# Treatment of Cheese Processing Wastewaters In Aerated Lagoons



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#### TREATMENT OF CHEESE PROCESSING WASTEWATERS

IN AERATED LAGOONS

Ву

Francis R. Daul

Project 12060 EKQ Program Element 1BB037 Roap/Task 21 BAD 26

Project Officer

Max W. Cochrane Environmental Protection Agency National Environmental Research Center Corvallis, Oregon 97330

Prepared for

OFFICE OF RESEARCH AND DEVELOPMENT U.S. ENVIRONMENTAL PROTECTION AGENCY WASHINGTON, D.C. 20460

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#### ABSTRACT

The treatment of cheese processing wastewaters in two-stage aerated lagoons was evaluated over a one year period. Aeration was provided by Polcon Corporation subsurface aerators (Helixors). The aeration system provided an average standard transfer rate of 4.0 lbs oxygen per kilowatt-hour at 20°C and 0.0 dissolved oxygen in tap water. Oxygen dispersion throughout the primary lagoon was normally adequate at a power input of 8.5 horsepower per million gallons. During the spring, however, dissolved oxygen concentrations in the primary lagoon were zero probably due to the solubilization of benthal deposits accumulated but not stabilized during the winter months. There was not sufficient power, however, to prevent suspended solids deposition. Horizontal velocity components varied from 1.0 fps within the central portion of the lagoon to less than 0.1 fps in the sloped peripheral zones. Over the one year period of this study, sludge accumulations ranged from 0 to 2 inches depending upon location.

The lagoon system performance during the one year period was correlated with temperature. The average total BOD removal was 97 percent producing an effluent BOD concentration of 52 mg/l. During the summer months secondary lagoon effluents were less than 20 mg/l, 70 percent of the time. Approximately 50 percent of the total effluent BOD was soluble. Effluent suspended solids were high, averaging 108 mg/l. Greater than 90 percent of the suspended solids were volatile. Total and fecal coliform reductions were normally greater than 99.9 percent throughout the study period.

During the early spring primary lagoon dissolved oxygen concentrations dropped to zero and remained below 0.1 mg/1 until mid July. Benthal deposits, accumulated during the winter period, were believed to account for the unusually high oxygen demands during this period. Demands in excess of three times greater than predicted by the BOD load to the lagoon were noted.

Costs for lagoon operation, maintenance, and amortization were estimated to be \$13,377 per year, \$2.15 per 1000 gallons, \$0.14 per 1b of BOD applied, and \$0.0033 per 1b cheese produced.

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#### SECTION I

#### CONCLUSIONS

The conclusions summarized below were the result of the analysis of data collected over a one year period from the two-stage aerated lagoons treating cheese processing wastewaters for the Kent Cheese Company.

Each of the two lagoons were of equal volume and provided detention times of from 50 to 82 days each during the first year of study. The average flow to the lagoons was approximately 17,000 gpd, ranging from 7,300 gpd to 35,500 gpd. Raw influent BOD loading averaged 285 lbs/d ranging from less than 15 lb/d to 554 lb/d resulting in a primary lagoon loading of from 0.117 to 4.34 lb BOD/1000 cu ft/d.

#### 1. Aerator Performance

- a. The oxygen transfer efficiency of the Polcon Corporation Helixors in the primary lagoon was 4.0 lbs oxygen/kw-hr or 3.0 lbs oxygen/hp-hr, for standard conditions (20°C, tap water, zero dissolved oxygen) at an air flow rate of 14.8 scfm per unit.
- b. An increase of the air flow rate by a factor of two per unit substantially reduced the oxygen transfer efficiency.
- c. The dispersion of oxygen in the primary lagoon at a power input of 8.5 hp/MG was nearly uniform.
- d. At the power input of 8.5 hp/MG, there was a tendency for suspended solids to accumulate in the peripheral regions where horizontal velocity components were normally less than 0.1 fps. The one year study was not long enough to evaluate the rate of solids accumulation within the lagoon.
- e. The relative oxygen transfer coefficient in the primary lagoon was approximately 81 percent of that expected in tap water. The oxygen saturation value in the primary lagoon content was approximately 98 percent of that in pure water.

#### 2. Lagoon Performance

a. An average of 97 percent removal of BOD was achieved in the two stage aerated lagoon system with the poorest performance (95 percent) occurring during the winter season. The average final effluent BOD concentration was 52 mg/l of which 27 mg/l was soluble. Probability distributions for treatment performance were prepared for the one year study.

- b. Temperature had an important effect on lagoon performance.
- c. Suspended solids removals in the primary lagoon were sporadic. The overall average removal of suspended solids in the lagoon system was 82 percent with an average concentration of 108 mg/l in the final effluent. Over 90 percent of the suspended solids were volatile.
- d. Approximately 29% of the volatile suspended solids in the final effluent contributed to the effluent BOD. A similar relationship for the primary effluent suspended solids indicated that 46 percent of the volatile suspended solids contributed to the BOD.
- e. High oxygen uptake rates occurring in the early spring resulted in zero D.O. concentrations in the primary lagoon. These high uptake rates are likely due to the solubilization of solids deposited during the winter months. The cycling of sludge accumulation during the cold periods and active biological stabilization of the benthal deposits during the warmer months will result in dynamic fluctuations in oxygen requirements and sludge deposits. Aerator sizing to account for these fluctuations must be provided. In this study oxygen demands in excess of three times the influent BOD were measured during the early spring.
- f. The staging of the wastewater treatment lagoons provided considerable attenuation of the fluctuating BOD and solids concentrations in the primary lagoon effluent.
- g. Total and fecal coliform reductions exceeded 99.9 percent in the two stage lagoon system without disinfection.
- h. Nitrogen concentrations in the raw wastewater were low with respect to carbon. The BOD to Nitrogen ratio was 100:0.7 indicating a substantial deficiency in nitrogen. Phosphorus concentrations were high resulting in a BOD:P ratio of 100:2.2.
- i. Total phosphorus removals of 52 percent observed during the one year study may be misleading insofar as a steady state with respect to phosphorus transformations has probably not been established. Total organic nitrogen removal of 47 percent were observed. Nitrate concentrations in the final effluent were highest during the warm summer months but were less than 0.4 mg/l during the rest of the year.

#### 3. Costs

The cost for wastewater treatment including operation, maintenance, & capital cost amortization during the first year of operation was \$2.15 per 1000 gal., \$0.14 per 1b BOD applied, and \$0.0033 per 1b of cheese produced.

#### SECTION II

#### RECOMMENDATIONS

Based on the operation of the two stage cheese processing wastewater lagoons at Kent, Illinois, the following recommendations have been proposed:

- 1. Efforts should be made to further reduce the settleable solids input to the lagoons. These solids cause clogging problems at the inverted siphon, interfere with flow measurement and sampling at the primary lagoon, and result in increased BOD loading to the primary lagoon. Insofar as solubilization of settleable solids during the warm months has resulted in severe depressions in dissolved oxygen concentrations in the primary lagoon, every measure taken to reduce influent settleable solids will help to alleviate that problem. It is suggested that a settling tank or imhoff tank be provided ahead of the primary lagoon for this purpose.
- 2. The high air pressures observed during dual compressor operation suggests that the air headers and inlet orifices are undersized for the increased air flow rates. Since oxygen uptake rates during the spring and early summer exceed current oxygen transfer rates, it is recommended that investigations be made to determine the necessity for enlarging existing air piping or providing additional Helixors in the primary lagoon.
- 3. The increased oxygen uptake rates in facultative aerated lagoons during the spring have been reported by a number of investigators. Further investigation should be conducted to provide a quantitative estimate of this increased demand. Such a study would require data collection over a number of years at the existing aerated lagoon site.
- 4. Nitrogen deficiencies in the raw wastewater will normally result in poorer performance of the biological system. It is recommended that additional nitrogen be added as ammonia or urea to more closely approximate the BOD to Nitrogen ratio of 100 to 5. Comparisons in lagoon performance should be noted during this period (at least one full year) so as to evaluate the value of this supplimentation.
- 5. The secondary lagoon effluent structure consists of only an upturned 4 inch cast iron pipe elbow. It is recommended that this effluent structure be modified so as to provide some baffling to eliminate gross solids and scum entrainment.

#### SECTION III

#### INTRODUCTION

The evaluations and findings reported herein were supported by EPA Project No. 12060 EKQ for post-construction studies over a 12-month period to demonstrate and evaluate the use of aerated lagoons for the treatment of cheese wastes. Initially the objectives of the study included the evaluation and demonstration of the effectiveness of reverse osmosis for the reduction of BOD of cheese whey wastes. The decision was made that the reverse osmosis method would not be included in this study.

The authors of this report were not engaged to design or size the system employed, select equipment utilized or were consulted regarding the process flowsheet applicable to this wastewater treatment scheme. The authors were retained to conduct the post construction study, direct frequency of sampling, recommend analyses to be performed, and evaluate and present findings of the performance of the aerated lagoon systems for the wastes received at the treatment site under the limitations of the existing budget.

The principal objectives of this study were to demonstrate the performance of a staged aerated lagoon treatment plant utilizing the Helixor\* type of submerged aeration system treating cheese processing wastes over a 12 month period of operation. Part of the evaluation was directed to the performance of the aeration equipment employed wherein mixing effectiveness in terms of liquid velocities produced, uniformity of dissolved oxygen levels and sludge accumulations could be ascertained under the conditions of operation and geometric configuration of the lagoons employed. In addition the oxygen transfer efficiency of the aeration system employed was evaluated under field conditions. A major part of this study was to determine the BOD removal rate functions for each stage of aerated lagoon treatment noting the influence of BOD loadings, temperature and seasonal variations in loading and performance obtained for this type of biological treatment. The theoretical and applied concepts of biological treatment related to low-solids aerated lagoon systems were employed to evaluate the treatment system. Lastly, the performance of this type of treatment was evaluated in terms of costs associated with the wastewater characteristics and the pounds of cheese produced.

<sup>\*</sup>subsurface aeration unit manufactured by Polcon Corporation, Montreal, Canada

#### SECTION IV

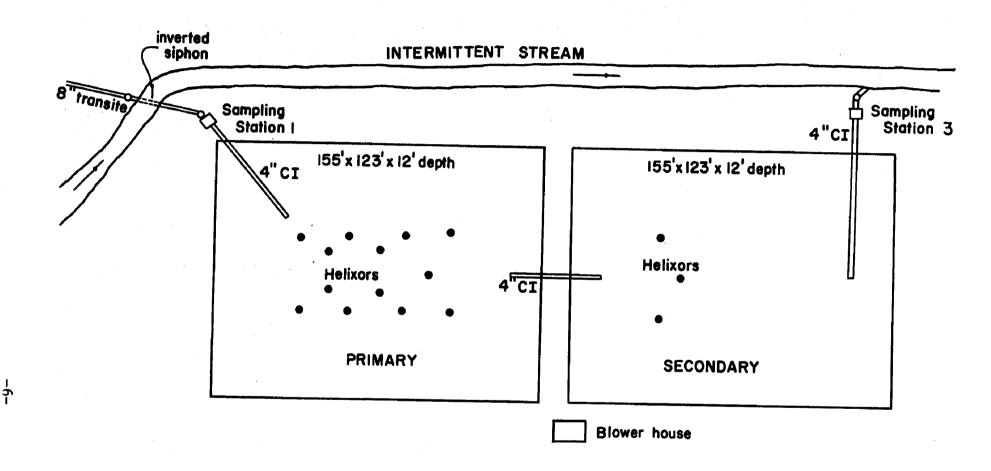
#### EXPERIMENTAL SYSTEM

The wastewater treatment system consisted of two equal volume aerated lagoons with a staged flowsheet wherein the effluent from Lagoon No. 1 is passed to Lagoon No. 2 before final discharge of the treated effluent. The lagoons had earth embankments with side slopes of 3 horizontal to 1 vertical, a water depth of 12 feet at the lowest central portion of the lagoon, and a length-width measurement of 157 feet by 123 feet at the water line for a volume of 955,000 gallons each (Figure 1). The first lagoon was provided with thirteen 6 foot long 18 inch diameter Helixors arranged in a pattern of 2 rows of 4 units equally spaced along the intersection of the flat bottom and side slopes of the long dimension of the lagoon. Five additional units were added to this lagoon between the 2 rows of four units such that 4 of the five units were equally spaced in 2 rows between the 1st and 2nd and 2nd and 3rd aeration units. The fifth additional aerator was equally spaced between the third and fourth unit of the initial eight unit arrangement (Figure 1). The second lagoon was provided with three aeration units arranged in a triangular pattern close to the inlet end of the lagoon in the lowest central section of the lagoon.

A separate blower building with two rated 240 cfm Gardner Denver rotary blowers (one standby), Model 3CDL5, provided the air supply to both lagoons. Usual operating conditions required only one compressor to be operative and it was assumed that the air flow was distributed between the two aerated ponds with an estimated 80% of the airflow supplied to Lagoon No. 1 and the remainder to Lagoon No. 2. The air flow regulation and distribution was effected by the number and size of air orifices providing air directly to each Helixor. The air header piping and valving was arranged to permit the control of air to a pair of aeration units in most instances; however, several aeration headers supplied air to a single unit. The valving in the aeration headers were used only for fully open or closed conditions wherein the inlet orifices controlled the distribution of air flow.

It appeared that the recommendation of the Polcon Corporation Specifications on concrete weights for holding down the air lines and Helixors was inadequate. An inspection of the air lines and Helixors after the initial start up period was performed in the spring of 1970 by a scuba diver determining the location of the air lines, Helixors, and concrete weights in reference to the bottom of the lagoon. It was noted that not all of the Helixors were positioned on the bottom of the lagoon. Rather, some were 18" - 24" above the bottom. Approximately 500 pounds of extra concrete weights were added to each individual lateral.

### KENT CHEESE WASTEWATER LAGOONS



FIGURE

Six additional Helixors were installed as per Polcon Corporation specifications utilizing 250 pound concrete weights under each additional Helixor. These six Helixors and lines remained in their original positions.

The wastewater was conveyed to the treatment lagoons through about 3000 ft. of 8 inch transite pipe, passes through a 4 inch inverted siphon to a Snyder-Teague sampling station and thence is discharged through a 4" cast iron pipe to a point of entry to Lagoon No. 1 at approximately mid-depth. The effluent from Lagoon No. 1 passed through a submerged 4 inch cast iron pipe to the secondary lagoon. The water surface elevation in both lagoons was controlled by the placement of 4 inch cast iron riser pipe with the inlet to this pipe, or overflow from the lagoons, at a fixed elevation to maintain the 12 foot water depth. The effluent from the secondary lagoon passed through another Snyder-Teague sampling station.

The sampling stations provided flow measurements and flow composited samples for the raw waste water and treated effluents. Serious operating difficulties resulted in the raw waste sampling station due to the accumulation of cheese solids upstream from the flow measuring control section. The sampling device operated on the basis of taking a fixed volume aliquot on the discharge side of the flow measuring weir. The sampling device was activated by the water depth on the upstream side of the flow measuring weir; consequently, when the upstream float was supported by accumulated solids in the absence of influent flow, the sampling device was activated but no wastewater aliquot could be collected for the composite sample at these times. Thus, the composited sample for quality determination was considered to be reliable, whereas the flow measurements required correction.

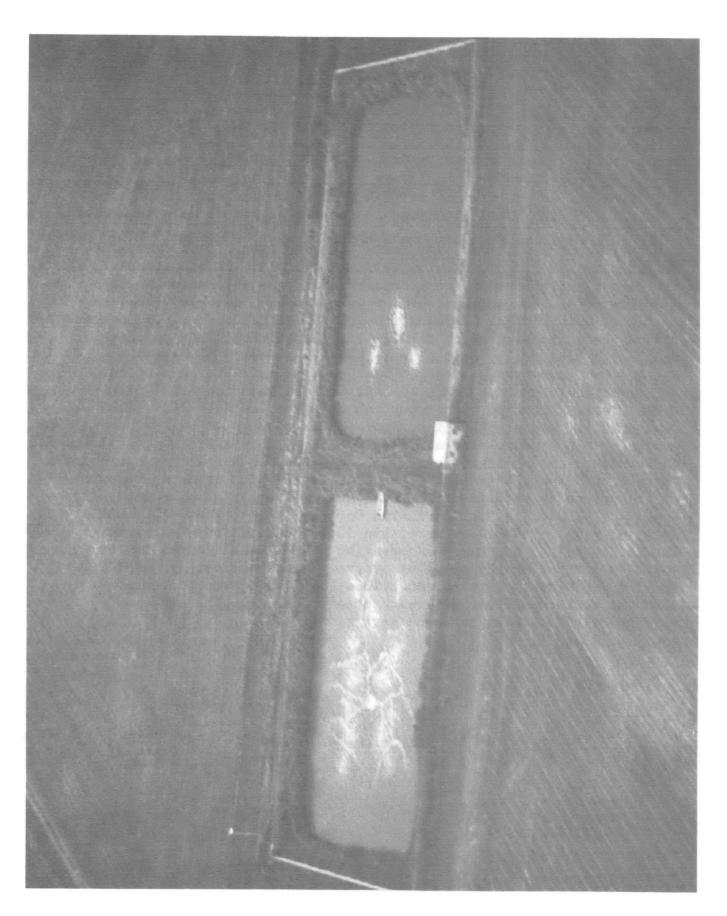
The accumulation of solids resulted in inaccurate flow measurements. The control section was modified somewhat but proved to present difficulties throughout the period of the study. In order to obtain reliable measurements of flow, it was necessary to meter water use at the cheese processing plant and to determine boiler feedwater requirements and the extent of infiltration of the conveying sewerage system for reliable flow information. Routine measurements of water pumped at the cheese processing plant were made throughout the post construction study. The well pumps were calibrated and clocks were employed to record total water used. Boiler feed water was also monitored routinely. Only after considerable effort had been exerted to correct the installed flowmeters at the lagoon site was it determined that corrected water use data would have to be employed for flow estimates to the lagoons. At that time infiltration studies were conducted over two weekends, Dec. 4-5 and Dec. 18-19, 1971, to provide an estimate of this flow contribution. During these two periods infiltration rates of approximately 690 and 900 gal/d were measured. The pipeline conveying the cheese process waters to the lagoon system was constructed in 1969 and employed a 12 inch diameter Armco truss pipe with chemical band joint. The manufacturer estimates a maximum infiltration rate with this type of construction

of 100 gal/inch diameter/day. This would result in a value of 875 gal/day based on 3,860 ft of 12 inch pipe. An average of 800 gal/d correction was employed to all water use data recognizing that this value would undoubtedly fluctuate with season. The corrections that were employed are tabulated in Appendix C.

A sampling station was not provided between the primary and secondary lagoons; thus, grab samples of the primary lagoon contents were taken near the effluent discharge pipe. Because of the long detention times experienced in each lagoon, 45-75 days, this procedure was deemed acceptable.

An aerial photograph of the wastewater treatment system appears in Figure 2.

FIGURE 2
AERIAL PHOTOGRAPH OF LAGOON SYSTEM



## SECTION V PROCEDURES

The treatment system handled the wastewater from a cheese making industry specializing principally in products of Ricotta, Parmesan, Romano, and Mozzarella cheese. The sources of wastewater were primarily from rinses and washes associated with the storage of milk, transmission lines, vats and pasteurizer with a limited amount of wastes of domestic origin. The by-product whey was collected and transported to another site for recovery and only whey wastes associated with washing and rinsing were discharged to this treatment system. The sources of wastewater treated by the aerated lagoon system are shown in Figure 3. The quantity or quality of rinsewater from each operation was not determined in this study, but rather the collective properties of the total discharge to the lagoon system were determined.

Certain measurements were made daily for operation of the treatment works such as D.O., pH, alkalinity, settleable solids, temperature, and flow quantity whereas more detailed analyses for evaluating the performance of the treatment system were obtained for raw wastewater influent, primary lagoon contents near effluent structure, and secondary lagoon effluent on an eight day sampling frequency. This permitted each day of the weekly operation to be sampled every 56 days throughout the one year study commencing on January 14, 1971. A sampling schedule is outlined in Table 1. All the analyses indicated in Table 1 were performed by Corning Laboratories, Inc., Cedar Falls, Iowa. After sample collection, appropriate volumes of well mixed sample were placed in separate sample bottles for coliform determination, for nitrogen and phosphorus analysis, and for BOD and solids analysis. Nitrogen and phosphorus samples were preserved with mercuric chlorides as prescribed by Methods of Chemical Analyses (1). Samples were shipped immediately after collection to the Corning Laboratories, Inc. where analyses were initiated no more than 10 to 12 hours after collection. The lagoon system had been in operation approximately eight months prior to initiating the sampling program; thus, the data presented does not represent start-up conditions, but there were indications that the performance did not reach steady state in all respects particularly regarding seasonal variations.

FIGURE 3

#### SOURCES OF CHEESE PLANT WASTEWATER

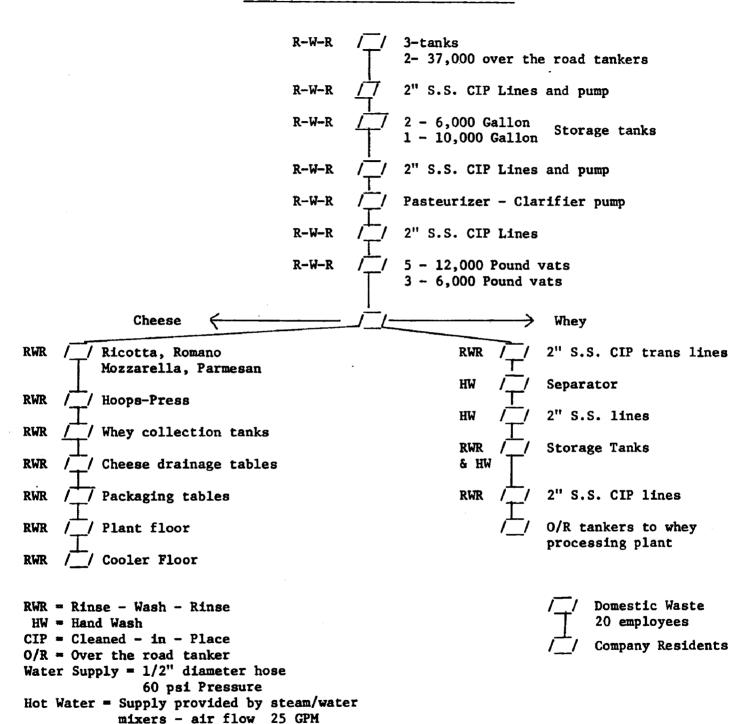


TABLE 1
SAMPLING SCHEDULE\*

Sampling Frequency	Parameter or Determination	Raw	Sampling Po Primary Effluent	oints Secondary Effluent
8 days	BOD 5 day 20°C Total	X	x	X
8 days	BOD 5 day 20° Soluble**	X	X	x
8 days	Total Suspended Solids	x	X	X
8 days	Total Volatile Suspended Solids	X	x	X
8 days	Total Coliforms	X		X
8 days	Fecal Coliforms	x		x
30 days	Total Phosphorus	x		x
30 days	Total Kjeldahl Nitrogen	X		X
30 days	Nitrate Nitrogen	x		X

<sup>\*</sup>Analyses performed according to Methods of Chemical Analysis (1)
\*\*Analysis performed on sample filtrates (Whatman #42 filter paper)

Additional operating and performance evaluations were performed during the experimental period including the following items:

- 1. D.O. measurements with respect to horizontal and vertical control in the primary lagoon,
- 2. measurements of velocity in the horizontal plane at various locations and depths within the primary lagoon,
- 3. measurement of oxygen uptake of the treatment lagoon contents,
- 4. determination of the oxygen transfer coefficients of  $\alpha$  and  $\beta$  to enable the evaluation of the aeration system employed.

Various other operating conditions and observations were recorded on a routine basis to assist in the overall evaluation of the performance of this treatment system (Appendix C).

Special measurement techniques employed for the purpose of evaluating the aeration system employed a grid established in the primary lagoon above the water surface to provide a measuring base for horizontal control. At the time oxygen transfer was evaluated, it was necessary to have a measureable D.O. in the lagoon and evidence that the D.O. was maintained at a uniform or steady state condition. The oxygen uptake or demand of the lagoon contents were determined by taking a number of representative samples of the mixture under aeration and placing them in bottles capable of excluding further oxygen transfer from the atmosphere and wherein the decrease in D.O. concentration was measured with respect to time with a Yellow Springs Instrument D.O. probe. In order to evaluate the influence of photosynthetic plankton on the oxygen transfer in the lagoon, oxygen uptake measurements were made both on rates observed under light and dark conditions. There was no discernible difference in the rates observed which indicated that oxygen transfer or supply from this source was negligible during the test periods.

Aerator efficiency was determined on the basis of line to water for the complete aeration system wherein power input was metered in each instance for amperage and voltage drawn. The results were presented in terms of pounds oxygen transferred per kw-hr. Thus, the reported efficiency included oxygen transferred from the atmosphere as well as oxygen transferred by virtue of the aeration system employed.

Aeration efficiencies were corrected to standard conditions of zero dissolved oxygen and 20°C with appropriate corrections for  $\alpha$  and  $\beta$ . Representative samples of the lagoon contents were tested in Madison laboratories in a simulated diffused air system to compare transfer capacities (a) for the waste mixture against tap water for two replicate samples. The value  $\alpha$  was obtained as an average of the ratio of the transfer rates in the wastewater and tap water. The value  $\beta$  was

determined on the basis of the highest D.O. obtained in the aerated lagoon contents for a given temperature, checked by the Winkler D.O. method, against the D.O. at saturation for tap water at the same temperature.

The velocity profiles in the aerated lagoon were made with the use of a Gurley Current Meter which was fixed in the horizontal plane by attachment to a vertical rod. The current meter was rotated in the horizontal plane at a predetermined depth and the maximum velocity and direction were noted to obtain a vectorial representation of the water movement in the lagoon. The velocity was observed in four directions parallel to the sides of the lagoon in the central portion of the aerated lagoon where a single maximum velocity component was not observed due to the highly undirected flow patterns evident within the area bounded by the aeration devices. An attempt was made to measure the vertical velocity component immediately above the discharge of the aeration device but the variation in fluid density as a result of high levels of air entrainment caused on the measurements to be somewhat erratic. Measurements were made one foot below the surface, mid-depth and one foot from the bottom.

#### SECTION VI

#### RESULTS AND DISCUSSION

#### Aerator Performance

The performance of the Polcon Corporation subsurface aerators (Helixors) was evaluated over the one year period in the primary lagoon only. As indicated earlier, the lagoon was sampled at a number of points on a grid (Figure 4) at selected depths to determine the oxygen uptake rates, dissolved oxygen concentration, and temperature. Tests were repeated five times over the grant period in order to evaluate performance under different waste loading and environmental conditions.

Oxygen Uptake Rates - The average oxygen uptake rates recorded in the primary lagoon over the one year study period are presented in Table 2. The values reported represent the average uptake measured at a number of selected points within the lagoon (see Appendix A). Normally, the uptake rates measured on any given day were within 10 percent of each other. The uptake rates reported were not converted to specific uptake rates (mg O<sub>2</sub>/gVSS/day) since the volatile solids fraction measured in this lagoon was not well correlated with the active biomass during the study period. Large amounts of suspended volatile matter non-biological in origin associated with the wastewater contributed significantly to the total volatile fraction measured. The influence of algal cells on the oxygen uptake rates in the primary lagoon were negligible as measured by both dark and light bottle uptake rates.

Examination of the uptake rates presented in Table 2 suggest that the biological activity in the primary lagoon was comparable to other aerated lagoon systems (2). The analysis obtained in May, 1971, occurred during the anaerobic period and indicates that the lagoon was extremely active. As discussed later, this high rate of oxygen consumption was believed to be due to the rapid solubilization of organic matter and synthesized cells which had settled in the lagoon and had subsequently been stored over the cold winter months.

Oxygen Transfer Capacity - The values of the oxygen transfer capacity of the primary lagoon contents, as measured by the ratio of the oxygen transfer coefficients in the waste to that in tap water, are reported in Table 3 as alpha. These values represent the average values obtained from several points within the lagoon. Alpha appeared to fluctuate seasonally being highest during the summer months. There was not sufficient data to establish any reliable correlations between alpha and lagoon BOD solids or temperature, but it is reasonable to assume that changes in the metabolic activity within the lagoon would result in changes in the oxygen transfer relationships.

TABLE 2

OXYGEN UPTAKE RATES

PRIMARY LAGOON

(Average Values)

DATE	TEMP.	D.O. mg/l	UPTAKE RATE mg/1/hr 1b/h			
11-19-70	6.3	2.8(43)	2.1 (5)	16.7		
5-23-71	17.0	0.0(43)	5.6 (5)	44.6		
7-21-71	24.0	4.4(39)	1.2 (4)	9.55		
8-26-71	23.5	4.4(27)	0.8 (4)	6.4		
10-7-71	16.5	2.3(41)	1.5 (3)	12.0		

( ) - Number of samples analyzed (see Appendix A & Table 4)

TABLE 3
OXYGEN TRANSFER DATA
PRIMARY LAGOON

DATE	TEMP.	D.O. mg/1	α	β	STD. 1b/kw-hr	TRANSFER <sup>1</sup> 1b/hp-hr
11-19-70	6.3	2.8	0.75	0.96	5.25	3.91
5-23-71	17.0	0.0	0.88	0.96		
7-21-71	24.0	4.4	0,90	0.98	3.18	2.47
8-26-71	23.5	4.4	0.94	0.98	0.98	0.73
10-7-71	16.5	2.3	0.70	0.98	3.45	2.59
Overall Ave	erage <sup>2</sup>		0.81	0.97	3.96	2.98

 $<sup>1 - 20^{\</sup>circ}C$ , D.O. = 0.0, Tap water - see Appendix B

<sup>2 -</sup> Excluding 8-26-71 data for reason on page 19.

The solubility of oxygen was only slightly affected by the wastewater characteristics within the primary lagoon. Normally values of beta in long detention-type processes such as the one studied here approach 1.0.

Aeration Efficiency - The aeration efficiency of the Polcon Corporation Helixors are presented in Table 3 for five test days. The aeration efficiencies were computed by employing the measured oxygen uptake rates, dissolved oxygen concentration, alpha, beta, and temperature. Power was metered during the test period. The value of the aeration efficiency was corrected to standard conditions of 20°C, tap water (alpha and beta equal 1.0), and a dissolved oxygen concentration of 0.0 mg/l. The approximate saturation value for oxygen for the 12 foot lagoon depth was estimated by using the relationship given by Oldshue(3):

$$C_s = C \begin{cases} \frac{P_b}{20.4} + \frac{O_t}{42} \end{cases}$$
 (1)

where C is the oxygen saturation value corresponding to the average partial pressure of oxygen in the gas stream entering and leaving the aerator, C is the oxygen saturation value of water at atmospheric pressure,  $P_b$  is the absolute pressure in psi at depth, d and  $O_t$  is the percent concentration of oxygen in the air leaving the lagoon. A sample calculation appears in Appendix B.

The overall aeration efficiencies in the primary lagoon were estimated by assuming that 80% of the air provided by the blower was directed to the primary lagoon, since 13 of the 16 helixors were located in the primary lagoon. During August 26, 1971, test, both blowers were in operation resulting in a doubling of the power input to the aerators. Yet the actual measureable oxygen transfer rate was not appreciably higher than for other test periods. It has been well documented in the literature (4,5) that aeration efficiency decreases with increased air flow rates and this analysis tends to verify that fact. In addition, however, it was determined that, during the operation of both compressors, the line pressures increased very significantly requiring the bleeding of air from one of the headers. This was practiced by the use of a "blow-off" or "by-pass" Helixor in the secondary lagoon. Thus, the poor transfer rates for August 26 were not included in the overall average for aeration efficiency.

The measured transfer efficiencies of the Helixor units were lower than the expected performance claimed by the manufacturer (6). Whether this was due to undersizing of the air headers, the geometry of the Helixor-basin configuration, or was, in fact, the actual field expectation for these units has not been determined but it suggests that caution must be exercised in sizing aerators for lagoon systems. An additional factor to be considered was the oxygen uptake occurring in the benthal deposit. No effort was made to ascertain what this contribution was, but significant uptake from this source would substantially increase the aeration efficiencies reported herein.

Oxygen Dispersion - An important function of aeration devices in biological processes is the dispersion of oxygen throughout the basin contents. Normally, rule of thumb criteria recommend that at least 6 to 10 horsepower per million gallons (hp/MG) be supplied to insure adequate oxygen dispersion (7). For the primary lagoon, 8.5 hp/MG were provided during the single blower operation. The results of dissolved oxygen analyses in the primary lagoon for the five field surveys are presented in Table 4. Examination of this data indicates that oxygen was well dispersed throughout the primary lagoon.

Mixing and Solids Suspension - The D.O. parameter is indicative of the dispersion of substances in the dissolved state such as soluble B.O.D. However, the distribution of particulate matter such as biological floc would be a function of the velocity distribution and the velocities necessary to keep the particulate matter in suspension. Normally in aeration tanks in wastewater treatment, velocities along the bottoms of these units would range from 1.0 to 2.0 fps depending upon the geometry of the aeration system employed. Velocity measurements within the lagoon can serve only to show flow regimes developed within the lagoon with the given aeration equipment for the placement arrangements employed and the geometry of the lagoon. Effectiveness of the system in terms of keeping particulate solids in suspension can be evaluated in a qualitative way by determining the location and extent of solids deposition on bottom surfaces of the lagoon.

The maximum velocities observed in the primary lagoon at the various depths are reported in Table 5 for one blower, 192 cfm, and two blowers, 384 cfm. These airflow rates correspond to 2.3 cfm per lineal foot and 4.6 cfm per lineal foot respectively along the longest bottom dimension of the lagoon. The vectorial representations of the resulting surface velocities are shown in Figures 5 and 6 for the single and dual blower operation.

An estimated surface wind of less than 5 mph as indicated on the diagram may have influenced the resulting velocity measurements. Because of the upward vertical velocity components in the immediate vicinity of the inlet and outlet of the Helixors and the necessary compensating downward movement of water between adjacent aeration units, no definitive flow pattern is discernible. However, in zones peripheral to the aeration section, where the bottom of the lagoon is sloped to intersect the water surface, horizontal velocities were measured near the water surface indicating a general circulation pattern towards the periphery of the aerated lagoon with velocities ranging from less than 0.1 to 0.8 fps. Likewise, to compensate for the outward movement of water toward the periphery of the lagoon near the surface, water movement must be in the opposite direction near the bottom of the lagoon, the measurements and magnitude of which was less discernible. Thus, a circulatory pattern of flow developed, in the outer prism shaped sections similar to that which is depicted in Figure 7.

2

# TABLE 4 DISSOLVED OXYGEN CONCENTRATIONS\* PRIMARY LAGOON

#### Station Number

Date	11	2	3	4	5	6	7	8	9	10	11	12	13	14	15
11/19/70	2.7	2.9	2.9	2.7	3.1	2.9	3.0	2.8	2.8	2.8	2.8	2.7	2.8	2.7	2.6
1 blower	2.7	2.8	2.9	2.7	3.0	2.9	2.9	2.8	2.8	2.8	2.7	2.7	2.7	2.7	2.5
		2.8	2.7	2.7	3.0	2.9	2.9	2.8	2.7	2.8	2.7	2.7	2.6	2.7	
5/14/71	0.1	0.1	0	0	0.3	0.1	0	0.3	0.1	0.1	0.3	0	0	0.1	0.1
1 blower	0.1	0.1	0	0	0.2	0.1	0	0.2	0.1	0.1	0.3	0	0	0.1	0
		0.1	0	Ō	0.2	0	0	0.1	0		0.3	0	0	0.1	0
7/21/71	4.4	4.4	4.4	4.3	4.3	4.4	4.6	4.7	4.2	4.8	4.4	4.3	4.9	4.7	4.4
1 blower	4.3	4.2	4.3	4.3	4.3	4.3	4.7	4.7	4.2	4.7	4.4	4.2	4.8	4.7	4.4
		4.2		3.0	4.3	4.2	4.7	4.5	4.2	4.7	4.3				
8/26/71	4.1	4.3	4.7	4.3	4.4	4.7		4.7			4.4		4.4	4.5	4.5
2 blowers	3.9	4.6		4.3	4.4	4.7		4.6			4.4		4.4	4.5	
		3.0		4.4	4.5	4.7		4.6			4.5		-	4.5	
10/7/71	2.2	2.2	2.7	2.3	2.3	2.4	2.3	2.4	2.3	2.3	2.1	2.3	2.4	3.0	
1 blower	2.0	2.1	2.7	2.4	2.3	2.4	2.3	2.2	2.3	2.3	2.1	2.3	2.3	2.6	
	2.1	2.0		2.3	2.3	2.4	2.3	2.2	2.3	2.3	2.1	2.3	2.3	2.5	

\*D.O. values in mg/1 at 1 ft

3 ft

5 ft below water surface

TABLE 5

VELOCITY DISTRIBUTIONS IN PRIMARY LAGOON

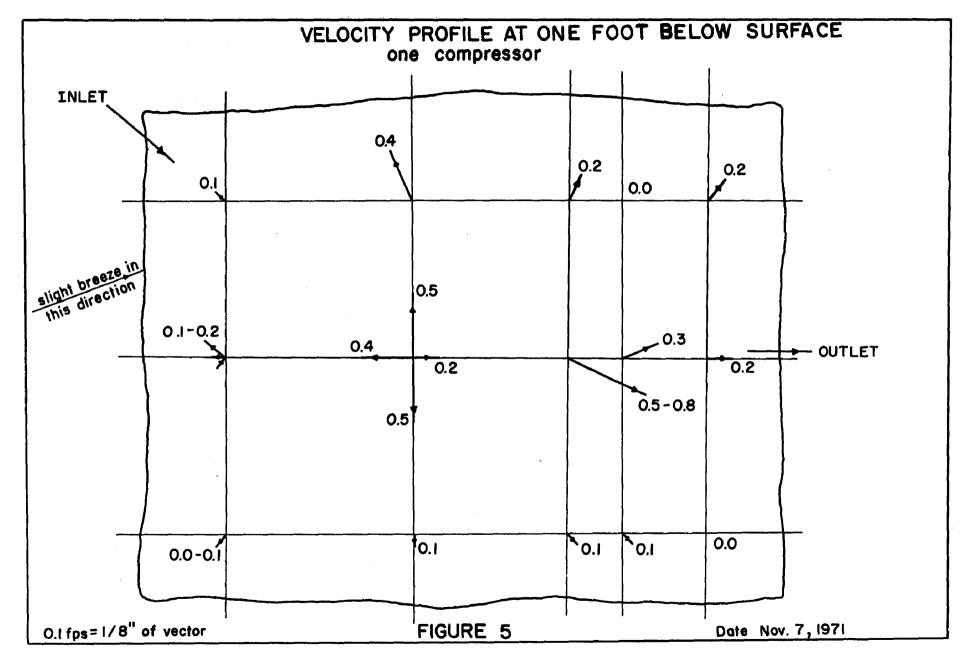
November 7, 1971

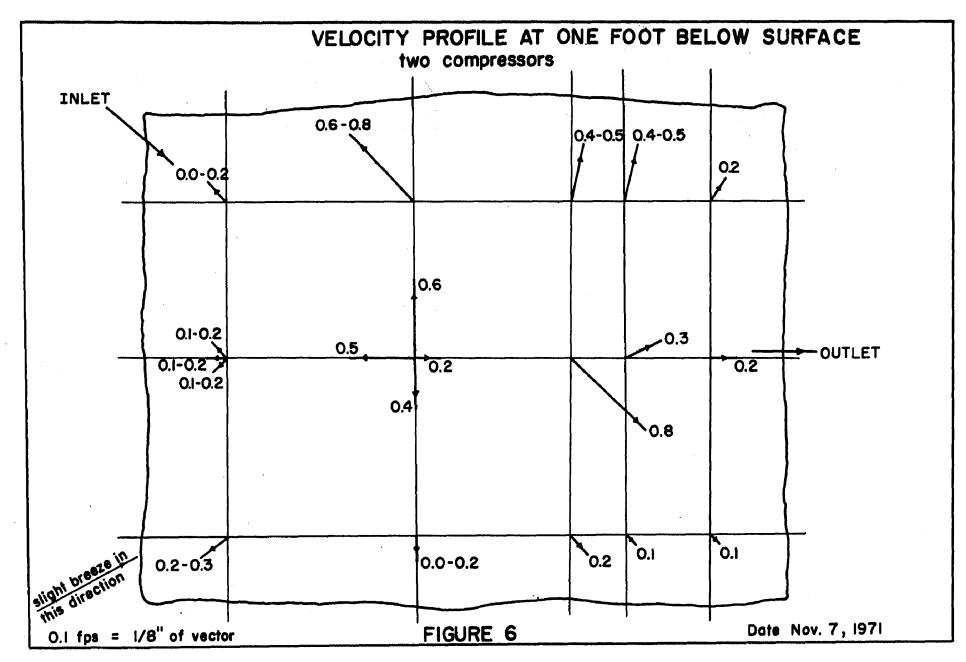
	On	e Compresso	r skal	Two Compressors			
Station	Surface*	Mid Depth	Bottom	Surfac	Mid Depth	Bottom**	
1	0.1	0.0	0.0	0.0-0.2	0.0-0.2	0.0	
2	0.1-0.2	0.0	0.0	0.1-0.2	0.0-0.1	0.1-0.2	
3	0.0-0.1	0.0-0.1	0.0	0.2-,0.3	0.2-0.3	0.0-0.1	
4	0.4	0.1	0.0	0.6-0.8	0.3-0.4	0.1=0.2	
5	0.2-0.5	0.2-0.3	0.1	0.2-0.6	0.2-0.3	0.0-0.2	
6	0.1	0.0	0.0	0.0-0.2	0.0-0.2	0.0-0.1	
7	0.2	0.0	0.0	0.4-0.5	0.0-0.2	0.0-0.1	
8	0.5-0.8	0.2	0.2	0.8	0.2	0.2	
9	0.1	0.0	0.1	0.2	0.1	0.1	
10	0.0	0.2	0.0	0.4-0.5	0.0-0.1	0.0	
11	0.3	0.2	0.1	0.3	0.2	0.1-0.2	
12	0.1	0.0	0.1	0.1	0.1	0.0	
13	0.2	0.0	0.0	0.2	0.0	0.0	
14	0.2	0.1	0.0	0.2	0.0	0.1	
15	0.0	0.0	0.0	0.1	0.0	0.0	
			I				

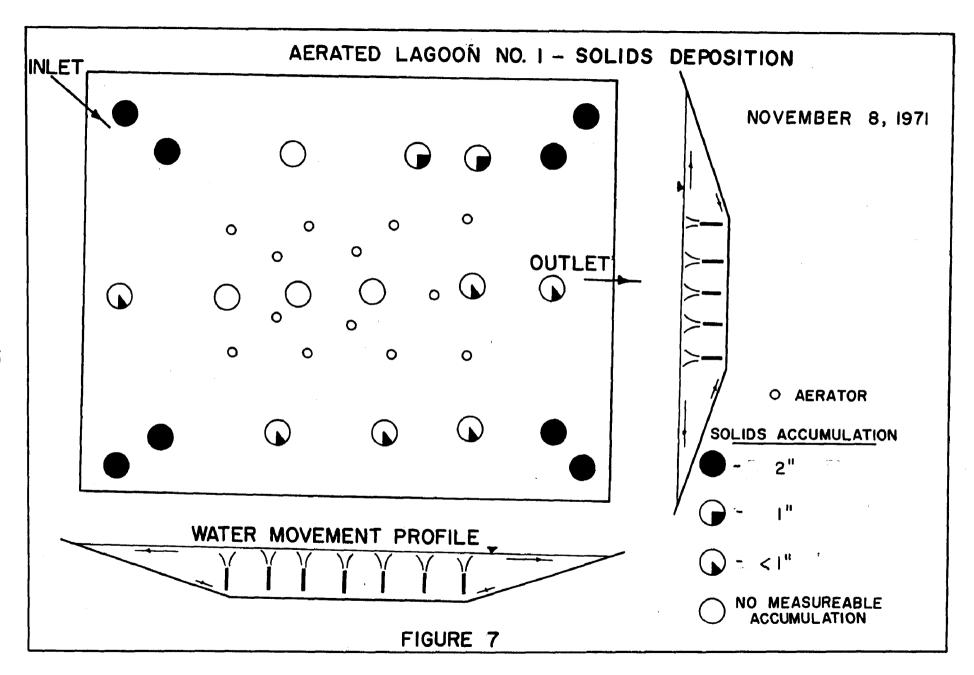
All values of velocity in ft/sec.

<sup>\*</sup> One foot below

<sup>\*\*</sup>One foot above







Perhaps a more meaningful observation that was made to ascertain the ability of the aeration system to keep solids in suspension was to estimate the accumulation of solids on the lagoon bottom. Utilizing a D.O. probe, a crude estimate of the sludge accumulation was made by lowering the probe to an elevation where the D.O. decreased to zero. This elevation was compared with that where a weight would come to rest. a point which was presumed to be at the bottom of the lagoon. results were plotted qualitatively by the graphical representations for deposit as shown in Figure 7. It is noted that the greatest accumulation reported on November 8, 1971, almost two years after lagoon operation, occurred in the non-aerated zones of the lagoon corresponding to the tapered prism sections representing sloped sections of the lagoon nearest the intersection of the two adjacent sloped bottom sections. least discernible deposits of solids occurred in the central aeration zone which may be due either to the higher levels of fluid turbulence to keep the solids in suspension or due to the greater diffusion of oxygen in the lighter accumulations of sludge at these points. longer periods of treatment plant operation, it is likely that greater accumulations will occur. However, at the time the determinations were made, it would appear that the accumulation of solids would not present a serious problem for this system.

Aeration System Operation - Maintenance of low ambient temperature in the compressor building during the months of July and August required the installation of a fan to maintain an equilibrium temperature between the inside and the outside of the building. During the extreme cold months of the winter, when the temperature of the water drops to approximately 4 degrees Centigrade, the brake horsepower of the fifteen horsepower motor climbed to 18-19 brake horsepower. This problem was corrected by utilizing the six extra Helixors that were initially installed to discharge the air whenever two compressors were required to be in operation in order to maintain residual dissolved oxygen in the springtime. This problem of increased brake power could also be handled by replacing the fifteen horsepower motors with twenty horsepower motors in this particular situation.

The recommendations of Polcon Corporation for stainless steel clamps with the ends bent over has proven to be inadequate. Since January, 1971, three Helixors have floated to the top because of the stainless steel clamp vibrating loose. A type of bolt-nut clamp with the threads damaged after the connections have been made would prevent the nuts from vibrating off the bolts.

#### Lagoon Performance

The performance of the staged lagoon system was interpreted in terms of several measured parameters: BOD (total and soluble), suspended solids, nitrogen, phosphorus, coliforms (total and fecal). The results reported herein are subdivided in accordance with these parameters. Tabulated results of lagoon performance appear in Appendix C.

The loading to the lagoon system is summarized for the one year study in Figure 8. It is apparent from this figure that flow and organic load to the lagoon is quite variable over the one year period. Hourly flows over a typical 24 hour period are presented in Figure 9 for two days, January 24 and 25, 1972. Analysis of this data indicates that the ratio of maximum to average flow for the day was 1.9. The average flow rate over the 12 month study was approximately 17,000 gallons per day including Sundays.

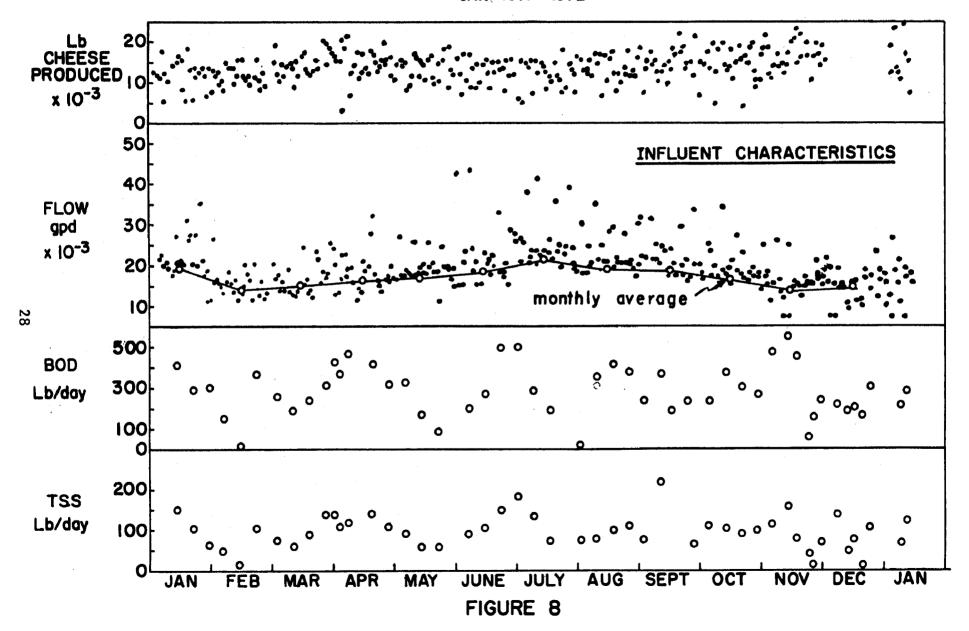
The apparent cyclic variation in BOD and solids loading to the lagoons (Figure 8) are due to the sampling schedule employed. As noted in Table 1, samples were collected every 8 days in order to obtain data over the entire week. The flows and pollutant loads noted, therefore, represent weekly cycles. Therefore, a trend line was difficult to describe. Average BOD loading over the 12 month study was 285 lb/day and suspended solids loading was 85 lb/day. An average of 13,000 lb cheese were produced per day (excluding Sunday) requiring approximately 1.3 gallons of water per lb of cheese and resulting in 22 lbs of BOD and 6.5 lb of suspended solids per 1000 lb of cheese produced.

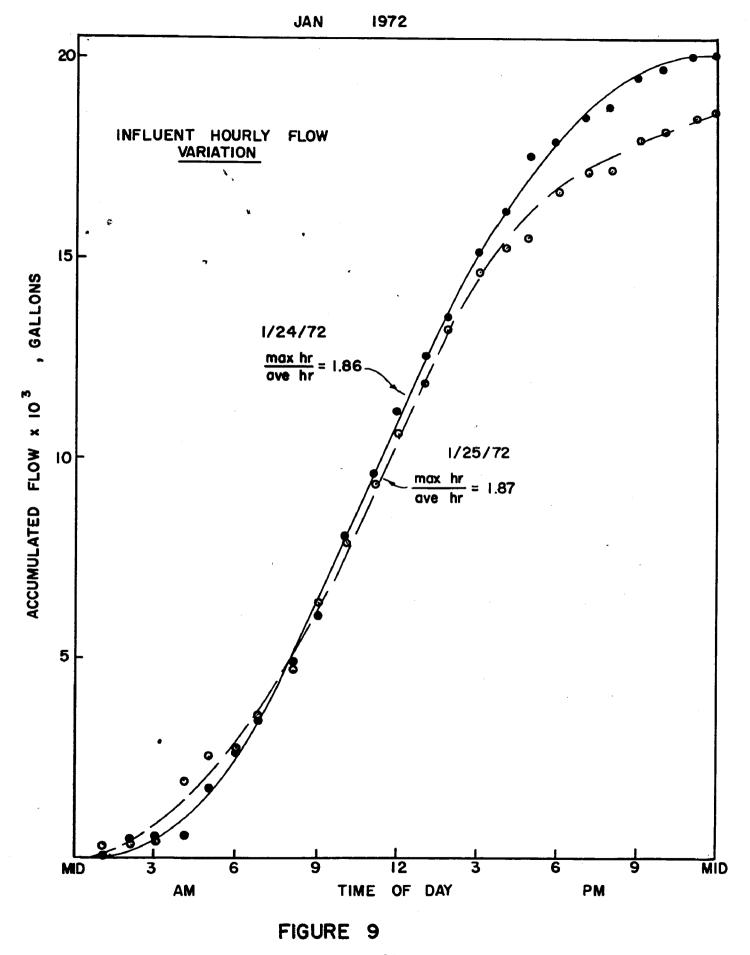
Biochemical Oxygen Demand - The performance of the lagoon system in removing biodegradable organic matter was measured by the five day BOD determination. Both total and filtered analyses were performed in order to reflect the influence of the pond system on actual biochemical stabilization as compared with physical separation of particulate organic matter. No long-term ultimate BOD analyses were performed nor were COD or Total Carbon Analyses; thus, the interpretation of the data in terms of mass balances of oxygen demanding materials through the system was limited.

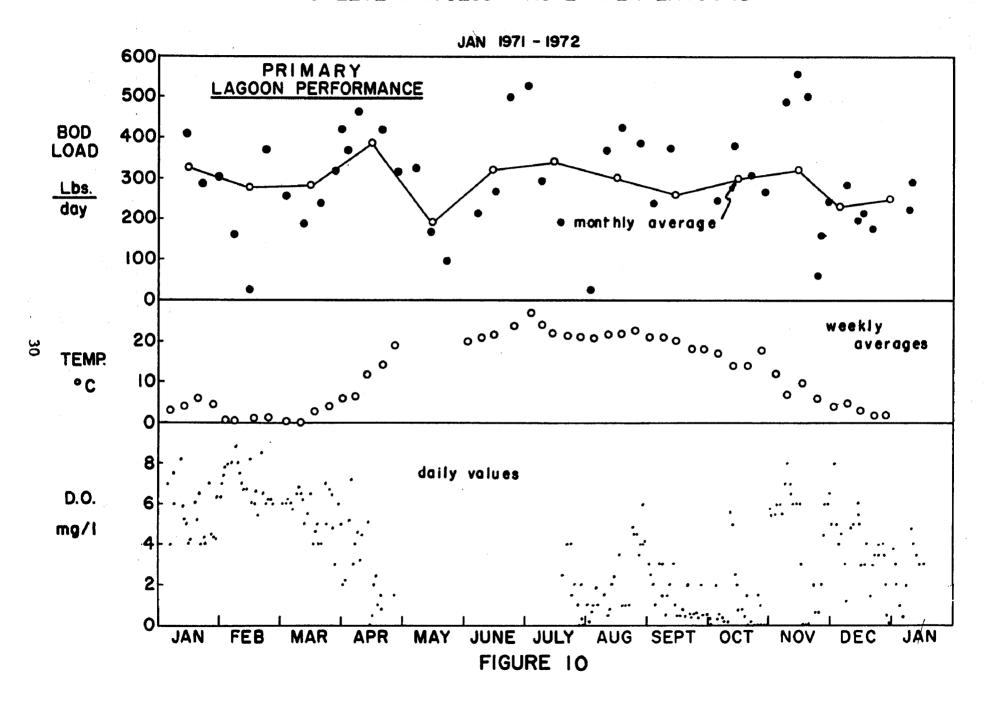
Although the BOD load to the lagoons varied widely over the one year period of study, the most apparent influence on lagoon performance was temperature. In Figure 10, BOD load and temperature are plotted along with the primary lagoon dissolved oxygen (D.O.) concentrations. It is immediately apparent that D.O. rapidly disappeared in the spring along with rising temperatures. In fact, D.O. reached 0.0 on April 10 and did not reappear until July 19. During the remainder of July, six of thirteen days the lagoon was devoid of oxygen. Also, 11 of 31 days in August, 15 of 30 days in September, 19 of 31 days in October, 6 of 30 days in November, and 2 of 31 days in December D.O. values were less than 1.0 mg/1.

#### CHEESE PROCESS WASTEWATER LAGOONS

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It might be assumed that this increased demand for oxygen during the spring was due to the increased biological activities brought about by the warmer temperatures. However, examination of the BOD load just preceding this period would indicate that the demands measured during this period could not be accounted for by the applied load.

For example, if one assumes that the ultimate oxygen demand is 1.47 times the 5 day value (deoxygenation constant,  $k_1 = 0.1/\text{day}$ ), the oxygen demand for a 5 day BOD load of 285 lb/day would be 419 lb/day. The oxygen demand measured in May, during the oxygen deficient period, was 1165 lbs/day, 746 lbs in excess of that predicted above. Furthermore, if one computes the oxygen supplied during this period based on a standard transfer rate of 4.0 lbs  $0_2$  per Kw hr., approximately 600 lbs of oxygen are transferred per day by this system at  $20^{\circ}\text{C}$  and zero D.O.

Therefore, it is reasonable to assume that significant amounts of biodegradable organics are being solubilized from the benthal deposits. During the cold months, anaerobic decomposition of the settled organic matter slows down resulting in accumulations of this material. As the temperature increases, biological activity also increases and solubilization of the benthal deposits begin to appreciably contribute to the organic load of the lagoon. This solubilization increases to a maximum value and then remains constant over the summer months. It may be assumed that this cyclic effect will continue, stabilizing the settled organic solids during the warmer periods and continue to produce both gas and soluble organic compounds.

The high oxygen demands occurring during the spring in waste stabilization lagoons has been attributed to benthal digestion by a number of investigators. Marais and Capri (8) have attempted to estimate the benthal demand in aerated lagoons. In a dynamic simulation of data collected for domestic wastewater lagoons, they showed that the rate of benthal decomposition was highly temperature dependent with an estimated value of the temperature coefficient theta of 1.35. They found that approximately 40 percent of the settled organic matter was resolubilized for the domestic waste.

The one year period of this study was insufficient to adequately quantify the benthal contribution to the lagoon organic load. The study period was too brief to evaluate the effectiveness of the anaerobic digestion in reducing accumulated sludge deposits. Steady state (or rather a quasi-steady state) may occur only after a number of years. Yet, based on this and other studies (2,8,9), it is reasonable to report that oxygen demands during the spring may exceed three times the applied BOD load.

Dissolved oxygen concentrations in the secondary lagoon were normally greater than 2.0 mg/l even during the April-July period of anaerobiosis in the primary lagoon. Only during two weeks in late June did dissolved oxygen levels drop below 2.0 mg/l and there was never a time when a value of zero D.O. was recorded. Dissolved oxygen concentrations were highest during the winter months, decreasing as oxygen uptakes increased in the spring and summer months.

The results of BOD analyses in the two lagoons appear in Figure 11 and average BOD-analyses by quarter are presented in Table 6. The removal of BOD was strongly correlated with lagoon temperatures and concomitant biological activity reaching a maximum late in the summer and early fall. Average BOD removal through both lagoons was 97.3 percent with the poorest performance occurring in the winter quarter with a value of 94.5 percent. During the 4 1/2 month period, July 1 through November 15, the effluent soluble BOD did not exceed 30 mg/1 and was 20 mg/1 70 percent of the time. Examination of the probability plots for total and soluble BOD in the primary and secondary lagoon effluents (Figures 12 and 13) indicates that the distributions of effluent BOD are skewed, being bounded at the lower end by the presence of organic matter relatively resistant to biological degradation. Considerable scatter of effluent BOD is demonstrated above the 50 percentile suggesting that, even with very long detention times, the reliability of the system to produce effluent BOD concentration less than 25 mg/l at all times is very improbable.

It is of interest to note that the soluble fraction of the effluent BOD represents on an average 50 percent of the total, with the highest values occurring during the spring and summer quarters.

Lagoon Modelling for BOD Removal - The mathematical modelling of biological wastewater treatment systems has become a popular method for compacting design data into a simple and useable form. Unfortunately, the biochemical, chemical, and physical reactions taking place in biological waste treatment processes are only poorly understood and the techniques of mathematical modelling normally become curve fitting processes (10). Most popular models currently in use for biological systems include the Monod equation (11, 12, 13, 14) and first order reaction models (15, 16).

Defining the appropriate model parameters for these mathematical models requires careful experimental design in order to examine the model over a rather broad range of the independent variables. In addition, the modeler must be aware of all independent variables which will influence the outcome of the reaction.

The study conducted and reported herein was not designed to provide this kind of information. The lagoons were designed to accept the flow and pollutant load that occurred and little or no control was available. Furthermore, funds were inadequate to make meaningful measurements of both dependent and independent variables. Therefore, it would seem inappropriate to try to develop any type of sophisticated model for this system. Furthermore, until one is able to define with considerable confidence, the independent variables which will provide a model with sufficient rigor to extrapolate process performance from one system to another, there is considerable hazard in presenting any type of mathematical expression which might be used in an effort to describe another system. Thus, by use of dynamic modelling, one may employ the data appearing in Appendix C to write a model for the experimental system described herein. Yet, use of this model to describe performance of any other similar



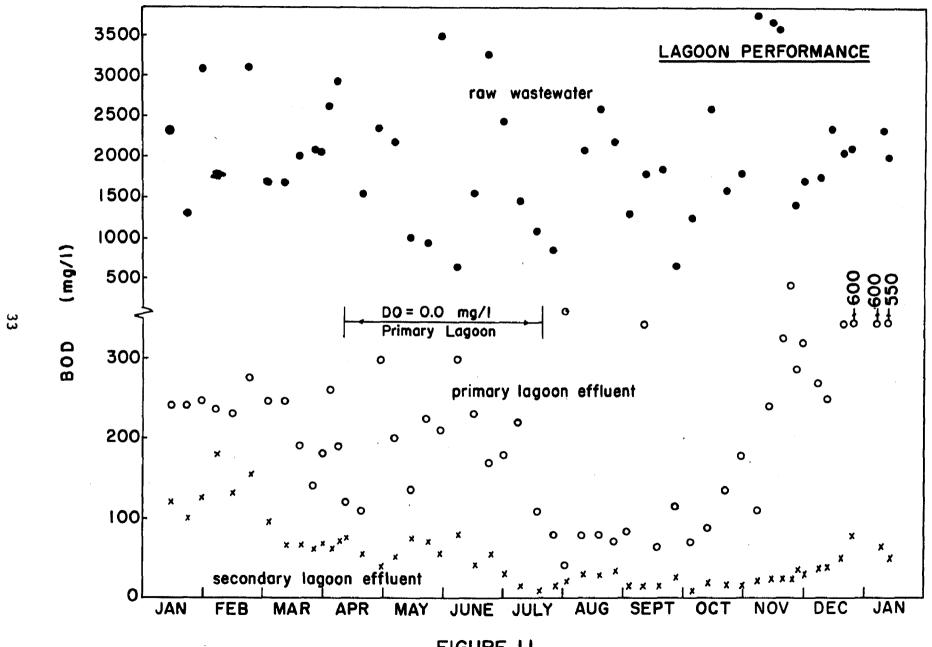


FIGURE 11

34

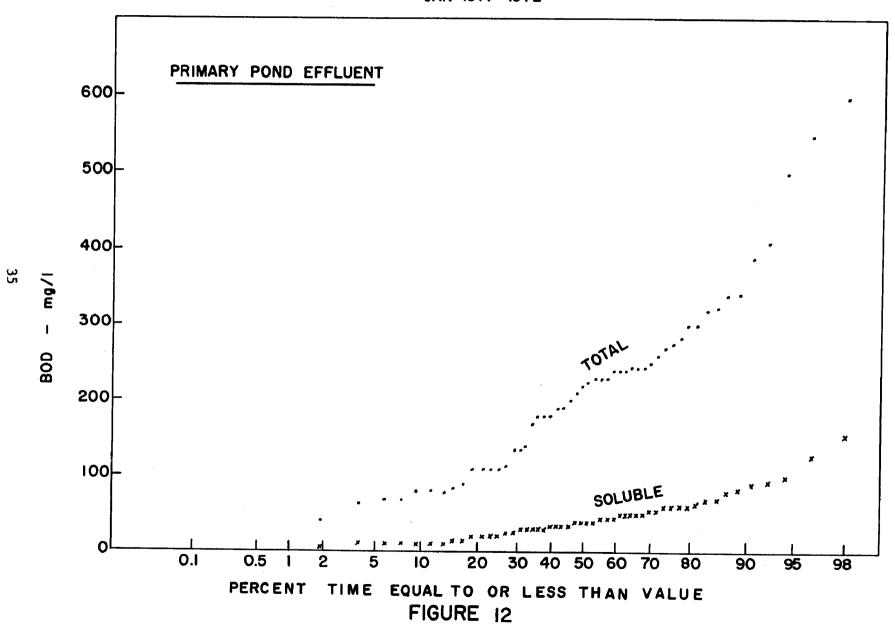
TABLE 6
AVERAGE BOD ANALYSES

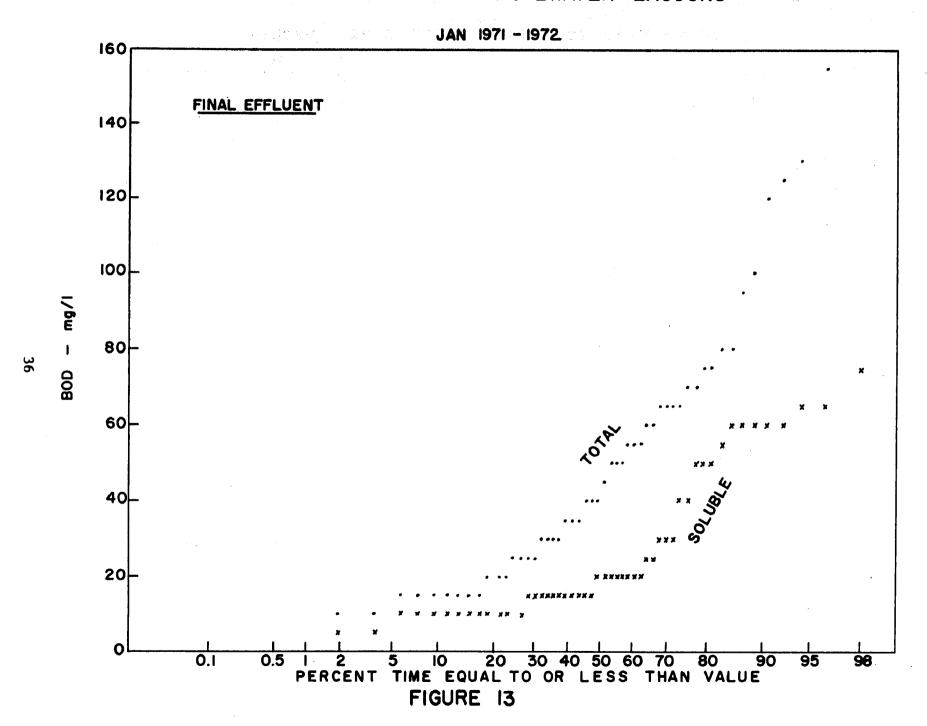
Quarter	Inf	luent	Pri	nary Efflu	ent		Secondar	y Effluent
1971	Total	Soluble	Total	Soluble	% R.* (Total)	Total	Soluble	% Overall R.* (Total)
Jan-Mar	1940	1060	224	49	88.5	106	51	94.5
Apr-June	2040	1110	204	45	90.0	61	32	97.0
July-Sept.	1530	870	122	37	92.0	21	13	98.5
Oct-Dec.	2100	1420	274	42	87.0	31	14 .	98.5
Avg.	1910	1140	209	43	89.0	52	27	97.3

All values of BOD in mg/l

\*BOD Removal based on total BOD

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system may be inappropriate unless the significant independent variables have been described.

The design parameters most often used for aerated lagoon design include BOD loading per unit volume, detention time, and BOD removal rate (usually expressed as a zero or first order kinetic parameter). For reasons stated above, a simplistic analysis of design parameters is provided. Table 7 summarizes by quarter the three design parameters most often associated with aerated lagoons.

Examination of Tables 6 and 7 suggests that BOD loadings, although increasing slightly during the summer months, are relatively uniform based on quarterly analysis. Day to day variations are of little importance in overall lagoon performance owing to the long detention periods. Note the significant increase in flow during the third quarter (July-Sept) resulting in a reduction in detention time of approximately 15 percent from the average.

The estimation of the first-order removal rate constant is of academic interest only. The assumption of completely-stirred conditions is probably reasonable for the soluble BOD fraction based on oxygen dispersion measurements, but total BOD undoubtedly is greatly influenced by the mixing patterns in the lagoon. No measurements were made to ascertain quantitatively the mixing properties of the lagoon system. A first-order reaction is also assumed, although there was no measurement made in this study to confirm it. First-order kinetics are commonly applied to lagoon systems (8,9,16). The magnitude of the removal rate constants are reasonable for this type of system. The seasonal influence on these rates are difficult to separate from the effects of increased flow rates. In any event, a temperature coefficient  $\theta_+$ , defined by the relationship

$$k_{T_1} = k_{T_2} \theta_t^{T_1 - T_2}$$
 (2)

was estimated for the primary lagoon employing the quarterly data presented in Table 7. By solving equation (2) for  $\theta$ , an average value for  $\theta$  for soluble BOD in the primary lagoon was 1.015 whereas a similar value for the total BOD was 1.012. These values are substantially lower than those normally reported for aerated lagoon systems (2, 16). Efforts to calculate  $\theta$  for the secondary lagoon were unsuccessful owing to the variation of the first-order rates calculated. It is important to note that the influence of temperature on biological processes is related primarily to changes in biochemical reaction rates. This can normally be detected through analysis of oxygen uptake rates, dehydrogenose activity, and other biochemical analyses. As stated earlier, volatile solids measurements in this study were meaningless indicators of biological activity owing to the high volatile solids of an inert nature occurring in the raw wastewater.

TABLE 7

LAGOON LOADING PARAMETERS SUMMARY\*

Quarter	Flow	1	Loading al/Solu Prim. Eff.		Volume Load Prim.		Hydrau Detent Prim.	ion Time	First-O BOD Re Prim.	moval	Temp.	Sec.
1971	(gal/day)		(1b/da		(1b/d/	' (1000f <del>1</del>	(D	ays)	(Day	s <sup>-1</sup> )	(*(	<b>C)</b>
Jan-Mar	14,900	270 143	$\frac{31.2}{6.7}$	$\frac{14.2}{7.2}$	$\frac{2.12}{1.12}$	0.24	64	64	$\begin{array}{ c c }\hline 0.12\\\hline 0.30\end{array}$	0.019	2.2	1.0
Apr-June	17,300	285 159	$\frac{35.9}{7.2}$	10.7 5.2	$\frac{2.24}{1.25}$	0.28	55	55	$\begin{array}{ c c }\hline 0.13\\\hline 0.39\end{array}$	0.042 0.007	17.8+	17.8+
July-Sept	20,000	300 169	26.9 8.4	$\frac{4.6}{2.8}$	$\frac{2.35}{1.33}$	$\frac{0.21}{0.07}$	48	48	$\begin{array}{ c c }\hline 0.21\\\hline 0.40\end{array}$	$\frac{0.100}{0.041}$	21.7	21.7
Oct-Dec	15,200	284 191	35.7 8.5	4.2 8.3	2.22 1.50	$\begin{array}{c} \underline{0.28} \\ \overline{0.07} \end{array}$	63	63	$\begin{array}{ c c }\hline 0.11\\\hline 0.34\end{array}$	0.119 0.0003	9.5	9.5
Ave.	16,900	285 166	32.4	8.4 5.9	2.24 1.30	0.25 0.06	57	57	0.14	0.05 0.005		

<sup>\*</sup>all values averaged from Appendix C

<sup>\*\*</sup>k =  $\frac{\text{Lo-Le}}{\text{Le}\theta}$  , based on BOD assuming first-order kinetics, CSTR.

<sup>+</sup>Estimated, based on estimated temperature for May of 20°C

Suspended Solids - The performance of the lagoon system with respect to suspended solids is considerably more difficult to quantify than BOD. Influent suspended solids were high (Table 8) and contained a large fraction of organic matter. Adequate data was not available on the settling properties of the influent solids and it is apparent from Figure 14 that suspended solids removals in the primary lagoon were very sporadic. During the oxygen deficient period, primary effluent solids often exceeded the influent solids level.

Again the advantage of staged lagoon operation is evident in Figure 14 where it is noted that final effluent suspended solids were considerably more attenuated. The probability plots (Figures 15 and 16) indicate that the effluent suspended solids are not distributed normally, being rather widely scattered at the high end of the distribution. The lowest values of final effluent suspended solids occurred during the period July 1 through September 19. In this period, suspended solids concentrations were less than 50 mg/l, 60 percent of the time. The average suspended solids concentration in the final effluent was 108 mg/l of which 91% was volatile matter (Table 8). It is interesting to note that the solids discharged were predominately volatile, especially during the July through December period. Furthermore, lowest effluent solids occurred during the period when greatest algal growth was evident.

The effluent discharge structure, as noted previously, was a vertical standpipe with no baffling. This discharge pipe was used to control lagoon elevation. Although no data was available, it is reasonable to assume that baffles may more significantly reduce suspended solids discharges.

Solids Contributions to Effluent BOD - The data presented above suggests that suspended solids play a significant role in the bio-degradable organic matter discharged from the lagoons. A simple mathematical expression for this contribution is:

$$L_e(total) = L_e(soluble) + C_v S_e,$$
 (3)

where S is the concentration of volatile suspended solids in mg/l, and L is the total or soluble BOD in mg/l and C is the fraction of the volatile solids contributing to the BOD (total).

Plots of effluent insoluble BOD versus volatile suspended solids (VSS) for both primary and secondary lagoons are presented in Figures 17 and 18. Although not well correlated, there is an indication that approximately 46 percent of the VSS contribute to the total BOD in the primary lagoon. This value is high indicating that the primary lagoon solids are not well stabilized. The contribution of VSS to the final effluent BOD is also quite variable, the average value of  $C_{\rm V}$  being 0.29.

60

TABLE 8

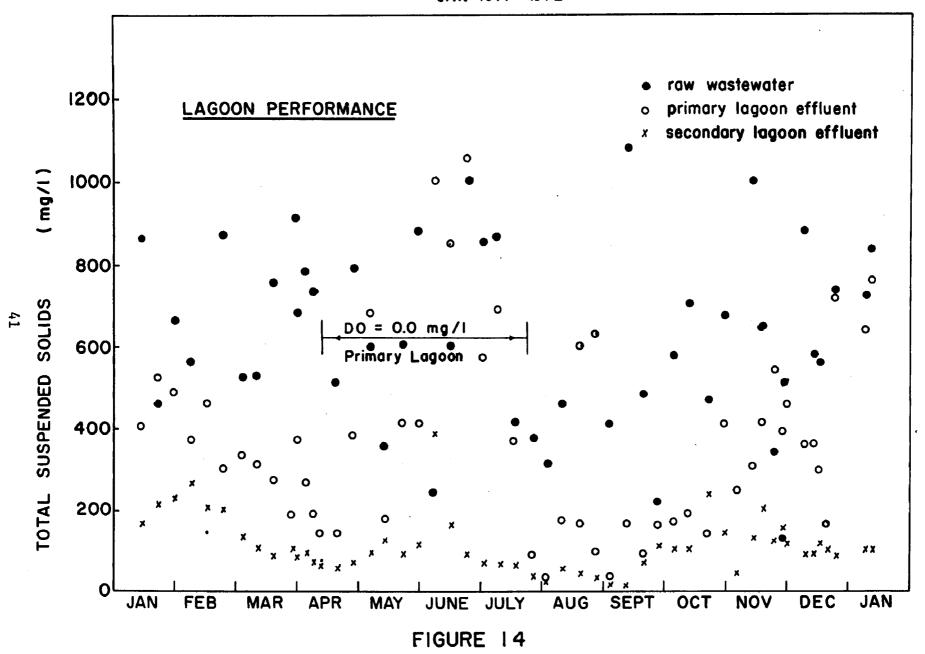
AVERAGE SUSPENDED SOLIDS ANALYSES

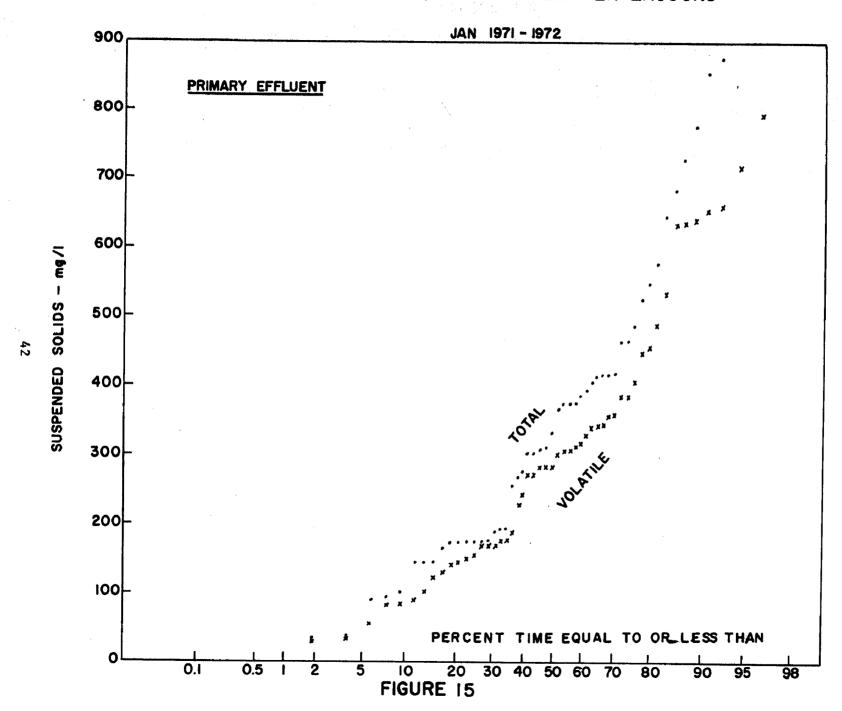
	Inf1	uent	Pr	imary Efflu	ent		Secondar	y Effluent
Quarter	Total	Volatile	Tota1	Volatile	% * (Total)	Total	Volatile	% Overall* (Total)
•	658	614	403	335	38.6	155	143	73.4
Apr-June	600	569	477	390	20.5	119	103	80.1
July-Sept	547	520	239	197	56.3	43	43	92.1
Oct-Dec	595	565	445	377	25.2	111	110	81.3
Average	602	567	395	328	34.4	108	98	82.0

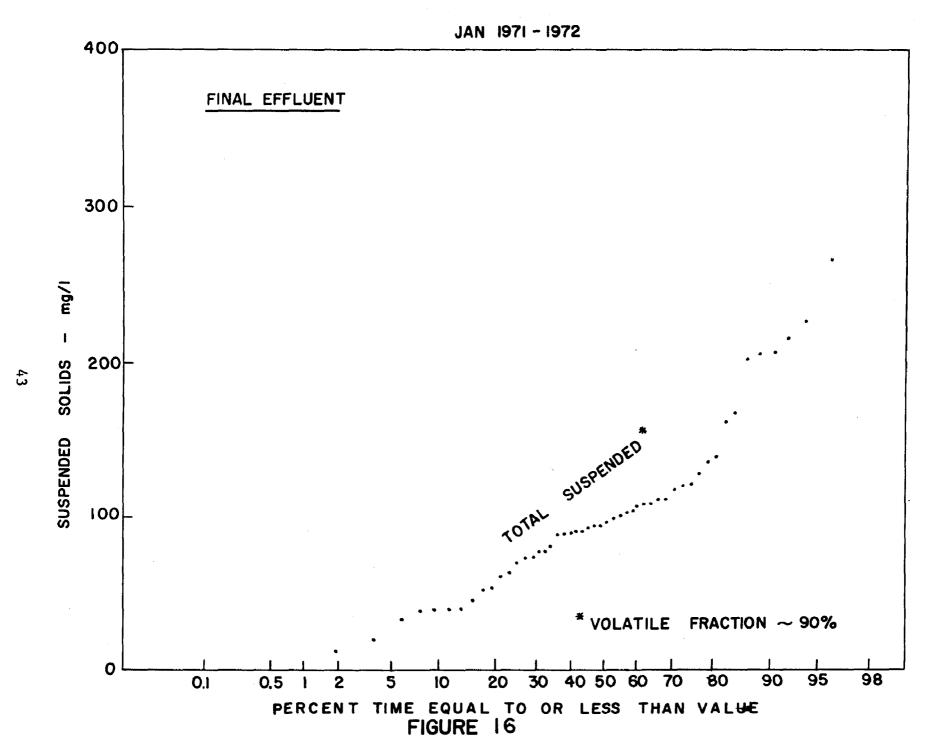
All values of suspended solids in mg/l

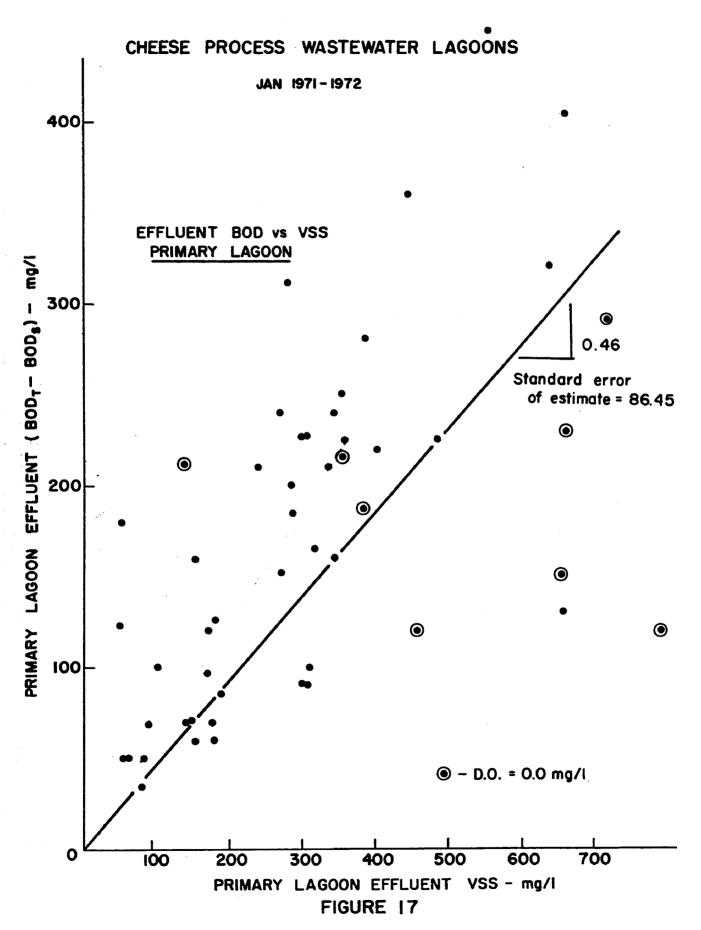
\*Percent removal based on total suspended solids

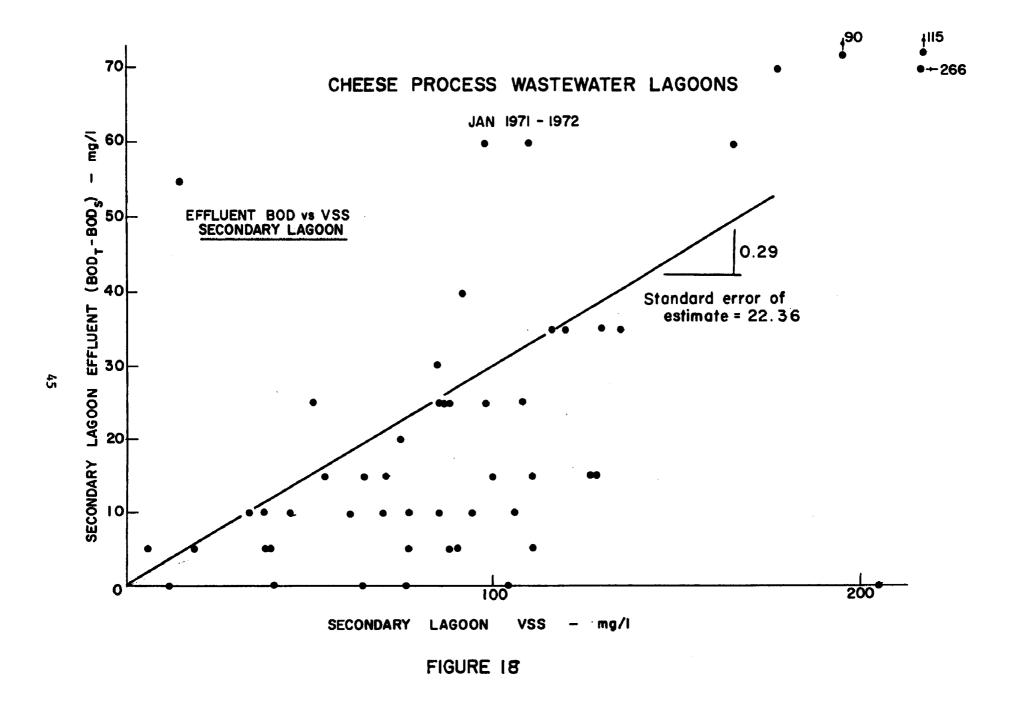
JAN 1971 - 1972











Nutrient Levels - The nutrients, nitrogen and phosphorus were determined monthly on 24 hour composite samples collected from the raw wastewater stream and the secondary lagoon. The results of these analyses appear in Table 9. Examination of Table 8 indicates that total phosphorus levels are high owing to the use of phosphate base cleaners. A substantial amount of phosphorus was removed in the pond system. It should be emphasized, however, that removal is the result of sedimentation of the phosphorus and that eventually substantial amounts of phosphorus may be resolubulized and be discharged in the final effluent. Thus, during the one year period, it is doubtful that a steady state with respect to phosphorus transformations has been achieved and, likely, effluent levels may rise significantly over those presently reported.

No data was available on the ammonia nitrogen levels in the system, but it is apparent that total Kjeldahl nitrogen is very low with respect to carbon. The BOD to Nitrogen ratio for the raw wastewater averaged 100:0.7, well below that normally required for adequate biological growth (100:5). The substantial removals of total Kjeldahl nitrogen through the pond system is indicative of both cell synthesis and sedimentation. It is expected that nitrogen recycle would play an important role in the biological system in these lagoons. Nitrate nitrogen in the final effluent was normally low, but warmer temperatures did result in significant nitrification between May and September. An examination of lagoon temperature during these months (Figure 10) indicate that when temperatures exceeded 20°C the rates of nitrification did increase.

Coliform and Fecal Coliform - The sanitary wastewaters from the cheese processing plant were discharged to the process sewers, thereby requiring the installation of chlorination equipment for disinfection of the final effluent. Both total and fecal coliforms were determined from 24 hour flow composited samples of the raw wastewater and final effluent preceding disinfection. Results of these analyses appear in Figures 19 and 20. During this study, no chlorination of the effluent was practiced.

Examination of Figures 19 and 20 indicate that better than 99.9 percent removal of the total coliforms and fecal coliforms was achieved in the lagoons. Temperature did not appear to significantly influence coliform disappearance in the lagoons during the survey period, although there was considerable variation in effluent coliform numbers.

Other Measurements - Values of pH and alkalinity were measured routinely throughout the one year study (Appendix C). Raw wastewater pH values were highest during January through May, averaging approximately 6.5, and then decreasing and leveling off to 6.2 for the remainder of the study. Similarly alkalinities of the raw wastewater averaged 330 mg/l during the January through April period and then continuously decreased to approximately 230 mg/l in September. It is possible that process changes of raw water quality accounted for these changes.

TABLE 9
NITROGEN AND PHOSPHORUS

Primary Lagoon Influent

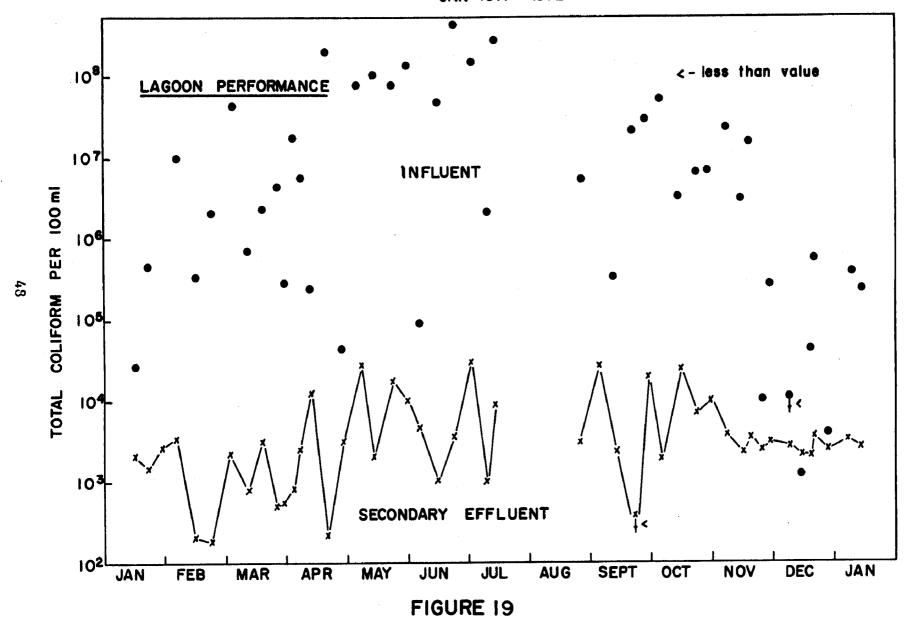
### Secondary Lagoon Effluent

Date 1971	Day	Total Phosph	orus	Total Kjeldal	nl Nitrogen	Nitrate Nitrogen	BOD:N:P**	Total Phosp	horus		Kjel- Nitrogen	Nitrate Nitrogen
	<del></del>	mg/1	1b/d*	mg/1	1b/d*	mg/1		mg/1	1b/d*	mg/1	1b/d*	mg/l
	F	37.0	8.5	7.5	1.7	0.1	100:0.6:0.3	35.0	8.0	4.0	0.9	0.3
2-23	Tu	35.0	4.1	9.2	1.1	1.0	100:0.3:1.1	27.5	3.3	5.9	0.7	0.5
	F	41.3	4.9	10.8	1.3	0.1	100:0.5:2.1	26.4	3.1	5.5	0.7	0.2
	м	7.4	0.7	10.2	1.0	0.4	100:6.8:4.9	28.0	2.8	6.0	0.6	0.4
	F	32.5	5.4	5.3	0.9	0.4	100:0.5:3.2	26.8	4.5	10.6	1.8	1.9
		76.9	13.2	16.0	2.7	4.4	100:1.0:5.0	10.7	1.8	4.2	0.7	24.4
		15.5	3.1	12.2	2.4	0.1	100:0.8:1.0	14.8	2.9	1.5	0.3	3.4
	Tu	33.0	5.8	3.6	0.6	0.1	100:0.2:1.6	11.6	2.0	1.0	0.2	4.0
		68.4	14.0	12.1	2.5	0.1	100:0.7:3.8	8.4	1.7	1.1	0.2	1.4
		55.7	8.2	2.0	0.3	0.1	100:1.1:2.2	14.8	2.2	0.2	0.0	0.4
		04.6	14.4	26.2	3.9	0.1	100:0.7:2.6	23.1	3.5	1.8	0.3	0.4
	Th	42.6	5.7	7.7	1.0	2.7	100:0.5:2.7	31.6	4.3	4.2	0.6	0.2
Avg.		45.0	7.3	10.2	1.6		100:0.7:2.2	21.6	3.3	3.8	0.6	

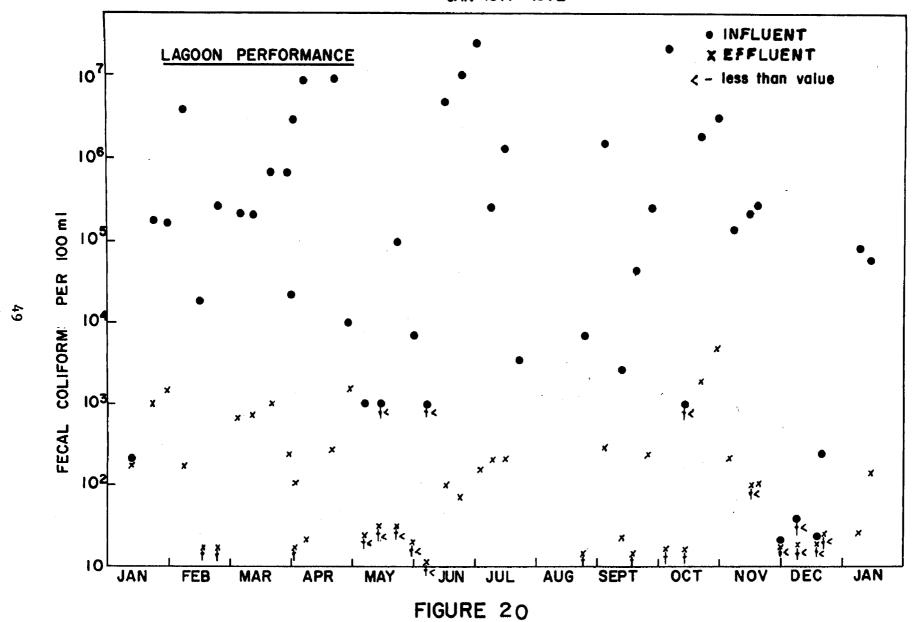
<sup>\*1</sup>b/d - Calculated by using flow rates on day parameter was measured

<sup>\*\*</sup> BOD:N:P - ratio of weights (or concentrations) of each component on day of analysis

JAN 1971 - 1972



JAN 1971 - 1972



The lagoons exhibited pH values of approximately 7.4 during the first one half of the year (January through June), which then sharply decreased to a value of about 7.2 for the remainder of the study. Alkalinities ranged from about 310 mg/l to as high as 540 mg/l during the year. No correlation was noted between pH and measured alkalinity and there was no particular trend in alkalinity values during the year. No correlation was noted between pH and measured alkalinity and there was no particular trend in alkalinity values during the year. It is assumed that changes in algal activities and microorganism respiration influenced these variations, but no data was available on algal activities to document these variations.

Settleable solids were monitored daily on the raw wastewater and on the primary and secondary lagoon effluents (Appendix C). Settleable solids' in the raw wastewater ranged between 0.1 and 5 ml/liter during the one year study, normally being less than 1.0. Secondary lagoon effluents never demonstrated significant amounts of settleable solids, but the primary lagoon effluent was high in settleable solids from mid-May through mid-July. Values as high as 120 ml/liter were recorded during this period when dissolved oxygen concentrations were zero. Floating mats of solids and rising sludge was noted throughout this period. Once dissolved oxygen was maintained in this lagoon, the settleable solids decreased to an immeasureable amount.

#### Costs

The estimated annual capital costs for wastewater treatment at Kent Cheese Co. are based on an interest rate of 7 1/2 percent for mechanical and equipment amortized over a five year period and costs for lagoon construction over a 30 year period. Whereas the estimated annual costs for operation and maintenance are based on costs associated with operating labor, electric power, repairs and replacements, supplies and administration. The annual costs presented do not include those costs associated with the post construction studies that are not representative of normal operating expenses. A summary of the costs related to annual operation for the period January 18, 1971, to January 17, 1972, of these treatment facilities are presented below.

#### Construction Costs

		Annual Costs
Contract	29,097.42	
Engineering	•	
Preconstruction Report	500.00	
Preparation of Plans	3,387.14	
Construction Supervision	916.00	
sub total	\$33,900.56	
(i = 7.5% 30 yr)		\$ 2,870.40
Mechanical Equipment	15,592.82	
(i = 7.5% 5  yr)	13,372.02	3,853.99
sub total		\$ 6,724.39
		, , , , , , , , , , , , , , , , , , ,
Operating and Maintenance Costs		
Operator labor	\$ 1,364.83	
Electric Power (2.676¢/kwm		
x 92575 kwhr)	2,477.71	
Repairs and Replacement	1,725.43	
Supplies (expendable)	347.47	
Administration	144.62	
sub total	6,060.06	\$ 6,060.06
Total Annual Costs		\$12,784.45

The estimated costs associated with the treatment of wastewaters for the period January 18, 1971, through January 17, 1972, are based on the interest rates and amortization periods selected for fixed costs, whereas the operating and maintenance costs are those costs actually incurred for the period of operation indicated. One item in the Repairs and Replacement costs category; that is the replacement of a single blower with associated costs, accounted for all but \$134.55 of the \$1,725.43 indicated in this category. Normally one would expect a longer period of performance for this type of equipment.

The costs prorated to pounds of BOD applied or 1000 gallons of wastewater treated are \$0.134 per pound of BOD and \$2.05 per 1000 gallons respectively. Based on cheese produced, the estimated cost is \$0.0032 per pound.

#### SECTION VII

### **ACKNOWLEDGMENTS**

This study was supported by the United States Environmental Protection Agency, Office of Research and Monitoring, and Kent Cheese Company, Inc., Kent, Illinois.

William C. Boyle, Ph.D., and Lawrence B. Polkowski, Ph.D., were contracted as research consultants and authors after the treatment plant was in operation and thereafter were responsible for the interpretation of data collected during the post-construction studies and for writing this report.

Lorne Gramms, Ph.D., Tom Jensen, E.I.T., and Jack Quigley, P.E., were employed as part-time assistants to the project.

Mr. Allen Fehr, P.E., Freeport, Illinois, designed and supervised construction of the treatment plant.

E.P.A. Representatives included William J. Lacy, H. G. Keeler, Robert Burm, and Max Cochrane.

Laboratory analysis were performed by Robert Corning, Chief Chemist, and Mrs. Susan Kelly, Laboratory Supervisor, Corning Laboratories, Cedar Falls, Iowa.

The Assistant Operator, plant operator, was Michael Green, Kent Cheese Company, Kent, Illinois.

Mrs. Kathy Watson and Mrs. Barbara Daul, Lena, Illinois, served as administrative assistants.

#### SECTION VIII

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#### SECTION IX

### GLOSSARY OF TERMS

Alpha,  $\alpha$  Ratio of overall oxygen mass transfer coefficients in waste to tap water

Beta,  $\beta$  Ratio of oxygen saturation value in waste to tap water

BOD Biochemical Oxygen Demand, Five-day, 20°C

C Oxygen concentration in water, mg/1

C Oxygen saturation concentration in water, mg/1

C<sub>v</sub> Fraction of VSS contributing to BOD

CIP Clean-in-place

cu ft Cubic foot

d Day

DO Dissolved oxygen concentration

ft Feet

fps Feet per second

gal Gallon

gpd Gallons per day

hp Horsepower

k First order removal rate of BOD, 1/days

k<sub>1</sub> BOD deoxygenation constant

kw Kilowatt

kw-hr Kilowatt-hour

L Five-day 20°C BOD of influent, mg/1

L Five-day 20°C BOD of effluent, mg/1

pound 1b MG Million gallons mg/1Milligrams per liter N Nitrogen  $N_{f}$ Field oxygen transfer efficiency, 1b 02 kw-hr Standard oxygen transfer efficiency at 20°C, zero D.O. in Ng tap water, 1b 02 kw-hr O/R Over-the-road trucking Percent concentration of oxygen in air leaving lagoon, % 0<sub>E</sub> P Phosphorus Absolute pressure at point of air release, 1b/in<sup>2</sup> P<sub>b</sub> Q MGD Flow rate, R Percent removal of pollutional constituent Sຼ VSS Concentration SS Suspended solids Sec Second Scfm Standard cubic foot per minute Detention time, V/Q, days θ Temperature correction coefficient θ, T Temperature, °C Volts v Volume VSS Volatile suspended solids

SECTION X

APPENDICES

APPENDIX A

OXYGEN UPTAKE RATES
Primary Lagoon

Date	Station (Figure 4)	Depth from water surface (ft)	Temp °C	Uptake Rate (mg/1/hr)
11-19-70	1	<b>5</b>	6.3	1.9
	1 3 5	5	6.3	2.3
	5	5 5	6.2	2.1
	13	5 5	6.2	2.0
	14	5	6.3	1.9
5-23-71	1	1	17.1	5.0
	1	5 1	17.1	5.8
	1 1 5 5	1	17.1	6.3
		6 2	17.0	6.0
	14	2	17.1	4.9
7-21-71	1	6	24.0	1.2
	1 5 8	6	24.1	1.2
	8	6	24.0	1.3
	13	6	24.0	1.1
8-26-71	3	6	23.5	0.9
	5	6		1.0
	12	6	23.5	0.8
	14	6	23.5	0.6
10-7-71	1	2	16.5	1.6
	14	6	16.5	1.4
,	8	6	16.5	1.6

Calculation of Oxygen Transfer Rate November 19, 1970

Lagoon No. 1 (Primary Lagoon)

Average Oxygen Uptake Rate = 2.1 mg/1/hr = 16.7 lb/hrLagoon Temp. =  $6.3^{\circ}$ C Dissolved Oxygen = 2.9 mg/11 compressor - 30 amps at 220 v = 6.6 kwAir Distribution to Primary Lagoon - 80% estimated

Field Transfer: (17, 18)

$$N_f = \frac{16.7 \text{ lb/hr}}{6.6 \text{ kw x 0.8}} = 3.16 \text{ lb/kw-hr}$$

Standard Transfer:

$$N_{s} = N_{f} \frac{C_{s}}{(C_{s} - C) \theta} T-20$$

$$C_{s} \text{ at } 20^{\circ}C = 9.2(\frac{P_{b}}{(29.4} + \frac{0_{t}}{42}))$$

$$= 9.2(\frac{14.7 + 12/2.3}{29.4} + \frac{18.3}{42})^{*}$$

$$C_{s}_{20^{\circ}} = 9.2(0.68 + 0.5)$$

$$= 10.8 \text{ mg/1}$$

$$C_{s}_{6.3^{\circ}C} = 14.8 \text{ mg/1} (2.12^{\circ} \text{ depth})$$

$$\alpha = 0.75, \beta = 0.96, \theta = 1.02^{(19)}$$

$$N_{s} = 3.16 \frac{10.8}{0.75(0.96 \times 14.8 - 2.8) \cdot 1.02} 6.3-20$$

$$N_{s} = 5.25 \cdot 1b/kw-hr$$

$$N_{s} = 3.91 \cdot 1b/hp-hr$$

\*where depth = 12' and estimated transfer efficiency is estimated at 10%

## APPENDIX C

Lagoon Data

										W	EE	KL	Υ	D/	<b>4T</b>	\							,		
DA	TE	DAY	LAGOON	ВО	D <sub>5</sub> (T	OTAL/	OLUB	LE)		TOTAL	SUS	PEND	ED S	OLIDS	3	VOL	ATILE	SUSPE	NOE	SOL.	IDS	TOTAL CO	LIFORMS	FECAL CO	OLIFORMS
	1	Ì	MFLUENT	S, F	? 1	3. F	2	S. P.		S. P.			? 2		? 3	8.1	•	8. F			? 3	5. P. I	S.R 3	S.P. I	S.P. 3
<u> </u>			(GPD)	<b>M6/</b> L	410	14/L		120	LO/D						LB/D			46/L			LE/P	NO./ 100		MO. / 10	
止	14	TH	21,000	1140	200	30	41.0	100	21.0 105 22.9	864	151	405	71	167	29.2	846	148	340	515	147	29.2	280	200	47,000	2400
1-	22	F	27,5∞	330	286. 15.5	20	55 4.58	3c	4.67	460	105	525	120	216	49.5	434	99.5	405	92.7	118	40.9	180,000	1000	120,000	1333
1-	<u>30</u>	১	11,800	1860		245	24.47	30	12.3	612	64.1	488	48,0	226	222	664	65.3	488	48.0	218	21.5	210,000	1500	× 106	3100
2	-7	Sul	10,300	1800	103	130	19.89	180	15.59	569	48.9	372	32.0	2466	22.8	540	46.4	359	8,05	266	22.8	5,400,000	200	10,100,00	5000
12	-15	М	13,200	190	11.5	230 50	3.50	130	17.3 8.25	146	16.1	462	50,8	205	22.6	24	2.64	56	6.16	15	1.65	17,000	HONE	450,000	280
2	23	T	14,200	3130	3-6	275	32.5 5.90	155	7.67	878	104	301	35.5	202	23.8	854	101	301	35.5	197	23.2	305,000	0.53	4,200,000	200
3	-3	W	17,500	1730	184	45	35.8	95 60	13,9 8,76	523	76,4	334	48.8	135	197	498	12.7	286	41.8	135	19.1	350,000	150	క్రద్దియయ	3,250
13	-11	TH	13,3∞	1685 1185	131	155	17.2	45	7.12	529	58.7	309	34.3	104	11.5	490	54.4	309	34.3	104	11.5	330,000	275	840,000	830
2	-19	F	14,300	1000		190	24:3	65	7.8	154	89.9	273	32.6	89	1C.6	105	84.1	273	32.6	89	10.6	800,000	looo	2,100,00	.4000
3	-27	5	18,000	2100 930	315	140	8.3	ص 20	9.0 2.5	915	137.4	190	28.5	100	15.9	768	115.3	190	28.0	156	15.9	<del>8</del> 00,000	330	6,600,000	610
3	-31	W	25,000	1230	433.7	180	37.5	65 55	م.ن.3 کیلا	<b>68</b> 0	141.8	172	35.9	89	18,6	614	128.0	172	35.9	85	17.7	30,000	100	33U 000	566
4	-4	74	14,700	2640 1500	34.2.2	2440 50	36.2	60 50	8.4	784	109.2	269	37.4	90	12:5	613	93.7	244	34.0	11	10,7	4,300,000	HONE	21,000,000	825
	-8	Ħ	19,300	2900 1630	466.8	190	30.0	100	11.3	139	1190	194	31.2	10	11.3	698	112.4	171	27.5	10	11.3	9,300,000	20	Secon	4000
	1-12	3	11,800	4 150 4 150	414.8	1150	19.8	75 49	749	85	8.4	144	14.2	73	1.2	15	7.4	130	12.8	45	6.4	NONE	NONE	270,000	1200
[4	-20	Т	32,500	1550	4120.1 230.1	110	19.8	35	14,4	513	139.0	143	38.8	54	14.6	485	131.5	123	33.3	54	14.6	9,600,000	<b>53</b> 0	116 × 106	200
	-28	£	14,200	1200	317.5	300	40.5	40	5.7	197	100.7	386	52.2	24	190	182	105.7	348	17.0	74	100	N <sup>0</sup> 000	1700	56,యు	4000
	56	되	17,500	1220	324.6	200	29.7	50 25	7.4	598	883	(હ્8ા	101.1	91	14.4	518	85.8	632	93.8	85	12,6	1000	<b>433</b>	11 000 000	<b>37,∞</b> 00
	7-14	П	20.000	1000	93.4	135	22.5	15 15	12.5	353	58.9	178	29.7	121	20.2	341	57.9	113	28.9	110	18.3	41000	<b>4</b> 50	84,000,000	2200
	7-22	3	11,600	930	9.7	325	23.8	38	4.8	<b>608</b>	58.8	415	40.1	92	8,4	584	56.7	388	37.5	92	8.9	∡ /∞,∞∞	<b>433</b>	හිට පත ගත	18,000
	5-30	3UN		3500		210		55 20		<b>9</b> 60		411		117		854		318		117		1000	£ 20	128 × 106	∡10,∞0
	-1	3	43.100	570		300	107.8	<b>8</b> 20	26.8 ३.५		37.7	เดอา	3420	390	140.2	197	70,8	717	257.7	266	95.4	41000	<b>4</b> 1	94,000	6,000
_	-15	$\overline{\tau}$	20,600	1550	1976	230	39.5	40	6.9	605	103.9	857	147.2	<del></del>	28,0	569	97.8	636	109.3		22,3	2000 000	/00	15,000,000	1000
	.23	*	18400	3250	498.7	170	240.1	55 50	8.4	1003	1539	1062	/63.0	90	13.8	981	150.5		/22	90	13.8	(O,∞∞,00	<i>8</i> 0	6/0×106	6000
	1-1	TH	25,900	1800 1800	317.9	160	38.9	30	6.5	860	186,8		124.4	69	14.9	833	179,9	456	98,5	51		28×106	110	170×106	42000
	1-9	F	23,600	1470	108.3		43.3	15-	3,0 3,0	691	134	879	173	U	13.2	527	103.7	653	118.5	40	7.9	100 000	200	2,600,000	1000
	7-17	5	21,700	10 <b>8</b> 0		110	19.9	10	1.8	419	15.8	372	67.3	64	116	392	10,9	312	56,5	64	11.6	1500,000	200	430×106	8600
	1-25	<b>327</b>		938		90	1	15		375	1-2-	90		39		369		90		39					
	3-2	3	30,000	100	25.0 18.8	40	10.C	20	5.0	313	18.3	-33	8.3	33	8.3	304	76.1	33	8,3	33	8.3				
	3-10	Τ	21,100	2013	300.5	မ်း	17:18	36 20	5.4	463		179	31.5	GI	10.7	456	801	119	31.5	61	19.7				
	3-18	8	19.700	75 40			13.1	30	49	601	98.1	121	28.1	45	7.4	601	98.7	150	24.6	45	7.4				
		HΤ	21,000	2190		10	12.3	35	4.4	634	111.0	1	17.5	37	65		105,8		14.5	31	6.5	7000	5	7,300,000	3000
	1-3		21 800	1300	234.4	B <sub>2</sub>	15.5	15	2.7					19			15.1		69	19		1,330,000		1280×10	29800
_	1-1	5	24,600	1800	369,3	3-10	69.8	15	3.1	/084	223,-			115	3.1		217.1		30,2	11	2.3			140,000	
	7-19			1850	1	65		15		189		93		13	<u> </u>	457		81		39		61,000		21 × 106	
_	1-27	3	33,900	475	190.8	115	31.5	25	7.1		64.2	167	47.2	113	31,9		59.4		40.1		22.9	380∞		30×106	
	2-5	7	23,400	12.50	243.9	125	13.7	10	2.0	581	113.4				21.1					16		16× 106		5100000	
		¥	23,400 17,700	7325	380.9	138	13:3	2¢ 15	3.0	211	105	195	28.8	109	Ko.1	691	102,0	180	16.6					45 1106	

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DATE	DAY	LAGOON	BO	Ds	(TOTA	L/SOL	UBLE	)	TOTA	L SU	BPEND	ED 8	OLID	B	VOLA	TILE :	SU SPE	NDED	SOL	D\$	TOTAL CO	LIFORMS	PECAL CO	<b>ALPONNS</b>
1		INFLUENT (GPD)	8.0	1	] <b>8</b> . F	2	8.R	3	\$. Ji		8.P.	2	S.R	3	8.6		S.R.			3	S.R I	8.R 3		8.R 3
4				13/2	Hey	18/0		7840		4							147L				NO./ 10		HO. 7 10	
10-21	TH	23200	1000	173.5	35	6.8	15	2.7	471	91.1	146	18.2	2-46	41.6		26.9		19.7	206		1.2×10	3200	7.3 . 10	
10-29		17.900	330	130.8	20	3,0	100	2.2	<b>ن</b> 18	101.2	115	<u>u.o</u>	MA	22.2	८इ५		345	51.5	111	16,6		6200	8×06	
11-6	5	15,400	2.00	105	110	14.1	15	1 000			252	32.4	44	5.1	874			274	٥	6.9	125,000	200	22×10	<del></del>
11-14		18,200	25.00	357 6 450 6 300 5 300 5	35	34.5	25	7.9	1011	1535		47.2	131	19.9	786		281	420	111	8.0	170,000	<b>∠ 100</b>	4.9 x 106	
11-18		15,015	<u> 2400</u>	300 5	25	17.2-19.3-1-19.11 19.2-19.3-1-19.11	15	3.1	659	82.S		52.3	206	25.8	<u>७अ</u>	79.4		33,9	128	16.0	290000	<u> ∠ 100</u>	13×10	5400
11-24		15,700	320	35.1	36	3.4	115 225	19-10-20	348	45.6		71.8	128	Ko.8		24.9	448	58.7	94	12.3	4200	4 100	10,000	2300
11-26	F	4,200	1200	Meri	35	4.0	20	لية	132	15,6	394	46.7	156	18.5	124	H.7	330	39.1	138	Va, 3	4 100	130	4 100	2700
11-30	工	17,000	1550	241.0	40	5.1	15	4.5	524	74.3		Wal	122	17.3	502	71.2	388	55,0	100	14.2	420	420	270,000	4000
12-8	W	19.310	1330	F. F.	~73	43.5	13	3.5	884	W2.4	368	59.3	8	15.5	874	1-10-8	300.	42.3	88	14.2	450	420	450	4400
12-13	M	9615	1350	218.8 199.4 189.5 149.5 210.2 171.8	250	10.1 15.6 1.0	40	3. L 6.7 500	582	41.0	376	30.3	94	7	560	452	340	79.0	පිරි	21	410	410	1200	2100
12-K	TH	14,155	1940	2 vo. 2	340	45.8	50 15	50,0	556	14.9	302	40,7	120	W.2	516	69.5	284	38.3	120	Va.2	410	410	66,000	2,200
12-20	M	10.325	1220	132.3	- SE	31.7	45	3.4	176	15.2	1580	136.1	108	9.3	176	15.2	1320	113.7	801	9.3	175	120	(000,000)	6,500
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1-12	W	17,415	1780	240.5	410	37.5	50	7.3	846	122,9	172	1121	108	15.7	816	118,5		300	84	12.2	54 000	120	200,000	35∞
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### MONTHLY OPERATION REPORT

	=	Ι.	PLANT PROCE	rss		Boiler	LAGOON	١,,		SETTLE	LABLE	SOLIC	78	pH4		ALKALIN			T	LAG	90N	AGO	OM V	VEAT	HER.
Comments	ă	À	Milk Prets.	Cheese Prots.	Plant	FLOW	INF. (CALC.)*	Ĭ	E E	,	Sample: Z	3	1	Sample 2	• 3	١,	Samples Z	3	5 g	2 5	É	251	u ke	in. E	É
JANUARY 1971	١.		lbs.	the.	Well (GPD)	GPD	` GPD	¥	0 2	m4/1	mi/1	""1/1		1	1	Mg/1	mg/1	IP92/1	•	֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	-	֓֟֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓֓	"	"  *	•
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-1007	7	T		5,702						2	١٠					380		440				11 3		$oxed{\mathbb{L}}$	$\Box$
2nd CEIL FROZEN OVER		F		12,319						1	٠١	_	7.0	7.1	7.4	370	444	470		٦.5	36	9.3	И		
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D.O. CHECKED FOR		W		16,307	28.500	2000	27 200			-	1	_	6.8	<i>1.3</i> 6	7.6	376	430	466		52	40	1.0 3	ธ	$\perp$	
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		۴		15,980	22,600	1800	51,400	<u> </u>		١	-	_	69	7.1	7.55	374	430	Aldo		_		7.63		1	$\perp$
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	_	7		5,868					Ш	١.		_	<u>د.5</u>				440	470				.03		$\bot$	Ļ
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	-	S		11,606	51.5∞	2100	WP PI		Ш	2,5			<u>د.ع</u>	7.4	7.5	320	436	Aldo		1.0	12/	603	7	$\bot$	L
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	+	T		13.404					Н	١٥			66			242	430	486		15	O)	e.53'	2	+-	<u> </u>
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LOW SOLIDS - NO	+	F		13,565					$\vdash$	.2		_=				326		484				7.03		₩	<u> </u>
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### MONTHLY OPERATION REPORT

	L	T	PLANT PROCESS		Boiler	LAgoni	A	SETTL	ZAGLE	BOLIE	-	pH		ALKALIN	TV			1440	ONE	ABOI	N W	EATH	ER -
Commonts .	Į	Ì	Mills Freits. Chasse	Profit. Plant	FLOW	(	1	ď.			ĩ.	2	3	,	Sompton 2	3	a					<b>~ 1</b>	1
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		E		xe.21 50					<del>  -</del>	-				296		480	<u> </u>	20				╀~	10
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	T_	Т	PLANT PROCESS		IBox -	. 1 4			TTLEA												<del></del> -			THER
Comments	ĮĘ	ž	Milk Prets. Cheese Pre	Its. Plant ,	Flow	LAgon	AR F		Sec	mples		1.	Sample	• .	ALKALIN	Semples	_	١.	1	١.	1 -	100N		_ 4
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	3	12	12.038	<u>3 18,8∞</u>	2100	17,500		\	1.5	_	_	6.5	7.4	7.5	316	364	380		6.1	32	6.0	32		8
	•	₽,	10,513	15,300	2100	14000			1			6.7	7.45	760	300	352	380		60	32	60	32	$\Box$	0
THAWING ICE		F	H 73	17,100	1800	16,100			٦.	_	_	6.7	7.5	7.5	300	صد	380				60		1	۵
" "		5	11,772	15,300	2100	How		1	2	_	_	6.6	7.4	7.5	296	302	384		5.7	32	6.1	32		ام
J	7	3		* *																Г	Г	П	T	T
THAWING ICE	_	M	4.18	20,000	200	19,300			۱۰	_]	_	6.5	7.4	2.5	290	374	390		6.5	32	6.5	32		0
11 11	•	7	14.80	4 15 322	2400	13700		$\Box$	ī	-	_	6.1	75	745	320	370	38%		68	32	65	32		5
11 11		144		OUTH P				$\Box$	1	-	_	_	75		3-00	370	388			-	6.3	_	2	
11 11	•	7		7 14.600				$\Box$	.6	_1	_		7.45		294	370	370	_	_	-	6.3	_	3	
" "	12	۴	15,29	7 14,500	1800	13.500			1	-1	_	64	7.5	,	294	3714	390				٤2	_	3	_
THAWING ICE		5	11,40	4 15,800	2100	H.500			2 ·		_	65	74	7.4	300	Ore	390				6.0		3	
	14	3		**		Ī		1														$\Box$	「	
THAWING ILE	15	M	Ko CA	25800	2100	24.300		Ι.	[	-1	_	'o.4	745	75	2%	374	2700		6.5	34	68	32	1	5
" "		上	1733	ص ۱۳ س				1	.5	_ [	-	6.4	7.4	7.5	298	370	366				55		24	<b>₁</b>
" "	_	W	1331	3 17500	2000	الدكص			1	[	-	6.3	745	75	305	370	صد		16	37	5.4	32	18	
		7	12 33	7 H50	2100	13200		١	.3	-1	-	6.7	7.35	7.4	<b>ు</b> డిం	370	<b>38</b> 0				6.0		34	
HEAVY SHOW	19	F	11.50			14 300				-	-	66	7.4	7.4	310	374	380		1.0	37	6.0	м	šc	$\mathbf{J}$
<u>-</u>	20	5	12,514	13600	දාග	12,300		1.	.2	-[			7.4	7.5	315	375	376		1.0	39	60	34	30	٦
THANING COND.		-3		* *															T	$\neg$	П	T	Т	$\top$
BRIGHT YELLOW		M	15,24	24520	2100	23,200			_	_	1	6.7	7.55	7.5	350	380	376		7	35	7.1	<b>Y</b> 6	30	2
SKUM APPEARED	-	T	17.45	3 73700	2400	21,400		١	.5		1	ن. ت			320	364	3400				40		15	<u> </u>
٥٨١ ځالاجمد		W		3 17100				1	2	_[	1	9	4.5	7.5	310	مليك	360				58		10	$\prod$
		7		19700				١.	. ما.	<b>-</b> [	I	6.5	1.5	75	340	360	360				5.2		20	<u>.                                    </u>
W D.O. CHELL	_	E	1984	3 18,000	1800	NOW		\	.5		_	66	75	٦.5	330	376	OVE		18	37	62	32	3.	Л
	27	₽	14.390	19 300	2100	18,000		2	٥,	_		65	74	745	316	376	380				55.		30	
ATPROX. 1" TEE ON	20			* *						$\perp$										$\Box$	П			П
LAST 1/2 OF 20th CELL		M	18,30	9 24700	2100	25400		1.	<u>ა   -</u>		_	<b>6.5</b>	7.5	7.4	330	.386	376		0	1	60	Z.	3.	ļП
		上	17.410	20,600	2400	19000		_ 2	.2 -	=1	_	6.6	7.45	7.55	<b>.38</b> 0	390	374		çd.	41	8.2	<b>%</b>	30	
	31	X	16.35						- ه.		_	6.7	7.65	7.7	કન્દ	388	346		12	13	78 :	37	1/5	
	100			1 501100				_	99 -	-[		17.1	8,65	202.2	8494	10,030	10,152	1	184	เป	172 8	B90	66	_
	Avg.			16,165			$\perp$		.2 -	- [	_]	_		1.5	315	371	376		55	16	64	<u>i3</u>	25	<del>,</del> [ ]
		1-1		26,700				2.	.2   -	_			_	7.7	380	<b>390</b>	390		<b>7</b> 0-	13	9,83	л	45	
	Min.		8.74	1 13,600	1800	12,300		<u> </u>	4-	-1	_	6.3	7.35	7.4	290	352	346		: م	32	5,2	32	0	$\prod$
	<u>_</u>	Ļ			<u> </u>			$\bot$		丄		l									$\perp$	$\perp$	$\perp$	$\Box$
	*( )	ME/	L-Boiler + 800 G	PD ENFILT.)	L	**(SUND)	Y FLO	M IN	kd. 1	1 1	MON	OVYŠ	)	[							I			I = I

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Communic	Ĕ	Š	PLANT PROCESS MISS Profits Chapse P	dia Plant		r Lagoon		s 4	_			. •	-			Semelet	•		٦٤			1 L		
	ā	٥	100 100.	Well (GPE	900	SHP.	11.	H.	<u>-</u> /1	1	÷	١.	2.	3		2	3,	2 T						1 .
April 1971		T	1634	1 21Ex	2100	29200			2		_	6.6	7.6	7.6	300	356	360			10			_	160
	•	F	15,71	4 Has	2/1800	13000			2	_		6.7				386	340			4				28
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DO N 350m - 4 pom	•	M	200	25 24 10	2100	22,800			Lol	_	-	64	15	7.6	380	400	344	<u> </u>		57			$\vdash$	20
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" Lan	,	7	21.3	48 18 4v	212	17,300			3	-	1	6.5	75	า.15	442	416	346	<u> </u>		43			$oldsymbol{oldsymbol{\sqcup}}$	10
· ·	•	T	66	19 20 4	a 5100	19 300		2	2.5	-	1	45	745	7.85	440	416	344	<u> </u>		*			<b>—</b>	12
CHAKED WELL		F		820.60	_			П	2.	-	_	65	7.1	7.8	388	412	360			50			$oldsymbol{\perp}$	50
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	11	_		0 22,80				T,	.5	_		6.5	74	75		440	366		2	SA	7.6	54		35
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	-	3	12,6	· * *		1,5,000	1	1	$\neg$		<del>                                     </del>		1.3	† · · · ·				1				59		ho
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	-	╁	1 1 1 2 2 2	62 5229	54 2	Acres	_	<del>   </del> ,	32.3			V 95	105	44	9824	11,130	9499	1	SO	4	100	HR.	$\vdash$	1211
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Springhes	·····		_									ABLE					ALKALIN			ι —	LAG	-	AGOS	200 W	EATI	ER
Com	nments	M	Š	PLANT PROCI	Chassa Práts,	Plant -	Flow	LAGOH EHF.	ā i			Semptos 2	3	I .	Sample	• _		Somptos	•							
. M	107.	٥	ľ	Mas.	Ma.	Well (GPD)	<b>GPD</b>	CALC.)	Į į	8	ml/1 ,	m/1	ml/1	١,	, 2	. 3	me/1	, K	3	5 €	185		1	y 100	4 ₹	, ē
MAY	1971	1.	5		14007	17.500	2100	16200		Ш	1.5			69	7.5	255	370	440	396		0	18	35 4	в_	丄	L
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		•	7		14.487						1.5	_	_	7	7.7	7.4	340	344	36 <sub>50</sub>		0		3.2	$\mathbf{I}$		
			×		14,589	09.81	2100	17,500			2	_	_	1	7.6	7.5	320	460	406		0		3.2	Τ	Τ	Γ
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		10	M		16,785	26400	2100	25.400				_	_	6.6	74	7.6	326	476	408		.5	Ş	32	I	П	
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		17	K		15.533	24.700	2100	25,400			-		-	62	7.4	6,9	ZUZ	500	410		0	_2	3		L	
		18	$ \tau $		16,573	21000	2400	19 400			•	Z	-	6.4	7.5	7.0	とんの	536	400		0	2	Z	丄	L	Ĺ
RAIN		19	W		11,231	21900	2100	20400			1.5	40	1	6.4	74	7.2	228	524	400		0		2		L	L
		20	T		9.853	<u> ∞Γ</u> ΡΙ	2100	18-400			۲	40	_	<u>6.3</u>	7.2	7.3	240	530	400		0		3	丄	↓_	L
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Comments	Ę	š	PLANT PROCE	i36 Choose Frets.	Plant	Boiler	(ديون)	ž ž		TLEAS Sen	piet.		٠. ١	pill Samplu		ALKALINI	Samples	_	_			746			_ 4
	3	۱۵	No.	No.	Well (GPD)	600	(cuc) 1	11.	8 II .	41		<b>2,</b>	١,	2	3	me/I	2	3,	<b>5</b>	8 5	扫	87.	11		,
JUNE 1971	Ŀ	T		14.013	מסרשו					<u>. 119</u>	<u>1-</u>		6.5	7.8	74	240	520	376						$\bot$	
RAIN	<u>.</u>	W		12,043	16800	2100	15,500			<u> 8</u>	<u> </u>		6.4	74	7.4	270	520	356				41			
	Ŀ	工		PS1.1	W-700	2100	15.400		2.	5 8	5 .	_	6.3	74	75	300	516	366				40			
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wast Brands as of or or or This A.M.	•	5		11,658	24800	2100	23,500			9	<u>;</u> [	_	6.3	74	7.5	300	540	430		0	68	2	68		
MICH AMPERACIÓ AND TEMPERATURE	•	5			* *						$\mathbf{I}$								l			Ш			
RAIN	,	7		14.782	44.400	2100	43.100		Τ.	- 13	٠ .	_]	6.4	75	16	200	540	120		0	20	3	68		
-		7		2978	20:00					2 14	• T		64	7.4	7.6	180	530	370		0	70	3.5	70		
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PRODUCES WITH HIGH AMPERACES ON OUR COMPRESSION		┰		_	22,200				1	5 13	٦.	$\overline{}$	6.3		74	206	536	374				5.3		$\neg$	丁
THEY ARE CLESOMY OF Or WID THE	10	W		12.301	23.200					2 13	-	_1	63	74	15	240	526	392		_	71	_	73	$\exists$	$\top$
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_		12	,	PLANT PROC			Baller	LAGOCA INF.	1 .		TTLEA	OLE :	<b>WLIE</b>	)S	914		ALKALIN			Γ	LAG	OON	LAGO	>014	MEV.	THE
	nments	PATE	À	Milk Prets. Ibs.	Choose Prots.	Plant .		CALC.)	×įį	S S S	Sac 1	mples 2 ml/1	3,	١,	Sample 2	* 3	١ ،	Samples 2	. Mg/1	28	9 5	į b	Q 5		ain.	I
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			3		12,773	21,900	2/00	20,600	<u> </u>		5 4	6	1	4.2	74	7.5	250	5∞	33C.		0	72	1.67	иΓ	Т	T
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	-	Ι.	PLANT PROCE	44		BOILET	LANDCH					90U0	6	984	-	ALKALINI	TY			LAG	DON	LAGG	XX	WE/	THE	R
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IPPI TEUDINA	١,	5	<b>Inc.</b>	Me.	Well (GPO)	<b>670</b>	GPD	r a	•	<b>~</b> "	<b></b> (/1	***		1	1	Mg/1	mg/1		•	•	ا ''ا				٦,	۴
	2	М		16.828	31500	2300	30.000		1				62	7.1	7.7	170	512	416		1.0	20	SO	7.	기	$\exists$	
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compressor, 4 Am		T			22,800				H	1.5	_		63	7.2		190	512	400				5.0		$\neg$	ヿ	_
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		F			21,900				Ħ	٠2٠	_	_	6.2	7.1	7.2	160	506	416				12		寸	╗	
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	18	3			**												1	i		Г					$\neg$	
	10	M		15.614	30300	2300	28,800			_	_	_	6.3	7.2	7.2	140	5∞	400		15	21	42	72	П	٦	
	17	T		17,517	31,100					2.5	_	_		7.2	_	280	488	380				4.3		П	$\neg$	
	10	W			21,400				П	1	-	_	6.3	7.1	7.2	260	490	400		1	71	4.5	12	$\Box$	$\Box$	
	19	7		7.754	12.80c	2000	21,600			.5	_	_	6.2	7.1	62	240	480	400		1	71	4.5	12		$\Box$	
	20	F		11.936	21,900	1900	20,600			.8	-	-	6.2	7.1	6.2	246	476	412			72	4.5	13	$\Box$	$\Box$	_
		5		7,945	21900	2100	20,600			.8	-	1	6.2	7.1	6.2	200	440	412		$\lfloor L \rfloor$	72	1.3	13			_
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	23	M		13426	18,9∞	2000	27,700				_	_	6.1	2.2	7.3	9	420	400		48	72	15	73	$\Box$		
	34	T		11,955	22,300	2400	20,700			8٠	-	_	6.2	7.2	7.2	200	400	400		45	72	1.5	13			
	26	W		17.297	21,900	2100	20,600			,2	1	-	4.2	7.2	7.2	180	412	400		15	72	4.5	73	$\Box$		
	26	T		11.742	22,800	2600	21,000			8,		1	6.2	7.3	7.2	wo	400	406		3.5	11	4.5	12			
METER REPLACED	27	F		12511	20,600	2100	19,300			, 2	1	1	6.1	7.2	7.2	180	400	400		10	21	4.5	21			
an tele		3		11, 639	18,800	2500	17,300			٠5		-	6.2	7.1	7.5	190	412	400		4.0	70	38	21			
· · · · · · · · · · · · · · · · · · ·	219	<u>s</u>		•	* *																		$\Box$		$\Box$	
	+	M		16,004	32,000	2500	30,3∞0		Ц	_	-	_		2.1		160	436	120		60	20	60	10	$\bot$	$\Box$	
	31	T		12,080	21,900	2400	20,100			.5	ı	_	6.2	7.1	7.2	200	430	416		-	_	56			$\bot$	
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	Ave	-			20,255		19,06	$ldsymbol{ld}}}}}}}}}$	$\sqcup$	<u>٩</u>	_	_	6.2	7.2	7.1	193	472	408				41			⅃	
<del></del>	****	1	ļ	17.517	34,5∞0	2700	35,200	$\sqcup$	Ц.	2.5	_	_	6.3	_	7.5	280	516	420				6.		$\dashv$	┙	
	nes.	╀_		6,797	18,800	1700	17,3c0		$\sqcup$	.2	_	_	6.1	1.1	6.2	140	400	380		12	وي	38	10	_	┙	_
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	<b> </b> •(\	NEL	L-Boler+	G9⊅ ∞8	INFILI.)	l	ek(Sund	MY F	\ow	( E.	ĸL.	IN	MON	(¿YAC	1 !				l		. 1	ıl	ı	- 1	- 1	

		T		D 44	***							ABLE				(EPO)		177			4.60-					MF -
Comments		¥	ě	PLANT PROC MHR Prets.	ESS Choose Prets,	Plant .	Denler Flow	- Lagoon INF.	ž ž	4 ¥1	4	Somple: 2	•	)\$ 	Sample Sample		ALKALIN	Samples		_	1 1	, ,	2		WEAT	- 4
SEPTEMBER	197L	1	l 1	ibs.	ibs.	Well (QPD)	GPD	(CALC.)	Į į	8	1 mi/1	72 mi/1	3 mi/i	١,	. 2	. 3	mg/1	2. mg/1	3 , ms/1	2 8	8 3	ĔĽ.	3 👌		] <b>\$</b>	Į
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			┢╾┤		4963	1.1100	i ibo	16,200	-	+	-1	-	_	6,2	1.1	7.1	اصاا	412	<u>4</u> ∞		46	뱌	<u> 1</u> 6	<u> </u>	╁┤	Н
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Comments	PA TE	Ì	Mills Profes.	Cheese Prots.	Plant	FLOW	ENF. (CALC)*		불합	٠, •	2 mi/1	3	۱, ۱	Lompini Ž	3		Samples 2. mg/l	3	<b>0</b> €	2.5	12 8	5 1	****	ī	1
OCTOBER 197			109.	100.	Well (GPD)	GPD 1	IE 200	2 ₹ 1	8 2	mi/1	~√/1 <b>1</b>	mi/1	6.2		1	mg/1	AUC	선사시		71	بدور	-0/4	- اے	*	١
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	<b>+</b>	127		17,573					$\dashv$	٠2		_	6.2			190	476	410	<del> </del>		676			╁	┢╾
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	1.	43		19.635					$\vdash$	.5	_		6.2	1.3	7.2	260	460	ملاام	<del> </del>		62 7			┿	₽
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	7,	·T		18.819	20,100					.5		_	6.2	7.2	7.2	240	476	444			56				1_
•	20	٠١,	/	15819	19 300				П	.3		-	4.2	7.2	7.2	240	474	444		٥	58	6 و	<u>a</u>	1	上
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	22	F		14 796	18.00				П	.3		_	6.3	7.2	7.1	230	476	440		0	60	<u> 2014</u>	2	丄	L
	2			14 534	17,500	1			П	.3	_	_	6.3	12	7.1	226	470	436		0	61	Sel	4	1	
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المارية	20	_		19.474	21,000				П	.4	_	_	6.1	7.2	7.2	2360	480	436		0	66	7.0 (	₁T	Т	Γ
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	2	<del></del>	1	10,380			17,300			.5	-	_	4,2	_	7.1	240	476	444			61			Т	Γ
		•   F	+	10,535			Ţ		П	.4	-	_	6.3	_	7.2	240	410	444		0	58	305	;7	T	Γ
	1	·   5	<del></del>	16.925						.5	_		_	_	1.0		474	440			658			T	Γ
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	٠,	_	+	278.874	535,300	54~	شما تسہ	1	Н	9.2		_	1/4 3	182	194	5912	12,380	11434		254	159/1	83 K	<u>s</u>	$\top$	T
<del></del>	A.	+		14.572					Н	.4		_	6.2		7.5	227	476	440	1	_	41	_		T	Г
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	94			4,136			14.500		Н	.2		_	6.1	1.1	1		460	430			52 5			1	Г
· · · · · · · · · · · · · · · · · · ·	+	7		~1,100	17,10	<u>, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,</u>	17,500	1	Н	<del>, ~</del>	<del> </del>		1.,,	† <del>'''</del>	1	1,00	<u> </u>	<del>                                     </del>	1	Ħ	<del>-</del>	Ť	十	†	T
	- L	<u> </u>	LL-Boluer 1	- Boo (- Bo	T. E. E.	<del>                                     </del>	* M(STH!			Tool	ـــــــا دور	<b>*</b>	-6104 	<u> </u>			<u> </u>	<b></b>	1	T	$\top$	十	+	1	<u> </u>
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-	2	,	PLANT PROC			المالحا	LAGON	4			MOLE		<b>)</b> \$	pHi		ALKALIN			Ī	LAG	OON	LAG	MOO	WEA	FHER
Commenty	1	Š	MHk Prets.	Choose Prots.	Plant .	FLOW	("A.C.)		E E	1	Sampler 2		10	Sample	* 3	١.	51m7frr 2	3	5€	95	للإنكا	Q 5	1	lain-	1 1
November 1971	١,	M	lbs.	1 <u>,</u> 1, 1, 1, 4, 4	24 700	1200	apo "	¥	ō	, 2	ml/1	mi/1 	١,,		ء ما	mg/1	mg/1	I ****	3					(41)	, F
70.	1	_					25,200		$\dashv$		_	-			7.2	1	484	4125	<del> </del>			80		+	╁
RAN	-	-		21,800	18,600				$\dashv$	.4		=		7.3		23c	480	1				88		+	4
	4	w		17,681	1660					.5		=	6.2		7.1	240	480	448		5.5				4	+
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		ш		11.204	11,800	1300	11,300			.2		<u> </u>	6.2	7.2	7.1	236	નક્ષ	446		<u>.</u> .O	52	10	52	$\perp$	$\perp$
	-	5		14,623	الملص	1000	15,-100		$\Box$	.5		<u> </u>	6.2	7.2	1.2	240	480	430		5.5	47	10	49	$\perp$	$\perp$
	-	5			**							<u> </u>	<u> </u>		<u> </u>									$\perp$	$\perp$
		M		11,071	28000	2800	24000			.1	_	L=	6.2	7.1	2.1	180	484	444		10	45	10	46	Т	Т
	•	7		13,165	18,900	3300	16400			.5	_	_	43	7.2	7.2	240	480	446		89		_	_	7	T
	10	W		H815	14.600	کرححی	13400			.5	_	-	6.2	1.2	1.2	240	476	440				12		$\top$	T
	11	_		12,225	13.400				7	٦.	_	_	6.2	_	7.2	230	418	440		65				十	+
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<del></del>	23	7		19,897						_			6.2			180	474	444		2				4	$\bot$
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		F		16394					4	_			6.2		7.2	2∞	480	440		1.5	14	11	40	丄	L
SNOW	27	_		19.315	22,875	2100	21,575			6			6,2	7.2	7.2	240	484	440		60	<u>41</u> 1	11.3	40	$\perp$	
	20	-			19925	2800	17.925													Т		$\Box$		T	П
SNOW	20			14430	12500	1800	16.500			_		-	6.2	7.1	7.2	160	480	444		.01	ы	12	19	T	Γ
	_	工		19092					$\Box$	.5	_	_	4.3	7.2	1.2	240	484	444		25/4				T	Г
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	Total			416,845	413,000	56,800	431,600		Ç	13	_	_	1615	187.1	187	5868	12,500	11,492		12.2 1:	24	رو ا	LA	十	
	Avę.			15,439	15,761				_	A	_	_			7.2	226	481	442		1.3 1				1	П
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	*(\	اهن	-Baler+8	SOO GPD T	MENT.		**(5140	WY F	اجت	Tn:	<u>،</u>	~~		روا					$\dashv$	+	十	十	+	+-	Н

	T <sub>p</sub>	T.	PLANT PROCESS		Boyle	r hAsaci	<del></del>		SETTL	EABLE	<b>50L</b> I	26	pti		ALKALIN	ITY .			LAG	000		100	( wi	LATHE	R
Comments	18	ž	Milk Prots. Choose Prots.	Plent	From	ENF.	al i	ğ	<b>1</b>	2	* 5		Sample 2	. 3	١,	Samples 2	•	_0	65		05	du	tain		1
DECEMBER 1971	١.	l,	1 100	Well (GPO)		15,195	<b>12 2</b>	181	.5	انجا	1		7.2		mg/i	, mg/1	456	4.3			125			<b>1</b>	ř
CHALSO CALIDATION	1			23.305	1	21.505	1	+-	.3		+=	6.2		1.2	234	520	4160				115			$\vdash$	-
OR GRA.	1	F				15,745	-	╁╴	.2	_	+=	42			230	520	462				11			╂╌┼	-
11: 11	•	3				19.750	_	†	.5		_	62	_	7.2	240	520	468				1			╁╌┼	-
IMPUTRATION TEST	•	_				8.925		+	-		1=	100	1.5	1.2	240	345	-100		17	20	₩	130	<del>                                     </del>	╁	_
ON SEWER LINE	•	M		9.155		2955	_	1		_	<u> </u>	4.3	1.2	7.2	180	520	4180			10	1.5	70	┝╌	╁┼	-
CHECK CALIBRATION		F		17,56			一	1-	.4		+=	6.3	_		240	520	480				11.5		├	╁	-
ran eff.	•	-		21,210		19310	<del>                                     </del>	十一	.5		+	6.2			2-10	522	480				11.6		┝╌	╂═┼╴	-
RAIN	•	+		<del></del>		143.5	<del> </del>	1	.3	_	<del>  _  </del>	6.2	_			520	416				12		⊣	╂╼┼	-
RAIN	10	+		16765		15,265	<del> </del>	╁	.5			63		7.2		530	426						╂╼╌	╂╌╂	-
CHELLED CALCUATION	11			15.980		14.460	<del> </del>		<u>د.</u>	-	-	0.3	1.2	1.5	236	516	480				125 125		₩	$\vdash$	-
EARLIER SHOWEALL	12	3		11,450	1600		╁	t	-			6.3	121	1.2	120	216	1480		۱3	4	113	72	┢	₩	-
HAS MULTUD (8")		М		9.875	1000	,	<del> </del>	$\vdash$	┝═┈		-	6.3	<del>                                     </del>		-01	1	1.00		-	-		-	H	╆	-
	1.0	_		17.805			<del> </del>		.5	_	+=	62	_	<del> </del>	194	520	480				13		┝╌┤	₩	_
RAIN (2")	110	+		17,540	2700		$\vdash$	╁╌	<del>د.</del> ع.		<del>                                     </del>	_	T .	7.2	190	520	460				14		<b> </b>	┢┷┼	-
	10	Ť		17.855	2500		<del>                                     </del>	+-			<del>  -</del>	6.2	<del>1</del> —	7.2-	23a	516	4160		_		14	_		⊢┼	-
<del></del>	_	+		15,835	2400		-	├	.4	_	-	6.2	7.3	7.2	220	520	4164				14		⊢	┢┵	_
<del></del>		5		13,270	1			$\vdash$	.3	-	_	4.2	7.5	7.2	120	500	440				М		⊢⊢	⊢∔	-
INFILTRATION TEST ON	_	_		_	2300		<del>                                     </del>	$\vdash$	.4	_	=		<del>                                     </del>		226	5∞	444		4.0	<u>33</u>	18	34	Ш	⊢	_
SEWER LINE	_	+-		13095			_	-	_		<del>  -</del>	6,2	+							لـــر	Ш	$\vdash$	$\vdash$	$\vdash$	_
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EAS.	22	<u>.                                    </u>		2010	2500			Н	.5		-	4.3	7.2			500	460				U		Щ	$\vdash$	_
C17-,	23			18.570		16910	-	$\vdash$	.5	_	_		_		240	480	490				112		Ш	$\vdash$	_
		-		15995		14595	ļ	-	ک,	_	_	6.3		7.2	230	484	480				11		Ш	$\vdash \bot$	_
	_	_		19,05	2500	12.405	<u> </u>	Н	ک	_	<u>                                     </u>	6.3	7.2	7.2	240	490	484		40	36	11	35		$oldsymbol{\perp}$	_
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	-	12		25,135					.2	_	<u> </u>	62	7.2	7.5	200	484	470				12		_	$\perp$	_
	_	M		13,215				Ш	.ح	_	_		7.2		236	484	476		3.5	37	11	<u>ع</u>			
CHECKED CAL. OF	_	T				1660		Ш	ی.	_	_	62	12	7.2	240	490	480				10	36			
E Fire,	Ī	₽~		20,955				Ш	.4	_	_	63	7.2	7.2	23C	490	484		.٤		1		$\Box$		
	*	T		20435				Ш	.د	_	_	4.2	7.2	7.2	240	492	484		.2	اح	9	3_			_
	31	۴		20,120				Ш	.4	_			7.2	1.2	230	490	484		3	36	9	عد	_]		_
·	Total	<u> </u>		505,905		461,000		Ц	101	_	-	<b>L&amp;</b> .3	1942	1944	4134	13,628	12,714		71.7	οł	34,	PEN	$\Box$	$oldsymbol{oldsymbol{oldsymbol{oldsymbol{\Box}}}$	
	*	-		16,320					4		_	د.ک	1.2	7.2	227	505	471				11.7		T		_
		<u> </u>		25,185				Ц	.8	_	_	6.3	1.3	7.3	240	522	490				15		$\Box$	$\Box$	_
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	لِـا	Ļ	<u> </u>	<u></u>					[										T	$\top$	T	$\exists$	丁	$\top$	
	*(v	4E/	L-Baler + 800 GP	O ENFAT.)																7	丁	$\top$	$\neg$	$\neg$	-

	Ŀ	Ţ	PLANT PROC	ESS		Beiter	LAGOON	<b>.</b> .					*	<b>pH</b>		ALKALIN	ITY			LAG	00N	LAG	PON	WEA	THE	R
JANUARY 1972	8	PAY.	Milk Prots. Ibs.	Choose Prots.	Plant .	FLEW	ENF. (CALL,)		ŧ i	ı	Somple: 2	3	1	Sample Z	3	١	Samples 2 mg/1		<b>5</b> €	2 5	<b>1</b>	25	ξŅ	tain-	Į	É
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#### APPENDIX D

## Cost Estimates

Annuity whose present value is 1

$$A_{n}^{-1}$$
 = i /(1-v<sup>n</sup>) =  $S_{n}^{-1}$  i +i

Mechanical Equipment - \$15,592.82

n = 5yr i = 0.075 (7 1/2%) 0.24716472 x \$15.592.82 = 3853.99

Construction - \$33,900.56

n = 30 yr i = 0.075 (7 1/2%) 0.08467124 x 33,900.56 = 2870.40

where

And i present value of annuity for unit periodic payment at interest rate i

i = rate of interest

n = number of conversion periods

v = present value (at compound interest) for unit principal, $= <math>(1 + i)^{-1}$ 

 $S = Amount (at compound interest) for unit principal = <math>(1 + i)^n$ 

# APPENDIX E

**Pictures** 

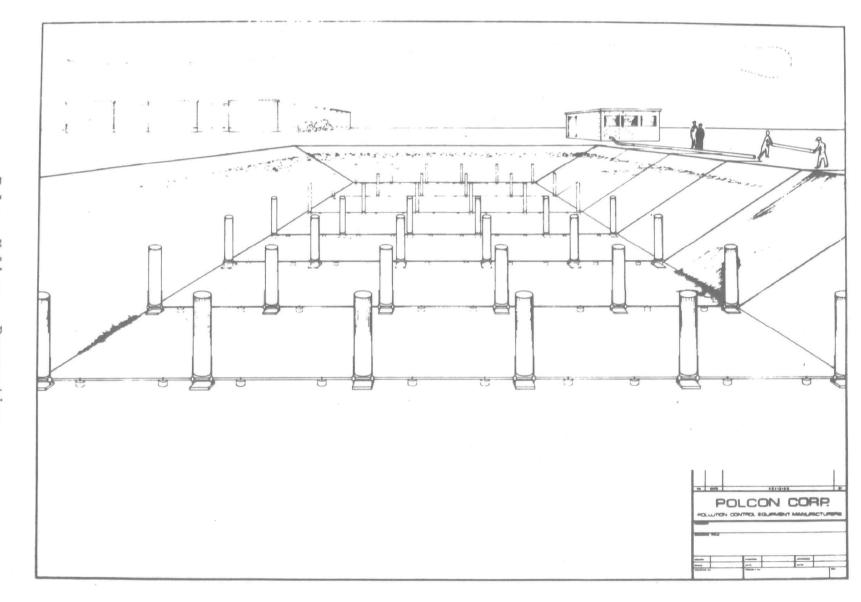


Primary Lagoon - Summer



Primary Lagoon - Winter

Polcon Helixor - Flow Diagram



Polcon Helixor - Perspective

SELECTED WATER		1. Report N	0. 2.	3. Accession No.
RESOURCES ABSTR	ACTS		ł	W
INPUT TRANSACTION	FORM			ww.
4. Title Treatment of	Cheese Processing		•	5. Report Date
in Aerated Lagoons		, mastewaters		δ. ·
				8. Performing Organization
7. Author(s) Francis R	. Daul, Grant Dir	ector contra	ted w riti	Report No.
final report to William	C. Boyle End & 1	awrence Polk	wski Phd	10. Project No.
9. Organization Kent Chee	se Co.	Olkowski		12060 EKQ
Kent, Ill	inois 61044			11. Contract/Grant No.
				13. Type of Report and Period Covered
12. Sponsoring Organization	Environmental Pro	tection Agenc	У	July 🏻 1969
15. Supplementary Notes				January <b>L</b> 972
Environmental Prot	ection Agency rep	ort number, E	PA-660/2-7	4-012,
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an average BOD remo	a one year period val of 97 percent Hydraulic detenti single greatest aintenance and am	<ul> <li>A two-stag at BOD loadi on time varie influence on ortization we</li> </ul>	e aerated ngs <b>rangin</b> d from 50 process ef	lagoon system provided g from 0.117 to 4.34 to 82 days per lagoon. ficiency. Costs for
This report was sub Cheese Co. under the Work was completed	e partial sponsor	ship of the E	ct No. 120 nvironment	60EKQ by the Kent al Protection Agency.
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17a. Descriptors *Aerated   Evaluation, *Aeration	Lagoons, *Industron, Treatment Cos	▼	Biological	Treatment,
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17c. COWRR Field & Group				Į.
		las No es	Send To:	
18. Availability	19. Security Class. (Report)	21. No. of Pages 79	Sena 10:	
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