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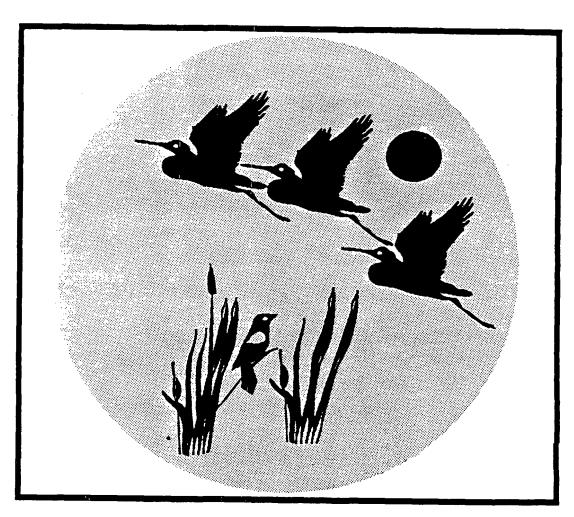
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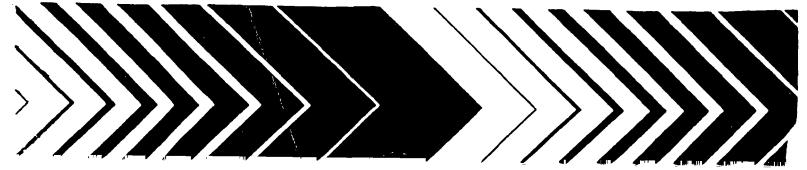
Research and Development



# WATER QUALITY CRITERIA TO PROTECT WILDLIFE RESOURCES

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# Water Quality Criteria to Protect Wildlife Resources

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#### **EXECUTIVE SUMMARY**

The U.S. Environmental Protection Agency (EPA) Environmental Research Laboratory-Corvallis (ERL-C) sponsored a Water Quality Criteria to Protect Wildlife Resources Workshop, co-chaired by the U.S. EPA Office of Water and the United States Fish and Wildlife Service (USFWS). The workshop was convened to identify and define the need for water quality criteria to protect wildlife species. The workshop's goals were to (1) generate a strategy for developing wildlife criteria based on available toxicological data, (2) recommend an approach to incorporating wildlife criteria into the regulatory process, and (3) identify research needs.

Although workshop participants believe that existing aquatic life water quality criteria will in general protect wildlife species, they identified several important exceptions. The recommended procedures are designed to develop a method for identifying chemicals likely to adversely affect wildlife and to provide a mechanism for developing protective criteria.

Workshop participants recommended an approach that includes modifying the existing National Water Quality Criteria Guidelines by incorporating a Final Wildlife Value into the "Guidelines" framework. They also identified procedures for developing the Final Wildlife Value.

The procedures recommended by workshop participants include two phases:

(1) prioritize chemicals based on potential adverse impact on wildlife species and (2) generate a procedure for developing Final Wildlife Values.

The process to prioritize chemicals is a risk assessment based on evaluation of the chemicals' hazard (e.g., toxicity data) and the probability

that wildlife species will be exposed to them (e.g., bioaccumulation, persistence, and discharge rate).

The recommended chemical screening model is an algorithm developed at the workshop and subsequently refined to evaluate the relative hazard of chemicals based on toxicity and tendency for a chemical to bioaccumulate. This algorithm uses basic measures for generating screening-level wildlife criteria designed to establish which of the current criteria may not be protective for wildlife species and to prioritize chemicals for developing wildlife criteria. The algorithm developed was based on the State of Wisconsin's Wild and Domestic Animal Criterion procedure (WDNR 1988). The general equation used in deriving the screening level criteria is:

$$SLWC = \frac{NOEL \times Wta \times SSF}{Wa + (Fa \times BCF)}$$

#### Where:

SLWC = screening-level wildlife criteria (mg/L).

Wa = average daily water consumption (L/day)

Fa = average food consumption (kg/day)

BCF = aquatic life bioconcentration factor (L/kg)

Wta = average weight of the animal (kg)

SSF = species sensitivity factor (0.01 to 1)

NOEL = no observed effect level (mg/kg-day)

The exposure assessment is necessary because wildlife are particularly vulnerable to toxicity resulting from indirect food-chain exposures to contaminants. Factors to be considered in the exposure assessment included persistence (i.e., rate of degradation), production quantities and use

patterns, and probability of release into ambient water bodies used by wildlife species.

Once chemicals are prioritized the next step would be to derive Final Wildlife Values to incorporate into the Ambient Aquatic Life Water Quality Criteria. Since it would be cost prohibitive to conduct toxicity tests such as those used for the aquatic life water quality criteria on all wildlife species and chemicals, workshop participants have recommended that physiologically based toxicokinetic models be used to develop Final Wildlife Values. An important consideration for this process is the incorporation of characteristics unique to wildlife species.

Physiologically based toxicokinetic models (PB-TK) are mathematical simulations of known anatomical and physiological functions used to predict the blood levels of toxic chemicals in organisms subjected to various dosage regimens. The models also attempt to predict the various organ and tissue levels and extra— versus intracellular concentrations (Bischoff 1987). The models would be based on three representative wildlife groups to derive acceptable ambient water and tissue concentrations for Final Wildlife Values. The PB-TK models would require a selective verification program.

Workshop participants recommended that the following five primary factors be considered when evaluating impacts on wildlife species:

- -- <u>Bioaccumulation</u>, especially for chemicals that may not be directly toxic but which may become toxic to wildlife species that are exposed to the chemicals through food webs
- -- <u>Persistence</u> of chemicals in the environment if prolonged exposure increases the potential for bioconcentration and biomagnification in wildlife species

- -- Potential <u>physiological differences</u> in metabolic mechanisms between wildlife species and aquatic life, as well as unique results of toxicity
- -- Unique <u>behavioral characteristics</u> that may increase exposure and result in different relative risks among wildlife species
- -- <u>Life history characteristics</u> that may expose certain wildlife species to higher levels of contamination or result in unique toxicity effects.

Workshop participants identified preliminary research needs that were intended to address the prioritization of chemicals and to refine Final Wildlife Value procedures. The primary purpose of the proposed research is to establish and validate the needed toxic effect and PB-TK models of representative wildlife species (i.e., a mammal, an avian, and a reptile or amphibian selected to meet ecological, toxicological, and experimental considerations) for generating Final Wildlife Values. These studies would also seek to relate wildlife residue concentrations to water/sediment contaminant levels (major omission in current water quality criteria) by predicting contaminant partitioning in the food chain and subsequently applying specific PB-TK models that predict residue uptake, accumulation, distribution, and elimination in the organism. The research needs are consistent with the role identified for the use of models in the development of Final Wildlife Values to be incorporated into the Ambient Water Quality Criteria Guidelines.

#### SECTION 1

#### INTRODUCTION

Wildlife are frequently impaired by exposure to environmental contaminants. Current ambient water quality criteria published by the U.S. EPA Office of Water Regulations and Standards are based on toxicological testing of aquatic organisms. Research conducted in the last 30 years indicates that the series of aquatic life toxicity tests used in the development of water quality criteria does not always account for exposures and mechanisms typical of wildlife species (e.g., DDT causing eggshell thinning). Some chemicals such as DDT, selenium, and PCBs have been evaluated for wildlife effects. However, many chemicals that may pose hazards to wildlife species have not been considered. In this document, the term wildlife denotes birds, mammals, reptiles, and amphibians that use aquatic resources. Aquatic life such as fish, shellfish, and benthic organisms are not included in this category. Unique physiological and behavioral characteristics of wildlife may increase their relative sensitivity to some chemicals, making current criteria inadequate for wildlife protection. Studies conducted by the U.S. Fish and Wildlife Service (USFWS) document adverse effects in natural wildlife populations from surface water contamination in National Wildlife Refuges (USFWS 1986). Toxicological studies by the USFWS and the U.S. Environmental Protection Agency (EPA) have also defined specific adverse effects on wildlife species from exposure to chemical contaminants (Smith 1987). It is imperative to develop wildlife criteria for all chemicals likely to impair wildlife.

The U.S. EPA Environmental Research Laboratory-Corvallis (ERL-C) sponsored the Water Quality Criteria to Protect Wildlife Resources Workshop,

co-chaired by EPA and the USFWS, on November 1-3, 1988. The workshop was conducted to identify and define the need for water quality criteria that are protective of wildlife. The workshop's goals were to (1) generate a strategy for developing wildlife criteria based on available toxicological data, (2) identify research areas to fill data gaps, and (3) recommend ways to incorporate these criteria into the regulatory process. This document summarizes the workshop results.

#### Workshop Organization

Twenty-six professionals from a variety of institutions (e.g., EPA, USFWS, State government, academia, and consultants) participated in the workshop. Participants with expertise in wildlife toxicology, aquatic toxicology, ecology, environmental risk assessment, and conservation were selected to provide a variety of perspectives on the wildlife criteria issue. (see Appendix A for a list of workshop participants). Workshop participants generated three objectives to accomplish workshop goals:

- 1. Propose a strategy to incorporate wildlife criteria into the water quality criteria program.
- Develop a technique for screening chemicals to prioritize criteria development efforts based on the chemical's potential to cause adverse effects for wildlife.
- 3. Develop a research plan for establishing wildlife criteria.

### Document Organization

This document is organized into five topic areas. Section 2 provides background on the legislative authority for wildlife criteria development and

a justification for this effort based on observed wildlife impacts. Section 3 outlines a strategy for incorporating wildlife criteria values into the current water quality criteria framework for regulation of chemicals toxic to wildlife. Section 4 provides a strategy for determining which chemicals are most likely to impact wildlife species; this will help direct research and regulatory efforts. Section 5 provides a recommended research strategy for generating actual wildlife criteria values. Section 6 briefly summarizes the current status of research efforts that will contribute to wildlife criteria development and outlines additional research needs.

#### SECTION 2

# NEED FOR WATER QUALITY CRITERIA TO PROTECT WILDLIFE

Public awareness of adverse effects to wildlife caused by exposure to environmental contaminants has increased. Articles documenting wildlife impairments caused by these contaminants have appeared more frequently in both scientific journals (e.g., Hunter et al. 1984, Ringer 1983, Veith et al. 1979) and popular literature (e.g., McIntyre 1989). Public awareness has also supported new legislation that requires protecting wildlife species from deleterious contamination.

#### Legislative Mandate

Section 101(a)(2) of the Clean Water Act (CWA) requires maintaining water quality for the protection and propagation of fish, shellfish, and wildlife. Section 304(a)(1) requires the Administrator to develop and publish criteria for water quality that accurately reflect the latest scientific knowledge on

the (1) kind and extent of all identifiable effects on health and welfare including, but not limited to wildlife, that may be expected from the presence of pollutants in any body of water; (2) concentration and dispersal of pollutants through biological, physical, and chemical processes; and (3) effects of pollutants on biological diversity, productivity, and stability. Section 304(a)(2)(B) requires the Administrator to develop and publish information on the factors necessary for the protection and propagation of shellfish, fish, and wildlife. In addition to the CWA, there are several other national legislative acts that specifically mandate the protection of wildlife from environmental hazards (Table 1).

To meet the requirements of the CWA to establish and publish water quality criteria, EPA developed the <u>Guidelines for Deriving Numerical National Water Quality Criteria for the Protection of Aquatic Organisms and Their Uses.</u>
EPA uses these guidelines to establish water quality criteria for a variety of compounds (EPA 1986). These criteria were based on toxicity testing of aquatic organisms. Although wildlife may be protected by these criteria, the criteria are designed to protect human health and aquatic life; they do not systematically incorporate unique characteristics of wildlife species.

#### GAO Report

In July, 1987, the Government Accounting Office (GAO) issued a report titled National Refuge Contamination is Difficult to Confirm and Clean Up (GAO 1987) that documented (1) clean-up activities in Kesterson National Wildlife Refuge (a site where selenium contamination has caused deformities in waterfowl), (2) adverse effects of contaminant problems at other refuges, and (3) limited federal efforts to develop water quality criteria to protect

Endangered Species Act

Fish and Wildlife Act of 1956

Migratory Bird Treaty Act of 1918

Bald Eagle Act

Wild and Scenic Rivers Act

Marine Mammal Protection Act

Fish and Wildlife Coordination Act

National Environmental Policy Act

Federal Insecticide, Fungicide and Rodenticide Act

Toxic Substances Control Act

Comprehensive Environmental Response Compensation and Liability Act of 1980

Superfund Amendments and Reauthorization Act of 1986

Land and Water Conservation Fund Act of 1965

wildlife and refuge habitat from adverse effects of contamination. The GAO report concluded, "progress in cleaning up contaminated sites was likely to be slow [because of] a lack of water quality criteria to determine when wildlife and refuge habitat are threatened". In response to the GAO report, EPA modified the water quality criteria for selenium to include wildlife effects.

#### Need for Criteria Specific to Wildlife

As the GAO Report suggested, water quality criteria for wildlife are needed by wildlife managers to determine what levels are safe for wildlife habitat. Wildlife managers are frequently faced with moderately contaminated

aquatic habitat that support wildlife populations. Environmental effects of the contamination may be subtle. Costly, extensive studies may be required to determine whether populations are affected and remediation is necessary. Water quality criteria for wildlife would be extremely useful to determine whether termination of pollutant discharge or cleanup of wildlife habitat is needed. With water quality criteria in their present form, the manager cannot be assured that water quality criteria will provide the protection that is needed.

For many chemicals it is likely that water quality criteria are protective of wildlife species. However, selenium, mercury, DDT, and PCB Ambient Aquatic Life Water Quality Criteria were determined to be insufficiently protective of wildlife. Based on empirical evidence of wildlife impairment, these criteria were subsequently modified to incorporate a wildlife component; the criteria became more stringent as a result. These criteria illustrate the need to incorporate wildlife considerations during development of the criteria before adverse impacts occur. Since wildlife data are not routinely generated for, or incorporated in, ambient aquatic life water quality criteria, existing criteria cannot always provide assurance to managers that wildlife are protected. Only when the guidelines for establishing criteria specifically considered wildlife will users be assured that the criteria are adequate for protecting wildlife and wildlife habitat.

Procedures for deriving criteria presently incorporate a Final Residue Value. The FDA Action Level is used (when available) as the tissue concentration in aquatic life that must not be exceeded due to bioaccumulation of an aquatic contaminant. However, the FDA Action Level has not yet been established for most chemicals. Even where it is set, it is based on typical

consumption patterns for humans. A more realistic scheme would estimate tissue concentration based on wildlife behavior (e.g., feeding, drinking, swimming) and sensitivity to contaminants.

Although aquatic species are immersed in surface waters and are therefore directly exposed, some contaminants may have a greater impact on wildlife species than aquatic species because of (1) contaminant bioaccumulation and (2) differences in species sensitivity to specific toxicants. Both of these factors need to be assessed to adequately protect wildlife. These factors are discussed in greater detail in Section 4.

#### SECTION 3

#### STRATEGY TO INCORPORATE WILDLIFE INTO WATER QUALITY CRITERIA

Two possible approaches for establishing criteria to protect wildlife include: (1) incorporate a Final Wildlife Value into the current framework for developing ambient aquatic life water quality criteria, or (2) develop separate wildlife criteria. Because the first option would require only minor revision to the Guidelines and a regulatory framework is already in place, this strategy was preferred by workshop participants. This decision was based on the assumption that acceptable ambient water concentrations could be derived by incorporating variables unique to wildlife species.

#### Framework for Ambient Aquatic Life Water Quality Criteria

In the current Guidelines (Figure 1), criteria are developed using an array of aquatic organisms in toxicity tests. A Final Acute Value for freshwater aquatic organisms and their uses is derived from acceptable acute test

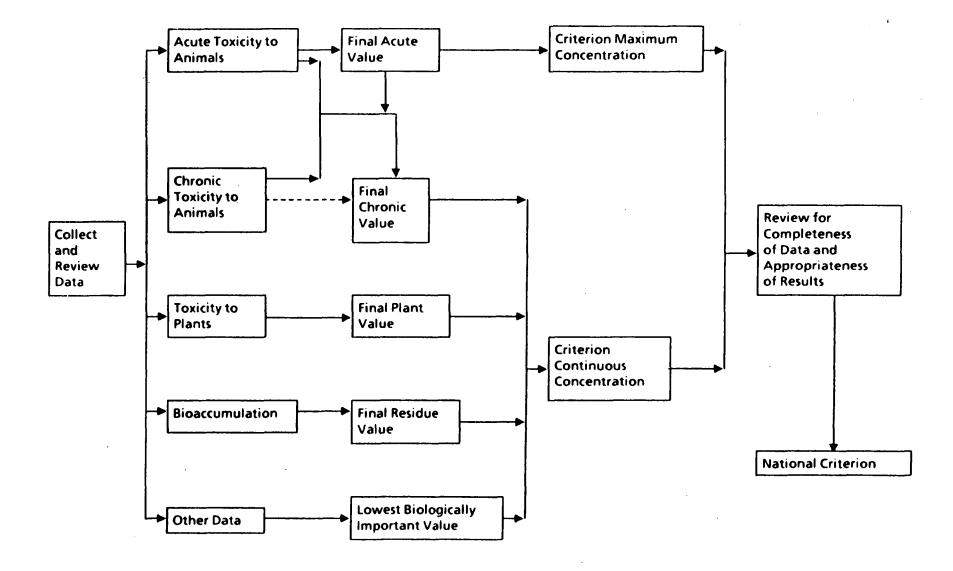


Figure 1. National Water Quality Criteria Guidelines to protect aquatic organisms and their uses.

results. These tests require at least one species of freshwater animal from at least eight different families (Salmonidae, Osteichthyes, a third family in the phylum Chordata, a planktonic crustacean, a benthic crustacean, an insect, a family in a phylum other than Arthropoda or Chordata, and a family in any order of insect or any phylum not already represented). Acute-Chronic Ratios are derived from chronic toxicity tests on three or more species (including one fish, one invertebrate, and one acutely sensitive freshwater species) that are compared to acute tests on the same complement of species. A Final Chronic Value may then be derived.

Criteria are generated from these and other tests (see Figure 1). The Criterion Maximum Concentration (CMC) is derived using the Final Acute Value. The Criterion Continuous Concentration (CCC) is derived from several components: Final Chronic Value, Final Plant Value, Final Residue Value, and Lowest Biologically Important Value. The value with the lowest concentration drives the CCC.

#### Final Wildlife Value

Workshop participants suggested incorporating a Final Wildlife Value into the CCC (see Figure 2). This value would be derived from estimates of chronic effect values for representative species and bioaccumulation. As with other final values, the Final Wildlife Value would drive the CCC, which would be lowered if wildlife were more sensitive to a contaminant than aquatic life. Recalculation may be required if aquatic organisms were shown to accumulate tissue concentrations of the contaminant that would be toxic if consumed by wildlife.

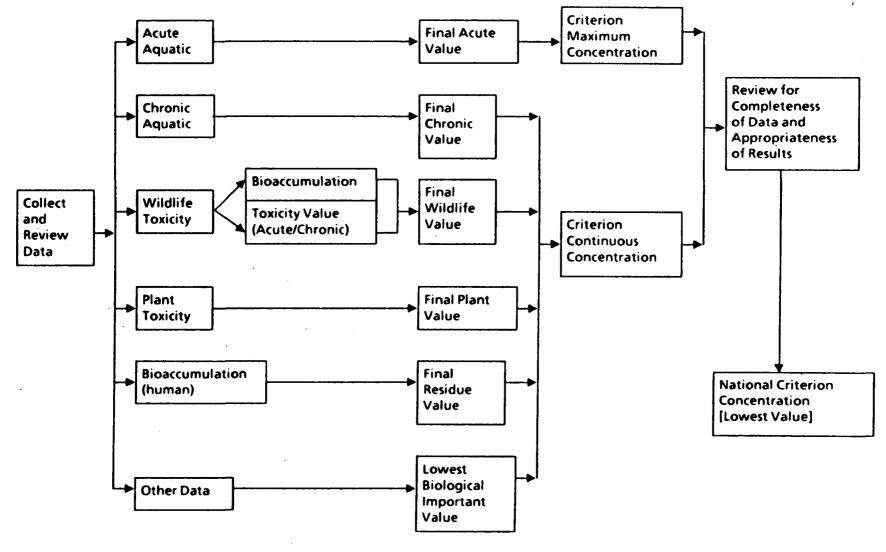


Figure 2. Proposed guidelines to derive numerical water quality criteria to protect all organisms.

Deriving a Final Wildlife Value may be more complex than deriving other final values used in current water quality criteria. Several factors including physiological, behavioral, and life history characteristics of wildlife species must be considered. The persistence and bioaccumulation potential of contaminants are also critical. These factors are further discussed in Section 5.

#### SECTION 4

#### STRATEGY TO IDENTIFY RISK AND PRIORITIZE CHEMICALS

No field studies have been conducted to verify that the water quality criteria are protective of wildlife. Studies are currently under way to investigate the concentrations of selenium needed for wildlife protection, and preliminary, unpublished data from field studies of PCBs in cormorants in the Great Lakes have been interpreted by some to suggest that aquatic criteria do not always protect wildlife (Kubiak, pers. comm.).

A risk assessment paradigm can be used to determine which chemicals are most likely to impair wildlife. To assess the risk a chemical presents to wildlife species, two factors must be considered: (1) evaluation of the hazard and (2) the probability of wildlife exposure to the chemical or its metabolites. Chemicals that could pose a risk to wildlife are those that are highly toxic and/or bioaccumulate. Chemicals that increase the probability of exposure are those that are frequently applied, ubiquitous, and/or persist in the environment and are most likely to come into contact with wildlife.

#### Hazard Assessment

Workshop participants concluded that the relative hazards of a contaminant to wildlife may be assessed and prioritized using three sources of information: (1) toxicity tests on aquatic, domestic and wildlife species: (2) determination of the bioaccumulation potential (bioconcentration factor) of a chemical; and (3) wildlife epidemiological studies. Only limited information for these three categories is currently available. Toxicity tests have been conducted by USFWS and EPA on selected chemicals on a limited number of species. The obvious need for additional toxicity tests on representative species and selected chemicals is discussed in Section 5. A data base of information exists for bioaccumulation potential in aquatic plants and fish for most criteria pollutants. Bioaccumulation potential may also be estimated from the chemical and physical properties of the chemical. Information from categories (1) and (2) above can be combined for use in a chemical screening algorithm. Field evidence suggests reproductive and physiological impairments from exposure to contaminants in natural wildlife populations. However, no rigorous wildlife epidemiological studies exist, and it is difficult to use the existing anecdotal evidence in hazard assessment. Therefore, wildlife epidemiological studies are not discussed in this report.

# <u>Toxicity Tests</u>

EPA's Quality Criteria for Water 1986 (EPA 1986) lists 136 chemicals for which a criterion or other value has been generated based on toxicological tests of aquatic organisms. Human health advisories generated from research on mammalian species are also available. Many of these chemicals are listed as a Lowest Observed Effect Level (LOEL) because insufficient data are

currently available for criteria development (see Appendix B). However, relative toxicity values for these chemicals allow an assessment of potential wildlife impact.

The USFWS prepared a list of 138 chemicals that are potential water quality hazards for wildlife (see Appendix C). USFWS personnel in research and operations added chemicals to the water quality criteria list and rated the combined list based on their experience with toxicity to wildlife, bioaccumulation potential, and extent of environmental exposure. The priorities given on the list are intended to suggest an order for incorporating wildlife information into EPA criteria or advisories.

#### Chemical Screening Algorithm

An algorithm was developed at the workshop and subsequently refined to evaluate the relative hazard of chemicals based on toxicity and tendency for a chemical to bioaccumulate. This algorithm uses basic measures for generating screening-level wildlife criteria designed to (1) establish which of the current criteria may not be protective for wildlife species, and (2) prioritize chemicals for developing wildlife criteria.

The algorithm developed was based on the State of Wisconsin's Wild and Domestic Animal Criterion procedure (WDNR 1988). The general equation used in deriving the screening level criteria is:

#### SLWC = NOEL x Wta x SSF Wa + (Fa x BCF)

#### Where:

SLWC = screening-level wildlife criteria (mg/L)

Wa = average daily water consumption (L/day)

Fa = average food consumption (kg/day)

BCF = aquatic life bioconcentration factor (L/kg)

Wta = average weight of the animal (kg)

SSF = species sensitivity factor (0.01 to 1)

NOEL = no observed effect level (mg/kg-day)

Data from applicable studies in the scientific literature are used to calculate SLWCs. The lowest SLWC of those calculated is used as the criteria. The SSF depends on applicability of the study to wildlife: SSF of 1.0 is used when the SLWC is derived from a wildlife species; smaller SSF values are used when the SLWC is derived from non-wildlife species, reflecting the uncertainty of extrapolation from non-wildlife to wildlife species. During a post-workshop meeting at the Environmental Research Laboratory Duluth (ERL-D), MN, in January 1989, a core committee from the original workshop proposed to obtain BCFs for organic chemicals from a quantitative structure activity relationship (QSAR) model developed at ERL-D (Bradbury et al. 1989). Utilization of this approach will allow rapid and consistent generation of SLWCs.

Chemicals to be used to calibrate and validate the screening algorithm are those known or expected to impair wildlife including DDT, dieldrin, PCBs, mercury, lead, pentachlorophenol, and selenium. It is generally accepted that chemicals of greatest immediate concern for wildlife are probably those with a

Log P > 2.5 (highly bioaccumulating) and those with documented reproductive effects.

ERL-Duluth generated SLWCs using the above algorithm for current criteria, advisory, and USFWS chemicals of concern. This effort, in conjunction with evidence outlined above, provides a basis for prioritizing chemicals that need wildlife criteria development. A chemical priority list will be generated from evidence of wildlife impairment, chemical toxicity ratings available from EPA and USFWS, and the chemical screening algorithm. The priority list is expected to be complete in October 1989 and will serve as the preliminary hazard assessment for wildlife.

#### Exposure Assessment

Contaminant impacts on wildlife may result from characteristics that cannot be accounted for by testing aquatic species. The physiological, behavioral, and life history characteristics of different wildlife groups will alter both their probability of exposure and their sensitivity to a contaminant. Wildlife are particularly vulnerable to toxicity resulting from indirect food-chain exposures to contaminants. Some factors to consider when deriving wildlife criteria are persistence (i.e., rate of degradation), production quantities and use patterns, and likelihood of release into ambient water bodies used by wildlife species. These characteristics largely determine the probability of significant exposure of wildlife populations. To determine the risk associated with a contaminant, its inherent toxicity must be combined with an exposure assessment. If exposure is very high, a moderately toxic chemical can be classified as a high environmental risk. Conversely, if the probability of wildlife exposure is extremely low, even a

highly toxic or bioaccumulating chemical may be classified as a low risk contaminant. Additionally, the environmental fate and transformation of toxic chemicals should also be considered in any risk assessment. This is important for such transformations as mercury to methyl mercury, selenium to selenomethionine, acephate to methamidophos, and others.

Factors and sources of information identified at the workshop that could contribute to an exposure assessment include the following:

- Exposure to chemicals that occur through oral (water, food, and sediment consumption) and dermal exposures -- Critical variables for assessing risk include exposure pathways for wildlife from bioconcentration/bioaccumulation through the food chain.
- 2. Production history Both historical and current production rates and use of a chemical should be evaluated. Production estimates can help determine whether a chemical is likely to be encountered frequently enough or in sufficient quantities to constitute a hazard. Some persistent chemicals like DDT have been banned from use in the United States (DDT is still in use in other countries). DDT also occurs sometimes as an impurity in other compounds. DDT and other persistent chemicals are still widespread in the environment.
- 3. Intended use Categories of intended chemical use will help determine potential routes of exposure. Chemicals with underestimated exposure values because the chemical is listed as an "inert" ingredient in a pesticide formulation (i.e., not the primary active ingredient of a compound) are of special concern.

- 4. Inventories of point source discharges These will aid in determining the types, quantities, and locations of discharged contaminants.
- 5. Sources of nonpoint contamination Nonpoint sources should be considered, particularly for pesticides in agricultural areas. Some naturally occurring contaminants may become a problem when naturally concentrated or released (e.g., metals and sulfur compounds from Mt. St. Helen's eruption). Naturally occurring compounds are also concentrated and released in hazardous quantities as a result of anthropogenic activities (e.g., mining activities and irrigation). For example, naturally occurring selenium leached from soil by irrigation and diversion of irrigation waste water surrounding the Kesterson Reservoir and Wildlife Refuge caused reproductive impairments in local waterfowl.
- 6. Potential contaminant "sinks" that may result in overestimated (and sometimes underestimated) risk of exposure Chemical properties that promote or restrict exchange from one matrix to another (i.e., sediment water column volatilization) will alter exposure potential values.
- 7. Seasonal effects Contaminant concentrations may change with the seasons (e.g., increased contaminant levels could be associated with seasonal snowmelt; low water that exposes contaminated sediment beds during the dry season could result in high exposures).
- 8. Critical life stages -- Seasonal changes in concentration may or may not coincide with critical life stages such as molting or egg laying. Different life stages of an organism can exhibit different

degrees of sensitivity to toxic chemicals; generally, young wildlife are more sensitive than adults.

Information from the monitoring networks of various agencies (e.g., U.S. Geological Survey (USGS), EPA, USFWS) can provide a basis for determining the extent and magnitude of contaminant distribution. Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA III) requires that owners and operators of certain facilities that manufacture, import, process, or otherwise use certain toxic chemicals annually report their releases of those chemicals to each environmental medium. This information aids research and development of regulations, guidelines, and standards. The most recent updated list from the Federal Register (February 16, 1988) is included as Appendix D.

#### SECTION 5

#### DEVELOPMENT OF FINAL WILDLIFE VALUES

It would be cost prohibitive to conduct a number of toxicity tests, such as the series of aquatic life toxicity tests used in the development of water quality criteria, on all wildlife species and chemicals. Workshop participants have recommended using physiologically based (PB-TK) toxicokinetic models to develop most of the data needed for formulating a Final Wildlife Value. However, some toxicity testing will also be required to verify modelderived numbers and to expand the comparative toxicology data base.

#### Wildlife Factors to Consider

There are five primary factors to consider when evaluating impacts on specific species: bioaccumulation, persistence, physiology, behavior, and life history characteristics.

Although some chemicals may not be directly toxic to aquatic life or wildlife, those that bioaccumulate to toxic levels may become a problem for wildlife species exposed to the chemical through food webs. Except for the Final Residue Value (BCFs based on FDA action levels for humans, when they are available) bioaccumulation is not considered under the Guidelines. Bioaccumulation is based on two mechanisms: bioconcentration and biomagnification. Bioconcentration occurs when organisms are directly exposed to a contaminant in the ambient water, and the contaminant is absorbed and stored in body tissue. Biomagnification occurs when predators consume and accumulate contaminants bioconcentrated in the tissues of prey species. Organic compounds with a log octanol: water partition coefficient (Log P) greater than 2.5 are likely to bioaccumulate. Metals of concern include organometallic forms (e.g.,methylated forms) and those that pose special problems because of their specific characteristics of transformation, sequestration, complexation, and speciation.

If a contaminant persists in the environment, wildlife exposure is prolonged and the potential for bioconcentration and biomagnification increases. One measure of environmental <u>persistence</u> is half-life (T1/2), which quantifies the effects of volatilization, photolysis, biodegradation, and other processes that diminish the concentration in aquatic ecosystems and, thus, the hazard to wildlife.

Potential <u>physiological differences</u> in metabolic mechanisms between wildlife species and aquatic life, as well as unique results of toxicity, should be evaluated. For example, DDT when metabolized to DDE can cause eggshell thinning, an important toxic effect in birds. Unique elimination pathways (e.g., milk) may affect reproduction in wildlife species. Wildlife that are directly exposed to a contaminant may not be the ultimate target of the toxic effect. Animals may also pass their contaminant burden on to their offspring, which are likely to be more sensitive than the adult.

Behavioral characteristics unique to wildlife species may increase exposure that results in different relative risk. For example, ducks consume sediments along with aquatic invertebrates and vegetation and may directly ingest concentrated contaminants. Predators more easily obtain poisoned and sick prey species. Manatee, attracted to warm currents, seek out highly contaminated effluent discharges. Wildlife use aquatic resources that may be exempt from existing water quality criteria. For example, the Galveston ship canal has an industrial use designation and is laden with contaminants, yet waterfowl are attracted to the area. Precious metal extraction processors often use large open pools of water containing cyanide in regions where there are few alternative landing sites for waterfowl. These sites attract and kill large numbers of birds each year. Although alterations in behavior caused by exposure to toxic chemicals may not always result in acute toxicity, it may make animals more susceptible to predation or upset normal patterns of foraging, habitat selection, and mate selection.

Some <u>life history characteristics</u> of wildlife expose them to higher levels of contamination or result in unique toxicity effects. For example, wildlife with semi-aquatic habits have several exposure pathways. Piscivorous

birds may be exposed through drinking water and consumption of fish. Newborn beavers may be exposed dermally from contaminated sediments in the lodge and orally by ingestion of contaminated milk. During periods of migration, estivation or hibernation, some wildlife species mobilize fat-soluble contaminants, thus significantly increasing potential exposure to contaminants.

# Procedures for Wildlife Criteria Development

New techniques for developing wildlife criteria are needed because of the high cost and time required to obtain acute toxicity data on each chemical for eight species and chronic toxicity data on each chemical for three species. The current format that relies on FDA action levels and aquatic toxicity is impractical and inadequate. Rather, PB-TK models that predict toxicity effects based on data from representative species are necessary to develop wildlife criteria because few wildlife species are suitable for laboratory culture and testing. Additionally, the cost and complexity of these models are significantly less. Figure 3 illustrates the approach recommended by workshop participants to develop wildlife criteria.

Workshop participants were concerned about relying on the model for simulations it cannot give, i.e., multiple trophic changes, life history, and behavioral interactions with the environment. Exposure pathways unique to wildlife may require an additional form of criteria that will account for contamination from both direct and indirect exposure. Wildlife criteria should be formulated from two values: one for water concentration and one for tissue concentration. Applying criteria to wildlife will be greatly enhanced by using both variables. For each variable, toxicity values for three species would be used, one each for mammals, birds, and amphibian/reptiles. Final

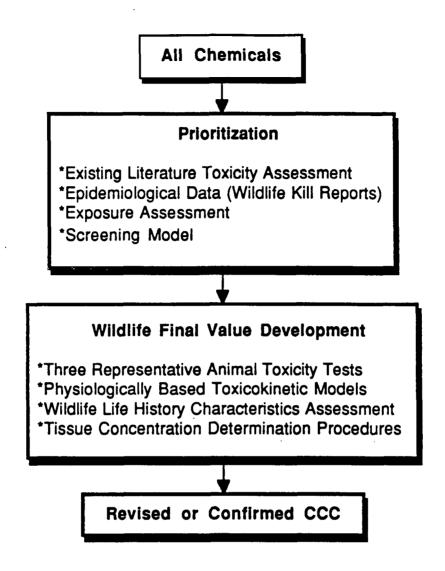


Figure 3. Conceptual framework for developing wildlife criteria.

Criteria Values will be derived from the three PB-TK models that predict toxic effects and steady-state concentrations from concentrations in the tissue of food organisms and the environment of the three groups. The model will extrapolate data generated from the physiologic responses of similar laboratory species. It will ultimately incorporate behavioral data (e.g., food consumption) available from the literature and scientific community in order

to determine input sources of exposure. For example, by basing model interactions on the physiologic basis of chemical effects, a model for rats might be adjusted to more closely resemble the physiological functioning of a mink. Values for contaminant exposure would be modified to more closely resemble the sources and quantities available to mink. The model would then be used to extrapolate between the rat and the mink in terms of the toxic effects and the steady-state concentration.

#### Water and Tissue Concentrations in Wildlife Criteria

Water quality criteria may not provide sufficient protection for wildlife species, because of multiple ambient exposure pathways. To understand this impact, an alternative tissue based criteria is needed. Direct wildlife exposure to aquatic contaminants results from consuming and swimming in or on contaminated water. Where water bodies are subject to regulation by available Ambient Aquatic Life Water Quality Criteria, wildlife species will generally be protected from direct exposure. However, wildlife come in direct contact with waters not designated for aquatic life (e.g., mine tailings ponds) and where few, if any, aquatic organisms may survive. Wildlife exposure from this contact may be significantly elevated above safe levels.

Indirect exposure through bioaccumulation of persistent chemicals is frequently the primary source of contamination to wildlife species. Although related to ambient water concentrations, characterizing the problem may not be possible from water concentration alone, particularly in wetlands. Measures of tissue burdens will promote characterization of contamination and trophic level impacts. Extrapolation from tissue concentrations to ambient water concentrations may be needed to regulate ambient water concentrations of

pollutants. The New York Department of Environmental Conservation extrapolated from tissue concentration, to solution chemistry, to a water quality criteria value by extrapolating back from critical tissue concentrations to estimate the water concentration. A contaminant dose was derived from this estimated water concentration (Newell et al. 1987). Uncertainty exists with this back calculation; this technique should receive further evaluation before being used in developing wildlife tissue criteria.

Establishing a tissue concentration criterion will also help account for other exposure routes (e.g., direct consumption through ingestion of sediments or grooming activity). Developing the residue-based criteria that protect wildlife from waterborne pollutants requires (1) defining the relationship between contaminant concentrations in water/sediment and those in wildlife species, and (2) developing the knowledge base to estimate the probability of harmful effects based on contaminant-residue levels in tissue.

Tissue concentration criteria can be used effectively in an assessment and regulatory framework. Residue-based tissue criteria will be valuable to the USFWS programs for assessing the impact to wildlife on and off USFWS lands. The EPA Wetlands program would benefit from tissue-based criteria because ambient water concentrations are not always measurable in wetlands. For contaminants that are readily metabolized, it is often necessary to look at the chemical breakdown products. Using this approach may also provide a basis by which risk assessments of contaminated and hazardous waste sites may be completed and remedial action implemented. However, serious problems exist in determining the sources of tissue contaminants, especially in global migrant species.

#### Three Representative Animals

Sensitivity to a toxic chemical can vary greatly among different species. Workshop participants identified three wildlife subgroups (mammals, birds, and amphibian/reptiles) for which predictive/extrapolative models should be developed to generate criteria (see Figure 3), and for which validation testing should be conducted. This approach will help account for species differences in physiology and behavior, including diet, food intake, length and intensity of exposure, storage, reabsorption, metabolism, periodic starvation and associated metabolic changes, and differences in hibernation or migration.

The models will account for the major variables that directly impact contaminant exposure, and will predict the kind and amount of anticipated exposure for each representative animal. Limited empirical research will be required to develop and verify model predictions; verification will be done with laboratory and field experiments. Water concentration criteria should be based on these representative animals.

The tissue concentration portion of the model predicts that each representative animal will bioaccumulate total body levels of contaminant based on variables unique to the animal group. Body burdens will result from contact with or consumption of sediments, water, and aquatic organisms. Also included is a grooming factor to account for possible consumption through removing contaminants on skin, feathers, or fur. Water concentration and tissue concentration extrapolations would be combined to derive a Final Wildlife Value.

# Physiologically Based Toxicokinetic Modeling to Generate Criteria

Performing toxicity tests with wild mammals and birds is expensive and often impractical; the workshop recommendation is to develop most wildlife criteria from predictive/extrapolative techniques based on PB-TK models. The models presently developed are mathematical simulations of known anatomical and physiological functions to predict the blood levels of various dosage regimens. The models also attempt to predict the various organ and tissue levels as well as extra- versus intracellular concentrations. PB-TK modeling establishes the uptake, elimination, distribution, and body burden of contaminants within an organism. Broad application of the models is based on the many similarities in the anatomy and physiology of mammalian species. The same blood flow diagram could be used for all mammals, and most organs and tissues are similar fractions of the body weight. Major qualitative differences, such as the absence of a gallbladder in some species, are the exception (Bischoff 1987). As noted above, the PB-TK models will be based on three representative wildlife groups to derive acceptable ambient water and tissue concentrations for Final Wildlife Values. The models would require a selective verification program (see Section 6). The implementation of a model-generated data program with verification is a major commitment that should be identified and discussed in more detail.

#### SECTION 6

#### RESEARCH

Once we prioritize our research effort to reflect the relative risk of wildlife contamination for individual chemicals, developing PB-TK models to

generate wildlife criteria will require understanding the relationship between body and/or tissue contaminant levels for both acute and chronic toxic effects.

#### Current Status

Preliminary efforts to establish wildlife criteria have begun. The wildlife criteria workshop was a major step toward establishing direction in the initiative and determining how to use ongoing research conducted by the USFWS and EPA laboratories. The EPA Environmental Research Laboratories in Athens, GA, and Duluth, MN, have developed and are refining fate and transport models that can be used to predict how contaminants partition in ecosystems. Sediment criteria research at USFWS and EPA laboratories is evaluating the relationship between water/sediment contaminants and food chain species. Research on toxicant accumulation is in progress at ERL-Corvallis. Development of toxic effects models in aquatic organisms and wildlife species are in progress at ERL-Duluth. These techniques and corresponding expertise provide a broad foundation of knowledge that can be applied toward developing wildlife criteria.

ERL-Duluth is prioritizing chemicals with the screening level wildlife criteria algorithm (see Section 4). To help verify this effort, work at ERL-Corvallis was initiated in spring 1989 to test the effects of dieldrin and other selected toxic compounds on mallard ducks and leopard frogs and to obtain acute 96-hr LC50 values, chronic NOEL values, and bioconcentration data from water-only (drinking, paddling, and preening) and feeding exposures.

Data will be compared to information in the criteria documents for fish and invertebrates and to screening level criteria predictions from ERL-Duluth.

The experimental design used in feeding studies will include both commercial and natural foods that contain dieldrin. In addition, the design will examine the impact of oral and dermal exposures by including drinking and paddling water both with and without dieldrin.

#### Research Needs

Contaminant residues in representative wildlife species (e.g., mammal, avian, and reptile or amphibian selected on ecological, toxicological, and experimental considerations) needs to be correlated to acute and chronic toxicity endpoints through the development of toxic effect models. In turn, wildlife residue concentrations will be related to water/sediment contaminant levels based on predicting contaminant partitioning in the food chain by applying specific PB-TK models that predict residue uptake, accumulation, distribution, and elimination in the organism. Models will not be able to predict uptake from sediment without some idea of sediment consumption by wildlife in the field. As far as is known, these values are not available in the literature.

Expanded research efforts will be required to establish and validate the needed toxic effect and PB-TK models for generating wildlife criteria. The comparative toxicology data base for wildlife species should be expanded to document correlations between dose and effect as well as dose and body burden. Future research will deal with selected representative chemicals and with a limited number of animals. Initial efforts will use one bird (such as the mallard) and one mammal (such as the mink). There is also a need for studies to develop methods to better estimate food chain bioaccumulation pathways.

Extensive literature searches will be conducted to obtain toxicity and exposure information on domestic animals (including ducks, chickens, mice, and rats) to compare with the limited available information for wild species.

Extrapolation from existing data to wild species will show us where research needs to be conducted to generate missing information. Residue/toxic effect data will be obtained from existing data bases, tests for specific toxic mechanisms, and ongoing residue monitoring studies. Physiological parameters for developing PB-TK models (e.g., tissue volumes, lipid content, and blood flow rates) will be derived from the literature and through empirical experimentation. The approach will be validated by comparing predicted residue effect levels and accumulation to those empirically determined in the laboratory.

A limited number of acute and chronic tests will be conducted to verify model predictions. There are no plans to conduct tests with each criteria chemical for several species of animals, as has been done in the past with fish and invertebrates. A major consideration in applying criteria, both for aquatic and wildlife species, is the uncertainty associated with using either empirically-based or mechanistic model-derived results under field conditions. Thus, as is being done for laboratory-derived aquatic life criteria, model-derived criteria for wildlife also should be assessed in controlled field studies.

Identifying new test species and developing new test procedures, especially for more diverse groups of waterfowl and amphibians, is needed.

Additionally, adapting existing fish exposure systems for waterfowl and amphibians can be very productive; some studies in this regard have been initiated at ERL-Corvallis. Verification of model predictions with laboratory

testing is critical to ensure model usefulness and acceptance by the scientific and industrial community. For example, chronic laboratory reproductive studies exposing mallards to selenium could simulate conditions at Kesterson Reservoir and provide valuable model-verification data.

#### REFERENCES

- Bischoff, K.B. 1987. Physiologically Based Pharmacokinetic Modeling:

  Pharmacokinetics in Risk Assessment, Drinking Water and Health, Vol. 8.

  National Academy Press. Washington, DC. 1987.
- Bradbury, S., P. Fitzsimmons, and E. Anderson. 1989. Screening study for Wildlife Criteria development. Prepared for U.S. EPA Office of Water, Office of Water Regulations and Standards. ERL-Duluth, Duluth, MN.
- Hunter, M.L., J.W. Witham, and H. Dow. 1984. Effects of a carbaryl-induced depression in invertebrate abundance on the growth and behavior of American black duck and mallard ducklings. Can. J. Zool. 62:452-456.
- Kubiak, T. 1989. U.S. Fish and Wildlife Service, East Lansing, MI.
  Telephone conversation with Dr. Suzanne Marcy. Office of Water Regulations and Standards, USEPA, Washington, DC.
- McIntyre, J. 1989. The common loon calls for help. National Geographic, 175:510-524, April.
- Newell, A.J., D.W. Johnson, and L.K. Allen. 1987. Niagara River biota contamination project: fish flesh criteria for piscivorous wildlife.

  Technical Report 87-3, Division of Fish and Wildlife, Bureau of Environmental Protection, New York State Department of Environmental Conservation.
- Ringer, R. 1983. Toxicology of PCBs in mink and ferrets. Pages 227-290.

  In: F. D'Itri and M. Kamrin, eds. PCBs: Human and Environmental

  Hazards. Butterworth Pub., Woburn, MA.
- Smith, G. J. 1987. Pesticide Use and Toxicology in Relation to Wildlife:
  Organophosphorus and Carbamate Compounds. U.S. Fish and Wildlife
  Service. U.S. Department of Interior Resource Publication 170.

- U.S. Environmental Protection Agency. 1986. Quality Criteria for Water 1986 (Gold Book). EPA 440/5-86-001. Office of Water Regulations and Standards, Washington, DC (with updated Criteria Documents and advisories as they become available).
- U.S. Fish and Wildlife Service. 1986. Preliminary Survey of Contaminant Issues of Concern on National Wildlife Refuges. Division of Refuge Management, Department of Interior. Washington, DC.
- U.S. General Accounting Office. 1987. National refuge contamination is difficult to clean up. Resources, Community, and Economic Development Division. GAO/RCED-87-128. Washington, DC.
- Veith, G., D. Kuehl, E. Leonard, F. Puglisi, and A. Lemke. 1979.

  Polychlorinated biphenyls and other organic chemical residues in fish from major watersheds of the United States, 1976. Pest. Monit. J. 13:1-20.
- Wisconsin Department of Natural Resources. 1988. Technical Support Document for Chapter NR105 of the Wisconsin Administrative Code. Madison.

## APPENDIX A WORKSHOP PARTICIPANTS

#### WORKSHOP LIST OF PARTICIPANTS

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<sup>\*</sup> co-chairperson

# APPENDIX B 1986 U.S. EPA WATER QUALITY CRITERIA SUMMARY

			Concentr	ationa In µgL		Unita Per Liter				# of States With	
	Priority Pollutant	Carcl- nogen	Fresh Acute Criteria	Fresh Chronic Criteria	Marine Acute Critoria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- eumption Only	Drinking Water M.C.L.	Date Refer- ence	Aquatic Life Stan- dard
Acenapthene Acrolein Acrylonitrile	Y Y Y	и и у	1,700.* 68.* 7,660.*	520.* 21.* 2,600.*	970.° 65.*	710.*	320.µg 0.068µg**	780.րg 0.65րg**		1980FR 1980FR 1980FR	1
Aldrin Alkalinity Ammonia	Y N	Y 14 14 14 14 14 14 14 14 14 14 14 14 14	3.0 Critene	20,000. pH and	1.3 Temperature	Dependent	0.074ng** See Document,	0.079ng**		1980FR 1976RB 1985FR	16 <sup>-</sup>
Antimony Arsenic Arsenic (PENT)	Y Y Y	N Y Y	9,000.* 850.*	1.600.* 48*	2.319*	13.*	146.µg 2.2ng**	45,000.µg 17.5ng**	0.05mg	1980FR 1980FR 1985FR	1 21 21
Arsenic (TRI) Asbestos Bacteria	Y Y N	Y Y N	360.	190. Recreation	69.	36. Bheillea	30kf/L.**	See Document.	<1/100ml	1985FR 1980FR 1986FR	21 56
Barium Benzene Benzidine	N Y Y	N Y Y	5,300.* 2,500.*		5,100.*	700.*	1.mg 0.66µg** 0.12ng**	40.g** 0.63ng**	1.0mg	1976RB 1980FR 1980FR	8 1 6
Beryllium BHC Cadmium	Y Y Y	Y N	130.° 100.° 3.9 +	6.3° 1.1+	0.34° 43.	9.3	6.8ng** 10.µg	117.ng**	0.010mg	1980FR 1980FR 1985FR	8
Carbon Tetrachloride Chlordane Chlorinated Benzenes	Y Y Y	Y Y Y	35,200.* 2.4 250.*	0.0043 60°	80,000° 0.09 160°	0.004 129°	0.4µg** 0.46ng**	6.94µg** 0.48ng**	·	1980FR 1980FR 1980FR	1 12 1
Chlorinated Naphthalenea Chlorine Chloroalkyl Ethera	Y N Y	222	1,600.* 19. 238,000.*	11.	7.6° 13.	7.8				1980FR 1985FR 1980FR	1 21 1

g = grams

Y = Yes

mg = milligrams

N = No

.....

µg = micrograms

ng = nanograms

f = fibers

+ = Hardness Dependent Criteria (100 mg L used).

. = Insufficient Data to Develop Criteria.

Value Presented is the L.O.E.L. - Lowest Observed Effect Level.

\*\* = Human Health Criteria for Carcinogene Reported for Three Risk Levels. Value Presented is the 10-6 Risk Level.

\*\*\* = pH Dependent Criteria (7.8 pH used).

RB = Quality Criteria for Water, 1976 (Redbook).

FR - Federal Register

M.C.L. - Maximum Contaminant Level

		Concentrations in pgl.						Unita Per Liter			of States With
	Priority Pollutant	Carci- nogen	Fresh Acute Criteria	Fresh Chronic Criteris	Marine Acute Criteria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- sumption Only	Drinking Water M.C.L.	Date Refer- ence	Aquatic Life Stan- dard
Chlorocthyl Ether (BIS-2) Chloroform Chloroisopropyl Ether (BIS-2)	Y Y Y	Y Y N	28,900.*	1.240.*			0.03μg** 0.19μg** 34.7μg	1.36µg** 15.7µg** 4.36mg		1980FR 1980FR 1980FR	ı
Chloromethyl Ether (BIS) Chlorophenol 2 Chlorophenol 4	и У	Y N N	4,380.*	2.000.*	29,700.*		0 00000378mg**	0.00184µg**		1980FR 1980FR 1980FR	1
Chlorophenoxy Herbicidos (2,4,5,·TP) Chlorophenoxy Herbicides (2,4-D) Chlorpyrifos	7 7 7	2 2 2	0.063	0.041	0.011	0.0056	10.μg 100.μg			1980FR 1976RB 1986FR	7
Chloro-4 Methyl-3 Phenol Chromium (HEX) Chromium (TRI)	и 7	и и и	30.* 16. 1,700.+	11. 210.+	1,100. 10,300.*	<b>5</b> 0.	50.μg 170.mg	3,433.mg	0.05mg 0.05mg	1980FR 1985FR 1985FR	24 24
Color Copper Cyanide	N Y Y	222	Narrative 18. + 22.	Statument 12. + 5.2	See Document 2.19 1.	2.9 1.	200.µg	·		1976RB 1985FR 1985FR	20 23
DDT DDTMetabolite (DDE) DDTMetabolite (TDE)	Y Y Y	Y Y Y	1.1 1,050.* 0.06*	0.001	0.13 14.* 3.6*	0.001	0.024ng**	0.024ng**		1980FR 1980FR 1980FR	16
Demeton DibutyIphthalate Dichlorobenzenes	Y Y Y	222	1,120.*	0.1 763.*	1,970.*	0.1	35.mg 400.µg	184.mg 2.6mg		1976RR 1980FR 1980FR	1
Dichlocorbenzidine Dichloroethane 1,2 Dichlorethylenes	Y Y Y	Y Y Y	118,000.° 11,600.°	20,000.*	113,000.° 224,000.°		0.01µg** 0.94µg** 0.033µg**	0.020μg** 243.μς** 1.8δμς**		1980FR 1980FR 1980FR	1

g = grams

Y = Yes

mg = milligrama

N = No

µg = micrograms

ng = nanograma

f = fibers

+ = Hardness Dependent Criteria (100 mg L used).

• = Insufficient Data to Develop Criteria.

Value Presented in the L.O.E.L. - Lowest Observed Effect Level.

\*\* = Human Health Criteria for Carcinogens Reported for Three Risk Levels. Value Presented is the 10-8 Risk Level.

\*\*\* = pH Dependent Criteria (7.8 pH used).

egister RB = Quality Criteria for Water, 1976 (Redbook).

FR = Federal Register

teria for Water 1976 (Redbook)

M.C.L. - Maximum Contaminant Level

		Concentrations In µgL						Unita Per Liter			of States With
	Priority Pollutant	Carcl- nogen	Fresh Acute Criteria	Fresh Chronic Criteris	Marine Acute Criteria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- sumption Only	Drinking Water M.C.L.	Date Refer- ence	Aquatic Life Stan- dard
Dichlorophenol 2,4 Dichloropropane Dichloropropene	N Y Y	222	2,020.° 23,000.° 6,060.°	366.* 6,700.* 244.*	10,300.° 790.°	3,040.*	3.09mg 87.µg	14.1 mg		1980FR 1980FR 1980FR	1 1
Dieldrin Diethylphthalate Dimethylphenol 2,4	Y Y Y	Y	2.6 2,120.*	0.0019	0.71	.0019	0.071ng** 350.mg	0.076ng** 1.8g		1980FR 1980FR 1980FR	16
Diniethylphthalato Dinitrotoluene 2,4 Dinitrotoluene	Y N Y	7 7 7					313.mg 0.11µg** 70.µg	2.9g 9.1µg** 14.3mg		1980FR 1980FR 1980FR	
Dinitrotoluene Dinitro-o-cresol 2,4 Dioxin (2,3,7,8-TCDO)	N Y Y	Y N Y	330.* 0.01*	230.° 0.00001°	690.*	370.*	13.4µg 0.000013ng**	785.µg 0.000014ng**		1980FR 1980FR 1984FR	1
Diphenylhydrazine Diphenylhydrazine 1,2 DI-2-Ethylhoxyl Phthalate	Y Y Y	222	270.*				42.ng** 16.mg	0.66µg** 80mg		1980FR 1980FR 1980FR	1
Endosulfan Endrin Ethylbenzene	Y Y Y	222	0.22 0.18 32,000.*	0.086 0.0023	0.034 0.037 430.*	0.0067 0.0023	74.µg 1.µg 1.4mg	159µg 3.28mg	0.0002mg	1980FR 1980FR 1980FR	10
Fluoranthene Gassos, Total Dissolved Guthion	Y N	222	3,960.* Narrativa	Biotemeni 0.01	40° See Dorument	16.° 0.01	42.µg	54.µg		1980FR 1976RB 1976RB	3
Haloethers Halomethanes Heptachlor	Y Y Y	Y Y	360.° 11,000.° 0.62	122.* 0.0038	12,000° 0.063	6,400.° 0.0036.	0.19μg** 0.26ng**	15.7μg** 0.29ng**		1980FR 1980FR 1980FR	12
Hexachloroethane Hexachlorobenzene Hexachlorobutadiene	N Y Y	727	980.* 90.*	540.° 9.3°	940.° 32.°		1.9μg 0.72ng** 0.45μg**	8.74µg 0.74ng 60.µg**		1980FR 1980FR 1980FR	2

g = grams

Y - Yes

mg = milligrams

N - No

µg = micrograms

ng = nanograms

f = fibers

- + Hardness Dependent Criteria (100 mg L used).
- = Insufficient Data to Develop Criteria.

Value Presented in the L.O.E.L. - Lowest Observed Effect Level.

M.C.L. - Maximum Contaminant Level

\*\* - Human Health Criteria for Carcinogens Reported for Three Risk Lovels. Value Presented is the 10-6 Risk Level.

\*\*\* - pH Dependent Criteria (7.8 pH used).

RB - Quality Criteria for Water, 1976 (Redbook).

	Concentrations In µgL Units Per Liter							í	ø of States With		
	Priority Pollutant	Carcl- nogen	Fresh Acute Critorin	Fresh Chronic Critoris	Marine Acute Criteria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- sumption Only	Drinking Water M.C.1.	Date Refer- ence	Aquotic Life Stan- dard
Hexachlorocyclohexane (Lindane) Hexachlorocyclohexane-Alpha Hexachlorocyclohexane-Beta	Y Y Y	Y Y Y	2.0	0.06	0.16		9.2ng** 16.3ng**	31.ng** 64.7ng**	0.004mg	1980FR 1980FR 1980FR	12
Hexachlorocyclohexane-Gama Hexachlorocyclohexane-Technical Hexachlorcyclopentadiene	Y Y Y	Y Y N	7.*	5.2°	7.•		18.6ng** 12.3ng** 206.µg	62.5ng** 41.4ng**		1980FR 1980FR 1980FR	3
iron Isophorone Lead	N Y Y	2 2 2	117,000.* 82.+	1,000. 3.2 +	12,900.° 140.	5.6	• 0.3mg 6.2mg 80.µg	520.mg	0.05mg	1976RB 1980FR 1985FH	15
Malathion Manganese Mercury	N N Y	2 2 2	2.4	0.1 0.012	2.1	0.1 0.025	50.µg 144.ng	100.µg 146.ng	0.002mg	1976RB 1976RB 1985FR	7 7 17
Methoxychlor Mirex Monochlorobenzene	N N Y	222		0.03 0.001		0.03 0.001	100.µg 486.µg		0.15mg	1976RB 1976RB 1980FR	12 7
Naphthalene Nickel Nitrates	Y Y N	222	2,300.* 1,400.+	620.* 160+	2,360.° 76	8.3	13.4µg 10.mg	100.µg	10mg	1980FH 1986FR 1976RB	1 10 6
Nitrobenzone Nitrophenols Nitrosomines	Y Y Y	227	27,000.* 230.* 5,850.*	180.*	6,680.° 4,860.° 3,300,000°		19.8mg			1980FR 1980FR 1980FR	1 1
Nitrosodibutylamine N Nitrosodiethylamine N Nitrosodimethylamine N	Y Y Y	Y Y Y					6.4ng** 0.8ng** 1.4ng**	587.ng** 1,240.ng** 16,000.ng**		1980FR 1980FR 1980FR	

g = grams

Y = Yes

mg = milligrams

N = Nu

µg = micrograms

g - micrograms

ng = nenograma

f = fibers

- + = Hardness Dependent Criteria (100 mg L used).
- \* ... Insufficient Data to Develop Criteria.

Value Presented in the L.O.E.L. - Lowest Observed Effect Level.

M.C.L. - Maximum Contaminant Level

\*\* = Human Health Criteria for Carcinogens Reported for Three Risk Levels. Value Presented is the 10-6 Risk Level.

\*\*\* = pH Dependent Criteria (7.8 pH used).

FR = Federal Register RB = Quality Criteria for Water, 1976 (Redbook).

			Concentr	ationa In µgL				Units Por Liter			Fof States With Aquatic Life Stan- dard
	Priority Pollutant	Carci- nogen	Fresh Acute Criteria	Fresh Chronic Criterin	Marine Acute Criteria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- sumption Only	Drinking Water M.C.L.	Date Refer- ence	
Nitrosodiphenylamine N Nitrosopyrrolidine N Oil and Gresse	Y Y N	N A A	Narrative	Statement	-See Dorument		4,900.ng** 16.ng**	16,100.ng** 91,900.ng**		1980FR 1980FR 1976RB	56
Oxygen Dissolved Parathion PCB's	N N Y	N Y	0.065 2.0	984 Coldwarer 0.013 0.014	Criteria Matria	-See Document 0.03	0.079ng**	0.079ng**		1986FR 1986FR 1980FR	56 8 16
Pentachlorinated Ethanes Pentachlorobenzene Pentachlorophenol	и И Ү	7 7 7	7,240.*	1,100.*	390.* 13.	281.° 7.9°	74.µg 101.mg	86.µg		1980FR 1980FR 1986FR	1 2
pH Phenol Phosphorus Elemental	и Ү и	222	10,200.*	6.8-9 2,560.*	<b>6,8</b> 00.*	6.5-8.5 0.1	3.5mg			1976RB 1980FR 1976RB	56 23
Phthalate Esters Polynuclear Aromatic Hydrocarbons Selenium	Y Y Y	2 7 2	940.* 260.	3.* 35.	2,944.* 300.* 410.	3.4° 54.	2.8ng** 10.µg	31.1ng**	0.01 mg	1980FR 1980FR 1980FR	6 1 15
Silver Solida Dissolved and Salinity Solida Suspended and Turbidity	Y 24 24	222	4.1 + Narrative	0.12 Statement	2.3 -See Document		80.µg 260.mg	·	0.05mg	1980FR 1976RB 1976RB	14 56 44
Sulfide-Hydrogen Sulfide Temperature Tetrachlorinated Ethanea	227	222	Species 9,320.*	2. Dependent	Criteria	2. -Bee Derument				1976R8 1976RB 1980FR	56
Tetrachlorobenzene 1,2,4,8 Tetrachloroethane 1,1,2,2 Tetrachloroethanes	Υ Υ Υ	2 4 %	9,320.*	2,400.*	9,020.*		38.μg 0.17μg**	48μg 10.7μg**		1980FR 1980FR 1980FR	1

g = grams

Y = Yes

mg = milligrams

N = No

µg = micrograme

ng = nanograms

f = fibers

- + = Hardness Dependent Criteria (100 mg L used).
- \* = Insufficient Data to Develop Criteria.

Value Presented in the L.O.E.L. - Lowest Observed Effect Level.

\*\* 
Human Health Criteria for Carcinogena Reported for Three Risk Levels. Value Presented is the 10-6 Risk Level.

\*\*\* = pH Dependent Criteria (7.8 pH used).

FR = Federal Register .

RB = Quality Criteria for Water, 1976 (Redbook).

M.C.L. - Maximum Contaminant Level

		Concentrátiona In µgL						Unita Per Liter			# of States With
	Priority Pollutant	Carci- nogen	Fresh Acule Criterin	Fresh Chronic Criteria	Marine Acute Criteria	Marine Chronic Criteria	Water and Fish Ingestion	Fish Con- sumption Only	Drinking Woter M.C.L.	Date Relei- ence	Aquatic Life Stan- dard
Tetrachloroethylene Tetrachlorophenol 2,3,6,6 Thailium	Y Y Y	Y 2 Z	6,280.° 1,400.°	840.* 40.*	10,200.*	460.° 440.°	0.8µg** 13.µg	8.85µg** 48.ng**		1980FR 1980FR 1980FR	1
Toluene Toxaphene Trichlorinated Ethanes	Y Y Y	N Y Y	17,500.* 0.73 18,000.*	0.0002	6,300.* 0.21	5,000.* 0.0002	14.3mg 0.71ng**	424.mg 0.73ng**	0.006mg	1980FR 1986FR 1980FR	1 17
Trichloroethane 1,1,1 Trichloroethane 1,1,2 Trichloroethylene	Y Y Y	N Y Y	46,000.*	9,400.* 21,900.*	31,200.° 2,000.°		18.4mg 0.6µg** 2.7µg**	1.03g 41.8μg°° 80.7μς°°		1980FR 1980FR 1980FR	1 1
Trichlorophenol 2,4,6 Trichlorophenol 2,4,6 Vinyl Chloride	N Y Y	N Y Y		970.*			2,600.µg 1.2µg** 2.µg**	3.6µg** 626.µg**		1980FR 1980FR 1980FR	
Zinc	Υ	N	120.+	110+	96	96			<u> </u>	1987FR	19

g = grams

Y = Yes N - No

nig = milligrams

µg = micrograms

ng = nanograms

f = fibers

+ = Hardness Dependent Criteria (100 mg Lused).

· = Insufficient Data to Develop Criteria.

Value Presented is the L.O.E.L. - Lowest Observed Effect Level.

\*\* = Human Health Criteria for Carcinogena Reported for Three

Risk Levels. Value Presented is the 10-8 Risk Level.

\*\*\* = pH Dependent Criteria (7.8 pH used).

FR = Federal Register

RB = Quality Criteria for Water, 1976 (Redbook).

M.C.L. - Maximum Contaminant Level

## APPENDIX C U.S. FISH AND WILDLIFE SERVICE CHEMICAL PRIORITIZATION LIST

CHEMICAL	RATING	CHEMICAL	RATING
Acenaphthene	L	Chloralkyl ethers	L
Acephate	М	Chloroform	Ĺ
-Acrolein	L	2-Chlorophenol	Ĺ
	•	•	_
Acrylonitrile	L	Chlorostyrenes	M
Alachlor	L	Chlorpyrifos	M
Aldicarb	М	Chlorsulfron	L
Aldrin/Dieldrin	M	Chromium	M
Aluminum	L	Copper	М
Ammonia	L	Creosote	М
Antimony	L	Cyanides	н
Arsenic	H	Cypermethrin	L
Asbestos	L	2,4-D	M
7.5003(0)	J	C, 1 D	4-4
Atrazine	М	2,4-DP	М
Azinphos methyl	L	DDT metabolites	H
Azoxybenzenes	H	Diazinon	М
Azobenzenes	H	Dibenzodioxins	Н
Barium	L	Dibenzofurans	H
Benzeae	M	Dicamba	L
Benzidine	L	Dichlorobenzidine	L
Beryllium	L	Dichloroethylenes	L
Boron	M	2,4-dichlorophenol	L
Borou	141	2,4 dicinorophenoi	• 44
Cadmium	н	Dimethoate	М
Carbaryl	L	2,4-dimethylphenol	L
Carbofuran	М	Dinitrotoluene	·M
	•		
Carbon tetrachloride	L	Diphenylhydrazine	М
Chlordane	Н	Disulfoton	M
Chlordimeform	L	Diuron	L
Chlorinated biphenyls	<b>H</b>	Endosulfan	М
Chlorinated benzenes	H	Endrin	M H
	H	Eptam Eptam	L
Chlorinated biphenylenes	n	Eptam	. L
Chlorinated diphenylethers	Н	Ethylbenzene	М
Chlorinated ethanes	L	Famphur	L
Chlorinated hydrazobenzene	s H	Fenvalerate	М
	••		
Chlorinated naphthalene	H		
Chlorinated phenols	M		
Chlorine	L		

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CHEMICAL	RATING	CHEMICAL	RATING
Fluroanthene	М	Phenol	М
Fluoride	М	Phosphate	L
Fluridone	М	Phthalate esters	Ĺ
Freon 113	L	Picloram	М
Glyphosate	L	Polybrominated biphenyl	
Guthion	М	Polynuclear aromatic	
		hydrocarbons	Н
Haloethers	М	Proban	L
Halomethanes	L	Propanil	L
Heptachlor	Н	Resmethrin	M
Hexachlorobutadiene	Н	Selenium	н .
Hexachlorocyclohexane	H	Silver	M
Hexachlorocyclopentadiene	e M	Sulfometuron methyl	L
Isophorone	М	Temephos	Н
Lead	<b>H</b> .	Terbufos	M
Linuron	L	Tetrachloroethene	L
Malathion	L	Tetrachlorethylene	L
Mercur <del>y</del>	H	Terphenyls	Н
Methomyl	М	Thallium	М
Methoxychlor	L	Thiobencarb	L
Methyl mercury	Н	Toluene	L
Mirex	Н	Toxaphene	М
Naphthalene	L	Trichloroethylene	L
Neburon	L	Trifuralin	М
Nickel	М	Vanadium	L
Nitrobenzene	М	Vinyl chloride	L
Nitrophenols	М	Xylene	L
Nitropyrenes	М	Zinc	M
Nitrosamines	· L		
Organotins	H		
Organolead	М		
Paraquat	M		
Parathion (ethyl and methy			
Pendimethalin	L		
Pentachlorophenol	М		
Permethrin	L.		
pН	L		•

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## APPENDIX D SARA TITLE III SUPERFUND LIST

### ENVIRONMENTAL PROTECTION AGENCY

40 CFR Part 372

[OPTS-400002A; FRL 3298-2]

Toxic Chemical Release Reporting; Community Right-to-know

AGENCY: Environmental Protection Agency (EPA).
ACTION: Final rule.

SUMMARY: This rule contains the uniform toxic chemical release reporting form as required by section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986. Section 313 requires that owners and operators of certain facilities that manufacture, import, process, or otherwise use certain toxic chemicals report annually their releases of those chemicals to each environmental medium. This rule also requires certain suppliers of toxic chemicals to notify recipients of such chemicals in mixtures and trade name products.

DATE: This rule is effective March 17.

FOR FURTHER INFORMATION CONTACT:
Sam K. Susnett, Deputy Project
Coordinator (TS-779), Emergency
Planning and Community Right-to-Know
Hotline, Environmental Protection
Agency, Room, WH 562A, 401 M Street.
SW. (Washington, DC 20460, [800-5350202), In Washington, DC and Alaska, 202-479-2449).

#### SUPPLEMENTARY INFORMATION:

#### I. Authority

The Agency is promulgating this rule pursuant to sections 313 and 328 of Title III of the Superfund Amendments and Reauthorization Act of 1986, Pub. L. 99-499 (42 U.S.C. 11013 and 11028). Title IU is also titled "The Emergency Planning and Community Right-To-Know Act of 1986." Section 313 of Title III requires owners and operators of covered facilities to report annually their releases of listed toxic chemicals. Section 313 also specifies that EPA must publish a uniform toxic chemical release form. In addition, section 328 provides EPA with the authority to promulgate such regulations as may be necessary to carry out the purposes of Title III.

#### II. Background

A. Regulatory History and Summary of Public Participation

On October 17, 1986, the President signed into law the Superfund Amendments and Reauthorization Act of 1986 (SARA), Pub. L. 99–499. The major function of this legislation is to

amend and reauthorize provisions of the Comprehensive Environmental Response. Compensation, and Liability Act of 1980 (CERCLA). However, Title III of SARA is a free-standing statute (not part of CERCLA) that is titled "The Emergency Planning and Community Right-To-Know Act of 1986." In general, Title III contains authorities relating to emergency planning, emergency notification, community right-to-know on chemicals, and a toxic chemical release inventory.

The focus of this rule is the toxic chemical release inventory provision contained in section 313 of Title III of SARA. Section 313 requires owners and operators of certain facilities that manufacture, process, or otherwise use a listed chemical to report annually their. releases of such chemicals to any environmental medium. The reports are to be sent to both EPA and the State in which the facility is located. The basic purpose of this provision is to make available to the public information. about releases of certain toxic chemicals that result from operations of certain facilities in their community.

EPA issued a proposed rule, published in the Federal Register of June 4, 1987 (52) FR 21152). The proposed rule contained the toxic chemical release inventory reporting form and interpretive. requirements for reporting. The preamble of the proposed rule outlined the public participation activities that led up to the development of the proposal. After publication, EPA received over 100 written comments on the proposed rule. In addition, EPA held. public meetings in Washington, DC, Chicago, II, and San Francisco, CA. Attendees at these meetings presented. oral comments representative of wide. range of interests including the affected industry, environmental and other public interest groups. State and local governments, and individual citizens.

In addition, EPA has held other meetings with, and received other communications from, interested parties.

#### B. Overview of Final Rule Requirements

The reporting requirements of this rule apply to owners and operators of covered facilities that manufacture, process, or otherwise use listed toxic chemicals. A covered facility is one that:

Has 10 or more full-time employees. Is in SIC codes 20 through 39.
Exceeds an applicable manufacture, process, or use threshold.

EPA interprets "in SIC codes 20 through 39" to relate to the primary SIC code of the facility. If the facility is comprised of multiple establishments, facility coverage is based on a relative comparison of the value of products.

shipped and/or produced at 20 through 39 establishments versus non-20 through 39 establishments in that facility.

EPA has included a definition of "fulltime employee" and guidance on determining SIC coverage.

EPA has not included a small business exemption in this rule different from that provided by section 313. However, the Agency is allowing reporting in ranges for releases to an environmental medium and for off-site transfers of wastes that are below 1.000 pounds per year. EPA expects that small businesses will benefit most from this provision. The range reporting is for calendar years 1987, 1988, and 1989 only.

The thresholds are those provided by the statute:

For manufacturing or processing as defined—75,000 pounds for 1987, 50,000 pounds for 1988, 25,000 pounds per year for 1989 and thereafter.

For toxic chemicals otherwise used the threshold is 10.000 pounds per year for all years.

Reports must be submitted annually on or before July 1 for the preceding year's data.

The chemicals subject to reporting initially are those chemicals as provided by section 313(c), with certain technical modifications.

Additions or deletions of chemicals from the list may result from petitions or EPA's own review of the list. Any such changes will be by notice and comment rulemaking, and EPA will identify the reporting years which they apply.

Mixtures and trade name products imported, processed, or used at a facility must be evaluated for the presence of listed toxic chemicals. However, EPA has applied a de minimis concentration limitation of 1 percent (or 0.1 percent if the chemical is a carcinogen) consistent with the Occupational Safety and Health Administration (OSHA) Hazard Communications Standard (HCS) in 29 CFR 1910.1200. Toxic chemicals present in concentrations below the de minimis limit do not have to be factored into threshold and release reporting calculations.

In relation to reporting on mixtures, EPA has developed a supplier requirement. Owners or operators of facilities in SIC codes 20 through 39 who supply mixtures or trade name products containing listed toxic chemicals must notify their customers about the presence and concentration of those chemicals in their products. However, the de minimis limit as described above also applies to this requirement. The supplier notification requirement takes effect with the first product shipment in 1969.

the product to be a trade secret. The notice would indicate that the toxic chemical is present in the mixture in a concentration of no more than 15 percent by weight. The upper bound value chosen must be no larger than necessary to adequately protect the trade secret.

(g) A person is not subject to the requirements of this section to the extent the person does not know that the facility or establishment(s) is selling or otherwise distributing a toxic chemical to another person in a mixture or trade name product. However, for purposes of this section, a person has such knowledge if the person receives a notice under this section from a supplier of a mixture or trade name product and the person in turn sells or otherwise

distributes that mixture or trade name product to another person.

(h) If two or more persons, who do not have any common corporate or business interest (including common ownership or control), as described in § 372.38(f), operate separate establishments within a single facility, each such persons shall treat the establishment(s) it operates as a facility for purposes of this section. The determination under paragraph (a) of this section shall be made for those establishments.

### Subpart D—Specific Toxic Chemical Listings

### § 372.65 Chemicals and chemical categories to which this Part applies.

The requirements of this Part apply to the following chemicals and chemicals

categories. This section contains three listings. Paragraph (a) of this section is an alphabetical order listing of those chemicals that have an associated Chemical Abstracts Service (CAS) Registry number. Paragraph (b) of this section contains a CAS number order list of the same chemicals listed in paragraph (a) of this section. Paragraph (c) of this section contains the chemical categories for which reporting is required. These chemical categories are listed in alphabetical order and do not have CAS numbers. Each listing identifies the effective date for reporting under § 372.30.

(a) Alphabetical listing.

	Chemical name	CAS No.	Effective date
Acetaldehyde		75-07-0	01/01/
cetamide		60-35-5	01/01/
celone		67-64-1	01/01/
			01/01/
-Acetylaminofluorene		53-96-3	01/01/
		107-02-8	01/01/
Acrylamide			01/01/
crylic acid	·	79–10-7	01/01/
crylonitrile	10-hexachloro-1,4,4a,5,8,8a-hexahydro-(1.aipha,,4.aipha,,4a	hete 5 sions 8 sions	01/01/
Sa.beta   ]		209-00-2	01/01/
llyl chlonde		107-05-1	01/01/
duminum (fume or dust)		7429-90-5	01/01/
Uuminum oxide		1344-28-1	01/01/
-Aminoanthraquinone		117-79-3	01/01/
-Aminoazobenzone		60-09-3	01/01/
-Ammobiphenyl		92-67-1	01/01/
-Amino-2-methylanthraquinone		. 82-28-0	01/01/
Immonia		7664-41-7	01/01/
Immonium nitrate (solution)		6484-52-2	01/01/
ummonium sulfate (solution)		7783-20-2	01/01/
niline		62-53-3	01/01/
- Anisiding		90-04-0	01/01/
Anisidine		104-94-9	01/01/
			01/01/
inthracene.		120-12-7	01/01/
ntimony		7440-36-0	01/01/
vsenic			01/01/
sbestos (Inable)			01/01/
			01/01/
			01/01/
lenzamide		55-21-0	01/01.
	·		01/01/
lenzidine		92-87-5	01/01/
enzoic trichloride (Benzotrichloride)		98-07-7	01/01/
lenzoyl chlonde		98-88-4	01/01/
enzoyi peroxide	· · · · · · · · · · · · · · · · · · ·	94-36-0	01/01/
enzyl chlonde		100-44-7	01/01/
erylium.			01/01/
iphenyl		92-52-4	01/01/
is(2-chloroethyflether		111-44-4	01/01/
lis(chloromethyl)ether		542-88-1	01/01/
			01/01/
			01/01/
romoform (Tribromomethane)		75-25-2	01/01/
			01/01/
			01/01
	***************************************		01/01
	·		01/01
	· • • • • • • • • • • • • • • • • • • •		01/01
			01/01
lutyl benzyl phihalate		85-68-7	01/01
			01/01
utyraldetryda		123-72-8	01/01

Chemical name	CAS No.	Effective date
C.1. Acid Blue 9, diammonium salt	2650-18-2	01/01/8
I. Acid Blue 9, disodium salt	3844-45-9	01/01/8
C.I. Acid Green 3	4680-78-8 569-64-2	01/01/8 01/01/8
I. Basic Green 4	989-38-8	01/01/8
.1. Direct Black 38	1937-37-7	01/01/8
.i. Orrect Blue 6.	2602-46-2	01/01/8
1. Direct Brown 95	16071-86-6	01/01/8
I. Disperse Yellow 3	2832-40-8	01/01/8
I. Food Red 5	3761-53-3	01/01/8
I. Food Red 15	81-88-9 3118-97-6	01/01/8
I. Solvent Yellow 3	97-56-3	01/01/8
! Solvent Yellow 14	842-07-9	01/01/8
I. Solvent Yellow 34 (Aunmine)	492-80-8	01/01/8
I. Vat Yellow 4	128-66-5	01/01/6
3dmwn	7440-43-9	01/01/8
skrium cyanamide	156-62-7 133-06-2	01/01/8
arbaryl(1-Naphthalenol, methylcarbamate)	63-25-2	01/01/8
arbon disultide	75-15-0	01/01/8
urbon tetrachlonde	56-23-5	01/01/8
rbonyi sulide	463-58-1	01/01/8
ilechol	120-80-9	01/01/8
Noramben (Benzoic acid,3-amino-2,5-dichloro-)	133-90-4 57-74-9	01/01/8 01/01/8
Nordanet 4,7-Methanoindan,1,2,4,5,6,7,8,8-octachioro-2,3,38,4,7,7a-nexanyoro-1	7782-50-5	01/01/8
Nome dioxide	10049-04-4	01/01/8
nioroacetic acid	79-11-8	01/01/8
Chloroacetophenone	532-27-4	01/01/8
Vorobenzene	108-90-7	01/01/8
horobenziate (Benzeneacetic acid, 4-chloro-alpha-(4-chlorophenyl)-alpha-hydroxy-, ethyl ester)horoethane (Ethyl chlonde)	510-15-6	01/01/8
ilorolorm	75-00-3 67-68-3	01/01/8
loromethane (Methyl chlonde)	74-87-3	01/01/6
loromethyl methyl ether	107-30-2	01/01/8
loroprene	126-99-8	- 01/01/8
Norothalonil (1,3-Benzenedicarbonitrile,2,4,5,6-tetrachloro-)	1897-45-6	01/01/8
топит	7440-47-3	01/01/8
obalt	7440-48-4	01/01/8
pper	7440-50-8 120-71-8	01/01/8
esol (mued somers).	1319-77-3	01/01/8
Cresol	108-39-4	01/01/8
Cresol	95-48-7	01/01/8
Cresol	106-44-5 98-82-8	01/01/8 01/01/8
mene hydroperoxide	80-15-9	01/01/8
pterron[Benzeneamine, N-hydroxy-N-nitroso, ammonium salt]	135-20-8	01/01/8
ciohexane	110-82-7	01/01/8
I-D [Acetic acid, (2,4-dichlorophenoxy)-1]	94-75-7	01/01/6
cabromodiphenyl oxide	1163-19-5	01/01/8
allate (Carbamothioic acid, bis(1-methylethyl)-, S-(2,3-dichloro-2-propenyl) ester]		01/01/8
I-Diaminoanisole sulfate		01/01/8
V-Diaminodiphenyl ether		01/01/8
aminotoluene (mixed isomers)	25376-45-8	01/01/8
-Diaminotoluene	1	01/01/8
IZOMEINANE		01/01/6
Penzoluran		01/01/8
-Dibromoethane (Ethylene dibromide)		01/01/8
outyl phihalate	84-74-2	01/01/
chlorobenzene (mixed isomers)		01/01/
-Dichlorobenzene		01/01/4
-Dichlorobenzene		01/01/
· Dichlorobenzidine		01/01/
hlorobromomethane	75-27-4	01/01/
-Dichloroethane (Ethylene dichloride)		01/01/
-Dichlorethylene		01/01/
Chloromethane (Methylene chloride)		01/01/
- Dichlorophenol		01/01/
- Orchigropropylene		01/01/
chloryos (Phosphoric acid, 2,2-dichloroethonyl dimethyl ester)	62-73-7	01/01/
		. 01/01/
epoxybutane		1
cotol (Benzenemethanol,4-chloro-,alpha,-(4-chlorophenyi)-,alpha,-(Inchloromethyi)-,} epoxybutane ethanolamine -(2-ethylpexylphthalate (DEHP)	111-42-2	01/01/

Chemical name	CAS No.	Elfectiv
ritiyi sullate	64-67-5	01/01/
- Ormethoxybenzidine	119-90-4	01/01/
Dimethylani/noazobonzene	60-11-7	01/01/
-Dimethylbenzidine (o-Tolidine)	119-93-7	01/01/
neinyicarbamyi chloride	79-44-7	01/01/
-Cimclify! hydrazine	57-14-7	01/01/
-Dimethylphenol	105-67-9	01/01/
nethyl phthalate	131-11-3	01/01/
nethyl sullate	77-78-1	01/01/
-Dinitro-o-cresol.	534-52-1	01/01/
-Dinitrophenol	51-28-5 121-14-2	01/01/
- Omitrotoluene	606-20-2	01/01/
pociyi phihalale	117-84-0	01/01/
- Dickrane	123-91-1	01/01/
-Diphenylhydrazine (Hydrazobenzene)	122-66-7	01/01/
chorohydra	106-89-8	01/01/
thoryetnanol	110-80-5	01/01/
yf acrylate	140-88-5	01/01/
y:benzene	100-41-4	01/01/
yl chioroformale	541-41-3	01/01/
ylene	. 74-85-1	01/01/
ylene glycol	107-21-1	01/01/
yleneimine(Azridine)	151-56-4	01/01
ylene oxide	75-21-8	01/01/
ylene thioures	96-45-7	01/01
ometwon (Urea, N,N-dimethyl-N'-(3-(Influoromethyl)pt.anyl)-)	2164-17-2	01/01
makidehyde	. 50-00-0	01/01
on 113 (Ethane, 1,1,2-inchioro-1,2,2-inhiooro-) plachior (1,4,5,6,7,8,8-Heptachioro-3a,4,7,7a-tetrahydro-4,7-methano-1H-indene]	76-13-1	01/01
		01/01
rachloroberzene	118-74-1	01/01
achiorocyclopentadiene	77-47-4	01/01
achioroethane	67-72-1	01/01
achioronaphihalane	1335-87-1	01/01
amethylphosphoramide	680-31-9	01/01
trazne	302-01-2	01/01
Kanne sullate.	10034-93-2	01/01
trochlone seid.	7647-01-0	01/01
Vogen cyanide	74-90-8	01/01
rogen (luonde	7664-39-3	01/01
krogumone	123-31-9	01/01
pulyraidelryde	78-84-2	01/01
ropyl alcohol (Only persons who manufacture by the strong acid process are subject, no supplier notifiction.)	67-63-3	01/01
Isoprocylidened-phenol	80-05-7	01/01
1	7439-92-1	01/01
Nane [Cyclohexane, 1,2.3,4,5,6-hexachloro-(1.alphai,2.alphai,3.betai,4.alphai,5.alphai,6.betai)-]	58-89-9	01/01
eic anhydride	108-31-6	01/01
neb (Carbamodithioic acid, 1,2-ethanedlyfbis-, manganese complex)	12427-38-2	01/01
rganese	7439-96-5	01/01
aminė	106-78-1	01/01
CVy	7439-97-6	01/01
hanol	67-56-1	01/01
Phoxychor (Benzene, 1,1 -(2,2,2-inchloroethylidene)bis(4-methoxy-)	72-43-5   109-86-4	01/01
hyl acrylate.	96-37-3	01/01
ry len-butyl ether	1634-04-4	01/01
Methylenebis(2-chloro amiline) (MBOCA)	101-14-4	01/01
Methylenebis(N,N-dimethyl) benzenamine	. 101-61-1	01/01
nylenobisiphenylesocyanate) (MBI)	101-68-8	01/01
nylene bromide	74-95-3	01/01
Metrylenedianiline	101-77-9	01/01
hyl etnyl ketone	78-93-3	01/01
hyl hydrazine	60-34-4	01/01
hyl rodide	, 74-88-4	01/01
hyl isobutyl ketone	108-10-1	01/01
nyi socyanate.	624-83-9	01/01
nyi mathacrytate	80-62-6 90-94-8	01/01 01/01
nier's keione	1313-27-5	01/01
rbdenum trioxide	505-60-2	01/01
nthalene	91-20-3	01/01
3-Naphihytamine	134-32-7	01/01
	91-59-8	01/01
	7440-02-0	01/01
-Naphthylamine		01/01
- Naphth <sub>y</sub> iamme	7697-37-2	01.0
P-Naphthylamine	7697-37-2 139-13-9	01/01
a-Naphthylamine		
P-Naphth; lamine	139-13-9	101/01

Chemical name	CAS No.	Ellective
Nitrogen mustard (2-Chloro-N-(2-chloroethyl)-N-methylethanamine)	51-75-2	01/01/87
Nutrogrycerin	55-63-0	01/01/87
2-Nitrophenol	86-75-5	01/01/87
4-hurophenot	100-02-7	01/01/87
2 Nitropropane	79-46-9	01/01/87
p-Nitrosodiphenylamine	156-10-5	01/01/87
N,N-Dimethylanifine.	121-69-7	01/01/87
N-Nitrosod- A-butylamne	924-16-3	01/01/87
M-Nitrosociethylamine	55-18-5	01/01/87
Al-Nitrosodiphenylamine	62-75-0 86-30-6	01/01/87
M-Nitrosociphenyiamine	621-64-7	01/01/87
N-Nitrosomethylmnylamine	4549-40-0	01/01/87
W-Nitroscmorpholine	59-89-2	01/01/87
W.Nitroso-Methyturea	759-73-9	01/01/87
N-Nitroso-N-methyturea.	684-93-5	01/01/87
	16543-55-8	01/01/87
N-Nitrosoniperidina	100-75-4	01/01/87
Octachioronaphihalene	2234-13-1	01/01/87
Osmium tetroxide	20816-12-0	01/01/67
Parathion (Phosphorothioic acid, O,O-diethyl-O-(4-nitrophenyl) ester)	56-38-2	01/01/87
Pentachlorophenol (PCP)	87-86-5	01/01/87
Peracetic acid	79-21-0	01/01/87
Phenol	108-95-2	01/01/87
p-Phenylenediamine	106-50-3	01/01/67
6 T T 6 T Y 1 V 1	90-43-7	01/01/87
Phosphanc acid	75-44-6 7684-38-2	01/01/87
Phosphorus (yellow or white)	7723-14-0	01/01/87
Phithatic annythide	85-44-0	01/01/87
Perc acid	88-89-1	01/01/67
Polychionnated biphenyls (PCBs)	1236-36-3	01/01/87
Propane sultone	1120-71-4	01/01/87
bela-Propiolacione	57-57-8	01/01/87
Propionalderivde	123-38-6	01/01/87
Proposur (Phenol, 2-(1-methylethoxy)-, methylcarbamate)	114-26-1	01/01/87
Propylene (Propene)	. 115-07-1	£ 01/01/87
Propylene-mine	75-55-8	01/01/87
Propylene oxide	75-56-9	01/01/87
Ynune	110-86-1	01/01/87
Ovnoline	91-22-5	01/01/87
Zunone	106-51-4 82-68-8	01/01/87
Dumtozene (Pentachloronitrobenzene) Saccharm (only persons who manufacture are subject, no supplier notification) [1,2-Benzisothiazol-3(2H)-one,1,1-dioxide]	81-07-2	01/01/87
Salroke	94-59-7	01/01/87
Selenium	7782-49-2	01/01/87
Silver	7440-22-4	01/01/87
Sodium hydroxide (solution)	1310-73-2	01/01/87
Socium sulfate (solution)	7757-82-6	01/01/87
lyrene	100-42-5	01/01/87
hyrene oxide		01/01/87
Sulfure acid	7664-93-9	01/01/87
erephthalic acid	100-21-0	01/01/87
.1.2.2-Tetrachloroethane	79-34-5 127-18-4	01/01/87 01/01/87
etrachiorvinohos (Phosphone acid. 2-chloro-1-(2,4,5-trichlorophenyl)ethenyl dimethyl ester)	961-11-5	01/01/87
hallium		01/01/87
hoace:amide	62-55-5	01/01/87
.4-Thedianiline		01/01/87
hiourea	62-55-6	01/01/87
horrum dioxide	1314-20-1	01/01/87
itanium dioxida		01/01/87
fanium tetrachlonde		01/01/87
oluene		01/01/87
Cluene-2,4-disocyanate		01/01/87
Oluene-2,6-dirsocyanate		01/01/87
- Toluidine hydrochloride		01/01/87
oxaphene		01/01/87
riaziquone (2,5-Cyclohexadiene-1,4-dione.2,3,5-tris(1-aziridinyl)-)		01/01/87
nchlorion (Pnosphonic acid. (2.2.2-irichloro-1-hydroxyethyl)-, dimethyl ester)	52-68-6	01/01/87
2.4-Trichlerobenzone		01/01/87
1.1-Trichloroethane (Methyl chloroform)		01/01/87
1.2-Trichloroethane	79-00-5	01/01/87
nchloracity/ene		01/01/87
4.5-Trichlorophenol		01/01/87
4.6-Trichicrophenol		01/01/87
Influentin (Benzencamine, 2,6-diratro-N.N-dioropy)— (trifluoromethyl)-1)		01/01/87
2,4-Trimethylbenzene		01/01/87
votrano (Etnyl carbamate)		01/01/87
	., 31-13-0	

Chemical name	CAS No.	Effective date
Vanadium (fume or dust)	7440-62-2	01/01/8
Vinyl acetate	108-05-4	01/01/87
Vinyl bromide	593-60-2	01/01/87
Vinyl chlonde	75-01-4	01/01/87
Vinylidene chloride	75-35-4	01/01/87
Xylene (mixed isomers)	1330-20-7	01/01/87
7- Xylene	108-38-3	01/01/87
>-Xylene	95-47-8	01/01/87
p-Xytene	106-42-3	01/01/87
2.6-Xylidine	87-62-7	01/01/87
Zinc (turne or dust)	7440-66-6	01/01/87
Zineb (Carbamodithioic acid, 1,2-ethanedrylbis-, zinc complex)	12122-67-7	01/01/8

#### (b) CAS Number listing.

LS No.	Chemical name
50-00-0	Formaldehyde
51-28-5	2.4-Dintrophenol
51-75-2	
51-79-6	Urethane (Ethyl carpamate)
52-68-8	
53-96-3	2-Acetylaminofluorene
55-18-5	A-Nitrosodiethylamine.
55-21-0	
55-63-0	
56-23-5	Carbon tetrachlonde
56-38-2	
57-14-7	
57-57-8	beta-Propolacione
57-74-9	
58-89-9	
59-89-2	Microsomorpholine
60-09-3	4-Aminoazobenzene
60-11-7	
60-34-4	Methyl hydrazine
60-35-5	
	Acetamos
85-22-3	Antine
62-55-5	Thicacetamide
62-56-6	Thourse
62-73-7	Dichlorvos (Phosphoric acid, 2,2-dichloroethenyl dimethyl ester)
62-75-9	N-Ni/rosodimethylamine
63-25-2	Carbaryl (1-Naphthalenol, methylicarbamate)
64-67-5	Diethyl sulfate
67-56-1	Methanol
67-63-0	Isopropyl alcohol (only persons who manufacture by the strong acid process are subject, supplier notification not required.)
67-64-1	Acetone
67-66-3	Chloroform
67-72-1	Hexachioroethane
58-76-8	Triaziquone [2,5-Cyclohexadiene-1,4-dione,2,3,5-tris(1-azindinyl)-]
71-36-3	n-Butyl alcohol
71-43-2	Benzene
71-55-6	1,1,1-Trichloroethane (Methyl chloroform)
72-43-5	Methoxycnlor [Benzene, 1,1'-(2,2,2,-trichloroethylidene)bis [4-methoxy-]
74-83-9	Bromomethane (Methyl bromide)
74-85-1	Ethylene
74-87-3	Chloromethane (Methyl chloride)
74-88-4	Methyl iodide
74-90-8	Hydrogen cyanide
74-95-3	Methylene bromide
75-00-3	Chloroethane (Ethyl chloride)
75-01-4	Virtyl chlonde
75-05-8	Acetonitrie
75-07-0	Acetaldehyde
75-09-2	Dichloromethane (Methylene chloride).
	Carbon disulfide.
75-21-8	
75-25-2	
75-27-4	Dichlorobromomethane
75-35-4	Virylidene chlonde
75-44-5	Phospane
75-55-8	Propylene mine
75-56-9	Propylene Oxide
75-65-0	181-Budy alcohol
77-13-1 76-44-8	Heptachlor (1.4.5,6,7,8,8-Heptachloro-3a,4,7,7a-tetrahydro-4,7-methano-1H-indene)

S No. Chemical name			
No.	Chemical name		
-78 .1	Ornettryl suitate		
-84-2	i lennuturaldenide		
	12-Dichloropropane.		
-92-2	i sec-Butyl alcohol		
-93-3	Methyl ethyl ketone		
-oc-s	1.1,2-Trichioroethane		
	Trichloroethylene		
-06-1	1 - 7 - 7 - 7 - 7 - 7 - 7 - 7 - 7 - 7 -		
-10-7			
-11-8	Chloroacetic acid		
-21-0	1,1,2,2-Tetrachloroethene		
-44-7	1. 1. 2. 1 ev admitrouvers		
46-9	Demathylcarbamyl chlonde		
-05-7	4.4 - scorcovirienedichend		
15-9	4.4'-Isopropylidenediphenol		
-62-6	Methyl methacyviate		
07-2	Secchann (only persons who manufacture are subject, no supplier notification) [1,2-Benzisothiazol-3(2H)-one,1,1-dioxide]		
e8-9	C.I. Food Red 15.		
28-0	1-Amino-2-methylanthragunone		
	Ountozene (Pentachlorontrobenzene)		
66-2	Detryl phthalate		
4-2	Diethyl phthalate Dibutyl phthalate Phinairc anhydnde		
	PRODUCTOR STRANGE CONTRACTOR CONT		
8-7 10-6	Outy vertay profises		
0-0 2-7	A-Nitrosodiphenylamine 2.6-Xylidine		
8-3	Merachines, 1 Subdeciene		
٠.	Pantachioranheani (PCP)		
6-2	2.4.6-Trichlorophenol 2-Nitrophenol		
5-5	2-Nitrophenol		
<b>3-1</b> !	Piono acid		
-0	o-Anisdine		
.7	2-Phenylphenol		
-8 -7	Michier's ketone		
	Attaches to the state of the st		
2-5	Naphhalene		
2-5	Nephralene  Ounoline  bete-Naphthylamine  3,3 -Dichlorobenzidine  Biphenyl		
1-1	Gete-Reprintations		
٦	Biphenyl.		
-1	4-Amnobiohenvi		
.5	Benzidine		
3	4-Nitrobiphenyl.		
٥	Banzoyl peronde		
-7	Salrole		
7	2.4-D (Acetic acid, (2.4-dichiorophenoxy)-)		
6	o-Xylene		
-1	1,2-Ci-chlorobenzene		
4	o-Toludine		
-6	1.2.4-Trimethylbenzene		
-7	2.4-Diaminololuene		
-4	2.4,5-Tnchlorophenol		
3	Styrene oxide		
-8	1,2-Dibromo-3-chloropropane (DBCP)		
-3	Methyl acrylate		
-7	Ethylene thoures		
-3 -7	C.I. Solvent Yellow 3  Benzoic Inchlonde (Benzotrichlonde)		
-/ -8	Cumene		
3	Benzal chloride		
إ 🚣	Benzoyl chloride		
í	Nitrobenzene		
2	5-Nitro-o-ansidine		
7-	5-Nitro-o-ansidine. 4-Nitrophenol		
ן כ	Terephthalic acid		
4	Ethylbenzane		
5	Styrene		
7	Benzyl chlonde		
4	V-Nitrosoppendine		
4	4.4 - Methylenebis(2-chloroanine) (MBOCA)		
-1 -8	4.4 - Methylenebis(N.N-dimethyl)benzenamne		
	Mathylenebis(phenylisocyanate) (MBI)		
- 1	4.4'-Diaminodiphenyl ether		
: د_	Bis(2-ethylhexyl) adipate		
1			
9	p-Anisidine.		
1 : 3 !			

5 No.	Chemical name
06-46-7	1,4-Dichlorobenzene
06-50-3	p-Phenylenediamine
06-51-4	Ounone
06-88-7	1.2-Butylene oxide
06-89-8	Epichlorohydrin
06-93-4	1.2-Oibromoethane (Ethylene dibromide)
06-99-0	1,3-Butadiene
7-02-8	Acrolen
07-05-1	Allyl chlonde
07-06-2	1.2-Dichloroethane (Ethylene dichloride)
07-13-1 07-21-1	Acrytontnie
07-30-2	Chloromethyl methyl ether.
05-4	Vinyl acetale
8-10-1	Mednyl isobutyl kelone
8-31-6	Maleic anhydride
8-38-3	m-Xylene
X6-39-4	m-Cresol
8-60-1	Bis(2-chloro-1-methylethyl)ether
8-78-1	Melamene
8883	Tollene
8-90-7	Chlorobenzene
18-95-2 19-86-4	Phenol 2-Methoxyethanol
0-80-5	2-Behoxyethanol 2-Ehoxyethanol 2-Eho
0-82-7	Cyclohexane
0-86-1	Printing
1-42-2	Diethanolamine
1-44-4	Bis/2-chloroethy/) ether
4-26-1	Propoxur (Phenol, 2-(1-methylethoxy)-, methylcarbamate)
5-07-1	Propylene (Propene)
5-32-2	Dicotol [Benzenemethanol, 4-chlorosipha-(4-chlorophenyl)alpha-(trichloromethyl)-].
7-79-3	2-Aminoant/vaquinone
7-81-7 7-84-0	Di(2-ethylhexyl) phthalate (DEHP)
8-74-1	Principle Princi
9-90-4	3,3'-Dimethoxybenzidine
9-93-7	3.T-Dimethylbenzidine (a-Tolidine)
0-12-7	Antivacene
0-71-8	p-Creatine
0-80-9	Catachol
0-82-1	Catechol
0-63-2	2.4-Dichlorophenol
1-14-2 1 <del>-69-</del> 7	Z.4-Ointrolouene
2 <del>-66-</del> 7	1.2-Dipherrythydrazine (Hydrazoberizene)
2-31-9	hydrogenone
3-38-6	Proponaldehyde
3-72-8	Bulyraldenyde
3-91-1	
6-72-7	
6-99-8	Chloroprene
7-18-4	Tetrachloroethylene (Perchloroethylene)
8-66-5 1-11-3	
2-64-9	Diberzofuran
3-06-2	
3-90-4	Chloramben (Benzoic acid, 3-amino-2,5-dichloro-)
4-29-2	o-Anisidine hydrochloride
4-32-7	alphe-Naphthylamine
5-20-6	Cupterron [Benzeneamine, N-hydroxy-N-nitroso, ammonium salt]
9-13-9	Nitrilotrisceuc acid
9-65-1	4.4'-Thiodaniine
0-88-5 1-32-2	
1-32-2 1-56-4	
6-10-5	Ethyteneimine (Aziridine)
6-62-7	Calcium Cyanamide
2-01-2	Hydrazine
9-00-2	Aldrin[1,4:5,8-Dimethanonaphthalene,1,2,3,4,10,10-hexachloro-1,4,4a,5,8,8a-hexahydro-(1.alpha.,4.alpha.,4a.beta.,5.alpha.,8.alpha.,8a.beta.)-]
4-88-3	Diammethane
3-58-1	Carbonyl sulfide. C.L Solvent Yellow 34 (Aurimine)
2-80-8	CL Solvent Yellow 34 (Aurimine)
5-60-2	
0-15-6	Chlorobenzilate[Benezeneacetic acid, 4-chloro-alpha-(4-chlorophenyt)-alpha-hydroxy-, ethyl ester]
2-27-4 4-52-1	
0-59-0 1-41-3	Ethyl chloroformate

S No.	Chemical name
542-75-6	1,3-Dichloropropytene
42-88-1	Bis(chloromethyl) ether
69-64-2	C.I. Basic Green 4
306-20-2 315-05-4	2.6-Dentrotoluene 2.4-Dentrotoluene 2.4-Dentroto
21-64-7	A-Nitrotod-n-propylamine
24-83-9	Methyl 150:ryanal9
36-21-5	o-Tolurane hydrochloride
80-31-9	Hexamethylphosphoramide
84-93-5	M-Nitroso-M-methyturea
59-73-9 42-07-9	M-Nitroso-N-ethytures C.I. Solvent Yellow 14
24-16-3	
61-11-5	A-Nitrosod-a-bunytamine
89-38-8	C.I. Basic Red I
20-71-4	Propane sultone
63-19-5	
10-73-2	Socium hydroxide (solution)
14-20-1	MolybJenum thoxide
19-77-3	Throium dioride
10-20-7	Xylene (mused somers)
32-21-4	Ashestos (mable)
35-87-1	Hexachloronaphthalene Polychlorinated bicheriyts (PCBs)
36-36-3   44-28-1	Aluminian goods
54-53-5	Departure
32-09-8	Trifluration (Benzeneamme, 2.6-dinitro-N.N-dipropyl-4-(trifluoromethyl)-)
4-04-4	Trifluratin (Benzeneamine, 2.6-dinitro-N.N-dipropyl-4-(trifluoromethyl)-)  Methyl ter: butyl ether
6-75-5	Netroton (Ranzene 2 Aufrichenn 1./Augrophenow).
7-45-6 7-37-7	Chlorothalonal [1-3-Benzenedicarbonitrile.2.4,5,6-tetrachloro-]
4-17-2	C.I. Direct Black 38
4-13-1	OCIZERO CONTROLLO CONTROLL
3-16-4	Diallate [Carbamothioic acid, bis(1-methylethyl)-, S-(2,3-dichloro-2-propenyl)ester]
2-46-2	C.I. Drect Blue 6
0-18-2	C1. Acd Blue 9, diammonum salt
2-40 <b>-8</b>   8-97 <b>-</b> 6	C.I. Solvent Crange 7
1-53-3	C.I. Food Red 5
اوحيه	C1 And Phis Q dentition eat
9-40-0	M-Nitrosomethylvinylamine C.L Acid Green 3
	C.I. Aod Green 3
4-52-2 9-90-5	Ammonium (itirate (solution)
-92-1	Lead
96-5	Mangariese
9-97-6	Mercury
3-02-0	Nickel
0-22-4 ! 0-28-0 !	Silver
-36-0	Antimony
-38-2	Arsonic
-39-3	Banum
-41-7	Beryllium
)-43-9   )-47-3	Cadmium
48-4	Coball
-50-8	Copper
-62-2	Vanadium (fume or dust)
)-66-6 )-45-0	Zinc (tume or dust)
-01-0	Titanium tetrachlonde
-38-2	Phosphone acid
-39-3	Hydrogen fluonde
-11-7	Ammonia
93-9	Sulfunc acid
37-2	Nitric acid
82-6	Phosphorus (yellow or white)
49-2	Selenum
-50-5	Chlorine
3-20-2	Ammonum sulfate (solution)
1-35-2 4-93-2	Toraphene
9-04-4	Hydrazine suifate
2-67-7	Zinab (Carbamodithioic acid, 1,2-ethancullybis-, zinc complex)
5-01-1 1	
7-38-2	Maneb [Carbamodithioic acid, 1.2-ethan-ediyibis-, manganese complex]
	Maneb [Carbamodithioic acid, 1,2-ethanedlyibis-, manganese complex]  Titanium dioxide

CAS No.	Chemical name	Effective date
25321-22-6 25376-45-8	Osmum tetraixide Dichlorobenzene (mixed isomers)	. 01/01/87 01/01/87 01/01/87 01/01/87

## (c) Chemical categories in alphabetical order.

Category name	Effective date.
ntimony Compounds: Includes any unique chemical substance that contains antimony as part of that chemical's infrastructure	01/01/87 01/01/87 01/01/87 01/01/87 01/01/87
romande de la companya de la compan Reconstruir de la companya de la co	
CI	
grand the TT following a statute of the common terms of the common terms of the state of the common terms	
`H(5-x)	even
and the second of the second o	
the common of	1 .
Where x = 1 to 5  romsum Compounds: Includes any unique chemical substance that contains chromium as part of that chemical's infrastructure  box Compounds: Includes any unique chemical substance that contains coball as old of that chemical's infrastructure	01/01/8
pper Compounds: Includes any unique chemical substance that contains copper as part of that chemical's infrastructure anide Compounds: X* CN* where X.= H* or any other group where a formal dissociation can be made. For example KCN, or Ca(CN).	01/01/8
ycol Ethers: includes mono- and di- ethers of ethylene glycol, diethylene glycol, and thethylene glycor	- 01/01/8
Approximate the Control of the Cont	
n=1, 2, or 3  R = alkyl or anyl groups.  R = R H, or groups which, when removed, yield glycol ethers with the structure:	
R = (OCH):	1
Aymens are excluded from this category.	
ad Compounds; Includes any unique chemical substance that contains lead as part of that chemical's infrastructure	01/01/8
Inganese Compounds: Includes any unique chemical substance that contains manganese as part of that chemical's infrastructure	_ 01/01/8

, <del></del>		
	Calegory name	Effective date.
Polytrominated Biphenyls (PBBs).		. 01/01/87
	B	<b>[X</b>
	H	IO-x)
a karantaran dari dari dari dari dari dari dari dari	กลางการทางการการการการที่ผู้เครื่องได้ อย่างการการก็เ ชาวสาทสามารถสอดของชาติสารสมารถก็วสามารถ เปลี่ยวการการการการการการการครับ	
Thallium Compounds: Includes any unique chemical Thallium Compounds: Includes any unique chemi	nicel substance that contains selenium as part of the I substance that contains silver as part of that chemical substance that contains thallium as part of that substance that contains zinc as part of that chemic	nical's intrastructure

#### Subpart E-Forms and Instructions

§ 372.85 Toxic chamical release reporting form and instructions.

(a) Reporting form.

BILLING CODE 6560-60-M

CAS No.	Chemical name	Effective date
25321-22-8 25376-45-8	Osmum tetroxide Dichlorobenzene (mixed isomers)	. 01/01/87 01/01/87 01/01/87 01/01/87

## (c) Chemical categories in alphabetical order.

Category name	Effective date.
Antimony Compounds: Includes any unique chemical substance that contains antimony as part of that chemical's infrastructure  Arsenic Compounds: Includes any unique chemical substance that contains arsenic as part of that chemical's infrastructure  Barium Compounds: Includes any unique chemical substance that contains barium as part of that chemical's infrastructure  Berytium Compounds: Includes any unique chemical substance that contains berytium as part of that chemical's infrastructure  Cadmium Compounds: Includes any unique chemical substance that contains cadmium as part of that chemical's infrastructure  Chlorophenols.	01/01/8 01/01/8 01/01/8 01/01/8 01/01/8
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Chromium Compounds: Includes any unique chemical substance that contains chromium as part of that chemical's infrastructure	01/01/
Copper Compounds: Includes any unique chemical substance that contains cobalt as part of that chemical's infrastructure	01/01/
Cysnide Compounds: X* CN* where X.= H* or any other group where a formal dissocution can be made. For example KCN_ or Ca(CN).	1 01/01/
Glycol Ethers: Includes mono- and di- ethers of ethylene glycol, diethylene glycol, and thethylene glycol	01/01
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Where:	1
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Polymers are excluded from this category.  Leed Compounds: Includes any unique chemical substance that contains lead as part of that chemical's infrastructure	01/01/
Manganese Compounds: includes any unique chemical substance that contains manganese as part of that chemical's infrastructure	01/01/
Mercury Compounds: Includes any unique chemical substance that contains mercury as part of that chemical's infrastructure	_ 01/01/

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Thallium Compounds: Includos any unique chemical substance that contains thallium as part of that chemical's infrastructure

#### Subpart E-Forms and Instructions :::

#### § 372.85 Toxic chamical release reporting form and instructions.

(a) Reporting form.

BILLING CODE 6560-60-M