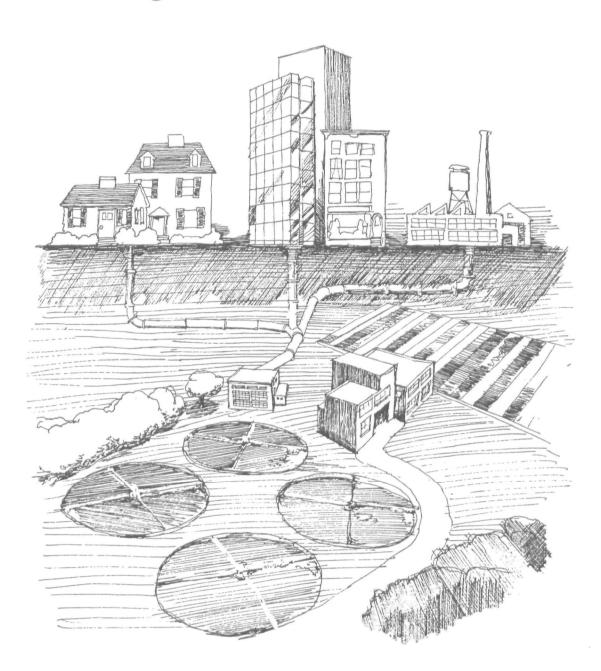
Sources of Toxic Pollutants Found In Influents To Sewage Treatment Plants

Washington, DC 20460

VI. Integrated Interpresentation



SOURCES OF TOXIC POLLUTANTS FOUND IN INFLUENTS TO SEWAGE TREATMENT PLANTS

VI. INTEGRATED INTERPRETATION

Report On
EPA Contract No. 68-01-3857

Mr. Donald Ehreth, Project Officer

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GLOSSARY

The following terms and abbreviations are used in this report.

POTW Publicly owned sewage treatment works.

INF

IND

POTW influent.

Influent

Tap Water Finished drinking water supply.

Industrial source.

RES Residential source.

COM Commercial source.

SUM Calculated sum of contribution for the RES, COM and

IND sources.

SMSA Standard metropolitan statistical area.

Level Refers to concentration level of pollutants.

Classicals The six conventionally measured parameters:

ammonia, oil and grease, total suspended solids (TSS), total organic carbon (TOC), chemical oxygen demand (COD) and biological oxygen demand (BOD). The classicals were always measured in mg/L units in contrast to the toxic

pollutant measurements in µg/L units.

Pollutant A series of reference numbers were assigned to the Reference pollutants for convenience in data storage and retrieval

Numbers as follows:

1XX Volatiles analysis category.

2XX Acids analysis category.

3XX Base/Neutral analysis category.

4XX Pesticides and PCB analysis category.

5XX Metals analysis category.

6XX Total cyanids and total phenols.

7XX Classicals.

µg Microgram.

mg Milligram.

. .

Kg Kilogram.

mg/L Concentration in micrograms per liter.

mg/L Concentration in milligrams per liter.

Lps Flow rate in liters per second.

MGD Flow rate in million gallons per day.

QA Quality assurance.

QC Quality control.

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We wish to acknowledge the considerable efforts and cooperation of the many people whose contribution helped in the successful completion of the work described in this report.

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I. SUMMARY

A study has been carried out to determine the relative significance of the major source type - residential, commercial, industrial - contributions of priority pollutants to POTW influents.

A service area in each of four cities—Cincinnati, St. Louis
Atlanta and Hartford—has been studied in detail. In each city, specific
sampling sites were selected to represent each of the major source cate—
gories. In total, 11 residential, 10 commercial and 5 industrial sites
have been sampled, in addition to the tap water and POTW influents.

This report presents a summary analysis of the data made possible by integrating the results obtained from each of the cities and treating the data by source category. The data have been analyzed to determine the frequency of occurrence of toxic pollutants (specifically the list of 129 priority pollutants), their concentration levels, the sources of these pollutants, and the impact of the source contributions on the POTW influent. Manganese and several classical parameters (7XX series) were also measured and included in the data base. The analysis has been carried out within the constraints imposed by the inherent characteristics of each of the major source categories - residential, commercial, industrial - concerning the range of discharge levels which was observed for each category.

The data available for analysis consist primarily of three types:

- 1. Total service area source type description and demography and similar data for each specific sampling site.
- 2. Entire service area and site specific flow data,
- 3. Chemical concentration data.

The general source descriptions and details of housing, population, SIC category industries, etc. was obtained from local agencies in each

^{*} Publicly Owned Treatment Works

of the four cities studied. Flow information for the commercial and industrial users in the entire service area was obtained from the water supply records.

Field sampling at each site occurred over a period of six days, resulting in 30-60 24-hour or 48-hour composite samples per city.

Flows were measured at each site for use in calculating mass discharge rates. Each of the samples was returned to the laboratory for complete chemical analysis according to the EPA protocol.

A full quality control program was implemented for the chemical analyses. The results of this program showed that most pollutants were analyzed with 80-90% accuracy and with a relative standard deviation of 10-30%.

The available data have been grouped according to the major source categories. For each category, the frequency of detection of a given pollutant and its average concentration has been determined. For residential sources, the per capita discharge rate (mg/person/day) has been calculated for each site and each pollutant. An average index value has been calculated for each source category such that, when the quantity of each source type was known, the POTW influent mass flow could be calculated from

POTW (Kg/day) =
$$V_R A_R + V_C A_C + A_I A_I$$

where V is the index value for each source type (R = residential, C = commercial, I = industrial) and A is the amount of source activity. For the residential sources, the population was used as an index base. For the commercial and industrial soruces, the total source type flow was used as the base. The indices calculated using this approach appear to be reliable for the residential and commercial sources, but can only be used as estimates for the industrial sources, because of the highly specific dependence of this index on industry type. The industrial index has been useful primarily for purposes of comparison with the other two source types. Perhaps the most important observation in this study is that relatively few toxic pollutants were found in the sources and many of those found were present at low concentration levels. A total

of 56 priority pollutants were observed in this study, grouped as indicated below by analysis category:

Volatiles 24
Acids 7
Base/Neutrals 11
Pesticides 2
Metals 12 (plus manganese)
Total Cyanides
Total Phenols

The pollutants given in Table 1 were never detected (within the limits of the chemical analysis) during the entire study.

The following 5 box plots (Figures 1-5) show the frequency of detection and the average source concentration values for pollutants which were observed more than 50% of the time and/or at source average concentrations greater than 10 μ g/L. The data have been grouped according to those chemicals observed at levels less than 10 μ g/L, 10-100 μ g/L.

The increase in numbers and concentration of chemicals is clear as one proceeds from tap water through residential and commercial to the industrial sources. The final result observed at the POTW influent does indeed appear to be a good integration of the individual source values because the POTW influent concentration levels are higher than just residential values and lower than industrial values and the requency of observation is increased at the POTW.

The data in Figures 1-5 relate to concentration only and cannot be used directly for projection to other areas.

In order to evaluate the potential impact of the individual source type contributions on the POTW, the average index values for each type were scaled by the flow (or population) for that source to calculate typical POTW loadings. The example in Table 2 shows the fraction contributed from each source type resulting in the indicated POTW loading (in Kg/day) for a hypothetical city whose characteristics were as follows:

Table 1
Pollutants (67 Total) Never Detected in Four Cities

*101	Chloromethane	340	Chrysene/Benzo(a)anthracene
102	Dichlorodifluoromethane	342	3,3'-Dichlorobenzidine
103	Bromomethane		Benzofluoranthenes**
107	Acrolein	345	Benzo(a)pyrene
122	Cis-1,3-dichloropropylene	346	Indeno (1,2,3-c,d)pyrene
202	Nitrophenol	347	Dibenzo(a,h)Anthracene
	2,4-dinitrophenol		Benzo(g,h,i)pervlene
209	4,6-dinitro-2-cresol	349	TCDD
211	4-Nitrophenol	401	alpha-BHC
304	Hexachloroethane	402	gamma-BHC
305	Bis(chloromethyl)ether	403	beta-BHC
306	Bis(2-chloroethyl)ether	405	delta-BHC
307	Bis(2-chloroisopropyl)ether	407	Heptachlor epoxide
308	N-Nitrosodimethylamine	408	Endosulian I.
309	Nitrosodi-n-propylamine	409	DDE
311	Hexachlorobutadiene	410	Dieldrin
313	2-Chloroethyl vinyl ether	411	Endrin
314	Bis(2-chloroethoxy)methane	412	DDD
316	Isophorone	413	Endosulfan II.
31.7	Hexachlorocyclopentadiene	414	DDT
318	2-Chloronaphthalene	415	Endrin aldehyde
319	Acenaphthylene	416	Endosulfan sulfate
320	Acenaphthene	417	Chlordane
321	Dimethyl phthalate	418	Toxaphene
322	2,6-Dinitrotoluene	419	PCB-1221
323	4-Chlorophenyl phenyl ether	420	PCB-1232
324	Fluorene	421	PCB-1242
325	2,4-Dinitrotoluene	422	PCB-1248
327	1,2-Diphenylhydrazine	423	PCB-1254
328	N-Nitrosodiphenylamine	424	PCB-1260
329	Hexachlorobenzene	425	PCB-1016
330	4-Bromophenyl phenyl ether	503	Beryllium
336	Benzidine		

^{*}The 101, etc., numbers paired with pollutants are referencing numbers for data storage.

^{**}Two compounds.

Figure 1: Concentration/Frequency of Occurrence: Tap Water

Figure 2: Concentration/Frequency of Occurrence: Residential

	Less Then 50%	Greater Than 30%
Greater Than 100 μg/L		Manganese Zinc
Between 10 μg/L and 100 μg/L	Trichlorœthylene Di-n-butylphthalate	1,1,2,2-Tetrachloroethylene Toluene Butylbenzylphthalate Copper Lead Chromium Nickel Total Phenols
Less Then 10 µg/L	(All others detected)	Chloroform Bromodichloromethane 1,1,1-Trichloroethane Benzene Ethylbenzene Silver

Figure 3: Concentration/Frequency of Occurrence: Commercial

Greater Than 100 μg/L		Ethylbenzene Phenol Dichlorobenzenes Butylbenzylphthalate Silver Copper Nickel Chromium Lead Manganese Zinc Total Phenols
Between 10 μg/L and 100 μg/L	1,1-Dichloroethylene Trans-1,2-dichloroethylene Carbon Tetrachloride 2,4-Dimethylphenol Pentachlorophenol Bis(2-ethylhexyl)phthalate Cadmium	Chloroform Trichloroethylene 1,1,1-Trichloroethane 1,1,2,2-Tetrachloroethylene Toluene Naphthalene Di-n-butylphthalate Total Cyanides
Less Than 10 µg/L	(All others detected)	Benzene Bromodichloromethane Dibromochloromethane Antimony

Figure 4: Concentration/Frequency of Occurrence: Industrial

	Cass Than 50%	Greater Than 50%
Greater Than 100 μg/L		Chromium Manganese Zinc
Between 10 μg/L and 100 μg/L	Naphthalene Butylbenzylphthalate Antimony	Trichloroethylene 1,1,1-Trichloroethane 1,1,2,2-Tetrachloroethylene Toluene Ethylbenzene Dichlorobenzenes Copper Lead Nickel Total Cyanides Total Phenols
Less Than 10 µg/L	(All Others Detected)	Chloroform Benzene Diethylphthalate Di-n-butylphthalate Cadmium Silver

Figure 5: Concentration/Frequency of Occurrence: POTW Influent

Table 2

Hypothetical Drainage Basin Calculation

(Flow 60% Residential, 20% commercial, 20% Industrial)

		Relative Fraction S			SUM
		RES	COM	IND	Kg/day*
110	1,1-DICH LOROETHYLENE	.00	.02	.98	. 30
111	1, 1-DICHLOROETHANE	.00	.03	.97	.04
112	TRAUS-1, 2-DICHLOROETHYLERE	.00	.08	.92	.33
113	CHLOROPORM	.34	.18	.48	. 65
114	1,2-DICHLOROETHANE	.08	.12	.80	.02
115	1,1,1-TRICHLOROLTHAGE	.03	.02	•95	2.32
116	CARBOU TETRACILLORIDE	.00	.00	1.00	• 74
117	BROMODICHLOROMETHANE	.06	. 28	.67	.06
120	TRICHLOROETHYLENE	.06	. 24	.70	. 94
121	BENZENE	.15	.50	.35	.09
123	DI BROMOCU LOROMETU ANE	.05	.28	.67	.04
	BROMOFORM	.00	.00	.00	.00
127	1,1,2,2-TETRACHLOROUTHYLEKE		.14	.68	2.66
128	TOLUENE	.12	.11	.77	1.76
129	CHLOROBENZEUE	.17	.02	.81	.03
130	ETHYL BEHZENE	.01	.02	.97	2.69
_	PHENOL	.17	.02	.82	4.31
	2,4-DIMETHYLPHENOL	.02	.00	.98	1.96
_	PENTACHLOROPHEROL	.35	.18	.47	. 55
	DICHLOROBENZENES	.02	.01	.97	
	RAPHTHALEHE	.05	.03	.92	1.43
	DIETHYL PUTHALATE	.93	.07	.00	1.44
	DI-N-BUTYL PHTHALATE	. 34	.07		2.95
	BUTYL BENZYL PHTHALATE	.12	.04	-	5.15
	BIS(2-LTHYLHEXYL)PHTHALATE	.32	.07	.61	1.84
	ANTINONY	.83	.02	.15	.29
	ANSERIC	.86	.05	.09	. 94
504	CADNIU%	.07	.02	.91	.59
	CHRONIUM	•06	.05	.89	20.80
	COPPER	.63	.08	.29	11.23
	LEAD	. 32	.06	.61	13.66
	MANGALIESE	.55	.18	. 27	21.97
509	MERCURY	. 34	.08	.58	80.
	WICKEL	.09	.06	.85	3.33
	SELLATUM	.83	.12	.05	.46
	SILVER	.06	.01	.93	4.20
	THALLIUM	.00	.49	.51	.00
	ZINC	.35	.06	• 59	37.81
	TOTAL CYANIDES	.06	.00	.94	2.51
	TOTAL PHENOLS	.33	.07	•59	8.91
	AMMONIA	.77	.09	.14	2.02
	OIL AND CREASE	.53	.19	.28	9,81
	TSS	.66	.09	. 25	22.65
	2°CC	.59	.14	•26	12.85
	COD	.57	.13	.30	46.05
708	EOD	• 56	. 14	.29	19.21

^{*}Classicals (7XX) in 10³ kg/day.

POTW Influent Flow:

Residential - Flow:

- population

Commercial Flow:

Industrial Flow:

1,000 Lps (Liters per second)

600 Lps

200 Lps

200 Lps

Although this example is just for a hypothetical city, the relative flow contributions chosen for each source type approximates the actual average values for 327 larger drainage basins with POTWs having secondary treatment.

The relative contributions indicated in the table show clearly that the industrial sources dominate the loading on the POTW for most pollutants, but, for an important number of pollutants the residential and commercial contributions are still important. In interpreting these results one should remember that the industrial component is quite industry specific and the industrial contribution could be higher or lower depending on the particular industries present. This example represents one way in which the data contained in this report may be used to evaluate the importance of source strengths on POTW influent toxic pollutant burden.

Several other factors are evident in examination of the data. Toxic pollutants are found slightly more frequently on weekdays than weekends. There is a higher per capita pollutant load from old versus new residential areas, especially for lead and phenol. There is a high degree of correlation in the amounts of aliphatic and aromatic hydrocarbons found in the samples. The quantities of aromatic hydrocarbons show similar trends and frequently follow the quantities of a number of other pollutants. The quantities of lead and zinc trend in the same manner.

For the limited rain event data collected in this study, the lead, zinc and manganese levels were observed to increase during the rain.

II. INTRODUCTION

It is the concern of the Office of Water Planning and Standards (OWPS) to develop a comprehensive strategy governing the toxic substances introduced into, and subsequently passing through, the publicly-owned sewage treatment works (POTWs). In order to supply the necessary basis for formulating guidelines, the Monitoring and Data Support Division (MDSD) has sponsored this study of cities across the country. In addition to assessing the extent to which priority pollutants may enter the environment via the POTWs, this POTW program is concerned with determining the sources of those pollutants. The objectives of the POTW source survey include defining the various types of source categories, describing those categories in terms of priority pollutant contributions, and determining the relationship of the individual source measurements to the pollutant burden at the POTW influent.

By using the data to calculate a set of pollutant specific indices corresponding to the residential, commercial and, to a lesser extent, industrial sources for each of the cities sampled, it is hoped that a general characterization of the pollutant load attributable to these categories can be made. In this way, the sources of the pollutants measured in the POTW influent of previously unsampled treatment basins may be estimated in such a way as to suggest valid routes to limit pollutant loads.

The details of the studies carried out in this program have been published in five preceeding reports (1-5). The sampling and analysis procedures employed in the POTW source survey were those outlined in the EPA Screening Protocol for Priority Pollutants 6. A detailed quality control program was implemented for this study patterned after the EPA recommendations of a QC program for verification studies 7. The data given in the individual city reports showed the analyses to be in control with respect to producing reliable concentration data, free from interference. The QC program also made it possible to consistently achieve low detection limits for the toxic pollutants.

A limited attempt had been made in each of the individual city reports to compare the importance of source types on the POTW burden. This report integrates the data from each of the individual studies, by source type, for a more comprehensive and reliable analysis of each of the factors which were goals of the study.

There were many objectives in this study, but of fundamental importance was the desire to determine which pollutants were present in sources (and which were not) at what frequency and the relative mass contribution of each source type for each pollutant. If possible, it was desirable to establish a discharge index for at least the residential and commercial source types so that their impact on the POTW could be compared to that of the industrial sources.

Other objectives included examining variances within and between source types: weekday/weekend effects, chemical to chemical correlations, etc. Some of these objectives could be addressed during this study, many of the other secondary objectives will require further study.

The overall approach used in the study was to collect specific mass discharge rate data (calculated from measured concentration and flow values) from a number of specific sites representing residential, commercial and, to a lesser extent, industrial sites. The methdology involved going to several cities and sampling portions of each source type in each city (when possible, depending on the city characteristics) such that the desired data base would be available after all of the cities had been sampled. Because of the high cost and considerable time associated with completing the study of a single city, it has only been possible to sample four cities. The cities, and drainage basins within the cities, were selected in an effort to reasonably reflect each of the major source types. It is felt that the residential and commercial sources are well represented in the data base in terms of overall POTW activity. It is also recognized that the industrial source data base is much more restricted in terms of overall representatives. While the conclusions which can be drawn from such

a limited data base are tentative, it is believed that the data summarized in this report provide a reliable base for future evaluations.

The individual city reports contain a great deal of information about the site and service area descriptions, the sampling and analysis procedures and detailed results. The purpose of this report is to summarize essential portions of the data in those reports in order to be able to interpret the total data base in terms of each of the objectives.

III. METHODOLOGY DATA BASE

A. Drainage Basin Selection Criteria

Selection of specific service areas (drainage basins) and sampling sites within the areas that meet all of the program goals turned out to be a much more difficult problem than had originally been anticipated. The criteria used in the selection process and the factors influencing the final decisions are described briefly in this section.

At the outset of this study, three constraints were imposed upon site selection:

- Only those plants employing secondary or better treatment technology were to be considered;
- 2) Only plants with average daily influent flows greater than or equal to 5.0 MGD were to be considered; and
- 3) Only those treatment facilities located within standard metropolitan statistical areas (SMSA's) were to be considered.

The reasons for setting these original constraints were that

- 1) Secondary treatment technology or better would be mandated by 1983 under the provisions of the Clean Water Act;
- 2) The impact of large flow variability is not as great as plant size increases;
- 3) The variety and amount of industrial activity frequently increases as plant size increases;
- 4) The sampling process would not interfere with normal operations in larger plants;
- 5) Most industrial activity occurs within SMSA's, urban treatment facilities have a larger variety of industrial dischargers than do rural POTW's.

In preparation for this study, EPA had (through its contractor, SRI International) formed a data base of 25,076 POTWs based upon information supplied in the 1976 Needs Survey. From that base, a subset of

324 plants with secondary treatment was selected for consideration. A further reduced list of 80 plants was constructed by random selection from the 324 plant list. That list of 80 plants represented the starting point for site selection for this study.

Some problems arose, however, in attempting to use only the "80" list for site selection. While the concept of multivariate regression analysis and random selection (used in developing the list of 80 plants) would have lent statistical credibility to the ultimate site selection process, such an approach must be founded upon a good data base. Subsequent to contacting several of the facilities identified on the list, it was learned that many plants in fact only had primary treatment. Similarly it was learned that many secondary plants that were desirable for this study's needs had been missed during the analysis because data obtained from the 1976 Needs Survey was incorrect. Therefore, the 80 and 324 lists were frequently used only as a first reference. Additional data was obtained from other sources (water pollution control federations, state departments of environmental protection, telephone conversations with sewage treatment authorities, etc.) to supplement these lists.

The process of screening the remaining POTW's was conducted by placing telephone calls to a number of candidate facilities, and by obtaining as much additional information about each as was possible. After completing this series of calls, all the additional data was reviewed prior to selecting two or three which were then visited. Once site visits had been completed the final selection of a test facility was made. This process was repeated for each city.

Specific issues examined during the screening selection process phase included:

- Geographic location of the facility
- Plant and drainage basin size
- Identification of proper sampling zones
- Availability of background information
- Availability of maps
- Convenience of the city, both with respect to internal

congestion and with respect to sample shipment

- Logistics support available within the area
- The perceived safety and accessibility of the sampling area, and
- The willingness of the local authorities to participate and assist with the study.

Prior to final site selection, initial contacts were made both over the telephone and during preliminary meetings to accumulate as much information as was possible.

Plant and basin size were important considerations because each appeared to have a bearing upon the diversity of socio-economic activity that existed within an area. As was learned early in this program, plants with small daily influent flows (5 to 10 MGD) were frequently located in areas where only one type of activity was present. For example, many of the basins that exhibited low influent rates (5 - 10 MGD) were comprised of virtually all (90-95%) residential activity, with very little (5-10%) commercial activity and almost no industrial component (0-1%). Conversely, one plant with an average daily influent of 12 MGD had a flow mix which was nearly 99% industrial. Any of these plants would have been acceptable if the project's goals were only to assess one source's contributions independent of the others; but inasmuch as an assessment of all three was desired concurrently, these types of sites were excluded from further consideration.

A second problem encountered in several of the smaller service areas (with respect to the area served) was that even when a basin was identified which contained all three activities (residential, commercial and industrial), it was frequently impossible to isolate these

activities in the collection system. This difficulty usually arose because the smaller basins were frequently interconnected by single interceptors, where wastewater from one activity would drain through another, prior to reaching the POTW.

The identification of proper sampling zones was also considered to be important. Since the final goal of this study was to enumerate the pollution burden of at least two socio-economic activities at a minimum in each basin, areas typifying both of these had to be identified, and segregated if possible. It was desirable to locate duplicate areas within a basin because this allowed for an immediate confirmation of results under conditions that were equivalent.

Another factor considered important to the selection of a test facility related to the availability of background or supportive data. Of particular importance was the availability of demographic information which is needed to describe the activity within the particular sampling zones selected and within the basin as a whole. However, supplementary data, such as 201 and 208 studies, facility plans, and inflow/infiltration assessment were also valuable.

Similarly, it was essential that the identified facility have detail or cadastral maps of the collection system. Without having access to these maps, it is virtually impossible to select appropriate sampling locations because the area drained cannot be outlined.

The geographic location of the facility was also used as a criteria in the selection process. Although one reason for including this issue pertained to the climate of the area during the analysis period, the main reason for considering it related to the issues of the variability of groundwater and soil chemistry across the country. Initially these factors were believed to have some possible bearing upon the distribution or fate of the priority pollutants within the collection system. For example, the pH of water (either drinking or groundwater) could influence the partitioning of organic acids and bases between the sediment, aqueous or gaseous phase. Comparably, interactions (particularly adsorption) between the priority pollutants and soil types could have an affect on measured concentrations.

To address this issue, the country was divided into seven regions based upon three factors:

- 1) Water chemistry as defined by the presence of anions such as $C1^-$, $S0_4^-$, and $C0_3^-$,
- 2) The ionic strength of the water as defined by the concentration of dissolved minerals, and
- 3) A comparison of soil types.

The initial goal was to select at least one test facility from each zone. Program constraints and the difficulty in locating areas which met all of the needs of the study resulted in the study of basins in three of the seven zones. One zone represented about 50 percent of the area of the Continental United States and the Cincinnati and St. Louis sites were in this zone.

Other factors which also influenced the basin selection related to logistical concerns. Some of the key factors considered at this juncture included the congestion of the area, as this related directly to the ability to maintain a four-hour rotation between all the remote sites; the proximity of a major airport, as this affected the shipment of samples back to the laboratory; and the availability of rental vehicles (trucks, automobiles, trailers) and supplies.

B. Drainage Basins Selected for Study

Four drainage basins have been sampled for this study. They are:
Muddy Creek Drainage Basin, Cincinnati, Ohio

Coldwater Creek Drainage Basin, St. Louis, Missouri

R. M. Clayton Drainage Basin, Atlanta, Georgia Hartford WPCP Drainage Basin, Hartford, Connecticut

Detailed descriptions of each basin are given in the individual reports.

A brief description of each basin is given here in order to help provide a perspective on the source character and mix of each of the areas.

1. Muddy Creek Drainage Basin, Cincinnati, Ohio

The Muddy Creek drainage basin is located in the western portion of of the greater Cincinnati, Ohio metropolitan area. It is bounded to the south by the Ohio River, to the west by the Ohio/Indiana state line and to the north by I-74. The drainage basin is roughly 29 square miles

in size and has a population of approximately 88,000. Contained within this area are newer housing developments, older residential communities, recreational areas, shopping centers and commercial districts, a small amount of industrial activity and a fair amount of open space or land currently undergoing development. Generally, the area may best be described as "suburban Cincinnati."

Major communities included in part or in total within the basin are Cincinnati, Cheviot, Addyston, Green Township, Miami Township and Delhi Township. Of these, the areas of Cincinnati, Cheviot, and Addyston are older (circa pre-1940). The three township areas are generally more open, but have been the location of recent developments.

Based upon estimates derived from a theoretical flow analysis of the basin, the blend of activity contained within the area is 90-92% residential, 7-9% commercial and 0-1% industrial. The average daily influent to the treatment plant is 9.5 million gallons. The collection system is a combination of both sanitary and combined sewers.

2. Coldwater Creek Drainage Basin, St. Louis, Missouri

The Coldwater Creek drainage basin lies to the north and west of the City of St. Louis, Missouri. None of the City of St. Louis is located within the area of the basin; however, all of the basin is located within St. Louis County. The basin is bordered to the north and west by the Missouri River, open land, and the community of Bridgeton; to the east by several communities contained within St. Louis County (Ferguson, Belridge); and to the south by the community of Olivette. Part or all of fifteen communities are contained within the basin, including Florissant, Berkeley and St. Ann. The total land mass of the drainage area encompasses approximately 34-36 square miles. The population of this area is estimated to be roughly 200,000.

Socio-economic activity contained within this area includes older residential which is concentrated along the southern and eastern borders and newer residential, predominently along the northern and northwestern borders. Furthermore, high density strip commercial zones are found running north to south along Lindbergh Boulevard and east to west along

St. Charles Rock Road. The largest shopping center is situated at the intersection of these two roads. Industrial activity is concentrated near the center of the basin, just north of Lambert Airport. Major industrial activities include an aircraft manufacturer, two automobile assembly or part facilities and a diesel engine (railroad) assembly plant.

Based upon estimates obtained from the theoretical flow analysis of the basin, the wastewater tributary to the Coldwater Creek Plant is comprised of roughly 78% residential, 10% commercial and 12% industrial flow. The treatment plant has an average daily influent flow of roughly 23.5 MGD and uses the activated sludge technology for wastewater cleanup. The collection system is sanitary only, with storm water being channeled directly to the numerous creeks within the basin.

3. The R. M. Clayton Drainage Basin, Atlanta, Georgia

The R. M. Clayton drainage basin is located in the metropolitan Atlanta, Georgia area. The overall size of the basin is approximately 130-140 square miles and the estimated population is roughly 385,000. With the exception of being bordered on the west by the Chattahcochee River, no well defined borders exist to describe the area. However, the basin encompasses most of the northern part of the City of Atlanta, the southern portion of Fulton County, a large amount of the mid-section of DeKalb County and a small amount of southwestern Gwinett County.

A rough breakdown of the distribution of land from each of these areas is listed below:

City of Atlanta	45-50 square miles
Fulton County	20-25 square miles
DeKalb County	60-65 square miles
Gwinett County	5-10 square miles

The breakdown of socio-economic activity within the area indicates that there are many industrial parks scattered throughout the basin.

Two of these are reasonably large: one situated along the southwestern border of the basin near the river, and the second located in

DeKalb County near the intersections of I-285 and the Buford Highway. Major industrial complexes contained in the basin include an automotive assembly plant, a steel mill, paint manufacturers and industrial laundries. Similarly, commercial activity is scattered throughout the basin. However, of these zones, the most extensive is concentrated in "downtown" Atlanta. Residential activity includes all strata of the economic spectrum. Typically, the older residential areas are most concentrated in or near the city limits, while the newer areas are found near the northern and eastern edges of the basin in both Fulton and Gwinett counties. Also included within the basin is a large amount of institutional (colleges, universities, hospitals, etc.) and municipal activity, as well as open space.

Based upon theoretical flow estimates, the blend of activity within the R. M. Clayton basin is roughly 61% residential, 21% commercial and 18% industrial. The average daily influent flow to the treatment plant is 80 MGD, and the treatment technology used is activated sludge. The collection system contains both combined and sanitary sections, with all of the combined lines being present within the Atlanta city limits.

4. Hartford WPCP Drainage Basin, Hartford, Connecticut

The hartford Water Pollution Control Plant's drainage basin is located in the greater Hartford, Connecticut metropolitan area. Portions of six communities (Hartford, Wethersfield, Newington, West Hartford, Bloomfield and Windsor) are served by the plant. The basin itself covers approximately 60-65 square miles of area and the current population is estimated to be 285,000. The basin is bordered on the east by the Connecticut River, on the west by the communities of Farmington and Avon, to the south by the community of Rocky Hill and to the north by East Granby.

The major activities present within the basin are principally residential and commercial, although a small industrial component is also included. Typically, older residential activity is concentrated in the City of Hartford, Wethersfield and Windsor, with newer residential areas

located principally in the community of Bloomfield. The major commercial district is in downtown Hartford. This area is comprised of both retail businesses and office building activity. The main commercial interest within this basin is insurance. The industrial component contained in the basin is scattered throughout the area. This activity is principally light in nature (warehousing, supply companies) although a number of tool and die shops, metal fabricators and platers are also included, Municipal activities also abound as Hartford is the State Capital of Connecticut.

Based upon flow estimates derived from a theoretical analysis, the breakdown by activity of this basin is 72% residential, 21% commercial, and 7% industrial. The average daily influent to the plant is 40-44 MGD. Once again, the collection system is comprised of both sanitary and combined sewers.

5. Summary of Source Characteristics

All of the data from the four cities was organized by source category. Table 3 lists the sites in each source category and their basic characteristics. The entire group of residential sources has been treated as one category for the majority of the analyses. Both old and new residential sites were sampled and some differences were seen between these subsets, but the information on the relative amounts of each of these sub categories was not available for the entire service area.

In addition to these sites, four sites were sampled but not used in the overall analysis because they were of mixed source character. They were Sylved and St. George (a hospital) in Cincinnati, Wabash in St. Louis, and Peachtree in Atlanta. Of course, the POTW influent was sampled in each city as well as the tap water. In St. Louis, both the influent and the effluent of the POTW were sampled and an analysis of treatment efficiency is given in the St. Louis report.

C. Demographic and Economic Data

An important aspect of the field sampling program was to obtain demographic and economic data needed to characterize the entire POTW treatment area as well as the individual sampling sites. This information was important in describing the sites and permitting comparisons among sites in different cities, as well as establishing a basis for comparison of pollutant loading, e.g., mass per capita, etc.

An attempt was made to obtain the following data for the POTW treatment area and each individual sampling site:

Table 3

Description of Source Sites Used in Overall Data Analyis

	% Flow				
Sources	RES	COM	IND	Population	Flows (Lps
Residential					
Kirkridge, C	100	0	0	1,056	12.2
Elco, C	_	-	_	600	10.8
Eppingham, S	99	1	0	3,300	15.9
Avocado, S	94	6	0	6,929	31.9
Brightwell, S	96	4	0	1,545	7.8
Northside, A	73	27	0	10,280	104.0
Warren, A	100	0	0	2,416	9.0
Franklin, H	85	15	0	30,762	259.0
Hillside, H	97	3	0	2,312	31.1
Tunxis, H.	100	0	0	1,285	13.9
Brentwood, H	100	0	0	1,527	10.6
Commercial	-				
DelFair, C	_	_	_	2,731	20.4
West Bourne, C	-	-	_	3,201	19.4
Cross Keys, S	82	18	С	2,124	15.2
Northwest, S	55	45	0	3,160	37.2
Lenox, A	21	79	0	1,852	20.0
DeKalb, A	72	26	2	1,868	6.8
Sixteenth, A	18	42	41	12,810	234.0
Clover, H	2	98	0	14	7.1
Potter, H	66	29	5	70,931	603.0
Seneca, H	46	54	0	293	3.5
Industrial					
Frost, S	89	1	10	11,222	119.6
Brown, S	0	2	98	0	6.1
Surrey, A	3	4	93	500	42.0
DeFoors, A	11	12	77	1,951	82.0
Ensign, A	36	43	20	3,533	54.0

C = Cincinnati source

S = St. Louis source

A = Atlanta source

H = Hartford source

- 1) Population -- current or 1978
- 2) Number and ages of residences -- single, multiple, apartment
- Land use -- residential, commercial, industrial, open, etc.
- 4) Characteristics of commercial areas -- number and types of firms including SIC categories where available, size, employment, etc.
- 5) Characteristics of industrial zones -- types of industrial firms, SIC categories, employment.

The goal in this program was to use published or publicly available data, extrapolating where necessary and confirming data sources through observation. In general, raw data were not collected.

1. Sources of Data

Sources of data which were common to most of the cities that were sampled included:

- 1) Census data, usually 1970, sometimes updated to 1975
- 2) Regional Planning Commissions
- 3) Municipal Planning Departments
- 4) Water or Sewer Departments
- 5) Industrial Councils, Chamber of Commerce, or local industry associations
- 6) Municipal housing, real estate, or zoning departments
- 7) 208 and 201 planning studies
- 8) Municipal industry directories

In most cases, the POTW's or sewer authority personnel either identified sources of demographic information or had collected this information for their own purposes.

2. Use of the Data

In general, census data were sufficient to provide both population and housing (residence) estimates. However, most of the census data were from 1970 and had to be updated to the current year, or 1978. The updating was generally accomplished through local population estimates made by planning commissions, municipal planning departments,

or water and sewer departments. Quite often, these departments had made yearly population estimates or had an estimate in 1975 or 1978 which could be used as a basis for extrapolation to the present. In many cases, population estimates were made and updated. However, the number of residences were not updated. In these situations, estimates of number of persons per single family residence and per apartment were used to ascertain the number of residences and apartments from updated population statistics. In other cases, planning departments had updated numbers of residences available but not population data. In these cases estimates of number of persons per single family or apartment residences were again used to obtain population estimates.

One of the difficulties in using census data is that the boundaries of the POTW treatment area and the individual sampling sites do not often correspond to census blocks or census tracts. As a result, estimation was required in determining what fraction of the census block or tract was in each sampling area. Because of the larger size of the POTW treatment area, extrapolation of census data was usually easier since the POTW treatment area usually contained entire tracts and blocks. In the smaller sampling sites, use of census data became difficult. However, in most cases it was found that updated population data were available through the city regional planning commission. In some cases, sewer department personnel or municipal personnel provided estimates of increase in population or residences for selective sampling sites.

Census data as well as data available within each city were usually sufficient to determine the overall age of the sampling site. In general, an old residential site was considered to be one where the dwellings were constructed prior to World War II. A new residential site was defined as one whose dwellings were generally constructed after World War II. In some sampling sites, dwellings were still in the process of construction. This led to uncertainty in the number of houses and the population figures. In several cities, both population and residence data were obtained from several sources; these were compared and averaged to be used in analyzing the pollutant data.

It was generally more difficult to obtain a descriptive characterization of the commercial and industrial zones. This information usually came from municipal planning departments or regional commissions or water and sewer departments. In some cities, planning departments had available lists of all commercial and industrial establishments, their locations, their SIC categories, and employment data. Addresses and employment sometimes were considered confidential information and not provided. In other cases, city agencies accumulated data across several SIC categories before providing them to us. Characterization of the commercial and institutional aspects of sampling zones was usually more difficult than defining the industrial zones. Most water and sewer departments had listings of industrial accounts which could form a basis for a description of the industrial sector. In some cases, visual observations were required to determine the degree of commercialization or the types For most shopping center type commercial zones t of establishments. were sampled, it was possible to obtain from the shopping center developer or planner listings of the types of establishments.

In many cities, additional detailed data were available either from housing and tax assessment officers or from surveys which had been made using proprietary data, for example, R. L. Polk data. These data were generally not used since the level of detail provided in real estate or tax assessment documents was never actually needed, and the Polk priority data were expected to be more expensive than was appropriate in view of the limited amount of information desired.

If detailed characterizations of the commercial and industrial zones were required, much more primary data would have been required and would have added significantly to the cost of the program.

In general, the population data for both individual sampling sites and the POTW treatment area are estimated to be accurate to within $\pm 5-10\%$. In general, the number of residences is estimated to be accurate to within $\pm 10-15\%$ since both sampling sites and cities vary somewhat in the number of individuals per dwelling.

D. Sample Collection

Throughout the performance of this study, all sampling was accomplished by means of manual collection methods. Typically, a field crew of between 12 and 14 people was deployed in the basin of interest for a period of eight days to complete all aspects of the required sampling. The field crew was divided into two alternating shifts, each of which worked a minimum of 12 hours per day. Each shift was further divided into 3 teams of either 2 or 3 people. Two of these crews were directly involved in completing all collection portions of the field work at up to five remote locations. The third crew was responsible for logistical concerns (i.e., sorting, logging in, repackaging of all collected sample increments), as well as maintaining the working status (by providing essential supplies, repairing equipment, etc.) of the remote crews. To a limited extent, the logistics crew also participated in sampling activities by being responsible for the collection of influent, effluent and tap water samples.

Actual collection was completed using a two liter stainless steel graduate (bucket) and a telescopic pole (extended length of 9.75 meters). Normally, the first aliquot obtained was used to determine pH, temperature and to determine whether oxidizing species were present (by means of a potassium-iodide, starch indicating paper test). This volume was then discarded and additional aliquots were obtained to fill a predetermined number of sample bottles. Prior to leaving a site to move onto the next site, an instantaneous flow measurement was made and the results recorded. These flow measurements were used in the laboratory to flow composite all appropriate increments into the final sample for chemical analysis.

E. Flow Measurement

Flow measurements were initially obtained using a depth of flow/
Manning equation approach. In practice, the measured depth of water contained within a pipe can be used to determine the rate of water flow, if certain physical parameters of the pipe (pipe diameter, slope, and roughness coefficient) are also known. However, subsequent to the first basin

studied, the accuracy of this approach, compared to those discussed below, was questioned because values obtained appeared to be unexplainably high. Similar observations were also obtained in the next two cities, but in these instances confirmation that the measured flows were too high was obtained by the results of theoretical flow balances.

The theoretical analysis was based on the assumption that the residential contribution to the basin flow was 100 gallons per day per person, and that all other activities (commercial, industrial, municipal, etc.) discharged as much as they consumed. By obtaining the water billing records of the area, it was possible to estimate a dry weather flow throughout a basin or for any individual site.

As a result of these theoretical analyses, additional flow measurement procedures were evaluated during the fourth city study. Included among the alternative procedures were a direct velocity determination/depth of flow approach, a Palmer-Bowles flume/Manning dipper approach and a Palmer-Bowles flume/depth of flow Manning equation approach.

The results of this study indicated that either of the first two approaches produced more reliable estimates of the actual flow rate than did the depth of flow/Manning equation technique. However, the flume/ Manning dipper technique was somewhat more difficult to implement due to the additional effort required to install both the flumes and the dippers. Therefore, the velocity/depth of flow method was used to correct or recalibrate all depth of flow/Manning equation results that had been obtained from the first three city studies. The flow data used for the analyses in the report are all based on the velocity measured (or corrected) flow for each sampling site.

F. Chemical Analysis

The analytical procedures used were those outlined in the EPA Screening Protocol for Priority Pollutants. A few of the procedures were modified during the studies of each of the individual basins. These modifications are documented in the reports on the four individual drainage basins sampled.

A Quality Control (QC) program was developed for this study in order to establish the reliability of the data. The program was based upon the EPA recommendations. Because the recovery and precision data were available from the QC program, it was possible to modify the analytical procedures where problems were indicated. Consequently, consistently low reporting levels were achieved throughout the study, independent of sample matrix interferences.

Included in Appendix A is a listing of reporting limits, recoveries and precision of measurement for each individual pollutant in the raw wastewater samples. Those data have been summarized by analysis category in Table 4. The data in this summary and in Appendix A demonstrate that the chemical concentration data for the samples have a high degree of reliability. It would not have been possible to achieve this degree of reliability or to document it without the QC program. For a few compounds the EPA screening protocol methods were problemmatic and these are indicated by footnote in Appendix A. There were only three priority pollutants for which it was not possible to obtain data, due to deficiencies in the analysis protocol. They were:

Bis(chloromethyl) ether
Hexachlorocyclopentadiene
2-Chloroethylvinyl ether

Table 4

Chemical Analysis Accuracy and Precision Summary

	METHOD RI	EFERENCE STANDARD*	RAW !	WASTEWATER
	Average	Average	Average	Average
Analysis Category	Recovery	Standard Deviation	Recovery	Standard Deviation
Volatiles	92	18	88	23
Acids	79	16	86	16
Base/Neutrals	79	21	72	19
Pesticides and PCB's	77	14	75	15
Total Cyanides	96	8	91	12
Total Phenols	97	7	96	11
Metals	100	26	94	18
Classical Parameters (7XX Series)	81	14		.

^{*}Standards spiked into pure distilled water.

IV. INTERPRETATION ANALYSIS OBJECTIVES

The entire POTW program has a large number of objectives ranging from an understanding of the sources, types and quantities of pollutants to knowledge of their treatment efficiency and the impact of plant design on that efficiency. This study has focused on those objectives which could be met by a study of the sources of pollutants. Table 5 lists briefly some of the objectives which were developed prior to the initiation of, and during the course of, this study.

The objectives have been grouped into three general categories to reflect the relative importance of each to the overall program goals. During this study it has been possible to directly address and supply information on each of the primary objectives. The characteristics of the source sites and schedule constraints allowed the examination of many, but not all, of the secondary objectives. The data reliability or QA/QC tertiary objective was addressed in detail for flow and concentration measurement and source descriptions. It will be necessary to conduct a study modified in several ways compared with the present study to address the remaining tertiary objectives. The next section (V) presents a detailed analysis of the data organized to address each of the objectives.

Table 5 Interpretation Analysis Objectives*

Primary

Frequency of detection

Presence or absence

Quantity of pollutant (concentration and mass)

Sources of pollutants

Index values for each pollutant—to allow projections

Mass Balances and examination of relative source concentrations

Secondary

Examination of weekday/weekend differences

Determination of site variance within source type

Source variance between sources—are source types different

City/City variance

Correlations between chemicals or parameters

Measurement and analysis (QA/QC) problems

Tertiary

Steady state versus slug flow
Separate/Combined sewers
Type of housing
Time of year
Runoff
Ultimate Source/SIC correlation for industrial sources

*For each of the toxic pollutants

V. RESULTS AND DISCUSSION

The information from this study has been grouped into several categories for analysis, organized primarily by:

Frequency of detection Concentration levels Mass flows and balances Examination of variances and correlations.

The frequency and concentration reviews are straight forward presentations of the basic data. In order to carry out the mass balance, indices of mass contribution have been calculated for each source category. These index values for each major category - residential, commercial, industrial - have been calculated from the sources for all four cities. These average values have been used both to compare with the actual POTW influent values for each of the cities studied and also to calculate some hypothetical mass flows for several different types of hypothetical cities varying in degree of flow from each source type.

The data from these studies are complex. The most accurate interpretations are carried out on a pollutant by pollutant basis. Some overall generalizations are possible, but with caution and with various exceptions.

The data have been grouped and reorganized in a variety of presentations in an effort to reveal the patterns implicit in the data.

A. Frequency of Detection

The number of times any pollutant was detected in each source type--residential (RES), commercial (COM), industrial (IND), tap water and influent--was tabulated and is summarized in Table 6 for each category. That data, coverted to percentages are presented in Table 7. Those percentages are also shown graphically in Figures 6-10.

Table & summarizes the frequencies of occurrence for the two major categories -- organics and metals.

Table 6 Total Number of Observations

	Residential		Commercial		Indostrial		Tap Water		Influent	
Total Number of Samples	47		42		21		12		18	
104. Vinyl Chloride					1					
105. Chlorosthene					1					
108. Trichlorofluoremethane									-1	
100. Acrylanitrile 110. 1,1-Dichlaraethylene			2		8				$-\frac{1}{3}$	
111, 1,1-Dichlorosthene					7				1	
112. Trans-1,2-dichlorosthylene			18		8				5	
113. Chloroform	42 4		42		21		12		18	
114. 1,2-Dishlorosthere	1.8		2		-3				-:-	
115. 1,1,1-Trichioraethene 116. Carbon tetrachioride	14 8		22	-	7				14	
117, Bromodishloremethene	2.0		21		12		12		2	$\neg \neg$
118. 1,2-Dichloropropone					1					
119. Trans-1,3-Dichlerogrepylene			1]				1	
120. Trichlarcethylene	5ª		6		21			——	12	
121. Benzene 123. Dibromochleromethene	10 *	-	21 18		16 12		-,		4	_
124. 1,1,2-Trichloreethene	-				1	-				
125. Bramaform							4			
126. 1,1,2,2-Tetrachlorosthere	10				2				1	
127. 1,1,2,2-Tetrachiorcethylene	36		41		21		3		15	├ ─-┥
129. Chlorobenzene	29 ª		38 2		3		-		14	
130. Ethylbenzene	8 4		21		16		1	 	12	
201, 2-Chlorephanol					2					
203. Phonol	18		17		11				6	
204. 2,4-Dimethylphenol	3				8		 -	<u> </u>	2	
205. 2,4-Dichlorophenol 206. 2,4,6-Trichlorophenol	 		├		2	-	-	-		\vdash
207. p-Chipro-m-greeti	 				-		_	 	1	1
210. Pentachiorophenol	2		6		4				4	
301. Dicklorofensones	6		13		12		L		10	
310. Mitrobenaune 312. 1,2,4-Trichforobenaune	1	<u> </u>	 			ļ		-		
315. Naghthelere	1 4		7		13		-		8	
328. Diethylphthelete	23		15				1		9	
331. Anthrecens/Phonenthrens	2		1		3					
333. Dien-butylpheheliste	16		18		12		3		12	
334. Fluorenthone 335. Pyrent	1	 	 		 	 -				
335. Pyrene 337. Bucythensylphthelete	22	-	23		111		 	 	8	╅
338. Bis (2-othylbary) /di-n-octyl	11	1	16		5	T	2	1	4	
shahelate	+	 				├	 	┼	1	
404. Heptschler 406. Aldrin	2	-	2	 	 	 	 	 		+-1
801. Antimony	164		3		6d		2		7_	
902. Americ	16*		16		9		3		3	
904. Codmium	74		5		8	<u> </u>	1	<u> </u>	10	lacksquare
906. Chrumium	29ª 46ª	┼	24b	 	21	 	11	┼	16	╂╼╼┤
808. Copper 807. Leed	384		36	 	21	 	111	+-	16	╅┈┤
208. Marganese	45=	1	42		21		6	L^-	18	
999. Marany	8.		1		7				3	
S10. Motel	28ª	_	33	ļ	21		3	 	15	
S11. Scholum S12. Shor	26ª		16 20c	 	18	├	1	┼	17	+
512 Thellium	+	+	4	 	1	 	+-	+-	+	1
. 614. Zine	464		42		21		7		18	
601. Total Cyanidae	2		1		16				11	
Out of A6 semiles	1 43	J	40	1	21	1	2	Ь	18	

b. Out of 41 samples c. Out of 39 samples d. Out of 12 samples

Table 7 Percentage Occurrence

dential

	Reside		3		Indus		Tap W		Influ	
Total Number of Samples	47		42		21		12		18	
104. Vinyt Chloride					5					
106, Chloroethane					5					_
108. Trichloroffuoromethans					5				_6	
109. Acrylonitrile									-6	
110. 1,1-Dichloroethylene			5		38				17	
111. 1,1-Dichloroethene			2		33					
112. Trans-1,2-dichloroethylene			43		38		100		28 100	
113. Chloroform	91ª 2ª		100		100		100		11	
114. 1,2-Dichloroethene 116. 1,1,1-Trichloroethene	30ª		52		71		-		78	
116. 1,1,1-Trichicroethene 116. Carbon tetrachicride	3		3		33					
117. Bromodichloromethane	44		50		57		100		11	
118. 1,2-Dichioropropene					5					
118. Trans-1,3-Dichloropropylene			2							
120. Trichloraethylene	118		14		100				67	
121, Benzene	24		50		76				67	
123. Dibromochloromethene	4		43		57		58		· 22	
124. 1,1,2-Trichloroethene					5					
125. Bromoform			1		1	<u> </u>	33	<u> </u>	1	
126. 1,1,2,2-Tetrachloroethane	24	L	 	L	10	 	-	 -	6	
127. 1,1,2,2-Tetrachloroethylene	78ª	<u> </u>	98		100		25	}	83	
128. Toluens	634	<u> </u>	90		100		8_		78	
129. Chlorobenzene	7.4	ļ	15	<u> </u>	14			├	6	——
130. Ethylbenzene	17 ^a	┞	50	├ ──	76	├	8		67	
201. 2-Chlorophenol	 	├	+	├	10	├	├	├	1 22	\vdash
203. Phenol	38	├—	40	├	52 38	├	├	├─	33	
204. 2,4-Dimethylphenol	-6	├	╂──	┼──	10	 	┼	 	 ** -	┝╼┪
208. 2,4-Dichlorophenal 208. 2,4,6-Trichlorophenal	\vdash	┼	+	 	10	 	 	 	 	\vdash
206. 2,4,6-Trichlorophenol 207. p-Chloro-m-cresol	├ ──	+-	+	 	+**	 	†	1	6	
210. Pentachiorophenol	6	┼─	14	†	19	1	1		22	
301. Dichlorobenzenes	13	†	31	1	57			1	56	
310. Nitrobenzerie	2	†	1	1						
312. 1,2,4-Trichlorobenzens	1 2	1	2		T					
315. Nephthelene	9		17		62		L_		44	
326. Diethylphthelate	49		36			<u> </u>	1.	↓	<u> 50</u> _	
331. Anthrecene/Phonenthrene	1		2	↓	14	 	 	 	-	
333. Di-n-buty/phthalete	34	1	43	↓	57_	↓	25_	┼	67	
334. Fluorenthene	1.2	┼	-	┿┈	- 	┼	╂	┼		┼
336. Pyrene	12	↓		╁	-} -	┿	┼	 	1	┼─
337. Butylbenzylphthelete	47	+	55	╁┯	- 52 -	+	+	+	44	
338. Sie (2-estrythexyl)/di-n-ostyl phtholete	23	1_	38	<u> </u>	24		17		22	
404. Heptachlor			5						6	4
408. Aldrin	4						1	1	1	
501. Antimony	354		7		50d		17	1_	39	₩
802, Aresnic	354		38	1	43		25	-	44	
804. Cadmium	158	_	12	4	38	4-	1.	 -	36	┼
806. Chromium	638		598	∔—	100	+	- 8	╅	89	+
908. Copper	1994		100	+-	100	┽—	192	+	1200	+
807. Leed	83		183	+-	100	+-	33	+	100	+
SOS. Manganese	1004	_	100	+-	100 33	+	150	+	17	+
908. Mercury	17*			+	100	+-	25	+-	83	+
51Q. Nichel	614	_	79	+-	14	+-	33	+	28	+-
511. Selenium	57ª		36 31¢	+	86	+-	133	+-	95	†
512. Silver 513. Theliters	+**	+	10	+-	5	+-	┪	1	1 7	+
\$14. Zinc	100	1	100	\top	100	1	58		100	1
801. Total Cyanidas	14	1	2	1	76	I^-	T	\mathbf{I}^{-}	61	\mathbf{I}_{-}
802. Total Phanels	93		95		100		17		100	$oldsymbol{\mathbb{T}}$
and the complete	-	-								

Out of 46 camples Out of 41 samples Out of 39 samples Out of 12 samples

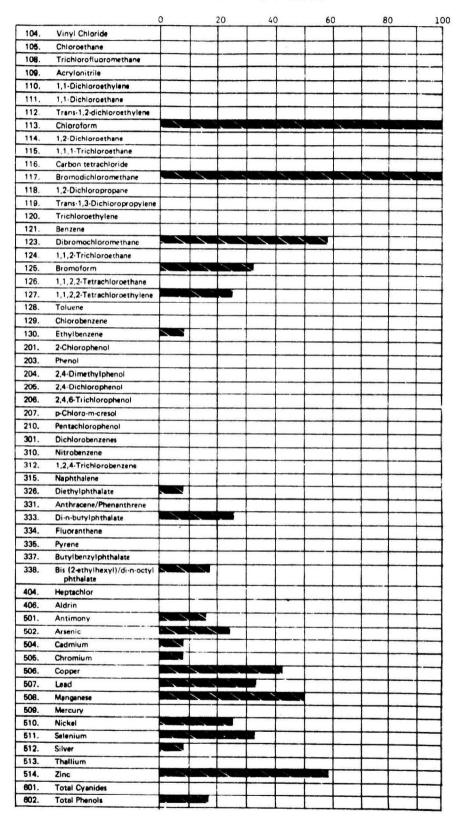


Figure 6: Frequency of Occurrence (%), Tap Water

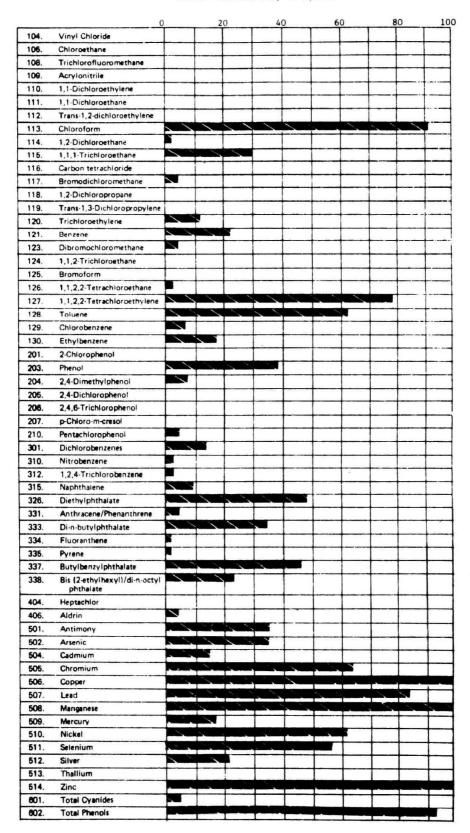


Figure 7: Frequency of Occurrence (%), Residential

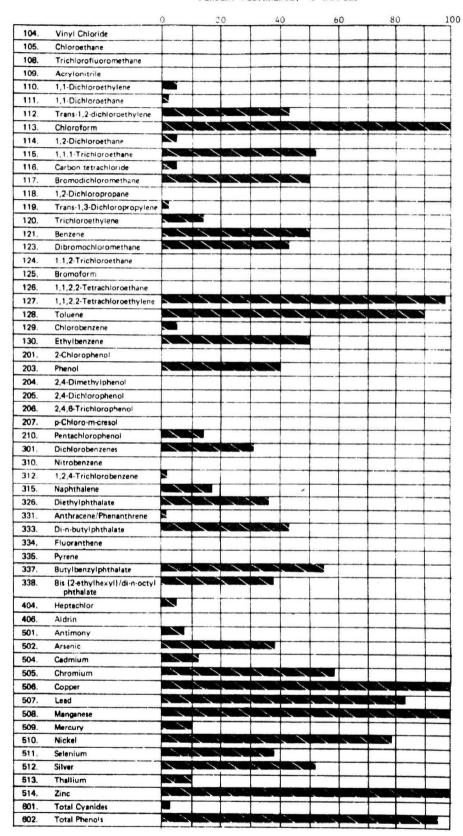


Figure 8: Frequency of Occurrence (%), Commercial

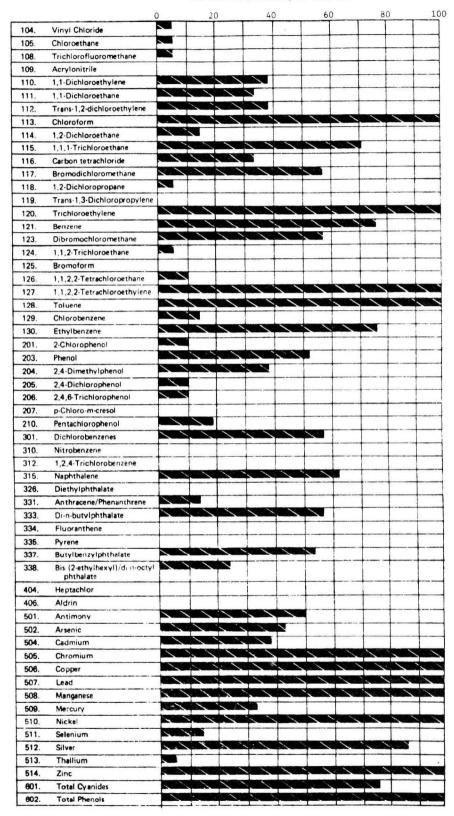


Figure 9: Frequency of Occurrence (%), Industrial

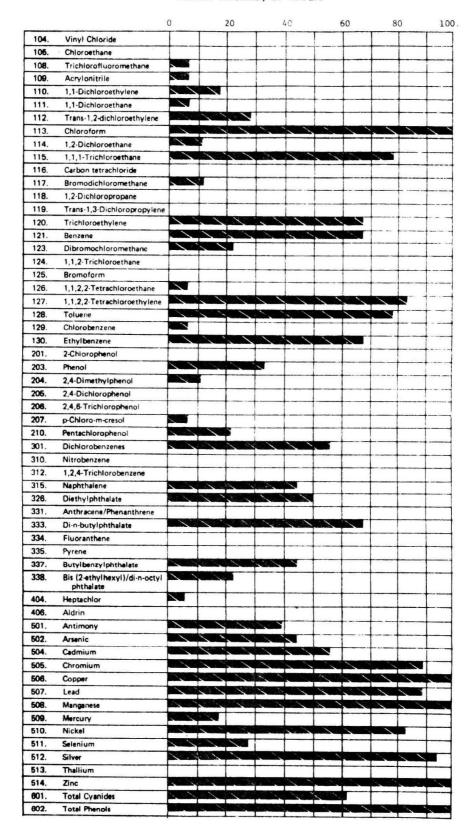


Figure 10: Frequency of Occurrence (%), POTW Influent

Table 8
Summary of Overall Frequency Observations

	C	rganics	$(42)^{1}$		•	Metals (etals (13) ² >90%			
	Total ³	<u>></u> 90%	<u>></u> 50%	<10%	Total ⁴			<10%		
Tap Water	10	2 ⁵	3	2	11	0	2	3		
Residential	26	1	3	14	12	3	7	0		
Commercial	27	3	8	9	13	3	7	1		
Industrial	33	4	5	13	6	8	1	0		
Influent	28	1	10	7	11	4	8	0		

¹⁴² organic pollutants (volatiles, acids, base/neutrals).

²13 metals (12 priority pollutants + manganese).

³Number of organic pollutants observed in each source category.

⁴Number of metals observed in each source category.

⁵Values are number of pollutants, out of the total in each analysis category, which were observed equal or greater than 90% of the time, equal or greater than 50% of the time or less than 10% of the time.

A total of 42 organic pollutants were observed at some time in one or another of the sources, but a maximum of 46 could be present—four of the organics are not resolved in the analysis scheme and are reported as groups.* The following list is a summary of the number of priority pollutants seen at some time in these sources.

Volatiles 24
Acids 7
Base/Neutrals 11 (could be 15*)
Pesticides 2
Metals 12 plus manganese
Total Cyanides
Total Phenols

The list of 67 priority pollutants given in Table 9 along with their reporting limits, were never detected in any of the samples in any city. In general the reporting limits refer to the concentration level which the analysis protocol was designed to measure reliably. Such is the case for instance for the pesticides at 1 μ g/L and most of the other pollutants at 10 μ g/L, and most of the other pollutants at 10 μ g/L. Values higher than 10 μ g/L represent detection limits.

Table 10 gives a list of pollutants (20) which were observed 0-3 times in at least one of the cities. Because the mass data for these chemicals was so sparse, they were excluded from the subsequent concentration and mass flow interpretations. The data for methylene chloride was excluded from these analyses because it is such a ubiquitous contaminant.

Table 11 gives a list of the 40 toxic pollutants which have been examined in detail in the subsequent sections of this report. The six classical parameters (7XX series) of ammonia, oil and grease, TSS, TOC, COD and BOD were also included in the detailed analysis. The data in Table 11 are given in terms of the number of times a pollutant was detected in a city. Only pollutants detected greater than three times in at least one city are included.

^{*}The unresolved groups are:
Dichlorobenzenes - 3 isomers
Anthracene and Phenanthrene
Bis(2-ethylhexyl)phthalate and di-n-octyl phthalate

Table 9
Sixty-Seven (67) Pollutants Never Derected in Four Cities

Compound	Reporting Limit µg/L	Compound	Reporting Limit $\mu g/L$
101 Chloromethane	5	340 Chrysene/Benzo(a)anthracene	5-10
102 Dichlorodifluoromethane	5	342 3,3'-Dichlorobenzidine	10
103 Bromomethane ^a	5	343 Benzofluoranthenes	1–5
107 Acrolein	1-7	345 Benzo(a)pyrene	5–10
122 Cis-1,3-dichloropropylene	1	346 Indeno (1,2,3-c,d)pyrene	5
202 Nitrophenol	10-15	347 Dibenzo(a,h)Anthracene	5–10
208 2,4-dinitrophenol ^b	20-40	348 Benzo(g,h,i)perylene	5–10
209 4,6-dinitro-2-cresolb	20-40	349 TCDD	
211 4-Nitrophenol ^b	10–25	401 alpha-BHC	1
304 Hexachloroethane	10-20	402 gamma-BHC	1
305 Bis(chloromethy1)ether ^a		403 beta-BHC	1
306 Bis(2-chloroethyl)ether	10-20	405 delta-BHC	1
307 Bis(2-chloroisopropyl)ethe	r 10	407 Heptachlor epoxide	1
308 N-Nitrosodimethylamine b	10-70	408 Endosulfan I	1
309 Nitrosodi-n-propylamine	10-20	409 DDE	1
311 Hexachlorobutadiene	10	410 Dieldrin	1
313 2-Chloroethyl vinyl ethera		411 Endrin	1
314 Bis(2-chloroethoxy)methane	10	412 DDD	1
316 Isophorone	10	413 Endosulfan II	1
317 Hexachlorocyclopentadiene ^a		414 DDT	1
318 2-Chloronaphthalene	10	415 Endrin aldehyde	1
319 Acenaphthylene	10	416 Endosulfan sulfate	1
320 Acenaphthene	10	417 Chlordane	1
321 Dimethyl phthalate	10	418 Toxaphene	1
322 2,6-Dinitrotoluene	10	419 PCB-1221	1
323 4-Chlorophenyl phenyl ether		420 PCB-1232	1
324 Fluorene	10	421 PCB-1242	1
325 2,4-Dinitrotoluene	10	422 PCB-1248	1
327 1,2-Diphenylhydrazine	10	423 PCB-1254	1
328 N-Nitrosodiphenylamine	10	424 PCB-1260	1
329 Hexachlorobenzene	10	425 PCB-1016	1
330 4-Bromophenyl phenyl ether	10	503 Beryllium	1-3
336 Benzidineb	10-20		

These compounds were not detected by the EPA method. bChromatographic problems encountered with these compounds.

Table 10

Priority Pollutants Never Observed Greater Than Three Times In Any One City*

	·									
		Number of Times Detected								
		Cin.	St.L.	Atl.	Htfd.					
104.	Vinyl chloride	-	-	1	-					
105.	Chloroethane	-	-	1	-					
108.	Trichlorofluoromethane	-	-	2	-					
109.	Acrylonitrile	-	-	1	-					
118.	1,2-Dichloropropane	-	2	1	-					
119.	Trans-1,3-dichloropropylene	-	2	-	_					
124.	1,1,2-Trichloroethane	-	-	1	-					
126.	1,1,2,2-Tetrachloroethane	2	- .	2	-					
201.	2-Chlorophenol	-	1	1 .	-					
205.	2,4-Dichlorophenol	-	1	1	•					
206.	2,4,6-Trichlorophenol	-	1	1	-					
207.	4-Chloro-3-cresol	-	-	-	1					
310.	Nitrobenzene	-,	-	-	1					
312.	1,2,4-Trichlorobenzene	-	-	-	2					
331.	Anthracene/Phenanthrene	1	1	3	1					
334.	Fluoranthene	-	-	-	1					
335.	Pyrene	-	1	-	-					
404.	Reptachlor	3	-	-	_					
406.	Aldrin	-	2	-	_					

^{*}Including influent, tap, and source samples. Dash means not detected.

Table 11

Pollutants Selected for Detailed Analysis - Frequency of Detection

Compounds detected greater than 3 times in at least one city^{a, h}

		Number of Times Detecte						
		C(38)	s(56)	A(32)	H(28)			
110.	1,1-Dichloroethylene	1	3	10	-			
111.	1,1-Dichloroethane	-	2	7				
112.	Trans-1,2-dichloroethylene	-	11	19	1			
113.	Chloroform	37	55	31	25			
114.	1,2-Dichloroethane	2	1 .	4	1.			
115.	1,1,1-Trichloroethane	10	35	9	9			
116.	Carbon tetrachloride	-	2	: 7 .	-			
117.	Bromodichloromethane	16	34	2.	4			
120.	Trichloroethylene	-	21	22	4			
121.	Benzene	16	41	10	1			
123.	Dibromochloromethane	13	37	1	-			
125.	Bromoform	4	1	-	-			
127.	1,1,2,2-tetrachloroethylene	24	55	31	18			
128.	Toluene	21	54	27	11			
129.	Chlorobenzene	1	6	4	-			
130.	Ethylbenzene	16	24	19	3			
203.	Phenol	. 13	28	14	3			
204.	2,4-Dimethylphenol	1	1	8	· -			
210.	Pentachlorophenol	2	-	18	-			
301.	Dichlorobenzenes	2	34	7	5			
315.	Naphthalene	7	15	. 8	2			
326.	Diethylphthalate	27	21	1	3			
333.	Di-n-butylphthalate	25	29	8	. 7			
337.	Butylbenzylphthalate	11	43	11	6			
338.	Bis(2-ethylhexyl)/di-n-octyl		7					
	phthalate	24	10	6-	1			
501.	Antimony	4	26	5				
502.	Arsenic	36	5	. 4	. 11			
504.	Cadmium	11	15	7	2			
505.	Chromium	8	53	30	9			
506.	Copper	37	56	32	27			
507.	Lead	20	56	32	18			
508.	Manganese	36	54	32	25			
509.	Mercury	3	9	8	2			
510.	Nickel	13	55	32	13			
511.	Selenium	22	38	3	3			
512.	Silver	20	2 5	2 3	5			
513.	Thallium	4	2	••	••			
514.	Zinc	35	55	32	26			
601.	Total Cyanides	4	20	12	1			
602.	Total Phenols	35	57	30	20			
0 U2.	TOPOT I HEHATS							

a_C = Cincinnati: 38 24-hr composite samples

Sources, influent and tap water samples included; field blanks not included.

49

S = St. Louis: 56 24-hr composite samples

A = Atlanta: 32 48-hr composite samples

H = Hartford: 28 48-hr composite samples

bMethylene Chloride (106) was observed as a contaminant in almost all of the samples.

B. Observed Pollutant Concentration Levels

1. Concentrations

The original concentration data for the 24 or 48 hour composite samples were averaged on a flow-weighted basis to produce a single number for the six-day sampling period at each site. These concentration values are summarized in Tables 12-16 along with the grand average value and, where appropriate the standard deviation, for the tap water, residential, commercial, industrial, and POTW influent samples. The per capita (mg/person/day) discharge for residential sources is given in Table 17.

For many of the pollutants in the residential and commercial categories, the standard deviation is about the same value as the average. Although an average industrial concentration value has been calculated for the purposes of testing mass balances, there may not be real significance to the concept of an average industrial value.

For the purposes of developing a projection model, each source type was considered to be part of the same overall population. Average concentration data were calculated for each category by averaging similar source sites within a city and then averaging between the cities. For the residential per capita values, the per capita rate was calculated for each site and then averaged on a population weighted basis, within the city, followed by a straight average of the cities.

The average concentration values for each of the source categories are shown in Table 18. From this table, it is clear that the industrial sources are the most intense for most of the chemicals, but the residential sources are important contributors of some pollutants such as diethyl phthalate and copper. Some of the pollutants which show residences as the most intense source (antimony, arsenic) are present at very low levels. A number of pollutants show equivalent intensity levels across source types, such as chloroform, manganese, and the classical parameters. The source comparisons are best done on a mass basis, as are presented in Part C of this section.

POLLUTANT	Cincinnati	St. Louis	Atlanta	ilartford	Average	Standard** Devia- tion
ilo 1,1-DICaloroeTayLene	. C	.0	.0	.0	•0	.0
111 1,1-DICaLOROLL'ANE	.0	.0	.0	, C	.0	•0
112 TRANG-1, 2-DICHLOROETHY LEWE	•0	•0	.C	.0	.0	.C
113 CalükGrünk	39.5	21.0	21.5	26.3	27.1	8. 6
114 1,2-DICALONUETHALL	. C	.0	.0	.0	.0	.C
115 1,1,1-In ICHLOROWI HANE	.C	.0	.0	2.5	.6	1.3
110 CARBON TETRACHLORIDE	• C	.C	.0	.0	.0	• C
117 EAGGOULCALOROMETHANE	10.8	12.5	3.5	2.5	8.8	6.9
120 Thicalchostailene	.0	.0	.0	.0	.0	.0
1/1 behabit	.0	•0	•G	.0	0.	•0
123 vIbhomochlohometrane	15.0	8.0	, <u>ē</u>	ن.	5.9	7.1
125 Bhowerdh	1.8	1.5	.0	.0	.8	.9
127 1.1.2,2-TETRACALOROETATLENE	.0	1.5	1.5	.0	.8	.9
126 TOLUENE	.0	1.0	.0	.0	.3	.5
129 ChLUNOBEN LENE	.0	• G	.C	.0	.0	.0
130 Erall bendene	.3	4.0	.0	.0	-1	.1
203 PhENOL	.c	.0	.0	•c	.0	• C
204 2, 4#DineThilPnENOL	.0	.0	.G	.0	.0	.0
216 PENTACHLUNOPHENOL	.0	.C	.0	.0	.0	•0
301 DICHLORODENZENES	.0	.0	.0	.0	.0	•0
315 WARNTNALENE	.C	.0	.C	.0	.0	.0
320 LIETHIL PHINALATE	3.3	.0	.0	.0	.8	1.6
333 LI-N-BUTTL PaTAALATE	14.3	.0	.0	3.8	4.5	6.7
337 BUTYL BENGYL PHTHALATE	.0	.0	.0	.0	0.	• 0
338 bls(?=ETHYLHEXYL)FnfhALATE	16.5	.0	.0	•0	4.1	8.3
501 ANT INCOM	.0	12.G	. C	. G	3.0	6.0
502 AKSENIC	ö. 3	.0	.0	3.	1.6	3.1
5Gu CADMIUM	2.0	.0	.0	.0	.5	1.0
505 CHKGI-LUM	10.0	.0	.0	.0	2.5	5.0
506 COPPER	28.0	6. 5	22.5	56.8	28.4	21.0
567 LEAD	• G	28.5	13.0	.0	10.4	13.5
500 MANGANESE	7.5	.0	6.5	6.3	5.1	3.4
509 MERCURY	.0	.0	.0	.ŭ	-0	.0
510 NICKEL	13.0	.0	4.0	.0	4.3	6.1
511 SELENIUM	3.5	8.G	• G	.0	2.9	3.8
512 SILVER	1.0	.0	.0	,6	.3	• 5
513 PhALLIUM	.0	•0	.0	.0	.0	.0
514 21 WC	27.5	14.5	210.0	15.8	66.9	95.6
6C1 TOTAL CYANIDES	.G	.0	.0	٥.	.0	-0
602 TOTAL PALNULS	8.0	.0	.0	.0	2.0	π-0
703 AMMONIA	.1	• n	.0	.3	.2	•2
704 UIL AND GREADE	.0	.0	.0	.0	.0	•0
705 <i>T55</i>	1.3	.0	.0	.0	.3	.6 3.2
766 10C	4.3	8.0	4,5	.3	.0	.0
707 COD	.0	.0	.0	•0	.0	.0
708 BOD	.G	.0	.0	.0	1 .0	•0

^{*} Classicals in mg/L. **Standard Deviation = $\frac{\sum \bar{X} - \bar{X}}{n-1}$

Table 13 Residential Concentration Summary $(\mu g/L)^*$

	•	Cian.		St. Lou	ite	At	lesta		Eert	ford			
: · ·	Kirkridge	filee	a de la companya de l	rocado	Fightwell	fortheids	Merron	Franklin	Hilleide	Purrocia	Frantiscod	A	Standard Devia-
POLLUTANT	śń	<u> </u>	<u> </u>	. 4								Average	
110 1,1-Dicklonouverland	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	
111 1,1-DECREOROFFRANCE	.0	.0 .0	.0 .0	.0	.0	.0	.0	.0 .0	.0 .0	.0	.0	.0	
112 TROUG-1, 2-DECELOROSTRYLENS 112 CELOROFORM	.0 1.5	2.6	2.9	3.4	3.3	3.7	4.7	3,3	4.6		2.7	3.0	.9
114 1.2-DECELORDETEARS			.0	.0	.7	.0	.0	.0	.0	.0	.0	.1	.1
115 1.1.1-TRECULOROUSHAME	.0	.2	4.6	.6	1.2	1.3	.0	.0	.9	.0	24.2	2.3	2.0
116 CARROW THYMACELORIDE	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0
117 BRONDSECTEONOMETRAIS	.0	.1	.0	.0	.3	.0	.0	.0	.0	.0	.0	.0	.1
120 THEILONOSSYTLENE	.0	.0	.0	1.6		2.1	.0	.0	.0	.0	.0	.4	.\$.4
121 BENEAUS	.0	.3	.7	.2	1.5	.0	.0	.0	.0 0.	.0 .0	.0	.2 .0	
123 DEBROICKLOROWYHAIR 125 DEGWOTODY	.0	.1 .0	.0 .0	.0 .0	.3	.0	.0 .0	.0 .0	.0	.0			
125 BRONOFORM 127 1,1,2,2-TETRACELOROESHYLINE	.0	1.7	21.3	15.0	15.9	8.6	2.0	3.9	1.6	.3	.7	6.3	7.6
126 FOLKERE		4.5	2.1	5.4	11.9	1.3	.5	1.9	.0	.0	.0	2.6	2.7
120 CHLOROWENEERS	.0	.2		.0	8.	.0	.0	.0	.0	.0	.0	.1	.1
130 KERYL MARKEN	. 2		.0	.0	3.4	.0	.3	.0	.0	.0	.0	.4	. 5
203 PHENOL	1.5	23.6	2.1	15.4	6.2	.0	5.1	.0	.0	.0	.0	5.8	5.6
204 2,4-DINETHILPHENOL	.0	.7	.0	2.1	0	.0	3.7	.0	.0	.0	.0	.7	.8 2.4
210 PRITACILOROPHUNOC	.0	.0	.0		.0	9.7	.0	.0	.0	.0	.0	1.2 2.6	5.5
301 DECELOROUSEURS	.0 .0	.0 1.2	.0 .0	12.8	20.3 19.2	.0	.0	.0 5, 3	.0 .0	.0 .0	.0	2.1	2.9
315 RAPUTRALENS 226 RESSYL PUTRALATE	16.8	35.6	13.3	11.3	3.4		.0	.0		.0	15.0	9.8	11.6
222 AZ-W-METL PETRALATE	18.6	13.1	8.5	23.8	7.1	.0	5.0	.0	.0	14.5	3.2	9.0	6.5
897 BUTTL BEHETL PETRALATE	7.5	9.8	13.2	11.9	24.7	.0	.0	.0	.0	.0	8.3	6.8	7.5
336 BIS(2-ETHYLHEXYL)PHYMALATE	7.4	13.9	2.9	.0	16.3	.0	.0	.0	.0	.0	.0	6.8	8.1
SOL ARTINOUT	.0	3.2	12.6	12.4	.0	0	1.9	.0	.0	.0	.0	2.7	3.8 8.7
SOR ARGENIC	19.6	15.9	.0	.0	.0	.0	.0	.0	.0	2.5	3.4	4.4	1.6
504 CADAZUN	.0	.5	1.4	1.0	6.2	.0	1.6	.0	0	.0 .0	13.9	1.8	6.1
SOS CERCILINA	8.5 139.6	23, 2 95, 5	10.5 30.9	7.6 35. 5	10.3 139.3	24.2 38.6	7.0 44.2	84.0 74.8	13.3 41.1	61.0	67.0	72.1	32.4
SOS COPPER SOT LEAD	.0	48.0	55.3	54. 6	792.1	41.7	41.5	94.8	5. 2	13.2	29.2	97.3	135.8
SOR MARCANERS	84.6	131.2	199.0	219.6	107.3	180.5	245.9	78.9	199.1	64.0	120.0	153.0	50.2
SOS MERCUET	.0	.3	.5	2.1	.9	.0	.0	.0	.0	.0	.7	.4	. 5
510 MECKEL	.0	4.0	9,8	8.1	10.1	3.0	5.6	2.8	.0	.0	1.7	4.2	3.7 4.7
511 SELENTON	3.9	5.3	7.8	13.7	1.5	.0	.0	.0	.7	.6	.0	3.8	2.5
S12 SILVER	.7	.2	4.6	11.4	7	5.5	.0	.0	.0	.0	.0	2.2	0
513 THALLIUM	.0	.0		.0	.0 1215.3	.0 133.6	.0 207.2		.0 49. 6	54.3	121.0	214.0	175.7
514 SINC 601 TOTAL CYANIDES	130.7 .0	134.2	102.8	97.2 .0	2.0	7.5	207.2	100.2	-3.0	.0	.0	1.1	1.8
602 TOTAL PRINCES	29.8	56.1	35.7	40.9	37.1	20.4	20.7	40.9	22.1	,0	24.9	30.8	11.3
703 AMERICA	24.4	21.7	16.2	21.2	16.6	9.1	9.0	4.7	2.5	7.7	12.4	14.2	7.6
704 OIL AND GREASE	53.7	69.0	41.5	54.1	489.8	29.3	33.9	33.8	15.2	19.8	17.1	77.4	80.3
705 <i>185</i>	151.1	330.6	101.6	133.8	147.7	77.0	294.0	44.9	18. 3	38.5	190.5	156.8	72.5 27.3
706 TOC	90.3	110.4	90.0	124.6	110.0	59,1	74.2	46.2	29.7	62.1	64.8	81.5	47.7
797 COD	277.0	356.4	194.5	292.5	310.2	361.8	182.3	170.3	112.4	221.0	299.7	263.8	42.1
708 <i>BOD</i>	117.9	176.0	115.9	183,6	159.9	75.4	90.2	60.0	16.3	75.1	138.0	113.3	700 0

^{*}Classicals in mg/L-

			Cinn.	۶t.	Louis		tlanta		Ha	rtford			
	POLLUTANT	Delfair	West Bourne	Cross Keys	Northwest	Lenox	DeKalb	Sixteenth	Clover	Potter	Seneca	Average	Standard Devia- tion
110	1.1-DICHLOROETHYLENE	.0	.1	1.3	. 8	.0	.0	.0	.0	.0	.0	.3	.5
	1.1-DICHLOROETHANE	.0	.1	.0	.5	.0	.0	.0	.0	.0	.0	.1	.1
	TRANS-1, 2-DICHLOROETHYLENE	. 4	.0	2.7	.0	2.0	9.2	2.2	.0	.0	.7	1.5	2.0
113		6.0	4.2	5.3	5.8	6.6	16.2	6.4	9.0	4.3	5.1	6.7	2.0
114	1,2-DICHLOROSTHANE	.0	.4	.0	.0	.0	.0	.0	.0	.0	1.0	.1	.2
115	1.1.1-TRICELOROETHANS	1.7	3.2	.8	10.3	2.4	1.7	.0	.0	7.1	.0	2.9	1.8
116	CARBON TETRACHLORIDE	.0	.0	.0	.0	.0	.0	1.1	.0	.0	.0	.1	.2
117	BRONODICH LORONETHAME	2.3	1.1	2.5	2.0	.0	.0	.0	.0	.0	.0	1.0	1.2
120	TRICHLOROSTHYLENE	.0	.0	.0	.0	.0	73.0	80.9	. 0	.3	.0	12.8	25.6
121		. 4	10.3	1.0	4.6	.0	.0	1.3	.0	.0	6.2	2.7	2.1
123		1.6	.7	2.0	1.5	.0	.0	.0	.0	.0	.0	.7	.9
	BRONOPORN	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0
	1, 1, 2, 2-TETRACHLOROETHYLENE	6.7	3.0	45.2	20.9	9.5	#1.6	2.6	6.3	25.0	18.0	21.4	13.3
126		7.4	27.5	6.0	24.4	2, 1	4.3	1.1	6.0	7.5	12.4	11.0	6.8
129	*	.0	.0	.0	.3	.0	.0	.0	.0	.0	.0	.0	.1
130		3.6	11.5	.2	3.6	5.9	.0	.0	.7	.6	1.0	3.0 4.5	3.1 .8
	PHENOL PROPERTY CONTROL	5.2	4.7	6.1	4.3	3,5	6.6	.0	13.7 .0	.0	.0	.0	.0
204		.0	.0	.0	.0	.0	.0	27.8	.0	.0	.0	5.8	11.5
210	PENTACH LOROPHENOL	.0	.0 .0	0 25.9	.0 22.9	41.5 .0	.0 7.0	.0	.0	6.6	.0	7.5	11.3
301	DICHLOROBENZENES NAPHTHALENE	2.4 1.6	13.0	.0	4.5	.0	.0	.0	.0	3.1	.0	2.6	3.2
326		13.0	15.6	12.1	5.3	.0	.0	.0	.0		.0	5.7	7.0
333		38.2	17.6	4.8	15.7	4.6	.0	4.5	17.3	.0	.0	11.7	11.2
337		13.6	5.7	16.3	13.5	.0	33.0	.0	12.3	4.6	3.6	10.6	3.3
338	BIS(2-ETHYLHEXYL)PHTHALATE	22.3	14.3	4.5	4.1	.0	20.5	.0	.0	4.3	.0	7.7	7.4
	ANTINONY	1.7	.0	.0	.0	.5	.0	.6	.0	.0	.0	.3	.4
502	ARSENIC	9.6	7.9	.0	.0	.0	.0	.0	.0	1.2	3.7	2.6	4.2
504	CADMIUN	2.0	1.8	.0	.0	.0	.0	.0	.0	1.2	.0	.6	.9
	CHRONIUM	.0	.0	6.8	20.8	423.8	6.6	113.1	.0	83.5	13.1	56.8	84.0
	COPPER	35.3	57.0	41.7	32.2	51.3	52.2	45.2	95.1	93.3	67.3	54.5	21.2
	LEAD	.9	31.4	87.1	88.3	49.3	38.7	108.6	24.7	58.7	5.6	49.8	32.8 106.2
	HANGANESE	449.3	318.0	91.5	269.2	178.2	154.9	183.7	25.0	130.1	333,3 2,7	224.8	.4
	MERCURY		.2	,7	.5	.0	.0	.0	.0 6.3	.0 36.7	5.2	12.4	3.9
	NICKEL	12.1	18.2	7.8	7.5	5.5 1.8	18.3 .0	a.7 1.0	•.u	30.7	.6	3.3	4.2
	SELENIUM SILVER	4.2 1.4	1.3 1.0	18.1 1.1	3.8	9.1	.0	12.6	.0	-1.6	3.5	2.9	3.0
-	THALLIUM	.4	.5		.3	.0	.0	.0	.5	.0	6,0	1 .1	. 2
	ZINC	142.5	133.7	114.9	75.5	171.9	134.5	172.	218.2	183.0	76.5	138.1	30.3
	TOTAL CYANIDES	.0	.0	,0	1.7	.0	.0	.0	.0	.0	.6	.2	.4
	TOTAL PHENOLS	38,0	35.7	32.0	30.7	43.5	40.5	36.0	38.5	17.3	62.5	27.0	4.0
	AMMONIA	14.2	14.8	8.7	2.9	22.6	5.8	4.3	7.5	7.6	9,2	10.7	2.2
_	OIL AND GREASE	300.7	121.5	48.7	41.4	123.8	69.6	52.0	125.2	139.5	19.5	109.0	71.4
	TSS	133.3	129.3	113.6	34.4	121.3	163.6	201.2	173.1	56.9	62.1	122.4	30.7
	TOC	84.3	74.2	91.0	120,7	190.3	184.0	60.6	179.9	55.9	73.0	106.2	26.0
707	COD	256.9	246.9	233.4	216.2	401.4	566.5	131.0	920.0	277.3	022.3	345.0	131.6
708	BOD	108.6	98.4	143.1	137.8	245.4	262.8	78.8	350.9	74.4	100.5	1 160.0	39.7

^{*}Classicals in mg/L.

Table 15 $\label{eq:Table 15} Industrial Concentration Summary \left(\mu g/L\right)^*$

•	St.	Louis		Atlanta		
POLLUTANT	Frost	Brown	Surrey	DeFoors	Ensign	Average
	.0	.0	3.0	49.2	17.3	11.6
110 1,1-DICHLOROSTHYLENS 111 1,1-DICHLOROSTHANS	.0	.0	1.0	7.9	.6	1.5
111 1,1-DICELOROSIMANS 112 TRANS-1,2-DICHLOROSTHYLENS	.2	.0	.1	56.B	13.0	11.7
113 CHLOROFORM	19.8	6.7	5.0	19.6	7.5	12.0
114 1,2-DICHLOROSTHANS	.0	.0	.0	1.9	1.6	
115 1,1,1-TRICHLOROETHANE	8.1	.0	73.2			85,1
116 CARBON TETRACHLORIDE	.5	.0	14.8	151.7	3.2	28.4
117 BRONODICHLORONETHAND	2.3	4.0	.0	.0	.0 18.2	
120 TRICHLOROSTHYLENS	22.5	19.2	4.0	67.6 1.8	1.0	
121 BENZENE	2.0	.6	. 8	.0	.0	1.2
123 DIBRONOCHLORONETHANB	1.7	3.0	.0 .0	.0	.0	
125 BRONOFORN	.0	.0 14.4	123.9	204.8	43.5	69.9
127 1,1,2,2-TETRACH LOROSTHYLENE	17.1 5.4	29.6	123.8	63,3	74.2	52.3
128 TOLUENE	.0	.0	.0	5.6	.0	.9
129 CHLOROBENZEVE	.5	1.9	258.3	228.9	111.9	
130 ETHYL BENZESE	7.3	.0	551.8	232.5	19.8	135.8
203 PHENOL 204 2,4-DIMETHY LPHENOL	.0	.0	301.3	130.9	11.6	74.0
210 PENTACHLOROPHENOL	.0	.0	9.2	51.3	.0	10.1
301 DICHLONOBENIEWS	27.6	16.6	.0	2187.7	5.2	•
315 NAPHTHALENE	9.2	11.4	194.9	78.6	.0	50.7
326 DIETHYL PHYHALATE	.0	.0	.0	.0	.0	.0
333 DI-W-BUTYL PHYMALATE	81.1	98.5	42.7		.0	1
337 MITYL HENZYL PHINALATS	17.6	1.1	604.0	.0	377.3	
338 BIS(2-BTWYLHEXYL)PUTHALATE	•0	.0	173.7	84.1	.0	43.0 1.7
501 ANTINONY	8.8	.0				1 -
502 ARSENIC	.0	5.8	2.0	8.5	.0	
504 CADMIUM	10.8	4.0	17.1	1.7 2136.9	33.0	713.2
505 CHRONIUM	134.4	18.3			342.5	124.0
506 COPPER	70.6	41.5	75.4 1224.5		91.4	323.7
SOT LEAD	95.9	78.3 60.3	165.5	388.1	441.2	232.1
508 MANGANESE	20 5. 0		1.8	5.9	3.4	1.9
509 MERCURY	10.3	5.8	596.7	22.1	8.8	108.7
510 NICKEL	3.0	.4	.0	.0	.0	9
511 SELENIUN	2.8	567.2	1.3	25.7	13.6	
512 SILVER 513 TBALLIUN	.3	.0	.0	.0	.0	
514 ZINC	652.3	122.4	3356.0	493.4	148.7	
601 TOTAL CYANIDES	56.2	1.7	236.4	48.8	172.4	
602 TOTAL PREMOLS	28.6	26.2	446.1		181.5	
703 AMOUZA	10.7	21.1	3.6	8,4	4.1	
704 OIL AND GREASE	21.6	5.5			62.0	
705 <i>TSS</i>	97.4	47.9	434.1		91.5	
706 20C	80.1	30,1				540.3
707 COD	192.6	144.3				216.1
700 800	59. 3	47.4	388.3	727.7	44714	. ,

Classicals in mg/L.

Table 16 POTW Influent Concentration Summary ($\mu g/L$) *

POLLUTANT	Cincinnati	St. Louis	Atlanta	Hartford	Average
110 1,1-DiCalokobraflebe	.0	. 9	6.0	.0	2.4
111 1,1-LICALURGETHANE	• G	.3	.G	• C	.1
112 TRANS-1, 2-DICaLUNGETHYLENE	•C	• 5	10.6	.0	4.8
113 Culokófókk	2.6	v. ?	7.1	3.6	4.5
114 1,2-DIChLGhulThAnd	• 11	.0	. 4	•C	.?
115 1,1,1-TRICHLOROETHANE	. 3	9. 3	95.9	10.3	28.9
116 CARBON TETRACALURIDE	•0	• <u>c</u>	•0	٥.	٥.
117 bkomudichlokombtaalie	.0	.7	•G	.0	.2
120 Thicalorosimilent	.0	20.0	164.9	8.4	50.5
171 BENAGNE 173 DIBNOWCALOKOMETAANE	3.7 .0	7.6 1.6	• 6	.0 .0	2.7
125 phonocoldonomerans	.0	.6	•ŭ •0	.0	.6
127 1.1.2.2-TETHACHLORGETATLERE	1.1	.U 45.U	239.4	20.2	77.9
128 IULUENS	1.5	66.2	25.5	76.2 15.6	25.6
125 CaLOkObbiraciib	.0	.7	. C	.0	.0
130 bluil Bengene	. 9	15.6	48.7	.0	16.3
203 Fabril	.0	10.5	18.6	.0	7.3
204 2.4-DIMEINILPHBNUL	.0	.0	9.9	.c	2.5
210 FEWT ACHLOKOPHENOL	3.8	.6	19.2	.0	5.7
301 DICULOROPENABUBO	.6	26.1	92.7	13.4	33.1
315 NAPATHALLUE	3. 8	9.9	32.9	.0	11.0
320 LIEINIL FAINALATE	11.0	7.0	5.0	3.6	0.8
333 LINNABUTYL PHTHALATE	12.6	15.8	4.4	4.2	9.3
337 BUTIL BEWAIL PATAALATE	.0	11.4	77.3	.0	22.2
336 blo(2-ETnilnEXYL)ruTuALATE	4.5	4.3	.0	.0	2.2
501 ANT LIGORY	.0	52.8	.6	.0	13.3
402 ARSENIC	21.6	•G	.G	1.9	5.9
14 CADMIUM	2.4	2. š	3.1	.0	2.1
505 CHRONIUM	151.6	135.4	72.1	65.4	106.1
506 CUPPER	62.1	46.6	50.4	96.6	ö3. ÿ
507 LEAU	15.9	210.2	135.6	35.6	99.3
508 MANGANESE	342.4	201.0	277.5	158.0	244.9
509 AEHCUNI	.4	.5	.8	.0	.4
51G NICABL	34.8	45.8	18.3	35.0	33.5
511 JOLBAIUM	5.3	4.3	.0	.0	2.4
512 51LV&R	3.7	16.0	12.4	3.3	8.8
513 InALLIUM	.0	.0 290.6	.0 364 2	.C 167 4	0.0
514 ZINC	372.0 39.7	14.0	353.2 4.5	157.4 4.0	293.3 15.8
601 TUTAL CIANIDES	35.7 24.d	5G.8	99.8	52.5	59.5
602 TUTAL ENENOLS 703 AbioCn1A	13. y	17.G	7.4	9.4	11.9
704 OIL AND GEBASE	#A*A	31.6	28.5	37.2	36.4
705 TSS	164.9	124.8	137.9	76.9	111.1
70° 133	43.0	97.2	67.9	42.5	62.6
767 <i>COD</i>	157.9	305.7	154.8	191.0	209.9
76 8 66D	47.6	154.0	100.9	68.1	92.6

^{*}Classicals in mg/L.

Table 17

Residential Per Capita Mass Discharge Rate Summary (mg/person/day)*

	Cinc	innati St. Louis		At I	lanta	Hertford							
POLLUTANT	Kirkridge	Elco	Eppinghen	Avocado	Brightwell	Northaide	Varren	Preskiin	Hillside	Tunxis	Brentwood	Average	Standard Devia- tion
110 1.1-DECHLOROETHYLES."	.0	.0	.0	.0	.0	.0	.0	.0	.0	0	.0	.0	.0
111 1.1-DECHLOROSTNAME	.0	.0	.0	.0	. 0	.0	.0	.0	.0	.0	.0	.0	.0
112 TRANS-1, 2-DICHLONGETHYLENE	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0	.0
113 CHLOHOPORN	1.5	4.0	1.2	1.3	1.4	2.4	1.5	2.4	5.4	.0	1.6	2.1	.6
114 1,2-DICHLOROSTHANS	.0	.0	.0	.0	.3	.0	.0	.0	.0	.0	.0	.0	.0
115 1,1,1-TRICHLOROSTHANS	.0	.3	1.9	.3	.5		.0	.0 .0	1.1	.0	14.5	1.3	1.7
116 CARRON TETRACHLORIDE 117 BRONDDICHLONOMETHANE	.0	.0 .2	.0	.0	.0 .1	.0	.0	.0	.0 .0	.0	.0	.0	.0 .0
117 BRONDDICHLONOMERWANE 120 TRICHLONOMERWYLEWS	.0	.0	.0	.6	.1	1.3	.0	.0	.ŏ	.0	.0	.2	.3
121 BENZENE	.0	.4	.3	.1	.7			.0	.0	.0	.0	. i	.2
123 DIBRONOCHLORONETHAVE		.1	.0	.0	.1	.0	.0	.0	.0	.0	.0	.0	.0
125 BROMOFORM	.0	.0	.0	.0	.0	.0	.c	.0	.0	.0	.0	.0	. 0
127 1.1.2.2-TETRACULOHOETHYLENG	.4	2.7	8.9	5.9	5.9	5.5	.6	2.8	1.8	.3	.4	3.3	2.7
128 TOLUENE	.5	7.0	. 9	2.1	5.2	.9	.1	1.4	.0	.0	.0	1.8	1.7
129 CHLOROSENZENE	.0	.3	.0	.0	. 1	.0	.0	.0	.0	.0	.0	.1	.1
190 ETHYL BEHERHE	. 2	1.2	.0	.0	1.5	.0	. 1	.0	.0	.0	.0	.3	• 4
203 PHENOL	1.5	36.9	. 9	6.1	2.7	.0	1.7	.0	.0	.0	.0	5.8	9.0
204 2,4-DINETHYLPHRNOL	.0	1.1	.0	. 8	.0	.0	1.2	.0	.0	.0	.0	.4	.3
210 PENTACHLOROPHENOL	.0	.0	.0	.0	.0	6.2	.0	.0	.0	.0	.0	8	1.6 2.3
301 DICHLOROGENZENES	.0	.0	.0	5.1	8.9	.0	.0	.0	.0	.0	.0	1.2	1.2
315 NAPHTHALENE	0	1.9 55.7	.0	.0 4.5	8.4	.0	.0 .0	3.8 .0	.0 .0	.0	9.0	10.6	17.2
326 DIETHYL PHINALATS 333 DI-N-BUTYL PHINALATS	16.8 18.6	20.5	5.5 3.5	9.5	1.5 3.1	.0	1.6	.0	.0	13.6	1.9	7.4	8.3
337 MITTL REVITE PHINALATE	7.5	15.3	5.5	4.7	10.8	.0	.0	.0	.0	.0	5.0	4.9	5.3
388 BIS(2-ETHYLHEXYL)PHTHALATE	7.4	21.8	1.2	٠.٥	20.2	.0		.0	.0	.0	.0	5.4	7.0
501 ANT DIONY		5.0	5.2	4.9	.0	.0	.6	.0	.0	.0	.0	1.6	1.7
502 ARSENIC	19.6	24.9	.0	.0	.0	.0	.0	.0	.0	2.3	2.0	5.8	10.9
SON CADMIUM	.0	. 0	. 6	. 4	2.7	.0	.5	.0	.0	.0	8.3	1.0	. 8
SOS CHBONIUN	8.5	36.3	4.4	3.0	4.5	15.5	2.3	61.0	15.4	.0	.0	13,6	8.6
506 COPPER	139.5	149.2	12.8	14.1	60.9	24.7	14.3	54.3	47.8	56.9	40.2	60.7	57.2
507 LEAD	.0	74.9	23.0	21.7	346.2	26.7	13.4	32.5	6.0	12.3	17.5	51.2 104.9	53.5 30.1
504 MANGANESE	84.6	205.1	82.7	87.3	46.9	115.6	79.4 .0	57. 3	231.7	59.7	72.0	.2	.2
509 MERCURY 510 MICKEL	.0	.4 5.3	.2 4.1	. 8 3. 2	.4 4.4	.0 1.9	1.8	2.0	.0	.0	1.0	2.4	1.4
511 SELEVIUN	3.9	8.3	3.2	5.4	3.8		.0	.0	.8	.5	.0	2.7	3.0
512 SILVER	.7	.3	1.9	4,5	.3	3.5	.0	.0	.0	.0	. 0	1.1	1.0
513 THALLIUM	.0	.6		.0	.0	.0	.0	.0	.0	.0	.0	.0	. 0
514 ZINC	130.7	209.7	42.7	38.6	531.2	85.5	66.9	72.8	57.8	50,6	72.6	128.5	59.4
601 TOTAL CYANIDES	.0	ناء	. 0	.0	.9	4.8	.0	.0	.0	.0	.0	.7	1.2
602 TOTAL PHEMOLS	29.8	67.6	14.8	16.3	16.2	13.1	6.7	29.7	25.7	.0	14.9	25.5	22.4
703 AMNONIA	24.4	33. 4	5.7	8.4	7.3	5.8	2.9	3.4	2.9	7.2	7.4	11.6	11.8
704 OZL AND GREASE	53.7	107.9	17.2	21.5	214.1	10.7	11.0	24.5	17.6	18.4	10.3	49.4	38.3
705 <i>TSS</i>	151.1	516.5	42.3	53.2	64.6	49.3	95.0	32.6	21.3	35.9	114.2	127.6	137.8 46.6
706 <i>TOC</i>	90.3	172.5	37.4	49.5	48.1	37.6	24.0 58.9	33.5 123.7	34.5 130.7	57.9 206.0	38.9 179.8	62.1 208.3	140.6
707 COD	276.5	556.9	50.8 48.1	116.3 73.0	135.6 59.9	231.6 48.6	29.1	43.5	19.0	70.0	82.8	8.2	72.9
708 <i>BOD</i>	117.8	275.C	40.1	.3.0	C.1. 3	70.0	43.4	70.0	13.0				. • • •

^{*}Classicals in g/person/day

Table 18
Overall Source Average Concentrations

	μ		
Pollutant	RES	COM	IND
110 1.1-DICHLOROETHYLENE	.0	.3	11.6
111 1.1-DICHLOROETHANE	.0	.1	1.6
112 TRANS-1.2-DICHLOROFTHYLENE	.0	1.5	11.7
113 CHLOROFORM	3.0	5.7	12.0
114 1.2-DICHLOROETHANE	.1	.1	.6
115 1.1.1-TRICHLOROETHANE	2.3	2.9	85.1
116 CARBON TETRACHLORIDE	.0	.1	28.4
117 BROMODICHLOROMETHANE	.0	1.0	1.6
120 TRICHLOROETHYLENE	.4	12.8	25.4
121 BENZENE	.2	2.7	1.2
123 DIBROMOCHLOROMETHANE	.0	.7	1.2
125 BROMOFORM	.0	.0	.0
127 1,1,2,2-TETRACHLOROETHYLENE	6.3	21.4	69.9
128 TOLUENE	2.6	11.0	52.3
129 CHLOROBENZENE	.1	.0	.9
130 ETHYL BENZENE	.4	3.0	100.4
203 PHENOL	5.8	4.5	135.8
204 2,4-DIMETHYLPHENOL	.7	.0	74.0
210 PENTACHLOROPHENOL	1.2	5.8	10.1
301 DICHLOROBENZENES	2.8	7.5	376.5
315 NAPHTHALENE	2.1	2.6	50.7
326 DIETHYL PUTHALATE	9.8	5.7	.0
333 DI-N-BUTYL PHTHALATE	9.0	11.7	67.1
337 FUTYL BEWZYL PHTHALATE	6.8	10.6	168.2
338 FIS(2-ETHYLHEXYL)PHTHALATE	6.8	7.7	43.0
501 ANTIMONY	2.7	.3	1.7 3.2
502 ARSENIC	4.8	2.6	20.7
504 CADMIUM	1.8	.6	713.2
505 CHROMIUM	16.3	56.8	124.8
506 COPPER	72.1	54.5	323.7
507 LEAD	97.3	49.8	232.1
508 MANGANESE	153.0	224.8	1.9
509 MERCURY	.4	.4 12 h	108.7
510 NICKEL	4.2	12.4 3.3	.9
511 SELENIUM	3.8	2.9	150.4
512 SILVER	2.2	.1	.1
513 THALLIUM	.0 214.0	138.1	860.0
514 ZINC	1.1	.2	90.7
601 TOTAL CYANIDES	30.8	37.0	204.1
602 TOTAL PHENOLS	30.8 14.2	10.7	10.6
703 AMMONIA	77.4	109.0	106.1
704 OIL AND GREASE	156.8	122.4	215.8
705 <i>TSS</i>	81.5	106.2	130.1
705 <i>TOC</i>	263.8	346.0	540.3
707 COD	113.9	160.0	216.1
708 BOD	7-44 4		

^{*}Classicals in mg/L.

2. Frequency/Concentration Relationships

For each pollutant, the frequency of detection data (in percent) have been paired with the average category concentration data by source category and is summarized in Table 19. For simplification in attempting to interpret this data, they have been grouped into general categories; pollutants detected greater than or less than 50 percent of the time, in concentration level groups of <10 $\mu g/L$, 10-100 $\mu g/L$ and >100 $\mu g/L$. These results were shown in Figure 1-5 of the Summary.

These displays clearly show the low levels of pollutants associated with tap water and the increase in contribution from residential to commercial to industrial sources. The POTW influent data do reflect the integration of these results as indicated by presence of most of the detected pollutants, but at lower levels than the industrial sources and with greater overall frequency.

Table 19

Detection Frequency/Concentration* Summary

			TAP	AP RES			OMI MO			INF	
		% Occur- rence	Average Conc.	% Occur- rence	Average Conc.	% Occur- rence	Average Conc.	% Occur-	Average Conc.	% Occur- rence	Average Conc.
110.	1,1-Dichloroethylene	0	0	0	0	5	0.3	38	11.6	17	2.4
111.	1,1-Dichloroethane	0	0	0	0	2	0.1	33	1.6	6	0.1
112.	Trans-1,2-dichloro- ethylene	0	0	0	0	43	1.5	38	11.7	28	4.8
113.	Chloroform	100	27.1	91	3.0	100	6.7	100	12.0	100	4.9
114.	1,2-Dichloroethane	0	0	2	0.1	5	0.1	14	0.6	11	0.2
115.	1,1,1-Trichloroethane	0	0	30	2.3	52	2.9	71	85.1	78	28.9
116.	Carbon tetrachloride	0	0	0	0	5	0.1	33	28.4	0	0
117.	Bromodichloromethane	100	8.8	4	đ	50	1.0	57	1.6	11	0.2
120.	Trichloroethylene	0	0	11	0.4	14	12.8	100	25.4	67	50.5
121.	Benzene	0	0	22	0.2	50	2.7	76	1.2	67	2.7
123.	Dibromochloromethane	58	5.9	4	0	43	0.7	57	1.2	22	0.2
125.	Bromoform	33	0.8	0	0	0	0	0	0	0	0
127.	1,1,2,2-Tetrachloro- ethylene	25	0.8	78	6.3	98	21.4	100	69.9	83	77.9
128.	Toluene	8	0.3	63	2.6	90	11.0	100	52.3	78	25.8
129.	Chlorobenzene	0	0	7	0.1	5	0	14	0.9	6	0
130.	Ethylbenzene	8	0.1	17	0.4	50	3.0	76	100.4	67	16.3
203.	Phenol	0	0	38	5.8	40	4.5	52	135.8	33	7.3
204.	2,4-Dimethylphenol	0	0	6	0.7	0	0	38	74.0	11	2.5
210.	Pentachlorophenol	0	0	4	1.2	14	5.8	19	10.1	22	5.7
301.	Dichlorobenzenes	0	0	13	2.8	31	7.5	57	376.5	56	33.1
315.	Naphthalene	0	0	9	2.1	17	2.6	62	50.7	44	11.6
326.	Diethylphthalate	8	0.8	49	9.8	36	5.7	0	0	50	6.8
333.	Di-n-butylphthalate	25	4.5	34	9.0	43	11.7	57	67.1	67	9.3
337.	Butylbenzylphthalate	0	0	47	6.8	55	10.6	52	168.2	44	22.2
338.	Bis(2-ethylhexyl)/di-n octyl phthalate	17	4.1	23	6.8	38	7.7	24	43.0	22	2.2
501.	Antimony	17	3.0	35	2.7	7	0.3	50	1.7	39	13.3
502.	Arsenic	25	1.6	35	4.8	38	2.6	43	3.2	44	5.9
504.	Cadmium	8	0.5	15	1.8	12	0.6	38	20.7	56	2.1
505.	Chromium	8	2.5	63	16.3	59	56.8	100	713.2	89	106.1
506.	Copper	92	28.4	100	72.1	100	54.5	100	124.8	100	63.9
507.	Lead	33	10.4	83	97.3	83	49.8	100		89	99.3
508.	Manganese	50	5.1	100	153.0	100	224.8	100	232.1	100	244.9
509.	Mercury	0	0	17	0.4	10	0.4	33	1.9	17	0.4
510.	Nickel	25	4.3	61	4.2	79	12.4	100	108.7	83	33.5
511.	Selenium	33	2.9	57	3.8	38	3.3	14	0.9	28	2.4
512.	Silver	8	0.3	22	2.2	51	2.9	86		94	8.8
513.	Thallium	0	0	0	0	10	0.1	5		100	0
514.	Zinc	58	66.9	100	214.0	100	138.1	100		100	293.3
601.	Total Cyanides	1 0	0	4	1.1	2	0.2 37.0	76		100	<u>- 15.8</u> 59.5
602.	Total Phenols	17	2.0	93	30.8	95	3/.0	100	204.1	100	

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C. Mass Flow Analysis

One of the objectives of this study is to be able to predict the relative mass contribution of residential and commercial sources, in particular, to POTW influents. One reason for doing this is to estimate the industrial contribution at any given POTW by measurement of its influent. The total mass flow to the POTW for any pollutant may be expressed as:

representing the total mass flow (e.g., in Kg/day) to the POTW from each of the three major source categories and inflow/infiltration, runoff, etc. Because it was not possible to measure the "other" values during this study, it has deliberately not been included. The impact of the "other" sources is, of course, implicitly included in unknown proportions in the data from each of the categories. Therefore, for the purposes of this study the total POTW mass flow has been represented by:

$$POTW = RES + COM + IND$$

For any new city, Q, if the total contribution from the residential and commercial sources can be estimated, then the industrial contribution can be calculated after measuring the POTW as follows:

$$IND_Q = POTW_Q - (RES_Q + COM_Q).$$

One means of checking the validity of the data, as it is being developed, is to carry out a mass balance calculation for the city (z) being studied by adding the relative contributions from each source type for comparison with the POTW:

$$POTW_z = RES_z + COM_z + IND_z$$
.

These goals could be attained if it were possible to determine an average index value (V) for each source category which could be scaled up for each POTW basin according to the relative amount of each type of source activity in the basin (A). In the general case, the equation would take the form

$$POTW = V_R A_R + V_C A_C + V_T A_T$$

indicating the quantities of each source type (R = RES, C = COM, I = IND).

The basic data available from each sampling site to use in developing this approach is concentration, flow, and population. For the POTW service area as a whole, it is usually possible to obtain reliable estimates of total population (from the land planning agency) and total commercial and industrial flow (from the water use records).

For the residential sites, it is reasonable to use the population as an index basis. Thus, for the residential sites, a per capita discharge rate can be calculated as follows:

$$mass/person/day = \frac{concentration \times flow}{population}$$

For reporting convenience, the residential values have been developed in units of mg/person/day. The total basin residential contribution may thus be estimated as:

RES(Kg/day) = Res. Ave. (mg/person/day) x Basin Population x
$$10^{-6}$$
 (10⁶ is the µg to Kg conversion factor)

For the commercial and industrial sites, the only index reliably available for all of the sites studied (and the basin) is the total flow. Thus, for these source types, an average concentration value has been calculated so that, when the average value is multiplied by the total basin source type flow, the total source contribution is obtained:

COM (Kg/day) = [Avg. Com. Conc. (
$$\mu$$
g/L)] x [Com. Flow(Lps)] x 8.64 x 10⁻⁵ IND(Kg/day) = [Avg. Ind. Conc. (μ g/L)] x [Ind. Flow(Lps)] x 8.64 x 10⁻⁵ (8.64 x 10⁻⁵ is the μ g/sec to Kg/day conversion factor)

The data obtained from the commercial sites do not show a wide range in type or quantity of pollutant between sites and suggest that an average commercial concentration is a valid concept. To the contrary, the industrial site data show a wide range of both types and concentration of pollutant indicating that an average industrial concentration is not a valid concept which can be applied generally. It is useful, however, within a basin to calculate this value so that a mass balance comparison between the sources and POTW may be made. Such a comparison provides a test of how well the sites sampled represent, quantitatively and qualitatively, the total of that source type within the basin.

The average index values (Tables 14,15,17) may be used to calculate the total mass flow from each of the source types within the drainage basic according to the equation:

SUM = RES x Population + COM x $Flow_C$ + IND x $Flow_I$ where RES, COM, and IND indicate the average value either on a per capita or concentration basis; population refers to the total drainage basin population; and $Flow_C$ and $Flow_I$ are, respectively, the total commercial and industrial flows in the basin. The values thus calculated may be compared with the POTW influent.

Two sets of analyses have been carried out using this approach. The first involved using the average values to calculate the relative significance of each source type for several hypothetical cities varying in degree of industrialization and residential/commercial mix. The second involved using the four city average values to calculate a total influent flow (SUM) to compare with the actual measured value for each of the cities studied.

Residential source averaging was done assuming that sites within a city were all part of the same overall statistical population and, therefore, the average residential value for each city was calculated on a per capita weighted basis. The assumption of homogeneity within a city's population implies that each site's population is a proper measure of that site's ability to represent the residences in the basin as a whole, allowing population weighted averages within a city.

For commercial and industrial sources, average concentration values were calculated as straight averages of sites to form city averages and as straight averages of city means to form an overall average. The assumption is that for commercial and industrial sources, different sites represent different aspects of the source type and, therefore, different sites are not from a single statistical population. Average residential values across cities were straight averaged on the assumption that the individual cities represent samples from different statistical populations.

1. Hypothetical Cities

One means of obtaining a perspective on the significance of relative source type contributions to POTW influents is to use the average data obtained in this study to calculate mass flows for several hypothetical cities. Such an analysis has been carried out for five cases (A, B, C, D, E) using the distribution of flow and population described in Table 20. These cases represent a range of industrialization of 0-50% (in terms of flow) and a residential flow of 90-30% (200,000-68,000 population) with varying levels of commercial activity (10-20%). The Case B distribution is about the same as was observed in St. Louis.

A review of influent flow for 324 POTWs having secondary treatment showed an overall average flow of about 1,000 Lps and this value was used for these Case calculations. Those same plants also had an average industrial flow of about 20%. The Case E example was calculated to represent this "typical" basin.

The calculations for these prototype cities are given in Tables 21-25. The relative impact of the sources changes between these Cases is perhaps seen most easily by comparing the ratios of the source categories to that with the highest value. Those ratios are given in Tables 26-30 for Cases A, B, C, D, and E, respectively.

The dominance of the residential category in Case A is obvious, with commercial sources dominant for only eight toxic pollutants. Commercial sources still play a small role in Case B, but the impact of even a small degree of industrialization is clear. This trend continues for Cases C, D, and E where the industrial category is dominant for most pollutants.

The overall impact of these source distribution changes can be seen further in Table 31, where the total mass flow (SUM) for each case is compared. Much higher mass flows are observed for most pollutants in Case D than in Case A. However, some pollutants, whose concentrations are not strongly source dependent, such as manganese, do not change much across the cases. Once again, the trends in these data for each pollutant can be seen in the ratios given in Table 32.

Some pollutants still have the highest mass flows in the most residential cases—diethyl phthalate, copper, and manganese, for instance. The reader is reminded to use these analyses with caution because they are limited in terms of estimating the industrial impact, based on the data from the industrial sources sampled during the study.

For each of these hypothetical cases with an influent flow of 1,000 Lps, a 1 $\mu g/L$ influent concentration would correspond to a mass flow of 0.08 Kg/day.

Table 20
Description of Hypothetical City Source Contribution

CITY	TOTAL FLOW (Lps)		RES	COM	IND
Case A	1,000	% flow pop.	90 900 200,000	10 100 -	0 0 -
Case B	1,000	% flow pop.	80 800 182,000	10 100 -	10 100 -
Case C	1,000	% flow pop.	50 500 114,000	20 200 -	30 300 -
Case D	1,000	% flow pop.	30 300 68,000	20 200 -	50 500 -
Case E	1,000	% flow pop.	60 600 136,500	20 200 -	20 200 -

Table 21

Hypothetical Ciry - Case A - Mass Flow

RES = 200,000 people, COM = 100 Lps, IND = 0 Lps

Kg/day SUM IND COM RES Pollutant .00 .00 .00 .00 110 1.1-DICHLOROETHYLENE .00 .00 .00 .00 111 1,1-DICHLOROETHAUE .00 .01 .01 .00 112 TRANS-1.2-DICHLOROETHYLENE .45 .00 .06 .39 113 CHLOROFORM .00 .00 .00 .00 114 1,2-DICHLOROETHANE .03 .13 .00 .10 115 1.1.1-TRICHLOROETHANE .00 .00 .00 .00 116 CAREON TETRACHLORIDE .01 .00 .01 .01 117 BROMODICHLOROMETHANE .21 .00 .10 .11 120 TRICHLOROETHYLENE .00 .05 .02 .02 121 BENZENE .01 .00 .01 .00 123 DIBROMOCHLOROMETHANE .00 .00 .00 .00 125 BROMOFORM .18 .00 1.03 .85 127 1,1,2,2-TETRACHLOROETHYLENE .48 .00 .09 .39 128 TOLUENE .00 .01 .00 .01 129 CHLOROBUNZEUE .08 .00 .03 .05 130 ETHYL BENZENE .00 .04 1.30 1.26 203 PHENOL .08 .00 .00 .08 204 2,4-DIMETHYLPHENOL .38 .00 .33 .05 210 PENTACHLOROPHENOL .00 .34 . 28 .07 301 DICHLOROBENZENES .00 .14 .02 .12 315 NAPHTHALENE .00 2.40 .05 2.35 326 DIETHYL PHTHALATE 1.87 .00 .10 1.77 333 DI-N-BUTYL PHTHALATE .00 1,16 .09 1.07 337 FUTYL BENZYL PHTHALATE .00 1.11 1.04 .07 338 BIS (2-ETHYLHEXYL)PHTHALATE .00 .42 .00 .42 501 ANTIMONY 1.45 .00 1.43 .02 502 ARSENIC .08 .08 .00 .00 504 CADMIUM .00 2.83 .49 2.34 505 CHROMIUM .00 12.85 .47 12.38 506 COPPER .00 8.16 .43 7.73 507 *LEAD* .00 23.12 1.94 21.17 **508 MANGANESE** .00 .05 .00 .05 509 MERCURY .62 .00 .11 .52 510 NICKEL .70 .03 .00 .68 511 SELEUIUM .47 .03 .00 .45 512 SILVER .00 .00 .00 .00 513 THALLIUM 24.24 .00 23.05 1.19 514 ZINC .00 .27 .00 .27 601 TOTAL CYANIDES 5.55 .00 5.23 .32 602 TOTAL PHENOLS .09 .00 2.82 2.73 703 AMMONIA 10.02 .00 .94 9.08 704 OIL AND CREASE 27.27 .00 1.06 26.21 705 TSS 14.33 .92 .00 13.41 706 TOC .00 48.72 45.73 2.99 707 COD 20.40 .00 1.38 19.G2 708 ROD

^{*}Classicals in 103 kg/day.

Table 22

Hypothetical City - Case B - Mass Flow RES = 182,000 people, COM = 100 Lps, IND = 100 Lps

		Kg/d	*	
Pollutant	RES	COM	IND	SUM
		20	4.0	4.0
110 1,1-DICHLOROETHYLEHE	.00	.00	.10	.10
111 1,1-DICHLOROETHANE	.00	.00	.01	.01
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.01	.10	.11
113 CHLOROFORM	. 36	.06	.10	.52
114 1,2-DIC!!LOROET!!ANE	.00	.00	.01	.01
115 1, 1, 1-TRICH LOROETHAME	.09	.03	.74	.85
116 CARBON TETRACHLORIDE	.00	.00	.25	.25
117 BROMODICHLOROMETHANE	.01	.01	.01	.03
120 TRICHLOROETHYLENE	•09	.11	.22	.42
121 BENZENE	•02	.02	.01	.06
123 DIBROMOCHLOROMETHANE	•00	.01	.01	.02
125 BRONOFORM	.00	.00	.00	.00
127 1,1,2,2-TETRACHLOROETHYLENE		.18	.60	1.56
128 TOLUENE	.35	.09	. 45	•90
129 CHLOROBENZENE	.01	•00	.01	.02
130 ETHYL BENZENE	.05	.03	.87	. 94
203 PHENOL	1.14	.04	1.17	2.36
204 2,4-DIMETHYLPHENOL	.07	.00	.64	.71
210 PENTACHLOROPHENOL	.30	.05	.09	.44
301 DICHLOROBENZENES	. 25	.07	3.25	3.57
315 RAPHTHALENE	.11	.02	.44	.57
326 DIETHYL PHTHALATE	2.14	.05	.00	2.19
333 DI-N-BUTYL PHTHALATE	1.61	.10	.58	2.29
337 BUTYL BENZYL PHTHALATE	.97	.09	1.45	2.52
338 BIS(2-ETHYLHEXYL)PHTHALATE	•95	.07	.37	1.38
501 ANTIMONY	.38	.00	.01	.40
502 ARSENIC	1.30	.02	.03	1.35
504 CADMIUM	.07	.00	.18	.25
505 CHROMIUM	2,13	.49	6 .16	8.79
506 COPPER	11.26	.47	1.08	12.81
507 LEAD	7.03	.43	2.80	10.26
508 MANGANESE	19.27	1.94	2.01	23.22
509 MERCURY	.05	.00	.02	.07
510 hickel	.47	.11	.94	1.52
511 SELENIUM	.61	.03	.01	.65
512 SILVER	.41	.03	1.30	1.73
513 THALLIUM	.00	.00	.00	.00
514 ZINC	20.97	1.19	7.43	29.60
601 TOTAL CYANIDES	. 24	.00	.78	1.03
602 TOTAL PHENOLS	4.76	.32	1.76	6.84
703 AMMONIA	2.48	.09	.09	2.67
704 OIL AND GREASE	8.26	.94	.92	10.12
705 <i>TSS</i>	23.85	1.06	1.86	26.77
705 <i>135</i> 706 <i>TOC</i>	12.21	.92	1.12	14.25
707 COD	41.62	2.99	4.67	49, 27
	17.31	1.38	1.87	20.56
708 <i>BOD</i>			- 	= : 3 = -

^{*}Classicals in 103 kg/day.

Table 23

Hypothetical City - Case C - Mass Flow

RES = 114,000 people, COM = 200 Lps, IND = 300 Lps

Kg/day* RES COM IND SUM Pollutant 110 1.1-DICHLOROETHYLENE .00 .00 .30 .30 111 1,1-DICHLOROETHANE .00 .00 .04 .04 .30 .33 112 TRANS-1, 2-DICHLOROETHYLEHE .00 .03 .12 .31 .22 .65 113 CHLOROFORM 114 1.2-DICHLOROETHANE .00 .00 .02 .02 .06 115 1,1,1-TRICHLOROETHANE .05 2.21 2.32 .00 .00 .74 116 CARBON TETRACHLORIDE .74 .04 .00 .02 117 EROKODICELOROMETHANE .06 .66 . 94 120 TRICHLOROETHYLENE .06 .22 .05 J0 .48 .22 .01 .13 .09 121 BUNZEKE .01 .03 .03 . 04 123 DIEROFOCHLOROMETHANE .01 .03 .00 .00 .37 1.81 .19 1.36 .01 125 BROMOFORM .00 127 1,1,2,2-TETRACHLOROETHYLENE .37 2.66 128 TOLUENE 1.76 .00 130 ETHYL BENZENE .02 .03 .05 2.60 2.69 203 PHENOL .08 3.52 4.31 .00 1.92 .10 .26 .13 9.76 .05 1.32 .10 .00 .20 1.74 204 2.4-DIMETHYLPHEROL .04 1.96 .19 .55 210 PENTACHLOROPHENOL .16 301 DICHLOROBENZENES 10,05 .07 315 MAPHTHALENE .07 1.34 1.43 326 PIETHYL PHTHALATE 1.44 333 DI-N-BUTYL PHTHALATE 1.01 2.95 337 BUTYL BENZYL PHTHALATE .61 .18 4.36 5.15 .13 1.11 .01 .04 .04 .08 .01 .54 .98 18.49 338 BIS(2-ETHYLHEXYL)PHTHALATE .59 1.84 . 24 501 ANTIMONY . 29 502 ARSENIC .82 . 94 504 CADMIUN . 04 .59 506 COPPER 507 LEAD 1.34 20.80 . 94 7.05 3.24 11.23 4.41 .86 8.39 13.66 **508 MANGANESE** 6.02 21.97 12.07 3.88 .01 .05 509 MERCURY .03 .08 .21 510 HICKEL .29 2.82 3.33 .38 . .06 511 SELENIUM .02 . 46 .05 .25 .00 512 SILVER 3.90 4.20 .00 .00 513 THALLIUM .00 13.14 22.29 37.81 514 ZINC 2.39 601 TOTAL CYANIDES .15 .00 2.35 2,51 .64 2.98 5.29 8.91 602 TOTAL PHENOLS .18 703 AMMONIA 1.55 .28 2.02 2.75 5.17 1.88 704 OIL AND GREASE 9.81 705 TSS 2.12 14.94 5.59 22.65 706 TOC 7.65 1.83 3.37 12.85 26.07 10.84 5.98 14.00 2.76 5.60 707 COD 46.05 708 BOD 19.21

^{*}Classicals in 103 kg/day.

Table 24

Hypothetical City - Case D - Mass Flow
RES = 68,000 people, COM = 200 Lps, IND = 500 Lps

	Kg/day*				
	RES	COM	IND	SUM	
110 1,1-DICHLOROETHYLENE	.00	.00	.50	. 50	
111 1.1-PICHLOROETHANE	.00	.00	.07	.07	
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.03	.51	.53	
113 CHLOROFORM	.13	.12	.52	.7 7	
114 1,2-DICHLOROETHANE	.00	•00	.03	.03	
115 1,1,1-TRICHLOROETHANE	.03	.05	3.68	3.76	
116 CARBON TETRACHLORIDE	.00	.00	1.23	1.23	
117 EROMODICHLOROMETHANE	•00	.02	.07	.09	
120 TRICHLOROETHYLENE	.03	.22	1.10	1.35	
121 BENZEHE	.01	.05	.05	.11	
123 DIBROMOCHLOROMETHANE	.00	.01	.05	.06	
125 BROMOFORM	.00	•00	.00	.00	
127 1,1,2,2-TETRACHLOROETHYLENE	. 29	.37	3.02	3.68	
128 TOLUENE	.13	.19	2.26	2.58	
129 CHLOKOBENZENE	.00	.00	.04	• 04	
130 ETHYL FENZEUE	.02	.05	4.34	4.41	
203 PHENOL	.43	.08	5.87	6.37	
204 2.4-DIMETHYLPHENOL	.03	.00	3.19	3.22	
210 PENTACHLOROPHENOL	.11	.10	. 44	.65	
301 DICHLOROBENZENES	•09	.13			
315 NAPHTHALENE	. 04	.05	2.19	2.28	
326 DIETHYL PHTHALATE	.80	.10	.00	.90	
333 DI-N-BUTYL PHTHALATE	.60	.20	2.90	3.70	
337 BUTYL BENZYL PHTHALATE	.36	.18	7.27	7.81	
338 BIS(2-ETHYLHEXYL)PHTHALATE	.35	.13	1.86	2.34	
501 ANTIMONY	.14	.01	.07	.22	
502 ARSENIC	.49	.04	.14	.67	
504 CADMIUM	.03	.01	.90	.93	
505 CHROMIUM	.80	•	30.81	32.59	
506 COPPER	4.21	.94		10.54	
507 <i>LEAD</i>	2.63	.86		17.47 21.11	
508 MANGANESE	7.20		10.03	.10	
509 MERCURY	.12	.01	.08 4.69	5.08	
510 NICKEL	.18	.21	.04	.32	
511 SELENIUM	.23	•06	6.50	6.70	
512 SILVER	.15	.05	.00	.01	
513 THALLIUM	.00	.00	37.15	47.37	
514 ZINC	7.84	2.39	3.92	4.01	
601 TOTAL CYANIDES	.09	.00 .64	8.82	11.23	
602 TOTAL PHENOLS	1.78		.46	1.57	
703 AMMONIA	.93	.18 1.88	4.59	9.56	
704 OIL AND GREASE	3.09	2.12	9.32	20.35	
705 <i>TSS</i>	8.91	1.83	5.62	12.02	
706 TOC	4.56	5.98	23.34	44.87	
707 COD	15.55 6.47	2.76	9.34	18.57	
708 EOD	0.47	2.70	3101	-2,000	

^{*}Classicals in 103 kg/day.

Table 25

Hypothetical City - Case E - Mass Flow
RES = 136,500 people, COM = 200 Lps, IND - 200 Lps

Kg/day* Pollutant RES COM IND SUM .00 .00 110 1.1-DICHLOROETHYLENE .20 . 20 .00 .00 .03 111 1, 1-DICHLOROETHANE .03 112 TRANS-1, 2-DICHLOROETHYLEHE .00 .03 .20 .23 114 1, 2-DICHLOROETHAHE
115 1. 1. 1-mp. 70... .12 . 27 . 21 . 59 .00 .00 .01 .01 .07 .05 1.47 1.59 .49 116 CARBON TETRACHLORIDE .00 .49 .02 .03 117 BRONODICHLOROMETHANE .05 . 22 .07 .44 120 TRICHLOROETHYLENE .73 .05 121 BENZENE .02 .02 .08 .02 .00 .01 123 DIBRONOCHLOROMETHANE .04 .00 .00 125 BRONOFORM .00 .58 .37 127 1, 1, 2, 2-TETRACHLOROETHYLENE 1.21 2.16 .19 128 TOLUENE . 26 .90 1.36 .01 .00 .02 129 CHLOROBENZENE . 02 130 ETHYL BENZENE .04 .05 1.74 1.82 . 86 .08 203 PHENOL 2.35 3.28 204 2.4-DIMETHYLPHENOL .05 .00 1.28 1.33 .23
.19
.08
.33 DI-N-BUTYL PHTHALATE
.337 BUTYL BENZYL PHTHALATE
.73
.338 BIS(2-ETHYLHEXYL)PHTHALATE
.501 ANTIHONY
.502 ARSENIC
.504 CADMIUM 210 PENTACHLOROPHENOL . 23 .10 . 50 .17 6.51 . 13 6.83 . 05 .88 1.00 .00 1.16 .10 1.70 . 20 2.57 . 18 3.82 .13 .74 1.59 .03 .01 .32 . 04 .06 1.08 .36 .01 .42 12.32 14.91 505 CHRONIUM 1.60 . 98 . 94 506 COPPER 8.45 2.16 5.59 11.55 507 *LEAD* . 86 5.28 11.73 508 MANGAHESE 14.45 22.35 3.88 4.01 .03 .03 509 MERCURY .01 . 07 . 35 510 NICKEL . 21 1.88 2.44 .06 . 46 .01 511 SELENIUM . 53 .05 .30 512 SILVER 2.60 2.95 513 THALLIUM .00 .00 .00 514 ZINC 15.73 2.39 14.86 32.98 601 TOTAL CYANIDES .18 .00 1.57 1.75 602 TOTAL PHENOLS 7.73 3,57 . 64 3.53 703 AHMONIA 1.86 . 18 2.23 .18 704 OIL AND-GREASE 1.83 6.20 1.88 9.91 705 TSS 17.89 3.73 2.12 23.73 706 TOC 2.25 9.16 1.83 13.24 9.34 707 COD 31.21 5.98 46.53 708 BOD 12.98 2.76 3.73 19.48

^{*} Classicals in 10³ kg/day.

Table 26

Relative Source Strength Comparison - Case A

			SUM .		
		RES	COM	IND	Kg/day
	1,1-DICHLOROETHYLENE	.00	1.00	.00	• 00
	1,1-DICHLOROETHANE	.00	1.00	.00	.00
112	TRANS-1, 2-DICHLOROETHYLENL	.00	1.00	.00	.01
113	CHLOROFORM	.87	.13	.00	. 45
114	1,2-DICHLOROETHANE	.69	.31	.00	.00
115	1,1,1-TRICHLOROETHAHE	.80	.20	.00	.13
116	CARBON TETRACHLORIPE	.00	1.00	.00	.00
117	BROMODICH LOROWETH ANE	.42	.58	.00	.01
120	TRICHLOROETHYLENE	.47	.53	.00	.21
121	RENZENE	.51	.49	.00	.05
123	DI BROMOCHLOROMETHANE	.37	.63	.00	.01
	BROMOFORM	.00	.00	•00	.00
127	1,1,2,2-TETRACHLOROETHYLENE	.82	.18	.00	1.03
128	TOLUENE	.80	.20	.00	.48
	CHLOROBENZENE	•96	.04	.00	.01
130	FTHYL BENZENE	.67	.33	.00	.08
203	PHENOL	.97	.03	.00	1.30
204	2,4-DIMETHYLPHENOL	1.00	.00	.00	.08
210	PENT ACHLOROPHENOL	.87	.13	.00	.38
301	DICHLOROBENZENES	.81	.19	.00	.34
315	NAPHTHALCHE	.84	.16	.00	. 14
326	DIETHYL PHTHALATE	.98	.02	.00	2.40
333	DI-N-BUTYL PHTHALATE	• 95	.05	.00	1.87
337	BUTYL BENZYL PHTHALATE	.92	.08	.00	1.16
338	BIS(2-ETHYLHEXYL)PHTHALATE	•94	•06	.00	1.11
501	ANTIMOUY	•99	.01	.00	.42
502	ARSENIC	•98	.02	.00	1.45
504	CADMIUM	. 94	.06	.00	.08
505	CHROMIUM	. 83	.17	.00	2.83
506	COPPER	•96	.04	.00	12.85
507	LEAD	• 95	.05	.00	8.16
508	MANGANESE	•92	.08	.00	23.12
509	MERCURY	. 94	.06	.00	.05
510	WICKEL	.83	.17	.00	.62
511	SELENIUM	• 96	.04	.00	.70
512	SILVER	• 95	.05	.00	.47
513	THALLIUM	• 00	1.00	.00	.00
514	ZINC	• 95	.05	.00	24.24
601	TOTAL CYANIDES	.99	.01	.00	.27
602	TOTAL PHENOLS	. 94	.06	.00	
703	AMMONIA	•97	.03	.00	2.82
704	OIL AND GREASE	.91	.09	•00	10.02
705	TSS	•96	.04	.00	27.27
706	TOC	. 94	.06	•00	14.33
707	COD	. 94	•06	.00	48.72
708	BOD	. 93	.07	•00	20.40

^{*}Classicals in 10³ kg/day.

Table 27

Relative Source Strength Comparison - Case B

,		Fraction		SUM
	RES	COM	IND	Kg/day*
110 1,1-DICHLOROETHYLEHE	.00	.02	.98	.10
111 1,1-DICHLOROETHANE	.00	.04	•96	.01
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.12	.88	.11
113 CHLOROFORM	.69	.11	.20	.52
114 1,2-DICHLOROETHANE	.27	.14	• 59	.01
115 1,1,1-TRICHLOROETHANE	.11	.03	.86	.85
116 CARBON TETRACHLORIDE	.00	.00	1.00	.25
117 BROMODICHLOROMETHANE	. 20	.31	.49	.03
120 TRICHLOROETHYLENE	.21	.27	•52	.42
121 BENZENE	.39	.42	.19	• 06
123 DIBROMOCHLOROMETHAILE	.17	.32	.51	.02
125 BROMUFORM	.00	.00	.00	.00
127 1,1,2,2-TETRACHLOROFTHYLENE	.49	.12	.39	1.56
128 TOLUENE	.39	.11	• 50	.90
129 CHLOROBENZENE	.49	.02	.49	.02
130 ETHYL BENZENE	.05	.03	.92	.94
203 PHENOL	.49	.02	.50	2.36
204 2,4-DIMETHYLPHENOL	.10	.00	.90	.71
210 PENTACHLOROPHENOL	.69	.11	.20	.44
301 DICHLOROBENZENES	.07	.02	.91	3,57
315 NAPHTHALENE	.19	. 04	.77	.57
326 DIETHYL PHTHALATE	.98	.02	.00	2.19
333 DI-I:-BUTYL PHTHALATE	.70	.04	.25	2.29
337 BUTYL BENZYL PHTHALATE	.39	.04	. 58	2.52
338 BIS(2-ETHYLHEXYL)PHTHALATE	.68	.05	.27	1.38
501 ANTIGORY	.96	.01	. 04	.40
502 ARSENIC	.96	.02	.02	1.35
504 CADMIUM	.27	.02	.71	. 25
505 CHROMIUM	.24	.06	.70	8.79
506 COPPER	.88	.04	.08	12.81
507 LEAD	.69	.04	.27	10.26
508 MANGANESE	.83	.08	.09	23.22
509 MERCURY	.70	.05	.25	.07
510 NICKEL	.31	.07	.62	1.52
511 SELENIUM	.94	.04	.01	•65
512 SILVER	.23	.01	.75	1.73
513 THALLIUM	.00	•59	.41	.00
514 ZINC	.71	.04	.25	29.60
601 TOTAL CYANIDES	. 24	.00	.76	1.03
602 TOTAL PHENOLS	.70	.05	.26	6.84
703 AMMONIA	.93	.03	.03	2.67
704 OIL AID GREASE	.82	.09	.09	10.12
705 <i>TSS</i>	.89	. 04	.07	26.77
706 TOC	.86	.06	.08	14.25
707 COD	. 84	.06	.09	49.27
708 BOD	.84	.07	.09	20.56
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^{*}Classicals in 10³ kg/day.

Table 28

Relative Source Strength Comparison - Case C

	F	SUM .		
•	RES	COM	IND	Kg/day*
110 1,1-DICH LOROETHYLENE	.00	.02	.98	.30
111 1.1-DICHLOROETHANE	•00	.03	.97	.04
112 TRAUS-1, 2-DICHLOROETHYLENE	.00	.08	.92	.33
113 CHLOROFORM	.34	.18	.48	.65
114 1.2-DICHLOROETHANE	.08	.12	.80	.02
115 1.1.1-TRICHLOROETHANE	.03	.02	•95	2.32
116 CARBON TETRACHLORIDE	.00	.00	1.00	.74
117 BROMODICHLOROMETHAME	.06	.28	.67	.06
120 TRICHLOROETHYLENE	.06	.24	.70	.94
121 BENZENE	.15	•50	.35	.09
123 DIBROMOCHLOROMETHANE	.05	.28	.67	.04
125 BROWOFORM	.00	.00	.00	• 00
127 1,1,2,2-TETRACHLOROETHYLENE	.18	.14	.68	2.66
128 TOLUENE	.12	.11	.77	1.76
129 CHLOROBENZEUE	.17	.02	.81	.03
130 ETHYL BEGZENE	.01	.02	.97	2.69
203 PHENOL	.17	.02	.82	4.31
204 2.4-DIMETHYLPHENOL	.02	.00	.98	1.96
210 PENTACHLOROPHENOL	.35	.18	.47	. 55
301 DICHLOROBENZENES	.02	.01	.97	10.05
315 NAPHTHALENE	.05	.03	.92	1.43
326 PIETHYL PHTHALATE	.93	.07	.00	1.44
333 DI-N-BUTYL PHTHALATE	. 34	.07	.59	2.95
337 BUTYL BENZYL PHTHALATE	.12	.04	.85	5.15
338 EIS(2-ETHYLHEXYL)PHTHALATE	.32	.07	.61	1.84
501 ANTIMONY	.83	.02	.15	.29
502 AKSENIC	.86	.05	•09	. 94
504 CADVIUM	.07	.02	.91	.59
505 CHROMIUM	.06	.05	.89	20.80
506 COPPER	.63	.08	.29	11.23
507 LEAD	.32	.06	.61	13.66
507 LEAD 508 MANGANESE	.55	.18	.27	21.97
509 MERCURY	. 34	.08	.58	.08
510 WICKEL	.09	.06	.85	3.33
511 SELENIUM	.83	.12	.05	.46
512 SILVER	.06	.01	.93	4.20
513 THALLIUM	.00	.49	.51	.00
514 ZINC	. 35	.06	. 59	37.81
601 TOTAL CYANIDES	.06	.00	. 94	2.51
602 TOTAL PHENOLS	.33	.07	.59	8.91
703 AMMONIA	.77	.09	.14	2.02
704 OIL AND CREASE	.53	.19	.28	9.81
705 TSS	.66	.09	. 25	22.65
	.59	. 14	.26	12.85
706 <i>TOC</i>	.57	.13	.30	46.05
707 <i>COD</i>	.56	. 14	.29	19.21
708 <i>EOD</i>				

^{*}Classicals in 103 kg/day.

Table 29

Relative Source Strength Comparison - Case D

		SUM .		
	RES	COM	IND	Kg/day*
110 1.1-DICHLOROETHYLENE	.00	.01	.99	.50
111 1,1-DICHLOROETHANE	• 00	.02	. 98	.07
112 TRANS-1, 2-DICHLOROETEYLENE	.00	.05	.95	.53
113 CHLOROFORM	.17	.15	.67	.77
114 1,2-DICHLOROETHANE	.03	.08	.89	.03
115 1,1,1-TRICHLOROETHANE	.01	.01	.98	3.76
116 CARBON TETRACHLORIDE	.00	.00	1.00	1.23
117 BROMODICHLOROMETHANF	.02	.20	.78	.09
120 TRICHLOROETHYLENE	.02	.16	.81	1.35
121 BENZENE	.08	.43	•50	.11
123 DIBROMOCHLOROMETHANE	.02	.20	.78	.06
125 BRONOFORM	.00	.00	.00	.00
127 1,1,2,2-TETRACHLOROETHYLENE	.08	.10	.82	3.68
128 TOLUENE	.05	.07	.88	2.58
129 CHLOROBENZENE	.07	.02	.92	.04
130 ETHYL BENZEHE	.00	.01	•98	4.41
203 PHENOL	.07	.01	.92	6.37
204 2.4-DIMETHYLPHENOL	.01	.00	.99	3.22
210 PENTACHLOROPHEGGL	.18	.15	.67	.65
301 DICHLOROBENZENES	.01	.01	.99	16.49
315 WAPHTHALENE	.02	.02	.96	2.28
326 DIETHYL PHTHALATE	.89	.11	.00	.90
333 DI-N-BUTYL PHTHALATE	.16	.05	.78	3.70
337 BUTYL BENZYL PUTHALATE	.05	.02	.93	7.81
338 BIS (2-ETHYLHEXYL) PHTHALATE	.15	.06	.79	2.34
501 ANT'IMONY	.64	.02	.33	.22
502 ARSENIC	.73	.07	.21	.67
504 CADMIUM	.03	.01	• 96	.93
505 CHROMIUM	.02	.03	•95	32.59
506 COPPER	.40	.09	.51	10.54
507 <i>LEAD</i>	.15	.05	.80	17.47
508 MANGANESE	. 34	.18	.48	21.11
509 MERCURY	.16	.06	.77	.10
510 HICKEL	.03	.04	.92	5.08
511 SELENIUM	.71	.18	.11	.32
512 SILVER	.02	.01	.97	6.70
513 THALLIUM	.00	.36	.64	.01
514 ZINC	.17	.05	.78	47.37
601 TOTAL CYANIDES	.02	.00	.98	4.01
602 TOTAL PHENOLS	.16	.06	.78	11.23
703 AMMONIA	. 59	.12	.29	1.57
704 OIL AND GREASE	.32	.20	.48	9.56
705 <i>TSS</i>	.44	.10	.46	20.35
705 <i>155</i> 706 <i>TOC</i>	.38	.15	.47	12.02
707 <i>COD</i>	.35	.13	.52	44.87
707 COD	.35	.15	.50	18.57
TOO LIUU	,	7	•	

^{*}Classicals in 10³ kg/day.

Table 30

Relative Source Strength Comparison - Case E

	<u> </u>	Fraction		
Pollutant	RES	COM	IND	SUM * Kg/day
110 1.1-DICHLOROETHYLENE	.00	.02	.98	. 20
111 1,1-DICHLOROETHANE	.00	. 04	. 96	.03
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.12	.88	. 23
113 CHLOROFORM	1. 5	.20	.35	• 59
114 1,2-DICHLOROETHANE	.12	.16	.71	.01
115 1.1.1-TRICHLOROETHANE	.04	.03	.92	1.59
116 CARBON TETRACHLORIDE	.00	.00	1.00	.49
117 BROMODICHLOROMETHANE	.09	.35	• 56	.05
120 TRICHLOROETHYLENE	.09	.31	.60	.73
121 BENZENE	.19	.55	. 26	.08
123 DIBROMOCHLOROMETHANE	.07	. 36	.57	.04
125 BROMOFORM	.00	.00	.00	.00
127 1,1,2,2-TETRACHLOROETHYLENE	.27	.17	.56	2.16
128 TOLUENE	.19	.14	.67	1.36
129 CHLOROBENZENE	.26	.03	.70	.02
130 ETHYL BENZENE	.02	.03	• 95	1.82
203 <i>PHENOL</i>	. 26	.02	.72	3.28
204 2,4-DIMETHYLPHENOL	.04	.00	• 96	1.33
210 PENTACHLOROPHENOL	.45	. 20	.35	. 50
301 DICHLOROBENZENES	.03	.02	. 95	6.83
315 NAPHTHALENE	.08	.05	.87	1.00
326 DIETHYL PHTHALATE	. 94	.06	.00	1.70
333 <i>DI-N-BUTYL PHTHALATE</i>	.47	.08	.45	2.57
337 BUTYL BENZYL PHTHALATE	.19	.05	.76	3.82
338 BIS(2-ETHYLHEXYL)PHTHALATE	.45	.08	.47	1.59 .32
501 ANTIMONY	. 89	.02	.09	1.08
502 ARSENIC	.91	.04	.05	.42
504 CADMIUM	.12	.02	.85 .83	14.91
505 CHROMIUM	.11	.07	.19	11.55
506 COPPER	.73	.08 .07	.48	11.73
507 LEAD	.45 .65	.17	.18	22.35
508 MANGANESE	.47	.09	. 44	.07
509 MERCURY	.14	.09	.77	2.44
510 WICKEL	.86	.11	.03	. 53
511 SELENIUM	.10	.02	.88	2.95
512 SILVER	.00	•59	.41	.00
513 THALLIUM	.48	.07	.45	32.98
514 ZINC	.10	.00	.89	1.75
601 TOTAL CYANIDES 602 TOTAL PHENOLS	.46	.08	.46	7.73
703 AMMONIA	.83	.08	.08	2.23
703 AMONIA 704 OIL AND GREASE	.63	.19	.19	9.91
704 OIL AND GREASE 705 TSS	.75	.09	.16	23.73
705 <i>155</i> 706 <i>TOC</i>	.69	. 14	.17	13.24
707 <i>COD</i>	.67	.13	.20	46.53
707 60D 708 BOD	.67	.14	.19	19.48
100 000				.

^{*}Classicals in 10³ kg/day.

Table 31 Total Mass Flow Comparison of Hypothetical Cities* (kg/day)**

		•				
		CASE	CASE	CASE	CASE	CASE
		A	В	С	D	E
110	1,1-DICHLOROETHYLENE	.00	.10	.30	. 50	. 20
		.00		.04		.03
	TRANS-1, 2-DICHLOROETHY LENE		.11	.33	.53	.23
	CHLOROFORM	.45		.65		
	1,2-DICHLOKOETHANE	.00		.02	.03	
	1,1,1-TRICHLOROETHANE	.13		2.32		
	CARBON TETRACHLORIDE	.00	.25	.74		
	BROMODICHLORONETHANE	.01	.03	.06	.09	.05
	TRICHLOROETHYLENE	.21	.42	.94		.73
	BENZENE	.05	.06	.09	.11	.08
	DIBROMOCH LOROMETH ANE	.01	.02	.04	.06	.01.
	BROM OF ORM	.00	.00	.00	.00	.00
	1,1,2,2-TETRACHLOROETHYLENE		1.56	2.66	3.68	
128	TOLUENE	.48	.90	1.76	2.58	1.30
	CHLOROBENZENE	.01	.02	.03	.04	.02
	ETHYL BENZENE	.08	.94	2.69	4.41	1 5 -
	PHENOL	1.30	2.36	4.31	6.37	3.28
	2,4-DIMETHY LPHENOL				3.22	1.33
		.08	.71	1.96		.50
	PENTACHLOROPHENOL DICHLOROBENZENES	.38 .34	.44 3.57	.55 10.05	.65	6.53
			3.57	10.05		1.00
	NAPHTHALENE	.14	.57		2.28	
	DIETHYL PHTHALATE	2.40	2.19	1.44	.90	1.70
	DI-N-BUTYL PHTHALATE	1.87	2.29	2.95	3.70	2.57
	BUTYL BENZYL PHTHALATE	1.16	2.52	5.15	7.81	3.82
	BIS(2-ETHYLHEXYL)PHTHALATE		1.38	1.84		1.59
	ANTIMONY	.42	.40	.29		.32
	ARSENIC		1.35	.94	. 67	1.08
	CADMIUM	.08	.25	.59	.93	.42
	CHROMIUM	2.83		20.80	32.59	
	COPPER	12.85	12.81	11.23		
	LEAD	8.16		13.66		
	MANGANESE	23.12	23.22	21.97		22.35
	MERCURY	. 05	.07	.08	.10	.07
	NICKEL		1.52		5.08	2.44
	SELENIUM	.70	.65	.46	.32	• 53
	SILVER	.47	1.73	4.20	6.70	2.95
	THALLIUM	•00	.00	.00	.01	.00
	ZINC	24.24	29.60	37.81	47.37	32.93
	TOTAL CYANIDES	. 27	1.03	2.51	4.01	1.75
	TOTAL PHENOLS	5.55	6.84	8.91	11.23	7.73
	AMMONIA	2.82	2.67	2.02	1.57	2.23
	OIL AND GREASE	10.02	10.12	9.81	9.56	9.91
	TSS	27.27	26.77	22.65	20.35	23.73
	TOC	14.33	14.25	12.85	12.02	13.24
	COD	48.72	49.27	46.05	44.87	46.53
	BOD	20.40	20.56	19.21	18.57	19.48
*					e a . /a	,

^{*} A flow of 0.08 kg/day would be equivalent to a concentration of 1 μ g/L. ** Classicals in 10³ kg/day.

Table 32

Relative Comparison of Hypothetical City Loadings*

		CASE	CASE	CASE	CASE	CASE
		A	<u> </u>	<u>C</u>	<u>D</u>	<u>E</u>
110	1.1-DICHLOROETHYLEHE	•00	.20	•60	1.00	.41
	1,1-DICHLOROETHAUE	•01	.21	.61	1.00	.41
112	TRANS-1, 2-DICHLOROETHYLEME	.03	.22	.62	1.00	.43
	CHLOROFO 244	.59	•68	. 85	1.00	•77
	1,2-DICHLOROETHANE	.13	.30	.67	1.00	.50
	1,1,1-TRICHLOROETHANE	.03	.23	.62	1.00	.42
	CARBON TETRACHLORIDE	.00	.20	•60	1.00	.40
	BROMODIC ALORO AETHANE	.17	.32	•70	1.00	• 56
120	TRICHLOROETHYLEU E	.15	.31	•69	1.00	• 54
121	BENZENE	• 44	•52	.85	1.00	.78
123	DIBROMOCHLOROMETHANE	.16	.31	.70	1.00	•55
	BROMOFORM	· C	0	0	0	0
127	1.1.2.2-TETRACHLOROETHYLEHE	.28	.42	.72	1.00	•59
	TOLUENE	.19	.35	.68	1.00	•53
	CHLOROBEN ZENE	.21	.37	•68	1.00	.52 .41
	ETHYL BENZENE	.02	.21	.61	1.00 1.00	.52
	PHENOL	.20	.37	.68	1.00	. 41
	2,4-DIMETHYLPHEHOL	.02	.22	.61	1.00	.7 7
	PEUTACULOROPHENOL	•59	.68	.85 .61	1.00	.41
	DICHLOROBENZENES	.02	.22	.63	1.00	.44
	<i>HAPHTHALEHE</i>	.06	.25 .91	.60	.37	.71
	DIETHYL PHTHALATE	1.00 .51	.62	.80	1.00	.69
	DI-N-BUTYL PHTHALATE	.15	.32	.66	1.00	.49
3 37	BUTYL BENZYL PHTHALATE	•15 •47	.59	.79	1.00	.68
	BIS(2-ETHYLHEXYL) PATHALATE	1.00	.95	.68	52	.76
	AHTIMOHY	1.00	.93	.65	.46	.74
	ARSENIC	.09	.27	.63	1.00	.45
	CADAIUM	.09	.27	.64	1.00	.46
	CHRONIUM	1.00	1.00	.87	. 82	.90
	COPPER	.47	.59	.78	1.00	•67
	LEAD MANGANESE	1.00	1.00	.95	.91	• 96
	MERCUKY	.52	.63	.80	1.00	.70
	ivic Kel	.12	.30	•65	1.00	.48
	SELENIUM	1.00	•92	•66	.46	. 76
	SILVER	.07	. 26	•63	1.00	•##
	THALLIUM	.18	.31	.75	1.00	•62
	ZINC	•51	.62	. 80	1.00	.70
	TOTAL CYANIDES	.07	.26	.62	1.00	.44
	TOTAL PHENOLS	.49	.61	.79	1.00	• 69
	AMMOUIA	1.00	•95	.71	. 56	.79
	OIL AND GREASE	• 99	1.00	.97	. 94	.98
	TSS	1.00	.98	. 83	.75	.87
	TOC	1.00	.99	• 90	.84	• 92
	COD	.99	1.00	.93	.91	•94 os
	BOD	.99	1.00	.93	•90	.95

^{*}Ratio of source SUMs to largest SUM.

2. Application to Cities Actually Sampled

In the detailed reports on the individual cities, attempts were made to conduct a mass balance analysis comparing sources and POTW influent. These comparisons were limited, however, due to having sampled only a relatively small fraction of a given source type from which to project or, as in the Hartford case, not having any industrial sites to represent the industrial component. One other means of carrying out the mass balance analysis is to use the average index values developed in this report to scale the basins sampled. This approach also provides an opportunity to compare the values projected from these source indices to actual influent values and thus serve as a test of the validity of the indices and the value in general of this type of approach.

Table 33 gives a summary of the basic characteristics of each of the cities which have been studied. These population and flow values have been used to scale the index values presented in part A of this section to give the mass contribution from each source type. From these values, an influent SUM was calculated and compared to the values actually observed at the POTW influent (INF). It must be borne in mind that these analyses are done using an "averaged" industrial value and are limited by the fact that the industrial contribution is a major component of the total and is industry specific.

The mass balance data for the cities using this approach involving the average data base values are given in Tables 34-37. The SUM/INF values for each have been summarized in Table 38 for convenience in analyzing the degree of "balance." SUM/INF values have been included only for those cases where the INF mass flow rate was at least 0.01 Kg/day. This value corresponds to an influent concentration of 0.1-0.35 µg/L, dependent on the city (see Table 33). Further, those SUM/INF values whose average influent concentrations were less than 1 µg/L, indicated by (), were excluded from interpretation because their values were too small to be reliable for the mass balance analysis. It is seen that some pollutants have higher source (SUM) levels than the influent (INF), while others are lower and some are about the same. Because of the uncertainties in each of the concentration and flow values used

Table 33
Summary of Discharge Characteristics for Cities Studie

POTW Mass Flow for		Population and Flow				
City	Influent Flow (Lps)	l μg/L Influent Concentration	R Population	Flow (Lps)	COM Lps	IND Lps
Cincinnati	427	0.04	87,900	385	32	4.3
St. Louis	1,022	0.08	200,000	876	100	122
Atlanta	4,072	0.35	385,000	1,687	847	729
Hartford	2,444	0.21	285,000	1,248	538	171

Table 34
Cincinnati Mass Balance Using Four City Averages

Kg/day* ** SUM Pollutant RES COM IND SUM INF INF .00 110 1.1 DICHLORDETHYLENE .00 .01 -00 -00 .00 .00 .00 111 1.1 DICHLOROBTHANE _GO .00 .00 .00 .01 112 TRANS-1, 2-LICHLOROETaYLEUE .00 .00 .10 2.01 113 Chloroform .17 .02 .00 .20 . .62 (.10).00 .00 .00 114 1,2mDICalokOETHANE .00 .01 (7.80)115 1.1.1 TKICaLOROETHANE .05 .01 .03 .08 .00 116 CARBON TETRACULORIUE .01 .00 .00 .01 700 .00 117 BKOMODICHLOKOMETHANE .00 .00 .01 .00 .04 120 TRICALOR OF THYLENE .09 .04 -01 .14 .14 121 BENZENE .01 .02 .01 .00 .00 123 LIBRONOCHLOROMETHAUE .00 .00 .00 .00 .00 .00 .00 .00 .00 125 bh OMORORN 10.86 127 1,1,2,2 **TETKACHLOKOFTHYLENE .37 .06 .46 .04 .03 .02 .22 .07 3.07 128 TOLUENE .17 .03 .00 .00 .00 129 CHLOROBENGENE .00 .00 .03 (2.15)130 ETUYL BENZENE .02 .04 .07 .01 .UG .55 203 PHENOL .01 .05 . 61 .03 .06 . CC 204 2,4 DIMETAYLPHENOL .03 .00 .14 1.19 .15 .17 210 PENTACHLOROPHENOL .00 .02 .28 .00 .12 .14 301 DICALOROBENZENES .02 .09 .14 .56 .05 .02 315 NAPHTHALEWE .01 2.45 326 DIETHYL PHTHALATE 1.03 .02 .00 1.05 .43 .79 1.8C 333 DIWNWBUTYL PHTHALATE .03 .02 .84 .40 337 BUTYL BENZYL PUTHALATE .47 .03 .06 . 56 .00 .15 3.00 338 BIS(2*FTHYLHEXYL)PHTHALATE .46 .02 .49 .02 .00 501 ANTIMONY .19 .00 .19 .00 .80 .30 502 ARSENIC .63 .01 .00 .64 .04 .01 504 CADMIUM .03 .09 .48 .00 .26 505 CHR JHIL'M 1.03 1.45 5.58 .20 .16 .05 5.44 5.64 2.40 .15 2.29 506 COPHER .59 0.24 3.40 .12 3.65 507 LEAD .14 .79 .09 10.01 508 MANGANESE 9.30 .62 12.62 509 MERCUKY .02 .00 .00 .02 .02 (1.56).24 510 HICKEL .23 .03 .04 .30 1.20 .00 .30 .01 .31 .20 1.56 511 SELENII'M .20 .01 .06 . 26 .14 512 SILVER 1.92 513 TuALLIUM .00 .00 .00 .00 .00 10.83 514 ZINC 10.13 .32 .79 .38 13.71 601 TOTAL CYANIDES .03 .15 .12 .00 1.46 .10 2.49 2.30 .08 .91 602 TOTAL PHENOLS .10 2.71 1.23 1.20 .00 703 AMNIONIA .03 .51 2.41 .30 704 OIL AND GREASE 3.99 -04 4.33 1.60 2.41 705 TSS .34 .08 11.93 11.52 3.87 3.09 706 TOC 5.89 .29 .05 6.24 1.58 3.94 .20 707 COD 20.09 .96 21.25 3.65 5.82 .08 8.88 708 BOD .44 1.75 5.06

^{*}Classicals in units of 10³ kg/day.

**Calculated for pollutants with INF >0.01 kg/day; values in () for INF
less than 0.04 kg/day.

Table 35
St. Louis Mass Balance Using Four City Averages

		Kg/day*					
	Pollutant	RES	СОМ	IND	SUM	INF	INF
110	1,1~DICaLOROETaYLENE	•00	.00	.12	.12	.08	1.62
	1, 1 w DIC n LOK OFT nAME	.00	.00	.02	.02	.03	(.56)
	THANSW1, 2 WDICHLORDETHY LENE	.00	.01	.12	.14	.04	(3.06)
	CHLOKOFORM	.39	.06	.13	.59	.54	1.06
	1,2#DICHLOROETHANE	.00	.00	.01	.01	.00	
	1,1,1.TKICHLOROETHANE	.10	.03	.90	1.03	.82	1.25
	CARBON TETRACHLORIUE	.00	.00	.30	.30	-00	
	BROMO DICHLOROMETHANE	.01	.01	.02	.03	.06	(.54)
	TRICaloroeTHYLENE	.10	.11	.27	.48	2.52	. 19
	beuzene	.02	.02	.01	.06	.62	.10
	DIBROMOCHLOROMETHANE	.00	.01	.01	.02	.09	.25
	BROMOFORM	.00	•00	.00	.00	.00	_
	1,1,2,2 TETHACHLORORTHYLENE	.85	.18	.74	1.77	3. 97	.45
	TOLUENE	.39	.09	.55	1.03	5.32	.19
	Culorobenzene	.01	.00	.01	.02	.01	(1.31)
	ETHYL BENZENE	.05	.03	1.06	1.14	1.38	.82
	Pn ENOL	1.26	.04	1.43	2.73	• 93	2.93
	2,4 DIMETHYLPHENOL	.08	.00	.78	.85	.00	
	PENT ACHLOROPHENOL	.33	.05	.11	.49	.00	4 .5
	DICHLOROBENZENES	.29	.07	3.97	4.31	2.36	1.87
	NAPATAALENE	.12	.02	.53	.68	.87	.77
	DIETHYL PHTHALATE	2.35	.05	.00	2.40	.62	3.90
	DIWNWHITYL PHIMALATE	1.77	.10	.71	2.58	1.40	1.85
	BUTYL BENZYL PHTHALATE	1.07	.09	1.77	2.93	1.00	2.93
	BIS (2wEThYLHEXYL) PhTHALATE	1.04	.07	•45	1.56	.38	4.12
	ANTINONY	.42	.00	.02	.44	4.66	.09
	AKSENIC	1.43	.02	.03	1.49	.00	
	CADMIUM	.08	.00	.22	.30	. 26	1.15
	CHROMIUM	2.34	.49	7.52	10.35	11.95	.87
	COPPEK	12.38	.47	1.32	14.16	4.11	3.45
	LEAD	7.73	.43	3.41	11.57	18.58	.62
	MANG ANESE	21.17	1.94	2.45	25.56	17.79	7.44
	MERCURY	.05	•00	.02	.07	•04	(1.66)
	NICKEL	•52	.11	1.15	1.77	4.05	.44
	SELENIUM	.68	.03	.01	.71	• 34	1.88
	SILVER	•45	.03	1.59	2.06	1.41	1.45
	THALLIUM	.00	.00	.00	.00	.00	
	ZINC	23.05	1.19	9.07	33.31	25.04	1.30
	TOTAL CYAUIUES	.27	.00	• 96	1.22	1.29	.95
	TOTAL PAENOLS	5.23	.32	2.15	7.70	5.37	1.43
	AMMONI A	2.73	-09	.11	2.93	1.50	1.90
	OIL AND GREASE	9.09	-94	1.12	11.14	2.74	4.07
	TSS	26.21	1.06	2.27	29.54	11.02	2.68
706		13.41	- 92	1.37	15.70	8.58	1.83
	COD	45.73	2.99	5.70	54.42	26.98	2.02
	BOD	19.02	1.38	2.28	22.68	13.00	1.66

^{*} Classicals in units of 10³ kg/day.

^{**}Calculated for pollutants with INF >0.01 kg/day; values in () for INF less than 0.08 kg/day.

Table 36 Atlanta Mass Balance Using Four City Averages

Kg/dav*

	Kg/day*					**
Pollutant	RES	COM	IND	SUM	INF	SUM
110 1,1eDICaLOROETaYLEWE	.00	.02	.73	.75	3.02	.25
111 1.1*DICaLOROETHANE	.00	.01	.10	.10	.00	
112 ThANSw1.2wDIChLORDEThYLENE	.00	.11	.74	.85	6.54	.13
113 ChLOROFORM	.76	.49	.75	2.01	2.51	.80
	.00	.01	.04	.05	.14	(.37)
114 1,2*DICaLOROFTAANE	.20	.21	5.36	5.7 <u>8</u>	33.73	.17
115 1,1,1*TKICHLOROETHANE		.01	1.79	1.80	• 🖒	
116 CARBON TETRACHLORIUE	.00		.10	.18	.00	
117 BROWDICALOR OWETHANK	.01	.07		2.73	58.02	.05
120 THICHLOROETHYLENE	.19	.94	1.60		.00	
121 BEUGEWE	•05	.20	.09	.32	.00	
123 DIBROMOCALOROMETHANE	.01	•05	.07	.13	.60	
125 BROMOFORM	.00	.00	.00	.00	84.21	.09
127 1,1,2,2 TETHACHLOROETHYLENE	1.63	1.57	4.40	7.60	3.97	.54
128 ICLUENE	.74	. 90	3.29	4.94	.00	• - 1
129 CalokobEWZEwE	.02	.00	.06	.08	17.15	.39
130 ETHYL BEWZENE	.10	.22	6.33	6.65	6.63	1.71
203 PhENOL	2.42	.33	8.56	11.31	3.48	1.35
204 2,4*DIMETnYLPnENOL	•14	-00	4.66	4.80		
210 PENTACALOROPAENOL	. 65	.42	.63	1.70	6.75	.25 .76
301 DICHLOROBENZENES	.54	•55	23.72	24.90	32.62	
315 NAPATRALENE	.23	.19	3.20	3.52	11.55	.31
326 DIETHYL PhInALATE	4.53	.42	.00	4.95	1.75	2.82
333 DIWAWBUTYL PATHALAPE	3.41	•86	4.23	٩.50	1.55	5.50
337 BUTYL BENZYL PHTHALATE	2.0ñ	•77	10.60	13.43	27.18	.49
339 BIS(2-ETHYLHEXYL)PHTHALATE	2.00	.56	2.71	5.27	.00	
501 ANTIMONY	.81	.02	.11	.94	.22	(4.36)
502 ARSENIC	2.76	.19	.20	3.15	.00	
504 CADMIUM	.15	-04	1.31	1.50	1.10	1.36
505 CakOMIUM	4.52	4.16	44.92	53 .60	25.37	2.11
506 COPPER	23.85	3.99	7.86	35.70	17. 71	2.02
507 <i>LEAD</i>	14.90	3.64	20.39	39.9 3	47.72	.82
508 MANGANESE	40.80	16.45	14.62	71.8 7	97.63	.74
509 MERCURY	.10	.03	.12	. 24	.28	(. oc)
510 NICKEL	1.00	.91	6.84	8.75	6.45	1.30
511 SELEUIUM	1.30	.24	.05	1.60	.00	
512 SILVER	.86	.21	9.48	10.55	4.37	2.41
513 THALLIUN	.00	.01	.00	.01	•0C	
514 2IMC	44.42	10.10	54.17	108.69	124.20	.87
601 TOTAL CYANILES	•51	.02	5.71	6.24	1.73	3.02
602 TOTAL PHENOLS	10.08	2.70	12.95	25.64	35.13	.73
703 AMMONIA	5.26	.78	.67	6.71	2.01	2.57
704 OIL AND GEEASE	17.50	7.98	6.69	32.16	10.04	3.20
705 ISS	50.51	8.96	13.59	73.06	48.51	1.51
706 <i>19C</i>	25.85	7.77	9.20	41.92	23.08	1.75
707 COD	99.13	25.32	34.03	147.48	υ5.03	2.27
708 <i>BOD</i>	36.65	11.71	13.61	61.97	35.49	1.75
*					•	- •

^{*} Classicals in units of 10³ kg/day.

Calculated for pollutants with INF >0.01 kg/day; values in () for INF less than 0.35 kg/day.

Table 37
Hartford Mass Balance Using Four City Averages

	Kg/day*					**
Pollutant	RES	COM	IND	SUM	INF	SUM
110 1,1-DICHLOROETHYLENE	.00	.01	.17	.18	.00	
111 1.1-DICHLOROETHANE	.00	.00	.02	.03	.00	
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.07	.17	.24	.00	
113 CHLOROFORM	. 56	.31	.18	1.05	.77	1.36
114 1,2-DICHLOROETHANE	.00	.01	.01	.02	.00	
115 1.1.1-TRICHLOROETHANE	.15	.14	1.26	1.54	2.16	.71
116 CARBON TETRACHLORIDE	.00	.00	.42	.42	.00	•
117 BROMODICHLOROMETHANE	.01	.05	.02	.08	.00	
120 TRICHLOROETHYLENE	.14	.60	.38	1.11	1.70	.52
121 BENZENE	.03	.12	.02	.18	.00	
123 DIBROMOCHLOROMETHANE	.01	.03	.02	.06	.00	
125 BROMOFORM	.00	.00	.00	.00	.00	
127 1, 1, 2, 2-TETRACHLOROETHYLENE	1.21	1.00	1.03	3.23	5.54	.58
128 TOLUENE	. 55	.51	.77	1.83	3.29	.56
129 CHLOROBENZENE	.01	.00	.01	.03	.00	
130 ETHYL BENZENE	.08	.14	1.48	1.70	.00	
203 PHENOL	1.79	.21	2.01	4.01	.00	
204 2,4-DIMETHYLPHENOL	. 11	.00	1.09	1.20	.00	
210 PENTACHLOROPHENOL	.48	.27	.15	.89	.60	
301 DICHLOROBENZENES	.40	.35	5.56	6.31	2.83	2.23
315 NAPHTHALENE	.17	.12	.7 5	1.04	.00	
326 DIETHYL PHTHALATE	3.35	.27	.00	3.62	.77	4.72
333 DI-N-BUTYL PHTHALATE	2.53	•55	.99	4.06	€8.	4.56
337 BUTYL BENZYL PHTHALATE	1.52	.49	2.49	4.50	.00	
338 BIS(2-ETHYLHEXYL)PHTHALATE	1.48	.36	.63	2.47	.00	
501 ANTIMONY	.60	.01	.03	.64	.00	_
502 ARSENIC	2.04	.12	.05	2.21	.41	5.38
504 CADMIUM	.11	.03	.31	• ##	.00	
505 CHROMIUM	3.34	2.64	10.54	16.52	13.81	1.20
506 COPPER	17.64	2.53	1.84	22.01	20.40	1.08
507 LEAD	11.01	2.31	4.78	18.11	7.51	2.41
508 MANGANESE	30.17	10.45	3.43	44.05	33.37	1.32
509 MERCURY	.07	.02	.03	.12	.00	
510 NICKEL	. 74	.58	1.61	2.92	7.39	.40
511 SELENIUM	. 96	.15	.01	1.13	.00	
512 SILVER	.64	.13	2.22	2.99	•69	4.34
513 THALLIUM	.00	.01	.00	.01	.00	4 56
514 ZINC	32.84	6.42	12.71	51.97	33, 23	1.56
601 TOTAL CYANIDES	.38	.01	1.34	1.73	. gu	2.07
602 TOTAL PHENOS.S	7.45	1.72	3.01	12.19	11.08	1.10
703 AMMONIA	3.89	.50	. 16	4.54	1.59	2.28
704 OIL AND GREASE	12.94	5.07	1.57	19.57	7.85	2.49
705 <i>TSS</i>	37.35	5.69	3.19	46.23	16.23	2. 85
706 <i>TOC</i>	19.12	4.93	1.92	25.97	8.98	2.89 2.21
707 COD	65.17	16.08	7.98	89.23	40.3t	2.62
708. <i>BOD</i>	27.10	7.44	3.19	37. 73	14.37	2.02

^{*} Classicals in units of $10^3 \ kg/day$.

^{**}Calculated for pollutants with INF >0.01 kg/day; values in () for INF less than 0.21 kg/day.

Table 38

Mass Balance Analysis For All Four Cities

SUM/INF

	DOLLY INT			
	Cincinnati	St. Louis	Atlanta	<u>Hartford</u>
-110-1,1-010mb0kosad 1b5k \$		1.62	.25	
111 1.1-DICTLORGETTANE		(.56)		
112 Indis-1,2-DienberdentEub		(3.06)	.13	
113 ChLOROEOEM	2.01	1.06	.80	1.36
114 1,2-biCabulOsThauk	(.10)		(.37)	
115 1.1.1-TRICHLOAGETHANE	(7.80)	1.25	.17	.71
116 CALBON ISAMAGULOKIDE				
117 BECHODICHLORDALTAAIL		(.54)		
376 rkiJHLOkOsimiLani		. 19	.05	.62
121 HEGGENE	.14	.10		
123 Jubnel ConLordinate		.25		
125 BROMORORI				
127 1,1,2,2-TETHACHLOROLTHILENE	10.86	.4 ^K	. 09	.58
178 YODUSHE	3.07	.19	. 54	.56
12% CnLohObbli24u2		(1.31)		
170 SINIS CONNENT	(2.15)	.82	.39	
203 ma.OL		2.93	1.71	
204 CA-Dianalization			1.38	
210 ESATECALOROFACAOL	1.19		.25	
301 bICalonesbuZEMbS		1.87	.76	2.23
315 & APRITHALENE	.56	.77	.31	
320 DlaFaft FaTaALATE	2.45	3.90	2.82	4.72
333 Stere Cottin anthaland	1.80	1.85	5.50	4.56
337 EULLE BEHALL EMINALATE		2.93	.49	
33c Lis (7-oral bushil) kai'asbai's	3.00	4.12		
501 auTlaOut		.09	(4.36)	
502 AhShuTC	. 50			5.38
504 CADITUM	.48	1.15	1.36	
505 Cantial Ula	. 20	. 67	2.11	1.20
Sia JureaR	2.46	3.45	2.02	1.08
507 <i>LEAD</i>	6.24	.62	.82	2.41
FOS LANGAHESE	.79	1.44	• 74	1.32
509 AshCURI	(1.56)	(1.66)	(.86)	
510 MICKED	.24	•44	1.36	.40
511 SELLEIUM	1.56	1.88		
512 SILVER	1.92	1.45	2.41	4.34
513 InALLIUM				
514 ZIGC	.79	1.30	.87	1.56
601 TOTAL CYALIDES	.10	.95	3.62	2.07
602 TUTAL FRENUIS	2.71	1.43	.73	1.10
763 AMAGNIA	2.41	1.96	2.57	2.28
704 OIL AND GREASE	2.41	4.67	3.20	2.49
765 753	3.09	2.68	1.51	2.85
760 1 <i>UC</i>	3.94	1.83	1.75	2.89
7G7 COU	3.6 5	2.02	2.27	2.21
70n 50D	5.06	1.66	1.75	2.62

to obtain these data, in addition to scaling errors, it is estimated through an accumulation of errors analysis that pollutants whose SUM/INF value falls in the 0.5-2.0 range are in balance.

In Table 39, various comparisons of the degree of balance between cities have been made (part A). Additional comparisons were made by analysis category (part B) to see if some types balanced better than others, and by examining those which balance in a given percentage of the cities (part C). On the whole, the priority pollutants balance one-half to two-thirds of the time, for the cases where the influent mass flow was high enough to conduct the comparison. A much larger number of pollutants "balance" if the error range is opened to a factor of 4, i.e., 0.25x - 4x. This range would appear to be suggested by the magnitude of the variance in the average index values, which is about equal to the value in most cases.

More than 50% of the volatiles data at the influent were too low to be treated in this manner, although many of these pollutants were observed in the sources. For cases where the INF volatiles levels are measurable, they balance 12 out of 26 times. They project high 2 out of 26 times and low 12 out of 26 times.

The acids and base/neutrals balance 10 out of 23 times and project high 11 out of 23 times. This pattern is reinforced by the classicals measurements which balance 7 out of 24 times and project high 17 out of 24 times. The classicals never project low.

These observations on the volatiles, acids, base/neutrals, and the classicals support the general considerations of raw wastewater collection systems which indicate that a large fraction of the "treatment" occurs in the collection system, in addition to that which occurs in the POTW. The data indicate that all of these groups are initially high at their source and undergo some degradation in the collection system before reaching the POTW. In many cases, the levels are low enough at the POTW not to be detected.

This hypothesis is supported by the data for the metals, which are always analyzed only as the element. These elements, therefore, should

Table 39
Summary of Mass Balance Comparisons

A. BY CITY

Number of Pollutants (40 total)	Cincinnati	St. Louis	Atlanta	<u>Hartford</u>
Balance-priority pollutants	10	16	13	10
-classicals	0	4	3	0
Sources Greater-priority pollutants	7	5	5	7
-classicals	6	2	3	6
Influent Greater-priority pollutants	4	7	8	1
-classicals	0	0	0	0
Too Small at Influent to Balance	4	5	3 ,	0
Less Than 0.01 Kg/day at Influent	15	7	11	22

В.	BALANCE BY ANALYSIS CATEGORY	Balance	Sources <u>Creater</u>	Influent <u>Greater</u>
	Volatiles (26 values)	12	2	12
	Acids, Base/Neutrals (5+18=23 values)	10	11	2
	Metals (36 values)	24	7	5
	Total Cyanides/Total Phenols (8 values)	4	3	1
	Classicals (24 values)	7	17	0

C. NUMBER OF POLLUTANTS WHICH BALANCE IN A GIVEN % OF CITIES

	Priorit	y Pollutan	ts	
	Organics	Metals	TÇN/TP	Classicals
% of Cities				
100	2	4	-	-
75	0	1	1	-
66	3	_	-	-
50	7	5	1	2
33	2	4	-	-
25	1	-	-	3
0	5	1	-	1

be present in the influent at about the same level as the sources, even though their molecular association may be different. The metals balance 24 out of 36 observations and are about evenly projected high (7/36) and low (5/36) a small fraction of the time. On the average, the high projections are 3.7 times the influent and the low projections are 0.26 times the influent. All but 2 of the total of these 12 out-of-balance values fall within a factor of 4 range of the influent value.

A further comparison of this data can be made, within the uncertainties imposed by the mass balance by comparing the relative contribution of each source type for each pollutant (similar to what was done for the hypothetical cases). These ratios for the sources within the cities are given in Tables 40-43, along with the SUM value in Kg/day (10³ Kg/day for the classicals). This type of comparison could be viewed as the analysis of basins whose mix of source types was actually as represented by the average character of the source sites sampled for each category and scaled by the actual source flows for these cities.

For "Cincinnati" (Table 40), the residential sources dominate the pollutant mass flow, but the area is predominantly residential. For 'St. Louis" (Table 41), residential sources are still important, but many toxic pollutants are dominated by the industrial category, even though it only has about 12% of the flow. The industrial category dominance of "Atlanta" (Table 42), is clear from this presentation. "Hartford" (Table 43), with a small (7%) industrial component, shows a balance in the importance of each of the source types.

Table 40

Cincinnati Distribution of Pollutant Loading
(91% Residential, 8% Commercial, 1% Industrial Flow)

	Fra	action of S	SUM		
Pollutant	RES	COM	IND	SUM (Kg	/day)*
110 1,1*DICaLORORTHYLENE	.00	.15	. A5	.01	
111 1, 1.DICaLORONIANE	.00	• 25	.75	.00	
112 Thans-1, 2-DICaloroftaYLENE		.50	.50	.01	
113 Calokorokia	.88	.10	.02	.20	
114 1,2-DICALOROFTHAUF	.66	•22	.13	.00	
115 1,1,1-TRICALOROFTHANA	•53	.10	.37	.09	
116 CARBON TETRACALORIUE	.00	.02	.98	. 01	
117 BKCMODICALOROMETAANE	.45	.45	.10	.01	
120 TAYCULOKOLTUYLENE	.49	.41	.11	.09	
121 BENZENE	.57	.40	.03	•02	
123 DIBA 249ChLOR 2MFTu Auf	.40	.49	.11	.00	
125 BKOMOFORE	.00	.00	.00	.90	
127 1, 1, 2, 2 TETA ACALOR OF THY LEWE	.81	.13	.06	. 46	
128 TOWENE	.77	.14	.09	.22	
123 CnLOROBENAKAF	.89	.03	.08	.00	
130 ETHYL BEHZEHE	.34	.12	.54	.07	
203 PaENOL	.30	.02	•08	.51	
204 2,4+DIMHInYLPnEWOL	•55	•00	.45	.06	
210 PENTACALOKOPANNOL	.88	.10	•62	. 17	
301 DICHLOKOBENZENES	.43	.07	.49	. 29	
315 NAPhThALE of	.67	.09	. 24	.08	
326 DIMINYL PHINALATE	.98	.02	.00	1.05	
333 DIwiwebUTYL PnTnALATA	•93	.04	.03	.84	
337 BUTYL BEWAYL PATRALATE	.84	•05	.11	.56	
339 EIS(2mkTmYLmEXYL)PmThALATE	•92	•04	.03	•49	
501 ANTIMONY	.99	.00	.00	.19	
502 ARSENIC	.99	.01	.00	.64	
504 CALMIUN	.79	•04	•18	•0#	
505 ChkOMIUW	.71	.11	.19	1.45	
506 COPPER	•97	.03	.01	5.64	
507 <i>LEAD</i>	.93	•04	.03	3.65	
509 MANGANESE	.93	•06	.01	10.01	
509 MERCURY	.93	.05	.03	.02	
510 WICKEL	.75	.11	.13	.30	
S11 SELENIUN	.97	.03	.00	.31	
512 SILVER	.75	.03	.22	.26	
513 THALLIUM	.00	.91	.09	.00	
514 ZINC	.94	-04	.03	10.93	
601 TOTAL CYANIDES	.77	.00	.22	.15	
602 TOTAL PHENOLS	.93	.04	.03	2.48	
703 AMMCNIA	.97 .92	.02 .07	.00 .01	1.23 4.33	
704 CIL AND GREASE	.96	.03	.01	11.93	
705 <i>TSS</i>	•95	.05	.01	6.24	
706 TOC 707 COL	.95	.05	.01	21.25	
707 692 708 <i>B</i> 2E	.93	.05	.01	8.98	
ענים איני	. 34	•05	•01	3679	

^{*}Classicals in 103 kg/day.

Table 41

St. Louis Distribution of Pollutant Loading (80% Residential, 9% Commercial, 11% Industrial Flow)

,	1	Fraction o	f SUM		
Pollutant	RES	COM	IND	SUM	(Kg/day)*
110 1.1 DICHLORORTHY DENE	.00	.02	.98	.12	
111 1, 1. DICALORONIA AND	.00	.03	•97	.02	
112 Thansal, 2-DICHLORDETHYLENE	.00	.10	.90	.14	
113 Culokokokm	.68	.10	.22	.58	
113 Culoroform 114 1,2 Licalorofitame	.26	.12	.62	.01	
115 1,1,1.TRICALORDETHARE	.10	.02	.88	1.03	
116 CARBON TETRACHLORIDE	.00	.00	1.00	.30	
117 BROWDDICALDROMETHANE	.20	.27	•53	.03	
120 TRICALORORTHYLEWE	.20	.23	•56	.48	
121 BENGEWE	.40	.38	•22	.06	
123 DIBROMOCHLOROMETHAWE	.17	.28	.55	.02	*
125 6k 2h0F0RM	•00	.00	.00	.00	
127 1,1,2,2 WIETK ACULOROFT HYLENE	*#B	.10	.42	1.77	
128 TOLUENE	.37	.09	.53	1.03	
129 CHLOROLENZENE	.46	.02	.52	.02	
130 ETHYL BENZENE	.05	.02	.93	1.14	
203 PARNOL	. 46	.01	.53	2.73	
204 2,4 abIMETaYLFaEnOL	.09	.00		.95	
210 PENTACALOROPHENOL	.69	.10		.49	
301 DICALOROBENZENES	.06	.02			
315 NAPATHALENE	.18	.03	.79	.68	
326 DIETNYL POTOALATE	.98	.52	.00	2.40	
333 DI =N =BUTYL PHTHALATE	.69	.04	.27	2.59	
337 BUTYL BENZYL PHTHALATE	•3b	.03	.60	2.93	
339 bis(2weThYLnEXYL)PnThALATE	.67	.64	. 29	1.56	
501 ANTIMONY	.95	.01	.04	.44	
502 AKSENIC	.96	.02	.02	1.49	
564 CADMILLA	.25	.02	.73	.30	
505 CahOMIUM	.23	.05		10.35	
506 COPPER	.97	.03		14.16	
507 Lh.AD	.67	.04		11.57	
508 MANGAINESE	.83	.08	.10	25.56	
509 GERCURY	. 6 9	•05		.07	
510 uICAEL	. 29	.05	.65	1.77	
511 SELEWIUM	. 95	.04	.01	.71	
512 SILVER	.22	.01	.77	2.06	
513 TaALLIUN	.00	.54	.46	.00	
514 aIuC	.69	.04	.27	33.31	
601 FOTAL CYANIDES	•22	.00	.78	1.22	
602 TOTAL PARNOLS	.ba	.04	.28	7.70	
703 AMADNIA	• 93	.03	.04	2.93	
704 OIL AND GREASE	. ⁸²	•08	.10	11.14	
705 TSS	.89	.04	•08	29.54	
706 <i>TOC</i>	.85	.06	.09	15.70	
707 COU	.84	.05	.10	54.42	
708 bOU	.84	•06	.10	22,69	
→ 					

^{*}Classicals in 103 kg/day.

Table 42

Atlanta Distribution of Pollutant Loading
(52% Residential, 26% Commercial, 22% Industrial Flow)

	Fr	action of	SUM	*
Pollutant	RES	COM	IND	SUM (Kg/day)
110 1, 1+DICalOnONTaYLENK	.00	.03	.97	.75
111 1.1mDICaLORONTHANE	.00	.05	•95	.10
112 THANS #1, 2 DICALORONTHY LERE	.60	.13	.87	.85
113 CHLOKOFORM	.38	.25	.39	2.01
114 1,2*DICuLORONTuANE	.10	.19	.71	.05
115 1,1,1-ThICHLOROETHANE	.03	.04	.93	5.78
115 CARBON TETRACHLORIUM	.00	.00	1.00	1.90
117 EKOMODICHLOKOMMIHAWM	.06	• 39	.54	.19
120 ThICHLOROETHYLEUE	.07	.34	•59	
121 DENGENE	.14	.61	. 24	.32
123 DIBROMOCALOROMETHANE	.05	•40	.55	. 13
125 BROMOFORM	.00	.00	.00	.00
127 1,1,2,2mFETKACdLOKOKTaYLEWE	.21	.21	.58	7.60
129 IOLIENE	.15	.17	.68	4.84
129 CHLOKOBENAENE	.22	•04	.74	.08
130 ETHYL BENZEWE	.02	.03	.95	6.65
203 PHENOL	.21	.03	.76	11.31
204 2,4 EDIMETHYLPHEADL	.03	.00	.97	4.80
210 PENTACHLOROPHENOL	.39	. 25	.37	1.70
301 DICALOROBENZEWES	.02	.02	.96	24.80 3.62
315 uAPHPaALEUE	.06	.05	.88	4.95
326 DIKINYL PHINALAIK	.92	.08	.00	8.50
333 DI THE BUTYL PATAALATE	•40	.10	.50 .79	13.43
337 BUTYL BENZYL PHTHALATE	.15	.06	.51	5.27
339 DIE (2-ETHY LOEXYL) POTGALATE	•38 oc	.11 .02	.11	.94
501 AUTIMONY	•86 90	.02	.06	3.15
502 AKSENIC	.98 .10	.03	.87	1.50
504 CADMIUN	.09	•04	.84	53.60
505 CukCNIUM	.67	.11	.22	35.70
506 COPPER	.38	.09	.52	38.93
507 LEAD	.57	.23	.20	71.97
508 MAUGANESE	.40	.12	.48	. 24
509 MERCURY	.11	.10	.78	8.75
510 mickel 511 selenium	.91	.15	.03	1.60
511 SHINION 512 SILVER	.09	.02	.90	10.55
513 THALLIUM	.00	.62	.34	.01
514 ZIÚC	.41	.09	.50	108.69
601 TOTAL CYANIDES	. OR	.00	.92	6.24
602 FOTAL PHENOLS	.39	.11	.50	25.64
703 AMACHIA	.78	.12	.10	6.71
704 OIL AND GREASE	.54	.25	.21	32.16
705 TSS	.69	.12	.19	73.06
706 <i>F2C</i>	.62	.19	.20	41.82
707 COD	.60	.17	.23	147.48
708 BDU	.59	.19	.22	61.97

^{*}Classicals in 10³ kg/day.

Table 43

Hartford Distribution of Pollutant Loading
(64% Residential, 27% Commercial, 9% Industrial Flow)

(1	Fraction of SUM			*
Pollutant	RES	COM	IND	SUM (Kg/day)
110 1.1-DICHLOROETHYLENE	.00	. 07	.93	.18
111 1,1-DICHLOROETHANE	.00	. 12	.88	.03
112 TRANS-1, 2-DICHLOROETHYLENE	.00	.29	.71	. 24
	.53	.30	.17	1.05
113 CHLOROFORM 114 1,2-DICHLOROETHANE	.20	.34	.46	.02
115 1,1,1-TRICHLOROETHANE	.09	.09	. 82	1.54
116 CARBON TETRACHLORIDE	.00	.01	.99	.42
117 BROMODICHLOROMETHANE	.11	.59	.30	.08
120 TRICHLOROETHYLENE	.12	• 54	.34	1.11
121 BENZENE	.19	.70	.10	.18
123 DIBROMOCHLOROMETHANE	.09	. 60	.31	.06
125 BRONOFORM	.00	.00	.00	.00
127 1, 1, 2, 2-TETRACHLOROETHYLEN E	.37	. 31	.32	3.23
128 TOLUENE	.30	.28	.42	1.83
129 CHLOROBENZENE	.44	.07	.49	.03 1.70
130 ETHYL BENZENE	. 04	.08	.87	4.01
203 PHENOL	.45		•50	1.20
207 217 22322323	.09		.91 .17	.89
ZIO I BNI ADNIBONOI NENO	.53	.30	.88	6.31
301 D1011D0110D311-1-1	.06	.06 .12	.72	1.04
212 MATHIMADUMD	.16 .93		.00	3.62
OZO DIUINID INIMA	.62	.13	.24	4.06
333 DI-N-BUTYL PHTHALATE	. 34	.11	.55	4.50
337 BUTYL BENZYL PHTHALATE	.60	.14	.26	2.47
338 BIS(2-ETHYLHEXYL)PHTHALATE	.94	.02	.04	.64
501 ANTIMONY	.92	.05	.02	2.21
502 ARSENIC	.25	.06	.69	• ##
504 CADNIUM	.20	. 16	.64	16.52
505 CHROMIUM 506 COPPER	.80	.12	.08	22.01
	.61	.13	.26	18.11
507 <i>LEAD</i> 508 <i>MANGANESE</i>	.68	. 24	.08	44.05
509 MERCURY	.61	.16	.23	.12
510 NICKEL	.25	. 20	•55	2.92
511 SELENIUM	.85	. 14	.01	1.13
512 SILVER	.21	.05	.74	2.99
513 THALLIUM	.00	.82	.18	.01
514 ZINC	.63	.12	.24	51.97
601 TOTAL CYANIDES	.22	.01	.77	1.73
602 TOTAL PHENOLS	.61	. 14	.25	12.19
703 AMMONIA	. 86	.11	.03	4.54 19.57
704 OIL AND GREASE	.66	.26	.08	46.23
705 <i>TSS</i>	.81	.12	.07 .07	25 . 97
706 <i>TOC</i>	.74	.19 .18	.07	89.23
707 COD	.73 .72	.20	.08	37 .73
708 BOD	• 12	• 20	• • • •	

^{*}Classicals in 10³ kg/day.

D. Examination of Variances and Correlations

The design of the sampling plan has provided an opportunity to examine some of the secondary objectives, while the restrictions imposed by the site characteristics or other factors, such as weather, limited the ability to examine other objectives. Some of the differences between weekday and weekend effects and old and new residential sources are summarized in this section. Some limited runoff results obtained during the Hartford study are discussed.

1. Weekday/Weekend Differences

An exploratory test of differences between weekday and weekend samples suggests that, in the aggregate, priority pollutants are found more frequently in weekday than weekend samples. This result is indicated for all source types and for the influent. A contingency table was formed separately for each source type over all cities and over all pollutants. These tables display the number of pollutants detected vs. the number of pollutants not detected, summed over all samples and all pollutants separately for weekday samples and weekend samples.

	detections	non- detections
weekdays	n ₁	n ₂
weekends	n ₃	n ₄

Under the null hypothesis that the day of the week does not affect the likelihood of any particular pollutant being present, the fraction of weekday detections $\begin{pmatrix} n_1 \\ \hline n_1 + n_2 \end{pmatrix}$ would be approximately equal to the

fraction of weekend detections
$$\left(\frac{n_3}{n_3 + n_4}\right)$$
.

For each source type, weekday samples slightly exceeded weekend samples in the frequency of detections as follows:

and the land	Residential	Commercial	Industrial	All Source Sites	Influent
weekday fraction of detections	14%	15%	24%	16%	19%
weekend fraction of detections	12%	13%	21%	14%	17%

This small but consistent difference is statistically significant at the 95% level (given the simplifying assumption of independence of all pollutants and all samples) when considering all aggregated source sites. This procedure only considers the relationship of day of the week with the absence or presence of pollutants and does not address the concentrations of pollutants.

2. Old vs. New Residential Comparisons

The sampled residential areas have been separated by the approximate age of housing into old and new residential areas. This initial comparison was performed on an average mass per capita basis separately for each pollutant, and on the basis of frequency of detection across all pollutants.

Table 44 shows the average mass per capita in mg/person/day for the 6 new residential sites and the 5 old residential sites sampled over the four cities. Also displayed is the ratio of the averages to the greater of the two averages for each pollutant. The majority of pollutants show higher per capita mass contributions from old residential areas.

A contingency table was formed to display the number of detections of pollutants at old residential vs. new residential sites.

	detections	non- detections
new residential	ⁿ 1	ⁿ 2
old residential	ⁿ 3	n ₄

If the age of housing were independent of the number of detections found at the sampling site, then the ratio of detections to total samples

Table 44
Old and New Residential Mass Discharge Rates

	Mg/person/day*		Ratio	
	New	Old	New	01d
110 1,1-DICHLOROETHYLENE	.00	.00		
111 1.1-DICHLOROETHAUE	.00	•00		
112 TRANS-1,2-DICHLOROETHYLENE	•00	•00		
113 CHLOROFORM	1.48	2.38	-62	1.00
114 1.2-DICHLOROETHANE	.00	.07	•00	1.00
115 1.1.1-TRICHLOROETHANE	2.36	.23	1.00	.10
116 CARBON TETRACHLORIDE	•00	.00		
117 BROMODICHLOROMETHANE	.00	.09	.00	1.00
120 TRICHLOROETHYLENE	.43	.03	1.00	.08
121 BENZENE	.03	.27	.12	1.00
123 DIBROMOCHLOROMETHANE	.00	• 06	.00	1.00
125 BROMOFORM	.00	.00		4 00
127 1, 1, 2, 2-TETRACHLOROETHYLENE	3.19	3.25	.98	1.00
128 TOLUENE	.75	3.42	•22	1.00
129 CHLOROBENZENE	.00	.12	•00	1.00
130 ETHYL BENZENE	.06	.70	•08	1.00 1.00
203 PHENOL	1.43	10.31	.14	1.00
204 2.4-DIMETHYLPHENOL	.14	.58	.24 1.00	.00
210 PENTACHLOROPHENOL	1.55	.00	.37	1.00
301 DICHLOROBENZENES	.82	2.22	.00	1.00
315 NAPHTHALENE	.00	3.46	.46	1.00
326 DIETHYL PHTHALATE	6.56	14.29	1.00	.77
333 DI-N-BUTYL PHTHALATE	8.24	6.32	.57	1.00
337 BUTYL BENZYL PHTHALATE	3.72	6.52 10.50	.18	1.00
338 BIS(2-ETHYLHEXYL)PHTHALATE	1.94	1.41	.84	1.00
501 ANTIMONY	1.19	6.22	.87	1.00
502 ARSENIC	5.43 1.24	1.01	1.00	.82
504 CADMIUN	6.82	25.22	.27	1.00
505 CHRONIUM	56.25	69.57	.81	1.00
506 COPPER	15.67	116.31	.13	1.00
507 <i>LEAD</i> 508 <i>MANGANES</i> E	86.88	100.23	.87	1.00
508 MANGANESE 509 MERCURY	.20	•21	•95	1.00
510 NICKEL	1.44	3.58	.40	1.00
510 NICKEL 511 SELENIUM	2.16	3.04	.71	1.00
512 SILVER	1.94	.15	1.00	.08
513 THALLIUM	.00	•00		
514 ZINC	79.13	219.89	.36	1.00
601 TOTAL CYANIDES	1.20	•22	1.00	.18
602 TOTAL PHENOLS	16.48	34.99	.47	1.00
703 AMMONIA	11.24	11.87	.95	1.00
704 OIL AND GREASE	26.37	89.24	.30	1.00
705 2'SS	81.43	176.99	•46	1.00
706 TOC	54.70	69.54	.79	1.00
707 <i>COD</i>	199.84	218.88	.91	1.00
708 <i>BOD</i>	76.18	103.99	.73	1.00

^{*}Classicals in kg/day.

$$\left(\frac{n_1}{n_1+n_2}\right)$$
 and $\left(\frac{n_3}{n_3+n_4}\right)$ should be approximately equal for new and old

residential areas, i.e., the presence of pollutants would be about the same. For all four cities and all pollutants aggregated, 12% of pollutants tested were detected in new residential areas versus 13% in old residential. This small difference is not consistent among the cities taken separately and is not statistically significant in the aggregate. Age of housing appears to affect the level of mass contribution from residential, but has little effect on the frequency of presence of pollutants.

3. Runoff

During the Hartford study, a limited amount of information was obtained on the effect of rain on the mass flow rate of certain metals in the collection system. The data base was very limited and thus the conclusions are tentative.

The mass flow rates for lead, zinc, and manganese were observed to increase during the rain event. Lead and zinc, and perhaps manganese, are known automotive sources and it is therefore not a surprise that they were found to increase.

VI. CONCLUSIONS

Perhaps the most important conclusion from this study is that relatively few toxic pollutants were found in the sources and many were at low concentration levels. Only fifty-six priority pollutants were observed. Sixty-seven pollutants were never detected and an additional twenty were detected less than ten percent of the time.

Tap water contributed only trihalomethanes and copper.

Residential sources had high zinc and manganese levels, plus some other metals. Commercial sources were quite similar to residential sources, but did have some additional pollutants and a few more metals. The industrial sources had high concentrations of many of the detected organic pollutants and all of the observed metals were present in this source category.

The data have been used to develop indices for each source category which could be used to compare the impact of different proportions of source types on POTW influents. The indices appear reliable for the residential and commercial sources, but are only approximate indicators for the industrial sources, because of the extremely variable and specific nature of industrial source types.

The indices have been used to calculate relative source strengths and loadings on POTW's for a number of hypothetical drainage basins. These calculations clearly reflect the impact of industrialization of a basin, but also show the dominant role played by residential and commercial sources for some pollutants. Reasonable success was achieved in applying the source indices to the four cities studied, to make a mass balance comparison with the measured POTW influent values.

The frequency of observation of pollutants is consistently lower on weekends than on weekdays. Old residential sources contribute higher levels of pollutants than new residential sources.

VII. RECOMMENDATIONS

Further analyses should be carried out on the data base which has been developed to search for other effects and correlations.

The indices and approaches developed in this study should be used to examine available data for industrialized cities.

This source data should be integrated with the POTW plant data to enable a complete analysis of the POTW situation.

Further sampling efforts should be designed to test the findings summarized in this report. Any future studies should also attempt to develop a more complete understanding of industrial sources and their impact on POTW loading.

VIII. REFERENCES

- 1. "Sources of Toxic Pollutants Found in Influents to Sewage Treatment Plants" I. Literature Review, EPA, MDSD, Final Report on Task Order No. 6, Contract No. 68-01-3857, Report No. ADL 81099-50, June, 1979.
- 2. "Sources of Toxic Pollutants Found in Influents to Sewage Treatment Plants" II. Muddy Creek Drainage Basin, Cincinnati, Ohio, EPA, MDSD, Final Report on Task Order No. 6, Contract No. 68-01-3857, Report No. ADL 81099-51, June, 1979.
- 3. "Sources of Toxic Pollutants Found in Influents to Sewage Treatment Plants" III. Coldwater Creek Drainage Basin, St. Louis, MO., EPA, MDSD, Final Report on Task Order No. 10 Contract No. 68-01-3857, Report No. ADL 81099-16, October, 1979.
- 4. "Sources of Toxic Pollutants Found in Influents to Sewage Treatment Plants" IV. R.M. Clayton Drainage Basin, Atlanta, Georgia, EPA, MDSD, Final Report on Task Order No. 13, Contract No. 68-01-3857, Report No. ADL 81099-26, October, 1979.
- 5. "Sources of Toxic Pollutants Found in Influents to Sewage Treatment Plants" V. Hartford WPCP Drainage Basin, Hartford, Connecticut, EPA, MDSD, Final Report on Task Order No. 13, Contract No. 68-01-3857, Report No. ADL 81099-46, November, 1979.
- 6. "Sampling and Analysis Procedures for Screening of Industrial Effluents for Priority Pollutants", U.S. EPA, EMSL, Cincinnati, Ohio, March, 1977, revised April, 1977.
- 7. "Quality Assurance Program for the Analyses of Chemical Constituents in Environmental Samples", U.S. Environmental Protection Agency, Environmental Monitoring and Support Laboratory, Cincinnati, Ohio, March, 1978.

APPENDIX A

Individual Pollutant Reporting Limits, Recovery and Precision Data

The data in these tables summarize the results that were obtained for each pollutant reviewed over all four cities, for which reference compounds were available. These data are the results obtained in the raw wastewater samples. The reporting limits were the same for each city where a single value is indicated, otherwise the range reported over the four cities is given. The average recovery values reported are the means over all four cities. The precision (relative standard deviation) is given as the range of values observed in all of the cities. In general, the precision of the data improved throughout the program; the recovery values stayed consistently high for all four cities and low reporting limits were consistently achieved.

The quality control data are very good for the majority of the priority pollutants. The priority pollutants for which the EPA Screening Protocol was problematic are listed below along with their respective problems. These priority pollutants are indicated by footnote, in the tables. The footnote definitions are:

(a) These priority pollutants were never detected using the EPA Screening Protocol. Therefore, if these compounds were present in the samples from the four surveys they would not have been detected.

Bis(chloromethyl)ether - very short half life in water.

2-Chloroethyl vinyl ether - volatile (bp 109°C) causing erratic recoveries during Kuderna Danish concentration.

Hexachlorocyclopentadiene - Possible high GC/MS reporting limit or degradation in the heated GC injector.

(b) These priority pollutants were sporadically not detected using the EPA Screening Protocol. Consequently, accuracy and precision data are poor. The analytical results for these compounds in the four surveys may not be reliable. The problem for most of these

compounds is related to poor chromatography. The compounds for which this was particularly problematic are:

- Benzidine poor chromatography, heat labile, unstable in methylene chloride (problematic in Cincinnati, St. Louis, Atlanta and Hartford).
- N-nitrosodimethylamine poor chromatography, high GC/MS reporting limit, poor extraction efficiency from water into methylene chloride (problematic in Cincinnati, St. Louis, Atlanta).
- 2,4-Dinitrophenol poor chromatography.
- 4,6-Dinitro-2-cresol poor chromatography (problematic in Atlanta, Hartford).
- 4-Nitrophenol poor chromatography (problematic in Cincinnati and Hartford).
- (c) These volatile priority pollutants were not detected until the PAT/GC/MS procedure was modified during the Atlanta study. Therefore if these compounds were present in samples from Cincinnati or St. Louis they would not have been detected. Also during the Atlanta study precision and accuracy were poor. Therefore the quantitative analytical results are not reliable.

Dichlorodifluoromethane - broke through sorbent trap (also occurred in Atlanta study).

Bromomethane - broke through sorbent trap.

Vinyl chloride - broke through sorbent trap.

Chloroethane - broke through sorbent trap.

The analytical method for those compounds just listed was improved for the last city and reliable data are available from the Hartford samples for these pollutants.

- (d) The analytical results for methylene chloride are erratic due to sporadic contamination from the field and laboratory. This problem was finally brought under control during the Hartford study.
- (e) Reference standards were never available for these priority pollutants. It may be implied from QC data or similar compounds that these priority pollutants would have been defected if they were present in the samples.

2,3,7,8-TCDD

Bis(2-chloroisopropyl)ether.

The reference standards that were not available for all four cities surveyed but available for some, are so indicated in the following Tables.

Table A-1

SUMMARY OF QUALITY CONTROL DATA

Volatiles

		Reporting Limit	Average	Range of Relative Standard
	Compound	μg/L	Recovery	Deviations
101.	Chloromethane C*	5*	118*	21*
102.	Dichlorodifluoromethane C	5*	194*	48*
103.	Bromomethane C	5*	113*	22*
104.	Vinyl chloride C	5*	123*	19*
105.	Chloroethane C	5*	108*	15*
106.	Methylene chloride d	1	138	12-259
107.	Acrolein	1-7	58	35-149
108.	Trichlorofluoromethane	1-6	78	11-83
109.	Acrylonitrile	1	94	8-24
110.	1,1—Dichloroethylene	1-5	72	4-134
111.	1,1—Dichloroethane	1-2	83	2-43
112.	Trans-1,2-dichloroethylene	1	73	3-73
113.	Chloroform	1	82	5-38
114.	1,2—Dichloroethane	1	96	4-35
115.	1,1,1—Trichloroethane	1	82	10-78
116.	Carbon tetrachloride	1	87	6-44
117.	Bromodichloromethane	1	88	5-24
118.	1,2-Dichloropropane	1	82	3-37
119.	Trans-1,3-dichloropropylene	1	82	4-11
120.	Trichloroethylene	1-2	98	2-78
121.	Benzene	1	89	5-17
122.	Cis-1,3-dichloropropylene	1	85	3-20
123.	Dibromochloromethane	1-2	95	4-24
124.	1,1,2-Trichloroethane	1	99	2-13
125.	Bromoform	1-3	74	7–37
126.	1,1,2,2—Tetrachloroethane	1	78	4–75
127.	1,1,2,2—Tetrachloroethylene	1	94	8–79
128.	Toluene	1	102	3-32
129.	Chlorobenzene	.1	106	2–29
130.	Ethyl benzene	1	113	2-47

^{*} Hartford data only

Table A-2

SUMMARY OF QUALITY CONTROL DATA

Acids

	Compound	Reporting Limit µg/L	Average Recovery	Range of Relative Standard Deviations
201.	2-Chlorophenol	10	86	6-29
202.	2-Nitrophenol	10-15	93	8-26
203.	Phenol	10	60	20-26
204.	2,4-Dimethylphenol	10	90	5-20
205.	2,4—Dichlorophenol	10	103	9-15
206.	2,4,6-Trichlorophenol	10	92	7–18
207.	4-Chloro-3-cresol	10	98	12-17
208.	2,4-Dinitrophenol b	20-40	41	26-155
209.	4,6-Dinitro-2-cresol b	20-40	57	23-112
210.	Pentachlorophenol	10-25	105	11-28
211.	4-Nitrophenol b	10-25	54	11-42

Table A-3
SUMMARY OF QUALITY CONTROL DATA

Base/Neutrals

Compound	Reporting Limit µg/L	Average Recovery	Range of Relative Standard Deviations
301. 1,3 Dichlorobenzene			
302. 1,4 Dichlorobenzene	10-30	71	11-30
303. 1,2 Dichlorobenzene			
304. Hexachloroethane	10-20	70	26-42
305. Bis(chloromethyl)ether a	-	-	-
306. Bis(2-chloroethyl) ether	10-20	78	10-39
307. Bis(2-chloroisopropyl) ether ⁶	10	_	-
308. N—Nitrosodimethylamine ^C	70 *	37*	78*
309. Nitrosodi-n-propylamine	10-20	89	9-27
310. Nitrobenzene	10-20	78	10-28
311. Hexachlorobutadiene	10	57	13-23
312. 1,2,4—Trichlorobenzene	10-20	74	16-20
313. 2-Chloroethyl vinyl ether a	-	-	_
314. Bis(2—chloroethoxy) methane	10	92	10-44
315. Naphthalene	10	81	14-43
316. Isophorone	10	82	8-35
317. Hexachlorocyclopentadiene	-	-	_
318. 2—Chloronaphthalene	10	81	15-27
319. Acenaphthylene	10	85	12-21
320. Acenaphthene	10	82	17-24
321. Dimethyl phthalate	10	67	9-40
322. 2,6—Dinitrotoluene	10	86	20-25
· 323. 4—Chlorophenyl phenyl ether	10	74	18-29
324. Fluorene	10	79	16-24
325. 2,4—Dinitrotoluene	10	62	19-43
326. Diethyl phthalate	10	91	19-34
327. 1,2—Diphenylhydrazine	10	75	23-28
328. N—Nitrosodiphenylamine	10	113	13-22
329. Hexachlorobenzene	10	64	22-43
330. 4—Bromophenyl phenyl ether	10	64	12-29

^{*} Hartford data only

Table A-3 (Continued)

SUMMARY OF QUALITY CONTROL DATA

Base/Neutrals

	Compound	Reporting Limit µg/L	Average Recovery	Range of Relative Standard Deviations
331.	Anthracene	5-10	82	14-23
332.	Phenanthrene	3-10		14-23
333.	Di-n-butyl phthalate	10	74	28-81
334.	Fluoranthene	5-10	66	11-35
335.	Pyrene	5-10	67	14-35
336.	Benzidine ^b	10-20	18	95-111
337.	Butyl benzyl phthalate	10	45	33–57
338.	Bis(2—ethylhexyl) phthalate	10	42	29-84
339.	Di-n-octyl phthalate	10		
340.	Chrysene	5-10	59	13-30
341.	Benzo(a)anthracene			
342.	3,3'-Dichlorobenzidine	10	80	15-27
343.	Benzo(b)fluoranthene	1-5	46	16-93
344.	Benzo(k)fluoranthene	1-3	40	10 75
345.	Benzo(a)pyrene	5-10	51	18-34
346.	Indeno (1,2,3-c,d) pyrene	5 *	29*	19*
347.	Dibenzo (a,h) Anthracene	5-10	50	17-60
348.	Benzo (g,h,i) perylene	5-10	40	20-245

^{*} Hartford data only

Table A-4

SUMMARY OF QUALITY CONTROL DATA

Pesticides

	Compound	Reporting Limit µg/L	Average Recovery	Range of Relative Standard Deviations
401. a	pha-BHC	1	77	8-28
402. ga	amma-BHC	1	78	7-43
403. H	eptachlor	1	67	7-70
40 4. b	eta-BHC	1	80	5-42
405. d	elta-BHC	1	89	7-31
406. A	Mdrin	1	76	5-20
407. H	leptachlor epoxide	1	80	5-18
408.	ndosulfan I.	1	64	11-51
409. D	DDE	1	84	5-26
410. D	Pieldrin	1	48	6-39
411. E	ndrin	1	77	9-26
412. D	DDD	1	78	10-31
413. E	ndosulfan II		76	10-31
414. D	DŢ	1	76	8-21
415. E	ndrin aldehyde	1*	60*	29 *
416. E	ndosulfan sulfate	1*	84*	18 *
417 C	Chlordane			
418. T	oxaphene			
419. P	CB-1221			
420. P	CB-1232			
421. P	CB-1242			
422. P	CB-1248			
423. P	CB-1254	1	86	7-18
424. P	CB-1260			
425. P	CB-1016			

^{*} Hartford data only

Table A-5
SUMMARY OF QUALITY CONTROL DATA

Metals, Total Cyanides, Total Phenols

Compound	Reporting Limit µg/L	Average Recovery	Range of Relative Standard Deviations
501. Antimony	1-3	73	25-48
502. Arsenic	2-4	101	11-36
503. Beryllium	1-3	69	6-13
504. Cadmium	1-3	85	15-63
505. Chromium	1-67	99	2-48
506. Copper	4–9	103	9-12
507. Lead	3–15	90	10-47
508. Manganese	3-11	100	4-9
509. Mercury	1-2	73	7-34
510. Nickel	1-30	105	3-60
511. Selenium	1-5	87	13-47
512. Silver	1-3	103	6-30
513. Thallium	1	96	5-14
514. Zinc	6-50	104	5-45
601. Total Cyanides	10-20	91	10-17
602. Total Phenois	10-20	96	6-16

Table A-6

QUALITY ASSURANCE DATA

Classical Parameters (7XX Series) Analysis*

	Compound	Spike Concentration mg/L	Mean % Recovery	Relative Standard Deviation, %
703.	Ammonia	4.1	94	2
704.	Oil and Grease	230	79	30
705.	TSS	70	42	75
706.	TOC	75	102	2
707.	COD	190	75	16
708.	BOD	105	117	17

^{*} Data from Atlanta study - method reference standards only.

APPENDIX B

Total Number of Pollutant Observations in Sources - by City

The tables in this Appendix report the number of times a pollutant was observed in each city, organized by source category. Data for all of the pollutants ever detected are included, except for methylene chloride, which was excluded because of its probable presence due to contamination. A blank indicates that it was not detected in that city, or at all.

Table B-1

TOTAL NUMBER OF OBSERVATIONS IN TAP WATER SAMPLES

					······································				
	=	İ	2				-		,
	Cincinnuti	- 1	Louts		5		Hartford		:
	2				At, Jauta		=		1
Ī	5		St.		-		로		Total
	4		2		<u> </u>		1 4		. 12
Number of Samples					┯┪		1		1
104. Vinyl Chloride								-+	
105. Chloroethene					-	_	 		
108. Trichlorofluoromethane							ļi		
109. Acrylonitrile					L		.		
110. 1,1-Dichloroethylene					<u> </u>		 	├	
111. 1,1 Dichloroethane							↓		
112. Trans-1,2-dichloroethylene					<u> </u>	L	<u> </u>		
113. Chloroform	4		2	L	2		14	├ ──┼	12
114. 1,2-Dichlorgethene			l		<u> </u>	<u> </u>		L	
115. 1,1.1-Trichloroethene					<u> </u>			├	+
116. Carbon tetrachloride			I		<u> </u>	<u> </u>	 _		
117. Bromodichloromethane	4		2		2	1	4	├	12
118. 1,2-Dichloropropane						!	 	├	
119. Trans-1,3-Dichloropropylene				<u> </u>		1	 	1	
120. Trichloroethylene								↓ 	
121. Benzene					↓	↓		1	
123. Dibromochioromethane	4		2	1	1		-	↓↓	7
124. 1,1,2-Trichloroethane		L		1	1_	1_		↓	
125. Bromoform	3		1			-		↓ — ↓	
126. 1,1,2,2-Tetrachioroethane				1		\downarrow _	-	1	
127. 1,1,2,2-Tetrachloroethylene			2	<u></u>	1			1	3
128. Toluene			1					1	1
129. Chlorobenzene					1			4	
130. Ethylbenzene	1	Ι		L		1_		1	1
201. 2-Chlorophenol		Ι.				Ь.		\bot	
203. Phenoi				<u> </u>				-	
204. 2,4-Dimethylphenol									
206. 2,4-Dichlorophenol		1							
206. 2,4.6-Trichlorophenol		T		L.				↓	
207. p-Chloro-m-cresol				L.,					
210. Pentachlorophenol			1						
301. Dichlorobenzenes									
310. Nitrobenzene									
312. 1,2,4-Trichlorobenzene	Γ							<u> </u>	I
315. Naphthelene						_			L
326. Diethylphthelete	1	1		┷		-	→		 '-
331. Anthracene/Phonenthrene	<u></u>	1	┷	-	┷-	4-			$+-+\frac{1}{3}$
333. Di-n-butylphthalate	2			┷			<u> </u>		
334. Fluoranthene		1	1	-		-			} }
336. Pyrene				_		-	-		
337. Butylbenzylphthelate				1_				4-	
338. Bis (2-ethylhexyl)/di-n-actyl	2					1		1	2
phthelate	+-	+	+-	+-	+-	+	\neg	1	
404. Heptachlor	+	+		+		+	_	1	
408. Aidrin	+	+-	1 2	+-	+	+-	+-	+	1 2
801. Antimony	+3	+	+-	+		+		 	$+-+\frac{1}{3}$
802. Arsenic	11	+-		+-	-+-	+	+-	+-	1 1
804. Cadmium	1	+-	+	-+-		-+-		-	1 1
S06. Chromium	+		+-			+	4	+-	11
606. Copper	+ 3	+	2 2	+	- 1 2		-+ -	+	4
867. Leed	1 2	+	+-	+	+ 2		1 2	+	1 6
908. Manganese	+-	+-		+	+-	+	╌	+	+-+-
808. Mercury	+-	+	+	+	2	. 	-+-	+	3
\$10. Nickel	1 2	+-	+-	-+-	+-				4
\$11. Salenium			+	+		+	\dashv		+++
512. Silver	1	+	+	+-	-+-	+			 -
\$12. Thellium	1	+	+	+		-	- 3	+-	1 7
S14. Zinc	+*		╌┼╧	-+-	-+ '	+	┯		+ - 1 ′
601. Total Cyanidas	12	-+-	+	+	_	+		_	1 2
602. Total Phenois									

Table B-2 TOTAL NUMBER OF OBSERVATIONS IN RESIDENTIAL SAMPLES

	Cincinnatí		St. Louis		Atlanta		Hartford		Total
Number of Samp	les 11		18		6		12		47
104. Vinyl Chloride									
105. Chloroethane									
108. Trichlorofluoro	methane								
109. Acrylonitrile		Γ							
110. 1,1-Dichloroeth	ylene								
111. 1,1-Dichloroeth	ane								
112. Trans-1,2-dichlo									
113. Chloroform	102	Ì	17		6		-9		42
114. 1,2-Dichloroeth		-	1						1
115. 1,1,1-Trichloroe			8		2		3		14
116. Carbon tetrachi		└ ─							
117. Bromodichloror		├ ─	┵┵						
118. 1.2-Dichloropro		├	}	——	-				
119. Trans-1,3-Dichle		 	-3		2				5
120. Trichtoroethyle		├	1 3		-				10
121. Benzene	methane 1ª	+	1		 				10
123. Dibromochloror		+	+		 		 		
125. Bromoform		1	 		 		\vdash		
125. Bromotorm 126. 1,1,2,2-Tetrachi	loroethane 1ª	+-	\vdash		 			-+	1
127. 1,1,2,2-Tetrach			18		6		7		36
128. Toluene	58	J	18		4		2	-t	29
129. Chlorobenzene	- 1 <u>1</u> a	1	1 2		 		-		3
130. Ethylbenzene		↓	1 3		 1				8
201. 2-Chlorophenol		+	+-	-	 				
203. Phenoi	7	+	10	_	1	-			18
204. 2,4-Dimethylph		+	 1		1	$\vdash \vdash$			3
205. 2,4-Dichloroph		+-	+	-	 	 	1		
206. 2,4,6-Trichloro		+-	1	-	1	-			
207. p-Chloro-m-cre		+	 		 	 			
210. Pentachlorophe		+-	†		2				2
301. Dichlorobenzer		 	6	1	1				6
310. Nitrobenzene		—	 	1	1		1		1
312. 1,2,4-Trichlord	benzene		1				1		1
315. Naphthalene	1	1	2				1		4
326. Diethylphthala	te 10		11		J		2		23
331. Anthracene/Ph	enanthrene		1		L	<u> </u>	1	L	2
333. Di-n-butylphth	alate 5	Ι_	8		1		2		16
334. Fluorenthene				<u> </u>	<u> </u>		1	 	
335. Pyrene			1	L	1	 	 	├ ──┼	1
337. Butylbenzylph		1	17		↓	 	2	╁╼╾┼	22
338. Bis (2-ethylhes	tyi)/di-n-octyl 7		4	}	1	1	1		11
phthalate		+-	1	1	1	T	1	1-1	
404. Heptachior 406. Aldrin		+-	2	1	1	1		1-1	2
		+-	12	1	2	 	1-	1 1	16
501. Antimony 502. Arsenic	11	+	†	 	+	1	50		16
504. Cadmium	1	+	4	1	1		10	1. 1	;
506. Chromium	3	1	18	1	6	I	20		29
508. Copper	111	1-	18	1	6	I	110		40
507. Lead	6	\top	18		6		80		36
508. Manganese	11	\top	18		6		11 ^b		40
509. Mercury	1		6				15		
510. Nickel	2		18		6		2	1	2/
511. Selenium	6	I	18				20	1	20
512. Silver	3		5		2	1		\bot	10
513. Thellium					1	1	-	╃╼╾╃	
514. Zinc	11		18	1	6	4-	110	1	41
801. Total Cyanide			1	\downarrow	1	4	4	↓ ↓	
602. Total Phenois	10		18		6	1	9	1_1	4:

a. 10 samples b. 11 samples

Table B-3 TOTAL NUMBER OF OBSERVATIONS IN COMMERCIAL SAMPLES

		Cincinnati		St. Louis		Atlantu		Hartford			Total
Num	ber of Samples	12		12		9		9			42
104.	Vinyl Chloride										
106.	Chloroethene		 		-						_
108.			-1								
	Trichlorofluoromethane		-					-			
100.	Acrylonitrile										
110.	1,1-Dichloroethylene		\longrightarrow	2							2
111.	1,1-Dichloroethane			1							1
112.	Trans-1,2-dichloroethylens	2		6		9		1			18
113.	Chloroform	12	1	12		9		9			42
114.	1,2-Dichloroethane	1						1			2
115.	1,1,1-Trichloroethane	7		10		2		3			22
116.	Carbon tetrachloride					2					2
117.	Bromodichloromethane	9		12							21
118.	1,2-Dichloropropene										I
119.	Trans-1,3-Dichloropropylene			1							1
120.	Trichloroethylene					5		1			6
121.	Benzene	8		11		1		1			21
123.	Dibromachloromethane	6		12							18
124.	1,1,2-Trichloroethene										
126.	Bromoform		-				_	-			
126.	1,1,2,2-Tetrachioroethene						·				
127.	1,1,2,2-Tetrachlorosthylene	12	 	12		9	 	8			41
128.	Toluene	12		12		8	 	6		 	 -
129.	Chlorobenzene	12	 			•	-	-	<u> </u>		38
130.		_		2			 				2
	Ethylbenzene	8	-	. 7		3_		1			21
201.	2-Chlorophenol						 -			-	-
203.	Phenol	6	1	5		3		3		 	17
204.	2,4-Dimethylphenol										L
206.	2,4-Dichlorophenol							L			L
204.	2,4,6-Trichlorophenel										
207.	p-Chilaro-m-cresal										
210.	Pentachlorophenol		<u> </u>			6					6
301.	Dichlorobenzenes	1		9		1		2		_	13
310.	Nitrobenzene										
312.	1.2,4-Trichlorobenzene							1			1
315.	Naphthalene	4		2				1			7
326.	Diethylphthelete	9		6							15
331.	Anthracene/Phenenthrane	1									1
333.	Di-n-butylphthelate	9		4		2		3			18
334.	Fluoranthene						 				1
336.	Pyrana				_		1				1
337.	Butylbenzylphthelate	6	1	11		2		4	├ -		23
336.	Bis (2-ethylhenyl)/di-n-octyl	<u> </u>	 								-
	phthelate	11		_3		1		1		1	16
404.	Heptachior	2									2
406.	Aldrin										1
501.	Antimony	1				2	 		 		3
802.	Arsenic	12	\vdash		 	 	_	4	 		16
804.	Cadmium	14	 		 	-	 		 	 	+
305.	Chremium	┝┷	 	1,2	 	-	 	1	 	├	1 3
986.	Copper	12	 	11ª 12	 	9	-	4	 	├	124
507.	Lood	7	 		 	9	├	9	 	 	42
908.	Mangatese	12	 	12	 	9	 	9		 	35
F00.	Mercury			12	 	 ' -	├-			├ ─	42
510.		6	 	12		-		1	-	 	14
510.	Michel			12		9	↓	6	_		33
512.	Sutenium Cit.	6	 	7		2	 	1	 	↓	16
	Silver	7		-	-	6	_	36	 	L	20
813.	Thellium	1.3				<u> </u>	!	ļ			1
814.	Zinc	12	_	12	L	9		9	<u> </u>		42
601.	Total Cyanidas	<u> </u>		1	L	L					1
602	Total Phonois	12		11		9					

a. 11 samples analyzed b. 6 samples analyzed

Table B-4

TOTAL NUMBER OF OBSERVATIONS IN INDUSTRIAL SAMPLES

		Louis			ıta				
		St. L			Atlanta				Total
V	r of Samples	12							Total 21
	r of Samples				9		1	τ	1
	yl Chloride proethane				$\frac{1}{1}$	 -	 	┼~~	$\frac{1}{1}$
	hiorofluoromethane	-			+-+-		 	1	1
	ylonitrile				1				
110. 1,1	Dichloroethylene				8				8
111. 1,1	Dichloroethane				7		<u> </u>	↓	7
	ns-1,2-dichloroethylene	1			1-7-	L	ļ	↓	1 8
	oroform	12			9	├	 	┥—	21
	Dichloroethane	6			3		┼	╅	15
	1-Trichloroethane bon tetrachloride	2			5		 	┪	+ -,
	modichloromethane	12					1	+-	12
	Dichloropropene				1		1	1	1
119. Tra	ns-1,3-Dichloropropylene								
120. Tri	chloroethylene	12			9				21
121. Bei	zene	9	L	L_	1.	<u> </u>	4	↓	16
	romochloromethane	12				}		-	12
	,2-Trichloroethane		 		1		┿		 '
	moform			├—	1 2	+-	+	+	
	,2,2-Tetrachioroethane ,2,2-Tetrachioroethylene	12		 	9	┼─	+	+	21
	uene	12	├	┼	9	1	+	+	2
	orobenzene		├─	1-	3	 	+	\top	1
	rylbenzene	7	 	†	9	1			10
	Chlorophenol	1			1				
203. Ph	mol	3			8				111
204. 2,4	-Dimethylphenol				8		┵	-	
206. 2,4	-Dichlorophenol	1		↓	1	↓_	┵	┥	
	,6-Trichlorophenol	1	ļ	↓	1	↓_		-	
	Chloro-m-cresol			╂	4	+	+	+	+,
	ntachlorophenol chlorobenzenes	8	┼	+	4	╅	╅┈	+-	12
	trobenzene	-	 	╁		+	+-	_	+
	2,4-Trichlorobenzene		1						T
	phthalene	7			6				1.
326. Di	ethylphthalate			L.			┸		
331. A	thracene/Phonanthrene	L	 	↓_	1 3	 			1
	n-buty/phthelate	8	-	—	4	┼			
	uoranthene	├	╁	╅╾-	-	 -	-	+	-+-
	rene itylbenzylphthalate	5	+	╀	6	+		+	1
	(2-ethylhexyl)/di-n-octyl	 	+-	+		+	1	-	
330. 0	phthelate	L	↓	—	5	┼—	-	-	_
404. H	ptachior	_	↓	╄		╃~	┰	+-	-
	drin	 	┼—	╁	NA.	+-		┵	-+,
	ntimony	5	┼	+-	4	+	+-	+-	+
	rsenic	1 4	+	┿	14	+-	-	_	-
	dmium hromium	12	+	+-	9	+-	_		2
	POPPER	12	+-	\top	9	1	I	\perp	2
	and .	12		$oldsymbol{oldsymbol{oldsymbol{oldsymbol{\square}}}$	9			工	2
	unganete	12			9	1	_	-	- 2
	ercury		1	4-		4-	-		-
510. N	ickel	12	4	4		+-			- 2
	dentum	3	—	+	 9	+-			-+-
	liver	1 9	+-	+-		+-	+	+	11/
	hellium	12	+	+-	9	+-	\top	+-	12
	inc	1 1/7	╁┈	1	7	_	1		
	otel Cyanides otel Phenois	12	+-	+-	1 5	\top	_	_	2

^{*} Out of 12 samples.

Table B-5

TOTAL NUMBER OF OBSERVATIONS IN INFLUENT SAMPLES

	Cincinnati		St. Louis		A.lante		Partford			lotal
Number of Samples	6		6		3	i	3			18
104. Vinyl Chloride										
105. Chloroethane										
108. Trichlorofluoromethane					1					1
109. Acrylonitrile					1					1
110. 1,1-Dichloroethylene			1	$\neg \neg$	2					3
111. 1,1-Dichloroethane			1							1
112. Trans-1,2-dichloroethylene			2		3					5
113. Chloroform	6		6		3		3			18
114. 1,2-Dichloroethane	1				1					2
115. 1,1,1-Trichloroethene	2		6		3		3			14
116. Carbon tetrachloride										
117. Bromodichloromethene			2							2
118. 1,2-Dichloropropene										
119. Trans-1,3-Dichleropropylene										
120. Trichloroethylene			6		3		3			12
121. Benzene	6		6							12
123. Dibromachloromethene			4							4
124. 1,1,2-Trichloroethene	· · · · ·									
126. Bromoform										
126. 1,1,2,2-Tetrachioroethane	1									1
127. 1,1,2,2-Tetrachigroethylene	3		6		3		3			15
128. Toluene	2		6		3		3			14
129. Chlorobenzene			1							1
130. Ethylbenzene	3		6		3					12
201. 2-Chlorophenol										
203. Phenol			4		2					6
204. 2,4-Dimethylphenol		1			2					2
205. 2,4-Dichlorophenol		1			1					
208. 2,4,6-Trichtorophenol		1								
207. p-Chloro-m-cresol		1					1			1
210. Pentachlorophenol	T				3					4
301. Dichlorobenzenes			5		2		3			10
310. Nitrobenzene										
312. 1,2,4-Trichlorobenzene		1								T
316. Naphthelene	2		4		2					8
326. Diethylphthelete	4		3		1		1			9
331. Anthracene/Phenenthrene		1								
333. Di-n-butylphthelete	4	1	6		1		1			12
334. Fluoranthene		1					I			I
336. Pyrene								$I_{}$		1
337. Butylbenzylphthelete	1		5	L	3					8
336. Bis (2-ethylhexyl)/di-n-octyl	٦.		2							4
phthelate	2	╂	 	+-	+	├	+	+	+-	+
404. Heptschier	1-	┼	₩	↓	+	├	┼─	+	+	+-
406. Aldrin	↓	4—	1-	↓ —	+1	├ ─	+-	+	+	+ 7
901. Antimony	1	—	6	 	+ -	↓_	+	+	+	8
802. Arsenic	6		+	↓_	+-	₩-	2	+	 	10
804. Codmium	14	-	6	╀	3	+-	3	╂	+-	16
505. Chromium	4	 		+	3	┼	13	╂	+	
808. Copper	6	+	6	+-		┼		+-	+	18
907. Lond	4	-	6	╁	3	+-	3	+-	+	16
808. Mangamese	6	+	6	╁—		+-	13	+	┼	18
808. Mercury	11	+	1	+-	1	+-	+-	+	+	13
510. Michai	13		 6	₩	13	+	╀┹-	+	+-	15
611. Sulenium	1 3	+-	12	+	+-	┼	+-	+	+	1.7
512. Silver	6	-	6	-	3	+-	2	+	+	17
513. Thefficen	+-	+	+	+-	+-	+	+	+	+	
\$14. Zinc	- 6	+	16.		++	+	++	+	+	18
601. Total Cyanides	1.	-	15	+-	 	┼	++	+-	+-	11
602. Total Phonois	6		6	1	13		11			118