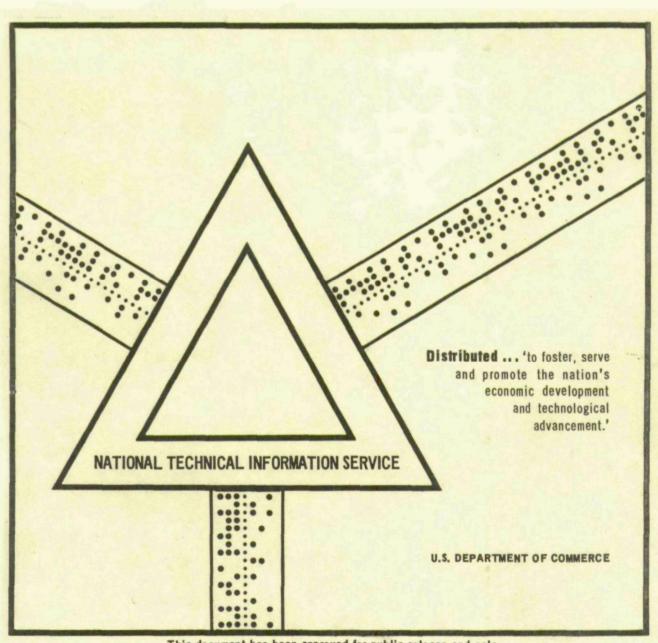
SYSTEMS ANALYSIS OF EMISSIONS AND EMISSIONS CONTROL IN THE IRON FOUNDRY INDUSTRY. VOLUME III. APPENDIX

A. T. Kearney and Company Chicago, Illinois

February 1971



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SYSTEMS ANALYSIS OF EMISSIONS AND EMISSIONS CONTROL IN THE IRON FOUNDRY INDUSTRY

VOLUME III - APPENDIX FEBRUARY, 1971

For

Division of Process Control Engineering Air Pollution Control Office Environmental Protection Agency

Contract No. CPA 22-69-106

Prepared by
A. T. Kearney & Company, Inc.
Chicago, Illinois

AIR POLLUTION CONTROL OFFICE

SYSTEMS ANALYSIS OF EMISSIONS AND EMISSIONS CONTROL IN THE IRON FOUNDRY INDUSTRY

VOLUME III - APPENDIX FEBRUARY, 1971

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APPENDIX B

DATA BANK

All the data compiled on the foundries are contained in the data bank which is a part of this appendix. The actual listing of the data is given in Exhibit I of this appendix.

Exhibit I is divided into six sections with each section dealing with a different aspect of the foundry data. Each section begins with the foundry identification number and this number, therefore, provides the key between sections when obtaining additional data listed for that foundry number.

If a foundry number is omitted in a section it can be assumed that the data was not available or did not apply as the case may be.

The headings for each section which describe the data fields are, for the most part, abbreviations. A listing of the abbreviations and the appropriate descriptions are given in the Format for the Data Bank, Exhibit II of this appendix. The format for the data bank is arranged in sections which correspond to the sections of the data bank, Exhibit I.

In several cases the data is provided as coded input. The explanation of the code is given after the appropriate description of the abbreviated heading in Exhibit II.

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GENERAL FOUNDRY DATA PAGE 7

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********************* FDRY CD FCE FCE LIN BLT BLT TOP CHG GAS AFT CHG FL OX FCE FCE HLD MLT MT PR OT *****BLAST*** SIZE HT CHG AFT AFT DIS POWR NO. TY NO TYP TYP DES HTG C/O DR IN EN USE DIA CAP RAT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LOC AFT SPLY 1-0 0024 10 1 1 86 000 32 1 0 0024 10 86 000 32 0025 10 84 000 20 13 ı 0026 10 8000 2 720 0000 54 000 10 00 5400 19 0026 10 2 720 0000 54 000 10 0027 10 1 1 0 72 000 14 8000 14 0000 0 000 0000 1 1 0028 10 000 4 0028 10 0 0000 . 8 Oύ 0029 10 78 000 15 14 1780 1 276 0000 0029 10 000 1000 6 00 0029 10 6 00 000 1000 0029 10 0000 0 6 00 000 1000 0030 10 . 5 0000 0 0030 10 0030 10 00000 00 0000 0 0031 10 96 000 35 000 0000 ì 0031 10 ı 96 000 35 000 0000 0032 10 56 000 16 5400 15 1000 2 31 0000 0032 10 00 10 5 00000 00 0000 0 000 0000 20 00 0033 10 108 000 50 0033 10 108 000 50 0033 10 ı 000 137 10 0000 0 000 4400 0033 10 000 137 10 000 4400 0033 10 000 137 10 000 4400

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******************************* FDRY CD FCE FCE LIN BLT BLT TOP CHG GAS AFT CHG FL DX FCE FCE HLD MLT MT PR DT ******BLAST**** SIZE HT CHG AFT AFT DIS POWR CTL DR IN EN USE DIA CAP RAT UT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LOC AFT SPLY SYS NO. TY NO TYP TYP DES HTG C/O T-0 0033 10 7 2 7 2 000 137 00 00 00 0000 0 0000 0 000 2200 0034 10 1 1 1 108 000 50 000 0000 0034 10 2 2 000 100 10 000 4000 0035 10 54 000 13 9 00 000 0000 0035 10 2 1 48 000 8 9 00 000 0000 0035 10 48 000 8 9 00 00 9 00 00000 00 000 420 0035 10 000 14 00 0036 10 42 000 4 0037 10 34 000 4 0038 10 78 000 23 0000 0 000 0000 1 0 0 0040 10 1 1 000 6 0040 10 2 1 000 6 0040 10 000 6 0040 10 00000 00 0041 10 1 1 000 4 0041 10 2 2 000 .75 00000 00 0041 10 000 .25 0041 10 000 .20 ı 0042 10 72 000 25 16 480 0000 0042 10 9 00 24 00000 0042 10 9 00 24 00000 00 0043 10 54 000 11 4500 2 0044 10 1 1 72 000 20 0044 10 2 1 72 000 20 0044 10 000 30 00 00 00000 00 000 750

**********FURNACE CLASSIFICATION******* FORY CO FCE FCE LIN BLT BLT TOP CHG GAS AFT CHG FL DX FCE FCE HLD MLT MT PR OT ******BLAST**** SIZE HT CHG AFT AFT DIS POWR CTL NO. I'Y NO TYP TYP DES HTG C/O T-0 DR IN EN USE DIA CAP RAT UT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LOC AFT SPLY SYS 0044 10 00000 00 0000 0 000 750 000 30 00 00 004 - 10 000 0000 84 000 18 0000 0 0045 10 84 000 18 0000 0 000 0000 0045 10 84 000 18 0000 0 000 0000 0045 10 000 0000 84 000 18 0045 10 0000 0 000 0000 84 000 18 0045 10 000 0000 U 84 000 18 0045 10 96 000 26 000 0000 1400 70 0000 0 0045 10 96 000 26 0000 0000 0046 10 103 15 13 16 000 9100 0046 10 000 9100 Э 103 15 13 16 0000 0 0046 10 103 15 13 16 000 9100 0046 10 000 9100 103 15 13 16 0046 10 83 00 00 15 000 1400 0046 10 Ω 00C 83 00 00 15 000 1400 0046 10 83 00 0047 10 54 000 10 500 35 0047 10 54 000 10 0048 10 103 20 8 00 00000 00 000 6000 0048 10 000 22 00 00 2 22 00000 0048 10 800 ' 000 22 00 00 2 22 00000 0000 0 000 0048 10 2 000 22 00 00 2 22 00000 00 0000 00 0049 10 ı 72 000 5 ì 0050 10 66 000 17 0051 10

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0052	10	2	1	1	2	3	2	2	8	0	1	0	1	1	60	000	16	16		5900	18	450	60		0	0000	0	000	0000	1
0052	10	3	1	ı	2	3	2	2	8	0	1	0	1	1	60	000	16	16		5900	18	450	60		0	0000	0	000	0000	0
0052	10	4	1	ı	3	1	2	2	8	4	1	0	0	1	48	000	8	8		3500	16		60		0	1000	4	480	0000	2
0053	10	1	4		0	0	0			0	0	0	0							00000	00	0000	00	00		0000	0	000		
0054	10	1	1	1	3	ı	1	2	1	0	1	0	0	1	60	000	15			8800						0000	0	000	0000	1
0055	10	1	1											1		000	2												0000	
0055	10	2	1											1		000	5												0000	
0056	10	1	1				ı	2						ı	54	000	9												0000	ι
0057	10	1	ı	1	3	1								1		000	9												0000	1
0058	10	1	1	4	i	3	ı	2	ı	2	1	0	0	1	96	000	30					1000							0000	1
0058	10	2	1	4	1	3	1	2	1	2	ı	0	0	1	96	000	30					1000							0000	2
0058	10	3	1	4	1	3	ı	2	1	2	ı	0	0	1	96	000	35					1000							0000	3
0058	10	4	1	4	ì	3	1	2	1	2	1	0	0	1	96	000	35					1000							0000	4
0058	10	5	ı	4	1	3	1	2	1	2	1	0	0	1	96	000	35					1000							0000	5
0058	10	6	1	4	ı	3	1	2	ı	2	1	0	0	L	96	000	35	•				1000							0000	6
0058	10	7	1											ı	96	000	35												0000	
0058	10	8	4	1	0	0	0			0	0	0	0			12				00000	00	0000	00	00		0000	0	000		
0059	10	1	i	4	1	3	1	2	1		1			ı	96	000	25					1000							0000	1
0059	10	2	1	4	1	3	1	2	1		1			1	96	000	25					1000							0000	· 2
0059	10	3	1	•	1	3	1	2	1		1			ı	96	000	25					1000							0000	3
0059	10	4	1	4	1	3	1	2 .	1		ı			ı		000						1000							0000	4
0059	10	5	1	4	ı	3 .	ı	2	ı		1			1	96	000	35					1000							0000	5
0059	10	6	1	4	ı	3	1	2	1		1			1	96	000	35					1000							0000	6

FORY OD FCE FCE LIN BLT BLT TOP CHG GAS AFT CHG FL DX FCE FCE HLD MLT MT PR OT ******BLAST**** SIZE HT CHG AFT AFT DIS POWR CTL NO. TY NO TYP TYP DES HTG C/O T-0 DR IN EN USE DIA CAP RAT UT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LOC AFT SPLY 0059 10 7 1 96 000 35 005. 10 96 000 35 0059 10 96 000 35 0059 10 2 120 20 00 00 0059 10 000 14 00 0059 10 12 000 14 00 0060 10 1 1 84 000 15 0060 10 84 000 15 0060 10 90 000 22 ı ı 0060 10 90 000 22 0060 10 90 000 22 0060 10 90 000 22 ı 0060 10 1 00 00 0060 10 1 00 0060 10 1 00 0060 10 4 00 0060 10 4 00 0060 10 Э 0061 10 8 00 10000 24 480 0000 ı 65 000 20 0 1500 l 0061 10 8 00 00000 000 1.5 1 0062 10 66 000 14 1000 48 0000 1 0062 10 ı 66 000 14 0063 10 ı 200 5000 0063 10 J 000 3500

000 8

0064 10

*********FURNACE CLASSIFICATION*******

					** ** *																											
ı	FDRY NO.				LAb FIA				CHG	GAS T-0	AFT	C HG D R	FL IN	DX EN	FCE USE	FCE	HLD CAP	ML T RAT	MT UT	PR UT	0 T	VDLUM	BLAS PRES	TEMP	SIZE CH DR	HT DR	CHG PRE				PDWR SPLY	SYS
	0064	10	2	ı	ì										1		000	21													0000	
	006-	10	1	1	4	ı	3	ı	2	ı	2	1	0	0	1	84	000	35				16000		1600	96		0	1500	1	600	0000	1
	0065	10	2	1	4	1	3	1	2	1	2	1	0	0	1	84	000	35				16000		1600	96		0	1500	1	600	0000	2
	0065	10	3	1	4	1	3	ı	2	1	2	1	0	0	1	102	000	40				16000		1000							0000	3
	0065	10	4	ı	ı	2	3	2	2	8		1	0	0	1	108	000	25				16000		750							0000	4
1	0065	10	5	i	1	2	3	2	2	8		1	0	0	1	108	000	25				16000		750							0000	5
1	0065	10	6	1	ı	2	3	2	2	8		1	0	0	1	108	000	25				16000		750							0000	6
1	0065	10	7	4		0	0	0			0	0	0	0			10					00000	00	0000	00	00		0000	0	000		
	0066	10	1	1	1										1		000	35													0000	
(0066	10	2	1	1										1		000	35													0000	
(0066	10	3	1	1										1		000	35													0000	
(0066	10	4	1	4	1	3	2	2	8	0	1	0	0	1	108	000	35				25000		1000				0000	0	000	0000	
(0066	10	5	1	4	1	3	l	2	2	0	1	0	0	1	108	000	40				25000	30	1000	97			0000	0	000	0000	Ł
(0066	10	7	4	1	Ó	0	0	1		0	0	0	0	2	96	15	00	00			00000	00	0000	00	00	0	0000	0	000		
(0066	10	8	2	6	0	0	0	0	0	0	0	0	0	2	000		00	00			00000	00	0000	00	00	0	0000	0	000		
(0067	10	1	1	4	ı	3	1	2	1	ı	i	0	0	1	114	000	45	16		00	25000	72	1200	99	46	0	6000	2		0000	1
(0067	10	2	1	4	1	3	1	2	ı	ı	1	0	0	1	114	000	50	16		00	25000	72	1400	51	35	0	9999	1		0000	2
(0067	10	3	1	4	1	3	2	2	8	0	1	0	0	1	102	000	45	8		00	25000	72	1200	57	37	0	0000	0	000	0000	3
(0067	10	4	3	3	0	5	0	1	0	0	0	0	0	1		33	10	10	2	4	00000	00	0000	00	00	2	0000	0	000	1300	0
(0067	10	5	2	7	0	5	0	ı	0	0	0	0	0	2		60	00	00	16	00	00000	00	3000	00	00	0	0000	0	000	1100	0
(0067	10	6	2	7	0	5	0	i	0	0	0	0	0	2		60	00	00	16	00	00000	00	0000	00	00	0	0000	0	000	1100	0
	0067		7	4	3	0	0	0	1	0	0	0	0	0	2		30	00	00	8	00	00000	00	0000	00	00	0	0000	0	000	3500	0
	0067		8	4	3	0	0	0	1	0	0	0	0	0	2		4	00	00	8	00	00000	00	0000	00	00	0	0000	0	000	3500	0
	8 600		1	1	1	2	3	2	2	8	0	1	0	0	1	102	000	45	16	0 0	00	20000	35	350	85	30	0	0000	0	000	0000	1
C	8 800	10	2	ı	ì	2	3	2	2	8	0	1	0	0	1	102	000	45	16	00	00	20000	35	350	85	30	0	100	l		0000	2

			* **	****	***F(URNA	E CL	ASS I	FICA	TION	1 * * * 1	***	***																		
			FCE						GAS T-O	AFT											****** VOLUM										SYS
0068	10	3	1	ı	2	3	2	2	8	0	1	0	0	1	102	000	45	16	00	00	20000	35	350	85	30	0	100	1		0000	3
0068	TO	4	1	ı	2	3	2	2	8	0	1	0	0	1	102	000	45		00	00	20000	35	350	85	30	0	100	1		0000	4
0068	10	5	3		0	0	0	0	0	ı	0	0	0	1	72	65	40	16	16	00	00000	00	0000	00	00	1	5000				5
0068	10	6	3		0	0	0	0	0	1	0	0	0	3	60	33	10		16		00000	00	c000	00	00	0	0000			1700	
0068	10	7	4	ı	0	0	0	0	0	1	0	0	0	3	108	18	3		16		00000	00	0000	00	00	0	0000			470	
0068	10	8	4	ı	0	0	0	0	0	1	0	0	0	3	108	18	3		16		00000	00	5000	00	00	0	0000			350	
00 58	10	9	4	ì	0	0	0	0	0	1	၁	0	0	3	108	18	3		16		00000	00	0330	0.0	00	0	0000			350	
0069	10	1	1	ı	1	3	ı	2	2	0	1	0	1	ı	114	000	55	16			28000	80	1000		51	0	0000	0	000	0000	5
0069	10	2	1	1	1	3	1	2	2	0	1	0	1	1	114	000	55	16			28000	80	1000		51	0	0000	0	000	0000	6
0069	10	3	1	1	1	3	ì	2	2	0	1	0	1	1	108	000	45	16			20000	56	1000		58	0	0000	0	000	0000	3
0069	10	4	1	ı	1	2	1	2	2	0	1	0	1	ı	108	000	45	16			22000	72	1000		58	0	0000	0	000	0000	4
0069	10	5	1	ì	1	2	1	2	2	0	ı	0	1	1	102	000	35	16			20000	56	1000		46	0	0000	0	000	0000	1
0069	10	6	1	1	1	2	2	2	2	0	1	0	ı	1	102	000	35	16			20000	56	1000		46	0	0000	0	000	0000	2
0069	10	7	1	1	3	1	2	2	8	0	1	0	0	1	72	0 00	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	8	1	ı	3	1	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	9	1	1	3	1	2	2	8	0	1	0	0	ı	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	10	1	ı	3	1	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	11	1	1	3	ı	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	12	1	1	3	1	2	2	8	0	1	0	0	ı	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	13	1	1	3	i	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	14	1	ı	3	1	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	15	1	ı	3	1	2	2	8	0	1	0	0	1	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	16	ı	ì	3	ı	2	2	8	0	1	0	0	ı	72	000	20	16			15000	40			25	0	0000	0	000	0000	
0069	10	17	4	1	0	0	0	0	0	0	0	0	0	3	96	17	00	00	15		00000	00	0000	00	00	0	0000	0	000	3500	
0069	10	18	4	ι	0	0	0	0	0	0	0	0	0	1	108	18	3	8	16	00	00000	00	0000	00	00	0	0000	0	000	6250	

A GENERAL CONTRACTOR CASE CLASSIFICATION REPRESENTANT FORY CONFIGENCE LIN BUT BUT TOP CHG GAS AFT CHG FE ON FCE FCE HLD MLT MT PROUT ******BLAST**** SIZE POOR ON AFT AFT DIS PORT OFF DR. IN EN USE DIA CAP RAT UT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LOC AFT SPLY. SYS NO. 14 NO TYP TYP DES HTG C/O 7-0 3 1 8 00 00 00000 000 1100 0069 10 0069 10 33 00 00 16 00 - 00 000 000 1100 72 33 00 01 60 00 00000 000 1100 0069 10 00 16 00 00000 000 1600 0069 10 22 4 6 00 000 0000 0070 10 0. 1 102 000 50 0' 0000 000 0000 2 . 0070 10 1 102 000 50 0 1000 10000 2,0070 10 1 102 000 50 30070 10 1 102 000 50 000 0000 000 0000 0070 10غ 102 000 50 40070 10 40070 10 20070 10 20071 10 000 0000 96 000 40 16 16 000 0000 0071 10 a 96 000 40 16 16 0071 10 96 000 40 .16 16 000 0000 0071 10 108 000 50 000 0000 0071 10 000 26 00 00 16 24 00000 0071 10 000 26 00 00 16 24 00000 0071 10 n 26 00 00 16 24 00000 0071 10 0000 c00 000 26 00 00 16 24 00000 .0071 10 26 00 00 16 24 00000 000~ 0. 0071 10 54 00 00 16 24 00000 0072 10 O 96 000 30 16 16 00 15000 30 000 0000 0072 10 96 000 30 16 16 00 15000 30 000 0000 0072 10 3 1 4 1 2 2 96 000 30 16 16 00 15000 30 0000 0000

******************* FDRY CD FCE FCE LIN BLT BLT TOP CHG GAS AFT CHG FL DX FCE FCE HLD MLT MT PR OT ******BLAST**** SIZE HT CHG AFT AFT DIS POWR CTL 1-0 NO. TY NO TYP TYP DES HTG C/O DR IN EN USE DIA CAP RAT UT UT UT VOLUM PRES TEMP CH DR DR PRE SIZE LDC AFT SPLY SYS 0072 10 96 000 30 16 16 00 15000 30 0.000 000 0000 O 0072 10 000 0000 92 000 24 16 16 00 12000 30 0072 10 84 000 21 16 16 00 12000 30 000 0000 0072 10 00000 00 000 3500 3 105 15 00 00 16 0072 10 000 3500 105 15 00 00 15 0072 10 105 15 00 00 16 00000 30 000 6500 0072 10 00000 93 ാഠ 000 2000 178 65 00 18 16 0073 10 0 0 1 102 000 45 0073 10 0 0 1 102 000 45 0074 10 48 000 10 0075 10 Λ ı 8 4 0000 0 000 5000 0076 10 ı 36 000 3 0076 10 OÜ 0000 0 000 0076 10 ı 0076 10 UU 0077 10 48 000 10 0077 10 48 000 10 0077 10 78 000 20 0077 10 1 0 0 78 000 20 0078 10 000 22 0080 10 00000 00 0081 10 36 000 4 000 0000 0082 10 68 000 20 000 0000 0082 10 58 000 20 0000 0 000 0000 0083 10 2. ı 42 000 8 ı

*********FURNACE CLASSIFICATION*******

			****	****	**F(JRNAC	E CL	ASSI	IF ICA	TION	***	***	**																		
FDRY NO.			FCE	LIY	BL T	BLT	TOP	CHG	GAS T-D	AFT	CHG	FL	ΟX	FCE USE	FCE DIA	HLD C AP	MLT RAT	MT	PR UT	0T UT	ADFAW	BL AS PRES	TEMP	SIZE CH DR	HT DR	CHG PRE	SIZE	AF T LOC	DIS	POWR SPLY	SYS
0083	10	2	ı	i			ı	2			ı			ı	42	000	8				5000			22						0000	2
008.	10	1	1	4	1	3	ı	2	1	2	1	0	0	ì	96	000	35	16			12000		1000	83		0	3500			00,00	1
0085	10	1	1		2	3	2	2	8	0	1	0	0	1	84	000	24				8000		500	52			0000	0	000	0000	
0086	10	1			3		0	0	0	0	0	0	0	i	000	3					00000	00	0000	00	00		0000	0	000		
0087	10	1	1	ı	3	ì	ı	2	ı	0	1	0	0	ı	48	000	8	3	2	00	4250	32		31		0	0000	0	000	0000	1
8800	10	1	ł	1	3	ı	2	2	8		1	0	0	1	64	000	17				7500			72						0000	1
0088	10	2	i	1	3	1	2	2	8		i	0	0	1	64	000	17				7500			72						0000	2
0089	10	ı	ı	1	1	2	2	2	8		1	0	0	1		000	3													0000	1
0090	10	1	1	1	2	3	ı	2	1	3	1	0	0	ì	72	000	25	16			13600	22	400	72	35	0	80	2	650	0000	1
0090	10	2	i	1	2	3	Ł	2	1	3	1	0	0	ı	72	00c	25	16			13600	22	400	72	35	O	80	0	660	0000	i
0090	10	3	1	1	2	3	ı	2	1	3	1	0	0	1	72	000	25	16			13600	22	400	72	35	0	80	0	660	0000	2
0090	10	4	1	1	2	3	1	2	1	3	1	0	0	1	72	000	25	16			13600	22	400	72	35	0	80	0	660	0000	2
0090	10	5	2	7	0	0	0	0	0	0	0	0	0	2			00	00	00	24	00000	00	0000	00	00	0	0000	0	000	600	
0090	10	6	2	7	0	0	0	0	0	0	0	0	0	2			00	00	00	24	00000	00	0000	00	00	0	0000	0	000	720	
0091	10	1	1	1	2	2	1	2	1	1	2	0	0	1	54	000	22	16	16		10800	32	350	57	37	0	1000	2	648	0000	1
0091	10	2	i	1	2	2	ı	2	1	1	2	0	0	1	54	000	22	16	16		10800	32	350	57	37	0	1000	2	648	0000	i
0091	10	3	1	1	2	2	ı	2	1	ı	2	0	0	ı	54	000	22	16	16		10800	32	350	57	37	0	1000	2	648	0000	2
0091	10	4	1	ì	2	2	1	2	1	1	2	0	0	1	54	000	22	16	16		10800	32	350	57	37	0	1000	2	649	0000	2
0091	10	5	1	ı	2	2	1	2	ı	1	2	0	0	ì	54	000	22	16	16		10800	32	350	57	37	0	1000	2	648	0000	3
0091	10	6	1	1	2	2	ı	2	ι	1	2	0	0	ı	54	000	22	16	16		10800	32	350	57	37	0	1000	2	649	0000	3
0092	10	1	1				1	2						1	66	000	15				7500			50						0000	1
0092	10	2	1											1		000	15													0000	
0092	01	4	3	1	0	5	0	0	0	0	0	0	0	1	000	22					00000	00	0000	00	00		0000	0	000		
0092	10	5	3	1	0	5 ,	0	0	0	0	0	0	0	1	000	22					00000	00	0000	00	00		0000	0	000	3850	
0093	10	1	1	1	2	3	2	2	8	2	ı	0	ì	1	72	000	22	8	00	00	12000	16	730	93	32	0	6000	1		0000	i

5004	٠.								FICA							0	M1 T		•		****			61.15		cuc	457	45.7	015	00.40	C T1
NO.						HTG		LHG	1-0	AFI	-										VOLUM									PO WR SPLY	SYS
0093	13	2	1	ı	2	3	2	2	8	2	ı	0	l	1	72	000	22	8	00	00	12000	16	730	93	32	0	6000	l		0000	2
0093	í)	3	3		0	0	0	0	0	0	0	0	0	3	000	14	00	00	00	8	00000	00	0000	00	00	0	0000	0	000	1825	0
0093	10	4	3		o	0	0	0	0	0	0	0	0	3	000	14	00	00	00	8	00000	00	0000	00	00	0	0000	0	000	1825	0
0094	10	1	1	4	ı	3	1	2	2	0	1	0	0	1	112	000	30	17	17		12000	30	900	99	43		0000	0	000	0000	1
0094	10	2	1	4	ı	3	1	2	2	0	1	0	0	1	112	000	30	17	17		12000	30	900	99	43		0000	0	000	0000	1
0094	10	3	4	ı	э	0	0	1	4	0	0	0	0	1	96	8	4	9	8		00000	00	0000	00	00	0	0000	0	000	3000	2
0095	10	1	1	1	3	1	ı	2	1	0	1	0	0	1	102	000	25				13000			55		0	0000	0	000	0000	1
0095	10	2	ı	ı	3	1	1	2	1	0	1	0	0	1	102	000	25				13000			55		0	0000	0	000	0000	ı
0095	10	3	1	1	3	1		2						ı	102	000	25													0000	
0095	10	4	1	1	3	1		2						1	102	000	25													0000	
0096	10	1	1	1			ı	2	1		2			1	102	000	15				2700			8						0000	1
0097	10	ı	1	1	3	1	1	2	1	0	1			1	28	000	. 4				1725			3			0000	0	000	0000	1
0098	10	ı	1	4	1	3	ı	2	1		1	0	0	1	102	000	40				20000		800	90						0000	1
0098	10	2	1	4	1	3	1	2	1		1	0	0	1	102	000	40				20000		800	90						0000	1
0098	10	3	4	1	0	0	0	0		0	0	0	0	2	240	60					00000	00	0000	00	00	0	0000	0	000	8000	
0098	10	4	3	1	0	5	0	0	0	0	0	0	0	1	000	23					00000	00	0000	00	00		0000	0	000	3200	
0098	10	5	3	1	Э	5	0	0	0	0	0	0	0	ı	000	23					00000	00	0000	00	00		0000	0	000	3200	
0099	10	1	1	4	1	3	1	2	2	0	1	0	0	1	54	000	12	20			6200	64	1000	69		0	0000	0	000	0000	ì
0099	10	2	1	4	ì	3	1	2	2	0	1	0	0	1	54	000	12	20			6200	64	1000	69		0	0000	0	000	0000	ı
0099	10	3	2		0	0	0	0	0	0	0	0	0	3	000	20					00000	00	0000	00	00	0	0000	0	000		
0100	10	1	6	0	0	0	0	0	0	0	0	0	0	1	000	80					00000	00	0000	00	00		0000	о	000		ř
0101	10	1	ı	2			1	2						1	60	000	14				8700			14						0000	ı
0101	10	2	1	2			ı	2						1	60	000	14				8700			14						0000	1
0102	10	1	1						1					ı	72	000	10				4000									0000	1
0103	10	1	1											1		000	10													0000	

									I F I CA																						
						BLT HTG		CHG	GAS I	AFT											VOLUM									POWR SPLY	SYS
0103	10	2	1											ı		000	12													0000	
0103	10	3	4	1	0	0	0			0	0	0	0			3					00000	00	0000	00	00		0000	0	000		
0103	10	4	4	1	0	0	0			0	0	0	0			•6					00000	00	0000	00	00		0000	0	000		
0104	10	1	ı	ı	3	1	1	2	ı	2	2	0	0	1	54	000	9	3			4700	20			35			2	360	0000	ı
0105	10	1	1				1	2						1	54	000	8				6000									0000	1
0105	10	2	1				ı	2						1	54	000	8													0000	
0105	10	3	3		0		0	0	0	0	0	0	0		000	. 5					00000	00	0000	00	00		0000	0	0.00		
0105	10	4	3		0		0	0	0	0	0	0	0		000	•5					00000	00	0000	00	00		0000	0	000		
·0106	10	ı	i	1			2	2	8		i			1	42	000	8							. 5						0000	i
01.07	10	1	1	1	3	ı	2	2	8	0	1	0	0	1	48	4	8	4	4	00	7000	21		27	8	0	0000	0	000	0000	
0108	10	ì	1	2			1	2			1			1		000	8				3800			10						0000	l
8010	10	2	1	i			1	2						1		000	7				3700			10						0000	£
0103	10	3	3		0		0	0	0	0	0	0	0		000		5				00000	00	0000	00	00		0000	0	000		
0109	10	1	1				1	2			1			1	36	000	5				4000			12						0000	1
0109	10	2	1				1	2			1			1	36	000	5				4000			12						0000	2
0109	10	3	1				ı	2			1			1	36	000	5				4000			12						0000	2
0109	10	4	4	1	0	0	0			0	0	0	0		36		5				00000	00	0000	00	00		0000	0	000		3
0110	10	1	1	ı	3	1	ı	2	1	0	1	0	0	1	84	0.00	18				9500						0000	0	000	0000	1
0110	10	2	ı	ı	3	1	ı	2	1	0	1	0	0	1	84	000	18				9500						0000	0	000	0000	1
0110	10	3	6	0	0	0	0	0	0	0	0	0	0	3	000	50	00	00			00000	00	0000	00	00	0	0000	0	000		
©110	10	4	6	0	0	0	0	0	0	0	0	0	0	1	000	50	00	00			00000	00	0000	00	00	0	0000	0	000		
0111	10	1	1											ı		000	12													0000	1
0112	10	1	1	1			2	2	8		ı			1	54	000	10				4600			55						0000	i
0113	10	1	1	1	2	3	2	2	8	0	1	0	1	ı	36	000	10	10	9		3950	16	600	18	27	0	0000	0	000	0000	ı
0113	10	2	1	ı	2	3	2	2	8	0	1	0	1	1	36	000	10	10	9		3950	16	600	18	27	0	0000	0	000	0000	2

FORY	C.D.	FCF				JR NAC Rit								FCF	ECE	HI D	MI T	MT	DQ I	ΩT	*****	RI AS	T***	\$17 F	нт	CHG	AFT	AFT	210	POWR	CTL
						HTG		0.10	1-0												VOLUM										SYS
0113	10	3	6	0	0	0	0	0	0	0	0	0	0	3	000	12	00				00000	00	0000	00	00	0	0000	0	000	0000	0
0113	10	4	2	6	0	0	0	0	0	0	0	0	0	2	000	25	00	10	18		00000	00	0000	00	00	0	0000	0	000	600	0
0114	* 0		1		2	3	1	2	,	0		0	^	,	4.0	000	10				18500		750	80		0	0000	0	000	0000	1
		•	-	1	_		_		ı		1		0	1			_										0000				_
0114	10	2	1	1	2	.3	1	2	l	0	1	0	0	ı	60	000	18				18500		750	80		0	0000	0	000	0000	1
0115	ſΟ	1	1	2							1			ì	36	000	5				2800			31			•			0000	1
0115	10	2	2	2							1			1	36	000	5				2800			31						0000	2
0115	10	3	2	2										1		000														0000	
0115	10	4	1	2										1		000														0000	
0116	10	1	1	4			2	2	8	3	ı	0	0	1	50	000	9	9			4400			54	30	0	1050	1		0000	ı
0116	10	2	1	4			2	2	8	3	1	0	0	1	50	000	9	9			4400			54	30	0	1050	1		0000	2
0117	10	1	ı	1	2	3	1	2	ı	2	ı	0	υ	ı		000	32						750							0000	
0117	10	2	1	1	2	3	,	2	1	2	1	0	0	1		000	32						750							0000	
		•	•	•	•	•	•	-	٠	2	•	Ū	J	•		000	32						750							0000	
0117	10	3	6	0	0	0	0	0	0	0	0	0	0	3	000	34	00	00			00000	00	0000	00	00	0	0000	0	000		
0117	10	4	6	0	0	0	0	0	0	0	0	0	J	3	000	34	00	00			00000	00	0000	00	00	0	0000	0	000		
0118	10	1	ì	4	2	3	2	2	8	0	1	0	С	1	54	000	12				6000			48			0000	0	000	0000	1
0118	10	2	1	4	2	3	2	2	8	0	1	0	0	1	61	000	14				6500			46			0000	0	000	0000	1
0118	10	3	ì	1	2	3	2	2	8		ı	0)	1	66	000	16				8000			61						0000	2
0118	10	4	i	1	2	3	2	2	8		1	0		1	66	000	16				8000			61						0000	2
0118	10	5	1	4	2	3	2	2	8	0	1	0	j	1	60	000	14				6500			55			0000	0	000	0000	3
0119	10	1	1											1		000	7													0000	
0119	10	2	1				1	2						1	78	000	10													0000	1
0120	10	1	1	1	3	1	1	2	ı		1	0	J	1	32	000	6													0000	1
0120	10	2	1	1	3	ı	1	2	1		1	0	0	1	32	000	6													0000	1
0120	10	3	6	0	0	0	0	0	0	0	0	0	,	3	000	6	00	00			00000	00	0000	00	00	0	0000	0	000		
0120	10	4	7		э		0	0		0	0	{										00			-						
-120		~	•		•		•	v		U	U	1				- 5					00000	00	0000	00	00		0000	0	000		

			FCE		OLT	BLT	TOP	CHG	GAS T-0		CHG										****** VOLUM										SYS
0121	10	1	1	1	3	1	2	2	8		L	0	0	ı	72	000	12				7500			53						0000	1
0121	10	2	1	ı	3	ı	2	2	8		1	0	0	1	60	000	12				7500			45			•			0000	2
0121	ſO	3	1	1	3	ı	2	2	8		1	0	0	ı		000	6													0000	
0121	10	4	4	1	3	0	0			0	0	0	0				3				00000	00	0000	00	00		0000	0	000		
0122	10	1	1	1	3	1	1	2	1	1	1	0	0	1	48	000	8				3800			22 .				1		0000	1
0123	10	1	1	1			1	2			ı			1	36	000	6							7						0000	1
0124	10	1	1	2			1	2			1			ı	36	000	4							8						0000	ì
0,125	10	1	1	4	ı	3	1	2	1	2	1	0	0	1	1 08	000	35	9			15500		1000	92		4	5250	ı	484	UU UO	1
0125	10	2	ì	4	ı	3	i	2	1	2	1	0	0	1	108	000	35	9			15500		1000	92		4	5250	ì	484	0000	ì
0125	10	3	1											1		000														0000	
0125	10	4	1											1.		000	20													0000	
01,25	10	5	1											1		000	20													0000	
0125	10	6	I											1		000	12													0000	
0126	10	ı	1	4	1	3	1	2	1	0	ì	0	0	1	102	000	40	17		00	20000	42	1200	99		0	0000	0	000	0000	1
0126	10	2	3		0	0	0	0	0	0	0	0	0	2	000	23		00	00	17	00000	00	0000	00	00	0	0000	0	000	1500	0
0126	10	3	3		0	0	0	0	0	0	0	0	0	2	000	23		00	00	17	00000	00	0000	00	00	0	0000	0	000	1500	0
0126	10	4	3		0	0	0	0	0	0	0	0	0	2	000	23		00	00	17	00000	00	0000	00	00	0	0000	0	000	1500	0
0126	10	5	3		0	Q,	0	0	0	0	0	0	0	2	000	23		00	00	17	00000	00	0000	00	00	0	0000	0	000	1500	0
0127	10	1	1	1			ı	2			1			1	4.0	000	6							8						0000	1
0128	10	1	1	1			ı	2			1			1	36	000	5				1600			5						0000	1
0159	10	1	1				l	2						1		000	9							30						0000	1
0130	10	1	Ţ	2			1	2	1		2			1	45	000	9				4000			30						0000	1
0132	10	1	Ł	1			ı	2	1		1			1	78	000	12	8			7500			35						0000	1
0132	10	2	1	1			1	2	1		1			1	60	000	12	8			7500			35						0000	1
0133	10	1	1	1	3	1	2	2	8	0	1	0	0	1	42	000	7										0000	0	000	0000	

FDRY	ÇΒ	FCE				URNA(BLT								FCE	FCE	HLD	MLT	мТ	PR	o T	*****	BLAS	T****	SIZE	нт	CHG	AF T	AFT	DIS	POWR	CTL
NO.	ŢY	NO	T YP	TYP	DE S	HTG	C/0		1-0		DR	ĮN	EN	USE	DIA	CAP	RAT	UT	υT	UT	VOLUM	PRES	TEMP	CH DR	DR	PRE	SIZE	LOC	AFT	SPLY	SYS
0133	10	2	1	ì	3	1	2	2	8	0	1	0	0	1	42	000	7										0000	0	000	0000	
0173	10	3	7	0	0	0	0	0	0	0	0	0	0			• 5					00000	00	0000	00	00		0000	0	000		
0134	10	1	1	4	1	3	1	2	1		1	0	0	1	84	000	60	16			15000		1000							0000	ì
0134	10	2	1	4	L	3	ı	2	1		1	0	0	1	84	000	60	16			15000		1000							0000	1
0135	10	1	1	2			1	2						1	52	000	4				3000									0000	1
0136	10	1	1	1	2	3	1	2	1	0	1	0	0	1	63	000	16	18		00	5500	14	700	45		0	0000	0	000	0000	1
0136	10	2	1	ı	2	3	1	2	1	0	1	0	0	1	63	000	16	18		00	5500	14	700	45		0	0000	0	000	0000	1
0136	10	3	6	0	0	0	0	0		0	0	0	0	3	000	30	14	00	16	00	00000	00	0000	00	00	0	0000	0	000	0000	0
0137	10	1	1	1	1	3	2	2	8		1	0	0	1	72	000	18				10700		1000	69						0000	1
0137	LO	2	1	1	ı	3	2	2	8		1	0	0	1	72	000	18				10700		1000	69						0000	2
0137	10	3	1	1	1	3								1		000	25						1000							0000	
0137	10	4	1	1	1	3								1		000	25						1000							0000	
0138	10	1	1		3	1	1	2						1	60	000	10				3600			17						0000	1
0138	13	2	1											1		000	6													0000	
0139	10	1	1	1	3	1	1	2						L	60	000	16				6400			50						0000	1
0140	10	1	1	1	3	ì	1	2	1		ı	(-	0	1	54	000	7				5300			37						0000	1
0140	10	2	1	1	3	1	1	2	1		1		0	1	54	000	7				5300			37						0000	1
0141	10	1	1	1	1		ı	2			1			1	54	000	12				6000			40						0000	1
0141	10	2	1	1	1		1	2			1			1	54	000	12				6000			40						0000	1
0141	10	3	4	1	0	0	0			0	0	į	0								00000	00	0000	00	00		3000	0	000		
0141	10	4	3		0		0	0	0	0	0		0		000						00000	00	0000	00	00		0000	0	000		
0141	10	5	3		э		0	0	0	0	0		0		000						00000	00	0000	00	00		0000	0	000		
0141	10	6	3		0		0	0	0	0	0		0		000						00000	00	0000	00	00		0000	0	000		
0141	10	7	3		0		0	0	0	0	0		0		000						00000	00	0000	00	00		0000	0	000		
0142	10	1	1											1		000	5													0000	1

			* **	****	****	JRNAC	E CL	ASS	IFICA	I ION	1 * * *	***	***																		
FDRY NU.				LAb				CHG	GAS T-O	AFT											AOF NW									POWR SPLY	CTL SYS
0143	10	1	1				1	2						1		000	8													0000	1
0:44	10	1	1		3	1	ı	2						1	43	000	9				4800			8						0000	1
0145	10	1	ı	1			l	2	1		1			Ł	48	000	5				4200			21						0000	1
0146	10	1	1	1	3	1	ı	2	1	0	2	0	ı	ı	108	000	55	16	00	00	16500	40		91	38	0	0000	0	000	0000	1
0146	10	2	1	i	3	ı	1	2	1	0	2	0	1	1	108	000	55	16	00	00	16500	40		91	38	0	0000	0	000	0000	1
0147	10	1	1	1			2	1	8	0	0	0	0	1	70	000	20										0000	0	000	0000	
0147	10	2	1	1			2	1	8	0	0	0	0	1	70	000	20										0000	0	000	0000	
0147	10	3	1	1			2	1	8	0	0	0	0	1	70	000	55										0000	0	000	0000	
0147	10	4	1	1			2	1	8	0	0	0	0	1	70	000	55										0000	0	000	0000	
0147	10	5	1	1			2	1	8	0	0	0	0	1	70	000	55										0000	0	000	0000	
0148	10	1	1											1		000	35													0000	
0148	10	2	1											1		000	35													0000	
0149	10	1	1	1										1		000	13													0000	
0149	10	2	1	1										1		000	13						•							0000	
0150	10	1	1	1	S	2	l	1	2	3	2	0	1	1	78	000	38	9			11000	40	700		21	0	4250	l	150	0000	1
0150	10	2	1	1	2	2	2	L	2	3	2	0	1	1	78	000	38	9			11000	40	700		21	0	4250	1	150	0000	1
0151	10	ı	1	4	ı	3	1	2	1	0	1	0	0	1	96	000	25				12500	24	1000	60			0000	0	000	0000	1
0152	10	ı	I	i	3	i	Ĺ	2	1		1	0	0	1	60	000	22				10500									0000	L
0152	10	2	1	1	3	1	1	2	1		1	0	0	l	60	000	22				10500									0000	1
0153		1	ı	ı	2	3	2	2	8	0	l	0	0	1	50	000	22						700				0000	0	000	0000	
0153		2	ı	1	2	3	2	2	8	0	1	0	0	1	50	000	22						700				0000	0	000	0000	
0153	10	3	6	0	0	0	0	0	0	0	0	0	0	3	000	35					00000	00	0000	00	00		0000	0	000		
0153		4	6	0)	0	0	0	0	0	0	0	0	3	000	35					00000	00	0000	00	00		0000	0	000		
0153		5	3		0		0	0	0	0	0	0	0		000	9	3				00000	00	0000	00	00		2000	0	000	1600	
0153	10	6	3		0		0	0	0	0	0	0	0		000	9	3				00000	00	0000	00	00		0000	0	000	1600	

				****														_								•					
	IA		TYP					CHG	GAS T-O	AFT											MUJCV									SPLY	SYS
0154	10	ı	1	ı	3	1	2	2	8	0	1	0	0	ı	48	000	6										0000	0	000	0000	
0.54	10	2	1	1	3	1	2	2	8	0	1	0	0	1	60	000	15									-	0000	0	000	0000	
0154	10	3	ı	1	3	1	2	2	8	0	1	0	0	1	60	000											0000	0	000	0000	
0155	10	1	ì	1										1		000	5													0000	
0156	10	1	1	1	2	3	2	2	8	2	1	0	0	1	66	000	23						750							0000	ı
0157	10	1	1				ı	2			2			1	37	000	4				2700			20						0000	1
0158	10	1	1	1	3	1	2	2	8		1	0	0	1	88	000	21				15000			60						0000	1
0159	10	1	1		3	1	1	2						1	42	000	6				3400			14						0000	l
0160	10	1	1	4	3	1	1	2	1	0	1	0	0	Ł	90	000	34	16	16	00	14000	54		99		0	0000	0	000	0000	1
0160	10	2	1	4	3	1	1	2	1	0	1	0	0	1	90	000	34	16	16	00	14000	54		99		0	0000	0	000	0000	1
0160	10	3	1	1	1		2	2	8	0	1	0	0	1	68	000	23	16	16	00		35		60		0	0000	0	000	0000	0
0160	10	4	1	1			2	2	8	0	1	0	0	1	68	000	23	16	16	00		35		60		0	0000	0	000	0000	0
0160	10	5	1	1	0		2	2	8	0	1	0	0	1	68	000	23	16	16	00	00000	35	0000	60		0	0000	0	000	0000	0
0160	10	6	1	1	9		2	2	8	0	ı	0	0	ı	68	000	23	16	16	00	00000	35	0000	60		0	0000	0	000	0000	0
0160	10	7	2	5	0	0	0	0	0	0	0	0	0	2	000	35	00	00	16	24	00000	00	0000	00	00	0	0000	0	000	600	0
0160		8	2	5	0	0	0	0	0	0	0	0	0	2	000	30	00	00	16	24	00000	00	0000	00	00	0	0000	0	000	600	0
0160		9	2	5	0	0	0	0	0	0	0	0	0	2	000	25	00	00	16	24	00000	00	0000	00	00	0	0000	0	000	600	0
0161		1	_				1	2						1		000					6000									0000	1
0161			1				_	_						1	54	000	4													0000	
0161			7				0	0		0	0	0	0			•5					00000	00	0000	00	00		0000	0	000		
0162		1	1	1		-	1	2		0		_	_	1		000					3500						0000	0	000	0000	1
0163		1	1		3	1		2			1	0	0	1		000					9000									0000	1
0163		2		1	3	1	0	2	1	•	1	0	0	1		000		^			9000	0.0	0000	0.0	00		0000	•	000	0000	1
		1	4			0	•	ı	3	0	0	0	0	1	96		5	8			00000	00	0000	00	00	0	0000	0		4500	
0165	TO	2	4	ı	0	0	0	T	3	0	0	0	0	1	96		5	8			00000	00	0000	00	00	0	0000	0	000	4500	ī

			***	****	***F	URNAC	E CI	LASSI	I F I CA	TION	***	***	***																		
FDRY	CO	FCE	FCE	LIN	BLT	BLT	TOP	CHG	GAS	AFT	CHG	FL	OX	FCE	FCE	HLD	HLT	MT	PR	OT	*****	*BLAS	T****	31 ZE	HT	CHG	AFT			POWR	CTL
NO.	1 A	NO	T YP	TYP	DES	HTG	C\0		1-0		DR	IN	EN	USE	DIA	CAP	RAT	UT	UT	UT	AOĻŪM	PRES	TEMP	CH DR	DR	PRE.	SIZE	LOC	AFT	SPLY	SYS
0165	10	3	2	6	0	0	0	0	0	0	0	0	0	2	000	20	00		8	16	00000	00	0000	00	00	0	0000	0	000	500	0
0165	10	4	•	6	o	0	0	0	o	0	0	0	0	2	000	20	00		А	16	00000	00	0000	00	00	0	0000	0	000	500	0
010)	10	7	~	U	J	·	U	•	v	J	•	U	v	_	000	20	UU		J		00000	00	0000	00	•	•	0000	v	000	700	·
0166	10	1	6	Ó	. 0	0.	0	0	0		0	0	0	1	48		10			•	00000	00	0000	00 -	00					0000	ı
`0167	10	1	1	1	2	3	2	2	8	1	1	0	0	1	70	000	22				10000		·550	40				ļ		0000	• 1
0167	10	2	1	ı	2	3	2	2	8	1	1	0	.0	1	70	000	22				10000		550	40				1		0000	2
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(ING OPERATION

FORY NO.			TOTAL MTL CH	REMELT		PURCH CST I		BR IQ	PUN/ TURN			SODA ASH			CARB COKE				ME/CO RATIO							СО	C 0 2	N2
0001	20	1													000				8									
0001	20	2													000				8									
0001	20	3													000				7									
0001	20	4													000				7									
00 02	20	1													000				9									
0002	20	2													000				9									
0004	21	2	6000	6300	0000	0000	00000	000	0000)	000	00	00	00	000	000	00	0 000	00	00	0	4 0	0	000	0	00	00	00
0004	21	3	39600	21500	0000	0000	16000	000	2000		200	00	00	00	450	500	00	0 000	00	00	0	2 6	0	000	0	00	00	00
0004	21	4	39600	21600	0000	0000	16000		2000						450	500	00	0 000	00	00	0	2 6	0	000	0	00	00	00
0004	21	5	39600	21500	0000	0000	16000		2000		200				450	500	00	0 000	00	00	0	2 6	0	000	0	00	00	00
0005	20	1													000				8									
0007	20	ı	2500	500	0000	1500	500	000	0000	0	000	00	00	00	000	000	00	0 000	9	•6	0	3 2	0	30	2			
0008	20	ı	4000	1500	0000	2200	300	000	0000	5	000	00	00	00	000	000	00	0 000	7	.6	0	3 2	0	100	2			
0009	20	1	22000	8360											000	150			10			3 2			1			
0009	20	2	22000	8360											000	150			10			3 2			1			
0011	20	1													000				8									
0012	20	ı	7000	1300	1000	4500	500	000	0000	10	000	00	00	00	000	000	20	0 000	12	•6	2	2 3	3	120	5	10	14	65
0012	22	2	5400	700	700	0000	4000	000	0000	ŋ	000	00	00	00	000	000	00	0 000	9	• 6	2	1 3	3	120	6	16	12	68
0012	20	3	2500	00000	900	0000	800	800	0000	ð	000	00	00	00	000	35	15	0 000	8	.6	0	1 3	3	120	6	18	L 4	63
0012	22	4	1500	00300	600	0000	900	000	0001	i	000	00	30	00	000	45	00	2 550	5	• 6	0	1 3	3	120	6			
0012	20	5	200	60	0000	0000	140	000	000	:	000	00	00	00	000	12	00	1 1	6	•6	0	1 3	3	120	6			
0013	20	1													000				18	•								SEC
0014	20	ı	2000	00000	687	1250	00000	* 63	000.)					000				7	.7								TIO
0014	20	2	2000	00000	687	1250	00000	* 63	000)					000				7	. 7								SECTION 3
0015	20	1													000				12									PAG

FURY CD F						PURCH STEEL	3R1Q	PUN/ TURN	LIM STN	DOLO	SODA ASH	FLGU SPAR	OTH ER	CARB COKE	FF- SIL	MN	A D	DIT LBS	ME/CO RATIO	SUL CON	D A	Q (; S	LI TI*	ITE-	UP ETH	co	CO2	N2
CO16 2J	ı													000					10										
0016 20	2													000					10										
0018 20	ı	1900	530	0000	0000	400	000	0000	40	000	00	00	00	000	000	00	3	2	,6										
0018 27	2	1300	500	0000	0000	400	000	0000	40	000	00	00	00	000	000	00	3	2	6										
0021 20	1						000	0000						000	000	00	0	000	10	•6	0	4 2	2 0	20	•	6			
0021 22	ì	2275	1000	75	9000	1200	000	0000	120	000	00	30	00	000	000	3		5	7	.6	0	4 ;	2 0	20)	6			
0021 20	2													000	000	00	0 (000	10	- 6	0	4 ;	2 0	20)	6			
0021 22	2	2275	1000	75	0000	1200	000	0000	120	000	00	30	00	000	000	3		5	7	. 6	0	4 2	2 0	20)	6			
0022 20	1	2500	1000	100	0000	1400	000	0000											00	00		C	1				00	00	00
0022 20	2	2500	1000	100	0000	1400	000	0000											00	00		C	1				00	00	00
0022 20	3	2500	1000	100	0000	1400	000	0000											00	00		(1				00	00	00
0022 20	4	2500	1330	100	0000	1400	000	0000											00	90		C	1				00	00	00
0025 20	1	4000	400	0000	0000	2000		0000	125	000	00	00	00	000	000	00	0 :	000	6			3					16	3	78
0026 20	1	2000	200	0000	1500	300	000	0000	75	000	00	2	00	000		12	0 (000	8	1.	0	3 ?	0	120)	1			
0026 20	2	2000	200	0000	1500	300	000	0000	75	000	00	2	00	000		12	Э (000	8	1.	0	3 ?	0	120)	1			
0027 20	1	4000	1630	800	0000	1500	100	0000	130	000	00	ა ა	00	000			0	000	7	•6	2	2	0	30		2		8	
0029 20	L	3000	750	* 62	1050	450	750	0000		125			25						6		0	3 ?	3 0	60)	6			
0031 23	1	4000	1300	1000	2000				80	000	00	00	00	000	000	00	0 :	000	10		0								
0031 20	2	4000	1000	1000	2000				80	000	00	00	00	000	000	00	0 (000	10		0								
0032 21	1	1935	900	0000	0000	1135	000	0000	60	000	5	00	00	000		65			12	. 6	0	1 2	2 0	60)	1			
0033 20	1													000					7										
0033 20	2													000					7										
0034 20	1			900	3600	4500			200					000					8										
0035 20	i	1800	400	230	200	00000	750	200	120	000	00	5	00	000	40	4	0 (000	8	.6	0	3 3	1 3			5			
0035 20	2	1800	550	230	200	00000	600	200	90	000	00	3	00	000	40	2	0 (000	8	.6	0	3 3	3			5			

PAGE 3 MELTING OPERATION FDRY CO FCE TOTAL REMELT PIG PURCH PURCH BRIQ PUN/ LIM DOLO SODA FLOU OTH CARB FE- MM ADDIT ME/CO SUL D Q C S LITE-UP CO CO2 N2 T LBS RATIO CON A S M P TIME METH TURN STN MITE ASH SPAR ER COKE SIL NO. TY NO MIL CH IRON CST I STEEL 40 2 0 000 .6 0 3 3 3 5 550 0035 20 3 1800 230 200 00000 600 200 90 000 00 3 00 000 8 000 0036 20 6 0037 20 000 90 1 ٨ 1 4 0 000 0350 2 0042 20 1 4000 1200 0000 0000 800 0000 000 12 00 00 00 000 136 0043 20 1 1500 475 500 00000 125 0000 80 000 000 6 00 0 000 0 3 2 0 60 400 00 00 00 0045 20 000 9 1 0045 20 000 0045 20 000 0045 20 000 0045 20 000 0045 20 000 0045 20 7 000 0045 20 000 00 0046 22 00 00 00 600 000 00 0 000 4 3 0 000 00 00 00 0046 22 2 30000 15000 4 3 0 000 00 00 0000 0000 15000 000 0000 000 000 00 00 00 600 000 00 0 000 00 00 0 3 30000 15000 0000 0000 4 3 0 000 0 00 00 00 0046 22 15000 000 0000 000 000 600 000 00 0 000 00 00 00 00 00 00 00 00 0046 22 00 00 00 600 000 00 0 000 00 00 4 3 0 000 0 0047 20 1 000 10 0047 20 000 10 0048 20 1 32000 15000 0000 5000 10000 000 1000 000 00 00 00 550 250 25 00 00 3 3 0 000 00 00 00 0049 20 1 000 3 12 14 74 0052 20 1 4000 90 2 1500 1600 800 00000 +75 0000 60 000 00 00 000 000 00 0 000 10 2 2 . . 6 0052 20 2 4000 1500 1600 800 90 2 12 14 74 00000 \$75 0000 60 000 00 00 00 000 000 00 0 000 10 .6 2 2 0 0052 20 3 4000 1500 1600 12 14 74 800 00000 *75 0000 60 000 000 00 0 000 10 .6 2 2 0 90 2 00 00 00 000

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FDRY			TOTAL MTL CH	REMELT			PURCH STEEL	BRIQ	PUN/ TURN				FLOU Spar		CARB COKE		MN	ADDIT T LBS	ME/CO RATIO	SUL	D A	Q (: S	LITE	-UP ME TH	co	CD2	N2
0061			3000	1300		300	€.	000	0000		60	00	00	00	000	000	6	0 000	9	. 6	0	1	0	90	2			
0062		1													000				10									
0062		2													000				10									
0065		1		00000	0000	3170	2700	000	0000	170	000	00	00	00	000	170	00	0 000	8			2	2 0					
0065		2		00000	0000		2700		0000			00	00	00	000			0 000	8				2 0					
0066			1 0000	4000	1450		2000		0000	• • •					000		••	• • • • •	-									
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006,7 0070		1													000				5	••					•			
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0072		5													000									120	2			00
0072		6																						120	2	00		00
0074		1													000									120	2	00	UU	00
74		1													000				8									
99.76 9981		1													000				7									
0083		1													000				5									
(j)															000				9									
0083	20	2													000				9									

MELTING OPERATION PAGE 5

FDRY CD F		TOTAL MTL CH	REMELT			PURCH STEEL	BRIQ			DOLO				CARB COKE											TE-UP		CO2	N2
0084 20	ı	4000	00000	0000	400	3600	000	0000	300	000	00	30		000	150	00	4	10	7	. 6						11	13	76
0085 20	1	2950	1000	900	0000	1050	000	0000	120	000	00	00	00	000	50	00	0 (000	11							3	13	82
0087 20	l	1000	450	275	0000	275	000	0000	26	000	00	00	00	000	5	3	0	000	8	. 5	0	4 1	. 1	90	6			
0088 20	ı	2000	00000	*250	150	1600	000	0000	70	000	00	4	00	000	8	12	0	000	8					45	ı			
0088 20	2													000					8									
0090 20	ı	4200	1475	1200	260	600	500	*125	125	000	00	00	00	000	000	00	0 (000	11	• 6	0	3 3	0	45	1			
0090 20	2	4200	1495	1200	260	600	500	+125	125	000	00	00	00	000	000	00	0 (000	11	.6	0	3 3	0	45	1			
0090 20	3	4200	1495	1200	260	600	500	•125	125	000	00	00	00	000	000	00	0 (000	11	.6	0	3 3	0	45	l			
0090 20	4	4200	1495	1200	260	600	500	* 125	125	000	00	00	00	000	000	00	0 (000	11	.6	0	3 3	0	45	1			
0091 20	1	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	.6	0	3 3	0	90	1			
0091 20	2	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	•6	0	3 3	0	90	1			
0091 20	3	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	.6	0	3 3	0	90	1			
0091 20	4	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	.6	0	3 3	0	90	1			
0091 20	5	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	• 6	0	3 3	0	90	1			
0091 20	6	4000	1400	900	500	1000	200	0000	110	000	00	00	00	000	26	20	5	30	11	. 6	0	3 3	0	90	1			
0092 21	1													000					9									
0093 20	1	5500	1100	0000	1650	2750	000	0000	70	000	00	00	00	000	75	00	0 (000	8	• 6	2	1 3	0	120	2		13	
0093 22	1	6000	1200	0000	0000	4800	000	0000	90	000	00	00	5	000	100	00	0 (000	18	. 6	2	4	0	120	2			
0093 20	2	5500	1100	0000	1650	2750	000	0000	70	000	00	00	00	000	75	00	0 (000	8	•6	2	1 3	0		2		13	
0093 22	2	6000	1200	0000	0000	4800	000	0000	90	000	00	00	5	000	100	00	0 (000	8	•6	2	4 3	0	120	2			
0093 20	3	0 0000	00000	0000	0000	00000	000	0000	000	000	00	00	13	000	000	00	3	11	00	00	0	0 (0	000	0	00	00	00
0093 22	3	0 00 00	00000	0000	0000	00000	000	0000	000	000	00	00	8	0 00	000	00	3	13	00	.00	0	0 (0	000	0	00	00	00
0093 20	4	00000	00000	0000	0000	00000	000	0000	000	000	00	00	13	000	000	00	3	11	00	00	0	0 (0	000	0	00	00	00
0093 22	4	00000	00000	0000	0000	00000	000	0000	000	000	00	30	8	300	000	00	3	13	00	00	0	0 (0	000	0	00	00	00
0094 20	ı	6500	1400	0000	2500	2600	000	0000	000	100	00	00	00	000	35	00	0	000	9	• 6	2	3	0	240	2	16	13	71

			TOTAL MTL CH	REMELT		PURCH CST I		BRIQ			DOLO				CARB COKE					ME/CO RATIO							ÇO	COS	N2
0094	22	1	5000	750	0000	0000	4250	000	0000	000	80	00	00	00	000	40	00	0	000	6	.6	2 2	3	0	240	2	16	13	71
0094	20	2	6500	1400	0000	2500	2600	000	0000	000	100	00	00	00	000	35	00	0	000	9	. 6	2 3	3	0	240	2	16	13	71
0094	22	2	5000	750	0000	0000	4250	000	0000	000	80	00	00	00	000	40	00	0 (000	6	.6	2 2	3	0	240	2	16	13	71
0094	20	3	7200	1550	1550	0000	4100	000	0 00 0	000	000	00	00	00		000	00	0 (000	00	00	0.0	3	0	000	0	00	00	00
0095	20	1													000							3							
0095	20	2													000							^3							
0096	20	1													000					8									
0097	2,3	1	800	00000	400	0000	400	000	0000						000														
0098		1													000					7									
0098		2													000					7									
0029	20	1	1100	200	0000	700	00000	200	0000	60	000	00	00	00	000	000	00	0 (000	6		0 3	3	0		2	3	2	80
0099		2	1100	200	0000	700	00000	200	0000	60	000	00	00	00	000	000	00	0 (000	6		0 3	3	0		2	3	2	С8
0102	20	1													000							3							
0104		1	2000	800	400	200	600			60	000	00	00	00	000					6	.6	3 4	2	0		2			
0106	20	1													000					8									
0107	20	1	1200	600	0000	0000	600	000	0000	12	. 5	4			000	8	4	0 (000	8	55	3 4			20	l			
0108	20	ı													000					9									
0108	20	. 2													000					9									
0109	20	1													000					10									
0109	20	2													000					10									
0109	20	3													000					10									
9110	21	1	2945	1800	0000	0000	1145	000	0000	60	000	00	00	00	000	55	00	6	4	10		3	•						
0110	21	2	2945	1800	0000	0000	1145	000	0000	60	000	00	00	00	000	55	00	5	4	10		3	3						
0113	21	1	1000	580	*80	0000	340	000	0000	20	000	•5	00	00	000	•75	.3	7	• 5	11	. 7	0 4	. 1	0	60	2			
<u>0113</u>	21	2	1000	580	*80	0000	340	000	0000	20	000	.5	00	00	000	.75	. 3	7	. 5	11	. 7	0 4	1	0	60	2			

MELTING OPERATION PAGE 7

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			TUTAL MTL CH	4E4ELT			PURCH STEEL	BRIQ					FLOU SPAR							ME/CO RATIO							CO	CO2	N2
0114	20	1	2000	1200	200	0000	600	000	0000	75	000	00	00	00	000	000	00	5	23	9			4						
0114	20	2	2000	1200	200	0000	600	000	0000	75	000	00	00	00	000	000	00	5	23	9			4.						
0115	20	1							•						000					6									
0115	20	2													000				•	6									
0116	20	1	1430	730	400	0000	300	000	0000	60	000	00	00	00	000					6							. 2	8	71
0116	20	2	1430	730	400	0000	300	000	0000	60	000	00	00	00	000					6							. 2	8	71
0122	20	1	1280	450	320	125	385	000	0000	60	000	00	00	10	000					8			4 ;	2		2			
0123	20	. 1													000					10									
0124	20	1													000					10									
0125	20	1													000					10			3				16	11	72
0125	20	2													000					10			3				16	11	72
0126	20	1	10000	500	0000	3500	5000	000	900	400	000	00	18	00	000	260	00	0	000	9	. 6	3	2 2	2 0		5			
0126	22	1	1 0000	1000	0000	0000	9000	000	000	400	000	00	18	00	000	210	00	0	000	9	.6	3	2 2	2 0		5			
0127	20	1													000					5									
0128	20	1													000					4									
0130	23	1													000					9									
0132	20	ı	1933	735	0000	1265	00000	33	0000	66	000	00	00	00	000					10									
0132	20	. 2	1933	735	0000	1265	00000	33	0000	66	000	00	00	00	000					10									
0136	23	1	2500	1250	400	750	00000	100	0000	80	000	00	4	00	000	000	00	5	25	12									
0136	21	1	2500	1650	0000	0000	750	100	0000	30	000	00	1	00	000	000	5	5	8	8	.5	2	2 :	3 0	60	2			
0136	2)	2	2500	1250	400	750	00000	100	0000	80	000	00	4	00	000	000	00	5	25	12									
0136	21	2	2500	1550	0000	0000	750	100	0000	30	000	00	1	00	000	000	5	5	8	8 -	. 5	2	2	3 0	60	2			
0137	20	1													000					10									
0137	20	2													000					ĩo									
0141	20	1													000					7									

MELTING OPERATION PAGE 8

FORY CO		TOTAL HTL CH	REMELT		PURCH CST I		BRIQ							CARB COKE		MN		ME/CO RATIO							co	C02	NZ
0141 20	2													000				7									
0145 20	1													000				10			4						
0146 20	1	8000	560	1760	3680	2000	000	0000	160	000	00	00	00	000	25	00	0 000	9	. 6	3	4 2	2 3	120	2			
0146 22	1	8000	800	3200	0000	4000	000	0000	160	000	00	00	00	000	25	00	0 000	8	.6	3	4 2	2 3	120	2			
0146 20	2	8000	560	1760	3680	2000	000	0000	160	000	o o	00	00	000	25	00	0 000	9	-6	3	4 2	2 3	120	2			
0146 22	2	8000	800	3200	0000	4000	000	0000	160	000	00	00	00	000	25	00	0 0,00	8	.6	3	4 ;	2 3	120	2			
0150 20	ı	6000	900	1500	2700	900	000	0000	130	000	00	00	00	000	000	00	7 20	11	•6	2	1	1 0		6	8	17	76
0150 22	1	5000	500	2250	0000	2250	000	0000	130	000	00	00	00	000	20	00	7 000	10	.6	2	4 1	0		6.			
0150 20	2	6000	900	1500	2700	900	000	0000	130	000	00	00	00	000	000	00	7 20	11	.6	2	1 !	١ ٥		6	8	17	76
0150 22	2	5000	500	2250	0000	2250	000	0000	130	000	00	00	00	000	20	00	7 000	10	.6	2	4 1	0		6			
0151 20	1	7060	5250	1050	700	00000	000	0000	200	000	00	00	00	000	60	00	0 000	9	-6		i	2					
0157 20	1													000				£									
0,158 20	1													000				9									
0160 20	1	5000	2300	600	0000	1900	500	0000	150	000	00	00	00	000		99		8		0	1 3	0		2			
0160 20	2	5000	2000	600	0000	1900	500	0000	150	000-	00	00	00	000		99		8		0	1 3	0		2			
0160 23	3	3764	1700	500	0000	1500	64	0000	90	000	00	00	00	000		30	0 000	9		0	1 3	3 0		2			
0160 20	4	3764	1700	500	0000	1500	64	0000	90	000	00	00	00	000		30	0 000	9		0	1 3	0		S			
0160 20	5	3764	1700	500	0000	1500	64	0000	90	000	00	00	00	000		30	0 000	9		0	1 3	0		2			
0160 20	6	3764	1700	500	0000	1500	64	0000	90	000	ΟÒ	00	00	000		30	0 000	9		0	1 3	3 0		2			
0162 20	t	1000	200	300	200	300	000	0000	000	000	00	00	00	000	000	00											
0163 20	1	2025							70					000				8									
0163 20	2	2025							70					000				8									
0165 20	1	1 0000	3500	1000	1700	3000	000	700	000	000	00	00	00	100	120	00	8 30	00	00	3	4 3	0	000	0			
0165 22	1	10000	4300	2000	0000	3000	000	700	000	000	00	00	00	100	120	00	8 20	00	00	3	4 3	0	000	0			
0165 20	2	1 00 00	3600	1000	1700	3000	000	700	000	000	00	00	00	100	120	00	8 30	00	00	3	4 3	0	000	0			

									MEL	TING	OPER.	47 E ON											PAGE	9	
			TOTAL MTL CH		_		PURCH BRIQ STEEL					_												CO CO2 N	12
0165	זי	2	10000	4300	2000	0000	3000 000	700	000	000	00	00	00	100	120	00	8 20	00	00	3	4 3	Ō	000 0		
0166	20	1	14000	6300	5600	0000	1600 LBP	ERHR				CO	AL	100	OL B	PE	R TON				4		3		
0167	20	1	2400	2250	0000	0000	150 000	0000	120	000	00	00	00	000	17	00	0 000	8			4 3	į	3		

4 3 3

CONTROL SYSTEM

FORY CO CTL SYS	FIES YEAR GAS GA CONT INST VOLUME TE	S PRES HEIGHT MP DROP EXH STK		CONSUMPTION GAS RECIRC				OUT (CATCH	COLL EFF	
0001 30 1 4	123 58 104000 4	00 8 87	0000	72	2 0	0		. 28		80	15
0002 30 1 27	12 54 45300 5	50 4 46	9999	25	2 0	0		.039		90	12
0004-30-1-5	345 200000 1	20 9 13	0000 0000	000 0000	2 3	3			6	99	
0005 30 1 26	1 51 11500 2	00 4	250		2 3	2		.008		99	7
0006 30 1 6	1 68		0000		3						
0007 30 1 26	1 67 54000 5	00 6 83	7400 0000	30 30	2 3	2	.86		26	99	20
0008 30 1 26	1 67 38000 5	00 7 46	3300 0000	20 20	2 3	2		-69	11	99	12
0009 30 1 24	12 67 84000 4	50 125	8000 0000	000 0000	2 0	0 .	•65	.13	11	80	
0009 30 2 24	3 67 84000 4	50 125	8000 0000	000 0000	2 0	0	.678	.163	13	75	13
6 1 66 0100	1 69		0000		3						
0011 30 1 26	1 65 46000 4	50 6	7800	20	2 3	2					
0012 30 1 4	12 55 101000 5	00 6 82	0000	200	2 0	0	•5	.075		85	40
0013 30 1 25	1 63 34000 1	63 12 60	500	140	2 0	0					18
0014 30 1 4	12 68 48000 4	00 125	00 00	000 0000	2 0	0	.311	. 153		51	6
0015 30 1 5	1 67 11000 15	00 15	0000	2200	0	0		. 25		80	6
0016 30 1 23	1 58	00 61	3300	175	2 0	0					
0016 30 2 23	2 58	00 61	3300	175	2 0	0					
0017 30 1 6	12 16500 5	00			3	2					
0018 30 1 3	1 56 32050 14	35 00 45	0000	20	0	0	1.06	.159		85	6
0018 30 2 3	2 56 32050 14	35 00 45	0000	20	0	0					
0019 30 1 3	1										
0019 30 2 3	2										
0021 30 1 35	12 66 24000	76	0000 175	250 0000	2 0	0		-269	6		18
0023 30 1 4	1 59 36000 5	00	0000		0	0					
0024 30 1 5	12 67		0000		0	e					

CONTROL SYSTEM PAGE 2

				FCES CONT					HEIGHT EXH STK				SUMPTEON RECIRC			AIR/CLOTH RATIO	INLET CONC	OUT CONC	CATCH	COLL EFF	MLT RTE
002	3 3 0	i	3	1	67													.314			20
002	3 30	1	24	12	68	46500	389	8	125	8000	0000			2	0	0	.46	.174	19	6 5	10
002	7 30	1	3	ı	68			00	76	0000	2 00	000	180	2	0	0					
002	30	1	235	1	69	82000	300	40	125		150	000	0000		0	0	1.06	-12	17	93	16
003	30	1	35	12	69	19500	150	45	100	0000	1100	200	1150		0	0		.177			30
0032	30	1	23	1	58		1000	00	70	2000	350	000	320	2	0	0		. 37	20	90	16
003	3 3 0	1	25	1	66	96000	180	58	107	4240			3000	1	0	0		.016			
0033	30	2	25	2	66	96000	120	3	107				3000	2	0	O					
0034	30	1	5	1	65	34500	122	70	124				300	2	0	0		• 05		95	50
003	30	1	25	1	64	25000	140	35	1 75		450		400	2	0	0	2.26	.13		95	12
003	30	2	3	2	47	20000		00		0000	70	000	0000	2	0	0					
003	30	3	3	3	47	20000		00		0000	70	000	0000	2	0	0					
0036	30	1	6	1	51	13200	450	6		60			12		3	2					
0037	30	1	6	1	50	13500	425	6						2	3		•6			99	4
0036	30	1	•	ı	58	84000	500	9	71					2	0	Ò					
0039	30	1	5	1	70										0	0					
0041	30	1	6	1	57										3						
0042	30	1	23	1	69	50000	2000		99	4200	360	20		2	0	0					
0043	30	1	7	1	70	20000	410	4	50	9000	0000	42	20	2	0	0 .	2.0	.029	38.	98	9
0045	30	1	5	78	63	50000	155	14	70	418			150	2	0	0	1.24			50	26
0046	30	1	Ĝ	12	68	100000	250	13	35	0000	0000	000	0000	2	2	· 3	.85			99	
0046	30	2	6	34	68	100000	250	13	35	0000	0000	000	0000	2	2	3	.85			99	
0047	30	1	24	12	58	60000	500	6	90	100					0	0					
0048	30	1	6	ı	65	32000	250	4	27	0000	0000	000	0000	2	2	2			18		7
0049	30	1	5	1	67			6							0	0					

CONTROL SYSTEM PAGE 3

										COM NO	. 313										-
			SAZ						HEIGHT EXH STK				UMPTION RECIRC			AIR/CLOTH RATIO	INLET	OUT CONC	CATCH	COLL EFF	MLT RTE
0050	30	1	5	1	68	27500		50							0	0	1.09	.034		97	
0051	30	1	6	12																	
0052	30	ı	3	2	69	26000	450	00	86	0000	150	000	0000	2	0	0		. 165			8
0052	30	2	23	4	69	24000	2200	00	98	0000	520	000	0000	2	0	0					
0054	30	1	5	1	69										0	0					
0056	30	1	4	1	69										0	0					
0058	30	1	4	1	57										0	0		-15			
0058	30	2	4	2	57										0	c		-15			
0061	30_	. 1	4	12	69	85000	500		80	750	0000	70		2	0	0			22	81	20
0062	30	ı	4	12	66	72000	500	4	85	750			40	2	0	0		.227		80	17
0063	30	1	6	12	67	52000															
0065	30	1	25	1	69	80000	185	80		3000	1600	400	1800		0	0		.047			30
0065	30	2	25	2	69	80000	185	80		3000	1600	400	1800		0	0		.035			30
0065	30	3	3	3											0	0					
0065	30	4	3	4											0	0					
0065	30	5	3	6											0	0					
0066	30	1	5	5											0	0					
0067	30	ì	125	1	66	27000	600	53	120	6000		999	785	1	0	0	12.9	.079		99	45
0067	30	2	1 25	2	69	50000	1900	53	98	9999			2200	ì	0	0	5.83				50
0067	30	3	3	3	55				98		600				0	0					
0068	30	1	3	1	64	30000	800		100	0000	600			2	0	9					
0068	30	2	23	2	64	30000	800		100		600			2	0	0					
0068	30	3	23	3	64	30000	800		95					2	0	0					
0068	30	4	23	4	64	30000	800							2	0	0					
0068	30	5	2	5	69	35000	1200							2	0	0					

CONTROL SYSTEM

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										00.11.150	. 3.3.										•
			SYS	FIES					HEIGHT EXH STK	AFTBURN SIZE			MECIRC			AIR/CLOTH RATIO	INLET	OL'T CONC	CATCH	COLL	MLT RTE
0069	30	1	236	5	70	35500	500	18	85		40			2	3	2	4.0	•02	15.	99	
0069	30	2	236	6	70	35600	500	18	85		60			2	3	2	4.0.	.02	15.	99	
0069	30	3	35	3	69	22400	170	72	120	0000	300	23	850	2	0	0	4.0	.05	15.	99	
0069	30	4	35	4	68	26600	170	72	120	0000	300	23	850	2	0	0	4.0	.05	15.	99	
0069	30	5	35	1	68	29000	170	72	120	0000	450	30	800	2	0	0	4.0	.05	15.	99	
0069	30	6	35	2	70	29000	170	72	120	0000	350	30	1000	2	0	o	4.0	•05	15.	99	40
0070	30	1	5	1	66										0	0		.05		99	
0070	30	2	5	2	67										0	0		.05		9	
0070	30	3	5	3	67										0	0		. 05		79	
0070	30	4	5	4	68										0	o		•05		99	
0071	30	ı	5	ı	65	027000	110	80	129	0000	1000		2700	2	0	0	6.67	.075		99	35
0071	30	2	5	2	65	02 7000	110	80	129	0000	1000		2700	2	0	0	8.8	.051		99	45
0071	30	3	5	3	65	027000	110	80	129	0000	1000		2700	2	0	0		-05			
0071	30	4	5	4	70	27000	110	80	150	0000	800		2500	2	0	0					
0072	30	1	3	1					81	0000	750			2	0	0					
00,72	30	2	3	2					81	0000	750			2	0	0					
0072	30	3	3	3					81	0000	750			2	0	0					
0072	30	4	3	4					81	0000	750			2	0	0					
0072	30	5	3	5					81	0000;	750			2	0	0					
0072	30	6	3	6					81	0000	750			2	0	0					
0072	30	7	6	10	68	110000	265	12	107	0000	0000			2	2	3					
0073	30	1	4	12	64	126000	500								0	0					
0074	30	1	3	1	57			00	50	50			200	2	0	0					
0075	30	1	6	1																	
0076	30	1	26	1	52	12500	450	6		500			3	2	3	2	•6				

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										CONTROL	. 313	167								, , , ,	,
			LAb 2A2		YEAR INST		GAS TEMP		HEIGHT EXH STK	AFTBURN SIZE	WATER DUST	CONS GAS	UMPTION RECIRC	NOISE CUNTR	F I L MED	AIR/CLOTH RATIO	INLET CONC	OUT CONC	CATCH	COLL EFF	MLT RTE
0077	30	1	4	12	57	84000	500								0	0					
0077	30	2	4	34	57										0	0					
0080	30	ı	6	1																	
0081	30	1	6	1	52	32000	500	6		2000			35	2	3	2					
0083	30	1	23	1	57			00	60	2300			400	2	0	0					
0083	30	2	4	2	57				60	2300			400	2	0	0					
0084	30	1	6	1		40000	175			7000	0000	000	0000		2	2	4.0	. 8	35	80	30
0085	30																8.35	.154	23		24
0087	30	1	5	ı	69	21100	1500	25	52	0000	420		395	2	0	0					
0088	30	1	3	1	57	69000	300	25	93						0	0		-144			17
0088	30	2	3	2	57	69000	300	00	93						0	0		. 144			17
0089	30	1	3	1																	
0090	30	1	24.	12	61	96 000	500	5	100	2400	0000	000	0000	2	0	0	1.98	.83	6.8	51	17
0090	30	2	24	34	61	96000	500	5	100	2400	0000	000	0000	2	0	0	1.98	.83	6.8	61	17
0091	30	1	24	12	59	84000	400	6	94	1000	0000	220	0000	2	0	0		.27		86	21
0091	30	2	24	34	60	84000	400	6	94	1000	0000	220	0000	2	0	0		. 30		84	21
0091	30	3	24	56	61	84000	.400	6	94	1000	0000	220	0000	2	0	0		.17		88	21
0092	30	1	4	1		60000	500								0	0	.97	.16		84	
0093	30	1	23	1	58				96		200		0000	2	0	0					
0093	30	2	23	2	60				96		200		0000	2	0	0					
0094	30	ı	5	12	67	33000	300	11	84					2				.7			
0094	30	2	5	3	55	16600	900	6			68	000	68	2							
0095	30	1	25	12	66	36000	130	26			500		450	2	0	0		.16			20
0096	30	1	26	1	64	16000	540	7	70	3000				2	3			-148			20
0097	30	1	5	1	68	7200	110	20	40	0000	-			2	0	0		.110			

FORY CO			S FOES P CONT					HEIGHT EXH STK	AFTBURN SIZE	WATER DUST	CONS GAS	SUMPTION RECIRC	NOISE CONTR	F I L MED	AIR/CLOTH RATIO	INLET		CATCH COL	
0098 30	0 1	26		61	100000		7	70						3	2			99	40
0099 30) 1	35	12	69	19380	80	25	104	0000	L90	60	250	2	0	0		.27		11
0101 30) l	25	12	67	30000	170	40							0	0	1.48	.04	98	
0102 30) 1	5	1	67										0	0				
0104 30) 1	26	1	60		420	4	60		0000		0000	2	3					
0105 30) 1	4	12	57	33000	500								0	0				
0106 30) i	3	1	66		1800	00	6 5				300	2	0	o [·]				9
01 08 30) 1	6	12	63	24000	530	6	61				25		3	2		. 243		9
0109 30	0 1	45	12	66	4500	1500	21	42				75		0	0			80	5
01.09 30	2	45	23	66	4500	1500	21	42				75		0	0			8 0	5
0109 30	3	45	4	66	4500	1500	28	42				75		0	0			80	5
0110 30	ı	5	12	66	25000	70	25			700		650		0	n		. 153		20
0112 30) 1	3	1	52			00	68				175		0	0				
0113 30) 1	1	1				00	52	0000	0000	oco	0000	2	0	0	.95			
0113 30	2	1	2				00	52	0000	0000	000	0000	2	0	0	.95			
0114, 30) 1	3	12	69			40	125						0	0				
0115 30) 1						00	56				200	2	0	0				
0115 30	2						00	56				200	2	0	0				
0116 30	ı	23	ı		23000	800		45	3150					0	0		.88		
0116 30	2	23	2		23000	800		45	3150					0	0		.88		
0118 30) 1	3	12	55		2000	00					250		0	0				
0118 30	2	3	34	63		2000	00					250		0	0				
0118 30	3	3	5	68			00							0	0				
0119 30) 1	5	2											0	0				
0120 30) i	6	12											3					

PAGE 7 CONTROL SYSTEM FORY CO CTL SYS FOES YEAR GAS GAS PRES HEIGHT AFTBURN WATER CONSUMPTION NOISE FIL AIR/CLOTH INLET GUT CATCH COLL MLT

NO. TY SYS TYP	CONT INST VOLUME T						RECIRC			RATIO	CONC		CAICH	EFF	RTE
0121 30 1 3	1 64	00					250		0	0					
0121 30 2 3	2 64	00					250		0	0					
0122 30 1 5	1 70 16600	158 25	81		330	000	300	2	0	0		•024			8
0123 30 1 6	1 51 13000	500	20				•	2	3						6
0124 30 1 26	1 66	500 3						2	3						4
0125 30 1 5	12 67 33000	80 88	78	9999	750		750		0	0		.05	24	99	35
0126 30 1 5	1 70 43800	150 26	114	0000	450	350	550	2	0	0					
0127 30 1 26	1 51 10800	440 4		80				2	3	2					
0128 30 1 26	1 50 12550	450 3		2000			15	2	3						
0129 30 1 6	1 60								3						
0130 30 1 26	1 60 28000	500 6	49	3000			20	2	3					90	9
0132 30 1 25	12 68 30000	165 45	62	1200			60	2	0	0	5.0	.038		99	10
0134 30 1 5	12 67 60000	50							0	0	2.43	.068		97	
0135 30 1 5	1 69	•							0	0					
0136 30 1 5	12 69 30000	250 7	150	0000	150	75	0000	2	0	0					
0137 30 1 23	1 53	00	77	640			450	2	0	0					
0137 30 2 23	2 53	00	77	640			450	2	0	0					
0138 30 1 5	1 68 18600								0	0		.114			
0139 30 1 24	1 67 72000 1	1400 7	85						0	Ö		.183		83	16
0140 30 1 5	12 69 26000	35							0	0					
0141 30 1 24	12 67 60000	350 8	125	1500			30	2	0	0					
0143 30 1 7	1								0	0					
0144 30 1 5	1 69 20000	25							0	0					
0145 30 1 5	1 67 27000	867 6	50				115	2	0	0					
0146 30 1 5	12 69 40000	155 20		0000	1300	400		2	0	0	1.84	.83	20		52

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CONTROL SYSTEM

FURY CD NO. TY				YEAR INST				HEIGHT EXH STK				UMPTION RECIRC			AIR/CLOTH RATIO	INLET CONC	OUT CÓNC	CATCH	CJLL EFF	MLT RTE
0150 30	1	26	12 -	61	48500	500-	- 9	32		0000	10	0000	2	3	. 2	2.04	•	18	99	37
0151 30	1	5	1	70	20000	1200	50		0000					0	0	1.09	.054		97	23
0152 30	ı	5	12	70	42000		50							0	0	1.19	.033		97	
0156 30	4	3	1,					-							· 					
0157 30	i	26	1	62	15000	480	15		3000			20	2	3	3					
0158 30	1	3	1	65	35000.	800	00	80				600	2	0	0					
0159 30	1	5	1	70	11800		25							0	Ö					
0160 30	1	5	12	68	35000	150	5	120	0000			300	2	0	0	2.16	. 257	34	88	16
0161 30	1	4	1	57	18000	500								0	0					
0162 30	1	5 ·	1		20000	160	16	50	0000	70	30		2 .	0,	. 0,	1-14	. 25		78	
0163 30	1	5	12	68	50700	171	30	82					2	0	0.		.074			
dî65 30		6	12	69	34700	120		50	0000	0000	000	0000	2	2	3			20		
0166 30	1	2	1					97					2	0	0		15.2	LBP	ER	HR
ÓÌ67 30	ı	3	t.			÷		79				300	2	0	O .,		1.43			22
0167 30	2	3	2					79				300	2	O .	0		1.43			2,2
0168 30	1	5	1 .	69				45	0000			25	2	0	0					

ANALYSIS

FDRY CO NO. TY		co					RCENT SO2		PARTICLE PROP. PERCENT :LESS THAN CATCH PROP. PERCENT :LESS THAN 2 5 10 20 50 100 200 500 1000 2 5 10 20 50 100 200 500
0009 40	ı								65 92 98 99
0009 40	2								30 50 65 82 90 99
0014 40	ì								64 82 98 99
0018 40	1	•2	9	10	77			4	2 12 34 92 99 99
0021 40	1	2	7	12	75		4		
0025 40	1	00	12	12	76				
0026 40	1								13 28 45 55 60 74 90 95 98 99 99
0032 40	1	2	4	7	86				54 86 98 99 99
0035 40	1								99
0046 40	1								99 99
0046 40	2								99 99
0061 40	1								8 16 23 31 99 99
0067 40	1	12	10	3					14 15 15 21 99 16 99
0067 40	2	3	16	1					19 25 99
0084 40	_	•2	10	11					
0099 40	-	ı	3	17	75	3			
0113 40									64 90 97 99 99
0113 40	_								64 90 97 99 99
0116 40					•				5 25 53
0116 40									5 25 53
0122 40				_					. 82
0125 40		00		7					
0146 40		00	13	7	80	00			99 99
.0151 40	Ţ								.6 2 3 8 99 99

0162 40 1

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CATCH PROP. PERCENT :LESS THAN

2 5 10 20 50 100 200 500

ANALYSIS

FORY CO CTL GAS ANALYSIS PERCENT PARTICLE PROP. PERCENT :LESS THAN NO. TY SYS CO CO2 O2 N2 H2 SO2 H2O 2 5 10 20 50 100 200 500 1000 C163 40 1 00 5 14 81

O166 40 1 86

O167 40 1 00 9 12 79

O167 40 2 00 9 12 79

ANALYSIS AND COSTS

FDRY NO.	1 A CD	CTL SYS	\$102	CAJ	C AL203	HEMIC MGO	AL ANAL FEO.FE	YSES MNO	PBO	ZND	SNO	SOX	BASIC EQUIP		ENG COST	INST COST	TOT COST	OP ER CO ST	MAENT COST	DEPR	OVE HE 4
0001	50.	1															150				
0002	50	1															375				
0003	50	1															8				
0004	50	1											153	21	000	7	181			14	
0005	50	1															60				
0007	50	1															420				
8000	50	ı											33	000	8	8	49	10	1		
0009	50	1															384				
0009	50	2															384				
0011	50	1															358				
0012	50	1		÷													999				
0013	50	1															44				•
0015	50	1															70				
0016	50	1															45				
0016	50	2															45				
0020	50																200				
0021	50	1											302	108			410				
0026	50	1											150	50	25	50	275	2	ι	28	2
0027	50	1											60	20	5	15	100				
0028	50	1															100				
0029	50	1											284	31	85		610	63	5	50	5 3
0030	50																35				
0031	50	1											80	160							
0032	50	1											26	6	2	5	39	1	3	2	1
0034	50	1					•										500				

ANALYSIS AND COSTS

FDRY CD NO. TY		2012	CAJ	AL203	HEMEC MGD	AL A	ANALYSIS FE MNO	P 80	ZNO	\$40	\$0x	C O 4 B U S	BASIC EQUIP	AUX EQUIP	ENG COST	INST COST	TOT COST	OP ER CO ST	MAINT COST	DEPR		EQUIP CHANG	
0001 50	1																150						
0002 50	ı																375						
0003 50	1																Ą						
0004 50	1												153	21	000	7	181			14			
0005 50	1														•		60						
0007 50	1																420						
0008 50	i												33	000	8	8	49	10	1			00	
0009 50	1																384						50
0009 50	2																384						50
0011 50	1																358						
0012 50	1																999						
0013 50	1																44						
0015 50	1																70						
0016 50	,1																45						
0016 50	2																45						
0020 50	, , ,																200						
0021 50	1												302	108			410						
0026 50	1												150	. 50	25	50	275	2	ı	28	2	1	34
0027 50	1												60	20	5	15	100						
0028 50	ì																100						
0029 50	1												284	31	85		610	63	5 i	50	58	00	1.76
0030 50																	35						APPE
0031 50	1				-								80	160									CTION 6
0032 50													26	6	2	5	39	1	3	2	1	00	15 " 1 _{EH}
0034 50	1																500						PAGE

FDRY CO CTL NO. TY SYS SIO2	CHEMICAL ANALYSIS OMM 37 ,037 CDM CALA CAL	XC2 °ON2 ON5 OE9	COM BASIC BUS EQUIP	AUX EN	G INST ST COST	TOT 0	PER MAINT		EQUIP TOT
0035 50. 1	95.		45	56 L	0 10	121	12 10		
QQ35 50 2			4	1	1 2	8	5 4		
0035 50 3			4	i_ 00	0 2	7	5 4		
0036 50 1						70			
0037 50 1			16	5	3	24	1 4	1 2	8
0038 50 1						120			
0040 50						3/1			
ób42 50 1			18	10 .	1 8	37			25
00'43 50 1						150			46
0045 50 1						125			
0046 50 1						250			
0046 50 2						250			
0047 50 1						150			
óó48 50 1					9	45	1		
ნინი 1									30
0052 50 1			28	40	5 32	105	4 5	5 15	00 29
0052 50 2			26	9 00	0 23	58	3 2	3 8	00 15
0053 50						125			•
0061 50 1			80	30	4 40	154	7 5	6.	00
0062 50 1						132			
0064 50						250			
0065 50 1			350						
0065 50 2			350						
0066 50 1 12.3	11-1								
0074 50 1						40			

ANALYSIS AND COSTS PAGE 3

FDRY C) NO. TY		\$102	CAÇ	C AL203	HEMICA Mgú	L ANAL FEO.FE	212Y Onm	P80	ZNO	SNO	sox		SASIC EQUIP	AUX	ENG COST	INST	TOT COST	OP ER COST	MAINT COST	DEPR		EQUIP CHANG	
0076 50	1																12						
0078 50)																45						
C080 50	1																75						
0081 50	1																81						
0083 50	1																100						
0085 50	1	28.7			1.3	14.7		1.4				24											
0086 50)																5						
0087 50	1												45	7		53	105 -	5	5				
0088 50	1																75						
0090 50	1	56.3	42.									•9					325	12	15	22	7		56
0090 50	2	56.3	42.									•9					325	12	15	22	7		56
0091 50	1												45	14	000		277	16	20	14	20		70
0091 50	2												45	14	000		277	16	20	14	20		70
0091 50	3				•								45	14	000		277	16	20	14	20		70
0094 50	1									, ·			100			90	300	6	8				14
0094 50	2												6	3	2	5	16	3	1				4
0095 50	1																219						
0096 50	1																40						
0097 50	1												30										2
0098 50	1																450						
0101 50																	226						
0106 50																	21						
0108 50																	100						
0109 50																	60						
0110 50	1												51	58	6								

FDRY CO CTL		СН	EMICAL ANA	LYSIS					COM	BASIC	AUX	ENG	INST	TOT	OPER	MAINT DE	PP. O	VER.	EQUIP	tor
NO. TY SYS	\$102 CA3	AL203	MGD FEO, FE	MNO	PBG	ZNO	SNO	SOX				COST	COST	COST	COST	COST	Н	EAD	CHANG	ANN
0112 50 1														38						
0113 50 1	31.8 3.1	-05	8.6	3.7				13.9	27											,
0113 50 2	31.8 3.1	•05	8.6	3.7				13.9	27											
6.15 50 1														25						
0115 50 2														25						
0116 50 1	10. 3.	5.	5. 10.	10.		1.			5										-	
Ó116 50 2	10. 3.	5.	5. 10.	10.		1.			5									,		
0118 50 1														997						
0118 50 2														997						
0118 50 3														997						
0121 50 1														175						
0Ĭ 21 50 2														175						
0122 50 1										45										3
0123 50 1														35						
0124 50 1														30						
0125 50 1										72	136	27		430						
0126 50 1										120	113		23	256						
0127 50 1														40						
0128 50 1														30						
0129 50 1														23						
0130 50 1														100						
0131 50														90						
0132 50 1										110				170						6
0136 50 1										110	45	20	35	210	6	10	6	14	00	36
0137 50 1														. 85						

ANALYSIS AND COSTS PAGE 5

FDRY NO.		CTL SYS	\$102	CAC			AL ANAL		P80	ZNO	SNO	sox		BASIC EQUIP		ENG COST					DEPR		EQUIP CHANG	
0137	50	2																85						
7139	50	1																94						
0141	50	ı																220						
0144	50	1																90						10
0145	50	1																52						
0146	50	ı	20.0	1.0			33.0	1.0	5.0	38.0	2.0			80	490	100		999	30	20	99	99		380
0150	50	1	30.1	1.1	1-4	.99	11.6	5.5	20.	14.7				300	25	25		500	25	10				
0151	50	1												250				389						
0155	50																	140						
0157	50	1																40						
0158	5 Q	,1																71						
0160	50													45	105		50	200	24	10			00	34
. 0162	50	1																150						15
0163	50	1																36 7						20
0165	50	1												60	5	5	30	100	5	1	5	5	•5	16
0166	50	1											3.	8			5	13						14

FORMAT FOR DATA BANK

SECTION I GENERAL FOUNDRY DATA

<u>Abbreviation</u>	Description
FDRY NO. CD TY LOC MET CST	Foundry No. Card Type Location (1st 3 Digits of Zip Code) Type Metal Cast
	Percent Cast
GI MI DI	Grey Iron Malleable Iron Ductile Iron
	Ton/Mo. Melt
GI MI DI	Grey Iron Malleable Iron Ductile Iron
SIZ CLA	Size Classification (Code) 1 Under 10 2 10 - 49 3 50 - 249 4 Over 250

Abbreviation	Description
IND CLA	Industry Classification (Code) 1 Automotive 2 Agricultural 3 Cast Iron Pipe 4 Industrial and Electrical Equipment 5 Valves and Fittings, Refrigeration 6 Jobbing 7 Railroad
WT RG	Weight Range of Castings
PRO CST	Basic Product Cast
NOD	Nodularization (Code) O None or does not apply 1 Yes

Abbreviation	Description
	Alloy Additions to the Ladle
T	Alloys Type (Code) 0 None or does not apply 1 Caloy 2 Calcium Carbide 3 Mag-Coke 4 FeSi 5 FeCr 6 Mg Fe Si 7 Inoculoy 63 8 Ce 9 Noduloy 5C
LB T	Addition, 1bs. Alloys Type (Code) (See Above)
LB T	Additions, 1bs. Alloys Type (Code) (See Above)
LB 1	Additions, lbs. #1 Other Additions (Code) 0 None 1 Graphite 2 SmZ 3 Bi
LB 2	Additions, 1bs. #2 Other Additions (Code) 0 None 1 Fe Si 2 Fe Mo 3 A1 9 See Questionnaire
LB	Additions, lbs.

Abbreviation	Description
CUP REP	Cupola Replaced last 10 yrs? (Code) 0 No 1 Yes
TYP FCE	If Yes, Type of Furnace (Code) C Does not apply 1 Cupola 2 Channel induction 3 Coreless induction 4 Direct Arc 5 Indirect Arc 6 Air Furnace 7 Other, see Questionnaire
VENT TAP	Ventilation During Fce. Tapping (Code) 1 General 2 Local
EFF	Effectiveness (Code) 1 Excellent 2 Good 3 Fair
VENT MOLD	Ventilation During Mold Pouring (Code) 1 General 2 Local
EFF	Effectiveness (Code) 1 Excellent 2 Good 3 Fair
EERP ATK	Foundries Visited by EERP/OLPA Foundries Visited by A. T. Kearney & Company, Inc.

SECTION 2 FURNACE DATA

Abbreviation	<u>Description</u>
FDRY NO. CD TY FCE NO. FCE TYP	Foundry No. Card Type Furnace No. Furnace Type (Code) 1 Cupola 2 Channel Induction 3 Coreless Induction 4 Direct Arc 5 Indirect Arc 6 Air Furnace 7 Other
LIN TYP	Lining Type or Coil Type (Code) 0 Does Not Apply 1 Acid 2 Basic 3 Neutral 4 Unlined 5 Single Channel Open 6 Single Channel Closed 7 Double Channel
BLT DES	Blast Description (Code) 0 Does Not Apply 1 Hot Blast > 800 F 2 Warm Blast 3 Cold Blast

Abbreviation	Description
BLT HTG	Blast Heating or Frequency (Code) 0 Does Not Apply 1 No Blast Heating 2 Recuperative Heating 3 Externally Fired 4 Recup. and Ext. Fired 5 Line Frequency 6 Medium Frequency 7 High Frequency
TOP C/O	Cupola Top - Closed or Open (Code) 0 Does not Apply 1 Closed Top 2 Open Top
СНС	Top or Side Charged (Code) 0 Does Not Apply 1 Top 2 Side
GAS T-O	Gas Take-Off (Code) O Does Not Apply 1 Above Charging Door 2 Below Charging Door 3 Gas Take-Off into Side Draft Hood 4 Gas Take-Off into Full Roof Hood 5 Gas Take-Off into Canopy Hood 6 Gas Take-Off into Snorkel 7 Direct Shell Evacuation 8 No Gas Take-Off
AFT	Afterburners (Code) O Does Not Apply or None No. Indicates Use and Quantity of Burners
CHG DR	Charging Door - Open or Closed (Code) 0 Does Not Apply 1 Door Open 2 Door Closed

Abbreviation	Description
FL IN	Fuel Injection (Code) O Does Not Apply or Without Fuel Injection l With Fuel Injection
OX EN	Oxygen Enrichment (Code) O Does Not Apply or Without Oxygen Enrichment 1 With Oxygen
FCE USE	Furnace Use (Code) 1 Melting 2 Holding 3 Duplexing
FCE DIA HLD CAP MLT RAT MT UT PR UT OT UT VOLUME PRES TEMP SIZE CH DR HT DR CHG PRE	Furnace Dia., Inches Holding Cap., Tons Melt Rate, Tons/Hours Melting Utilization, Hour/Day Pouring Utilization, Hour/Day Other Utilization, Hour/Day Blast Volume, SCFM Blast Pressure, Oz H20 Blast Temperature, OF Size of Charge Door, Sq. Ft. Height Charge Door Sill Above Floor, Ft. Charge Preheated and/or Dried (Code) O No or Does Not Apply 1 Preheated 2 Dried 3 Both
SIZE AFT AFT LOC	Afterburner Size, BUT/Hr x 10 ³ Afterburner Location (Code) 0 Does Not Apply 1 Above Charge Door 2 Below Charge Door 3 In Gas Take-Off 4 Above and Below Door
DIS AFT POWR SPLY CTL SYS	Distance Afterburner to Gas Take-Off, Inch. Power Supply, KW or KVA Control System No.

SECTION 3 MELTING OPERATION

Abbreviation	Description
FDRY NO. CD TY	Fourdry No. Card Type (Code) 20 Grey Iron 21 Malleable Iron 22 Ductile Iron
FCE NO TOTAL MTL CH	Furnace No. Total Metallic Charge, 1bs. Remelt, 1bs.
PURCH CST I PURCH STEEL BRIQ PUN/TURN LIM STN	Pig Iron, 1bs. (With *, Silvery Pig) Purchased Cast Iron, 1bs Purchased Steel, 1bs. Briquettes, 1bs. Punchings and/or Turnings, 1bs. Limestone, 1bs.
FLOU SPAR	Dolomite, lbs. Soda Ash, lbs. Flourspar, lbs. Other, lbs. Carbo-Coke, Lbs.
FE-SIL MN T	Ferrosilican, lbs. Manganese, FeMn, SiMn. lbs. Additive, Type (Code) 0 None 1 Copper 2 Silvery Iron 3 Silicon 4 Calcium Carbide 5 Silicon Carbide 6 Carnell (CaF ₂) 7 Ferrophosphorous 8 Graphite
LB ME/CO RATIO SUL CON D A	Additive, lbs. Metal to Coke Ratio Sulphur Content of Coke, % Desulphurizing Agents (Code) 0 Does Not Apply or None 1 Caustic Soda NaOH 2 Soda Ash Na ₂ CO ₃ 3 Calcium Carbide

Abbreviation	Description				
Q S	Quality of Scrap (Code) O Does Not Apply 1 Rusty 2 Dirty 3 Oily 4 Clean				
C M	Charging Method (Code) O Does Not Apply 1 Skip Hoist Side Discharge Bucket 2 Skip Hoist Bottom Discharge Bucket 3 Crane Type Charger 4 Belt Conveyor 5 Vibrating Feeder 6 Magnet 7 Manual				
S P	Scrap Preparation (Code) O Does Not Apply or None 1 Shot Blast 2 Degreasing 3 Cut to Size				
TIME METH	Average Length of "Light-Up", Min. Per Day Method of light-up (Code) O Does Not Apply 1 Wood 2 Gas 3 Oil 4 Electric 5 Other 6 More than one of above				
CO CO ₂ N ₂	Carbon Monoxide, % Carbon Dioxide, % Nitrogen, %				

Appendix B Exhibit II Section 4

SECTION 4 CONTROL SYSTEM

CONTENT

Abbreviation Description FDRY NO. Foundry No. CD TY Card Type CTL SYS Control System No. SYS TYP Control System Type (Code) 0 No Control System 1 Fly Ash and Spark Arrester 2 Afterburner 3 Wet Cap 4 Mechanical Collector 5 Wet Scrubber

6 Fabric Filter7 Electrostatic Precipitator

FCES CONT YEAR INST GAS VOLUME GAS TEMP PRES DROP HEIGHT EXH STK AFTBURN SIZE DUST GAS RECIRC NOISE CONTR	Furnaces Controlled Year Installed Rated Gas Volume at Exh. Inlet, CFM Gas Temperature at Exh. Inlet, of Pressure Drop, in H2O Height Exhaust Stack, Ft. Afterburner Size, BTU.Hr. x 10 Water Consumption, Dust Collector, GPM Water Consumption, Gas Cooling, GPM Water Consumption, Recirculated, GPM Noise Control (Code) 1 Yes
FIL MED	Pilter Media (Code) 0 Does Not Apply 1 Natural Fibre
·	2 Synthetic Fibre 3 Glass Fibre

Air to Cloth Ratio
Inlet Concentration, GR/SCF
Outlet Concentration, GR/SCF
Catch, 1b. Dust/Ton Melt
Collection Efficiency, %
Melt Rate at Which Test was Made, TPH

INLET CONC OUT CONC COLL EFF MLT RTE

SECTION 5 ANALYSIS

CONTENT

Abbreviation

FDRY NO. CD TY CTL SYS

<u>Description</u>

Four	ndry	No.		
	d Typ			
Cont	rol	Syst	em	No.
Gas	Ana 1	ysis	%	CO
Cas	Ana1	ysis	%	CO2
Gas	Ana 1	ysis	%	02
Gas	Ana l	ysis	%	N_2
Gas	Anal	ysis	%	H_2
Gas	Ana1	ysis	%	$S\delta_2$
Gas	Ana1	ysis		

Particle Prop. %

Less	Than	2 Microns
Less	Than	5 Microns
Less	Than	10 Microns
Less	Than	20 Microns
Less	Than	50 Microns
Less	Than	100 Microns
Less	Than	200 Microns
Less	Than	500 Microns
Less	Than	1,000 Microns

Catch Prop., %

Less	Than	2	Microns
Less	Than	5	Microns
Less	Than	10	Microns
Less	Than	20	Microns
Less	Than	50	Microns
Less	Than	100	Microns
Less	Than	200	Microns
Less	Than	500	Microns

SECTION 6 ANALYSIS AND COSTS

Abbreviation	Description		
FDRY NO. CD TY CTL SYS	Foundry No. Card Type Control System No.		
	Chemical Analysis		
COMBUS BASIC EQUIP AUX EQUIP ENG COST INST COST TOT COST OPER COST MAINT COST DEPR EQUIP CHANG TOT ANN	SiO ₂ CaO Al ₂ O ₃ MgO FeO, Fe ₂ O ₃ , Fe MmO, Mn ₃ O ₄ PbO ZnO SnO SO _x Combustibles Basic Equipment Costs Aux. Equipment Costs Engineering Costs Installation Costs Installation Costs Total Cost Operating Cost Maintenance Cost Depreciation Overhead Process and Equipment Changes Total Annual Costs	<i>ው-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ-ሙ</i>	x 103 x 103 x 103 x 103 x 103 x 103 x 103 x 103 x 103 x 103

MATERIAL AND HEAT BALANCE

Appendix C consists of three exhibits, each of which deals with material and heat balances of foundry melting furnaces. The first exhibit of this appendix is the Cupola Material and Heat Balance Model. This exhibit discusses the development of the mathematical model and the nature of input required. A sample of the material balance and heat balance outputs and the chemical reactions considered in the model are given.

Exhibit 2 of this appendix is a listing of the FORTRAN IV computer program version of the material and heat balance.

The final exhibit of Appendix C is a series of material and heat balances for cupolas and electric furnaces. A material and heat balance is given for different furnace classifications. Each classification, as discussed earlier in this report, represents a unique furnace type and operating characteristics found in practice.

CUPOLA MATERIAL AND HEAT BALANCE MODEL

Because the major single source of air emissions from the foundry is the cupola, a material and heat balance model has been developed to compute (for given operating conditions) the estimated emissions generated by the cupola.

The model requires the following general inputs:

- Size and certain design characteristics of the cupola.
- The type and hourly melting rate of iron being produced.
- The makeup of the charge, including metallics, fuel, flux and additives.
- 4. The rate and temperature of the blast air including any oxygen enrichment.
- 5. The temperature of the top gases immediately above the burden.

Using these inputs, the computerized version of the model will calculate a material and heat balance for the cupola and print out a summary report showing these results. An example of the output report is shown in a later section of this appendix.

The approach taken in constructing the model has been to concentrate on those aspects of cupola design and operation which will influence the emissions characteristics of the top gases.

Those aspects which relate to such matters as detailed metallurgical characteristics of the tapped iron but do not influence emissions have not been included in the model.

A search of the literature and previous research was undertaken to collect the known physical and chemical relationships in the cupola that influence emissions characteristics in terms of composition and quantity.

Most of the relationships included in the model are based on known chemical and physical laws and their accuracy with regard to the computed results is limited only by the accuracy of the input data. It has been necessary, however, to include in the model several empirical relationships cited in the literature to permit the calculation of estimated quantities in the top gases of particulate matter and sulfur dioxide. Furthermore, in order to simplify the task of specifying the input values and to limit the required data to figures commonly available in foundry operation, the average or typical composition of charge materials has been incorporated into the model. For example, when specifying for the model that the tapped iron is gray iron, the model uses an average composition for gray iron in its calculations. has been done to avoid having to specify the actual chemical composition of charging materials and tapped iron in order to use In practice, the slight variations in the actual the model. composition of materials from the averages used in the model will have very little effect on the results calculated by the model as far as emissions are concerned.

The average compositions for materials used in the model have been taken from the literature and are shown in the following table. It will be noted that the chemical compositions used are not rigorously complete and only those components which can significantly affect the calculated emissions are included.

The balance of the documentation of the model included in the report consists of a description of the chemical and physical relationships incorporated in the model. A listing of the FORTRAN IV computer program version of the model is given in Exhibit 2 of this appendix.

Average Composition of Materials

<u>Used in the Cupola Model</u>

(Figures Given Are Weight Percentages)

A. Metallics

Gray Iron Malleable Iron Ductile Iron Pig Iron Silvery Pig Iron Scrap Steel Scrap	C 3.2% 2.5 3.8 3.5 2.5 3.3 0.4	Si 2.0% 1.2 2.4 0.6 10.0 2.1 0.2	Mn 0.6% 0.6 0.6 1.5 0.7 0.6	S 0.12% 0.12 .03 .05 .05 .12
Ferrosilicon	0.05	50.0	0.0	0.00

B. Fluxes

						CO2	
	SiO_2	(Ca, Mg)0	CaF ₂	CaC2	Na ₂ O	(asCO3)	$A1_{2}0_{3}$
Limestone & Dolomite	1.0%	54.0%	-		-	45.0%	
Fluorspar	2.0	8.0	84.0	-	-	6.0	-
Fdry Carbide	3.0	14.0	-	70.0	-	11.0	2.0
Soda Ash	-	•	-	-	57.0	43.0	-

C. Foundry Coke Ash

SiO ₂	A1 ₂ 0 ₃	Fe ₂ O ₃	(Ca, Mg)0	Misc.(1)
50.0%	35.0%	8.0%	3.0%	4.0%

D. Cupola Refractory Linings

	$\frac{\text{SiO}_2}{}$	A1 ₂ 0 ₃	MgO	$\frac{\text{Misc}(1)}{}$
Acid	60.0%	30.0%		10.0%
Basic	20.0	20.0	60.0	-

Note: (1) Miscellaneous category includes various other metallic oxides and trace elements not listed.

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SPECIFIC INPUTS REQUIRED FOR THE MODEL

Nature of Cupola and Metal Produced

- 1. Diameter of the cupola in inches
- 2. Type of lining
 - a. Acid
 - b. Basic
 - c. None (water cooled)
- 3. Type of iron being produced
 - a. Gray
 - b. Malleable
 - c. Ductile
- 4. Rate of molten iron produced in tons per hour
- 5. Temperature of the tapped iron in degrees Fahrenheit

Characteristics of Charge

- 1. Metallics (fraction or relative weight of each)
 - a. Pig iron
 - b. Silvery pig
 - c. Scrap
 - 1) Steel
 - 2) Iron
 - d. Foundry returns
 - 1) Gray
 - 2) Malleable
 - 3) Ductile
 - e. Ferrosilicon (50%)

2. Fuel

- a. Metal to coke ratio
- b. Coke analysis
 - 1) Fixed carbon (fraction)
 - 2) Ash content (percent)
 - 3) Sulfur content (percent)
- c. Other fuels (if any)
 - 1) Natural gas (SCFM)
 - 2) Fuel oil (gallons/min.)
- 3. Flux (percent to metal of each)
 - a. Limestone and/or dolomite
 - b. Flourspar
 - c. Foundry carbide
 - d. Soda ash

Characteristics of the Blast

- 1. Air
 - a. Quantity of air (SCFM)
 - b. Inlet air temperature ([°]F)
 - c. Relative humidity of air (fraction)
- Oxygen enrichment (SCFM)
- 3. Blast temperature (°F)

Characteristics of Exhaust Gases

- 1. Top gas temperature (top of burden-F)
- 2. Stack gas above charge door
 - a. Volume of stack gas (CFM)
 - b. Temperature of stackgas for volume given (°F)
 - c. Carbon monoxide in stack gas (%/vol.)

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CHEMICAL REACTIONS CONSIDERED IN THE MODEL

Oxidation Zone

1. C (coke) +
$$0_2 \rightarrow C0_2$$

2.
$$CH_{\Delta}$$
 (natural gas) + 2·0₂ -> CO_2 + 2· H_2O

4.
$$Si + O_2 \rightarrow SiO_2$$

5.
$$2 \cdot (\text{Fe}, \text{Mn}) + 0_2 \rightarrow 2 \cdot (\text{Fe}, \text{Mn}) 0$$

6.
$$2 \cdot CaC_2$$
 (foundry carbide) + $5 \cdot O_2 \rightarrow 2 \cdot CaO + 4 \cdot CO_2$

Reduction Zone

7. C (coke) +
$$CO_2 \rightarrow 2 \cdot CO$$

8. C (coke) +
$$H_2O \rightarrow CO + H_2$$

Preheating Zone

9. (Ca, Mg)CO3
$$\rightarrow$$
 (Ca, Mg)O + CO2

10.
$$Na_2CO_3 \rightarrow Na_2O + CO_2$$

11.
$$S + Fe \rightarrow FeS$$

Slagging Reactions

12. (Ca, Mg, Fe, Mn)0 +
$$SiO_2$$
 -> (Ca, Mg, Fe, Mn)SiO₃

14. CaS + FeSiO₃ +
$$2 \cdot \text{MnO}$$
 -> CaSiO₃ + Fe + $2 \cdot \text{Mn}$ + SO₂

MATERIAL BALANCE RELATIONSHIPS CONSIDERED IN THE MODEL

The material balance relationships employed in the model use the input quantities (specified on the preceding pages) of materials and the average compositions of materials shown earlier in this exhibit. The source of empirical relationships used are shown for reference. The quantities are all computed to a base of 2,000 pounds of tapped iron.

- 1. Silicon Oxidized = Net Silicon Loss from Metalics Net Silicon Loss From Metalics = Weight of Silicon in Metalic Charge - Weight of Silicon in Tapped Iron
- 2. Manganese Oxidized (same approach as for silicon above)
- 3. Carbon From Coke Into Iron = Net Carbon Gain of Tapped Iron Net Carbon Gain = Weight of Carbon in Tapped Iron - Weight of Carbon in Charged Metalics
- 4. Oxygen Input = Weight of Oxygen in Blast Air + Oxygen
 Enrichment
- 5. Water Input = Water Content of Blast Air as a Function of
 Temperature and Relative Humidity
- 6. Oxygen Available For Fuel Combustion = Oxygen Input Oxygen

 Used to Oxidize Silicon,

 Manganese and Foundry

 Carbide
- 7. Oxygen Available to Burn Coke = Oxygen Available For Fuel

 Combustion Oxygen Consumed

 to Burn Natural Gas and/or

Fuel Oil

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- 8. Coke Consumed in Complete Burning
 - Carbon Required to Combine with Available Oxygen /Carbon Content of Coke
- 9. Total Water to React With Coke
 - Water Input in Air + Water Generated in Combustion of Natural Gas and Fuel Oil
- 10. Coke Available to Reduce Carbon Dioxide
 - = Total Coke Input Coke Consumed in Complete Burning
 - Coke Required to Reduce Total Water
- 11. Resulting Carbon Dioxide
 - = CO₂ from Complete Combustion of Coke
 - CO2 Reduced by Remaining Coke
 - + CO₂ from Calcining of Carbonates
- 12. Resulting Carbon Monoxide
 - CO from Water Reduction + CO from CO₂ Reduction
- 13. Resulting Hydrogen = Weight of Hydrogen in Total Water
- 14. Silica From Cupola Lining = Weight of Lining Melted as a function of Cupola Diameter*

 × Silica Content of Lining
- 15. Magnesium Oxide From Cupola Lining (same approach as for silica)
- 16. Total Silica For Slag = Silica From Coke Ash + Silica

 Fluxes + Silica From Cupola Lining
- 17. Total Calcium and Magnesium Oxide for Slag
 - = (Ca, Mg)O from Fluxes, Cupola Lining, and Coke Ash
- 18. Slag Basicity Ratio = Total (Ca, Mg)0/SiO₂

 *The Cupola and Its Operation (third edition; Des Plaines, Illinois: American Foundrymen's Society, 1965), p. 233.

- 19. Sulfur in Slag = Function of the Basicity Ratio* of the Slag
- 20. Sulfur To Top Gas (SO_2) = Total Sulfur Input in Coke and Metalics Sulfur Output in Tapped Iron Sulfur Output in the Slag
- 21. Oxidation of Iron To Slag = Assumption that Slag Contains

 Two Percent FeO Derived from

 Oxidation of Iron*
 - *Note: No method could be found in the literature to predict the amount of iron oxidized and going into the slag. Typical compositions suggest that two percent is a good average value.
- 22. Particulate Matter in Top Gases = Empirical Function**of:
 - a) Blast Volume
 - b) Coke Ratio
 - c) Melting Rate
- 23. Total Slag Output = Weight of Slag Forming Constituerits
 in Coke Ash and Fluxes
 - + SiO_2 , MnO, and FeO from Metalics
 - + Cupola Lining Melted
 - Emissions Dust
- 24. Total Metallic Inputs = Tapped Iron Weight
 - + Si, Mn, Fe Loss From Metallic Charge
 - C and S Gain in Tapped Iron
- *The Cupola and Its Operation, p. 236
 **Engels and Weber, Cupola Emission Control, trans. (Cleveland, Ohio: Gray and Durtile Iron Founders' Society, 1969, Original, 1967), pp. 52-54.

HEAT BALANCE RELATIONSHIPS CONSIDERED IN THE MODEL

The heat balance relationships are calculated in B.T.U.'s per hour of cupola operation.

Heat Inputs

- 1. Potential Heat of Fuels = Heat of Total Combustion for Weight of Carbon in Coke plus Natural Gas and Fuel Oil Combustion.
- 2. Heat of Oxidation of Metallics
 - = Heats of Oxidation for Weight of Iron, Silicon, Manganese, and Carbide Oxidized per Hour
- 3. Sensible Heat of Air Blast
 - = Temperature x Specific Heat
 x Weight of Air Blast Per Hour

Heat Outputs

- 1. Heat Content of Tapped Iron
 - = Temperature x Specific Heat
 x Weight of Molten Iron Tapped
 Per Hour
- 2. Heat for Calcining Carbonates
 - Heat of Reaction for CalciningX Weight of CarbonateFluxes Per Hour
- 3. Heat of Slagging = Heat of Reaction of Calcium and

 Magnesium Oxides with Silica To

 Form Silicates in the Slag

- 4. Heat Content of Slag = Temperature x Specific Heat

 x Weight of Slag Heat of

 Slagging Reactions
- 5. Heat of Water Decomposition
 - = Heat of Reaction with Carbon to Reduce Water to Hydrogen and Carbon Monoxide
- 6. Sensible Heat of Top Gas
 - = Temperature x Specific Heat
 x Weight of Top Gases
- 7. Latent Heat of Top Gas = Potential Heat of Combustion of

 Carbon Monixide x Weight of Carbon

 Monoxide in the Top Gas Per Hour
- 8. Heat Radiation From the Cupola
 - = Sum of Heat Inputs
 - Sum of Heat Outputs (above)

SAMPLE OUTPUT MATERIAL BALANCE

<u>INPUTS</u>	POUNDS	PERCENT
METAL CHARGE PIG IRON RETURNS STEEL SCRAP	2015 288. 959. 672.	51.50 7.36 24.52
IRON SCRAP FERROALLOYS	0. 96.	17.17 0.00 2.45
COKE NATURAL GAS FUEL OIL	208. 0. 0.	5.33 0.00 0.00
FLUX AND ADDITI	VES 27.	0.68
AIR OXYGEN	1641. 0.	41.95 0.01
CUPOLA LINING	20.	0.52
TOTAL INPUT MTL	S 3912.	100.00
OUTPUTS		
MOLTEN IRON	2000.	51.13
SLAG	64.	1.62
EMISSIONS DUST	16.	0.40
TOP GASES NITROGEN CARBON DIOXIDE CARBON MONOXIDE HYDROGEN SULFUR DIOXIDE		46.85 68.41 20.44 10.99 0.07 0.09

SAMPLE OUTPUT HEAT BALANCE

INPUT HEAT	B.T.U.'s (000/HR.)	PERCENT
POTENTIAL HEAT OF FUEL	56208.	91.51
SENSIBLE HEAT OF THE BLAST	2554.	4.16
HEAT FROM OXIDATION OF MN, FE, SI	2664.	4.34
TOTAL INPUT HEAT	61426.	100.00
OUTPUT HEAT		
HEATING AND MELTING OF IRON	24212.	39.42
HEAT CONTENT OF THE SLAG	1085.	1.77
CALCINING OF LIMESTONE	419.	0.68
DECOMPOSITION OF WATER	713.	1.16
TOP GASES -SENSIBLE HEAT -LATENT HEAT	8112. 19168.	13.21 31.20
HEAT RADIATION FROM THE CUPOLA	7717.	12.56
VOLUME OF TOP GAS (MCF) ACTUAL 1280. STANDARD 508. STOP		

```
5 REAL I, I1, I2, I3, I4, I5, I6, I7, I8, I9, U, M1, M2, M3, M4, M5, M8
6 REAL K,K1,K2,K3,K4,K5,K6,K7,K8,K9,L1,L2,L3,L4,L5,N2,N3
  15 REAU (1,16)D,L1,L2,17,18,19
   16 16 FORMAT (10F10.0)
   20 READ (1,16)M1,T4,I1,I2,R3,I3,I
   25 READ (1,16)14,15,16,K1,K2,K7,K8,G1,G2
   30 REAU (1,16)F1,F2,F3,F4,A1,T1,H3,O2,T2,T3
32 READ (1,16) V3, T5, COP
   35 J=I1+I2+I3+I4+I5+I6+I+R3
   40 Il=I1/J
   45 12=12/0
   50 I3=I3/J
   55 I4=I4/J
   60 I5=I5/J
   65 16=16/J
  70 I=I/J
   75 K3=R3/J
   59=(.006\%11+.1\%12+.002\%13+.49\%R3+.021\%1+.02\%14+.012\%15+.024\%16)
   05 S9=M1#2000.0*S9
   90 S9=S9-2000.0*M1*(.02*I7+.012*I8+.024*I9)
   92 IF ($9.LT.2.*M1) $9=2.0*M1
   95 Cd=.032*17+.025*10+.038*19
  100 C8=(C8-.035*11-.025*12-.004*13-.033*1-.032*14-.025*15-.038*16)
  105 C8=2000.0*M1*C8
  110 M8=2000.0*M1*(.009*I1+.002*(1.0-I1))
  115 01=(.209"A1+02)".0347"60.0
  120 U3=01-20.0*M1*.86*F3-1.14$9-.41*M8
  125 H4=8.631"EXP(.03598"T1)"H3".0763"A1"60.0/7000.0
  130 K3=(03-(10.03\%G1)-1450.0\%G2)/(2.66\%K2)
  135 C2=1.30*O3
  140 H1=5.72*G1+519.0*G2+H4
  145 K4=.667#H1/K2
  150 H=.111 H1
  155 K5=2000.0*M1/K1
  160 K6=K5-K3-K4-C8/K2
  165 C3=4.66#K6#K2
  170 C2=C2-3.66*K6*K2
  175 C1=1.55*H1
  180 C4=100.0°C2/(C3+C2)
  185 C5=20.0*M1*(.45*F1+.06*F2+1.06*F3+.43*F4)
  190 C2=C2+C5
  195 Cl=Cl+C3
  200 N2=3.52#A1
  205 L4=(M1×20.0×465.0/(D/2.0)××1.75)×(L1+L2)
  210 S5=.5*K5*K7/100.+L4*(.6*L1+.2*L2)+20.0*M1*(.01*F1+.02*F2+.03*F3)
  211 S5=S5+2.14*S9
  215 M5=.0003*K7*K5+L4*(.6*L2)+20.0*M1*(.54*F1+.08*F2+.75*F3)
  220 B1=M5/S5
  225 S1=Kd*K5/100.0+.0076*G1+1.34*G2
  230 S2=2000.0*M1*(.0005*(I1+I2+I3+I6)+.0012*(I4+I5+I))
  235 $2=$2-2000.0\m\1\(\cdot\)(17+18)+.0003\(\cdot\)19)
  240 S0 = L4 + K7 \times K5/100.0
  245 $8=$8+20.0%M1%(.55%F1+.94%F2+.8%F3+.57%F4)+2.14%$9+1.29%M8
  250 So=1.02*So
  255 $3=(.0000659*(B1*100.0)**2.65)*$8/10000.0
  260 So=So+S3
```

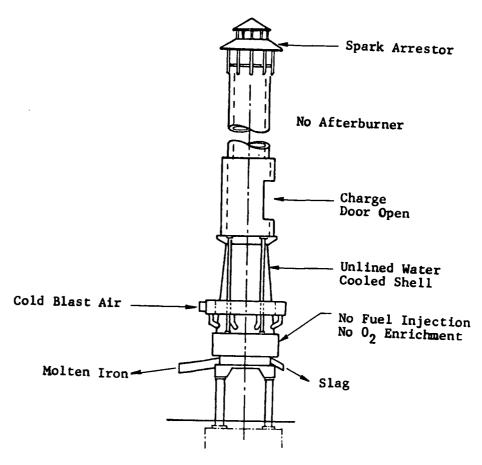
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265 $4=2.0%($1+$2-$3)
270 Kl=.02#S8#.78
275 M2=M1 x2005.0-C8+S9+M8+R1+S2
200 B3=(A1+02+G1)/((D/24.0)**2*3.1416)
205 U3=.0625#(4.0+(B3/32.8-7.0))
290 \nu4=200.0/K1-16.0
292 U5=56.0-29.2*M1/((D/24.0)**2*3.14)
300 \text{ V1=}(N2/28.0+C2/44.0+C1/28.0+H/2.0+S4/64.0)*.01073*(T3+460.0)/14.7
305 V2=520.0*V1/(T3+460.0)
310 U6=U3*V2/M1
315 D7=.5*U6+.25*U4+.25*U5
320 So=Sd-D7*M1
325 Q1=14.45\%((K5-K4)\%K2-C8)+60.0\%G1+8470.0\%G2
330 Q2=127.4"F3"M1+2.05"R1+13.1"S9+3.02"M8
335 Q3=(T2-60.0)*(.249*N2+.224*01+.453*H4)/1000.0
340 M3=M2/M1
345 F5=(F1+F2+F3+F4) M2/M1/100.0
350 A5=(N2#1.3+H4)/M1
355 O6=O2*5.08/Ml
360 K=K5/M1
365 G4=2.54#G1/M1
370 G5=446.0×G2/M1
375 L5=L4/M1
300 W1=M3+F5+A5+06+K+G4+G5+L5
385 M4=2000
390 S6=S3/M1
395 N3=N2/M1
400 C6=C2/M1
405 C7=C1/M1
410 H7=H/M1
415 S7=S4/M1
420 W2=N3+C6+C7+H7+S7
425 W6=W2/100.0
430 W3=2000.0+56+D7+W2
435 W4=W1-W3
440 S6=S6+W4
445 W1=W1/100.0
450 R5=(14+15+16)*M3
455 R6=(12+R3)*M3
460 Q4=(T4-60.0)*.2086*2.0*M1
465 Q5=2.16*.750*(C5-20.0*M1*F3*.95)
470 CS=S5*(1.-DIM(1.,(61.*M5/52./S5)))
472 Q6=.02*M5-.44*CS
475 Q7=(T4-60.0)".321"S6"M1/1000.0
400 Q7=Q7-Q6
405 Q8=2.090"H1
490 Q9=(T3-60.0)*(.254*N2+.243*C2+.256*C1+3.466*H+.226*D7*M1)/1000.0
495 P1=4.346*C1
500 Q=Q1+Q2+Q3
505 P2=Q-Q4-Q5-Q7-Q8-Q9-P1
510 Z = Q/100.0
700 KM32=M3/W1
705 kIll=Il*M3
710 RI12=RI11/W1
715 K52=R5/W1
720 KI31=I3*M3
```

1000 WRITE (9,140)A5,A52

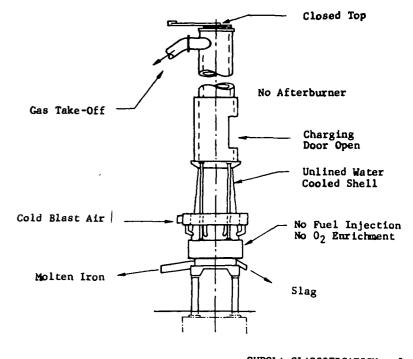
```
725 KI32=RI31/W1
730 KIO1=I*M3
735 KIO2=RIO1/W1
740 R62=R6/W1
745 RK02=K/W1
750 G42=G4/W1
755 G52=G5/W1
760 F52=F5/W1
765 A52=A5/W1
770 U62=06/W1
775 KL52=L5/Wl
700 W11=100. W1
705 W12=W11/W1
790 RM42=M4/W1
795 S62=S6/W1
800 D72=D7/W1
005 W2Z=W2/W1
010 RN32=N3/W6
015 C62=C6/W6
020 C72=C7/W6
025 H72=H7/W6
030 S72=S7/W6
035 Q12=Q1/Z
040 Q32=Q3/Z
845 Q22=Q2/Z
050 QU2=Q/Z
d55 Q42=Q4/Z
d60 Q72=Q7/Z
Ø65 Q52=Q5/Z
070 Qd2=Qd/Z
075 Q92=Q9/Z
000 P12=P1/Z
do5 P22=P2/Z
090 WRITE (9,30)
d95 30 FORMAT (5X16HMATERIAL BALANCE//6HINPUTS/20X17HPOUNDS
                                                                  PERCENT
900 WRITE (9,40)M3,RM32
905 40 FORMAT (12HMETAL CHARGE8X, F7.0, F9.2)
910 WRITE (9,50)RI11,RI12
915 50 FORMAT (8HPIG IRON12X, F7.0, F9.2)
920 WRITE (9,60)R5,R52
925 60 FORMAT (7HRETURNS13X, F7.0, F9.2)
930 WRITE (9,70)RI31,RI32
935 70 FORMAT (11HSTEEL SCRAP9X, F7.0, F9.2)
940 WRITE (9,80)RIO1,RIO2
945 80 FORMAT (10HIRON SCRAP10X, F7.0, F9.2)
950 WRITE (9,90)R6,R62
955 90 FORMAT (11HFERROALLOYS9X, F7.0, F9.2)
960 WRITE (9,100)K,RK02
965 100 FORMAT (/4HCOKE16X,F7.0,F9.2)
970 WRITE (9,110)G4,G42
975 110 FORMAT (11HNATURAL GAS9X, F7.0, F9.2)
900 WRITE (9,120)G5,G52
905 120 FORMAT (8HFUEL OIL12X, F7.0, F9.2)
990 WRITE (9,130)F5,F52
995 130 FORMAT (/18HFLUX AND ADDITIVES2X, F7.0, F9.2)
```

```
1005 140 FORMAT (/3HAIR17X,F7.0,F9.2)
1010 WRITE (9,150)06,062
1015 150 FORMAT (6HOXYGEN14X, F7.0, F9.2)
1020 WRITE (9,160)L5,RL52
1025 160 FORMAT (/13HCUPOLA LINING7X, F7.0, F9.2)
1030 WRITE (9,170)W11,W12
1035 170 FORMAT (/16HTOTAL INPUT MTLS4X, F7.0, F9.2)
1040 WRITE (9,180)M4,RM42
1045 100 FORMAT (//7HOUTPUTS//11HMOLTEN IRON9X, F7.0, F9.2)
1050 WRITE (9,190)$6,$62
1055 190 FORMAT (/4HSLAG16X,F7.0,F9.2)
1060 WRITE (9,200)U7,U72
1065 200 FURMAT (/14HEMISSIONS DUST6X, F7.0, F9.2)
1070 WRITE (9,210)W2,W22
1075 210 FORMAT (/9HTOP GASES11X.F7.0,F9.2)
1080 WRITE (9,220)N3,RN32
1085 220 FORMAT (8HNITROGEN12X, F7.0, F9.2)
1090 WRITE (9,230)C6,C62
1095 230 FORMAT (14HCARBON DIOXIDE6X.F7.0,F9.2)
1100 WRITE (9,240)C7,C72
1105 240 FORMAT (15HCARBON MONOXIDE5X, F7.0, F9.2)
1110 WRITE (9,250)H7,H72
1115 250 FORMAT (8HHYDROGEN12X, F7.0, F9.2)
1120 WRITE (9,260)57,572
1125 260 FORMAT (14HSULFUR DIOXIDE6x, F7.0, F9.2)
1130 WRITE (9,265)
1135 265 FORMAT (///5x12HHEAT BALANCE//10HINPUT HEAT10x7HB.T.U.S)
1140 WRITE (9,270)
1145 270 FORMAT (20X20H(000/HR.)
                                     PERCENT)
1150 WRITE (9,280)Q1,Q12
1155 200 FORMAT (/14HPOTENTIAL HEAT/10H OF FUEL10X, F7.0, F9.2)
1160 WRITE (9,290)Q3,Q32
1165 290 FORMAT (/13HSENSIBLE HEAT/15H OF THE BLAST5X, F7.0, F9.2)
1170 WRITE (9,300)Q2,Q22
1175 300 FORMAT (/19HHEAT FROM OXIDATION/16H OF MN, FE, SI4X,F7.0,F9.
1100 WRITE (9,310)Q,Q02
1185 310 FORMAT (/19H
                       TOTAL INPUT HEATX, F7.0, F9.2)
1190 WRITE (9,320)
1195 320 FORMAT (//11HOUTPUT HEAT)
1200 WRITE (9,330)Q4,Q42
1205 330 FORMAT (/19HHEATING AND MELTING/10H OF IRON10X, F7.0, F9.2)
1210 WRITE (9,340)Q7,Q72
1215 340 FORMAT (/12HHEAT CONTENT/14H OF THE SLAG6X, F7.0, F9.2)
1220 WRITE (9,350)Q5,Q52
1225 360 FORMAT (/13HDECOMPOSITION/11H OF WATER9X, F7.0, F9.2)
1230 WRITE (9,360)Q8,Q82
1235 350 FORMAT (/12HCALCINING OF/12H LIMESTONE 8X, F7.0, F9.2)
1240 WRITE (9,370)Q9,Q92
1245 370 FORMAT (/9HTOP GASES/16H -SENSIBLE HEAT4X, F7.0, F9.2)
1250 WRITE (9,380)P1,P12
1255 300 FORMAT (14H -LATENT HEAT6X, F7.0, F9.2)
1260 WRITE (9,390)P2,P22
1265 390 FORMAT (/19HHEAT RADIATION FROM/13H THE CUPOLA7X, F7.0, F9.2)
1270 WRITE (9,400)V1,V2
1275 400 FORMAT (//23HVOLUME OF TOP GAS (MCF)/6HACTUAL4X, F7.0/
1200 + ohstandard2x, F7.0
1960
         Stop
1970
        End
```

Insufficient Data for Feat and Material Balance Calculations

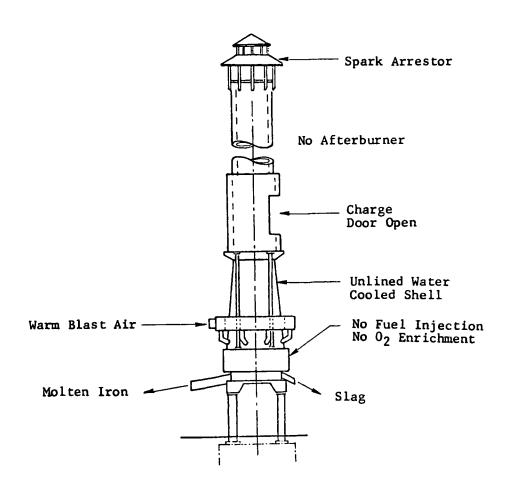


INPUT HEAT	B.T.U.S (000/HR.)	PERCENT	MATERIAL BALAN	CE	
			INPUTS		
POTENTIAL HEAT				POUNDS	PERCENT
ØF FUEL	100823.	98.35	METAL CHARGE	1992.	47.42
			PIG 1RØN	233.	5.56
SENSIBLE HEAT			RETURNS	996.	23.71
OF THE BLAST	. 158.	0.15	STEEL SCRAP	739.	17.60
			IRON SCRAP	0.	0.00
HEAT FROM ØXIDATION			FERHOALLOYS	23.	0.56
OF MN, FE, SI	1537.	1.50			
			COKE	253.	6.03
TØTAL INPUT HEAT	102518.	100.00	NATURAL GAS	0.	0.00
			FUEL ØIL	0.	0.00
OUTPUT HEAT			FLUX AND ADDITIVES	58•	1.38
HEATING AND MELTING			AIR	1898.	45.18
OF IRON	38157.	37.22	ØXYGEN	0.	0.00
HEAT CONTENT			CUPOLA LINING	0.	0.00
OF THE SLAG	598•	0.58			
			TOTAL INPUT MTLS	4201.	100.00
CALCINING OF					
LIMESTONE	1438.	1 - 40			
			ØUTPUTS		
DECOMPOSITION					
OF WATER	1388.	1.35	MØLTEN IRØN	2000•	47.60
TOP GASES			SLAG	44.	1.04
-SENSIBLE HEAT	14714.	14.35	•		
-LATENT HEAT	32509.	31 • 71	EMISSIONS DUST	19.	0.45
HEAT RADIATION FROM			TOP GASES	2139.	50.90
THE CUPOLA	13714.	13.38	NITROGEN	1449.	67.77
			CARBON DIOXIDE	468.	21.87
			CARBON MONOXIDE	220.	10.29
VØLUME ØF TØP GAS (MCF)		HYDRØGEN	2.	0.07
ACTUAL 2309.			SULFUR DIØXIDE	-0-	-0.00
STANDARD 917.					

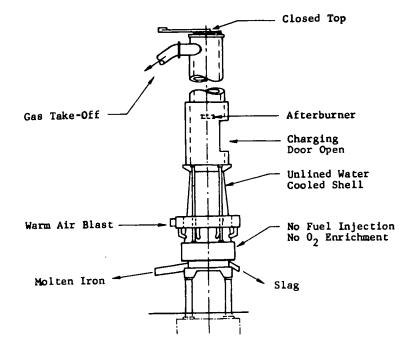




Insufficient Data for Heat and Material Balance Calculations



HEAT BALANCE					
INPUT HEAT	B.T.U.S (000/HR.)	PERCENT	MATERIAL BALAN	CE	
	(000)		INPUTS		DEDCENT
				POUNDS	PERCENT 30.73
POTENTIAL HEAT	56685•	85.19	METAL CHARGE	2004.	0.00
OF FUEL	.360034		PIG IRØN	0.	
			RETURNS	982•	15.05
SENSIBLE HEAT	9301 •	13.98	STEEL SCRAP	295	
OF THE BLAST	73014	10070	IRON SCRAP	687•	10.54
			FERROALLOYS	41.	0.62
HEAT FROM OXIDATION	556•	0.84	• =		
OF MN, FE, SI	220.	0004	COKE	333•	5-11
		100.00	NATURAL GAS	0.	0.00
TOTAL INPUT HEAT	66541+	100.00	FUEL OIL	0.	0.00
			, 022 000		_
			FLUX AND ADDITIVES	100.	1.54
OUTPUT, HEAT			, Zow with the second		
			AIR	4085•	62 • 63
HEATING AND MELTING	j	23.19	ØXYGEN	0.	0.00
ØF IRØN	15432.	23.17	6x102		
			CUPOLA LINING	0.	0.00
HEAT CONTENT		0.23	00.02.0		
OF THE SLAG	156.	0.23	TOTAL INPUT MTLS	6523•	100.00
			TOTAL TIME		
CALCINING OF		1 • 48			
LIMESTONE	984•	1 • 40	<u> </u>		
			<u> 9011 013</u>		
DECOMPOSITION			MOLTEN IRON	2000•	30.66
OF WATER	680•	1.02	HOLIEN THOM		
			SLAG	56.	0.86
TOP GASES			JLAG		
-SENSIBLE HEAT	11973.	17.99	EMISSIONS DUST	32.	0.49
-LATENT HEAT	-15611•	-23.46	E41331842 DOT.		
			TOP GASES	4435•	67.99
HEAT RADIATION FRO	M		NITROGEN	3129•	70.55
THE CUPOLA	52926•	79.54	CARBON DIOXIDE	1573.	35.46
			CARBON MONOXIDE	-266.	-6.00
			HYDROGEN	2.	0.04
VOLUME OF TOP GAS	(MCF)			-3.	-0.06
ACTUAL 1793.			SULFUR DIOXIDE	•	
STANDARD 712.					



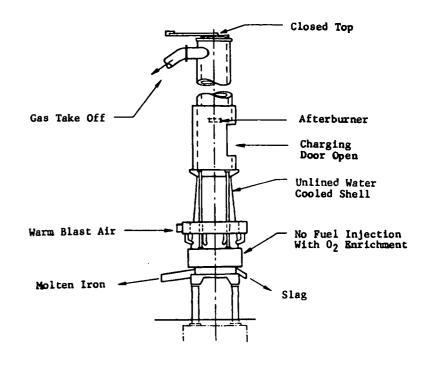


STANDARD

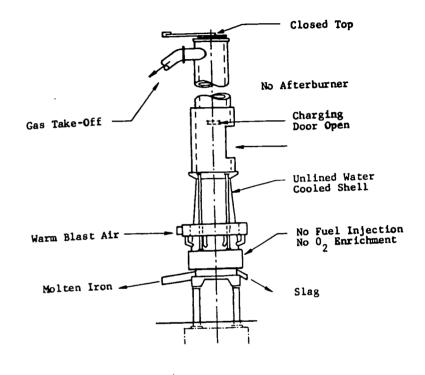
346.

MATERIAL BALANCE

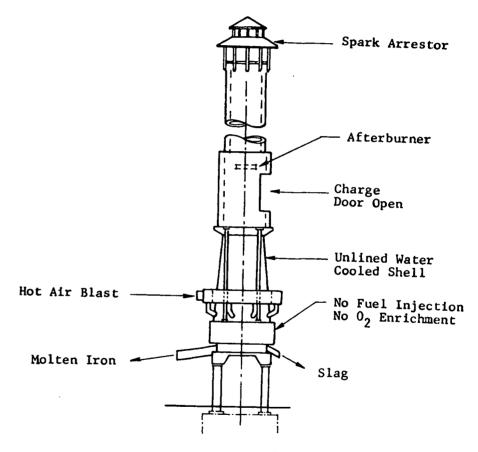
TARRET MEAT	B.T.U.S		MATERIAL BALAN	<u>ice</u>	
- INPUT HEAT	(000/HR.)	PERCENT	INPUTS	PØUNDS	PERCENT
PØTENTIAL HEAT			METAL CHARGE	2030•	46.23
ØF FUEL	40947.	86.57	PIG IRON	256.	5.83
0 022			RETURNS	1507.	34.32
SENSIBLE HEAT			STEEL SCRAP	0.	0.00
OF THE BLAST	3247.	6.86	IRON SCRAP	223.	5.07
		-	FERROALLOYS	45.	1.01
HEAT FROM OXIDATION			PERMUNELUIS	730	1.01
OF MN. FE. SI	3106.	6.57	COKE	250.	5.69
			NATURAL GAS	0.	0.00
TOTAL INPUT HEAT	47300 •	100.00	FUEL ØIL	0.	0.00
			roce ore	••	0.00
			FLUX AND ADDITIVES	136.	3.10
OUTPUT HEAT					
			AIR	1893.	43.13
- HEATING AND MELTING			ØXYGEN	81.	1.85
ØF IRØN	14028.	29.66			
			CUPOLA LINING	0.	0.00
HEAT CONTENT					
ØF THE SLAG	806.	1 - 70	TOTAL INPUT MTLS	4390•	100.00
CALCINING OF					
LIMESTONE	1221.	2.58			
LIMESTONE	12211	2.50	OUTPUTS		
DECOMPOSITION			MØLTEN IRØN	2000•	45.55
OF WATER	292.	0.62	HOLIEN INDIA	2000•	73733
			SLAG	120.	2.74
TOP GASES			22.0		
-SENSIBLE HEAT	5652.	11.95	EMISSIONS DUST	19.	0.44
-LATENT HEAT	10785.	22.80		• • •	
			TØP GASES	2251.	51.27
HEAT RADIATION FROM			NITRØGEN	1450.	64-43
THE CUPOLA	14515.	30.69	CARBON DIOXIDE	600•	26.65
			CARBON MONOXIDE	199.	8.82
			HYDRØGEN	1.	0.04
VOLUME OF TOP GAS	(MCF)		SULFUR DIØXIDE	1.	0.06
ACTUAL 872.					



HEAT BALANCE					
	B.T.U.S		MATERIAL BALAN	<u>CE</u>	
INPUT HEAT	(000/HR.)	PERCENT			
	(000)		INPUTS	POUNDS	PERCENT
				1981 •	55.05
POTENTIAL HEAT	44940.	89.05	METAL CHARGE	0.	0.00
OF FUEL	• • • •		PIG IRON	869•	24.14
			RETURNS	1043.	28.97
SENSIBLE HEAT OF THE BLAST	4694.	9.30	STEEL SCRAP	0.	0.00
A) THE BENS.			IRON SCRAP	70.	1.93
HEAT FROM OXIDATION	l		FERROALLØYS		
OF MN, FE, SI	834•	1 • 65		200•	5.56
Ab Mas Les a.			COKE	0.	0.00
TOTAL INPUT HEAT	50468•	100.00	NATURAL GAS Fuel Øil	0.	0.00
TOTAL TIME			PUEL DIL		
			FLUX AND ADDITIVES	44.	1.21
OUTPUT HEAT			FON MILE MODELLE		
			AIR	1374.	38.18
HEATING AND MELTIN	G	46.25	AXYGEN	0.	0.00
OF IRON	23343•	46.23	<i>9</i> 77.02		
			CUPBLA LINING	0•	0.00
HEAT CONTENT	706.	1.40			100.00
OF THE SLAG	700•		TOTAL INPUT MTLS	3599•	100.00
CALCINING OF	326.	0.64			
LIMESTONE	3200	*	<u>OUTPUTS</u>		
				2000•	55.57
DECOMPOSITION	615•	1.22	MØLTEN IRØN	2000•	5000
OF WATER	•			45•	1.26
TOD CASES			SLAG	430	
TOP GASES -SENSIBLE HEAT	6892 •	13.66	COLONE DUST	14.	0.38
-LATENT HEAT	14898 -	29.52	EMISSIONS DUST	• • • •	
-CATENI III			772 CASE\$	1540•	42.79
HEAT RADIATION FRO	3M		TØP GASES Nitrøgen	1049•	68.13
THE CUPOLA	3688•	7.31	CARBON DIOXIDE	324.	21.05
			CARBON MONOXIDE	165.	10.70
			HYDROGEN	1.	0.07
VOLUME OF TOP GAS	(MCF)		SULFUR DIØXIDE	1.	0.04
ACTUAL 1059.			502, 0 512.		
STANDARD 405.					



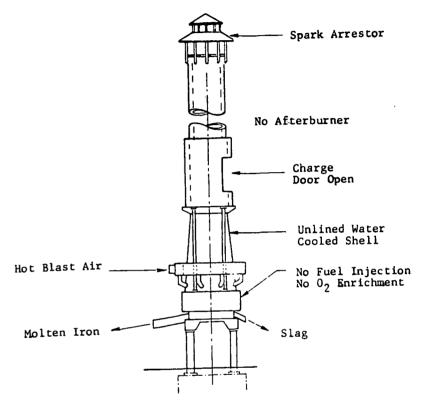
Insufficient Data for Heat and Material Balance Calculations



CUPOLA CLASSIFICATION - 7



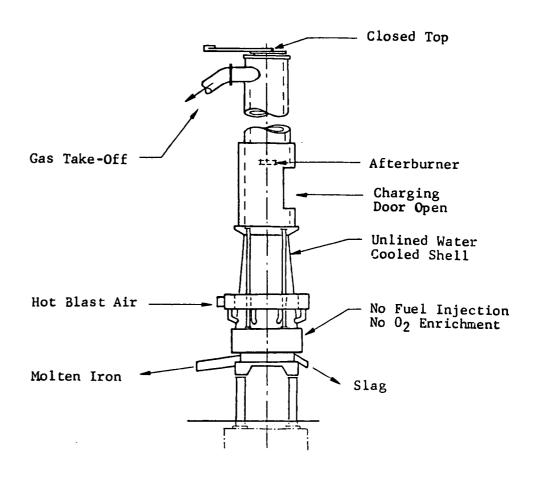
HEAT BILLIAN					
INPUT HEAT	B.T.U.S (000/Hh.)	PERCENT	MATERIAL BALAN	<u>CF.</u>	
			INPUTS	POUNDS	PERCENT
POTENTIAL HEAT				2023•	39.54
OF FUEL	64255•	76.64	METAL CHARGE	0.	0.00
0,			PIG INON	567•	11.47
SENSIBLE HEAT			KETUKNS	391.	7.65
OF THE BLAST	12745.	15.20	STEEL SCHAP	978.	19.12
-			IRON SCRAP	67.	1.30
HEAT FROM OXIDATION			FERRUALLOYS	0	
OF MN. FE. SI	6840•	8.16		267•	5.21
			COKE	0.	0.00
TOTAL INPUT HEAT	83840•	100.00	NATUKAL GAS	0.	0.00
• • •			FUEL OIL		
OUTPUT HEAT			FLUX AND ADDITIVES	61•	1.19
				2766.	54.06
HEATING AND MELTING	82863•	27.27	AIK DXYGEN	0.	0.00
HEAT CONTENT		1.52	CUPBLA LINING	C.	0.00
OF THE SLAG	1271.	1 • 32	TOTAL INPUT MTLS	5117•	100.00
CALCINTHE OF		1.04			
LIMESTONE	ช 7 5•	1.04	ALITERIT C		
			OUTPUTS		
DECOMPOSITION OF WATER	1190.	1 • 42	MOLTEN INON	5000•	39.09
			SLAG	ხ4∙	1.63
TOP GASES	12113.	14.45	SLAU		
-SENSIBLE HEAT	356•	0.42	EMISSIONS DUST	26•	0.50
-LATENT HEAT	330•		FMISSIMS See		
The second second second second	ım		TOP GASES	3008∙	58 • 78
HEAT RADIATION FRO	45172•	53.88	NITROGEN	2112.	70.22
THE CUPULA	45.16.		CARBON DIOXIDE	გგ 7∙	29.46
			CARBON MONDXIDE	4.	0.14
- NO TAN CAU	(MCE)		HYDROGEN	2•	0.08
VOLUME OF TOP GAS ACTUAL 1653. STANDARD 736.	CHUP		SULFUN DIDAIDE	2.	0.08



CUPOLA CLASSIFICATION - 8



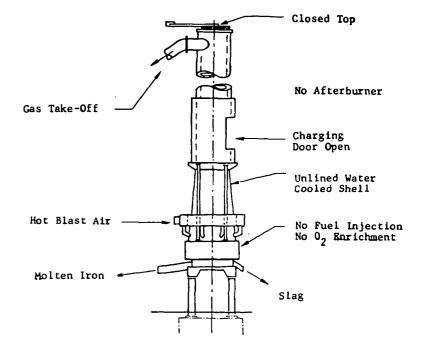
Insufficient Data for Heat and Material Balance Calculations





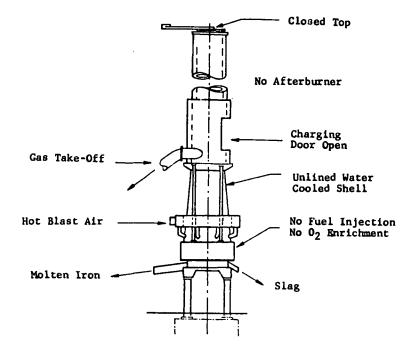
MATERIAL BALANCE

					
INPUT HEAT	B.T.U.S				
	(000/HR.)	PERCENT	INPUTS		
				PØUNDS	
POTENTIAL HEAT			METAL CHARGE	2190.	40.38
OF FUEL	39360.	56.01	PIG IRØN	0.	0.00
			RETURNS	640.	11.81
SENSIBLE HEAT			STEEL SCRAP	0.	0.00
OF THE BLAST	6373.	9.07	IRON SCRAP	1121.	20.66
or the beast	00704	,,,,,	FERROALLOYS	429.	
HEAT FROM ØXIDATION					
OF MN, FE, SI		34.93	COKE	320.	5.90
A HIN PE 31	243434	34.73	NATURAL GAS	0.	0.00
			· · · · · · · · · · · · · · · · · · ·	0.	0.00
TOTAL INPUT HEAT	70278.	100.00	FUEL UIL	٠.	0.00
			FLUX AND ADDITIVES	119.	2.20
OUTPUT HEAT			TEON AND FEBRUARES		2.20
OUTPUT HEAT			AIR	2766.	50.99
HEATING AND MELTING			ØXYGEN	0.	0.00
OF IRON	11223.	15 07	AVIOEN	•	0.00
משאז זש	11223.	12.77	CUPOLA LINING	29.	0.54
			CUPBEA LINING	27.	0.54
HEAT CONTENT				5.05	100 00
ØF THE SLAG	4148.	5.90	TOTAL INPUT MTLS	5425.	100.00
CALCINING OF					
	795.		OUTHUTE		
LIMESTONE	195.	1-13	OUTPUTS		
DECOMPOSITION			MOLTEN IRON	2000.	36.87
OF WATER	595.	0.45	HOLIEN INDIV	2000	
EL MUIEV	3734	0.03	SLAG	501.	9.24
TOP GASES			JENO	3011	,,,,
-SENSIBLE HEAT	£ 0.00		EMISSIONS DUST	27.	0.49
			ENITACIONA DOSI	210	0.47
-LATENT HEAT	24225.	34.47	701) 04555	2897.	53.40
			TOP GASES	_	
HEAT RADIATION FROM			NITRUGEN	2112.	72.91
THE CUPOLA	23390.	33.28	CARBON DIOXIDE	555.	7.67
			CARBON MONUXIDE	557.	17.24
			HYDRUGEN		0.08
VOLUME OF TOP GAS (MCF)		SULFUR DIØXIDE	3.	0.10
ACTUAL 971.					
STANDARD 386.					





INPUT HEAT	B.T.U.S (000/HR.)	PERCENT	MATERIAL BALAN	ICE_	
POTENTIAL HEAT OF FUEL	94371.	88.08	INPUTS METAL CHARGE PIG IRØN	PØUNDS 1989.	PERCENT 51.90 0.00
SENSIBLE HEAT OF THE BLAST		10.63	RETURNS STEEL SCRAP IRON SCRAP	426. 791. 761.	11.12 20.65 19.85
HEAT FROM ØXIDATION OF MN, FE, SI	1386.	1.29	FERROALLOYS	11.	0.28
-TOTAL INPUT HEAT	107146.	100.00	COKE Natural Gas Fuel Oil	233. 0. 0.	6.07 C.00 O.00
BUTPUT HEAT			FLUX AND ADDITIVES	31.	0.80
HEATING AND MELTING OF IRON	39279.	36.66	A IR Øxygen	1581 • 0 •	41.24
HEAT CONTENT OF THE SLAG	505.	0 • 47	CUPOLA LINING	0.	0.00
CALCINING ØF	786.	0.73	TØTAL INPUT MTLS	3833.	100.00
DECOMPOSITION			<u>OUTPUTS</u>		
OF WATER	1190.	1-11	MOLTEN IRON	5000•	52 • 18
TØP GASES -SENSIBLE HEAT	12693.	11.85	SLAG	27.	0.70
-LATENT HEAT	39035.	36.43	EMISSIONS DUST	16.	0.42
HEAT RADIATION FROM THE CUPOLA	13658.	12.75	TØP GASES Nitkøgen Carbøn diøxice	1790. 1207. 324.	46.70 67.43 18.09
VOLUME OF TOP GAS (ACTUAL 2018. STANDARD 801.	MCF)		CARBON MONOXIDE Hydrogen Sulfur Dioxide	257. i. 1.	14.34 0.07 0.08

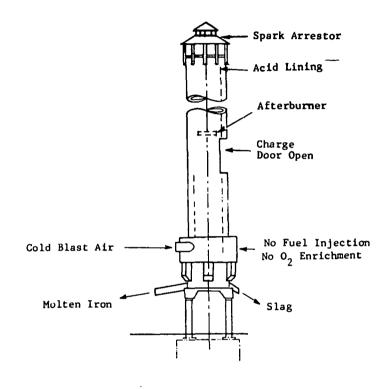


CUPOLA CLASSIFICATION - 11



MATERIAL BALANCE

INPUT HEAT	B.T.U.S		PINTENTAL BALAN	102	
INFO! BEK!	(000/HR+)	PERCENT			
	(000) 11/17	1 ENG CIVI	INPUTS		
				POUNDS	PERCENT
POTENTIAL HEAT	00040	B / 30	METAL CHARGE	2004.	52.08
ØF FUEL	29060•	86.72	PIG IRØN	0.	0.00
_			RETURNS	802.	20.83
SENSIBLE HEAT			STEEL SCRAP	1137.	29.56
OF THE BLAST	61.	0.18	IRON SCRAP	0.	0.00
			FERROALLØYS	65.	1.69
HEAT FROM OXIDATION	1				,
OF MN, FE, SI	4390•	13.10	COKE	167.	4.33
			NATURAL GAS	29.	0.75
TØTAL INPUT HEAT	33511.	100.00	FUEL ØIL	0.	0.73
			PUEL BIL	0.	0.00
•			ELLIV AND ADDITING		
GUTPUT HEAT			FLUX AND ADDITIVES	65•	1.69
DOTTOT HEAT					
HEATING AND MELTING			AIR	1556.	40.43
OF IRON	17623.	52.59	ØXYGEN	0.	0.00
0F 1.40W	11023	32.437			
HEAT CONTENT			CUPØLA LINING	27.	0.71
OF THE SLAG	274.	0.82			
OF THE BLAG	214.	0.02	TOTAL INPUT MTLS	3848.	100.00
4					
CALCINING OF		0.05			
LIMESTONE	756.	2.25	<u>ØUTPUTS</u>		
					
DECOMPOSITION			MOLTEN IRON	2000.	51.97
ØF WATER	3569•	10.65			
			SLAG	32.	0.83
TOP GASES					0.00
-SENSIBLE HEAT	6182•	18.45	EMISSIONS DUST	14.	0.37
-LATENT HEAT	6880+	20.53			0.37
			TØP GASES	1802.	46.84
HEAT RADIATION FROM	l		NITRØGEN	1188.	65.91
THE CUPOLA	-1772.	-5.29	CARBON DIOXIDE		
				507•	28.10
			CARBON MONOXIDE	99.	5.49
VOLUME OF TOP GAS (MCF)		HYDRØGEN	9•	0.47
ACTUAL 952.			SULFUR DIOXIDE	0.	0.02
STANDARD 375.					
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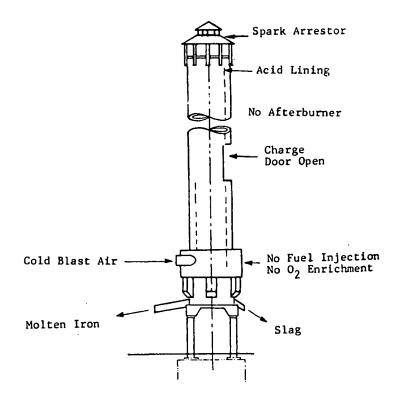
INPUT REAT

STANDARD

509.

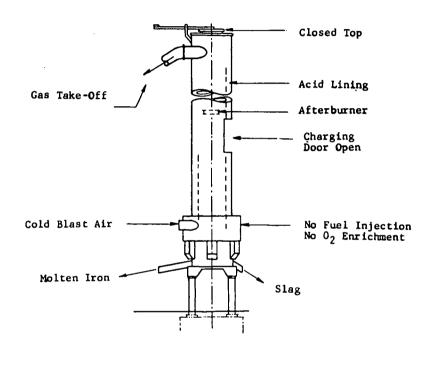
MATERIAL HALANCE

	(000ZHK.)	PERCENT	INPU1S		
				POUNDS	PEFCENT
PUTENTIAL HEAT			METAL CHARGE	1994.	39.73
OF FUEL	51194.	95.50	PIG IKØN	3ს9•	7.75
			KETUKNS	8C3.	15.99
SENSIBLE HEAT			STEEL SCHAP	730.	14.53
OF THE BLAST	90.	0.17	INUN SCHAP	49.	0.97
			FERKOALLOYS	24.	U • 4n
HEAT FROM MXIDATION					
OF MN. FE. SI	691•	1.33	CIKE	30∺.	6.13
			NATUKAL GAS	U•	0.00
TOTAL INPUT HEAT	51976.	100.00	FLEL GIL	U•	0.00
OUTPUT HEAT			FLUX AND ADDITIVES	66•	1.31
WOTFOT HEAT			Alis	2634.	52 • 45
HEATING AND MELTING			9XYGEN		0.00
OF INON	15712.	20. 62	38.105%	· ·	0.00
WF INVIN	137124	30.23	CUPOLA LINING	18.	0.35
HEAT CONTENT					
OF THE SLAG	511.	0.98	TOTAL INPUT MILS	5020•	100.00
CALCINING OF					
LIMESTONE	674.	1.30	OUTPUTS		
DECOMPOSITION			MOLTEN IRON	2000-	39.64
OF WATER	793•	1.53			
			SLAG	64.	1.27
TOP GASES					
-SENSIBLE HEAT	8281•	15.93	FMISSIONS DUST	24.	0.47
-LATENT HEAT	9439.	18.16			
			TOP GASES	2932.	56 • 42
HEAT RADIATION FROM			NITRUGEN	2011.	68.59
THE CUPOLA	16567.	31.87	CARBON DIBKIDE	762.	25.99
			CARBON MONUXIDE	155.	5.29
			HYDK@GEN	8•	0.07
VØLUME OF TOP GAS C	MCF)		SULFUN DIGAIDE	1.	0.05
ACTUAL 1283.					
CTANDALD 500					



MATERIAL BALANCE

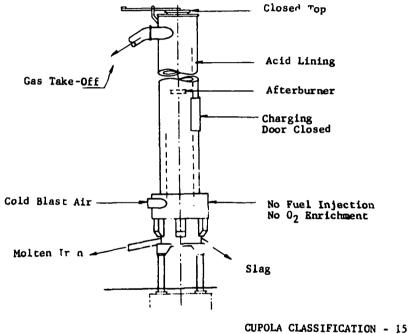
INPUT HEAT	B.T.U.S (000/HK.)	PERCENT	INPUTS	POUNDS	PERCENT
			METAL CHARGE	2005.	45.94
POTENTIAL HEAT	1		PIG IRON	0.	0.00
OF FUEL	31164.	98.54	RETURNS	201 •	4.59
Dr rocz	*		STEEL SCRAP	301.	6.89
SENSIBLE HEAT			IRON SCRAP	1504.	34.45
OF THE BLAST	49•	0.16	FERROALLOYS	0.	0.00
HEAT FROM GXIDATION			CØKE	250•	5.73
OF MN, FE, SI	414.	1.31	NATURAL GAS	0•	0.00
V			FUEL ØIL	0.	0.00
TOTAL INPUT HEAT	31628+	100.00	FOEL DIE		
101112 1111 1111			FLUX AND ADDITIVES	75•	1.73
OUTPUT HEAT			4.1D	2005•	45.94
BUITUS HEAT			AIR	0.	0.00
HEATING AND MELTING	1		ØXYGEN	•	
OF IKON	11160.	35.29	CUPOLA LINING	29.	0.67
HEAT CONTENT			TOTAL INPUT MILS	4365•	100.00
OF THE SLAG	459•	1 • 45			
CALCINING OF		1.73	<u>ØUTPUTS</u>		
LIMESTONE	547•	1.73			45.82
			MOLTEN IRON	5000•	43.06
DECOMPOSITION		1.36			1.77
OF WATER	431 •	1.36	SLAG	77.	1 • 7 7
					0.43
TOP GASES		1 A AU	EMISSIONS DUST	19•	0.43
-SENSIBLE HEAT	4581 •	14.48			51.99
-LATENT HEAT	9802•	30.99	TOP GASES	2269•	67.47
			NITRØGEN	1531•	
HEAT RADIATION FROM			CARBON DIOXIDE	507.	22.34
THE CUPBLA	4647.	14.69	CARBON MONOXIDE	226.	9.94
			HYDRØGEN	2.	0.07
			SULFUR DIOXIDE	4.	0.18
VØLUME ØF TØP GAS	(MCF)		· -		
ACTUAL 719.					
STANDAKO 285.					





 	.ANC	•

INPUT HEAT	8.T.U.S		MATERIAL BALAN	CE	
THE OF THEM	(000/HR.)	PERCENT	INPUTS		
				PØUNDS	PERCENT
POTENTIAL HEAT	35065.	08.72	METAL CHARGE	1999.	38 • 79
OF FUEL	33063.	70 • 73	PIG IRØN	400.	7.76
CENCIONE VEAT			RETURNS	800.	15•52 11•64
SENSIBLE HEAT OF THE BLAST	53.	0.15	STEEL SCRAP	600•	11.64
AL THE BENST	33.	00.15	IRØN SCRAP	200•	
HEAT FROM OXIDATION			FERRØALLØYS	0.	0.00
OF MN, FE, SI		1.12			
OF PINS PES 31	3774		COKE		6.93
TOTAL INPUT HEAT	35518.	100.00	NATURAL GAS	0.	0.00
INTAL THEOL NEWS	333.01		FUEL ØIL	0•	0.00
GUTPUT HEAT			FLUX AND ADDITIVES	60.	1.16
			AIR	2709.	52.55
HEATING AND MELTING OF IRON	9312.	26.22	ØXYGEN	٥.	0.00
HEAT CONTENT			CUPOLA LINING	29.	0.56
OF THE SLAG	420•	1.18	TOTAL INPUT MTLS	5154.	100.00
CALCINING OF					
LIMESTONE	350•	0.99	<u>@UTPUTS</u>		
DECOMPOSITION			MØLTEN IRØN	2000•	38.81
OF WATER	466.	1.31			•
TOP GASES			SLAG	72.	1 • 40
-SENSIBLE HEAT	4945.	13.92			
-LATENT HEAT	12530.	35.28	EMISSIONS DUST	26.	0.51
HEAT RADIATION FROM			lor GASES	3055.	
THE CUPOLA	7495.	21.10	NITKØGEN	2068.	67-68
THE CUPBLA	74731	21110	CARBON DIOXIDE	623.	20.38
•			CARBON MONOXIDE		11.79
VOLUME OF TOP GAS	MCE)		HYDRØGEN		0.07
ACTUAL 780. STANDARD 310.			SULFUR DIƏXIDE	2.	0.07



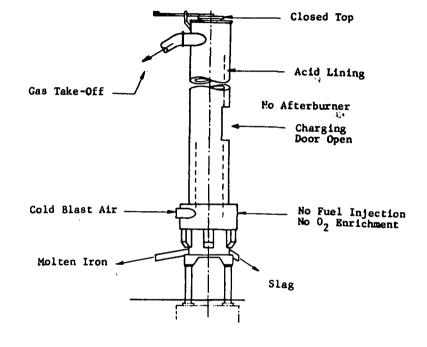


STANDARD

266.

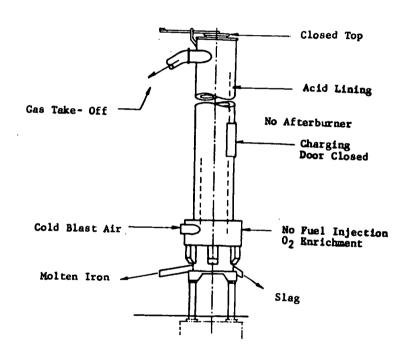
MATERIAL BALANCE

INPUT HEAT	B.T.U.S				
	(000/HR.)	PERCENT	<u>INPUTS</u>		
			·	POUNDS	PERCENT
POTENTIAL HEAT			METAL CHARGE	2002.	43.10
OF FUEL	25435.	98.08	PIG IRON	548•	11.79
			RETURNS	896.	19.30
SENSIBLE HEAT			STEEL SCRAP	548•	11.79
OF THE BLAST	48.	0.19	IRØN SCRAP	0.	0.00
•			FERROALLOYS	10.	0.21
HEAT FROM OXIDATION	l				
* OF MN, FE, SI	450.	1.74	COKE	250•	5.38
			NATURAL GAS	0.	0.00
TOTAL INPUT HEAT	25934.	100.00	FUEL ØIL	0.	0.00
		•	FLUX AND ADDITIVES	52.	1.12
OUTPUT HEAT			AIR	2305.	49.63
ζ΄.			ØXYGEN	0.	0.00
HEATING AND MELTING		36.78	BATGEN	٠.	0.00
· OF INON	9539.	36.78	CUPOLA LINING	36.	0.77
HEAT CONTENT					
OF THE SLAG	485.	1.87	TOTAL INPUT MTLS	4644.	100.00
CALCINING OF					
LIMESTONE	322.	1.24	OUTPUTS		
DECOMPOSITION			MOLTEN IRON	2000.	43.06
- OF WATER	421.	1.63			
			SLAG	76.	1.64
TOP GASES					
-SENSIBLE HEAT	4365.	16.83	EMISSIONS DUST	21.	0.45
-LATENT HEAT	2615.	10.08			
•		•	TOP GASES	2547•	54.85
HEAT RADIATION FROM	1		NITRØGEN	1760.	69.09
THE CUPOLA	8186.	31.57	CARBON DIOXIDE	714.	28.03
			CARBON MONOXIDE	71.	2.78
			HYDRØGEN	2.	0.07
VOLUME OF TOP GAS (MCF)		SULFUR DIBXIDE	1.	0.03
ACTUAL 671.					
CZANDADD OCC					

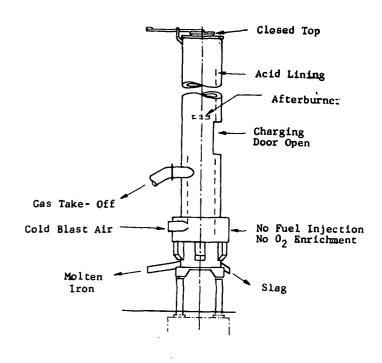




INPUT HEAT	B.T.U.S (000/HR.)	Becamin	MATERIAL BALA	NCE	
	(000) AK.)	PERCENT	INPUTS		
POTENTIAL HEAT			1013	_	
ØF FUEL	144207.	97.67	METAL CHARGE	POUNDS	PERCENT
		,,,,,,	PIG IRØN	1984.	53.30
SENSIBLE HEAT			RETURNS	861.	23.13
OF THE BLAST	194.	0.13	STEEL SCRAP	186.	5.00
		0113	IRON SCRAP	931.	25.01
HEAT FROM OXIDATION			FERROALLOYS	0.	0.00
OF MN, FE, SI	3240.	2.19	. THEFT IS	6.	0.16
		,	COKE		
TOTAL INPUT HEAT	147641.	100.00	NATURAL GAS	242.	6.51
		.00100	FUEL OIL	0.	0.00
			, occ oic	0.	0.00
SUTPUT HEAT			FLUX AND ADDITIVES	40.	1.07
HEATING AND MELTING			AIR		
OF IRON	62872.	42.5B	9XYGEN	1383.	37.16
		72130	AVIGEN	61.	1 - 64
HEAT CONTENT OF THE SLAG	00.40		CUPOLA LINING	12.	0.33
or the serie	2249.	1.52			000
CALCINING OF			TOTAL INPUT MTLS	3722.	100.00
LIMESTONE	1604.				
	1004.	1.09			
DECOMPOSITION			OUTPUTS		
OF WATER	1636.		***		
	1030.	1.11	MOLTEN IRON	2000.	53.73
TOP GASES			***		
-SENSIBLE HEAT	18321.	10	\$LAG	57.	1.52
-LATENT HEAT	49633.	12.41	PMSGGGGGG		
	47033.	33.62	EMISSIONS DUST	15.	0.40
HEAT RADIATION FROM			TOD CARRO		
THE CUPOLA	11324.	7.67	TOP GASES Nitrogen	1651.	44.35
		****		1056.	63.97
			CARBON DIOXIDE	382.	23.17
VOLUME OF TOP GAS (M	CF)		CARBON MONOXIDE Hydrogen	208.	12.58
ACTUAL 2863.	. ,			1.	0.07
STANDARD 1137.			SULFUR DIGXIDE	4.	0.21
					



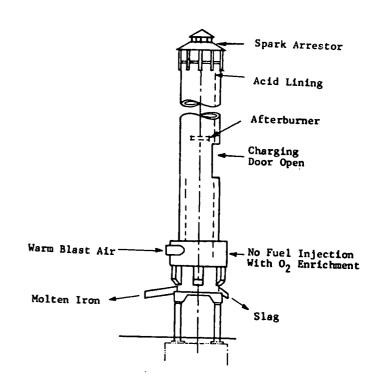
INPUT HEAT	B.T.U.S		MATERIAL BALAN	ICE.	
	(000/Hk·)	PERCENT	INPUTS		
POTENTIAL HEAT				POUNDS	PERCENT
	53096.	98.37	METAL CHARGE	5005.	56.15
DY YOEL	330700	70.37	PIG IRØN	0.	0.00
SENSIBLE HEAT			KETUKNS	400.	11.23 11.23 33.69
OF THE BLAST	62.	0.12	STEEL SCHAP	400.	11.23
or the beast			IRUN SCRAP	1201.	33.69
HEAT FROM OXIDATION		*	FERROALLOYS	0.	0.00
	816.	1.51			
J. 1 125 G.	•••		COKE	222.	6.23
TOTAL INPUT HEAT	53975.	100.00	NATURAL GAS	0.	0.00
TOTAL IMOT NEAT	30770		FUEL OIL	0.	0.00
OUTPUT HEAT			FLUX AND ADDITIVES	56.	1.57
HEATING AND MELTING			AIL	1268.	35.56
	22445.	41.59	ØXYGEN	0.	0.00
HEAT CONTENT			CUPJLA LINING	18.	0.49
	885.	1 • 64	TOTAL INPUT MTLS	25.4	100.00
			ININE INFO MILS	3300•	100.00
CALCINING OF					
LIMESTONE	816.	1.51	OUTPUTS		
DECOMPOSITION					
OF WATER	545.	1.01	MOLTEN IRON	2000.	56.09
TØP GASES			SLAG	68•	1.90
-SENSIBLE HEAT	6038.	11-19			
-LATENT HEAT			EMISSIONS DUST	14.	0.40
HEAT RADIATION FROM			TOP GASES	1484.	41.61
THE CUPOLA	-8559.	-15.86	NITKØGEN	968.	65.24
			NITKOGEN CARBON DIOXIDE CARBON MONOXIDE	147.	9.90
VOLUME OF TOP GAS (MCF)		HYDKØGEN		0.07
ACTUAL 985. STANDARD 391.			SULFUR DIOXIDE	2.	0.15





MATERIAL BALANCE

INPUT HEAT	B.T.U.S		MATERIAL BALA	NCE	
	(000/HR.)	PERCENT	INPUTS	_	
POTENTIAL HEAT				POUNDS	PERCENT
OF FUEL	68407.	90.79	METAL CHARGE	2010.	44.99
	00-07.	70.77	PIG IRØN Returns	0.	0.00
SENSIBLE HEAT				1873.	41.91
OF THE BLAST	5504.	7.31	STEEL SCRAP IRON SCRAP	125.	2.79
		7.51	FERROALLOYS	0.	0.00
HEAT FROM OXIDATION			L CUMBALLUIS	13.	0.29
OF MN, FE, SI	1434.	1.90	COKE		
			NATURAL GAS	250.	5.59
TOTAL INPUT HEAT	75345.	100.00	FUEL OIL	0.	0.00
			, ozt bit	0.	0.00
BUTPUT HEAT			FLUX AND ADDITIVES	101.	2.25
			AIR	2089.	
HEATING AND MELTING			ØXYGEN		46.75
ØF IRØN	25332.	33.62		0.	0.00
HEAT CONTENT			CUPBLA LINING	18.	0.41
OF THE SLAG	1103.	1 • 46	TOTAL INPUT MTLS	4468.	100.00
CALCINING OF					
LIMESTONE	1604.	2-13	OUTPUTS		
DECOMPOSITION OF Water	567.	0.75	MOLTEN IRON	2000.	44.76
TOP GASES			SLAG	93.	2.08
-SENSIBLE HEAT	11610.	15.41	EMISSIONS DUST	00	_
-LATENT HEAT	17324.	22.99		20.	0.46
			TOP GASES	2355.	
HEAT RADIATION FROM			NITROGEN	1600.	52.70
THE CUPOLA	17804.	23.63	CARBON DIOXIDE	572.	67.94
			CARBON MONOXIDE	181.	24.27
tion time and and			HYDRØGEN	101.	7.69
VOLUME OF TOP GAS (M	CF)		SULFUR DIØXIDE	1.	0.04
ACTUAL 1734. Standard 644.			377	1.	0.05

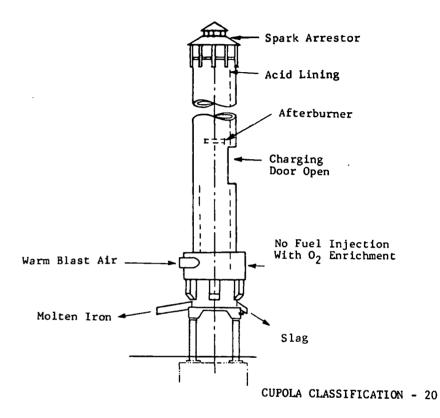


STANDARD

746.

MATERIAL BALANCE

TANDET MEAT	B.T.U.S		MATERIAL BALANCE			
INPUT HEAT	_	PERCENT	TAILUITE			
	(000/11110/		INPUTS	U.O.L.A.D.O.		
RØTENTIAL HEAT			METAL CHALCE		PERCENT	
WF FUEL	69898-	87.22	METAL CHARGE	1983.	43.78	
WY FUEL	0,0,0	0	PIG INON	0.	0.00	
SENSIBLE HEAT			KETUKNS		8.60	
	9237.	11.53	STEEL SCRAP	974.		
OF THE BLAST	7237.	11.33	INON SCHAP		12.90	
			FERROALLØYS	35.	0.76	
HEAT FROM OXIDATION		. 05				
OF MN, FE, SI	1004+	1.25	COKE	250.	5.52	
_			NATURAL GAS	0.	0.00	
TOTAL INPUT HEAT	80140.	100.00	FUEL WIL	U •	0.00	
			FLUX AND ADDITIVES	26.	0.57	
AUTPUT HEAT						
			AIk	2213.	48.85	
HEATING AND MELTING			のみYGEN	41.	0.90	
ØF IKØN	29100.	36.31				
			CUPULA LINING	18.	0.39	
HEAT CONTENT						
OF THE SLAG	695•	0.87	TOTAL INFUT MILS	4530•	100.00	
CALCINING OF						
LIMESTONE	474.	0.59	OUTPUTS			
		•	0011013			
DECOMPOSITION			MOLTEN IRØN	2000•		
OF WATER	1190.	1 - 48	HOLIEN INUN	2000.	44.15	
			SLAG	2.4		
TOP GASES			JEAN	36.	0.79	
-SENSIBLE HEAT	12431.	15.51	EMISSIONS DUST			
-LATENT HEAT	-3301.	-4.12	EM1221002 DO21	55.	0 • 48	
EATER HEIL	00011		TOP GASES	0.50		
HEAT KADIATION FROM			——————————————————————————————————————	2472.	54•5ช	
THE CUPULA		49.35	NITHOGEN	1690.	68.34	
IIIL COLUCA	373321	7/103	CARBON DIOXIDE	810.	32.76	
			CAPRON WONOXIDE	-30.	-1.23	
VOLUME OF TOP GAS (VCE1		HYDKØGEN	S•	0.07	
	WIGF J		SULFUR DIMXIDE	1 •	0.06	
ACTUAL 1879.						





MATERIAL BALANCE

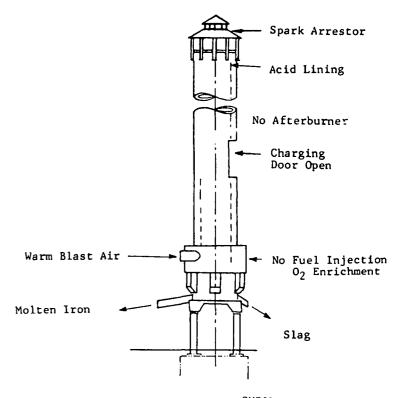
INPIT HEAT	B.T.U.S.		MATERIAL BALANCE		
	(000/HR.)	PERCENT	INPUTS		
POTENTIAL HEAT				POUNDS	PERCENT
OF FUEL	9985.	70.	METAL CHARGE	2001.	25.42
6 8 9 9 9 9 9 9 9 9 9 9	, , e J •	79.64	PIG IRØN	0 •	0.00
SENSIBLE HEAT			RETURNS	566.	7.19
OF THE BLAST	1337.	13	STEEL SCRAP	1321.	16.78
110.45		10.66	IRON SCRAP	0.	0.00
HEAT FROM OXIDATIO	N		FERRØALLØYS	113.	1 • 44
OF MN, FE, SI	1216.	0	eau e		
TOTAL .		9 • 70	COKE	308.	3.91
TOTAL INPUT HEAT	12538.		NATURAL GAS	0.	0.00
durous		100.00	FUEL OIL	0.	0.00
SUTPUT HEAT			PLUV AND ADD		
UDATING AND			FLUX AND ADDITIVES	120.	1.52
HEATING AND MELTIN	G		AIR		
OF IRON	3367.	04.05		5361.	68.12
UFAT CONTENT		26.85	ØXYGEN	0.	0.00
HEAT CONTENT			CUDOLA LANGUE		
OF THE SLAG	401.	3.20	CUPOLA LINING	81.	1.03
CALCINING OF		3.80	TATAL INDUSTRIA		
LIMESTONE			TOTAL INPUT MTLS.	7871.	100.00
LIMESIANE	262.	2.09	OUTPUTS		
DECOMPOSITION		2.09	WOLFOLS		
OF WATER			MOLTEN IRON		
WATER	198.	1.58	HOLIEN TRING	2000.	25 - 41
TOP GASES		1 • 30	SLAG		
- SENSIBLE HEAT				178.	2.26
- LATENT HEAT	3371.	26.89	EMISSIONS DUST	. .	
- TENT NEAT	13044.	104.04	20101010 0131	54.	0.69
EAT RADIATION FROM			TOP GASSES	5 4 9 9	
THE CUPOLA			NITROGEN	5639.	71.64
= 0 % DEH	17982.	143.42	CARBON DIOXIDE	4107.	72.82
		. 5-76.	CARBON MONOXIDE	2529.	44.85
OLUME OF TOP GAS (MCC		HYDROGEN	-1000.	-17.74
CTHAL 487.	11.F)		SULFUR DIOXIDE	3.	0.04
TAMBARD 193.			- STRAIDE	1.	0.03

Spark Arrestor Acid Lining No Afterburner - Charge Door Open Warm Air Blast -With Fuel Injection No O₂ Enrichment Molten Iron -Slag



MATERIAL BALANCE

INPUT HEAT	.B.T.U.S (000/Hk.)	PEKCENT	INPUTS	Laund	LEDGENT
				FOUNDS	PERCENT
POTENTIAL HEAT			METAL CHARGE	2020• 793•	50 • 68 19 • 90
OF FUEL	43896.	92.59	PIG IKØN		19.90
-			RETURNS	793 • U •	0.00
SENSIBLE HEAT			STEEL SCRAP		
OF THE BLAST	8600.	5 • 48	IRØN SCRAP	396.	9.95
			FERRØALLØYS	37.	0.93
HEAT FROM ØXIDATIØN	l		2211	011	5 (14)
OF MN. FE. SI	915.	1.93	COKE	211.	5.28
			NATURAL GAS	0 • 0 •	0.00
TOTAL INPUT HEAT	47411.	100.00	FUEL ØIL	0.	0.00
•			FLUX AND ADDITIVES	30•	0.76
			FEON AND ADDITIVES	30.	0.70
BUTPUT HEAT			Alk	1700.	42.67
			ØXYGEN	0.	0.00
HEATING AND MELTING			. DATELY	• • • • • • • • • • • • • • • • • • • •	0.00
ØF IRON	17956.	37.87	CUPOLA LINING	24.	0.61
HEAT CONTENT					
OF THE SLAG	718.	1.51	TØTAL INPUT MILS	3985.	100.00
OF THE SEAG	710+	1 • 31			
CALCINING OF					
LINESTONE	350.	0.74	<u>ØUTPUTS</u>		
		•	==		
DECOMPOSITION			MULTEN IKON	2000•	50.19
OF WATER	585.	1.23		F.1.	
			SLAG	58∙	1 • 45
TØF GASES			ENICCIONE DU UT		
-SENSIBLE HEAT	6186.	13.05	EMISSIUNS DUST	16.	0.40
-LATENT HEAT	15923.	33.59	TOU CAUSE	1011	43 05
			TOP GASES	1911.	47.95
HEAT RADIATION FLOM			NITHUGEN	1298.	67.93
THE CUPOLA	5693.	12.01	CARBON DIOXIDE	381.	19.94
			CARBON MONUXIDE	229•	11.98
			HYDKØGEN	1.	0.07
VOLUME OF TOP GAS (MCF)		SULFUR DIOXIDE	1.	0.07
ACTUAL 978.	•				
STANDARD 388.					



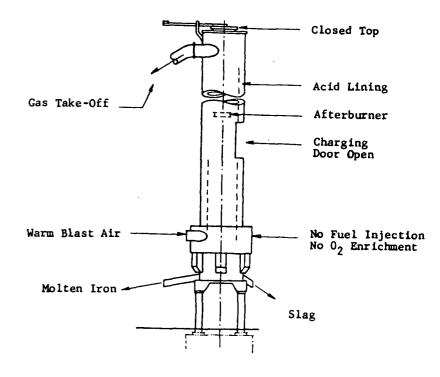


STANDARD

348.

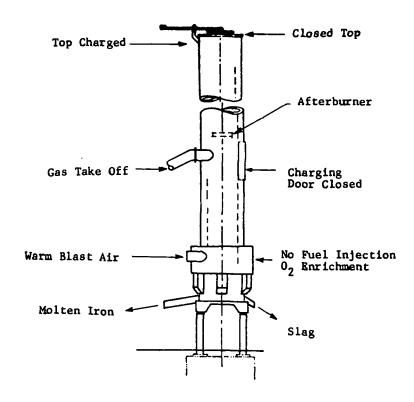
MATERIAL BALANCE

	MATERIAL BALANCE						
INPUT HEAT	B.T.U.S (000/HK.)	PERCENT	INPUTS	Launs	UELCENT		
POTENTIAL HEAT			METAL CHARGE	POUNDS 2024•	PERCENT 52.36		
OF FUEL	33777•	83.55					
OF FUEL	337776	03.33	PIG IKUN	319.	8.25		
SENSIBLE HEAT			KETUKNS	996• .			
OF THE BLAST	3977.	9.84	STEEL SCHAP	0.			
WE THE BLAST	37110	7.04	IKON SCHAP	677•			
HEAT FROM OXIDATION			FERHUALLUYS	32.	0.62		
		6 • 61					
OF MN, FE, SI	2673.	0+01	COKE	167.	4.31		
TOTAL INDUT HEAT	40.40.	100 00	NATURAL GAS	0.	0.00		
TOTAL INPUT HEAT	40427.	100.00	FUEL ØIL	0.	0.00		
GUTUUT UEAT			FLUX AND ADDITIVES	68∙	1.76		
<u>ØUTPUT HEAT</u>							
			AIR	1585.	40.99		
HEATING AND MELTING			ØXYGEN	0.	0.00		
ØF IRØN	18123.	44.83					
			CUPWLA LINING	55.	0.57		
HEAT CONTENT							
OF THE SLAG	1157.	2.86	TOTAL INPUT MTLS	3866•	100.00		
CALCINING OF							
LIMESTONE	751•	1.86	<u>autputs</u>				
DECOMPOSITION			MØLTEN IKØN	2000•	51.73		
ØF WATER	545.	1.35			223.0		
			SLAG	95.	2.46		
TOP GASES			52	, • •			
-SENSIBLE HEAT	5668•	14.02	EMISSIONS DUST	14.	0.35		
-LATENT HEAT	5977.	14.78	Emissions bosi	•	0.00		
		•	TOP GASES	1757.	45.45		
HEAT KADIATION FROM			NITRØGEN	1210.	68.87		
THE CUPOLA	8205.	20.29	CARBON DIGXIDE	459•	26.11		
,			CARBON MONOXIDE	66.	4.89		
			HYDRØGEN	1.	0.07		
VOLUME OF TOP GAS (MCF)		SULFUR DIOXIDE	1.	0.07		
ACTUAL 878.			SOLFER DIBNIDE	, •	0.08		



MATERIAL BALANCE

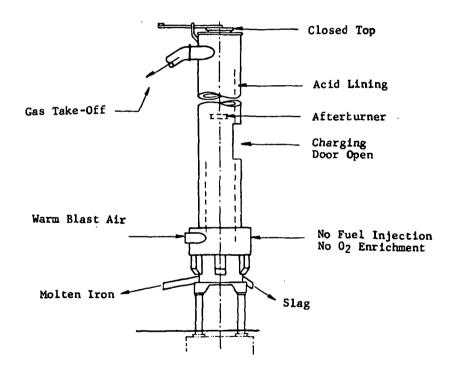
INDUT USAT	MATERIAL BALANCE						
INPUT HEAT	8.T.U.S (000/HR.)	PERCENT	INPUTS		0505547		
POTENTIAL HEAT				PØUNDS	PERCENT		
ØF FUEL	95 775	80 A/	METAL CHARGE	2009•	55 • 42		
OF FUEL	85775.	07 • 46	PIG IRØN	502•	13.86		
SENSIBLE HEAT			RETURNS	301 •	8.31		
OF THE BLAST	4170	0.50	STEEL SCRAP	301 •	8.31		
OF THE BLAST	8173.	8 • 52	IRØN SCRAP	904•	24.94		
HEAT FROM OXIDATION			FERROALLOYS	0•	0.00		
OF MN, FE, SI		2.02					
OF HIMS PES 21	1937.	5.05	CØKE	182.	5.02		
TOTAL INDUT NEAT	05004		NATURAL GAS	0.	0.00		
TOTAL INPUT HEAT	95886•	100.00	FUEL ØIL	0.	0.00		
OUTPUT HEAT			FLUX AND ADDITIVES	44.	1.22		
INSH TUTION					54.40		
HEATING AND MELTING			AIR	1335.	36 - 82		
OF IRON	43043.	44.89	ØXYGEN	40 •	1 - 1 1		
or India	43043•	44.89			0 43		
HEAT CONTENT			CUPØLA LINING	15.	0.42		
OF THE SLAG	1 450 •	1.51		2405	100.00		
or the serio	1430.	1.31	TOTAL INPUT MTLS	3625.	100.00		
CALCINING ØF							
LIMESTONE	1219.	1.27	ØUTPUTS				
			2017 013				
DECOMPOSITION			MOLTEN IRON	2000•	55.17		
ØF WATER	1091 •	1.14	THE TEN THEM	2000	30.1		
			SLAG	56.	1.56		
TØP GASES			32				
-SENSIBLE HEAT	11927.	12.44	EMISSIONS DUST	12.	0.34		
-LATENT HEAT	23741 •	24.76	2	• -			
•			TØP GASES	1556.	42.92		
HEAT RADIATION FROM			NITRØGEN	1019.	65.49		
THE CUPOLA	13415.	13.99	CARBON DIOXIDE	391 •	25.14		
			CARBON MONOXIDE	144.	9.24		
			HYDRØGEN	1.	0.07		
VOLUME OF TOP GAS (M	ICF)		SULFUR DIBXIDE	1.	0.06		
ACTUAL 1852.							
STANDARD 735.							





MATERIAL BALANCE

INPUT HEAT	B.T.U.S				
	(000/Hk.)	PERCENT	INPUTS		
				POUNDS	PERCENT
POTENTIAL HEAT			METAL CHARGE	5009•	42.04
OF FUEL	54422.	89.26	PIG IRØN	574.	12.01
			KETURNS	954.	19.97
SENSIBLE HEAT			STEEL SCRAP	287.	6.01
OF THE BLAST	5225.	6.57	IRØN SCRAP	134.	2.80
			FERROALLØYS	60•	2.80 1.25
HEAT FROM OXIDATION					
OF MN. FE. SI	1323.	2.17	COKE	182.	3.80
			NATUKAL GAS	0•	0.00
TOTAL INPUT HEAT	60970.	100.00	FUEL ØIL	0.	0.00
			FLUX AND ADDITIVES	62•	1.30
OUTPUT HEAT					
			AIK	2506•	
HEATING AND MELTING			ØXYGEN	0.	0.00
OF IKON	28057•	46.02			
			CUPOLA LINING	18.	0.37
HEAT CONTENT					
OF THE SLAG	797•	1.31	TOTAL INPUT MILS	4779.	100.00
CALCINING OF					
LIMESTUNE	1130.	1.85	BUTPUTS		
DECOMPOSITION			MOLTEN IKON	5000•	41.85
	1349.	2.21			
			SLAG	58.	1.22
TOP GASES					
-SENSIBLE HEAT	13526.	22.19	EMISSIONS DUST	21.	0.45
-LATENT HEAT	-29511.	-48-40			
			TOP GASES	2699.	56.48
HEAT EADIATION FROM	I		NITRØGEN	1915.	70.95
THE CUPGLA	45622.	74.83	CARBON DIOXIDE	1054.	39.04
			· CARBON MONUXIDE	-272.	-10.06
			HYDRØGEN		0.08
VOLUME OF TOP GAS C	MCF)		SULFUR DIOXIDE	0.	
ACTUAL 2000.					
STANDAKD 794.					



CUPOLA CLASSIFICATION - 25

MATERIAL BALANCE

INPUT HEAT	B.T.U.S	PERCENT	INPUTS	POUNDS	PERCENT
	(000/HR.)	PERCENT		2001	44.08
			METAL CHARGE	429.	9.44
POTENTIAL HEAT			PIG IRØN	698.	15.37
OF FUEL	48492•	91.27	RETURNS		10.98
			STEEL SCRAP	498•	7.55
SENSIBLE HEAT			IRON SCRAP	343•	0.75
OF THE BLAST	3539•	6-66	FERROALLOYS	34•	0.75
FROM AVIDATION			CØKE	190•	4.20
HEAT FROM OXIDATION	1102.	2.07	NATURAL GAS	0.	0.00
OF MN, FE, SI	1.02		FUEL BIL	0.	0.00
	53133.	100.00	PUEL DIL		
TOTAL INPUT HEAT	23133+	100100	FLUX AND ADDITIVES	56•	1.23
				2263.	49.85
OUTPUT HEAT			AIR Øxygen	0.	0.00
HEATING AND MELTING OF IRON	24690•	46-47	CUPOLA LINING	29•	0.64
HEAT CONTENT OF THE SLAG	949.	1.79	TOTAL INPUT MTLS	4540•	100.00
CALCINING ØF			OUTPUTS_		
LIMESTONE	898•	1 - 69			44.05
			MOLTEN IRON	2000•	44.03
DECOMPOSITION OF WATER	1071•	2.02	SLAG	65•	1 - 43
TOP GASES	10840	20.41	EMISSIONS DUST	23.	0.50
-SENSIBLE HEAT	10842	-30-15		2453.	54.02
-LATENT HEAT	-16022.	-30.13	TOP GASES	1728	70.46
			NITROGEN	890.	36.28
HEAT RADIATION FRO			CARBON DIOXIDE	-168.	-6.83
THE CUPBLA	30705•	57.79	CARBON MONOXIDE		0.08
			H YDRØGEN	2•	0.03
			SULFUR DIØXIDE	1 -	0.03
VOLUME OF TOP GAS	(MCF)				

VOLUME OF TOP GAS (MCF) ACTUAL 1618. STANDARD 642. Gas Take-Off

T Afterburner

Charging Door Closed

No Fuel Injection 02 Enrichment

Molten Iron

Slag

CUPOLA CLASSIFICATION - 26

Closed Top



ACTUAL

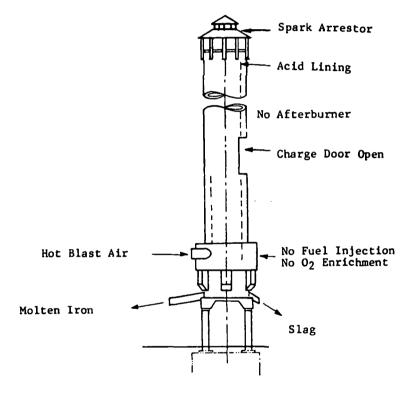
STANDARD

2917.

1158.

MATERIAL BALANCE

INPUT HEAT	B•T•U•S		INPUTS		
		PERCENT		POUNDS	
		2	METAL CHARGE	_	47.94
PUTENTIAL HEAT	• •		PIG IKØN	287.	
ØF FUEL	97700.	80.01	KETUKNS	292•	_
			STEEL SCHAP	143.	_
SENSIBLE HEAT			IKØN SCKAP	1290.	
OF THE BLAST	22327.	18.28	FERROALLOYS	0•	0.00
HEAT FROM OXIDATION			COKE	174.	4.14
OF MN. FE. SI		1 - 71	NATURAL GAS	0.	0.00
01 11117 127 31	2003.		FUEL ØIL	0•	0.00
TOTAL INPUT HEAT	122112.	100.00	FLUX AND ADDITIVES	56•	1.34
411.1117.1174.7			Alk	1947.	46.38
BUIPUT HEAT			ØXYGEN	0.	0.00
UEATING AND MELTING			SATOLIA .		
HEATING AND MELTING OF IKON	51441•	A9 . 19	CUPOLA LINING	8•	0.19
OF INDIA	214414	42.13			
HEAT CONTENT			TOTAL INPUT MTLS	4196.	100.00
OF THE SLAG	1122.	0.92			
3 Jan.		0072			
CALCINING OF			<u>ØUTPUTS</u>		
LIMESTONE	1837.	1.50			
			MOLTEN IKON	2000.	47.66
DECOMPOSITION					
OF WATER	1884.	1.54	SLAG	49.	1.17
TOP GASES			EMISSIONS DUST	16.	0.37
-SENSIBLE HEAT	19276.	15.79			
-LATENT HEAT		-B.77	TOP GASES	2132.	50.80
GIVE HEAT	10114	00,,	NITROGEN	1466.	69.72
HEAT KADIATION FROM			CARBON DIOXIDE	698•	32.75
	57267.	46.90	CARBON MONOXIDE	-55•	
00.02.	3.2011	-3170	HYDRØGEN	2.	
			SULFUR DIØXIDE	0.	0.05
VOLUME OF TOP GAS	MCF)				

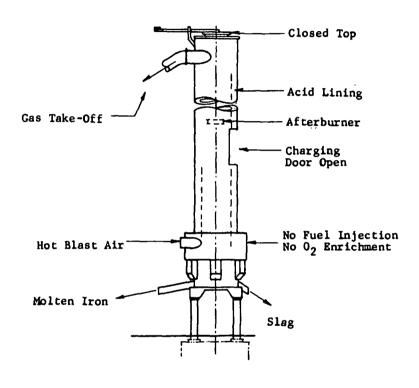


CUPOLA CLASSIFICATION - 27

EXHIBIT 3 PAGE 27

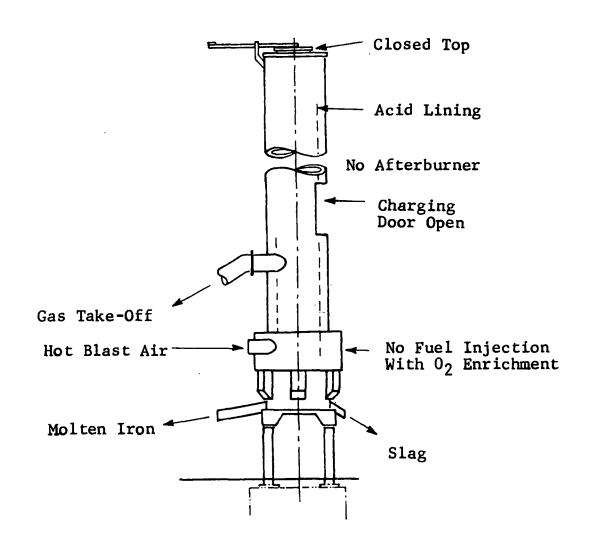
MATERIAL BALANCE

INPUT HEAT	B.T.U.S		***************************************		
	(000/HK•)	PERCENT	INPUTS	PAUNDS	PERCENT
POTENTIAL HEAT			METAL CHARGE	2001.	45.19
OF FUEL	52620.	82.89	PIG IKUN	400.	9.04
WF FOEL	520201	02407	KETUKNS	867.	19.58
SENSIBLE HEAT			STEEL SCRAP	534.	12.05
OF THE BLAST	9878.	15.56	IKON SCRAP	200.	4.52
of the beast	70707	13,030	FERROALLOYS	0.	0.00
HEAT FROM OXIDATION					
OF MN. FE. SI	961.	1.54	COKE	222.	5.02
0, 1,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		, , ,	NATUKAL GAS	0.	0.00
TOTAL INPUT HEAT	63478•	100.00	FUEL DIL	0.	0.00
			FLUX AND ADDITIVES	40.	0.90
MUTPUT HEAT					
			AIR	2144.	48 • 42
HEATING AND MELTING			UXYGEN	0.	0.00
OF IRON	23029.	36.28	CINICIA A TAITNA	21.	0 47
USAT ACUTOUT			CUPØLA LINING	21.	0.47
HEAT CONTENT OF THE SLAG	789.	1.24	TOTAL INPUT MTLS	4428.	100.00
CALCINING OF					
LIMESTONE	583•	0.92	<u>@UTPUTS</u>		
DECOMPOSITION			MØLTEN IKØN	2000•	45.17
OF WATER	922.	1 • 45			
			SLAG	53.	1.19
TOP GASES					
-SENSIBLE HEAT	9487.	14.95	EMISSIONS DUST	20.	0.45
-LATENT HEAT	1458.	2.30			
			TØP GASES	2355•	53.19
HEAT RADIATION FROM			NITRØGEN	1637.	69.50
THE CUPOLA	27209.	42.86	CARBON DIOXIDE	699•	29.67
			CARBON MONOXIDE	17.	0.71
			HYDRØGEN	2.	0.07
VOLUME OF TOP GAS (MCF)		SULFUR DIWXIDE	1 •	0.04
ACTUAL 1450.					
STANDARD 576.					





Insufficient Data for Heat and Material Balance Calculations



MATERIAL BALANCE

			MAISUANE DACAG	CE		
INPUT HEAT	B.T.U.S (000/Hk.)	PERCENT	INPUTS	POUNDS	PERCENT	
POTENTIAL HEAT			METAL CHARGE	2030.	45 • 69	
OF FUEL	23751.	92.57	PIG IKON	256.	5.77	Spark Arrestor
OF FOEL	2373.4	,200,	KETUKNS	1506.	33.90	
SENSIBLE HEAT			STEEL SCHAP	U.	0.00	
OF THE BLAST	37.	0.15	INON SCHAP	283.	5.02	Basic
or the bensi	3.4		FERRMALLOYS	45•	1.00	; Lining
HEAT FROM ØXIDATION						
OF MN, FE, SI	1870.	7.29	COKF	247.	5.56	
OF THE SI	10704	*****	NATURAL GAS	0.	0.00	
TATAL INPUT HEAT	25658.	100.00	FUEL ØIL	0.	0.00	No Afterburner
INTAL THEOT REAL	23030.	100100				الله ا
CHILING LIEAT			FLUX AND ADDITIVES	102•	2.29	Charge Door Open
AUTPUT HEAT			AIK	2028.	45 • 66	
			ØXYGEN	0.	0.00	11 ! 17
HEATING AND MELTING	8323+	32.44	AN LOUIN	•	0.00	$\{1, i, i\}$
af Ikan	6323+	32 • 44	CUPULA LINING	36.	0.60	
HEAT CONTENT			COINER EINING	304	0000	
HEAT CONTENT	Euu	2.29	TOTAL INPUT MILS	4443.	100.00	
UE THE SLAG	588∙	2.27	TOTAL THEOT WILLS	44454	100100	1: 1 1
GALCINING ØF						Cold Blast Air No Fuel Injection
LIMESTONE	547.	2.13	MUTPUTS			No O2 Enrichment
FINESTONE	3410	2.13	<u></u>			111 111 111
DECOMPOSITION .			MOLTEN IKON	2000.	45.08	$A \bowtie A$
OF WATER	327•	1.28	THE CLASS THE CONTRACT OF THE	20001	45.01.	
AL MAIEN	32 14	1 • 2 0	SLAG	137.	3.09	Molten Iron
TOP GASES			BEAG		0.07	
-SENSIBLE HEAT	3468•	13.52	EMISSIONS DUST	19.	0.42	Slag
-LATENT HEAT	8656•	33 • 62	Elited Inter	• • • •	0112	
-PAIEM NEWS	0 O E. O ●	33 + 6E	TOP GASES	2256.	51 • 47	
HEAT RADIATION FROM			NITRØGEN	1549.	67.74	i i
THE CUPULA	3779.	14.73	CARBON DIOXIDE	470.	20.55	' '
THE COPPER	31170	14073	CARBON MONOXIDE	265.	11.57	
			HYDROGEN	2.	0.07	CUPOLA CLASSIFICATION - 30
VULUME OF TOP GAS (MCE.		SULFUR DIWXIDE	1.	0.06	OULOW CARROLLING TO
ACTUAL 547. STANDARD 217.	mc F J		SULPUN DINATUE	1 •	0.05 .	
STUADAND ST.						

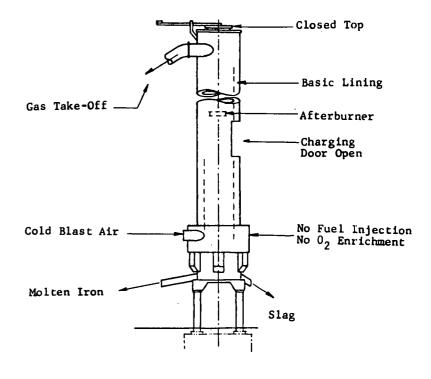


STANDARD

335.

MATERIAL BALANCE

INPUT HEAT	B.T.U.S	PERCENT	INPUTS		
	(0007HK.)	PERCENT		POUNDS	PERCENT
DOTELLTIAL LICAT			METAL CHARGE		45.30
POTENTIAL HEAT	39465.	09 70	PIG IKON	0.	0.00
OF FUEL	39463•	76 • 70	KETUKNS	753.	
SENSIBLE HEAT			STEEL SCHAP	151.	3 • 40
OF THE BLAST	56.	0.14	IKON SCHAP	1105.	24.91
AL IUE DEWO!	30.	0	FERHOALLOYS	0.	0.00
HEAT FROM UXIDATION			COKE	286.	6.44
ØF MN, FE, SI	462.	1.16	NATUKAL GAS	0.	0.00
	•		FUEL ØIL	0.	0.00
TOTAL INPUT HEAT	39964.	100.00	FUEL WIL	0.	0.00
1			FLUX AND ADDITIVES	58•	1.31
GUTPUT HEAT			AIR	2058•	46.40
			ØXYGEN	0.	0.00
HEATING AND MELTING			DATTE	•	
ØF IRON	12686.	31.73	CUPULA LINING	24.	0.55
HEAT CONTENT			TOTAL INPUT MTLS	4436.	100.00
OF THE SLAG	413.	1.03	INIME INFO MIES	44361	100.00
CALCINING OF					
LIMESTONE	474.	1.18	<u>autputs</u>		
			Marten Tran	2000•	45.09
DECOMPOSITION			MØLTEN IRØN	2000•	43.07
OF WATER	496•	1.24	SLAG	75.	1 • 69
TOP GASES				20	0.44
-SENSIBLE HEAT	11694.	29.25	EMISSIONS DUST	20.	0.46
-LATENT HEAT	16365.	40.93	Tall 04555	02.40	52.77
			TOP GASES	2340•	
HEAT RADIATION FROM			NITROGEN	1571.	67.14
THE CUPALA	-2144.	-5.36	CARBON DIOXIDE	430•	18.35
			CARBON MONOXIDE	336.	
			HYDRØGEN	2.	0.07
VOLUME OF TOP GAS O	MCF)		SULFUR DIØXIDE	2.	0.07
ACTUAL 1455.					

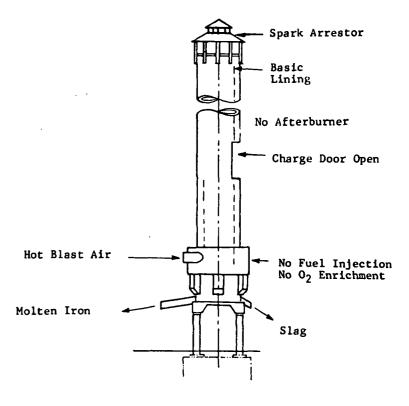




MATERIAL BALANCE

			INPUTS		
INPUT HEAT	B.T.U.S		1117 013	POUNDS	PERCENT
	(000/HK.)	PERCENT	METAL CHARGE	1963.	
UGTENTIAL HEAT			PIG IRUN	255.	4.46
POTENTIAL HEAT	*0000	22 44	KETUKNS		4.46
OF FUEL	79902.	77•46		1454.	
CENSTRE LEAT			IKUN SCKAP		0.00
SENSIBLE HEAT OF THE BLAST	01/11	00 05	FERROALLOYS	0.	
AL THE BEAST	21611.	20.93		•	0.00
HEAT FROM OXIDATION			COKE	235.	4.12
	1634.	1 60	NATURAL GAS	0.	0.00
OL MAY LED 21	10341	1 • 30	FUEL ØIL	Ü•	0.00
TOTAL INPUT HEAT	103147-	100.00		-	
ININC INIOI NEMI	1031474	100.00	FLUX AND ADDITIVES	90•	1.58
			A *14	2.41.4	5 to 36
OUTPUT HEAT			AIR	3414.	
			ØXYGEN	0.	0.00
HEATING AND MELTING			CUDAL A LINENC	8.	
OF -IRON	40009•	3b•79	CUPBLA LINING	n•	0.14
			THIAL INPUT MILS	5711•	100.00
HEAT CONTENT			TOTAL TWO THILS	3/11•	100.00
ØF THE SLAG	617.	0.60			
641 6 THING 6 5			ØUTPUTS		
CALCINING ØF	00.43	0.06			
LIMESTONE	2347.	2.28	MØLTEN IRØN	2000•	35.02
DECOMPOSITION			1102127 177014	2000	00.00
DECOMPOSITION	1.433	1.43	SLAG	64.	1.12
ØF WATER	1473.	1 • 43		• • •	
TOP GASES			EMISSIONS DUST	25.	0.44
-SENSIBLE HEAT	25238.	24.47			
	-93602.		TOP GASES	3622•	63 • 42
			NITHØGEN	2615.	72.20
HEAT RADIATION FROM			CARBON DIOXIDE	1625.	44.86
· · · ·	127064.	123.19	CARBON MONOXIDE	-615.	-16.99
000 020		,	HYDRØGEN	2.	0.04
			SULFUR DIØXIDE	-4.	-0.12
VOLUME OF TOP GAS (MCF)				

VØLUME ØF TØP GAS (MCF) ACTUAL 3651. STANDARD 1449.





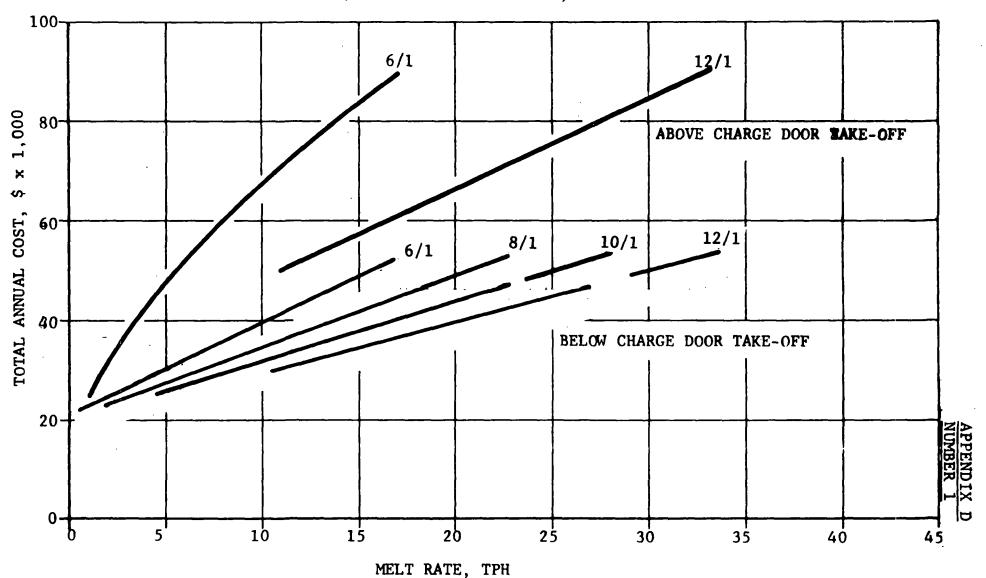
APPENDIX D

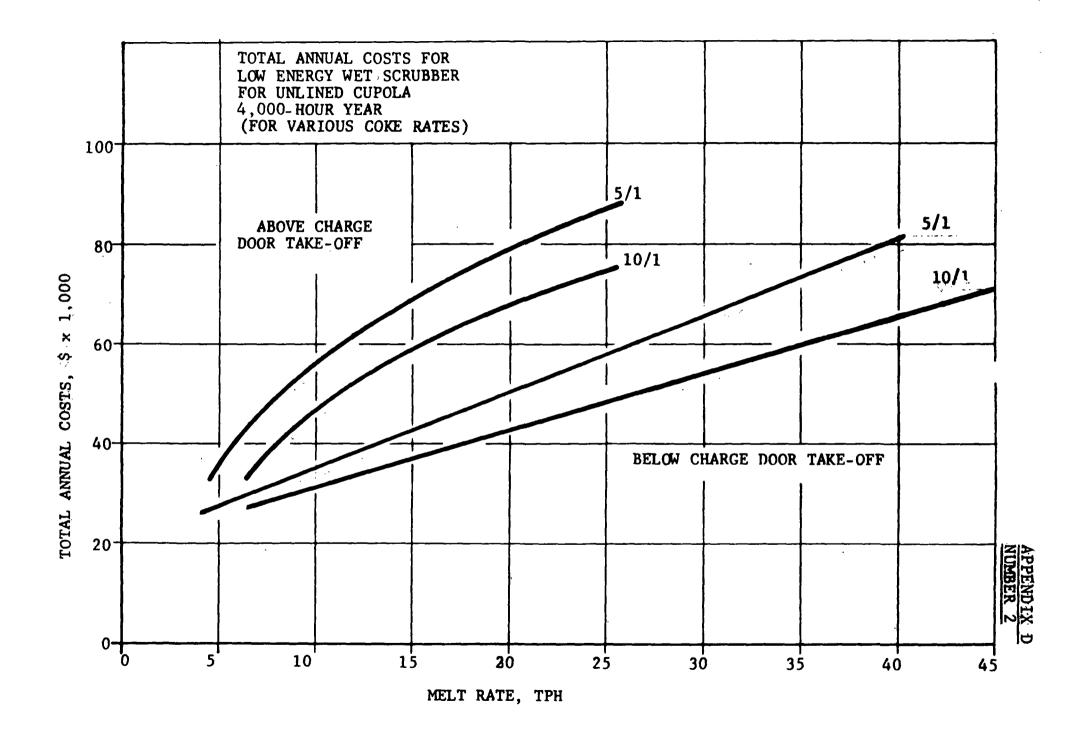
DETAIL ECONOMIC COST CURVES

This appendix contains the detail curves used to determine the cost of pollution control equipment. Summaries of these curves appear as exhibits in Section VIII of the report.

Following the economic cost curves are the detail data for determining the melt shop costs for the model foundries.

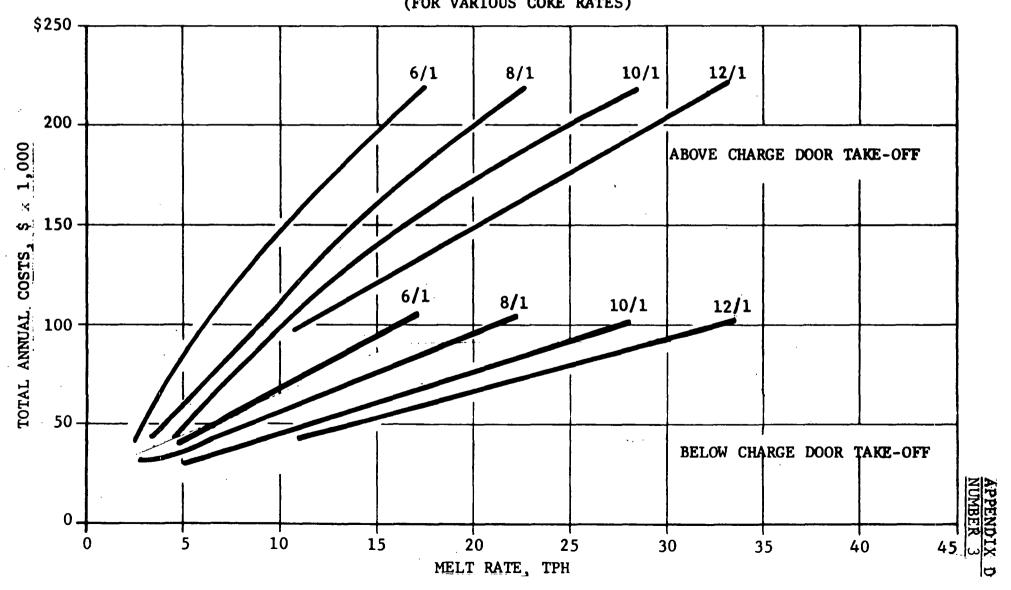
TOTAL ANNUAL COSTS FOR LOW ENERGY WET SCRUBBER FOR LINED CUPOLA 4,000-HOUR YEAR (FOR VARIOUS COKE RATES)

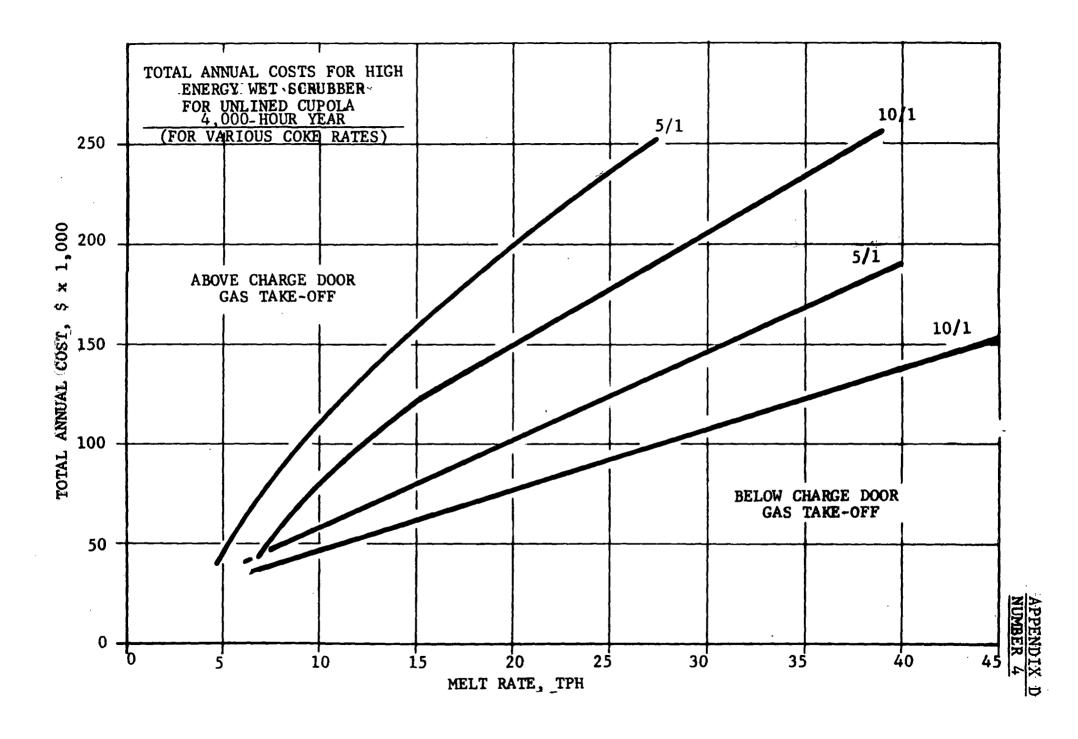


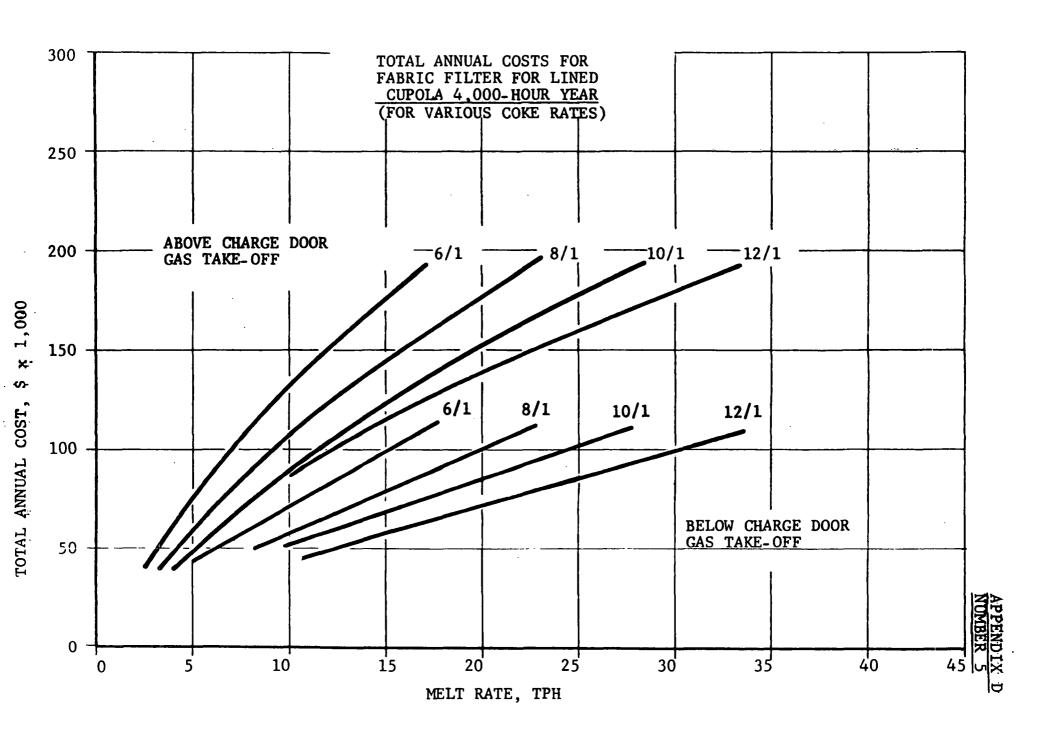


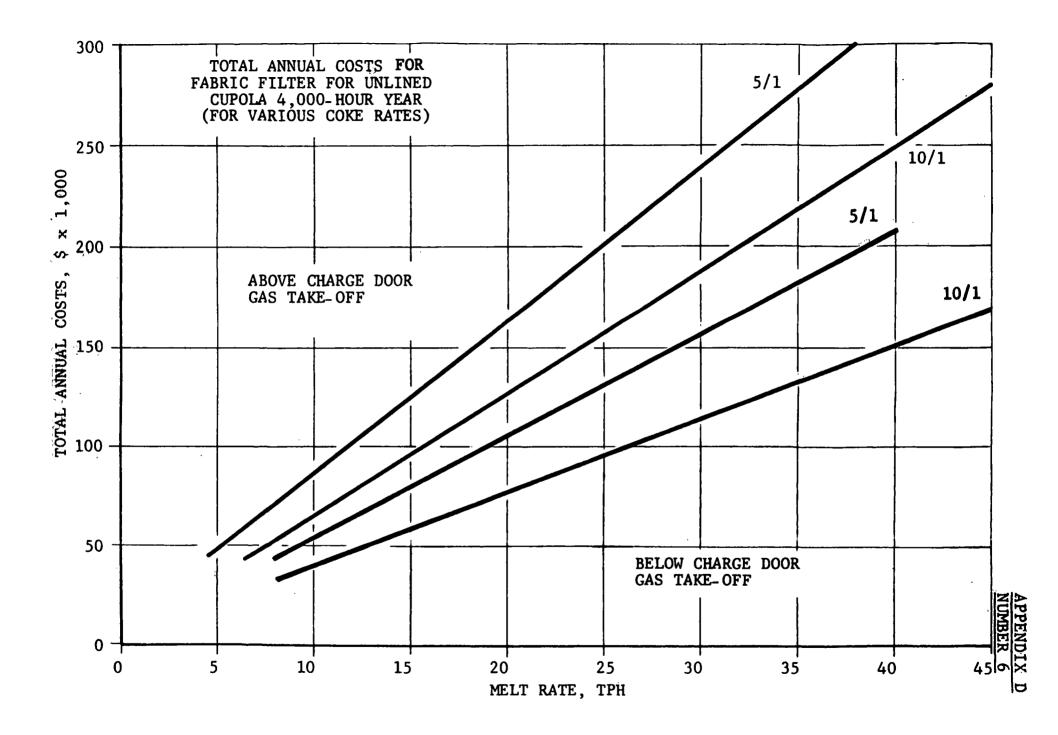
TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER FOR LINED CUPOLA

4,000-HOUR YEAR
(FOR VARIOUS COKE RATES)

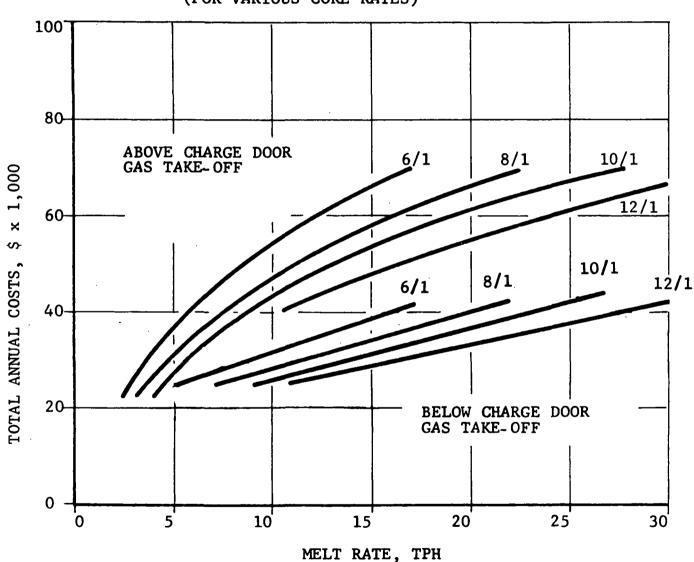




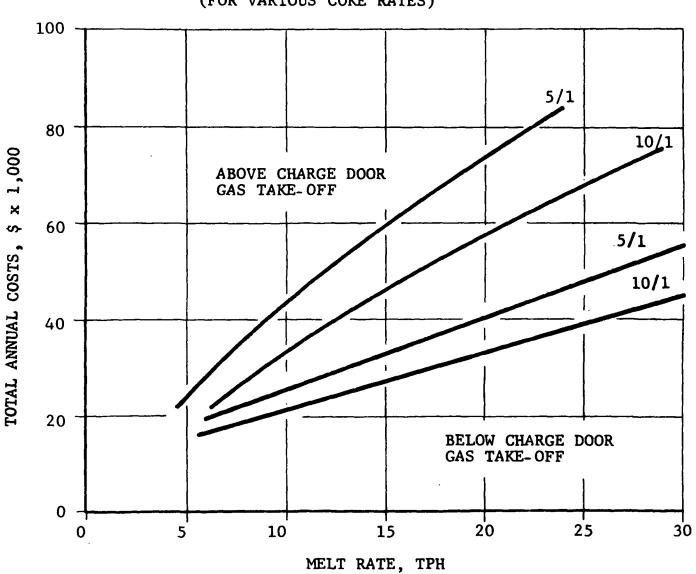




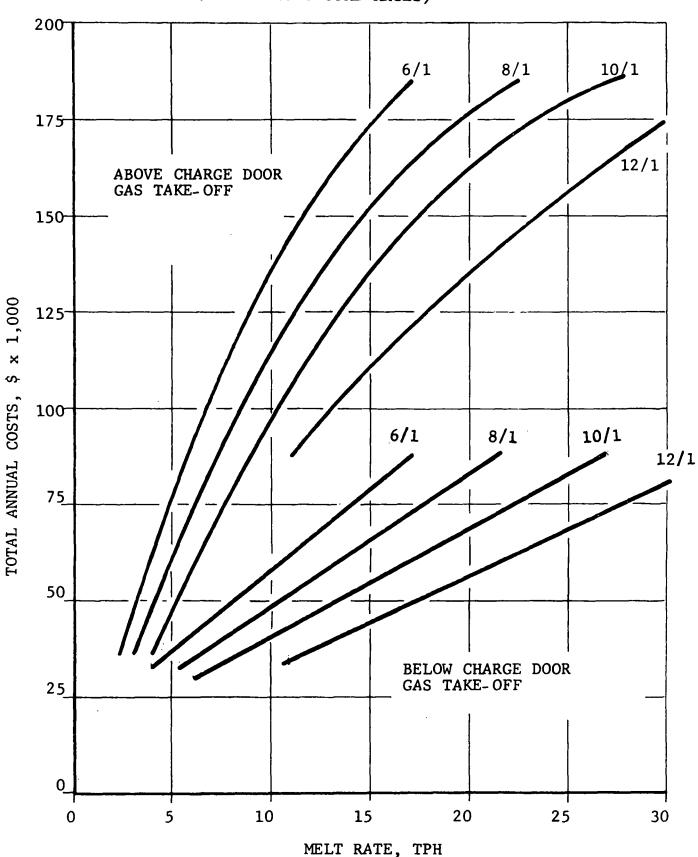
TOTAL ANNUAL COSTS FOR LOW ENERGY WET SCRUBBER ON LINED CUPOLA 2,000-HOUR YEAR (FOR VARIOUS COKE RATES)



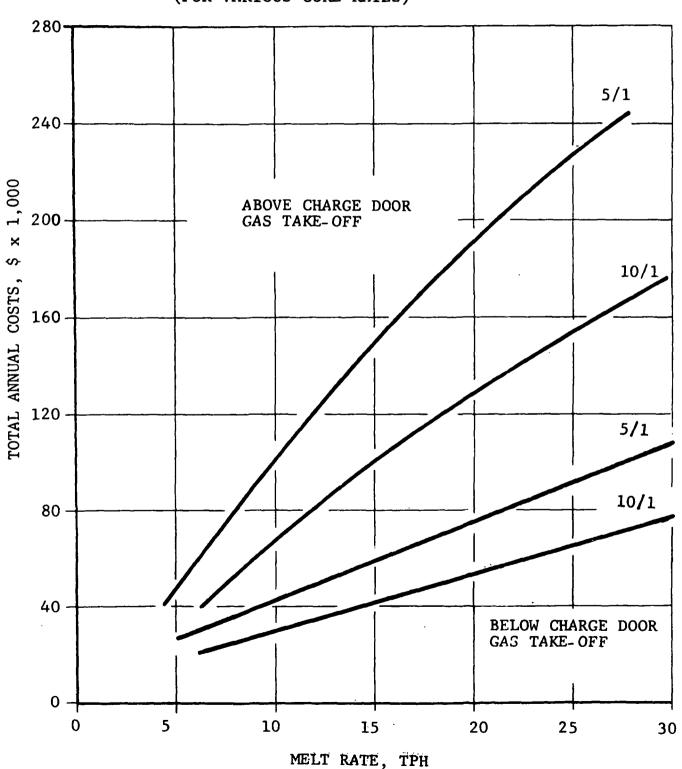
TOTAL ANNUAL COSTS FOR LOW ENERGY WET SCRUBBER ON UNLINED CUPOLA 2,000-HOUR YEAR (FOR VARIOUS COKE RATES)

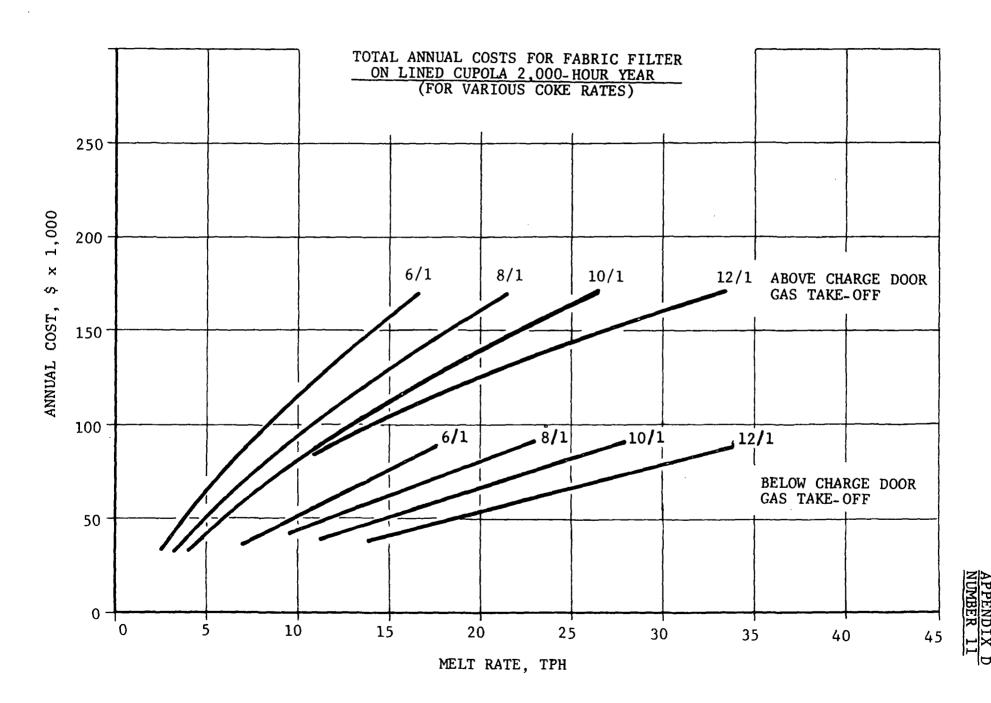


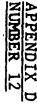
TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON LINED CUPOLA 2,000-HOUR YEAR (FOR VARIOUS COKE RATES)



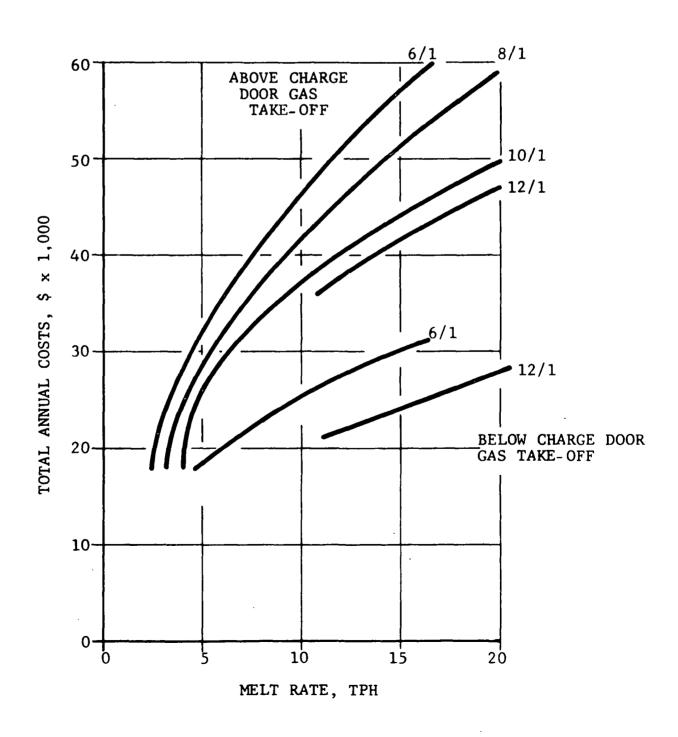
TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON UNLINED CUPOLA 2,000-HOUR YEAR (FOR VARIOUS COKE RATES)



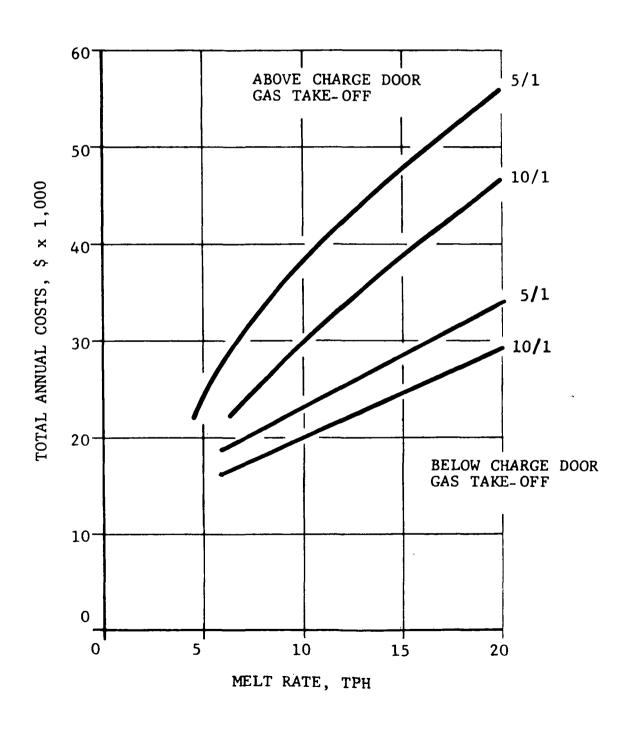




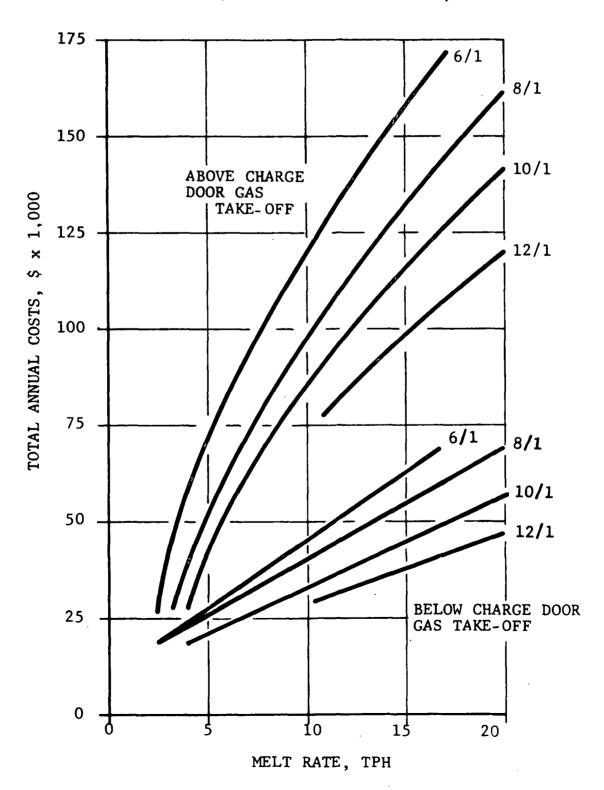
TOTAL ANNUAL COSTS FOR LOW ENERGY WET SCRUBBER ON LINED CUPOLA 1,000-HOUR YEAR (FOR VARIOUS COKE RATES)



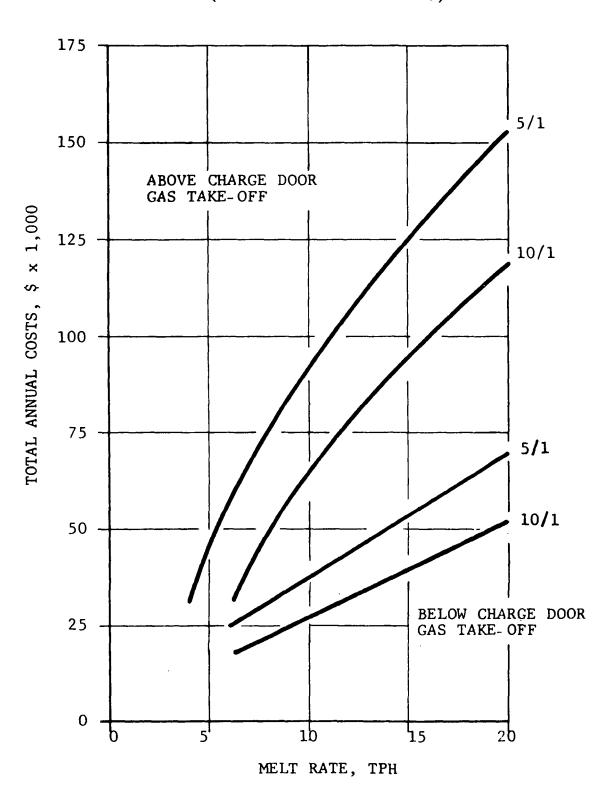
TOTAL ANNUAL COSTS FOR LOW ENERGY WET SCRUBBER ON UNLINED CUPOLA 1,000-HOUR YEAR (FOR VARIOUS COKE RATES)



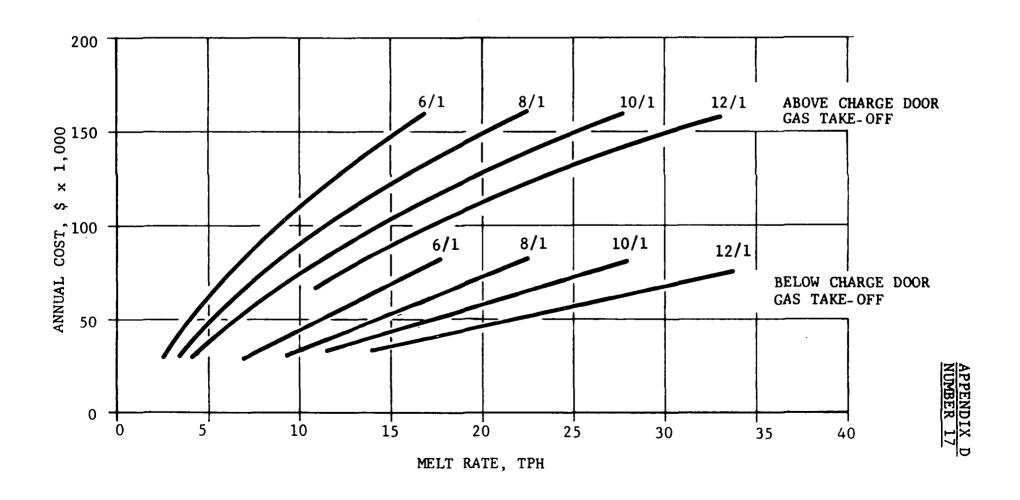
TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON LINED CUPOLA 1,000-HOUR YEAR (FOR VARIOUS COKE RATES)

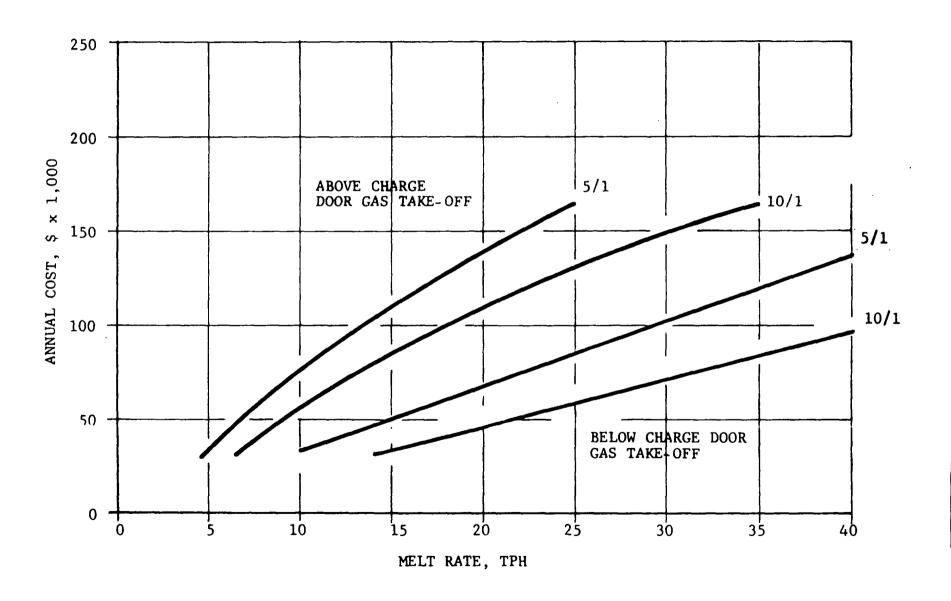


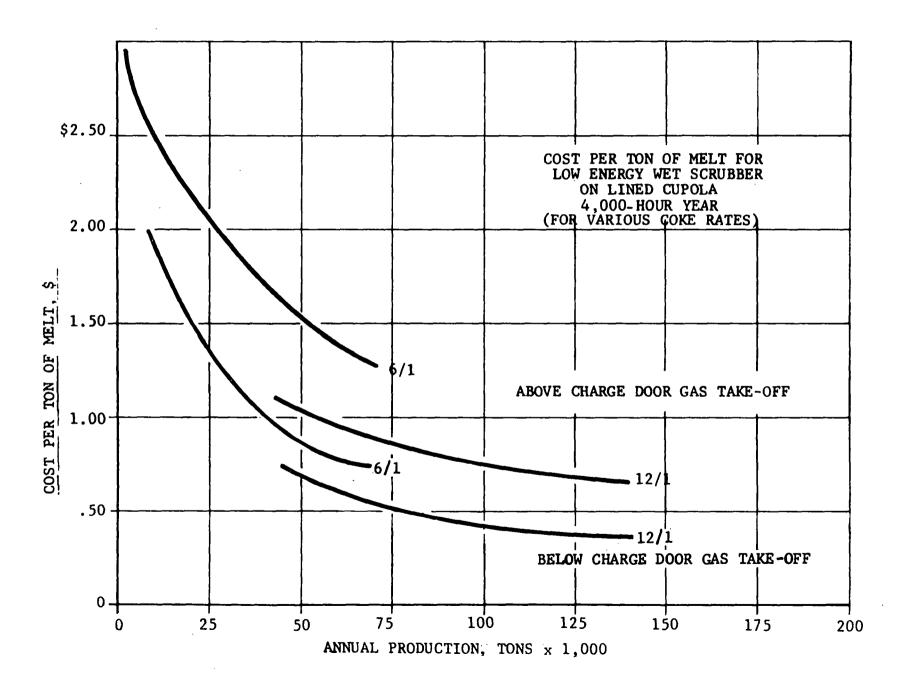
TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON UNLINED CUPOLA 1,000-HOUR YEAR (FOR VARIOUS COKE RATES)



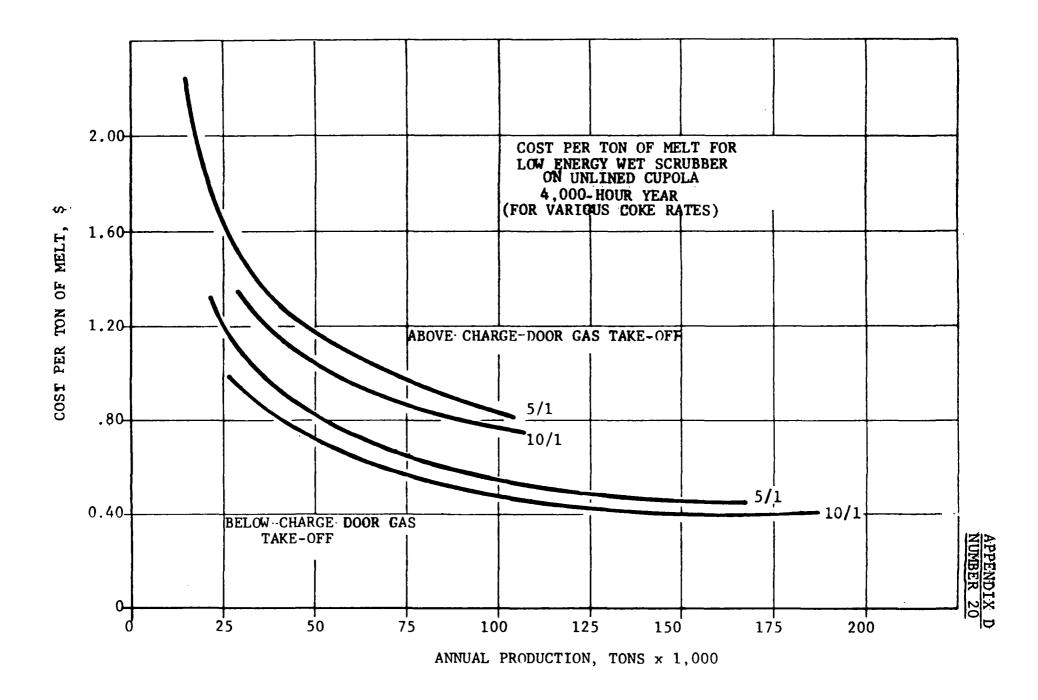
TOTAL ANNUAL COSTS FOR FABRIC FILTER ON LINED CUPOLA 1,000-HOUR YEAR (FOR VARIOUS COKE RATES)

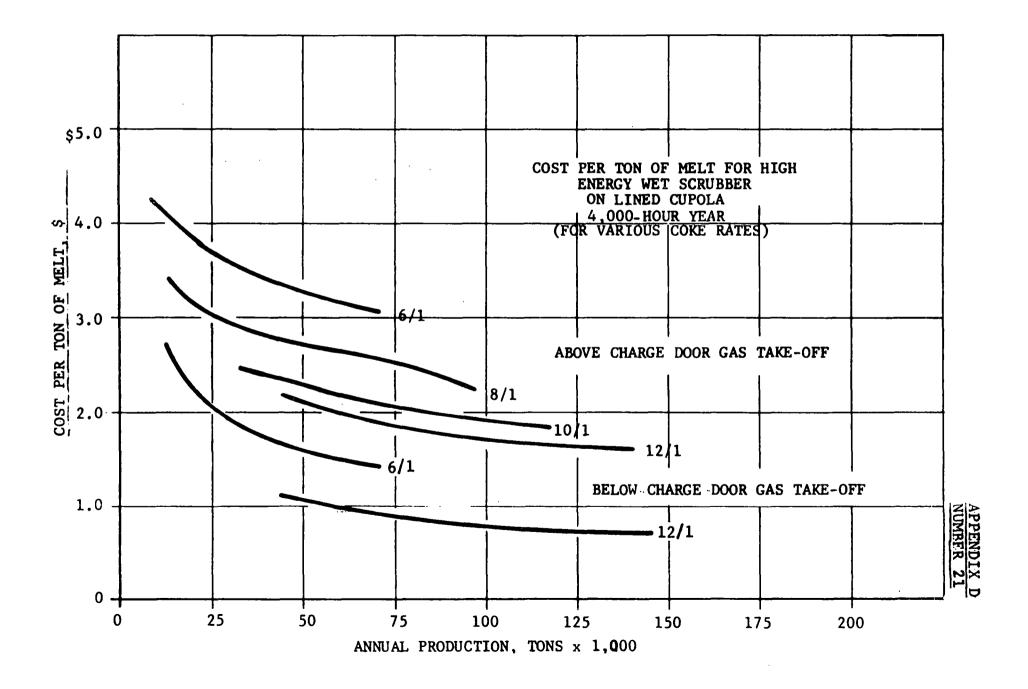






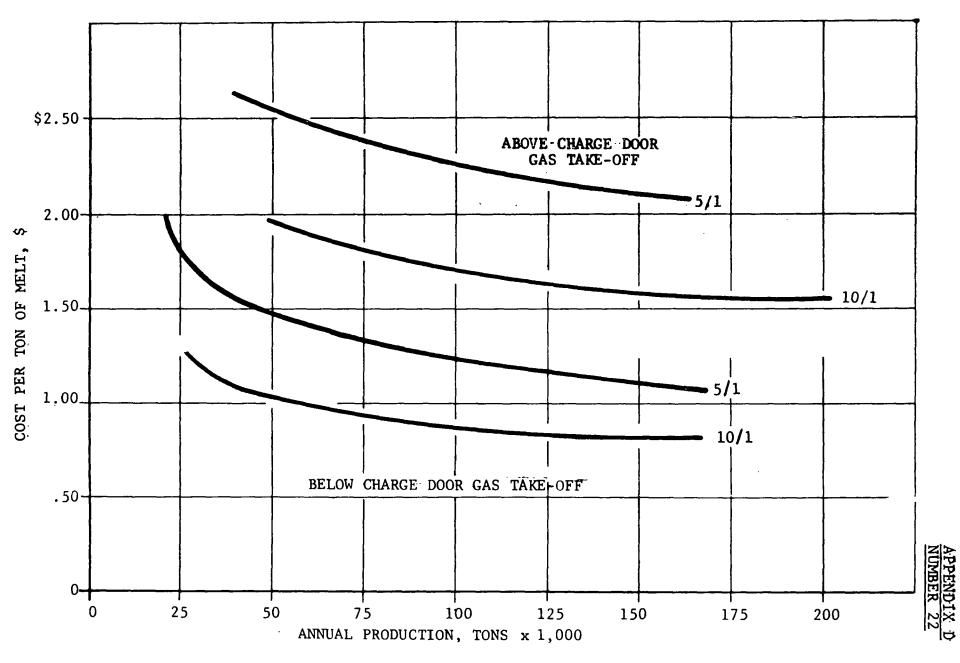
UMBER 19



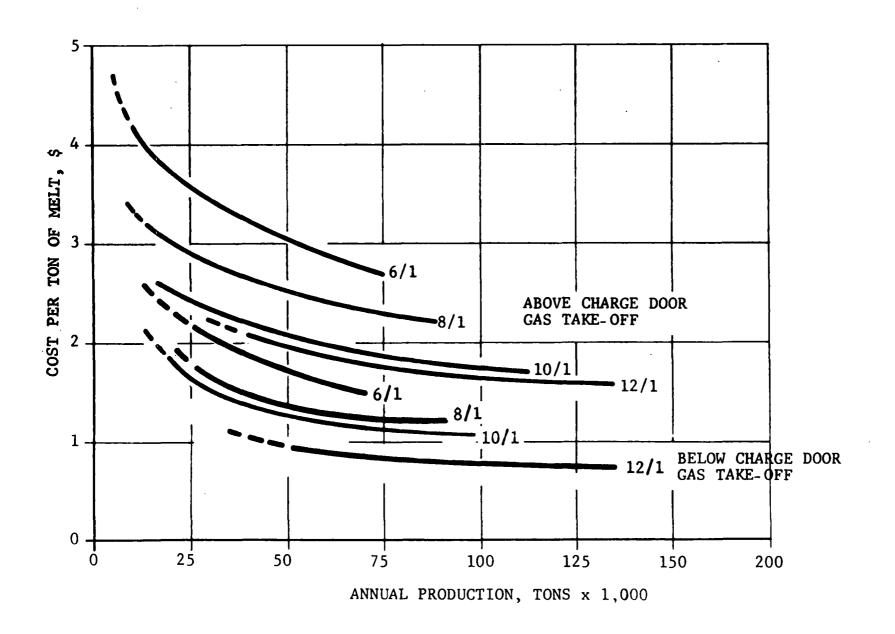


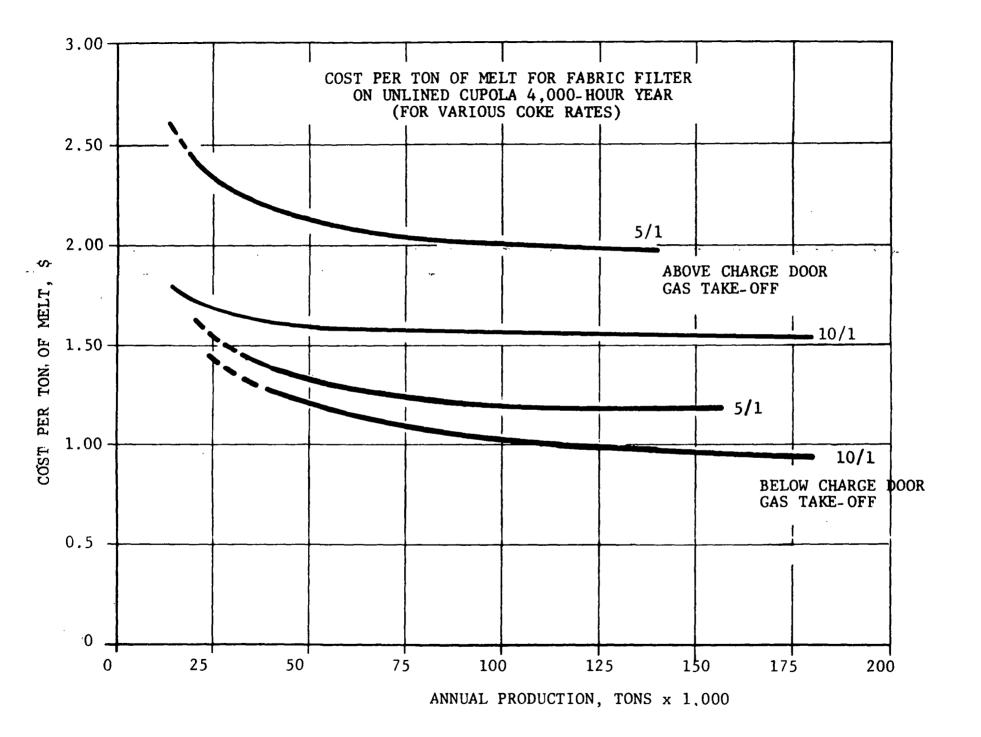
COST PER TON OF MELT FOR HIGH ENERGY WET SCRUBBER ON UNLINED CUPOLA

4,000-HOUR YEAR (FOR VARIOUS COKE RATES)

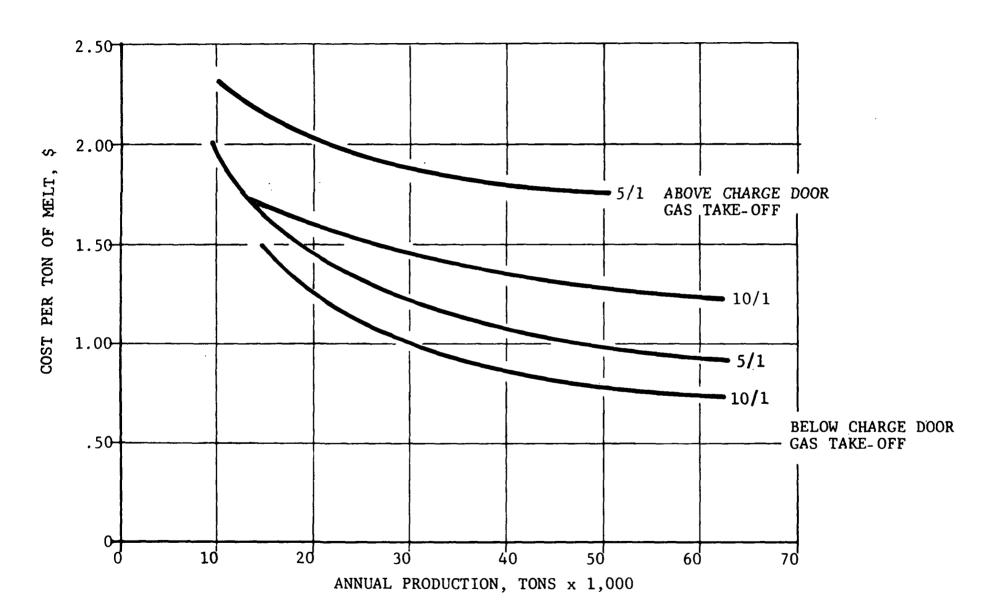


COST PER TON OF MELT FOR FABRIC FILTER ON LINED CUPOLA 4,000-HOUR YEAR (FOR VARIOUS COKE RATES)





APPENDIX D NUMBER 25



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COST PER TON OF MELT FOR

ANNUAL PRODUCTION, TONS x 1,000

40

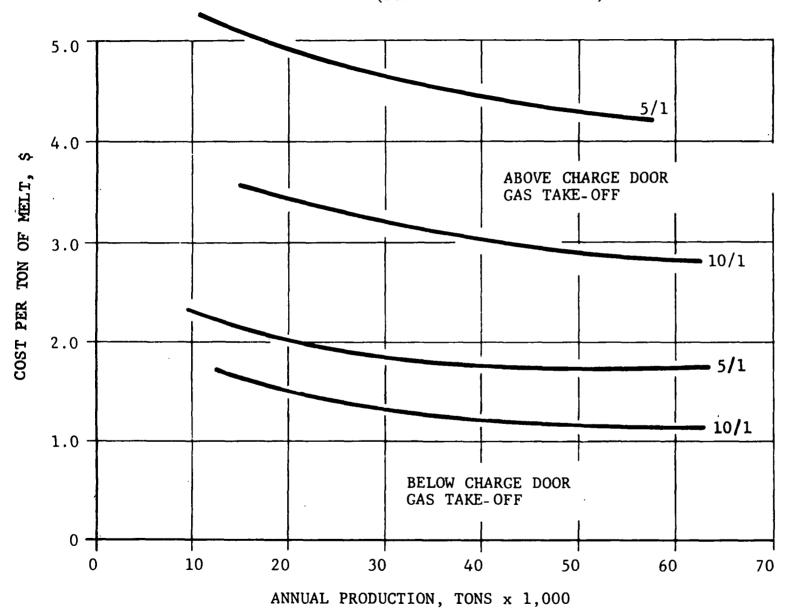
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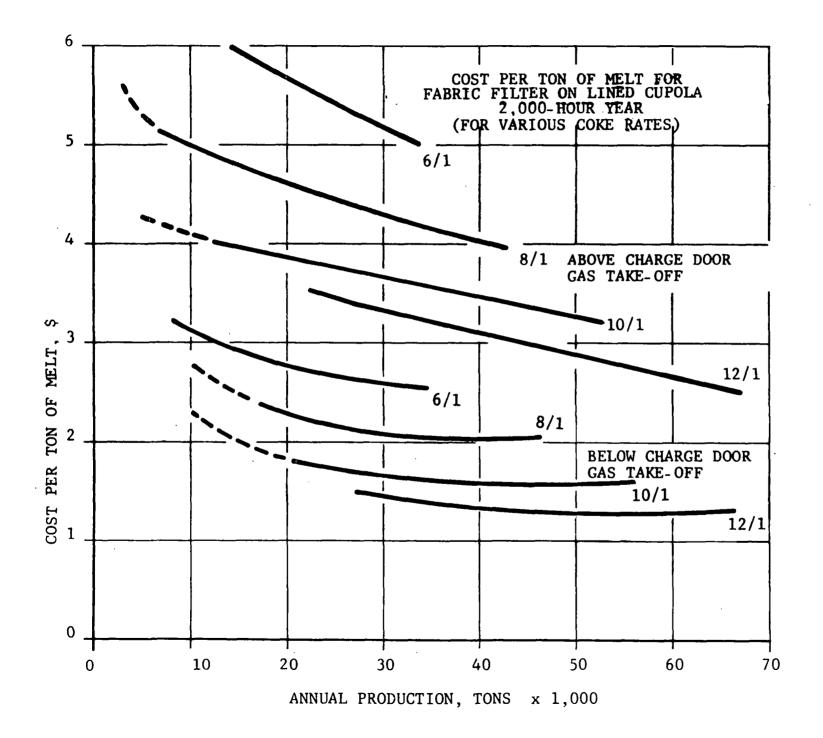
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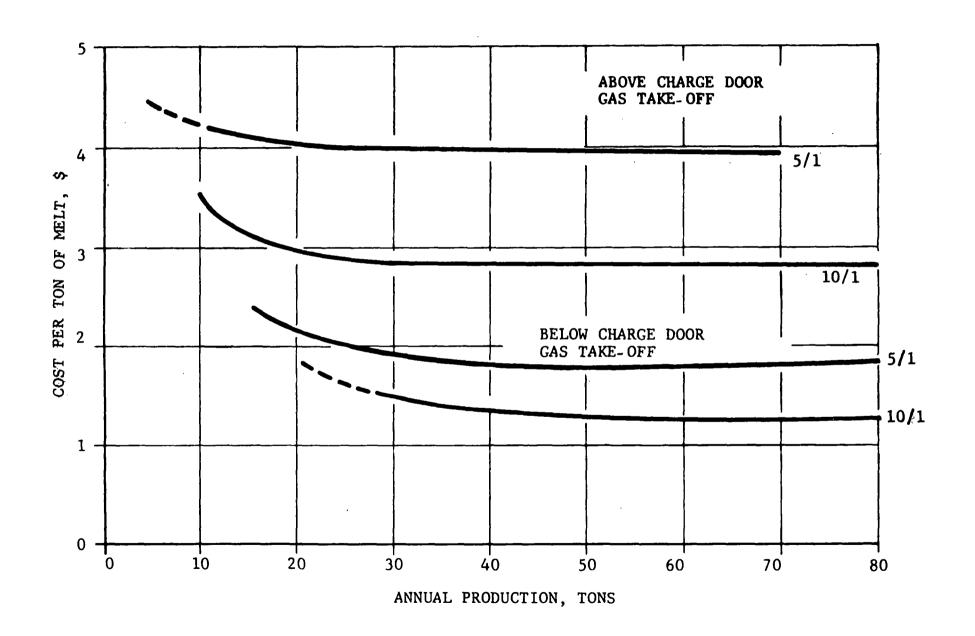
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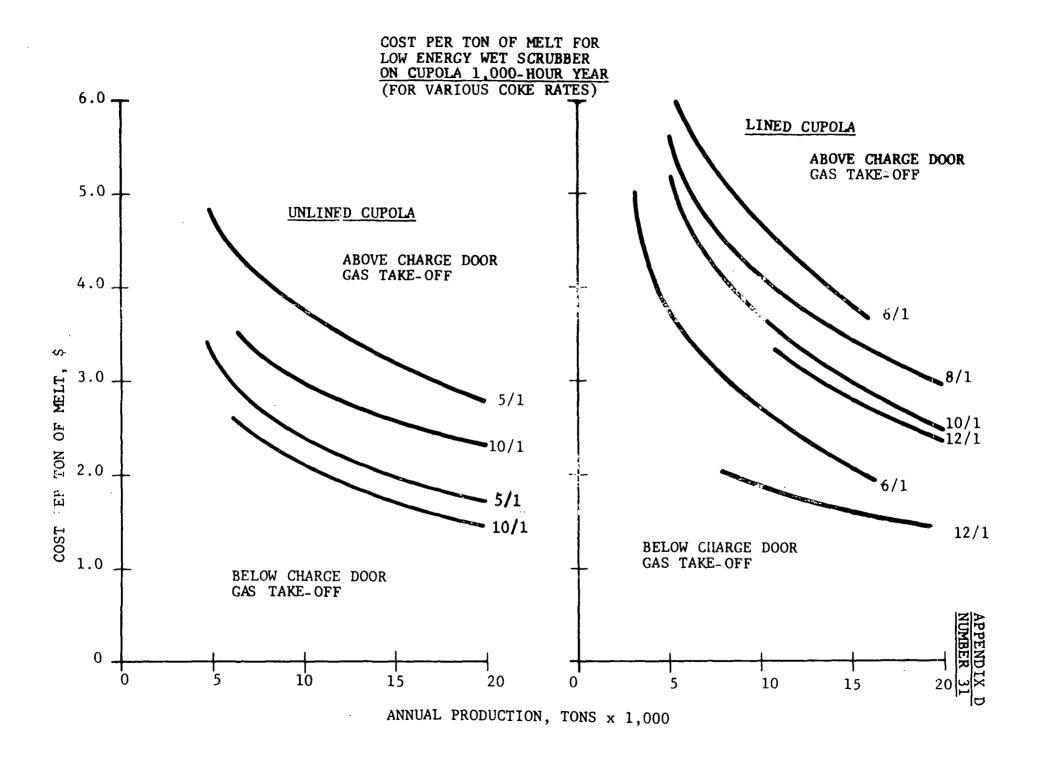
COST PER TON OF MELT FOR HIGH ENERGY WET SCRUBBER ON UNLINED CUPOLA 2,000-HOUR YEAR

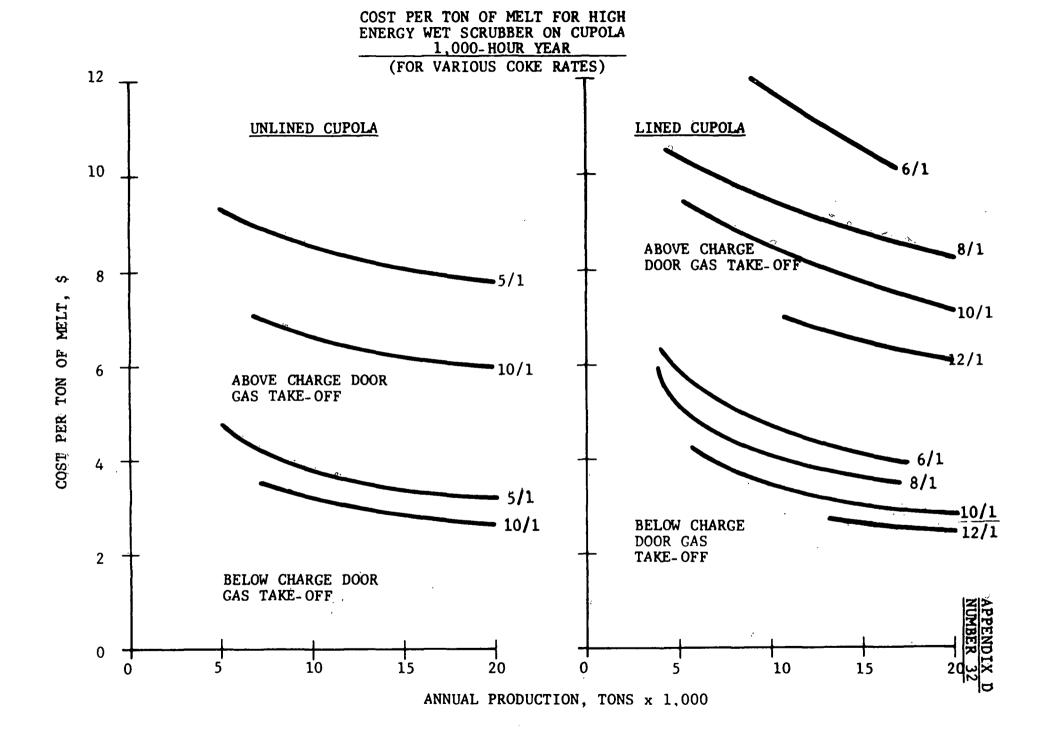
(FOR VARIOUS COKE RATES)





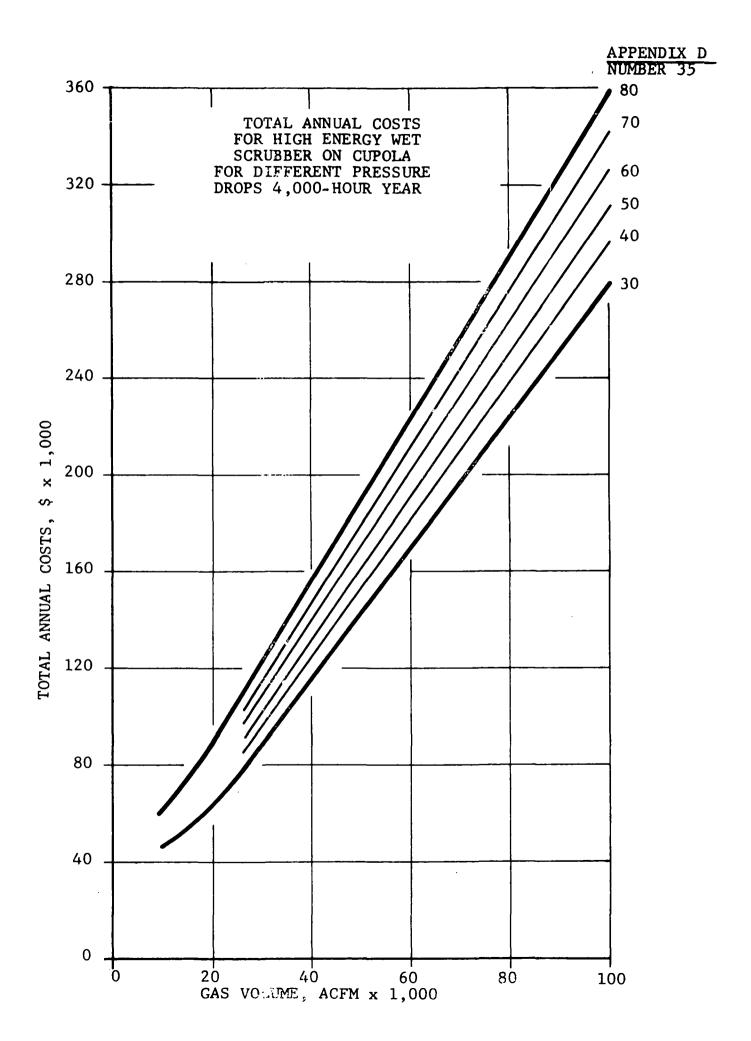




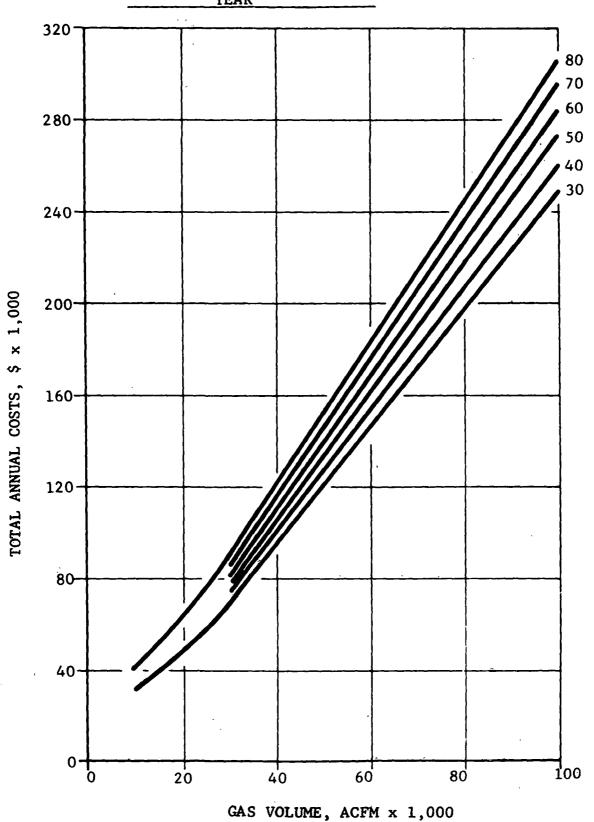


NUMBER 33

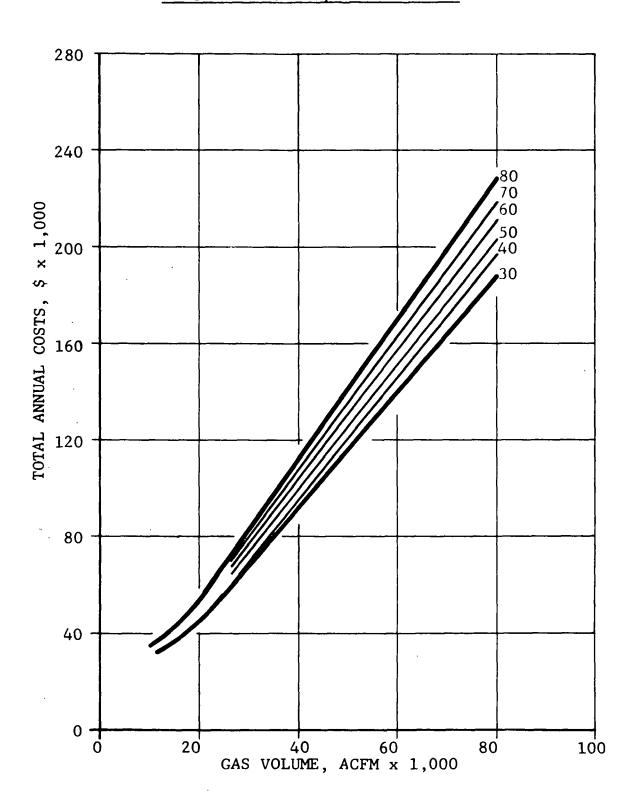




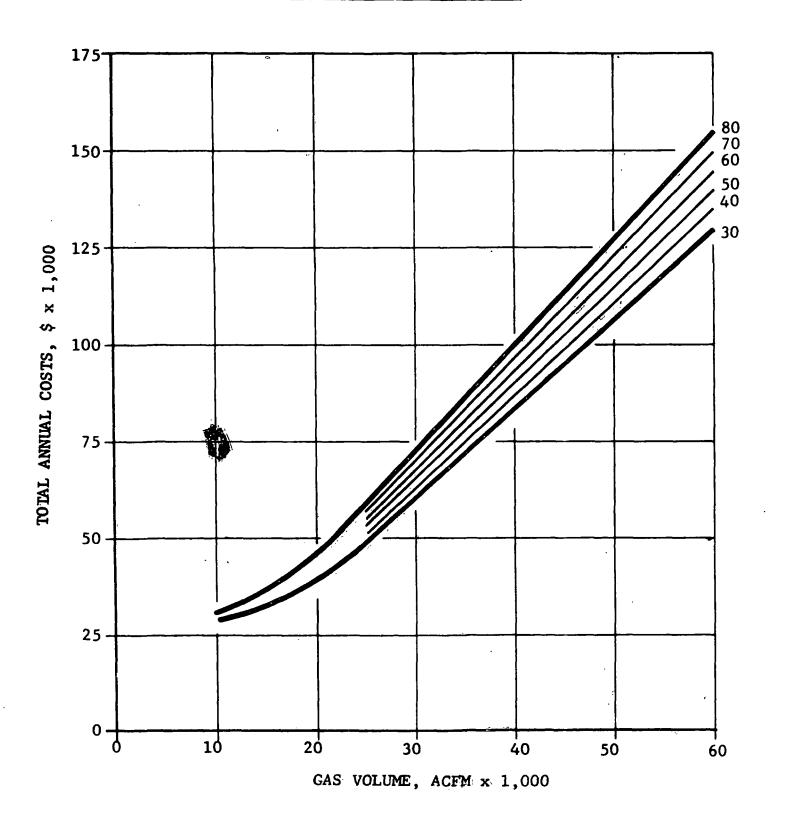
TOTAL AMNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON CUPOLA FOR DIFFERENT PRESSURE DROPS 2,000-HOUR YEAR



TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON CUPOLA FOR DIFFERENT PRESSURE DROPS 1,000-HOUR YEAR



TOTAL ANNUAL COSTS FOR HIGH ENERGY WET SCRUBBER ON CUPOLA FOR DIFFERENT PRESSURE DROPS 500-HOUR YEAR



		FURNA F
		N an
Α.	Buildings	
	 Cupola Building 500 & 1,000 Hours per Year, \$25/Sq. Ft. Cupola Auxiliary Building \$10/Sq. Ft. 	2,500 750
В.	Scrapyard Equipment	
	 Scrapyard Crane Crane Runway Metal Trim Platform & Weigh Hopper Coke & Stone Weigh Hoppers & Feeders Coke & Stone Bins Coke & Stone Unloading Equipment Magnet Platform Scale 	1-5T 12,500
c.	Charging Equipment	
٠.	1. Skip Charger	
	2. Charge Buckets	
D.	Melting Equipment	
	 Cupolas @ 500 & 1,000 H/A Blower & Motor (SCFM Basis) Air Piping, Piping, Valves & Weight Control Forehearth Runners Refractories for Cupola & Runners Slag Buckets Slag Disposal Equipment Piping, Header, & Sewer Connections 	1-56" 1-3,10
E.	Holding Equipment (Not Required)	
	Subtotal $\#1$ (B + C + D + E)	
F.	Spares & Freight (3½% of B, C, D, E)	
	Subtotal $#2$ (S.T. $#1 + A + F$)	
G.	Engineering 5%	
	Subtotal	
Н.	Contingencies 10%	
	Total	

A

Cost/Ton 500 Hours 1,000 Hours

CUPOLA, COLD BLAST, NO HOLDING FURNACE, FABRIC FILTER EMISSION CONTROL FOR 500 & 1,000 HOURS PER YEAR

	5 ГРН		15_TP	Н	
	Number and Size	Cost	Number <u>and Size</u>	Cost	
per Year, \$25/Sq. Ft. Ft.	2,500 Sq. Ft. 750 Sq. Ft.	\$ 63,000 8,000	4,000 Sq. Ft. 1,800 Sq. Ft.	\$100,000 18,000	8,0 ¹ 3,0 ¹
r ers	1-5T 12,500 Open	60,000 19,000 20,000	1-7½T 16,000 Covered	83,000 128,000 55,000 95,000	2-7 24,
	1-45"	15,000 10,000 6,000 10,000	1-65"	40,000 8,000	2-6
	-	24,000	1-15 ТРН	134,000	2-1
ht Control	1-56" 1-3,100	14,000 26,000 25,000 5,000 3,000 10,000 5,000	1-90" 1-9,200	34,000 36,000 40,000 6,000 4,000 29,000	2-9 ¹ 2-9
ns		10,000		20,000	
		\$ <u>262,000</u>		\$733,000	
	·	\$ <u>9,000</u>		\$ 25,000	
')		\$342,000		\$876,000	
		\$ 17,000		\$ <u>44,000</u>	
		\$ <u>359,000</u>		\$ <u>920,000</u>	
		\$ 36,000		\$ <u>92,000</u>	
		\$ <u>395,000</u>	\$	1,012,000	
		\$158 79		\$67.4	7

MENT REQUIREMENTS

OLD BLAST, NO HOLDING IC FILTER EMISSION CONIROL 1,000 HOURS PER YEAR

'PH	15 ТРН		30 ТРН		50 TPH	
Cost	Number and Size	Cost	Number and Size	Cost	Number and Size	Cost
\$ 63,000 8,000	4,000 Sq. Ft. 1,800 Sq. Ft.	\$100,000 18,000	8,000 Sq. Ft. 3,000 Sq. Ft.	\$200,000	10,000 Sq. Ft. 5,000 Sq. Ft.	\$250,000 50,000
60,000 19,000 20,000	1-7½T 16,000 Covered	83,000 128,000 55,000 95,000	2-7½T 24,000 Covered	166,000 192,000 70,000 143,000	2-10T 30,000 Covered	200,000 240,000 90,000 240,000
15,000 10,000 6,000 10,000	1-65"	40,000 8,000	2-65"	52,000 16,000	3-65"	75,000 24,000
24,000	1-15 TPH	134,000	2-15 TPH	238,000	2-30 ТРН	330,000
14,000 26,000 25,000 5,000 3,000 10,000	1-90" 1-9,200	34,000 36,000 40,000 6,000 4,000 29,000	2-90" 2-9,200	67,000 72,000 50,000 12,000 6,000 58,000	2-108" 2-12,500	80.000 175,000 72.000 15,000 8,000 70,000
5,000 - 10,000		21,000 20,000		27,000 25,000		36,000 30,000
\$ <u>262,000</u>		\$ <u>733,000</u>	\$	<u>1,194,000</u>	\$	1,685,000
\$ 9,000		\$ <u>25,000</u>	Ş	42,000	\$	59,000
\$ <u>342,000</u>		\$ <u>876,000</u>	\$	\$ <u>1,466,000</u>	\$	2,044,000
\$ <u>17,000</u>		\$ <u>44,000</u>	\$	\$ 73,000	Ş	102,000
\$359,000		\$ <u>920,000</u>		\$ <u>1,539,000</u>	Ş	2,146,000
\$ <u>36,000</u>		\$ <u>92,000</u>		\$ <u>154,000</u>	\$	215,000
\$ <u>395,000</u>	\$	1,012,000	\$	\$ <u>1,693,000</u>	\$	2,361,000
\$158 79		\$67.4	7			

CUPOLA FURN.

FOR 2,0

EQ

Increase For 2,000 & 4,000/ Number lours per Year and Size

A. Buildings

80% Cupola Building

Charging Equipment

Swivel Skip Charger 15%

D. Melting Equipment

1.	Cupolas	100%
3.	Air Piping & Valves	10
6.	Runners	50
7.	Refractories	100
9.	Slag Disposal Equipment	15
5.	Forehearth	100

Subtotal #1 (C + D)

E. Spares & Freight 3-1/2%

Subtotal (#1 + A + E)

Engineering 5%

Subtotal

Contingencies 10%

Subtotal

Total from 500 & 1,000 Hours Page 1.

Grand Total

Cost/Ton

2,000 Hours 4,000 Hours

CUPOLA, COID BLAST, NO HOLDING FURNACE, SUPPLEMENTARY COSTS FOR 2,000 & 4,000 HOURS PER YEAR

Increase For	5.TP1	15 TPH		
2,000 & 4,000/ <u>Hours per Year</u>	Number and Size Cost	Number and Size	Cost	Numbe and Si
80%	\$ 50,000	\$	80,000	
15%	3,000		20,000	
100%	14,000	•	34,000	
10 50	3,000 2,000		4,000 2,000	
100 15	10,000 1,000		2,000 29,000 3,000	
100	5,000		6,000	
	\$ 38,000	\$	98,000	
	\$ <u>1,000</u>	\$	<u>3,000</u>	
	\$ 89,000	\$	181,000	
	\$ 4,000	\$	9,000	
	\$ <u>93,000</u>	\$	190,000	
	\$ <u>9,000</u>	\$	<u>19,000</u>	
	\$ <u>102,000</u>	\$	209,000	
· • 1.	\$ <u>395,000</u>	\$ <u>1</u>	,012,000	
,	\$ <u>497,000</u>	\$ <u>1</u>	,221,000	
	\$49.	70	\$40.70 20.35	

T REQUIREMENTS

) BLAST, NO HOLDING UPPLEMENTARY COSTS ,000 HOURS PER YEAR

	15 TPH	30 ТРН		50 TPI	Ī
Cost	Number and Size Cost	Number	Cost	Number and Size	Cost
\$.50,000	\$ 80,00	00 \$	160,000	\$	200,000
3,000	20,00	00	36,000		50,000
14,000 3,000 2,000 10,000 1,000 5,000	34,00 4,00 2,00 29,00 3,00 6,00	00 00 00 00	67,000 5,000 3,000 58,000 4,000 12,000		80,000 7,000 4,000 70,000 5,000 15,000
\$ <u>38,000</u>	\$ <u>98,0</u> 0	<u>00</u> \$	185,000	\$	231,000
\$ <u>1,000</u>	\$ 3,00	00 \$	6,000	\$	9,000
\$ 89,000	\$ <u>181,00</u>	900	351,000	\$	440,000
\$ 4,000	\$ 9,00	00 \$	18,000	\$	22,000
\$ <u>93,000</u>	\$ <u>190,00</u>	900	369,000	\$	462,000
\$ 9,000	\$ 19,00	00 \$	37,000	\$	46,000
\$ <u>102,000</u>	\$ <u>209,00</u>	00 \$	406,000	\$	508,000
\$ <u>395,000</u>	\$ <u>1,012,0</u>	<u>00</u> \$ <u>1</u>	,693,000	\$ <u>2</u>	361,000
\$ <u>497,000</u>	\$ <u>1,221,00</u>	<u>00</u> \$ <u>2</u>	,099,000	\$2	,869,000
\$49.70		40.70 20.35	\$34.98 17.49		\$28.69 14.35

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2,000

A. Buildings

l.	Melting Building	@ \$25/Sq. Ft.	
2.	Cupola Auxiliary	Building @ \$10/Sq.	Ft.

B. Scrapyard Equipment

1.	Scrapyard Crane	1-5T
1A.	Crane Runway & Stockyard	12,50
2.	Metal Trim Platform & Weigh Hopper	Lump
4.	Coke & Stone Weigh Hoppers, Feeders, Scales & Bins	•
5.	Coke & Stone Unloading Equipment	Lump
6.	Magnets	1-45'
7.	Platform Scale	Lump

C. Charging Equipment

- Skip Charger-Swivel Type
 Charge Buckets

D. Melting Equipment

2. Air 3. Blo 4. For 5. Run 6. Ref 7. Gas 8. Hot 9. Pip 10. Sla	ola-Unlined-Water Cooled Piping, Gates, Charge Indicator wer & Motor ehearth-Lined ners ractories for Runners Piping & Meter Blast Heater ing, Headers, Sewer Connections g Disposal Equipment er Cooling Tower & Pumps	1-36' Lump 2,300 1 Lump Lump Lump Lump Lump Lump Lump
---	---	---

E. Holding Equipment

- Channel Induction Furnace & Controls
 Spare Furnace Body
 Refractories for Holding Furnaces

Subtotal #1 (B + C + D + E)

F. Spares & Freight $(3\frac{1}{2}\% \text{ of } B + C + D + E)$

Subtotal #2 (St. #1 + A + F)

G. Engineering (5% of St. #2)

Subtotal #3 (G + St. #2)

Contingencies (10% of St. #3)

Total (St. #3 + H)

Cost/Ton 2,000 Hours 4,000 Hours

CUPOL HOT BLAST, WATER-COOLED, CHANNEJINDUCTION HOLDING FURNACE, WET:CRUBBER EMISSION CONTROL

	5 T	PII	15	ГРН	
	Numbe		Number		N
	and Sze	Cost	and Size	Cost	an
:. 310/Sq. Ft.	2,000 1,000	\$ 50,000 10,000	1-2,500 Sq. Ft.		1-4,0 1-3,0
Hopper Feeders, Scales & Bins ment	1-5T 12,500 Oen Lump Sum Lump Sum 1-45" Lump Sum	60,000 19,000 20,000 - 10,000 6,000 10,000	1-7½ T-70' 16,000 Sq. Ft Lump Sum Lump Sum Lump Sum Lump Sum 1-65"	83,000 128,000 55,000 95,000 40,000 8,000	2-7½ 25,0C Lump Lump 2 -65
		24,000	1-15 TPH	134,000	1- 30
licator	1-36" Lump Sum 2,300 SCM 1 Lump Sum Lump Sum	40,000 50,000 25,000 5,000 3,000 1,500	1-66" Lump Sum 7,700 SCFM 1 Lump Sum Lump Sum	80,000 66,000 30,000 6,000 3,000 2,000	1-90' Lump 14,3C 1 Lump Lump
tions	Lump Sum Lump Sum Lump Sum Lump Sum Lump Sum Lump Sum	7,500 65,000 15,000 5,000 20,000	Lump Sum Lump Sum Lump Sum Lump Sum Lump Sum Lump Sum	10,000 75,000 20,000 21,000 30,000	Lump Lump Lump Lump Lump
Controls		-	1-13½T 500KW	100,000	1-20/
aces		<u> </u>	Lump Sum Lump Sum	30,000 20,000	Lump Lump
E)		\$ <u>366,000</u>		\$1,036,000	
D + E)		\$ <u>13,000</u>		\$ 35,000	
+ F)		\$ <u>439,000</u>		\$ <u>1,150,000</u>	
		\$ <u>22,000</u>		\$58,000	
		\$ <u>461,000</u>		\$ <u>1,208,000</u>	
		\$ <u>46,000</u>		\$ <u>121,000</u>	
		\$ <u>507,000</u>		\$ <u>1,329,000</u>	
		\$50.7	20	611	20

\$50.70

POLE HOT BLAST, WATER-COOLED, NNEIINDUCTION HOLDING FURNACE, WET CRUBBER EMISSION CONTROL

5 <u></u>	PH	15	грн		TPh	50 7	ГРН
imbe d Ste	Cost	Number and Size	Cost	Number and Size	Cost	Number and Size	Cost
	\$ 50,000 10,000]-2,500 Sq. F 1,500 Sq. Ft.	t. \$ 63,000 15,000	1-4,000 Sq. F 1-3,000 Sq. F		1-5,000 Sq. Ft 5,000 Sq. Ft.	\$125,000 50,000
0 Oen Sum Sum	60,000 19,000 20,000 10,000 6,000 10,000	1-7½ T-70° 16,000 Sq. Ft Lump Sum Lump Sum Lump Sum 1-65"	83,000 128,000 55,000 95,000 40,000 8,000	2-7½ T 25,000 Sq. Ft Lump Sum Lump Sum Sum 2 -65"	166,000 192,000 70,000 143,000 52,000 16,000	2-10T 30,000 Sq. Ft. Lump Sum Lump Sum Lump Sum 3-65"	200,000 240,000 90,000 240,000 75,000 24,000
•	24,000	1-15 TPH	134,000	1-30 TPH	165,000	1-50 ТРН	225,000
Sum SOI Sum Sum Sum Sum Sum Sum	40,000 50,000 25,000 5,000 3,000 1,500 7,500 65,000 15,000 5,000 20,000	1-66" Lump Sum 7,700 SCFM 1 Lump Sum	80,000 66,000 30,000 6,000 3,000 2,000 10,000 75,000 20,000 21,000 30,000	1-90" Lump Sum 14,300 SCFM 1 Lump Sum	180,000 84,000 52,000 12,000 4,500 3,000 12,000 12,000 25,000 27,000 40,000	1-108" Lump Sum 20,600 SCFM 1 Lump Sum	265,000 150,000 120,000 15,000 5,000 15,000 195,000 30,000 36,000
	-	1-13 ^L T 500KW Lump Sum Lump Sum	100,000 30,000 20,000	1-20/8 800KW Lump Sum Lump Sum	200,000 80,000 45,000	1-60T 1,200KW Lump Sum Lump Sum	350,000 125,000 90,000
	\$ <u>366,000</u>		\$ <u>1,036,000</u>		\$ <u>1,680,500</u>		\$ <u>2,545,000</u>
	\$ <u>13,000</u>		\$ 36,000	2 -	\$ 59,000		\$ 89,000
	\$439,000		\$1,150,000		\$ <u>1,869,500</u>		\$2,809,000
	\$ 22,000		\$ 58,000		\$ <u>93,000</u>		\$ 140,000
	\$ <u>461,000</u>		\$ <u>1,208,000</u>	£	\$ <u>1,962,500</u>		\$2,949,000
	\$ 46,000		\$ 121,000	· ·	\$ 196,000		\$ 295,000
	\$ <u>507,000</u>		\$ <u>1,329,000</u>		\$ <u>2,158,500</u>	ĺ	\$ <u>3,244,000</u>
	\$50.7	70	\$44. 22.		\$35.9 17.9	98 99	\$32.44 16.22



Α.	Buildings
	Nelting Building \$13/Sq. Ft. Transformer Rooms \$10/Sq. Ft.
В.	Scrapyard Equipment
	 Scrapyard Crane Crane Runway & Stockyard Platform Scales Magnets
c.	Charging Equipment
	 Charge Bucket Transfer Cars & Track Charge Buckets Charging Crane
D.	Melting Equipment
	1. Arc Furnaces 2. Transformers
	 Furnace Refractories Electrodes Power Feeders, Piping Headers, Sewer Connections Slag Buckets
E.	Holding Equipment
	 Channel Induction Furnaces & Electrics Spare Furnace Body Refractories for Holding Furnaces
	Subtotal
F.	Spares & Freight
	Subtotal
G.	Engineering 5%
	Subtotal
Н.	Contingencies 10%
	Total
	Cost/Ton 500 Hours 1,000 Hours 2,000 Hours
	1,000 Hours 2,000 Hours 4,000 Hours

10, 1,2

1-5 1-3

1-4

ź

2-3' 1-5

Lum; Lum;

None

1



ELECTRIC ARC FURNACE, CHANNEL INDUCTION HOLDING, FABRIC FILTER EMISSION CONTROL

		5 TPH		15_ТРН		
	Number and Size	Cost	Number and Size	Cost	a	
	10,000 Sq. Ft 1,200 Sq. Ft.	. \$ 130,000 12,000	18,000 1,500 Sq. Ft.	\$ 234,000 15,000	25,0 2,10	
	1-5T 1-3T 1-45"	60,000 19,000 15,000 6,000	1-7½T-80' 1-65"	83,000 128,000 20,000 8,000	2-7½ 2-65	
ack	2	10,000 10,000	Lump Sum 3 1-25/45	15,000 18,000 85,000	3	
·	2-3T 1-5,000KVA	410,000 Included in D-1	2-11"-15T 2-7,500 KVA	830,000 Included in D-1	3"-1 3-10	
ewer Connections	Lump Lump 1	9,000 2,000 10,000 3,000	Lump Lump Lump 2	20,000 6,000 20,000 8,000	Lump Lump	
ctrics ·s	None -	\$554,000 \$ 19,000 \$715,000 \$ 36,000 \$751,000 \$ 75,000 \$ 826,000	1-30T 600 KW. 1 Lump	325,000 115,000 45,000 \$1,726,000 \$ 61,000 \$2,036,000 \$ 102,000 \$2,138,000 \$ 214,000 \$2,352,000	2-40 1 Lump	
		\$330 165				

165 82.50

, .

\$156.80 78.40 39.20

LECTRIC ARC FURNACE, CHANNEL UCTION HOLDING, FABRIC FILTER EMISSION CONTROL

5.TPH		15 7	ГРН		Ttri	50 TPH		
mber Size	Cost	Number and Size	Cost	Number and Size	Cost	Number <u>and Size</u>	Cost	
Sq. Ft.	\$ 130,000 12,000	18,000 1,500 Sq. Ft.	\$ 234,000 15,000	25,000 Sq. Ft 2,100 Sq. Ft.		30,000 Sq. Ft. 3,000 Sq. Ft.	\$ 540,000 30,000	
	60,000 19,000 15,000 6,000	1-7½T-80' 1-65"	83,000 128,000 20,000 8,000	2-7½T 2-65"	166,000 192,000 30,000 16,000	2-10T 30,000 Sq. Ft. 2 3-65"	200,000 240,000 50,000 24,000	
	10,000 10,000	Lump Sum 3 1-25/45	15,000 18,000 85,000	3	20,000 18,000 100,000	2 4 1	25,000 24,000 130,000	
)KVA	410,000 Included in D-1	2-11"-15T 2-7,500 KVA	830,000 Included in D-l	3"-15T 3-10,000 KVA	1,275,000 Included in D-1	4-11'-15T 4-13,000 KVA	1,816,000 Included in D-1	
	9,000 2,000 10,000 3,000	Lump Lump Lump 2	20,000 6,000 20,000 8,000	Lump Lump 3	30,000 9,000 25,000 9,000	Lurp Limp 4	40,000 12,000 30,000 12,000	
	- -	1-30T 600 KW. 1 Lump	325,000 115,000 45,000	2-40T 800 KW 1 Lump	Ea. 700,000 125,000 90,000	2-50T 1200 KW 1 Luap	Ea.800,000 145,000 180,000	
	\$ <u>554,000</u>		\$ <u>1,726,000</u>		\$2,815,000		\$3,778,000	
	\$ <u>19,000</u>		\$ 61,000		\$ 99,000		\$ 132,000	
	\$ <u>715,000</u>		\$ <u>2,036,000</u>		\$3,260,000		\$ <u>4,480,000</u>	
•	\$ 36,000		\$ 102,000		\$ 163,000		\$ 224,000	
•	\$ <u>751,000</u>		\$2,138,000		\$3,423,000		\$ <u>4,704,000</u>	
	\$ 75,000		\$ 214,000		\$ 342,000		\$ 470,000	
•	\$ <u>826,000</u>		\$ <u>2,352,000</u>		\$ <u>3,765,000</u>		\$ <u>5,174,000</u>	
	*							

\$330 165 82.50

> \$156.80 78.40 39.20

\$62.75 31.38 \$51.74 25.87



CORELI PREHE/

Numbe __and Si

Α. Buildings

Melting Building \$17/Sq. Ft. Transformer Building \$10/Sq. Ft.

5,000 1,000 Sq.

В. Scrapyard Equipment

- Scrapyard Crane
- Crane Runway 2.
- Platform Scales 3.
- 4. Magnets
- Charge Bucket Transfer Cars & Track

1-45"

2-10 Ton KW 2,100

1

1-5T

0pen

C. Charging Equipment

Charging Monorail

D. Melting Equipment

- 1. Coreless Induction Furnaces
- 3. Preheater with Charge Bucket
- Slag Box
- Piping, Headers, Sewer Connections
- Holding Equipment None Specified

Subtotal #1 (B + C + D + E)

Spares & Freight $(3\frac{1}{2}\% \text{ of B, C, D, E})$

Subtotal #2 (ST. #1 + A + F)

Engineering (5% of ST. #2) G.

Subtotal #3 (G + ST: #2)

Contingencies (10% of ST. #3) Н.

Total (ST. #3 + H)

Cost/Ton 500

1,000 2,000

1,000 2,000 4,000

2,000

4,000

CORELESS INDUCTION FURNACES, WITH PREHEATERS, NO HOLDING FURNACES, AFTERBURNER ON PREHEATER

5 TPI	<u>-i</u>	15 TP		
Number and Size	Cost	Number and Size	Cost	Nu and
			-	
5,000	\$ 85,000	9,000 Sq. Ft.	\$ 153,000	13,500
1,000 Sq. Ft.	10,000	1,200 Sq. Ft.	12,000	1,800
1-5T	60,000	1-7½T-80' Span	83,000	2- 7½T
Open	19,000	Covered	128,000	Covere
1-45"	15,000	1	20,000	2 (5!)
1-45	6,000 8,000	1-65" Lump	8,000 10,000	2-65"
		·		
	40,000		50,000	
2-10 Ton KW 2,100 Each	400,000	2-20 Ton 4,100 KW Each	850,000	3-25 T 6,250
1	28,000	2 2 2 Each	64,000	3
	2,000 10,000		5,000 20,000	
,	\$588,000		1,238,000	
	\$ 21,000	\$ \$		
				
	\$704,000		1,446,000	
. •	\$ <u>35,000</u>	, \$,	72,000	
	\$ <u>739,000</u>	\$.	1,518,000	
	\$_74,000	\$.	152,000	
	\$813,000	\$	1,670,000	
	4205			
	\$325	. 0		

325 162.50 81.25

> \$111.33 55.67 27.84

SS INDUCTION FURNACES, WITH TERS, NO HOLDING FURNACES, TERBURNER ON PREHEATER

5 TPH		15 ТРН		30 1	TF.T	50 TPH		
r ze	Cost	Number and Size	Cost	Number and Size	Cost	Number and Size	Cost	
Ft.	\$ 85,000 10,000	9,000 Sq. Ft. \$ 1 1,200 Sq. Ft.	153,000 12,000	13,500 Sq. Ft. 1,800 Sq. Ft.	, \$ 230,000 18,000	20,000 Sq. Ft. 3,000 Sq. Ft.	\$ 340,000 30,000	
	60,000 19,000	1-7½T-80' Span Covered	83,000 128,000	2-7½T Covered	166,000 192,000	2-10T Covered 30,000 Sq. Ft.	200,000 240,000	
	15,000 6,000 8,000	1 1-65" Lump	20,000 8,000 10,000	2-65"	30,000 16,000 15,000	3 2 2	50,000 ~ 24,000 20,000	
	40,000		50,000		60,000		75,000	
Each	400,000 28,000 2,000 10,000	2-20 Ton 8 4,100 KW Each 2	64,000 5,000 20,000	3-25 Ton 6,250 KW Each 3	1,600,000 129,000 7,000 25,000	4-30 Ton 6,750 KW Each 4	2,200,000 172,000 10,000 30,000	
ż	\$588,000	\$ <u>1,2</u>	238,000		\$2,240,000		\$3,021,000	
	\$ <u>21,000</u>	\$	43,000		\$ 78,000		\$ 106,000	
	\$704,000	\$ <u>1,</u> 4	446,000		\$2,566,000	:	\$ <u>3,497,000</u>	
	\$ 35,000	\$	72,000		\$ <u>128,000</u>		\$ 175,000	
	\$ <u>739,000</u>	\$ <u>1,5</u>	518,000		\$ <u>2,694,000</u>	:	\$ <u>3,672,000</u>	
	\$ 74,000		152,000		\$ 269,000	:	\$ 367,000	
	\$ <u>813,000</u>	\$ <u>1,6</u>	570,000		\$ <u>2,963,000</u>	:	\$ <u>4,039,000</u>	

\$325 162.50 81.25

> \$111.33 55.67 27.84

\$ 49.38 24.69

\$ 40.39 20.20



DIRECT M

Metallics	Cost per Pound	<u>L</u> <u>Percent</u>	ined Cupol Pounds	La <u>Cost</u>	Perc
Pig Iron #2 Heavy Melting Steel Scrap #1 Heavy Melting Steel Scrap Borings-Briquettes Borings-Loose Iron Scrap-Remelt Iron, Cast Scrap Sil Mn. Briquettes Sil. Carb. Briquettes Fe. Si. 85% Fe. Si. 50%	\$.03326 .01451 .01813 .01506 .00924 .02121 .02121 .1050 .075 .1945 .1530	5% 30 5 15 - 34 10 .40 .40 .20	104 623 104 312 - 706 208 8 4	\$3.46 9.04 1.89 4.70 14.97 4.41 .84 .60	37 5 15 - 32 10
			2,077		
Nonmetallics					
Coke Sil. Mn. Briquettes Carbo-Graphite Soda Ash Limestone Sil. Carb. Briquettes	.02475 .1050 .04 .03 .00388 .075		360 1.5 8.5 2.5 60 9	\$8.91 .16 .34 .08 .23 68	
Cost per Ton				\$ <u>51.09</u>	

DIRECT MATERIAL COST

L	ined Cupol	.a	Wate	r-Cooled (Cupola	Electi	ric Furna	ce_
t	Pounds	Cost	Percent	<u>Pounds</u>	Cost	Percent	<u>Pounds</u>	
	104 623	\$3.46 9.04	37	800	\$11.61	٠. ٠		
	104 312	1.89 4.70	5 15	104 312	1.89 4.70	50% -	1,045	\$
	706 208	14.97 4.41	32 10	706 208	14.97 4.41	15 33 -	300 680	
	8 4 4 -	.84 .60 .78	.40 .40 .20	8 8 4	.84 .60 .78	.4	8 30	
	2,077	٠.		2,150	·	- - · ·	2,063	
	360 1.5 8.5	\$8.91 .16 .34		250 1.5	\$6.19 .16		60	
	2.5 60 9	.08 .23 <u>.68</u>		2.5 60 9	.08 .23 <u>.68</u>			
		\$ <u>51.09</u>			\$ <u>47.14</u>			\$

TERIAL COST

ater-Cooled Cupola		Electric Furnace			Indu	Induction Furnace		
nt	Pounds	Cost	Percent	<u>Pounds</u>	Cost	Percent	Pounds	Cost
·	800 104 312 - 706 208	\$11.61 1.89 4.70 14.97 4.41	50% - 15 33	1,045 300 680	\$18.95 2.77 14.42	27 15 - 34 22	550 302 686 445	\$ 9.97 4.55 14.55 9.44
0 0	8 8 4 —————————————————————————————————	.84 .60 .78	1.4	$\frac{8}{30}$ 2,063	1.56 4.59	1.4	8 30 2,021	1.56 4.59
	250 1.5 - 2.5 60 9	\$6.19 .16 .08 .23 68		60	\$2.40		60	\$2.40
		\$ <u>47.14</u>			\$ <u>44.69</u>			\$ <u>47.06</u>

SUMMARY OF CO CUPOLA, COLD BLAST FABRIC FILTER

	Melt Rate 5 Tons Per Hour			
Costs Per Ton	Operation 500	ng Hours 1,000	Per Year 2,000	
Direct and Indirect Labor	\$18.00	\$16.00	\$14.00	
Salaried Personnel	4.32	3.60	2.88	
Depreciation	16.00	8.00	5.00	
Capital Charges (Interest, Insurance, Taxes)	20.40	10.20	6.50	
Electrical Power	.07	. 05	.04	
Gas	.05	.05	.05	
Supplies and Maintenance Material	3.00	3.00	3.00	
Allocated Costs	4.00	4.00	4.00	
Total	\$ <u>65.84</u>	\$ <u>44.90</u>	\$ <u>35.2</u> 7	

SUMMARY OF CONVERSION COSTS

CUPOLA, COLD BLAST NO HOLDING FURNACE, FABRIC FILTER EMISSION CONTROL

Melt Rate Melt Rate						Melt Ra	
	ns Per Hou			ns Per Ho		30 Tons I	
<u>00</u>	ng Hours 1 1,000	2.000	Operating 1,000	2,000	4,000	Operating 2,000	HOT
8.00	\$16.00	\$14.00	\$ 8.67	\$ 7.33	\$ 6.36	\$ 5.83	!
4.32	3.60	2.88	1.92	1.92	1.44	.96	
6.00	8.00	5.00	6.73	4.07	2.03	3.50	
0.40	10.20	6.50	8.80	5.30	2.65	4.55	
.07	.05	.04	.05	.03	.03	.03	
.05	.05	.05	.05	.05	.05	.05	
3.00	3.00	3.00	3.00	3.00	3.00	3.00	
4.00	4.00	4.00	4.00	4.00	4.00	4.00	
<u>5.84</u>	\$ <u>44.90</u>	\$ <u>35.27</u>	\$ <u>33.22</u>	\$ <u>25.70</u>	\$ <u>19.56</u>	\$ <u>21.92</u>	٠

NUMBER 44

VERSION COSTS

NO HOLDING FURNACE, MISSION CONTROL

Melt Rate 15 Tons Per Hour Operating Hours Per Year			Melt Rate 30 Tons Per Hour Operating Hours Per Year			Melt F 50 Tons Operating	Year	
1,000	2,000	4,000	2,000	4,000		2,000	4,000	
\$ 8.67	\$ 7.33	\$ 6.36	\$ 5.83	\$ 4.58		\$ 4.30	\$ 3.35	
1.92	1.92	1.44	.96	.72		.58	.43	
6.73	4.07	2.03	3.50	1.75		2.87	1.44	
8.80	5.30	2.65	4.55	2.28		3.73	1.87	
.05	.03	.03	.03	.02		.02	.02	
.05	.05	.05	.05	.05		.05	.05	
3.00	3.00	3.00	3.00	3.00		3.00	3.00	
4.00	4.00	4.00	4.00	4.00		_4.00	4.00	
\$ <u>33,22</u>	\$ <u>25.70</u>	\$ <u>19.56</u>	\$ <u>21.92</u>	\$ <u>16.40</u>		\$ <u>18.55</u>	\$ <u>14.16</u>	

SUMMARY OF CO

CUPOLA, HOT BLAST, N CHANNEL INDUCTION F ENERGY WET SCRUBI

	Melt Rate 5 Tons per Hour			
_		lours per Yea		
Cost per Ton	2,000	4,000		
Direct and Indirect Labor	\$17.00	\$15.00		
Salaried Personnel	2.88	2.16		
Depreciation	5.07	2.54		
Capital Charges (Interest, Insurance, Taxes)	6.59	3.29		
Electrical Power	.05	.04		
Water	.20	.20		
Gas	.25	.25		
Supplies and Maintenance Material	2.00	2.00		
Allocated Costs	4.00	4.00		
Total	\$ <u>38.04</u>	\$ <u>29.48</u>		

SUMMARY OF CONVERSION COSTS

CUPOLA, HOT BLAST, WATER-COOLED, UNLINED, CHANNEL INDUCTION HOLDING FURNACE, HIGH ENERGY WET SCRUBBER EMISSION CONTROL

Melt 5 Tons perenting Ho 2,000		15 Tons	Rate per Hour ours per Year 4,000	Melt Rate 30 Tons per Operating Ho 2,000
\$17.00	\$15.00	\$ 7.33	\$ 5.83	\$ 5.50
2.88	2.16	1.92	1.44	.96
5.07	2.54	4.43	2.22	3.60
6.59	3.29	5.76	2.88	4.68
.05	.04	.62	.40	.52
.20	.20	.20	.20	.20
.25	.25	.25	.25	.25
2.00	2.00	2.00	2.00	2.00
4.00	4.00	4.00	4.00	4.00
\$ <u>38.04</u>	\$ <u>29.48</u>	\$ <u>26.51</u>	\$ <u>19.22</u>	\$ <u>21.71</u>

NUMBER 45

VERSION COSTS

TER-COOLED, UNLINED, LDING FURNACE, HIGH R EMISSION CONTROL

15 Tons	t Rate per Hour	Melt Rate	Hour	Melt Ra 50 Tons pe	er Hour
<u>Operating H</u> <u> 2,000</u>	lours per Year 4,000	Operating Ho	ours per Year 4,000	Operating Ho	ours per Year 4,000
\$ 7.33	\$ 5.83	\$ 5.50	\$ 4.41	\$ 3.90	\$ 3.10
1.92	1.44	.96	.72	.58	.43
4.43	2.22	3.60	1.80	3.24	1.62
5.76	2.88	4.68	2.34	4.21	2.11
.62	.40	.52	.33	.45	.29
.20	.20	.20	.20	.20	.20
.25	.25	.25	.25	.25	.25
2.00	2.00	2.00	2.00	2.00	2.00
4.00	4.00	4.00	4.00	4.00	4.00
\$ <u>26.51</u>	\$ <u>19.22</u>	\$ <u>21.71</u>	\$ <u>16.05</u>	\$ <u>18.83</u>	\$ <u>14.00</u>

SUMMARY OF CELECTRIC ARC FURNACE WITH CELECTRIC FILTE

	Melt 5 To Operati	- 0		
Costs Per Ton	_5 <u>00</u>	1,000	2,000	$\frac{\overline{\overline{0}}}{\overline{0}}$
Direct and Indirect Labor	\$18.00	\$14.00	\$12.00	
Salaried Personnel	4.32	3.60	2.88	•
Depreciation	33.04	16.52	8.26	
Capital Charges (Interest, Insurance,	Taxes)42.95	21.48	10.74	
Electrical Power	36.00	19.90	11.76	
Water	.05	.05	.05	
Gas	.05	.05	.05	
Electrodes	2.85	2.85	2.85	
Supplies and Maintenance Material	2.50	2.50	2.50	
Allocated Costs	4.00	4.00	4.00	
Total	\$143.76	\$84.95	\$55.09	\$

SUMMARY OF CONVERSION COSTS

ELECTRIC ARC FURNACE WITH CHANNEL INDUCTION HOLDING FURNACE, FABRIC FILTER EMISSION CONTROL

Melt Rate 5 Tons Per Hour Operating Hours Per Year			15 Tor	Rate is Per Ho	Melt Rate 30 Tons Per Hour		
<u>500</u>	1,0 <u>00</u>	2,000	1,000	18 Hours 2,000	4 <u>,000</u>	Operating Hou	4,
\$18.00	\$14.00	\$12.00	\$7.66	\$6.67	\$5.50	\$5.33	Ç
4.32	3.60	2.88	1.92	1.92	1.44	.96	
33.04	16.52	8.26	15.67	7.84	3.92	6.28	
kes)42.95	21.48	10.74	20.40	10.20	5.10	8.17	
36.00	19.90	11.76	19.68	11.86	7.92	11.25	
.05	.05	.05	.05	.05	.05	.05	
.05	.05	.05	.05	.05	.05	.05	
2.85	2.85	2.85	2.85	2.85	2.85	2.85	
2.50	2.50	2.50	2.50	2.50	2.50	2.50	
4.00	4.00	4.00	4.00	4.00	4.00	4.00	_
\$ <u>143.76</u>	\$ <u>84.95</u>	\$55.09	\$74.78	\$ <u>47.94</u>	\$ <u>33.33</u>	\$ <u>41.44</u>	\$2 =

SUMMARY OF CELECTRIC ARC FURNACE WITH CELECTRIC FILTE:

	Melt Rate 5 Tons Per Hour				
Costs Per Ton	<u>Operati</u> _500	ng Hours I,000	Per Year 2,000		
Direct and Indirect Labor	\$18.00	\$14.00	\$12.00		
Salaried Personnel	4.32	3.60	2.88		
Depreciation	33.04	16.52	8.26		
Capital Charges (Interest, Insurance,	Taxes)42.95	21.48	10.74		
Electrical Power	36.00	19.90	11.76		
Water	.05	.05	.05		
Gas	.05	.05	.05		
Electrodes	2.85	2.85	2.85		
Supplies and Maintenance Material	2.50	2.50	2.50		
Allocated Costs	4.00	4.00	4.00		
Total	\$ <u>143.76</u>	\$ <u>84.95</u>	\$ <u>55.09</u>		

SUMMARY OF CONVERSION COSTS

TRIC ARC FURNACE WITH CHANNEL INDUCTION HOLDING FURNACE, FABRIC FILTER EMISSION CONTROL

Rate ns Per Ho)))		Rate ns Per Ho	112	Melt Rate 30 Tons Per 1	Jour
ng Hours			ng Hours		Operating Hou	
1,000	2,000	1,000	2,000	4,000	2,000	4.
\$14.00	\$12.00	\$7.66	\$6.67	\$5.50	\$5.33	Ç
3.60	2.88	1.92	1.92	1.44	.96	
16.52	8.26	15.67	7.84	3.92	6.28	
21.48	10.74	20.40	10.20	5.10	8.17	
19.90	11.76	19.68	11.86	7.92	11.25	
.05	.05	.05	.05	.05	.05	
.05	.05	.05	.05	.05	.05	
2.85	2.85	2.85	2.85	2.85	2.85	
2.50	2.50	2.50	2.50	2.50	2.50	
4.00	4.00	4.00	4.00	4.00	4.00	_
\$ <u>84.95</u>	\$ <u>55.09</u>	\$ <u>74.78</u>	\$ <u>47.94</u>	\$ <u>33.33</u>	\$ <u>41.44</u>	\$2

APPENDIX D NUMBER 46

DNVERSION COSTS

ANNEL INDUCTION HOLDING FURNACE, EMISSION CONTROL

eratir	ns Per Houng Hours	Per Year	Operat	s Per I	ıra Per Y	ear		s Per ng Ho	urs Per	Year
000	2,000	4,000		000	4,000		2,0		4 <u>,000</u>	
7.66	\$6.67	\$5.50	:	\$5.33	\$4.50		\$4	. 10	\$3.45	
1.92	1.92	1.44		.96	.72			. 58	.43	
5.67	7.84	3.92		6.28	3.14		5	.17	2.59	
0.40	10.20	5.10		8.17	4.09		6	.73	3.37	
9.68	11.86	7.92	. 1	11.25	7.57		11	.11	7.50	
.05	.05	.05		.05	.05			.05	.05	
. 05	.05	.05		.05	.05			.05	.05	
2.85	2.85	2.85		2.85	2.85		2	.85	2.85	
2.50	2.50	2.50		2.50	2.50		2	. 50	2.50	
4.00	4.00	4.00		4.00	4.00		_4	.00	4.00	
4.78	\$ <u>47.94</u>	\$ <u>33.33</u>	\$4	41.44	\$ <u>29.47</u>		\$ <u>37</u>	.14	\$ <u>26.79</u>	

SUMMARY OF CO

CORELESS INDUCTION FURNAL NO HOLDING FURNACE, A

	Mel <u>5 Ton</u> Operati	<u>]</u>		
Costs Per Ton	<u>500</u>	1,000	2,000	$\frac{\overline{O}_{I}}{1}$
Direct and Indirect Labor	\$18.00	\$15.00	\$14.00	\$
Salaried Personnel	4.32	3.60	2.88	
Depreciation	32.52	16.26	8.13	1
Capital Charges (Interest, Insurance,	Taxes)42.27	21.14	10.57]
Electrical Power	39.79	22.18	13.37	1
Water	.05	.05	.05	
Gas	. 30	.30	.30	
Supplies and Maintenance Material	3.00	3.00	3.00	
Allocated Costs	4.00	4.00	4.00	_
Total	\$ <u>144.25</u>	\$ <u>85.53</u>	\$ <u>56.30</u> `	S _i

لهنو

SUMMARY OF CONVERSION COSTS

CORELESS INDUCTION FURNACE WITH CHARGE PREHEATER, NO HOLDING FURNACE, AFTERBURNER ON PREHEATER

<u>'on</u>	t Rate s Per Hours 1,000	Per Year 2,000	; !	15 Tons	Rate S Per Hours 12,000		Melt Rate 30 Tons Per Operating Hou 2,000	Ho
10	\$15.00	\$14.00		\$ 8.00	\$ 7.00	\$ 6.36	\$ 4.84	\$
;2	3.60	2.88		1.92	1.92	1.44	.96	
-2	16.26	8.13		11.13	5.57	2.78	4.94	
.7	21.14	10.57		14.47	7.24	3.62	6.42	
9	22.18	13.37		14.86	9.40	6.64	9.64	
۰5	.05	.05		.05	.05	.05	.05	
0	.30	.30		.30	.30	.30	.30	
0	3.00	3.00		3.00	3.00	3.00	3.00	
<u>0</u>	4.00	4.00		4.00	4.00	4.00	4.00	
<u>5</u>	\$ <u>85.53</u>	\$ <u>56.30</u>		\$ <u>57.73</u>	\$ <u>38.48</u>	\$28.19	\$ <u>34.15</u>	\$.

APPENDIX D NUMBER 47

WERSION COSTS

E WITH CHARGE PREHEATER, TERBURNER ON PREHEATER

Me l t	Rate		Melt Rate	2	Melt Rate
	s Per Hou		30 Tons Per		50 Tons Per Hour
	ng Hours		Operating Hou	ırs Per Yea	r Operating Hours Per Year
00	2,000	4,000	2,000	4,000	2,000 4,000
3.00	\$ 7.00	\$ 6.36	\$ 4.84	\$ 4.16	\$ 3.90 \$ 3.40
1.92	1.92	1.44	.96	.72	.58 .43
1.13	5.57	2.78	4.94	2.47	4.04 2.02
4.47	7.24	3.62	6.42	3.21	5.25 2.62
4.86	9.40	6.64	9.64	6.76	8.68 6.28
.05	.05	.05	.05	.05	.05 .05
.30	.30	.30	.30	.30	.30 .30
3.00	3.00	3.00	3.00	3.00	3.00 3.00
4.00	4.00	4.00	4.00	4.00	4.00 4.00
<u>7.73</u>	\$ <u>38.48</u>	\$ <u>28.19</u>	\$ <u>34.15</u>	\$ <u>24.67</u>	\$ <u>29.80</u> \$ <u>22.10</u>

ANNUAL FOR EMISSION CO

		5 Tons/Hour				
		500	_1,000	2,000		
1.	Fabric Filter on Cupola (Cold Blast	<u>:</u>)	·			
	Number Cupolas ACFM Each	9,200	9,200	9,200		
	Annual Operating Cost Annual Tons Cost/Ton	\$25,000 2,500 \$10	\$30,000 5,000 \$6	\$35,000 10,000 \$3.50		
2.	Wet Scrubber on Cupola (Hot Blast)					
	ACFM Annual Operating Cost Annual Tons Cost/Ton			6,800 \$35,000 10,000 \$3.50		
3.	Fabric Filter on Electric Arc					
	Furnace Diameter Annual Operating Cost/System Number Fabric Filter Systems	7'3" \$28,000 2	7'3'' \$31,000	7 ' 3'' \$37,500		
Total Annua	Total Annual Operating Cost Annual Tons Cost/Ton	\$56,000 2,500 \$22.40	\$62,000 5,000 \$12.40	\$65,000 10,000 \$6.50		
4.	Afterburner on Coreless Induction					
	Annual Operating Cost Tons Cost/Ton	\$1,150 2,500 \$.46	\$1,350 5,000 \$.27	\$1,730 10,000 \$.17		

ANNUAL OPERATING COSTS FOR EMISSION CONTROL EQUIPMENT SYSTEMS

		15 Ton 1,000 2,C											
	500	1,000	2,000	4,000	1,000	<u> 2, C</u>							
Cold Blast)													
	9,200	9,200	9,200		27,300	27,							
	\$25,000 2,500 \$10	\$30,000 5,000 \$6	\$35,000 10,000 \$3.50		\$60,000 15,000 \$4.00	\$70, 30, \$2							
ot Blast)													
	•		6,800 \$35,000 10,000 \$3.50	6,800 \$50,000 20,000 \$2.50		22, \$60, 30, \$2							
<u>Arc</u>													
ystem stems Cost	7'3" \$28,000 2 \$56,000 2,500 \$22.40	7'3" \$31,000 2 \$62,000 5,000 \$12.40	7'3" \$37,500 2 \$65,000 10,000 \$6.50		\$41,000 2 \$82,000 15,000 \$5.47	\$49, \$98, 30, \$3							
nduction	•												
	\$1,150 2,500 \$.46	\$1,350 5,000 \$.27	\$1,730 10,000 \$.17		\$3,040 15,000 \$.20	\$4; 30;							

)PERATING COSTS 'TROL EQUIPMENT SYSTEMS

,000	15 Tons/Hour 1,000 2,000 4,000		30 Tons/Hour		50 Tons/Hour		
,000	1,000	2,000	4,000	_2,000	4,000	2,000	4,000
	1 27,300	27,300	27,300	27,300	27,300	2 37,200	2 37,200
		ŕ	·	Each	Each	Each	Each
	\$60,000 15,000	\$70,000 30,000	\$80,000 60,000	\$140,000 60,000	\$160,000 120,000	\$180,000 100,000	\$200,000 200,000
	\$4.00	\$2.33	\$1.33	\$2.33	\$1.33	\$1.80	\$1.00
6,800		22,800	22,800	43,000	43,000	62,000	62,000
50,000		\$60,000	\$75,000	\$110,000	\$140,000	\$165,000	\$195,000
20,000 \$2.50		30,000 \$2.00	60,000 \$1.25	60,000 \$1.83	120,000 \$1.17	100,000 \$1.65	200,000
	111	111	111	111	111	11!	11!
	11' \$41,000	11' \$49,000	11' \$64,000	11' \$49,000	11' \$64,000	11' \$49,000	11' \$64,000
	2 \$82,000	2 \$98,000	\$128,000	3 \$147,000	\$192,000	4 \$196,000	4 \$256,000
	15,000	30,000	60,000	60,000	120,000	100,000	200,000
	\$5.47	\$3 . 27	\$2.13	\$2.45	\$1.60	\$1.96	\$1.28
	\$3,040	\$4,150	\$6,380	\$7,260	\$11,640	\$11,060	\$18,280
	15,000 \$.20	30,000 \$.14	60,000 \$.11	60,000 \$.12	120,000 \$.10	100,000 \$.11	200,000 \$.09
	,	,	,	,		•	

EMISSION TEST PROCEDURES

A standard recommended procedure for testing particulate emissions from iron foundry cupolas did not exist until the end of 1970. At that time the "Recommended Practice for Testing Particulate Emissions from Iron Foundry Cupolas" was adopted by the American Foundrymen's Society and the Gray and Ductile Iron Founders' Society, Inc. This industry standard procedure, broadly based on the American Society of Mechanical Engineers Performance Test Codes 21-1941 and 27-1957, recognizes the unique problems of cupola testing and recommends procedures to deal with them in a satisfactory manner. The recommended practice is reproduced in Exhibit 1 of this appendix for information purposes only.

In a more specific manner, Exhibit 2 presents detailed sampling and analytical techniques for individual components of cupola emissions. These techniques are widely accepted by chemical and testing laboratories and used in analytical work. Techniques are included for identification and quantification of particulate matter such as arsenic, beryllium, cadmium, fluorides, lead, mercury, and zinc, and two gaseous components, nitrogen oxides and sulfer dioxide.

RECOMMENDED PRACTICE FOR TESTING

PARTICULATE EMISSIONS

FROM

IRON FOUNDRY CUPOLAS

Edited by

A Joint Committee of

American Foundrymen's Society

and

The Gray & Ductile Iron Founders' Society, Inc.

The recommended procedure discussed in this section has, as of this date, not been endorsed by any bodies other than A.F.S. and G.D.I.F.S. and is presented for information only.

INTRODUCTION

The iron foundry industry has had many air pollution studies conducted on cupola emissions at their various plants. Of great concern to the industry and to the individual firms that have conducted such testing are the many varied and diverse test methods and test procedures used by the variety of independent organizations conducting such tests. The diverse methods and equipment used in performing such tests have made comparison and evaluation of results impractical or a near impossibility. Many of the tests conducted have shown marked inconsistencies between individual test runs by the same test group and also in comparing the results on the same system by different testing organizations.

A number of the procedures used in cupola testing suffer from obvious inadequacies when they are carefully scrutinized. Consequently, it has been deemed desirable and necessary that a recommended test procedure and testing method be made available to assist the metalcasting industry in achieving the maximum in emission control with the minimum of wasted and misdirected effort and expense. Since the industry is unique in the large, nonproductive investments needed to gain compliance with air pollution control requirements, it is especially significant that its emissions be evaluated by test methods and procedures able to produce consistently reliable results detailing these emissions, but do not unnecessarily and unfairly penalize the plant.

Particulate emission tests of cupola stack gases are done under varied conditions and in several different locations, depending on the test objective. Both location and objective influence the test equipment employed although the two usual purposes will be:

- to determine nature and/or quantity of emissions released in the raw cupola gases
- 2) to determine nature and/or quantity of emissions on the cleaned gas side of a control unit.

Raw gas test locations:

- a) In cupola stack, above charging door. This is the most difficult location for testing. Gas flow is extremely uneven and the flow rate is relatively low; gas temperature is high often 1200°-2200°F and fluctuating; dust loading is extremely uneven because of channeling caused by indraft of much cold outside air drawn into the cupola stack through the charge door. This test location is necessary where a cupola has no control systems or has a wet cap type collector.
- b) In inlet duct ahead of dust collector. This is an easier location if a reasonably straight duct run is available. Duct velocities and dust loadings are more uniform and confined in a smaller cross section. Normally gases will be cooled to 500°F or lower at the sampling point from evaporation of cooling water. The added volume of water vapor must be measured and considered in gas density calculations and dust loadings if reported in grains per standard cubic feet dry gas. Inlet and outlet samples should be supplemented wherever possible by using the catch as a check for the test data. Catch can be more readily obtained from dry collector types especially for a complete melting cycle.
- c) Catch plus outlet loadings. Where dry collectors are employed, the entire test procedure is simplified by actual weighing of collected material. The higher the efficiency of the collecting device the more nearly the catch will represent the raw sample. Chances for error are diminished because of quantity

of collected material available although it will be difficult to obtain accurate catch quantities except for a complete melting cycle - thus providing an averaging of the peaks and valleys of emission concentrations.

See comments for outlet loading under "Cleaned Gas Locations".

Cleaned Gas Locations:

- a) After dry collector. Conventional dust sampling techniques will be satisfactory for such locations. Coarse particles will be removed by a dust collector so the importance of a large diameter sampling probe diminishes. Water vapor content of the gas should cause no condensation problems with 350° to 550°F gas temperatures. Collecting device in sampler can be influenced by intended analysis gross weight, particle size distribution, chemical composition, particle count, etc.
- b) After wet collector. Sampling problems are more complicated than after dry collectors because gas stream is saturated or nearly so. Close coupling of sampling components is essential and heating of the sampled air often required.

 Exception: Wet cap type of collectors have too short a contact time to bring gas stream close to saturated conditions. Sampling after the collector will be questionable value unless gases are gathered in a discharge stack of several diameter lengths.

In recognition of these differences in purpose and location for testing emissions the following procedure is divided into three sections.

Section I deals exclusively with sampling raw particulate emissions in the cupola stack.

Section II deals exclusively with sampling raw particulate emissions in the inlet duct connecting the cupola to the dust collector.

Section III deals exclusively with sampling cupola gases after they have been cleaned.

REASONS FOR SAMPLING A CUPOLA

Basically sampling is done for three reasons:

- to determine if a collecting device is of a high enough efficiency so that its effluent does not exceed a predetermined level.
- 2) to meet regulatory requirements that specify a minimum efficiency of removal of particulate from the gas stream, expressed as a percentage of uncontrolled emission.
- 3) to obtain information regarding particulate emission which will be used for designing gas cleaning devices.

Officials of local, regional or state regulatory bodies should be consulted prior to testing except when the testing is being done for purely informational data for the cupola owner or operator.

If source testing is being done to determine compliance with legal requirements the appropriate control officials should be consulted. If the control body has experience and is equipped to perform cupola testing, they may wish to perform their own tests to determine compliance.

Generally control bodies will not accept the results of tests performed by the owners, operators or vendors of collection devices unless standard procedures were followed and test data and reports show evidence that experienced personnel conducted the tests.

In most cases it will be necessary for the owner or operator of a cupola to employ the services of an organization capable of performing these tests. When this is done the control authorities should have given prior approval of the testing organizations capabilities and acceptability of their test results. In any event, it is advisable to notify the proper authorities in advance so that they may have on site observers present if they so desire.

. . .

The foundryman should select a testing organization with proven capability, a good reputation and in whom he has complete confidence. As test data can have major economic consequences and as the foundryman usually cannot check the quality of the testing procedures confidence in the organization is a prerequisite.

The next step is consultation with the appropriate control authorities. The foundryman along with the testing organization must involve themselves in this because regulations are sometimes not easily understood, and frequently interpretation is modified by political and community attitudes. Authorities will be aware of changes in enforcement policies, or pending changes in legislation, and the foundryman cannot expect outside testing organizations to be cognizant of these considerations.

The number and type of tests to be taken must be agreed on in advance by all parties concerned. Frequently, meeting the specifications of the pertinent code dictate the number and kind of samples to be run. At other times the purchase agreement between vendor and foundryman specifies testing methods. If discretion can be used the use of several short tests is recommended over one longer one. When several results can be compared, any large differences are evident. If these differences are not as a result of operational changes or adjustments they may indicate error in the test procedure or malfunction of the test equipment. One test of long duration gives only one answer with no basis

for comparison. Accuracy and precision of testing is controlled as much by the care exercised and quality of the testing personnel as it is by the test procedure.

Errors in each manipulation such as weighing, measuring gas volume, and calculating results must not exceed 1 percent and should be kept under that if possible. In this way cumulative errors can be held to little more than 1 percent.

CUPOLA OPERATING AND TEST CONDITIONS

Due to the various possible modes of operation of cupolas and cupola systems, it is recommended that cupola emissions be evaluated under conditions that characterize normal or average cupola operations at any particular plant.

Particulate matter emitted via raw cupola stack gases consists principally of iron oxides and silica from the charge metal and impurities adhering to the charge metal plus combustible matter. Secondary combustion in the upper portion of a cupola stack will tend to reduce the combustible portion of the particulate emissions to ash if temperature and retention time are sufficient.

Cupola stack gas will also contain some vapor from substances which reaches the melting zone and is volatilized. These substances include silicon, zinc and silica (sand). The degree of volatilization will depend on melting zone temperature which is influenced by changes in the fuel (coke and/or gas) ratio, preheating of the blast air or scrap and enrichment of the blast air with oxygen. Consequently, it is of utmost importance that the factors affecting melt zone temperature be normal before testing begins. Equally as important, materials that can cause fuming, such as galvanized iron, sand, and silicon, for example, be added in normal amounts during the test period. Changes from normal melt process can result in emissions which are markedly better or worse than will be obtained during everyday operation. Either result will be unsatisfactory.

The particulate matter emanating from a cupola has a wide range of particle size distribution which influences the correct choice of stack testing method. For many cupolas, peaks in particle sizes can be found on distribution curves at three ranges. These are in the 200 to 500 micron range, the 20 to 50 micron range and the 5 micron and below range.

Many factors influence the particulate emission rates of a cupola system. These include the rate of cupola operation, the character, cleanliness and method of introduction of the charge material, the type, size and amount of the coke used, the frequency, length of time and number of periods when tuyere blast air is operative or inoperative during any period, the type of metal being melted, the method and type of alloy introduction, and other diverse factors.

It is necessary, therefore, that each cupola and cupola system be individually analyzed to determine conditions under which stack or source emission tests are needed to define the full range and character of its emissions.

One of the factors having a most profound effect in cupola emissions is the rate of cupola melting; as cupola blast air and coke input is increased to accommodate higher melting rates, cupola emissions increase significantly. It is important, therefore, that cupola source-emission tests be conducted at melting rates approaching the normal expected rate of cupola operation if the results are expected to characterize emissions for the system. Often times it is not practical to operate at maximum melting rates since melting rates must reflect current production and pouring schedules. It should be appreciated, however, that cupola charging and melting rates have a profound influence on cupola emissions.

If for any reason tests during either start-up or burn down periods are made such tests should be kept and evaluated separately from each other as well as all others.

If various metals are produced at various times from the same cupola (such as gray and cutile iron) it is desirable that the emissions be evaluated for each type produced if there is a difference in melting conditions. The melting conditions that would tend to require evaluation in terms of differences in emissions would be reflected by variations in blast air rates, coke rates and charge metal characteristics.

Prior to any field test period the testing firm should be consulted for recommendations as to the number of days and number of test runs to be conducted to define the full range of cupola emissions consistent with cupola operating practices and other pertinent considerations. It is important that the plant's full range of operations be evaluated consistent with the stated objectives of the emission test program.

OBTAINING MEANINGFUL TEST DATA

For short run jobbing cupolas, it is recommended that a minimum of three dustloading test determinations be conducted of cupola emissions as part of any emission study. A volumetric determination should be conducted for each of the three test periods. To make the emission data be the most meaningful it is necessary and desirable that detailed records be kept of cupola operating conditions concurrent with the emission studies.

The emission test program can usually be conducted in one to three days of field sampling by an experienced testing organization. The following minimum information is considered necessary in establishing and fixing cupola operating conditions. It is necessary that these cupola operating data be secured concurrently with stack emission studies:

- 1) Nature, weight and constituents of all cupola charges.
- 2) Number and time of all cupola charges made on the test date(s).

- 3) Cupola blast air record showing volume changes during test. Verify that records indicate volume introduced but not quantities diverted as a means of throttling.
- 4) Presence of, type, number, capacity and location of afterburners.
- 5) Existence of gas ignition in the stack.

Ample precedents exist for evaluating the emission performance of only one cupola in a bank of two cupolas that are operated on alternate days. This situation is particularly valid if both cupolas are of the same size, oerate from the same tuyere blast air supply, are used in the production of similar types of iron and are operated at the same approximate rates.

If there are marked variations or changes in the operation of a 2-bank cupola system, particularly with respect to the factors outlined above, it is recommended that each cupola be evaluated individually for its emission potential. The design of a single emission control system serving a dual bank of cupolas must be predicated on achieving conformance with regulations for the most severe conditions of cupola operation during the normal production part of the melt cycle. For the larger job-shop cupola-operators and for the production foundry it is recommended that a minimum of two days field testing of cupola emissions be conducted. This type of test program will permit the operation and evaluation of both cupolas in a two unit bank.

The cupolas themselves should be operated at normal melting rates during the test period. Test dates should be selected when foundry pouring schedules will permit normal operation.

It is not necessary to obtain a gas analysis to determine gas density from the cupola because the difference in weight between air and the combustion gases is insignificant for exhaust volume calculation purposes.

SAMPLING PROCEDURES AND EQUIPMENT

A major problem in sampling and analysis is that high accuracy and precision must be obtained in a working foundry, where conditions are not conducive to laboratory-type manipulations. To achieve effective installation and operation of a sampling train in a foundry requires someone who is not overly worried with minute detail. On the other hand, when the critical analytical measurements and manipulations are made, the greatest attention to cleanliness, accuracy, and detail is required.

The sampling equipment required for this work must fit the same pattern. It must be simple, rugged, and yet capable of high accuracy. In general, it must be highly portable. Reliable equipment is available from several vendors, and all qualified testing groups have their own.

a. Filtering Media

A good filtering medium is a prerequisite to accurate sampling. Efficiency of collection must be at least 99 percent for all particulates encountered. An ideal filter medium should be very light so that accurate weight differences can be obtained from small samples. The filter should also be strong and resistant to both heat and moisture.

No medium available has all these properties so a compromise must be made. Readily available media and some of their characteristics are listed below. Reliable suppliers will give the characteristics of their products on demand.

FILTER PAPER

Conventional filter paper, made from cellulose, comes in hundreds of grades; most of them are not suited to fine particulate filtration, but some are specially designed for

this service. They have good mechanical strength, good resistance to moisture, and reasonable heat resistance. Conventional paper must be dried and desiccated before each weighing, and must weighed on a balance from which moisture can be excluded. Ideally the paper should be allowed to reach equilibrium in a constant-humidity room, and should be weighed there.

GLASS FIBER FILTER PAPER

Glass fiber filter paper will withstand higher temperature than conventional paper, but it should be remembered that a plastic binder is used in the manufacture of most of this paper and that the binder lowers temperature resistance.

Some paper is made without binder and this is much more resistant to temperature. However, this material lacks mechanical strength, and the unbonded variety is particularly weak. Glass fiber filter paper has the great advantage that it is not sensitive to humidity and so can be used where a dessicator is not available.

THIMBLES

The Soxhlet thimble has been used widely in the past. The thimbles are made of two materials, paper and ceramics. The paper thimbles have the same strengths and weaknesses as ordinary paper, and the same precautions apply. The ceramic ones come in a variety of porosities. If the pores are small enough for this work, rates of filtration will be extremely small. In addition, ceramic thimbles are very heavy so that large samples must be weighed to obtain accuracy. Thimbles of any type are not recommended for this work.

A variety of cloth materials are used for filtering particulates. Usually efficient filtration results only after a coating of particulate has been built upon the cloth. This buildup occurs most rapidly when the sampled gases contain large amounts of particulate, hence sampling error is minimized.

When particulate loading is low, such as when sampling cleaned gas, significant error can be introduced unless the fabric is 99 percent efficient on the first material that deposits.

b. Weighing

The first steps in sampling is weighing the filter paper, or other medium. Each paper should be marked with a number before weighing. The common practice of writing the weight on the paper after it has been obtained creates an error equal to the weight of the ink used. Much larger errors can result from the handling required to write on the paper.

Lastly, and most importantly, the practice is poor technique, and, if allowed, will encourage other slovenly practices.

The atmosphere in an ordinary analytical balance can be dried to some extent if a small beaker of concentrated sulfuric acid or container of silica gel is placed inside and the doors are kept closed.

If filter papers are weighed on one balance initially, and on a second when loaded, the second balance should be checked for consistency with the first. This can best be done by checking the weight of pre-weighed paper, and

applying a correction if required. Accuracy on the total weight is not vital, but the difference between initial and final weights, which represents the weight of the sample, is critical.

c. Flow Measuring

Volume flow rate measuring devices must be preceded by system components to minimize the surging or pulsating effects normal in cupola operation and sampling. The use of flow rate measuring devices in testing effluents from a dynamic system, such as a cupola, requires that frequent readings be taken (2 or 3 minutes reading cycle should be the maximum time period between readings) and that all readings must be conducted on a stopwatch timed basis.

It is recommended that sampling volume flow rate measurements be taken using two different flow measuring mechanisms in any high volume sampling train. The average of the two sampling volume rate measurements and computed sampling volumes should then be used in the subsequent dustloading calculations.

d. Flushing the Sampling Train

At the end of each dustloading test run it is imperative that the sampling train (nozzle, connecting tube or hose and sampler) be thoroughly cleaned and flushed. Distilled water should be used and introduced into the nozzle at high velocities to aid in scrubbing the sampling train.

The particulates flushed from the sampling train should be handled, weighed and separately determined. Significant quantities of particulates are deposited in any sampling train, so it is important that this material be included with the

sampled catch when computing the dustloading test results.

e. Velocity-Volumetric Tests

An S-type pitot tube is preferred to a regular pitot tube because it is not as prone to plugging. Stainless steel, Iconel or other high temperature resistant material should be used in the pitot tube construction. A ruggedly constructed inclined draft gage is recommended for use in the velocity and dustloading tests. The pitot tube should be checked and calibrated according to the manufacturers recommendations at regular intervals to establish the proper correction factor to be used in the volumetric calculations.

f. Sampling Trains and Sampling Equipment

Few emission sources offer the trying field test conditions attendant to sampling as do cupola systems. The need also cannot be emphasized enough for rugged field sampling equipment for this testing. A schematic diagram of sampling apparatus for the static balanced tube method of sampling, incorporating recommendations for cupola effluent source emission sampling is presented in Fig. 1.

A possible commercial source for various components of the sampling train is indicated in Table 2. This is not to be construed as an endorsement of any particular manufacturer but is illustrative only of the rugged type of test equipment recommended for cupola source sampling.

When assembling sampling equipment, joint sealing materials should not be exposed to the sampled gas stream where adherence of the particulate could occur. Long-radius bends should be used

instead of elbows to facilitate cleaning. The probe should be just long enough for the task at hand. The rest of the train should be assembled and tested for leaks. If the meter is a dry gas meter, it is to be calibrated before each use. If an orifice meter, or flow-meter type, is used it must also be calibrated each time, and it must, in addition, have enough sensitivity so that readings can be read to less than 1 percent. Finally, if volume is obtained by multiplying an instantaneous reading by the time of operation, fluctuations must be kept to 1 percent.

The vacuum pump or compound air ejector must be the last element of the sampling train unless it can be proved that there is no leakage through the packing, etc., under the worst conditions that can be visualized.

g. Analysis of Captured Particulate

It is recommended that the procedure for weighing and determining size distribution of the captured particulate be used as stated in the ASME PTC 27-1957 Section 4 Paragraphs 75-79 (see Appendix).

Fine particulate matter should be sized and analyzed within 24 hours after the sample is taken to minimize agglomeration and a possible change in character. It is most desirable if the sample is dried immediately after the test has been run to prevent degradation.

The minus 44 micron fraction of the collected particulate must be carefully handled and analyzed because of the strong tendency to agglomerate.

SECTION I - SAMPLING RAW PARTICULATE EMISSIONS IN THE CUPOLA STACK Recommended Test Method and Procedures

A thorough and complete review of the available test methods and procedures used in the conduct of source emission studies has resulted in the recommendation of the following basic requirements as essential to an acceptable evaluation of the test methods:

- 1) In order to obtain a truly representative sample of coarse particulates from the gas stream a large volume sampling train should be used. The sampling nozzle should be constructed of stainless steel having a minimum inside diameter of 3/4 inches, since raw cupola emissions cover a broad range of particle sizes, with individual particles not uncommonly ranging up to 3/8 inch diameter or larger.
- 2) Particulate matter is defined consistent with the definition accepted by the dust collection industry and as adopted in the American Society of Mechanical Engineers Performance Test Code 21-1941, Dust Separating Apparatus and Performance Test Code 27-1957, Determining the Dust Concentration in a Gas Stream. See item 1 in the Appendix. In essence, this defines particulate matter as all filterable solids present at standard temperature in an effluent gas stream.
- 3) It is necessary that a truly "isokinetic" sample of gases
 and solids be secured by the sampling system. This requirement
 is a practical consideration dictated by the wide range of

- particle sizes involved and, therefore, the special need for securing a truly isokinetic sample of the effluent solids.
- 4) When sampling in the cupola stack, water-cooled corrosion-resistant, sampling probes and sampling nozzles are required. This is a practical requirement since cupola temperatures in excess of 1200°F are common, and sample contamination by corrosion products formed in the nozzles and probes of the sampling system must be prevented. Water-cooling also serves to preserve the sampling probes from deterioration and distortion.
- Test Code 27-1957, Determining Dust Concentration in a
 Gas Stream, with modifications as outlined below offers the
 best and most practical test method and test procedures
 for the conduct of source emission studies from cupolas.

 The following are additional important considerations in
 the sampling of cupolas and cupola systems when utilizing as
 a broad base the test procedures and techniques embodied in
 ASME PTC 27-1957, Determining Dust Concentration in a Gas
 Stream. See item 2 in the Appendix. The criteria supplement
 the methods and procedures contained in ASME PTC 27, when
 applied to cupola source emission testing:

a. Test Location and Test Openings

A test location in a cupola stack must at best be a compromise. The location should be as far above the top of the charge door opening as practical but be at least one equivalent cupola inside daimeter below the top of the cupola stack. This location will require that protective shelter be provided since test personnel and equipment may be subjected to possible fallout of particles.

Test ports should consist of two six inch pipe nipples (schedule 40) installed radially in the cupola shell and cupola lining at 90 degrees to each other. Both 90 degree test ports must be accessible from the sheltered test platform. Six-in test ports are usually required to accommodate high volume sampling nozzles. An acceptable test platform can usually be constructed using temporary steel scaffolding. Corrugated metal sheeting can be used for the roof of the test platform. The six-inch pipe nipple test ports should protrude out a few inches from the cupola shell and should be flush with the inside of the cupola lining. The test port nipples should be fillet-welded to the cupola shell. The threads of the pipe nipples should be graphited and six-inch pipe caps installed hand tight so that they can be readily removed during the test period.

b. Method of Subdividing Cupola Stack

The cupola stack cross-sectional area should be measured at the test elevation. Due to refractory erosion and/or the

buildup of slagged deposits which affect the cupola cross section, it is important that the cupola cross section and cross sectional area be determined at the test elevation. The ASME PTC 27 test code prescribed procedure (see item 2 in Appendix) should be followed in determining the location of the test points to be used in both the volumetric or pitot tube traverses and during the test runs.

A <u>minimum</u> of 12 points should be used as sampling locations in traversing a cupola stack in the dustloading test runs. Additional sampling points should be used when the maximum to minimum velocity variation in the velocity profile approaches, or exceeds, a 2 to 1 figure.

It is important that dust sampling be conducted at each test point and that the dustloading test-data sheet reflects the sampling conditions at each test point in traverse of the cupola from each test port. The practice of using a much smaller number of test points during the dustloading test runs, as compared to a large number of points used in the velocity checks, is almost certain to bias the test results and cause the results to be of a questionable nature with respect to securing a representative cupola sample.

c. Number and Duration of Test Runs

Test runs shall consist of a minimum of 60 minutes actual dust sampling. Based upon a minimum of 12 points of dust sampling of the cupola cross section from the two 90

degree test ports, an acceptable minimum sampling schedule would consist of sampling for 5 minutes at each of the 12 points. The field test data sheets and the test report must clearly reflect the location and the time of sampling at each of the sampling points used. A minimum of three sets of flow, temperature and pressure readings should be taken at each sampling point. The field data shall be logged and should reflect the dynamic conditions of cupola flows and sampling rates at each test point.

Readings of sampling flow rates, temperatures, pressures, gas analyses and other pertinent test data which are part of each dustloading test run should be taken on a 2 (maximum 3) minute cycle at each sampling point during each dustloading test run. The total sampling program should be conducted under stopwatch timing precision.

Three dustloading test runs and 3 velocity-volumetric test runs should be conducted in a single day of field sampling, as previously mentioned.

d. Sampling Probes

Sampling probes used in the dustloading test runs of raw gas should be of water-cooled, stainless steel construction. The sampling probes should be a minimum of 3/4 inch inside diameter, and preferably of larger inside diameter for tests conducted on raw gas emissions. Conventional smaller diameter test probes are suitable for use on the downstream side of dust collectors, but should not be used in raw gas sampling.

Either a standard or null type probe may be used for sampling raw cupola gases. Whichever probe is employed, a truly isokinetic sample must be taken at all test points during all test runs.

A null sampling probe of either the balanced static pressure type or balanced impact pressure type can be used. Null type probes are prone to introduce minimum error as their diameters increase and as the velocity of the flow system increases.

A null sampling probe must be calibrated and of such a size as to give the minimum sampling error (deviation from isokinetic) for the expected sampling velocity range.

Either type probe presents certain shortcomings which must be compensated for under the adverse, dynamic and widely varying flow conditions attendant to normal cupola operation.

Cupola velocities can be expected to range from 600 to 2400 ft/min. depending upon the size of the cupola and the rate of cupola operation. Normal operating velocity ranges can be expected to be 1000 to 1800 ft/min.

Fixed rate sampling trains, based upon an occasional velocity determination made at some fixed time, are unacceptable for cupola source sampling since such methods completely ignore the dynamic nature of the cupola melting process.

e. Filter Media

Due to the need for a large diameter sample probe and the necessity of isokinetically sampling the gas stream, a high volume sampling train is mandatory. The filtering media used for removing particulate from the gas stream must be of

sufficient size to maintain the sampling rates necessary without imposing undue pressure drop restrictions on the sampling train.

The advantages and disadvantages of some of the various filtering media that can be used in removing the particulates from the sampled gas stream are stated in ASME PTC 27 Section 4 Paragraph 59 (See Appendix).

FOOTNOTE:

Cloth is often used as the filtering media because of its high collection efficiency, good flow permeability, ability to be shaped or adapted to any sampler configuration, and freedom from plugging or excessive pressure buildup under minimum condensation conditions.

In the event that a cotton sateen fabric is selected it must be thoroughly washed and rinsed prior to use to be free of starch and sizing materials. This filter medium has as its most serious limitation a humidity or moisture pickup tendency. This problem can be adequately dealt with by proper and skilled weighing and handling techniques using an enclosed single pan desiccated analytical balance.

Sampler units housing the filter medium should be made or lined with corrosion resistant material and must permit ready and free insertion and removal of the filter medium. Sampler units must consist of airtight enclosures to ensure that all sampled gases pass through the filter medium, be capable of easy field cleaning and of conserving the sampled dusts with a minimum of sample loss in filter handling.

To facilitate transfer of collected material and prevent the possibility of incandescent particles from contacting the final filter medium it may be desirable to incorporate a small stainless steel cyclonic collector ahead of the ultimate filter medium. Such cyclones tend to remove the larger particulates and prolong the sampling period before the pressure buildup on the filter medium restricts isokinetic sampling, due to reduced sampling flow rate capability.

Such cyclones offer the additional advantage of providing a convenient method of measuring the gas sampling flow rate. This can be accomplished by calibrating the pressure drop across the cyclone collector unit entailing the measurement of the pressure differential across the cyclone, the temperature and the static pressure at that location.

FOOTNOTE:

Scrubber (impinger) or condensing systems are considered unsatisfactory for particulate filtration in cupola sampling trains. Such
systems promote and cause the formation of reaction products which
were not present in the cupola gas stream. Since most available
impinger or wet collecting apparatus, are associated with low
volume sampling rates (not to exceed 1.0 cfm), it can be seen
that they do not lend themselves well to high volume rate
sampling without the use of multiple, parallel units.

While it may be of interest in some instances to determine if condensible material is present in cupola effluents such determinations are beyond the scope of this recommended practice for particulate.

When a gas analysis is desired it is recommended that a continuous carbon dioxide and/or a continuous oxygen analyzer be used to measure these gas constituents. Periodic checks can also be made using an Orsat gas analyzer to verify the performance of the continuous gas analyzer or to check on the total gas composition (CO_2, O_2, CO, N_2) . The continuous gas analyzer should be read on a two or three minute cycle throughout each test run and the time noted. Results of each Orsat gas analysis conducted should be clearly indicated on the field data sheets and in the test report.

Sulfur oxide emissions from cupola systems are of such a low order that it is usually unnecessary to measure them in light of present day standards.

f. Sampling Volume Flow Rate

The need for a high volume sampling system to secure representative samples from cupola raw gas effluents often mitigates against the use of an integrating gas meter for measuring the sample gas volume although such are available to handle the flow ranges covered by 3/4 to 2 inch inside diameter dust sampling nozzles. However, portability requirements for such meters leave much to be desired and adverse field conditions in cupola sampling often preclude the use of such meters.

Sampling volume flow rate measurements can be made by flowrator systems, calibrated pressure drop mechanisms such as orifices, venturis or other similar flow measuring devices.

SECTION II - RAW GAS TEST LOCATION IN DUCT AHEAD OF COLLECTOR

Tests are often conducted to determine performance of collectors installed for cupola gas cleaning. Often a sample location in the connecting duct will have advantages over that of a cupola stack location because -

- 1. Gases will be cooled, usually by evaporation of water, to temperatures below 500°F.
- 2. Location more accessible.
- 3. Dustloadings and gas velocity more uniform thru cross section of sampling area.

(Duct velocities usually in teh 3000 - 5000 fpm range.)
In such locations:

- a. Number of sample points can correspond to ASME PTC 27 and need not be the minimum of 12 recommended for the cupola stack.
- b. Gas volume will include substantial proportion of water vapor and influence gas density.
- c. Some dust, especially of the coarser fractions, can bypass the sample area if there is substantial runoff of cooling water or for dust fallout in cooling towers, external combustion chambers, etc.

Whenever possible, catch from collector should be obtained and checked against calculated collected quantity from inlet and outlet samples. It is often difficult to get a sample covering only the test period, but often feasible to obtain quantity collected during a complete melting cycle. In the latter case, daily average data can be compared to short test runs of the sampling equipment.

Comparison of coarse fraction in the catch with the quantities reported by sampling will also give an indication of effectiveness of the sampling technique of such fractions. When indicated, catch from the collector needs to be augmented by inclusion of fallout in preceding system elements as noted in Item c.

SECTION III - SAMPLING CLEANED CUPOLA GASES

General

Sampling behind a gas cleaner alleviates some of the problems experienced when sampling raw cupola gases. Extremely large particles are no longer present permitting the use of conventional 1/4" or 3/8" diameter sampling probes and lower sampling volumes. The violent velocity fluctuations experienced in a cupola stack have been moderated; and the high temperatures of raw cupola gases have been reduced. On the other hand, a different problem is accentuated. Gas cleaning equipment is expensive, and is usually sold to meet a specified emission standard. Since performance curves for emissions become asymptotic, small changes in performance can cause large expenditures in equipment alteration, therefore accuracy of testing becomes more critical.

Because the gas sampled is hot and humid, the probe or filter holder must be heated to stop condensation on the walls of the apparatus from occuring. Such condensate will interfere with the filtration of particulate.

Cupola off-gases are almost always cooled by direct contact with water, so it can be a-sumed that they are humid after they have passed through a cleaning device, whether a wet scrubber or not. Consequently, a condenser must be inserted in the filtering train. This serves two purposes. First, it removes excess water which may condense and damage the gas meter. Secondly, and of vital importance, a condenser gives assurance that the gas passing through the train is saturated at an identifiable point. This provides the basis for exact calculation of the volume of dry gas metered, converted to standard conditions.

An acceptable procedure for testing is "Determining Dust Concentration in a Gas Stream", PTC 27-1957, published by the American Society of Mechanical Engineers.

While isokinetic sampling is not as critical for cleaned gases because of the small particle sizes involved, its use is recommended, following the same procedure of test locations, sample time, pitot traverse and data log recommended in Section I and II.

SCHEMATIC DIAGRAM OF SAMPLING APPARATUS STATIC BALANCED TUBE METHOD OF SAMPLING

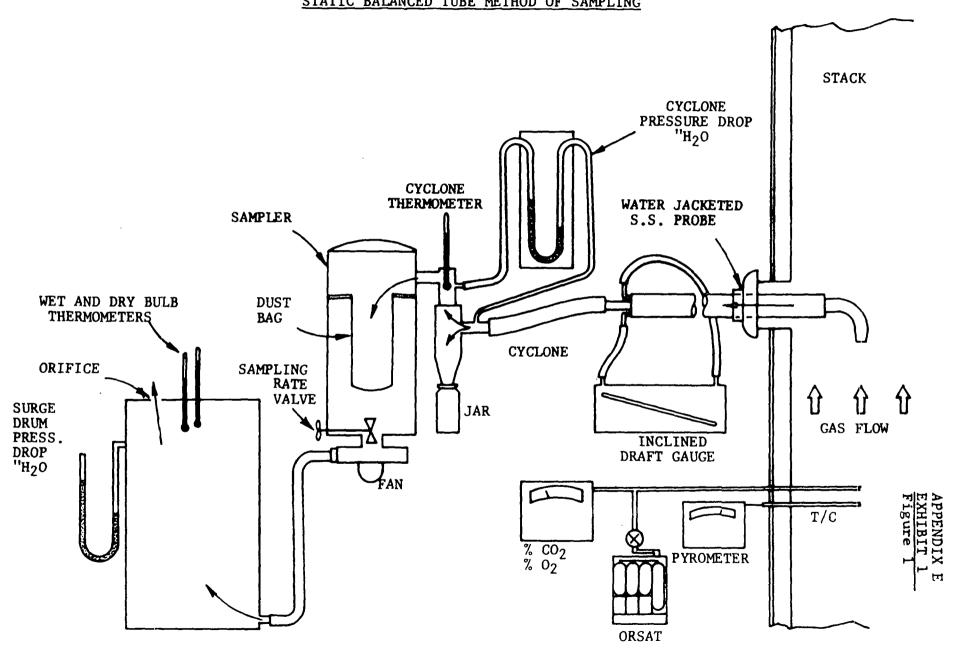


TABLE 2 - REPRESENTATIVE SOURCES OF COMMERCIALLY MANUFACTURED COMPONENTS

Component

Source

1.	Stainless steel, water-cooled
	static balanced, tube sampling
	nozzle-3/4 in. minimum ID.

Individually designed and constructed to meet nozzle diameter and probe length needs. Fitted with 6 in. pipe cap and pipe sleeve. Nozzles are to be calibrated to effect isokinetic sampling with minimum sampling error at the optimum velocity range for each different probe diameter. An acceptable nozzlehead design is schematically illustrated in ASME PTC 27 (Fig. 2).

- Inclined draft gage and holder, pitot tube.
- Industrial Engineering Instrument Co., Allentown, Pennsylvania
- 3. Stainless steel cyclone

UOP Air Corrections Division Darien, Connecticut

4. Sampler

Fabricate to meet filter media confinement and handling requirements.

5. Manometer

The Meriam Instrument Co., Cleveland, Ohio

6. Thermometer

Weston Electrical Instrument Corp., Newark, New Jersey

7. Industrial exhauster

Clements Manufacturing Co., Chicago, Illinois

8. Continuous gas analyzer

Thermco Instrument Corp., LaPorte, Indiana

9. Orsat gas analyzer

Hayes Corp., Michigan City, Indiana

10. Pyrometer and thermocouple

Alnor Instrument Co., Division Ill. Testing Laboratories, Inc. Chicago, Illinois

SAMPLING AND ANALYTICAL TECHNIQUES

INTRODUCTION

Sampling and analytical techniques for the determination of emission rates from industrial processes have been standardized for many specific particulate and gaseous materials. The techniques described in the following paragraphs are those most widely used in the testing of iron foundry emissions testing. The format and wording for most procedures correspond to the source indicated for each procedure.

SAMPLING TECHNIQUE

Scope

The primary objective of stack testing is to determine the nature and/or quantity of emissions being released into the atmosphere. Sampling procedures that follow are applicable to the cleaned gas side of the control unit.

Apparatus

The accuracy of emission testing results is dependent upon qualified personnel conducting the test and the use of the proper apparatus for the material to be collected. Figure 1 illustrates information on sampling locations and apparatus most commonly involved in stack testing.

Sampling Principles

The location and number of sampling points are based on size and shape of the duct, uniformity of gas flow in the duct,

availability of an adequate sampling port, and the space required to set up the equipment. Unfortunately, ideal conditions are seldom found in field testing and agreement on these factors must be reached before conducting the test.

To insure constancy of test conditions and results, complete information must be developed as to continuous or cyclic operation; nature, weight and composition of materials; gas volume and fluctuations; pressure; temperature and humidity; presence of other devices such as afterburners; and related conditions affecting the operation and equipment. These factors will regulate the time, number and duration of test runs.

Stack Gas Velocity

To determine particulate concentration in an exhaust stack, isokinetic source sampling must be used. This is the condition that exists when the velocity in the nozzle of the sampling tube is exactly the same as that in the stack. Isokinetic sampling is not mandatory when only gaseous substances are to be assayed.

In isokinetic sampling, the traverse area of the duct must be divided into equal areas and a pitot traverse taken. The use of the S-type pitot is recommended where particulates are involved to avoid any possibility of partial plugging and faulty readings. The velocity at each point must be calculated, and the volume of flow required to maintain that

velocity in the sampling tip should volume fluctuate. Provisions must be made so that the volume can be recalculated immediately each time the pressure changes at the meter. However, when sampling is downstream from a gas cleaner, the volume is controlled by the system's fan and remains relatively constant and this procedure may not be necessary.

Detailed procedures on conducting velocity measurements are given in Bulletin WP-50 of the Western Precipitation

Company, ASME Performance Test Code 27-1957 and the Industrial Ventilation Manual of the American Conference of Governmental Industrial Hygienists.

Concurrent with conducting the pitot traverse, it is essential to determine the temperature of the stack gas. The measuring device will be dependent on the temperatures involved.

Sample Probe

In assembling the sampling probe, teflon tape should always be used instead of pipe dope to prevent adherences of particulates. Long radius bends should be used instead of elbows to facilitate cleaning. The probe should be just long enough for the task at hand. The rest of the train should be assembled and tested for leaks.

Temperature and Humidity

If the gas sampled is hot and humid, condensation may occur in the probe or in the filter holder. The probe or filter holder must be heated to stop condensation from occurring

because the water formed will trap water on the walls of the apparatus and will interfere with the filtration of particulates. Temperature control baths may be required for gas absorbers. In some cases the probe can be provided with a water cooling jacket.

Condensation

A condenser in the sampling train is required if the gas is humid. This serves two purposes. First, it removes excess water which may condense and damage the gas meter. Second, and of vital importance, a condenser gives assurance that the gas passing through the train is saturated at an identifiable point. This provides the basis for exact calculations of the volume of dried gas metered and conversion to standard conditions.

Collection Devices

The characteristics of the material in the stack will determine the collection method required. Dry filter mediums, of a variety of types, are most commonly used for particulate matter. Although in some cases the wet impingement method followed by a thimble is used. Gases are collected in absorbers with a proper absorbing solution. Grab sample units are available for spot sampling.

Flow Meters

If a dry gas meter is used, it must be calibrated before each use. If an orifice meter, or flow-type meter, is used

it must also be calibrated each time, and it must have enough sensitivity so that readings can be obtained to less than one percent. Finally, if volume is obtained by multiplying an instantaneous reading by the time of the operation, fluctuations must be kept to one percent.

Vacuum Source

A vacuum source is required to draw the sample from the stack through the sampling train. A variety of pumps or ejectors are available for this purpose. Their capacity must be sufficient to draw the gas through the sample train at the required volume. The range is from one liter to several cubic feet per minute.

Sampling time will be dependent upon the factor of obtaining a representative sample of the operation. It may vary from several long continuous integrated samples of 30 to 60 minutes or a number of short samples of 5-10 minutes.

ANALYTICAL PROCEDURES

Introduction

Analytical procedures for a number of materials are given in the sections that follow. All calculations must be according to standard procedures and the standard conditions of temperature at 70 degrees Fahrenheit and an atmospheric pressure of 29.92 inches of mercury.

Particulate Matter

(a) Scope

The definition of particulate matter accepted by the dust collection industry is given in the ASME Performance Test Codes 21-1941 and 27-1957. In essence, this defines particulate matter as all filterable solids present at standard temperature in an effluent gas stream.

(b) Auxiliary Apparatus

- Filter Media Efficiency of collection must be at least 99% for all particulates encountered and must be resistant to both heat and moisture.
- Balance Macro analytical balance or equivalent.
- Drying Oven Suitable for drying filters for about 5 hours at 105° C.
- Desiccation To retain dried filters before weighing.

(c) Sampling Procedure

The first step in sampling is to prepare the filtering medium. An identification number should be provided for each filter and recorded on a separate data sheet. Prior to weighing, the filter should be dried for about 5 hours at about 1050 C and then weighed immediately. This weight should be recorded on the data sheets and not on the filter. In order to keep weighing errors at a minimum, careful handling of the filters is required.

Preferably the pitot traverse, temperature and humidity readings should be taken not more than one-half hour before sampling is begun. Assemble the sampling train as shown in Figure 1 and proceed with the sampling by inserting the probe into the test stack. Continual observation of the sampling train during the entire sampling period is required to record any changes in pressure, temperature and airflow. This information, along with barometric pressure, sampling time and rate, is recorded on the sampling data sheet. Complete information on the process should also be noted on the sampling data sheet.

Length of the sampling time, at any specific point in the stack, will be contingent upon changes, if any, in the process or fluctuations of air volume. The sampling time should at least cover a complete cycle and will vary from 30-60 minutes. If airflow is not uniform in the stack, 5- to 10- minute samples at each of the traverse points should be obtained. Samples taken during start-up and burn-down periods should, as a rule, be considered separately from those taken during the production cycle of the cupola.

After a run is completed the probe must be cleaned of retained particulate matter. An acceptable procedure is to brush with a long flexible brush while the sample train is pulling in clean air. For other contaminants, follow procedures, if any, indicated for the specific material.

(d) Sample Preparation

Collected samples should be dried and placed in a desiccator to reach equilibrium before weighing. The difference between the original weight and final weight is the total amount of particulate matter collected.

(e) Calculations

The total particulate matter collected is expressed in grams. From this value, calculations can be made to express the findings in grains/SCF, pounds/hour, or pounds/1,000 pounds of gas, using the following constants:

One (1) gram = 15.43 grains
One (1) pound = 7,000 grains
One (1) gram = 0.002205 pounds
One (1) standard cubic foot of air = 0.075 pounds

1. Grains/SCF

$$\frac{\text{Grains/SCF}}{\text{Total SCF sampled}}$$

2. Pounds/Hour

Pounds/hour = 60 (grains/SCF) (total gas volume to atmosphere - SCFM) 7,000

3. Pounds/1,000 Pounds Gas

Pounds/1,000 Pounds gas =
$$\frac{\text{(grams) (2.205)}}{\text{(0.075) (total SCF sampled)}}$$

Arsenic

Source: American Conference of Governmental Industrial Hygienists.

(a) Scope

Stack sampling for arsenic is based on the reaction of arsine with silver diethyldithiocarbamate. The amount of arsenic, in the air sample, is read directly from the calibration curve.

(b) Auxiliary Apparatus

- Greenberg-Smith Impinger.
- Beckman DU Spectrophotometer with photomultiplier or equivalent
- Arsine Generator (See Figure 2)

(c) Reagents

Silver Diethyldithiocarbamate - a cooled solution of silver nitrate (1.7 g in 100 ml distilled water) is added to a cooled solution of sodium diethyldithiocarbamate (2.25 g in 100 ml distilled water). The lemon yellow precipitate is filtered off, washed thoroughly with distilled water and dried in a vacuum desiccator below 20° C.

Pyridine - Mallinckrodt reagent grade pyridine is passed through an alumina column 1 inch in diameter and 6 inches in depth, at the rate of approximately 150 ml per hour. This process may remove a considerable quantity of colored material.

Arsine Absorbing Solution - Dissolve 1 g of silver diethyldithiocarbamate in 200 ml of chromatographed pyridine and filter the solution.

Hydrochloric acid - Baker's analyzed, specific gravity
1.19.

Potassium Iodide Solution - Dissolve 15 g reagent grade potassium iodide in 100 ml distilled water.

Stannous Chloride Solution - Dissolve 40 g stannous chloride dihydrate in 100 ml hydrochloric acid.

Zinc - Baker's analyzed; granular 20 mesh.

Lead Acetate - Dissolve 10 g reagent grade lead acetate in 100 ml distilled water.

Arsenic Standard Stock Solution - Dissolve 1.320 g arsenic trioxide in 10 ml of 40% sodium hydroxide and diluted to 1 liter with distilled water. (Various strengths of standard solutions are prepared by further diluting this stock solution with suitable volumes of water, triple distilled in glass.)

Nonag - Stopcock grease, Fischer Scientific Co.

(d) Sampling Procedure

Assemble sampling train of probe, impinger with 100 ml of distilled water, flow meter and vacuum pump. Sampling rate is at 1 CFM for a period long enough to provide a minimum of 30 cubic feet at standard conditions.

(e) Analytical Procedure

Calibration curve - known microgram amounts of arsenic (1-15 micrograms) in the form of standard arsenic solution are pipetted into 125 ml Erlenmeyer flasks. Distilled water is added to make the total volume 35 ml. To the flasks are added 5 ml hydrochloric acid, 2 ml 15% potassium iodide solution, and 8 drops of stannous chloride solution. The flasks are swirled and allowed to stand for 15 minutes.

Three ml of the pyridine solution of silver diethyldithiocarbamate are placed in the absorbing tube, which is attached to the scrubber containing glass wool impregnated with lead acetate. (See Figure 2.)

The ground joints are lubricated with "Nonag" stopcock grease, 3 g of granulated zinc are added to the solution in the flask, and the receiving tube is inserted immediately. Arsine evolution is completed in about 30 minutes.

At the end of this time, the absorbing solution is transferred to a 1 cm square cell and the absorbance measured at 560 millimicrons in the Beckman spectrophotometer. Plotting measured absorbances against micrograms of arsenic taken produces the standard curve.

Air samples, after the previously described preparation treatment, are treated in the same manner as the standards.

(f) Calculations

Arsenic, in the form of arsine, displaces an equivalent amount of silver from silver diethyldithiocarbamate.

$$\begin{array}{lll} \text{mg As/M}^3 &=& \underbrace{\text{V·Y}} \\ \hline 1,000\cdot\text{v·Va} \\ \\ \text{Where} & \text{v} = & \text{aliquot (ml)} \\ \text{V} = & \text{total sample (ml)} \\ \text{Y} = & \text{micrograms in v} \\ \text{Va} = & \text{gas sample volume, in cubic meters,} \\ \text{at standard conditions} \\ \end{array}$$

Beryllium

Source: Michigan Department of Public Health.

(a) Scope

This method describes a procedure for determining beryllium in stack gases.

(b) Auxiliary Apparatus

- Millipore filters and holder.
- Bausch & Lomb Large Littrow Emission Spectrograph or equivalent.

(c) Reagents

Platinum Internal Stock Solution - Purchase directly from Jarrell-Ash Company a 10% platinic chloride solution. This calculates out to be 57.88 mg platinum in 1 ml solution.

Platinum Internal Standard Working Solution - Pipette 1 ml of platinum stock solution containing 57.88 mg Pt per ml into a

25 ml volumetric flask, take to volume with water giving a solution containing 116 micrograms platinum/.05 ml.

Standard Beryllium Solutions:

- 1. <u>Beryllium stock solution</u>. Dissolve .0982 g of BeS04·4H₂0 in 10 ml of redistilled 1:1 hydrochloric acid and dilute to 100 ml with distilled water. Solution contains 5.0 mg beryllium per 100 ml or 2.5 micrograms Be/.05 ml.
- 2. Working beryllium standard solutions. These should be prepared from the stock solution just before use. Suggested concentrations are from .003 to .5 microgram Be/.05 ml.

Nitric Acid - To clean all laboratory glassware.

(d) Sampling Procedure

Assemble sampling train of probe, millipore filter and holder, flow meter and vacuum pump. Sampling rate at 1 CFM for a period long enough to provide a minimum of 10 CF at standard conditions.

(e) Analytical Procedure

The millipore filter containing the sample is transferred to a chemically clean 125 ml beaker. The filter and sample are wet ashed with nitric acid. The residue is then dissolved in 3 ml of concentrated nitric acid and 1-2 ml of distilled water. Transfer to a graduated centrifuge tube, rinse the beaker with water and add the rinsing to the sample solution. Evaporate to

a volume of 0.2 ml and if an appreciable amount of salt is present, a volume of more than 0.2 ml may be required.

The standard curve is plotted on log-log paper and micrograms Be per .05 ml is plotted versus the intensity ratio of Be 2348.6 line over Pt 2357.1 line. The standard curve is usually set up in the range of .003 microgram Be/.05 ml to .5 microgram Be/.05 ml. Six beryllium concentrations used to establish the working curve are prepared as follows:

For the first 3 concentrations, the stock solution containing 50 micrograms Be/ml is diluted 1 ml to 100 in distilled water giving a working solution of .5 microgram Be/ml.

- 1. .003 microgram Be/.05 ml. Pipette 1.2 ml of working standard beryllium solution (.5 microgram Be/ml) into a 10 ml volumetric flask and take to volume with water.
- 2. .005 microgram Be/.05 ml. Pipette 2 ml of working standard beryllium solution (.5 microgram Be/ml) into a 10 ml volumetric flask and take to volume with water.
- 3. .01 microgram Be/.05 ml. Pipette 4 ml of working standard beryllium solution (.5 microgram Be/ml) into a 10 ml volumetric flask and take to volume with water.
- 4. .05 microgram Be/.05 ml. Pipette .2 ml of stock beryllium solution (50 micrograms Be/ml) into a 10 ml volumetric flask and take to volume with water.

- 5. .1 microgram Be/.05 ml. Pipette .4 ml of stock beryllium solution (50 micrograms Be/ml) into a 10 ml volumetric flask and take to volume with water.
- 6. .5 microgram Be/.05 ml. Pipette 2 ml of stock beryllium solution (50 micrograms Be/ml) into a 10 ml volumetric flask and take to volume with water.

Spectrographic apparatus, materials and exposure conditions are as follows:

- 1. Optical conditions 10 micron slit is used in the spectrograph.
- 2. Densitometer Non-recording National Spectrograph Spec Reader.
- 3. Electrodes Upper Electrode (cathode) United Carbon Products Company, 3/16" diameter, sharpened to a point in a regular de-leaded pencil sharpener. Lower Electrode (anode). United Carbon Products Electrode, catalog No. 100-L, 1/4" diameter, crater is 3/16" diameter and 5/32" deep.
- 4. Exposure conditions 220 volts DC arc, operating at 7.5 amperes with a constant gap of 5 mm maintained between the anode and cathode, exposure time is until burn-out of lithium chloride buffer.
- 5. Photographic Eastman Kodak Spectrum Analysis
 No. 1 Plate, developed 3.5 minutes in Eastman D-19 Developer
 at 68° F and fixed for 8 minutes in Eastman Koda Fixer (National
 Spectrographic Developing machine). Emulsion is calibrated by
 use of the two-step filter in front of the slit. The density of

the filter section is given by Bausch and Lomb Company, makers of the filter.

6. Nitrogen - AirCo dry nitrogen, flow rate regulated by F. W. Dwyer Manufacturing Company flow meter, maximum flow rate 6 liters per minute, regulator 3,000 pounds. The nitrogen flow around the electrode is between 3-4 liters per minute.

Preparation of the electrodes for both standard curve and sample analysis is as follows: A 1/4" diameter electrode is waterproofed by immersion in Dow Corning silicone solution (2% in acetone), and air dried for at least 30 minutes. A 10 mg charge of lithium chloride-graphite buffer is placed in the electrode and packed by tapping gently on the table top.

Into the electrodes prepared as described above is pipetted .05 ml of the platinum internal standard working solution (116 micrograms/.05 ml). The electrodes are placed in a 60° C oven and allowed to dry. Upon removal from the oven, .05 ml of the standard beryllium solution is pipetted into the appropriate electrodes. From the centrifuge tubes, where the samples have been evaporated down, is pipetted .05 ml into the appropriate electrodes. The electrodes are then returned to the 60° C oven and maintained at that temperature until dry. The temperature is then brought up to 105° C and maintained at that temperature for 1 hour. The electrodes are now removed from the oven and are ready for analysis. After the spectrograph and power

supply have been set as previously described, the electrodes are placed in the respective electrode holders. The nitrogen flow is turned on and set at a rate of between 3-4 liters per minute around the lower electrode. With the shutter open during the entire exposure the arc is lit and allowed to run until burn-out of the lithium chloride buffer which is indicated by a vanishing of the red lithium color.

After the plate has been developed and dried as described previously, it is placed on the densitometer and the percent transmission set to 100 on a clear portion of the plate. The percent transmittance value of Be 2348.6 and the background adjacent to this line is read. The percent transmittance value of Pt 2357.1 line is also read. Through the use of the gamma curve the percent transmission values of the bismuth line and the background adjacent to it and the Pt line are transformed to I values and a ratio taken of I value Be 2348.6 over I value Pt 2357.1 made. Each one of the varying concentrations of beryllium standard curve and of the sample is run in triplicate and an average of these taken for the final calculation. The amount of beryllium per .05 ml sample is read from the standard curve.

(f) Calculation

micrograms
$$Be/M^3 = \frac{V \cdot Y}{V \cdot Va}$$

where $V = \text{aliquot (ml)}$
 $V = \text{total sample (ml)}$
 $Y = \text{micrograms in } V$
 $Va = \text{gas sample volume, in cubic meters,}$

at standard conditions

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Cadmium

Source: Michigan Department of Public Health.

(a) Scope

Stack testing for cadmium can be accomplished by the polarograph method using a dropping-mercury electrode with the sample as the electrolyte.

(b) Auxiliary Apparatus

Sargent Polarograph - Model XXI, recording type or equivalent.

(c) Reagents

Standard Lead Solution - Dissolve approximately 25 grams of C.P. Pb(NO₃)₂ in minimum of hot water and cool with stirring. Filter with suction on small Buchner funnel. Repeat recrystallization. Dry crystals at 100°-110° C to constant weight, cool in desiccator and store in tightly stoppered pyrex bottle. The product has no water of crystallization and is not appreciably hygroscopic. Weigh exactly 0.1599 grams of recrystallized C.P. Pb(NO₃)₂, put into 500-ml volumetric flask, and take to volume with 0.1 N HCl. This gives a standard lead solution containing 200 micrograms Pb/ml with 0.1 N HCl as the electrolyte. The 0.1 N HCl should be prepared from constant boiling hydrochloric acid.

Standard Cadmium Solution - Weigh exactly 0.2744 grams of Cd(NO₃)₂·4H₂0 into a 500-ml volumetric flask and take to volume with 0.1 N HCl. This gives a standard cadmium solution containing 200 micrograms cadmium per ml with 0.1 NHCl as the electrolyte. As in the lead solution the 0.1 N HCl should be prepared from constant boiling hydrochloric acid.

Oxygen Absorbent for Purification of Nitrogen - Pass nitrogen through a first scrubbing flask (a midget impinger) containing concentrated NH40H and copper turnings. Caution:

Make certain hole in impinger is not plugged before turning nitrogen under pressure on. Then pass nitrogen through a second scrubbing flask containing concentrated sulfuric acid, again making certain this is not plugged before applying pressure.

0.2 N hydrochloric acid - Prepare this from constant boiling hydrochloric acid according to outline in Lange's Handbook.

Clean, Dry Mercury - Purchase from Eberback & Son

(d) Sampling Procedure

Assemble sampling train of probe, impinger with 100 ml of 5% nitric acrid, flow meter and vacuum pump. Sample at rate of 1 CFM for a period long enough to provide a minimum of 30 cubic feet at standard conditions.

(e) Analytical Procedure

Sample Preparation - Transfer the collecting solution from the impinger into a 250 ml beaker, wash out impinger with hot 5% nitric acid and all taken down to dryness on a hot plate. Cool and add 25 ml of 0.2 N HCl. Heat just to boiling and transfer to a 50 ml volumetric flask. Dilute to volume with distilled water which will dilute the 0.2 N HCl to 0.1 N HCl which is the electrolyte.

Transfer a 10-ml aliquot from the 50-ml volumetric flask into the polarographic cell, add 1 ml of 200 micrograms Pb per ml solution, and remove oxygen from the cell by bubbling nitrogen, which is being purified as described under reagents, through for three to five minutes. The initial voltmeter is set at .3 volts, the span voltmeter is set at .6 volts, thereby giving a range from -.3 volts to -.9 volts. This is sufficient as lead "comes off" at approximately -.44 volts and cadmium at approximately -.66 volts. The sensitivity setting might have to be found by trial and error; 0.020 suffices for most samples although if the cadmium is low the sensitivity will have to be increased (decreasing the number of microamperes/mm.).

If there is a possibility that Pb is present in the sample an aliquot of the sample should be run in the polarographic cell first, without any internal standard added. If there is Pb present in the sample, this must be taken into account when Pb, the internal standard, is added.

Standard Curve - Into the polarographic cell is introduced 1 ml of 200 micrograms Pb per ml solution, 1 ml of 200 micrograms Cd per ml solution and 9 ml of 0.1 N HCl. This gives a total amount of solution in the cell of 11 ml, thereby enabling a later removal of 10 ml of the sample and 1 ml of 200 micrograms Pb per ml internal standard solution. Also, there is an electrolyte in the cell of 0.1 N HCl. Both the volume of liquid in the cell and the electrolyte for standard curve and sample are critical for a proper analysis.

On the standard curve the heights of the Pb and Cd curves are measured in mm. The Cd to Pb ratio is found, which is divided by the number of micrograms of Cd used giving a factor for 1 microgram Cd versus 200 micrograms Pb. It is suggested that 200 micrograms Pb be used as an internal standard in each sample for Cd thereby simplifying the calculations. The factor for 1 microgram Cd versus 200 microgram Pb, found at the beginning of the series of samples being analyzed, will be used for the calculations throughout this series.

(e) Calculations

For the sample "polarogram" the heights of the Pb and Cd curves are measured in mm. and the Cd to Pb ratio found in the same manner as the standard curve. The ratio found here is divided by the factor found in the standard curve for 1 microgram Cd versus 200 micrograms Pb giving the number of micrograms of Cd in the aliquot put into the polarographic cell.

$$mg \ Cd/M^3 = \frac{V \cdot Y}{1,000 \cdot v \cdot Va}$$

Where v = aliquot (m1)

V = total sample (m1)
Y = micrograms in v

Va = gas sample volume, in cubic meters,
 at standard conditions

at Standard Condition

Fluoride

Source: Talvitie method modified by Michigan Department of Public Health.

(a) Scope

This method describes a procedure for determining fluoride in stack gases.

(b) Auxiliary Apparatus

- Standard impinger with fritted glass bubbler.
- 250 ml Claissen flasks.
- 100 ml Nessler Tubes.

(c) Reagents

Standard Sodium Fluoride - Make a solution containing 1 mg of fluoride per ml (2.21 g of sodium fluoride to 1 liter).

Take 10 mls of this solution and dilute to 1 liter; 1 ml of this dilution contains .01 mg fluoride.

Color Forming Reagent - Dissolve 36.99 g of sodium sulfate in about 500 ml of hot distilled water and 17.7 g of sodium formate in about 200 ml of hot distilled water. Mix together and when cooled, add 0.1436 g thorium nitrate tetrahydrate and 11 ml of 90% formic acid.

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Alizarin monosodium sulfonate indicator 128.25 mg dissolved in 1 liter of distilled water.

Nitric Acid - About 5 ml concentrated acid, diluted to a liter with distilled water.

Sodium Hydroxide - .5 N. (20 g dissolved in 1 liter of water).

Silver Sulfate.

Concentrated Sulfuric Acid.

(e) Sampling Procedure

Assemble sampling train of probe, impinger with fritted glass bubbler containing 100 ml of a 2% sodium hydroxide solution, flow meter and vacuum pump. Sample at a rate of 1 CFM for a period long enough to provide a minimum of 15 cubic feet at standard conditions.

(f) Analytical Procedure

Sample Preparation - Transfer the collecting solution from the impinger into a Claissen flask. Slowly add 35 ml of concentrated sulfuric acid (using small long stem funnel) to content, submerging and swirling flask in cool-cold water while adding the acid--this offsets the loss of HF. Add boiling chips and silver sulfate (to cover the end of a spatula). Close the flask with a two-hole rubber stopper, through which passes a thermometer and a 6 mm O.D. glass tube drawn to

capillary size and extends down into the solution. Connect tube to a separatory funnel containing water. This is to slowly add water to both cool the flask and to replenish the water boiled off due to distillation in the Claissen flask.

The distillation flask should be placed on a pad of transite or asbestos, or on a plate of aluminum with a hole about 2 inches in diameter made to fit the flask perfectly.

Regulate the heat under the steam distillation flask so that the distillate being collected remains cool. Adjust the application of heat to the still so that a temperature of 165° C is maintained. Collect the distillate in a 250-m1 volumetric flask or in a 250-ml beaker, and then make up to exactly 250 ml in a volumetric flask. Stopper the flask and mix. Pipette 25 ml into a 100-ml-long form Nessler tube. Add 5.0 ml of alizarin indicator. Titrate carefully with a .5 N sodium hydroxide until the solution changes from yellow to a decided pink. titrate with the dilute nitric acid until the solution changes to a pure yellow. Dilute to about 90 ml, add 3 ml of thorium reagent, make up to exactly 100 ml and mix well. After 30 minutes, compare with the standards. If the same is beyond the range of the standards, use a smaller aliquot. If it is too close to the standard containing no fluorine, double or treble the aliquot.

A blank must be carried through all the steps of the procedure, using the same amounts of reagents as are used in the samples. An aliquot of 75 ml is usually necessary to determine the amount of fluorine present in the blank.

(f) Calculations

Calculate the total amount of fluorine present in the blank and subtract this from the total fluorine found in each sample.

$$\begin{array}{ll} \text{mg } F/M^3 = & \frac{V \cdot Y}{1,000 \cdot v \cdot Va} \\ \\ \text{where } v = \text{aliquot } (\text{ml}) \\ V = \text{total sample } (\text{ml}) \\ Y = \text{micrograms in } v \\ Va = \text{gas sample volume, in cubic meters,} \\ \text{at standard conditions} \end{array}$$

Lead

Source: Michigan Department of Public Health.

(a) Scope

Stack testing for cadmium can be accomplished by the polarograph method using a dropping-mercury electrode with the sample as the electrolyte.

(b) Auxiliary Apparatus

Sargent Polarograph - Model XX1, recording type, or equivalent.

(c) Reagents

Standard Lead Solution - Dissolve approximately 25 grams of C.P. $Pb(NO_3)_2$ in minimum of hot water and cool with stirring. Filter with suction on small Buchner funnel. Repeat

recrystallization. Dry crystals at $100^{\circ}-110^{\circ}$ C to constant weight, cool in desiccator and store in tightly stoppered pyrex bottle. The product has no water of crystallization and is not appreciably hygroscopic. Weight exactly 0.1599 grams of recrystallized C.P. Pb(NO₃)₂, put into 500-ml volumetric flask, and take to volume with 0.1 N HCl. This gives a standard lead solution containing 200 micrograms Pb/ml with 0.1 N HCl as the electrolyte. The 0.1 N HCl should be prepared from constant boiling hydrochloric acid.

Standard Cadmium Solution - Weight exactly 0.2744 grams of $Cd(NO_3)_2 \cdot 4H_20$ into a 500-ml volumetric flask and take to volume with 0.1 N HCl. This gives a standard cadmium solution containing 200 micrograms cadmium per ml with 0.1 N HCl as the electrolyte. As in the lead solution, the 0.1 N HCl should be prepared from constant boiling hydrochloric acid.

Oxygen Absorbent for Purification of Nitrogen - Pass nitrogen through a first scrubbing flask (a midget impinger) containing concentrated NH40H and copper turnings. Caution: Make certain hole in impinger is not plugged before turning nitrogen under pressure on. Then pass nitrogen through a second scrubbing flask containing concentrated sulfuric acid, again making certain this is not plugged before applying pressure.

0.2 N. Hydrochloric Acid - Prepare this from constant boiling hydrochloric acid according to outline in Lange's Handbook.

Clean, Dry Mercury - Purchase from Eberbach and Son.

(d) Sampling Procedure

Assemble sampling train of probe, impinger with 100 ml 5% nitric acid solution, flow meter and vacuum pump. Sample at rate of 1 CFM for a period long enough to provide a minimum of 30 cubic feet at standard conditions.

(e) Analytical Procedure

Sample Preparation - Transfer the collecting solution to a 250-ml beaker, wash out impinger with 5% hot nitric acid and all taken down to dryness on a hot plate. Cool and add 25 ml of 0.2 N HCl. Heat just to boiling and transfer to a 50-ml volumetric flask. Dilute to volume with distilled water which will dilute the 0.2 N HCl to 0.1 N HCl which is the electrolyte.

Transfer a 10-ml aliquot from the 50-ml volumetric flask into the polarographic cell, add 1 ml of 200 micrograms Cd per ml solution, and remove oxygen from the cell by bubbling nitrogen which is being purified as described under reagents, through for three to five minutes. The instrument used is a Sargent Polarograph - Model XXI and the settings are as follows: A.C. switch down (on), D.M.E. - up (negative), Damping - down (off), Initial E.M.F. - up (additive), D.C. E.M.F. - down (1.5 V span), Chart drive - up (on), Operation - up (E.M.F. increasing). The initial voltmeter is set at .3 volts, the span voltmeter is set at .6 volts, thereby giving a range from -.3 volts to -.9

volts. This is sufficient as lead "comes off" at approximately -.44 volts and cadmium at approximately -.66 volts. The sensitivity setting might have to be found by trial and error, 0.020 suffices for most samples although if the lead is low the sensitivity will have to be increased (decreasing the number of microamperes/mm).

If there is a possibility that Cd is present in the sample, an aliquot of the sample should be run in the polarographic cell first, without any internal standard added. If there is CD present in the sample this must be taken into account when Cd, the internal standard, is added.

Standard Curve - Into the polarographic cell is introduced 1 ml of 200 micrograms Pb per ml solution, 1 ml of 200 micrograms Cd per ml solution and 9 ml of 0.1 N HCl. This gives a total amount of solution in the cell of 11 ml thereby enabling a later removal of 10 ml of the sample and 1 ml of 200 micrograms Cd per ml internal standard solution. Also, there is an electrolyte in the cell of 0.1 N HCl. Both the volume of liquid in the cell and the electrolyte for standard curve and sample are critical for a proper analysis.

On the standard curve the heights of the Pb and Cd curves are measured in mm. The Pb to Cd ratio is found, which is divided by the number of micrograms of Pb used giving a factor for 1 microgram Pb versus 200 micrograms Cd. It is suggested that 200 micrograms Cd be used as an internal standard in each sample for Pb thereby simplifying the calculations. The factor

for 1 microgram Pb versus 200 micrograms Cd, found at the beginning of the series of samples being analyzed, will be used for the calculations through this series.

(f) Calculations

For the sample "polarogram" the heights of the Pb and Cd curves are measured in mm and the Pb to Cd ratio found in the same manner as the standard curve. The ratio found here is divided by the factor found in the standard curve for 1 microgram Pb versus 200 micrograms Cd giving the number of micrograms of Pb in the aliquot put into the polarographic cell.

Mercury

Source: American Conference of Governmental Industrial Hygienists.

(a) Scope

Divalent mercury forms an orange-yellow complex with dithizone in dilute acid solution which can be extracted by chloroform. An additional extraction in the presence of chloride and bromide ions eliminates the interference of other metals.

(b) Auxiliary Apparatus

- Beckman DU Spectrophotometer or equivalent.
- Squibb separator funnels.
- Cuvettes.

(c) Reagents

HC1-0.1 N.

Meta Cresol Purple Indicator - Dissolve 0.05 g of the power in 6 ml of 0.05 N NaOH; then dilute to 100 ml with distilled water.

Buffer Solution - Dissolve 300 g anhydrous Na_2HPO_4 and 75 g K_2CO_3 in distilled water to make 2 liters of solution (MacIlvaine's Buffer Solutions).

Treated Chloroform - Chloroform treated with hydroxylamine hydrochloride as per the method of Hubbard, Industrial Engineering Chemistry, Anal. Ed., 9, 493 (1937).

Dithizone Solutions - Make up a stock solution containing 0.5 mg dithizone per ml of chloroform. Other strength dithizone solutions can be made up as needed. It is advisable to allow the dithizone solutions to stabilize overnight before use.

Potassium Bromide Solution - 40% KBr in distilled water.

Ammonium Citrate - 40%. Mix 40 g citric acid, monohydrate, with about 20 ml distilled water. Add sufficient ammonium hydroxide slowly with constant stirring to make solution

alkaline to phenol red and make to volume with water. Purify by shaking with dithizone in chloroform and clear with pure chloroform.

Mercury Standard Solutions - Dissolve 0.1354 g mercuric chloride, C.P., special reagent grade in 1 N HCl and make up to 100 ml with the acid. This solution contains 1 mg Hg per ml and is quite stable. If any cloud or sediment develops, it should be discarded. Other strength solutions can be made by dilution with distilled water as the need arises.

Hydroxylamine Hydrochloride - 20% solution in distilled water.

(d) Sampling Procedure

Assemble sampling train of probe, impinger with 100 ml of 0.25% iodine in a 3% aqueous solution of potassium iodide.

Sampling rate of 1 CFM for a period long enought to provide a minimum of 30 cubic feet at standard conditions.

(e) Analytical Procedure

Sample Preparation - The contents of the impinger flask and washings are made up to a known volume with distilled water. A proper aliquot is taken to place the mercury concentration within range of the method. Add 5 ml of ammonium citrate, 1 ml hydroxylamine hydrochloride and shake. Add 2 drops of phenol red indicator. (Always add the hydroxylamine hydrochloride before the phenol red.) Titrate with ammonium

hydroxide to the full color end point pH of 8.5. Extract with 5 ml portions of 20 mg/liter dithizone solution, withdrawing the chloroform layers into another 250 ml Squibb separatory funnel, into which has been placed 50 ml of 0.1 N HCl. Continue to extract with and withdraw 5-ml portions until the dithizone in the chloroform layer does not change color.

Shake the above dithizone extract with 50 ml 0.1 N HCl for 2 minutes. Draw off the chloroform into a clean separatory funnel. Wash the aqueous phase with two, 3-5 ml portions of treated chloroform and add to the extracts. Discard the aqueous phase. To the chloroform extracts, add 50 ml of 0.1 N HCl and 10 ml of the 40% KBr reagent. Shake for 2 minutes. The Hg goes into the aqueous phase as H2HgBr4 while the Cu and Bi remains in the dithizone which is discarded. Wash the aqueous phase with a few ml of treated chloroform. Let the phases separate well and discard completely all chloroform droplets. An aliquot of the stripping solution may be taken if necessary so that the amount of Hg will fall on the standard curve. If an aliquot is taken, make up to 50 ml volume with 0.1 N HCl.

Add 10 ml buffer solution to bring the pH to 6, and 10 ml of 10 mg/liter dithizone solution. Shake well for 2 minutes. Avoid any exposure to direct sunlight or exceedingly bright artificial light.

NOTE: If the separatory funnel was not washed thoroughly with distilled water, the dithizone may be oxidized.

By means of a cotton swab on an applicator stick, remove any traces of moisture from the stem of the funnel after the stopcock has been opened for a second to allow the chloroform to fill the bore. Loosely insert a small cotton plug in the stem of the funnel. Rinse a cuvette twice with 1-2 ml portions of the chloroform layer and draw off the remaining dithizone into the cuvette. Place in the spectrophotometer and read at point of maximum light absorption (485 millimicron) against distilled chloroform. A blank on reagents should be carried through the entire procedure and this blank subtracted from the final result.

Standard curve - Suitable concentrations of mercury to give coverage over the entire range are used to establish a particular curve. Three or four points are sufficient.

Place 5 ml of the 40% KBr reagent, 10 ml of the buffer solution and the proper amount of standard mercury solution in a 125 ml Squibb separatory funnel. Add enought 0.1 N HCl to make the final volume 65 ml. Then add 10 ml of 10 mg/liter dithizone solution and shake for 2 minutes. Flush the stem of the separatory funnel and remove moisture by means of a cotton swab, withdraw the chloroform layer and read in the spectrophotometer as described above.

The 10 mg/liter dithizone solution is of sufficient strength to cover the range from 0 to 15 micrograms of mercury.

By using 20 ml instead of the standard 10 ml of this reagent,

the concentration range covered can be doubled. It is not recommended to add more than 20 ml of 10 mg/liter dithizone to any sample.

For only an occasional mercury analysis, it is better to bracket the sample with standard amounts rather than prepare an entire curve.

(f) Calculation

<u>Zinc</u>

Source: Michigan Department of Public Health.

(a) Scope

Stack testing for zinc can be accomplished by the polarograph method using a dropping-mercury electrode with the sample as the electrolyte.

(b) Auxiliary Apparatus

Sargent Polarograph - Model XX1, recording type, or equivalent.

(c) Reagents

Stock Zinc Solution - Weigh exactly 5.0 grams of dry reagent zinc (30 mesh or finer) into a 500-ml volumetric flask and add a minimum amount of constant boiling hydrochloric acid to get the zinc in solution. Boil until solution is complete and make up to volume with distilled water. The solution contains 10.0 mg zinc per ml.

Working Standard Zinc Solution - Pipette 5.0 ml of stock zinc solution (10.0 mg zinc per ml) into 500-ml volumetric flask and take to volume with 0.2 M KCl. The solution contains 100 micrograms zinc per ml with 0.2 M KCl as the electrolyte.

0.2 M KCl Solution - Weigh 14.9 grams reagent grade KCl into 1 liter volumetric flask and take to volume with distilled water.

Standard Cadmium Solution - Weigh exactly 0.2744 grams of $Cd(NO_3)_2 \cdot 4H_20$ into a 500-ml volumetric flask and take to volume with 0.2 M KCl. The solution contains 200 micrograms Cd per ml with 0.2 M KCl as electrolyte.

Oxygen Absorbent for Purification of Nitrogen - Pass nitrogen through a first scrubbing flask (midget impinger) containing concentrated NH40H and copper turnings. Caution: Make certain hole in impinger is not plugged before turning nitrogen on under pressure. Then pass nitrogen through a

second scrubbing flask containing concentrated sulfuric acid, again making certain this is not plugged before applying pressure.

Clean, Dry Mercury - Purchase from Eberbach & Son.

(d) Sampling Procedure

Assemble sampling train of probe, impinger with 100 ml 5% nitric acid solution, flow meter and vacuum pump. Sample at rate of 1 CFM for a period long enough to provide a minimum of 30 cubic feet at standard conditions.

(e) Analytical Procedure

Sample Preparation - Transfer the collecting solution from the impinger into a 250 ml beaker, wash out impinger with 5% hot nitric acid and all taken down to dryness on a hot plate. Add 2 ml concentrated nitric acid, wetting the sample thoroughly. Add 6 drops perchloric acid and swirl to mix. Evaporate to dryness on a hot plate at 350°-400° C. Repeat the acid treatment to obtain complete digestion. Cool and add 10 ml of 0.2 M potassium chloride solution. Loosen the solids with a rubber policeman, rinse policeman and beaker walls with 3-5 ml of 0.2 M potassium chloride solution. Cover with a watch glass and boil 2-3 minutes. Filter the solution into a 50-ml volumetric flask washing the filter with 0.2 M KCl. Dilute to volume with 0.2 M KCl giving the sample in 50 ml with 0.2 M KCl as the electrolyte.

Transfer 10 ml aliquot into polarographic cell, add l ml of 200 micrograms Cd per ml solution, and remove oxygen from cell by bubbling nitrogen through for three to five minutes. The initial voltmeter is set at .4 volts, the span voltmeter is set at 1 volt, thereby giving a range from -.4 volts to -1.4 volts. This is sufficient as cadmium "comes off" at approximately -.66 volts and zinc at approximately -1.05 volts. The sensitivity setting will vary depending on the amount of zinc present. The setting used for the standard curve is 0.02 microamperes/mm.

If there is a possibility that Cd is present in the sample an aliquot of the sample should be run in the polarographic cell first, without any internal standard added. If there is Cd present in the sample this must be taken into account when CD, the internal standard, is added.

Standard curve - Into the polarographic cell is introduced 1 ml of 100 micrograms Zn per ml solution, 1 ml of 200 micrograms Cd per ml solution, and 9 ml of 0.2 M KCl solution. This gives a total amount of solution in the cell of 11 ml thereby enabling a later removal of 10 ml of the sample and 1 ml of 200 micrograms Cd per ml internal standard solution. Also, there is an electrolyte in the cell of 0.2 M KCl. Both the volume of liquid in the cell and the electrolyte for standard curve and sample are critical for a proper analysis.

On the standard curve the heights of the Zn and Cd curves are measured in mm. The Zn to Cd ratio is found which is divided by the number of micrograms of Zn used giving a factor for 1 microgram Zn versus 200 micrograms Cd. It is suggested that 200 micrograms Cd be used as an internal standard in each sample for Zn thereby simplifying the calculations. The factor for 1 microgram Zn versus 200 micrograms Cd, found at the beginning of the series of samples being analyzed, will be used for the calculations through this series.

(f) Calculations

For the sample "polarogram" the heights of the Zn and Cd curves are measured in mm and the Zn to Cd ratio found in the same manner as the standard curve. The ratio found here is divided by the factor found in the standard curve for 1 microgram Zn versus 200 micrograms Cd giving the number of micrograms of Zn in the aliquot put into the polarographic cell.

$$mg Zn/M^3 = \frac{v \cdot y}{1.000 \cdot v \cdot Va}$$

where v = aliquot (m1)

V = total sample (m1)

Y = micrograms in v

Va = gas sample volume, in cubic meters,

at standard conditions

Nitrogen Oxides, Phenoldisulfonic Acid Method

Source: Public Health Service.

(a) Scope

When sulfur dioxide, ammonia, iron or other compounds that interfere with the hydrogen peroxide method are present in the gas to be sampled and/or the concentration of the nitrogen oxides is below about 100 ppm, this method is used. Accuracy below 5 ppm is questionable. This test is unsuitable for atmospheric sampling.

(b) Apparatus

- Sampling Probe Stainless steel (type 304 or 316) or glass tubing of suitable size (1/4-inch-OD, 6-footlong stainless steel tubing has been used).
- Collection Flask A 2-liter round-bottom flask with an outer 24/40 joint for integrated samples or a 250-ml MSA sampling tube for grab samples.
- Orifice Assembly The size of the glass capillary tubing depends on the desired sampling period (flow rates of about 1 liter per minute have been used). Use of this orifice is not mandatory.
- Adapter with Adapter for connecting col-Stopcock lection flask to sampling "T".
- Three-way Stopcock.

- Manometer A 36-inch Hg manometer or accurate vacuum gage.
- Spectrophotometer Beckman Model "B" or equivalent.

(c) Reagents

Thirty Percent Hydrogen Peroxide - (reagent grade).

Three Percent Hydrogen Peroxide - Dilute 30% $\mathrm{H}_2\mathrm{O}_2$ with water at 1:10 ratio. Prepare fresh daily.

Concentrated Sulfuric Acid.

0.1 N (approximate) Sulfuric Acid - Dilute 2.8 ml concentrated $\rm H_2SO_4$ to 1 liter with water.

Absorbing Solution - Add 12 drops 3% H_2O_2 to each 100 ml 0.1 N H_2SO_4 . Make enough for required number of tests.

1 N (approximate) Sodium Hydroximde - Dissolve 40 gm NaOH pellets in water and dilute to 1 liter.

Concentrated Ammonium Hydroxide.

Fuming Sulfuric Acid - 15 to 18 weight percent free sulfuric anhydride (oleum).

Phenol (reagent grade).

Phenoldisulfonic Acid Solution - Dissolve 25 grams of pure white phenol in 150 ml concentrated $\rm H_2SO_4$ on a steam bath. Cool and add 75 ml fuming sulfuric acid. Heat to $100^{\rm O}$ C for 2

hours. Store in a dark stoppered bottle. This solution should be colorless if prepared with quality reagents.

Potassium Nitrate (reagent grade).

Standard Potassium Nitrate Solution - Solution A: Dissolve 0.5495 gram KNO_3 and dilute to 1 liter in a volumetric flask. Solution B: Dilute 100 ml of Solution A to 1 liter. One ml of Solution A contains the equivalent of 0.250 mg NO_2 and of Solution B, 0.0250 mg NO_2 .

(d) Sampling Procedure

Integrated Grab Sample - Add 25 ml freshly prepared absorbing solution into the flask. Record the exact volume of absorbing solution used.

Set up the apparatus as shown in Figure 3, attach the selected orifice. Purge the probe and orifice assembly with the gas to be tested before sampling begins by applying suction to it. Evacuate the system to the vapor pressure of the solution: this pressure is reached when the solution begins to boil. Record the pressure in the flask and the ambient temperature. Open the valve to the sampling probe to collect the sample. Constant flow will be maintained until the pressure reaches 0.53 of the atmospheric pressure. Stop before this point is reached. During sampling, check the rate of fall of the mercury in one leg of the manometer in case clogging, especially of the orifice, occurs. At the end of the sampling period, record the pressure, temperature, and barometric pressure.

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An extended period of sampling can be obtained by following this procedure. Open the valve only a few seconds at regular intervals. For example: Open the valve for 10 seconds and close it for 50 seconds; repeat every 60 seconds.

Grab Sample - Set up the apparatus as shown in Figure 4 for high concentrations (200-3000 ppm) or the apparatus as shown in Figure 4 for low concentrations (0-200 ppm) but delete the orifice assembly. The same procedure is followed as in the integrated method except that the valve is opened at the source for about 10 seconds and no orifice is used.

Calibration curves are made to cover different ranges of concentrations. Using a microburette for the first two lower ranges and a 50-ml burette for the next two higher ranges, transfer the following into separate 150-ml beakers (or 200-ml casseroles).

- 1. 0-100 ppm: 0.0 (blank), 2.0, 4.0, 6.0., 8.0, 10.0, 12.0, 16.0, 20.0 ml of KNO₃ Solution B.
- 2. 50-500 ppm: 0.0 (blank), 1.0, 1.5, 2.0, 3.0, 4.0, 6.0, 8.0, 10.0 ml of KNO₃ Solution A.
- 3. 500-1500 ppm: 0.0 (blank), 5.0, 10.0, 15.0, 20.0, 25.0, 30.0 ml of KNO₃ Solution A.
- 4. 1500-3000 ppm: 0.0 (blank), 15.0, 30.0, 35.0, 40.0, 45.0., 50.0, 55.0, 60.0 ml KNO₃ Solution A.

Add 25.0 ml absorbing solution to each beaker. Follow as directed in the Analytical Procedure section starting with the addition of 1 N NaOH.

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After the yellow color has developed, make dilutions for the following ranges: 50 to 500 ppm (1:10); 500 to 1,400 ppm (1:20); and 1,500 to 3,000 ppm (1:50). Read the absorbance of each solution at 420 millimicron.

Plot concentrations against absorbance on rectangular graph paper. A new calibration curve should be made with each new batch of phenoldisulfonic acid solution or every few weeks.

(e) Analytical Procedure

Shake the flask for 15 minutes and allow to stand over night.

Transfer the contents of the collection flask to a beaker. Wash the flask three times with 15-ml portions of $\rm H_2O$ and add the washings to the solution in the beaker. For a blank add 25 ml absorbing solution and 45 ml $\rm H_2O$ to a beaker. Proceed as follows for the bank and samples.

Add 1 N NaOH to the beaker until the solution is just alkaline to litmus paper. Evaporate the solution to dryness on a water bath and allow to cool. Carefully add 2 ml phenoldisulfonic acid solution to the dried residue and triturate thoroughly with a glass rod, making sure that all the residue comes into contact with the solution. Add 1 ml $\rm H_{2O}$ and four drops concentrated $\rm H_{2SO_{4}}$. Heat the solution on the water bath for 3 minutes, stirring occasionally.

Allow to cool and add 20 ml H₂0, mix well by stirring, and add 10 ml concentrated NH₄OH, dropwise, stirring constantly. Transfer the solution to a 50-ml volumetric flask, washing the beaker three times with 4- to 5-ml portions of H₂0. Dilute to mark with water and mix thoroughly. Transfer a portion of the solution to a dry, clean centrifuge tube and centrifuge, or filter a portion of the solution.

Read the absorbance of each sample at 420 millimicron. If the absorbance is higher than 0.6, make a suitable dilution of both the sample and blank and read the absorbance again.

(f) Calculations

$$ppm NO_2 = (5.24 \times 10^5) (C)$$
 V_s

Where C = concentration of NO₂, mg (from calibration chart)

Vs= gas sample volume at 70° F and 29.92 in Hg, ml.

Sulfur Dioxide and Sulfur Trioxide, Shell Development Company Method

Source: National Air Pollution Control Administration Publication 999-AP-13.

(a) Scope

This method describes a procedure for determining sulfur dioxide and sulfur trioxide in stack gases.

(b) Apparatus

-	Sampling	Probe	-	Glass tubing (preferably boro-
				silicate or quartz) of suitable
				size with a ball joint at one

end and a removable filter at the other (a 1/2-inch-OD, 6foot-long tube has been used.)

- Filter

- A filter is needed to remove particulate matter, which may contain metal sulfates and cause interference during analysis. Borosilicate glass wool, Kaolin wool, or silica wool are suitable filters for removing particulate matter.

- Adapter

- Six plug-type connecting tubes T 24/40, one with a 90° bend and a socket joint.

- Heating Tape

- An insulated heating tape with a powerstat to prevent condensation in exposed portion of probe and adapter. Alternative: glass wool or other suitable insulators.

- Dry Gas Meter

- A 0.1-cubic-foot-per-revolution dry gas meter equipped with a fitting for a thermometer and a manometer. Alternately, a calibrated tank or a rotameter calibrated at the operating pressure may be used.

- Vacuum pump.

- Thermometers

- One $10^{\circ}-50^{\circ}$ C, $\pm 1^{\circ}$ C; and one $0^{\circ}-300^{\circ}$ C $\pm 5^{\circ}$ C are suitable.

- Manometer

- A 36-inch-Hg manometer

- Absorbers

 Two U-shaped ASTM D 1266 lamp sulfur absorbers with coarsesintered plates. - Filter Tube

- One 40-mm-diameter Corning medium-sintered plate.

 Scrubber for Purifying Air - An ASTM D 1266 lamp sulfur absorber with coarse-sintered plate.

- Teflon Tubing

- Teflon tubing, 1/4-inch ID, for connecting absorbers.
Alternative: 8-mm pyrex tubing with butt-to-butt connections held together with Tygon.

(c) Reagents

Water - Distilled water that has been deionized.

Isopropanol, Anhydrous.

Eighty Percent Isopropyl Alcohl - Dilute isopropanol with water at a ratio of 4 to 1.

Thirty Percent Hydrogen Peroxide - (reagent grade).

Three Percent Hydrogen Peroxide - Dilute 30% hydrogen peroxide with water at a ratio of 10 to 1. Prepare fresh daily.

Barium Chloride - (BaCl₂·2H₂0, reagent grade).

0.0100 N Alcoholic Barium Chloride - Dissolve 1.2216 grams BaCl₂·2H₂O in 200 ml of water and dilute to 1 liter with isopropanol. Standardize this solution with 0.01 N alcoholic sulfuric acid solution.

(As an alternate titrating solution to 0.01 N alcoholic barium chloride, in American Petroleum Institute Study Group uses 0.01 N alcoholic barium perchlorate because they believe that it gives a sharper end point during titration.)

Thorin Indicator - 1-(0-arsonophenylazo)-2 naphthol-3, 6-disulfonic acid, disodium salt.

0.2 Percent Thorin Indicator - Dissolve 0.2 gram thorin indicator in 100 ml water. Store in polyethylene bottle.

(d) Sampling Procedure

Set up the apparatus as shown in Figure 5. Place 30 ml of 80% isopropyl alcohol in the first absorber and 10 ml in the filter tube. The add 50 ml of 3% hydrogen peroxide to the second absorber. A light film of silicone grease on the upper parts of the joints may be used to prevent leakage. Wind the heating tape in a uniform single layer around the exposed portion of the probe and adapter and cover the heating tape with asbestos tape wound in the opposite direction. Place a thermometer between the heating tape and asbestos as near the adapter joint as possible. Connect the heating tape to a powerstat, switch on the current, and maintain the probe and adapter at a temperature at which no condensation will occur (about 250° C). Sample at 0.075 cubic foot per minute until 2 cubic feet or a suitable volume of gas has been sampled. Record the meter readings, temperatures and pressures at

10-minute intervals. Note the barometric pressure. Do not sample at a vacuum of more than 8 inches Hg.

Disconnect the asbestos tape, heating tape, probe, and adapter and allow them to cool. Connect the scrubber for purifying air to the inlet of the isopropyl alcohol absorber and add 50 ml of 3% hydrogen peroxide. Replace the water in the ice bath with tap water. Draw air through the system for 15 minutes to transfer residual sulfur dioxide to the hydrogen peroxide absorber. Disconnect the purifying air scrubber. (Although the use of air for removal of sulfur dioxide from isopropyl alcohol should not result in oxidation of sulfur dioxide to sulfur trioxide, the American Petroleum Institute Joint Study Group uses 99% nitrogen to preclude any possibility of oxidation.) Remove the filter and wash the probe and adapter with 80% isopropyl alcohol. Place the washings in the isopropyl alcohol absorber.

Disconnect the hydrogen peroxide absorber and transfer the contents and the water washings to a 250-ml volumetric flask. Dilute the water to the mark. Analyze for sulfur dioxide.

Stopper the isopropyl alcohol absorber and apply suction to the filter end. Remove the suction line and allow the partial vacuum in the absorber to draw the solution from the filter. Rinse the filter tube with 80% isopropyl alcohol before the suction is lost. Transfer the contents of the isopropyl

alcohol absorber and its washings to a 250-ml volumetric flask and dilute to the mark with 80% isopropyl alcohol. Analyze for sulfur trioxide.

(e) Analytical Procedure

Sulfur Trioxide - Pipette a suitable aliquot to a flask and dilute to 100 ml with 80% isopropyl alcohol. Add a few drops of thorin indicator (enough to give a yellow color). Titrate with 0.01 N BaCl₂ to the pink end point. Make a blank determination in parallel.

Sulfur Dioxide - Transfer a suitable aliquot to a flask and add 4 times this volume of isopropyl alcohol. Dilute to 100 ml with 80% isopropyl alcohol, add enough thorin indicator to give a yellow color, an titrate with standard 0.01 N BaCl₂ to the pink end point. Run a blank determination in parallel.

(f) Calculations

ppm S0₂ or S0₃ by volume = $\frac{24(A-B)(N)(F)(T)}{(V_O)(P)}$

Where A = 0.01N BaCl₂ used for titration of sample

B = ml 0.01N BaCl₂ used for titration of blank

N = exact normality of BaCl₂

F = dilution factor

T = average meter temperature, OR

Vo = observed volume of gas sample, cu ft P = average absolute meter pressure, in. Hg

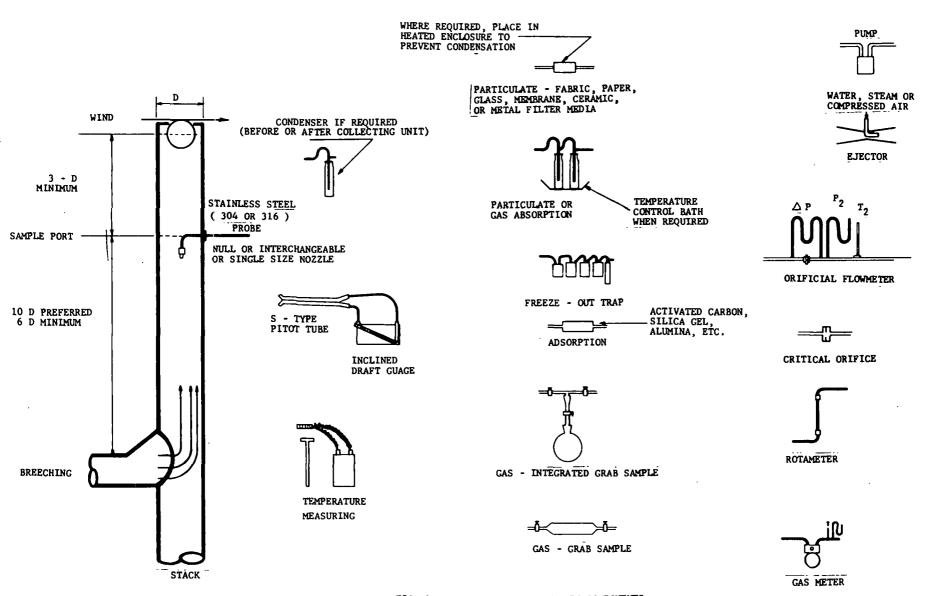


FIG. 1 SAMPLING LOCATION & TRAIN COMPONENTS

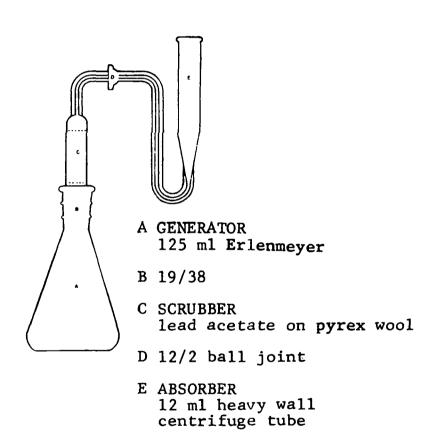


FIGURE 2
Arsine Generator

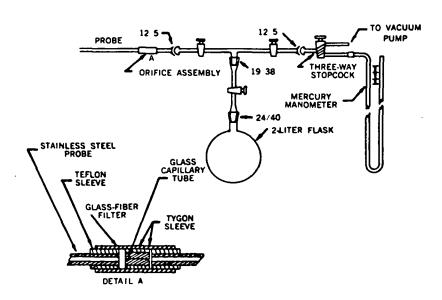
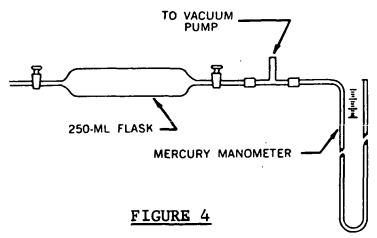


FIGURE 3

APPARATUS FOR INTEGRATED GRAB SAMPLES



APPARATUS FOR GRAB SAMPLES

Filter Tube

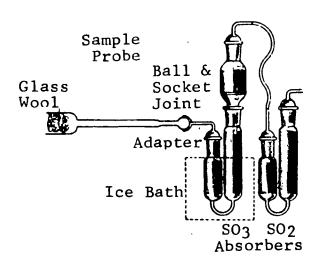


FIGURE 5
Sulfur dioxide - sulfur trioxide sampling train.

GLOSSARY OF TERMS

- ACFM Actual cubic feet per minute; refers to the volume of gas at the prevailing temperature and pressure.
- Acid Lining A refractory furnace lining essentially of silica.
- Additive A substance added to another in relatively small amounts to impart or improve desirable qualities, or suppress undesirable qualities. As additives to molding sand, for example, cereal, sea coal, etc.
- Aerosol Small particles, liquid or solid, suspended in the air. The diameters vary from 100 microns down to 0.01 microns or less; for example, dust, fog, smoke.
- Afterburner A device for burning combustible materials that were not oxidized in an initial burning process.
- Agglomeration Gathering together of small particles into larger particles.
- Air Cleaner A device designed for the purpose of removing atmospheric airborne impurities such as dusts, gases, vapors, fumes and smokes.
- Air Filter Any method used to remove gases and particulates from the environment and stack emission; it may be of cloth, fibers, liquid spray, electrostatic, etc.
- Air Furnace A reverberatory-type furnace in which metal is melted by heat from fuel burning at one end of the hearth, passing over the bath toward the stack at the other end.
- Air

 Pollution The presence in the outdoor atmosphere of one or more air contaminants or combinations thereof in such quantities and of such duration that they are or may tend to be injurious to human, plant or animal life, or property, or that interfere with the comfortable enjoyment of life or property or the conduct of business.
- Anneal A heat treatment which usually involves a slow cooling for the purpose of altering mechanical or physical properties of the metal, particularly to reduce hardness.

Baghouse - A large chamber for holding bags used in the filtration of gases from a furnace to recover metal oxides and other solids suspended in the gases. It's a form of dust collector and the bags may be constructed of natural, synthetic, or glass fibers.

Baked Core - A core which has been heated through sufficient time and temperature to produce the desired physical properties attainable from its oxidizing or thermal setting binders.

Balanced Arrangement of tuyeres in a cupola which provides for distributing or balancing the blast as required between upper and lower levels of the melting zone.

Basic Lining - In a melting furnace, the inner lining and bottom composed of materials that have a basic reaction in the melting process, usually either crushed burned dolomite, magnesite, magnesite bricks or basic slag.

Bed - Initial charge of fuel in a cupola upon which the melting is started.

Blast - Air driven into the cupola furnace for combustion of fuel.

Blast Volume - The volume of air introduced into the cupola for the burning of fuel. This volume governs the melting rate of the cupola and approximately 30,000 cubic feet of air is required per ton of metal melted.

Briquette - Compact cylindrical or other shaped block formed of finely divided materials by incorporation of a binder, by pressure, or both. Materials may be ferroalloys, metal borings or chips, silicon carbide, coke breeze, etc.

Burden - A collective term of the component parts of the metal charge for a cupola melt.

Burned Sand - Sand in which the binder or bond has been removed or impaired by contact with molten metal.

Canopy Hood - A metal hood over a furnace for collecting gases being exhausted into the atmosphere surrounding the furnace.

Cantilever Hood - A counterbalanced hood over a furnace that can be folded out of the way for charging and pouring the furnace.

Cast Iron -

Essentially an alloy of iron, carbon and silicon in which the carbon is present in excess of the amount which can be retained in solid solution in austenite at the eutectic temperature.

Catalytic Combustion -

A device for burning combustible gases, vapors, aerosols and odorous substances, reducing them to water vapor and carbon dioxide.

Centrifuging -

A method of casting, employing a core and depending on centrifugal force to make the metal more dense and strong in the outer portion of the casting. The mold cavities are usually spaced symmetrically about a central sprue, and the whole assembly is rotated about that axis during pouring and solidification.

Cereal Binder - A binder used in core mixtures and molding sands, derived principally from corn flour.

Charge -

The total ore, ingot, metal, pig iron, scrap, limestone, etc. introduced into a melting furnace for the production of a single heat.

Charging Door - An opening in the cupola or furnace through which the charges are introduced.

Coke -

A porous gray infusible product resulting from the dry distillation of bituminous coal, which is used as a fuel in cupola melting.

Coke Breeze -

These are fines from coke screenings.

Convection -

The motion resulting in a fluid from the differences in density and the action of gravity due to temperature differences in one part of the fluid and another. The motion of the fluid results in a transfer of heat from one part to the other.

Cope ~

The upper or topmost section of a flask, mold, or pattern.

Core -

A separate part of the mold which forms cavities and openings in castings which are not possible with a pattern alone. Cores are usually made of a different sand from that used in the mold and are generally baked or set by a combination of resins.

Core Binder - Any material used to hold the grains of core sand together.

Core Blower - A machine for making cores by blowing sand into the core box by means of compressed air.

Core Oven - Specially heated chambers for the drying of cores at low temperatures.

Core Sand - Sand for making cores to which a binding material has been added to obtain good cohesion and porosity after drying.

Crucible - A vessel or pot made of a refractory such as graphite or silicon carbide with a high melting point and used for melting metals.

Cupola - A cylindrical straight shaft furnace usually lined with refractories, for melting metal in direct contact with coke by forcing air under pressure through openings near its base.

Cupola, Hot A cupola supplied with a preheated air blast.
Blast -

Cupola Stack - The overall top column of the cupola from the charging floor to the spark arrestor.

Cyclone - A device with a control descending vortex (centrifugal created to spiral objectionable gases and dusts to the bottom of a collector cone for the purpose of collecting particulate matter from process gases.

Cyclonic Radial liquid (usually water) sprays introduced into cyclones to facilitate collection of particulates.

Density - Ratio of the weight of gas to the volume, normally expressed as pounds per cubic foot.

Desulfurizing - The removal of sulfur from molten metal by the addition of suitable compounds.

Direct Arc An electric arc furnace in which the metal being Furnace - melted is one of the poles.

Drag - The lower or bottom section of the mold, flask or pattern.

- Ductile Iron Iron of a normally gray cast type that has been suitably treated with a nodularizing agent so that all or the major portion of its graphitic carbon has a nodular or spherulitic form as cast.
- Duplexing A method of producing molten metal of desired analysis. The metal being melted in one furnace and refined in a second.
- Dust Small solid particles created by the breaking up of larger particles by processes such as crushing, grinding, drilling, explosion, etc.
- Dust An air cleaning device to remove heavy particu-Collector - late loadings from exhaust systems before discharge to outdoors.
- Dust Loading The concentration of dust in the gas entering or leaving the collector, usually expressed as pounds of particulate per 1,000 pounds of dry gas or grains per standard cubic foot.
- Efficiency With regard to dust collectors, it is the ratio of the weight of dust trapped in the collector to the weight of dust entering the collector. This is expressed as a percent.
- Effluent The discharge entering the atmosphere from the process.
- Electrostatic A dust collector utilizing a high voltage Precipitator- electrostatic field formed by negative and positive electrodes; the positive, uncharged electrode attracts and collects the gas-borne particles.
- Elutriation The sizing or classifying of particulate matter by suspension in a fluid (liquid or gas), the larger particulates tending to separate by sinking.
- Emission The total pollutants emitted into the atmosphere usually expressed as weight per unit of time such as pounds per hour.
- Endothermic Designating, or pertaining to a reaction which occurs with the absorption of heat from the surroundings.
- Equivalent
 Opacity The determination of smoke density by comparing the apparent density of smoke as it issues from a stack with a Ringelmann chart. In effect, it is a measure of the light obscurity capacity of the plume.

Exothermic Chemical reactions involving the liberation of heat; such as burning of fuel and deoxidizing of iron with aluminum.

Fabric A dust collector using filters made of synthetic, relation or glass fibers within a baghouse for removing solid particulate matter from the air or gas stream.

Facing Sand - Specially prepared molding sand mixture used in the mold adjacent to the pattern to produce a smooth casting surface.

Fines - A term the exact meaning of which varies.

- 1. Those sand grains that are substantially smaller than the predominating grain size.
- 2. That portion of sieved material that passes through the mesh.

Flask - Metal or wood frame without top or without fixed bottom used to retain the sand in which a mold is formed; usually consists of two parts, cope and drag.

Flux - Material or mixture of materials which causes other compounds with which it comes in contact to fuse at a temperature lower than their normal fusion temperature.

Fly Ash - A finely divided siliceous material, usually oxides, formed as a product of combustion of coke. A common effluent from the cupola.

Forehearth - Brick lined reservoir in front of and connected to the cupola or other melting furnaces for receiving and holding the melted metal.

Foundry Waste material in water or air that is discharged Effluent - from a foundry.

Fourth Hole In air pollution control, using a fourth hole Ventilation - in the roof of an electric furnace to exhaust (Direct Tap) fumes.

Fume - A term applied to fine solid particles dispersed in air or gases and formed by condensation, sublimation, or chemical reaction.

Gas • Formless fluids which tend to occupy entire space uniformly at ordinary temperatures and pressures.

Gate - The portion of the runner in a mold through which molten metal enters the mold cavity.

Gray Iron - Cast iron which contains a relatively large percentage of its carbon in the form of graphite and substantially all of the remainder of the carbon in the form of eutectoid carbide.

Green Sand - A naturally bonded sand or a compounded molding sand mixture which has been tempered with water and additives for use while still in a damp or wet condition.

Griffin A method operating in two stages, to recoup and preheat air by using the latent heat of cupola gases.

Heat Balance • A determination of the sources of heat input and the subsequent flow of heat usually expressed in equation form so that heat input equals heat output.

Heat A combination of heating and cleaning operations timed and applied to a metal or alloy in the solid state in a manner which will produce desired properties.

Heel - Metal left in ladle after pouring has been completed. Metal kept in induction furnaces during standby periods.

Holding A furnace for maintaining molten metal, from a larger melting furnace, at the proper casting temperature.

Hood - Projecting cover above a furnace or other equipment for purpose of collecting smoke, fume or dust.

Hot Blast - Blast which has been heated prior to entering into the combustion reaction of a cupola.

Indirect Arc An electric arc furnace in which the metal bath is not one of the poles of the arc.

Induction A melting furnace which utilizes the heat gen-Furnace - erated by electrical induction to melt a metal charge.

Inlet The quantity of gas entering the collector from the system it serves (in cubic feet per minute at a specified temperature).

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Inoculant - Material which when added to molten metal modifies the structure changing the physical and mechanical properties of the metal.

Inoculation - The addition to molten metal substances designed to form nuclei for crystallization.

Ladle The addition of alloying elements to the molten Addition - metal in the ladle.

Latent Heat - Thermal energy absorbed or released when a substance changes state; that is, from one solid phase to another, or from solid to liquid or the like.

Lining - Inside refractory layer of firebrick, clay, sand or other material in a furnace or ladle.

Magnesium The addition of magnesium to molten metal to form nodular iron.

Malleable
Iron amounts of silicon, manganese, sulfur and phosphorous, which, after being cast as white iron, is converted structurally by heat treatment into a matrix of ferrite containing nodules of temper carbon, and substantially free of all combined carbon.

Material A determination of the material input to the cupola and the output to fully account for all material.

Melting Rate - The tonnage of metal melted per unit of time, generally tons per hour.

Micron - A unit of measurement which is 1/25,000 of an inch or a millionth of a meter. Often designated by the Greek letter mu.

Mist - Visible emission usually formed by a condensation process or vapor-phase reaction, the liquid particles being sufficiently large to fall of their own weight.

Mold - The form, usually made of sand, which contains the cavity into which molten metal is poured to produce a casting of definite shape and outline.

Muller - A type of foundry sand mixing machine.

Nodular Cast Iron -

(See Ductile Iron)

Opacity -

The state of a substance which renders it partially or wholly impervious to rays of light. Opacity as used in an ordinance refers to the obscuration of an observer's view.

Outlet Volume - Quantity of gas exhausting from the collector (in cubic feet per minute at a specified temperature).

Oxidizing
Atmosphere -

An atmosphere resulting from the combustion of fuels in an atmosphere where excess oxygen is present, and with no unburned fuel lost in the products of combustion.

Oxidation Losses - Reduction in amount of metal or alloy through oxidation. Such losses usually are the largest factor in melting loss.

Particulate Matter -

Solid or liquid particles, except water, visible with or without a microscope, that make up the obvious portion of an exhaust gas or smoke.

Parting Compound -

A material dusted, brushed or sprayed on patterns or mold halves to prevent adherence of sand and to promote easy separation of cope and drag parting surfaces when cope is lifted from drag.

Pattern -

A form made of wood, metal or other materials around which molding material is placed to make a mold for casting metals.

Plume -

A visible, elongated, vertical (horizontal when windblown) column of mixed gases and gas-borne particulates emitted from a smoke stack.

Pollutant -

Any foreign substance in the air or water in sufficient quantities and of such characteristics and duration as to be injurious to human, plant, or animal life or property, or which unreasonably interferes with the enjoyment of life and property.

Preheater - A device used to preheat the charge before it is charged into the furnace.

Process Weight - The total weight of raw materials, except air, introduced into any specific process, possibly causing discharge into the atmosphere.

Recuperator - Equipment for transferring heat from hot gases for the preheating of incoming fuel or air.

Reducing An atmosphere resulting from the incomplete Atmosphere - combustion of fuels.

Refractory - Heat resistant material, usually nonmetallic, used for furnace linings, etc.

Reverberatory A large quantity furnace with a vaulted ceiling that reflects flame and heat toward the hearth or the surface of the charge to be melted.

Ringelmann's
Scale (chart)

A system of optical charts reading from all clear to solid black for grading the density of smoke emissions.

Riser - An opening in the top of a mold which acts as a reservoir for molten metal and connected to the casting to provide additional metal to the casting as it contracts on solidification.

Rotary A furnace using pulverized coal, gas or oil;
Furnace - of cylindrical shape with conical ends, mounted so as to be tipped at either end to facilitate charging, pouring and slagging.

SCFM - Units standing for Standard Cubic Feet per Minute.
The volume of gas measured at standard conditions, one atmosphere of pressure and 70° F.

Sea Coal - A term applied to finely ground coal which is mixed with foundry sands.

Sensible Heat - That portion of the heat which changes only the temperature, but does not cause a phase change.

Shakeout - The operation of removing castings from a sand mold.

Shell Molding - A process for forming a mold from thermosetting resin bonded sand mixtures brought in contact with preheated metal patterns, resulting in a firm shell with a cavity corresponding to the outline of the pattern.

Shotblasting - Casting cleaning process employing a metal abrasive propelled by centrifugal force.

Slag - Nonmetallic covering which forms on the molten metal as a result of the flux action in combining impurities contained in the original charge, some ash from the fuel and silica and clay eroded from the refractory lining.

Smoke -

A type of emission resulting from incomplete combustion and consisting predominantly of small gas-borne particles of combustible material present in sufficient quantity to be observable independently of the presence of other solids in the gas stream.

Spark Arrestor - Device over the top of the cupola to prevent the emission of sparks.

Sprue -

The channel, usually vertical, connecting the pouring basin with the runner to the mold cavity. In top pour casting the sprue may also act as a riser.

Standard Air - Air with a density of .075 pounds per cubic foot, generally equivalent to dry air at 70° F and one atmosphere of pressure (14.7 psia).

Superheating - Heating of a metal to temperatures above the melting point of the metal to obtain more complete refining or greater fluidity.

Tapping - Removing molten metal from the melting furnace by opening the tap hole and allowing the metal to run into a ladle.

Tuyere - The nozzle openings in the cupola shell and refractory lining through which the air blast is forced.

Vapor - The gaseous form of a substance normally in the solid or liquid state and which can be returned to these states either by increasing pressure or decreasing temperature.

Ventilation In the foundry, the exhaust ventilation and dust control equipment for the health, safety, comfort and good housekeeping of those who work there.

Venturi In air pollution control, a high velocity gas stream directed into the throat of a venturi of a wet scrubber to separate out particulates.

Wet Cap - A device installed on a cupola stack that collects emissions by forcing them through a curtain of water. The device requires no exhaust fan but depends upon the velocity pressure of the effluent gases.

Wet

In air pollution control, a liquid spray device, usually water, for collecting pollutants in escaping foundry gases.

Scrubber -

Wind Box -

The chamber surrounding a cupola through which air is conducted under pressure to the tuyeres.