Initial Gas Meter Reading 988.200 ft³
 Total Condensate in Impingers 96 ml (109)
 Moisture in Silica Gel 250.8 - 237.8 = 13 gm
 Silica Gel Container No. 304 Filter No. 72-039
 Orsat: CO₂ .2
 O₂ 20.5
 CO 0
 N₂ 79.3
 Excess Air —
 Test Conducted by: J. Dollar
A. Arroyo
R. Wicker
 Remarks: See Run #1

Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
1			988.200									
2		0905	991.900	.29	.91	.91	102	75	75	13	75	2.9
3		0911	995.000	.34	1.05	1.05	102	75	75	6	75	3.3
4		0917	998.500	.35	1.1	1.1	102	75	75	12	75	3.0
5		0923	1001.9	.29	.91	.91	102	75	75	3	75	3.5
6		0929	1005.1	.30	0.95	0.95	102	75	75	9	75	3.0

[illegible]

ENVIRONMENTAL ENGINEERING, INC.
Gainesville, Florida

SOURCE SAMPLING FIELD DATA SHEET

Plant W. R. GRACE

Sampling Location STATIONED

Date 1/5/72 Run No. 3

Time Start 1200 Time End 1400

Sampling Time/Point 6 min @ 20 ft = 120 tot min

DB — °F, WB — °F, VF @ DP — "Hg

Moisture 6.7 %, FDA 93.3, Gas Density Factor —

Barometric Press 30 "Hg, Stack Press 29.68 "Hg

Weather overcast

Temp. 75 °F, W/D —, W/S —

Sample Box No. — Meter Box No. 1

Meter ΔH@ 1.74 Pitot Corr. Factor 0.83

Nozzle Dia. .25 in., Probe Length 4 ft

Probe Heater Setting —

Stack Dimensions: Inside Diameter 32.4 in

Inside Area 5.72 ft²

Height — ft

See Run #1

Mat'l Processing Rate —

Final Gas Meter Reading 146.515 ft³

Initial Gas Meter Reading 66.254 ft³

Total Condensate in Impingers 108.5 ml

Moisture in Silica Gel 256 - 242.6 = 13.4 gm

Silica Gel Container No. 307 Filter No. 72-042

Orsat: CO₂ 0.4

O₂ 20.4

CO 0

N₂ 79.2

Excess Air —

Test Conducted by: J. Dollar

A. Arroyo

R. Wicker

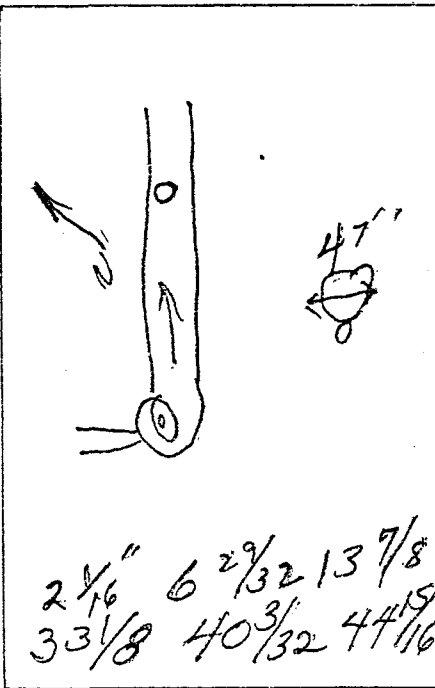
Remarks: See Run #1

Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
1	See Run #1		66.254									
2		12:00	69.8	.35	1.12	1.12	105	82	82	15	87	2.5
3		12:06	74.54	.65	2.0	2.0	105	82	82	6	78	2.8
4		12:12	78.88	.55	1.75	1.75	105	83	82	12	78	5
5		12:18	82.05	.26	.77	.77	105	86	83	3	82	2.2
6		12:24	85.48	.30	.94	.94	105	86	84	9	81	2.8

[illegible]

SOURCE SAMPLING FIELD DATA SHEET

Plant W R. Grace (Wet Acid)
Sampling Location Station (E) (Outlet)
Date 1/4/72 Run No. 1
Time Start 2:15 ^{14:15} Time End 4:15 ^{16:15}
Sampling Time/Point 20 min ^(120 total)
DB 93 °F, WB _____ °F, VF @ DP _____ "Hg
Moisture 12 %, FDA 0.78, Gas Density Factor _____
Barometric Press 30 "Hg, Stack Press 30 "Hg
Weather Clear
Temp. _____ °F, W/D _____, W/S _____
Sample Box No. _____ Meter Box No. _____
Meter ΔH@ _____ Pitot Corr. Factor 0.83
Nozzle Dia. .25 in., Probe Length 6 ft
Probe Heater Setting _____
Stack Dimensions: Inside Diameter 47 in
Inside Area 12.04 ft²
Height _____ ft



Mat'l Processing Rate _____
Final Gas Meter Reading 758.882 ft³
Initial Gas Meter Reading 702.589 ft³
Total Condensate in Impingers 48 ml.
Moisture in Silica Gel (242.5)(254.6) = 12.1 gm
Silica Gel Container No. 303 Filter No. 72-036

Orsat: CO₂ 0
O₂ 20.9
CO 0
N₂ _____

Excess
Air _____

Test Conducted by: G. Allen
E. Johnson

Remarks: Optimum # of Sampling Points
is 12, only 1 point is Available
So 6 pts.

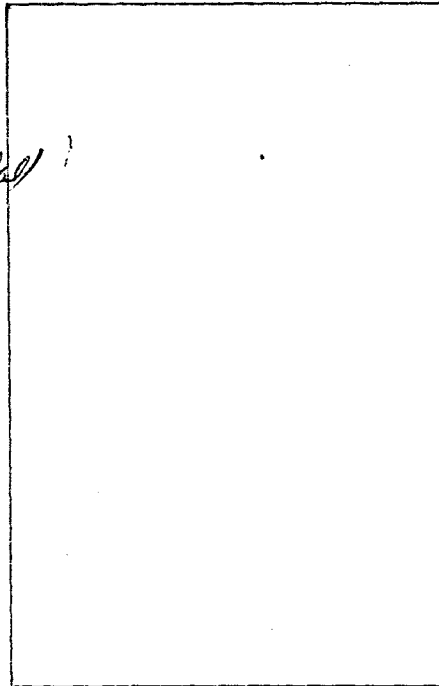
Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
1)	1	2:15	706.7	0.14	0.45	0.45	93	79	87	0		3.0
		2:25	710.9	0.14	0.45	0.45	93	81	84	10		3.6
2	6 ²⁹ / ₃₂	2:35	715.5	0.18	0.58	0.58	92	84	83	5		3.0
		2:45	720.3	0.19	0.61	0.61	92	85	82	15		3.1
3	13 ⁷ / ₈	2:55	725.1	0.185	0.59	0.59	92	85	82	10		3.0
		3:05	729.8	0.185	0.59	0.59	92	86	83	5		3.0
4	33 ¹ / ₈	3:15	734.4	0.18	0.58	0.58	96	86	82	15	85	3.0

[illegible]

ENVIRONMENTAL ENGINEERING, INC.
Gainesville, Florida

SOURCE SAMPLING FIELD DATA SHEET

Plant WR Grace wet phos acid
Sampling Location Station E outlet
Date 1/5/72 Run No. 2
Time Start 9.00 Time End 11.00
Sampling Time/Point 20 min (120 total)
DB °F, WB °F, VP @ DP "Hg
Moisture 5 %, FDA 95, Gas Density Factor
Barometric Press 30 "Hg, Stack Press 30 "Hg
Weather cloudy
Temp. 70 °F, W/D , W/S
Sample Box No. Meter Box No. 5
Meter ΔH₀ 166 Pitot Corr. Factor 0.83
Nozzle Dia. 1/4 in., Probe Length 6 ft
Probe Heater Setting
Stack Dimensions: Inside Diameter 47 in
Inside Area ft²
Height ft



Mat'l Processing Rate
Final Gas Meter Reading 822.485 ft³
Initial Gas Meter Reading 758.892 ft³
Total Condensate in Impingers 58 ml
Moisture in Silica Gel 248.1231.9 [16.2 gm]
Silica Gel Container No. 301 Filter No. 72-040
Orsat: CO₂
O₂
CO
N₂
Excess Air
Test Conducted by: C. Allen
E. Johnson
Remarks:

Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F)	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
			758.892									
1		9.00	763.4	0.15	0.58	0.58	93	72	73	0	-	3.1
		9.10	768.0	0.15	0.58	0.58	93	72	72	10	80	3.0
2		9.20	773.0	0.18	0.68	0.68	93	72	72	5	83	3.1
		9.30	778.0	0.18	0.68	0.68	93	72	72	15	84	3.2
3		9.40	783.4	0.21	0.78	0.78	94	73	72	10	-	3.9
		9.50	788.9	0.22	0.85	0.815	95	74	72	5	-	4.0
4		10.00	794.7	0.24	0.90	0.90	95	74	72	15	75	4.7

[illegible]

ENVIRONMENTAL ENGINEERING, INC.
Gainesville, Florida

SOURCE SAMPLING FIELD DATA SHEET

Plant W. R. Grace Wex phos Acid.
 Sampling Location Station E outlet
 Date 1/5/71 Run No. 3
 Time Start 12.00 Time End 2.00 1400
 Sampling Time/Point 20 min 6 pts 120 tot.
 DB _____ °F, WB _____ °F, VF @ DP _____ "Hg
 Moisture 05 %, FDA 0.95, Gas Density Factor _____
 Barometric Press 30 "Hg, Stack Press 30 "Hg
 Weather Clear Hot
 Temp. 80 °F, W/D _____, W/S _____
 Sample Box No. _____ Meter Box No. 5
 Meter ΔH_3 168 Pitot Corr. Factor 0.83
 Nozzle Dia. 1/4 in., Probe Length 6 ft
 Probe Heater Setting _____
 Stack Dimensions: Inside Diameter 47 in
 Inside Area _____ ft²
 Height _____ ft

Mat'l Processing Rate _____
 Final Gas Meter Reading 886.763 ft³
 Initial Gas Meter Reading 822.485 ft³
 Total Condensate in Impingers 60 ml
 Moisture in Silica Gel 249.6 235.8 13.8 gm
 Silica Gel Container No. 500 Filter No. 72-048
 Orsat: CO₂ 8.0
 O₂ _____
 CO _____
 N₂ _____
 Excess Air _____
 Test Conducted by: C. Allen
E. Johnson
 Remarks: _____

Port and Traverse Point No.	Distance from End of Port (in.)	Clock Time	Gas Meter Reading (ft ³)	Stack Velocity Head ("H ₂ O)	Meter Orifice Press. Diff. ("H ₂ O)		Stack Gas Temp. (°F)	Gas Sample Temp. @ Dry Gas Meter (°F)		Sample Box Temp. (°F) <i>Stop Watch</i>	Last Impinger Test (°F)	Vacuum on Sample Train ("Hg)
					Calc.	Actual		In	Out			
			<u>822.485</u>									
1	1	12.00	827.3	0.17	0.65	0.65	95	78	79	0	-	3.1
		12.10	832.1	0.16	0.605	0.605	96	79	78	10	-	3.0
2		12.20	832.3	0.19	0.715	0.715	96	79	78	5	-	3.6
		12.30	842.6	0.20	0.75	0.75	97	80	78	15	-	3.8
3		12.40	848.2	0.23	0.86	0.86	97	81	79	10	-	4.2
		12.50	854.1	0.23	0.86	0.86	97	82	80	5	-	4.3
4		1.00	859.5	0.23	0.86	0.86	96	83	80	15	82	4.4

[illegible]

GRAB SAMPLE COLLECTION

GRAB SAMPLE DATA SHEET

Plant W.R. GRACE, BARTON, FLA

EPA Sample No.	351PD	351PD	351PD	351PD
Run No.	1	1	1	1-Composite
Date	1-4-72	1-4-72	1-4-71	1-4-72
Time	2:30	3:20	4:00	4:10
Sampling Point	A	A	A	A
Temperature, °F	76	77	77	75
pH	1.8	1.80	1.85	1.95
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W.R. Grace Bartow, Fla

EPA Sample No.	352 PD	352 PD	352 PD	352 PD
Run No.	1	1	1	1-Composite
Date	1-4-72	1-4-71	1-4-71	1-4-71
Time	2:30	3:20	2:00	4:15
Sampling Point	B ₁	B ₁	B ₁	B ₁
Temperature, °F	90	90	90	81
pH	1.85	1.70	1.95	1.95
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W R Grace Barlow Florida

EPA Sample No.	353 PD	353 PD	353 PD	353 PD
Run No.	1	1	1	1-composite
Date	1-4-72	1-4-72	1-4-72	1-4-72
Time	2:30	3:20	4:00	4:20
Sampling Point	B ₁	B ₁	B ₁	B ₁
Temperature, °F	90	88	91	83
pH	1.65	1.65	2.0	1.95
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W R Grace Barlow Fla.

EPA Sample No.	354PD	354PD	354PD	354PD
Run No.	1	1	1	1
Date	1-4-72	1-4-72	1-4-71	
Time	2:30	2:20	4:00	
Sampling Point	Cake			
Temperature, °F				
pH				
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

Plant W R Grace Bartow

Remarks

GRAB SAMPLE DATA SHEET

Plant _____

EPA Sample No.	367PD	367PD	367PD	367PD
Run No.	1	1	1	1
Date	1-4-72	1-4-72	1-4-72	1-4-72
Time	2:40	3:40	4:20	4:35
Sampling Point	G	G	G	G-comp
Temperature, °F	122	138	121	197
pH	1.9	1.95	2.15	2.05
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant WR Grace Bartow Fla.

EPA Sample No.	371PD	371PD	371PD	371PD
Run No.	2	2	2	2-Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:00	10:00	10:50	11:00
Sampling Point	A	A	A	A
Temperature, °F	74	74	74	74
pH	2.1	2.0	2.0	2.0
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W. R. Grace Bartow Fla.

EPA Sample No.	372PD	372PD	373PD	373PD
Run No.	2	2	2	2-101
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:10	10:00	10:50	11:00
Sampling Point	B ₁	B ₁	B ₁	B ₁ -cont.
Temperature, °F	89	90	89	83
pH	2.1	2.0	2.0	2.0
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W R Graca Bartow Fla

EPA Sample No.	373PD	373PD	373PD	373PD
Run No.	2	2	2	2 Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:10	10:00	10:50	11:00
Sampling Point	B ₁	B ₁	B ₁	B ₁ -Comp
Temperature, °F	87	90	90	83
pH	2.1	2.1	2.0	2.0
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W R Grace Bartow Fla

EPA Sample No.	374PD	374PD	374PD	374PD
Run No.	2	2	2	2-Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:25	10:35	10:55	11:10
Sampling Point	F	F	F	F-Comp
Temperature, °F	90	90	86	83
pH	2.1	2.1	2.0	2.0
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks Sample taken from each Hotwell
and mix for sample

GRAB SAMPLE DATA SHEET

Plant (1) R Grace - Bartow Fla

EPA Sample No.	375PD	375PD	375PD	375PD
Run No.	2	2	2	2-Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:25	10:35	10:55	11:15
Sampling Point	G	G	G	G-Comp
Temperature, °F	125	121	110	95
pH	2.1	2.1	2.1	2.1
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant U R Grace Birtow Fla

EPA Sample No.	376PD	376PD	376PD	376PD
Run No.	2	2	2	2
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	9:00	10:00	10:50	10:50
Sampling Point				
Temperature, °F				
pH				
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks Ca K

GRAB SAMPLE DATA SHEET

Plant UIR Graco Barton Fla.

EPA Sample No.	390PD	390PD	390PD	390PD
Run No.	3	3	3	3-Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:15	1:15	1:50	2:00
Sampling Point	A	A	A	A
Temperature, °F	75	75	75	75
pH	1.9	2.0	2.0	1.95
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant WR Grace, Bartow Fla

EPA Sample No.	39/PD	39/PD	39/PD	39/DD
Run No.	3	3	3	3-contin
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:15	1:15	1:50	2:05
Sampling Point	B ₁	B ₁	B ₁	B ₁
Temperature, °F	88	89	87	84
pH	1.9	1.8	1.9	1.9
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant WR Grace Bartow Fla

EPA Sample No.	392PD	392PD	392PD	392PD
Run No.	3	3	3	3-Cont
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:15	1:15	1:50	2:10
Sampling Point	B ₁	B ₁	B ₁	B ₁ -Cont
Temperature, °F	91	93	92	85
pH	1.8	1.7	1.9	1.9
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks _____

GRAB SAMPLE DATA SHEET

Plant W P Grace

EPA Sample No.	395PD	395PD	395PD	395PD
Run No.	3	3	3	3
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:15	1:15	1:50	1:50
Sampling Point				
Temperature, °F				
pH				
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks Cg/Kc

GRAB SAMPLE DATA SHEET

Plant W. R. Grace Barlow Fl

EPA Sample No.	393PD	393PD	393PD	393PD
Run No.	3	3	3	3-Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:20	1:20	1:55	2:20
Sampling Point	F	F	F	F-Comp
Temperature, °F	90	91	91	87
pH	1.9	1.9	1.9	1.9
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks Sample taken from two Hot wells
and combined

GRAB SAMPLE DATA SHEET

Plant W.R. Grace Boston Fl.

EPA Sample No.	394/P1	394/P1	394/P1	394/P1
Run No.	3	3	3	3 - Comp
Date	1-5-72	1-5-72	1-5-72	1-5-72
Time	12:20	1:20	* 2:30	2:35
Sampling Point	G	G	G	G
Temperature, °F	113	118	127	120
pH	1.9	1.9	1.9	1.9
Fluorides				
P ₂ O ₅				
Trace Metals				

Remarks * Had to take one extra sample
one taken at 1:55 (1.9) so 1.

APPENDIX C

Standard Analytical Procedures

ENVIRONMENTAL PROTECTION AGENCY
Research Triangle Park, North Carolina 27711

Reply to
Attn of:

Date: 12-21-72

Subject: Summary of Fluoride Analysis

To: R. Neulicht, EMB, IRL

This memorandum is in response to your request for a brief summary of our SPADNS-Zirconium Lake procedure for determination of fluoride in stack emission samples.

Samples received in our laboratory are filtered through fluoride free paper filters to yield water soluble and water insoluble portions. The water insoluble particulate collected on the filter is rinsed thoroughly to be sure that all water soluble fluoride is rinsed through. The water soluble fraction is distilled from sulfuric acid to a maximum temperature of 180°C. If chloride is suspected in the sample Ag_2SO_4 is added to the still. SPADNS solution is added to an aliquot of the distillate and the absorbance is read at 570 nm. The concentration of the sample is determined from a calibration curve prepared from standard fluoride solutions. It is very important that the temperature of the samples be the same as that of the standards when absorbances are recorded.

The water insoluble fraction of the sample is evaporated to dryness in the presence of a slurry of CAO, and then fused with NaOH. The fusate is dissolved with distilled water, neutralized with dilute H_2SO_4 , distilled and analyzed as described for the soluble portion.

Paper filters containing particulate are cut into small pieces, suspended in a slurry of CAO, evaporated to dryness and ashed prior to the alkali fusion and distillation.

If you have any questions about this procedure, let me know.

Howard L. Crist

Howard L. Crist
Chief, Source Sample Analysis Section
SSFAB, QAEML

cc: R. E. Lee

Phosphorous Pentoxide Determination

Colorimetric Molybdovanadophosphate Method

An aliquot of sample is hydrolyzed in the presence of HCl and HNO_3 acids by boiling almost to dryness.

The sample is cooled to room temperature, transferred to a 250 ml volumetric flask and diluted to volume with distilled water. A 20 ml aliquot is transferred to a 100 ml volumetric flask, 20 ml of molybdovanadate reagent is added and the flask is diluted to volume.

The absorbance of the yellow color is determined after ten minutes at 400 nm. The concentration of phosphorous pentoxide is determined from a calibration curve prepared with standard solutions.

APPENDIX E
Project Participants

PROJECT PARTICIPANTS AND TITLES

John Dollar	Project Manager
Bill Demery	Environmental Specialist
A.L. Wilson	Environmental Specialist
George Allen	Environmental Specialist
Bob Durgan	Environmental Specialist
Eric Johnson	Environmental Specialist
Russ Wicker	Environmental Specialist
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