

CADMIUM

FACT SHEET ON A DRINKING WATER CHEMICAL CONTAMINANT

GENERAL INFORMATION

Synonyms

None

Chemical Description:

 Naturally occurring element (transition metal) found in most rock, coal, and lead or zinc ores as various salts

Properties:

- Elemental form is insoluble in water
- Solubility of cadmium salts is compound specific

Production and Use:

- Commercially obtained as a byproduct during the processing of zinc-bearing ores
- Used in the manufacturing of batteries, stabilizers, pigments, metal alloys, and electroplating for corrosion resistance

ENVIRONMENTAL PROFILE

Occurrence:

- Naturally occurring in water supplies from erosion of mineral deposits at levels in the range of 0 001-0 01 mg/L
- Naturally occurring in soil, and as consequence, in food
- Occurs as a byproduct of corrosion of some galvanized plumbing and distribution system materials

Releases:

- Enters the environment primarily through industrial effluents and landfill leaching
- Can enter surface waters from natural sources and from a variety of manufacturing operations that involve cadmium or zinc that contains a cadmium impurity

Environmental Fate:

- Highly persistent in water
- High potential for bioaccumulation, especially fish and other aquatic organisms
- Persistent in many tissues such as muscles, kidneys, and liver

HEALTH EFFECTS

Humans:

- Symptoms of acute oral toxicity include nausea, vomiting, diarrhea, muscular cramps, salivation, sensory disturbances, liver injury, convulsions, shock, and/or renal failure and cardiopulmonary depression
- Long-term citil exposure to high levels of codminiminary result in damage to the kidneys, liver, bones, and blood

Experimental Animals:

- Short-term, high-dose oral exposures to various cadmium compounds resulted in damage to the nervous system, kidney, liver, bone, blood formation system, cardiovascular system
- Long-term, high-dose exposures by drinking water had adverse effects upon functions of the liver and kidneys
- High oral doses of cadmium had detrimental reproductive and teratogenic effects on rats
- Equivocal evidence of mutagenic potential
- Carcinogenic via inhalation, but limited evidence of carcinogenicity following chronic oral exposure

REGULATORY HISTORY

Existing Standards:

- ·Clean Air Act (CAA): Not regulated
- •Clean Water Act (CWA):

Criteria established

•Resource Conservation and Recovery Act (RCRA):

Not regulated

- · Superfund (CERCLA):
 - · Hazardous waste
 - •SARA: Toxic substance
- •Federal Insecticide, Fungicide, and Rodenticide Act (FIFRA):

Registered (Cadmium compounds)

•Toxic Substances Control Act (TSCA):
Not regulated

HEALTH INFORMATION

Maximum Contaminant Level Goals (MCLG):

- Non-enforceable levels based solely on an evaluation of possible health risks and exposure, and taking into consideration a margin for public safety
- Set at 0 005 mg/L to protect against kidney and liver damage

MCLG for Cadmium = 0.005 mg/L (effective July 1992)

Maximum Contaminant Levels (MCL):

- Legally enforceable levels for contaminants in public drinking water supplies
- Based on health risks associated with the contaminants, analytical methods for their assay, and water treatment feasibility and practicality aspects
- Exceedance of the MCL in drinking water may result in adverse effects which will depend upon the contaminant concentration in water, amount of water/contaminant ingested, length of exposure, and other biological parameters
- Current MCL = 0.01 mg/L

MCL for Cadmium = 0.005 mg/L (effective July 1992)

EPA Health Advisories (HA):

- Short-term HAs: Provide acceptable concentrations of contaminants in water for up to 10 day exposures, primarily to evaluate the public health risk resulting from an accidental spill or an emergency contamination situation
- Longer-term HAs: Provide guidance for persistent water contamination situations to cover a period of up to 7 years
- Lifetime HAs: Derived in the same way as an MCLG

Health Advisories:

Short-term HA for a child = 0.04 mg/L Longer-term HA for a child = 0.005 mg/L Longer-term HA for an adult = 0.02 mg/L Lifetime HA = 0.005 mg/L

ANALYTICAL METHODS

- Graphite Furnace Atomic Absorbtion EPA Method 213 2
- Inductively Coupled Plasma EPA Method 200 7A

WATER TREATMENT

Permanent Treatment:

Best Available Technology (BAT):

- Coagulation/Filtration
- Lime Softening
- Reverse Osmosis
- Ion Exchange

SHORT-TERM HAZARD ELIMINATION

 If the drinking water standards are exceeded, install BAT or use an alternative drinking water supply such as bottled water

ADDITIONAL HELP

- State or county health officials can indicate a certified laboratory for testing
- Experts in the state Department of Environmental Protection or Natural Resources may also be of help
- The EPA has toll-free numbers for further information on drinking water quality, treatment technologies, for obtaining Health Advisories, and for other regulatory information
- EPA Hotlines are available Monday through Friday

Safe Drinking Water: 800-426-4791
National Pesticides: 800-858-7378
Superfund/RCRA: 800-424-9346

- For information on the Clean Water Act, call (202) 260-7301
- For information on the Toxic Substances Control Act, call (202) 554-1404
- For information on the Clean Air Act, call (919) 541-2777