# DEVELOPMENT DOCUMENT for

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### PROPOSED EFFLUENT LIMITATIONS GUIDELINES AND NEW SOURCE PERFORMANCE STANDARDS

### FOR THE

### ORGANIC CHEMICALS AND PLASTICS AND SYNTHETIC FIBERS INDUSTRY

VOLUME I

#### (BPT)

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# NOTICE MAR 3 1 1983

On February 28, 1983, EPA proposed effluent limitations guidelines and standards for the organic chemicals and plastics and synthetic fibers (OCPSF) point source category. The <u>Federal Register</u> notice of this proposal was printed on March 21, 1983 (48 FR 11828 to 11867).

Information received by the Agency after proposal indicates that the total OCPSF industry estimated annual discharges of toxic pollutants are too high. The Agency will be reevaluating these estimates when additional information becomes available prior to promulgation of a final regulation. In the interim, the Agency advises that there should be no reliance on the annual total toxic pollutant discharge estimates presented in the <u>Federal Register</u> notice, the February 1983 OCPSF Development Nocument, and February 10, 1983 OCPSF Regulatory Impact Analysis.

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#### SECTION I

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#### INTRODUCTION

#### PURPOSE AND LEGAL AUTHORITY

The Federal Water Pollution Control Act Amendments of 1972 established a comprehensive program to restore and maintain the chemical, physical and biological integrity of the nation's waters [Section 101(a) ]. by July 1, 1977 existing industrial direct dischargers were required to achieve effluent limitations requiring the application of the best practicable control technology currently available (BPT) [ Section 301(b)(1)(A) ]. Bv July 1, 1983 these dischargers were required to achieve effluent limitations requiring the application of the best available technology economically achievable (BAT), which will result in reasonable further progress toward the national goal of eliminating the discharge of all pollutants [ Section 301(b)(2)(A) ]. New industrial direct dischargers were required to comply with Section 306 new source performance standards (NSPS) based on best available demonstrated technology. New and existing dischargers to publicly owned treatment works (POTW) were subject to pretreatment standards under Sections 307(b) and (c) of the Act. The requirements for direct dischargers were to be incorporated into National Pollutant Discharge Elimination System (NPDES) permits issued under Section 402 of the Act. Pretreatment standards were made enforceable directly against dischargers to POTWs (indirect dischargers).

Although Section 402(a)(1) of the 1972 Act authorized local authorities to set requirements for direct dischargers on a case-by-case basis, Congress intended that for the most part control requirements would be based on regulations promulgated by the EPA Administrator. Section 304(b) of the Act required the Administrator to promulgate regulatory guidelines for direct discharger effluent limitations, setting forth the degree of effluent reduction attainable through the application of best practicable control technology (BPT). Moreover, Sections 304(c) and 306 of the Act required promulgation of regulations for NSPS, and Sections 304(f), 307(b) and 307(c) required promulgation of regulations for pretreatment standards. In addition to these regulations for designated industry categories, Section 307 (a) of the Act required the Administrator to promulgate effluent standards applicable to all dischargers of toxic pollutants. Finally, Section 501(a) of the Act authorized the Aministrator to prescribe any additional regulations necessary to carry out his or her functions under the Act.

The EPA was unable to promulgate many of these regulations by the dates contained in the Act. In 1976 EPA was sued by several environmental groups. In settlement of this lawsuit, EPA and the plaintiffs executed a settlement agreement which was approved by the Court. This agreement required EPA to develop a program and adhere to a schedule for promulgating, for 21 major industries, BAT effluent limitations guidelines, pretreatment standards and new source performance standards for 65 "toxic" pollutants and classes of pollutants.

On December 27, 1977 the President signed into law the Clean Water Act of 1977. Although this law makes several important changes in the federal water pollution control program, its most significant feature is its

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incorporating into the Act several of the basic elements of the settlement agreement program for toxic pollution control. Sections 301(b)(2)(A) and 301(b)(2)(C) of the Act now require the achievement by July 1, 1984 of effluent limitations requiring application of BAT for toxic pollutants, including the 65 priority pollutants and classes of pollutants which Congress declared toxic under Section 307(a) of the Act. Likewise, EPA's programs for new source performance standards and pretreatment standards are now aimed principally at toxic pollutant controls. Moreover, to strengthen the toxics control program, Congress added Section 304(e) to the Act, authorizing the Administrator to prescribe best management practices (BMPs) to prevent the release of toxic and hazardous pollutants from plant site runoff, spillage or leaks, sludge or waste disposal, and drainage from raw material storage associated with, or ancillary to, the manufacturing or treatment process.

In keeping with its emphasis on toxic pollutants, the Clean Water Act of 1977 also revised the control program for "conventional" pollutants (including biochemical oxygen demand, suspended solids, fecal coliform, oil and grease, and pH) identified under Section 304(a)(4). Instead of BAT for conventional pollutants, the new Section 301(b)(2)(E) requires by July 1, 1984 achievement of effluent limitations requiring the application of the best conventional pollutant control technology (BCT). The factors considered in assessing BCT include the reasonableness of the relationship between the costs of attaining a reduction in effluents and the effluent reduction benefits derived, and the comparison of the cost and level of reduction for an industrial discharge with the cost and level of reduction of similar parameters for a typical POTW [ Section 304(b)(4)(B) ]. For nontoxic, nonconventional pollutants, Sections 301(b)(2)(A)and 301(b)(2)(F) require achievement of BAT effluent limitations within three years after their establishment or after July 1, 1984 (whichever is later), but not later than July 1, 1897.

This document presents the technical bases for the application of revised BPT, BCT and conventional pollutant new source performance standards (NSPS) for the organic chemicals and plastics and synthetics (OCPS) manufacturing point source category. The technical bases for toxic pollutant related limitations are presented in the "BAT" Development Document which is being published jointly with this report.

#### PRIOR EPA REGULATIONS

EPA promulgated effluent limitation guidelines and standards for the Organic Chemicals Manufacturing Industry, in two phases, in 40 CFR Part 414. Phase I, covering 40 product/processes (a product that is manufactured by the use of a particular process--some products may be produced by any of several processes), was promulgated on April 25, 1974 (39 FR 12076). Phase II, covering 27 additional product/processes, was promulgated on January 5, 1976 (41 FR 902).

EPA also promulgated effluent limitation guidelines and standards for the Plastics and Synthetics Industry, in two phases, in 40 CFR Part 416. Phase I, covering 31 product/processes, was promulgated on April 5, 1974 (39 FR 12502). Phase II, covering 8 additional product/processes, was promulgated on January 23, 1975 (40 FR 3718).

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Several industry members challenged the above regulations. On February 10, 1976 the Court, in Union Carbide v. Train, 541 F.2d 1171 (4th Cir. 1976), granted the parties' motion to remand the Phase I Organic Chemicals regulations. The Court also directed EPA to withdraw the Phase II Organic Chemical regulations, which EPA did on April 1, 1976 (41 FR 13936). Pursuant to an agreement with the industry petitioners, however, the regulations for butadiene manufacture were left in place. The court in FMC Corp. v. Train, 539 F.2d 973 (4th Cir. 1976). remanded the Phase I Plastics and Synthetics regulations. In response, EPA withdrew both the Phase I and Phase II regulations, which had not been addressed in the lawsuit.

Today, there are no promulgated regulations for the Organic Chemicals and Plastics and Synthetics Industries, except for the butadiene and pH regulations mentioned above.

This report presents a summary of the data collected by the studies undertaken since 1976, and the analyses used to support the proposed regulations. Section II presents a summary of the findings presented in this document, along with the proposed regulations. Sections III through VIII present the technical data and the supporting analyses used as the bases for the proposed regulations, and Sections IX through XI include the actual numerical development of the national limitations. Detailed data displays are included in Appendices A-G.

#### SECTION II

#### SUMMARY AND CONCLUSIONS

#### SUMMARY

EPA is proposing effluent limitations guidelines basd on the application of the best practicable technology (BPT), best conventional technology (BCT), best available technology (BAT), new source performance standards (NSPS) and pretreatment standards for existing and new sources (PSES and PSNS).

These proposed regulations apply to wastewater discharges resulting from the manufacture of organic chemicals, plastics and synthetic fibers. The organic chemicals industry is generally included within the U. S. Department of Commerce, Bureau of the Census Standard Industrial Classification (SIC) Major Groups 2865 and 2869. The plastic and synthetic fibers industry is generally included in SIC Groups 2821, 23823, and 2824. Due to the interdependence of these two industries, EPA studied them in combination and is including both of them in a single set of proposed regulations.

When finally promulgated, these regulations will supersede the existing regulations for butadiene manufacture and the pH limitation for the manufacture of plastics and synthetic fibers.

Some plants have OCPS operations that are a minor portion of and ancillary to their primary production. In some such cases, effluent guidelines for the primary production category (e.g., the guidelines for the petroleum refining, pesticides, and pharmaceuticals industries) include subcategories for the discharge of combined wastewaters from the primary production and the OCPS processes. In such cases, to avoid duplication and potential inconsistencies, these OCPS discharges are excluded from coverage by the proposed OCPS regulations and remain subject to the other applicable regulations.

The proposed regulations also do not apply to discharges from the extraction of organic chemical compounds from natural materials. Natural materials used to make organic chemical compounds include a variety of parts of plants (e.g., trees and seaweed) and animals. These proposed regulations address the manufacture of organic chemicals via chemical synthesis. Readers should note that extraction of chemical compounds from natural materials is included in many other industrial categories, e.g., Adhesives and Sealants, Pharmaceuticals, and Gum and Wood Chemicals. Readers should also note that discharges from the synthesis of organic chmical compounds that have been extracted from natural materials are covered by these proposed regulations.

The OCPS industry is large and diverse, and many plants in the industry are highly complex. The industry includes approximately 1,200 facilities which manufacture their principal or primary product or group of products under the OCPS SIC Groups. Some plants are secondary producers, with OCPS products ancillary to their primary manufacture. Various sources studied by EPA indicate that the number of secondary OCPS plants is in the range of 320 to approximately 900 plants. Thus, the total number of plants in the OCPS industry may be as high as 2,100. This range is attributed to the difficulties inherent in segregating the OCPS industry from other chemical producing industries such as petroleum refining, inorganic chemicals, pharmaceuticals and pesticides as well as chemical formulations industries such as adhesives and sealants, paint and ink, and plastics molding and formulating. Even though over 25,000 different organic chemicals, plastics and synthetic fibers are manufactured, only 1,200 products are produced in excess of 1,000 pounds per year. As mentioned above, except for certain specified exceptions, all discharges from OCPS operations at these plants are covered by these proposed regulations.

Some plants produce chemicals in large volumes, while others produce only small volumes of "specialty" chemicals. Large-volume production tends toward continuous processes, while small-volume production tends toward batch processes. Continuous processes are generally more efficient than batch processes minimizing water use and optimizing the consumption of raw materials in the process.

Different products are made by varying the raw materials, chemical reaction conditions and the chemical engineering unit processes. The products being manufactured at a single large chemical plant can vary on a weekly or even daily basis. Thus, a single plant may simultaneously produce many different products in a variety of continuous and batch operations, and the product mix may change frequently.

Total production of organic chemicals in 1980 was 291 billion pounds, with sales of \$54 billion. Production of plastics and synthetic fibers in 1980 was 60 billion pounds, with sales of \$26 billion.

For the 1200 facilities whose principal production relates to the OCPS industry, approximately 40 percent are direct dischargers, approximately 36 percent are indirect dischargers (plants that discharge to publicly owned treatment works), and the remaining facilities use zero or alternative discharge methods. The estimated average daily flow per plant is 2.31 MGD (millions of gallons per day) for direct dischargers and 0.80 MGD for indirect dischargers. The remainder use dry processes, reuse their wastewater, or dispose of their wastewater by deep well injection, incineration, contract hauling, or evaporation or percolation ponds.

As a result of the wide variety and complexity of raw materials and processes used and of products manufactured in the OCPS industry, an exceptionally wide variety of pollutants are found in the wastewaters of this industry. This includes conventional pollutants (pH, BOD, TSS and oil and grease), toxic pollutants (both metals and organic compounds), and a large number of organic compounds produced by the industry for sale).

To control the wide variety of pollutants discharged by the OCPS industry, OCPS plants use a broad range of in-plant controls, process modifications and end-of-pipe treatment techniques. Most plants have implemented programs that combine elements of both in-plant control and end-of-pipe wastewater treatment. The configuration of controls and technologies differs

from plant to plant, corresponding to the differing mixes of products manufactured by different facilities. In general, direct dischargers treat their wastes more extensively than indirect dischargers.

The predominant end-of-pipe control technology for direct dischargers in the OCPS industry is biological treatment. The chief forms of biological treatment are activatd sludge and aerated lagoons. Other systems, such as extended aeration and trickling filters, are also used, but less extensively. All of these systems reduce BOD and TSS loadings and, in many instances, incidentally remove toxic and nonconventional pollutants. Biological systems biodegrade some of the organic pollutants, remove bio-refractory organics and metals by sorption into the sludge, and strip some volatile organic compounds into the air.

Other end-of-pipe treatment technologies used in the OCPS industry include neutralization, equalization, polishing ponds, filtration and carbon adsorption. While most direct dischargers use these physical/chemical technologies in conjunction with end-of-pipe biological treatment, at least 39 direct dischargers use only physical/chemical treatment.

In-plant control measures employed at OCPS plants include water reduction and reuse techniques, chemical substitution and process changes. Techniques to reduce water use include the elimination of water use where practicable, and the reuse and recycling of certain streams, such as reactor and floor washwater, surface runoff, scrubber effluent and vacuum seal discharges. Chemical substitution is utilized to replace process chemicals possessing highly toxic or refractory properties by others that are less toxic or more amenable to treatment. Process changes include various measures that reduce water use, waste discharges, and/or waste loadings while improving process efficiency. Replacement of barometric condensers with surface condensers, replacement of steam jet ejectors with vacuum pumps, recovery of product or by-product by steam stripping, distillation, solvent extraction or recycle, oil-water separation and carbon adsorption, and the addition of spill control systems are examples of process changes that have been successfully employed in the OCPS industry to reduce pollutant loadings while improving process efficiencies.

Another type of control widely used in the OCPS industry is physical/ chemical in-plant control. This treatment technology is generally used selectively on certain process wastewaters to recover products or process solvents, to reduce loadings that may impair the operation of the biological system or to remove certain pollutants that are not removed sufficiently by the biological system. In-plant technologies widely used in the OCPS industry include sedimentation/clarification, coagulation, flocculation, equalization, neutralization, oil/water separation, steam stripping, distillation and dissolved air flotation.

Many OCPS plants also use physical/chemical treatment after biological treatment. Such treatment is used in the majority of situations to reduce solids loadings that are discharged from biological treatment systems. The most common post-biological treatment systems are polishing ponds and multimedia filtration.

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At approximately 5 percent of the direct discharging plants surveyed, no treatment is provided. At another 20 percent, only physical/chemical treatment is provided. The remaining 75 percent utilize biological treatment. Approximately 36 percent of biologically treated effluents are further treated by polishing ponds, filtration or other forms of physical/ chemical control.

At approximately 52 percent of the indirect discharging plants surveyed, no treatment is provided. At another 39 percent, some physical/chemical treatment is provided. Nine percent have biological treatment.

#### CONCLUSIONS

#### BPT

Biological teatment has been identified as the best practicable control technology currently available for each of the four proposed subcategories. In general, the long-term median BPT final effluent BOD, and TSS concentrations were calculated for each subcategory by using the performance of plants which attain 95% BOD, reduction or a final effluent BOD, concentration less than or equal to 50 mg/l.

Maximum 30-day and daily maximum effluent limitations were determined by multiplying long-term median effluent limitations by appropriate variability factors which were calculated through statistical analysis of long-term BOD<sub>5</sub> and TSS daily data. This statistical analysis is described in detail in Section VII.

Proposed BPT limitations are presented in Table 2-1.

#### BCT

The 1977 amendments added Section 301(b)(2)(E) to the Act, establishing "best conventional pollutant control technology" (BCT) for discharges of conventional pollutants from existing industrial point sources. Section 304 (a)(4) designated the following as conventional pollutants: BOD, TSS, fecal coliform and pH. The Administrator designated oil and grease as "conventional" on July 30, 1979, 44 FR 44501.

EPA has proposed a BCT cost-reasonableness test which provides that BCT is cost-reasonable if: (1) the incremental cost per pound of conventional pollutant removed in going from BPT to BCT is less than \$.27 per pound in 1976 dollars, and (2) this same incremental cost per pound is less than 143% of the incremental cost per pound associated with achieving BPT.

All the incremental costs per pound ratios were found to fail this first part of the BCT "cost-reasonableness" test (\$0.33 per pound in 1979 dollars). Therefore, EPA did not perform the second part of the BCT "costreasonableness" test, and is proposing BCT effluent limitations which are equal to the BPT effluent limitations for each of the proposed BPT subcategories.

# TABLE 2-1

# BPT EFFLUENT L'IMITATIONS .(mg/l or ppm)

	LONG TER	M MEDIAN	MAXIMUM	30-DAY	MAXIMUM	DAILY
SUBCATEGORY	BOD5	TSS	BOD5	TSS	BOD	TSS
Plastics Only	14.5	24	22	36	49	117
Oxidation						
o High Water Use o Low Water Use	≥ 26 36	62 89	42 58	84 120	106 146	246 353
Type I	24,5	34.5	40	47	100	137
Other Discharges	5 17	29	28	39	69	115

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The basis for new source performance standards (NSPS) under Section 306 of the Act is the best available demonstrated technology. At new plants, the opportunity exists to design the best and most efficient production processes and wastewater treatment facilities. Therefore, Congress directed EPA to consider the best demonstrated process change, in-plant controls and end-of-pipe treatment technologies that reduce pollution to the maximum extent feasible. It is encouraged that at new sources reductions in the use of and/or discharge of wastewater be attained by application of in-plant control measures.

The technologies employed to control conventional pollutants at existing plants are fully applicable to new plants. In addition, no other technologies could be identified for new sources which were different from those used to establish BPT effluent limitations. Thus, the technology basis for NSPS is the same as that for BPT effluent limitations. For detailed information on the technology basis for BPT effluent limitations, refer to Section IX of this document.

Since the Agency could identify no additional generally applicable technology for NSPS, and since the technology basis for NSPS is the same as that identified for BPT effluent limitations, EPA has established NSPS effluent limitations equal to the proposed BPT and BCT effluent limitations.

#### NSPS

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#### SECTION III.

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#### DESCRIPTION OF INDUSTRY

#### INTRODUCTION

The organic chemicals industry had very modest beginnings in the middle of the 19th century. The production of coke, used both as fuel and a reductant in blast furnaces for steel production, generated coal tar as a by-product. Although these tars were initially regarded as wastes, with the synthesis of the first coal tar dye (mauve) by Perkin in 1856, chemists and engineers began to look for ways to recover and use all industrial by-products.

With increasing numbers of chemical compounds possessing valuable properties being identified, commercial routes to these compounds became necessary. Not surprisingly, the early products of the chemical industry were those most desired by society: dyestuffs, explosives, and pharmaceuticals. The economic incentive to find markets for industrial wastes and by-products continued to be a driving force behind these in-The chlorinated aromatic chemicals industry, for example, dustries. developed mainly out of: (1) the need to use the large quantities of chlorine formed as a by-product from caustic soda production, (2) the availability of benzene derived from coal tar, and (3) the discovery that such compounds could serve as useful intermediates for production of other, more valuable materials, e.g., phenol and picric acid. In time, specialty products such as surfactants, pesticides, and aerosol propellants were also developed.

The plastics and synthetic fibers industry began only somewhat later. The first commercial polymers, rayon and bakelite, were produced in the early 1900s from feedstocks manufactured by the organic chemicals industry. While the organic chemicals and plastics and synthetic fibers industries are regarded as separate, the latter is clearly an outgrowth of the organic chemicals industry. The variety of plastic and synthetic fiber products developed in the last decades and the diversity of markets and applications of these products have made the plastics and synthetic fibers industry the largest consumer of organic chemicals on a volume basis.

Coal derived chemicals were the principal feedstocks of the early industry (though ethanol, derived from fermentation, served as a source of some aliphatic compounds). The growth in the markets for organic chemicals and plastics and synthetic fibers led, in time, however to changes in the source of feedstocks for the industry. By World War II, the modern organic chemicals and plastics and synthetic materials industry based on petrochemicals was firmly established in the United States.

Today the industry is comprised of production facilities of two distinct types: those facilities whose primary function is chemical synthesis and plants that recover organic chemicals as a by-product from unrelated manufacturing operations such as steel production. The bulk of the industry is comprised of plants in the former category: plants that process chemical raw materials into a wide variety of products that permeate virtually every industrial and consumer market. Approximately 90% of the precursors, which are the primary feedstocks for all of the industry's thousands of products, are derived from petroleum and natural gas. The remaining 10% is supplied by plants that recover organic chemicals from coal tar condensates generated by coke production.

The apparent complexity and diversity of the organic chemical manufacturing industry can be simplified by recognizing that approximately 2,500 distinct chemical products are synthesized from only seven parent compounds--methane, ethylene, propylene, butane/butenes, benzene, toluene, and o,p-xylenes. These seven compounds are processed into derivatives which in turn are marketed or used as feedstocks for the synthesis of other derivatives. However, the product line of the industry is very complex with approximately 1,200 products that are produced in excess of one thousand pounds per year, and probably several thousand more that are produced in lesser quantities. Because these products are produced by one or more manufacturers using different synthetic routes, few plants are exactly alike in terms of either product or processes.

The early chemical industry used an assortment of general purpose equipment and operated very labor intensive batch processes that required relatively little capital investment. As the demand grew, around the time of World War II, the chemical production shifted to large scale continuous processing units because of technological improvement and also because of the economies of scale associated with large production facilities. This changed the industry to a high-capital-intensive, lowlabor basis.

Although there is still a large number of small organics producers utilizing batch processes, these producers are usually dedicated to the manufacture of fairly small volumes of high-priced specialty products which may contribute substantially to the total value of organic chemical production, but is only a small portion of chemical production volume.

Since organic chemicals are produced both by large manufacturing complexes made up of continuous major processing units and by smaller batch process plants producing many different products, there is a wide variability of products and process units from one complex to another with treatment facilities typically servicing the complex rather than the individual process units. Among the hundreds of products made by the industry, there are derivative and coproduct relationships that result in groups of products commonly being made together.

#### DEFINITION OF THE INDUSTRY

It is difficult to profile the organic chemicals and plastics and synthetic fibers industries due to their complexity and diversity. However, traditional profiles can provide useful descriptions of the chemical industry. The following profile factors are discussed briefly in the ensuing subsections: Standard Industrial Classification System (SIC) Production and Sales Geographic Location Size of Plant Age of Plant

#### Standard Industrial Classification System

One industrial profile commonly employed for collection of economic data for manufacturing industries is the Standard Industrial Classification (SIC) System. The organic chemicals and plastics and synthetic materials industrial categories are nominally described under SIC 2865 and 2869 for organic chemicals, and SIC 2821, 2823, and 2824 for plastics and synthetic materials. SIC codes as established by the U.S. Department of Commerce are "classifications of establishments by type of activity in which they are engaged." Each plant is "assigned an industry code on the basis of its primary activity which is determined by its principal product or group of products." However, as a practical matter, many plants can also have secondary, tertiary, or subsequent order SIC codes assigned to classify those activities in which they engage beyond their primary activities. Thus the inclusion of establishments with one of these SIC codes as primary, secondary, or subsequent classification would provide an all inclusive listing of establishments producing organic chemicals including such operations as steel mills, which are not intended to be controlled under the organic chemical industry guidelines. This classification system is oriented towards the collection of economic data related to gross production, sales, number of employees and geographic location.

#### Production and Sales

Estimates of the production volume and sales for the OCPS industry were made using the 1981 U. S. Department of Commerce statistics and are shown in Table 3-1. These estimates of production and sales include secondary as well as primary production. Primary products are those materials that comprise the largest portion of a facility's total production. Secondary production involves those products manufactured in smaller volumes as co-products, by-products or as raw materials for primary products. Therefore, these estimates reflect some double counting since certain secondary products are derived from products also included in the total (e.g., ethylene dichloride is included as well as the ethylene from which it is produced). Furthermore, the ITC presents statistics on products or groups of products within a specific use category. These use categories can contain products from more than one SIC code. Where possible, adjustments were made to exclude products not applicable.

The production volumes of the 29 organic chemicals included in the Chemical and Engineering News' 1980 Top 50 List of Chemicals are listed in Table 3-2. The total volume of production for these 29 organic compounds was 78.75 million kkg (173.66 billion 1bs) or 60 percent of all organic chemicals productions (as shown in Table 3-1). Table 3-3 gives the production volumes of the "top" products in the plastics and synthetic fibers categories.

	SIC CODE	Production (million kkg)	Sales (billion dollars)
Organic			
Chemicals	2865	132	11.0
	2869		43.2
Plastics and			
Synthetic	2821	27	16.1
Materials	2823		1.2
	2824		8.7
TOTAL		159	80.2

## ANNUAL PRODUCTION AND SALES BY SIC CODE

SOURCE: U. S. Department of Commerce, 1981.

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## ANNUAL PRODUCTION VOLUME OF ORGANIC CHEMICALS IN "TOP 50" LIST 1980

Rank	Chemical	Production (million kkg)
 б	Ethylene	12.50
13	Urea	6.51
14	Propylene	6.22
15	Toluene	5.12
16	Benzene	4.98
17	Ethylene dichloride	4.53
18	Ethylbenzene	3.45
20	Methanol	3.18
21	Styrene	3.13
22	Vinyl chloride	2.93
23	Xylene	2.91
24	Terephthalic acid	2.69
:5	Formaldehyde	2.62
.7	Ethylene oxide	2.25
8	Ethylene glycol	1.92
0	p-Xylene	1.74
1	Cumene	1.43
2	Butadiene (1,3-)	1.31
3	Acetic acid	1.28
6	Phenol	1.12
8	Acetone	0.96
9	Cyclohexane	0.89
1	Vinyl acetate	0.87
2	Acrylonitrile	0.83
3	Isopropyl alcohol	0.81
4	Propylene oxide	0.80
6	Acetic anhydride	0.67
9	Ethanol	0.55
0	Adipic acid	0.55
TOTAL	·	78.75

SOURCE: Chemical and Engineering News 1981

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# ANNUAL PRODUCTION VOLUME OF PLASTICS AND SYNTHETIC FIBERS 1980

Resin/Fiber	Production (million kkg)
Thermosetting resins	
Phenolic and other tar acid resins	0.68
Polyesters (unsaturated)	0.41
Urea resins	0.53
Expoxies (unmodified)	0.15
Melamine resins	0.08
Thermoplastic resins	•
Low-density polyethylene	3.31
Polyvinyl chloride and copolymers	2.48
Polystyrene and copolymers	2.06
High-density polyethylene	2.00
Polypropylene and copolymers	1.66
Cellulosics	
Ravon	0.22
Acetate	0.15
Noncellulosics	
Polvester	1.81
Nylon	1.07
Glass fiber	0.39
Acrylic	0.35
Olefin	0.34
TOTAL	17.69

SOURCE: Chemical and Engineering News 1981

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#### Manufacturing Sites and Geographic Distribution of Industry

The number of plants operating under each of the five primary SIC codes and classifications and the total number of organics and plastics and synthetic materials plants are shown in Table 3-4. Table 3-5 presents the distribution of these plants by state. It is not surprising that most organic chemical plants are located in the coastal regions near sources of raw materials. The plastics and synthetic materials industries generally follow this trend to minimize transportation costs of monomer feedstock. However, a significant number of plastic plants are situated near end product markets (i.e., large population centers) for the same reason.

The first column in Table 3-4 utilizes Economic Information System data which are based mainly on U. S. Department of Commerce statistics from the Bureau of Census on Manufacturers. These statistics concentrate on primary production facilities and estimates are used to predict the number of smaller facilities below certain employee levels. The second column represents an estimate of all OCPS facilities which attempt to take into account secondary production facilities. In estimating these plant counts, a number of information sources were used, including:

- Permit listings supplied by NEIC-Denver and EPA's Office of Water Enforcement
- 2. 308 Questionnaire plant listings
- 3. EGD Telephone Survey of Plastics and Synthetic Materials facilities
- 4. Plant listings from the economic contractor
- 5. Economic Information Service (EIS) plant listings
- 6. Plant listings from EPA's Permit Compliance System (PCS)
- 7. Dunn & Bradstreet
- 8. TSCA inventories

In compiling plants from the above listings, the number of direct discharge plants was first obtained by cross-checking each of the available data sources. Non-direct discharge plants were projected utilizing ratios of non-direct to direct discharge plants from the 308 Questionnaire plant listings and the direct discharge plants as determined above. SIC code information as well as 308 Questionnaire and Telephone Survey data were used to group all plants into three broad industry segments: (1) plants manufacturing only organic chemicals, (2) plants manufacturing only plastics and synthetic materials, and (3) plants manufacturing both organic chemicals and plastics and synthetic materials in the same facility.

Except for EIS (which utilizes Census of Manufacturers statistics), each of these information sources is independent of the others and provides a

Industry	SIC Code	Number of Plants*	Projected Estimate of Number of Plants
Organic Chemicals	2865	195	1045
Only	- 2869	457	
Plastics and	2821	484	879
Synthetic Materials	2823	19	
Only	2824	62	
Organic Chemicals & Plastics and Syntheti Materials (Combined)	c		176
TOTAL		1217	2100

# OCPS PLANT DISTRIBUTION BY SIC CODE

\* SOURCE: Economic Information Service (1981)

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## PLANT DISTRIBUTION BY STATE

	Organics Chemicals Industry			Plastics and Synthetic Fibers Industry			
	SIC CODE		SIC CODE				
	2865	2969	- Total	2821	2323	2324	Total
STATE							
Alabama	4	5	9	7	1	2	10
Alaska	0	1	1	0	0	0	0
Arizona	0	1	1	1	0	0	1
Arkansas	1	3	4	5	· 0	1	6
California	6	30	36	45	0	1	46
Colorado	1	4	5	3	0	0	3
Connecticut	0	8	8	11	0	3	14
Delaware	1	5	6	12	0	1	13
Florida	3	6	9	7	0	2	9
Georgia	1	6	7	7	1	5	13
Hawaii	0	0	0	0	Ō	0	0
Idaho	Ō	0	Ō	Ō	0	0	0
Illinois	12	32	44	28	1	1	30
Indiana	4	4	8	7	1	-1	9
Iowa	0 0	2	2	4	0	0	Ĺ
Kansas	2	3	5	1	Õ	õ	1
Kentucky	2	8	10	5	ñ	0 0	5
Louisiana	3	33	36	10	0	Ő	10
Maine	0	0	0	10	Õ	n n	1
Maryland	õ	4	4	3	1	. 0	5
Massachusette	7	14	21	32	Ô	2	34
Michign	, ,	16	19	16	ĩ	Ō	17
Minnesota	1	Ĩ	4	3	ò	ů n	3
Missouri	1	7	8	, R	ň	0 0	8
Mississinni	3	1	4	6	ñ	1	ט ר
Montana	0 0	ì	1	1	n	0	, 1
North Caroline	12	11	23	12	n	11	22
North Dakota	0	0	0	0	1	<u>^</u>	1
Nebraska	n	2	2	ñ	n	ñ	
New Hampshire	1	2	2	4	ñ	ñ	6
New Tersey	48	67	115		1	1	47 67
New Jerbey	40 A	1	1	00	U Y	л Т	0Z
New MEXILU	0	1 2	1	1	0 A	0 0	U I
Nevaua Nev Vork	10	27	27	1	U 1	1	1 26
NEW IOLK	10	<b>Z</b> /	57	44	1	1	20

# TABLE 3-5 (Continued)

### PLANT DISTRIBUTION BY STATE

	Organic Chemicals Industry		Plastics and Synthetic Fibers Industry			tic	
	SIC CODE			SIC CODE			
	2865	2869	Total	2821	2323	2324	Total
STATE							
Ohio	18	19	37	43	2	1	46
Oklahoma	0	2	2	5	0	0	5
Oregon	0	4	4	4	0	0	4
Pennsylvania	12	21	33	31	2	0	33
Puerto Rico	3	9	.12	4	0	3	7
Rhode Island	5	6	11	1	0	0	1
South Carolina	10	9	19	5	1	15	21
South Dakota	0	0	0	0	0	0	· 0
Tennessee	1	5	6	7	2	4	13
Texas	17 ·	57	74	35	0	0	35
Utah	1	0	1	2	0	0	2
Virginia	1	4	5	5	2	7	14
Vermont	0	0	0	1	0	0	1
Washington	2	4	6	5	1	0	6
Wisconsin	0	6	6	9	0	0	9
West Virginia	1	11	12	7	0.	1	8
Wyoming	<u>1</u>	<u>o</u>	<u>1</u>	<u>0</u>	<u>0</u>	<u>0</u>	<u>0</u>
TOTAL	198	466	664	488	19	65	572

SOURCE: Continental United States (EIS 1981); Puerto Rico (Bureau of the Census 1977)

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fairly accurate estimate of both primary and secondary production facilities.

#### Plant Size

Sales volume, number of employees, area of plant site, plant capacity (design or "nameplate" capacity) and production rate are factors that logically would be considered to define plant size. However, none of these completely describes plant size in a manner satisfactory for all purposes. Each of these definitions are discussed below.

Often, number of workers employed will be used as an indication of the relative size of a facililty. However, continuous plants producing commodity (i.e., high volume) chemicals typically employ fewer workers per unit of production than do plants producing specialty (i.e., relatively low volume) chemicals. Also, the area of a plant site can be very misleading when considering it for determining plant size. Some plants are built on enormous lots of land but only take up a small portion of that land, while other plants may utilize the entire lot. Sales volume does not accurately define plant size since it is totally dependent on the demand for certain products or the demand for goods produced from those chemical products. Demand may then be dependent on prices and the economy, with sales volume fluctuating because of outside variables, and therefore not relating to a plant or its size.

Table 3-6 and Figure 3-1 present the plant distribution of the organic chemicals and plastics and synthetic materials industries based upon number of employees. Table 3-7 and Figure 3-2 present plant distribution based on sales volume.

For the purposes of this report, plant size cannot be sufficiently defined based on plant or design capacity due to the often broad differences between a plant's design capacity or rate and its average production rate per year. Therefore, plant size for this evaluation is best described by the average production (lbs/day) while operating, as reported in the 308 Questionnaire. Production data on an industry-wide basis is not available. However, a summary and analysis of the 308 production data is presented in Section IV.

#### Plant Age

Plant age within the organic chemicals and plastics and synthetic materials industries is difficult to define since such plants evolve over extended periods of time by additions of product/processes, increases in production rates or changes in technology for the existing product lines. Because new products are continually being introduced by the industry, process units are added to satisfy a growing product demand. Plant age is problematic at such plants, i.e., which process should be chosen to define plant age? Typically, the oldest process in current operation is used to define plant age. Information concerning plant age is not available in the literature and has been compiled from the 308 data base. Table 3-8 and Figure 3-3 illustrate the age (as defined above) of manufacturing facilities within these industries.

	Num Organic P1	Number of Plastics and Synthetic Fibers Plants			
Number of Employees	SIC CODE 2865 2869		SIC CODE 2821 2323 2824		
20-49	77	181	184	4	7
50-99	45	96	107	4	6
100-249	38	79	101	0	9
250 <b>499</b>	23	53	45	1	8
50 <b>0-999</b>	9	28	30	5	7
1000-2499	3	14	16	3	17
2500 <b>-9999</b>	0	6	1	2	8

## PLANT DISTRIBUTION BY EMPLOYMENT SIZE

SOURCE: EIS 1981

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# PLANT DISTRIBUTION BY NUMBER OF EMPLOYEES



FIGURE 3-1

TABLE 3-7
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	Number of Plants			
Sales (million dollars)	Organic	Plastics		
1-10	217	287		
10-20	137	86		
20-30	76	53		
30-40	42	29		
40-50	28	14		
50-60	17	13		
60-70	20	8		
70-80	19	5		
80-90	11	6		
90-100	9	6		
100-110	7	5		
110-120	8	4		
120-130	- 5	4		
130-140	9	4		
140-150	2	3		
150-160	- 2	2		
160-170	-	3		
170-180	4	2		
180-190	3	· · · · · · · · · · · · · · · · · · ·		
190-200	2	3		
200-210	-	1		
210-210	1	-		
220-220	2			
220-230	1	1		
250-240	2	· <u> </u>		
240-250	. 2	4		
250-280	-	1		
270-290	5	· 1		
270-280		2		
280-290	-	2		
290-300	1	2		
300-310	-	-		
310-320	2	2		
320-330	2	-		
330-340	1	-		
340-350	1	• 💻		
350-360	1	1		
360-370	-	2		
370-380	-	-		
380-390	1	-		
390-400	-	-		

### PLANT DISTRIBUTION BY SALES VOLUME
## TABLE 3-7 (Continued)

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	Number of Plants			
Sales (million dollars)	Organic	Plastics		
400-410	1	1		
410-420	1	3		
450-460	1	-		
470-480	1	-		
480-490	1	-		
580 <b>-59</b> 0	1	-		
640-650	-	1		
670-680	1	-		
690-700	1	-		
730-740	-	1		
780-790	1	-		
920-930	1	-		
1240-1250	1	-		
1850-1860	1	-		

### PLANT DISTRIBUTION BY SALES VOLUME

## PLANT DISTRIBUTION BY SALES VOLUME



FIGURE 3-2

TABLE 3-8	
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Age in Years	Number of Plants
0-5	18
6-10	44
11-15	61
16-20	53
21-25	44
26-30	22
31-35	17
36-40	11
41-45	5
46-50	2
51-55	3
56-60	1
61-65	1
TOTAL	282

### DISTRIBUTION BY AGE OF ORGANIC CHEMICALS AND PLASTICS AND SYNTHETIC MATERIALS PLANTS



## DISTRIBUTION BY AGE OF ORGANIC CHEMICALS AND PLASTICS/SYNTHETIC FIBERS PLANTS

FIGURE 3-3

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#### PRODUCT/PROCESS DESCRIPTION

Synthetic organic chemicals are derived from petroleum, natural gas, and coal by some type of chemical reaction (e.g., oxidation, hydrogenation, halogenation, alkylation). The chemical process and its variations can produce an enormous number of potential organic products from a simple list of starting materials.

Petrochemicals are the major raw materials used to produce many organic products. Five major sources (methane, ethylene, propylene, and higher aliphatics and aromatics) are utilized in organic chemical production.[3-1] This list is extended when such aromatics as benzene, toluene, and xylenes used for manufacture are included. A small number of these aromatics are derived from coal, but most raw materials evolve from petroleum and natural gas. In fact, 90 percent (by weight) of all organics are derived from these latter two sources.[3-2] Other raw materials are derived from coal and some naturally occurring renewable sources, notably fats, oils, and carbohydrates. Obscure natural products used as raw materials contribute to specialty chemical production within the organics industry.

Methane, one of the seven basic raw materials, is one of the least complex of the organic chemicals. Even using this simple compound, however, a series of increasingly complex chemicals can be made (see Figure 3-4).

As the chemical complexity of a raw material increases, the variety and number of potential products and chemical intermediates tend to increase also (see Figures 3-5 through 3-8 for the products and intermediates from the raw materials ethylene, propylene,  $C_4$  hydrocarbons and higher aliphatics, and the aromatics). Most of the organic chemicals and plastics and synthetics produced in the U.S. are derived from relatively few basic raw materials, which come almost entirely from petroleum and natural gas.

Even though a portion of the raw materials is derived from other sources (such as coal), these materials are subjected to similar chemical manipulations and appear in the same series of intermediates and products.

Delineation between raw materials and products is difficult to determine at best, since the product from one manufacturer can be the raw material for another manufacturer. This lack of distinction is more pronounced as the process series approaches the ultimate end product, which is normally the fabrication or consumer stage. Also, many products/intermediates can be made from more than one raw material (a specific example of this is acetone which is produced from such raw materials as propylene,  $C_4$  hydrocarbons, and aromatics). Frequently, there are alternate processes by which a product can be made from the same basic raw material.

Another characteristic which makes profiling the OCPS industry by raw material, process, or product difficult is the high degree of integration in the manufacturing units. Since the bulk of the basic raw materials are derived from petroleum or natural gas, many of the organic chemical manufacturing plants are incorporated into petroleum refiner-



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### FIGURE 3-4 - SOME PRODUCTS DERIVED FROM METHANE

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FIGURE 3-5 - SOME PRODUCTS DERIVED FROM ETHYLENE

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# FIGURE 3-6 - SOME PRODUCTS DERIVED FROM PROPYLENE

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### FIGURE 3-7 - SOME PRODUCTS DERIVED FROM C4 AND HIGHER ALIPHATICS

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PIGURE 3-8 - SOME PRODUCTS DERIVED FROM AROMATICS

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ies, and may produce to almost any point in a process from any or all of the basic raw materials. Normally, relatively few organic chemicals manufacturing facilities are single product/process plants unless the final product is near the fabrication or consumer product stage.

This generalized configuration, of which the bulk of the organics industry is comprised, is commonly referred to as a petrochemical complex. Processing arrangements within petrochemical complexes can be quite similar on a worldwide basis since a variety of raw materials, intermediates or finished products is relatively common in the larger scale manufacturing facilities. Furthermore, several processing units are characteristically integrated in such a fashion that the relative amounts of products can be varied as desired over wide ranges.

The capacity of individual plants can change over time. Plants are often modified to produce other products, increase capacity, or produce the same product by a different synthesis route. Some plants or companies exhibit a pronounced degree of vertical integration, while other plants or companies may only produce a limited number of products from one basic chemical raw material. Plant capacities are highly variable even among those plants that use the same unit process to produce the same product.

#### Wastewater Generation

Chemical and plastics manufacturing plants share an important characteristic: chemical processes never convert 100 percent of the feed stocks to the desired products, since the chemical reactions/processes never proceed to total completion. Moreover, because there are generally a variety of reaction pathways available to reactants, undesirable byproducts are often generated. This produces a mixture of unreacted raw materials, products and by-products that must be separated and recovered by operations that generate residues with little or no commercial value. These losses appear in process wastewater, in air emissions, or directly as chemical wastes. The specific chemicals that appear as losses are determined by the feedstock and the process chemistry imposed upon it. The different combinations of products and production processes distinguish the wastewater characteristics of one plant from that of another.

#### Plastic Plants vs. Non-Plastic Plants

In contrast to organic chemicals, plastics and synthetic fibers are polymeric products. Their manufacture directly utilizes only a small subset of either the chemicals manufactured or processes used within the Organic Chemical Industry. Such products are manufactured by polymerization processes in which organic chemicals (monomers) react to form macromolecules or polymers, composed of thousands of monomer units. Reaction conditions are designed to drive the polymerization as far to completion as practical and to recover unreacted monomer. Unless a solvent is used in the polymerization, by-products of polymeric product manufactures are usually restricted to the monomer(s) or to oligomers (a polymer consisting of only a few monomer units). Because the mild reaction conditions generate few by-products, there is economic incentive to recover the monomer(s) and oligomers for recycle. The principal yield loss is typically scrap polymer. Thus, smaller amounts of fewer organics chemical co-products (pollutants) are generated by the production of polymeric plastics and synthetic fibers, than are generated by the manufacture of the monomers and other organic chemicals. A logical first subcategorization step is to separate production of plastics from all other production processes. The subcategorization of the remaining organics and mixed plastics-organics processes is evaluated below.

#### Generic Processes and Product/Processes

Despite the differences between individual chemical production plants, all transform one chemical to another by chemical reactions and physical processes. Though each transformation represents at least one chemical reaction, production of virtually all the industry's products can be described by one or more of 41 generalized chemical reactions/processes shown in Table 3-9. Subjecting the basic feedstocks to sequences of these 41 generic processes produces all the commercial organic chemicals and plastics.

Each chemical product may be made by one or more combinations of raw feedstock and generic process sequences. Specification of the sequence of product synthesis by identification of the products and the generic process by which it is produced is called a "product/process." There are thousands of product/processes within these industries. Data gathered on the nature and quantity of pollutants associated with the manufacture of specific products within the Organic Chemicals and Plastics/ Synthetic Fibers Industries have been indexed by product/process.

Thus, while the industry may be examined on the basis of a plant's capacity, age, size, location, or number of employees, it is the mixture of products and the processes by which they are made that distinguishes the wastewater characteristics of one plant from that of another. Product/ processes are a fundamental descriptor by which data concerning the nature and quantity of pollutants associated with the manufacture of specific products have been gathered. There are, however, thousands of industrial product/process combinations which would have to be evaluated to define the pollutant discharge potential for the entire industry. Evaluation of each for overall wastewater yield losses, to say nothing of identifying the pollutant loadings in the plant effluent, is unnecessarily difficult and burdensome.

The premise of the generic approach is that a generic process once characterized in one or more plants for generation of process wastes (yield losses) can be extended to similar generic processes throughout the industry. Given that biological treatment is widely practiced by direct dischargers (and ultimately by indirect dischargers as well), there is a strong inference that pollutant loadings characteristic of generic processes have similar treatabilities. The bulk of chemical processes employed commercially, moreover, can be limited to a number of generic processes, and this procedure can serve as the basis for relatively simple characterization of the OCPS industry. The great advantage of a generic approach, as applied to effluent regulation within the organic chemicals and plastics and synthetic materials industries, is the struc

### TABLE 3-9

GENERIC CHEMICAL PROCESSES AND CODES

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36.

- Oxidation (C) 26. Sulfonation (M) 1. Peroxidation (8) 2. 27. Nitration (L)
- 3. Acid Cleavage (9)
- Condensation (A) 4.

Isomerization (22) 5.

Esterification (G) 6.

7. Hydroacetylation (13)

8. Hydration (12)

9. Alkoxylation (5)

10. Hydrolysis (E)

11. Carbonylation (0)

- 12. Hydrogenation (F)
- Neutralization (24) 13.
- 14. Amination (6)
- 15. Ammonolysis (K)
- 16. Oximation (10)
- 17. Dehydration (Q)
- Ammoxidation (N) 18.
- 19. Electrohydrodimerization (19)
- 20. Cyanation/Hydrocyanation (7,18)
- 21. Epoxidation (21)
- Etherification (14) 22.
- 23. Polymerization (D)
- 24. Alkylation (I)
- 25. Dehydrogenation (J)
- 37

### OTHER CLASSIFICATIONS

- Non OCPS Product/Processes (3)
- Cannot Be Classified (2)

- 37, Oxyhalogenation (S)
- Hydrohalogenation (P)

35. Fiber Production (23)

Halogenation (B)

Hydrodealkylation (U)

Extractive Distillation (15)

34. Crystallization/Distillation (17)

Pyrolysis (H)

Cracking (T)

33. Extraction (16)

Distillation (2)

- 40. Chlorohydrination (20)
- 41. Phosgenation (V)

- 39. Dehydrohalogenation (R)
- 38.

turing of existing data within a framework which allows extrapolation to processes not explicitly evaluated.

The extent to which process yield losses can be correlated with generic process types depends on the ability to evaluate the chemical reaction system. The evaluation must include a full consideration of the process chemistry with a wide range of feedstocks and reaction conditions. Some of the fundamental concepts of this procedure are presented in the following discussion.

Manufacture of a chemical product necessarily consists of three steps: (1) combination of reactants, under suitable conditions, to yield the desired product, (2) separation of the product from the reaction matrix (e.g., by-products, co-products, reaction solvent), and (3) final purification of the product. Among the basic concepts that can be employed to limit the scope of pollutants expected from a plant are: (1) conservation of mass, (2) principles of thermodynamics, and (3) kinetic or mechanistic analyses.

In general, chemical species do not react via a single reaction pathway. Depending on the nature of the reactive intermediate, there are a variety of pathways which lead to a series of reaction products. Often, and certainly the case for reactions of industrial significance, one pathway may be greatly favored over all others, but never to total exclusion. Thus, by appropriate process design and proper control of reaction conditions, product yield is maximized. There are two fundamental sources of pollutants within a process: pollutants formed as the result of alternate reaction pathways; and reaction, by either the main or alternate reaction pathways, of impurities present in feedstocks. With regard to the latter, it is important to realize that even though feedstock impurities may be inert under a given set of reaction conditions, the direct discharge of such impurities to the environment may still represent a significant pollution potential.

Potentially, an extremely wide variety of compounds could form within a given process. The formation of expected products from known reactants is controlled thermodynamically while the rate at which such transformations occur depends upon the existence of suitable reaction pathways. Detailed thermodynamic calculations are of limited value in predicting the entire spectrum of products produced in a process. Both the identity of true reacting species and the assumption of equilibrium between reacting species are often speculative. Also, kinetic data concerning minor side reactions are generally unavailable. Thus, neither thermodynamic nor kinetic analyses alone can be used for absolute prediction of pollutant formation. However, these analyses do provide a framework within which pollutant loadings may be considered and generalized.

The direction of reactions in a process sequence is controlled through careful adjustment and maintenance of conditions in the reaction vessel.

The physical condition of species present (liquid, solid, or gaseous phase), conditions of temperature and pressure, the presence of solvents and catalysts, and the configuration of process equipment dictate the kinetic pathway by which a particular reaction will proceed. From this

knowledge it is possible to identify reactive intermediates and thus anticipate species (potential pollutants) formed.

To produce a complete and valid descriptor using the generic methodology, the initial feedstock and each generic process used to produce a final product must be specified. For commodity chemicals, generally it is sufficient to specify a feedstock and a single generic process. Nitration of benzene to produce nitrobenzene, for example, is sufficient description to predict composition of process wastewaters: nitrophenols will be the principal process wastewater constituents. Other compounds, however, may involve several chemical reactions and require a fuller description. For example, acetic acid and its anhydride can be produced by first manufacturing acetaldehyde by the hydration of acetelene, followed by the oxidation of acetaldehyde to acetic acid and acetic anhydride.

This example is relatively simple and manufacture of speciality chemicals is more complex. Thus, as individual chemicals become further removed from the basic feedstocks of the industry, fuller description is required for unique specification of process wastewaters. Limited plant data, however, were available by which to assign generic processes to a product, and in many cases the product was specified while the feedstock was not.

In such cases a generic process assignment was made on the basis of process chemistry and engineering, i.e., judgment was made as to the feedstock and chemistry employed at the plant. In no case, however, was more than one generic process assigned to a given product within a production line.

Appendix A presents the product/process frequency counts for the 308 Summary Data Base for direct dischargers, and zero dischargers and alternative disposal plants by each of the 41 generic product/processes.

#### DATA BASE PROFILE

#### Introduction

Despite the wide range of plant sizes, the diversity of plant specific product/processes, and the dynamic nature of technological innovations and market conditions, the OCPS industry is characterized using the latest available data from the industry and published sources.

Most of the data used for the engineering analysis in this report are extracted from the industry responses to the 1976 BPT questionnaire and the subsequent 1977 BAT questionnaire. The data from these questionnaires were transcribed on a plant-by-plant basis to a computer tape. The transcribed data for each plant were then computer printed and the individual data were submitted from December 1979 to January 1980 to the plants for review and comments. Also, long-term daily pollutant raw waste and final effluent data were collected and transcribed to the computer at this time. Additionally, some qualitative information on the generation of wastewater and mode of discharge at 301 plastics manufacturing facilities was obtained by a supplemental telephone survey. It was also determined whether these plants produced resins and polymers which should be included in the data bases or whether the operations were limited to extrusion and/or fabrication of plastics.

The data sources included in this analysis are as follows:

1. The Telephone Survey data from the 301 plastic manufacturing plants, 251 of which were not covered in the 308 Data Base. This survey consistently determined the mode of discharge (direct, indirect, zero), location, and general product type. Thirty seven of the plants contacted were identifiable as extruding or otherwise fabricating plastics from purchased polymers. At the present time, the data from these 37 plants are included only in the mode of discharge portion of the data bases.

2. The daily data from plants contained in the 308 Data Base. A following subsection details the decisions involved in the selection of the plants included in the long-term Daily Data Base.

3. The original 308 data tape was used as the basis for the current specialized data bases. The following subsection describes the changes made to the original data and the parameters and terms used to profile the data.

The 308 Questionnaire was designed to collect information that would adequately describe and characterize the OCPS industry. Requested information related to such items as products manufactured, processes used, production rates, age, size, water consumption, wastewater generation, treatment technologies employed, and influent and effluent characteristics.

The responses varied in respect to completeness of response and detail of information. Some plants misinterpreted the units requested, did not give complete responses, provided data in units other than those requested, or otherwise responded in a manner which required either recalculation of the data, follow-up contacts for clarification, or in some cases rejection of the data. This may be explained in part by the fact that some companies simply did not keep records of information as was requested by the questionnaire, and consequently could not respond fully on all items of interest.

The data acquired from the questionnaire were necessary to assure that the industry was adequately described and to determine the need for subcategorization of the industry. Some specifics of the problems associated with the raw data and the corrective steps required for clarification are discussed in the following sections of this report.

The names applied to the data bases used in this report and a brief description of their contents are shown in Table 3-10 and Figure 3-9. These data base names will be used where the data bases are referred to in the text of this report.

### TABLE 3-10

### DATA BASE DESIGNATION

Data Base File Name	Description					
308 Data Base	Original data base containing all data extracted from 308 Questionnaires					
Daily Data Base	Contains long-term influent d effluent data from 50 plants					
Summary Data Base	Updated version of 308 data base covering the 291 direct & zero discharge plants					

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(see note 4)

### NOTES:

- (1) 308 Data Base contains information on 566 plants
- (2) Daily Data Base contains information on 50 plants
- (3) Summary Data Base contains information of 291 plants
- (4) Telephone Survey Data Base contains information on 301 plants

### FIGURE 3-9 - DATA OVERLAPS

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In addition to the above data bases, there are numerous files of data that have been generated and stored in the computer to allow segregation and manipulation of special or selected data. These files contain such data as plant numbers, product/processes and treatment systems for indirect discharge plants, and plants rejected from the direct discharge and zero discharge (Summary) data base ("gray" plants) because the majority of each plant's production was not associated with the OCPS industry.

#### Final Data Base Development

The Summary Data Base is a corrected and updated version of the original data found in the 308 Data Base. Since this report covers only zero and direct dischargers, the 343 plants shown in the 308 Data Base as using those discharge modes were used as the initial list of plants for the Summary Data Base. A review of the information in the files of the remaining 223 indirect dischargers (including the responses from 1979 mailing to industry for data update) indicated that another 35 plants could no longer be classified as indirect dischargers. This brought the number of plants to be used in the Summary Data Base to 378. This left 188 plants marked as indirect dischargers in the 308 Data Base.

Data on product/processes, plant location and age, production, percent operating capacity, mode of discharge, treatment unit operations, influent and effluent wastewater flow and concentrations, age, and period of data collection were obtained from the original data printouts for each of the 378 plants in the total direct/zero discharge data base. The file for each plant was examined and the data were modified to reflect any corrections to the original data and to incorporate the plant's responses to the 1979 mailing. After these final corrections, the data were placed in a System 2000 Data Base Mangement System (DBMS) on EPA's UNIVAC Computers.

Examination of these data, however, pointed out problems which led to the deletion of 87 of the plants from the initial Summary Data Base. Forty-two of the deleted plants were rejected from the Summary Data Base because they were found to be indirect dischargers whose status had changed from direct or zero dischargers; eight of these indirect discharge plants also utilize some type of zero discharge technique. Eleven additional plants had data which were not representative of this industry. These included plants which have since been shut down, plants which have been sold or no longer make the products described in the BAT mailing, and one plant whose influent includes a large and unquantifiable amount of municipal sewage. Thirty-four plants were rejected because their products do not fall under the SIC codes being studied. These plants were divided into two groups. One group consists of 23 plants which clearly are not primarily in the SIC codes under study (e.g., refineries, paper mills, tall oil plants, welding gas plants, and plastics extrusion and compounding plants which do not polymerize on site). The other group consists of plants which make organics, but which are primarily inorganic plants. These plants typically have only one treatment system for all plant operations, with the wastewater from organics processes accounting for less than 10 percent of the total

plant flow. This group of 11 rejected plants ("gray" plants) is segregated from the other group and may be studied separately or later with the rest of the organics industry. The final Summary Data Base, which after rejection of these 87 plants contains data from 291 plants, was entered into a System 2000 DBMS on the UNIVAC Computer.

The final Summary Data Base contains detailed information on 291 plants which are direct dischargers or zero discharge/alternate disposal facilities. They were selected from the 566 plant data base (308 Data Base) for the reasons given previously. In addition to the 566 plants, some information is available on 243 plants from the EGD Telephone Survey, giving a total of 809 plants which are represented in some way in the data bases. This means that about 67 percent of approximately 1200 plants (or 40 percent of approximately 2100 plants estimated to be included in the OCPS industry by EPA) are directly covered in the combined data base.

SIC Code Applicability - As a result of the complexity of many plants in the chemical industry, several unrelated SIC codes may be applicable to a single plant. Consequently, the boundaries of the OCPS and related industries may not be sharply defined using SIC codes.

As a result there exists an overlapping of SIC code coverage in the Summary Data Base. Although the data included in the Summary Data Base are for the OCPS industries, some plants which manufacture primarily other materials, but also produce organic chemicals or plastics (e.g., production of alkyd or urethane resins in paint plants and formaldehyde production in adhesive plants), have been included in the Summary Data Base where the relevant development document specifically left such production for limitation by the OCPS regulations.

Additionally, where a separate wastewater treatment system exists for the OCPS portion of a mixed product (SIC code) plant, that plant's data were included in the Summary Data Base. Plants that specifically included manufacture of OCPS products in their regulations (petroleum refining, production of rosin resins in gum and wood chemicals, and pharmaceuticals) have been excluded from the Summary Data Base since this production is already covered by those regulations. Figure 3-10 presents these data base and industry guideline overlaps.

<u>Stream/Plant Distinctions</u> - The 291 plants in the Summary Data Base actually represent 377 different wastewater streams. A wastewater stream in this context is defined as a discrete disposal method used for the disposition of some of a plant's wastewater; dry processing and recycling of wastewater count as a stream each. For example, if a plant had two wastewater streams going to one activated sludge system, three to deep well injection, two processes which discharge wastewater untreated, and two dry processes (i.e., processes which neither use nor generate process contact water), it would be defined as having four waste streams: activated sludge, deep well, no treatment, and dry processing. However, if the two wastewater streams going to one activated sludge system were instead going to two separate activated sludge systems and had separate influent and effluent data for each, data for <u>five</u> streams would exist: two with activated sludge, one with deep well injection,



#### NOTES:

- (1) No identified direct discharge Data Base overlap with organic chemicals and plastics and synthetic materials industries.
- (2) Identified direct discharge Data Base overlap with organic chemicals and plastics and synthetic materials industries.
- (3) Overlapping plants excluded from Data Base because organic chemical production is covered by categorical regulations through petrochemical and chemical synthesis subcategories in the appropriate industries.

FIGURE 3-10 - DATA BASE AND RELATED INDUSTRY GUIDELINES OVERLAP

one with no treatment, and one with dry processing. Separation of the plant processes in this manner allows each process to be linked with the influent and effluent of the treatment system to which it goes, rather than simply be considered as a contribution to an overall plant average loading.

Of these 377 streams, 212 are direct, 162 are zero or alternate disposal and three are of unknown disposition. The majority of plants (225) have only one discharge. The remaining 66 plants account for the other 152 waste streams. The tallies of plants and the associated streams are given in Table 3-11.

Some of the tables in this report are presented in terms of plants, some are presented in terms of streams, and some are presented in terms of both. It should be noted whether the word "plant" or the word "stream" appears in the title or content of each table.

<u>Stream Combining</u> - Early emphasis on the data evaluation was put on the determination of overall plant wastewater treatment efficiencies and effluent qualities. Since, by inspection of the 308 data, it was evident that biological treatment, especially activated sludge, was the most prevalent method of treatment, these plants were the first examined. Where more than one treatment system existed at a plant, the data over the systems were combined by calculating a total composite influent load and a total composite effluent load, and then the overall removal of a given pollutant parameter achieved by the plant was calculated.

In subsequent data evaluation efforts required to demonstrate the efficiency of a particular treatment technology, it became apparent that this procedure of combining streams to arrive at an overall influent and effluent loading over multiple treatment systems (including the "no treatment" discharges) was not suited for the study of individual treatment system performance because it led to gross over or under estimation of the efficiency of a specific technology. For example combining the characteristics of the effluent from a well operated biological treatment plant with an untreated stream could mask the effectiveness of the biological treatment plant.

To avoid the potential misrepresentation of treatment efficiencies, stream combination was abandoned except for five plants which utilize either dual biological or non-biological treatment. Each of these plants have multiple treatment systems for which the data presented were not detailed enough to allow separation of the data to evaluate the individual treatment system's performance. However, since the treatment technologies employed at each of the plants are similar within the multiple treatment installation at that plant, it is judged that no substantial data errors are generated by combining the streams and using the resultant data. For example, the data from a biological treatment system are not being combined with the results from a non-biological treatment system.

Finally, streams have been combined where product/processes could be specifically allocated to each stream. For example, if a plant sends its wastewater to north and south wastewater sewers without specifying

#### <u>All</u> Direct Discharge Zero Discharge Unknown Discharge For All Plants 291 94 2 Total Number of Plants 195 67 2 Number of Plants with One Stream 225 156 Number of Plants with Multiple 66 27 Streams 39 Number of Plants with Zero Discharge Streams 127 33 94 Number of Plants with Direct Discharge Streams 195 195 Number of Plants with Unknown 3 Discharge Streams 3 -For Plants Of Any Number Of Streams Total Number of Streams 377 251 124 Number of Direct Discharge Streams 212 212 \_ Number of Zero Discharge Streams 162 38 124 Number of Unknown Discharge Streams 3 1 2 FOR PLANTS WITH MULTIPLE STREAMS 152 95 57 Total Number of Streams Number of Direct Discharge Streams 56 56 -Number of Zero Discharge 95 38 57 Streams Number of Unknown Discharge Streams 1 1

#### TABLE 3-11 - PLANTS AND THEIR ASSOCIATED STREAMS

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which process has its wastewater sent to which sewer, and if both streams are treated with oil/water separation, it was assumed that those two separators had been combined.

<u>Cooling Water</u> - Often, effluent data for a plant is gathered after noncontact cooling water is mixed with the effluent from the treatment system. This dilution will decrease the apparent effluent concentration from the treatment processes. To factor out the effects of this dilution, each plant's effluent flow was reduced by the amount of the cooling water. The necessary assumption is that the total pounds of pollutant discharged are due to the effluent from the treatment system and no pollutants were contributed by the noncontact cooling water. If data on the cooling water were available, they were used for back calculation instead of assuming the water to be uncontaminated. The practice of reporting the plant effluent on the basis of total discharge (i.e., including commingled noncontact cooling water) is very common in this industry since most state regulatory agencies require the information on discharges to be based on total discharges and the quality thereof.

Of all the plants in the Summary Data Base, a total of 49 plants had cooling water commingled with the treatment plant discharge, thus requiring calculation to eliminate the diluting effects of the cooling water.

The assumption of uncontaminated cooling water will result in slight underestimates of treatment efficiency since the cooling water will not actually be completely free of contamination. It will also result in conservative (i.e., slightly high) estimates of effluent concentrations from the treatment facilities. However, it should be noted that cooling water can contribute relatively high TSS loadings, especially to the typically low strength plastics and synthetic materials wastewaters. If a cooling water stream was combined at the influent of a treatment system after the influent sample point, a composite influent stream was also developed as described above.

Offsite Treatment - One of the more confusing issues concerning the designation of treatment facility type was offsite treatment. Offsite treatment refers to that method used by a plant which discharges its wastewater to a privately or jointly owned treatment work. This designation was a source of confusion since some plants employing "offsite" treatment were originally regarded as indirect dischargers. Subsequent analysis determined that a plant discharging to a treatment work not owned by a governmental entity would not be covered by pretreatment standards for existing sources or pretreatment standards for new sources, and therefore would be covered by this study.

The offsite treatment plants were first differentiated from the 22 plants which use contract removal. Contract removal was considered to be removal in drums or trucks. Offsite-treated wastewater is defined as being wastewater piped directly to the treatment system handling the wastewater. Two more plants were removed because they had been purchased by the plant treating the wastewater. The product/processes and all other parameters associated with the purchased plants were combined with those from the parent plant.

After removing these two types of plants, there were still six plants which pumped their wastes either to a jointly or a privately owned treatment work. These six plants are described as utilizing offsite treatment.

<u>Generic Processes</u> - In addition to cataloging the products at each plant by product/process numbers, data was also sorted using generic chemical processes. For this effort, a list of 41 major generic processes was examined. General categories were also established for inorganic operations on organic chemicals, items which are called chemical processes but which really are not (cooling tower blowdown, etc.), products for which insufficient data exist to characterize the process, and products outside the SIC codes of interest to this report. A list of the generic codes used in the industry is shown in Table 3-9.

Each of the product/processes was then examined to determine its appropriate generic unit process. Where more than one generic process was required to characterize any particular product/process, engineering judgment was exercised to assign the step in the overall process most likely to generate wastewater.

<u>Possible Sources of Inaccuracy</u> - As is the case whenever a large compilation of diverse types of data are accumulated from a large number of varying sources, there are potential sources of error both in the data accumulated and in the interpretation of the data. Errors can arise because of questionnaire ambiguities and technical misinterpretation by the respondees. Some examples of these possible errors are:

1. The assumption that MGD was interpreted as million gallons per day, a commonly recognized term used by people in the wastewater field. However, in many responses "M" was interpreted as the Roman numeral for thousand, a practice also fairly common in many fields of engineering including those of the chemical industry.

2. Misinterpretation of treatment technology definition. This was most evident in the lack of consistency in referring to treatment processes which have subtle differences such as aerobic lagoon vs. aerated lagoon, the several options of activated sludge processes, and the use of colloquial or "house" names for such technologies as tertiary lagoons, polishing ponds and similar installations.

3. Failure of the respondents to fill in the questionnaire completely, or the submittal of conflicting or contradictory information.

To alleviate the effect of the possible errors, engineering judgments and calculations were made to determine reasonable values based on the data supplied, or follow-up contacts were made to plant personnel to clarify the data in question.

A source of inaccuracy in the data is the reporting of identical influent and effluent flows. This is a very common practice in industry where the effluent values for flow are reported, and for control of the waste treatment system influent flow is equated. Although this procedure may create errors by not accounting for slight amounts of water removed in the sludge, evaporation or other miscellaneous losses, the discrepancies introduced by equating influent and effluent flows are of more theoretical than practical interest and would be meaningful only for the most highly sophisticated material balance studies.

Another possible source of error is the reporting of net pollutant values. The NPDES permits at some plants allow them to offset the pollutant concentration of intake waters and allow them to report only the increase in pollution due to the plant. Many plants have this kind of permit and, since the 308 Questionnaire asked for pounds per day rather than concentration of pollutants, therefore reported the net increase in lbs/day used for their NPDES Discharge Monitoring Reports (DMRs) rather than the total amount of discharge. Unless specifically stated as such in the questionnaire response, plants reporting net BOD (or other parameter) discharges could be detected only where negative discharge loads were reported. Where detected, the values for net reporting were adjusted to the gross value prior to entry in the Summary Data Base. One plant (113) reported a negative value and three plants indicated the use of net values.

An additional potential error source could lie in the lack of differentiation between the filtered (soluble) BOD and unfiltered (total) BOD values. The customary practice in industry is to report total BOD since practially all reporting is normally done for permit purposes concerned with the total pollutant discharged. A review of the Summary and Daily Data Base plants showed only four plants which specified the method of reporting or analyzing for BOD: one plant specified unfiltered BOD results, another plant reported four months filtered and the remainder unfiltered, and two plants reported total BOD.

Another possible source of error in BOD values is the use of chlorination by some plants before the effluent sample point. Although there may be some reduction in BOD due to chlorination, the main effect of chlorination can be interference with the BOD test procedure. Residual chlorine in the wastewater sample can inhibit the growth of the bacterial seed used in the BOD test. This results in a BOD value lower than the true oxygen demand of the pollutants contained in the sample. However, since the effect of chlorine on BOD determinations is well known, the laboratory procedures (Standard Methods) employed in practically all wastewater laboratories provide methods for the removal of the chlorine before the BOD analyses are carried out. Examination of the data indicates that the respondents followed standard analytical procedures in BOD determinations.

The values for COD and TOC used in the Summary Data Base were collected from the data provided in the 308 Questionnaire responses. The values furnished could have been derived from laboratory determination, BOD/COD/TOC ratios, or values taken from literature. Since any source other than laboratory determinations or a properly derived and applied statistical correlation of parameter ratios may lead to erroneous conclusions, every effort was made to exclude COD and/or TOC values which could not be verified as being derived from acceptable procedures. Only These sources of error, if ignored, could seriously affect the quality of the data bases. Since these sources of error can and do exist in the accumulated raw data, a diligent review of the 308 Questionnaires and supplemental information was made in an effort to identify and either reject or correct discrepancies. Only after this review was data incorporated into the Summary Data Base.

#### Daily Data Base Development

One of the major purposes of this study is the development of long-term daily pollutant data. These data are required to derive variability factors which characterize wastewater treatment performance and provide the basis for derivation of proposed effluent limitations guidelines and standards. Hundreds of thousands of data points have been collected, analyzed, and entered into the computer.

The first effort at gathering daily data involved the BPT and BAT mailings. These questionnaires asked each plant for backup information to support the long-term pollutant values reported. Many plants submitted influent and effluent daily observations covering the time period of interest in the BPT questionnaire (January 1, 1976 to September 30, 1976). Additionally, there were some other conventional and nonconventional pollutant daily data in the files from the period of verification sampling. Some plants also submitted additional data with their responses to the 1979 mailing. Data from these three sources were examined and interpreted.

Approximately 50 plants were identified as those routinely taking influent and effluent daily observations of parameters of interest. These plants were contacted, and in many cases visited. The contacts usually resulted in accumulation of long-term (sometimes six years) influent and effluent data and detailed information on plant operations. Other data were obtained from various EPA offices which provided long term daily data for a total of 56 plants. Data from fifty of the plants were transcribed, keypunched, and loaded into the computer. Data from six of the plants were never used due to deficiencies in data.

After the data from plants were available on the computer, further investigation resulted in the reconsideration of some of the plants and/or data. However, the data from these flagged plants may be utilized in some of the statistical evaluations. Reasons for flagging of the transcribed daily data plants are:

- 1. Incomplete data (either influent or effluent data missing).
- 2. Major process changes during data collection period.
- 3. Dilution of effluent stream before sample point or influent stream after sample point which causes some difficulty in analyzing certain stream's data for each day.

4. The existence of conditions which indicate poor operation of a biological treatment system. Examples of these conditions include low MLSS for activated sludge plants, extreme carryover of suspended solids, and unconventional design parameters such as inadequate detention times.

A detailed acceptance/rejection analysis of Daily Data plants, for use in development of variability factors, is presented in Section VII.

The final Daily Data Base consists of data from 50 plants. These data are available in two forms on the UNIVAC computer. They are available in the units in which they were measured at the plant (some in mass per unit time and some in concentration) and in a file which has been preprocessed to a consistent set of flow (million gallons per day) and concentration (milligrams per liter) units. Each day's data consists of the plant number, the date and the influent and effluent parameters for flow BOD, COD, TSS, TOC, ammonia, oil and grease, phenol, chromium, pH, and temperature, where available.

#### Mode of Discharge

There are three basic discharge modes utilized by the industry: direct, indirect and zero or alternative disposal/discharge. Direct dischargers are plants which have a contaminated effluent, treated or untreated, which is discharged directly into a surface water. Plants with only noncontact cooling water or sanitary sewage effluents are not considered to be direct dischargers for purposes of this report. Indirect dischargers are plants which route their effluents to publicly owned treatment works (POTWs) and are therefore subject to pretreament standards. Discharge of wastewaters into the system of an adjoining manufacturing facility or to a treatment system not owned by a government entity is not considered indirect discharge, but is termed offsite treatment (see Offsite Treatment). Indirect dischargers are outside the scope of this report. Zero or alternative disposal/dischargers are plants which discharge no wastewater to surface streams or to POTWs. For the purposes of this report, these include plants which generate no wastewaters, plants which recycle contaminated waters, and plants which use some kind of alternate disposal technology (e.g., deep well injection, incineration or contractor removal).

Some plants with insufficient information to determine discharge mode were termed unknown dischargers (see Table 3-12).

The final Summary Data Base contains 291 plants, 195 of which are direct dischargers, 94 of which are zero or alternative disposal/dischargers, and two of which are unknown. No indirect dischargers are included. These 291 plants contain 377 waste streams, 212 of which are direct, 162 of which are zero or alternative disposal/dischargers, and three of which are unknown. The 195 direct discharge plants include 33 plants which also utilize zero or alternative disposal discharge techniques (see Table 3-12).

### TABLE 3-12

Summary Data Base (291 Plants)	Num All	ber o: Dir	f Plan Zero	<u>Unk</u>	Num Al	ber of <u>1 Dir</u>	Strea Zero	ams <u>Unk</u>
A11	291	195	94	2	377	212	162	3
Organic - Only	62	41	20	1	89	52	36	1
Plastic - Only	113	67	45	1	146	77	67	2
Organics/Plastics	116	87	29	-	142	83	59	-
Daily Data Base							,	
A11	50	50	-	-	-	-	-	-
Organic - Only	6	6	-	-	-	-	-	-
Plastic - Only	17	17	-	-	-	-	-	-
Organics/Plastics	27	27	÷	-	-	-	-	-

### NUMBER AND TYPES OF PLANTS AND STREAMS IN THE DATA BASES

Informal Telephone Survey

All 301 Nonduplicated Plants 251

### Although there are several possible ways to describe plant size, the 308 Questionnaire did not ask for sales, employment, or acreage data. Therefore, the only possible remaining description for size in the 308 Data Base is capacity and actual production. Since data available on the industry do not include capacity, and since one would not expect design capacity to determine the pollutant load, those numbers were not included in the Summary Data Base.

The Summary Data Base includes values for actual production as shown in the 308 Questionnaire and for percent of operating capacity being used in each plant. These data are presented on a stream basis as the total production of all products made at the plant whose wastewaters are directed to that stream and include production of all products contributing wastewater, including inorganics and other products not covered under the SIC codes of interest to this study. Table 3-13 presents the total production or organic chemicals and plastics in the 308 Data Base, the total OCPS production for industry determined by the Bureau of Census, and the Summary Data Base production values.

#### Age

Plant age could have an impact on pollutant loadings since water use, process technology, waste treatment technology, and plant maintenance techniques have vastly improved over the years since industry beginnings. Age was defined for purposes of this study as the year of installation of the oldest remaining unit at the plant. Table 3-14 presents the distribution of plant ages in the Summary and Daily Data Base. Plant ages range from two to 64 years, with most plants between 9 and 24 years old.

#### Products

The OCPS industry may be described in terms of the number and variety of products manufactured. This can be done by listing the manufactured products in a broad categorization such as "organics" and "plastics and synthetics" or by listing the separate products made at each plant.

The latter approach would provide useful information concerning the prediction of the presence of toxic pollutants at each plant but would not contribute significantly to the study of conventional and nonconventional pollutant parameters found in end-of-pipe wastewater. Therefore, the former method of using broad product categories was used for grouping data. Many of the tables in this section have included information based on plants which make only organic chemicals, plants which make only plastics, and plants which make both.

The final Summary Data Base contains 62 plants which are organic only producers, 113 plants which are plastic-only manufacturers, and 116 plants which make both (see Table 3-12). Approximately 1200 products exist in the 308 Data Base, while 31 percent (373) exist in the Summary Data Base.

#### <u>Size</u>

### **TABLE 3-**13

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#### PRODUCTION COMPARISONS

	All Products (billion lb/yr)	Plastics ( <u>billion lb/yr</u> )	Organics ( <u>billion lb/yr</u> )
Total U.S. *	350	60	<b>29</b> 0
Total 308 Data Base	198.4 (57%)	44.6 (74%)	153.8 (53%)
Summary Data Base**			
All Streams	230	N/A	N/A
Direct Streams	190	N/A	N/A
Zero Streams	40	N/A	N/A

\* per Table 3-i

\*\* These data include manufacture of all products contributing to wastewater loading, including inorganics and other products not under study here.

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N/A Data not collected in such a way as to make these numbers available

	Streame						Plants			Plants			
	411	<b>D</b> 4 -	7	Ilak		D1 -	7	11-1-	Dailj				
, <u>,</u>		DIT	Zer.	UNK.		DIF	zer.	Unk	Data				
2	1	1	-	-	1	1	-	-	-				
3	6	4	1	1	5	4	1	-	1				
4	8	6	2	-	8	6	2	-	2				
5	4	1	3	-	4	1	3	-	-				
6	8	4	4	-	7	4	3	-	-				
1	6	3	3	-	4	2	2	-	1				
8	7	2	5	-	7	3	4	-	-				
9	14	11	3	-	13	10	3	-	4				
0	14	8	6	-	13	7	6	-	-				
1	16	8	8	-	12	7	5	-	1				
2	14	9	5	-	12	1	5	-	1				
3	10	8	2	-	9	7	2	-	2				
4	11	4	6	1	10	5	4	1	3				
5	20	16	3	1	18	16	1	1	4				
6	9	6	- 3	-	8	6	2	-	2				
7	10	5	5	-	10	5	5	-	· 1				
8	13	1	6	-	12	7	5	-	1				
)	14	7	7	-	14	7	7	-	1				
כ	10	5	5	-	9	4	5	-	2				
	13	7	6	-	11	7	4	-	2				
2	11	6	5	-	9	4	5	-	-				
	10	8	2	-	9	7	1	-	•				
•	10	7	3	-	9	1	2		1				
	7	5	2	-	7	5	2	-	-				
i i	4	2	2	-	4	נ	1	-	-				
	4	4	-	-	4	4	-	-	1				
r	5	4	1	-	5	4	1	-	1				
	5	2	3	-	5	2	3	-	-				
l	4	2	2	-	4	2	2	-	1				
•	2	-	2	-	2	-	2	-	:				
2	8	5	3	-	5	4	1	-	:				
I.	6	5	1	-	6	5	1	-	•				
<b>i</b>	3	3	-	-	3	3	-	-	:				
	1	1	-	-	1	1	-	-	-				
2	2	2	-	-	2	2	-	-	-				
1	4	4	-	-	4	4	-	-	2				
J	6	5	1	-		5	-	-	1				
	1	1	-	-	1	1	-	-	-				
2	1	1	-	-	1	L	-	-	-				
1	3	Z	1	-	د	2	1	-	-				
•	0	Ű	-	-	0	υ	-	-	1				
6	1	1	-	-	1	1	-	-	1				
)	1	1	-	-	1	1	-	-	-				
1	2	2	-	-	2	2	-	1	2				
2	1	1	-	-	1	1	-	-	-				
6	1	1	-	-	1	1	-	-	-				
4	1	-	1	-	1	-	1	-	-				
0	65	15	50	-	9	7	2	-	4				
	372	212	14.9										

TABLE 3-14 NUMBER AND TYPES OF PLANTS AND STREAMS BY AGE

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#### Processes

Another important way to describe the data bases is in terms of processes. Process is a more significant description than product because one product may be made by numerous different processes, each of which uses different raw materials and reaction conditions. Data on products and on processes were taken in combinations which yield what are termed product/processes. A product/process is one product made by a particu-For example, cyclohexanone manufacture by oxidation of lar process. cyclohexane is one product/process, and production of cyclohexanone (the same product) by dehydrogenation of cyclohexanol is a different product/ process. Product/ process designations were given to all products and processes found in the data bases, including products not in the SIC codes of interest. There are approximately 2100 product/processes associated with the 1200 products in the 308 Data Base. The Summary Data Base contains 858 different product/processes. Attempting to relate each individual product/ process with its associated pollutant loading would be nearly impossible, not because the mechanics of such an effort would be difficult, but because the results would be of little value. This is true for two reasons. First, each product/process occurs an average of 2.1 times in the data base. This means that conclusions would have to be drawn on specific product/processes based on very little data. Second, each plant in the data base utilized an average of 6.2 product/processes and each stream contains an average of 4.8 product/processes. The fact that the average plant in the data bases makes so many products means that the end-of-pipe data collected on each plant will be a combination of the pollutant loads from all product/ processes existing at that plant. Attempting to relate the end-of-pipe data to any one of the processes present cannot be done since all of the product/process data are for total end-of-pipe discharges.

Two methods have been used to make the study of processes more useful: complexity and generic product/processes.

<u>Complexity</u> - Plant complexity is a description of the variety of products and processes represented at each plant. In the OCPS study, complexity is defined as the number of product/processes available at each plant. The distribution of plants and streams among the various numbers of product/processes is indicated in Table 3-15. Plants range in complexity from one to 51 product/processes. Seventy-three percent of the streams have five or less product/processes going to each, while 45 percent of the plants have between twenty and thirty product/processes.

<u>Generic Processes</u> - To make the process information more meaningful, data were developed for generic process groups as described in "Generic Processes." Distributions of the streams among these generic groups are presented in Table 3-16. All of the generic groups in Table 3-9 are represented in the Summary Data Base.

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No. of Product/         Dir         Zer         Unk         All         Dir         Zer         Unk         All         Dir         Zer         Unk.         Dat           1         65         31         33         1         42         21         20         1           2         43         27         15         1         26         15         11         -           3         55         30         24         1         22         14         8         -         1           4         25         19         6         -         8         5         3         -         -           6         17         10         7         -         3         2         - <th></th> <th></th> <th>St</th> <th>reams</th> <th></th> <th></th> <th>Plant</th> <th>3</th> <th></th> <th></th>			St	reams			Plant	3		
Processes         All         Dir         Zer         Unk         All         Dir         Zer         Unk.         Da           1         65         31         33         1         42         21         20         1           2         43         27         15         1         26         15         11         -           3         55         30         24         1         22         14         8         -         1           4         25         19         6         -         8         5         3         -         -           5         26         19         7         -         5         3         2         -         -         -           6         17         10         7         -         3         2         -         1         <	No. of Product	1								Daily
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Processes	<u>A11</u>	Dir	Zer	Unk	A11	Dir	Zer	Unk.	Data
2       43       27       15       1       26       15       11       -         3       55       30       24       1       22       14       8       -       1         4       25       19       6       -       8       5       3       -       1         5       26       19       7       -       3       2       1       -       -         6       17       10       7       -       3       2       1       -       -         7       8       6       2       -       1       1       -       -       -         9       8       6       2       -       1       1       -	1	65	31	22	1	42	21	20	1	٦
3       55       30       24       1       20       10       11       11         4       25       19       6       -       8       5       3       -       1         5       26       19       7       -       5       3       2       -       -         6       17       10       7       -       3       2       1       -         7       8       6       2       -       2       2       -       -         9       8       6       2       -       1       1       -       -         10       8       5       3       -       7       5       2       -       1         11       10       8       2       -       10       8       2       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -	2	43	27	15	1	26	15	11	_	<b>כ</b> ר
4       25       19       6       1       1       6       1       7       6       1       7       7       3       2       1       -       7       6       1       7       -       3       2       1       -       -       7       6       1       1       7       -       3       2       1       -       -       7       6       1       1       -       -       -       3       2       1       -       -       -       3       2       -       -       1       -       -       -       3       2       -       -       1       1       -       -       -       3       2       -       1       1       -       -       3       3       -       -       3       3       -       -       3       3       -       -       3       3       -       -       3       3       -       -       3       3       -       -       3       3       -       -       3       3       -       -       1       1       -       -       1       1       1       1       1       1       1       1	3	55	30	24	1	20	14	8	_	10
5       26       19       7       -       5       3       2       -         6       17       10       7       -       3       2       1       -         7       8       6       2       -       2       2       -       -         9       8       6       2       -       1       1       -       -         9       8       6       2       -       1       1       -       -         10       8       5       3       -       7       5       2       -         11       10       8       2       -       10       8       2       -         11       10       8       2       -       10       8       2       -         12       3       3       -       -       4       4       -       -         13       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -       -         16       3       3       -       -       13 <td>4</td> <td>25</td> <td>19</td> <td>6</td> <td>_</td> <td>8</td> <td>5</td> <td>3</td> <td>-</td> <td>10</td>	4	25	19	6	_	8	5	3	-	10
6       17       10       7       -       3       2       1       -         7       8       6       2       -       2       2       -       -         9       8       6       2       -       1       1       -       -         9       8       6       2       -       10       8       2       -       11       -       -         10       8       5       3       -       7       5       2       -       11         11       10       8       2       -       10       8       2       -         11       10       8       2       -       10       8       2       -         13       4       2       2       -       2       1       -       -         14       4       4       -       -       3       3       -       -       -       -       -       -       -       -       -       1       -       -       -       -       1       -       -       -       1       -       -       1       -       -       1       1	5	26	19	7	-	5	3	2	-	2
7       8       6       2       -       2       2       -       -         8       12       8       4       -       1       1       -       -         9       8       6       2       -       1       1       -       -         10       8       5       3       -       7       5       2       -         11       10       8       2       -       10       8       2       -         12       3       3       -       -       3       3       -       -         13       4       2       2       -       4       4       -       -         14       4       4       -       -       4       2       -       -         14       4       4       -       -       4       2       -       -       1       -       -       1       -       -       1       -       -       1       -       -       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1	6	17	10	, 7	-	3	2	1	-	1
8       12       8       4       -       1       1       -       -         9       8       6       2       -       1       1       -       -       -         10       8       5       3       -       7       5       2       -         11       10       8       2       -       10       8       2       -         11       10       8       2       -       10       8       2       -         12       3       3       -       -       3       3       -       -         13       4       2       2       -       4       4       -       -         14       4       4       -       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -       -         18       3       3       -       -       13       12       1       -         20       2       2       -       - </td <td>7</td> <td>8</td> <td>6</td> <td>2</td> <td>-</td> <td>2</td> <td>2</td> <td>-</td> <td>-</td> <td>1</td>	7	8	6	2	-	2	2	-	-	1
9         8         6         2         -         1         1         -         -           10         8         5         3         -         7         5         2         -           11         10         8         2         -         10         8         2         -           12         3         3         -         -         3         3         -         -           13         4         2         2         -         2         1         1         -           14         4         4         -         -         4         2         2         -           16         3         3         -         -         3         3         -         -           17         2         1         1         -         2         1         -         -         1         1         -         -         1         1         -         -         1         1         -         -         1         1         -         -         1         1         -         -         1         1         -         -         1         1         - <td< td=""><td>8</td><td>12</td><td>8</td><td>4</td><td>-</td><td>1</td><td>1</td><td>_</td><td>-</td><td>2</td></td<>	8	12	8	4	-	1	1	_	-	2
10       8       5       3       -       7       5       2       -         11       10       8       2       -       10       8       2       -         12       3       3       -       -       3       3       -       -         13       4       2       2       -       2       1       1       -         14       4       4       -       -       4       4       -       -         14       4       4       -       -       4       2       2       -         16       3       3       -       -       3       3       -       -         16       3       3       -       -       3       3       -       -         16       3       3       -       -       3       3       -       -       -       1       1       -       -       1       1       -       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1       -       -       1       1 <t< td=""><td>9</td><td>8</td><td>6</td><td>2</td><td>-</td><td>1</td><td>ī</td><td>-</td><td>-</td><td>3</td></t<>	9	8	6	2	-	1	ī	-	-	3
11       10       8       2       -       10       8       2       -         12       3       3       -       -       3       3       -       -         13       4       2       2       -       2       1       1       -         14       4       4       -       -       4       4       -       -         15       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       3       3       -       -         20       2       2       -       -       11       6       5       -         21       2       2       -       -       13       12       1       -         23       -       -       13       12       1       -       -       13       13       1         24       1       1       - <td< td=""><td>10</td><td>8</td><td>5</td><td>3</td><td>_</td><td>7</td><td>5</td><td>2</td><td>-</td><td>1</td></td<>	10	8	5	3	_	7	5	2	-	1
12       3       3       -       -       3       3       -       -         13       4       2       2       -       2       1       1       -         14       4       4       -       -       4       4       -       -         15       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       3       3       -       -         20       2       2       -       -       11       6       5       -         21       2       2       -       -       13       10       3       -         22       2       2       -       -       13       10       3       -         23       -       -       13       10       3       -       -       -       -       -       -       -       -       -       -       - <td>11</td> <td>10</td> <td>8</td> <td>2</td> <td>-</td> <td>10</td> <td>8</td> <td>2</td> <td>-</td> <td>1</td>	11	10	8	2	-	10	8	2	-	1
13       4       2       2       -       2       1       1       -         14       4       4       -       -       4       4       -       -         15       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       4       4       -       -         20       2       2       -       -       11       6       5       -       -         21       2       2       -       -       13       12       1       -         23       -       27       13       13       1       -	12		3	_	-	3	3	_	-	1
14       4       4       -       -       4       4       -       -         15       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       4       4       -       -         20       2       2       -       -       11       6       5       -         21       2       2       -       -       13       12       1       -         23       -       -       13       10       3       -       -       -         23       -       -       13       10       3       -       -       -         24       1       1       -       -       13       10       3       -       -         26       2       2       -       -       16       10       6       -       -         31       1       1       -       <	13	4	2	2	-	2	1	1	-	-
15       4       2       2       -       4       2       2       -         16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       3       3       -       -         20       2       2       -       -       1       6       5       -         21       2       2       -       -       13       12       1       -         23       2       2       -       -       13       13       1       -         24       1       1       -       -       13       10       3       -       -         25       1       1       -       -       16       10       6       -         27       2       2       -       -       16       10       6       -         28       1       1       -       -       1       1       -       -       -         31       1       1       - <t< td=""><td>14</td><td>4</td><td>4</td><td>· _</td><td>_</td><td>4</td><td>4</td><td>· _</td><td>-</td><td>7.</td></t<>	14	4	4	· _	_	4	4	· _	-	7.
16       3       3       -       -       3       3       -       -         17       2       1       1       -       2       1       1       -         18       3       3       -       -       3       3       -       -         20       2       2       -       -       1       6       5       -         20       2       2       -       -       11       6       5       -         21       2       2       -       -       13       12       1       -         21       2       2       -       -       13       12       1       -         22       2       2       -       -       13       10       3       -         23       -       -       13       10       3       -       -       -         26       2       2       -       -       16       10       6       -         27       -       6       5       1       -       -       -       -       -       -       -         31       1       1       <	15	4	2	2	-	4	2	2	-	4 1
17 $2$ $1$ $1$ $ 2$ $1$ $1$ $ 18$ $3$ $3$ $  3$ $3$ $  20$ $2$ $2$ $2$ $  4$ $4$ $  21$ $2$ $2$ $  11$ $6$ $5$ $ 22$ $2$ $2$ $  13$ $12$ $1$ $ 23$ $2$ $2$ $  13$ $10$ $3$ $ 24$ $1$ $1$ $  13$ $10$ $3$ $  25$ $1$ $1$ $  13$ $10$ $3$ $  23$ $18$ $5$ $   2$ $1$ $1$ $  11$ $8$ $3$ $   1$ $1$ $  1$ $1$ <th< td=""><td>16</td><td>3</td><td>3</td><td>_</td><td>-</td><td>3</td><td>3</td><td>_</td><td>-</td><td>1</td></th<>	16	3	3	_	-	3	3	_	-	1
18       3       3       -       -       3       3       -       -         20       2       2       -       -       1       6       5       -         21       2       2       -       -       11       6       5       -         22       2       2       -       -       13       12       1       -         23       -       -       13       10       3       -       -       -         24       1       1       -       -       13       10       3       -       -         25       1       1       -       -       23       18       5       -       -         26       2       2       -       -       16       10       6       -       -         27       -       6       5       1       -	17	2	1	1	-	2	1	1	-	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	18	3	3	_	-	3	3	_	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	20	2	2	-	-	4	4	-	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	21	2	2		-	11	6	5	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	22	2	2	-	-	13	12	1	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	23					27	13	13	1	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	24	1	1	-	-	13	10	3	-	-
26 $2$ $2$ $  16$ $10$ $6$ $ 27$ $6$ $5$ $1$ $  11$ $8$ $3$ $ 28$ $11$ $8$ $3$ $ 9$ $6$ $3$ $ 28$ $29$ $1$ $1$ $  1$ $1$ $  33$ $  33$ $  33$ $  33$ $1$ $1$ $  1$ $1$ $  33$ $  33$ $1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $  1$ $1$ $-$ <	25	1	1	-	-	23	18	5	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	26	2	2	-	-	16	10	6	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	27					6	5	1	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	28					11	8	3		-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	2 <b>9</b>	1	1	-	-	9	6	3	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	31	1	1	-	-	1	1	-	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	33	1	1	-	-	1	1	-	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	34					1	1	-	-	_
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	39	1	-	1	-	1	-	1	-	-
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	41	1	1	-	-	1	1	-	-	1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	45	. 1	1	-	-	1	1	-	-	1
Not Reported $60 9 51 - 3 3$	51	1	1	-	-	1	1	-	-	-
	Not Reported	60	9	51		3	3	-	-	1
TOTAL C 277 212 162 2 201 105 04 9 5										
	TOTALS	377	212	162	3	291	195	94	2	50

### TABLE 3-16

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### OCCURRENCES OF GENERIC PROCESSES

				Occurrences
	Occurrences	Occurrences in	Occurrences	in Unknown
Generic Class*	in the	Direct Discharge	Zero Discharge	Discharge
Codes	Sum. Data Base	Streams	Streams	Streams
A	63	48	15	-
В	99	85	14	-
С	139	102	37	-
D1	378	231	144	3
D2	140	121	19	-
E	51	46	5	-
Fl	37	30	7	-
F2	16	14	2	-
G	117	92	25	-
Н	71	57	14	-
I	6	5	1	-
<b>I</b> ]	3	3	-	-
12	17	17	-	-
J	2	2	-	-
<b>J</b> 1	10	10	-	-
J2	19	15	4	-
К	26	22	4	-
L	24	24	-	-
М	35	32	. 3 .	. –
N	5	3	2	-
0	42	32	10	-
P	35	33	2	-
Q	10	7	3	-
R	18	16	2	-
5	6	6	-	-
Т	1	1	-	-
U	5	5	-	-
V	16	16	-	-
Z	6	6	-	-
12	20	17	3	-
13	1	-	1	-
14	9	9	-	-
15	27	23	4	-
16	6	3	3	-
17	8	8	-	-
18	4	3	1	-
19	1	l	-	-
2	48	43	5	-
20	8	8	<b>-</b>	
21	4	2	2	-
22	1	1	-	-
23	18	17	1	-
24	9	2	. 7	-

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## TABLE 3-16 (Continued)

Generic Class* Codes	Occurrences in the Sum. Data Base	Occurrences in Direct Discharge Streams	Occurrences Zero Discharge Streams	Occurrences in Unknown Discharge Streams
3	203	150	52	· 1 .
4	12	6	6	-
5	63	54	9	-
6	20	16	4	-
7	2	2	_ '	-
8	4	4	-	-
9	1	1	-	-
TOTAL	1866	1451	411	4

### OCCURRENCES OF GENERIC PROCESSES

\* For description of codes see Table 3-9.
#### SECTION IV.

#### SUBCATEGORIZATION

#### INTRODUCTION

Sections 304(b)(1)(B) and 304(b)(4)(B) of the Clean Water Act require EPA to assess certain factors in establishing effluent limitations guidelines based on the best practicable control technology (BPT) and best conventional pollutant control technology (BCT). These factors include the age of equipment and facilities involved, the manufacturing process employed, the engineering aspects of the application of recommended control technologies including process changes and in-plant controls, non-water quality environmental impacts including energy requirements and other factors as determined by the Administrator.

To accommodate these factors, it may be necessary to divide a major industry into a number of unique and homogeneous groups or subcategories. This allows the establishment of uniform national effluent limitations guidelines and standards while at the same time accounting for the individual characteristics of different groups of facilities.

The factors considered in the subcategorization of the Organic Chemicals and Plastics and Synthetics Point Source Categories (OCPS) include:

- 1. Facility Size
- 2. Geographical Location
- 3. Age of Facility and Equipment
- 4. Raw Wastewater Characteristics
- 5. Treatability
- 6. Raw Materials
- 7. Manufacturing Product/Processes
- 8. Nonwater Quality Environmental Impacts
- 9. Energy Requirements

The impacts of these factors have been evaluated to determine if subcategorization is necessary or feasible. These evaluations are discussed in detail in the following sections.

#### STATISTICAL METHODOLOGY

Two major statistical techniques were used to determine an appropriate subcategorization scheme for the OCPS industry: the Spearman Rank Correlation [4-1] and the Terry-Hoeffding Test.[4-2] Both techniques are non-parametric, thus making the fewest assumptions about the nature of the underlying data.

The Spearman Rank Correlation was used to determine the existence of any relationships among the factors which must be considered for subcategorization of the OCPS industry. A detailed explanation of the Spearman Rank Correlation technique and an example of its use are presented in Appendix B. The Terry-Hoeffding test was used to test whether two populations of plants differ in terms of median levels of a parameter of interest (e.g., median influent BOD concentration). If the test indicates that two groups of plants are different, then the groups could represent a basis for subcategorization. A detailed explanation of the Terry-Hoeffding test and an example of its use are presented in Appendix B.

#### TECHNICAL METHODOLOGY

All nine factors mentioned previously were examined for technical significance in the development of the proposed subcategorization scheme. However, in general, the proposed subcategorization is based primarily on significant differences in raw waste characteristics, since many of the other eight factors could not be examined in appropriate technical and statistical depth due to the intricacies of the data base. Therefore, variations in raw waste characteristics were utilized to evaluate the impact of the other eight factors on subcategorization. For example, the ideal data base for evaluating the need for subcategorization and the development of individual subcategories would include raw wastewater and final effluent pollutant data for facilities which employ only one generic manufacturing process or multiple product plants which segregate and treat each process raw waste stream separately. In this manner, each factor could be evaluated independently. Specifically, to evaluate the significance of facility size, the ideal data base would contain fifty or more plants using only one generic process and all varying in size (i.e., production rates of 10 kilograms per year to 1,000,000 kilograms per year). In addition, all 50 plants would be located in one geographic region and be of the same age. In this manner, the effects of size would not be masked or enhanced by the effects of geographic location or plant age. Therefore, to evaluate each factor ideally, the data base would need to contain plants that would allow isolation of each of the factors as described above for size.

However, the available information consists of historical data collected by individual companies primarily for the purpose of monitoring the performance of end-of-pipe wastewater treatment technology and compliance with NPDES permit limitations. The OCPS Industry is primarily comprised of multi-product/process, integrated facilities. Wastewaters generated from each product/process are collected in combined plant sewer systems and treated in one main treatment facility. Therefore, each plant's overall raw wastewater characteristics are affected by all of the production processes occurring at the site at one time. The effects of each production operation on the raw wastewater characteristics cannot be isolated accurately from all of the other site specific factors. Therefore, a combination of both technical and statistical methodologies had to be used to evaluate the significance of each of the subcategorization factors. In the methodology that was employed, the results of the technical analysis were compared to the results of the statistical efforts to determine the usefulness of each factor as a basis for subcategorization. The combined technical/statistical evaluations of the nine factors are presented below.

#### RAW WASTEWATER CHARACTERISTICS

Raw wastewater load (RWL) was selected as the dependent variable to be used to evaluate the significance of all of the subcategorization factors discussed in this section. RWL for the purposes of subcategorization is a measure of flow, BOD and TSS and was used as the basis for comparison to the other eight subcategorization factors.

Flow, for the purpose of this report, is measured in million gallons per day (MGD), and includes only process wastewater. This includes contact cooling waters, vacuum jet waters, wash waters, reaction media and contact steam. Wastewater flow does not include storm water, non-contact cooling water and sanitary wastewaters. Wastewater flow can be affected by facility size, efficiency of water use, methods of production (e.g., solvent or aqueous based), methods of cooling and vacuum generation, as well as other factors.

BOD is a measure of the wastewater's organic content (see Section V). Plants that use highly soluble organic materials, or use contact waters extensively, usually have higher BOD loadings than plants that use dry process techniques or solvent based reactions.

TSS is a measure of both organic and inorganic solid materials (see Section V). It is a measure of the insoluble phase of the wastewater. Higher TSS values can be associated with precipitation products, wash waters, contaminated storm water, as well as other sources.

#### MANUFACTURING PRODUCT/PROCESSES

Because this rulemaking involves the combination of two industries, (Organic Chemicals and Plastics & Synthetic Materials), an initial subcategorization involving the following broad industry segments was selected:

- 1. Plants manufacturing only plastics and synthetic materials
- 2. Plants manufacturing only organic chemicals
- 3. Plants manufacturing both organic chemicals and plastic and synthetic materials at the same facility.

Due to the nature of the raw materials and production processes, organic chemicals plants would be expected to have higher raw waste loads than plants manufacturing only plastics and synthetic materials, with combined organics and plastics plants lying between these two groups. This is confirmed in Figure 4-1, which shows the cumulative distribution of raw waste BOD for the three initial industry segments: Plastics Only, Organics Only, and Combined Plastics and Organics. Figure 4-2 presents the least squares fit of the data shown in Figure 4-1. As shown in these figures, the points generated from plotting the three cumulative distributions on log probability scale show the Plastics Only plants have considerably lower raw waste loads than the other two groups. This is further substantiated by the application of the Terry-Hoeffding Test for raw waste BOD for the groups Plastics Only and Not Plastics Only (combined groups 2 and 3). The test statistic T = 3.765 for a sample size of 123 yielded a probability level of zero. A level less than 0.05



FIGURE 4-1 - PROBABILITY PLOT OF SUBCATEGORY INFLUENT BOD



FIGURE 4-2 - LEAST SQUARES FIT OF FIGURE 4-1

is considered significant. Thus, Plastics Only is significantly different from the other two groups in terms of BOD. (There was no significant difference in the groups for TSS.)

The other two groups offer some statistical analysis problems. As shown in Figures 4-1 and 4-2, the Organics Only and Plastics and Organics groups yield cumulative curves which intersect, indicating an interrelationship between the groupings' organics contributions. Unfortunately, due to data deficiencies, it was not possible to proportion the individual organics and plastics raw waste contributions for combined organics and plastics producers by flow or production. As a result, another approach based on the product/process chemistry exhibited by plants in the combined groups (Not Plastics Only) was examined.

As detailed in Section III, the OCPS industry produces thousands of organic chemical products and in many cases, one product can be produced by a number of processes. Therefore, subcategorizing by specific product would result in an unmanageable number of subcategories. However, since BPT regulations will limit such broad based pollutant parameters as BOD, TSS, and pH, subcategorizing by type of production process and their tendencies to produce high or low quantities of these pollutants can result in a manageable, yet appropriate method of subcategorization. In general, certain production factors may affect the concentration of BOD or TSS in the raw wastewater generated by an OCPS industry facility. Factors that might contribute to a relatively higher BOD or TSS loading include: the use of aqueous reaction media that may require subsequent disposal, the general yield of the process (if a process does not retain a high percentage of reaction products, and instead product and reactant find their way to the waste stream, a relatively higher raw waste load may be observed), the absence of toxic materials in the raw wastewater that might inhibit the BOD test procedure, and the use of vacuum jet water, steam ejector condensate or contact cooling waters and their discharge to the process sewer.

Also contributing to relatively higher BOD or TSS raw waste characteristics is the use of raw materials or the manufacture of products that contain oxygen, nitrogen, or phosphorous. Generic processes which generate oxygenated by-products may be expected to produce wastewaters which are more biodegradable, and thus exert a higher biological oxygen demand than process wastewaters which do not. This is because enzymatic catabolic pathways generally follow a sequence of hydroxylation and subsequent oxidation to keto or carboxylic acid derivatives; and the greater the degree of oxidation (whether chemically or biologically induced) the shorter the pathway to ultimate biodegradation as well as a greater choice of existing catabolic enzymatic pathways. The generic process of oxidation therefore would be expected to generate wastewaters that exert a relatively high biochemical oxygen demand.

An intermediate 5-day biochemical oxygen demand may be expected for chemical species which occupy an intermediate position in metabolic pathways (i.e., compounds which require scission of bonds other than carbon-hydrogen). Substituted amines and similar nitrogen containing species, for example, are generally biodegradable, although at rates somewhat less than those of oxygen containing species. Processes that

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generate wastewaters containing nitrogen in a reduced form (amines, oximes, nitriles, etc.), or compounds that require scission of a carbon-oxygen bond prior to oxidative degradation (ethers), are predicted to exert an intermediate 5-day biochemical oxygen demand.

Other generic processes may be expected to generate wastewaters of relatively low biochemical oxygen demand for one of two reasons:

- 1. refractory chemical species predominate in the wastewater, or
- 2. relatively few chemical species are present in the wastewater

Generic processes such as nitration and sulfonation generate wastewaters of a refractory nature and, therefore, exert a low 5-day biochemical oxygen demand for the first reason. Other less refractory potential raw materials and products associated with the OCPS industry include aromatics, primary aliphatics, PCBs and halo-ethers. Generic processes producing high flow wastewaters that contain relatively few chemical species (for example, most polymerization processes) may also be expected to exert a low biochemical oxygen demand. Based on the presence or absence of species in industry wastewaters, and the generic process factors described above, it is reasonable to attempt to aggregate generic process wastewaters by their biodegradability on theoretical grounds. Table 4-1 summarizes expected 5-day biochemical oxygen demand by generic process group.

Type I includes those generic manufacturing processes expected to have high BOD values, whereas Type IV processes are expected to generate wastewaters lower in BOD. Type II and III processes complete the range of high to low BOD values.

The Terry-Hoeffding test was applied to the above product/process structure. As shown in Table 4-2, Type I vs. Not Type I was the only potential scheme which showed statistically significant differences, with raw waste BOD showing the greatest difference (T = 2.251, P = 0.024). This suggested that the Not Plastics Only initial grouping be further divided into two subgroups:

- Plants manufacturing organic chemicals only and organic chemicals and plastics and synthetic materials in the same facility and which employ Type I generic chemical processes whether or not other Type processes are used at the facility (Not Plastics Only - Type I)
- Plants manufacturing organic chemicals only and organic chemicals and plastics and synthetic materials in the same facility and which do not employ Type I generic chemical processes (Not Plastics Only - Not Type I)

Upon further engineering analysis of the production process chemistry, it was believed that the oxidation generic product/process segment, which is part of Type I, contributed higher raw waste BOD loadings than any of the other processes. Thus, it was decided to further subcategorize by the presence or absence of the oxidation generic product/process. This decision was statistically confirmed by applying the Terry-

## TABLE 4-1

## EXPECTED 5-DAY BIOCHEMICAL OXYGEN DEMAND BY GENERIC PROCESS GROUP

TYPE I - High 5-Day Biochemical Oxygen Demand

Oxidation	Hydration
Peroxidation	Alkoxylation
Acid Cleavage	Hydrolysis
Condensation	Carbonylation
Isomerization (maleic → fumaric acid)	Hydrogenation (butyraldehyde →
Esterification	Neutralization
Hydroacetylation	

## TYPE II - Intermediate 5-Day Biochemical Oxygen Demand

Electrohydrodimerization	
Cyanation/Hydrocyanation	
Epoxidation (unsat'd esters)	
Etherification (alkycellulose)	
Polymerization (condensation)	

TYPE III - Lower 5-Day Biochemical Oxygen Demand

Alkylation (phenol →	Dehydrogenation (isobutanol >
nonyl phenol)	acetone)
Hydrogenation	Sulfonation
(nitrobenzene > aniline)	Nitration

# TYPE IV - Lowest 5-Day Biochemical Oxygen Demand

Alkylation (phenol >	Polymerization (bulk & addition)		
nonyl phenol)	Fiber production		
Hydrodealkylation	Halogenation		
Isomerization	Oxyhalogenation		
Pyrolysis (steam)	Hydrohalogenation		
Cracking (catalytic)	Dehydrohalogenation		
Dehydrogenation (ethyl	$(1,2-dicholorethane \rightarrow vinyl Cl.)$		
benzene > styrene)	Chlorohydrination		
Distillation	Phosgenation		
Extractive distillation	Extraction		
Crystallization/Distillation			

### TABLE 4-2

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## TERRY-HOEFFDING TEST FOR NOT PLASTICS ONLY PLANTS

## Raw Waste BOD

TYPES	TEST STATISTIC	SAMPLE SIZE	SIGNIFICANCE LEVEL
TYPE I vs. NOT TYPE I	T = 2.251	N = 74	<b>P = .</b> 024
NOT TYPE I BUT TYPE II vs. NOT TYPE I OR TYPE I	T = .710 I	N = 11	P = .478

Raw Waste TSS

TYPE I vs. NOT TYPE I	T = .784	N = 47	P = .433
NOT TYPE I BUT TYPE II vs. Not type I or type II	<b>T = .56</b> 0	N = 13	P = .576

Note: See Table 4-1 for definition of Type I and II

Hoeffding Test to Type I With Oxidation versus Type I Without Oxidation. The test statistic was T = 2.706, sample size n = 63 and the significance level was P = 0.007. For TSS no significant differences were found.

The Terry-Hoeffding test was also used to investigate the applicability of the four subcategories to the parameters COD and TOC. The test results are shown in Table 4-3. For COD, significant differences were found between Plastics Only and Not Plastics Only plants, and between Type I and Not Type I plants in the Not Plastics Only group. For TOC, no significant differences were found. Based on these results it appears that the four subcategories are compatible with the COD and TOC data, and prior to considering the other factors listed previously, the initial subcategorization is:

- 1. Plants manufacturing only plastics and synthetic materials (Plastics Only).
- 2. Plants manufacturing organic chemicals only, organic chemicals and plastics and synthetic materials in the same facility, and other SIC code products which commingle their wastewater with the above OCPS wastewaters and employ Type I generic chemical processes including oxidation (Not Plastics Only - Type I With Oxidation).
- 3. Plants manufacturing organic chemicals only, organic chemicals and plastics and synthetic materials in the same facility, and other SIC code products which commingle their wastewater with the above OCPS wastewaters and employ Type I generic chemical processes, but do not include oxidation (Not Plastics Only -Type I Without Oxidation).
- 4. Plants manufacturing organic chemicals only, organic chemicals and plastic and synthetic materials in the same facility, and other SIC code products which commingle their wastewaters with the above OCPS wastewaters and employ Type II, III or IV generic chemical processes (Not Plastics Only - Not Type I).

#### FACILITY SIZE

Although sales volume, number of employees, area of plant site, plant capacity and production rate might logically be considered to define facility size, none of these factors completely describes size in a satisfactory manner. Section III discusses the elimination of each of these factors as an adequate definition of facility size. Specifically, measuring a facility's size by using the sum of its production quantities does not account for all characteristics encompassed in plant size. For example, Plant A may have a relatively fixed market for a given product and therefore manufactures this product with dedicated equipment, 24 hours per day, 365 days per year. However, Plant B, which lists its production rate as identical to Plant A, may manufacture the same product on a specification basis in six to eight weekly campaigns

<sup>(\*)</sup> See Section III, pp. 21 to 28.

## TABLE 4-3

## TERRY-HOEFFDING TEST FOR SUBCATEGORIZATION BASED ON COD AND TOC

## RAW WASTE COD

Category	Test Statistics	Sample Size	Significance Level
Plastics vs. Not Plastics Only	3.516	107	0.000
Not Plastics Only			
Type I vs. Not Type I	2.114	62	0.034
Type I and C vs. Type I and Not C*	1.347	49	0.178

RA	AW WASTE TOC		
Category	Test Statistics	Sample Size	Significance Level
Plastics vs. Not Plastics Only	0.738	48	0.461
Not Plastics Only			
Type I vs. Not Type I	1.205	40	0.228
Type I and C vs. Type I and Not C*	0.576	32	0.564

\* Type I w/oxidation vs. Type I w/o oxidation

per year. In addition, Plant A has invested R & D funds in this production process and has developed continuous production methods, while Plant B still utilizes batch production techniques. Therefore, although products are produced in the same annual quantities, Plant B will most likely have higher strength wastes due to less efficient production (lower yields) and much higher variability due to the campaign aspect of its operation. A statistical evaluation of size as defined by production also confirms that size is not a factor for subcategorization. In the Summary Data Base, the only production data available is on an indi-Table 4-4 presents the number of waste vidual waste stream basis. streams per facility size or production rate grouping for each of the initial subcategories. Table 4-5 and Figures 4-3 through 4-10 present correlations ranging from -0.3 to +0.19 for raw waste BOD and TSS in the initial subcategorization scheme. In all cases, the null hypothesis (Ho) is accepted; that is, raw waste BOD and TSS is independent of size as defined by production rate in pounds per day.

Therefore, there is no adequate method to define facility size, and it cannot be used as a technical basis for subcategorization. In addition, using production as an indication of facility size (because of data availability), as defined by production, was not a statistically significant factor for subcategorization.

#### GEOGRAPHICAL LOCATION

Companies in the OCPS industry usually locate their plants based on a number of factors. These include:

- 1. Sources of raw materials
- 2. Proximity of markets for products
- 3. Availability of an adequate water supply
- 4. Cheap sources of energy
- 5. Proximity to proper modes of transportation
- 6. Reasonably priced labor markets

In addition, a particular product/process may be located in an existing facility based on availability of certain types of equipment or land for expansion. Companies also locate their facilities based on the type of production involved. For example, specialty producers may be located closer to their major markets, whereas bulk producers may be centrally located to service a wide variety of markets. Also, a company may locate its plants based on its planned method of wastewater disposal. A company which has committed itself to zero discharge as its method of wastewater disposal has the ability to locate anywhere, while direct dischargers must locate near receiving waters, and indirect dischargers must locate in a city or town which has an adequate POTW capacity to treat OCPS wastewaters.

Because of the complexity and interrelationships of the factors outlined above affecting plant locations, no clear basis for subcategorization according to the plant location could be found. Therefore, location is not a basis for subcategorization of the OCPS industry.

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## TABLE 4-4

## (DIRECT SYSTEMS) NUMBER OF WASTE STREAMS WITHIN PRODUCTION RATE RANGES

(#/day)	A11	Plastics	Not Plastics	Not Plastics	Not Plastics
Size	Streams	Only	Туре 1 6 С*	Type I Not C**	Not Type I
1-9,999	1	0	0	1	0
10,000-49,999	5	2	0	2	1
50,000-99,999	7	5	1	1	0
100,000-249,999	21	15	2	1	3
250,000-449,999	27	16	5	3	3
450,000-599,999	20	8	7	2	3
600,000-999,999	28	14	7	4	3
1,000,000-1,999,999	33	12	7	6	8
2,000,000-4,999,999	29	1	13	8	7
Over 5,000,000	31	0	19	6	6
Missing	10	4	3	1	2
TOTAL	212	77	64	35	36

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\* Type I w/oxidation\*\* Type I w/o oxidation

## TABLE 4-5

## SPEARMAN CORRELATION COEFFICIENTS (R) FOR RAW WASTE BOD AND TSS vs. SIZE (Production Rate)

	A11	All Plastics Not Plastics Only		Not Plastics Only	
	Plants	Only	I w/oxidation	I w/o oxidation	No Group I
BOD	0.21	0.06	-0.12	-0.02	-0.31
	(0.02)	(N.S.)	(N.S.)	(N.S.)	(N.S.)
TSS	0.02	-0.00	-0.13	-0.42	-0.20
	(N.S.)	(N.S.)	(N.S.)	(N.S.)	(N.S.)

Note: Results of testing (Ho: C=O) are indicated as:

a) N.S. - Not significantly different from zero (P>.05) b) <.01 - Significantly didfferent from zero (P<.01) c) Actual probability (.01 <P<.05)

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FIGURE 4-4





FIGURE 4-5

\* Type I w/oxidation



SPEARMAN CORRELATION COEFFICIENT = .06 (P = .67, N = 49)







\* Type I w/o oxidation

RANK CORRELATION FOR THE PLASTIC & ORGANIC CHEMICAL INDUSTRIES DIRECT DISCHARGE SYSTEMS NOT PLASTICS ONLY/TYPE I & C\* TSS 20 Ô ð \*\*\* **SIZE** (Production Rate) SPEARMAN CORRELATION COEFFICIENT = -0.13 (P = .58, N = 20)



\* Type I w/oxidation



SPEARMAN CORRELATION COEFFICIENT = -0.00 (P = .90, N = 42)

The effects of temperature, which is related to geographical location, are discussed in detail in Section VII, Control and Treatment Technology. It is concluded in Section VII that the effects of temperature are inconclusive.

#### AGE OF FACILITY AND EQUIPMENT

The age of an OCPS plant is difficult to accurately define. This is because production facilities are continually modified to meet production goals and to accommodate new product lines. Therefore, actual process equipment is generally modern (i.e., 0-15 years old). However, major building structures and plant sewers are not generally upgraded unless the plant expands significantly. Facility age, for the purposes of this report, and as reported in the 308 Questionnaire, is defined as the oldest process in operation at the site. Table 4-6 presents the number of waste streams per age grouping for each of the initial subcategories.

Older plants may use open sewers and drainage ditches to collect process wastewater. In addition, cooling waters, steam condensates, wash waters, and tank drainage waters are generally collected in these drains due to their convenience and lack of other collection alternatives. These ditches may run inside the process buildings as well as between manufacturing centers. Therefore, older facilities are likely to exhibit higher wastewater discharge flow rates than newer facilities. In addition, since the higher flows may result from the inclusion of relatively clean noncontact cooling waters and steam condensates as well as infiltration/inflow, raw wastewater concentrations may be lower due to dilution effects. Furthermore, recycle techniques and wastewater segregation efforts normally cannot be accomplished with existing piping systems, and would require the installation of new collection lines as well as the isolation of the existing collection ditches. However, due to water conservation measures as well as ground contamination control, many older plants are upgrading their collection systems. In addition, the energy crisis of recent years has caused many plants to upgrade their steam and cooling systems to make them more efficient.

Figures 4-11 through 4-18 present BOD and TSS raw waste rank correlations versus facility age for each of the initial subcategories. A11 rank correlations shown in Table 4-7 show no clear trend. The only apparent correlation appears for the Not Plastics Only-Type I With Oxidation group with a rank correlation of R = -0.49 and P = < 0.01 for raw This negative correlation reinforces the argument waste BOD and age. that higher raw waste levels in newer plants can be attributed to more rigorous modern water conservation techniques. This is again supported by raw waste flow versus age rank correlations for the Not Plastics Only-Type I With Oxidation group, which shows a rank correlation of R =0.5 and P = 0.0001. Thus, older plants within the same grouping tend to have higher flows which dilute the strength of their raw wastewaters. Therefore: (1) a plant's age for the purposes of regulation would be difficult to accurately measure, and (2) the relationship between facility age and RWL characteristics is greatly affected by many external factors, eliminating facility age as a feasible basis for subcategori-

# TABLE 4-6

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(Years) Age	All Streams	Plastics Only	Not Plastics Type I & C <sup>*</sup>	Not Plastics Type I Not C**	Not Plastics Not Type I
1-5	12	. 9	1	1	1
6-9	20	7	10	0	3
10-12	25	7	9	5	4
13-15	28	8	8	6	6
16-18	18	5	4	6	3
19-20	12	6	2	1	3
21-23	21	7	10	2	2
24-30	26	10	5	4	7
31-40	25	8	10	3	4
41-0ver	10	4	2	2	2
Missing	15	6	3	5	1
TOTAL	212	77	64	35	36

# (DIRECT SYSTEMS) NUMBER OF STREAMS PER AGE GROUP

Type I w/Oxidation
 II Type I w/o Oxidation

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\* Type I w/oxidation



AGE

SPEARMAN CORRELATION COEFFICIENT = 0.01 (P = .96, N = 18)  $\frac{7}{7}$ 



\* Type I w/o oxidation

FIGURE 4-14









\* Type I w/oxidation







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## TABLE 4-7

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### SPEARMAN CORRELATION COEFFICIENTS (R) FOR RAW WASTE BOD AND TSS vs. AGE

	A11	<b>Plastics</b>		Not Plastics Only	
	Plants	Only	I w/oxidation	I w/o oxidation	No Group I
BOD	-0.19	-0.19	-0.49	0.01	0.06
	(0.04)	(N.S.)	.00	(N.S.)	(N.S.)
TSS	-0.13	-0.18	-0.35	-0.08	0.12
	(N.S.)	(N.S.)	(N.S.)	(N.S.)	(N.S.)

Note: Results of testing (Ho: C=O) are indicated as:

a) N.S. - Not significantly different from zero (P>.05)

b) <.01 - Significantly different from zero (P<.01)</li>
c) Actual probability (.01<P<.05)</li>

zation. Nevertheless, because there is a negative correlation between RWL and age in the Not Plastics Only - Type I With Oxidation group, age and the subsequent impact of water usage may be important in this group. Therefore, evaluation of this phenomenon to accommodate these statistically significant factors based on water usage within this particular subcategory, is appropriate and is discussed in detail in Section VII.

#### RAW MATERIALS

Synthetic organic chemicals can be defined as derivative products of naturally occurring materials (petroleum, natural gas and coal) which have undergone at least one chemical reaction such as oxidation, hydrogenation, halogenation or alkylation. This definition, when applied to the larger number of potential starting materials and the host of chemical reactions which can be applied, leads to the possibility of many thousands of organic chemical compounds being produced by a potentially large number of basic processes having many variations. There are more than 25,000 commercial organic chemical products derived principally from petrochemical sources. These are produced from five major raw material classifications: methane, ethylene, propylene, C, hydrocarbons, and higher aliphatics and aromatics. This major raw materials list can be expanded by further defining the aromatics to include benzene, toluene and xylene. These raw materials are derived from natural gas and petroleum, although a small portion of the aromatics are derived from Currently, approximately 90 percent by weight of the organic coal. chemicals used in the world are derived from petroleum or natural gas. Other sources of raw materials are coal and some naturally-occuring renewable material of which fats, oils and carbohydrates are the most important. The third source also includes more obscure natural products (consisting of small quantities of very specialized chemicals) which contribute to highly specialized segments of the industry.

Regardless of the relatively limited number of basic raw materials utilized by the organic chemicals industry, process technologies lead to the formation of a wide variety of products and intermediates, many of which can be produced from more than one basic raw material either as a primary reaction product or as a by-product. Furthermore, primary reaction products are frequently processed to other chemicals which categorize the primary product from one process as the raw material for a subsequent process.

Delineation between raw materials and products is nebulous at best, since the product from one manufacturer can be the raw material for another manufacturer. This lack of distinction is more pronounced as the process approaches the ultimate end product, which is normally the fabrication or consumer stage. Also, many products/intermediates can be made from more than one raw material. Frequently, there are alternate processes by which a product can be made from the same basic raw material.

Another characteristic of the OCPS industry which makes subcategorization by raw material difficult is the high degree of integration in manufacturing units. Since the bulk of the basic raw materials are derived from petroleum or natural gas, many of the organic chemical manufacturing plants are either incorporated into or continguous to petroleum refineries, and may formulate a product at almost any point in a process from any or all of the basic raw materials. Normally, relatively few organic manufacturing facilities are single product/process plants unless the final product is near the fabrication or consumer product stage.

Because of the integrated complexity of the largest (by weight) single segment of the organics industry (petrochemicals), it may be concluded that subcategorization by raw materials is not feasible for the following reasons:

- 1. The organic chemicals industry is made up primarily of chemical complexes of various sizes and complexity.
- 2. Very little, if any, of the total production is represented by single raw material plants.
- 3. The raw materials used by a plant can be varied widely over short time spans.
- 4. The conventional and nonconventional wastewater pollutant parameter data gathered for this study were not collected on a raw materials orientation, but rather represent the mixed end-ofpipe plant wastewaters.

#### TREATABILITY

Treatability of OCPS wastewaters is discussed in great detail in Section VII. The treatability of a given wastewater is affected by the presence of inhibitory materials (toxics); availability of alternative disposal methods; and pollutant concentrations in, and variability of, the RWL. However, all of these factors can be mitigated by sound waste management, treatment technology design, and operating practices. Examples of these are:

- The presence of toxic materials in the wastewater can be controlled by in-plant treatment methods. Technologies such as steam stripping, metals precipitation, activated carbon, reverse osmosis, etc. can eliminate the presence of materials in a plant's wastewater which may inhibit or upset biological treatment systems.
- o Although many plants utilize deep well injection for disposal of highly toxic wastes to avoid treatment system upsets, other alternative disposal techniques such as contract hauling and incineration are available to facilities which cannot utilize deep well disposal. In addition, stricter groundwater regulations may eliminate the option of deep well disposal for some plants, or make it uneconomical for others, forcing facilities to look more closely at these other options.
- o RWL variability can easily be controlled by the use of equalization basins. In some plants, "at process" storage and equaliza-

tion is used to meter specific process wastewaters, on a controlled basis, into the plant's wastewater treatment system.

 Raw waste concentrations can be reduced with roughing biological filters or with the use of two-stage biological treatment systems. These techniques are discussed in more detail in Section VII.

OCPS wastewaters can be treated by either physical-chemical or biological methods, depending on the pollutant to be removed. Also, depending on the specific composition of the wastewater, any pollutant may be removed to a greater or lesser degree by a technology not designed for removal of this pollutant. For example, a physical-chemical treatment system designed to remove suspended solids will also remove a portion of the BOD of a wastewater if the solids removed are organic and biodegradable. It is common in the OCPS industry to use a combination of technologies adapted to the individual wastewater stream to achieve desired results. These concepts are discussed in detail in Section VII.

In general, the percent removals of BOD and TSS are consistent across all initial subcategories. It is also possible for plants in all initial subcategories to achieve high percent removals (greater than 95%) for both BOD and TSS (data supporting these removals are presented and discussed in Section VII). Therefore, based on the consistency of these removal data and the ability of plants in all initial subcategories to achieve high removals of pollutants, it is concluded that subcategorization based on treatability is not justified.

#### ENERGY AND NON-WATER QUALITY ASPECTS

Energy and non-water quality aspects include the following:

- 1. Sludge production
- 2. Air pollution derived from wastewater generation and treatment
- 3. Energy consumption due to wastewater generation and treatment
- 4. Noise from wastewater treatment

The basic treatment step, used by virtually all plants in all subcategories that generate raw wastes containing basically BOD and TSS, is biological treatment. Therefore, the generation of sludges, air pollution, noise and the consumption of energy will be homogeneous across the industry. However, the levels of these factors will relate to the volume of wastewater treated and their associated pollutant loads. Since the volumes of wastewater generated and the RWL from each pollutant were considered in earlier sections, it is believed that all energy and nonwater quality aspects have been adequately addressed in the proposed subcategorization scheme.
## SUMMARY - SUBCATEGORIZATION

Based on the preceding technical and statistical evaluation of the OCPS industry, four subcategories have been established. These subcategories are as follows:

## Subcategory 1 - Plastics Only

Discharges resulting from the manufacture of plastics and synthetic fibers only.

## Subcategory 2 - Oxidation

Discharges resulting from the manufacture of organic chemicals only, or both organic chemicals and plastics and synthetic fibers, that include wastewater from the oxidation process.

## Subcategory 3 - Type I

Discharges resulting from the manufacture of organic chemials only, or both organic chemicals and plastics and synthetic fibers, that include wastewater from any of the following generic processes (referred to in the BPT Development document as "Type I" processes) but not from the oxidation process:

Peroxidation Acid Cleavage Condensation Isomerization Esterification Hydroacetylation Hydration Alkoxylation Hydrolysis Carbonylation Hydrogenation Neutralization

## Subcategory 4 - Other Discharges

All OCPS discharges not included in Subcategories 1-3.

## SECTION V.

#### SELECTION OF POLLUTANT PARAMETERS

## WASTEWATER PARAMETERS

Specific conventional and nonconventional wastewater parameters were determined to be significant in the Organic Chemicals and Plastics and Synthetic Materials Industries and were selected for evaluation based on: (1) an industry characterization, (2) data collected from sampling efforts, (3) historical data collected from the literature, and (4) data provided by industry questionnaires (308 Portfolio).

Conventional pollutant parameters chosen for evaluation include 5-day biochemical oxygen demand (BOD<sub>5</sub>), total suspended solids (TSS), pH, and oil and grease (O&G). Nonconventional pollutant parameters selected are chemical oxygen demand (COD) and total organic carbon (TOC).

## CONVENTIONAL POLLUTANT PARAMETERS

## 5-day Biochemical Oxygen Demand (BOD<sub>c</sub>)

The 5-day Biochemical Oxygen Demand (BOD<sub>5</sub>) test traditionally has been used to determine the strength of domestic and industrial wastewaters. It is a measure of the oxygen required by biological organisms to assimilate the biodegradable portion of a waste under aerobic conditions. [5-1] Substances that may contribute to the BOD include carbonaceous materials usable as a food source by aerobic organisms; oxidizable nitrogen derived from organic nitrogen compounds, ammonia and nitrites that are oxidized by specific bacteria; and chemically oxidizable materials such as ferrous compounds, sulfides, sulfite, and similar reducedstate inorganics that will react with dissolved oxygen or that are metabolized by bacteria.

The BOD of a wastewater is a measure of the dissolved oxygen depletion that might be caused by the discharge of that wastewater to a body of water. This depletion reduces the oxygen available to fish, plant life, and other aquatic species. Total exhaustion of the dissolved oxygen in water results in anaerobic conditions, and the subsequent dominance of anaerobic species that can produce undesirable gases such as hydrogen sulfide and methanol. The reduction of dissolved oxygen can be detrimental to fish populations, fish growth rates, and organisms used as fish food. A total lack of oxygen can result in the death of all aerobic aquatic inhabitants in the affected area.

The BOD<sub>5</sub> (5-day BOD) test is widely used to estimate the oxygen demand of domestic and industrial wastes and to evaluate the performance of waste treatment facilities. The test is widely used for measuring potential pollution since no other test methods have been developed that are as suitable or as widely accepted for evaluating the deoxygenation effect of a waste on a receiving water body.

The BOD test measures the weight of dissolved oxygen utilized by microorganisms as they oxidize or transform the gross mixture of chemical

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compounds in the wastewater. The degree of biochemical reaction involved in the oxidation of carbon compounds is related to the period of incubation. When municipal sewage is tested, BOD, normally measures only 60 to 80 percent of the total carbonaceous biological oxygen demand of the sample. When testing OCPS wastewaters, however, the fraction of total carbonaceous oxygen demand measured can range from less than 10 percent to more than 80 percent. The actual percentage for a given waste stream will depend on the degradation characteristics of the organic components present, the degree to which the seed is acclimated to these components, and the degree to which toxic or inhibitory components are present in the waste.

## Total Suspended Solids TSS

Suspended solids can include both organic and inorganic materials. The inorganic materials include sand, silt and clay and may include insoluble toxic metal compounds. The organic fraction includes such materials as grease, oils, animal and vegetable waste products, fibers, microorganisms and many other dispersed insoluble organic compounds. [5-2] These solids may settle rapidly and form bottom deposits that are often a mixture of both organic and inorganic solids.

Solids may be suspended in water for a time and then settle to the bottom of a stream or lake. They may be inert, slowly biodegradable materials, or they may be rapidly decomposable substances. While in suspension, they increase the turbidity of the water, reduce light penetration, and impair the photosynthetic activity of aquatic plants. After settling to the stream or lake bed, the solids can form sludge banks, which, if largely organic, create localized anaerobic and undesirable benthic conditions. Aside from any toxic effect attributable to substances leached out by water, suspended solids may kill fish and shellfish by causing abrasive injuries, clogging gills and respiratory passages, screening light, and by promoting and maintaining noxious conditions through oxygen depletion.

Suspended solids may also reduce the recreational value of a waterway and can cause problems in water used for domestic purposes. Suspended solids in intake water may interfere with many industrial processes, and cause foaming in boilers, or encrustations on exposed equipment, especially at elevated temperatures.

### рH

The term pH describes the hydrogen ion-hydroxyl ion equilibria in water. Technically, pH is a measure of the hydrogen ion concentration or activity present in a given solution. A pH number is the negative logarithm of the hydrogen ion concentration. A pH of 7.0 indicates neutrality or a balance between free hydrogen and free hydroxyl ions. A pH above 7.0 indicates that a solution is alkaline; a pH below 7.0 indicates that a solution is acidic.

The pH of discharge water is of concern because of its potential impact on the receiving body of water. Wastewater effluent, if not neutralized before release, may alter the pH of the receiving water. The critical range suitable for the existence of most biological life is quite narrow, lying between pH 6 and pH 9.

Extremes of pH or rapid pH changes can harm or kill aquatic life. Even moderate changes from acceptable pH limits can harm some species. A change in the pH of water may increase or decrease the relative toxicity of many materials to aquatic life. A drop of even 1.5 units, for example, can increase the toxicity of metalocyanide complexes a thousandfold. The bactericidal effect of chlorine in most cases lessens as the pH increases.

Waters with a pH below 6.0 corrode waterworks structures, distribution lines, and household plumbing fixtures. This corrosion can add to drink ing water such constituents as iron, copper, zinc, cadmium, and lead. Low pH waters not only tend to dissolve metals from structures and fixtures, but also tend to redissolve or leach metals from sludges and bottom sediments.

Normally, biological treatment systems are maintained at a pH between 6 and 9; however, once acclimated to a narrow pH range, sudden deviations (even in the 6 to 9 range) can cause upsets in the treatment system with a resultant decrease in treatment efficiency.

## Oil and Grease (0 & G)

Oil and grease analyses do not actually measure the quantity of a specific substance, but measure groups of substances whose common characteristic is their solubility in freon. Substances measured may include hydrocarbons, fatty acids, soaps, fats, oils, wax and other materials extracted by the solvent from an acidified sample and not volatilized by the conditions of the test. As a result, the term <u>oil and grease</u> is more properly defined by the conditions of the analysis rather than by a specific compound or group of compounds. Additionally, the material identified in the O&G determination is not necessarily free floating. It may be actually in solution but still extractable from water by the solvent.[5-3]

Oils and greases of hydrocarbon derivative, even in small quantities, cause troublesome taste and odor problems. Scum lines from these agents are produced on water treatment basin walls and other containers. Fish and water fowl are adversely affected by oils in their habitat. Oil emulsions may cause the suffocation of fish by adhering to their gills and may taint the flesh of fish when microorganisms exposed to waste oil are eaten. Deposition of oil in the bottom sediments of natural waters can serve to inhibit normal benthic growth. Oil and grease can also exhibit an oxygen demand.

Levels of oil and grease that are toxic to aquatic organism vary greatly depending on the oil and grease components and the susceptibility of the species exposed to them. Crude oil in concentrations as low as 0.3 mg/l can be extremely toxic to freshwater fish. Oil slicks prevent the full aesthetic enjoyment of water. The presence of oil in water can also in crease the toxicity of other substances being discharged into the receiving bodies of water. Municipalities frequently limit the quantity of oil and grease that can be discharged to their wastewater treatment systems by industry, since large quantities of O&G can cause difficulties in biological treatment systems.

There are several approved modifications of the analysis for oil and grease. Each is designed to increase the accuracy or enhance the selectivity of the analysis. Depending on the procedure and detection method employed, the accuracy of the test can vary from 88 percent for the Soxhlet Extraction Method to 99 percent for the Partition-Infrared Method.

## NONCONVENTIONAL POLLUTANT PARAMETERS

#### Chemical Oxygen Demand (COD)

COD is a chemical oxidation test devised as an alternate method of estimating the oxygen demand of a wastewater. Since the method relies on the oxidation-reduction system of a chemical reaction rather than a biological reaction, it is more precise, accurate, and rapid than the BOD<sub>5</sub> test. The COD test is sometimes used to estimate the total oxygen (ultimate rather than 5-day BOD) required to oxidize the compounds in a wastewater. In the COD test strong chemical oxidizing agents under acid conditions, with the assistance of certain inorganic catalysts, can oxidize most organic compounds, including many that are not biodegradable. [5-4]

The COD test measures organic components that may exert a biological oxygen demand and may affect public health. It is a useful analytical tool for pollution control activities. Most pollutants measured by the BOD<sub>5</sub> test will be measured by the COD test. In addition, pollutants resistant to biochemical oxidation will also be measured as COD.

Compounds resistant to biochemical oxidation are of great concern because of their slow, continuous oxygen demand on the receiving water and also, in some cases, because of their potential health effects on aquatic life and humans. Many of these compounds result from industrial discharges and some of the compounds have been found to have carcinogenic, mutagenic, and similar adverse effects. Concern about these compounds has increased as a result of demonstrations that their long life in receiving water (the result of a low biochemical oxidation rate) allows them to contaminate downstream water intakes. The commonly used systems of water purification are not effective in removing these types of materials and disinfection with chlorine may convert them into even more objectionable materials.

It should be noted that the COD test may not measure the oxygen demand of certain aromatic species such as benzene, toluene and pyridine.

## Total Organic Carbon (TOC)

TOC measures all oxidizable organic material in a waste stream, including the organic chemicals not oxidized (and therefore not detected) in BOD and COD tests. TOC analysis is a rapid test for estimating the total organic carbon in a wastestream. When testing for TOC, the organic carbon in a sample is converted to carbon dioxide  $(CO_2)$  by catalytic combusion or by wet chemical oxidation. The CO<sub>2</sub> formed can be measured directly by an infrared detector or it can be converted to methane  $(CH_2)$  and measured by a flame ionization detector. The amount of CO<sub>2</sub> or CH<sub>4</sub> is directly proportional to the concentration of carbonaceous material in the sample. TOC tests are usually performed on commercially available automatic TOC analyzers. Inorganic carbons, including carbonates and bicarbonates, interfere with these analyses and must be removed during sample preparation. [5-5]

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## SECTION VI.

## WATER USE AND WASTEWATER CHARACTERIZATION

## WATER USE AND SOURCES OF WASTEWATER

Water use and wastewater generation occur at a number of points in OCPS manufacturing processes and ancillary operations, including: (1) direct and indirect process contact, (2) contact and noncontact cooling water, (3) utilities, maintenance and housekeeping, and (4) air pollution control systems such as Venturi scrubbers.

An example of direct process contact water is the use of aqueous reaction media. The use of water as a media for certain chemical processes becomes a major high strength wastewater source after the primary reaction has been completed and the final product has been separated from the water media, leaving unwanted by-products formed during secondary reactions in solution.

Indirect process contact waters, such as those discharged from vacuum jets and steam ejectors, involve the recovery of solvents and volatile organics from the chemical reaction kettle. In using vacuum jets, a stream of water is used to create a vacuum, but also draws off volatilized solvents and organics from the reaction kettle into solution. Later, recoverable solvents are separated and reused while unwanted volatile organics remain in solution in the vacuum water which is discharged as wastewater. Steam ejector systems are similar to vacuum jets with steam being substituted for water. The steam is then drawn off and condensed to form a source of wastewater.

The major volume of water use in the OCPS industry is cooling water. Cooling water may be contaminated, such as contact cooling water from barometric condensers, or uncontaminated noncontact cooling water. Frequently, large volumes of cooling water may be used on a once-through basis and discharged with process wastewater. Many of the effluent values reported by plants in the data bases were based on flow volumes which included their cooling water. An adjustment of the reported volumes of the effluents was therefore required to arrive at performance of treatment systems and other effluent characteristics. This adjustment was made by eliminating the uncontaminated cooling water volume from the total volume, to arrive at the contaminated wastewater flow. Concentrations also were adjusted using the simplifying judgment that the uncontaminated cooling water did not contribute to the pollutant level. However, it should be noted that in some cases cooling water can contribute relatively high TSS loading, especially to typically low strength plastics and synthetic materials wastewaters.

Tables 6-1 and 6-2 present treated effluent wastewater and raw waste flows reported by direct discharge plants and zero or alternative discharge disposal plants, respectively, for each of the four proposed subcategories. The adjusted flows presented were calculated from all raw and treated wastewater streams reported with the number of observations corresponding to the total number of reported waste stream flows. It

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## EFFLUENT FLOWS FOR PROPOSED SUBCATEGORIES DIRECT DISCHARGERS ONLY

	PLASTICS ONLY	NOT PLASTICS TYPE I w/ OXIDATION	NOT PLASTICS TYPE I w/o OXIDATION	NOT PLASTICS NOT TYPE I
	E F F F L O\\ MGD	E F I F I. O' MG (	F EFF V FLOW D MGD	EFF FLON MGD
	10.	700 29	.000 32.100	40.000
MEAN	۱.:	357 2	.771 2.464	3.346
MINIMUM	0.0	034 0	.008 0.020	0.007
MEDIAN	0.0	612 1.	.010 0.852	0.950
NUMBER OF Observatio	S NG	74	63 34	a 35

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TABL	E	6-	2
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## INFLUENT AND EFFLUENT FLOWS FOR PROPOSED SUBCATEGORIES ZERO DISCHARGE/ALTERNATIVE DISPOSAL ONLY

PLASTICS	NOT PLASTICS	NOT PLASTICS	NOT PLASTICS
ONLY	TYPE I w/	TYPE I w/o	NOT
	OXIDATION	OXIDATION	TYPE I

	INF(1) FLOW MGD	EFF FLOW MGD	INF (1) FLOW MGD	EFF FLOW MGD	INF (1) FLOW MGD	EFF FLOW MGD	INF(1) FLOW MGD	EFF FLOW MGD
MAXIMUM	10.700	•	5.329	•	4.030	•	2.600	•
MEAN	0.530	•	1.009	•	0.475		0.500	•
MINIMUM	0.000	•	0.000	•	0.000	•	0.001	•
MEDIAN	0.012	•	0.271	•	0.152	•	0.118	
NUMBER OF OBSERVATION	S 29		20		14		14	

(1) Since effluent flow for these plants is by definition zero, influent flows are presented

should be noted that the number of streams does not correspond to the number of plants due to the existence of multi-stream plants.

## WASTEWATER CHARACTERIZATION

A number of different pollutant parameters are used to characterize wastewater discharged by OCPS facilities. These include:

- 1, Biochemical Oxygen Demand (BOD)
- 2. Suspended Solids (TSS)
- 3. pH
- 4. Chemical Oxygen Demand (COD)
- 5. Total Organic Carbon (TOC)
- 6. Oil and Grease (O&G)

BOD is one of the most important gauges of the pollution potential of a wastewater and varies with the amount of biodegradable matter which can be assimilated by biological organisms under aerobic conditions. Large, complex facilities tend to discharge a higher BOD mass loading, although concentrations are not necessarily different from smaller or less complex plants. The nature of specific chemicals discharged into wastewater affects the BOD due to the differences in susceptability of different molecular structures to microbiological degradation. Compounds with lower susceptability to decomposition by microorganisms tend to exhibit lower BOD values even though the total organic loading may be much higher than compounds exhibiting substantially higher BOD values.

Raw wastewater TSS is a function of the products manufactured and their processes, as well as the manner in which fine solids that may be removed by a processing step are handled in the operations. It can also be a function of a number of other external factors including stormwater runoff, runoff from raw material storage areas, and landfill leachates which may be diverted to the wastewater treatment system. Solids are frequently washed into the plant sewer and removed at the wastewater treatment plant. The solids may be organic, inorganic or a mixture of both. Settleable portions of the suspended solids are usually removed in a primary clarifier. Finer materials are carried through the system, and in the case of an activated sludge system, become enmeshed with the biomass where they are then removed with the sludge during secondary clarification. Many of the manufacturing plants show an increase in TSS after wastewater leaves the treatment plant. This characteristic is usually associated with biological systems and indicates an inefficiency of secondary clarification in the removal of secondary solids. However, in plastics and synthetic materials wastewaters, formation of biological solids within the treatment plant may cause this solids increase due to the low strength nature of the waste.

Raw wastewater pH can be a function of the nature of the processes contributing to the waste stream. This parameter can vary widely from plant to plant and can also show extreme variations in a single plant's raw wastewater, depending on such factors as waste concentration and the portion of the process cycle discharging at the time of measurement. Fluctuations in pH are readily reduced by equalization followed by a neutralization system, if necessary. pH control is important regardless

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of the disposition of the wastewater stream (i.e, indirect discharge to a POTW or direct discharge) to maintain favorable conditions for biological treatment organisms.

COD is a measure of oxidizable material in a wastewater as determined by subjecting the waste to a powerful chemical oxidizing agent (such as dichromate) under standardized conditions. Therefore, the COD test shows the presence of organic materials that are not susceptable to attack by biological microorganisms. As a result of this difference, COD values are almost invariably higher than BOD values for the same sample. The COD test cannot be substituted directly for the BOD test because the COD/ BOD ratio is a factor which is extremely variable and is very dependent on the specific chemical constituents in the wastewater. However, a COD/ BOD ratio for the wastewater from a single manufacturing facility can be established. This ratio is applicable only to the wastewater from which it was derived and cannot be utilized to estimate the BOD of another plant's wastewater. It is often established by plant personnel to monitor process and treatment plant performance with a minimum of analytical delay. As production rate and product mix changes. however, the COD/BOD ratio must be revalidated for the new conditions. Even if there are no changes in production, the ratio should be reconfirmed periodically.

TOC measurement is another means of determining the pollution potential of wastewater. This measurement shows the presence of organic compounds not necessarily measured by either BOD or COD tests. TOC can also be related to the BOD and COD by ratio, but it too is only applicable to the specific wastewater for which the ratio is derived. TOC determination is also useful for day-to-day control of treatment operations.

Oil and grease determinations do not measure the quantity of a specific substance but measure substances whose common characteristic is their solubility in freon. Treatment of oil and grease involves dissolved air flotation and skimming practices. If these procedures are implemented and efficiently used and maintained, oil and grease values should be substantially lowered. Therefore, plants discharging high oil and grease values may reflect limited use of available treatment technologies and limited source controls for oil and grease abatement.

Tables 6-3 through 6-10 present raw wastewater characteristics for each of the four proposed subcategories. Minimum, maximum, mean and median concentration values as well as mass loading in pounds per day for all pollutant parameters of interest (BOD, TSS, COD, TOC and O&G) are presented for direct discharge plants and zero or alternative discharge/ disposal facilities.

Each set of observations shown in Tables 6-1 thru 6-10 should be considered a separate data subset, independent of other data subsets presented. Calculations which involve more than one data subset (i.e., determining BOd/COD ratios) may not be meaningful, since data subsets do not reflect the same group of plants. Similarly, multiplying the median concentration for some parameter by the mean flow will not correspond to the median pounds for that parameter.

	INF BOD MG/L	INF 1800 19/04 y	INF TSS MG/L	INF TSS LB/DAY	INF COD MG/L	INF CCD LUZDAY	INF TOC MG/L	INF TOC LB/DAY	11F 0&G ₩G/L	INF OEG LB/DAY
	3520.0	25262.0	2898.0	16082.3	4338.0	51392.6	2751.0	5560.8	242.0	2355.7
MEAN	506.2	3693.4	394.4	2263.0	1101.6	8315.9	705.4	1720.7	82.3	710.8
MINIMUM	2.0	13.0	5.0	1.5	27.0	60.5	9.0	13.0	23.0	72.9
MEDIAN	349.0	1985.8	80.0	803.3	857.0	4889.5	362.0	949.9	32.0	207.3
NUMBER OF OBSERVATIONS	49	49	42	42	45	45	8	8	4	4

## TABLE 6-3 RAW WASTEWATER CHARACTERISTICS - PLASTICS ONLY SUBCATEGORY DIRECT DISCHARGERS ONLY

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Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

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## RAW WASTEWATER CHARACTERISTICS - PLASTICS ONLY SUBCATEGORY ZERO DISCHARGE/ALTERNATIVE DISPOSAL ONLY

				_						
	INF 000 MG/L	INF BOD LB/DAY	INF TSS MG/L	THE TSS LB/DAY	NF COD MG/L	1NF COO L670AY	1NF TOC NG/L	INF TOC LO/DAY	INF O&G MG/L	14F 08G 18/04Y
	12542.0	4174.0	1055.0	351.1	18549.0	6173.1	 1938.0	201.6		
MEAN	4110.7	1232.3	317,7	105.6	5303.4	2068.4	923.0	82.3		
IN IN I MUM	380.0	111.7	0.3	۱.7	48.0	6.0	31.0	16.0	•	•
MEDIAN	1378.0	327.6	237.0	29.1	1422.0	1107.4	800.0	26.6	•	
NUMBER OF OBSERVATIONS	6	6	5	5	5	5	3	. 3	0	0

Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

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## RAW WASTEWATER CHARACTERISTICS -TYPE I WITH OXIDATION SUBCATEGORY

## DIRECT DISCHARGERS ONLY

	THF DOD MG/L	1%F 800 LB/DAY	1NF 155 MG/L	INF TSS LH/DAY	1 N F COD MG/L	INF COD LB/OAY	INF TOC MG/L	INF TOC LB/DAY	INF 08G MG/L	INF O&G LB/OAY
	5961.0	249714.3	4110.0	34195.2	21178.0	202706.2	3202.0	115196.9	17.4	351.8
MEAN	1702.1	25975.2	441.6	3949.8	4414.6	40369.9	836.4	24792.7	11.4	180.1
MINIMUM	43.0	939. <b>7</b>	9.0	19.0	53.0	1409.6	20.0	49.9	0.8	42.1
MEDIAN	1036.5	13289.5	72.0	672.9	3302.5	19714.5	513.0	13334.0	16.0	146.4
NUMBER OF OBSERVATIONS	42	42	21	21	34	34	21	21	З	Э

Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

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## RAW WASTEWATER CHARACTERISTICS -NOT PLASTICS TYPE I WITH OXIDATION SUBCATEGORY ZERO DISCHARGE/ALTERNATIVE DISPOSAL ONLY

	INF BOD MG/L	INF UOO LB/DAY	INF TSS NG/L	1 N F 1 S S L B / D A Y	INF COD NG/L	INF COD Lu/Day	LNF TOC MG/L	INF TOC LU/DAY	INF OBG MG/L	[NF 08G 18/04Y
MAXIMUM	52554.0	374100.5	459.0	7904.6	67095.0	5926 <b>50.2</b>	11005.0	201826.6	539.0	5829.0
MEAN	14532.4	71520.6	207.7	4150.1	26441.4	147345.6	5746.8	85417.3	539.0	5029.0
MINIMUM	450.0	2566.9	61.0	659.0	960.0	7474.0	282.0	21808.1	539.0	5829.0
MEDIAN	6022.0	20550.7	103.0	3796.0	23744.0	74623.0	5049.5	56977.2	539.0	5029.0
NUMBER OF OBSERVATIONS	7	7	3	3	9	9	4	4	1	۱

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Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

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## RAW WASTEWATER CHARACTERISTICS -NOT PLASTICS TYPE I WITHOUT OXIDATION SUBCATEGORY

## DIRECT DISCHARGERS ONLY

	INF 800 MG/1	INF GOD LB/DAY	INF TSS MG/L	[NF TSS LU/DAY	INF COD MG/L	INF COD LB/DAY	LNF TOC NG/L	INF TOC LØ/DAY	1 N F 08 G MG/L	INF 08G LB/DAY
	2725.0	30913.8	2666.0	16925.5	32476.0	71448.8	5226.0	31009.9		
MEAN	783.6	9384.2	479.7	2624.7	4782.5	11982.2	940.8	6343.0	335.0	365.4
MINIMUM	9.0	6.0	1.0	0.7	23.0	18.6	66.0	342.1	17.0	94.0
MEDIAN	467.0	5624.3	170.0	594.0	1022.0	6139.0	272.0	2109.1	418.0	410.2
NUMBER OF OBSERVATIONS	21	21	13	13	15	15	11	11	3	З

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Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

## RAW WASTEWATER CHARACTERISTICS -NOT PLASTICS TYPE I WITHOUT OXIDATION SUBCATEGORY

## ZERO DISCHARGE/ALTERNATIVE DISPOSAL ONLY

	INF 90D MG/L	INF 000 LB/0AY	INF TSS MG/L	INF TSS LB/DAY	[HF COD MG/L	1 NF COO LB/DAY	lnF TOC MG/L	1NF TOC LB/DAY	INF OSG MG/L	INF O&G LB/DAY
	3044.0	7597.8	267.0	755.3	11966.0	29867.1	364.0	663.2		
NEAN	1019.0	2410.4	106.3	397.2	3600.5	0720.6	223.0	523.1		•
MINIMUM	42.0	29.7	32.0	82.4	14.0	9.9	130.0	411.0	•	•
MEDIAN	495.0	1007.1	63.0	375.6	1211.0	2514.6	175.0	495.0	•	•
NUMBER OF OBSERVATIONS	4	4	4	4	4	4	Э	. 3	0	0

Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

## RAW WASTEWATER CHARACTERISTICS -NOT PLASTICS NOT TYPE I SUBCATEGORY

## DIRECT DISCHARGERS ONLY

	INF BOD MG/L	INF BOD LB/DAY	LINF TSS MG/L	INF TSS LB/DAY	1xf CCO MG7L	INF CGD LB70AY	INF TCC MG/L	INF TOC LB/DAY	INF OSG MG/L	INF O&G LB/DAY
	1743.0	59904.0	1266.0	 36940.8	5498.0	66653.8	1455.0	64563.2	420.0	6900.0
ΜΕΛΝ	421.7	9035.4	456.5	7383.6	1623.5	15229.4	376.6	10759.9	235.0	3606.9
MINIMUM	95.0	435.4	17.0	14.9	218.0	223.7	35.0	72.0	50.0	225.1
MEDIAN	318.0	3083.9	138.0	1723.4	590.0	5899.1	184.5	1992.7	235.0	3606.9
NUMBER OF OBSERVATIONS	11	11	13	12	13	13	8	8	2	2

Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

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## RAW WASTEWATER CHARACTERISTICS --NOT PLASTICS NOT TYPE I SUBCATEGORY

## ZERO DISCHARGE/ALTERNATIVE DISPOSAL

	1 N F BOD MG/L	1 NF 800 18/04 y	NF TSS MG/1.	INF ISS UB/DAY	INF COD MG/L	185 COD L870AY	LNF TOC MG/L	INF TOC LO/DAY	1 N F D&G MG/L	LHF O&G LB/DAY
								<b>.</b>		
MAXIMUM	1439.0	5646.0	33170.0	103043.3	27872.0	154442.1	9592.0	1197.1	286.0	6186.8
MEAN	631.0	1997.5	10230.3	46790.1	9580.8	37921. <b>0</b>	4076.5	660.2	286.0	6186.0
MUNIMUM	193.0	167.0	81.0	70.1	546.0	472.4	161.0	139.3	286.0	6106.0
MEDIAN	261.0	179.6	3031.0	1639.5	2956.0	2607.0	4876.5	660.2	206.0	6196.8
NUMBER OF OBSERVATIONS	Э	З	4	4	5	5	2	2	۱	۱

Note: Each set of observations represents a separate data subset. Calculations which combine subsets may not be meaningful. (See last paragraph page 109.)

#### SECTION VII.

#### CONTROL AND TREATMENT TECHNOLOGIES

#### GENERAL

This chapter addresses control and treatment technologies currently used or available to the OCPS industries for BPT. The treatment methodologies presented in this section are divided into in-plant technologies, including source control and in-plant treatment, and end-of-pipe (EOP) technologies.

Wastewaters from the OCPS industries are disposed of by one of three methods: (1) direct discharge, (2) indirect discharge, and (3) zero or alternative discharge. Direct discharge refers to the release of treated or untreated wastewater to a receiving stream. Indirect dischargers transport wastewater to a publicly owned treatment works (POTW). Zero or alternate discharge refers to situations in which generated wastewater is either disposed of on plant property or transferred to an alternate location where it is disposed of on-site or discharged after treatment.

Table 7-1 lists the zero or alternate discharge practices and the principal direct discharge end-of-pipe treatment technologies reflected by the Summary Data Base. The principal disposal/treatment practices are grouped by the number of waste streams and the number of plants for the four potential subcategories discussed in Section IV.

A total of 71 plants are single stream zero or alternate dischargers, and 23 plants are multiple stream zero or alternate dischargers. The largest group of plants in the data base are the 157 single pipe direct dischargers. Five other plants have multiple direct discharge streams. An additional 33 plants have both zero or alternate and direct discharge streams. The disposal method used at 2 plants could not be determined. Although the Summary Data Base was only developed for direct and zero or alternative discharge, six plants are currently indirect dischargers because they have diverted their effluents to a POTW. Data collected at these plants while they were direct dischargers has been retained in the Summary Data Base.

## IN-PLANT SOURCE CONTROLS

In-plant source control refers to process or operating techniques used to either reduce the quantity or improve the quality of a waste stream within a plant. Some in-plant control methodologies are capable of completely eliminating a waste stream, while others recover valuable by-products of the manufacturing process.

In-plant controls provide several advantages. Beyond the potential for recovery of saleable material, in-plant control may reduce EOP treatment plant costs, which often offset the in-plant treatment costs. In-plant control can also remove pollutants inhibitory or not amenable to EOP treatment schemes.

#### TABLE 7-1

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PHILNCIPLE TREATHENT/DISPOSAL PRACTICE

 	I I PLAST: ;	1 3 DHL7 1	I NOT PLA	1 8 C %	I NOT PLA	NOT C++	I NOT PL. I NOT T	ASTICE   YPE I	I ALL I STR	NABTE EAMS
DIS- PRINCIPAL I POSAL DESIGNATION I CODE I	I ND, OF	NO, OFI MASTE I STREAMSI	I NO, OF	NO, OFI WASTE I BTREAMSI	I ND, OF	NO, OFI HASTE I STREAMSI	I NO, OF	I NO, DFI I WASTE I ISTREAMSI	I NO, DF I PLANTS	I ND. 01 I WASTE ISTREAM
ZERO DISCHANGE	1		1		1	1	1	1 .1	1	1
	! .	! !				! . !	! .	! . !	! _	!
CON CONTRACT MADE							! !	! { !		
			1 3				: :		1 12	1 22
DRY DRY PROFESS (REPORTED)	1 12		1 1		i		: :	: : :	20	15
EVP EVAPORATION	1 2	21	1 0			ĩ				
INP IMPOUNDMENT	i 3	i 7i	1 4	i ši	i i		i ž	iži	i 12	i 18
LAP LAND APPLICATION	1 3	1 <b>3</b> i	1 0	0 1	1 0	i o i	i õ	i i i	i 3	1 3
OFS DEF-SITE TREATMENT	1 1		1 2	. 21	1 0 1	) 0 j	1 1	1 3 1	1 4	I
RTE PECYCLE I	1 9 1	i 10 i	1 11	1 11 1	1 2	1 2 1	I 2	1 2 1	1 24	1 25
TOTAL ZERO I	1 46	 	1 22 1	   52	1	20 1	1 15	1 23 1	1 94	   _162
DIRECT DISCHARGE I							1		1	 
••••••••	•	ı <b>ı</b>	I . I		•		I _	1 1	1	1
ALA AFRATED LAGOON I	1 8		1 9	91	1 4 1		1 5	1 6 1	1 24	1 27
ANL ANAEHODIC								! !!	1 2	1 Z
ARE AEROBIC LAGOUN								! ! !	4	
VAL THOM I STODEL 1				42 !	14		1 10		104	1 107
PRC POIATING BIDLOGICAL CONTACION.							! !	! ! !		
TRE TRICKLING FILTER	1 7 1		i				1		: :	
				ונכ	22		1 19		1 146	1 1 2 1
ACR ACTIVATED CARBON	i o i	i ei	i 1 (	i si	1 2		i 3	i si		; ,
CLR CLARIFICATION	1 5	L 7 İ	1 0	i é i	i I i	i ii	i ž	i 2 i	1 8	1 10
DAF DISSOLVED AIR FLOATATION	1 0 1	1 0 1	1 0	0	1 0		1 1	1 1 1	1 1	i 1
PHF HULTI-HEDIA FTLTRATION	1 0		• •	1	1 0 1		I 0	1 0 1	1 1	1 1
NEU NEUTRALIZATION	1 0	1 1	1 0	0 1	1 3	3 1	1 🔶	• • •	•	1 10
OLS OIL WATER SEPARATION	2	5 1	!!!				! !	! <u>?</u> !	1 9	!!!
PLF FFI, CLAGOLAIIUN (FILIMATION )										
STR STEAM STRIPPING	1 0	, , , ,	1 1				1 2	1 2 1	1 3	: :
TOTAL NON-BIOLOGICAL	1	1 17	1 4	5 1	1	10	1 15	j	1	48
	!	! . !	! . !				•	! !	!	!
UNK UNKNOWN I	1 2		1 0				1 0		1 2	
GRAND TOTAL	1 116		1 80						. 291	377

\* Type I w/Oxidation

\*\* Type I w/o/Oxidation

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Many of the newer chemical manufacturing plants are being designed with a reduction in water use and consequent minimization of contamination as part of the overall planning and plant design criteria. In addition, improvements have been made in existing plants to control pollution from their manufacturing processes and other activities, prior to discharge. In-plant source controls that have been effective in reducing pollution loads in the OCPS industries are described in the following paragraphs.

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## Process Modification

Older plants were sometimes designed without regard for raw material or water conservation. As costs have increased and local environmental regulations have become more stringent, some plants have modified their manufacturing processes. For example, plants which once used batch processes have gone to continuous operation. By doing so, wastewaters containing spent solvents or caustic which are generated by between batch cleanup are eliminated. As a consequence, production yields increase and overall wastewater generation is reduced.

## Instrumentation

An important source of pollutant loading in the OCPS industries is occasional process upset resulting in discharge of products, raw materials or by-product. For example, reaction kettles occasionally become over pressurized, resulting in a burst rupture-disc and subsequent discharge.

The in-plant control best suited to eliminate these occurrences is the installation of more sophisticated instrumentation. Alarms, pH and flow sensors and similar devices are capable of early detection of process upsets. Use of this type of instrumentation, coupled with added operator training, can measurably reduce pollutant loading.

#### Solvent Recovery

The recovery of waste solvents has become a common practice among plants using solvents in their manufacturing processes. However, several plants have instituted further measures to reduce the amount of waste solvents discharged. Such measures include incineration of solvents that cannot be recovered economically, incineration of bottoms from solvent recovery units, and design and construction of better solvent recovery columns to strip solvents beyond the economical recovery point. The economical recovery point has been reached when the cost of recovering additional solvent (less the value of the recovered solvent) is greater than the cost of treating or disposing of the remaining waste solvent.

## Water Reuse, Recovery, and Recycle

The use of barometric condensers can result in significant water contamination, depending upon the nature of the materials entering the discharge water streams. As an alternative, several plants use surface condensers to reduce hydraulic or organic loads. Water-sealed vacuum pumps often create water pollution problems. Several plants use a recirculation system as a means of greatly reducing the amount of water being discharged.

Reduction of once-through cooling water by recycling through cooling towers is a common industrial practice which results in a decreased total discharge volume. Stormwater runoff from manufacturing areas can contain significant quantities of pollutants. Separation of stormwater from process wastewater has been practiced throughout the industry and often facilitates the isolation and treatment of contaminated runoff.

Process modifications allowing for enhanced wastewater recycle have also been applied within the OCPS industry. Twenty-four facilities in the 291 plant Summary Data Base indicate that, through wastewater recycle, they achieve zero discharge.

## IN-PLANT TREATMENT

Besides implementing source controls to reduce or eliminate the waste loads generated within a manufacturing process, another alternative is available. In-plant treatment is directed toward removing certain pollutants before they are combined with the plant's overall wastewaters and consequently diluted. In a general sense, in-plant treatment processes are designed to treat specific waste streams. Although in-plant technologies can remove a variety of pollutants, they are usually designed to treat toxic or priority pollutants.

Generally speaking, in-plant treatment is employed to avoid undesirable impacts on a plant's end-of-pipe (EOP) treatment. Indirect dischargers may utilize in-plant treatment to remove components which could detrimentally affect the POTW, or materials which could pass through a POTW without receiving adequate treatment. In-plant treatment is also used to take advantage of the more efficient treatment of low volume, concentrated and homogenous waste streams generated by specific unit operations.

The basis for any decision to employ in-plant treatment is governed by the presence of:

- Pollutants toxic to the biota of an EOP biological treatment system.
- Biologically refractive pollutants.
- Highly concentrated pollutants.
- Pollutants that may offer an economic recovery potential (solvent recovery).
- Pollutants that are hazardous if combined with other chemicals downstream.
- Pollutants generated in small volumes in remote areas, precluding conveyance to centralized treatment.

- Corrosive pollutants that are difficult to transport.

- Pollutants that would contaminate EOP waste sludge, limiting disposal options.

Many demonstrated technologies are available for the removal of specific pollutants found in the wastewaters from organics and plastics manufacture. The selection of a specified in-plant treatment scheme depends both on the nature of the pollutant to be removed and on the engineering and cost comparisons of the options available.

The following paragraphs provide brief summaries of technologies either in use as in-plant treatment technology, or available to the OCPS industry. In that in-plant treatment is primarily used to remove toxic materials (i.e., metals, cyanide, solvents, etc.), the reader is referred to the BAT Development Document for further details on these treatment processes.

#### Activated Carbon Adsorption

Adsorption on granular activated carbon (GAC) is an effective, and moreover, a commercially established means of removing dissolved organic species from aqueous waste streams. Contaminants are removed from solution by a three-step process involving (1) transport to the exterior of the carbon, (2) diffusion within the pores of the activated carbon, and (3) adsorption on the interior surfaces bounding the pore and capillary spaces of the activated carbon. Eventually the surface of the carbon is saturated. When this occurs, replacement of the adsorber system with fresh (i.e., virgin or reactivated) carbon is required.

Both powdered activated carbon (PAC) and GAC are capable of efficiently removing many pollutants, including toxic and refractory organics. Powdered carbon is most frequently added to biological treatment processes and is not recovered.

Table 7-2 was taken from a recently published study [7-1] of carbon adsorption systems which have been in use in the chemical industry. The table lists more than 100 examples of full scale activated carbon adsorption systems.

#### Metals Removal

Heavy metals are of importance since their presence even at very low levels can inhibit biological activity and thereby lower the efficiency of the biological treatment system.

Technologies are well established for a number of metals removal methods. Hydroxide and sulfide precipitations, for example, are the most common methods of metal ion removal used. Many metals ions form insoluble hydroxides and sulfides at a high pH when treated with either caustic soda, lime, or soluble sulfides. The precipitates may be removed from the waste stream by such methods as settling or filtration. Other technologies applicable to metals removal are ion exchange and

## TABLE 7-2

# COMPANIES REPORTED TO HAVE EXPERIENCE WITH FULL-SCALE [7-5] GRANULAR ACTIVATED CARBON SYSTEMS

Company	Location	Principal Product
Alkcolac	Sedalia, MO	Surface active agents
Allied Chemical	Fairfield, AL	Creosote oils, tars, pitches
	South Point, OH	Formaldehyde
	Frankford, PA	Organic chemicals .
	Buffalo, NY	Inorganic chemicals
	Syracuse, NY	Monochlorobenzene, o-dichlorobenzene
	Hopewell, VA	Organic chemicals
	Moundsville, WV	Toluene diisocyanate, methylene dianiline
Amerada Hess	a	Refinery products
American Aniline	a	b
American Color and Chemical	Lockhaven, PA	Dyes
American Cyanamid	Bound Brook, NJ	Organic chemicals
Atlantic Richfield <sup>C</sup>	Carson, CA	Refinery products
Atlantic Ríchfield/ Polymers, Inc.	Monaca, PA	Diethylebenzene, divinyl benzene
BASF Wyandotte	Geismar, LA	Chlorine, hydrogen, sodium hydroxide
	Washington, NJ	Polypropxy ethers, polypropylene glycol
Beckman Instrument Co.	Porterville, CA	Ъ
Borden, Inc.	Bainbridge, NY	Plastics and resins
British Petroleum Corp. <sup>C</sup>	Marcus Hook, PA	Refinery products
C. M. Masland	Waskefield, RI	Textiles
Ciba Geigy	St. Gabriel, LA	Pesticides
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Crompton and Knowles Corp.	Gibraltar, PA	Dyes
Diamond Shamrock	Houston, TX	Pesticides
Dow Chemical	Gales Ferry, CT Plaquemine, LA Midland, MI	Plastics and resins Organic chemicals Phenol, acetic acid
Du Pont	Beaumont, TX Richmond, Va Belle, WV	Pharmaceutical chems. Textile fibers Organic chemicals
EPA Emergency Response Unit	Mobile Unit	Chemical spill cleanup
Fike Chemical	Nitro, WV	Speciality organic chemicals
First Chemical Corp.	Pascagoula, MS	Aniline, nitrobenzene, nitrotoluene
FMC	Baltimore, MD So. Charleston, WV Nitro, WV Middleport, NY Bayport, TX	Pesticides Organic chemicals Organic chemicals Pesticides Plasticizers, glycerin
General Electric	Pittsfield, MA Fort Edwards, NY <sup>d</sup> Selkirk, NY	Plasticss and resins Plastics and resins
Georgia Pacific	Conway, NC	Phenolic resins
Hardewicke chemical	Elgin, SC	Specialty organic chemicals
Hercules, Inc.	Hattiesburg, MS	Terpen oils, hydrocarbon resins
Hooker	Hahnville, LA	
Houston Chem. Div. of PPC	Beaumont, TX	Ethylene glycol, ethylene dibromide
ICI Americas	Goldsboro, NC	Pesticide research
Iowa Army Ammunition Plant	Burlington, IA	Explosives
Joliet Army Ammunition Plt	Joliet, IL	Explosives
Kansas Army Ammunition Plt	Parsons, KS	Explosives

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Liquified Coal Development Corp.	Captina, WV	Anthracene-derived solvents
Lone Star Army Ammunition Plant	Texarkana, TX	Explosives
Louisiana Army Ammunition Plant	Shreveport, LA	Explosives
Matlack	Swedesboro, NJ	Tank truck washing
Mobay	Cedar Bayou, TX	Organic chemicals
	New Martinsville, WV	Isocyanates, polyols, polyesters
Monsanto	Anniston, AL Sauget, IL St. Louis, MO Alvin, TX Texas City, TX Luling, LA Nitro, WV	Polynitrophenol Organic chemicals Organic chemicals Organic chemicals Organic chemicals Organic chemicals Cyclohexanol Pesticides
Neville Chemical	Neville Island, PA	Plastics, resins
Olin Corp.	McIntosh, AL Bradenberg, KY Rochester, NY Ashtabula, OH	Pesticides Organic chemicals Organic chemicals Organic chemicals
Owens Corning	Anderson, SC	Plastics and resins
Palisades Industries	Peace Dale, RI	Textiles
Pennwalt	Houston, TX	Organic chemicals
Pfizer Chemical	Terrahaute, IN South Port, NC Brooklyn, NY Greensboro, NC	Pharamaceutical chems. Citric acid Organic chemicals Organic chemicals
Proctor and Gamble	Chicago, IL Baltimore, MD Kansas City, KS Dallas, TX	Fatty acids Fatty acids Fatty acids, alcohols Fatty acids
Reichhold Chemicals	Tuscaloosa, AL	Phenol, pentaerythritol, resins
Republic Steel	Cleveland, OH	Coke

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Rhodia (Rhone-Poulenc)	Freeport, TX <sup>C</sup> Portland, OR	Organic chemicals Pesticides
Rocky Mountain Arsenal	Denver, CO	b
Rogers Corporation	Manchester, CT	Plastics and resins
SCA Chemical Waste Services	Lewiston, NY	Chemical waste disposal
Schenectady Chemical	Schenectady, NY	Phenolic resins
Sherwin Williams Co.	Chicago, IL	Para-cresol
Stauffer Chemical	Bucks, AL <sup>d</sup>	
	Lemoyne, AL	Sulfur
	Richmond, CA	Pesticides
	Dominguez, CA	Inorganic chemicals
	San Jose, CA	Flavor and fragrance chemicals
	Delaware City, DE	Carbon disulfide
	Louisville, KY	Organic chemicals
	Geismar, LA <sup>d</sup>	
	Henderson, WV	Organic chemicals
	Skaneateles Falls,NY	Detergents
	Galliopolis Ferry,WV	Syn. lubricants, plasticizers, esters
	Green River, WY <sup>d</sup>	
Stepan Chemical	Fieldsboro, NJ	Detergent intermediates
Stephen-Leedom Carpet	Southhampton, PA	Carpets
Tooele Army Ammunition Plt	Tooele, UT	Explosives
TRA	Irving, TX	b
Union Carbide	Hahnville, LA <sup>d</sup> Ponce, PR Greenville, SC <sub>d</sub> Woodbine, GA	Synthetic fibers

Velvet Textile Co.	Blackstone, VA	Textiles
Vicksburg Chemical	Vicksburg, MS	Toxaphene, methyl parathion
Yorktown Naval Weapons Sta.	Yorktown, VA	Explosives
Unidentified (1) <sup>C</sup>		Pesticides
Unidentified (2)		Organic chemcials
Unidentified (3)		Explosives
Unidentified (4)		Chlorobenzene, dichlorobenzene
Unidentífied (5)		Toxaphne, DNBP, cyanazine
Unidentified (6)		Dalpon
Unidentified (7)		2,4-D, 2,4-DB, MCPA
Unidentified (8)		Parachloronítrobenzene, terrazole
Unidentified (9)		Dicofol
Unidentified (10)		Trifluralin, isopropanol, ethalfluralin
Unidentified (11)		DEET, piperonyl butoxide, thanite
Unidentified (12)		Carbofuran
Unidentified (13)		Atrazine

a b Location not given in data source. c Unit known to be shut down. d No plant listed at this location in 1979 Directory of Chemical Producers. Neither company nor location identified in data source.

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membrane processes such as reverse osmosis and ultrafiltration. These technologies are normally employed by industry as in-plant treatment; however, none were reported in the Summary Data Base.

## Steam Stripping

Steam stripping is a variation of distillation whereby steam is used as both the heating medium and the driving force for the removal of volatile materials. For employment of steam stripping, steam is introduced into the bottom of a tower. As it passes through the wastewater, the steam vaporizes and removes volatile materials from the waste and then exits via the top of the tower. Although most commonly employed as an inplant technology for solvent recovery, steam stripping has been reported as a wastewater treatment process. Data from three plants using steam stripping as the primary treatment step are summarized in Table 7-3.

## Liquid-Liquid Extraction

Liquid-liquid extraction (LLE) is a process that can separate certain components from a solution by contacting the solution with an immiscible liquid that has a higher solubility for the components of the solution than it does for the solution contacted. LLE operating and capital expenses involve the liquid-liquid contactor with its peripheral equipment and the solvent regeneration equipment. Although liquid-liquid extraction is a common process operation, it is normally applied as an inplant treatment to utilize the highest available concentration gradient.

No data were available for LLE.

#### Oxidation

Oxidation as a treatment practice is accomplished by either wet or chemical oxidation. Wet oxidation is a common process in which an aqueous waste can be oxidized in a closed, high-temperature, highpressure vessel. Wet oxidation has been used to treat a variety of wastes including pulping waste and acrylonitrile liquor. A percent reduction in excess of 99.8 has been reported for some of the toxic pollutants.[7-2] This process is applicable particularly as in-plant and EOP treatments of wastes with a high organic content.

The application of chemical oxidation to industrial wastes is well established for cyanides, sulfide, ammonia, and other such harmful substances in waste streams. Chemicals commonly used as oxidizing agents include chlorine, hypochlorite, hydrogen peroxide, potassium permanganate, ozone, and chlorine dioxide. Although several plants in the Summary Data Base reported using chlorination as part of their EOP treatment, it was used as a sterilizing medium rather than as a chemical oxidation process.

No data were available for oxidation.

#### TABLE 7-3

# STEAK LIRIPPING, ALL PASTE STREAMS IN UNBER OF STHEAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTERATER TREATMENTS 3

	EFF Flom	BUD Inf	FALL .	900 1	133 1nF	138 EFF	155 1	C UD I NF	COD Eff	CU9 X	ULG INF	OLG EFF	046 X
	MGU	MG/L	MG/L	AED	MG/L	MG/L	RED	MC/L	MG/L	RED	MG/L	₩6/L	RED
ALE INUM	0,346	579"0	208.0	84.9	1198.0	140.0	<b>18.5</b>	•	•	•	•	•	•
MEAN	0.375	238.0	112.0	84,9	<b>441.5</b>	<b>61,3</b>	84,0	•	•	•	•	•	•
elelene	0,280	238.0	12.0	84,7	85.0	18.0	49,4	•	•	•		•	•
HEDIAN	0.300	238.0	34.0	84.7	441.5	26,0	84.0	•	•	•		•	•
NUMBER OF	INS 3	1	3	1	2	3	2	0	0	0	0	•	

## STEAM STRIPPING, ALL MASTE STREAMS

## NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTS 3

	10C 1nF MG/L	10C EFF MG7L	TOC 1 RED	PHENOL INF MG7L	PHENOL EFF HG7L	PHENOL X RED	NH3N 1NF MG7L	NH3N EFF HG7L	NH3N 1 RED	CR INF HG/L	CR EFF Mg/L	CR 1 RED
		•••••										
ATTHON	492,0	360.0	87.4	•	•	•	•	•	•	•	•	•
EAN	492.0	145.3	87.0	•	•	• .	•	•	•	•	•	•
INTHON	492.0	15,0	87.6	•	•		•	•	•	•		•
EDIAN	492.0	61.0	87.6	•	•		•		•		•	•
UMBER OF	1	3	1	0	0	0	0	0	0	0	0	0

a. These data are from plants that use this technology as the principal component of their wastewater treatment system.

#### END-OF-PIPE TREATMENT

End-of-pipe treatment refers to those processes that treat a combined plant waste stream for pollutant removal prior to discharge. Adequately designed, operated, and maintained EOP facilities allow manufacturers to discharge their wastewater directly to a receiving body of water.

EOP technologies covered in this report are classified as primary, secondary and tertiary processes. Depending on the nature of the pollutants to be removed and the degree of removal required, different combinations of the available treatment technologies may be used.

Primary treatment usually involves physical separation processes. These technologies include clarifiers, oil skimmers, dissolved air flotation and similar devices which may use flocculants to assist in the removals. Depending on the nature of the suspended solids in the wastewater, this treatment may remove a significant amount of the BOD attributable to suspended solids or floating materials from industrial wastewaters.

Secondary treatment is utilized when the primary system cannot improve the wastewater to a sufficient degree to permit discharge. Secondary treatment usually consists of biological processes capable of removing the soluble pollutant constituents. Biological processes are widely used in industrial waste treatment and, as measured by BOD, are very successful in removing biodegradable organics. Factors which influence the design and operation of biological systems for industrial wastes include the sensitivity of these systems to influent composition changes and the potential inhibitory effects of certain industrial chemicals on the microorganisms. Design techniques which accommodate such factors are discussed in the section entitled "Design, Operation and Management Practices."

Tertiary treatment refers to treatment following the biological or other secondary treatment system. The technologies available for tertiary treatment vary, but normally relate to the removal of specific pollutant parameters not effectively removed in secondary treatment. Some tertiary treatment unit processes are also applicable to in-plant or primary treatment schemes.

## Primary Treatment

In the following paragraphs the primary treatment processes used by the 291 OCPS industry plants in the engineering data base are discussed.

Equalization - Equalization consists of a wastewater holding vessel or pond large enough to dampen flow and/or pollutant concentration variation and permit a nearly constant discharge rate and wastewater quality. The holding tank or pond capacity is determined by wastewater volume and composition variability. The equalization basin may be agitated or may utilize a baffle system to prevent short circuiting. Equalization is employed prior to wastewater treatment processes that are sensitive to fluctuations in waste composition or flow. No plants in the Summary Data Base reported equalization as the only treatment technology used. However, 124 plants included equalization as a part of their total treatment system.

<u>Neutralization</u> - Neutralization is practiced in industry to raise or lower the pH of a wastewater stream. Alkaline wastewaters may be neutralized with hydrochloric acid, carbon dioxide, sulfur dioxide, and, most commonly, sulfuric acid. Acidic wastewaters may be neutralized with limestone or lime slurries, soda ash, caustic soda, or anhydrous ammonia. Often a suitable pH can be achieved through the mixing of acidic and alkaline process wastewaters. Selection of neutralizing agents is based on cost, availability, ease of use, reaction byproducts, reaction rates, and quantities of sludge formed.

Nine plants in the 291 plant Summary Data Base reported using neutralization as their principal treatment method. In addition, 104 other plants used neutralization as part of their treatment system.

Clarification - Clarification, in this context, may be defined as the removal of solid particles from a wastewater through gravity settling. The nature of the solids and their concentration are the major factors affecting the settling properties.

Among plants in the Summary Data Base, eight employ clarification as the principal component of their treatment system. Performance data for these plants is presented in Table 7-4. In addition, 94 other plants use some form of clarification as part of their treatment system, either with or without the use of precipitation, coagulation or flocculation.

<u>Precipitation/Coagulation/Flocculation</u> - Gravity clarification may be supplemented by precipitation, coagulation or flocculation providing enhanced suspended solids removal. Precipitation, coagulation or flocculation may also be used as a primary treatment step to protect biological secondary treatment processes from upset due to toxic metallic pollutants.

Simple clarification is usually accomplished with standard sedimentation tanks (either rectangular or circular). If additional solids removal, removal of colloidal solids, or removal of dissolved metallic ions is required, precipitation, coagulation or flocculation are added. Coagulation is usually accomplished by adding an appropriate chemical (alum, lime, etc.) followed by a rapid mix and finally a slow agitation to promote floc particle growth. A polymeric coagulant aid is sometimes used in these systems.

A total of 3 plants in the Summary Data Base report using precipitation, coagulation, or flocculation as the principal component of their treatment system. Data reported for these systems is presented in Table 7-5. A total of 15 plants use some form of coagulation as part of their treatment system.

Flotation - Flotation is used to remove oils and other suspended substances with densities less than that of water or, in the case of dissolved air flotation, particles that may be slightly heavier than water. As with conventional clarifiers, flocculants are frequently employed to

TABLE 7- 4

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#### CLARIFICATION, ALL HASTE STREAMS

NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTERATER TREATMENTS 10

	EFF Flom Mgo	BUD 1nf Ng/L	800 E/F Mg/L	800 1 <sup>R</sup> E0	133 1NF Mg/L	158 EFF MG/L	TSS 1 R <sub>ED</sub>	COD 1nf Mg/L	COD Eff Mg/L	COD 1 RED	OLG INF Mg/L	OLG EFF Mg/L	OLG X RED
	3.700	\$72.0	115.0	62.7	170.0	90.0	92,4	1219,0	455.0	62.7	••••••	1.0	••••••
MEAN	0,867	190.0	49.8	48,5	62.3	23, <b>6</b>	51,5	475,5	165.6	28,4	•	1.0	•
IN CHUM	0,057	9,9	۰.۰	10.9	31.0	7.0	7.2	42.0	42.0	0.0	•	0.4	•
EDJAN	0.344	89.5	37.0	52.0	64.0	14.5	53.2	320.5	58,0	25.4		1.0	•
NUMBER OF DBSERVATE	DMg 10	٩	٠	3	٩	8	4	4	,	4	•	2	

#### CLARIFICATION, ALL MASTE STREAMS

NUMBER OF STREAMS REPORTING THIS IECHNOLOGY AS MAJOR MASTEWATER TREATMENTS 10

	TOC Inf MG/L	10C EFF #G/L	10c 1 RED	PHENUL INF MG/L	PHENOL EFF MG/L	PHENOL X RED	NH3N 1NF MG7L	NHJN LFF MG/L	NH3N 8 RED	CR INF MG/L	CQ Eff Mg/L	CR X RED
MAXIMUM	143.0	118.0		••••••	••••••	• • • • • •	••	• • • • • • •	••••••	····· 1,5	0,3	99,2
MEAN	76,0	67.5	17.5	•	•		•	•	•	1.0	0.1	69.6
HINIHUM	9.0	17.0	17.5	•		•		•		0,5	0.0	40.0
MEDIAN	76.0	67.5	17.5	•	•	•	•	•	•	1.0	0.0	69.6
NUMBER OF Observations	2	2	1	a	•	•	0	0	٥	2	J	2

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a These data are from plants that use this technology as the principal component of their vestowator treatment system.

#### TABLE 7- 5

# PPT., COAGULATION, FILTRATION, ALL MASTE STALEAS

NUMBER OF STREAMS REPORTING THIS SECHNOLOGY AS MAJUR RASTEWATER TREATMENTS 3

	EFF Floa Mgd	BUD Inf MG/L	800 EFF MG/L	BOD 1 RED	T33 -1NF MG/L	133 EFF HG/L	T33 X RED	CUD Inf Mg/L	COD 877 MG/L	CuD 1 RED	ULG Inf Mg/L	OLG EFF NG/L	OLC L RED
MAXINUM	32,100	62.0	632.0	72.6	•	894.0	•	•	523.0		•	•	•
MEAN	13,973	62.0	324,5	72.4	•	406.0		•	923.0	•			•
4141HUM	4,400	62.Q	17.0	72.4	•	8,0		•	523.0				•
MEDIAN	5.220	62.0	324.5	72.6	•	406.0	•	•	523,0		•	•	•
NUMBER OF Deservatio	DNS 3	ı	z	ı	•	z	0	0	1	0	0	•	0

## PPT,,COAGULATION,FILTRATION, ALL HASTE STREAMS

NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTE &

	10C [NF MG/L	10C EFF MG/L	TOC 1 RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL X RED	NH3N INF Hg7L	NH3N EFF HG/L	NH3N X RED	CA 1nf Hg/L	CR Eff HG/L	CR X RED
MUMIXIMUM	•	276.0	•	•	0,1	•	•	782.0	•	•	0.8	•
MEAN	•	276.0		•	0,1	•		782.0	•	•	0,8	•
HIMIMON	•	276,0			0,1	•	•	782.0	•		0,8	•
MEDIAN	•	276.0		•	0.1	•	•	782.0	•	•	0.8	•
NUMBER OF DBSERVATIONS	0	1		, o	1	0	0	1	0	0	1	٥

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

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enhance the efficiency of the flotation units. Although flotation is sometimes referred to in the context of dissolved air flotation, such other technologies as oil/liquid skimming and solids skimming are also flotation operations, and are sometimes an integral part of standard clarification.

For the one OCPS industry plant having flotation as its primary method of treatment, summary performance data is presented in Table 7-6. An additional 6 plants use flotation as a part of their total treatment system.

## Secondary Treatment

Technologies classified as secondary treatment are generally biological processes and serve the primary function of removing dissolved carbonaceous pollutants as represented by BOD, COD, and TOC measurements. Biological systems may also be designed to remove some nitrogenous pollutants. Biological systems can remove limited amounts of heavy metals and refractory organic toxic chemicals through adsorption, biomass uptake and biodegradation, if properly acclimated to the waste. Nevertheless, these processes are usually designed to treat large quantities of dissolved carbonaceous wastes and any other pollutant removal or treatment is often incidental.

Biological Treatment - All biological treatment systems are designed to expose wastewater containing biologically degradable organic compounds to a suitable mixture of microorganisms, in a controlled environment which contains sufficient essential nutrients for the biological reaction to procede. Under these conditions the reduction of biologically assimilable pollutants will take place in a reasonably predictable manner. Biological treatment is based on the ability of mircoorganisms to utilize organic carbon as a food source. The treatment is classified as aerobic, anaerobic, or facultative. Aerobic treatment requires the availability of free dissolved oxygen for the bio-oxidation of the waste. Anaerobic treatment is intolerant of free dissolved oxygen and can utilize "chemically bound" oxygen (such as sulfates) in breaking down the organic material. Facultative organisms can function under aerobic or anaerobic conditions as the oxygen availability dictates.

Although the definitions of the processes are distinct, in practice both aerobic and anaerobic conditions may exist in the same treatment unit, depending on degree of aeration, degree of mixing, effects of photosynthesis, and other factors which contribute to the supply and distribution of oxygen to the treatment system. Facultative lagoons are designed to utilize both aerobic and anaerobic mechanisms as a means of reducing the net sludge production.

Biological treatment processes are widely used, and if properly designed and operated, are capable of high BOD removal efficiencies. Such systems given sufficient reaction time, can reduce the concentration of any degradable organic material to a very low concentration. Any organic material which will respond to the standard BOD test procedure is by definition a degradable substrate.
#### DISSOUVED AND FLOATATION, ALL MASTE STREAMS

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS HAJOR MASTERATER TREATMENTS &

	EFF Flûm Mgd	800 Inf Mg/L	BUD EFF NG/L	BOD k RED	T33 INF Mg/L	133 EFF MG/L	153 1 RED	COD INF Mg/L	COD EFF MG/L	CUP 1 RED	ULG INF MG/L	OLG EFF Mg/L	DLG 1 RED
	0,334	•		•••••	••	42,0	••••••	••••		••••••	•	•	•
PEAN	0.334		122.0	•	•	42,0	•		162,0	•		•	•
HINIHUM	0.334	•	122.0	•.	•	42.0		•	165.0		•	•	•
HEDJAN	0.334	•	122.0	•	•	42.0	•	•	145.0	•	•	•	•
NUMBER OF Observatio	DNS I	ė	1	0		ı	•	•	1	•	0	0	•

#### DISSOLVED AIR FLOATATION, ALL HASTE STREAMS

NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJUR<sup>4</sup>MASTEWATER TREATMENTE I

	10C 1NF MG/L	TUC EFF MG/L	10C 1 RED	PHENUL INF MG/L	PHENOL EFF MG7L	PHENOL 1 RED	NH3N 1NF MG7L	NH3N EFF MG/L	NH3N 1 AED	C# 1N# MG/L	CR EFF MG/L	CR X RED
	•		•••••	•	0,3	•••••••	•	•••••			2,0	•••••
HEAN	•	62.0	•	•	0.3	•	•		•	•	2.0	•
HINIHUM	•	e5°0	•	•	0,3	•	•		•	•	2.0	•
MEDIAN	•	62.0	•	•	0.3	•	•	•	•	•	2.0	•
NUMBER OF Observations	0	1	(	<b>,</b> 0	1	0	a	۰	٥	0	1	0

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

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It was previously noted in Section IV that properly designed and operated biological treatment systems will produce similar effluent BOD, concentrations even though influent BOD, concentrations may be significantly different. This is illustrated by Figures 7-1 and 7-2. These figures were prepared by plotting the lognormal frequency distribution of plants achieving various effluent BOD, concentrations. The criteria, as explained later in this section, for establishing that a plant was well designed and operated was defined by including data from only those plants which achieved 95 percent BOD, removal or which achieved effluent BOD, concentrations of 50 mg/l or less. Prior to graphing, the data was sorted by influent BOD, concentration into the four ranges shown in the figure. The minimum, maximum, and median values for each set of data are also shown. Figure 7-1 presents data from Plastics Only plants, Figure 7-2 presents the data from all other plants for which influent and effluent BOD, concentrations were available, and which met the criteria for being well designed and operated.

The figures illustrate that good effluent quality can be achieved over a wide range of influent concentrations. For the Plastics Only plants, median effluent  $BOD_{r}$  varies between 9 and 21 mg/l although influents range over two orders of magnitude. For OCPS plants producing products other than Plastics Only, median effluent  $\dot{B}OD_{\varsigma}$  concentrations range from 10 to 20 mg/l for influent BOD<sub>5</sub> concentrations up to 1000 mg/l, and a median effluent BOD<sub>5</sub> of 44 mg/l for influents greater than 1000 mg/l. The higher median effluent obtained for influents greater than 1000 mg/l does not necessarily indicate that high strength influents are any less degradable than the lower strength influents previously presented. The three lower plots in Figure 7-2 represent a relatively narrow range of influent concentrations, specifically 0 to 1000 mg/l. The uppermost plot presents data from 19 plants with influent BOD concentrations which range from 1076 mg/l to 5710 mg/l. Because of the signifcantly wider range of influents in this group, the spread between minimum and maximum effluent values does not necessarily contradict the theoretical assumption that a similar limiting effluent concentration can be achieved. Although the 19 plants with influent BOD concentrations greater than 1000 mg/l generally achieve the highest percentage BOD removals, typically 96 to 98+ percent, they do not necessarily degrade the organic material to the maximum degree possible. Without access to the design basis for each of the 19 plants, it cannot be determined if the plant was designed to achieve the maximum removal possible by a biological system, or a specified level of treatment (i.e., some percentage of BOD reduction) which was judged adequate for a specific site.

Although most biological systems can ultimately reduce effluent BOD to similar limiting concentrations, the rate of reaction will depend on a variety of design considerations. These considerations do limit the direct transfer of design and operating conditions from one industrial plant to another although the chemical product lines may be similar. Techniques are available, however, to optimize design and operating conditions to ensure adequate treatment for all industry wastes.

Biological systems operate most efficiently under so called "steady state" conditions. Unfortunately, industrial wastewater is frequently found to be extremely variable in composition and compentration. Waste









BIOLOGICAL SYSTEMS - NOT PLASTICS ONLY

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equalization is typically used prior to biological treatment to address this consideration.

Toxic or inhibitory compounds frequently present in industrial wastes can impair the biological process. Proper acclimatization can develop strains of organisms which are tolerant to normally toxic substances. However, once a specialized strain is established, care must be exercised to avoid changes in concentration of the chemicals for which the microorganisms have developed a tolerance. Increases or decreases in concentration over a narrow range can result in a complete loss of the specialized organisms and failure of the treatment process. Reestablishment of a suitable microbial population can be a lengthy procedure.

It is generally accepted that temperature affects the performance of the biological treatment process since the biodegradation rate is temperature dependent. The relationship usually employed is:

 $k_{T} = k_{20} \circ_{C} \times \Theta^{(T-20)}$ where:  $k_{T} = kinetic rate at temperature T (°C)$  $K_{20}^{T} = kinetic rate at 20(°C)$  $\Theta^{20} = temperature coefficient$ 

It should be recognized that the temperature of significance is the temperature in the reaction system, and a thermal balance must be computed considering the ambient air temperature and influent wastewater temperature. The sensitivity of the reaction rate to temperature is defined by  $\Theta$ , a dimensionless coefficient.\* A value of  $\Theta$  equal to 1.00 would imply that the reaction kinetics are unaffected by changes in temperature. As the value of  $\Theta$  increases above 1.0 the reaction becomes increasingly sensitive to changes in operating temperature. The value of  $\Theta$  for several organic-chemical wastewaters has been reported [7-3] to vary from 1.055 to 1.10. The effect of temperature on BOD removal in an organic chemicals plant, as reported by Eckenfelder, et al., is shown in Figure -7-3. The figure shows that although treatment efficiency decreases with decreasing temperature, a high degree of BOD removal can be achieved even at very low temperatures if suitable food to microorganism ratios are maintained. Lower F/M ratios than those shown in Figure 7-3 can be used to obtain even higher BOD removals. Increasing MLSS concentrations and optimizing sludge ages will also help in improving BOD removals.

Other references show conflicting results in evaluating the effects of temperature on wastewater treatment plant performance.

Berthouex, et al. [7-4] developed linear and time series models relating effluent BOD<sub>5</sub> to influent BOD<sub>5</sub>, MLSS, temperature and hydraulic retention time based on three years of data compiled at the Madison Sewage Treatment Plant (Wisconsin). They found no significant effect of temperature on performance when gradual changes in temperature (4-24°C) occurred.

<sup>\*</sup> which must be determined empirically



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B.A. Sayigh [7-5] conducted activated-sludge laboratory studies with continuous stirred-tank reactors and concluded that the effects of temperature using domestic sewage, organic-chemicals wastewaters and petrochemical wastewaters depend on the specific type of wastewater being treated. The author also found that the higher the sludge age, the less the susceptibility of the process to variations in temperature.

Work done by Del Pino [7-6] using wastewaters from three organic chemical plants showed that low temperature operation did reduce treatment efficiency, but this could generally be compensated for by operation at higher MLSS concentrations.

Spearman correlation coefficients were used to statistically determine if temperature, as measured by geographic location in degree-days, was significant for biological effluents in the EGD Summary Data Base. Table 7-7 presents a summary of this analysis for BOD, COD and TSS. In all cases, effluent quality (measured as effluent mg/l) was found to be statistically independent of location using degree-days as a surrogate for temperature.

The principal difficulty encountered when evaluating the impact of temperature on treatment system performance is that temperature is only one of several characteristics which affect the operation of the system. Changes in temperature (both seasonal and short term), raw waste load, product mix, flow, food to microorganism ratio, dissolved solids and suspended solids will all have some impact on treatment. In reviewing full scale plant operating data, it is difficult to isolate temperature effects from changes caused by variables other than temperature. This problem can be overcome in laboratory scale studies where temperature can be controlled and other variables held constant, but the usefulness of applying temperature data collected in this manner to the operation of a full scale system is questionable. This is particularly true in the OCPS industry where raw waste load variability is significant due to batch operations, frequent product mix changes, and raw materials variations.

In summary, analysis of this data would generally confirm the observations which appear in the literature. Specifically, temperature can have an impact on the treatment efficiency in some cases. However, temperature is only one of several factors which impact treatment. Waste load variations, biomass acclimation, flow variations, waste treatability and temperature of the wastewater during treatment must all be taken into consideration when developing a treatment sequence for a specific industrial site. The interaction between these factors makes it difficult to isolate any one, such as temperature, separately. Thus, temperature considerations must be viewed as specific to a given site, rather than as specific to any given region or geographic area.

Regardless of the above restrictions and limitations on the applicability of biological treatment systems, technologies and operating techniques exist, which if properly applied, can overcome these limitations. Just as two organic chemical plants producing the same product may have different process chemistry which reflects differences in feedstocks, treatment systems must be designed and operated to reflect the specific

TABLE	7-7
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# DEGREE-DAYS VS. EFFLUENT QUALITY SPEARMAN CORRELATION COEFFICIENTS FOR EFFLUENTS

		Coefficient	
Data Set	BOD	COD	TSS
PLASTICS ONLY	-0.13	0.10	0.04
ORGANICS ONLY	-0.04	-0.03	-0.10
PLASTICS AND ORGANICS	0.07	0,12	-0.21
ALL PLANTS	-0,06	0.03	-0.11

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characteristics of the wastewater they process. By considering each wastewater stream to be treated individually, and judiciously selecting the optimum combination of source control, pretreatment and treatment technologies, treatment of OCPS wastewaters to very low effluent BOD levels will be possible in all but the most extreme cases. Specific means of mitigating temperature aspects are discussed later in this chapter.

Aerated Lagoons

Aerated lagoons are stabilization basins to which air is added either through diffusion or mechanical agitation. The air-provides the oxygen required for aerobic biodegradation of the organic waste. In some designs the air addition will provide sufficient mixing to maintain the biological solids in suspension so that they can be removed in a secondary sedimentation tank. After settling, sludge may be recycled to the head of the lagoon. When operated in this manner, the aerated lagoon is an activated sludge process. The viable biological solids level in an aerated lagoon is normally low when compared to that of an activated sludge unit. The aerated lagoon relies primarily on detention time for the breakdown and removal or organic matter. Aeration periods of 3 to 8 days are common.

Twenty-seven of the 291 plants included in the Summary Data Base reported using aerated lagoon treatment. A summary of the performance of these plants is presented in Tables 7-8 thru 7-12.

Aerobic Lagoons

Aerobic lagoons are shallow ponds which contain bacteria and algae in suspension, with aerobic conditions prevailing throughout the depth of the basin. Waste is stabilized as a result of the symbiotic relationship between aerobic bacteria and algae. Supplemental oxygen is provided through natural reaeration. Bacteria break down waste and generate carbon dioxide and nutrients (primarily nitrogen and phosphorus). Algae in the presence of sunlight utilize the nutrients and inorganic carbon and, in turn, supply oxygen that is utilized by aerobic bacteria. Aerobic lagoons are usually less than 4-6 feet deep and can be periodically mixed to maintain their aerobic conditions. Algae do not settle well using conventional clarification. In order to achieve effective pollutant removals with aerobic lagoons, some means of removing algae (coagulation, filtration, multiple-cell design) is sometimes necessary.

A total of four OCPS industry plants use aerobic lagoons as the principal component of their treatment system. A performance summary for these four plants is presented as Table 7-13.

Anaerobic Lagoons

An anaerobic lagoon is deoxygenated throughout its depth and has the advantages of low sludge production and operating costs. Treatment results from a combination of precipitation and anaerobic decomposition of organics, initially to organic acids and cell tissue, and ultimately to carbon dioxide, methane and other gaseous end products. Anaerobic

#### AERATED LAGOON, ALL MASTE STREAMS

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEWATER TREATMENTS 27

	EFF FLO# MGD	800 1 NF MG/L	BUD EFF MG/L	BOD 1 Red	133 1NF MG/L	133 EFF MG/L	tas 1 Red	COD INF MG/L	C 00 E P F MG/L	COD % RED	ULG Inf MG/L	OLG EFF MG/L	OLG 1 RED
				•••••••									
MAXIMUM	14.400	1137.0	274.0	98.4	2454.0	395,0	98.7	21178	1178.0	98.8	570.0	112.0	60.4
MEAN	3,158	479,8	54.4	90.2	550.8	÷2,6	51.2	4229,5	276,4	82,4	570.0	30,5	80,4
HINIMUM	0,008	94.0	7.0	45.8	23.0	2,0	-56,5	150.0	30.0	66,5	570,0	1.0	80,4
TEDIAN	0,812	514.5	15.0	¥3.0	285.0	33,0	<b>*1.1</b>	1327.0	135.0	83.5	570.0	11.5	80.4
NUMBER OF													

#### AERATED LAGOON, ALL WASTE STREAMS

NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS HAJOR MASTEMATER TREATMENTE 27

	10C ]NF MG/L	10C EFF MG7L	10c 1 RED	PHENOL JNF MG7L	PHENDL EFF MG7L	PHENOL X RED	NH3N INF MG/L	NH 3N LFF MG/L	NH SN 1 RED	CR INF MG/L	CR EFF Hg/L	CR X RED
******			*****		••••••					*****		
MAXINUM	2056.0	133.0	85.1	\$342.0	14,0	99,7	19.0	240.0	80,5	0.4	1.3	54.0
MEAN	503.7	48.1	61.9	954,1	4,3	99.5	10,4	54,4	52.7	0.4	0.3	54.6
MININUM	20.0	14.0	25,0	2,5	0.0	99,2	1.8	1,1	22,25	0.4	0,1	54.0
MEDIAN	• 332.0	34,0	63.9	709.5	0,8	99.6	10.4	3.7	54,0	0.4	0.2	54.6
NUMBER OF Observations	,	7	•		•	3	4	,	4	1	8	1

a These data are from plants that use this technology as the principal component of their vas:cwater treatment system.

# ALANICU LADOCHA PLASTICS ONLY D Mumber of Streams Reporting this technology as major vastevatem treatments B

	EFF FLOW HOD	800 INF	800 EFF MG (1	80D 8	TSS INF	155 EFF	155 8	COD INF	C00 EFF	COD	040 INF	OLG EFF	040 N
MUNTANT	8.950	447.0	86.0	<b>40,4</b>	443,0	38.0	97.0	531.0	334.0	84,0	•	14.0	•
HEAN	1+214	220.0	28.1	49.0	443.0	18.5	97,0	358.0	126.0	77.2	•	7.5	•
ALN THUM	0.167	94.0	7.0	s#,2	643,0	11.0	97.0	179.0	30.5	70.4		1.1	•
EDIAN	0.684	119+0	15.0	88,3	643.0	15.0	47.0	355.0	104.5	77.2	•	7.5	•
NUMBER OF		3	7	t	1	٠	1	2	٠	2	0	2	

# AERATED LAGODN, PLASTICS ONLY

# A NUMBER OF STHEAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTS 8

	10C INF MG/L	TOC LFF MG/L	TOC X RED	PHENOL INF MG7L	PHENOL EFF MG/L	PHENOL X RED	NH3N 1NF HG7L	NH3N EFF MG/L	NH 3N 1 RED	CR INF MG/L	CR EFF Mg/L	CR T RED
			••••••		••••••		••••••		••••••		 0.4	•••••
MEAN	•	•	•	•	0,1	•	•	•	•	•	0,4	•
ATHING .	•	•		•	0.1		•	•	•		0.4	•
EDIAN	•	•	•	•	0.1	•		•	•	•	0.4	•
NUMBER OF Deservations	0	0	(	, o	1	0	0	•	0	0	1	0

a. These data are from plants that use this technology as the principal component of their wastewater treatment system.

TA	DI	E.	- 7	- T -	<u> </u>
1 A	ЪL	,Ľ.	- / -	- ±'	υ

#### ACHATED LAUGUNE NOT PLASTICS ITYPE 1 & CH

#### NUMBER OF STREAMS REPORTING THIS TECHNOLLOGY AS MAJOR WASTEWATER TREATMENTE .

	FLOW MGO	800 1NF H0/L	00 277 40/L	800 8 8 039	133 INF M0/L	T\$\$ EFF H0/L	TSS REp	COD 1NF Ha/L	CON EFF MO/L	CDD B RED	049 INF #8/L	010 277 M0/L	OLG B PED
					********	***				******			
MAXIMUM	18.800	823.0	251.0	48,4	66B0	395.0	61.1	21178	1178.0	44,8	•		•
HEAN	3.210	534.7	72.3	87.2	2146	121+4	17.2	9239.0	413.9	67,3	•	9.0	•
NINIMUM		199.0	7.0	65,8	830	20.0	+56,8	1327.0	135.0	41+1	•	9.0	•
MEDIAN	1.010	582.0	41.0	97.4	730	62.0	25.9	7225.8	256.0	88.7	•	9.0	•
NUMBER OF	• 21	,	7	3		•	5	•	7	•		١	

#### AERATED LAGOODN, NOT PLASTICS (TYPE I & C)

# NUMBER OF STREAMS REPORTING THIS TECHNELLOGY AS HAJOR MASTEMATER TREATMENTS 9

	TOC 1NF MG7L	TOC EFF MQ/L	TOC X RED	PHENC INF MG/L	PHENDL EFF MG7L	PHENOL 1 RED	NH3N 1N7 MG7L	NH 3N EFF MOZL	NH 3N T RED	CR INF MG/L	CR EFF MOVL	CA I RFD
MAXINUN	513.0	64.0	88.1	974?	1.4	99.7	19.0	125.0	80.5	0.4	1.3	54.4
MEAN	288.3	46.7	64,6	709.35	0,7	99,7	15,5	34.0	65.3	0.4	0.4	54.6
NINIMUN	50.0	15.0	25.0	445	0.0	99,7	12.0	1.1	50,0	0,4	0,1	54,6
MEDIAN	332.0	61.0	80.7	709.55	0.7	99.7	15.5	4,8	65.3	0.0	0.2	54.0
NUMBER OF Observations	3	3	3	22	z	1	2	4	2	1	5	1

\* 1 Type I w/oxidation

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

### ACTAILS LAUGH, AUX PLASISES (TIPE & AUT C)\*

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTS 4

	EFF	BOD	404	BOD	153	138	T 5 3	C 00	C O D	C 0 9	044	DIC	049
	FLOM NGD	INF HG /I	EFF Mc/I	Ren.	INF Hc/L	EFF NG/L	1 Arn	INF MCZI	EFF MC/I	1 Rco	INF No/I	EFF	I
													*******
NUNTXAN	3.400	1137.0	80.0	98.0	2686.0	÷2,0	98.7	5546.0	354.0	96.5	570.0	112.0	80.4
HEAN	1,200	813,9	37.8	93.8	1390,3	31.0	74,9	2148.3	141,3	82.0	570.0	56.5	80.4
WINING	0.030	647.0	11.0	87.6	494.0	13,0	67.4	120.0	30.0	66,7	570.0	1.0	80.4
MEDJAN	0,762	655 <b>.</b> 0	30.0	95.9	991.0	24.5	98,T	1443.5	40.0	82.9	570.0	56.5	80.4
NUMBER OF						-							

#### AERATED LAGOON, NOT PLASTICS (TTPE I NOT C)\*

#### NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR<sup>4</sup>HASTEMATER TREATMENTE &

	TOC INF MG/L	TUC EFF MG7L	TOC I RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL 1 RED	NH3N INF MG/L	NH3N EFF MG7L	NH 3N 1 RED	CR ]NF MG/L	CR EFF MG/L	CR X RED
MAXIMUM	2056.0	14.0	• • • • •	2,5	 14,0	99,2	1.8	1.4	22,2	•	•••••	• • • •
MEAN	2056.0	14.0		2,5	7.0	99.2	1.8	1.4	22.25	•	0,1	•
HINIHUM	2056.0	14.0		2,5	0.0	99,2	1.8	1.4	\$5.5	•	0.1	•
MEDIAN	2056.0	14,0	•	2.5	7.0	99.2	1.8	1.4	22.2	•	0.1	•
NUMBER OF DBSERVATIONS	1	1	(	<b>b</b> 1	2	1	1	1	ı	0	1	•

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

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7 Type 1 w/s Oxidaties

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TABL	E 7	-12
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ACRATED LAGOON, NOT PLASTICS INOT TYPE IS a NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR WASTEWATEM TREATMENTE 6

	EFF Flow MgD	800 Inf H0/L	800 877 80/1	NOD N RED	T59 INF H0/L	155 EFF W0/L	TSS RED	C00 187 40/L	000 277 40/1	COD N Fred	040 185 46/L	010 677 M0/L	OLA N RED
,									-* -*	•*			
	19,900	95.0	279.0	91.6	138,0	32,9	89,1	266.0	1045,0	64,5	•	44,0	
HEAN	7.665	93.8	99.7	91.6	86,0	16.3	47.5	246.0	241.3	60.5	•	46.0	•
KINIMUM	0.008	95.0	8.0	91,6	34,0	2.0	5,9	266,0	40.0	66,5	•	44.0	•
EDIAN	4.090	95,6	12.0	91,6	86.0	15.0	47,5	266.0	89.0	66.5	•	46.0	•
NUHBER OF													

#### AERATED LAGOON, NOT PLASTICS (NOT TYPE I)

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJUR MASTEMATER TREATMENTS &

**	TOC INF MG/L	10C EFF MG/L	10c I RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL I RED	NH 3N INF MG/L	NH 3N EFF MG/L	NH3N I RED	C <sup>R</sup> ]nf MG/L	CR EFF MG/L	CR I RED
											•••••	
HAXIHDH	502.0	133.0	73,5	2395.0	10.0	44.6	8.8	240,0	28.0	•	0.1	•
MEAN	201.7	٥ <b>٥</b>	59,3	\$342.0	10.0	49.6	6,8	151.8	58.0	•	0,1	•
WIFINAH	35.0	16.0	50.0	2395.0	10.0	99.6	8.8	3.7	58.0	•	0.1	•
MEDIAN	68.0	34.0	54.3	2395.0	10.0	99.6	8.8	121.8	58.0	•	0.1	•
NUMBER OF Observations	3	3	3	1	۱	1	i	2	1	0	1	•

a These data are from plants that use this technology as the principal component of their vastewater treatment system.

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#### ARMULIC LAUDING ALL MADIE SINEAME

#### NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR WASTEMATER TREATMENTS S

	EFF FLOM Mgu	BOD Inf MG/L	800 EFF MG/L	800 1 RED	133 JNF MG/L	135 EFF MG/L	T33 I RED	COD INF HG/L	COD Eff Mg/L	COD X R(D	ULG INF MG/L	OBG EFF Mg/L	DIG I RED
MAXIMUM	3.320	78.0	•2.0	• 88.5		37,0	••••••	••••••	252.0	•	• • • • •	•.۰	•••••
MEAN	1,479	78,0	17.6	88.5		16.4	•	•	168,5	•		8.0	•
WINIHUM	0,155	78,0	7.0	88,5	•	4,0		•	85,0	•	•	8.0	•
MEDJAN	1.500	78.0	10.0	ŧ8,5	•	16,0		•	168,5	•		8.0	•
NUMBER OF OBSERVATION:	s 5	1	5	1	0	5	0	•	2	٠	•	1	٠

#### AERDBIC LAGDON, ALL MASTE STREAMS

# A NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS HAJOR MASTEMATER TREATMENTI S

	TOC 1NF MG/L	TUC EFF MG/L	TOC 1 RED	PHENUL INF MG/L	PHENOL EFF HG/L	PHENOL X RED	NH SN INF MG/L	NH3N Eff HG7L	NH 3N 1 RED	CA ]nf MG/L	CR EFF Mg/L	CR I RED
· · · · · · · · · · · · · · · · · · ·			71.2		6.6			0.4		••••••		55,1
MEAN	66.0	33,5	71.2	•	0.4			0,4	•	0,1	0,1	55.1
HIWIHUM	66. <sup>0</sup>	19.0	71.2		0.1	•		0,4	•	0,1	0.0	55,1
MEDIAN	66,0	33.5	71.2	•	0.4	•	•	0.4	•	6.1	0.0	55.1
NUMBER OF DBSERVATIONS	1	2	i	0	2	0	0	1	0	1	3	1

a These data are from plants that use this technology as the principal component of their Mastewater treatment system.

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lagoons are constructed with depths to 20 feet and steep side walls to minimize surface area (relative to total volume) to allow a natural organism cover (pellicle) to form and help retain heat, suppress odor, and maintain anaerobic conditions. Wastewater enters near the bottom and the discharge point is located opposite and below the pellicle. Sludge recirculation is not necessary because gasification and the inlet-outlet flow pattern provide adequate mixing. Anaerobic lagoons are sometimes used to digest the waste sludge from an activated sludge plant. Anaerobic lagoons as a principal wastewater treatment technology are used at two plants in the 291 plant Summary Data Base. BOD removals greater than 90 percent were reported by these plants. Additional performance data is presented in Table 7-14.

Activated Sludge

Activated sludge is an aerobic biological process. Its basic processes include an aerated biological reactor, a clarifier for separation of biomass, and a piping arrangement to return separated biomass to the biological reactor. Aeration provides the necessary oxygen for aerobic biodegradation and mixing to maintain the biological solids in suspension.

Activated sludge process modifications commonly in use include conventional, step-aeration, tapered-aeration, modified-aeration, contact stabilization, complete-mix, extended-aeration and oxygen activated sludge. Activated sludge is the most common end-of-pipe treatment employed in the OCPS industry Summary Data Base. Of the 291 plants making up the data base, a total of 104 use activated sludge, treating 107 separate waste streams. Tables 7-15 thru 7-19 summarize the performance of these activated sludge plants by OCPS industry proposed subcategories.

Pure oxygen activated sludge was reported to be the principal treatment technology at three plants. A performance summary of these plants is presented in Table 7-20.

Attached Growth Biological Treatment

In attached growth biological systems the biomass adheres to the surfaces of rigid supporting media that contact the wastewater. Systems of this type that are in common use in the OCPS industry include trickling filters, packed towers and rotating biological contactors. While the physical structures differ, the biological process is essentially the same in all attached growth systems.

As wastewater contacts the supporting medium, a thin-film biological slime develops and coats the surfaces. The film consists primarily of bacteria, protozoa, and fungi that feed on the waste. As the slime grows, it separates or sloughs off. The sloughed biomass is then removed in a secondary clarifier.

Trickling filters are classified by hydraulic or organic loading as "low rate" or "high rate." Low-rate filters generally have a depth of six to ten feet and no recirculation. High-rate filters have a depth of three to ten feet and a recirculation rate of 0.5 to 4.0. High-rate filters

#### ANAERUDIC, ALL NASTE STREAMS

# NUMBER OF STHEAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTS 2

	EFF FLOM MgD	BOD 1 NF Mg/L	800 6FF Mg/L	BOD N RED	TS& INF Mg/L	138 EFF Mg/L	135 1 RED	CUD 1nf MG/L	COD EFF MG/L	CUD X RED	UEG INF Mg/L	OLG EFF Mg/L	OLG 1 RED
MTXIMA	12,200	970.0	22.0	•7.7	•	86.0	•	2189.0	147.0	93,3	••••••	• • •	• • • • • • •
MEAN	8,225	694,5	20.5	95.9		86,0	•	1151.0	87,5	70,2	•		•
MININUM	4,250	319.0	19.0	¥4,0	•	86,9	•	53.0	28.0	47.2	•	•	•
MEDIAN	8.225	644,5	20.5	95.9		86.0	•	1121.0	87.5	70.2	•		•
NUMBER OF Observation	s 2	2	2	- 2	•	ı	•	2	2	2	٥	0	•

#### ANAEROBIC, ALL HASTE STREAMS

#### a NUMBER OF STREAMS REPURTING THIB TECHNOLOGY AS MAJOR WASTEWATER TREATMENTS 2

	10C INF MG/L	TOC EFF MG/L	10C 1 RLD	PHENOL INF MG7L	PHENDL EFF MG/L	PHENOL X RED	NH3N INF MG/L	NH 3N EFF MG7L	NH 3N 1 RED	CR INF MG/L	CR EFF Hg/L	CR I RED
			****									
MUNIIAM	695,0	74.0	89,4	•	•	•		•	•	•	•	•
MEAN	411.5	51.0	83.7	•			•		•		•	•
HINIHUM	128.0	28.0	78.1		•			•	•		•	•
MEDIAN	411.5	51.0	83.7	•		•	•	•		•	•	•
NUMBER OF DB3ERVATIONS	2	2		0	0	٥	٥	. 0	0	0	•	

a These data are from plantsthat use this technology as the principal component of their wastewater treatment system.

#### ACTIVATED SLUDGE, ALL MASTE STREAMS

#### NUMBER OF STHEAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTERATER TREATMENTS 107

	EFF Flum Mgd	BUD Inf Mg/L	800 LFF MG/L	BDD 1 Red	T33 Inf Mg/L	ISS EFF MG/L	T 8 S Red	COD Inf Mg/L	CUD EFF MG/L	C D D 1 R E D	ULG Inf NG/L	OLG EFF MG/L	DLG 1 RED
							•••••			••••			
ME & M	2.472	1214.4	37.4	93.4	512.9	69.1	43.3	32476	459.2	81.3	58.1	10.8	59.3
WINIHUM	0,008	14.0	3,0	47.5	17.0	7,0	-271,4	210.0	36,0	18,1	14,0	0.4	12,6
MEDIAN	1.070	754.0	24.0	95.6	114.9	45.0	62.9	1686.0	186.0	84.7	23.0	6.0	67.6
NUMBER DF Orservati	ONS 197	85	101		54	98	54	73	79	70	7	17	٠

#### ACTIVATED SLUDGE, ALL HASTE STREAMS

#### B NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTS 107

••••••	TOC Inf MG/L	TUC EFF MG/L	10c 1 RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL 1 RED	NH3N INF MG/L	NH3N EFF MG/L	NH 3N 1 RED	CR 1NF HG/L	CR EFF Mg/L	CR T RED
HAA1HUM	5226.0	605.0	•••••	 747.0	 36,0	100.0	390.0	274,0	93.8	2,3	10.0	•7,2
	1025.8	123.2	80.7	145.6	2.3	92.2	75.3	33.4	10.9	0.6	0.5	-80,2
13N1MUM	67,0	7.0	36.5	0,1	0.0	61,5	0,8	0.4	-347,4	٥.٥	0.0	+1324
EDJAN	505.0	69.0	89.6	14.3	0.1	99,7	30.0	6.9	27.0	0,2	0.1	73.0
NUMBER OF Observations	29	29	24	18	28	18	53	40	22	12	21	11

a These data are from plants that use this technology as the principal component of their vastevater treatment system.

#### الاسلام ويقياه وللمانين المناه المراجع

NUMBER OF STREAMS REPORTING THIS TECHNOLODY AS HAJOR VASTEWATER TREATMENTS 40

	EFF FLOW MgD	800 1NF HG/L	800 EFF HG/L	800 8 800	155 1NP H8/L	155 EFF HO/L	165 9 80	COD 1NF H0/L	COD EFF MG/L	COD N RED	049 INF Mg/L	040 877 H0/L	OLG N AED
*****									••				
MAXIMUN	10.700	3920.0	58.0	99.7	2898, 0	179.0	1 99,2	4338.0	620.0	98.1	242.0		73.9
MEAN	1+638	616.8	17.8	94,3	803.L	44,6	52.1	1302.1	128.8	84.5	102.0	18.5	69.7
KINIHUM	0.034	14.0	3.0	78.1	22, D	7.0	-45,5	210.0	36.0	48,2	83.0	1.2	44 . T
HEDIAN	0,683	300.0	10.5	96,5	101.0	30.0	66,0	900.0	97.0	87.0	41.0	7.6	48,3
NUHRER OF	0N3 40	35	38	35	3.1	39	31	33	34	31	.)		3

#### ACTIVATED BLUDGE, PLASTICS ONLY

#### a NUMBER OF STREAMS REPORTING THIS TECHNIILOGY AS MAJOR NASTEWATER TREATMENTS 40

	10C 14F MG/L	10C EFF MG/L	TOC 1 RED	PHENIOL INF MG711	PHENDL EFF MG/L	PHENDL X HED	NH3N INF MG/L	NH3N EFF MG7L	NH 3N X RED	CR INF MG/L	CR EFF MG/L	CR I RED
	2751.0	96.0	99.5	426.0		100.0	49.0	109.0	89.6	2.0	0.3	•7.2
EAN	1103.4	48.2	86,7	716	0,1	86.5	16,5	22,8	-27,0	0.6	0.1	53,5
<b>VINIHUM</b>	\$60"0	15,0	66,9	0,1	0.0	61,5	5.0	0,4	-347,4	0,1	0.0	-50,6
EDIAN	464.0	43.5	90.3	0.1	0.0	92.4	8.0	6.0	25.0	0.2	0.0	83.4
NUMBER <b>OF</b> DBSERVATIONS	5	•	4	•	9		10	16	•	•	7	•

a These data are from plantsthat use this technology as the principal component of their wastewater treatment system.

# AC:174-20-00000 Not -241710, 1772 1 6 01

# NUMBER OF STREAMS REPORTING THIS TECH-OLOGY AS MAJOR WASTEWATER TREATHENTS 42

	FLOW MGD	BOD INF Mg/L	80D EFF HG/L	BOD S RED	TSS INF HG/L	135 EFF H0/L	TSS NCD	COD 1NF HØ/L	COD 877 #0/L	COD N RED	040 105 Hg/L	010 677 H0/L	DLO S RED
										*			
NUMIXAN	29.000	\$961.0	468.0	99,6	4110.0	235.0	98,9	9229.0	4079.0	98.1	17+4	15.2	12.6
KEAN	2.859	1974.8		93,0	428.5	84.2	26.7	4242,9	867.8	79.8	14.7	7.2	12.6
INIMUN	0.008	183.0	4.0	47.8	28.0	9,0	-271.4	562.0	55.0	10.1	16.0		12.6
REDIAN	1+128	1396.0	80.0	95,6	84.0	71.0	57,1	4014.0	334.5	80.5	16+7	6.5	12.6
		-											
DRAFHAVIO	42 EF	34	41	34	13	31	13	25	28	84		•	1

# ACYIVATED SLUDGE, NOT PLASTICS (TYPE I & C)\*

O NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER TREATMENTE 42

	TOC INF MG/L	TDC EFF HG/L	TOC 1 RED	PHENOL INF MG/L	PHENOL EFF MG7L	PHENOL 1 RED	NH 3N INF MG/L	NH3N EFF Hg/L	NH3N 1 RED	CA INF HG/L	CA EFF Ma/L	CA X RED
	3202.0	605.0	97.7	747.0	36,0	100.0	390.0	274.0	91.1	2,3	0.5	80.0
(EAN	1056.9	156.4	78.9	269.7	5,5	94.6	137.1		37.1	0.8	0.2	22.9
INIMUM	268.0	23.0	36.5	1.4	0.0	80.0	2.4	1.8	-17.0	0.1	0.0	-66,7
EDIAN .	719.0	86.5	89.9	121.5	0.2	97.1	53.0	16.0	33.8	0.3	0.2	34,1
NUMBER OF Deservations	15	14	13	8	11	8	ę	17	ę	5	8	4

a These data are from plants that use this technology as the principal component of their vasiewater treatment system.

A Type I w/ Oxidation

# ACTIVALED SCUDGE, NOT FLASHED ETTRE & HUN 63 \*

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTENATER TREATMENTE IS

	EFF Flom Mgd	AOD 1nf Mg/L	800 EFF HG/L	BOD 1. Red	T38 ]nF Mg/L	133 EFF HG/L	T33 X Red	COD 1nf MG/L	COD EFF MG/L	COD 1 RED	OLG Inf Hg/L	OKG EFF Mg/L	OLG 1 RED
*******		••••••											••••••
MUNITAN	4,310	2725.0	780.0	97.7	780.9	851,0	85.0	32476	11000	98.3	17.0	15.0	35.3
MEAN	1,533	1267.7	112,2	90.4	345,2	136,1	41,2	8883.4	1813.5	82,5	17.0	13.0	35,3
HINIMUM	0.020	66.0	11.0	70.8	133.0	18.0	-9.1	333.0	40.0	66.1	17.0	11.0	35,3
MEDIAN	1.440	1015.5	28.0	92.7	177.0	71.0	40.1	3449.0	357,5	79.1	17.0	13.0	35.3
NUMBER OF Deservation	Ng 15	10	13	10	3	11	5	7	8	7	1	2	ì

#### ACTIVATED SLUDGE, NOT PLASTICS (TYPE I NOT C)

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER THEATMENTE IS

	10c Inf Mg/L	10C EFF MG7L	10c 1 RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL X RED	NH 3N 1NF MG/L	NHJN LFF MG/L	NH 3N 1 RED	CR ]NF MG/L	CR Eff Mg/L	CR X RED
MAXIMUM	5226,0	501.0	97.5	18.0	1.0	99.1	233.0	80.0	••••••		0.4	•
4EAN	1144.1	131,5	80.4	14,3	0.7	95.8	132,5	28,2	79.7	•	0.4	•
NINIHUM	67.0	7.0	48.4	10.7	0,2	92.6	32.0	1,8	65.7	•	0.4	•
TEDIAN	431.0	79.0	79.7	14.3	0.8	95.8	132.5	15.5	79 <b>.7</b>	•	0.4	•
NUMBER OF DBSERVATIONS	7	a	7	2	3	z	2	4	2	0	1	0

★ Type I v/o Oxidation

a These data are from plants that use this tech ology as the principal or popert of their last we er treatmont system.

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ACTIVATED SLUDGE+ NOT PLASTICS (NOT TYPE I) A Number of streams reporting this technology as haugr vastevater treatments ig

********													
·····	EFF Flow MgD	H0\L 101 900	тол <i>Ерр</i> на/с	POD B RED	155 JNF M0/L	135 <i>EFF</i> H0/L	TSS B RED	C00 I№# H0/L	COD EFF Hg/L	UOD B RED	014 JNF M0/L	010 277 H0/L	DLA B RED
		•								*-***		·F	
MUNIKAN	40.000	920.0	27.0	97.0	1266.0	101.0	97.3	5488.0	413.0	¥#.0	50.0	3,0	98.7
PEAN	5.598	353.5	15.6	15,5	437.0	41.E	33.7	1709.3	159.2	72.6	50+8	1.5	98,7
HINIKUN	0.335	162.0	3.0	91.5	17.0	18.0	-70.6	218.0	65.0	44,5	50.0	0.7	<b>*#.</b> T
PEOTAN	1+445	324+0	14.0	96,1	121.0	27.0	35,5	442.0	111.0	72.5	50.0	0	48.7
NUMBER OF OBSERVATIO	NS 10	•	٠	6	5	,	5		•	•	1	3	1

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#### ACTIVATED SLUDGE, NOT PLASTICS (NOT TYPE 1)

B NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER THEATMENTS 10

	10C 1nf MG/L	10C EFF MG/L	10C 1 RED	PHE 40L INF MG/L	PHENOL EFF HG/L	PHENOL 1 RED	NH 3N [NF MG/L	HH3N EFF HG7L	NH 3N X RED	CR 1 <i>NP</i> HG/L	CR EFF MG/L	CR X RED
	194,0	41.0	••	• 4.0	0,2	96.9		64.1	2.3	•••••	10.0	90.5
EAN	184.5	41.0		2,2	0.1	95,9	33.2	22,4	-5.1	0.3	z.1	-396.0
THINUM	175.0	41.0		9.3	0.0	95.0	0.8	0.9	-12,5	0.0	0,0	-1329
-EDJAN	184.5	41.0		2.2	0.1	95.9	33.2	2.2	-5.1	0,2	0.1	50.0
VUMBER OF Deservations	2	. 1		) 2	5	2	2	3	2	3	5	3

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

# Pure Caugen Activated Sludge

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#### ALL MASIE UIPEAMS

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR RABIERATER TREATMENTS 3

	EFF Flow Mgd	BUD Inf Mg/L	BUD EFF Mg/L	800 X R[D	133 Inf Mg/L	133 EFF MG/L	158 1 Red	COD Inf MG/L	COD EFF MG/L	COP 1 RED	OLG Inf MG/L	OLG EFF Mg/L	OLG X RED
••••			•••••				•••••						
MAXINUM	7.160	467.0	20.0	97.2	204.0	38.0	94,1	345.0	105.0	85,2	•		•
4E AN	3.830	270,0	18.7	90.8	131.0	25.0	÷4,3	322,5	78,0	75.1		•	•
INTHIN	0,950	143.0	13.0	67.0	58.0	12.0	34,5	300.0	51,0	65,0	•	•	•
EDIAN	3,300	200.0	17.0	88.1	131.0	25.0	64.3	322.5	78.0	75.1			•
HUMBER OF	DNg 3	ì	3	,	2	2	2	z	2	2	0	0	•

#### Pure Oxygen Activated Sludge

#### ALL HASTE STREAMS

#### A NUMBER OF STREAMS REPORTING THIS TECHNULOGY AS MAJOR MASTEMATER TREATMENTE S

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	10C ]NF 467L	10C EFF MG/L	IDC I RED	PHENUL 1NF MG/L	PHENOL EFF MG/L	PHENOL X RED	NH3N INF MG/L	NH3N EFF MG/L	NH3N R RED	CR INF HG/L	CR EFF MG/L	CR X RED
		•••••		•••••		••••••				•••••		
MUHIKAM	75.0	26,0	65,3	8.3	0,4	99.0	•	•	•	•	•	•
MEAN	75.Q	20.0	65.3	5.1	0.2	96.8	•	•	•	•	•	•
HINIMON	75.Q	20.0	٥5 <b>,</b> 3	2,0	0,0	94,7	•		•	•	•	•
MEDIAN	75.0	26.0	65.3	5.1	0.2	96.8	•	•	•	•	•	•
NURBER OF Observations	ı	1	1	2	2	2	0	0	•	0	0	•

a These data are from plantsthat use this technology as the principal component of their wastewater treatment system.

can be single or two stage. The most suitable medium in both the low and high-rate filters is crushed rock.

In the OCPS 291 plant Summary Data Base three plants reported using trickling filters as their principal technology for EOP treatment. A performance summary for these three plants is presented as Table 7-21.

Packed towers are much like conventional trickling filters, but use a manufactured medium instead of crushed rock or gravel. The manufactured medium can be corrugated plastic packing or rough-sawn redwood slats. These media have high specific surfaces  $(ft^2/ft^3)$ , a high percentage of void volume, uniformity for better liquid distribution, chemical resistance, light weight facilitating construction of deeper beds, and the ability to handle high-strength and unsettled wastewaters. Packed towers are used in flow patterns similar to normal high-rate, natural-media filter systems.

In rotating biological contactor systems a series of disks constructed of corrugated plastic plate and mounted on a horizontal shaft are placed in a contour-bottomed tank and immersed to approximately 40 percent of the diameter. The disks rotate as wastewater passes through the tank and a fixed-film biological growth similar to that on trickling filter media adheres to the surface. Alternating exposure to the wastewater and the oxygen in the air results in biological oxidation of the organics in the wastes. Biomass sloughs off (as in the trickling filter and packed tower systems) and is carried out in the effluent for gravity separation. Direct recirculation is not generally practiced with the rotating biological disks.

Four plants in the OCPS Summary Data Base use rotating biological contactors as their principal form of treatment. All four of these plants are "plastics only" facilities. A summary of their performance is presented in Table 7-22.

#### Tertiary Treatment

In some instances, where secondary treatment does not produce a satisfactory effluent, polishing or tertiary treatment is utilized. The addition of a tertiary unit process does not always result in an effluent of higher quality than can be achieved with biological treatment. Often tertiary treatment is used to compensate for inadequately designed or improperly operated biological systems. Depending on the nature of the pollutant to be removed and the degree of removal required, the polishing or tertiary treatment system can consist of a one unit operation or multiple-unit operations in series. Some of the unit operations used in tertiary treatment may also be used as in-plant treatment options.

Polishing Ponds - Polishing ponds serve as polishing steps following other biological treatment processes. They primarily serve the purpose of reducing suspended solids. Water depth generally is limited to two or three feet. Polishing ponds are commonly used as a final process.

<u>Powdered Activated Carbon</u> Treatment - Powdered activated carbon treatment (PAC) refers to the addition of powdered carbon to the aeration

#### Angunuan - LIERS ALL - "WILL WINLAW 8

NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJUR MASTEMATER THEATMENTS 3

	EFF	800 I NF	BUD EFF	800	133 [NF	TSS EFF	183 1	COD Inf	C00 [ff	C 00 I	ULG INF	DEG	OLC I
	#GD	MG7L	=G/L	#ED	#G/L	MG/L	RED	HG/L	MG/L	RED	MG/L	MG/L	RED
MAXINUM	3,570	481.0	31.0	95,2	1225.0	55,0	95.5	1979.0	250.0	67.4	•	•	•
MEAN	1.731	310.0	26.0	40.0	627,5	34.0	76.1	1021.7	163.0	79.1	•	•	•
WININNW	0.423	170.0	23,0	89.9	30.0	13.0	56.7	210,0	63.0	70.0	•	•	•
MEDIAN	1.200	279.0	24.0	68.9	427.5	34.0	76.1	876.0	176.0	79,4	•		•
NUMBER OF Observatio	DNS 3	3	3	3	z	2	2	3	3	3	0	•	•

# TRICKLING FILTER, ALL MASTE SIREAMS a Number of Streams Reporting This Technology as major kastemater treatments 3

	10C 1NF MG/L	10c EFF MG/L	TOC 1 RED	PHENUL INF MG/L	PHENOL EFF MG/L	PHEHOL X RED	NH3N Inf MG7L	NH3N EFF MG/L	NH3N 1 RED	CR 1nf MG/L	CR EFF MG/L	CR X RED
MATINUN	••	••••••	• • • •	•	•••••••	•	3.0	1.0		•	•••••	• •
MEAN	•	•			•	•	3.0	1.0	66.7	•		•
WIFIHON	•	•	•	•	•	•	3,0	1.0	66.7		•	•
MEDIAN	•					•	3,0	1.0	<b>66.</b> 7	•	•	•
NUPBER OF Observations	0	•	c		0	0	ı	1	1	٥	0	0

a These data are from plants that use this technology as the principal component of their wastewater treatment system.

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# HUIR, BIGLUUICHE CUITREICHT ALL RADIE UIREANS

#### NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOR MASTEMATER THEATMENTS 4

•	EFF FLOH HGD	BUD 1nf #67L	800 Eff MG/L	BOD 1 R(D	TSS INF Mg/L	193 EFF HG/L	193 1 RED	CUD Inf Mg/L	COD EFF Mg/L	COD 1 RED	UIG INF NG/L	016 EFF Mg/L	OLG R PED
**************************************	2,740	 1200.0	52.0	•-•• 95.7	49,0	•3.0	25.0	2376.0	188.0	 92,1	•••••••	2,3	••••••
MEAN	0.904	434,0	30.5	80,4	34.5	29,3	18.0	859,8	99.0	73.5	•	2.3	•
wiwiwna	0,054	51.0	4.0	41.0	20.0	15.0	12.2	97.0	15.0	31,2		2,3	•
MEDIAN	0.411	242.5	33.0	92,5	34.5	30.0	18.4	483.0	96.5	85.4	•	2.3	•
NUMBER OF Observations		4		4	2	3	2	4	4		o	1	

#### ROTATING BIOLOGICAL CONTACTOR, ALL MASTE STREAMS

# NUMBER OF STREAMS REPORTING THIS TECHNOLOGY AS MAJOH MASTENATER THEATMENTE &

	10c jnf 467L	10C EFF HG/L	TÚC 1 RED	PHENOL JNF MG/L	PHENOL EFF MG7L	PHENOL X RED	NH3N INF MG/L	NH 3N EFF MG7L	NH SN B RED	CR INF MG/L	CR EFF MG/L	CA I RED
	100.0	71.0	29.0	••••••	••••••	••••••	••••••	29,0	••••••	•.1	0.1	
MEAN	100.0	71.0	29.0			•	•	29,0	•	0,1	0,1	88.9
MINIHUM	100.0	71.0	29,0		•	•	•	29.0	•	0.1	0.0	68 <b>.</b> 9
HEDIAN	100.0	71.0	29.0	•	•	•	•	29.0	•	0.1	0.1	88.9
NUMBER OF Observations			,	0	0		0	ı		1	2	1

a These data are from plants that use this technology as the principal component of their Wastewater treatment system. basin in the activated sludge process. It is a recently developed process that has been shown to upgrade effluent quality in conventional activated sludge plants. In the PAC treatment process the carbon concentration in the mixed liquor is generally equal to or greater than the volatile mixed liquor suspended solids level. The carbon and adsorbed substances are removed as part of the waste biological sludge.

Activated Carbon Adsorption - The use of activated carbon adsorption can be confined to the removal of specific compounds or classes of compounds from wastewater streams, or for the removal of such parameters as COD, BOD and color. Although more common as in-process treatment, it is also used for tertiary treatment.

An aspect of granular carbon carbon columns that is currently receiving attention is the role and possible benefits of biological growth on the carbon surfaces. In some applications much of the removal has been found to result from biodegredation rather than from adsorption.

Six plants in the Summary Data Base reported using activated carbon as their principal EOP treatment. The performance of these systems is summarized in Table 7-23.

Filtration - Filtration may be employed to polish an existing biological effluent, to prepare wastewater for a subsequent advanced treatment process, or to enable direct reuse of a discharge. Filtration of a secondary effluent will remove additional BOD and TSS, and reduce turbidity.

<u>Reverse Osmosis/Ultrafiltration</u> - Reverse osmosis is a physical separation process that relies on applied pressure at a level greater than osmotic pressure to force flow through a semi-permeable membrane. The process is capable of removing suspended particles and substantial fractions of dissolved impurities, including organic and inorganic materials. The process results in two effluents, one relatively pure and the other containing the concentrated substances. Reverse osmosis systems generally require extensive pretreatment (pH adjustment, filtration, chemical precipitation, activated carbon adsorption) of the wastewater stream to prevent rapid fouling or deterioration of the membrane surface.

Ultrafiltration is similar to reverse osmosis and relies on a semipermeable membrane and an applied driving force to separate suspended and dissolved materials from wastewater. The membranes used in ultrafiltration have pores large enough to eliminate osmotic pressure as a factor and to allow operation at pressures as low as five to ten psi. Sieving is the predominant mechanism of removal and the process is usually applicable for the removal of materials that have a molecular weight above 500 and that have very small osmotic pressures at a moderate concentration.

Combined Secondary and Tertiary Treatment System - In practice primary, secondary and tertiary processes are often used in series to treat OCPS industry wastewater. In fact, of the 146 plants employing biological treatment in the Summary Data Base, 58 use a form of treatment after

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# NUMBER OF STHEAMS REPORTING THIS TECHNOLOGY AS MAJOR DWASTEWATER TREATMENTS 7

	EFF FLOM MGD	BOD Inf Mg/L	BUD LFP MG/L	800 1 800	133 INF MG/L	193 EFF HG/L	133 1 REO	LOD ]nf Mg/L	CUD LFF MG/L	C00 1 AED	ULC Inf Ng/L	OLG EFF MG/L	OLG X RED
••••••													
HAXIMUN	0,334	1743.0	772.0	55,7	847.0	37.0	97.5	4556.0	1326.0	78.0	•	1.5	•
MEAN	0.135	1273.0	509.5	54.9	a33,0	23.0	51,4	2006.0	410.8	75.0		1.2	•
HIMIMOM	0.014	801.0	8.0	54,2	19.0	16.0	5,3	286.0	68.0	70,9	•	1.0	
EDIAN	9.127	1273.0	38.0	54.9	433.0	19,5	51.4	2976.0	171.0	76.2		1.2	•
UMBER OF OBSERVATIONS	•	2	•	2	2		2	3	•	3	•	2	•

#### ACTIVATED CARBON, ALL HASTE STREAMS

# NUMBER OF STREAMS REPORTING THIS LECHNOLOGY AS MAJOR MASTEMATER TREATMENTE T

	10C INF 46/L	10C EFF PG/L	1UC 1 RED	PHENOL INF MG/L	PHENOL EFF MG/L	PHENOL X RED	NH3N Inf MG7L	NH3N EFF MG/L	NH3N 1 RED	CR INF MG/L	CR EF <b>F</b> MG/L	CA X RED
MAXIMUM	1455.0	315.0	78,4	173.0	······ 2.2	99,8	••	••••••	••••••	• • • •	•.•	•
MEAN	713.5	111.0	78.4	127.0	1.0	99.0			•		0.0	•
нініном	42.0	24.0	76.4	88,0	٥,2	97.5			•	•	0.0	•
MEDJAN	773.5	52.5	78.4	120.0	0.6	99.7	•	•	•	•	0.0	•
NUMBER OF DBSERVATIONS	2	٩	1	s	3	3	0	0	•	0	1	0

a These data are from plause that use this technology as the principal component of rheir was evaluated treatment system.

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secondary. The most prevalent tertiary process in the industry Summary Data Base is polishing ponds. A total of 34 plants reported using polishing ponds. Filtration and a combination of filtration and polishing ponds are the next most common tertiary processes in use with 11 streams utilizing this process.

#### Design, Operation and Management Practices

The need for good engineering design, good operating practices and conscientious waste management is as important in waste treatment as in chemical manufacturing. The design of the system must be site specific in that it must consider raw waste components, organic and hydraulic load variations, manufacturing practices, waste temperature, operator capabilities and other considerations which may be unique to the site. Operating practices must be based on a thorough understanding of mechanisms at work and probable response to changes in operating conditions. Waste management must be considered when planning for production campaigns, prodution shutdowns and new product addition, and should also include contingency planning for mechanical failures, inadvertent discharges and treatment system upsets.

As previously stated, optimum treatment system performance is usually obtained under so called "steady state conditions." This condition could be approached in wastewater from a single product, continuous process manufacturing operation. Such a situtation is unfortunately uncommon in OCPS plants. Many OCPS plants produce a variety of products, often on a campaign basis, using production operations which may be either continuous or batch. This frequently results in wastewater which varies significantly in composition and quantity.

Equalization and storage is the primary design approach taken to minimize this problem. It may be possible in some instances to modify production schedules to avoid simultaneous multiple batch discharges or cleanup operations to avoid excessive peak loads. Treatment plant operators should be advised of known or anticipated waste load changes so that they may respond accordingly, i.e., increase aeration, divert and hold accidental discharges, increase chemical feed rates, etc.

Plants operating in cold weather conditions should recognize that unnecessarily excessive storage prior to treatment may reduce the temperature of the biotreatment system. Cold temperature operation may require insulation of treatment units, covering of open tanks, and tracing of chemical feed lines. Insulation of treatment units may include installing tanks inground rather than above ground, using soil around the walls of above-ground units to prevent heat loss, or providing enclosures around treatment units. Operators should recognize that during colder periods it may be necessary to maintain higher MLSS concentrations, which may in turn require greater operator attention to effluent solids concentrations.

Plants operating in hot weather climates may be required to reduce waste temperatures to maintain a suitable treatment environment. Natural or mechanically induced evaporation may be used to reduce waste temperatures. Control of toxic and inhibitory waste components may be required to avoid treatment system upsets. This may be accomplished by separation and segregation of the material at the point of waste generation, or destruction or removal of the material through the use of a pretreatment system. Waste components typically handled in this manner include cyanides, heavy metals and metallic sulfides. In the case of inhibitory components, equalization and subsequent dilution may be sufficient to eliminate the inhibition.

#### Upgrading Biological Treatment Systems

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Many treatment systems in the OCPS industry have undergone one or more major modifications to upgrade performance since the initial installation of the system. The four most common reasons for plant upgrading are:

- 1. To accommodate changing environmental regulations
- 2. To accommodate higher loads from expanded production facilities
- 3. To accommodate wasteloads associated with the manufacture of new products
- 4. To address inadequacies in the treatment system design

Because of the modular nature of most treatment systems, upgrading is most commonly accomplished by adding additional modules. When the upgrading is done to increase treatment system capacity, it is commonly done by adding modules similar to those already installed, i.e., additional aeration basins or clarifiers. Upgrading to accommodate more stringent treatment requirements usually involves adding new treatment process unit operations to an existing treatment train, e.g., addition of multi-media filtration following secondary clarification or addition of a coagulant feed system to a primary clarifier.

The nature of this evaluation of treatment system capability is best illustrated by use of an example. Consider a hypothetical chemical plant whose treatment system initially consists of a simple aerobic lagoon. The first level of upgrading could include providing aeration to convert the aerobic lagoon to an aerated lagoon. The next level of upgrading might include the addition of secondary clarification to return solids, thus converting the lagoon to an activated sludge system, providing some initial equalization capacity, and providing an aerobic digestor to stabilize waste secondary solids. The next level of upgrading could include the installation of a dissolved air flotation system to reduce influent suspended solids or oils, addition of multi-media filtration to reduce effluent solids, and the addition of solids dewatering equipment to allow for landfill disposal of waste activated sludge.

The OCPS industry provides several examples of similar upgrades:

Plant 53 - This plant was upgraded in June 1977. Facilities originally consisted of an equalization basin and aerated lagoon. The lagoon was converted to activated sludge by the addition of two clarifiers and additional aeration basin capacity. Solids handling equipment and an aerobic solids disgestor were also added. Change in operating performance was as follows:

	Before Upgrade	After Upgrade
Effluent BOD	316 mg/1	28 mg/l
Effluent TSS	63 mg/1	74 mg/l

<u>Plant 292</u> - This plant was upgraded in late 1977. The existing activated sludge system was upgraded by adding multi-media filters followed by granular activated carbon contactors. Other changes included an improved solids handling system comprised of gravity thickening, vacuum filtration and multiple hearth incineration.

	Before Upgrade	After Upgrade
Effluent BOD	20 mg/1	12 mg/1
Effluent TSS	56 mg/1	34 mg/1

This plant, and plant 53, are somewhat unusual in that they exhibit negative TSS removals. This will occur in any biological system where the loss of biological solids from the secondary solids separation system to the effluent is greater than the influent TSS received by the biological system. When a treatment system is achieving good secondary solids capture, typically 50 mg/l or less effluent TSS, the occurrence of a negative TSS removal percentage is not significant.

<u>Plant 60</u> - This plant was upgraded in 1977. Treatment originally consisted of equalization, neutralization, primary clarification, aerated lagoon and final clarification. The system was converted to completely mixed activated sludge. Aerobic digestion, gravity thickening, pressure filtration and an onsite landfill were also provided. Change in operating performance was as follows:

		Before Upgrade	After Upgrade
Effluent	BOD	510 mg/1	41 mg/1
Effluent		No Data	No Data

<u>Plant 45</u> - This plant originally had an activated sludge system with primary clarification. The plant was upgraded by the addition of a 3.5 million gallon equalization basin and mixed-media filtration. Additional aeration capacity was installed and a second secondary clarifier was added. New sludge handling facilities consisting of two pressure filters were installed to accommodate increased solids production.

	Before	After
	Upgrade	Upgrade
Effluent BOD	46 mg/l	3 mg/l
Effluent TSS	91 mg/1	24  mg/1

<u>Plant 109</u> - In 1976 this activated sludge system was upgraded through the addition of an extended aeration basin and multi-media filtration. Both units were added downstream of the existing treatment plant. In 1977, additional blower (aeration) capacity was added.

		Before Upgrade	After Upgrade
Effluent B	dop	12 mg/1	3 mg/1
Effluent 1	SS	83 mg/l	No Data

<u>Plant 118</u> - This plant was originally operated as a single stage trickling filter plant. In 1977 it was ugraded by the addition of a dissolved air flotation system to accomplish primary treatment, and added a UNOX pure oxygen system as a second stage biological treatment unit.

	Before Upgrade	After Upgrade
Effluent BOD	293 mg/l	13 mg/l
Effluent TSS	No Data	No Data

<u>Plant 269</u> - This treatment facility originally consisted of clarification, neutralization and activated sludge. In late 1977 it was upgraded by the addition of increased primary sedimentation capacity, the addition of an equalization basin prior to the activated sludge system, additional instrumentation, the addition of another secondary clarifier and improvements to the sludge handling facilities.

	Before Upgrade	After Upgrade
Effluent BOD	255 mg/l	66 mg/l
Effluent TSS	No Data	No Data

<u>Plant 281</u> - The original activated sludge system at this plant was upgraded in 1977. System improvements included the addition of an emergency holding basin and a stormwater holding basin, the addition of equalization upstream of existing treatment units, chromium reduction on a boiler blowdown stream and some in-plant flow reductions. An additional secondary clarifier was also added.

	Before	After
	Upgrade	Upgrade
Effluent BOD	15 mg/l	11 mg/1
Effluent TSS	46 mg/1	No Data

#### OCPS EFFLUENT QUALITY

Effluent quality in the OCPS industry, as defined by conventional pollutant parameters, is determined by several factors. Those factors with the greatest influence on effluent quality include: the origin of the wastewater, the type of treatment system used, and the design and operation of the treatment system.

The origin of the wastewater, which takes into account the type of products manufactured and manufacturing processes used, has already been discussed in Section IV. In that section a subcategorization scheme based on wastewater origin was developed.

In determining the effluent quality achievable by OCPS plants, biological treatment has been evaluated as the principal treatment practice within the industry. Of the 185 plants for which treatment system information is available, 146 use some form of biological treatment. Although nonbiological treatment systems are often used to produce high quality effluents, only biological treatment has been sufficiently applied to be considered as applicable across the broad spectrum of the OCPS industry.

The various biological treatment technologies differ only in the mechanical means by which the wastewater, biomass and essential nutrients are brought together. Although an activated sludge system and rotating biological contactor appear very different, the biological processes are similar. Therefore, it follows that all biological systems, including air and pure oxygen activated sludge, trickling filters, aerobic and aerated lagoons, and rotating biological contactors should, given sufficient detention time, achieve essentially the same effluent BOD<sub>5</sub> concentration. This is illustrated by Table 7-24 which presents summary performance data by subcategory for all biological systems, activated sludge systems and all biological systems other than activated sludge.

Although some variations are apparent, particularly where the data base in a given subcategory is limited, the data generally tend to support the above statement. On this basis, biological treatment in general may be considered the best technology for treatment of OCPS wastewaters. The specific mechanical system used to accomplish biological treatment will depend on cost, available space, climate and other site-specific considerations.

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# BIDLOGICAL SYSTEMS BODS EFFLUENT CONCENTRATIONS

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# ACTIVATED SLUGGE UMLY BODS EFFLUENT COMCENTRATIONS

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\* Type I w/Oxidation \*\* Type I w/o Oxidation

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The 146 plants in the Summary Data Base which use some form of biological treatment, treat a total of 176 different waste streams. Single stage biological systems are used to treat 138 waste streams. The remaining 38 waste streams are treated using two separate biological units in series. Table 7-25 compares the performance of single and two stage biological systems. It is apparent that plants using two stage biological treatment do not achieve lower effluent concentrations than single stage systems. In several subcategories, single stage systems produced lower effluent BOD, concentrations than two stage systems. This apparent contradiction may have resulted from the fact that the second stage of many two stage systems was added to upgrade a single stage system which was either poorly designed, poorly operated or overloaded follow ing installation. However, due to the absence of interstage monitoring data, this assumption cannot be verified.

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Use of biological treatment as the principal treatment practice will produce high quality effluents as shown in the previous tables. An evaluation was done to determine if additional treatment processes, i.e., polishing ponds and filtration, would further improve effluent quality. Tables 7-26 and 7-27 present summary data by subcategory which illustrate effluent BOD, concentrations from plants using biological treatment with polishing ponds and filtration processes. A comparison of the median concentrations in each category indicates that plants with additional processes do not achieve significantly different effluent BOD concentrations than plants with only biological treatment.

Although biological treatment has been demonstrated to achieve low effluent BOD<sub>5</sub> concentrations, the median value obtained for all biological systems in the industry is not considered to represent the best level of treatment which could be achieved. While some plants in the data base are well designed and operated, others are operating at less than optimum performance. In order to segregate good performers from bad performers, it was necessary to develop a statistical test to distinguish the better plants from those operating less efficiently.

Table 7-28 presents a summary of the BOD percent removals for biological systems in the Summary Data Base. The median for all systems (108 streams) is 95.2% BOD removal. The medians for all subcatgories are also approximately 95%. Table 7-29 presents effluent data for those plants achieving 95% removal. These show significantly better effluents than for all biological systems shown in Table 7-23. Based on this analysis, 95% BOD removal has been determined to represent well operated systems.

It is also recognized that use of the 95% removal criteria eliminates some plants which achieve lower effluent BOD<sub>5</sub> concentrations, but by virtue of having very low influent BOD<sub>5</sub> concentrations, do not achieve 95% removal. In addition, there are well operated plants that have not reported percent removal data. In an attempt to address this potential inconsistency, a new segment was evaluated which included all plants which achieved 95% BOD<sub>5</sub> reduction or which achieved an effluent BOD<sub>5</sub> concentration of 50 mg/l or less. The resulting frequency distribution of plants as a function of effluent BOD<sub>5</sub> concentration and the summary statistics are presented in Table 7-30. A second analysis was

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#### BIOLOGICAL SYSTEMS WITH FOLISHING BODS EFFLUENT CONCENTRATIONS

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#### BIOLOGICAL SYSTEMS WITH MMF BODS EFFLUENT CONCENTRATIONS

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I       II MINIMUM       3,0 II MINIMUM       6,0 II MINIMUM       13,0 II MINIMUM       5,0         I       II MAXIMUM       52,0 II MAXIMUM       154,0 II MAXIMUM       82,0 II MAXIMUM       24,0         I       II MEAN       16,0 II MEAN       43,3 II MEAN       34,3 II MEAN       14,0         I       II MEAN       9,5 II MEDIAN       33,0 II MEDIAN       34,3 II MEAN       14,0         I       II MEDIAN       9,5 II MEDIAN       33,0 II MEDIAN       25,5 II MEDIAN       12,0         I       II N       20 II N       23 II N       6 II N       11         II       II       II       II       II       11	.0    MINIMUME 3.0   .0    MAXIMUME 154.0   .0    MEAN 29.5   .0    MEDIAN 21.0   6    N 55    )
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\* Type I w/oxidation \*\* Type I w/o/oxidation

## BIOLOGICAL SYSTEMS BODS EFFLUENT CONCENTRATIONS X REMOVAL >= 95% or effluent bod <= 50 mg/L

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1	BOD5 EFFLUENT	+ I + I	PLAS	TIC:	B ONLY		1	NUT	PLJ	AND C	     	   	NUT F Type	יר ז ו	STICS NUT C**	     	I NOT P		STICS PE I	     	ALL Sti	WRE	ABTE Ams	
	( <=MG/L )		CÚM FR	EQ	CUM	X	ICUM	FRE	01	CUM X		ICUM	FREG	11	CUM X		ICUM FREQ		CUM X	-     -	CUM FRED		CUM X	1
1	20	11	35		68.	6	1	12	- 1	36.4		1	1	1	43.8	-	1 8	1	57.1	- 1	62	1	54.4	1
1	30	11	40	1	78.	4 1	1	17	1	51,5	5 4	1	12	1	75.0	1	13	i.	92.9	i	82	i	71.9	i
i.	40	11	46	1	90	2 1	1	24	1	12,1	' I	1	14	1	87,5	- İ I	13	Ì.	92.9	1	97	i i	85.1	Ì
1	50	- 11	50	L.	98.	0 1	1	27	1	81.8	1 1	1	15	1	93.8	1	14	1	100.0	1	I 106	1	93.0	1
1	100	E F F	51	1	100.	0 1	1	31	1	93.9	) ł	1	16	1	100.0		14	1	100.0	1	112 (	1	98.2	I
1	200	11	51	- I	100.	0 1	1	33	1	100.0	) [	1	16	1	100.0		14	I.	100.0	1	114	1	100.0	1
1	300	11	51	1	100.	0 1	1	33	- ł	100.0	1	1	16	1	100.0		1 14	1	100.0		114	1	100.0	T
1	400	11	51	1	100.	0 1	1	33	1	100.0	) <b>i</b>	1	16	1	100.0	11	1 14	L	100.0		114	1	100.0	1
1	500	11	51	1	100.	0 1	I I	33	1	100.0	)	1	16	1	100.0	11	14	Ł	100.0		114	t	100.0	1
1	600	11	51	1	100	0 1	1	33	I	1.00.0	1	1	16	I.	100.0	11	14	I	100.0	Ì	114	1	100.0	1
1	700	11	51	l I	100.	0 1	i -	33	1	100.0	1	1	16	I.	100.0	11	14	1	100.0	11	114	1	100.0	1
1	800	11	51	l I	100.	0 1	1	33	1	100.0	1	1	16	1	100.0	11	14	1	100.0	1	114	1	100.0	I.
••	-4	••••		• • • •		* = 4 +				3UH	 MAH	 Y 51		110	;8			•••	******		,	••	***	

\* Type I w/oxidation \*\* Type I w/o/oxidation

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made using the criteria of plants which achieve 95% BOD reduction or which have an effluent BOD<sub>5</sub> concentration of 30 mg/l or less. The results of this analysis are presented in Table 7-31.

In reviewing the data, it appeared that well operated Type I and C plants exhibited a wider range of effluent BOD values than the plants in the other subcategories. Subsequent investigations showed that this effect appeared to be related to the efficiency of water use by plants in the Type I and C subcategory. Water use efficiency was defined as a plant's daily water usage divided by its daily production level. The resulting water use, in gallons per pound of production, was plotted as a function of effluent BOD, concentration for the plants in the subcategory identified as achieving 95% removal or effluent BOD less than 50 mg/l. This plot is presented as Figure 7-4. The figure indicates that plants with low water use, i.e., less than 0.2 gallons per pound, generally do not achieve effluent BOD<sub>e</sub> concentrations as low as do plants with higher water usage. A more détailed analysis showed this effect to be limited to those plants in the first quartile of water use. This included plants with water use equal to 0.165 gallons per pound or less. This would suggest that the Not Plastics, Type I and C subcategory be further divided into low flow (less than or equal to 0.165 gallons per pound) and high flow (greater than 0.165 gallons per pound) subcategories. If this is accomplished, the resulting median effluent BOD concentrations of plants with greater than 95% removal or 50 mg/l effluent  $BOD_{e}$  would be 26 mg/l for high flow and 36 mg/l for low flow plants.

It was previously shown that biological plants followed by additional unit processes do not achieve significantly lower effluent BOD<sub>5</sub> concentrations than biological systems alone. This is not the case for effluent TSS concentrations. Table 7-32 through 7-35 present a comparison of summary statistics for effluent TSS concentrations based on biological systems meeting the 95/50 criteria, and 95/50 biological systems followed by additional unit processes such as polishing ponds, multimedia filters and activated carbon. These tables show that systems which include polishing ponds or filters achieve lower median effluent TSS concentrations than those systems which do not. In addition, these data indicate that polishing ponds and filters achieve comparable effluent TSS levels.

#### EFFLUENT VARIABILITY

It is well known that biological wastewater treatment systems produce an effluent of varying quality, much of this attributable to the inherent nature of the treatment process. The range of this variation is dependent on many process characteristics. In some cases significant changes in effluent quality may be associated with specific causes such as shock loads, mechanical failures, poor design or operation, or errors in sampling or analysis.

The term variability, as used in this context, is defined as the variation in the effluent quality from a properly designed and operated biological treatment system which is attributable to the basic nature of the treatment process. Minimizing the impact of the cited specific causes of variability through the use of proper design and management techniques can reduce to a minimum the effluent variability which the

BIDLOGICAL 3Y3TEM BOOS EFFLUENT COMCENTR Removal 24 95% om Effluent		A T I DM 5	BOD <- 30 HG/L
P4	BIOLOGICAL SYSTEM	BOOS EFFLUENT CONCENTR	I REMOVAL ># 95% DH EFFLUENT

BUDS EFELICAT	= :	5 7 1 1 5 T 5 T 6	Y 100 1	Ξ	401 P	1.45	11 <b>C3</b> *D r *	= :		ABTIC NOT			101 PL	1411C3	= :	Ţ	HA.	315
NCENTRAT				: -		· !						;						
1/9H=>		CUM FHEAL	כטי ג		UM FREQ	1	CUM X	ŭ	UN FRED			H D U I	FREGI	CUH X		H FRE	ā	X XD
20	= =	35	5 61	:=	12		414	; ; _	-	50	0		- -	61.5		~~ 9	-	62.0
30	-	1 40	90.9	-	17	-	58.6	=	12	59	-	-	-	100_0		62	-	82.0
607	Ξ	41	5,2	=	2 <b>2</b>	_	75.9	-	2	- - 		-	-	100.0	=	8 8	-	99.0
50	=	43 1	1 • 1 6	-	ر <i>ح</i>	-	2.67	=	15	26	-	-	-	100,0	=	26	_	92,0
100	-		100.0	-	27	_	93.1	-	77	001		-	-	100.0	-	86	-	98.0
200	-		100.0	=	۰ ۲	-	100.0	=	14	100	- 0.	-	-	100.0	=	100	_	00.0
300	-	1 77	100.0	=	5.5	_	0.001	=	1 * 1	001	- 0.	-	-	100.0	=	001	-	00.00
101	-	1 51	100.0	=	74	-	0.001	=	3-	100	- 0.	-	<u> </u>	100.0	=	001	_	0.00
500	Ξ	50	100 0	=	24	_	0 0 0 0 1	=	7	100	-	-	-	100.0	-	100	_	0,00
600	Ξ	- 73	100.0	=	2 2	_	100.0	=	14	100	-	-	-	100.0	Ξ	100	_	0.00
700	Ξ	1 77	100.0	_	42	_	100.0	=	14	100	- ^ •	-	-	100.0	Ξ	100	_	0.00
800	Ξ	1 77	100.0	-	29	_	100.0	Ξ	-	100	- 0.	-		100.0	Ξ	001	_	0.00

SUMMARY STATISTICS

1 0 .	154.0 1	22.6 1	15.0 1	1001	-
3°0    HINIHUH	ZT.0 II MAXIMUM=	IS. G II MEAN	14.0 II MEDIAN -	13 I N I I	-
-HUMININ IT 0.2	TO TI MAKINGME	Z4.7 IJ MEAN	20.0 II MEDIAN		=
-+OHINIM    0*9	154 0 11 MAXIMUM-	36,3 11 NCAN =	Z4.0 II MEDIAN .	29 11 N .	Ξ
3.0 11 MINIMUM	52.0 11 MAXINUM	14.7 11 NEAN *	10.0 II MEDIAN .	44 11 Va	-
BHOHININ II	-MUHIXAM II	H MEAN	I MEDIAN -	• z =	=

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\* Type I w/oxidation
\*\* Type I w/o/oxidation





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## RIQLOGICAL SYSTEMS TSS EFFLUENT CONCENTRATIONS BOD ZKEMOVAL >=95% OR EFFLUENT BOD <=50MG/L

1	TSS FFFLUENT	    	PLASTI		ONLY		NOT F Type	PL/ I	STICS AND C	11 11 11	NOT Type	PL/	ASTICS NOT C	    	NOT PL. Rot T	ASTICS YPE I	11	AI_L STI	4 RE/	ASTE AMS	     
-11	( <=HG/1 )		CUM FRED	21	CUM X	- 11	CUM FRE	31	CUH Z	÷ù	CUM FRE	01	CUN X	÷	CUM FREQI	CUN X	÷Б	CUM FRER		CUH Z	÷
÷.		-11		-   -		-11		- 1 -		~11		-1-		-11			-11				- i
i	20	11	19	i	39.6	-11	3	i	10.7	11	4	i	33.3	11	- 4 i	28.6	ii.	30	i	29.4	i
i	30	11	29	i.	60.4	11	6	Í.	21.4	11	5	1	41.7	11	7 İ	50.0	11	47	i	46.1	Ì
÷	40	11	34	Ì	70.8	11	10	I.	35.7	11	7	1	58.3	11	10 1	71.4	11	61	i	59.8	T
T	50	11	3.5	T	75.0	11	12	Ŧ	42.9	11	9	1	75.0	11	11 1	78.6	11	68	1	66.7	1
Ł	100	11	46	t	95+8	- 1 1	21	I.	75.0	11	11	1	91.7	11	13 1	92.9	11	91	L	89.2	1
Т	200	11	48	T	100.0	11	28	-t	100.0	11	12	1	100.0	11	14	100.0	11	102	1	100.0	t
1	300	11	48	1	100.0	11	28	1	100.0	11	12	1	100.0	11	14	100.0	11	102	t i	100.0	1
Т	400	E F F	49	1	100.0	11	28	1	100.0	11	12	- F	100.0	11	14 1	100.0	11	102	I	100.0	1
T	500	11	48	I.	100.0	11	28	T	100.0	11	12	1	100.0	11	14 1	100.0	11	102	1	100.0	E.
- E	600	11	48	1	100.0	11	29	Т	100.0	11	. 12	ł	100.0	11	14 /	100.0	11	102	1	100.0	1
T	700	11	48	1	100.0	11	28	1	100.0	11	12	1	100.0	11	14 1	100.0	11	102	ł.	100.0	1
1	800		48		100.0		29	1	100.0	11	12		100.0	11	14 I	100.0	11	102	!	100.0	1
									SUMM	ARY	STATIS	TI	CS								
			MINIMU	 M=	4.0		MINIMU	 M=	9.0		MINTHU	 M=	13.0	11	MINIMUM=	2.0	+1	MINIMUM	- ··· =	2.0	
i		11	MAXTHU	M=	127.0	11	MAXTHU	M =	189.0	11	HAXTHU	H=	105.0	11	MAXINUH	101.0	11	HAXTHUM	=	187.0	1
i		11	MEAN	=	36.1	11	MEAN	=	71.3	11	HEAN	=	40.6	11	MEAN =	36.4	11	MEAN	2	46.3	I.
Ì		- 11	MEDIAN	=	24.5	11	MEDIAN	=	\$3.5	11	MEDIAN		34.5	11	MEDIAN =	30.5	11	MEDIAN	=	33.0	L
i		11	N	•	48	11	N	=	28	11	N	=	12	11	N =	14	11	N	3	102	I.
i		11			. –	- 11				11				11			11				1

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\* Type I w/Oxidation
\*\* Type I w/o/Oxidation

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## BIOLOGICAL SYSTEMS WITH FOLISHING TSS EFFLUENT CONCENTRATIONS BOD TREHOVAL >=957 OR EFFILIENT BOD <=50HG/L

1	TSS FFFLUENT	++	FLAS	1108	0NL.Y	    	N T	 0 T YPE	F1.6	ASTICS AND C	   	 	NOT F	1.1	ASTICS NOT C	11	том том	PLA TY	STICS FE J		ALL ST	U RF	ASTE AMS	
10	ONCENTRATI	ONII-					····-	••••			1	1		-	• · • • • • • • • • • • • • • • • • • •	-								
1	( <=MG/L )	110	UM FRI	FQI	CUM %	- 11	CUH	FRF	Q I	CUH	Z I	1.01	IN FREE	1	CUH %	ELCU	H FRE	N 1	CUH X	LICUM	FRFG	11	CUM X	- t
1-	· · · · · · • • • • · · •			1-	<b></b>	+			-1-		I	1		1		-		- 1 -		-11		1-		-1
I	20	14	5	1	38.5	- 14		2	1	25.	0 1	1	0	L	0.0	11	2	1	28.6	11	9	1	30.0	1
1	30	11	10	1	76.9	11	ł.	3	1	37.	5 I	1	0	Т	0.0	1.1	4	1	57.1	11	17	1	56.7	- 1
I.	40	11	11	1	84.6	11	l .	4	- t	50.	0 1	1	1	I.	50.0	11	6	1	85.7	11	22	I.	73.3	I
1	50	11	12	1	92.3	11	l	5	1	1.2 .	5 1	E -	t	÷.	50.0	11	E.	1	85.7	11	24	1	80.0	ł
1	100	11	13	1	100.0	11	1	ß	1	100.	0 1	1	2	÷.	100.0	11	7	1	100.0	11	30	1	100.0	1
1	200	11	13	1	100.0	11	l I	8	1	100.	0 I	1	2	Т	100.0	11	7	1	100.0	11	30	1	100.0	1
I.	300	11	13	1	100.0	- 11	l i	8	ſ	100.	0 1	1	2	1	100.0	1.1	7	1	100.0	L F	30	1	100.0	1
ŧ	400	11	13	1	100.0	11	I	8	I.	100.	0 1	1	ŗ.	Т	100.0	11	7	Т	100.0	11	30	1	100.0	- E
T.	500	E F	1.3	1	100.0	11	1	9	1	100.	0 1	1	2	1	100.0	11	7	+	100.0	11	30	1	100.0	1
1	600	- E E	13	1	100.0	11	I	8	1	100.	0 1	1	7	Т	100.0	11	7	I.	100.0	11	30	4	100.0	1
T	700	11	13	ł	100.0	_ <b>1</b>	l 4	ß	1	100.	0 1	1	2	T	100.0	11	7	1	100.0	11	30	I.	100.0	1
۲	800	11	13	I.	100.0	1	l	8	ł	100.	0 1	I	?	I	100.0	11	7	I	100.0	LT	30	I.	100.0	I.

#### SUMMARY STATISTICS

														• ••		· · · ·						-
I.	11	HU	NTHU	M =	9.0	- 11	HINTH	11JM =	9.0	11	нініни	H=	31.0	11	нтнтни	1=	2.0	11	HINTHUP	1=	2.0	I.
+	11	HA:	XIHU	M =	76.0	11	HAXIH	1UM=	\$5.0	14	HAXTHU	M =	62.0	11	HAXIHU	1=	55.0	11	HAXINUN	1=	95.0	I.
1	11	HE	AN	=	26.3	- 11	MEAN	=	49.5	11	MEAN	=	46.5	11	HEAN	=	27.7	11	MEAN	=	34.2	I.
1	11	HE.	NAIG	=	23.0	11	HEDTA	= N/	42.0	11	HEDTAN	=	45.5	14	MEDTAN	*:	26.0	11	HEDTAN	Ŧ	29.5	L
1	11	N		=	13	- 11	N	=	8	11	N	=	2	11	N	=	7	11	N	=	30	1
1	11					11				11				11				11				T
									<b></b>													-

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\* Type I w/Oxidation
\*\* Type I w/o/Oxidation

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## E1000GICAL SYSTEMS WITH MNF TSS EFFLUENT CONCENTRATIONS BOD %REMOVAL >=95% OR EFFLUENT ROD <=50MG/L

1	TSS							1 1	NUL	PL	ASTIC		11	NOT	е. Е.	ASTICS			21 61	STICS	11	Δi i	 ม	457F	
i	EFFLUENT	- ii	PL.	ASTI	cs	ONL Y	i	1	ΤΥΡΕ	I	AND (		11	ΤΥΓΕ	I	NOT C	ii –	NOT	TY	FE I	ii	S	IRE	ANS	
1	CONCENTRATIO	- I I NC					1	I ·					11-			- ·- <i>·</i> ··· <i>•</i> ····						<b>. .</b> .		·····	• • •
ł	( <=MG/L )	110	UM	FRED	11	CUH X	1	I CUM	FRE	Q I	CUM	z	110	UN FRE	0 I	CUH X	11008	I FREG	11	CUM X	LICUM	ERE	11	CUN X	
1		11-			- 1 -		1	I ·		-1			11-		1 -	••••••	-11		-			·· <b></b> -	- 1 -		
L	20	11		1	1	33.3	1	1	1	1	33.	3	EE.	0	1	0.0	11	1	1	100.0	11	3	1	37.5	
I.	30	11		3	1	100.0	- 1	I .	2	1	66.	7	11	0	1	0.0	11	1	1	100.0	11	6	1	75.0	
I.	40	11		3	1	100.0	- 1	I .	2	1	66.	7	11	0	t	0.0	11	1	1	100.0	11	6	1	75.0	
L	50	- 11		3	1	100.0	1	I	2	1	66	7	11	1	1	100.0	11	1	1	100.0	11	7	1	67.5	
L	100	11		3	1	100.0	- 1	I	3	- 1	100	0	11	1		100.0	11	1	1	100.0	11	8	1	100.0	
ł	200	- 11		3	1	100.0	- I	1	3	1	100	0	E E -	1	I.	100.0	11	1	1	100.0	11	8	1	100.0	
1	300	11		3	ł	100.0	1	I	3	ł	100	0	11	1	1	100.0	14	1	1	100.0	11	C	1	100.0	
1	400	11		3	1	100.0	1	L	.3	l	100	0	IL.	1	1	100.0	E F	1	1	100.0	11	ß	I.	100.0	
Ł	500	11		3	1	100.0	1	1	31	1	100	0	11	1	1	100.0	11	1	1	100.0	11	8	1	100.0	
Ł	600	11		3	1	100.0	1	1	3	1	100	0	11	1	1	100.0	11	1	1	100.0	11	8	1	100.0	
I.	700	11		3	÷.	100.0	1	1	3	I.	100	0	+ I -	1	1	100.0	11	1	1	100.0	11	3	+	100.0	
ŧ	800	11		3	1	100.0	1	1	3	1	100	. 0	11	1	1	100.0	11	1	1	100.0	11	8	1	100.0	

#### SUMMARY STATISTICS

															<b>.</b>								_
1	11	нτ	NTHU	M =	18.	0 1	L	HINTHU	1=	19.0	11	MINTHU	H =	18.0	11	атитни	n=	12.0	11	MINTHU	<b>1</b> =:	12.0	١
1	11	HA	XINU	H =	24.	0 1	I.	HAXINU	4=	74.0	11	MAXIMU	M-	48.0	11	HAXIMU	M =	12.0	11	ΜΛΧΤΗΙ	M=	74.0	Ł
1	11	HE	AN	=	21.	31	ł.	MEAN	=	40.7	11	MFAN	=	48.0	11	MEAN	=	12.0	11	HEAN	=	30.8	Ł
1	11	HE.	DIAN	=	22.	0 1	1	MEDIAN	=	29.0	LE	MEDIAN	=	48.0	11	HEDIAN	Ξ	12.0	11	HEDIAN	=	23.0	ι
1	11	N		=		3 I	Т	N	=	3	11	N	=	t	11	N	=	1	11	N	=	8	1
1	11					1	I.				11				11				:1				I.

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\* Type I w/Oxidation
\*\* Type I w/o/Oxidation

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#### DIOLOGICAL SYSTEMS WITH ACTIVATED CARBON TSS EFFLUENT CONCENTRATIONS BOD XREHOVAL >=75% OR EFFLUENT BOD <=50HG/L

																		• • • • • • •			••• •• •• ••	• - •• -					
I.	TSS	11					1	1	ΝПΤ	FI./	ASTICS	. 1	1	NOT	PL.	AGTICS	• • •	11	NOT	FLA	ASTICS	5	11	AI.L	. W	ASTE	1
1	EFFLUENT	11	- F1	AST)	105	ONL Y	' I	1	TYFI	F I	ANTI C1	R	1	TYPE	I	NOT C 3	<b>★</b> ★	1.1	NOT	. ТI	FE I		11	S1	I R E	ANS	1
10	DNCENTRAT	IONII						1					1								• • • • • •		11	<b></b>	• • • •		- 1
I.	( (=HG/L )	) II	CUH.	FRE	1 1	CUM	X I	1001	1 F.R.I	01	CUM 2	۱ ک	I CUM	I FKE	01	CUH X	. 1	псин	FRE	R I .	CUN	X	LICUN	FREC	11	CUN X	- 1
4 -		!!			- 1 -		1	1				1	1		-1-					-1-			11		-   -		- 1
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#### SUMMARY STATISTICS

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\* Type I w/ oxidation
\*\* Type I w/o oxidation

system can achieve. Simply stated, this level is the minimum variability which can be practically obtained assuming proper system design, management, operational control, sampling and measurement.

Effluent variability can be characterized by the statistical analysis of daily data from well-operated treatment systems. The daily data computer file available for this analysis contains daily BOD, TSS, COD and TOC influent and effluent measurements over variable periods of records from three months to five years. The data base includes records for 50 plants. Although some records were as short as three months, most were for a period of at least one year. Before performing the variability analysis, however, it was necessary to screen the 50 plants to identify those exhibiting acceptable treatment system performance and using proper sampling procedures. This data screening involved the complementary statistical and engineering analyses described below.

The statistical screening for each plant was based on summary statistics, plots of daily concentrations versus time, and plots of moving summary statistics. Examples of the plots produced are given in Figures 7-5 and 7-6. The most useful statistical screening tool was the plot of the moving 12-month 99th percentile illustrated in Figure 7-6. The first point represents the estimated 99th percentile of data from the first 12 months. Succeeding points represent estimated 99th percentiles of data from months 2-13, 3-14, etc. This plot reflects changes in either the mean or variance of effluent concentrations over time. Plots for both BOD and TSS were produced for each plant based on the lognormal distribution.

Three types of plots were identified in studying the moving 99th percentile (Y0.99) plots:

- Type I: Performance improved over time (there was a time after which Y0.99 decreased).
- Type II: Performance worsened over time (there was a time after which Y0.99 increased).
- Type III: There was a data gap due to modifications in the treatment system (generally followed by improved performance).

Table 7-36 shows the classification for each plant for BOD and TSS, along with beginning and ending dates and dates of data gaps (if any). Data from plants with Type I or III graphs for both BOD and TSS were tentatively accepted for further engineering analysis. Data from plants with Type II graphs were examined to determine if some cause for the worsening performance could be identified.

An engineering analysis also was conducted on each of the 50 candidate plants with daily data. The analysis consisted of a detailed review of each plant diagram and other information relevant to plant operation and performance. Specific points of interest included the relationship of the effluent sampling site to mixing points for stormwater runoff, untreated process water and cooling water, as well as modifications to the



## MOVING SUMMARY GRAPH TYPES







FIGURE 7-6 PLOT OF ESTIMATED 9916 PERCENTILES VIL HOVING TIME INTERVALS

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## SUMMARY OF PLANT SCREENING

BOD

Plant	Stream	Туре	Begin Date	Date Gap	End Date
1	<b>A</b> 01	I	01/02/75		09/29/76
1	A02	III	01/01/75	10/76 - 05/79	07/30/79
3	A01	III	01/04/77	09/76 - 12/76	05/29/80
9	A01	I	01/15/75		09/22/76
292	A01	Ι	01/01/7 <b>8</b>		12/29/78
15	A01	I	01/01/78		05/31/80
18	A01	II	07/02/75		09/20/76
293	A01	I	01/03/75		09/24/76
27	A01	III	04/04/78	05/78 - 08/78	08/21/80
28	A01	II	09/03/75		09/30/76
42	A01	II	01/17/79		07/17/80
44	A01	I	01/01/79		05/30/80
45	<b>A</b> 01	II	01/02/79		06/26/80
53	A01	I	01/02/75		09/30/76
60	A01	II	01/01/78		04/30/80
61	A01	I	01/03/75		09/30/76
73	. A01	II	05/02/75		06/25/76
75	A01	I	08/02/74		05/31/76
89	A01	II	05/01/74		09/30/76
90	A01	I	01/02/75		09/30/76
96	A01	I	01/01/75		09/27/76
106	A01	I	01/07/75		10/31/76
109	A01	I	05/01/74		09/30/76
110	A01	II	01/03/75		09/30/76
111	A01	I	01/04/77		12/29/77
113	A01	I	01/01/78		12/31/79
118	A01	I	01/01/79		12/31/79
120	A01	I	04/01/76		09/29/76
124	A01	11	01/01/75		09/28/76
126	A01	I	01/02/75		09/30/76
138	A01	I	01/05/79		12/30/79
146	A01	III	01/02/75	10/76 - 04/79	07/31/79
170	A01	II	06/04/78		05/29/80
175	A01	II	01/03/78		07/11/80
176	A01	I	08/01/78		10/31/78
220	A01	11	01/02/75		09/24/76
229	A01	I	09/01/74		09/30/76
234	A01	I	01/03/78	<b></b>	12/27/79
236	A01	111	05/01/78	03/79 - 05/79	06/25/80
245	A01	I	05/21/77		04/29/80
268	A01	II	01/01/79		07/31/80
269	A01	<b>II</b>	01/04/75		09/29/76
274	A01	I	01/01/75		06/30/80
281	A01	I	07/02/78		06/29/80

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## TABLE 7-36 (Continued)

## SUMMARY OF PLANT SCREENING

TSS

Plant	Stream	Type	<u>Begin Date</u>	Data Gap	End Date
1	A02	111	01/01/75	10/76 - 05/79	07/31/79
3	A01	II	01/01/77		05/29/80
9	A01	I	01/15/75		09/22/76
292	A01	I	01/01/78		12/30/78
18	A01	II	07/02/75		09/29/76
293	A01	I	01/03/75		09/24/76
27	A01	III	04/04/78	04/79 - 05/80	05/27/80
28	A01	II	09/02/75		09/30/76
44	A01	II	01/01/79		05/30/80
45	A01	I	01/01/79		06/30/80
53	A01	II	01/02/75		07/11/76
73	A01	II	05/05/75		06/25/76
89	A01	I	05/06/74		09/30/76
90	A01	I	01/01/75		09/30/76
96	A01	I	01/07/75		09/29/76
109	A01	I	06/18/74		07/26/74
110	A01	I	01/02/75		09/28/76
111	A01	I	01/01/77		12/31/77
113	A01	I	04/01/79	, <b></b> -	06/30/79
120	A01	I	04/01/76		09/29/76
123	A01	I	06/01/75		09/30/76
124	A01	I	01/01/75		09/28/76
126	A01	I	01/02/75		09/30/76
138	A01	I	01/01/70		12/31/79
146	A01	III	01/01/75	10/76 - 04/79	07/31/79
176	A01	I	0 <b>8/01/78</b>		10/31/78
220	A01	I	01/03/75		09/27/76
229	A01	I	09/01/74		09/30/76
236	A01	I	06/01/78		06/29/80
245	A01	I	06/21/77		04/29/80
274	A01	III	01/01/75	09/76 - 05/78	06/30/80
294	A01	I	01/01/79		12/31/79

treatment system during or after the period of data collection. A summary of the engineering analysis is presented in Table 7-37. This table provides the engineering comments and the nature of the reason for exclusion, where applicable, for each of the 50 candidate plants. A poor performer has been defined as a plant not achieving 95% BOD removal or an effluent of 50 mg/l BOD.

Based on the engineering and statistical analyses, 17 of the 50 plants were retained for further analysis. Appendix D contains a description of these plants and the reasons for exclusion. Because treatment system performance generally improved over time in the selected plants, the data retained in the final Daily Data Base for each plant were limited to samples collected within 12 months of the last sampling date.

Having selected daily data representing the performance of well-operated treatment systems, the next step was to find a statistical model to characterize effluent variability. The distributional model most commonly employed for daily measurements of BOD and TSS is the lognormal distribution. This model tends to be appropriate because distributions of daily pollutant measurements have a lower bound of zero, are positively skewed, and have standard deviations proportional to their means. To ensure that the lognormal model was appropriate for the data in the Daily Data Base, distributions of daily data were plotted and goodnessof-fit tests were run for each plant/pollutant data set. The goodnessof-fit test employed is described in Appendix B. The results of these analyses supported the use of the lognormal model.

Finally, effluent variability was characterized for BOD and TSS for each plant in terms of the variability factor--the ratio of the 99th percentile of the concentration to its long-term average for the daily maximum and the ratio of the 95th percentile of the concentration to its longterm median for the 30 day average. The methods used to estimate daily and 30-day average variability factors are described in Appendix B. The daily variability factors were based on the lognormal model, and the 30day variability factors on the Central Limit Theorem (taking day-to-day correlation into account). The daily and 30-day variability factors calculated for each parameter and plant-specific data set in the Daily Data Base are presented in Tables 7-38 and 7-39. These summary tables provide the number of observations (N), the estimated mean, 99th percentile (P0.99), 95th percentile (P0.95), daily variability factor

The variability factors were summarized separately for two categories of plants, Plastics Only and Not Plastics Only. The decision to use two categories, rather than determine a single set of factors for the entire OCPS industry, was based on the lower effluent BOD, levels achieved by Plastics Only plants relative to the rest of the OCPS plants. Further partitioning of the Not Plastics Only segment for the purpose of calculating variability factors did not appear warranted since all plants use biological treatment as the major treatment technology. In addition, further partitioning would have effectively reduced the available data base for each group.

## SUMMARY OF ENGINEERING ANALYSIS

PLANT	>95% or <50 mg/1	EXCLUDED FROM ANALYSIS	ENGINEERING COMMENTS AND/OR NATURE OF AND REASON FOR EXCLUSION
1	yes	yes	Treatment system upgraded during first data collection period; second data period too short for analysis.
3	yes	уев	Plant had upsets and bypasses continuously; poor operation, system modification dur- ing data period.
9	yes	no	None
292	-	yes	Receives significant amount of municipal waste of unknown quantity.
15	yes	no	None, BOD only
18	уев	yes	Treatment system under construction during period of performance, effluent data not representative.
293	no	yes	Plant is a poor performer <sup>(o)</sup> due to inade- quate solids control (66% removal TSS)
20	-	yes	No effluent data on BOD, COD or TSS
24	-	yes	No effluent data on BOD, COD or TSS
27	-	no	None
28	yes	yea	Chlorination before trickling filter, no longer manufactures organic chemicals.
42	no	yes	Poor performance plant; <sup>(o)</sup> poor operating practices, upset operations during data period.
44	yes	no	Treatment system changed prior to 1/79.
45	уев	no	None
53	уев	<b>y</b> es	Treatment system inadequate for average loadings, resulting in poor performance and subsequent upgrade of system after data period.
60	no	yes	BOD only

## TABLE 7-37 Continued

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	>95%R or	EXCLUDED FROM	ENGINEERING COMMENTS AND/OR
PLANT	<50 mg/1	ANALYSIS	NATURE OF AND REASON FOR EXCLUSION
61	no	уеб	Poor performance plant; <sup>(0)</sup> filtration unit added after period of performance in order to improve solids removal.
72	-	уев	No effluent data on BOD and TSS
73	уеб	уев	Plant phased out a process during data per- iod, resulting in drop of 70% BOD load dur- ing period of record.
74	-	yes	No effluent data on BOD, COD or TSS
75	yes	yes	Effluent sample point downstream of point where stormwater and cooling water are mixed with contaminated wastewater.
87	-	yes	Sample point is downstream of nonbiotreated effluent dilution, BOD and TSS data unavailable.
89	уев	yes	Sample point downstream of untreated process water dilution.
90	yes	yes	Sample point downstream of cooling water dilution.
96	yes	no	None
103	-	yes	No effluent data on BOD or TSS
106	уев	yes	Sample point downstream of cooling water dilution.
109	yes	yes	Effluent contains unquantified dilution from a "consolidated" sump.
110	уев	no	None -
111	уев	no	None
113	уев	no	None
118	yes	no	None
120	yes	yes	Treats refinery wastewater with OCPS waste- water.

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## TABLE 7-37 Continued

PLANT	>95%R or <50 mg/1	EXCLUDED FROM ANALYSIS	ENGINEERING COMMENTS AND/OR NATURE OF AND REASON FOR EXCLUSION
123	-	уев	Sample point downstream of stormwater dilu- tion (no BOD data).
124	уев	уев	Sample point downstream of stormwater mixing point.
126	уев	no	None
138	no	yes	Poor performer <sup>(o)</sup>
146	yes	yes	Sample point downstream of stormwater dilution.
170	yes	no	BOD data only
175	yes	no	None
176	yes	yes	Only three months of data available
220	yes	no	None
234	уев -	no	None, data is for secondary system only.
236	yes	no	None
245	yes	yes	Non-biological treatment
268	no	yes	None
269	no	yes	Poor plant performance <sup>(o)</sup> appears due to plant being undersize; plant has upgraded treatment since data period.
274	-	уев	Significant portion of wastewater treated is sanitary flow.
294	-	уев	Poor performance <sup>(o)</sup> is due to poor solids control, especially for plant with sand filtration.
281	yes	no	None

 Poor performance is defined as not achieving 95% BOD removal or 50 mg/l effluent BOD.

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	<del></del>		Effluent	BOD					Effluen	t TSS		
Plant	N	x	PO.99	PO.95	VF(1)	VF(30)*	   N	x	PO.99	PO.95	VF(1)	VF(30)
9	24	5.84	17.88	_	3.06	-	24	29.72	97.58	-	3.28	-
44	261	8.97	29.51	11.84	3.29	1.32	260	11.58	80.50	19.76	6.95	1.71
45	156	3.07	10.67	4.28	3.47	1.39	364	18.95	124.62	30.87	6.57	1.63
96	105	2.31	7.86	-	3.41	-	66	12.34	45.26	16.05	3.67	1.30
111	157	6.19	18.38	9.25	2.97	1.49	347	10.42	54.12	14.84	5.19	1.42
126	249	6.05	23.70	10.23	3.92	1.69	253	23.28	78.35	31.10	3.37	1.34
				AVG	3.35	1.47	(   			AVG	4.84	1.48

## ESTIMATES OF VARIABILITY FACTORS "PLASTICS ONLY" PLANTS

\* Where there was insufficient data to estimate day-to-day correlation, no VF(30) value is given.

Effluent BOD						Effluent TSS						
Plant	N	x	PO.99	PO.95	VF(1)	VF(30)*	   N	x	P0.99	P0.95	VF(1)	VF(30
15	363	16.88	63.55	22.80	3.77	1.35					·	
26	160	17.55	68.26	26.69	3.89	1.52	1 158	21.86	76.32	28.99	3.49	1.33
110	247	5.91	21.82	8.37	3.69	1.42	218	10.09	45.33	14.37	4.49	1.42
113	332	17.47	75.06	28.95	4.30	1.66	91	22.41	100.27	31,22	4.48	1.39
118	365	12.17	61.34	25.25	5.04	2.07	 					
170	103	33.90	145.75	74.51	4.30	2.20	 					
175	36Í	39.09	181.49	69.66	4.64	1.78	1					
220	55	55.13	291.35	-	5.28	-	   149	94.09	441.22	-	4.69	-
234	157	11.58	41.60	16.74	3.59	1.45						
236	162	32.19	93.17	39.75	2.89	1.23	362	59.72	159.29	72.55	2.67	1.21
281	205	8.44	26.73	11.62	3.17	1.38	{   				<del></del>	
				AVG	4.05	1.61	1			AVG	3.95	1.34

## ESTIMATES OF VARIABILITY FACTORS "NOT PLASTICS ONLY" PLANTS

\*Where there was insufficient data to estimate day-to-day correlation no VF(30) value is given.

## TABLE 7-39

#### WASTEWATER DISPOSAL

The method of treatment for the direct dischargers was discussed under the previous heading. Under this heading the treatment processes and disposal methods associated with zero or alternate discharge in the OCPS industry are described.

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#### Zero or Alternate Discharge

Zero or alternate discharge is defined as no discharge at the OCPS plant of contaminated process wastewater to either surface water bodies or to POTWs. Means by which zero or alternate discharge may be achieved are described in the following paragraphs.

#### Deep Well Disposal

Deep well injection is a method frequently used for disposal of highly contaminated or very toxic wastes not easily treated or disposed of by other methods. Deep well injection is limited geographically because of the geological requirements of the system. There must be a substantial and extensive impervious caprock strata overlying a porous strata which has no utility as a water supply or other withdrawal.

Because of the potential hazard of contaminating useable aquifers, some states prohibit the use of deep well disposal. Contamination of these aquifers can occur: (1) from improperly sealed well casings which allow the waste to flow up the bore hole, and (2) from unknown faults and fissures in the caprock which allow the waste to escape into the useable stratum. The latter is conceivable even though the fault may be miles from the well and the migration of the waste material to the fault might take many years. This problem could be enhanced by the increased subterranean pressure created by the injection well and could be further enhanced if a substantial withdrawal of water from the useable aquifer were made in the vicinity of the caprock flow.

Deep wells are drilled through impervious caprock layers into such unusable strata as brine aquifers. The wells are usually more than 3,000 ft deep and may reach levels over 15,000 ft. Pretreatment of the waste for corrosion control and especially for the removal of of suspended solids is normally required to avoid plugging of the receiving strata. Additional chemical conditioning could be required to prevent the waste and the constituents of the receiving strata from reacting and causing plugging of the well.

Because of the relatively high pressures required for injection and dispersion of the waste, high pumping costs for deep well disposal may be incurred.

A total of 20 plants in the Summary Data Base practice deep well injection. The wastes disposed of in this manner are fairly concentrated with mean BOD, TSS and COD of 3368 mg/1, 4301 mg/1 and 14,242 mg/1, respectively.

### Contract Hauling

Another method of achieving zero discharge is contract removal and disposal. This method involves paying a contract hauler/disposer to pick up the wastes at the generation site and to haul them to another site for treatment or disposal. The hauling may be accomplished by truck, rail or barge.

Contract hauling is usually limited to low volume wastes, many of which may require highly specialized treatment technologies for proper disposal. Although plants utilizing this technology are defined as zero dischargers, an impact on the environment may not be eliminated since the wastes are relocated only from the generating site and may be treated and discharged elsewhere. Reported data regarding contract disposal indicates that there are 15 plants using this disposal method in the Summary Data Base.

#### Offsite Treatment

Offsite treatment refers to wastewater treatment at a cooperative or privately owned centralized facility. Offsite treatment and disposal are used by plants that do not choose to install and operate their own treatment facilities. The rationale for utilization of offsite treatment usually is economically oriented and governed by the accessibility of suitable treatment facilities willing to treat the wastes (usually on a toll basis). Sometimes adjacent plants find it more feasible to install a centralized facility to handle all wastes from their facilites. The capital and operating costs usually are shared by the participants on a pro-rata basis.

Depending on the nature of the waste and/or the restrictions imposed by the receiving treatment plant, wastes sent for offsite treatment may require pretreatment at the generating plant. Four plants in the Summary Data Base practice off-site treatment.

#### Incineration

Incineration is a frequently used zero-discharge method in the OCPS industry. Depending upon the heat value of the material being incinerated, incinerators may or may not require auxiliary fuel. The gaseous combustion or composition products may require scrubbing, particulate removal, or another treatment to capture materials that cannot be discharged to the atmosphere. This treatment may generate a waste stream that ultimately will require some degree of treatment. Residue left after oxidation will also require some means of disposal.

Incineration is usually used for the disposal of flammable liquids, tars, solids, and/or hazardous waste materials of low volume and not amendable to the usual EOP treatment technologies. In all, seven plants in the Summary Data Base employ incineration. Only three data points were reported for incineration. The average of those data showed a waste BOD of approximately 25,000 mg/l.

#### Evaporation

Evaporation is used in the OCPS industry to reduce the volume of waste water and thereby concentrate the organic content to render it more suitable for incineration or disposal to landfill. This technology is normally used as in-plant treatment or pretreatment for incineration or landfill.

Evaporation equipment can range from simple open tanks to large, sophisticated, multi-effect evaporators capable of handling large volumes of liquid. Typically, steam or some other external heat source is required to effect vaporization. Therefore, the major limitations to mechanical evaporation is the amount of energy required.

Only two OCPS plants, both exclusively plastics, reported evaporation as their principal disposal method.

#### Impoundment

Impoundment generally refers to wastewater storage in large ponds. Alternate or zero discharge from these facilities relies on the natural losses by evaporation, percolation into the ground, or a combination thereof. Evaporation is generally feasible if precipitation, temperature, humidity and wind velocity combine to cause a net loss of liquid in the pond. If a net loss does not exist, recirculating sprays, heat or aeration can be used to enhance the evaporation rate to provide a net loss. The rate of percolation of water into the ground is dependent on the subsoil conditions of the area of pond construction. Since there is a great potential for contamination of the shallow aquifer from percolation, impoundment ponds are frequently lined or sealed to avoid percolation and thereby make the basins into evaporation ponds. Solids which accummulate over a period of time in these sealed ponds will eventually require removal. Land area required for impoundment is a major factor limiting the amount of flow disposed by this method.

Twelve plants in the Summary Data Base use impoundment for wastewater disposal. The wastewaters handled in this way are relatively concentrated having average BOD, TSS and COD levels of about 2700, 2000 and 7500 mg/l, respectively.

#### Land Disposal

There are two basic types of land disposal: landfilling and land application (or spray irrigation). Landfilling consists of dumping the wastes into a pit and subsequently burying them. Land application requires spraying the wastes over land. Both disposal methods require care in selecting the site to avoid any possibility of contaminating ground and surface water. The type of pollutant being disposed by land application also must be considered. For instance, if the land is to be used for growing crops at a later time, some of the pollutants present at the time of application may persist in the soil for long durations and later may be assimilated by the crops and find their way into the food chain.

Four plants in the Summary Data Base practice land disposal.

#### SECTION VIII.

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#### ENGINEERING COSTS AND NON-WATER QUALITY ASPECTS

#### INTRODUCTION

This section addresses the cost, energy requirements and non-water quality environmental impacts associated with meeting BPT effluent guidelines. The cost estimates represent incremental expenditures required to supplement the control and treatment technology presently in place. Cost estimates have been prepared using a modified version of the CAPDET computer costing algorithm. Non-water quality aspects reviewed include the potential for (a) air pollution, (b) solid waste generation, (C) RCRA considerations, (d) noise pollution, and (e) energy requirements.

#### COST DEVELOPMENT

In order to estimate the industry expenditures required to meet alternative effluent targets, and as a partial basis for economic studies, a plant-by-plant cost analysis has been made. Capital, operating and annual costs were developed for each of the plants that supplied sufficient information to the Summary Data Base. For each plant, the cost of applying various treatment technology alternatives to meet selected target concentrations for BOD and TSS was conducted.

The basic calculation tool used to develop alternative engineering costs (with the exception of RBC costs) is the computer program, CAPDET (Computer Assisted Procedure For the Design and Evaluation of Wastewater Treatment Facilities). The CAPDET computer model was developed jointly by the Corps of Engineers' Waterway Experiment Station, Vicksburg, Mississippi and the EPA Office of Water and Waste Management. The major purpose of the CAPDET model is to provide for the rapid design, cost estimating and ranking by cost of municipal sewage treatment plant alternatives for the EPA Construction Grants Program. The model may be used for the design of industrial wastewater treatment systems by modifying selected computer program default values. A detailed discussion of the design and application of the CAPDET program is presented in Appendix E.

#### CAPDET MODIFICATIONS

Development of the specific costs applicable to this study requires adaptation of many of the factors in the program from their default values for municipal systems to values more appropriate to chemical industrial wastewater and to industrial plant situations. Table 8-1 summarizes the quantitative bases and default values employed, reflecting those adjustments considered necessary for an industrial wastewater system.

Because of the varying complexity and biodegradability, as well as the generally lower biodegradability of industrial wastewaters as opposed to the essentially constant treatability of domestic wastewater, reaction rate coefficients must be adjusted from CAPDET default values. A TABLE 8-1

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### ADJUSTMENTS TO CAPDET DEPAULT DATA AND RESULTS

PROCESS	CAPDET VALUE	ADJUSTED VALUE	DEPAULT OR RESULT
ASĽ	k = 0.00135 1/mg/hr	$k = \frac{5 / day / 1}{day / 24hr/So mg} So = BOD Influe$	DEFAULT ent Concentration
ALA	k = 0.001 1/mg/hr	k = <u>5 / day / 1</u> day/24/hr/So mg	DEFAULT
PRIMARY CLARIFIER ("IN" "ASL" AND "ALA" TRAINS)	BOD % Removal = 32%	BOD % Removal = 10%	DEFAULT
DAF, CLAR	Land Result Based on \$1000/acre	Land Result Based on Dimensions of Unit at \$10,000 per acre	R ESU LT
	Laboratory Labor Based on Flow	Laboratory Labor Dused on Proration of \$50,000/yr for 1 man per an entire waste Treatment Plant	RESULT
	Administrative Costs Based on Flow	Administrative Costs Equal to 15 percent of of/maint. labor	R ESU LT
ALL UNITS	201 Planning Indirect Cost Equal TO 3.5% of Construction Cost and Contingency Equal to 8%	Delete 201 Planning and Adjust Contingency to 11.5% of Construction Costs	

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reaction rate coefficient of 5.0 days<sup>-1</sup> is used to represent organic chemical wastewater. In the primary clarification segment of the activated sludge process the percent BOD reduction is taken as 10 percent, rather than the default value of 32 percent. These changes adapt CAPDET's approach to the design of organic chemical industrial wastewaters. Alternative cost factors were needed for certain limited scope treatments to avoid the flow-loaded factors applicable to complete systems. Administrative costs for clarification and dissolved air flotation were set at 15 percent of the operating and maintenance labor costs.

Laboratory labor was also adjusted to reflect discrete unit manpower requirements for clarification and dissolved air flotation processes. Laboratory charges are based on the use of one full-time individual for an entire waste treatment system. The approximate annual cost is \$50,000. Costs are prorated over the various treatments and assumed constant, whether the flow volume treated was high or low. The hourly laboratory charges include analyst's base pay plus overtime, fringe benefits, lab equipment and materials and miscellaneous overhead burdens.

The availability of land and its valuation varies markedly with site specific conditions. If land is available on the site, its potential future use and opportunity cost should be considered. If land is not available, purchase of the necessary acreage must be considered. The CAPDET default value of \$1000 per acre is judged to be substantially on the low side, possibly several times too low. Where appropriate, if land costs were a factor in the technology costs, both acreage and cost per acre were separately estimated and inserted to override in the program. In these cases estimated acreage costs of \$10,000 per acre were used to represent the industrial value of land.

In developing the cost analysis, the cost of upgrading an existing treatment facility is assumed to be approximated by the cost of second stage treatment as an addition to the existing facilities. This represents a conservative approach to cost development in that it reflects the maximum cost of upgrading an existing system. In many cases, less expensive treatment system modifications, improved operating practices, or application of in-plant source control techniques may achieve equal or better results at a lower cost. However, in the absence of extensive plant-by-plant design details, second stage or add-on treatment was used as the cost basis. True costs will depend significantly on factors site specific for each plant.

Other considerations when applying CAPDET to industrial treatment estimates are as follows:

1. Unless changed by the user, CAPDET uses default values for conventional pollutant concentrations and other stream characteristics. Also, for each treatment process, specific relationships exist for the removal of the conventional pollutants. For example, the long term mean (LTM) TSS in the activated sludge model is reported at the raw wastewater influent as 200 mg/l (LTM) and at the final clarifier effluent as 20 mg/l (LTM). For industrial wastewaters, a final effluent of 20 mg/l (LTM) may be optimistic and is more likely to be in the range of 30 to 50 mg/l (LTM) unless flocculants or other settling aids are used. Some of the CAPDET values were found to be suitable for industrial wastewater treatment calculations and consequently were not revised. The influent and effluent values for the other characteristics (COD, 0&G, TOC, etc.) are reported as noted in Tables 8-2 and 8-3; however, these values do not affect the cost estimates and are included in CAPDET primarily for municipal planning purposes.

The cost estimates are generated for average flows. A common 2. engineering design practice is to determine the flow and other parameter variability, and design on a basis of an 80 to 95 percentile value. The exact design value chosen is determined by the range of variability of the parameter and the subjective opinion of the engineer. Built into the evaluation is a long range corporate objective or plan pertaining to future expansions, changes in product lines and similar factors which might affect the various parameters associated with the design. Since these factors were not available, the average flow values were used as a basis for the cost estimates. However, since the cost curves show ranges of parameter values, costs for any flow can be derived for any individual plant. Furthermore, CAPDET has built-in Excess Capacity Factor equations which allow for peak flow vs. average flow performance in calculating the detention times and other design factors of the technologies.

3. The cost generated by CAPDET for activated sludge and aerated lagoons appear to be insensitive to changes in target effluent BOD concentrations when the influent concentration is less than 500 mg/l. This phenomenon is created by the fact that the CAPDET model introduces a second equation for aeration detention time calculation at the 500 mg/l influent concentration and selects the larger value of the two.

The formula used for the detention time calculations when the influent BOD concentration is greater than 500 mg/l is:

$$t = (1 - S_{a}/S_{a}) / (S_{a}/S_{a} + X_{a})$$

(1)

Where:

- t = detention time, hours
- S = effluent BOD concentration, mg/l
- $S_{1} = influent BOD concentration, mg/1$
- k = reaction rate constant 1/mg hr
- Xv = mixed liquor volatile suspended solids, mg/l

When the specific influent BOD concentration is below 500 mg/liter, the following equation is added, the retention time calculated by both methods, and the greater time is reported:

## TABLE 8-2

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## CAPDET DEFAULT INFLUENT WASTE CHARACTERISTICS

TEMPERATURE	18.0	°c
SUSPENDED SOLIDS	200	<b>mg</b> /1
VOLATILE SOLIDS	60	% of SS
SETTLEABLE SOLIDS	15	mg/1
BOD <sub>5</sub>	250	mg/l
SBOD	75	mg/l
COD	500	mg/l
SCOD	400	mg/1
pH	7.6	
CATIONS	160	mg/l
ANIONS	160	mg/l
PO <sub>4</sub>	18	mg/l
TKN	45	mg/1
NH <sub>3</sub>	25	mg/l
NO <sub>2</sub>	0	mg/l
NO <sub>3</sub>	0	mg/1
OIL AND GREASE	80	mg/1

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### TABLE 8-3

#### WASTE CHARACTERISTIC REMOVAL DEFAULT VALUES FOR CAPDET PROCESSES

			WASTE CHARACTI	ERISTIC REMO	VALS			
PROCESS	BODS	TSS	COD	OIL & Grease	TKN	PHOS	с <sup>ни</sup>	SETTLEABL <b>E</b> Solids
Dissolved Air Flotation	30 <b>X</b>	80%	30%	-	102	-	-	-
Primary Clarification	32%	582	402	-	52	52	-	-
Activated Sludge	USER INPUT INFLUENT AND EFFLUENT	USER INPUT To secondary Clarifier	1.5 x BOD <sub>EFF</sub>	0	30 <b>2</b>	30 <b>%</b>	SET EQUAL TO TKN	0
Aerated Lagoon	USER INPUT Influent And Effluent	USER INPUT To secondary Clarifier	ASSUME		SAME AS	S ASL		
Multi Media Filtration	SET EFFLUENT EQUAL TO BOD SOLUBLE	60%	SET EFFLUENT EQUAL TO <sup>COD</sup> SOLUBLE	0	PASS ON THROUGH	PASS ON THROUGH	PASS Through	o
	INFLUENT		INFLUENT					

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$$t = (24 S_{0}) / (X_{1} (F/M))$$

Where:

- t = detention time, hours
- $S_{1} = influent BOD concentration, mg/l$
- $X_{..}$  = mixed liquor volatile suspended solids, mg/l
- F/M = food to microorganism ratio, lbs/day/lb usually 0.3-0.6

In comparing equations (1) and (2) it should be noted that the detention time (t) in equation (1) is influenced by influent and effluent substrate concentrations, the reaction rate "k," and the reactor MLVSS. The detention time determined by equation (2) is a function of the influent substrate concentration, MLVSS and the F/M ratio. Since the detention time determined by equation (2) is theoretically independent of the effluent concentration, a single cost is determined for all effluent target levels considered.

Figure 8-1 compares the calcuated detention time determined by the two methods. The solid lines represent the result of applying equation (1) at various target levels. The dashed line presents the results of equation (2) with the same applied influent concentration. Note that the time curves from both equations merge as the influent BOD decreases in concentration. Since the CAPDET program uses the greater of the two values calculated for detention time, cost estimates for aerators will be slightly overestimated for wastewaters with influent BOD concentrations less than 500 mg/l (LTM) and effluent concentrations of 50 mg/l (LTM).

#### ESTIMATING DISCRETE UNIT COSTS

Engineering cost estimates are presented for the following wastewater treatment processes:

- 1. Dissolved air flotation
- 2. Clarification
- 3. Activated sludge
- 4. Aerated lagoon
- 5. Multimedia filtration

The cost estimates were prepared using the CAPDET model which was modified as previously indicated to reflect industrial rather than municipal treatment costs.

An example of the engineering costs determined by CAPDET for a typical treatment application are shown for each of the above unit processes in Tables 8-4 through 8-8. The installed cost of the machinery is shown by unit, in the second column, followed by the amortization cost for the individual equipment pieces. Because the life expectancy of different

(2)



So = Influent BOD Concentration (100 mg/l)

## FIGURE 8-1 - AERATOR RETENTION TIME SENSITIVITY ANALYSIS

#### TABLE 8-4-COST SUMMAY, FLOTATION

Unit	Installed Equipment Cost Ş	Amort. Cost \$/YR	Oper . Labor Cost \$/YR	Maint, Lawor Cost \$/YR	Power Cost \$/YR \$0.04/KWH	Material Cost \$/YR	Chemical Cost \$/YR	Plant O&M Cost \$/YR
Flotation	283,767	29,114	6,664	1,259	7,841	2,887	-0-	18,651
Total	288,767	29,114	6,664	1,259	7,841	2,887	-0 <b>-</b>	18,651
								•

TOTAL CONSTRUCTION COST \$

288,767

63,528

352,295

#### TOTAL OW COST \$/1R

18,651 12,500

1,188

32,339

Plant OSM Cost Laboratory Cost

Administration

Total

Cost)

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EQUIVALENT ANNUAL COST \$/7.

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\$26,889/YR

PLANT	DESIGN	PASIS
FLOW	4.0	100

INF TSS 200 mg/l EFF TSS 40 mg/l

INDIVECT		
นากต่	1,000	(0.1 acres)
Musc. Non Construction	17.614	
A/E Design Fee	31.733	(9.01% Const.
Inspection	7.045	
Technical Costs	7.045	
Admin/Legal	7.045	
Contingencies	40,513	
Total Indirect	111,995	

DIRECT

Total Direct

Installed Equipment Contractor OH & Profit

Total Direct and Indirect 464,290
### TABLE 8-5-COST SUMMARY, CLARIFICATION (SEDUMINTATION)

Unit	instalied Equipment Cost Ş	Amort. Cost \$/YR	Oper. Labor Cost \$/YR	Maint . Labor Cost \$/YR	Power Cost \$/YR \$0.04/KWH	Material Cost \$/YR	Cnenical Cost \$/YR	Plant O&M Cost \$/YR
Prim Cia	202,253	18,579	4,288	1,532	346	2,022	-0-	8,188
Total	202,253	18,579	4,288	1,532	346	2,022	-0-	8,188
	TOTAL CON	STRUCTION COST	<u>\$</u>	TOTAL OSM	COST <u>\$/yr</u>	EQUI	VALENT ANNUAL	COST \$/YR
<u>DIRECT</u> Installed Eq Contractor O Total Direct <u>INDIRECT</u>	uipmeat M & Profit	202,253 <u>44,495</u> 246,748		Plant OSM Cost Laboratory Cos Administration Total	8,188 55 12,500 873 21,561	<u>PLAN</u> FLOW INF EFF	\$60,100/YR T DESIGN BASI 4.0 MCD TSS 200 mg/1 TSS 80 mg/1	<u>5</u>
Land Mise. Non Co A/E Design F Inspection Technical Co Admin/Legal Contingencie Total Indire	enstruction ee sts s	2,000 (0.2 a 12,337 24,055 (9,755 4,934 4,934 4,934 26,375 81,569	acre) & Const. Cost)					
Total Direct	& Indirect	328,317						

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Unit	Installed Equipment Cost \$	Amort. Cost \$/YR	Oper Labor Cost \$/YR	Maint Labor Cost \$/YR	Power Cost \$/YR \$0.04/KWH	Material Cost \$/YR	Chemical Cost \$/YR	Plant OSM Cost \$/YR
Prin Cin	167 816	9 90%	2 5/6	1 204	200	1 078	-0-	5 129
Purping	97 868	9 926	3,200	2 3 70	2 5 7	650	-0-	8 699
S Sec C	145 376	13 354	3 18/	2,170	300	3 / 51	-0-	6 404
Comp Mix	1 410 317	168,006	36,063	17 526	32) 666	6 637	-0-	379 892
Gray file	61 984	5 694	2 288	1 534	267	619	-0	4 688
Dry Beds	152 994	17 970	19 216	7 958	-0-	1 376	-0-	28 550
lloul & Lf	81,402	36,706	1,771	-0-	-0-	11,234	-0-	13,005
Total	2,072,777	261,563	66,271	31,855	325 <b>,855</b>	23,050	-0-	446,367
	TOTAL CONST	RUCTION COST \$	i	TOTAL	O&M COST \$/YR		EQUIVALENT ANNL	WL COST \$/YR

Plant OSM Cost 446,367

Laboratory Cost 19,149

6,911

472,427

Administration

Total

#### TABLE 8-6 COST SUMMARY, ACTIVATED SLUDGE

2,528,787 Total Direct INDIRECT 9,485 (.95 acre) Land Misc. Non Construction 126,439 A/E Design Fee 169,827 (6.72% Const. Cost) Inspection 50,575 Technical Costs 50,575 Admin/iegai 50, 575 Contingencies 390,809 848,285 Total Indirect

2,072,777

456,010

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c	$\supset$	
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## \$875,296/YR

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PLANT DESIGN BASIS

INF BOD 1000 mg/1 EFF BOD 30 mg/1 1 MGD FLOW EFF TSS 50 mg/l

Total Direct & Indirect 3,377,072

DIRECT

Installed Equipment

Contractor OH & Profit

#### TABLE 8-7-COST SUMARY, AERATED LACCON

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Unit	Installed Equipment Cost \$	Arort. Cost \$/YR	Oper. Labor Cosc \$/YR	Maint. Labor Cost \$/YR	Power Cost \$/YR \$0.04/KWH	Material Cost \$/YR	Chemical Cost \$/YR	Plant O&M Cost \$/YR
Prim Trut Aer Lago L Prm Cl Two St L L Prm Cl Grav Thc Dry Beds Houl & Lf	170,064 211,090 107,817 556,842 107,817 47,850 90,568 68,410	16,098 31,646 9,904 65,406 9,904 4,395 10,638 30,847	13,915 3,965 2,546 -0- 2,546 1,596 11,181 1,030	5.741 587 1,202 16,229 1,202 1,140 4,616 -0-	1,859 42,888 300 -0- 300 210 -0- -0-	4,251 3,099 1,078 1,824 1,078 478 815 11,234	-0- -0- -0- -0- -0- -0- -0-	25,766 50,539 5,126 17,953 5,126 3,424 16,612 12,264
Total	1,360,458	178,838	36,7 <b>79</b>	30,617	45,551	23,857	-0-	136,810
	TOTAL CONST	RUCTION COST \$	<u>.</u>	TOTAL OLM (	OST \$/YR		EQUIVALENT ANNU	AL COST \$/YR
<u>DIRECT</u> Installed Eq	uipment 1,3	6 <b>0</b> ,458		Plant O&M Cost Laboratory Cost Administration	136,810 28,649 28,414	·	\$467,705	
Total Direct	. 1,6	59,759		Total	19 <b>3,873</b>		PLANT DESIGN BA	<u>SIS</u>
INDIRECT								
Land Mise. Non Co	onstruction	22,000 (2.2 ac 82,988	eres)					

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Misc. Non Construction	82,988
A/E Design Fee	117,157 (7.06% Const. Cost)
Inspection	33,195
Technical Costs	33,195
Admin/Legal	33, 195
Contingencies	190,871
Total Indirect	512,601

Total Direct & Indirect 2,172,360

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#### TABLE 8-8-COST SUMARY, MULTIMEDIA FILTRATION

Unit	Installed Equipment Cost Ş	Amort Cost \$/YR	Oper. Labor Cost \$/YR	Maint. Labor Cost <b>Ş/YR</b>	Power Cost \$/YR	Material Cost \$/YR	Cnemical Cost \$/YR	Plant O&M Cost \$/YR
Filtrati Pumping	345,396 142,710	40,570 15,254	1,292 3,825	605 2,202	1,264 10,684	10,040 998	-0- -0-	13,201 17,709
Total	488,107	55,824	5,117	2,808	11,948	11,039	-0-	30,910

TOTAL CONSTRUCTION COST \$

### TOTAL OWN COST \$/YR

EQUIVALENT ANNUAL COST \$/YR

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DIRE	ECT			Plant O&M Cost	30,910	\$164,561/YR
Inst	talled Equipment	458,107		Administration	17,458	
Cont	tractor OH & Prolit	107, 373		Total	73, 304	PLANT DESIGN BASIS
Tota	al Direct	595,490				
Lanc Misc A/E Ins; Tecl Admi Cont	d c. Non Construction Design Fee pection limical Costs in/Legal tingencies	14,261 29,774 48,106 11,909 11,909 11,909 68,481	(8.05% Const.	Cost)		INF TSS 40 mg/l EFF TSS 16 mg/l AVG FLOW 4.0 MGD
Tota	al Indirect	196, 349				
Tota	al Direct & Indirect	791,839				

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pieces of equipment varies quite widely, CAPDET uses different amortization periods for different equipment. For example, the operating life expectancy of a pump may be 2 to 10 years, but a concrete structure such as a clarifier may have a life expectancy of 50 years. A listing of the life expectancy used by CAPDET for various treatment system components is presented in Table 8-9. This table also serves to define the unit process keywords shown in Tables 8-4 thru 8-8.

The lower part of the tables summarize the construction costs (direct and indirect), the operation and maintenance costs and the equivalent annual costs. Also included is the basis of the unit process design. Discrete unit cost curves for all treatment processes evaluated are in Appendix F.

#### Dissolved Air Flotation

In wastewater treatment, dissolved air flotation (DAF) is used as a clarification process to remove suspended solids, or oil and grease. It may also be used as a thickening process to concentrate various types of flotable sludges or scums.

The principal components of this system are a pressurizing pump, chemical mix tanks, air injection facilities, a retention tank, a backpressure regulating device and a flotation unit. The influent data used for the model includes wastewater flow and suspended solids concentration in the feed. Variations in both flow and concentration occur in industrial situations and consequently are considered in most designs; however, the governing parameter in the design of flotation units is the flow rate. Except in extreme cases, the solids concentration does not influence the size or operating cost of the unit.

CAPDET approximates BOD and COD removals in primary clarification and flotation at 30 percent each, with default influent values of 250 mg/l and 500 mg/l, respectively. These values are for estimation purposes for municipalities only and do not affect the costs of dissolved air flotation, which are primarily a function of flow rate.

Costs for polymer addition are included in this exercise as an option. The suggested usage of dissolved air flotation for solids removal is limited to effluent values of 50 mg/l (LTM) TSS. CAPDET assumes 80 percent removal of TSS and 30 percent removal of BOD. In practice, the reduction of suspended solids ranges from 70 to 95 percent with incidental BOD removal ranging from 10 to 50 percent. Table 8-4 is an example summary sheet for cost data at 4 MGD, and Figure 8-2 presents the cost versus flow curves from dissolved air flotation. Figure 8-2 presents costs with and without chemical addition.

#### Clarification

Clarification (sedimentation) is a solids-liquid process designed to remove suspended particles that are heavier than water. Primary clarifiers are normally used in conjunction with biological wastewater treatment systems to remove the settleable solids and a fraction of the BOD

### TABLE 8-9

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### CAPDET UNIT PROCESS REPLACEMENT SCHEDULE

Unit Processes	Key word	R Cost Item	leplacement Schedule (vears)
	<u>,</u>		()(010)
Activated Sludge Units Complete Mix	COMPLE	Mechanical Aerator (RSSA)	20
Contact Stabilization	CONTAC	Diffuser (RSPD)	30
Extended Aeration	EXIEND	Swing Arm Diffuser (KSPH)	25
Plug Flow	PLUG F	Structural (RSST)	40
Air Flotation	AIR FL	Air Flotation Unit (RSFS)	30
		Structural (RSST)	40
Aerated Lagoon	AERATE	Mechanical Aerator (RSSA)	15
		Liner (RSLL)	15
		Structural (RSST)	40
Chemical Feed Systems	None	Alum System	40
(Service life cannot		Iron Salts System	40
be changed)		Lime System	40
		Polymer System	40
Drying Beds	DRYING	4-in. Pipe (RSCP)	20
		6-in. Pipe (RSCP)	20
<i>,</i>		8-in. Pipe (RSCP)	20
		Structural (RSST)	40
Equalization	EQUALI	Floating Aerator (RSSA)	15
		Liner (RSLL)	15
		Structural (RSST)	40
Filtration	FILTRA	Filter Unit (RSSF)	20
		Pump (RSPS)	25
		Package Filter Unit (RSSF	) 20
		Structural (RSST)	40
Gravity Thickening	GRAVIT	Thickener (RSTS)	40
		Structural (RSSt)	40
Sludge Hauling and	HAULIN	Vehicle (RSSV)	6
Land Filling		Structural (RSST)	40
Lagoon	LAGOON	Steel Pipe (RSSP)	20
		Butterfly Valve (RSSV)	20
		Liner (RSLL)	15
		Structural (RSST)	40.
Microscreening	MICROS	Microscreen (RSSM)	15
		Structural (RSST)	. 40

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### TABLE 8-9 (Continued)

Unit Processes	Key word	Cost Item	Replacement Schedule <u>(years)</u>
Primary Clarification			10
Primary	PRIMAR	Mechanism (RSMS)	40
Two-Stage Lime	L PRIM	Structural (RSST)	40
Pumping	PUMPIN	Pump (RSPS)	25
		Structural (RSST)	40
Secondary Clarification			
General	CLARIF	Mechanism (RSMS)	40
Activated Sludge	A SECO		
Denitrification	D SECO	Structural (RSST)	40
Nitrification	N SECO		
Oxidation Ditch	O SECO		
Pure Oxygen	P SECO		
Trickling Filter	T SRCO		





and thereby reduce the load on the biological systems. Secondary clarifiers are used after the biological system to remove the biomass for disposal or recycle.

Clarifier costs are related to wastewater flow, since the overall cost of a clarification unit is not greatly affected by influent and effluent TSS levels. Suspended solids removal through sedimentation typically ranges from 60 to 90 percent, with incidental BOD removal ranging from 10 to 40 percent.[8-1] As previously indicated, CAPDET has been modified to use a constant 10% BOD removal through primary clarification. Clarification with floculation can producet TSS levels as low as 30 mg/1. [8-2] The cost of polymer addition is included as an option.

Secondary clarification units following biological treatment systems represented in the summary data base achieve an effluent TSS concentration of about 35 to 60 mg/l (LTM). The cost of clarification for various flow rates is presented in Figure 8-3. A cost summary example for a 4 MGD waste stream is listed in Table 8-5. CAPDET uses a circular clarifier in developing a process design.

#### Complete Mix Activated Sludge

Activated sludge is the most commonly used biological treatment process for removing soluble and colloidal contaminants from process wastewaters.[8-3] One of several possible unit process sequences for complete mix activated sludge treatment of industrial wastewater is shown in Figure 8-4. This typical design includes a complete mix activated sludge unit with primary and secondary clarification, sludge recycle, gravity thickening, sludge drying and hauling, and landfilling.

The input data required by CAPDET for this technology includes influent BOD, flow and a target effluent BOD. Average values are considered and the flow is assumed to be constant. A detailed calculation procedure is employed by CAPDET using coefficients and constants, adjusted as necessary for industrial application, as shown in Table 8-1.

The use of average input values is judged to provide cost estimates of sufficient accuracy to provide the basis for an economic impact evaluation. Although in actual site specific design practice the variability in flow and raw wastewater characteristics would be considered, the excess capacity included in the design calculations depends upon a detailed statistical study of the variability of all parameters. Once this is made, the design basis is influenced by the intuitive and subjective evaluation of the design engineer, and in some cases by corporate policy.

As an alternative to exhaustive site specific analysis, CAPDET provides an Excess Capacity Factor which automatically oversizes the activated sludge plant by the equation:

 $(ECF)t = 1.3 - 0.002 Q_{AVG}$ 

No Chemicals



FIGURE 8-3 - CAPITAL, OPERATING AND ANNUAL COSTS, SEDIMENTATION



FIGURE 8-4 - ACTIVATED SLUDGE PROCESS CONSIDERED FOR COST ESTIMATION

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Where:

(ECF)t ≈ excess capacity factor for tank volume of aerator

Q<sub>AVC</sub> = average daily wastewater flow MGD

(ECF)t is never less than 1.1

In similar calculations other components of the system also have excess capacity built in to calculations to accommodate peak demands and emergencies. Thus, without the rigorous variability analysis of the parameters, it is judged that peak demands are adequately addressed by the Excess Capacity Factor term in the model.

CAPDET predicts costs for BOD effluent levels less than 10 mg/l (LTM). However, considering the complex nature of industrial waste water and the relative accuracy of the test for BOD (+15 percent), it is reasonable to limit the general use of these costs to a minimum effluent target of 10 mg/l.

Several sample sets of curves are presented in Figures 8-5 through 8-7. Each set of curves lists capital, operating, or annual costs for a variety of effluent BOD concentrations and a fixed influent concentration. An example cost summary sheet used to plot one set of data points is presented in Table 8-6.

#### Aerated Lagoons

Aerated lagoons can provide a cost effective alternative to activated sludge treatment where sufficient land area is available. Because aerated lagoons utilize much longer detention times with lower biological solids concentrations they are less sensitive to variations in organic loading and flow.

The aerated lagoon system can approach or equal the organic removal capability of an activated sludge process, provided the unit is properly designed and operated. As indicated in Section VII, 26 plants in the Summary Data Base report using aerated lagoons as the major wastewater treatment process. Median effluent BOD and TSS for these plants are 15 mg/l and 33 mg/l, respectively.

Figure 8-8 shows the aerated lagoon system process diagram used for cost estimation. Table 8-7 shows a sample cost summary sheet, and Figures 8-9 through 8-11 graphically present a set of capital, operating and equivalent annual cost curves for aerated lagoons.

#### Multimedia Filtration

Removing finely divided suspended materials from wastewater (effluent polishing) is a growing technology of modern wastewater treatment. Two common methods are multimedia filtration and microstraining.[8-4] The design of filters depends on influent wastewater characteristics, process hydraulic loadings, method and intensity of cleaning; nature, size and depth of the filtering material, and the required quality of the

INFLUENT = 1,000 mg/L BOD



FIGURE 8-5-CAPITAL COSTS, ACTIVATED SLUDGE (1,000 mg/l)

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INFLUENT = 1,000 mg/L BOD



FLOW (MILLION GAL/DAY)

FIGURE 8-6-OPERATING COSTS, ACTIVATED SLUDGE (1,000 mg/l)

BOD INFLUENT = 1,000 mg/L



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# FIGURE 8-7-ANNUAL COSTS, ACTIVATED SLUDGE (1,000 mg/l)



AERATED LAGOON

FIGURE 8-8 - AERATED LAGOON PROCESS CONSIDERED FOR COST ESTIMATION

INFLUENT

PRIMARY

CLARIFIER

INFLUENT = 1000 mg/L BOD



# FIGURE 8-9-CAPITAL COSTS, AERATED LAGOONS (1,000 mg/l)

INFLUENT = 1000 mg/L BOD



FIGURE 8-10 - OPERATING COSTS, AERATED LAGOONS (1,000 mg/l)

INFLUENT = 1000 mg/L 800



FIGURE 8-11-ANNUAL COSTS, AERATED LAGOONS (1,000 mg/l)

final effluent. However, the costs of filtration are primarily dependent on flow rate. In general, multimedia filters are more effective than most filters and are easier and less expensive to operate for the treatment of wastewaters. Typically, either multimedia or dual-media filtration will reduce suspended solids to about 5 to 19 mg/l (LTM). Multimedia filtration has been shown to reduce TSS levels by 55 to 99 percent in various cases.[8-5] In addition to suspended solids removal, resulting incidental BOD removals ranging from 40 to 90 percent have been accomplished using multimedia filtration.[8-5]

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The design considered for cost estimation of multimedia filtration includes a layered filter with anthracite, sand, garnet sand and gravel. Pumps are used to provide pressure backwash. Surge tanks are provided to control the return of waste filter backwash water to treatment units. Surface sprayers and air blowers are included (see Figure 8-12).

Table 8-8 lists a sample cost summary sheet. The capital, operating and annual costs are plotted versus flow in Figure 8-13.

#### Polishing Ponds

Where sufficient land is available, polishing ponds may present an economically attractive alternative to multimedia filtration or microscreening as a means of reducing effluent TSS. It was determined that CAPDET could not be conveniently used to estimate polishing pond costs. Therefore, a manual estimating procedure similar to that used by CAPDET was used to determine the cost of applying polishing pond technology to four The four plants which were selected represented a wide OCPS plants. range of flow conditions. In addition, the four selected plants included one from each of the four proposed subcategories. Table 8-10 presents a cost summary for the application of polishing ponds at each of the four plants. Table 8-11 shows a comparison of the capital, operating and annual costs for polishing ponds with those costs generated by CAPDET for multimedia filtration. As indicated in the table, polishing pond costs are significantly lower. There are some minor differences in the cost estimating methods used, for example, polishing pond estimates do not include laboratory, administrative or inspection costs. These costs are, however, not significant in relation to the total cost of applying the technology. Inclusion of these additional cost items will still result in polishing pond technology being more cost effective than either multimedia filtration or microscreening.

Multimedia filtration, although more costly, is a more universally applicable technology. The significantly smaller land requirement may make filtration more attractive to plants where space is limited.

Since polishing ponds may not represent an implementable option at all OCPS manufacturing plants, multimedia filtration was used in preparing the plant-by-plant cost estimates. As indicated in Table 8-11, this represents a very conservative engineering approach in developing industry costs. Therefore, many of the OCPS industry plants will be able to provide solids control at a lower cost than that reflected by the plant-by-plant cost estimates.



FIGURE 8-12 - MULTIMEDIA FILTRATION PROCESS

228



FLOW (MILLION GAL/DAY)

#### FIGURE 8-13 - CAPITAL, OPERATING AND ANNUAL COSTS, MULTIMEDIA FILTRATION

### TABLE 8-10

### COST SUMMARY, POLISHING PONDS

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Plant: 9 Flow: 0.7 MGD Subcategory: PLASTICS

Item	Unit Cost	Amount	Total Cost(\$)
Excavation (3,000 min) Earth Prep (3,000 min) Liner	1.20 \$/yd <sup>3</sup> 0.54 \$/ft <sup>2</sup>	346 yd <sup>3</sup> 3900 ft <sup>2</sup>	3,000 3,000 2,100
Subtotal			8,100
Miscellaneous (15% of sub Subtotal	total)		<u>1,215</u> 9,315
Engineering (15%)(5,000 m Contingencies (15%) Total Installed Cost	in)		5,000 <u>1,400</u> 15,715
Land (2X pond) TOTAL CAPITAL C	10,000\$/acre OST	0.18 acre	$\frac{1,800}{17,515}$

### Operating Cost

Maintenance (10%) Sludge Disposal Total Operating Cost	7,600 \$/yr MGD	1,750 <u>530</u> 2,280
TOTAL ANNUAL COST 12.5% T.C.C. & T.O.C.		4,470

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### TABLE 8-10 (Continued)

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### COST SUMMARY, POLISHING PONDS

Plant: 97 Flow: 0.86 MGD Subcategory: NOT PLASTICS/NOT TYPE I

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Item	Unit Cost	Amount	Total Cost(\$)
Excavation	1.20 \$/yd <sup>3</sup>	4258 yd <sup>3</sup>	5,100
Liner Subtotal	0.54 \$/ft <sup>2</sup>	34225 ft <sup>2</sup>	18,480 28,680
Miscellaneous (15% of a Subtotal	ubtotal)		4,300 32,980
Engineering (15%)(5,000 Contingencies (15%) Total Installed Cost	) min)		5,000 4,950 42,930
Land (2X pond) TOTAL CAPITAL	10,000\$/acre . COST	1.6 acre	<u>16,000</u> 58,930
Operating Cost			-
Maintenance (10%) Sludge Disposal Total Operating Cost	7,600 \$/yr MGD		5,900 <u>6,540</u> 12,440
TOTAL ANNUAL COST			19,800

12.5% T.C.C. & T.O.C.

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### TABLE 8-10 (Continued)

COST SUMMARY, POLISHING PONDS

Plant: 171 Flow: 1.44 MGD Subcategory: NOT PLASTICS/TYPE I W/O OXIDATION

Item	Unit Cost	Amount	Total Cost(\$)
Excavation	1.20 \$/yd <sup>3</sup>	7130 yd <sup>3</sup>	8,560
Earth Prep	2	2	8,560
Liner	0.54 \$/ft <sup>-</sup>	55225 ft <sup>~</sup>	29,820
Subtotal			46,940
Miscellaneous (15% o Subtotal	f subtotal)		<u>7,040</u> 53,980
Engineering (15%)(5,	000 min)		8,100
Contingencies (15%) Total Installed Cost			70,180
Land (2X pond) TOTAL CAPI	10,000\$/acre TAL COST	2.6 acre	$\frac{26,000}{96,180}$

### Operating Cost

Maintenance (10%)		9,620
Sludge Disposal	7,600 \$/yr MGD	10,940
Total Operating Cost		20,560
TOTAL ANNUAL COST		32,600
12.5% T.C.C. & T.O.C.		

### TABLE 8-10 (Continued)

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### COST SUMMARY, POLISHING PONDS

Plant: 60 Flow: 5.07 MGD Subcategory: NOT PLASTICS/TYPE I WITH OXIDATION

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Item	Unit Cost	Amount	Total Cost(\$)
Excavation	1.20 \$/yd <sup>3</sup>	25100 yd <sup>3</sup>	30,120
Larth Frep Liner Subtotal	0.54 \$/ft <sup>2</sup>	180625 ft <sup>2</sup>	97,540 157,780
Miscellaneous (15% of s Subtotal	subtotal)		23,670 181,450
Engineering (15%)(5,000 Contingencies (15%) Total Installed Cost	) min)		27,200 27,200 235,850
Land (2X pond) TOTAL CAPITAI	10,000\$/acre 2 COST	8.4 acre	84,000 319,850
•			
Operating Cost			
Maintenance (10%) Sludge Disposal Total Operating Cost	7,600 \$/yr MGD		31,985 38,530 70,515
TOTAL ANNUAL COST 12.5% T.C.C. & T.O.C.			110,500

### TABLE 8-11

### COMPARISON OF POLISHING POND AND MULTIMEDIA FILTER COSTS

### MULTIMEDIA FILTRATION

Plant	Capital Cost	Operating Cost	Annual Cost
9	240,000	22,000	52,000
60	900,000	80,000	190,000
97	520,000	42,000	100,000
171	600,000	50,000	120,000
	PO	LISHING PONDS	
9	17,515	2,280	4,470
<b>6</b> 0	319,850	70,515	11 <b>0,</b> 500
97	58,930	12,440	1 <b>9,8</b> 00
171	96,180	20,560	32,600

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#### BENCH MARK ANALYSIS

A bench mark analysis was performed to compare the wastewater treatment technology cost estimates generated by CAPDET to actual industry experience. The objective of such an analysis is to determine the reasonableness of relying on the modified CAPDET costing model to estimate OCPS industry wastewater treatment costs.

Appropriate cost data were available from a total of four facilities, three from the organic chemicals and plastics and synthetics resins industry, and one from the petroleum refining industry. The data tabulated from these facilities, shown in Table 8-12, were selected because of their similarity to treatment system configurations utilized in the CAPDET estimates.

In all cases the costs were adjusted to the same cost year dollars to avoid distortions caused by changes in the construction cost index.

Although there are differences in the cost comparisons between the CAPDET plants and the industry plants, there is no definitive pattern to the differences in either magnitude or direction.

It is judged that cost differences may be due to variations in the cost accounting and cost estimating procedures which vary from one company to another.

In reference to the capital cost differences, the variations between the CAPDET estimates and the industry actuals are within the range normally associated in industrial practice with the preliminary engineering cost (+30%). To obtain more precise values requires substantially more detailed information than is available from the industrial costs studied.

It is therefore judged that CAPDET is a useful model with sufficient accuracy in cost estimating to permit an economic impact analysis to be made, providing that the industrial factors are used in the model as required.

#### EFFLUENT TARGET LEVELS

During the initial phases of the regulatory development, a series of effluent target levels for the OCPS industries were defined based on the performances demonstrated by the plants represented in the data base. Targets were selected to range from the minimum treatment level judged necessary to avoid serious potential adverse impacts on receiving waters, to the maximum degree of treatment shown to be achievable by well designed and operated plants within the industry group.

The least stringent BOD target concentration has been defined as 50 mg/l (LTM). Of the direct discharge plants in the data base, 74 percent have effluent BOD concentrations equal to or less than this value. For Plastics Only plants the most stringent target considered for BOD is 10 mg/l (LTM). This corresponds to an effluent BOD concentration slightly below the median obtained for well run Plastics Only plants (see Table 7-27). For Not Plastics Plants, the lowest effluent BOD target has been

#### TABLE 8-12-BENGE MARK ODEPARISONS

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Plant No., Treatment Typ 'and Design Parameters	e,	Reported Costs	CAPDET Costs	Difference (CAPDET - Reported)	% Difference Compared to <u>Reported Cost</u> CAPDET IS
/146 ASL @ 0.8 MCD BOD INF 720 mg/1 BOD EFF 20 mg/1	Capital,\$ Operating,\$/YR Eq. Annual,\$/YR	2,421,000 376,800 661,600	2,300,000 310,000 580,000	-121,000 - 66,800 - 81,600	5% low 18% low 12% low
#42 ASL @ 0.13 MCD BOD INF 6,000 mg/l BOD EFF 380 mg/l	Capital,\$ Operating,\$/YR Eq. Annual,\$/YR	1,082,000 190,000 317,000	900,000 140,000 260,000	-182,000 - 50,000 - 57,000	17% low 26% low 18% low
#178 ASL @ 3.5 NCD WOD INF 1,000 mg/1 BOD EFF 50 mg/1	Capiral,\$ Operating,\$/YR Eq. مرابطا,\$/YR	7,960,000 837,000 1,778,000	8,000,000 1,200,000 2,140,000	+ 40,000 +363,000 +362,000	0.5% high 43% high 20% high
Petroleum Refinery ASL @ 2.2 MCD BOD INF 134 mg/1 BOD EFF 12 mg/1	Installed Equipm Costs Only, \$	ent 630,000	868,000	+238,000	38% high
CLAR @ 2.2 MCD	Installed Equipm Costs Only, \$	ent 190,000	155,000	- 35,000	18% low
DAF without Chemical Feed @ 2.2 MCD	Installed Equipm Costs Only, \$	ent 155,000	205,000	+ 50,000	32% high

defined as 15 mg/l (LTM). This represents the median value obtained by the "Not Type I" segment of the Not Plastics Plants.

Three suspended solids targets were evaluated for engineering cost estimations. The highest effluent TSS target considered was 50 mg/l (LTM), which represents the level generally achievable using conventional clarification. Sixty-six percent of direct dischargers in the data base reported effluent TSS concentrations equal to or less than this value. The minimum target concentration considered is 20 mg/l (LTM), which is judged to be attainable using multimedia filtration or microscreening technology.

Based on these considerations, a series of target concentrations for BOD and TSS were defined. These targets, and the plants to which they are applied are presented in Table 8-13.

#### TABLE 8-13

#### EFFLUENT TARGET LEVELS

Target Level	(BOD/TSS)mg/1	Applied to
I	(50/50)	All Plants
II	(30/30)	All Plants
III	(20/20)	All Plants
IVa	(10/20)	Plastics Plants
IVb	(15/20)	Not Plastics Plants

For the plant-by-plant analysis, a tabulation of costs was prepared for each plant listing the treatment technology alternatives and corresponding estimated costs required to meet the four proposed effluent target levels for BOD and TSS. An example cost summary sheet for achieving Targets I, II and III for plant 203 is presented in Table 8-14. This example provides a reference for the following discussion.

The treatment alternatives considered for this plant-by-plant analysis include clarification, dissolved air flotation, activated sludge, aera-ted lagoons and multimedia filtration.

Based on an anlysis of the data available for engineering cost analysis, there are plants that will require (a) biological treatment for the removal of BOD and/or TSS, and/or (b) solids removal treatment for the reduction of TSS and/or BOD levels, or (c) no further treatment.

To achieve further BOD removal, the following criteria based on the generally attainable treatment levels (see Section VII) are established for the plant-by-plant analysis:

1. For plants with BOD concentrations greater than the targets and with reported solids less than the targets, biological systems are required except in the case of Criterion 3.

#### 8-14 SAMPLE PLANT-BY-PLANT COST ANALYSIS TABLE

#### PLANT BY PLANT COST ANALYSIS

#### ILPOLID WASTEWATLE DATA

TOD LIIIvent Fiana fi. TSS Ellivent Reported 203 (Million GPD), 0.106 (-9/1); وومناحب 18 (= 0/1). Treasments AS 14

ALTERNATIVE UNDEADE COSTS

Fragment Effluent Targer F

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\$ialegical	Annimelia	Capital Cart (1)	Operating Cost (1/2-)	Court1/pi	Salida	155 Allelnable	Captial Cait (3)	Operaileg Cau (1/24)	Annal Con (1/p)	Suggested Total Plant Texatories	Coplial Could 11	Operating Cast (5/Yi)	Con (S/Yr)
1. AIL	(\$0/50)	ನ್ನಳುಗ	50 00	130,000	1. CIAL	(50)	64,002	12.000	25,000	1. ASL	510.00	ຽບເມຍ	130,000
2, 14	{30/100}	<u> </u>	100, CC	3,0,00	7. DAF	(50)	75,002	14.12	2300	1. ALAKCAR	1473,000	114,000	A23,000
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Fispoid Li	(lyan) laiya				i	00/133 -(	JOV JOI							
finlegical	7100715	Capital	Operating	Annal	2012-	155	Capital	Openuling	Annual	Suggested Total	Capital	Operating	- lover	
	Arielasble	Cmi (5)	Contillyd	Contlyrt		Allelooble	Cont (1)	Con (1/p)	Cou (1/1)	flani Treaiment	Con (1)	Cur 15 Yrl	Con (15/Yr)	
1. ASL	(00.130)	20,00	SUUQ	130,000	1. MMF	(20)	226600	25:00	51,002	1. ASL/MMF	70,000	_Z=me_	182,600	
7. ALA	(30/100)	LYNAV	20,00	run	7, MICRO	(70)	250.00	35.00	51,00	1. ALA/MUE	1.150000	125000	351 000	-

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I	. 111	(70/30)	<b>ភ</b> រçឈ	5100	144,00	1. MM	(20)	250,00	<u> 75.000</u>	52.000	(1. ASL/MME)	Man	13,0012	197,000
1		(20/100)	1.160,000	JUDICUD	320,000	2, MICLO	(20)	250,002	25,600	51,000	2. ALA INME	1.65000	125,000	351.000

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Target 10 (15/20) some as Target 111

- 2. For plants with BOD concentrations greater than the targets (but not exceeding the targets by an amount equal to 60 percent of the target concentration) and with TSS concentrations greater than the targets, BOD reduction can be achieved with solids removal.
- 3. For plants with BOD concentrations within the 60 percent range of targets specified in Criterion 2 and with TSS concentrations less than the targets, BOD reduction by solids removal may be achieved if the amount of BOD to be removed is less than or equal to 75 percent of the TSS available for removal. This assumes 75 percent of the solids removed are biological and thus contribute to effluent BOD. Multimedia filters have been shown to reach effluent solids concentration of 5 to 19 mg/1 (LTM).
- 4. For plants requiring the addition of biological systems, the solids leaving the proposed secondary clarifiers are assumed to meet the TSS Target of 50 mg/l for activated sludge systems. However, for aerated lagoons some form of secondary solids separation will be required to achieve the 50 mg/l TSS Target. Additional solids treatment beyond conventional secondary clarification is required for all biological systems to achieve The exception to this rule includes cases Targets II and III. where a biological system is added to a reported biological system of the same type which produce solids concentrations that meet the proposed targets. For example, an activated sludge system added to a reported activated sludge system with excessive BOD and sufficiently low TSS is assumed to produce solids with similar settling characteristics and equivalent effluent TSS levels.
- 5. For each target, three biological alternatives are presented for the cases requiring BOD reduction by means of biological treatment. If solids removal alternatives are also presented, there are several possible combinations of bio-solid treatment alternatives. For example, an aerated lagooon system may be combined with additional clarification, dissolved air flotation, or multimedia filtration. There is no specific combination intended by placement of biological and solids alternatives on the same line of the cost sheet for each plant.

To achieve further TSS removal the following criteria are established:

1. For plants with TSS concentrations greater than the targets and also requiring biological treatment, the achievable TSS effluent concentrations are determined by the biological system (activated sludge: 50 mg/l (LTM), aerated lagoon: 100 mg/l (LTM). For Target I, additional solids removal alternatives are required for aerated lagoons, but not for activated sludge systems. Additional solids removal alternatives are required for all biological systems for Targets II and III except as noted. Clarification and dissolved air flotation are suitable alternatives to achieve Target I, but for Targets II and III, multimedia filtration or microscreening is necessary.

- 2. For plants with TSS concentrations above the proposed target levels and without additional biological system requirements, further solids removal can be achieved to reach Target I by clarification or dissolved air flotation, or for Targets II and III by multimedia filtration or microscreening.
- 3. For plants with TSS concentrations exceeding 200 mg/l (LTM), clarification or dissolved air flotation should precede multimedia filtration (or microscreening).
- 4. For plants with insufficient TSS data, but sufficient BOD data, costs based on a few additional assumptions are provided. The costs for plants with high BOD levels are not affected since the solids effluent is based on the biological treatment chosen. For plants with BOD levels within 60 percent of the proposed target, it is assumed that BOD reduction via solids removal is appropriate. This type case exists frequently among plants with sufficient BOD and TSS data. For plants meeting BOD targets, solids targets are assumed also to be within the target limits. These assumptions are considered reasonable based on the cases existing in the plants with sufficient data.

There are five plants with flows greater than 10 MGD (the maximum flow on the cost curves) and with concentrations above the targets. The flow rates are 10.7, 15.8, 16.7, 19.9, and 40 MGD. Costs for these plants were obtained from the graphs through a simple extrapolation of the curve.

The accuracy of the extension for the plant with a flow of 40 MGD is questionable; however, the only costs for this plant are for the installation of multimedia filtration. The modular nature of filtration systems suggests that simple extrapolation of costs will tend to overestimate rather than underestimate the cost of the system. In any case, the possible error which could result from this approach is small in relation to the total estimated cost for that portion of the industry requiring additional treatment.

The treatment codes used in the plant-by-plant analysis, as illustrated in Table 8-14, are defined as follows:

- 1. ASL refers to activated sludge
- 2. ALA refers to aerated lagoons
- 3. CLAR refers to clarification
- 4. DAF refers to dissolved air flotation
- 5. MMF refers to multimedia filtration
- 6. MICRO refers to microscreening

For plants with multiple streams, the costs are reviewed for treating each stream both individually and mixed. The least annual cost is considered the suggested alternative for each target. It is assumed for estimating purposes that mixing is a viable alternative even though mixing possibly would be difficult for some plants (e.g., a plant where inadequate piping between streams exists).

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#### PLANT-BY-PLANT COST RESULTS

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Costs for the preliminary plant-by-plant analysis are summarized in Table 8-15 and 8-16.

A review of the total costs indicates that for a 170 plant data base, the annual costs are \$10,346,000 per year for Target I, \$21,887,000 per year for Target II, \$28,963,000 per year for Target III and \$33,131,000 per year for Target IV.

#### BPT COST ESTIMATES

A second group of plant-by-plant cost estimates were prepared to determine the cost of complying with two sets of effluent limitations which could potentially be applied to the OCPS industry. These targets, identified as BPT (I) and BPT (II), have been developed separately for each proposed subcategory. Limitations are based on the effluent concentrations achieved by well designed and operated plants in the Summary Data Base. A more detailed explanation of the rationale for these potential limitations is presented in Sections IX and X.

The proposed effluent limitations used to develop these cost estimates are shown in Table 8-17. As the table indicates, the BPT (I) and BPT (II) differ in the level of effluent TSS control required. The costing methodology used for preparing the cost estimates was identical to that used earlier (see pages 235 thru 240). As previously explained, multimedia filters were costed as the technology used for effluent solids control. Use of polishing ponds, although not applicable to every OCPS plant, would result in significantly lower costs.

The plant-by-plant cost analysis to achieve BPT (I) and BPT (II) is presented in Table 8-18. Plant-by-plant costs are listed by subcategory.

Table 8-19 presents a summary of the total costs for additional treatment to meet the two sets of potential guidelines. Plant costs are presented by subcategory.

Table 8-20 indicates the percentage of plants in the Summary Data Base which require additional treatment to achieve BPT (I) limits. Table 8-21 presents the same information for BPT (II).

Table 8-22 presents the percentage of annual costs by subcategory to meet BPT (I) and BPT (II). Also shown for each target is the percentage of the total etimated annual cost attributable to each subcategory.

It should be noted that the plant counts in the earlier cost estimates do not agree with counts in the "BPT" costing exercise. This is due to the following factors: (1) the project data base has been updated since the earlier efforts, changing the count of direct dischargers, and (2) when the plants were re-subcategorized into 5 subcategories streams from multi-stream plants were evaluated differently. TABLE 8-15 PLANT-BY-PLANT SUGGESTED TREATMENTS AND COSTS, NON-PLASTICS

.

		LISYKELI	(38/85)			TARGET	195/951			TARCET	(91/61)	
Plant Na.	Suggested Treatment	Ceptel Ceste Ceste	Operating Costs (1/yr)	Annuel Costa (3/yr)	Suggested Treatment	Cepital Custo (1)	Operating Costs (5/ <u>y(</u> )	Annual Costs (1/yr)	Suggested Treatment		Doeraling Coarts [[/][]	Annuel Conta (1/ <u>yr</u> )
-	CLAR	145.000	18.000	000.00	- d MW	470.000	40 000	<b>65</b> ,000	MMP	470,000	0. D01	85.000
-	NO TREAT	•	•	0	NO TREAT	•	•	•	A M M	000, 210	75,000	170,000
=	ALA/CLAR	\$,510,000	000.261	1.200,000	ALA/WHF	9,150,000	D00'1ES	1,270,000	ALA/MMP	9,050,000	100,742	1,370,000
12	FO THEAT	•	•	•	NO TREAT	•	•	0	NO TREAT	•	•	•
2:	NO THEAT	•	-	0	NO TRLAT	0		a (	ASL/MMF	1.6(0,000	200,000	625,400
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22	CLAR	290,000	21,000	000,36	Nu P	720,000	65,000	150,000	N N F	7 20,000	65,000	000.021
5	NO TREAT	ç	•	•	NO TREAT	•	٥	•	AI M P	600,000	50,000	170,000
60	CLAR	000,010	000,62	002'35	r n l	000°228	50,000	100,000	M N L	000'005	<b>00, 00</b>	000'051
61	ASI,	1,150,000	120.000	263, 600	ASLIMMP	1,756.020	156,000	000.315	ASU/MMP	1,756,000	161,000	000.616
•	PO TREAT	•	•	•	NO TREAT	•	0	•	TABAT OH	•	•	•
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5	NO THEAT	0	Ð	•	NO TREAT		٥	•	NO TREAT	•	•	0
:	CLAR	105,000	11,000	30,003	N: NI P	000,016	32,020	73,000	N 41 P	340,000	500°CC	75,000
•6	CLAI	1,500,000	C00, 24	252,009	N: M F	1000 UL2 5	000,025	000°03¶	N: NI P	5,030,030	150.000	150,000
16	CLAR	165,000	16.000	31,000	41 M F	522,063	000'2+	000'00'	N N P	120,000	41,000	100,200
201	HO TUEAT	•	0	•	515 14	100,001	62,000	140,000	M AI F	100,000	62,000	143, CPD
011	HO THEAT	0	0	•	NO TREAT		•	•	KO TREAT	•	0	•
Ξ	A1 M P	500,000	41.000	11,000	A SL	900,003	95,000	000°C1Z	ASL/MHF	1. • 30,000	116,000	307,000
=	NO TREAT	•	D	ø	NO TREAT	•	•	•	ASL/MMF	7,300,000	200,001	0000,032,1
Ξ	CLAN	000,045	37,000	B4,00B	L X 2	150,030	000.11	170,000	M M: F	650, COD	17,000	170.000
120	HO TREAT	•	•	•	HO TREAT	•	۵	•	NMP	1,500,C00	130,000	010,014
121	A5L	1,200,000	115,000	280,000	ASL/MWF	000,001,1	159.000	000 050	ASU/WNP	1,960,000	174,000	120,000

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Plant /	Suggested Treatment	Capitel Coste (\$)	Operating Costs (\$/yr)	Annual Coste (\$/yr)
1	HOF	470,000	40,000	95,000
18	мр	825,000	75,000	170,000
16	ALA/1947	5,850,000	537,000	1,270,000
21	No Treat	0	0	. 0
28	ልያይ/ተያው	2,640,000	283,000	. 605,000
35	No Treat	0	0	0
36	ASL/MM	840,000	78,000	195,000
50	MMF	320,000	30,000	72,000
52	No Treat	0	0	0
53	HHLP.	675,000	60,000	135,000
57	POUR	720,000	65,000	150,000
59	<del>በ</del> ድር <b>ያ</b>	600,000	50,000	120,000
60	нмр	900,000	80,000	190,000
62	ASL/MHF	1,750,000	168,000	375,000
64	No Treat	0	0	٥
76	MAF	2,800,000	165,000	490,000
80	No Treat	0	0	0
85	ыс	340,000	32,000	75,000
94	H24F	5,000,000	250,000	850,000
97	HAP	520,000	42,000	100,000
102	<del>በር</del> ያ	700,000	62,000	140,000
110	No Treat	0	0	0
112	ASL/MMP	1,430,000	136,000	307,000
113	ASL/MMP	7,300,000	780,000	1,550,000
114	MMP	850,000	77,000	170,000
120	M	2,500,000	150,000	450,000
127	ASL/HOLP	1,960,000	174,000	420,000

### TARGET IV (15/20)

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TABLE 8-15 (cont), NON-PLASTICS

		TARGETT	(30/50)			TANCET II (30730)			TAROET 61 (20/20)			
Plant No.	Surgerlad Treatment	Capital Costa (1)	Operating Costa (1/yr)	Annual Costa (1/yr)	Suggesled Treatment	Capital Costa (1)	Operating Costa (S/yr)	Annual Costa (\$/yr)	Suggesled Treatment	Capital Conta (1)	Operating Costs (1/yr)	Annual Costs (\$/yr)
178	NO TREAT	0	0	0	MMP	620,000	69,000	150,000	NMP	850,000	68,000	150,00
130	NO TREAT	0	0	0	NO TREAT	0	0	0	NO TREAT	0	0	· 0
138	89L	800,00	105.000	236,00D	ASU/MAIF	1,220,000	141,000	315,000	ASL/MMP	1,445,000	151,000	335,000
144	HO TREAT	0	0	0	M M P	540,000	42,000	100,000	MMP	540,000	43,030	100,000
145	NO TREAT	0	0	0	MMP	600,000	50,000	120,000	MALE	600,000	50,000	130,000
160	85L	760,000	8C,000	175,000	ASL/MMF	1,180,000	118,000	260,000	ASU/MMP	1,180,000	118,000	260.000
171	NO TREAT	0	• 0	0	NIM F	600,000	50,000	120,000	AT ALC	600,000	50,000	120.000
178	ASL	1,000,000	200,000	500,000	ASL/MMP	2,700,000	263,000	640,000	ASL/MMP	3,100,000	303,000	723.000
163	NO TREAT	0	C	0	NO TREAT	0	D	0	NO THEAT	0	0	0
186	NO THUAT	0	·0	0	NO TREAT	0	0	٥	NO TREAT	0	0	0
188	NO TREAT	0	0	0	ALALF	540,000	42,000	100,000	MME	540,000	47,000	103.000
216	NO TREAT	0	0	0	NO TREAT	0	0	0	ALMP	220,000	24,000	55,200
218	ASI.	380,000	43,000	103,000	ASU/MMF	\$65,000	60,000	141.000	A51,/MMF	\$65,000	60,000	141,000
21 9	CLAR	55,000	16,000	27,000	ALM F	230,000	20,000	48,000	MME	230.002	20,000	43,000
220	DAF	76,000	15,000	23,000	MMF	250.000	25,000	\$6,000	MMP	250,000	25,000	56,000
211	A\$1.	860,000	110,000	\$30,000	ASL/MMF	1,210,000	142,000	306,000	ASL/MMP	1,450,000	152,000	346.000
276	CLAR	120,000	16,000	41,000	MMF	560,000	48,000	110,000	MME	\$60,000	45,000	110,00
228	CLAR	350,000	21,000	60,000	MINIF	750,000	10,000	160,000	MMP	750,000	70,000	160,00
231	ASL	800,000	150,000	190,000	ASL/MMF	1,120,000	146,000	282,000	ASL/MMP	1,270,000	156,000	335,60
239	NO TREAT	0	C	o	MME	163 000	16,000	35,000	MMP	165,000	16,000	22.02
247	NO TREAT	0	0	0	NO TREAT	0	D	0	MMP	2 2 5 . 0 6 0	65,000	120,00
256	NO TILEAT	0	0	0	MMP	270,000	26,000	64,000	MMP	270,000	36,000	64,00
757	ASL	FT0,000	90.000	210,000	ASL/MMP	1,330,000	129,000	302,000	ASL/MMP	1,330,000	129,000	330,00
263	NOTREAT	0	o	o	MMP	205,000	17,000	41,000	ASI,/MMP	605,000	57,000	148,00
264	NO TILEAT	0	0	0	NO TREAT	c	0 Ó	ວ່	MMP	630,000	\$3,000	130.00
270	NO TREAT	0	0	0	NO TREAT	0	0	0	NO TREAT	0	0	0
271	CLAR	185.000	18,000	40,000	MMF	560,000	44,000	110,000	MMF	360,000	44,000	110.001

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Plant #	Suggested Treatment	Capital Costs (\$)	Operating Costs (\$/yr)	-Annual Costs (\$/yr)
128	MALE	650,000	68,000	150,000
130	No Treat	0	0	0
138	ASL/MMF	1,445,000	151,000	335,000
144	NCLP	540,000	42,000	100,000
145	M0+C7	600,000	50,000	120,000
160	ASL/HMF	1,180,000	118,000	260,000
171	MCMP	600,000	50,000	120,000
178	ASL/HMF	3,100,000	303,000	720,000
183	No Treat	0	<b>0</b>	0
180	No Treat	0	. 0	0
188	ነውሮ	540,000	42,000	100,000
216	104 P	250,000	24,000	55,000
218	ASL/MMF	565,000	60,000	2 141,000
219	ммр	230,000	20,000	48,000
220	<del>ለን</del> ሆ	250,000	25,000	56,000
222	ASL/MMF	1,450,000	152,000	346,000
226	ææ	560,000	46,000	110,000
228	H:HF	750,000	70,000	160,000
231	ASL/MMP	1,270,000	156,000	332,000
2 3 9	MMF	165,000	16,000	35,000
247	юœ	725,000	65,000	150,000
256	MMF	270,000	26,000	64,000
277	ASL/MMF	1,330,000	129,000	320,000
263	ASL/MMP	605,000	57,000	146,000
264	ኵኇ	630,000	53,000	130,000
270	No Treat	0	o	0
271	къ	560,000	44,000	110,000

## TARGET IV (15/20)

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	TARGET 1 (50750)			TXRCET II (30/30)				TARGET III (70710)				
Plant Ng,	Suggested Treatment	Capital Costa (i)	Operating Costs (1/yr)	Annua) Costs (1/yr)	Suggested Treatment	Capital Costs (1)	Operating Costs (1/yr)	Annua) Corts (1/yr)	Suggested Trestment	Capital Costa <u>(5)</u>	Operating Couls (\$/yr)	Annual Costa (1/yr)
269	CLAR	340,000	22,000	67,000	V. Y P	600,000	75,000	170,000	ммр	803,60	73,000	110,000
272	CLAR	340,000	172,000	62,000	M N P	103,000	75,000	170,000	MMP	800,00	75,000	112,000
275	ASL	540,000	58,000	135,000	ASU/MMF	800,000	84,000	195,000	ASL/MMP	800,00	1 14,000	145,000
• 15	NO TREAT	0	0	C	NO TREAT	0	0	0	AL ALE	670,000	60,000	132.000
• 20	ASL	480,000	48,000	120,000	A5.JMMP	720,000	70,000	170,000	ASL/MMP	740.000	5 74,000	180,000
• •1	ASL	(90,000	54,000	175,010	ASU/MMP	720,000	73,000	170.000	ASL/MAP	810,000	) 15,000	195,000
- BL	<b>ASL</b>	1,550,000	170,000	252,200	AS./SMP	2,773,000	230,020	495,000	ASL/NNP	3,530,000	000,001	493,000
• 84	CLAR	435,000	25,000	75,000	MATE	1,050,000	85,000	210,000	ASLAMP	3,800,000	420,000	840,000
• 103	NO TREAT	0	o É	e	NO TREAT	0	0	0	NO TREAT	D	0	•
• 118	FO TREAT	0	0	0	NO THURK	D	0	0	NO TREAT	0	0	0
• 170	NO TREAT	0	0	C	NO CILAT	0	0	D	ммр	700,00	64,000	145.000
+ 175	NO TREAT	Ó	٥	0	MMP	\$20,000	44,000	98,000	ALMP	570,00	D 44,000	88,000
. 177	NOTREAT	0	0	D	NO TREAT	D Í	0	• ·	NO TREAT	4	0	0
• 114	NO TREAT	0	0		NO TREAT		0		NO TREAT	0		
TOTALS		73,773,000	7,776,000	5,158,000		47,267,000	4,185,000	10.018,000	1	78,015,00	0 0,386,000	14,879,000

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TABLE 8-15 (cont), NON-PLASTICS

"INSUPPICIENT TSS DATA.

Plant /	Suggested Treatment	Capital Costs (\$)	Operating Costs (\$/yr)	Annual Costs (\$/yr)
269	104 <b>F</b>	800,000	75,000	170,000
272	MHF	800,000	75,000	170,000
275	ASL/MHF	800,000	84,000	195,000
15	MMF	670,000	60,000	135,000
20	AS1/MMF	740,000	74,000	180,000
42	ASL/POP	810,000	75,000	185,000
61	ASL/MMF	2,220,000	230,000	495,000
84	ASL/HUF	3,900,000	420,000	890,000
103	No Treat	. 0	0	0
118	No Treat	٥	0	٥
170	<mark>የድ</mark> ዋ	700,000	84,000	145,000
175	MMP	520,000	44,000	98,000
177	No Treat	0	0	0
234	No Treat	0	0	. 0

TARGET IV (15/20)

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TABLE 8-15 (cont), NON-PLASTICS

		TARCETI	(50/50)		TADCET II (30730)				TANGET III (20/30)			
Plant No,	Suggested Treatment	Capital Costs (S)	Operating Costa (\$/yr)	Annual Costs (\$/yr)	Suggesled Treatment	Capital Costa (S)	Operating Costs (\$/yr)	Annual Costs (\$/yr)	Suggested Treatment	Capital Conts (\$)	Operating Costs (3/yr)	Annuel Coste (\$/yr)
	Mix Streems				Treet STM.I				Treal STM.1			
	HIT: NO.TRT	0	0	0	ASL/MMP	740,000	75,000	179,000	ASL/MMP	740,000	75,000	179,000
31	CLAR	185,000	18,000	40,000	MMP	\$\$0,000	44,000	108,000	MALP	550,000	44,000	108,000
23	A SL	410,000	44,000	106,000	ASL/MMP	620,000	61,000	148.000	A9L/MM <b>P</b>	\$20,000	81,000	146,000
49	NO TREAT	0	0	0	M M P	640,000	42,000	100,000	MMP	540,000	42,000	100,000
	Treat SIM		•		More HUI				Mix III			
63	ASL	1,100,000	160,000	300,000	ASU/MMF	4,200,000	449,000	960,000	ASL/MMP	4,200,000	449,000	980,000
\$6	CLAR	210,000	18.000	44,000	CLAR/MMF	E10,000	66,000	155,000	CLAR/MMP	810,000	66,000	159,000
81	ASL	750,000	35,000	190,000	ASL/MMF	1,090,000	120,000	265,000	ASL/MMP	1,330,000	176,000	283,000
P6	NO TREAT	0	0	0	ASU/MME	1,660,000	167.000	370,000	ASL/MMP	1,100,000	367,000	370,000
	CLAR	200,000	18,000	43,000	MMF	590,000	47,000	115,000	MMF	\$90,000	47,000	112,000
92	NO TREAT	0	0	0	NO TREAT	C	0	0	NO TREAT	0	0	0
98	MATE	650.000	56,000	130,000	MMF	E 5', 000	56,000	130,000	ASL/MMP	650,000	56,000	130,030
117	NO TREAT	0	0	c	NO TREAT	C	0	0	<b>NO TREAT</b>	0	٥	0
119	NO TREAT	o	0	0	NO TREAT	6	0	0	MALE	625,000	52,000	140,000
121	NO TREAT	0	0	0	NO TILEAT	0	0	0	NO TREAT	0	0	0 '
177	NO TREAT	0	0	0	MME	(50,000	\$6,000	130,000	MMF	650,000	56,000	130,000
151	HO TREAT	0	0	0	ΝΟ ΤΠΕΛΤ	0 .	0	0	ΝΟ ΤΠΕΑΤ	0	0	0
153	ASL	700,000	72,000	170,000	ASL/MMF	1,645,000	104,000	245.000	+ASL/MMP	1,045.000	104,000	345.001
158	CLAR	110,000	17.000	31,000	MME	370,000	34,000	80,000	MMP	370,000	34,000	80,00
159	NO TREAT	0	0	0 Ó	NO TREAT	0	0	0	NO TREAT	0	0	0
164	ASL	700,000	72,000	170,000	ASL/MMF	1,000,000	106,000	350,000	ASL/MMF	1,000,000	106,000	320,00
176	CLAR	190,000	18.000	40,000	MME	510,000	44,000	110,000	ммΓ	\$60,000	44,000	110,00
182	NO TREAT	0	o	0	N:M P	450,000	38,000	90,000	ASU/MMP	1.250,000	177,000	280,00
1.17	ASL	1,500,000	150,000	363,000	ASU/MMP	2,125 000	202,000	480,000	ASL/MMP	2,325,000	222.000	\$40,00
192	DAF	76.000	15,000	24,000	MMF	750,000	25.000	60,000	MMP	250,000	25,000	60,00

		TARGET IV (1	5/20)	
Plant 🖡	Treatment	Capital \$	Operational \$	Annual \$
8	ASL/HMF	740,000	75,000	179,000
31	MMF	550,000	44,000	108,000
32	ASL/HHF	620,000	61,000	148,000
49	MMF Mix I 6 II	540,000	42,000	100,000
63	ASL/MMP	4,200,000	449,000	960,000
66	CLAR/MMP	B10,000	66,000	159,000
81	ASL/MMP	1,220,000	126,000	285,000
85	ASL/MMP	1,660,000	167,000	370,000
88	ASL/MMP	1,150,000	120,000	260,000
92	No Treat	٥	0	0
98	ASL/HMP	650,000	56,000	130,000
117	No Treat	٥	0	0
119	MMF	625,000	52.000	140,000
121	No Treat	0	0	0
122	MMF	650,000	56,000	130,000
151	MMF	700,000	64,000	148,000
153	ASL/MMF	1,045,000	104,000	245,000
158	ASL/MMF	700,000	73,000	170,000
159	MMF	460,000	39,000	90,000
164	A SL/MMF	1,060,000	106,000	250,000
176	MUP	560,000	44,000	110,000
182	ASL/MMP	1,250,000	122,000	280,000
187	ASL/HMF	2,325,000	222,000	540,000
192	MLE	250,000	25,000	60,000

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TABLE 8-15 (cont), NON-PLASTICS

		TANCET I	(50/50)		TARCET II (30730)				TA NCET III (10/20)			
Plant No.	Suggested Treatment	Capital Costa (\$)	Operating Costs (\$/yr)	Annus) Costs (\$/yr)	Suggesled Treatment	Capital Costs (\$)	Operating Costs (\$/yr)	Annue) Costs (\$/yr)	Suggested Treatment	Capital Costa (\$)	Operating Costs (\$/yr}	Annuel Costs (\$/yr)
	ASL	850,000	110,000	270,000	ASL/MMP	1,235,000	141,000	303,000	ASI./MMP	1,535,000	171,000	373,000
195	NO TREAT	0	0	0 Ó	MMP	1,050,000	85,000	710,000	MMP	1.050.000	85,000	310,000
201	ASL	480,000	57,000	120,000	ASL/MMF	518,000	72,000	139,000	ASI./MMP	710,000	83,000	173,000
203	ASL	510,000	50,000	130.000	ASL/MMP	760,000	75,000	187,000	ASL/MMP	790,000	79,000	197,000
705	CLAR	30,000	14,000	18,000	MATP	160,000	15,000	34.000	MMP	160,000	15,000	34,000
206	CLAR	350,000	27,000	64,000	MME	850,000	75.000	170,000	MME	850,000	75,000	170,000
208	NO TILEAT	0	0	C	MMP	400,000	30,000	e5,000	MMP	400,000	36,000	\$5,000
230	ASL	700,000	77,000	170,000	ASI./MMP	1,050,000	104,000	246.000	ASL/MMP	1.050.000	104,000	246.000
235	ASL	2,300.000	270,000	560,000	ASL/MMP	3,700,000	350,000	750,000	ASL/MMP	3,200,000	350,000	750,000
236	CLAR	330,000	22,000	62,000	MANE	800,000	75,000	170,000	MMP	800,000	75,000	170,000
245	NO TREAT	D	0	0	NO TREAT	0	0		NO TREAT	0	0	0
248	ASL.	1.300,000	130,000	290,000	ASU/MATE	1,975,000	181,000	410,000	ASL/MMP	1,525,000	181,000	410,000
249	NO TREAT	0	0	c	NOT TREAT	0	0	•	NOTREAT	0	0	0
258	NO TREAT	0	0	0	ASL/MMP	550,000	59,000	136,000	ASL/MMP	550,000	59,000	138,000
• 6	ASL	510,000	58,000	130,000	ASU/MMF	755.000	81,000	182,000	ASL/MMP	815,000	83,000	194.000
* #1	ASL	510,000	58,000	130,000	ASL/MMF	755,000	81,000	182,000	ASL/MMF	\$15,000	83,000	194,000
• 163	ASL	410,000	44,000	107,000	ASLIMMP	015.000	61,000	147.000	ASL/MMP	615,000	61,000	141,000
• 204	NO TREAT	0	0	0	NO TREAT	0	Q	0	ммр	370,000	16,000	67,000
• 259	NO TREAT	0	0	0	NO TREAT	0	0	0	NO TREAT	0	0	0
• 268	ASL	3,200,000	320.000	580,000	ASL/MMF	2,880,000	380,000	720,000	ALA/MMP	3,880,000	350,000	840,000
• 781	NO TREAT			P	NO TREAT	_0		0	NOTREAT	0	0	°
TOTALS		17,253,000	1,973,000	4,229,000		35,483,000	3,607,000	8,050,00		39,130,000	3,815,000	8,780,000

INSUPPICIENT TSS DATA.

Plant #	Treatment	Capital \$	Operational \$	Annual \$
193	ASL/HMF	1,535,000	171,000	373,000
195	MAP	1,050,000	85,000	210,000
201	ASL/MMF	710,000	83,000	173,000
203	ASL/MMP	790,000	79,000	197,000
205	have	160,000	15,000	34,000
206	ASL/MMF	2,100,000	250,000	510,000
208	HMF	400,000	36,000	85,000
230	ASL/MMF	1,050,000	104,000	246,000
235	ASL/MMF	3,200,000	350,000	750,000
2 36	ASL/MMF	2,000,000	230,000	490,000
245	No Treat	D	0	0
248	ASL/HHF	1,925,000	181,000	410,000
249	No Treat	0	0	0
258	ASL/MMP	550,000	59,000	136,000
6	ASL/MMF	815,000	83,000	194,000
81	ASL/MP	815,000	83,000	194,000
163	ASL/MMF	615,000	61,000	147,000
204	HHF	270,000	26,000	62,000
25 <b>9</b>	No Treat	0	0	0
268	ALA/MMP	3,880,000	350,000	840,000
281	юлг	540,000	42,500	100,000
TOTALS		44,170,000	4,401,500	10,013,000

TARGET 1V (15/20)

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TABGLT I (55/56)				талсет II (36736)				TANGET IN (23792)				
Plant No.	Surgested Trentment	Cep(1) Costs (1)	Operating Costs (Myr)	Annunl Costs (\$/yr)	Polesler Trestment	Capital Costa (1)	Operating Costa (1/yr)	Annual Costa (1/yr)	Suggested Treatment	Capital Costa (1)	Operating Custs (1/yr)	Annual Conta (1/yr)
;	NO TREAT	0	0	0	NO TREAT	0	0	0	ммр	840,000	54,020	117,000
3	NO TREAT	0	0	c	ASU/MMP	590,000	89,000	203,000	ASU/MMP	\$90,000	95,000	333,000
ş	NO THEAT	ť	C	D	NO THEAT	0	0 <sup>°</sup>	U I	MMP	340,000	23,666	11,010
11	NC THEAT	0	0	t t	NO THEAT	0	D	D	NO TREAT	0	o	0
1 *	NO THEAT	C	t	0	ASL/MACC	L20,000	67,JOO	110,000	ASUMME	6.90,000	67,000	1.0,000
15	NO TRUAT	ũ	í	C	NO DREAT	U	0	ເ່	SO TREAT	D	0	:
::	NO TRUAT	t	0	3	NC TRPAT	0	0	0	MIN' P	1,750,000	120,000	1.1,03
:•	CUAR	• \$5,000	25.00	23,000	81 M P	1,750.000	8:,000	245,000	M St E	1,350,000	65,000	2+5,000
14	TA21/T OA	0	9	Û	NO TREAT	0	0	Q	At 51 11	140,000	10,100	50,003
: 9	FOTRIAT	D	ð	0	SO TREAT	c	5	0	NO CREAT	c	0	0
	FO THEAT	t	C	(	NU TREAT	ů	:	C	NO TREAT	a	a	D
45	NO TREAT	D	0	9	ROTANAT	C	:	ſ	MACE	E50,000	36,CGD	113,000
5.6	NO THEAT	D	C	0	M M P	190,030	:a.co:	£1,000	N"NL P	260,000	26,000	15,630
£1	NO THEAT	c	6	c	52 ** P	1,259,200	31,200	245,000	MINE P	1,350,000	81,000	a+5, 010
72	HO TIENT	0	0	Ú	NO TREAT	0	2	5	NO TREAT	C	٥	C
2.7	NO THEAT	0	0	0	NO TREAT	G	0	0	NO TREAT	0	D	C
19	CLAR	220,000	15,000	46,000	M M F	L25,C00	52,000	170,030	MMP	625,000	57,000	1:0,000
60	NOTHEAT	0	0	0	NO TREAT	3	0	0	NO TREAT	C	0	t
51	CLAR	256,000	\$E,000	50,000	N' 51 P	650,000	50.000	130,000	MMF	810,000	35,000	110,012
52	DAP	80,000	15,600	24,000	Nº Aº P	260,000	26,00 <b>0</b>	60,020	MME	260,000	21,000	60,000
<u>; </u>	NO TREAT	٢.	c	0	SC TREAT	Ū	C	c ُ	NO TREAT	C I	C	2
1::	SO TREAT!	C	¢	c	NO TREAT	Q	0	с	NO TREAT	D	0	C
124	CLAR	220,000	IP,COD	(5,000	N" 51 T	600,000	50,COO	136,000	MME	600,000	50,000	110,000

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# TABLE 8-16 PLANT-BY-PLANT SUGGESTED TREATMENT AND COSTS, PLASTICS

The multiple streams for aptimum annual cost.

Plant /	Trestment	Capital \$	Operational \$	Annusi Ş
2	MMP	640,000	54,000	127,000
3	ASL/MMF	990,000	99,000	233,000
9	HOLP	240,000	22,000	52,000
10	No Treat	٥	0	0
17	ASL/MMP	680,000	67,000	160,000
19	No Treat	0	0	0
27	ASL/MMF	5,400,000	680,000	1,400,000
29	MD:F	1,250,000	95,000	245,000
34	ASL/MMF	450,000	49,000	115,000
39	No Treat	. 0	0	0
44	MMF	470,000	40,000	92,000
45	MAF	650,000	56,000	130,000
54	<del>የ</del> ጎፓ	280,000	28,000	68,000
65	ASL/MMF	3,400,000	410,000	830,000
73	No Treat	0	D	0
77	No Treat	D	0	0
89	ASL/MMF	1,300,000	140,000	300,000
90	No Treat	0	0	0
91	MMF	650,000	56,000	130,000
93	KMF	260,000	26,000	60,000
96	No Treat	0	0	٥
100	No Treat	0	٥	0
104	MMF	600,000	50,000	120,000

# TARGET IVA (10/20)

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		TARGET I	(50/50)			TARCET II (16710)			TAROET NI (30/20)			
Plent No.	Suggeslad Treatment	Capitat Costs (1)	Operating Costs (1/yr)	Annual Costa (\$/yr)	Suggested Treatment	Capital Costa (\$)	Operating Costa (S/yr)	Annual Costs (I/yr)	Suggested Treatment	Capital Casta (1)	Operating Costa (1/yr)	Annuel Coste (1/yr)
103	NOTREAT		•	 0	M M P	343,000	32,000	15,D00		340,008	33,000	75,000
107	CLAR	370,000	70,000	52,000	MALP	700,000	87,000	140,000	A1 55 P	700,000	43,000	140,000
109	CLAR	220,000	19,000	45,000	MMP	4C0,000	50,000	120,000	MATE	000,000	50,000	170,000
111	NO TREAT	0	•	D	NO TREAT	0	0		NO TR <b>eat</b>	0	0	0
124	NO TREAT	0	.0	0	ASL/MMF	1,070,000	100,000	248,000	ASL/MMP	1,070,000	108,300	148,000
125	NO TREAT	0	C	0	ASL/MMP	1,710,000	<b>116</b> ,000	270,000	ASL/NMP	1,310,000	119,000	370,600
126	NO TILEAT	0	0	5	NO TREAT	0	0	0	MMP	380,000	34,600	80,000
132	NO TREAT	0	0	0	NO TREAT	D	0	•	NO TREAT	0	0	0
146	NO THEAT	0	0	٥	NO TILEAT	0	0		MAYE	360,000	34,000	80,000
147	NO TREAT	٥	0	0	NO TREAT	0	0	•	MM 9	330,000	31,000	73,000
150	NO TREAT	0	0	0	ASL/MMF	1,300,000	136,000	785,DOB	ASL/MMP	1,300,000	178.00	385,000
152	NO TREAT	C	0	D	NO TREAT	0	0	0	NO TREAT	D	0	Û
157	NO TILEAT	D	0	Û	ALM F	600, <b>000</b>	46,000	115,000	MMP	680,000	46,000	110,000
174	NO TREAT	D	Q	D	NO TREAT	Q	•	0	NO TREAT	0	0	0
119	NO TREAT	0	۵	a	ммГ між 1•Ц	420,000	38,000	85,000	MALP Mile F.4. II	430,000	39,000	15,000
184	ΝΟ ΤΛΕΛΤ	0	٥	0	NO TREAT	0	0	0	MNP	360,000	48,000	110,000
119	A51,	680,000	10,000	160.000	ASU/MMF	1.020.000	102,000	235,000	ASL/MMP	1,020,000	102,000	332.000
194	NO TREAT	D	0	D	NO TREAT	٥	0	0	NO TREAT	0	0	0
196	NOTREAT	C	0	0	NO TREAT	0	0	0	NO TREAT	. 0	٥	0
202	NO THEAT	D	a	۵	NO TREAT	0	0	D	NO TILEAT	•	0	0
310	NO THEAT	D	0	0	NO TREAT	٥	0	Q	NO TREAT	Ð	Q	0
217	NO TREAT	. 0	0	0	NO THEAT	0	0	0	NO TREAT	0	0	0

# TABLE 8-16 (cont), PLASTICS

Plant #	Treatment	Capital \$	Operational \$	Annual \$
105	MMF	340,000	32,000	75,000
107	ASL/HMF	1,600,000	175,000	370,000
109	MMF	600,000	50,000	120,000
111	No Treat	0	0	D
124	ASL/MMF	1,070,000	108,000	248,000
125	ASL/MMF	1,210,000	118,000	270,000
126	MHP	360,000	34,000	80,000
132	ASL	640,000	66,000	155,000
146	MMP	360,000	34,000	80,000
147	MMF	330,000	31,000	73,000
150	ASL/HMF	1,300,000	126,000	285,000
152	No Treat	0	0	0
157	M17	600,000	46,000	110,000
174	No Treat	o	0	0
179	MMP Mir I & II	420,000	38,000	85,000
184	104 P	560,000	46,000	110,000
189	ASL/HHP	1,020,000	102,000	235,000
194	No Treat	0	0	0
196	No Treat	0	0	0
202	ASL	540,000	58,000	135,000
210	No Treat	0	0	0
217	ASL	420,000	44,000	107,000

# TARGET IVA (10/20)

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		TA AGET 1 (50/50)			TANCET (1 (18718)				TANCET III (10/10)			
Pis∧t Ha,	Siggested Treatment	Copila) Custa (\$)	Operating Costs (\$/yr)	Annini Costa (\$/yr}	Suggested Treatment	Capilal Costs (1)	Opriating Costs (1/yr)	Annunt Casta (\$/yr)	Suggested Trestment	Capital Costa (\$)	Opensting Costs (1/yr)	Anquel Costs (1/yr)
113	NO TREAT	0	0	0	NO TREAT	0	•		NO TREAT		•	
224	DAF	78,00B	14,000	23,000	MMF	250,000	23,000	52,000	ммР	250,000	33,000	32,000
329	NO TREAT	0	•	0	NO TREAT	0	0	•	NO TREAT	0	•	0
748	A96	780,000	84,000	190,000	ASU/MMF	1,190,000	121,000	375,000	ASL/MMP	1,100,000	111,000	275,000
254	NO TREAT	D	'0	0	ΛΟ ΤΠΕΛΤ	Q	Q	Q D	NO TRUAT	0	•	Q
267	CLAIL	105,000	17,000	30,000	PINP .	340,000	37,000	75,000	MMP	240,000	33,000	75,000
273	CLAN	330,000	12,000	61,000	MME	115,000	70,000	161,000	MMP	175,008	70,000	160,000
277	NO TREAT	Q	°.	0	MINE	675,000	60,000	135,000	MMP	675,000	80,000	135,000
787	NO TREAT	D	٥	0	NO TREAT	0	o'	0	NÚ TREAT	0	0	٥
• 75	NO TREAT	0	0	0	SO TREAT	Û	٥	0	MMP	390,000	35,000	80,000
- 105	NO TREAT	0	C	0	NO TREAT	D	0	٥	NO TREAT	0	٥	C C
• 123	NO TREAT	0			M M F	«SD,000	39.000	<u> </u>	NIN] P	450,000	79,000	90,000
TOTALS		3,716,000	344,000	809,0 <b>00</b>		17.525,000	1,525,000	3,738.000		\$5,885,00	0 2.331,000	5.483,000

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TABLE 8- 16(cont), PLASTICS

\*INSUPPICIENT TSS DATA.

		Capital 5	Treatment	Plant V
94,000	40,000	480,000	ASL	223
120,000	52,000	470,000	ASL/HHT	224
0	0	٥	No Treat	229
275,000	121,000	1,190,000	ASL/MHP	246.
96,000	41,000	510,000	HMP	254
160,000	68,000	660,000	ASL/MMP	262
480,000	230,000	200,000	ASL/MMP	273
135,000	60,000	675,000	10:17	277
0	0	0	No Treat	287
175,000	75,000	730,000	ASL/MHE	75
0	0	0	No Treat	106
190,000	85,000	830,000	ASL/MMP	233

36,175,000

3,752,000

8,325,000

## TARGET IVA (10/20)

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TOTALS

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# POTENTIAL BPT EFFLUENT LIMITATIONS

# BPT (I)

Subcategory	BOD mg/1	TSS mg/l
Plastics	14.5	24
Not Plastics -Type I with Oxidat:	ion	
High Flow Low Flow	26.0 36.0	62 89
Not Plastics Type I w/o Oxidation	n 24.5	34.5
Not Plastics NOT Type I	17.0	29

# BPT (II)

Subcategory	BOD mg/1	TSS mg/l							
Plastics	14.5	23							
lot Plastics - Type I with Oxiation									
High Flow Low Flow	26.0 36.0	42 42							
Not Plastics Type I w/o Oxid	ation 24.5	27							
Not Plastics NOT Type I	17.0	26							

## PLANT-BY-PLANT COST ESTIMATES

SUBCATEGORY: PLASTICS ONLY

<u>BPT (1)</u>				BPT (II)				
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual
Plant #	Treatment	Cost (\$)	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)
2	No Treat.	0	0	0	l MMF	640 000	54 000	127 000
3	ASL/MMF	990,000	99,000	233 000	ASL/MME	990,000	99,000	233,000
9	MMF	240,000	22,000	52,000		240,000	22,000	52,000
10	No Treat.	0	0	0	No Treat	0	0	0
17	ASL/MMF	680,000	67,000	160,000	ASL/MME	680,000	67 000	160,000
19	No Treat.	0	0	0	No Treat.	0	0,000	0
29	MMF	1.250.000	95,000	245,000	MMF	1.250.000	95 000	245 000
39	No Treat.	0	0	0	No Treat.	0	0	0
45	No Treat.	0	0	0	No Treat.	Õ	Ő	õ
54	ASL/MMF	880,000	88.000	218,000	ASL/MMF	880,000	88,000	218,000
73	No Treat.	Ó	0	0	No Treat.	0	0	0
77	No Treat.	0	0	0	No Treat.	0	0	0
90	No Treat.	0	0	0	No Treat.	0	Õ.	Õ
91	MMF	650,000	56,000	130,000	MMF	650,000	56,000	130,000
93	MMF	260,000	26,000	60,000	MMF	260,000	26,000	60,000
96	No Treat.	ó	Ó	0	No Treat.	0	0	0
100	No Treat.	0	0	0	No Treat.	0	Õ	0 .
104	MMF	600,000	50,000	120,000	MMF	600,000	50,000	120.000
105	MMF	340,000	32,000	75,000	MMF	340,000	32,000	75.000
109	MMF	600,000	50,000	120,000	MMF	600,000	50,000	120,000
111	No Treat.	Ó	Ó	ó	No Treat.	0	0	0
124	ASL/MMF	1,070,000	108,000	248,000	ASL/MMF	1,070,000	108.000	248,000
125	ASL/MMF	1,210,000	118,000	270,000	ASL/MMF	1,210,000	118,000	270,000
126	MMF	360,000	34,000	80,000	MMF	360.000	34,000	80,000
146	No Treat.	Ó	Ó	0	No Treat.	0	0	0
147	MMF	330,000	31,000	73,000	MMF	330,000	31,000	73.000
150	ASL/MMF	1,300,000	126,000	285,000	ASL/MMF	1.300.000	126,000	285,000
152	No Treat.	Ō	Ó	ó	No Treat.	0	0	0
157	MMF	600,000	46,000	110.000	MMF	600.000	46,000	110.000

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## PLANT-BY-PLANT COST ESTIMATES

SUBCATEGORY: PLASTICS ONLY

	<u>BPT (I)</u>				BPT (II)			
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual
Plant #	Freatment	Cost (\$)	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)
174	No Treat.	0	0	0	No Treat.	0	0	0
179	MMF ·	420,000	38,000	85,000	MMF	420,000	38,000	85,000
184	MMF	560,000	46,000	110,000	MMF	560,000	46,000	110,000
189	ASL/MMF	1,020,000	102,000	235,000	ASL/MMF	1,020,000	102,000	235,000
196	ASL/MMF	2,640,000	290,000	615,000	ASL/MMF	2,640,000	290,000	615,000
210	No Treat.	0	- 0	0	No Treat.	0	0	0
229	No Treat.	0	0	0	No Treat.	0	0	0
246	ASL/MMF	1,190,000	121,000	275,000	ASL/MMF	1,190,000	121,000	275,000
277	MMF	675,000	60,000	135,000	MMF	675,000	60,000	135,000
287	No Treat.	0	0	0	No Treat.	0	0	0
27	MMF	1,750,000	120,000	300,000	ASL/MMF	7,150,000	800,000	1,700,000
34	ASL/MMF	690,000	64,000	165,000	ASL/MMF	690,000	64,000	165,000
44	No Treat.	0	0	0	No Treat.	0	0	0
52	No Treat.	0	0	0	No Treat.	0	0	0
65	MMF	1,250,000	95,000	245,000	MMF	1,250,000	95,000	245,000
75	ASL/MMF	1,120,000	110,000	255,000	ASL/MMF	1,120,000	110,000	255,000
89	ASL/MMF	1,925,000	192,000	420,000	ASL/MMF	1,925,000	192,000	420,000
107	ASL/MMF	1,900,000	237,000	510,000	ASL/MMF	1,900,000	237,000	510,000
119	No Treat.	0	0	0	MMF	125,000	52,000	140,000
134	MMF	320,000	31,000	71,000	ASL/MMF	960,000	97,000	226,000
192	MMF	250,000	25,000	60,000	ASL/MMF	750,000	78,000	190,000
20 <b>2</b>	MMF	290,000	26,000	60,000	ASL/MMF	830,000	84,000	195,000
217	MMF	210,000	16,000	41,000	MMF	210,000	16,000	41,000
223	No Treat.	Ó	0	0	MMF	480,000	40,000	94,000
224	ASL/MMF	720,000	75,000	172,000	ASL/MMF	720,000	75,000	172,000
233	ASL/MMF	980,000	119,000	280,000	ASL/MMF	980,000	119,000	280,000
254	No Treat.	Ó	Ó	Ó	MMF	510,000	47,000	96,000
262	ASL/MMF	1,000,000	100,000	235,000	ASL/MMF	1,000,000	100,000	235,000
273	ASL/MMF	2,775,000	300,000	640,000	ASL/MMF	2,775,000	300,000	640,000

# PLANT-BY-PLANT COST ESTIMATES

SUBCATEGORY: TYPE I W/ OXIDATION - HIGH FLOW

	BPT (I)					BPT (II)				
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual		
Plant #	Treatment	Cost (\$)	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)		
		<b>5</b> 00 000	5.0.000	-						
20	ASL	500,000	52,000	130,000	ASL/MMF	740,000	74,000	180,000		
31	CLR	185,000	18,000	40,000	MMF	550,000	44,000	108,000		
36	ASL	600,000	58,000	145,000	ASL/MMF	840,000	78,000	195,000		
49	No Treat.	0	0	0	No Treat.	0	0	0		
57	MMF	720,000	65,000	150,000	MMF	720,000	65,000	150,000		
60	MMF	900,000	80,000	190,000	MMF	900,000	80,000	190,000		
61	ASL	1,550,000	170,000	360,000	ASL/MMF	2,220,000	230,000	495,000		
62	ASL	1,150,000	120,000	260,000	ASL/MMF	1,750,000	168,000	375,000		
63	ASL	3,100,000	360,000	740,000	ASL/MMF	4,200,000	449,000	960,000		
76	No Treat.	0	0	0	No Treat.	0	0	Ó		
80	No Treat.	0	0	0	No Treat.	0	0	0		
84	ASL	2,850,000	335,000	680,000	ASL/MMF	3,900,000	420,000	890,000		
98	ASL	1,400,000	150,000	330,000	ASL/MMF	2,050,000	206,000	460,000		
102	No Treat.	0	Ó	Ó	No Treat.	0	0 0	Ó		
103	No Treat.	0	0	0	No Treat.	0	0	0		
110	No Treat.	0	0	0	No Treat.	0	0	0		
112	ASL	930,000	95,000	210,000	ASL/MMF	1,430,000	136,000	307.000		
113	No Treat.	Ó	Ó	· ó	No Treat.	0	0	0		
114	No Treat.	0	0	0	MMF	850,000	77.000	170.000		
127	ASL	1,400,000	130,000	310,000	ASL/MMF	1.960.000	174,000	420,000		
175	MMF	520,000	44,000	98,000	MMF	520,000	44,000	98,000		
176	MMF	560,000	44,000	110,000	MMF	560,000	44,000	110.000		
187	ASL	1,700,000	170,000	420,000	ASL/MMF	2.325.000	222,000	540,000		
193	ASL	1,200,000	140,000	300,000	ASL/MMF	1,535,000	171.000	373,000		
195	No Treat.	0	Ó	0	No. Treat.	0	0	0		
216	No Treat.	0	0	0	No Treat.	0	0	0		
220 ·	ASL	450,000	55,000	120.000	ASL/MMF	700.000	80,000	176.000		
222	ASL	1.100.000	120,000	270,000	ASL/MMF	1.450.000	152,000	346,000		
228	ASL	1.900.000	220,000	450,000	ASL/MMF	2.650.000	290,000	610,000		
234	No Treat.	0	0	0	No Treat.	_,,0	0	0		

PLANT-BY-PLANT COST ESTIMATES SUBCATEGORY: TYPE I W/ OXIDATION - HIGH FLOW

		BPT (I)		1		BPI	(11)	
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual
Plant #	Treatment	Cost (\$)	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)
	·							
235	ASL	2,300,000	270,000	560,000	ASL/MMF	3,200,000	350,000	750,000
248	ASL	1,300,000	130,000	290,000	ASL/MMF	1,925,000	181,000	410,000
249	No Treat.	, <b>O</b>	0	0	No Treat.	0	0	0
257	ASL	870,000	90,000	210,000	ASL/MMF	1,330,000	129,000	302,000
272	CLR	340,000	22,000	62,000	MMF	800,000	75,000	170,000
88	ASL	1,150,000	120,000	260,000	ASL/MMF	1,740,000	167,000	375,000
158	ASL	700,000	73,000	170,000	ASL/MMF	1,070,000	107,000	250,000
206	MMF	850,000	75,000	170,000	MMF	850,000	75,000	170,000
236	MMF	800,000	75,000	170,000	MMF	800,000	75,000	170,000
SUBCATEO	ORY: TYPE I	W/ OXIDATIO	N - LOW FLOW	1				
42	ASL	490,000	54,000	125,000	ASL/MMF	720,000	73,000	170,000
50	No Treat.	0	0	0	MMF	320,000	30,000	72,000
81	ASL	750,000	88,000	190,000	ASL/MMF	1,090,000	120,000	265,000
81	ASL	510,000	56,000	130,000	ASL/MMF	765,000	81,000	188,000
81 (tot	al 2 streams	)		-				
		1,260,000	144,000	320,000		1,855,000	201,000	453,000
138	ASL	800,000	105,000	230,000	ASL/MMF	1,220,000	141,000	315,000
160	ASL	760,000	80,000	175,000	ASL/MMF	1,180,000	118,000	260,000
163	ASL	410,000	44,000	107,000	ASL/MMF	615,000	61,000	147,000
177	No Treat.	Ó	Ó	0 1	No Treat.	Ò	Ó	0
188	No Treat.	0	0	0	No Treat.	0	0	0
218	ASL	380,000	43,000	103,000	ASL/MMF	565,000	60,000	141,000
219	CLR	55,000	16,000	22,000	MMF	230,000	20,000	48,000
239	No Treat.	0	0	0	No Treat.	Ō	0	0
268	ASL	2,200,000	320,000	580,000	ASL/MMF	2,880,000	380,000	720,000
271	CLR	185,000	18,000	40,000	MMF	560,000	44,000	110,000

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# PLANT-BY-PLANT COST ESTIMATES

SUBCATEGORY: Type I w/o Oxidation

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	BPT (I)					BPT (II)				
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual		
Plant #	Treatment	<u>   Cost (\$)</u>	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)		
							_			
1	ASL/MMF	1,330,000	178,000	295,000	ASL/MMF	1,330,000	178,000	295,000		
15	No Treat.	0	0	0	No Treat.	0	0	0		
16	ALA/MMF	5,850,000	537,000	1,270,000	ALA/MMF	5,850,000	537,000	1,270,000		
28	MMF	740,000	68,000	155,000	MMF	740,000	68,000	155,000		
32	ASL/MMF	620,000	61,000	148,000	ASL/MMF	620,000	61,000	148,000		
35	No Treat.	0	0	0	No Treat.	0	0	0		
53	MMF	675,000	60,000	135,000	MMF	675,000	60,000	135,000		
64	No Treat.	0	0	0	No Treat.	0	0	0		
85	MMF	340,000	32,000	75,000	MMF	340,000	32,000	75,000		
117	No Treat.	0	0	0	No Treat.	Ó	0	Ó		
118	No Treat.	0	0	0	No Treat.	0	0	0		
128	No Treat.	0	0	0	MMF	650,000	68,000	150,000		
130	No Treat.	0	0	0	No Treat.	Ő	Ó	Ó		
164	ASL/MMF	1,060,000	106,000	750,000	ASL/MMF	1,060,000	106,000	750,000		
171	MMF	600,000	50,000	120,000	1 MMF	600,000	50,000	120,000		
170	MMF	700,000	64,000	145,000	MMF	700,000	64,000	145,000		
178	ASL/MMF	3,100,000	303,000	720,000	ASL/MMF	3,100,000	303,000	720,000		
183	No Treat.	0	Ó	Ó	No Treat.	0	0	0		
201	ASL/MMF	710,000	83,000	173,000	ASL/MMF	710,000	83,000	173,000		
203	ASL/MMF	790,000	79,000	197,000	ASL/MMF	790,000	79.000	197.000		
204	No Treat.	ó	ó	0	No Treat.	0	0	0		
230	ASL/MMF	1,050,000	104,000	246,000	ASL/MMF	1.050.000	104,000	246,000		
256	No Treat.	0 0	ó	0	MMF	270,000	26,000	64,000		
259	No Treat.	0	0	Ō	No Treat	0	0	0		
263	MMF	205,000	17.000	41,000	MMF	205 000	17.000	41 000		
264	No Treat.	0	0	0	No Treat	205,000	0	0		
269	ASL/MMF	2,900,000	315,000	670,000	ASI./MMF	2.900 000	315 000	670 000		
275	ASL/MMF	800,000	84,000	195,000	ASI./MMF	800,000	84,000	195 000		
159	No Treat.	0	0	0	No Treat	0	0,000	0		
281	No Treat.	Õ	Õ	Õ	No Treat	ŏ	ŏ	õ		

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# PLANT-BY-PLANT COST ESTIMATES

SUBCATEGORY: NOT TYPE I

BPT (I)					BPT (II)				
	Suggested	Capital	Operating	Annual	Suggested	Capital	Operating	Annual	
Plant #	Treatment	Cost (\$)	Cost(\$/yr)	Cost(\$/yr)	Treatment	Cost(\$/yr)	Cost(\$/yr)	Cost(\$/yr)	
				•					
6	ASL/MMF	815,000	83,000	194,000	ASL/MMF	815,000	83,000	194,000	
8	ASL/MMF	890,000	90,000	208,000	ASL/MMF	890,000	90,000	208,000	
18	No Treat.	0	0	0	No Treat.	0	0	0	
21	No Treat.	0	0	0	No Treat.	0	0	0	
59	MMF	600,000	50,000	120,000	MMF	600,000	50,000	120,000	
66	MMF	810,000	66,000	159,000	MMF	810,000	66,000	159,000	
86	ASL/MMF	1,660,000	167,000	370,000	ASL/MMF	1,660,000	167,000	370,000	
92	No Treat.	0	0	0	No Treat.	0	0	0	
94	MMF	5,000,000	250,000	850,000	MMF	5,000,000	250,000	850,000	
97	MMF	520,000	42,000	100,000	MMF	520,000	42,000	10,000	
120	No Treat.	0	0	0	MMF	2,500,000	150,000	450,000	
121	No Treat.	0	0	0	No Treat.	0	0	0	
122	MMF	650,000	56,000	130,000	ASL/MMF	1,700,000	166,000	370,000	
144	MMF	540,000	42,000	100,000	ASL/MMF	1,020,000	94,000	220,000	
145	ASL/MMF	1,800,000	180,000	400,000	ASL/MMF	1,800,000	180,000	400,000	
153	ASL/MMF	1,045,000	104,000	245,000	ASL/MMF	1,045,000	104,000	245,000	
182	ASL/MMF	1,250,000	122,000	280,000	ASL/MMF	1,250,000	122,000	280,000	
186	No Treat.	0	0	0	No Treat.	0	0	0	
205	MMF	160,000	15,000	34,000	MMF	160,000	15,000	34,000	
208	MMF	400,000	36,000	85,000	ASL/MMF	830,000	81,000	195,000	
226	MMF	560,000	46,000	110,000	MMF	560,000	46,000	110,000	
231	ASL/MMF	1,270,000	156,000	332,000	ASL/MMF	1,270,000	156,000	332,000	
245	No Treat.	0	0	0	No Treat.	0	0	0	
247	No Treat.	0	0	0	No Treat.	0	0	Ο.	
258	ASL/MMF	550,000	59,000	136,000	ASL/MMF	550,000	59,000	136,000	
270	No Treat.	Ó	Ô	0 I	No Treat.	0	Ó	0	
151	MMF	700,000	64,000	148,000	MMF	700,000	64,000	148,000	

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# TOTAL COSTS, PLANT-BY-PLANT ANALYSIS

		BPT (I)		BPT (II)				
	Capital Cost (\$)	Operating Cost (\$/Yr)	Annual Cost (\$/Yr)	Capital   <u>Cost (\$)</u>	Operating Cost (\$/Yr)	Annual Cost (\$/Yr)		
PLASTICS	33,045,000	3,215,000	7,388,000	41,875,000	4,265,000	9,665,000		
TYPE I W/ OXID.								
-High Flow	31,025,000	3,281,000	7,205,000	43,565,000	4,363,000	9,682,000		
-Low Flow	6,540,000	824,000	1,702,000	10,145,000	1,128,000	2,436,000		
TYPE I W/O OXID.	21,470,000	2,141,000	5,335,000	22,390,000	2,235,000	5,549,000		
NOT TYPE I	19,220,000	1,628,000	4,001,000	   23,680,000 	1,985,000	4,921,000		
TOTAL	111,300,000	11,089,000	25,631,000	141,290,000	13,950,000	32,253,000		

## PERCENTAGE OF PLANTS REQUIRING ADDITIONAL TREATMENT FOR BPT (1)

Sub	category	No A <u>Tr</u>	dditional eatment	ASL/MMF	MMF	ASL	CLAR
Plastics Only		38	29	33	0	0	
Not W/	Plastics - Typ Oxidation	eΙ					
	-High Flow -Low Flow		31 31	0 0	15 0	49 54	5 15
Not W/O	Plastics -Type Oxidation	I	47	30	20	0	0
Not	Plastics NOT T	ype I	33	30	37	0	0

3% require ALA/MMF (Not Plastics - Type I W/O Oxidation)

TABLE	8-21
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PERCENTAGE OF PLANTS REQUIRING ADDITIONAL TREATMENT FOR BPT (II)

Subcategory	No Additional Treatment	ASL/MMF	MMF	ASL	CLAR
Plastics Only	31	36	33	0	0
Not Plastics - Typ W/Oxidation	εI				
-High Flow -Low Flow	28 23	49 54	23 23	0 0	0 0
Not Plastics -Type W/O Oxidation	I 40	30	27	0	0
Not Plastics NOT T	уре I 30	40	30	0	0

3% require ALA/MMF (Not Plastics - Type I W/O Oxidation)

# PERCENTAGE OF ANNUAL COSTS BY SUBCATEGORY

Subcategory	<b>BP</b> T (1)	BPT (II)
PLASTICS		
58 Plants		
(35% of data base)	29	30
NOT PLASTICS - TYPE I W/ OXIDATIC	N	
-High Flow		
39 Plants		
(23% of data base)	28	30
-Low Flow		
13 Plants		
(8% of data base)	7	8
NOT PLASTICS - TYPE I W/O OXIDATI 30 Plants	ON	
(18% of data base)	21	17
NOT PLASTICS - NOT TYPE I		
27 Plants		
(16% of data base)	15	15
167 Total Plants (100%)	100%	100%

#### CAPDET REPRODUCIBILITY

Following completion of the cost estimates used to prepare this report, an attempt was made to verify the reproducibility of the CAPDET program. In conducting this evaluation it was determined that the CAPDET program had been revised since the original cost analyses. Neither the EPA nor the Corps of Engineers has available documentation of the earlier version of CAPDET from which the costs were generated.

To determine the extent of the revisions, a set of activated sludge system costs were run using input values identical to those used in an earlier run for which printouts were available. The technologies designed and costed included primary clarification, complete mix activated sludge (with secondary clarifier), gravity thickening, sludge drying and contract hauling.

Both series of runs using the old and revised CAPDET showed identical designs and costs for primary and secondary clarifier, gravity thickener, drying beds and similar costs for hauling. However, the designs and costs for the complete mix activated sludge system were different. Comparing a 1000 mg/l BOD influent at 0.2 MGD and a 20 mg/l (LTM) effluent target, the sizing of the reactor by the revised version was larger, but cost less. A 4.86 MG vessel with costs of \$1.07 million for the revised CAPDET versus 3.69 MG vessel costing \$1.31 million for the older version.

#### FURTHER CHANGES TO CAPDET

Following this analysis some further revisions were made to CAPDET. The major difference in the design of OCPS industry wastewater treatment systems, compared with domestic or sanitary wastewater treatment systems, is the slower rate at which biological oxidation reactions proceed. Typically, the kinetic rate for the BOD degradation of OCPS wastewater is one-half to one-fifth the rate for domestic wastewater. In addition, the influent BOD concentration for a domestic wastewater treatment system usually falls within a relatively narrow range, typically from 200 to 350 mg/l. Influent BOD concentrations for OCPS industry treatment systems can range from less than one hundred to several thousand mg/l. On this basis, the kinetic model used by CAPDET was modified as shown in Table 8-1. As an alternative to the use of this model, the necessary programming changes were made to incorporate a kinetic model specifically intended for industrial wastewater.

Research, provided by Gloyna, Grau, and any others in the industrial waste field has found the Grau kinetic model more closely simulates actual biochemical removal or organics.[8-8] Union Carbide has completed extensive study on the use of Grau kinetics to model its chemical plants. The resulting conclusion is that BOD removal has been determined to be first order as a function of soluble BOD and volatile solids but inversely proportionate to the influent soluble BOD concentration. The work is described in an article by Cyron T. Lawson, <u>et al</u>. in "Comments on Selected Aspects of Activated Sludge Treatment Technology Based on Recent Union Carbide Experience". The use of soluble, rather than total, influent BOD concentrations represents a constraint when applying the Grau model to develop industrywide treatment costs. Very few plants in the OCPS data base reported soluble influent BOD characteristics. To apply the Grau model to plants which reported only total influent BOD data, some relationship between total and soluble BOD would have to be assumed. Errors in this assumption may offset the benefits of using a more accurate kinetic modeling technique.

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#### NON WATER QUALITY ASPECTS

The use of wastewater treatment systems to alleviate water pollution problems may result in adverse impacts in other environmental areas. Elements of other environmental concerns that must be considered include:

- 1. Air pollution
- 2. Solid waste generation
- 3. RCRA considerations
- 4. Noise pollution
- 5. Energy requirements

## Air Pollution

If solvents or other volatile hydrocarbons are subjected to an evaporative process (such as an evaporation pond) where vapor condensation is impractical, volatile materials will be lost to the atmosphere. Volatile materials can be lost to a lesser extent through aeration, dissolved air flotation and other treatment operations. The extent of the pollution potential depends on the nature and concentration of the volatile components and on the weather conditions at the site of disposal or treatment, as well as the treatment technology itself.

Landfilling is a fairly common method of disposing of settled solids, floating oils and biological sludges. If the landfill is not properly designed and maintained, volatile components in the nonwater wastes can evaporate and contribute to air pollution.

Incineration is another commonly employed method for the disposal of nonaqueous wastes. Improperly designed or operated incinerators can discharge particulates, hydrocarbons or noxious gases to the atmosphere. Most of these emissions can be controlled by the use of scrubbers. These scrubbers will, however, create an additional source of contaminated wastewater.

#### Solid Waste Generation

Solid wastes, for the purposes of this report, include the solids and skimmings produced by the treatment technologies involved in the primary, secondary and tertiary end-of-pipe treatment of the OCPS industry's wastewater.

The solids include such materials as grit and solids from primary clarifiers, sludges and skimmings generated by the various separation technologies, biological sludge from biological treatment plants, spent activated carbon and residues from incineration. Most of the above solids generated cannot be quantified without a specific on-site evaluation; either by the individual plant records, if available, or by a study to determine the quantity of solid waste generated.

The major quantity of solid waste generated, however, is the excess biological sludge. This material can be approximtely quantified because, in biological treatment, the excess sludge is directly related to the BOD removed. CAPDET used 0.73 pounds of sludge generated per pound of BOD removed. Figure 8-14 shows the sludge production as a function of the change in BOD concentration.

## RCRA Considerations

Processing operations and treatment facilities in OCPS plants generate solid and liquid wastes which in some cases may be classified as "hazardous wastes" by the definition and characteristics outlined in Section 3001 of the Resource Conservation and Recovery Act (RCRA). Storage, transport, treatment and disposal methods for these wastes are regulated by RCRA interim status standards.

OCPS hazardous waste generators using contract removal, offsite disposal and sales must comply with the transportation guidelines for hazardous waste. The guidelines include standards for manifest, labels, containers, marking and placarding of wastes before removal. The receiver of the wastes would then be responsible for meeting treatment, storage and disposal requirements.

Onsite treatment of hazardous waste by OCPS generators does not have to comply with transportation guidelines for hazardous wastes. They must, however, meet standards for treatment, storage and disposal of these wastes (RCRA Section 3004) and must obtain a permit (RCRA Section 2005). OCPS generators treating onsite include EOP systems and zero discharge performers using deep wells, incineration followed by scrubbing, evaporation ponds and land disposal.

#### Noise Generation

None of the alternate technologies presented in this report are judged to be generators of excessive noise levels. Most industrial installations, including the OCPS plants, do generate a background noise level which, if excessive, must be accommodated under OSHA regulations. Furthermore, practically all machinery of recent manufacture is constructed or installed to comply with OSHA noise level requirements.

None of the technologies proposed, nor the equipment associated with the technologies, are unique or radically different from other industrial machinery in respect to noise generation.

On this basis it is judged that noise pollution does not pose a potential problem with the implementation of the suggested technologies.



FIGURE 8-14 - SLUDGE PRODUCTION FOR COMPLETE MIX ACTIVATED SLUDGE

#### Energy Requirements

Due to the importance of today's need for energy conservation and the inceasing costs arising from energy shortages, energy usage must be considered before implementing treatment technologies.

In-plant technologies may require high energy consumptions. Activated carbon adsorption requires a large energy utilization (3,000 BTU/lb of carbon) to regenerate spent carbon.[8-9] Membrane technologies require energy for use in producing pressure to force liquid wastes through the film. Steam stripping utilizes energy to produce and move the steam. Wet oxidation requires large energy amounts for generating high pressures and for fueling the operation to promote extreme temperatures unless the organic content of the wastewater is high enough to sustain auto-oxidation.

Energy requirements for EOP treatments using aerators range from 0.6 to 1.15 hp/1000 ft<sup>-1</sup>.[8-10] Energy requirements for complete<sub>3</sub>mix activated sludge calculated for an average value of 0.88 hp/1000 ft<sup>-1</sup> are shown in Figure 8-15 as a function of flow and residence time. These costs also include pumping requirements in addition to aeration and mixing. The residence time can be obtained from Figure 8-1 discussed previously. Other energy requirements are needed for clarifier operation, some types of filtration requiring pressure, dissolved air flotation and activated carbon regeneration. Ultimate treatment of sludge, residues, scums and liquid wastes by incineration would require additional energy. Figures 8-16 through 8-18 list the energy requirements utilized by CAPDET for clarification, dissolved air flotatation and multimedia filtration.

Energy requirements for zero discharge treatment, except incineration, are mainly for pumping liquids or for operating and transporting vehicles. Recycling may require additional energy to return the recycled wastes to the process. Incineration would rely on energy to heat and oxidize wastes. Energy used to maintain evaporation process equipment must also be considered.



FIGURE 8-15 - ENERGY REQUIREMENTS FOR COMPLETE MIX ACTIVATED SLUDGE



FIGURE 8-16 - ENERGY REQUIREMENTS FOR CLARIFICATION

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FIGURE 8-17 - ENERGY REQUIREMENTS FOR DISSOLVED AIR FLOTATION



FIGURE 8-18 - ENERGY REQUIREMENTS FOR MULTIMEDIA FILTRATION

## SECTION IX.

## EFFLUENT REDUCTION ATTAINABLE THROUGH THE APPLICATION OF BEST PRACTICABLE CONTROL TECHNOLOGY CURRENTLY AVAILABLE

#### GENERAL

The effluent limitations which were required to be achieved by July 1, 1977, are based on the degree of effluent reduction attainable through the application of the best practicable control technology currently available (BPT). The best practicable control technology currently available generally is based upon the average of the best existing performance, in terms of treated effluent discharged, by plants of various sizes, ages, and unit processes within an industry or subcategory. Where existing performance is uniformly inadequate, BPT may be transferred from a different subcategory or category. Limitations based on transfer technology must be supported by a conclusion that the technology is, indeed, transferable and a reasonable prediction that it will be capable of achieving the prescribed effluent limits (see Tanners' Council of America v. Train, 540 F. 2d 1188 (4th Cir. 1976)). While best practicable control technology currently available focuses on endof-pipe treatment technology rather than process changes or internal controls, it can include process changes or internal controls when the changes or controls are normal practice with an industry.

BPT considers the total cost of the application of technology in relation to the effluent reduction benefits to be achieved from the technol-The cost/benefit inquiry for BPT is a limited balancing, which ogies. does not require the Agency to quantify benefits in monetary terms (see, e.g., American Iron and Steel v. EPA, 526 F 2d 1027 (3rd Cir. 1975)). In balancing costs in relation to effluent reduction benefits, EPA considers the volume and nature of existing discharges, the volume and nature of discharges expected after application of BPT, the general environmental effects of the pollutants, and the costs and economic impacts of the required pollution control level. The Act does not require or permit consideration of water quality problems attributable to particular point sources or industries, or water quality improvements in particular water bodies (see Weyerhaeuser Company v. Costle, 11 ERC 2149 (D.C. Cir. 1978)).

#### REGULATED POLLUTANTS

Pollutants proposed for regulation under BPT are BOD5, TSS and pH.

#### IDENTIFICATION OF THE BEST PRACTICABLE CONTROL TECHNOLOGY

In determining the best practicable control technology, biological treatment has been evaluated as the principal treatment practice within the OCPS industry. Of the 185 plants for which treatment system information is available, 146 use some form of biological treatment. Although nonbiological treatment systems are often used to produce high quality effluents, only biological treatment has been sufficiently applied to be considered across the broad spectrum of the OCPS industry. On this basis, biological treatment, in general, may be considered the

best technology for treatment of OCPS wastewaters and is, therefore, chosen as best practicable control technology.

#### RATIONALE FOR SELECTION OF THE BEST PRACTICABLE CONTROL TECHNOLOGY

Use of biological treatment as the best practicable control technology will produce high quality effluents as shown in Section VII. Evaluations to determine if the addition of tertiary treatment processes such as polishing ponds, filtration, carbon adsorption, etc., would improve effluent quality indicated that plants with tertiary processes do not achieve lower effluent  $\mathtt{BOD}_{\varsigma}$  concentrations than plants utilizing only biological systems. This apparent contradiction may have resulted from the fact that, because the data base is limited to performance data for entire treatment systems, performance data are not available for individual unit treatment processes. This tends to mask the effect of plants which install additional end-of-pipe treatment technologies to compensate for poorly designed or poorly operated existing treatment systems. Therefore, it was determined that the evaluation of levels of performance by the treatment technology utilized by OCPS plants was not appropriate.

Of the 143 plants in the Summary Data Base which employ biological treatment and have effluent BOD<sub>5</sub> data, only 65 plants, or 46%, comply with applicable BOD<sub>5</sub> effluent limitations, while the remaining 78 plants, or 55%, were not in compliance. In addition, of the 129 plants in the Summary Data Base which employ biological treatment and have effluent TSS data, only 60 plants, or 47%, comply with applicable TSS effluent limitations, while the remaining 69 plants, or 53%, were not in compliance. This is an indication that, while biological treatment has been demonstrated to achieve low effluent concentrations, the median values obtained for all biological systems in the industry are not considered to be representative of the average level of treatment which can be achieved. While some plants in the data base are well designed and operated to the maximum potential, others are operating at less than optimum performance. In order to segregate good performers from bad performers, it was necessary to develop a test to distinguish the better plants from those operating less efficiently.

The first test used to define the well operated plants was to consider only those plants which achieved BOD, removals equal to or greater than 95%. As presented in Section VII, plants which meet this test criteria achieve lower median values than do all plants with biological systems. However, it was also recognized that use of the 95% removal criteria eliminates some plants which achieve lower effluent BOD, concentrations, but by virtue of having very low influent BOD, concentrations, do not achieve 95% removal.

In an attempt to address this potential inconsistency, a new segment was evaluated which included all plants which achieved 95% BOD<sub>5</sub> reduction or which achieved a final effluent BOD<sub>5</sub> concentration of less than or equal to 50 mg/l. This also allowed the inclusion of effluent BOD<sub>5</sub> data which had no corresponding influent value.

In addition, while reviewing the data for this segment, it appeared that well operated plants in the Not Plastics Only-Type I with Oxidation subcategory had a wide range of effluent values. Subsequent investigations showed that this effect appeared to be related to the efficiency of water use by plants in this subcategory. Water use efficiency was defined as a plant's daily water usage divided by its daily production level. As described in detail in Section VII, a further analysis showed this effect to be limited to those plants in the first quartrile of water use. This included plants with water use less than or equal to 0.2 gallons per pound of product. This suggested that the Not Plastics Only -Type I with Oxidation subcategory be further divided into low water use (less than 0.2 gallons per pound) and high water use (greater than 0.2 gallons per pound) subcategories. When this was done, the resulting long term median final effluent BOD5 and TSS concentrations for plants with at least 95% BOD<sub>c</sub> removal or 50 mg/l final effluent BOD<sub>5</sub> concentrations were 26 mg/l and 62 mg/l for high water use facilities and 36 mg/l and 89 mg/l for low water use facilities.

After the calculation of long term median effluent concentration for each proposed subcategory, EPA grouped for analysis all plants which performed better than the subcategory long term median and all plants which performed worse. This analysis was performed to determine if certain plant characteristics would cause plants to perform better or worse than the subcategory long term median. The results showed that both groups have similar mixes and numbers of generic processes, similar ranges in the number of specific product processes, similar raw waste concentration distributions, similar contributions from secondary production of non-OCPS products and geographical mix (as a measure of temperature effects). Therefore, it can be concluded that different types of plants were not improperly combined within a subcategory.

The implementation of additional end-of-pipe treatment technologies can require additional land. In addition, spatial relationships and the physical characteristics of available land can affect construction costs. However, as detailed in Section VIII, the cost of land acquisition has been included in cost estimates, where appropriate. In addition, the impact of plant-by-plant variations has been lessened by costing least land intensive options available (i.e., activated sludge versus aerated lagoons).

#### BPT EFFLUENT LIMITATIONS

BPT effluent limitations are presented in Table 9-1.

#### METHODOLOGY USED FOR DEVELOPMENT OF BPT EFFLUENT LIMITATIONS

Biological treatment has been identified as the best practicable control technology currently available for each of the four proposed subcategories. The long term median BPT final effluent BOD<sub>5</sub> and TSS concentrations were calculated for each subcategory by using the performance of plants which attain 95% BOD<sub>5</sub> reduction or a final effluent BOD<sub>5</sub> concentration less than or equal to 50 mg/l. In addition, after a review of data and information submitted by the industry, it was determined that four plants which met the above BOD criteria had installed more advanced
# TABLE 9-1

# BPT EFFLUENT LIMITATIONS (mg/l or ppm)

	LONG TER	MEDIAN	MAXIMUM	30-DAY	MAXIMUM	DAILY
SUBCATEGORY	BOD <sub>5</sub>	TSS	BOD5	TSS	BOD <sub>5</sub>	TSS
Plastics Only	14.5	24	22	36	49	117
Oxidation						
o High Water Use o Low Water Use	e 26 36	62 89	42 58	84 120	106 146	246 353
Туре І	24.5	34.5	40	47	100	137
Other Discharges	s 17	29	28	39	69	115

pH - All subcategories - Within a range of 6.0 to 9.0 at all times

wastewater treatment technology in order to meet permit conditions based on water quality standards. It was determined that the performance of these four plants (three plants in the Plastics Only subcategory and one plant in the Other Discharges subcategory) was not representative of this technology-based regulation and therefore, were removed from the calculation of effluent limitations. Remaining plants were determined to achieve the best existing performance.

Maximum 30-day and daily maximum effluent limitations were determined by multiplying long-term median effluent concentrations by appropriate variability factors which were calculated through statistical analysis of long term BOD<sub>5</sub> and TSS daily data. This statistical analysis is described in detail in Section VII.

#### COST OF APPLICATION AND EFFLUENT REDUCTION BENEFITS

#### Summary Data Base

The total cost (1979 dollars) of attainment of BPT effluent limitations for the 130 direct discharging plants in the Summary Data Base which do not meet these limitations has been estimated to be about 111.30 million dollars in capital cost with associated total annual cost of about 25.63 million dollars per year.

Conventional pollutant removals from current discharge loadings associated with Summary Data Base plants have been estimated to be about 9.79 million kg/yr (21.58 million lbs/yr) of BOD<sub>5</sub> and 14.59 million kg/yr (32.17 million lbs/yr) of TSS. These removals represent a cost per pound of BOD and TSS removed of about \$0.48/lb. of BOD and TSS removed. Percent removals of BOD and TSS from current raw waste loads are estimated to be about 94.8 percent and 74.6 percent, respectively. Table 9-2 presents the annual pounds removed figures for each proposed subcategory.

#### Total Industry

The total cost of attainment of BPT effluent limitations for all direct discharging plants in the OCPS industry has been projected to be about 316 million dollars in capital costs with associated total annual costs of about 105 million dollars per year. These projections were made for use in the Economic Impact Analysis (EIA) and are documented in the EIA document.

As discussed in previous sections of this document, many different estimates have been made on the number of plants in the OCPS industry. Estimates range from about 1200 plants to as many as 2100 plants. Due to the many estimates of the number of direct discharging plants in the OCPS industry, a direct projection of the Summary Data Base to the entire industry was not possible and a general analysis was utilized to project the conventional pollutant removals to the entire OCPS industry. Weighted averages of current BOD<sub>5</sub> and TSS concentrations and current flows for each subcategory were utilized to calculate current total industry BOD and TSS loadings. Using the same weighted average for current flows, total industry BOD and TSS loadings were calculated using

# TABLE 9-2

# EFFLUENT REDUCTION BENEFITS AND NON WATER QUALITY IMPACTS FOR SUMMARY DATA BASE PLANTS IN EACH OF THE PROPOSED SUBCATEGORIES

	Million Remo	n lbs/yr oved	Energy Requirements	Additional Solid Waste		
	BOD	TSS	(Barrels/yr)	(Tons/yr)		
PLASTICS ONLY	1.75	6.01	29,455	3,311		
OXIDATION				•		
o High Water Use	6.50	4.34	84,137	3,307		
o Low Water Use	2.72	0.65	6,771	800		
TYPE I	9.68	11.09	24,188	7,239		
OTHER DISCHARGERS	0.93	10.08	12,444	5,203		
TOTALS	21.58	32.17	156,995	19,860		

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the proposed effluent limitations. The difference in these two total industry loadings was used as the total BOD and TSS removals from current discharge levels. A summary of this case study is shown in Table 9-3. As can be seen in Table 9-3, conventional pollutant removals from current discharge levels based on attainment of BPT effluent limitations have been projected to be about 67.59 million kg/yr (149 million pounds) of BOD and 46.27 million kg/yr (102 million pounds) of TSS. These removals represent a cost per pound of BOD and TSS removed of about 0.42/1b. of BOD and TSS removed. Percent removals of BOD and TSS from current raw waste loads are similar to those for the Summary Data Base.

#### NON-WATER QUALITY ENVIRONMENTAL IMPACTS

Non-water quality environmental impacts have also been considered in Section VIII. The impacts associated with attainment of BPT effluent limitations by Summary Data Base plants are discussed below. Non-water quality impacts based on total industry attainment of the BPT effluent lmitations are not discussed below due to difficulties associated with projecting Summary Data Base impacts. However, general conclusions on impacts can be drawn from the Summary Data Base statistics.

#### ENERGY

Attainment of BPT will require the use of the equivalent of approximately 32.69 million liters (157 thousand barrels) of residual fuel oil per year for Summary Data Base plants. Table 9-2 presents a breakdown of the energy requirement by subcategory.

#### SOLID WASTE

Attainment of BPT will result in an additional 18.02 thousand kkg/yr (19,860 tons/yr) of wastewater treatment solids for plants in the Summary Data Base. Table 9-2 presents the amount of additional solids generated by subcategory.

#### AIR AND NOISE

Attainment of BPT will have no measurable impact on air or noise pollution.

# TABLE 9-3

# CASE STUDY FOR ESTIMATING TOTAL INDUSTRY DIRECT DISCHARGE LOADINGS FOR BOD<sub>5</sub> AND TSS

	mg/1	million lbs/year <sup>3</sup>
BOD <sub>5</sub>		
Mean raw waste <sup>l</sup>	945	3,153.9
Current effluent	63	208.3_
BPT effluent <sup>2</sup>	18	59.5_
TSS		
Mean raw waste <sup>1</sup>	427	1,401.7
Current effluent <sup>1</sup>	63	208.3_
BPT effluent <sup>2</sup>	32	105.8

l mean of all plants in Summary Data Base

- $^2$  median value of all biological systems with BOD\_5 removal  $\geq$  95% or effluent BOD\_5  $\leq$  50 mg/1
- <sup>3</sup> based on 330 operating days per year, 2.31 MGD per plant, and 520 direct discharge plants

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#### SECTION X

## EFFLUENT REDUCTION ATTAINABLE THROUGH THE APPLICATION OF BEST CONVENTIONAL POLLUTANT CONTROL TECHNOLOGY EFFLUENT LIMITATIONS GUIDELINES

#### GENERAL

The 1977 amendments added Section 301(b)(2)(E) to the Act, establishing "best conventional pollutant control technology" (BCT) for discharges of conventional pollutants from existing industrial point sources. Section 304(a)(4) designated the following as conventional pollutants: BOD, TSS, fecal coliform and pH. The Administrator designated oil and grease as "conventional" on July 30, 1979, 44 FR 44501.

BCT is not an additional limitation, but replaces BAT for the control of conventional pollutants. In addition to other factors specified in Section 304(b)(4)(b), the Act requires that BCT limitations be assessed in light of a two-part "cost-reasonableness" test. EPA published a methodology for determining BCT on August 29, 1979 (44 FR 50732). In <u>American Paper</u> <u>Institute</u> v. EPA, 660 F. 2nd 954 (4th Cir. 1981), EPA was ordered to revise the cost test.

The court held that EPA must apply a two-part test. The first test compares the cost for private industry to reduce its conventional pollutants with the costs for Publicly Owned Treatment Works to attain similar reductions in their discharge of these pollutants. The second test examines the cost-effectiveness of additional industrial treatment beyond BPT. EPA must find that limitations are "reasonable" under both tests before establishing them as BCT. In no case may BCT be less stringent than BPT.

In response to the court order, EPA has proposed a revised BCT costreasonableness test at 47 FR 49176 (October 29, 1982). The proposed test provides that BCT is cost-reasonable if: (1) the incremental cost per pound of conventional pollutant removed in going from BPT to BCT is less than \$.27 per pound in 1976 dollars, and (2) this same incremental cost per pound is less than 143% of the incremental cost per pound associated with achieving BPT.

#### REGULATED POLLUTANTS

Pollutants proposed for regulation under BCT are BOD, TSS and pH.

#### IDENTIFICATION OF THE BEST CONVENTIONAL POLLUTANT CONTROL TECHNOLOGY

The Agency has considered an incremental technology level of conventional pollutant control beyond BPT. The technology, additional solids control such as polishing ponds and filtration, is already practiced by approximately one third of the plants in the industry.

#### BCT EFFLUENT LIMITATIONS

BCT effluent limitations are presented in Table 10-1.

# TABLE 10-1

# BCT EFFLUENT LIMITATIONS (mg/l or ppm)

	LONG TERM	MEDIAN	MAXIMUM	30-DAY	MAXIMUM DAILY		
SUBCATEGORY	BOD <sub>5</sub>	TSS	BOD5	TSS	BOD <sub>5</sub>	TSS	
Plastics Only	14.5	24	22	36	49	117	
Oxidation							
o High Water Use o Low Water Use	26 36	62 89	42 58	84 120	106 146	246 353	
Type I	24.5	34.5	40	47	100	137	
Other Discharges	17	29	28	39	69	115	

# RATIONALE FOR THE SELECTION OF BEST CONVENTIONAL POLLUTANT CONTROL TECHNOLOGY

The Agency performed the first part of the BCT "cost-reasonableness" test which compares the cost for private industry to reduce conventional pollutants with the costs for Publicly Owned Treatment Works to attain similar reductions in their discharge of conventional pollutants. To perform this first test, EPA calculated the incremental conventional pollutant removals beyond BPT and the incremental costs associated with the installation of the BCT technology. Based on this information, cost per pound of BOD and TSS removed ratios were calculated for each of the four BPT subcategories. The results of this analysis are presented in Table 10-2.

All the incremental costs per pound ratios were found to fail this first part of the BCT "cost-reasonableness" test (\$0.33 per pound in 1979 dollars). Therefore, EPA did not perform the second part of the BCT "cost reasonableness" test, and is proposing BCT effluent limitations which are equal to the BPT effluent limitations for each of the proposed BPT subcategories.

#### METHODOLOGY USED FOR DEVELOPMENT OF BCT EFFLUENT LIMITATIONS

The methodology used for development of BCT effluent limitations is the same as that used in the development of BPT effluent limitations since BCT effluent limitations are equal to the BPT effluent limitations. For a detailed description of this methodology, refer to Section IX of this document.

## COST OF APPLICATION AND EFFLUENT REDUCTION BENEFITS

There are no incremental costs or effluent reduction benefits associated with the attainment of BCT effluent lmitations since BCT effluent limitations are equal to the BPT effluent limitations. For detailed information on the costs and effluent reduction benefits associated with the BPT effluent limitations, refer to Section IX of this document.

#### NON-WATER QUALITY ENVIRONMENTAL IMPACTS

Incremental non-water quality environmental impacts are associated with the attainment of BCT effluent limitations since BCT effluent limitations are equal to the BPT effluent limitations. For detailed information on the non-water quality impacts associated with the BPT effluent limitations, refer to Section IX.

# TABLE 10-2

Subcategory	Incremental BOD and TSS Removed (pounds/year)	Incremental Cost (1979 \$/yr.)	Cost Per Pound of BOD <sub>5</sub> and TSS Removed 5(1979 \$/yr.)
PLASTICS ONLY	161,600	2,277,000	\$14.09/15.
OXIDATION			
o High Water Use o Low Water Use	2,199,000 414,000	2,477,000 734,000	\$ 1.13/1b. \$ 1.77/1b.
TYPE I	461,000	214,000	\$ 0.46/1b.
OTHER DISCHARGES	604,286	920,000	\$ L.52/1b.

# BCT "COST-REASONABLENESS" TEST RESULTS

## SECTION XI

#### NEW SOURCE PERFORMANCE STANDARDS

#### GENERAL

The basis for new source performance standards (NSPS) under section 306 of the Act is the best available demonstrated technology. At new plants, the opportunity exists to design the best and most efficient production processes and wastewater treatment facilities. Therefore, Congress directed EPA to consider the best demonstrated process change, in-plant controls and end-of-pipe treatment technologies that reduce pollution to the maximum extent feasible. It is encouraged that at new sources, reductions in the use of and/or discharge of wastewater be attained by application of inplant control measures.

#### **REGULATED POLLUTANTS**

#### Conventional Pollutants

Conventional pollutants proposed for regulation under NSPS are the same as for BPT:  $BOD_5$ , TSS and pH.

#### IDENTIFICATION OF THE TECHNOLOGY BASIS OF NSPS

The technologies employed to control conventional pollutants at existing plants are fully applicable to new plants. In addition, no other technologies could be identified for new sources which were different from those used to establish BPT effluent limitations. Thus, the technology basis for NSPS is the same as that for BPT effluent limitations. For detailed information on the technology basis for BPT effluent limitations, refer to Section IX of this document.

#### NEW SOURCE PERFORMANCE STANDARDS

New source performance standards for conventional pollutants are presented in Table 11-1.

#### RATIONALE FOR THE SELECTION OF THE TECHNOLOGY BASIS FOR NSPS

Since the Agency could identify no additional generally applicable technology for NSPS and since the technology basis for NSPS is the same as that identified for BPT effluent limitations, EPA has established NSPS effluent limitations equal to the proposed BPT and BCT effluent limitations.

#### METHODOLOGY USED FOR DEVELOPMENT OF NSPS EFFLUENT LIMITATIONS

The methodology used for the development of NSPS effluent limitations is the same as that used for the development of BPT effluent limitations. For detailed information on the methodology used to develop the BPT effluent limitations, refer to Section IX of this document.

# TABLE 11-1

# NSPS EFFLUENT LIMITATIONS (mg/l or ppm)

1	LONG TER	M MEDIAN	MAXIMUM	30-DAY	MAXIMUM	DAILY
SUBCATEGORY	BOD <sub>5</sub>	TSS	BOD5	TSS	BOD5	TSS
Plastics Only	14.5	24	22	36	49	117
Oxidation						
o High Water Use	26	62	42	84	106	246
o Low Water Use	36	89	58	120	146	353
Type I	24.5	34.5	40	47	100	137
Other Discharges	17	29	28	39	69	115

pH - All Subcategories - Within a range of 6.0 to 9.0 at all times

## COST OF APPLICATION AND EFFLUENT REDUCTION BENEFITS

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There are no incremental costs or effluent reduction benefits associated with the attainment of NSPS since NSPS effluent limitations are equal to the BPT effluent limitations. For detailed information on the costs and effluent reduction benefits associated with the attainment of BPT effluent limitations, refer to Section IX of this document.

#### NON-WATER QUALITY ENVIRONMENTAL IMPACTS

No incremental non-water quality environmental impacts are associated with attainment of NSPS since NSPS effluent lmitations are equal to the BPT effluent limitations. For detailed information on the non-water quality impacts associated with the attainment of BPT effluent lmitations, refer to Section IX of this document.

## SECTION XII.

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# APPENDIX A

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# GENERIC PROCESS CODES WITH FREQUENCY OF OCCURRENCE

A-i

PP_CODE	GEN_CODE	DIR	7640	UNK	PP_TEXT
0005=01	01	6	•	•	ARS RESIN/EMULSION POLYMERIZATION
0005=02	D1	1	•	•	ARS RESINIMASS POLYMERIZATION
0005-03	Dj	2	•	•	ABS RESIN/SUSPENSION POLYMERIZATION
0005-90	3	3	•	•	ARS RESIN/FINISHING PROCESS
0010-90	1	•	•	•	ABS/SAN/FINISHING PROCESS
0012=01	2	1	•	•	ACENAPHTHENE/BY-PRODUCT OF PROPANE PYROLYSIS
20 <b>0</b> 0200	С	1	2	•	ACETALDEHYDE/OXIDATION OF ETHYLENE WITH CUCLE CATALYST
0030-06	C	1	•	•	ACETALDEHYDE/RY-PRODUCT OF ACROLEIN BY PROPYLENE OXID
0030-07	J1	1	•	•	ACETALDEHYDE/CATALYTIC DEHYDROGENATION OF ETHANOL
0070-00	4	٠	•	٠	ACETIC ACTD/
0070=02	C	•	2	•	ACETIC ACID/CATALYTIC OXIDATION OF BUTANE
0070-04	C	1	2	•	ACETIC ACID/OXIDATION OF ACETALDEHYDE
0070-05	0	٠	2	•	ACETIC ACIO/CARBONYLATION OF METHANOL WITH CO AND H2
0070-07	Z	•	•	•	ACETIC ACID/BY-PRODUCT OF P-AMINOPHENOL BY ACID CLV
0070=08	13	•	•	•	ACETIC ACID/BY-PRODUCT OF DIATRIZOIC ACID
0070-09	G	•	1	٠	ACETIC ACID/TRANSESTERIFICATION-METHYLACETATESPORMICACID
0070-12	Dì	•	•	•	ACETIC ACID/BY-PRODUCT POLYVINYL FORMAL
0070#13	Ε	1	•	•	ACETIC ACID/BY-PRODUCT OF POLYVINYL ALCOMOL (HYDROLYSIS OF POLYVINLE ACETATE)
0070=14	3	•	•	•	ACETIC ACID/RECOVERY FROM POLYOL PROCESS
0070=15	3			•	ACETIC ACID/RECOVERY FROM SULFITE PULP WASTEWATER
0070-16	C	1	•		ACETIC ACID/COPRODUCT OF TPA BY OXIOAT OF ACETALDEHYDE
0075+00	24	•	•	•	ACETIC ACID SALTS/ ACETIC ACID + METAL OXIDE OR HYDROXIDE
0075-99	24	1	1	•	ACETIC ACID SALTS (TOTAL)/ ACETIC ACID + METAL OXIDE (HYDROXIDE)
0080-01	Q	2	1	•	ACETIC ANHYDRIDE/THERMAL CRACKING OF ACETIC ACID
0080 <b>-02</b>	0	2	1	•	ACETIC ANHYDRIDE/FROM ACETIC BY ACID RETENE PROCESS
0090-01	C	6	•	•	ACETONE/CUMENE PEROXIDATION AND ACID CLEAVAGE
0090=03	J1	4			ACETONE/DEHYDROGENATION OF ISOPROPANOL
0090=08	С	•	1	•	ACETONE/VAPOR-PHASE OXIDATION OF RUTANE/PROPANE
0090=11	С	•	•	•	ACETONE/BYPRODUCT OF H2O2 BY OXIDATION OF ISOPROPANOL
0100-01	Z	1	•	•	ACETONE CYANOHYDRIN/RXN OF ACETONE WITH HYDROCYANIC ACID
0110=01	ĸ	1	•	•	ACETONITRILE/ NH3 + ACETIC ACID.DEHYDRATION OF ACETAMIDE
0110-02	N	•	1	•	ACETONITRILE/AV-PRODUCT OF ACRYLONITRILE BY AMMOXIDATION OF PROPYLENE
0120+01	С	. 1	•	•	ACETOPHENONE/AVOPRODUCT PHENOL BY CUMENE PEROXIDATION AND ACTO CLEAVAGE
0130=01	JŻ	3		•	ACETYLENE/PARTIAL OXIDATION OF METHANE
0130+02	Ε	•	•	•	ACETYLENE/FROM CALCIUM CAPBIDE
0130-03	H	3	1	•	ACETYLENE/BY-PRODUCT OF BY PROPANE PYROLYSIS
0140-01	С	2		•	ACROLEIN/OXIDATION OF PROPYLENE
0150+01	E	1	•		ACRYLAMIDE/CATALYTIC HYDRATION OF ACRYLONITRILF
0153-01	01	٠	3	•	ACRYLIC LATEX/EMULSION POLYMERIZATION
0155-01	01	2	2	•	ACRYLIC RESINS/EMULSION POLYMERIZATION
0155+02	01	2	1		ACPYLIC RESINS/SUSPENSION POLYMERIZATION
0155=03	01	5	3	•	ACRYLIC RESINS/SOLUTION POLYMERIZATION
0155=06	nj	1	2	•	ACPYLIC RESINS/RULK POLYMERIZATION
*Noto.	Frequency	ounte -		بنا مم	
note.	riequency c	ouncs r	eriect	only th	ie Summary Data Base which includes 195 Direct Dischargers.
•	94 Zero Dis	charger	rs and 2	unknov	vns. Product/Processes without frequency counts are from
	indirect di	scharge	ers.		
		- ···· 3-			

FREQUENCY	OF	OCCURRENCE	FOR	FACH	PRODUCT	PPACESS
FREWUERCT.	vr	VUUUNNERGE	r vn			- FRUCED3

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
0155-10	D1	. 1	•	•	ACRYLIC RESINS/BULK POLYMERIZATION TO CAST SHEET
0155-11	01	1		•	ACRYLIC RESINS/EMULBION OR SOLUTION POLYM. TO COATINGS
0155=14	D1	1	•	•	ACRYLIC RESINS/POLYACRYLAMIDE BY SOLUTION POLYMERIZATION
0155-99	D1	1	•	•	ACRYLIC RESINS/PROCESS IN REVIEW
0156-01	D1	2	•	•	ACRYLIC FINER(85% POLYACRYLONITPILE)/SUSP POLYANET SPINN
0156-02	Dl	2		•	ACRYLIC FIHER (85% POLYACRYLONITRILE)SUSP POLY-DRY SPINN
0160+03	0	1			ACPYLIC ACID/FROM ACETYLENE CARBON MONOXIDE AND WATER
0160-04	С	1		•	ACRYLIC ACID/OXIDATION OF ACROLEIN
0160-80	С	Ž	1	•	ACRYLIC ACID/OXIDATION OF PROPYLENE VIA ACROLEIN
0165+01	G ·	3	2		ACRYLIC ACID ESTERS/ACRYLIC ACID ESTERTE OF HISC ALCOHOL
0165+04	Ğ	ī	-		ACTYLIC ACID ESTERS/HYDROXY ALKYL ACRYLATE BY ACRYLIC AL
0165=05	Ğ	ī	-		ACRYLIC ACID ESTERS/ETHL 2-CYANG ACRIATE FROM FORMALD.FT
0165406	Ğ	i			ACRYLIC ACID ESTERS/METHI 2=CYANO ACRIATE ERON EORMALD-E
0165-07	ñ	i	•	•	ACRYLIC ACTO ESTERSIALLY PECYANO ACRIATE ERON FORMALDE
0165=09	Ğ	i	•	•	ACRYLIC ACID ESTERS/N-BUTYL ACRYLATE-ACRY ACID-N-BUTANOL
0145-10	9		•	•	ACRY TO ACTO ESTERS/FILME ACRY ATE-ACRY ACTO A ETHANOL
0165-11	ő	<u>د</u>	•	•	
0148-12	6	;	•	•	ACTIVE ACTO ESTERSICIAL ACTIVE ACTIVE ACTIVE ACTIVE ACTIVE
0148-13	ŏ	1	•	•	ACTURE ACTOR ESTERS/ ESTERIATE ACTURE ACTOR DECOR
0105-13	Ň	1	•	•	ACTIVE ACTO ESTERIZZETTE ACTIVEZZENDOUTEED NETE TOUES
0105414	Ň	ţ	•	•	ACTULE ACTO ESTENSAMEINT ACATLEMADDIFIED ACADE BOACHE
0103413		1	•	•	ACALET ACTO ESTEMBLOTTE ACALEMICANOULLED MELLE HADES
	•	:		•	
0170=01	N	3	1	•	ACTION INTELEPTROPYLERE AMOUNTATION
0175401	0	1	•	•	ADIFIC ACIDIDI (ZEEINVLHEATLIESTENZ
0178401			•	•	ADIMIC ACTUBILIEISODECTE ESTERATSTERIFICATION OF ADIMIC A
0177401	G	1	•	•	ADIPIC ACIU, DIVINIUECTE ESTENZESTENIPICATION OF ADIPIC A
0114001	0	1	1	٠	ADIPIC ACID ESTENSIESTENFICATION OF ADIPIC ACID
0180=01	C ·	:	1	٠	ADIPIC ACTUVOTION OF CYCLOMEXAND
0180+02	E	1		•	ADIPIC ACID/DEPOLTMENIZATION OF NYLON 6
0180-03	ç	•	2	٠	ADIPIC ACID/04IDATION OF CYCLOHERANOL/ONE HIX
0180-80	C	1		•	ADIPIC ACIDA OXIDATION OF CYCLOHEXANE VIA OLAONE
0185+01	B	•	1	•	ADIPONITRILEZCHLORINATION + CYANATION OF BUTADIENE
0185=03	18	1	•	٠	ADIPONITHILE DIRECT HYDROCYANATION OF BUTADIENE
0185=04	ĸ	٠	1	•.	ADIPONITRILEZAMMONOLYSIS OF ADIPIC ACID, DEHADRATION OF DIAMIDE
0185#95	19	1	•	•	ADIPONITRILE/ELECTROMYDRODIMERIZATION OF ACRYLONITRILE
0186-01	5	1	•	٠	ALKOXY ALKANOLS/ ALKOXY ALKANOLS FROM ALKYLENE OXIDE AN
0189=70	D1	•	1	•	ALKYD RESINS/
0189=01	01	5	15	1	ALKYD RESIN/CONDENSATION POLYMERIZATION
0191-01	12	2	•	٠	ALKYL PENZENES/ALKYLATION OF BENZENE WITH ALPHAWOLEFINS
0192-01	F1	•	1	•	ALKYL AMINES/HYDROGENATION OF FATTY NITRILE
0192=03	ĸ	1	•	•	ALKYL AMINES/AMINATION OF ALCOHOLS
0192=04	Z	1	•	•	ALKYL AMINES/C-13+C19 FROM OLEFIN & HCN & HZ
0195-01	11	1	•	•	ALKYL PHENOLSZNONYL-OCTYL ALKYLATION OF PHENOL

PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
0195+02	11	2	•		ALKYL PHENOLS/MIXED ALKYLATION OF PHENOL
0200-01	Ž	ĩ		•	ALLYL ALCOHOL/REDOX OF ACROLEIN AND SEC-RUTANOL (REDUCTION BY ALUMINUM BUTOXIDE)
0200-03	Ē	i		•	ALLYL ALCOHOL/HYDPOLYSIS OF ALLYL CHLORIDE
0210-01	A	2			ALLYL CHLORIDE/CHLORINATION OF PROPULENE
0230=01	5		i		AMINOPTHYL FTHANOLAMINE/RXN OF STHYLENEDIAMINE & STH-OXID
0235-05	7				PHAMINOPHENOL/REDUCTION OF NITROBENJENE-ACTO BEARDANGE
0240=01	Ğ. ·		ĩ		AMYL ACTTATES/REN OF ACTTC ACTO & AMYL ALCOUDLS
0300+03	F2				ANTI INF/RY-PRODUCT OF P-ANTNOPHENOL
0300=04	#2				ANILINE/NITROBENTENE HYDROGENATION
0320=01	T1		•		ANTSIDINE/HETHYLATION AND REDUCTION
0320-99			•	•	
0335=01	2	i	•	•	ANTHACTOR ZODAL TAR DISTILLATION
0350-99	č	i		:	ANTHRADUINONE/ OXIDATION OF ANTHRACENE
0358=00	Ĵ	-		•	
0358+01	Ĵ				ASPIRIN/ACETYLATION OF SALICYLIC ACID
0358-99	Ğ	i			ASPIRIN/ACETYLATION OF SALICYLIC ACID
0360=03	č	i		•	BENZAL DEHYDEZOX TDATION OF TOLVENE
0380=00	Ň	i			BENJENEZ STEAM PURDI VSTS OF I PO
0380-01	U	j			BENZENEZHYDRODEAL KYLTZATION OF TOLUENE ANDZOR XYLENE
0380=02	2	ž		•	AFNZENE/DIST OF BIX FXTRACT, CAT. PEROPHATE
0380=04	Ň			•	HENZENE/DIST. OF ATX EXTRACT-COAL TAB LIGHT OIL
0380=08	с	i			BENZENE AV WERDBUCT OF PHENDI WERE BY CUMPNE ON TOATION (BECOVERED DAW MATERIAL)
0380+09	ž	6	i		BENZENZATIST OF ATA EXTRACT-PYPOLYSTS GASALINE
0380=11	2	ĩ			AFNZENEZBY-PRODUCT OF SIL LONE MANUFACTURE
0380+12	ū	i			AFNTENE/AV-PRODUCT OF STYRENE BY FINY AFNY DEMYDROGENATI
0380-13	Ō	i			BENZENEZAV-PRODUCT OF ACRYLATE MANUFACTURE (BERDES)
0381=01	2	i			ATX-BENZENF TOLUENE XYLENE (HTXPO) / PYROLYSTS AASOLTNE PRO
0430-03	č	i			SENTOIC ACID/OXIDATION OF TOLUENE
0445=01	M ,	ī			RENZO-A-RYRENE (RY-PRODUCT OF ACETYLENE MEG) / STEAM PYROL VSIS
0495=99	8	ĩ	-		BENZOYI PEROXIDE/ BENZOYI CHI OPIDE + SODIUM PEROXIDE
0530=01	B	i			HENZYL CHIORIDE/CHIORINATION OF TOULENE
0550+99	Ť			•	BIPHENYL/HENZENE PASSED THRU HOT TURE
0554=01	20	i		-	BIS-(2+CHLORDISOPROPYL) FTHER/BY-PDCT PROPY OX CHLORDHYD
0560+01	A .	Ā			RISPHENOL -A/CONDENSATION OF ACETONE WITH PHENOL
0575-01	8	i			HROMOFORM/ACETONE + SODIUM HYPORROWITE (HALOFORM REACTION)
0590=01	15	11	ż		BUTADIENE (1+3)/FAT-DIST. OF CHA PYROLYZATES
0590=03	JZ	i	ī		BUTADIENE (1.3) / BY DEMYD. OF N-BUTANE AND-OR BU
0590=05	15				BUTADIENE (COPRODUCT OF ETHYLENE BY PYROLYSIS (1770-07)) / BY EXTRACTIVE DISTILLATION
0592-01	2	ž			BUTANEZ NATURAL GAS BY-PRODUCT
0592=02	F2	-	ī		BUTANE/BYPRODUCT OF BUTADIENE MANUFACTURE
0592=03			•	•	
	. <b>T</b>	1			QVIANE LAII PUENAJZNAFNINA LAIALYIJU BEFURBINN
0592404	3	1	•	•	BUTANE (ALL FORMS)/REFFINERY BY-PRODUCT

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FREQUENCY OF OCCURRENCE FOR EACH PRODUCT PROCESS

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
0640-01	F1	1	•	•	N-RUTYL ALCOMOL/BY PRODUCT OF 1+3-BUTYLENE GLYCOL BY HYD
0640-02	Fi	6	1	•	N#BUTYL ALCOHOL/HYDROGENATION OF N#BUTYRALDEHYDE+ OXO PR
0640=05	12	1	•	•	N-BUTYL ALCOMOL/DISTILLATION OF DILUTE AQUEOUS BUTANOL
0650+01	12	3	•	•	SECHBUTYL ALCOHOL/INDIRECT HYDRATION OF BUTENES
0860-99	12	1	•	•	TERT-BUTYL ALCOHOL/FROM ISOBUTYLENE
0710-01	<b>F1</b>	2	•	•	1+3 BUTYLENE GLYCOL/HYNROGENATION OF ACETALDOL
0720=01	15	5	2	•	BUTYLENES/BY EXTRACTIVE DISTILLATION OF C4 PYROLYZATES
0720=99	15	•	•	•	BUTYLENES/ FROM PYROLYZATE BY EXTRACTIVE DISTILLATION
0730-99	I	•	•	•	TERT-BUTYLPHENOL/ ALKYLATION OF PHENOL WITH ISOHUTYLENE
0750-01	0	2	1		N-BUTYRALDEHYDE/HYDHOFORHYLATION OF PRDPYLENE;0X0 PROCES
0760=01	C	1	•	•	N-BUTYPIC ACID/DXIDATION OF BUTYPALDEHYDE
0760-02	C	•	1	•	NHUTYRIC ACID/CO-PRODUCT OF BUTANE OXIDATION
0780-99	6	1	•	•	N-RUTYRONITRILE/ BUTANOL + NM3, DEHYDRATION
0785=00	٠	•	•	•	
0785=06	Z	•	•	•	CAPROLACTAM/FHOM PHENOL VIA CYCLOHEXANONE OXIME/
0785-07	Z	•	•	•	CAPROLACTAM/DEPOLYMERIZATION NYLON 6
0785=09	C	2	•	•	CAPROLACTAM/FROM CYCLOHEXANE VIA CYCLOHEXANONE AND OXIME
0790=99	3	1	•	•	CARBON DISULFIDE/PROCESS UNDER REVIEW
0810=01	8	2	•	•	CAPBON TETRACHLORIDE/CHLORINATION OF METHANE
0810=02	8	3	2	•	CARBON TETRACHLORIDE/CHLOR. OF METHYL CHLORIDE (FROM HYDROCHLORINATION OF METHANOL)
0810=03	8	•	•	•	CARHON TETRACHLORIDE/CHLORINATION OF CARBON DISULFIDE
0810=04	8	4	2	•	CAPBON TETRACHLORIDE/CO-PRODUCTION OF TETRACHLOROETHYLEN
0810-05	8	1	•	•	CARBON TETRACHLORIDE/BY-PRODUCT OF PHOSGENE MANUFACTURE
0814=01	01	•	•	•	CARBOXMETHYL CELLULOSE/ETHERIFICATION OF CELLULOSE
0815=00	E	•	•	•	CASTOR OIL (INCLUDING USP)/
0815=99	4	•	•	•	CASTOR OIL (INCLUDING USP) PROCESS UNDER REVIEW
0816=01	G	3	٠	•	CELLOPHANE/VISCOSE PROCESS
0820=01	23	5	•	•	CELLULOSE ACETATES FIBERS/SPINNING FROM ACETYLATED CELLU
0820-03	01	5	•	. •	CELLULOSE ACETATES RESIN/ACETYLATION OF CELLULOSE
0821=01	G	1	•	•	CELLULOSE ACETATE/BUTYPATES
0823=01	G	1	•	•	CELLULOSE ACETATES/PROPIONATES/ESTERIFICATN OF CELLULOSE
0824=01	L	1	•	•	CELLULOSE NITRATE/NITRATION OF CELLULOSE
0825=01	G	1	•	•	CELLULOSE SPONGE/VISCOSE PROCESS
0840-99	8	•	•	•	CHLOROACETIC ACID/ CHLORINATION OF ACETIC ACID
0890+01	8	2	•	•	CHLORORENZENE/CHLORINATION OF BENZENE
0890=99	•	•	•	•	CHLOROBENZENE/
0921-01	P	1	•	•	CHLORODIFLUCHOMETHANEZHYDROFLUORINATION OF CHLOROFORM
0930=01	8	3	2	•	CHLOROFORM/CHLORINATION OF METHYLCHLORIDE (FROM MYDROCHLORINATION OF METHANOL)
0930-02	8	2	•	•	CHLOROFORM/CHLORINATION DF METHANE
0930=03	b d	•	•	•	CHLOROFOPM/BY=PRODUCT OF CARHON TETRACHLORIDE PRODUCTION
0930=04	C	1	•	•	CHLOROFORM/ BY-PRODUCT OF ACETALDEMYDE PRODUCTION
09 <b>49=01</b>	8	1	•	٠	M-CHLORONITROBENZENE/CHLORINATION OF NITROBENZENE
0950-99	L	•	٠	٠	O-CHLORONITROBENZENE/ NITRATION OF CHLOROBENZENE

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0951	L	1	٠	•	P-CHLORONITHORENZENE/ NITHATION OF CHLORORENZENE
0961=01		1	•	•	2-CHLOPOPHENOL/CHLOWINATION OF PHENOL
0963=01	U	1	•	•	S-CHLOROPHENYL PHENYL ETHER/CHLOR OF PHENYL PHENYL ETHER
0993+00	6		•	•	CHOLINE CHLOWINE/ EINYLENE OXIDE + TRIMETHYLAMINE + HCL
0993-01	6	1	•	٠	CHOLINE CHLORIDEZ ETHYLENE CHLOROHYDRIN + TRIMETHYL AMINE
0993-99	4	•	٠	•	CHOLINE CHLORIDE/PROCESS UNDER REVIEW
0994-99	3	•	•	•	CHROMIUM(TOTAL)/PROCESS UNDER REVIEW
0997=01	H	•	1	•	COAL TAR/COKING OF COAL
0998=99	3	1	•	٠	COPPER (TATAL) + PROCESS UNDER REVIEW
1005=01	2	2 -	1	•	COAL TAR PRODUCTS (MISC+)/COAL TAR DISTILLATION
1007=01	H	•	•	٠	CREOSOTE/DIST. OF COAL TAR LIGHT OIL
1010-01	H	•	•	•	N+CRESOL/REFINING OF MIXED CRESOLS
1021=01	2	•	1	•	CRESALS+MIXED/TAP ACID RECOVERY AND REFINING
1030-01	2	•	1	•	CRESYLIC ACID/TAR ACID RECOVERY REFINING
1060-01	15	2	٠	•	CUMENE/ALKYLATION OF BENZENE BY PROPYLENE
1080-99	7	1	• .	•	CYANOACETIC ACID/ CHLOROACETIC ACID + NACN
1100-99	3	•	•	•	CYANURIC ACIDZ HEAT UREA
1120-02	FZ	2	•	•	CYCLOHEXANE/HYDROGENATION OF BENZENE
1122-01	FZ	1	•	•	CYCLOHEXANE DIMETHANOL/HYDROBENATION OF DIMETHYL TERAPHY
1130=02	¥5	•	•	•	CYCLOHEXANOL/HYDRORENATION OF PHENOL-DISTIL
1135+01	C	1	2	•	CYCLOHEXANOL/ONE (MIXED)/OXIDATION OF CCYCLOHEXANE
1140+01	C	•	1	•	CYCLOHEXANONE/OXIDATION OF CYCLOHEXANE
1140-02	F2	٠	•	•	CYCLOHEXANONE/HYDROGENATION OF PHENOL=DISTIL
1140=03	С	1	•	•	CYCLOHEXANONE/OXIO OF CYCLOHEXANE-DISTIL-DEHYDROG OF OL
1140=04	JI	1		٠	CYCLOHEXANONE/DEMYDROGENATION OF CYCLOHEXANOL
1170-99	DZ	1	•		CYCLOOCTADIENE/ DIMERIZATION OF BUTADIENE (NICKLE CAT,)
1171=01	15	2	٠	•	CYCLOPENTADIENE DIMER/EXTR DIST C5 PYROLYZATES + DIMERIZ
1190+01	<b>A</b>	2	•	•	DIACETONE ALCOMOL/ALDOL CONDENSATION OF ACETONE
1215#01	8	1	•	•	MODICHLORORENZENE/CHLORINATION OF BENZENE
1216-01	8	• 2	•	•	O-DICHLORORENZENE/CHLORINATION OF RENZENE
1216=02	V	1	٠	•	OGDICHLOROBENZENE/BYGPRODUCT OF POLYMERIC MDI OR THI MANUFACTURE (PHOSGENATION SOLVENT)
1220+01	8	3		•	P-DICHLOROHENZENE/CHLORINATION OF BENZENE
1221=01	ρ	٠	1	•	DICHLORODIFLUOROMETHANE/HYDROFLUORINATION OF CARBON TETRACHLD
1225+02	P	1		•	1+1 DICHLOPOETHANE/REACT OF HCL & VINYL CHLORIDE
1236=00	8	1	•		1+2+TRANS-DICHLOROETHYLENE/ CHLORINATION OF ETHYLENE
1244=01	8	10	1		1+2-DICHLORDETHANE/DIRECT CHLORINATION OF ETHYLENE
1244=02	9	1		•	112 DICHLOROETHANE/ CHLORINATION OF ETHYLENE
1244=03	S	6	-		1.2 DICHLOROETHANE/OXYCHLORINATION OF FTHYLENE
1257=01	R	1	•		DICHLORONITROBENZENE/CHLORINATION OF NITROBENZENE
1265-01	8	1		•	2+4-DICHLOROPHENOL/CHLORINATION OF PHENOL
1271-01	ě	-			1.2-DICHLOROPROPENE/BY-POCT OF ALLYL CHLORIDE HANUFACTUR
1281-99	20	ī	-	•	DICHLOROPROPANEZ BY PRODUCT CHLOROMYDRINATION OF PROPYLENE
1300-01	5	Ģ	ī	-	
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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TFXT
1300-04	5	•	2	•	DIETHYLENE OLYCOL (REFINING OF DISTIL TAILS) / ETHYLENE GLYCOL + E.O.
1300-05	E	•	•		DIETHYLENE GLYCOL/CO-PROD OF HYDROLYSIS -ETHYLENE OFIDE
1365=01	6	•	1	•	DIETHYLENE TRIAMINE/ETHYLENE DIAMINE + EDC + NH3
1440-99	02	1		•	DIISORUTYLENE/PROCESS UNOER REVIEW
1442-01	12	1	•	•	DIISOPROPYL BENZENE/HY-PROD OF CUMENE
1450=01	<b>A</b>	2		•	DIKFTENE/DIMERIZATION OF KETENE-ACETIC ACID
1465-99	5		•		2-DIMETHYLAHINDETHANOL/ ETHOXYLATION OF DIMETHYLAMINE
1470-01	<b>A</b>	•	•		NON-DIMETHYLANILINE/CATALYTIC CONDENSATION OF ANILINE
1480-99	<b>A</b>	1	•	•	DIMETHYL ETHER/DEHYDRATION OF METHANOL
1490-00	ĸ	1	•		NONDINETHYLFORMAMINE / DIMETHYLAMINE · FORMALDEHYDE
1490-99	ĸ	1	•		N.N.DIMETHYLFORMAMIDE/ DIMETHYLAMINE + FORMIC ACID
1500-99	8	i			DIMETHYL SULFATE/ METHANOL + SULFURIC ACID
1510-99	1 e	ī			HETHYL SULFIDE/ K METHYL SULFATE +K25
1530=01	G	Ś			DIMETHYL TEREPHTHALATE/ESTERIFICATION OF TPA+METHANOL
1530-02	Ċ	2			DIMETHYL TEREPHTHALATE/OXIDATION OF P-XYLENP METHYL ESTE
1535-99	Ľ	ī			DINITROBENZENE (M.O.P)/ NITRATION OF BENZENE
1550=01	Ē	Ă	-		DINITROTOLUENE (MIXED)/NITRATION OF TOLUENE
1551=01	Ē	1			2+4 DINITROTOLUENE/NITRATION OF TOLUENE
1552=01	ĩ	ĩ			2+6 DINITROTOLUENE/NITRATION OF TOLUENE
1580=01	Ā	ĩ			DIPHENYLAMINE/CATALYTIC CONDENSATION OF ANTLINE
1590=99	<b>A</b>	-	ī		DIPHENYL FTHER ZCHLOROBENZENE + SODIUH PHENOLATE
1610-01	ŝ	;		•	DIPROPYLENE & YCOLARIN OF PROP. & YCOL + PROPY OXIDE
1610-02	Ē	-	•	•	DIPROPULENE OLYCOL/CO-PROD OF WYDROLYSIS-PROPULENE OXIDE
1612-99	Å	•	•	•	2 2 - DITHIOHISHENZOTHIAZOUEZ AND EZ ANTONOEZ ANTERADIORENZOTHIAZOUE
1634=01				•	DODECYL BUANDINE ACETATE/CONDENSATION OF DODECYL AMINE AC
1635=99		i	i	•	DODECYINERCAPTAN/ DODECHE A USS
1641-99	1	i	•	•	DVERLYE INFEMENTATESJUAT DVER ARD DVER
1650-01	20	i		•	FPICHIOROHYDRIN/FROM ALLYL CHIORIDE VIA DICHIOROHYDRIN
1653=01	21	2	;	•	FOXIDIZED ESTERSTOTAL ZE OXIDATION OF UNCATINGATED ESTED
1656=00	01	-	-	•	FPOLY RESINS/ EPICH OROHORIN & RISPHENDIA
1656-01	ni	i	i	•	
1656-02	DI	ĩ	ĩ	•	
1656=03	01	i		•	FPOXY RESINS/FPICHLOROHYDRIN AND NOVOLAK RESINS
1656-04	D		•	•	FOLLY DESINS/FRONTIONTION OF DOLYMERS
1656907	· D1	•	•	•	FPOLY RESINS/PURCHED FPOLY (FPI-RIS) RESIN & RISPHENGL A
1656=08	ni	i	•	•	FPOLY RESINS/PURCHED FOLY DESINGLADING ALFATTY ACTO
1656-09	01		•	•	FPOXY BEGING/MONTEER FOXYECTED
1656=10	nî	•	ī		FPOLY RESTNS/PURCHASED FPOLY(FPT=RTS) RESTNADIETMANDLAMIN
1659-01	H H	i			FTHANF/CRACKING OF NAPHTA
1659=02	2	i	-	•	ETHANE & NATURAL GAS BY-PRODUCT
1659=03	i	•	•	•	
1660=01	1.2		2	•	ETHANDI ANDEAT MUNDATIAN AF ETHVIENE
1660-02	16	3	E 1	٠	ETHAND ADDRALLOT OF BITANE OFTOTATION
1000-05	L	•	+	•	FINANC'S COLUCION OF BUIENC OFFICETION

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PP_CODE	GEN_CODE	OTR	ZERO	UNK	PP_TEXT
1661-01	ĸ	4	•	•	ETHANOLAMINES/AMMINOLYSIS OF ETHYLENE OXIDE
1666=01	5	4	•	•	ETHOXYLATES/FROM ALKYLENE OXIOE AND ALKANOL
1666-02	5	2	•	•	ETHOXYLATES, (HISC) / ALKYLPHENOL-PHENOL & ETHYLENE OXIDE
1666-03	5	2	1	•	ETHOXYLATES/C11+C12+LINEAR ALCOHOLS AND ETHYLENE OXIDE
1670-01	G	1	2	•	ETHYL ACETATE/ESTERIF OF ACETIC ACID WITH ETHANOL
1670-02	С	•	1	•	ETHYL ACETATE/COPRODUCT OF BUTANE OXIDATION
1670-03	Ε	•	•		ETHYL ACETATE/FROM POLYVINYL ACETATE AND ETHANOL
1670-04	<b>A</b>	1	•		ETHYL ACETOACETATE/ ACETALDEHYDE CONDENSATION
1700-01	ĸ	1	•		ETHYLAMINE /AMMONOLYSIS OF ETHANOL
1710+01	15	5	٠	•	ETHYLBENZENE/BENZENE ALKYLATION
1710+02	H	3	•		ETHYLBENZENE/SEPARATION FROM BIX EXTRACT
1710-03	I	1	1		ETHYLBENZENE / ALKYLATION OF BENZENE WITH ETHYLENE
1710-04	2	1	•	•	ETHYLBENZENE/DISTILLATION FROM COAL TAR
1730-01	14	1	٠	•	ETHYL CELLULOSE/CELLULOSE + ETHYL CHLOPIDE
1740=01	P	7	•	•	ETHYL CHLORIDE/HYDROCHLORINATION OF ETHYLENE
1760-99	G	1	•	•	ETHYLCYANOACETATE/ ETHANOL ESTER OF CYANDACETIC ACID
1770-01	H	11	4	•	ETHYLENE/PYROLYSIS OF ETHANE/PROPANE/BUTANE/LPG
1770-02	H	5	1	•	ETHYLENE/PYROLYSIS OF NAPHTHA AND+OR GAS OIL
1770=03	H	2	•	•	ETHYLENE/PYROLYSIS OF ETHANE
1770=04	н	2	•	•	ETHYLENE/PYPOLYSIS OF NAPHTMA;PROPANE;ETHANE;BUTANE
1770-05	2	1	•	•	ETHYLENE/FRACTIONATION OF REFINERY LIGHT ENDS
1800-01	ĸ	2	1		ETHYLENE DIAMINE/AMINATION OF ETHYLENE DICHLORIDE
1810-99	0	•	•	•	ETHYLENE DIBROHIDE/ ETHYLENE + BROMINEW
1830+01	Ľ	11	1	•	ETHYLENE GLYCOL/HYDRDLYSIS OF ETHYLENE OXIDE
1030-02	E	2	1	•	ETHYLENE GLYCOL/HYDROLYSIS OF ETHLENE OXIDE
1920-99	G	1	٠	•	ETHYLENE GLYCOL MONOETHYL ETHER ACETATE ETHANOL + ETHYLENE OXIDE + ACETIC ANHYDRIDE
1930+01	A .	•	٠	•	ETHYLENE GLYCOL MONOMETHYL ETHER/ METHANOL + ETHYLENEOXIDE
1930-02	5	1	•	•	ETHYLENE GLYCOL MONOMETHYL ETHER/ETHOXYLATION OF ALCHOLS
1930-99	5	5	•	•	ETHYLENE GLYCOL MONOMETHYL ETHER/ METHANOL + ETHYLENE OXIDE
1940=99	G	1	•	•	ETHYLENE GLYCOL MONOMETHYL ETHER ACETATE/ METHANOL + E.O. + ACETIC ANHYDRIDE
1980=01	C	11	2	•	ETHYLENE OXIDE/DIRECT OXIDATION OF ETHYLENE
1980-02	R	1	•	•	ETHYLENE OXIDE/VIA ETHYLENE CHLOROHYDRIN PROCESS
1985+01	20	8	1	•	POLYETHYLENE & POLYVINYL ACETATE COPOLYMERS/
1995+01	C	1	•		2 ETHYL HEXANOIC ACID/OXIDATION OF 2-ETHYLHEXALDEHYDE (FROM ALDOL COND OF N-RUTYRALDEHYDE)
1995-02	Ç	•	•	•	Z=ETHYLHEXANOIC ACID/OXIDATION OF 2=ETHYLHEXALDEHYDE
2000=01	<b>A</b>	•	٠	•	2-ETHYL HEXANOL/ALDOL CONDENSATION-HYDROG OF N-BUTALDEHY
2010+99	14	1	•	٠	ETHYL ORTHOFORMATE/ CHLOROFORM + NAETHANOLATE
2035-01	DZ	3	1	•	FLUOROCARBON RESINS/POLYMERIZATION OF FLUORINATED OLEFINS
2040401	C	11	9	•	FORMALDEHYDE/OXIDATION OF HETHANOL-SILVER CATALYST
2040403	C	7	9	•	FORMALDEHYDE/OXIDATION OF METHANOL-METAL OXIDE PROCESS
2050-99	ĸ	1	•	•	FORMAMIDE/ FORMIC ACIO + AMMONIA
2060-01	C	٠	2	•	FORMIC ACID/RY+PRODUCT OF BUTANE OXIDATION
20000002	3	•	•		FORMIC ACID/RECOVERY FROM SULFITE PULP WASTEWATER

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	PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
	2070+01	22	1	•	٠	FUMARIC ACID/ISOMERIZATION OF MALEIC ACID
	2080=00	24	•	•		GLUTAMIC ACID, MONOSODIUM SALT/
	2080=01	24	•	•	•	GLUTAMIC ACID, MONOSODIUM SALT/BLUTAMIC ACID + NAOH
	2080-99	1	•	•	•	GLUTAMIC ACID, MONOSODIUM SALT/NEUT OF GLUT ACID BY NAOH
	2090-01	12	1	•	•	GLYCERINE (SYNTHETIC)/HYDROXYLATION OF ALLYL ALCOMOL
	2090=03	E	2		•	GLYCERINE (SYN)/HYDROL OF EPICHLOROHY VIA ALLYL CHLORIDE
	2120-99	C	2	•	•	GLYOXAL/ DZONATION OF BENZENE
	2136-01	н		•	•	HEPTANE/BTX SOLVENT EXTRACTION AND ADSORPTION
	2145+01	B		1		HEXACHLORDBENZENE/CHLORINATION OF BENZEME
	2145-02	<b>A</b>	. 1	•	•	HEXACHLOROBENZENE/BY PRODUCT OF TETRACHLOROETHYLENE (CHLORINATION OF EDC)
	2145-03	8	1		•	HEXACHLOROBENZENE/BY=PRODUCT OF CHLOROSILANES
	2150=01	8	1	•	•	HEXACHLOROETHANE/ETHANE CHLORINATION
	2165+01	ĸ	•	1	•	HEXAMETHYLENEDIAMINE/AMMONOLYSIS OF 1=6 HEXANEDIOL
	2165=02	#1	2	2	•	HEXAMETHYLENEDIAMINE/HYDROGENATION OF ADIPONITRILE
	2165+03	E	1	•	•	HEXAMETHYLENEDIAMINE/DEPOLYMERIZATION OF NYLON 66
	2100-99	<b>A</b>	1	•	•	HEXYLENF GLYCOL/ ALDOL CINDENSATION OF ACETONE
	2170-99	<b>F</b> 1	•	1	•	HEXAMETHYLENE GLYCOL(1.6-HEXANEDIOL)/
	2180-99	A	•	2	•	HEXAMETHYLENETETRAMINE / FORMALDEHYDE + NH3
	2181-01	н	•	•	•	HEXANE/RTX SOLVENT EXTRACTION AND ADSORPTION
-	2181-02	3	•	•		HEXANE / REFINERY BY-PRODUCT
T	2185-99	3	1	•		HYDRAZINE SOLUTIONS/PROCESS UNDER REVIEW
ω	2190-99	3	1	2	•	HYDROGEN CYANIDE / PROCESS UNDER REVIEW
	2200-01	C	1	•	•	HADBOGNINONELOXIDATION OF ANIFINE AIV GNINONE
	8212+01	- 5	1	•	•	HYDROXYETHYL CELLULOSE/ETHOXYLATION OF ALKALI CELLULOSE
	2215-99	3	1	•	٠	HYDROXYLAMINE/PROCESS UNDER REVIEW
	2216=01	01	•	•	•	HYDROXYPROPYL CELLULOSE/ETHERIFICATION OF CELLULOSE
	2225+99	<b>A</b>	•	1	•	IMINODI-ACETIC ACIO/ NH3 + FORMALDEMADE + NACN
	2240-99	15	1	•	•	ISOAMYLENE / EXTRACTIVE DISTILLATION OF BTX RAFFINATE
	2250-01	. ₹1	5	1	•	ISOBUTANOL/HYDROG OF ISOBUTYRALDEHYDE=OXO PROCESS
	2250-02	A	٠	1	•	ISOBUTANOL/FROM ISO-BUTYRALDEMYDE BY ALDOL CONDENBATION
	2250-03	<b>A</b>	1	٠	•	ISOHUTANOL/ CHUDE ISOBUTANOL BY ALDOL CONDENSATION/ HYDR
	2260-99	G	1	1	•	ISORUTYL ACETATE/ISOBUTANOL + ACETIC ANMYDRIDE
	2265-01	15	2	•	•	ISOBUTYLENE/EXTRACT FROM C4 PYROLYZATE
	2265-02	· •	1	•	٠	ISOBUTYLENE/DEHYORATION OF PURCHASED TERT-BUTANOL
	5599-01	D <b>2</b>	1	•	٠	ISORUTYLENE POLYMERS/POLYMERIZATION OF ISORUTYLENE
	2270-01	0	2	1	٠	ISORUTYRALDEHYDE/HYDROFORMYLATION OF PROPYLENE=OXO PROCE
	2280-01	С	1	•	•	ISORUTYRIC ACID/ AIR OXIDATION OF ISORUTYRALDEHYDE
	2300+99	0	1	٠	•	ISODECANOL/ CARBONYLATION OF OLEFIN OLIGOMERS + H2
•	2320-99	0	1	•	•	ISOOCTYL ALCOHOL/CARBONYLATION OF OLEFIN OLIGOMERS + H2
	2321=99	2	•	•	•	ISOPENTANE/ DISTILLATION FROM CA/C5 HYDROCARBON MIX
	2330=01	A	1	٠	٠	ISOPHORONE/CATALYTIC GAS PHASE RXN OF ACETONE
	2340-99	C	1	•	•	ISOPHTHALIC ACID/ OXIDATION OF M-XYLENE
	2350+02	15	2	•	•	ISOPRFNEZEXTRACTIVE DIST CS PYROLYZATE

	PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
	2360=01	12	3	1		ISOPPOPANOL/HYDRATIONS OF PROPALENE
	2360-02	F1	1		•	ISOPROPANOL/CATALYTIC HYDROGENATION OF ACFTONE
	2360=03	12	ī	•		ISOPROPANOL/SOLVENT-WATER AZEOTROPIC DISTILLATION
	2370-99	Ğ	•	ī	•	ISOPROPYL ACETATE/ISOPROPANOL + ACETIC ANHYDRIDE
	2415=99	3	•	•	•	LEAD(TOTAL)/PROCESS UNDER REVIEW
	2420-00	12	•		•	HALIC ACID/ HYDRATION OF HALFIC ACID
	2430-01	Ċ	3	•	•	MALEIC ANHYDRIDE/RENZENE OXIDATION
	2430+02	Ċ			•	MALEIC ANHYDRIDE/BY-PRODUCT OF PHTHALIC ANHYDRIDE BY OXI
	2441+01	Â	i	•	•	HELAMINE/TRIMERIZATION OF UREA
	2442-99	3		•	•	MELAMINE CRYSTAL/ CONDENSATION OF URFA
	2443-01	01	Ť	3		MELAMINE RESINS/POLYCONDENSATION OF MELAMINE WITH FORMAL
	2450-01	A T	1			MESITYL OXIDE/DEHYDRATION OF DIACETONE ALCOHOL
	2455-99	F2	•			METANILIC ACID/ HYDROGENATION OF 3-NITROBENZENE SULFONIC ACID
	2460+01	10	ż		-	HETHACRYLIC ACID/ACETONE CYANOMYDRIN PROCESS
	2470-02	Ğ	ž		-	NETHACPYLIC ACID ESTERS/BUTYL HETHACPYLATES-ESTERTETC OPM
	2470-03	Ğ	ī			HETHACDYLIC ACID ESTERS/ETHYL HETHACDYLATE BY HETHACDT A
	2470+04	Ğ	i		-	METHACRYLIC ACID ESTERS/2-FINI METVI METUACOLTE DY MEA
	2470-05	Ğ	ī			METHACRY IC ACID ESTERS/HIGHED METHACRY ATER BY METH ACT
	2470-06	å	i			METHORY I C ACID ESTER/JACETOACETOXYETHI METHACOLTE EDM
	2470-07	ā	i			HETHACPYLIC ACID ESTER/HETHACPYLIC ACID ESTERIETE NO
>	2470+08	Ğ	ī			METHACOVITC ACTO PETERS/BITVI METHACOVIATE BY ACH & MENI
C	2500+01	õ	Ś	;	•	WETWORKEY WEY CONTRESS FORM NAT ALC VIA RUDAL
	2500+02	õ	2	-	•	
	2500=03	ř.	-		•	METMANNI/WANDANENATIAN AF CADRAN MANAYINE
	2500=04	<b>F</b> 1		•	•	METHINGEFFICHOGENATION OF CARDON HONORIDE
	2500+06	F	:		•	
	2500+07	D1		ī		METHANOL/CET PRODUCT POLYES
	2500-09	č	•	i	•	
	2500=10	ž	ī	•	•	
	2530-00	6				
	2530+01	ĸ	i		•	METHYLANINES (TOTAL) (CONDENSATION OF METHANOL AND AMMONIA
	2545+01	P	ĩ	•	•	METAL REALTO THE FORMAL OF ANTION OF METALANOL
	2555=01	14	2		•	
	2560=00	16	-	ĩ	•	METAVI CULIDIDE PROBALI VIGEO AS COLORIDE
	2560-01	P	÷	:		
	2560+03	A	,		•	
	2560-05	ř	;	•	•	METUNI CULORISE / SUBDALIST OF FLUXIES OF TA 1985-A1
	2620=01	Ä	i	2	•	METHY FNE CHECKIDE / BUY HUDDINGTION OF NET HYLCHICOLDE (980M HYDROCHICOTNATION OF METHANOLA
	2420-02	Ā	ĩ		•	WENTLENE ON OF DEFONDED ATTACH OF METAADE
	2620=03	č	;	•	•	METHYENE CHEMINES CHEMINES WITHE ATTRE 1884-41
	2630-01	Ă	;	•	•	METHICIC CALUTION OF ACTION OF FOOMAL DEWYSE AANTIING
	2631-00	Ā		•	•.	ALAINETUS SIGNILINES REACTION OF FORMALIENTIE MANIETRE
	2635=01	-	;	•	•	METWYIENE DIDUEWYI DITECAYANAT#/DUGGENATUDA

PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
2640-01	JI	2		•	METHYL ETHYL KETONE/DEMYDROGENATION OF SEC-BUTANOL
2640=06	C		2	•	METHYL ETHYL KETONE/RY-PRODUCT OF HUTANE OXIDATION
2640-07	Jl	1	•	•	METHYL ETHYL KETONE/REDOX. OF ACPOLEIN & SEC-BUTANOL
2640-08	JI	1	•	•	METHYL ETHYL KETONE/ DEHYDROGENATION OF SEC-RUTANOL
2645+99	G	ī	1	•	METHYL FORMATE/ FORMALDEHYDE + CAUSTIC (CANNIZARO)
2650-99	A	1	•	•	METHYL ISOBUTYL CARBINOL / ALDOL CINDENSATION OF ACETONE
2660-01	Fl	3	•	•	METHYL ISOBUTYL KETONE/HYDROGENATION OF MESITYL OXIDE
2065-02	G	3		•	METHYL METHACRYLATE/METHANOLYSIS OF ACETONE CYANOHYDRIN
2665-03	G	•	•	•	METHYL METHACRYLATE/METHACRYLIC ACID + METHANOL
2665-04	н		•		METHYL METHACRYLATE/POLYMER CRACKING
2680-01	G	•	•	•	METHYL SALICYLATE/ESTERIFICATION OF SALICYLIC ACID
2690=01	С	Ĵ	•	•	A-HETHYLSTYRENE/AY-PROD OF ACETONE&PHENOL BY CUNENE DXID
20-0692	JZ	•		•	ALPHAOMETHYLSTYRENE/DEMYDROGENATION OF CUMENE
2691-01	Ū1	- 1	•	•	MODACRYLIC RESIN/RESIN-POLYACRYLONITRILE & COMONOMER
2691-07	D		•	•	MODACRYLIC RESIN/RESIN+POLYACRYLDNITRILE & COMONOMER
2691+11	D1	ī	•		MODACRYLIC RESIN/FIRER-POLYACRYLONITRILE & COMONOMER
2695-99	6	-	•	•	2. (HORPHOLINO-THIO) -RENZOTHIAZOLE/ MORPHOLNE + 2-HERCAPTOBENZOTHIAZOLE
2701-01	ž	i	•	•	NAPHTHALENE/SEPARATION FROM COAL TAR DISTILLATE
2701-02	ž	2			NAPHTHALENE/DISTILLATION FROM PYROLYSIS GAS
2750=01	õ	ī			NEOPENTANOIC ACID/FROM ISOBUTYLENE VIA OXO PROCESS AND OXIDATION
2756=01	ĸ	ī			PONITROANILINE/AMMONOLYSIS OF PARAONITROCHLOROBENZENE
2757+01	ĸ	i			PONITPOANTLINE/AMMONOLYSIS OF PARAUNITROCHLOPORENZENE
2770-01	É.	Ă.			NITROBENZENE/NITRATION OF BENZENE
2770=10					NI TROBENZENE/
2792-00	ĩ	ĭ	-		2-NTTROPHENOL / NTTRATION OF PHENOL
2792-99					PANITROPHENOL/PROCESS UNDER REVIEW
2793-99	Ĩ.	i			PONTROPHENOL & SONTUH SALTZ NTRATION OF PHENOL
2800-01	ī	2	•		NT POTOLUENE /NI TRATION OF TOLUENE
2804-99	ē	-	•		NeNT ROSODTPHENT AMINE / DIPHENT AMINE + NT ROUS OF TOP
2824-01	ž				NYLON SALT/ADIPIC ACID & MMDA+AQUEDUS SOLUTION
2824=02	7		i		NYLON SALT/ADIPIC ACTO & HMDA-HETHANOL SOLUTION
2825-00				-	NYL ON/
2825-01	D1	i.	ż		NYLON 6/RESIN BY POLYCONDENSATION FROM CAPPOLACTAM
2825=02	01	7	ī		NYLON A/FIRER BY POLYCONDENSATION FROM CAPROLACTAM
2825-03	01	2	ī		NYLON 66/RESIN BY POLYCONDENSATION FROM NYLON SALT
2825+04	<b>D1</b>	ā	i		NYLON 66/FIBER BY POLYCONDENSATION FROM NYLON SALT
2825-05	01	i	•		NYLON 612/RESINE POLYCONDENSATION FROM HEDA & CISCIACIDS
2825-08	DI	i	•		NYLON 6 & 66 COPOLY/POLYCOND OF NYLON SALT & CAPPOLACTAM
2825-09	01	i	•		NYLON/HISCELLANEOUS NYLON PRODUCTS-RESINS
2025-10	õi	i -			NYLON/NYLON FORMAL DEHYDE RESIN
2825+11	01	•	-		NYLON COATING SOLUTION/
2825-22	21	1	•	•	
2825-90	21	i			NVI ONINI ON FINISHING PROFESES
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PP_CODE	BEN_CODE	DIR	ZERO	UNK	PP_TEXT
2831=01	0	•	٠	•	OXO ALDEHYDES-ALCOHOLS/MISC ALDEHYDES
2831-02	0	•	1	•	DXO ALDEHYDES-ALCOHOLS/AMYL ALCOHOL
2831=05	0	•	1		OXO ALDEHYDESHALCOHOLS/C11+C14 ALC FROM C10-C+C13 OLFFIN
2831-07	0	1	•	•	OXO ALDEHYDES/ALCOHOLS/NEOPENTANOL FROM ISORUTYLENE
2831-08	0		1		ONO ALDEHYDES-ALCOHOLS/AMYL ALDEHYDE (MIXED)
2832-01	н		•		OCTANE/RTX SOLVENT EXTRACTION AND ADSORPTION
2841+01		1	•		PENTACHLOPORENZENE/BY-PRODUCT OF RENZENE CHLORINATION
2843-01	B	1	•		PENTACHLOROPHENOL/CHLORINATION OF PHENOL
2843-99	9	•	1	•	PENTACHLOROPHENOL/ CHLORINATION OF PHENOL
2850=0]	A -	3	•	•	PENTAERYTHRITOL/ALDOL COND. OF ACETALDEHYDE & FORMALDEHYD
2853=01	3			•	PENTANE/REFINERY BY-PRODUCT
2856=01	02	Ĵ	1	•	PETROLEUM HYDROCARBON RESINS/FROM CS-CB UNSATURATES
2857+01	C	ĩ	•		PERACETIC ACID/PEROXIDATION OF ACETALDEHYDE
2858-01	G	2	2		PHTHALIC ESTER/PHTHALIC ACID AND ALCOHOL ESTERIFICATION
2859=01	G	2	•	•	RIS 2-ETHYLHEXYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIET
2860-01	G	•	•	•	BUTYLOCTYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATION
2862-01	G	•	•	•	C7-C10 PHTHALATE ESTER/2940 AND ALCOHOL ESTERIFICATION
2863-01	G	1	•		C11=C14 PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATION
2865+01	G	2	•	•	DIAUTYL PHTHALATE ESTER/2960 + ALCOHOL ESTERIFICATION
2867-01	G	•	•	•	DI-DECYL PHTHALATE ESTER/2960 AND ALCOHOL ESTERIFICATION
2868+01	G	•	•	•	DI-N-HEXYL PHTHALATE ESTEP/2960 AND ALCOHOL ESTERIFICATI
2870-01	G	i	•	•	DI-ISODECYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATIO
2871-01	G	1	•	•	DI-ISOOCTYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATIO
2872=01	G	•	•		DIPHENYL PHTHALATE ESTER/PHENOLAPHTHALYL CHLORIDE ESTERI
2873-01	G	1	•	•	DI-TRIDECYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATIO
2874-01	G	1	•		N-HEPTYLNONYLUNDECYL PHTHALATE ESTER/2960 & ALCOHOL ESTE
2875=01	G	1	•	•	N-HEXYL-2-ETHYLHEXYLISODECYL PHTHALATE ESTER/2960 & ALCO
2876=01	G	1	•	•	N-HEXYL-2-ETHYLHEXYL PHATHALATE/2960 & ALCOHOL ESTERIFIC
2877-01	G	1	•	•	NOMEXYLHEPTYLNONYLUNDECYL PHTHALATE ESTER/2960 & ALCOHOL
2878=01	G	1	•	•	N-HEXYLOCTYLDECYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIF
2879-01	G	1	•	•	MIXED ALCOMOL PHTHALATE ESTER/2960 & ALCOHOL ESTERFICATI
2880=01	G	•	٠	٠	OCTYLDECYL PHTHALATE ESTER/2960 & ALCOHOL ESTERIFICATION
2883-01	6	2	•	•	DIGETHYL PHTHALATE ESTER/2960 AND ALCOHOL ESTERIFICATION
2084=01	G	•	•	•	DI-N-OCTYL PHTHALATE ESTER/2960 AND ALCOHOL ESTERIFICATI
2885=0]	G	3	•	•	DIHETHYL PHTHALATE ESTER/2960 AND ALCOHOL ESTERIFICATION
2886-01	G	1	•	•	BUTYL9ENZYL PHTIHALATE ESTER/2960 + BUTANOL + BENZYL ICORIDE
2905-01	01	14	24	•	PHENOLIC RESINS/POLYCONDENSATION OF PHENOL WITH FORMALDENYDE
20+0162	C	6	•	•	PHENOL/CUMENE PEROXIDATION AND ACID CLEAVAGE
2910-03	C	1	•	•	PHENOL/OXIDATION OF TOLUENE
2910-05	M	1			PHENOL/BENZENE SULFONATION+HYDROLYSIS
2910-08	С	1	•	•	PHENOL/LIQUID PHASE OXIDATION OF RENZOIC ACID
2910-09	16	•	1	•	PHENOL/TAR ACID RECOVERY REFINING
2910-10	16	1	•	•	PHENOL/RECOVERY FROM PYROLYSIS GASOLINE

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
2910-12	•	1		•	PHENOL/BY-PRODUCT OF DIPHENYL OXIDE
2910-13	<u> </u>	i	•	•	PHENOL/BY-PRODUCT OF ISOCYANATES (FROM NITORENZENE AND DINITROTOLUPNE PROURSORS)
2950-00	3			•	PHOSGENE/ CHLORINE + CARHON MONOXIDE
2950+01	i 📮	6			PHOSEENE/CHLORINATION OF CARBON MONOXIDE
2951-01	G	1			PHOSPHATE ESTHERS/PHOSPHORUS OXYCHLORIDE AND PHENOL/ALC
2951-02	G	i	•		PHOSPHATE ESTERS /DIPHENYLISODECYL . POCL38PHENOLSISODEC
2960-01	C	ĩ.			PHTHALIC ANHYORIDE/OXIDATION OF NAPHTHALENE
2960-02	- C	Ž		•	PHTHALIC ANHYDRIDE/OXIDATION OF Q=XYLENE
2981+01	Ĥ	•	•		PITCH TAR RESIDUE/SEP.FROM COAL TAR LIGHT OIL DISTILLATE
2990-99	D1	i			POLYCETAL RESINS/PROCESS UNDER REVIEW
2992=99	02	•			POLY-ALPHA-METHYL STYRENE/PROCESS UNDER REVIEW
2996=01	01	-	•	ĩ	POLYAMIDES FROM ETHYLENE AMINES AND FATTY ACIDS
2996=02	01			•	NYLONS/FROM ADIPIC ACID, DIETHYLENETRIAMINE, EPICHLOROMYDR
2996+05	01		ī		POLYAMIDES-ALSO SEE NYLONS/FROM DIBASIC ACID AND AMINES
2996=99	01		ī	-	POLYAMIDES/ DICARBOXYLIC ACID + DIAMINE
3000-99	. 02		ĩ		POLYBUTENES/ SOLUTION POYMERIZATION OF BUTYLENES
3004=00	V	i	•		POLYCAPRONATES/GENERAL
3004-99	V	2		•	POLYCARBONATES/PROCESS UNDER REVIEW
3006=05	3		•		POLVESTER/FIBER - MELT SPINNING FROM PURCHASED RESIN
3006-20	D1	Ź			POLYESTER/RESIN BY POLYCOND. FROM DHT & 1,4=CYCLOHEXANOL
3006=21	D1	6	Ž	•	POLYESTER/RESIN BY POLYCOND. FROM TPA & ETHYLENE GLYCOL
3006-22	D1	1		•	POLYESTER/RESIN BY POLYCOND. FROM PHTHALIC AND ANHYDR.
3006-23	01	7	2	•	POLYESTER/RESIN BY POLYCOND. FROM DMT & ETHYLENE GLYCOL
3006+26	01	1	•	٠	POLYESTER/RESIN BY POLYCON, FROM TPA OR DMT & ETHYLOLYC,
3006=28	01	2		•	POLYESTER/RESIN BY POLYCON, FROM DMT AND BUTANEDIOL
3006=29	01	3	1	1	POLYESTER/RESIN BY POLYCOND. FROM VARIOUS ACID & ALC.
3076-30	01	1		•	POLYESTER/FIBER BY MELT SPINNING FROM DMT AND 1+4
3006=31	01	•	•	•	POLYESTEP/FIBER BY MELT SPINNING FROM TPA AND ETHY GLYCL
3006=33	01	10	•	•	POLYESTER/FIBER BY MELT SPINNING FROM DHT AND ETHY OLYCL
3006=34	01	1	•	•	POLYESTER/FIBER BY MELT SPINNING FROM TPA OR DMTGETHYBLY
3006=35	23	•	•		POLYESTER/FIBER BY MELT SPINNING FROM PURCHABED RESIN
3006-37	01	1	•	•	POLYESTER/FIRER
3006=40	D1	1	•	•	POLYESTER/FILM FROM OMT AND 106 CYCLOMEXANE DIMETHANOL
3006-90	3	٠		•	POLYESTER/FINISHING PROCESS
3006-99	D1	•	1	•	POLYESTER/
3008-01	02	9	1	•	POLYETHYLENE RESINS/SOLUTION POLYMERIZATION(HDPE)
3008-03	20	5	•	•	POLYETHYLENE RESIN/SUSPENSION POLYM(PARTICLE FORM)(HDPE)
3008-04	D2	15	3	•	POLYETHYLENE RESINS/HIGH_PRESSURE POLYMERIZATION (LDPE)
3006+08	02	2	•	•	POLYETHYLENE RESINS/GAS PHASE POLYMERIZATION (HDPE)
3008-90	3	1	•.	٠	POLYETHYLENE/FINISHING PROCESS
3008-99	02	٠	•	٠	POLVETHVLENE/PROCESS_UNDER_REVIEW
3009-00	02	5	•	•	POLYETHYLENE COPOLYMERS/
3010≠00	· 5	1		•	POLYOXYETHYLENE GLYCOL/ETHOXYLATION OF ETHYLENE GLYCOL

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
3010⇒01	5	2	•	•	POLYETHYLENE GYCOL/ ETHYLENE OXIDE
3011=01	6	2	•	•	POLYETMYLENE POLYAMINES/ETHYLENE DIAMINE + EDC + NH3
3013=01	01	2	•	•	POLYMERIC METHYLENE DIANILINE/REACTION OF ANIL & FORMALM
3015=01	▲	2	•	•	POLYMERIC METHYLENE DIPMENYL DIISOCYANATE/FROM POLY METH
3015+02	▲	1	•		POLYMERIC METHYLENE DIPHENYL OIISOCYANATE/METHYLENE DIAN
3015+80	<b>A</b>	1	•	•	POLYMERIC METHYLENE DIPHENYL DIISOCYANATE/RXN OF FORM+AN
3020=01	23	3	•	•	POLYPROPYLENE /FIBER BY MELT SPINNING FROM PURCHAS RESIN
3020#03	02	5	1	•	POLYPROPYLENE/RESIN BY SOLUTION POLYMERIZATION
3020=04	3	•	•	•	POLAPHOPALENE/POLAMER EXTRUSION
3020=05	D2	2	•	•	POLYPROPYLENE/RESIN BY SUSPENSION POLYMERIZATION
3020=06	20	1	•	•	POLYPROPYLENE/GAS PHASE POLYMERIZATION
3020-90	3	•	•	•	POLYPROPYLENE/FINISHING PROCESS
3025-01	5	4	•	•	POLYOXYPROPYLENE GYCOL/ REACT OF PROPYLENE GYCOL & PROPYLENE OXIDE
3025+02	5	2	•	•	POLYOXYPROPYLENE GLYCOL/PHOPOXYLATION OF GLYCEPINE
3030-02	D1	5	٠	•	POLYSTYRENE + COPOLYMERS/SUSP POLYMERIZATION WHO RUBBER
3030-03	50	6	2	•	POLYSTYRENE & COPOLYMERS/HULK POLYMERIZATION WITH RUBBER
3030-04	D2	•	4	•	POLYSTYPENE + COPOLYMERS/BULK POLYMERIZATION W=O RUBBER
3030-06	02	•	•	٠	POLYSTYRENE AND COPOLYMERS/
3030=90	3	1	٠	٠	POLYSIYPENE + COPDLYMERS/FINISHING PROCESS
3031=01	DZ	•	1	•	POLYSTYPENE DEADANDED/POLYMERIZATION OF POLYSTYPENE
3033+99	DI	I	•	•	POLYSULFONE RESINS/FROM SODIUM RISPMENOLATE
3036-00	01	1	•	•	POLYURETHANE PESINSA POLYOLS + DIISOCYANATE
3036=01	<u>01</u>	5	17	•	POLYURETHANE RESINS/ POLYOL + DIISOCYANATES
3036-02	3	•	1	•	POLYURETHANE RESINS/COMPOUNDING
3040=01	DI	6	2	•	POLYVINYL ACETATE RESINS/EMULSION POLYMERIZATION
3040=03	<u>D1</u>	1	•	•	POLYVINYL ACETATE PESINS/SOLUTION POLYMEPIZATION
3042-01	E	2	•	•	POLYVINYL ALCOHOL/HYDROLYSIS OF POLYVINYL ACETATE
3042408	DI	1	•	•	POLYVINYL ALCOHOL/PESIN-SOLUTION POLYM (METHANOL) OF VINYL ACETATE, HYDROLYSIS OF POLYME
3042-80	D	•	•	•	POLYVINYL ALCOHOL/RESIN-SOLN POLYM (METHANOL) OF VINYLACET
3043=01	01	1	2	٠	POLYVINYLACETATE+ACRYLIC COPOLYMERS/LATEX=EMULSION POLYM
3044#03	DI ·		•	•	POLYVINYL HUTYRAL/POLYVINYL ACETATE AND BUTYRALDEHYDE
3044499	01	1	•	٠	POLYVINYL BUTYRAL/PROCESS UNDER REVIEW
J045-01 J045-03	01	3	:	•	POLYVINYL ACETATE & PVC COPOLYMERS/SUSPEN POLYMERIZATION
3043402	DI	•	1	•	POLIVINUL ACETATE & PVC COPULYMENS/SOLUTION POLYMENIZATI
3045-03	01	1	•	٠	POLYVINYL ACETATE & PVC COPOLYMERS/EMULSION POLYMERIZATI
3045=04	01	1	•	•	POLYVINYL ACETATE & PVC COPOLYMERS/STAPLE FIRER FROM HES
3047401	DI .	1	•	•	COPOLTMERS OF POLYVINYL ACETATE/EMULSION POLYMERIZATION
3047402 3047402	01	1	•	•	POLIVINYL ACETATE COPOLYMERS/SOLPOLY WT VIN PYROLLIDINON
3047403	DI	1	٠	٠	CONOTHERS ON NOTATINAT VCELVLGNOTANEL ALLENG
3048-01	02	7	٠	•	PULYVINYL CHLOHIDE/EMULSION POLYMEHIZATION
2040405	DZ	16	٠	٠	POLYVINYL CHLOFIOE/SUSPENSION POLYMERIZATION
J048-04	02	3	٠	•	POLYVINYL CHLORIDE/BULK POLYMERIZATION
3048-11	3	٠	٠	•	POLYVINYL CHLOHIDE/FILM OR FIBER BY CALENDERING

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FREQUENCY OF OCCURRENCE FOR EACH PRODUCT PROCESS

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3048-90	3	٠	•	•	POLYVINYL CHLORIDE/FINISHING PROCESS
3050-01	· 0	1	1	•	PROPIONALDEHYDE/HYDROFORMYLATION OF ETHYLENE=OXO PROCESS
3052+01	02	1	•	•	POLYVINYL CHLORIDE COPOLYMERS/SUSPENSION POLYMERIZATION
3055+99	·· 01	1		•	POLYVINYL PYRROLIDONE/POLYMERIZATION OF VINYL PYRROLIDONE
3060=01	502		i		POLYVINYLIDENE CHLORIDE/EMULSION POLYMERIZATION
3040-99	02	ī		•	POLYVINYLIDENE CHLORIDES/PROCESS UNDER REVIEW
3063-01		-			PROPANE/REFINERY BY-PRODUCT
3063=02	ž	ż			PROPANE/NATURAL GAS BY-PRODUCT
3063-03			-		PROPANE/BUTANE PYROLYSIS
3066=01	Ċ	i	ī		PROPIONIC ACID/AIR OXIDATION OF PROPIONAL DEHYDE
3066-02	ř		2	-	PROPIONIC ACID/CO-PRODUCT OF BUTANE OXIDATION
3066-04	č		1		PROPIONIC ACID/AY-PRODUCT OF NITROPARAFFINS
3068=01	Ă	ī	i		N-PROPYL ACETATE/REACTION OF ACETIC ACTO & N-PROPANOL
3070-01	<b>F</b> 1		i	•	
3090-02	<b>.</b>	10	•	•	PRODU ENFINENCE OF ETHANE/PROBANE/RUTANE/ PRO
3090-04		Б	ĩ	•	PRODULERE/PUPOLISTS OF NADATHA AND OD GAS OT
3090-08	2	1	•	• .	DOON'S ENF FORTIONATION OF DESTINGUE AND OF GED OF
3000-11	E	2	•	•	DODALENE/FREGIAMETION OF HEFINEN LIGHT ENDS
3100-01	20	1	•	•	PROFILERET FINELISS OF NERSTANET AND ADAMANANA
31100001	20	;	•	•	PROPUENCE DICHIADIAKININICHACIINICI IN CELUNIATANA PROC
3110-01	20	;	•	•	PROFILENC DICHLORIDE/CRU-RADAT OF DDORY OVIDE AV AN ADDUVD
3111-01	20		٠	•	
2111-01	Ľ	3	•	•	PROPILENE OLIVOLATIVOLITII OF PROPILENE UNIDE
3120-02	HT	3		•	PROPYLENE UNIDESFRUM PROFYLENE VIA CRUNNINUMIN
3132401	м	•		•	PTWOLTSIS GASULINE/CRACKING ETHANE (PROPANE (BUTANE)LPG
3133-03		1	•	•	PYHOLYSIS GASULINE/CHACKING ETHANE/PHUPANE/BUTANE AND NA
3142401	Ę	2	•	•	NATON'A 13COSE MAGESS
3170401		•	٠	•	SALICYLIC ACID/CARBOATLATION OF DAY SOUTH PRENATE
3170444	C	1	٠	•	SALICYLIC ACIU/LARDUNUAYLATIUN UP PRENULATE
3176401	01	•	:	•	SAN RESINFUSPENSIUN POLITRENIZATION
3172-02	01	1	1	•	SAN HESIN/MASS POLYMENIZATION
3175401		3	•	•	SILICONE SYSTELCONE MUMMERS (CHECKOSIEANES) CHECKINATION OF SILICON DIONIDE
317502	Ľ	•	•	•	SILICONESSILICONE FLUIDS (NYDHOLYSIS AND CYCLIZATION)
3175-03	E	1	1	•	SILICUNESSILICUNE RESINS
3175404	Ę	•	٠	•	SILICONES SILICUNE HUNBERS
J175=05	E	•	٠	•	SILICONES/ELASTONER PRODUCTION
3175=06	E	Z	•	٠	SILICONEST SILICONE SPECIALTIES (GREASE, DISPERSION AGENT
J181+01	Z	1	٠	•	SODIUM BENZOATEZNEUTRALIZATION OF RENZOIC ACID
3200-99	0	3		•	SODIUM FUNMATE/ FUNMIC ACID + CAUSTIC
3230401	Je	7	2	•	SITHENE ULENTINUGENATION OF ETHTLBENZENE
3230=02	2	1	•	•	STYPENE/SEPARATION FROM PYPOLSIS GASOLINE
JZ35401	20	5	1	•	STYHENE-HUTADIENE RESIN/ EMULSION PROCESS
3236=01	DI	1	•	٠	STYPENE MALEIC ANNYORIDE RESINS/COPOLYMERIZATION OF STYRENBMAL.ANH.
3237#01	02	1	٠	٠	STYRENE=METHYL METHACRYLATE COPOLYMER\$/SUSPENSION PROCES

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3251-01	M	•	•	•	SULFANILIC ACIO/SULFONATION OF ANILINF
3260-99	16	i			SULFOLANE / BUTANE + SULFUR
3280-01	C	6			TEREPHTHALIC ACID/CATALYTIC OXIDATION OF P-XYLENE
3280-03	E	1	•		TEREPHTHALIC ACID/HYDROLYSIS OF DIMETHYL TEREPHTHALATE
3286=01		ī		-	1234-TETRACHLOROBENZENEZAY-PRODUCT BENZENE CHLORINATION
3287-01	8	ī	•		1245-TETRACHLOROBENZENE/AY-PRODUCT BENZENE CHIORINATION
3288=01	A	ī			1235-TETRACHLOHORENZENE/BY-PRODUCT BENZENE CHLORINATION
3291-01	8	ī		-	1.1.2.2-TETRACHI OROFTHANF/CHI ORINATION OF FTHULFNF
3295=01	S	-			TETRACHLORGETHYLENE/ OXYCHIORINATION OF HYDROCARBONS
3295-02	A	Š			TETRACHLOROETHYLENE/CHLORINATION OF FOC & OTHER CHLOR HC
3295=04		2	i		TETRACHLOROETHYLENE/CHLORINATION OF HYDROCARBONS
3300+01	B	ī	-		TETRACHLOROPHTHALIC ANNYDRIDE/CHLORINATN OF PHIMALIC ANNYDRIDE
3307-01	6		i		TETRAETHYLENE PENTAMINE/FIHYLENE DIAMINE + FOC + NH3
3310=01	12	4	-		TETRAETHYL LEAD/ALKYL HALTDE + SODTUM-IFAD ALLOY
3312-01	P	1			TETRAFI UORODICHI OROFIHANEZHYDROFI UORIN OF TETRACHI OROFINYI ENF
3315=99	F1	ĩ	•		TETRAHYDROFURAN/ HYDROGENATION OF MALFIC ANHYDRIDE
3325-01	5	ī			TETRAFTHYLENE QUYCOL/COPDCT OF FTHY GLYCOL FROM FTHYLENE
3325-02	5	-	ī		TETRAETHYLENE GLYCOL/FROM FTHYLENE GLYCOL STILL BOTTOMS
3338-01	12	3	-		TETRAMETHYL LEAD/ALKYL HALIDE + SODIUM-IFAD ALLOY
3349-00	H H	ī	ĩ		TOLUENEY STEAM PYROLYSIS OF LPG
3349=01	2	3	-		TOLUENE/DIST OF BTA EXTRACT-CAT REFORMATE
3349-02	ž	-			TOLUENEZDIST. OF BIX EXTRACT-COAL TAR LIGHT OT
3349=04	J2 .	4	i		TOLUENE/BY-PRODUCTOF STYRENE MEG
3349=07	້	6	ī	-	TOLUENEZDIST OF BTH EXTOPYROLYSTS RASOLINE
3350+01	F2	Ă.		-	2.4-TOLUENE DIAMINE/CATALYTIC HYDROGENATION OF DINTIBOTOLUENE
3351=01	F2	1	•	-	TOLUENE DIAMINE (MIXTURE)/CATALYTIC MYDROGENATION OF DINITROTOLUENE
3354+01	v -	3			2.4-TOLUENE DIISOCYANATE/PHOSGENATE OF 2.4-TOLUENE DIAMI
3355=01	v				TOLUENE DIISOCYANATES (HIXTURE) / PHOSOFNATION OF TOLUDIAMS
3360-99	6	•			TOLUENESULFONAMIDE / TOLUENE SULFONYL CHIORIDE ANH3
3380-99	ĨŇ.	•			TOLUENESULFONYL CHLORIDEZ TOLUENE SULFONTC ACTD + CHLOROSULFONTC ACTD
3381-01	72	i i			TOLUIDINES/CATALYTIC REQUCTION OF Q=NITROTOLUENE
3390-01	8	ī			TRICHLOROBENZENE/BY-PRODUCT OF BENZENE CHLORINATION
3392=01	9	ī			1.2.3-TRICHLOROGENZENEZAY-PRODUCT OF BENZENE CHIORINATION
3393+00	4	:	ī		1.2.4. TRICHLOPORENZENE/
3393-01	P,	i	•		1-2-4-TPICHLORORENZENE/CHLORINATION OF 1-4-DICHLORORENZ.
3393=02	A				124 TRICHLOROBENZENE AVERODUCT OF BENZENE CHIORINATION
3394-01	A	i			135-TRICHLOROBENZENF/RY-PRODUCT OF RENZENF CHLORINATION
3395+01	8	i			1.1.1 TRICHLOPOFTHANE/ CHIORINATION OF FINYLENE DICHLOPD
3395-03	ē	i			1.1. TETCHIOFTHANF/CHIORTNATION OF FTHANF
3395-04	P	ī	•	-	1.1.1.TRICHLOROETHANEZHYDROCHLORINATION OF VINYL CHLORIDE
3400-01	•	2		•	1.1.2 TRICHLOROFTHANE/CHLORINATION OF VINYL CHLORIDE
3400-02	B	-			1.1.2-TPICHLOROFTHANEZCHIORINATION OF FTHVIENE DICHLORID
3400=03	R	•		•	1.1.2-TPICHLOPOTHANE/AVPORTATION OF VINYI CHLOPIDE MANUSACT
	-	-		•	TATES TO BE STUDIED INC. A TILE CUPOLIDE UNINARY

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
3410-02	\$	•	•	•	TRICHLOROETHYLENE/OXYCHLORINATION OF HYDROCARBONS
3410=03	A	2	•	•	TRICHLOROETHYLENE/CHLOR. OF EDC AND OTHER CHLORINATED HC
3411=01	P	4	1	•	TRICHLOROFLUOROMETHANE/FLUORINATION OF CARBON TETRACHLOR
3415=01	Ą	1	•	•	2+4+6=TPICHLOPDPHENOL/CHLORINATION OF PHENOL
3430-01	P	1	•	٠	1.1.2=TRICHLORO=1.2.2=TRIFLUOROETHANE/COPROD OF 3312=01
3460=00	5	1		•	TRIETHYLENE GLYCOL/ETHYLENE GLYCOL + E.O.
3460-01	5	8	1	•	TRIETHYLENE GLYCOL/COPR.OF ETHYLENE GLYCOL FROM ETH OXD.
3460-02	E		•	•	TRIETHYLENE GLYCOL/COPROD OF HYDROLYSIS-ETHYLENE OXIDE
3460-03	5		1		TRIETHYLENE GLYCOL/FROM ETHYLENE GLYCOL STILL BOTTOMS
3475=01	ĸ		1		TRIETHYLENETETRAMINE/AMINATION OF ETHYLENE OICHLORIDE
3477-01	P	ī	•		TRIFLUORODICHLOROETHANE/HYDROFLUORINAT OF TETRACHLOROETHYLENE
3487=01	5	ī	•		TPIPROPYLENE GLYCOL/RXN OF PROPY GLYCOL + PROPY OXIDE
3488=00	<b>A</b>	1	•	•	2,2,4-TRIMETHYL-1,3-PENTANEDIOL/ ALDOL CONDENSATION ISOBUTYPALDEHYDE
3500-99	3	•	2	• •	UREA/ NH3 + CO2
3501=00	D	•	•	•	UNSATURATED POLYESTER RESINS/
3501=01	D1	3	12		UNSATURATED POLYESTER RESIN/REACT MALEIC/PHTHALIC/GLYCOL
3503-99	01	•	1	•	URETHANE PREPOLYMERS/PROCESS UNDER REVIEW
3506-00	02	•	•	•	UREA RESINS/GENERAL
3506=01	01	13	22	•	UPEA PESINS/POLYCONDENSATION OF UREA WITH FORMALDENYDE
3510+01	G	1	1	•	VINYL ACETATE/LIQUID PHASE ETHYLENE & ACETIC ACID
3510-03	13	•	1	•	VINYL ACETATE/ACETYLENE + ACETIC ACID
3510-05	G	1	2	٠	VINYL ACETATE/VAPOR PHASE RX OF ETHYLENE & ACETIC ACID
3520=03	R	6	1		VINYL CHLORIDE/THERMAL CRACKING OF ETHYLENE DICHLORIDE
3520-80	R	2	•		VINYL CHLORIDE/FROM ETHYLENE VIA EDC RY CHLORHOXY CHLOR
3530-02	P	2	•		VINYLIDENE CHLORIDE/DEHYDROCHLOR. OF TRICHLOROETHANE
3540-01	D <b>2</b>	•	1	•	VINYL TDLUENE/POLYMERIZATION
3541=01	17	2	•	•	XYLENES=MIXED/BOTTOM BTX EXT=PYROLYSIS GASOLINE
3541=03	17	2	٠	•	XYLENES®MIXED/BOTTOM BTX EXTRACT-CAT REFORMATE
3541=04	H	•	•	•	XYLENES+MIXED/BOTTOM BTX EXTRACT-COAL TAP LIGHT OIL
3541=08	17	1	•	•	XYLENES+MIXED/M+P+=XYLENES++BOTTOMS XYLENE SEPARATION
354]=09	2	•	•	•	XYLENES+MIXED/CRUDE P+XYLENE BY ISOMERIZATION OF COIS
3550-02	· 2	٠	٠	•	M-XYLENE (IMPURE)/FRACTIONATION OF MIXED XYLENES
3560-01	2	3	•	•	O-XYLENE/DISTILLATION FROM MIXED XYLENE
3570-01	17	1	٠	•	P-XYLENE/CRYSTALIZATION FROM MIXED XYLENE
3570-02	17	2	•	•	·P=XYLENE/ISOMERIZAT=CRYSTALLIZAT OF MIXED XYLENES
3570+05	2	2	•	•	MIXED XYLENES/ FROM BIX EXTRACT
3580-01	н	•	٠	•	XYLENOL • MIXED/TAR ACID RECOVERY AND REFINING
3587-00	M	1	•	• '	XYLENESULFONIC ACID: SODIUM SALT/
3587-99	M	2	٠	٠	XYLENESULFONIC ACID, SODIUM SALT/ SULFONATION OF XYLENE
3600-99	3	•	. •	٠	ZINC(TOTAL)/PROCESS UNDER REVIEW
9601-00	3	•	•	•	AMMONTUM CHLORIDE/
9601-01	3	•	٠	٠	AMMONIUM BICARBONATE/
9603=00	M	1	٠	٠	CYCLOHEXYL MERCAPTAN/ DODECENCNE + HZS

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
9604-00	01	•	•	•	DH - SOB RAYON BRADE/
9604-01	16		i	•	2.4-DIMETHYL PHENOL/ EXTRACTION, DISTILLATION OF REFINERY SPENT CRESYLATES
9608-00	M	i	•	•	NoHEXADECYL MERCAPTAN/ OLEFIN +H2S
9608=01	M	i	•	•	N-HEXYL MERCAPTAN/ OLEFIN + H25
9612-00	M	1		•	D-LIMONENE DIMERCAPTAN/ OLEFIN + M25
9615=00	1		i	•	POTERTOCTYL PHENOL/
9616=00	Ď1	1	•	•	POLYAMIDE RESINS/
9616=01	3	•	•	•	PHOSPHORIC ACID/
9619-00	Ĵ				SODIUM NITRATE/
9619=01	3	•	•	•	SODIUM BICAPRONATE/
9619=02	3	•	1	•	SONIUM MYDROSULFIDE/
9619-03	3	•	•	•	SODIUM SULFINE/
9619=04	E	2	•	•	SODIUM SULFATE/RECOVERY AS PART OF VISCOSE PROCESS
9619=05	3		•	•	SODIUM TETRASULFIDE/
9626=00	•	•	•	•	
9801-01	н	2	•	•	ACETYLENICS/ STEAM PYROLYSIS
9801-02	3	1	ĩ		AGPICULTURE CHEMICALS/
9801-04	н	1	1	•	AROMATIC CONCENTRATE/ STEAM PYROLVSIS OF CRUDE OIL CUTS. OR LPG
9801-05	01	2	•	•	ACRYLAMIDE RESINS/
9801-06	0	3	•	•	GLYCOLS, MIXED/ OLEFIN OLIGOMERS + OXO, MYDROGENATION
9801-09	16	•	•	•	AROMATIC SOLVENTSZ BTX EXTRACTION FROM PYROLYSIS BASOLINE
9801-11	6	1		•	ANTIOXIDANTS/
9801=12	3	•	•	•	ASPHALT/REFINERY PRODUCT
9801-13	2	•	•	•	ALIPHATIC SOLVENTS/ DISTILLATION FROM PYROLYSIS GASOLINE
<b>9801=14</b>	6	1	•	•	ACCELERATORS(TRADE NAME)/ MERCAPTOBENZOTHIAZOLES
9801-15	6	1	•	•	ANT IOZONANTS/
9801=16	01	' 1	•	٠	ACETAL RESIN (CELCON)/
9801-18	5	1	•	•	MISCELLANERUS ALKOXYLATES/
9801-19	6	2	1	•	MISCELLANEOUS AMIDES/ FATTY ACIDS + ALKANOLAMINES
9801-21	τ	2	•	•	ANTI KNOCK BLENDS/ LEAN ALKYLS. ETHYLENE DICHLORIDE
9801-23	н	1	٠	•	AROMATIC TAR/ STEAM PYROLVSIS RESIDUALS
9801-24	H	1	•	•	AROMATIC DISTILLATES/ BTX
9801-25	6	2	•	•	AMINES /METHANOL + NH3
9801-26	I	1	•		ALUMINUM ALKYLS/ OLEFIN + ALUMINUM + MYDROGEN
9801-27	3	1	1	1	ADHESIVES/

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	9801-29	FZ		1	•	MISC. ALKANES/
	9801-30			i		ALKENYL SUCCINIC ANHYDRIDES/
	9801-31	5				ALKATERGE (TRADE NAME) / ETMOXYLATION OF ALKYL PHENOL
	9801=34	A				AMANTADINE HYDROCHLORIDE/
	9801+35	D1		i		ALKYD MOLDING COMPOUND (TRADE NAME)/
	9801-36	•				ALPHA 841 PRODUCTS (TRADE NAME)/
	9801+40			ĭ		AMING ALCOHOLS/ ALDOL CONDENSATION OF NITROPARAFFINS WITH FORMALDEHYDE, REDUCTION
	9801-42	3	-	i		ACRYLONITPILE CATALYSTY BISHUTH PHOSPHOMOLYBDATE. SA-U207-SILICA GEL
	9801-43	6	i			ALKYLOLAMIDES/ FATTY ACIDS + ALKANOLAMINES
	9801-47	3				ANTIBIOTICS/
	9801+48	<b>D</b> 1				ACETONE FORMALDEHYDE RESINS/
	9801-49	•				ADDITIVES MISCELLANEOUS/
	9801+51	1.	ī			ALKYL VINYL ETHERS/ ACETYLENE + ALCOHOL
	9801-53	•				AMEX (TRADE NAME)/
	9801-58	F2				ACETYL P-ANINOPHENOL (P-HYDROXYACETANILIDE) / NITRATION, HYDROGENATION ACETYLATION OF PHENOL
	9802+04	12	i	•	-	N-BUTYL FORMOFIZ SOLUTION OF FORMALDENYDE IN BUTANOL
	9802-05	1	•	•	-	
	9802-11	i				
	9802-12	i	;	•	•	
Э	9802-13	1	1	•	•	
	9802-14		;	•	•	
8	9802-16		:	•	•	DENLETEZ BIELDAALAMINARVALAHEYNI.MEYNÄNEZ ANITINE A FAANN DENVAR, HYDAARNAYAN
	9002019	<b>.</b>		•	•	DISTRATE BODDIETS/
	9002-14	3	•	٠	•	BULLIVALIA FRUDULIJA BULLIVALIA FRUDULIJA
	9002010	6	:	٠	•	
	9002-21		ļ	•	•	NEGOINT MERCAPIANY ULEFIN & MES
	9802-13			٠	•	- Current MERCARANA CIEFIN & NEG
	7002423		1	•	•	THOUTTE HERCARIANY VERTIN - NES
	9802-34	<b>U</b> .	;	•	•	BATTA REJIM (FULTAL TLUNI) HILE BANAIR REJIN) Batta Rejim (Fultal Tluni) a footaal deuvor
	9602038	-	3	•	•	DUITMEDIDE/ ACTIVILEME & FORMALDENTOE 1.4 Duitmedide/ Activileme attain activileme utta formal deuvor, foi longo av uvororenation
	9802-34	<b>.</b>	E 1	•	•	14 BOLMANDIOL/ CONDENSATION ACCITENCE WIN FORMALDENTIDE FOLLOWED WY WYDROGENATION
	9802438	J		•	•	SUBULTRUE LIGNEZ DENTURGENATION OF INTRODUCATEDIOL (COPPER CAT)
	9807-02		1	•	•	Carling and a start of the star
	9803-04			•	•	CADBON DIACK EFFDERENT DISTILATION FROM BURDING AND AND INF
	9803-05	2	•	:	•	CARDON BEACH FEEDSINCH OF USILEAINA FAN FINDESIS BASULINE
	9803-04		٠		•	CHEVROBOLAUTENEZ CHEVRINATION DENTUNUCHUN MATION OF HUTADIENE 1- (1-CHU OBOLI VI 1-1-E-TOLIA 1-1-1-TONTADIANANTANE CHU ODIDE 1-1-1-DICHI ODODDOBNESE DOMAI DENVOESAMMONTA
	9803-11	-	•	1	•	Let An OUT ANE DATA TO A DATA AND A THE TEACH TANDALAN AND A CHECK THE TANDAL CHECK OK OK OK AND AND AND AND AND AND AND AND AND AND
	9803-13	, ,	1	•	•	CLAT FOLTMEN/CLAT FOLTMEN
	980301J	2		:		CUATED FILM/
	9803-14			1	•	CHICONDITINE - CHICONDITINI INI VI JAVICELUN DITATIN CE BITADIENE
	70VJ-17			•	•	CALCHING FORMATES CO. A LANE
	7003410	U A	1	•	٠	
	7403-14	0	٠	٠	•	Creding received a rection of the received and the received a re
	10V]#10	3	•			CICTOSCAINE (# # Padian # 3 # ISNYKRAFININANE)

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9803-19	6	•	•	•	CHOLINE BICARBONATE/ CHOLINE + SODIUM BICARBONATE
9803-22	Ð	1 -	•	•	CHLORINATED WAX/
9803+25	3		2	•	COATINGS/
9803-26	3	•	•		CALCIUM PROPIONATE/
9803-27	<b>A</b>	•	1		CHELATING AGENTS/ EDTA - ETHYLENEDIAMINE + PORMALDEHYDE + NACN
9803-28	ĸ	•	1	•	CLEANING COMPOUNDS/ VEGETABLE OIL SULFONATES. ETHORYLATES. QUATS
9803=39	F1	•			CROTON OIL ALCOHOL/ HYDROGENATION OF CROTONALOEHYDE
9803=42	15		•	•	C=5 UNSATURATES/ FROM PYROLYZATE BY EXTRACTIVE DISTILLATION
9803+45	14	•		•	CHEMICAL COTTON/ CELLULOSE + ETHYLENE CHLOROMYDRIN
9803=46	Ğ	1	•	•	CORPENT/ PENTAERYTHRITOL DERIVATIVE
9803-53	3			•	COAL ASH/
9804=02	6	ī		•	DIAMINO DIPHENYL METHANE/ ANTLINE + FORMALDEMYDE
9804=04	6			•	NINIDITSOPROPYL 2 BENZOTHTAZOLE/DITSOPROPYL ANTINE + 2-MERCAPTORENZOTHTAZOLE
9804-06	4			-	DISTILLATE + LIGHT +N-BUTANF DEHYDRO./RECOV.
9804=08	3	•			DIESEL FUEL/
9804-10	3	i			DIURON (TRADE NAME)/
9804-12	ĸ	ī	•	•	DIMETHYLACETAMIDE/ DIMETHYLAMINE + ACETIC ACID
9804-13	<b>A</b>		•	•	DIMER ACIDS/ FROM PINE ROSIN
9804=14	6		•	•	DIAMINES/
9804=15	0	ī			DIETHYL ANILINE/ ANILINE + FTHANOL + H2SOA
9804=17	4	ĩ	•		DACA/
9804=19	13	•			DIATRIZOIC ACID/
9804-20	Ō	1			DIETHYL MALONATE/ ETHANDL ESTER OF MALONIC ACID
9804-21	G	ī			DIMETHYL MALONATE/ METHANOL ESTER OF MALONIC ACID
9804-22	12	•	•		DENATURED ALCOHOL/
9804+24	Ğ	•	ī	•	DYEING ASSISTANTS/ POLYESTERS
9804-25	ĸ	•	ī	•	DETERGENTS AND SCOURS/ VEGETABLE OIL SULFONATES. FTHOXYLATPS, QUATS
9804-27	01		ī		DIALLYLPHTHALATE HOLDING COMPOUND/
9804-29	M	ī			NUDECYL MERCAPTAN/ OLEFIN + H28
9804+30	M	ī		-	NOTRIODECH MERCAPTANZ OLEETN + H25
9804=31	24		i		DISODIUMETHYLENEDIAMINETETRAACETATEZ NEUTRALIZATION OF FOTA
9804=36		ī			DICHLORODIPHENYL BULFONE/SO3 +THIONYL CL + CHLOROBENZENE
9804-47	9	ī			DIMETHYL BENZYL AL COHOL / ACID CLEAVAGE OF CUMENE HYDROPEROYIDE
9704=48	D1	ī		•	DICYANODIAMINE RESIN/
9804-49	M	ž			DODECYLBENZENE SULFONIC ACTO SALTS/ SULFONATION OF DODECYLBENZENE
9904-52	G				DIBUTYL PHENYL PHOSPHATE/ BUTANOL + PHENOL + POCL 3
9804-53	6				2.6 DIETHYL-N-(METHOXYMETHYL)-2-CHLOROACETANIL 10F/
9804-54	ŏ	•		-	2+6+DIETHYLPHENYL AZOMETHANE/
9804-57	Ŭ	ī	,		DINGNENE/ FROM TERPINENE BY DEALKYLATION
9804-58	5	ī	-		DIETHANOL AMMONIUM LAURYL SULFATE/ LAURYLAMINE + PTMYLENE OXIDE
9805-03	6	-		•	NOETHYL ANILINE/ ETHANOL + ANILINE
9805-04	- -	-		•	2-FTHYL HEXYL CHLORIDE/
9805-05	Ğ	ž			MISCELLANEOUS ESTERS (POSSIBLY ADIPATES) / ADIPIC ACID + 2-FTHYLMERANDI
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9805+08	01	1	•		EASTOBONDS (TRADE NAME) / POYSTYRENES AND CELLULOSICS (ADMESIVES FOR POLYOLEFINS)			
9805-09	24	•	•	•	ETHYLENEDIAMINE - DIHYDROIODIDE/ HI + ETHYLENEDIAMINE			
9805-12	DÌ	ĩ	•	•	ELASTOMER LATEX/			
9805-14	01	•	ĩ	•	EMULSION POLYMERS/			
9805-17	M	i		-	ETHYL MERCAPTAN/ OLEFIN + H2S			
9805-18	14	ĩ			DI-ETHYL SULFIDE/ K ETHYLSULFATE + K2S (AQ)			
9805+19	A .		i		ETHYLENEDIAMINETETRAACETIC ACID/ ETHYLENEDIAMINE + FORMALDEHYDE + NACN			
9805-22	5	ĩ	:		MONDETHYL GLYCOL ETHER/			
9805-29	18	•	ĩ		ETHYLENE CYANOHYDRIN/ ACETALDEHYDE + HCN			
9805+31	Å	i			ETHYLFNE UREA/ ETHYLENE DIAMINE + CO2			
9805=36	01	•	ż		ALKYD. PHENOLIC. POLYESTER. POLYURETHANE RESINS			
9806#01	3	ž	-		FUEL GAS HZ AND CHA (MEG)/			
9806-02	Ā				FUEL ADDITIVES/			
9806-03	3				FLOUR ADDITIVES/			
9806-04	,	ĩ	i		FATTY ACTOS & DERIVATIVES/ HYDROLYSIS OF GLYCERIDES			
9806-05	ŏ		ī		FATTY NITRILES DEHYDRATION OF FATTY AMIDES			
9806-06	3				NUMAER 2 FUEL OIL/			
9806-08	P	i	•		FREON (GENERAL) ZHYDROFLUORINATION			
9806-09	01		i		FLEXTAL/ ALKYD RESINS			
9806-11	P	ī	•		FLUOROCARBON BLENDS/ HE + CHLORINATED METHANE			
9806-12	3		•		FURAZOLIDONEZ REDUCTION OF NITROFURANTOIN			
9806-13	Ă				FLOWCO FAMILY/			
9804-14	01		;	•	FUDFUDAL DESIN (INC. FURAN)/			
9804-15	01		•		FOAM PESTNS (POLYURETMANES)			
9806-17	01	ī	•		FOME=COB/			
9804-19	1	i	•	•	FUNGICIDE AND INSECTICIDE/			
9807-01		•	•	•	SENERGI 100 EXTRACTION			
9807-02	1	•	•	•	GENEROL 105 FLAKING/			
9807+04	ž	i	•		GLYCOLONITRILF/ ETHYLENE CHLOROHYDRIN + NACN			
9807-04	<b>'</b>	i	•	•				
9807=07	14	ĩ	•		GASOLINE BLEND STOCK / FROM PURDLUZATE BY EXTRACTIVE DISTILLATION			
9807-08	<b>n</b> 1	i	•	•	GANTHEZ ANZ POLYMERIZATION OF VINYL PTHER AND ACRYLONITRILF			
9807=09	Ğ.	i	•		GANTREZ ESTERIFICATION PROPAGYL ALCOHOL + ACID			
9807-12	02	i	•		POLYMER GASOLINE/RY-PRODUCT ISORUTYLENE			
9807-13	5	i	•		GYCOLS (HISC.) / ALCHOHOL . GLYCOL . ETHYLENE OXIDE			
9808=02	Å	•	ī		HEXAMETHYLENEDIAMINE / 1.6 - HEXANEDIOL + AMMONIA (NICKLE CAT.)			
9808-05	1	i	•	•	HYDROGEN SIII FIDE/			
9808008	12	i			HYDROXYACETIC ACID/ CHLOROACETIC ACID + CAUSTIC			
9804=09	02	i			HYPOLON(TRADE NAME) / CHLOROSULFONATED POLYETHYLENE PUBBER			
9808-10	02	i	•		HYDAN (TRADE NAME)/			
9408=11	F	•	•		HYDROLYZED VEGETABLE PROTEIN/			
9808-12	1		•					
0000012	,	• •	•	•	WORDAY STRARTC ACTD AND DERIVATIVES/			
20/0-10		•	•	•	HIGHORY BIERLA HELD BUD DEUTAHITAER.			

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9808-17	3	•	1	•	HYDROXYLAMMONIUM ACID SULFATE/
9808-18	3		1		HYDROXYLAMMONIUM SULFATE/
9808-28	н	Ż	•	•	LIGHT HYDROCARRONS/
9808=31	м	Ī	•	•	NYDROTROPE/ SULFONATION OF ALKYL BENZENES (ALKYWMETNY) ETNYLIISOPROPYL)
9809=01	A	•	1	•	ISOPHOPYL ETHYTHIONO CARRAMATE/ ALPHASMETHYL PROPIONAMIDE + SODIUM ETHYLMERCATIDE
9809=03	D1	i	•		TON EXCHANGE RESINS (PROBABLY ACRYLIC RESINS)
9809=04	3			•	ISOMERASE/
9809-07	G		i	•	ISOPHTHALATE ESTER/ESTERIFICATION
9809-11	v	Ż			MISCELLANEOUS ISOCYANATES/ PHOSGENTATION OF ANTI INF-FORMAL DEWYDE DERIV.
9909-19	F1	1	ī		ISOBUTYPONITRILE/ FROM ALPHA-METHACRYLONITRILE BY HYDROGENATION
9810-01	3				JFT FUFL JP-4/
9811-01	6			-	KETONE PEROXIDE (DIACETONE AL COHOL PEROXIDE) / PEROXIDATION OF DIACETONE AL COHOL
9811-02	3		•	-	KEROSENE/
9811=03	01				KETONE RESINS/
9811-04	01	i			KEVLAR (ARAMID RESIN AND FIBER)/ISOPHTHALOYL CHIORIDE + 1.3-DIANTI INF (TYDICAL) AN ADOMATIC NYLON
9812-04	01				LAMINATING RESINS/ CRESTLIC ACID-PHENOL + FORMAL DEHYDE
9812-05	Ĵ.	i			
9912-07	j				
9812-08	ā	•	i		A COLER ( GENERAL ) /
9812=09	Ă		i		LAUROVIL SARCOSINATE (30 PERCENT SODIUM SALT) / RAPORETNE + LAURAL DEWYDE
9812+10	3				LUBRICANTS / ORGANIC PERIOLIDES. PEROLYCARRONATES. FTC
9812-12	Ř	i			LINE DEGI (TRADE NAME) / ORGANIC PENDATORA, PERDAVCARBONATES, FTC
9A12=13	Â	i			LUPERCO (TRADE NAME) / ORGANIC PERIOXIDES, PEROXYCARGONATES, FTC
9812-14	Ă	i	•		LIDEAL (TRADE NAME)/
9812=15	ž	i			
9812-16	01		ī		ATEX (UNKNOWN TYPE) / POLYVINYLAGETATE
9813=01	10				METHYL FTHYL KETOXIME /METHYL FTHYL KETONE + HYDROXYLANINE (NH20H) 24H25041
9A13=02	6		ī		NALE ACID FSTERS/ESTERIE LCATION
9813=04	24				2-MERCAPTOBENZOTHIAZOU FA ZING SAUTZ ZING OXIDE A 2-MERCAPTOENZOTHIAZOU F
9813-05	24				> HERCAPTOBENZOTHIAZOLE. SODIUH SALT/ CAUSTIC + 20HERCAPTOENZOTHIAZOLE
9813-06	Ā				2+21 METMYLENERTS (A-T-RUTY) -A-FMTYL PUENOL / A-FTMYL-A-T-RUTYVL PUENOL - EODMAL DEMYDE
9813+07	12	i			METHYL FORMEFLY SOLUTION OF FORMALDEHYDE IN METHANOL
9813=08	3				MOTOR GASOLINE/
9813-09	2	i			MINED CA COMPOSINGS/ DISTILLATION FROM BIX RAFEINATE
9813+10	i i	i			MOTOR MIX 1/
9813-11	j	i			
9813-12	Ā	i	i		METHYL MERCAPTAN/ METHYL CHLOPIDE & SODIOM HYDDOGHLEFIDE
9813-13	ä	i	•		HTIMMY /
9813+14	ä	i			
9813-14	01	-	,	-	MOLDING COMPOLINGS/ (POLYACRY) 1C RESINS)
9813-20	01	1			
9A11e22	<u> </u>	;	•.		HETUVI CVANDACETATE/ NETUANDI ESTED DE CVANDACETTE ACTO
9813-23	14		•		METURE COMPARED AND AND AND AND AND AND AND AND AND AN
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	D E	D E	I R	R O	N K	A Constraint of the second sec
	9813-24	6	•			MORPHOLINE DERIVATIVES OF NITRO ALCOHOLS/
	9813-25	02	ī	•	•	MICROTHENE (POWDERED PLOYETHYLENE RESIN)
	9813=26	Z	1	•	٠	METHANE/
	9813-29	6	•	1	•	CASION OIL DENIVATIVES/ (ESTERS)
	9813427		•	1	•	INTELLIAIS ESTERISTERITICATION
	9813-31	Ĝ	-	i	-	MALEANIZED OIL / OIL + MALFIC ANHYDRIDE
	9813=39	G	ī	•		TERANOL BENZOATE/ ESTERIFICATION OF BENZOIC ACID
	9813=41	G	1		•	METHOXYETHYL CARBAMATE/ 2-METHOXYETHANOL + CARBAMOYL CHLORIDE (HCL+UREA)
	9813-43	0	•	1	•	HALEATED OILS / MALEIC ANHOPIDRIDE + OILS WITH HYDROXYL GROUPS
	9813-45	0	1	•	٠	NOTETHYLOZOPYKHOLIDONE/ 30BUTYNOLACTONE + NETHYLAMINE
	9813448	- <b>1</b>	;	•	•	METALL ARTI ALCONDER ALCOL CONDENSATION OF ALCONE
	9813+49	Ă	:			2+2* MFTHYLENERIS (6=T=RUTY) = P=CRESOL )/
	9813=50	A	ī	•	-	METHOXY DINYDROPYRAN/
,	9814=01	<b>A</b>	ī	•	•	NITRILOTRIACETATE/ NH3 + FORMALDEHYDE + NACN
5	9814=02	14	1	•	•	NAPHTHA OXIDE OILS/ SODIUH PHENOLATE + CHLOROGENZENE
	9814=03	<b>A</b>	:	•	٠	41NITRO-ORTHO-XYLENE DIETHYL KETONE BLEND/ MIX OF 2 PRODUCTS
	9814-06	01		٠	٠	PARAFORMALDEHYOE/ POLYMERIZATION OF FORMALDEHYDE (ALDOL: CANNIZARO)
	9814-08	1		٠	•	NURDEL (TRADE NAME)/ETHYLENE O PHOPYLENE COPULYMENS
	9814=11	Ľ	1			NTLON TARMYFROM FURCHASED RESIN 5-Nito0-0-Toluifn Sui Fonte Acto/ Sui Fonation of toluifne, nitoation
	9814=13	ĩ	i			1-NITRO-3,4-DICHLOROBENZENE/ NITRATION OF 0-DICHLOROBENZENE
	9814+15	DI	1		•	NAMEX (ARAMID RESIN, FIBER AND SHEET)/ ISOPHTHALOYL CHORIDE + 1,3-DIANILINE (TYPICAL) AN AROMATIC NYLON
	9814-16	3	٠	1	•	NRD=47/
	9814-17	01	1	•	•	NYLON/DACRON COSPUN FIBER/(DACRON POLYESTER)
	9814=18	02	2	•	٠	NEOPRENE/ (3-CHLORD-1, 3-BUTADIENE)
	7014414 9814-21	-	•	•	•	NII"U AND AMINU ALCUHULY ALDO CUNDENSATION UF NIIHUPARAFFINS WITH FORMALDENDE
	9814022	v	;	•	•	ALIMOVONAMICIAY ISAMAYESEVIN Y METHYI AMINE A DUNGENE A INABUTUNI
	9814=24	Ň	i		-	NONYL MERCAPTAN/ ALPHANOLEFINS + H2S
	9814-25	L	•	ī	•	NITROPARAFFINS/ NITRATION OF METHANE, ETHANE, PROPANE (HIGH TEMP, VAP PHASE)
	9814-26	<b>A</b>	•	1	•	NITROALCOMOLS/ ALDOL CONDENSATION OF NITROPARAFFINS WITH FORMALDEMYDE
	9814=27	3	•	•	•	NUSOL/EXTRACTION FROM SULFITE PULP MILL WASTEWATER
	9814-28	3	1	1	•	NITPOGEN FERTILIZER SOLUTIONS/
	48] <b>4</b> +31	14	٠	•	•	P=NITHOPHENETOLE/ P=NITHOPHENOLATE + ETHYL CHLORIDE

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PP_CODE	GEN_CODE	OTR	ZERO	PP_TEXT				
9814-32	G	•	•	•	NONYLECUMYL PHENYL DIPHENYL PHOSPHATE/ ALKYL PHENOLS, PHENOL + POCLS			
9814-3A	2	•	•	•	SOLVENT NAPHTHA/ DISTILLATION FROM COALTAR CONDENSATE			
9814=39	L	1	•	•	NITROCHLOROBENZENE/ NITRATION OF CHLOROBENZENE			
<b>9815=01</b>	3	•	3	•	MISC. REFINERY DILS/			
9815+02	I 1	•	•	•	OCTYLATED DIPHENYLAMINE/ALKYLATION			
9815-03	8	•	•	•	ORGANIC PEROXIDE/			
9015-08	50	•	٠	•	OLEFINS MIXED/ ETHYLENE OLIGOMERS E			
9815-09	D1	•	1	•	OXAWAXES / RESIDUES FROM OXAZOLIDINES PRODUCTION			
9815-10	3	•	•	•	OXAZOL IDINES/			
9815-12	ε	•	•	•	OCTADECADIENOIC ACID MIX/			
9815-14	M	1	•	•	T-OCTYL MERCAPTAN/ ALPHA-OLEFINS + H2S			
9815+15	м	1	•	•	N-OCTYL MERCAPTAN/ ALPHA-OLEFINS + H2S			
9815-16	4	2	•	•	MISC. ORGANICS/			
9815=19	C	•	•		ORGANIC ACIDS (HIXED)/ OXIDATION OF PROPIONALDEHYDE CONDENSATES			
9815-22	10	•			OXIMES.MISC./ CARBONYL CMPD + HYDRXYL AMINE			
9816=01	D1	i	•		POLY METHYLENE DIPHENYL DIISOCYANATE/(POLYMERIC MDI)			
9816-03	02	2	•	•	ATACTIC POLYPROPYLENE/BYPRODUCT OF 3020-03			
9816=05	0	•	•		PEROXY ESTERS (T-BUTYL ESTERS OF PERRENZOICS PEROCTANOICS PERACETIC ACIDS			
9816=07	6	•	•	•	POLYAMINES/			
9816=08	50	2	•		POLYBUTADIENE RESINS/			
9816-09	D1	2			POLYBUTYLENE TEREPHTHALATE (PBT)/			
9816-11	23	1	•	•	POLYESTER YARN/			
9816-13	3	•	i		PIGHENTED FINISHES/			
9816=14	3	i	•		PARAFFINS			
9816=15	D	•	•	•	POLYETHYLENE FOAH/			
9816=16	01	•	Ż		POLYESTER IMIDE (TML)/			
9816=21	3	2	•	•	POLYPROPYLENE FILM/			
9816-22	02		•	•	POLYELECTROLYTE/ POLYACRYLAMIDE			
9816-23	6	ī			PHENYL GLYCINE/ANILINE + CHLOROACETIC ACID			
9816-25	3	•			PROTFIN GLUF/			
9816-27	3		ż	•	PAINT-GENERAL/			
9816=29	5	i		•	POLYETHYLENE OXIDE/			
9816=31	50	•			PICCOYAR RESIN/			
9816=36	02				PICCO 6100 RESIN/			
9816=37	50	•	•	•	PICCOMER/			
9816#3R	01	i	i		PHENOLIC HOLDING COMPOUND/			
9816=42	50	•	•		POLYETHYLENE XYLENE MIXTURF/			
9816+43	50	•	•		POLYETHYLENE TOLUENE MIXTURE/			
9816-44	15	ī			POLYOLS/ ALKOXYLATION OF FATTY ALCOHOLS. PROPYLENE GLYCOL AND BLYCEDOL			
9816=45	▲	ĩ		-	PHENYLETHYLPHENYL METHANE/			
9816=46	M	ī	•	-	N-PROPYL MERCAPTAN/ ALPHA-OLEFINS + H25			
9816-47	M	ī	-	•	I-PROPYL MERCAPTAN/ ALPHAUOLFFINS + HZS			
9816-4R	₹1	ĭ	-	-	Q-PHENYL PHENOL / FROM DIAFNZOFURAN			
	-		-	-				

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PP_CODE	GEN_CODE	DIR	ZERO	UNK	PP_TEXT
9816=49	24		1		PENTASODIUM DIETHYLENEDIAMINE PENTAACETATE / EDPA + CAUSTIC
9816-50	3	•	•	•	PHENOLIC COATED PAPERS/
9816-51	3	•	٠		POLYESTER COATED PAPERS/
9816-54	3	•	1	•	PLYWODD OVERLAYS/
9816=56	3	•	•	•	PAPER GENERAL/
9816-5A	5	2	•	•	POLYETHER/ PROPOXYLATION OF PROPYLENE GLYCOL
9816=69	01	1	•	•	POLY DIETHYLENE GLYCOL ADIPATE (POLYESTER)
9816=71	5	•	1	•	ETHOXYLATED PHENOLS/PHENOLS + ETHYLENE OXIDE
9816 <b>#72</b>	3	•	•	•	PENICILLIN/
9816=75	G	•	1	٠	PLASTICIZERS (PATHALATES, ADIPATES, SEBACATES) / PHTHALIC ANHYD, + ALCOHOL
9816=77	01	•	2	•	POLYVINYL FORMAL/
9816+80	3	٠	•	•	PARATHIONS/ '
9816-83	01	1	•	•	POLYETHER POLYOL RESINS/
9816-87	ĸ	1	•	•	2=PYRROLIDONE/3=BUTYROLACTONE + NH3
9816 <b>-93</b>	- 11 <b>- 6</b>	٠	•	•	POLY+N (TRADE NAME)/ ORGANO-UREA POLYMER
9816-97	3	•	•	•	POLYETHYLENE COMPOUNDS/COMPOUNDING
9816-9P	3	•	•	•	PLASTIC COMPONENT/
9816-99	3	1	•	٠	PLASTIC BATTERY SEPARATORS/
9817=01	6	٠	1	•	QUATERNARY AMINES/ ALKYL CHLORIDE + FATTY AMINE
<b>9816=01</b>	02	1	•	•	RESIN PRISTILL BOTTOMS FROM STEAM PYROLYSIS OF NAPTHAI GAS OIL
9818-02	н	1	•	•	RESIN OIL (AROMATIC)/STEAM PYROLYSIS OF GAS OIL
9616=05	H	1	•	٠	REACTIVE DISTILLATES FROM STREAM CRACKING/
9818+06	6	٠	•	•	RESINS SOLUTIONS/
9818-07	3	•	•	•	ROSINS AND DEPIVATIVES/
9818=08	01	•	1	•	ROSIN DERIVATIVE RESINS/
9818=09	3	•	1	•	PEADSORBER OFF=GASES/
9818=10	E	•	•	•	RICINOLEIC ACID/FROM CASTOR OIL BY MYDROLYSIS
9818-11	0	٠	•	•	RICINOLEATES AND DERIVATIVES/ RICINOLEIC ACID + ALCOHOL
9818+14	16	•	•	•	RAFFINATE/ BTX EXTRACT OF COAL TAR COWOENSATE
9818-15	01	•	1	•	RURAER RESINS(POLYURETHANE ELASTOMERS)
9818-16		1	•	٠	
9818+17	01	1	•	•	RUBBER+CYCLIZED/
9816 <b>-1</b> 9	01	1	•	•	RESINS (GENERAL)/
9819=01	0	3	1	•	SURFACTANTS/ SULFOSUCCINIC ACID ESTERS = SUCCINIC ACID + ALCOHOL + SODIUM BISULFITE
9819+02	3	•	•	•	SULFAMETHAZINE/ CONDENSATION OF SULFAGUANIDINE AND ACETYLACETONE + SODIUM BISULFITE
9819=04	3	1	•	•	SOLVENT BLENDS, MISCELLANEOUS/
9819=09	5	1	•	•	SODIUM METHYLATE/METHANOL + SODIUM
9819 <b>=1</b> 0	01	1	1	•	SPANDEX FIBERS (85% SEGMENTED POLYURETHANE) / POLYOL + D1150CYANATES
9819=11	M	1	•	•	SODIUM LAURYL SULFATE/ SULFONATION OF LAURYL ALCOHOL
9819=14	M	1	•	•	SODIUM STYRENE SULFONATE/ SULFONATION OF STYRENE
9819-15	4	1	•	•	SPECIALTY LUBRICANTS/
9819+17	24	•	•	•	SODIUM PROPIONATE/ PROPIONIC ACID + CAUSTIC
9819 <b>-</b> 18	<b>A</b>	1	2	•	TEXTILE SOFTENERS/ UREA + FORMALDENYDE + GLYOXAL

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	FREQUENCY	OF	OCCURRENCE	FOR	EACH PRODUCT	PROCESS
75 RQ	UNK	PP	TFXT			

0	PP_CODE	BEN_CODE	DIR	ZERO	UNK	PP_TEXT			
0819-22       0	9819=19	G	•	•	•	MISCELLANEOUS STEARATES/ STEARIC ACID + ALCOHOL			
<pre>9819-26 H i</pre>	9819-22	G	•	•	•	SYNTHETIC LURRICANTS/			
9819-27       H       LUKYL SULFONATES/ SULFONATES/ SULFONATES         9819-28       H       LINKAR ALKYLATE SULFONATES/ SULFONATES/ SULFONATION OF ALKYLBENZENES         9819-35       SOLVENT HASE COATING/       SULFONATES/ SULFONATES/ SULFONATES/ SULFONATION OF ALKYLBENZENES         9819-36       SOLUTIONS (HISCELLANEOUS)/       SOLUTIONS (HISCELLANEOUS)/         9819-37       SOLUTIONS (HISCELLANEOUS)/         9819-38       SOJUW NITROBENZENSULFONIC ACID/         9819-40       ZA       SOJUW NITROBENZENSULFONIC ACID/         9819-43       SUSTITUEO PHENDLS (HISCELLANEOUS)/         9819-43       SUSTATIONED FLANAMEDE TRANDED PUSTIMENE         9820-03       C       TRIMELLITIC ANHUDEDS TRANDED PUSTIMENE (1.2.4-FRIMETHYL BENZENE)         9820-03       C       TRIMELLITIC ANHUDEDS (HISCELLANEOUSLE)         9820-04       I       SUSTATION OF PSEUDOCUMENE (1.2.4-FRIMETHYL BENZENE)         9820-05       I       TRIMELLITIC ANHUDEDS (HISCE)/NITANE SULFONUE (1.2.4-FRIMETHYL BENZENE)         9820-06       I       TRIMETHYLOLEPHANE/ PODMOLS (HISCE)/NULLENTLE + FORMALDEHYDE (ALDOL, CANNIZAND)         9820-07       I	9819+26	Ň	i	•	•	FATTY ESTER SULFONATES/ SULFONATION OF FATTY ACIDS			
9919-28 H 1 . LINEAR ALKYLATE SULFONATION OF ALKYLBENZENES 9919-28 J . SOLVENT HASE COATING/ 9919-35 J . SOLVENT HASE COATING/ 9919-35 J . SOLVENT HASE COATING/ 9919-35 J . SOLVENT HASE COATING/ 9919-36 J . SOLVENT HASE COATING/ 9919-36 J . SOLVENT HASE COATING/ 9919-37 H . SOLVENT HASE COATING/ 9919-38 J . SOLVENT HASE COATING/ 9919-39 H . SOLVENT HASE COATING/ 9919-39 H . SOLVENT HASE COATING/ 9919-30 L . SOLVENT HASE COATING/ 9919-30 L . SOLVENT HASE COATING/ 9919-30 L . SOLVENT HASE COATING/ 9919-30 L . SOLVENT HASE COATING/ 9919-30 L . SOLVENT HASE COATING/ 9919-30 L . SUBSTITUTEO PHENOLS (HISCIN/HITRATION OF PALVELENE (1.2.4-FRIMETHYL BENZENE) 9920-03 L . SUBSTITUTEO PHENOLS (HISCIN/HITRATION 9920-04 L . SUBSTITUTEO PHENOLS (HISCIN/HITRATION 9920-05 A	9819-27	M	1	•		LAURAL SULFONATES / SULFONATION OF DDDECENE			
0A19-35       3       1       SOLVENT MASE COATING/         0A19-36       02       1       SVNTHETIC SPECIALTY POLYMERS (POLYSTYRENE)         0A19-36       4       1       SOLUTIONS (MISCELLANEOUS)/         0A19-36       4       1       SOLUTIONS (MISCELLANEOUS)/         0A19-36       4       1       SOLUTIONS (MISCELLANEOUS)/         0A19-40       24       5       SOLUM HIRAF MOLDS (MISCAL/NEWE SULFONATE/ SULFONATION OF ALKYLBENZENE         0A19-43       3       1       SPECIALITY PAPERS/       ONLOTATION OF DATANDO         0A19-43       1       SUBSTITUTEO PHENOLS (MISCA)/WITRATION       FORMALDEMYDE (ALDOL, CANNIZARO)         0A20-05       1       TRIMELLITIC ANMYDRIDEY (SUFONL) PYRIDINEY       BERZENE)         0A20-05       1       TRIMELLITIC ANMYDRIDEY (SUFONL) PYRIDINEY         0A20-05       1       TRIDANEY FORMALDEMYDE TAIMER         0A20-05       1       TRIMELLITIC ANMYDRIDEY (SUFONL) PYRIDINEY         0A20-05       1       TRIMELLITIC ANMYDRIDEY (SUFONL) PYRIDINEY         0A20-05       1       TRIPALIX THURAR MADSULTIOK         0A20-05       1       TRIPALIX THURAR MADSULTIOK         0A20-05       1       TRIPALIX THURAR MADSULTIOK         0A20-05       1       TRIPALIX TH	9819=28	M	1	•		LINEAR ALKYLATE SULFONATE/ SULFONATION OF ALKYLBENZENES			
9819-36       02       1       SVUTHETIC SPECIALTY PORTHETS (POLYETRENE)         9819-38       4       1       SOLUTIONS (MISCELLANEOUS)/         9819-39       H       1       SOLUTUNS (MISCELLANEOUS)/         9819-39       H       1       SOLUTUNS (MISCELLANEOUS)/         9819-40       24       SODIUM LINEAP ALKYL BENZEME SULFONG CAID         9819-43       3       1       SPECIALTY PAPERS/         9819-43       1       SUBSTITUTED PHENOLS (MISCI/WITHATION)         9820-03       1       TETRANIAL (THAUDED LEVANDET PAPERS/         9820-04       1       TETRANIAL (THAUDE NEWSENCE)         9820-05       1       TETRANIAL (THAUDE NEWSENCE)         9820-06       1       TETRANIAL (THAUDE NEWSENCE)         9820-01       1       TETRANIAL (THAUDE NEWSENCE)         9820-02       1       TETRANIAL (THAUDE NEWSENCE)         9820-03       1       TETRANIAL (THAUDE NEWS	9819 <b>-3</b> 5	3	•	1	•	SOLVENT HASE COATING			
0819-38       4       1       SOLUTIONS (MISCELLANCOUST)         9819-39       H       1       SODIUM (INCAP ALKYL BENZENE SULFONCATE/ SULFONATE/ SULFONATE/         9819-30       24       SODIUM (INCAP ALKYL BENZENE SULFONIC ACID/         9819-33       1       SPECIALITY PAPERS/         9819-43       1       SURSTITUTED PMENDES/(NISC.)/NITRATION         9819-43       1       SURSTITUTED PMENDES/(NISC.)/NITRATION         9820-03       C       1       TAIMELITIC ANMYDRIDE/ OXIDATION OF PSEUDOCUMENE (1.2.4-STRIMETHYL BENZENE)         9820-04       A       1       TAIMELITIC ANMYDRIDE/ OXIDATION OF PSEUDOCUMENE (1.2.4-STRIMETHYL BENZENE)         9820-05       A       1       TAIMELITIC ANMYDRIDE/ OXIDATION OF PSEUDOCUMENE (1.2.4-STRIMETHYL BENZENE)         9820-06       3       2       TETRAHIX (TRADE NAME)/       BENZENE)         9820-07       M       1       2.3.5.5-TETRACHLORD-A-(METHYL SULFONL) PYRIDIME/         9820-08       M       1       2.3.5.5-TETRACHLORD-A-(METHYL SULFONL) PYRIDIME/         9820-01       3       1       TETRAHIX (TRADE NAME)/       9820-11         9820-12       3       1       TETRALIX (TRADE NAME)/       9820-12         9820-13       1       TETRALIX (TRADE NAME)/       9820-13 <td< td=""><td>9819=36</td><td>50</td><td>1</td><td>•</td><td>•</td><td>SYNTHETIC SPECIALTY POLYMERS (POLYSTYRENE)</td></td<>	9819=36	50	1	•	•	SYNTHETIC SPECIALTY POLYMERS (POLYSTYRENE)			
9819-39       H       1       SODIUM LIPRAFALEVIC BEVZEME SULFOWATE/ SULFOWATE/SULFOWATE	9819=38	▲	•	ī		SOLUTIONS (MISCELLANEOUS)/			
9819+40       24       SODIUM NITRORENZE NULCONIC ACTO,         9819+20       22       1       STROFAM, KITAUDED (INCL) ACTOR         9819+33       3       1       SPECIALITY PAPERS, N. BUTNALDENUEN (INCL)ANTRATION         9819+35       1       SUBSTITUTED PHENDES (MISC.)/NITRATION         9819-35       1       SUBSTITUTED PHENDES (MISC.)/NITRATION         9820-03       C       1       TRIMELLITIC ANHYDRIDE/ OXIDATION OF PSEUDOCUMENE (INCL), CANNIZARO)         9820-04       A       1       TRIMELTITIC ANHYDRIDE/ OXIDATION OF PSEUDOCUMENE (INCL), CANNIZARO)         9820-05       A       1       TRIMELTITIC ANHYDRIDE/ N. BUTNALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-05       A       1       TRIMELTITIC ANHYDRIDE/ N. BUTNALDEHYDE (ALDOL, CANNIZARO)         9820-06       3       2       TETRAHIN (TRADE NAMEL)/         9820-07       M       1       2:3:5-8-TETRACHLOROGe-(NETRYL SULFONL) PYRIDINE/         9820-08       M       1       2:3:5-8-TETRACHLOROGe-(NETRYL SULFONL) PYRIDINE/         9820-10       3       1       TATAZINE DIDME/CONSULFIDE/         9820-13       1       TRIPACHLOROGE-(NEDESATION OF UREA, DEHYDROGENATION         9820-14       1       TRIPACHLOROGE-(NEDESATION OF UREA, DEHYDROGENATION         9820-15       1	9819-39	M	i	•		SODIUM LINEAR ALKYL BENZENF SULFONATE/ SULFONATION OF ALKYLBENZENE			
9819-42       D2       1       STYDOPOAH/ EXTOUDED TRANDED POLYSTYRENE         9819-43       3       1       SUBSTITUTED PHENDS(MISC.)/NITRATION         9819-43       1       SUBSTITUTED PHENDS(MISC.)/NITRATION         9820-03       C       1       TRIPELLITIC ANNYDDIDE/ XIDIDE/ XIDIDE/ SUBSCUMENE (1.2.4.4TRIMETHYL BENZENE)         9820-04       A       1       TRIPELLITIC ANNYDDIDE/ XIDIDE/ XIDIDE/ SUFONTLD PREDIDE/         9820-05       A       1       TRIPELTITIC ANNYDDIDE/ XIDIDE/ XIDIDE/         9820-06       3       1       TRIPATURE NAME/         9820-07       H       1       2:3:5:4=TERACHLORG-=(NETMYL SULFONVL) PRIDINE/         9820-08       H       1       2:3:5:4=TERACHLORG-=(NETMYL SULFONVL) PRIDINE/         9820-01       3       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-12       3       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-13       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-14       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-15       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-16       1       TETRAKTHYL THIURAH MONOSULFIDE/         9820-17       D2       1       TERMACHLARCHARE/         9820-18       1 <t< td=""><td>9819-40</td><td>24</td><td></td><td>•</td><td></td><td>SODIUM NITRORENZENESULFONIC ACID/</td></t<>	9819-40	24		•		SODIUM NITRORENZENESULFONIC ACID/			
OPAIGNALSPECIALITY PAPERSY0819-831SPECIALITY PAPERSY0819-83L10819-83L10820-03C10820-03C10820-04A10820-05A10820-06320820-06320820-07M10820-08A10820-09A10820-06320820-06320820-07M10820-08M10820-09A10820-01310820-1031123.35.67ECHARCH_DRA-(METHALE0820-113111ETRAMETHYL0820-12311TETRAKILY0820-1311TETRAKILY0820-1431TETRAKILY0820-1520820-1641TETRAKILY0820-1821TETRAKILY0820-2011TETRAKILY0820-2131TETRAKILY0820-2211TETRAKILY0820-2321TETRAKILY0820-2411TETRAKILY0820-2511TETRAKILY0820-2611TETRAKILY0820-2711TETRAKILY </td <td>9819=42</td> <td>50</td> <td>ĩ</td> <td>ī</td> <td>•</td> <td>STYROFOAM/ EXTRUDED EXPANDED POLYSTYRENE</td>	9819=42	50	ĩ	ī	•	STYROFOAM/ EXTRUDED EXPANDED POLYSTYRENE			
981945       I       SUBSTITUTED PHEROLS(MISC.)/MITRATION         9820-03       C       I       TRIMELLITIC ANHYDRIDE/ OXIDATION OF PSEUDOCUMENE (1:2:4=TRIMETHYL BENZENE)         9820-04       I       TRIMELTYCLOPROPANE/N = BUTYRALDEHYDE * FORMALDEHYDE (ALDOL, CANNIZARO)         9820-05       A       I       TRIMETHYCLOPROPANE/N = BUTYRALDEHYDE * FORMALDEHYDE (ALDOL, CANNIZARO)         9820-06       J       TRIMETHYCLOPROPANE/N = BUTYRALDEHYDE * FORMALDEHYDE (ALDOL, CANNIZARO)         9820-07       H       I       Z;3:5:6=TETRACHURDO-4=(METHYL SULFONYL) PYRIDINE/         9820-10       J       IETRAMETHYL THIURAM MONOSULFIDE/         9820-12       J       TETRACHURDO-4=(METHYL SULFONYL) PYRIDINE/         9820-13       J       TETRACHURDEY       THIDAHYDE/ENDE/         9820-14       TETRACHURDEY       TETRACHURDO-4       TETRACHURDO-4         9820-15       J       TETRACHURDEY       TETRACHURDO-4       TETRACHURDO-4 <td>9A19=43</td> <td>3</td> <td>ī</td> <td>ī</td> <td></td> <td>SPECIALITY PAPERS/</td>	9A19=43	3	ī	ī		SPECIALITY PAPERS/			
9820-03       C       I       TRIMETHYLOLPROPANE/N = BUTYRALDEWIDE (1+2+*TRIMETHYL BENZENE)         9820-05       A       I       TRIMETHYLOLPROPANE/N = BUTYRALDEWIDE * FORMALDEHYDE (4LDOL, CANNIZARO)         9820-05       A       I       TRIMETHYLOLPROPANE/N = BUTYRALDEWIDE * FORMALDEHYDE (4LDOL, CANNIZARO)         9820-05       A       I       TRIMETHYLOLPROPANE/N = BUTYRALDEWIDE * FORMALDEHYDE (4LDOL, CANNIZARO)         9820-05       A       I       TRIMETHYLOLPROPANE/N = BUTYRALDEWIDE * FORMALDEHYDE (4LDOL, CANNIZARO)         9820-06       3       I       TETRANIX (TRADE NAME)/         9820-07       H       I       ZissistTETRACHLOROMA=(METHYL SULFONYL) PYRIDINE/         9820-08       H       I       ZissistTETRACHLOROMA=(METHYL SULFONYL) PYRIDINE/         9820-01       3       I       TETRAMETHYL THURAM HONOSULFIDE/         9820-13       I       TETRACHLOROMA=(NAPROPYL SULFONYL) PYRIDINE/         9820-14       3       I       TETRACTINL THURAM DISULFIDE/         9820-15       I       TETRACTINL THURAM DISULFIDE/       BUTYRACHLOROMADENTION         9820-16       I       TRIMETHYL THURAM DISULFIDE/       BUTYRACHLOROMADENTION       BUTYRACHLOROMADENTION         9820-18       I       TETRACTINL THURAM DISULFIDE/       BUTYRACHLOROMADENTION       BUTYRACHLOROMADENTION<	9819=45	Ĺ	i	•		SURSTITUTED PHENOLS (MISC.)/NITRATION			
0820-04       A       I       TRINETAVLODPORATEON = BUTYRALDEMYDE * FORMALDEMYDE (ALDOL* CANNIZARD)         0820-05       A       I       TRINETAVLOPROPAREON = BUTYRALDEMYDE * FORMALDEMYDE (ALDOL* CANNIZARD)         0820-06       3       Z       TETRANIX (TRADE NAME)/         0820-07       M       I       Zijss6-TETRACHLORO-4-(METHYL SULFONYL) PYRIDINE/         0820-07       M       I       Zijss6-TETRACHLORO-4-(METHYL SULFONYL) PYRIDINE/         0820-07       M       I       Zijss6-TETRACHLORO-4-(METHYL SULFONYL) PYRIDINE/         0820-010       3       I       TETRANETHYL THURAM HONOSULTIDE/         0820-12       3       I       TETRANETHYL THURAM HONOSULTIDE/         0820-13       3       I       TETRATYL THURAM HONOSULTIDE/         0820-14       3       I       TETRATYL THURAM HONOSULTIDE/         0820-15       3       I       TETRATYL THURAM HONOSULTIDE/         0820-16       4       I       THINER/         0820-17       D2       I       TYVEK (SPUNBONDED POLYOLEFINI/ HO POLYETMYLENE SMEET         0820-18       2       I       TETRATYLOTEHYLOLETHANE/ PROPIONALDEMYDE * FORMALDEMYDE (ALDOL* CANNIZARD)         0820-20       A       I       TETRATYLOTEHYLOLETHANE/ PROPIONALDEMYDE * FORMALDEMYDE (ALDOL* CANNIZARD) <td>9820-03</td> <td>Ē</td> <td>ī</td> <td></td> <td></td> <td>TRIMELLITIC ANNYDRIDE/ OXIDATION OF PSEUDOCUMENE (1.2.4-TRIMETHYL BENZENE)</td>	9820-03	Ē	ī			TRIMELLITIC ANNYDRIDE/ OXIDATION OF PSEUDOCUMENE (1.2.4-TRIMETHYL BENZENE)			
9820-05       A       I       TRIDIANE/ TORMALDEHYDE TRIMER         9820-06       3       2       TETRANIX (TRADE NAME)/         9820-07       H       1       2:3:5:6=TETRACHLOR0=4=(NETHYL SULFONYL) PYRIDINE/         9820-08       H       1       2:3:5:6=TETRACHLOR0=4=(NETHYL SULFONYL) PYRIDINE/         9820-01       3       1       IETRAMETAYL THIURAH NONCOULFIDE/         9820-12       3       1       TRIAINE DIONE/CONDESATION OF UREA; DEHYDROGEMATION         9820-13       1       TRIAINE DIONE/CONDESATION OF UREA; DEHYDROGEMATION         9820-14       3       1       TRIAINE DIONE/CONDESATION OF UREA; DEHYDROGEMATION         9820-15       3       1       TRIAINE DIONE/CONDESATION OF UREA; DEHYDROGEMATION         9820-16       4       1       TRIAINE DIONE/CONDESATION OF UREA; DEHYDROGEMATION         9820-17       D2       1       TRIAINETHYL THIURAH DISULFIDE/         9820-18       2	9820=04	Ă	ī			TRIMETHYLOLPROPANE/ N - BUTYRALDEHYDE + FORMALDEHYDE (ALDOL + CANNIZARO)			
9820-05       3       2       TETRANIA (TRADE)ANAME)/         9820-07       H       1       2:3:5:6*TETRACHLORO-4(METHYL SULFONYL) PYRIDINE/         9820-08       H       1       2:3:5:6*TETRACHLORO-4(METHYL SULFONYL) PYRIDINE/         9820-10       3       1       FETRAMETHYL THIURAM MONOSULFIDE/         9820-12       3       1       FETRAMETHYL THIURAM MONOSULFIDE/         9820-13       3       1       TETRAMETHYL THIURAM MONOSULFIDE/         9820-14       3       1       TETRATHYL THIURAM MONOSULFIDE/         9820-15       3       1       TETRATHYL THIURAM DISULFIDE/         9820-16       4       1       TETRATHYL THIURAM DISULFIDE/         9820-17       D2       1       TETRATHYL THIURAM DISULFIDE/         9820-18       2       1       TETRATHYL THIURAM DISULFIDE/         9820-19       3       1       TOTAL OILS/         9820-20       A       1       TETRATHYL THIURAM DISULFIDE/         9820-30       J       1       TOTAL OILS/         9820-31       F1       1       TETRATHYL SULFATE         9820-32       H       1       TETRATHYL THURAME SULFATE         9820-33       J       1       TOTAL OILS/	9820+05		ī			TRIOXANE/ FORMALDEHYDE TRIMER			
9820-07 H 1 2735%6-TETRACMLORO+4-(HETHYL SULFONYL) PYRIDINE/ 9820-08 H 1 2735%6-TETRACMLORO+4-(HETHYL SULFONYL) PYRIDINE/ 9820-08 H 1 2735%6-TETRACMLORO+4-(HETHYL SULFONYL) PYRIDINE/ 9820-01 3 1 TETRATING LINGAM (DAUSULFIDE/ 9820-11 3 1 TETRATING DIONE/CONDESATION OF UREA, DEHYDROBENATION 9820-13 3 1 TETRATINE DIONE/CONDESATION OF UREA, DEHYDROBENATION 9820-14 3 1 TETRATINE DIONE/CONDESATION OF UREA, DEHYDROBENATION 9820-15 3 1 TETRATINE DIONE/CONDESATION OF UREA, DEHYDROBENATION 9820-16 4 1 THINNER/ 9820-17 D2 1 TETRATINE DIONE/CONDED POLYOLEFINI/ HO POLYETHYLENE SHEET 9820-18 2 TETRATINE JOINE/CONDED POLYOLEFINI/ HO POLYETHYLENE SHEET 9820-20 A 1 TETRATINE JOINE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO) 9820-20 A 1 TETRATINE TOTAL OILS/ 9820-27 E 1 TETRATINE MUTAYENZOIC ACTO/ TRIBROMOBENZOIC ACTO + DIMETHYLSULFATE 9820-30 J TI TETRATYRENZOIC ACTO/ TRIBROMOBENZOIC ACTO + DIMETHYLSULFATE 9820-31 F1 TETRADEYNENK + SULFUR 9820-32 H 1 TETRATYRENZOIC ACTO/ TRIBROMOBENZOIC ACTO + DIMETHYLSULFATE 9820-33 Z4 I TETRADOTHIOPHENK/ THIOPHENE + HYDROGEN 9820-33 Z4 I TETRADOTUM THENK/ THIOPHENE + HYDROGEN 9820-35 Z4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN +H2S 9820-36 Z4 I TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN 9820-36 Z4 I TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN 9820-37 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN TRIACETATE/ 9820-37 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN HIZ 9820-37 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN TETRAACETATE/ 9820-37 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN HIZ 9820-37 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN HIZ 9820-35 7 4 I TETRADECYL MERCAPTAN/ ALPHA-OLEFIN HIZ 9820-55 7 1 FILOLOH NOHYDROFENC CHLORINATION OF TRIMELLITIC ANHYDRIDE 9820-55 7 1 FILOLOH NOHYDROFENC CHLORINATION OF TRIMELLITIC ANHYDRIDE 9820-55 7 1 FILOLOH PONDENT OF TRIMELLITIC ANHYDRIDE	9820-06	Ĵ	2			TETRAMIX (TRADE NAME) /			
9820-08       H       1       2:35=TRICHLOROP4=IN=PROPYL_SULFORYL) PYRIDINE/         9820-10       3       1       TETRAMETHYL_THIURAM HONOSULFTOE/         9820-11       3       1       TETRAMETHYL_THIURAM HONOSULFTOE/         9820-12       3       1       TETRAMETHYL_THIURAM HONOSULFTOE/         9820-13       3       1       TETRAMETHYL_THIURAM HONOSULFTOE/         9820-14       3       1       TETRAETHYL_THIURAM HONOSULFTOE/         9820-15       3       1       TETRAETHYL_THIURAM DISULFTOE/         9820-16       4       1       TETRAETHYL_THIURAM HONOSULFTOE/         9820-16       1       TETRAETHYL_THIURAM DISULFTOE/         9820-17       D2       1       TETRAETHYL_THIURAM DISULFTOE/         9820-18       2       1       TETRETHYL_THIURAM DISULFTOE/         9820-19       3       1       TOTAL OTLS/         9820-20       A       1       TOTAL OTLS/         9820-31       9       1       TETRAMETHYL THIUPAMEY         9820-32       J       1       THIPHETHYLOCLETHANE/ PROPIONALDEMYDE + FORMALDEMYDE (ALDOL: CANNIZARO)         9820-33       J       1       TERPENE/         9820-31       FI       TEO ROTTOMS/ <td< td=""><td>9820=07</td><td>M</td><td></td><td>ī</td><td></td><td>2.3.5.6-TETRACHLORO-A-(METHYL SULFONYL) PURIDING/</td></td<>	9820=07	M		ī		2.3.5.6-TETRACHLORO-A-(METHYL SULFONYL) PURIDING/			
9820+10       3       1       TETRAMETHYL THIURAM HONGSULFIDE/         9820+11       3       1       THIPAN(INSECTICIDE,VULCANIZER)/         9820+12       3       1       TETRACIL/         9820+13       3       1       TETRACIL/         9820+14       3       1       TETRACIL/         9820+14       3       1       TETRACIL/         9820+14       3       1       TETRACIL/         9820+16       4       1       TETRACINUL THIURAM DISULFIDE/         9820+17       D2       1       TETRACINUL THIURAM DISULFIDE/         9820+18       2       TETRACINUL THIURAM DISULFIDE/         9820+19       3       1       TETRACINUL THIURAM DISULFIDE/         9820+20       A       1       TETRACINUL THIURAM DISULFIDE/         9820+218       2       TETRACINUL THIURAM DISULFIDE/         9820+20       A       1       TETRACINUL THIURAM DISULFIDE/         9820+20       A       1       TETRACINUL THIURAM DISULFIDE/         9820+219       3       1       TETRACINCLE ACIDY TRIBROMOBENZOIC ACID + DIMETHYLSULFATE         9820+20       A       1       TETRACINCLE ACIDY TRIBROMOBENZOIC ACID + DIMETHYLSULFATE         9820+219       J	9820-08	M		ī		2.3.5-TRICHLORO-4-(N-PROPAL SULFONAL) PARIDINF/			
9820-11       3       1       THIPAM (INSECTICIDE, VULCANIZER)/         9820-12       3       1       TERRACIL/         9820-13       3       1       TERRACIL/         9820-14       3       1       TETRACIL/         9820-14       3       1       TETRACIL/         9820-14       3       1       TETRACIL/         9820-16       4       1       THINARC/         9820-17       D2       1       TETRACINE (SUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SHEET         9820-18       2	9820+10	3	i	-		TETRAMETHYL THIURAM MONOSULEIDEZ			
9820-12       3       1       TERHACIL/         9820-13       3       1       TRIAZINE DIONE/CONDESATION OF UREA, DEHYDROGENATION         9820-14       3       1       TRIAZINE DIONE/CONDESATION OF UREA, DEHYDROGENATION         9820-14       3       1       TRIAZINE DIONE/CONDESATION OF UREA, DEHYDROGENATION         9820-14       3       1       TRIAZINE DIONE/CONDED POLYDEFIN)/ HO POLYETHYLENE SHEET         9820-16       4       1       TYVEK (SPUNBONDED POLYDEFIN)/ HO POLYETHYLENE SHEET         9820-17       D2       1       TYVEK (SPUNBONDED POLYDEFIN)/ HO POLYETHYLENE SHEET         9820-18       2       TERPENE/         9820-20       A       1       TOTAL OILS/         9820-20       A       1       TRIMETHYLOUETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820-27       E       1       TRIMETHYLOUETHANE/ SULFUR         9820-20       A       1       TRIMETHYLOUETHANE/ SULFUR         9820-21       F1       1       TERMENCY         9820-22       H       1       TRIMETHYLOUETHANE + SULFUR         9820-231       F1       1       TERMAYORTHIOPHENE/ THIOPHENE + HYDROGEN         9820-33       24       1       TERMAYOROTHIOPHENE/ THYLENEDIAMINE TERAACETATE/ EOTA + CAUSTI	9820=11	3	ī			THTPAM (INSECTICIDE + VUL CANTZFR) /			
9820-13       3       1       TRIAZINE DIONE/CONDESATION OF UREA, DEHYDROGENATION         9820-14       3       1       TETRAETHYL THJURAM DISULFIDE/         9820-16       4       1       THINNER/         9820-16       4       1       THINNER/         9820-17       D2       1       TYVEK (SPUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SMEET         9820-18       2       TERPENE/       9820-18         9820-20       A       1       TOTAL OILS/         9820-20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-216       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-22       4       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-23       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL, CANNIZARO)         9820-27       E       1       TERPENE/         9820-33       J       1       TETRANDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820-33       24       1       <	9820=12	3	ĩ			TERNACIL/			
9820-14       3       1       TETRETHYL THIURAM DISULFIDE/       OULY OLANOWSLAW TOWN         9820-16       4       1       THINNER/         9820-16       4       1       THINNER/         9820-17       D2       1       TYVEK (SPUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SHEET         9820-18       2       TERPENE/         9820-20       A       1       TOTAL OILS/         9820-20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEMYDE + FORMALDEMYDE (ALDOL+ CANNIZARO)         9820-20       A       1       TRIMETHOXYMENZOIC ACIO/ TRIBROMOBENZOIC ACIO + DIMETHYLSULFATE         9820-20       A       1       TRIMETHOXYMENZOIC ACIO/ TRIBROMOBENZOIC ACIO + DIMETHYLSULFATE         9820-20       A       1       TRIMETHOXYMENZOIC ACIO/ TRIBROMOBENZOIC ACIO + DIMETHYLSULFATE         9820-20       A       1       TRIMETHOXYMENZOIC ACIO/ TRIBROMOBENZOIC ACIO + DIMETHYLSULFATE         9820-20       A       1       TRIMETHOXYMENZOIC ACIO/ TRIBROMOBENZOIC ACIO + DIMETHYLSULFATE         9820-20       J       1       THIOPHENE/ BUTANE + SULFUR         9820-30       J       1       TETRAVDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820-31       F1       1       TETRAVDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820-33	9820-13	5	i		•	TRIATINE DIGNE/CONDESATION OF URFAM DEMYDROGENATION			
9820=16       4       1       THINNER/         9820=17       D2       1       TYVEK (SPUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SHEET         9820=18       2	9820-14	ī	i			TETRAETHYL THIURAM DISULFIDE/			
9A20=17DZ1TYVEK (SPUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SHEET9820=182•TERPENE/9820=193•1TOTAL OILS/9820=20A1•TRIMETHYLOLETHANE/ PROPIONALDEHYDE • FORMALDEHYDE (ALDOL, CANNIZARO)9820=20A1•TRIMETHYLOLETHANE/ PROPIONALDEHYDE • FORMALDEHYDE (ALDOL, CANNIZARO)9820=20A1•TRIMETHYLOLETHANE/ PROPIONALDEHYDE • FORMALDEHYDE (ALDOL, CANNIZARO)9820=27E1•TRIMETHOLYRENZOIC ACID/ TRIBROMOBENZOIC ACID • DIMETHYLSULFATE9820=30J1•TETRANDOTHIOPHENE/ BUTANE • SULFUR9820=31F11•TETRANDOTHIOPHENE/ THIOPHENE • HYDROGEN9820=33Z4•1•9820=34A1•TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EOTA • CAUSTIC9820=35Z4•1•TRISODIUM NITRILOTRIACETATE, MONOMYDRATE/ NH3 • FORMALDEHYDE • NACN9820=36Z4•1•TETRASODIUM NITRILOTRIACETATE, MONOMYDRATE/ NH3 • FORMALDEHYDE • NACN9820=3741•TETRASODIUM NITRILOTRIACETATE, MONOMYDRATE/9820=3931•TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/9820=4561•TRIAMINO CRYSTALS/9820=51R1•TRIAMINO CRYSTALS/9820=55T•••TERPHENYLS/ RENZER/ CHLORINATION = DEHYDROCHLORINATION OF PROPYLENE	9820-16	Ă.		i		THINNER/			
9820+18       2       TERPENE/         9820+19       3       1       TOTAL 0[L\$/         9820+20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820+20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820+20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820+20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820+21       E       1       TRIMETHOUYMENZOIC ACID/ TRIBROHOBENZOIC ACID + DIMETHYLSULFATE         9820+22       E       1       TERPENE/         9820+23       J       1       TETRANTORTHIOPHENE/ THIOPHENE + HYDROGEN         9820+31       F1       1       TETRANTORTHIOPHENE/ THIOPHENE + HYDROGEN         9820+32       M       1       TETRANTORTHIOPHENE/ THIOPHENE + HYDROGEN         9820+33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EOTA + CAUSTIC         9820+33       24       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9820+35       24       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9820+36       24       1       TRISODIUM N	9820-17	30	ī	-		TYVER (SPUNBONDED POLYOLEFIN)/ HO POLYETHYLENE SHEFT			
9820=19       3       1       TOTAL OILS/         9820=20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820=20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820=20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820=20       A       1       TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL+ CANNIZARO)         9820=21       E       1       TEG ROTTOMS/         9820=30       J       1       THIPHYLE/BUTANE + SULFUR         9820=31       F1       1       TETRAHYDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820=32       M       1       TETRAHYDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820=33       Z4       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EOTA + CAUSTIC         9820=34       A       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 • FORMALDEHYDE + NACN         9820=35       Z4       1       TRISODIUM NITRILOTRIACETATE, SOLUTION/         9820=36       Z4       1       TRISODIUM NITRILOTRIACETATE, SOLUTION/         9820=37       4       1       TERASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TERASODIUM ETHYLENEDIAMINE	9820+18	2	•			TERPENE/			
9820=20       A       1       TRIMETHYLOLETMANE/ PROPIONALDEMYDE + FORMALDEMYDE (ALDOL + CANNIZARO)         9820=26       14       .       TRIMETHYLOLETMANE/ PROPIONALDEMYDE + FORMALDEMYDE (ALDOL + CANNIZARO)         9820=26       14       .       TRIMETHYLOLETMANE/ PROPIONALDEMYDE + FORMALDEMYDE (ALDOL + CANNIZARO)         9820=27       E       1       .       TEG ROTTOMS/         9820=30       J       1       .       TETRANYDROTHIOPHENE + SULFUR         9820=31       F1       1       .       TETRANYDROTHIOPHENE + HYDROGEN         9820=32       M       1       .       TETRANYDROTHIOPHENE + HYDROGEN         9820=33       24       1       .       TETRANDECYL MERCAPTAN/ ALPMA=OLEFIN +H2S         9820=34       A       1       .       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EOTA + CAUSTIC         9820=35       24       1       .       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ MH3 + FORMALDEHYDE + NACN         9820=36       24       1       .       TRISODIUM NITRILOTRIACETATE, SOLUTION/         9820=36       24       1       .       TRISODIUM N=HYROXYETMYLENYLENYLENYLENYLENALDEMATE/         9820=37       4       1       .       .       TETRASODIUM ETHYLENEDIAMINE TETRACETATE/         9820=36       24 </td <td>9820-19</td> <td>3</td> <td>•</td> <td>i</td> <td></td> <td>TOTAL OILS/</td>	9820-19	3	•	i		TOTAL OILS/			
9820=26       14       .       TRIMETHOXYRENZOIC ACID/ TRIBROMOBENZOIC ACID + DIMETHYLSULFATE         9820=27       E       1       .       TEG ROTTOMS/         9820=30       J       1       .       THIOPHENE/ BUTANE + SULFUR         9820=31       F1       1       .       TETRAHYDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820=32       M       1       .       TETRANDECYL MERCAPTAN/ ALPHA=OLEFIN +H2S         9820=32       M       1       .       .         9820=33       24       1       .       .         9820=34       A       1       .       .         9820=35       24       1       .       .         9820=35       24       1       .       .         9820=36       24       1       .       .         9820=35       24       1       .       .         9820=36       24       1       .       .       .         9820=37       4       1       .       .       .       .         9820=36       24       1       .       .       .       .       .         9820=37       4       1       .       .       . <td>9820-20</td> <td><b>A</b></td> <td>i</td> <td></td> <td>•</td> <td>TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL + CANNIZADO)</td>	9820-20	<b>A</b>	i		•	TRIMETHYLOLETHANE/ PROPIONALDEHYDE + FORMALDEHYDE (ALDOL + CANNIZADO)			
9820=27       E       1       TEG ROTTOMS/         9820=30       J       1       THIOPHENE/ BUTANE + SULFUR         9820=31       F1       1       TETRAHYOROTHIOPHENE/ THIOPHENE + HYDRÖGEN         9820=32       M       1       N#TETRADECYL MERCAPTAN/ ALPHA=OLEFIN +H2S         9820=33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EOTA + CAUSTIC         9820=33       24       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9820=35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820=36       24       1       TRISODIUM NITRILOTRIACETATE/ MONOHYDRATE/         9820=37       4       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820=37       4       1       TRISODIUM NEMYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820=37       4       1       TRISODIUM STHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TRISODIUM STHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TRISODIUM STHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TRIAMINO CRYSTALS/         9820=37       6       1       TRIAMINO CRYSTALS/         9820=51       R       1       TRICHLOROPRO	9820-26	14	•	•	•	TRIMETHOXYRENZOIC ACID/ TRIGROMOGENZOIC ACID + DIMETHYLSULFATE			
9820=30       J       I       THIOPHENE/ BUTANE + SULFUR         9820=31       F1       I       TETRAHYDROTHIOPHENE/ THIOPHENE + HYDROGEN         9820=32       M       I       N#TETRADECYL MERCAPTAN/ ALPMA#OLEFIN +H2S         9820=33       24       I       TETRASODIUM ETHYLENEDIAHINE TETRAACETATE/ EOTA + CAUSTIC         9820=33       24       I       TETRASODIUM ETHYLENEDIAHINE TETRAACETATE/ EOTA + CAUSTIC         9820=34       A       I       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9820=35       24       I       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820=36       24       I       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820=37       4       I       TETRASODIUM ETHYLENEDIAHINE TETRAACETATE/         9820=37       4       I       TETRASODIUM ETHYLENEDIAHINE TETRAACETATE/         9820=39       3       I       TRIAMINO CRYSTALS/         9820=45       6       I       TPI=OCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE         9820=51       R       1       TRICHLOROPROPENE/ CHLORINATION + DEHYDROCHLORINATION OF PROPYLENE         9820=55       T       TERPHENYLS/ RENZENE THRU HOT TURE	9820-27	E	1	•		TEG ROTTOMS/			
9820-31       F1       1       TETRAHYDROTHIOPHENE/THIOPHENE/THIOPHENE + HYDROGEN         9820-32       M       1       N#TETRADECYL MERCAPTAN/ALPMABOLEFIN +H2S         9820-33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/EDTA + CAUSTIC         9820-33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/EDTA + CAUSTIC         9820-34       A       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/NH3 + FORMALDEHYDE + NACN         9820-35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820-36       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820-37       4       1       TRISODIUM NEHYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820-39       3       1       TRIAMINO CRYSTALS/         9820-45       6       1       TPI-OCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE         9820-51       R       1       TRICHLOROPROPENE/ CHLORINATION + DEHYDROCHLORINATION OF PROPYLENE         9820-55       T       .       TERPHENYLS/ RENZENE THRU HOT TURE	9820-30	J	3		•	THIOPHENE/ BUTANE + SULFUR			
9820-32       M       1       N#TETRADECYL MERCAPTAN/ ALPMAGOLEFIN #H2S         9820-33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EDTA * CAUSTIC         9820-33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EDTA * CAUSTIC         9820-34       A       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 * FORMALDEHYDE * NACN         9820-35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820-36       24       1       TRISODIUM NEHYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820-37       4       1       TRISODIUM ETHYLENEDIAMINE TETRAACETATE/         9820-39       3       1       TRIAMINO CRYSTALS/         9820-45       6       1       TPI-OCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE         9820-51       R       1       TRICHLOROPROPENE/ CHLORINATION + DEHYDROCHLORINATION OF PROPYLENE         9820-55       T       .       TERPHENYLS/ RENZENE THRU HOT TURE	9820-31	<b>#1</b>	ī	•	•	TETRAHYDROTHIOPHENE/ THIOPHENE + HYDROGEN			
9820-33       24       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ EDTA + CAUSTIC         9820-34       A       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9820-35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820-36       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820-36       24       1       TRISODIUM N=HYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820-36       24       1       TRISODIUM N=HYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820-37       4       1       TRIASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820-39       3       1       TRIAMINO CRYSTALS/         9820-45       6       1       TRIAMINO CRYSTALS/         9820-51       R       1       TRICHLOROPROPENE/ CHLORINATION + DEHYDROCHLORINATION OF PROPYLENE         9820-55       T       .       TERPHENYLS/ RENZENE THRU HOT TURE	9820-32	M	1		•	NOTETRADECYL MERCAPTAN/ ALPHADOLEFIN HHZS			
9R20=34       A       1       TRISODIUM NITRILOTRIACETATE, MONOHYDRATE/ NH3 + FORMALDEHYDE + NACN         9R20=35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9R20=36       24       1       TRISODIUM N=HYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9R20=37       4       1       TRISODIUM N=HYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9R20=37       4       1       TRIADOUM ETHYLENEDIAMINE TETRAACETATE/         9R20=37       4       1       TRIANINO CRYSTALS/         9R20=45       6       1       TRIAMINO CRYSTALS/         9R20=51       R       1       TRICHLOROPROPENE/ CHLORINATION + DEHYDROCHLORINATION OF PROPYLENE         9R20=55       T       .       TERPHENYLS/ RENZENE THRU HOT TURE	9820-33	24	•	1		TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ COTA + CAUSTIC			
9820=35       24       1       TRISODIUM NITRILOTRIACETATE SOLUTION/         9820=36       24       1       TRISODIUM N=MYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820=37       4       1       TRISODIUM N=MYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820=37       4       1       TRIADDIUM ETHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TRIASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TRIAMINO CRYSTALS/         9820=45       6       1       TRIAMINO CRYSTALS/         9820=51       R       1       TRICHLOROPROPENE/ CHLORINATION = DEHYDROCHLORINATION OF PROPYLENE         9820=55       T       .       TERPHENYLS/ RENZENE THRU HOT TURE	9820-34	Ā	•	1	•	TRISODIUM NITRILOTRIACETATE + MONOHYDRATE / NH3 + FORMALDEHYDE + NACH			
9820=36       24       1       TRISODIUM N=HYROXYETHYLETHYLENEDIAMINE TRIACETATE/         9820=37       4       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820=37       4       1       TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/         9820=39       3       1       TRIAMINO CRYSTALS/         9820=45       6       1       TRIOCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE         9820=51       R       1       TRICHLOROPROPENE/ CHLORINATION = DEHYDROCHLORINATION OF PROPYLENE         9820=55       T       *       TERPHENYLS/ RENZENE THRU MOT TURE	9820=35	24		1		TRISODIUM NITRILOTRIACETATE SOLUTION/			
9820-37 4 1 TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/ 9820-39 3 1 TRIAMINO CRYSTALS/ 9820-45 G 1 TRIAMINO CRYSTALS/ 9820-45 G 1 TRIADING TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE 9820-51 R 1 TRICHLOROPROPENE/ CHLORINATION - DEHYDROCHLORINATION OF PROPYLENE 9820-55 T TERPHENYLS/ RENZENE THRU HOT TURE	9820=36	24	•	ī		TRISODIUM NOMYROXYETHYLETHYLENEDIAMINE TRIACETATE/			
9420+39 3 1 TRIAMINO CRYSTALS/ 9820+45 G 1 TRIACTURE CALORINATION OF TRIMELLITIC ANHYDRIDE 9820-51 R 1 TRICHLOROPROPENE/ CHLORINATION OF DEMYDROCHLORINATION OF PROPYLENE 9820+55 T T TERPHENYLS/ RENZENE THRU HOT TURE	9820-37	-	, ,	1		TETRASODIUM ETHYLENEDIAMINE TETRAACETATE/			
9820-45 G I TPI-OCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE 9820-51 R I TRICHLOROPROPENE/ CHLORINATION - DEHYDROCHLORINATION OF PROPYLENE 9820-55 T TERPHENYLS/ RENZENE THRU HOT TURE	9A20+39	3		ī	•	TRIAMINO CRYSTALS/			
9820-51 R 1 TRICHLOROPROPENE/ CHLORINATION - DEHYDROCHLORINATION OF PROPYLENE 9820-55 T TERPHENYLS/ RENZENE THRU HOT TURE	9820-45	G	i	•	-	TPI-OCTYL TRIMELLITATE/ ESTERIFICATION OF TRIMELLITIC ANHYDRIDE			
9820-55 T TERPHENYLS/ RENZENE THRU HOT TURE	9820=51	R	ī		-	TRICHLOROPROPENEZ CHLORINATION - DEHYDROCHLORINATION OF PROPYLENE			
	9820-55	T	•	•	•	TERPHENYLS AENZENE THRU HOT TURE			

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PP_CODE	GEN_CODE	DIW	ZERO	UNK	PP_TEXT				
9820=56	01	1		•	TERATE RESINS/ POLYESTER				
9820-57	5	i	•	•	TRIETHANOL AMMONIUM LAURYL SULFATE/ LAURYL AMINE + ETHYLENE ORIDE				
9820-58	н	1	•	•	TRISODIUM SULFO SUCCINATE/ MALEIC ANMYDRIDE + SODIUM BISULFITE				
9820+59	5	1	•	•	TRIETHANOLAMINE LINEAR ALKYLBENZENE SULFONATE/ ETHOXYLATION				
9820=60	8	٠		•	TAMOL (TRADE NAME) (POSSIBLY EPOXIDIZED SOYA OILS)				
9820=62	F1	1	•	٠	1+2+3+6 TETRAHYDROBENZALDEHYDE/ HYDROGENATION OF BENZALDEHYDE				
9821=01	6	٠	•	٠	URAN (TRADE NAME)/ UREA DERIVATIVE				
9521-04	<b>A</b>	1	•	•	URETHANE (MISC)/ ISOCYANATE + POLYOL				
10=5589	G	1	•	•	VAZO(AZOBISISOBUTYRONITRILE)/ METHACRYLONITRILE + HYDRAZINE, DEHYDROGENATION				
9822=02	D1	•	•	•	VARNISH RESIN (ROSIN AND ROSIN ESTERS)				
9822=04	<b>A</b>	٠	•	•	VOPITES (TRADE NAME) (URETHANE PREPOLYMERS)				
<b>9822-07</b>	3	•	•	•	VULCANIZED FIBRE/				
9822=08	E	•	1	٠	VEGETABLE OILS-GENERAL/				
9822+09	3	•	1	•	PRINTING INK VARNISHES/				
9822-11	Q	1	•	٠	N=VINYL=2=PYHROLIDONE/ 3=BUTYROLACTONE + ETHANOLAMINE, DEHYDRATION				
9822+12	01	1	•	•	VINYL-ACPYLIC SHEET/				
9823=01	D1	•	4	•	WAX EMULSIONS/ FORMULATED FROM CAPTIVE FORMALDEHYDE . PHENOL, UREA, ETC RESINS				
9823-02	3	٠	1	٠	WATEPRORNE COATING/				
9823=03	3	1	•	•	WOODFLOUR/				
9823-05	4	•	1	•	WATER REPELLENT/				
9824=01	0	•	1	٠	XANTHATES OF C2+C5_ALCOHOLS/ ALCOHOLIC KOH + CS2				
9825+01	3	1	•	٠	SPUN YARN DRY PROCESS/				
9826=01	A	•	1	•	ZINC AMMONIUM VERSENATE/ ETHYLENE DIAMINE + CHLOROACETIC ACID				
9826=02	3	٠	•	•					
9826=04	24	•	•	٠	ZINC UNDECYLENATE/ UNDECYLENIC ACIO + ZINCOXIDE				
9826405	24	•	1	•	ZINC AND CALCIUM STEARATE/				
9826=07	24	•	1	٠	ZINC DISODIUM ETHYLENEDIAMINE TRIACETATE/				
9901-01	3	6	3	٠					
9901-04	3	•	1	٠	AMMONIUM NITRATE/				
9901=05	3	3	<u>.</u>	•	AMMONIUM SULFATE				
4401-07	3	•	3	٠	AMMONIA ANHYBHOUS/HXN NITROBEN HYDROBEN				
9901=08	3	1	•	•	ALUMINUM SULFATE/				
9901009	ĸ	1	٠	•	AMMONIUM SALTS-PATTY R-OH ETHER SULFATE/				
9401#10	3	:	•	•	ALOWINDE FLUCKIDE/				
9902002	3	1	:	•	RROHINEZ				
9903-02	J	1	1	٠	CATALYST				
9903403	3	6	2	•	CHLONINE-CAUSTIC/				
9903-00	3	•.	•	٠					
990J-14	5	٠	•	٠					
9903-12	J	:	:	•	CADRON MONTOLING				
31#EVFF	3		1	٠					
77UJ=13	J	2	2	•	CAUSTIC SUNAN				
AAN 7414	3	1	•	•	CELLULUSE MATTEMY SEPAMATONS/				

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PP_CODE	GEN_CODE	DIR	ZERO	· UNK	PP_TEXT
9903-15	3	2	•		CALCIUM CHLORIDE/
9903=16	3				CALCIUM CARBIDE/
9903-18	3	•	ī	•	CARBON DIOXIDE/
9903=19	3				CALCIUM HYDROKIDE/
9904=01	3	•	•	•	DIAMMONIUM PHOSPHATE/
9906-01	3	1	•		FLUORSPAR
9908-02	3	11	Ĵ	•	HYDROCHLORIC ACID/
9908-03	3	5	5		HYDPOGEN/
9908=04	3	1		•	HYDROFLOURIC ACID SALTS/
9908=07	3	1	•		HYDROGEN PEROXIDE/
9908=08	3	•	•		HYDROGEN SULFIDE/
9908-09	3	Ž		•	HYDPOGEN CHLORIDE/
9908-10	3	2			HYDROGEN CYANIDE/
9909-02	3	1		•	INDUSTRIAL GASES-HYDROGEN NITROGEN/
9909=03	3	- E - 1		•	IODINE/
9912=01	3	•	•		LUDOX-SILICA/
9913-01	3		•		MIKED ACIDS (NITRIC & BULFURIC)/
9913-03	3	i	i	•	MURIATIC ACID (LOW GRADE HYDROCHLORIC ACID)/
9914-02	3	3	Š	•	NITRIC ACID/
9914=04	3	•	•		NON-PIGMENTED PRODUCT/
9914=05	3	i	-		NITROGEN/
9915-01	3	2	ĩ		OXYGEN/
9916=05	3	-			PHOSPHOROUS ACID/
9916-08	Ĵ	ĩ	•		POLYSTYRENE (OPS) SHEET/
9916=12	3	1			POTASSIUM ACID PHTHALATE/
9916=20	3	-			PHOSPHOROUS PENTASULETDE/
9916-21	3				POTASSIUM CARRONATE (POTASH) /RYPROD OF MSG (2080-99)
9917=01	6			•	QUARTERNARY AMMONIUM CHLORIDES/DUANTERNIZED DIMER DIAMINES.TETRAMINES
9919=02	3		ī		SODIUM CHROMATE/
9919=04	3	ž	•		SODIUM SULFATE/
9919=05	3	7	ī		SULFURIC ACID/
9919=06	3		•	•	SODIUM NITRITE/
9919-0A	3	•		•	SODIUM CHLORATE/
9919-10	3	•	•		SODIUM SILICATE/
9919+11	3	•			SYNTHESIS GAS/
9919=12	3	Z			SDDIUM MEYAL/
9919=15	3	1			SODIUM CARBONATE (NACH + FLUE GAS)/
9919-16	3		•	-	SULFUR MONOCHLORIDE/
9919=17	3		•		SODIUM CHLORIDE/
9919-22	3	2	•	•	SODIUM HYPOCHLORATE/
9919=25	. 3	-	•	-	SULFUR (CPYSTEX)/
9919=26	3	ī	-		SODIUM SALTS-FATTY ALCH ETHER SULFATE/
9919-31	3	Ĩ	•	•	SODIUM FERROCYANIDE/

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PP_CODE	GEN_CODE	DTR	ZERO	UNK	PP_TEXT
9919=32	3	1	•	•	SULFUR ANHYDRIDE/
9920-01	3	1		•	TITANIUM DIOXIDE/
9923+01	3	1			WIRE AND CARLES
9926-02	3	1	•	•	ZINC SULFATE/

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## APPENDIX B

## BPT STATISTICS

## BPT Statistical Appendix

### DESCRIPTIVE STATISTICS

Some of the more commonly employed descriptive statistics are defined as follows:

 N - number of valid observations used in a particular analysis (e.g., the total number of effluent samples at a particular plant for a particular pollutant)

(2) Mean - arithmetic average: 
$$\overline{X} = \sum_{i=1}^{N} X_i / N_i = 1$$

(3) Variance - standard unbiased estimate:  $S^2 = \frac{1}{N-1} \sum_{i=1}^{N} (X_i - \overline{X})^2$ (The standard deviation is  $S = \sqrt{S^2}$ .)

(4) Minimum - the smallest value in a set of N observations

(5) Maximum - the largest value in a set of N observations

- (6) Range the minimum subtracted from the maximum
- (7) Median the middle value in a set of N observations. If N is odd
   (N = 2k 1 for some integer k), the median is the kth order statistic,
   C(k). If N is even (N = 2k), the median is

$$1/2[C(k) + C(k + 1)].$$

### MOVING STATISTICAL MEASURES

Over a year's data were available for some plants. The question of whether treatment system performance at those plants was consistent over time was investigated by examining moving statistical measures of performance. Let  $X_1$ , ...,  $X_N$  denote the N daily observations available from a plant listed in the order they were obtained. Then the moving mean and variance on day t based on observations for the latest n < N days are defined as

$$\overline{X}_{t} = \frac{1}{n} \sum_{i=1}^{n} \overline{X}_{t-i+1}$$

and

$$S_{t}^{2} = \frac{1}{(n-1)} \sum_{i=1}^{n} (X_{t-i+1} - \overline{X}_{t})^{2},$$

where  $t \ge n$ .

If the distribution of X is lognormal (so  $\log_e(X)$  is normal with parameters  $\mu$  and  $\sigma^2$ ), then the 99th percentile of X is

$$P99 = e^{\mu} + 2.326\sigma$$

The moving estimate of P99 at time t based on the lognormal model, therefore, is

$$\overline{X}_{t} + 2.326S_{t}$$
  
P99<sub>t</sub> = e

with  $\overline{X}_t$  and  $S_t^2$  defined above.

Moving estimates of the 99th percentiles of effluent concentrations were plotted over time for each plant to evaluate the consistency of its treatment performance (see Appendix D). Note that the moving 99th percentile will reflect changes in both average effluent levels (through  $\bar{X}_t$ ) and day-to-day effluent variation (through  $S_t$ ).

#### GOODNESS-OF-FIT TESTS

The statistical model used to describe effluent data assumes that y = log(C) is normally distributed, where C is the daily effluent BOD or TSS concentration. Goodness-of-fit tests for this model were run using the studentized range test based on the statistic

$$U = R/S$$
,

with the range (R) and standard deviation (S) defined above. Critical values of the U-test are given in <u>Biometrika Tables for Statisticians</u>, Vol. 1, page 200, for selected sample sizes (N). An upper tail test was used to guard against alternative distributions with heavier tails than the lognormal distribution; the lognormal model would tend to underestimate the 99th percentile if such alternatives were appropriate.

A significance level of  $\alpha = 0.01$  was employed in each test. Since there were a total of 28 data sets tested (17 for BOD and 11 for TSS), this choice of  $\alpha$  ensured that the overall probability of rejecting the lognormal model, when it was appropriate, was reasonably small.

Table B-1 shows the results of the goodness-of-fit tests. The model was rejected for only two out of twenty-eight data sets (BOD for plant 236 and TSS for plant 27). The impact on the 99th percentile estimate for the two rejected cases was evaluated by comparing model-based estimates with nonparametric estimates; namely, the next to largest of the 162 BOD observations and the next to the largest of the 158 TSS observations. For BOD at plant 236, the model gave P99 = 93 versus the nonparametric estimate of P99 = 70. For TSS at plant 27, the model gave P99 = 76 compared to the nonparametric estimate of P99 = 80. In neither case was the lognormal estimate substantially lower than the nonparametric estimate.

The goodness-of-fit of the lognormal model also was checked through a graphical procedure called a probability plot. Let  $X_1$ , ...,  $X_n$  denote the n observed daily values of the parameter of interest (the BOD or TSS measurements from a given plant). Denote the rth largest of the n values by X(r), and define a corresponding score called the "probit" by

Probit 
$$[X_{(r)}] = \Phi^{-1}[r/(n + 1)],$$

where  $\Phi^{-1}(\cdot)$  is the inverse of the standard normal cumulative distribution function. The probit score is the normal deviation (z-value) equivalent to the value X<sub>(r)</sub>. Probit scores are useful because plots of X values versus corresponding probit scores tend to be straight lines when X is normally distributed; this fact is the basis for probability plots. If X has a lognormal distribution, a log-scale plot of X values versus probit scores tends to be a straight line. Daniel and Wood (1971) give simulated examples of probability plots to indicate the degree of random departure from a straight line to expect for different sample sizes when X is normally distributed. Probability plots for BOD and TSS are presented in Figures B-1 to B-28.

Based on the results of the studentized range test and the probability plots, it was concluded that the lognormal distribution could be used to model the data.

	[ .	BOD			TSS		
PLANT	TYPE	N	U	P*	N	·U	P*
9	Р	24	4.84	N.S.	24	3.73	N.S.
15	NP	363	6.97	N.S.	0	-	-
27	NP	160	5.77	N.S.	158	7.36	<0.01
44	P	261	5.86	N.S.	260	6.00	N.S.
45	P	156	4.03	N.S.	364	3.86	N.S.
96	Р	105	3.96	N.S.	66	4.83	N.S.
110	NP	247	4.73	N.S.	218	4.70	N.S.
111	Р	157	4.36	N.S.	347	5.69	N.S.
113	NP	332	5.77	N.S.	91	4.45	N.S.
118	NP	365	4.52	N.S.	0	-	-
126	Р	249	5.09	N.S.	253	5.49	N.S.
170	NP	103	3.21	N.S.	0	-	-
175	NP	361	4.98	N.S.	0	-	-
220	NP	55	3.92	N.S.	149	5.22	N.S.
234	NP	157	5.71	N.S.	0	-	-
236	NP	162	7.28	<0.01	362	6.43	N.S.
281	NP	203	5.22	N.S.	0	-	-
		1					

# TABLE B-1. GOODNESS-OF-FIT TESTS FOR BOD AND TSS LOG<sub>e</sub> OF DAILY DATA

\* Critical values for the studentized range test (upper tail,  $\alpha = 0.01$ ) are

N 	U.99
25 50 65 90 100 150 200 500	5.06 5.77 6.01 6.27 6.36 6.64 6.84 7.42

N.S. = Not significant (U value below critical level).

Reference: Biometrika Tables for Statisticians, Vol. 1, page 200.





PROBABILITY PLOT









FIGURE 8-3. PROBABILITY PLOT FOR BOD



PLANT-44

FIGURE B-4. PROBABILITY PLOT FOR BOD

B-7







FIGURE B-6. PROBABILITY PLOT FOR BOD

B-8



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FIGURE B-7. PROBABILITY PLOT FOR BOD



FIGURE E-8. PROBABILITY PLOT FOR BOD

B-9



FIGURE B-9. PROBABILITY PLOT FOR BOD

PROBABILITY PLOT







FIGURE B-11. PROBABILITY PLOT FOR BOD





FIGURE B-12. PROBABILITY PLOT FOR BOD B-11

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FIGURE B-13. PROBABILITY FLOT FOR BOD





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FIGURE B-15. PROBABILITY PLOT FOR BOD

PROBABILITY PLOT

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FIGURE B-16. PROBABILITY PLOT FOR BOD



FIGURE 8-17. PROBABILITY PLOT FOR BOD



FIGURE 8-18. PROBABILITY PLOT FOR TSS



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FIGURE B-19. PROBABILITY PLOT FOR TSS



FIGURE 8-20. PROBABILITY PLOT FOR TSS

**B-15** 





PROBABILITY PLOT















FIGURE B-25. PROBABILITY PLOT FOR TSS



FIGURE B-26. PROBABILITY PLOT FOR TSS



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FIGURE B-27. PROBABILITY PLOT FOR TSS

PROBABILITY PLOT



FIGURE 8-28. PROBABILITY PLOT FOR TSS

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B-19

Assuming that the distribution of the concentration c is lognormal, then  $y = \log(c)$  is normally distributed with mean y and variance  $\sigma^2$  (Aitchison and Brown, pages 8-9). Thus the 99th percentile on the natural log scale is

$$y_{0.99} = \mu + 2.326 \sigma$$

and the 99th percentile on the concentration scale is

$$c_{n,qq} = \exp(y_{n,qq}) = e^{\mu + 2.326 \sigma}$$
 (1)

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The mean and variance on the concentration scale are:

$$\mu + 1/2 \sigma^2$$

$$\mu_c = e$$

and

$$\sigma_c^2 = e^{2\mu + \sigma^2} (e^{\sigma^2} - 1).$$

Hence, the daily variability factor under the lognormal model is:

$$VF(1) = \frac{c_{0.99}}{\mu_c} = e^{2.326 \sigma - 1/2 \sigma^2}$$
(2)

Estimates of any of the above quantities are calculated by substituting the mean and variance of natural logs of the observations for  $\mu$  and  $\sigma^2$ , respectively.

To determine the variability factor for 30-day average concentrations, VF(30), it was necessary to take day-to-day correlation into account. Positive autocorrelation between concentrations measured on consecutive days means that such concentrations tend to be similar. The medians of plant-specific autocorrelations for one-day-apart concentrations were about 0.7 and 0.4 for BOD and TSS, respectively. An average of positively correlated concentration measurements is more variable than an average of independent concentrations.

A rigorous time series analysis to model the autocorrelation structure of each data set was not possible because of the many missing days' data in most data sets. Therefore, the correlation (p) between consecutive days' measurements (i.e., the lag-l autocorrelation) was estimated for each plant using the available data. Then using the first-order autoregressive model commonly found to be appropriate in water pollution modeling, the mean and variance of an n-day average were approximated by:

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$$\mu_{\overline{c}} = e$$
 (3)

and

$$\sigma_c^2 = \frac{\sigma_c^2}{n} fn(\rho)$$
 (4)

with

$$fn(\rho) = \frac{1+\rho}{1-\rho} - \frac{2\rho(1-\rho^n)}{n(1-\rho)^2}.$$

It can be seen in (4) that  $\sigma_c^2$  equals the variance of an average of n uncorrelated observations,  $\sigma_c^2/n$ , times a factor fn(p) that adjusts for the presence of autocorrelation. The correlation-adjustment factor is derived as follows using the fact that the covariance between concentrations k days apart is  $\rho^k \sigma_c^2$  under the first-order autoregressive model. Since

$$\overline{c} = \frac{1}{n} \sum_{i=1}^{n} c_i,$$

$$\sigma_c^2 = \frac{1}{n^2} \sum_{i=1}^{n} \sum_{j=1}^{n} \operatorname{cov}(c_i, c_j)$$

$$= \frac{1}{n^2} [n \sigma_c^2 + 2 \sum_{k=1}^{n-1} (n - k) \rho^k \sigma_c^2]$$
$$= \frac{\sigma_c^2}{n} [\frac{2}{n} \sum_{k=0}^{n-1} (n - k) \rho^k - 1].$$

The expression in brackets reduces to  $fn(\rho)$  with the help of the summation formula for arithmetico-geometric progressions:

$$\frac{n-1}{\sum_{k=0}^{\infty} (a + kr)q^{k}} = \frac{a - [a + (n - 1)r]q^{n}}{1 - q} + \frac{rq(1 - q^{n-1})}{(1 - q)^{2}}$$

taking a = n, r = -1, and  $q = \rho$  (Gradshteyn and Ryzhik, page 1).

Finally, since  $\overline{c}$  is approximately normally distributed by the Central Limit Theorem, the 95th percentile and variability factor of a 30-day average are approximately

$$\bar{c}_{0.95} = \bar{\nu}_{\bar{c}} + 1.645 \sigma_{\bar{c}}$$
 (5)

and

$$VF(30) = \bar{c}_{0.95}/\nu_{c}$$

$$= 1 + 1.645[(e^{\sigma^2} - 1)f_{30}(\rho)/30]^{1/2}$$
(6)

with  $\nu_{\overline{c}}$  and  $\sigma_{\overline{c}}^2$  defined by equations (3) and (4). Estimates of  $\overline{c}_{0.95}$  or VF(30) are calculated by substituting estimates of  $\nu$ ,  $\sigma^2$ , and  $\rho$  into the formulas above.

#### SPEARMAN RANK CORRELATION TECHNIQUE

Let  $(X_1, Y_1)$ ,  $(X_2, Y_2)$ ,..., $(X_n, Y_n)$  be a bivariate random sample of size n. The rank of  $X_i$ ,  $R(X_i)$ , as compared with the other X values, for i = 1, 2, ..., nis the position of  $X_i$  as the X values are ordered from smallest to largest. Thus, if  $X_k$  is the smallest X value,  $R(X_k) = 1$  and if  $X_1$  is the largest X value,  $R(X_1) = n$ . Similarly the values for Y can be ranked for i = 1, 2, ..., n. Once ranked, the data can be replaced with the rank pairs  $(R(X_1), R(Y_1))$ ,  $(R(X_2), R(Y_2))$ ,..., $(R(X_n), R(Y_n))$ . The Spearman rank correlation coefficient is calculated as follows:

$$R = \frac{\sum_{i=1}^{n} R(X_i)R(Y_i) - [1/2 (n + 1)]^2}{\frac{n(n^2 - 1)}{12}}$$

Based on R the following hypothesis can be tested:

- H<sub>o</sub>: The X<sub>i</sub> and Y<sub>i</sub> are mutually independent (i.e., their correlation is zero)
- H1: Either (a) there is a tendency for the larger values of X to be paired with the larger values of Y, or (b) there is a tendency for the smaller values of X to be paired with the larger values of Y.

By using influent or effluent concentrations for the X's and subcategorization variables for the Y's, the above hypothesis becomes a statistical test for significant subcategorization factors. Throughout this chapter the term "null hypothesis" refers to the hypothesis  $H_0$ : the X<sub>i</sub> and Y<sub>i</sub> are mutually independent.

Aside from the fact that a rank correlation statistically tests whether two variables are independent, it also does not assume a linear relationship between the variables. Consider Table B-2, where X and Y are two variables that exhibit a nonlinear relationship. The Pearson correlation coefficient, r, which assumes a linear relation between X and Y, is 0.6, where

$$r = \sum_{i} (X_{i} - \overline{X})(Y_{i} - \overline{Y}) / [\sum_{i} (X_{i} - \overline{X})^{2} \sum_{j} (Y_{j} - \overline{Y})^{2}]^{1/2}$$
  
$$\overline{X} = \frac{1}{n} \sum_{i} X_{i}$$

while the nonparametric Spearman Rank Correlation coefficient, R between R(X) and R(Y), is 1. Correlation coefficients are numbers which range between -1 and +1. Values of +1 indicate perfect associations or correlations, while a value of zero indicates no relationship.

 $\overline{Y} = \frac{1}{n} \sum_{i} Y_{i}$ 

For each of the rank correlation coefficients calculated, a graph has been attached which plots the rank pairs  $(R(X_i), R(Y_i))$ . A least squares line has been superimposed on the plot of the rank pairs to indicate graphically the degree of correlation. Each graph is labeled with the appropriate industry subcategory (e.g., Plastics Only).

At the bottom of each figure is the Spearman rank correlation coefficient, the sample size N, and the probability, p, that the null hypothesis,  $H_0$ , is true. Values of p of less than 0.05 indicate that a relationship exists, as specified in the alternative hypothesis,  $H_1$ . For ease of interpretation, Figures B-29 through B-33 show the theoretical lines for a sample size of 50 and Spearman rank correlations of -1, -0.5, 0, 0.5, and 1, respectively. As can be seen from these graphs, correlations of -1 and -0.5 indicate that as R(Y) decreases, R(X) increases; a correlation of 0 indicates that no relationship exists between R(X) and R(Y); and correlations of 0.5 and 1 indicate

<u> </u>	<u> </u>	R(X)	R(Y)
1	1	1	1
10	1.5	2	2
20	2	3	3
25	4	4	4
30	10	5	5
35	50	6	6
50	70	7	7
100	90	8	8
200	95	9	9
600	100	10	10

TABLE B-2. AN EXAMPLE OF VARIABLES WITH A NONLINEAR RELATIONSHIP

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FIGURE 5-30. REGRESSION LINE FOR RANK CORRELATION OF -0.5

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FIGURE B-31. REGRESSION LINE FOR RANK CORRELATION OF 0



FIGURE E-32. REGRESSION LINE FOR RANK CORRELATION OF 0.5



FIGURE B-33. REGRESSION LINE FOR RANK CORRELATION OF 1

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that as R(Y) increases, R(X) increases. It should be noted that the Spearman rank coefficient only indicates a dependent relationship between R(X) and R(Y). Derivation of the functional relationship between X and Y requires additional statistical techniques.

#### TERRY-HOEFFDING TEST

A common problem in statistics is the "two-sample problem" in which two populations are compared based on random samples of observations from each. This is precisely the problem one faces when trying to determine subcategories for a guideline: if two populations are different, then they represent different subcategories. For example, a statistical test which shows that Plastic plants' influent BOD levels are different from those of Not Plastic plants demonstrates the need for separately analyzing Plastic and Not Plastic plants.

In statistical terms, the problem is to test the null hypothesis that two population distributions have the same mean or median value of a property of interest. Random samples of sizes  $n_1$  and  $n_2$  are taken from the populations, a test statistic is computed from the sample data, and the value of the test statistic is used to decide whether the null hypothesis of identical population distributions should be rejected.

The most commonly used test for differences between population means is the two-sample Student's t test. Let  $y_i(i = 1, ..., n_1)$  and  $z_i(i = 1, ..., n_2)$ represent the sample observations from the two populations. Then Student's t statistic is

$$\mathbf{t} = (\bar{\mathbf{y}} - \bar{\mathbf{z}})/s \sqrt{n/n_1 n_2}$$
(7)

where

is the pooled sample size,

$$\overline{\mathbf{y}} = \frac{1}{n_1} \sum_{i=1}^{n_1} \mathbf{y}_i$$

and

$$\overline{z} = \frac{1}{n_2} \sum_{i=1}^{n_2} z_i$$

are the sample means, and

$$s^{2} = (n-2)^{-1} \left[ \sum_{i=1}^{n_{1}} (y_{1} - \overline{y})^{2} + \sum_{i=1}^{n_{2}} (z_{i} - \overline{z})^{2} \right]$$

is the pooled sample variance. The observed value of the t statistic is compared to tabled critical values of the t distribution with n-2 degrees of freedom to determine whether to reject the null hypothesis of equal population means.

The t test assumes that the population values are normally distributed with equal variances. When either of these assumptions fails to hold, conclusions of the test may be invalidated. One problem that may result is that the null hypothesis may be rejected with higher probability than assumed when it actually is true (i.e., the probability of a Type 1 error ( $\alpha$ ) may exceed the nominal  $\alpha$ -level). Another possible problem is that the t test may fail to detect existing population differences as well as it would if the assumptions held (i.e., its statistical power may be reduced). Either of these problems would have unfortunate consequences in subcategorization: false rejection of the null hypothesis could lead to unnecessary subcategories; failure to detect differences could result in failure to recognize needed subcategories. In order to avoid incorrect conclusions that could be caused

B-30

by failing to satisfy the assumptions behind the t test, a different test based on less restrictive assumptions was used.

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The Terry-Hoeffding test corresponds closely to the two-sample t test, but it assumes only that observations are drawn randomly from two continuous population distributions. For large samples, the Terry-Hoeffding test can be thought of as a two-sample t test based on "normal scores" rather than on the original observations. That is, before performing the t test, one replaces the rth largest observation in the pooled sample of n observations with the expected value of the rth largest observation from a random sample of size n from a standard normal distribution (E(r, n)). For large n, it is convenient to approximate normal scores by

$$E(r, n) \stackrel{*}{=} \phi^{-1}[r/(n+1)],$$
 (8)

since the inverse of the cumulative normal distribution function,  $\Phi^{-1}(\cdot)$ , is readily available in computer systems. For small n, values of E(r, n) are tabled in nonparametric statistics books (e.g., Bradley, page 326). The intuitive idea of the Terry-Hoeffding test is that replacing original observations (which may not be normally distributed) with normal scores leads to a more robust test (one whose validity is not as limited by underlying assumptions).

Because the sum of normal scores for the pooled sample is zero, the t statistic based on normal scores simplifies to

$$t = \sqrt{n-2} S / \left[ \frac{n_1 n_2}{n} \sum_{r=1}^{n} E(r, n)^2 - S^2 \right]^{1/2}, \qquad (9)$$

where S is the sum of normal scores for the sample with fewer observations (Bradley, page 152). The observed value of t is compared to critical values of the t distribution with n-2 degrees of freedom (just like the classical t test). A simpler approximation to this test for large samples is based on comparing

$$T = \left[\frac{n-1}{n_1 n_2}\right]^{1/2} S$$
 (10)

to critical values of the standard normal distribution (the mathematical justification for this approximation is given in Kendall and Stuart). For small samples, the Terry-Hoeffding test compares observed values of S to tabled critical values. Bradley (pages 327-330) gives critical values of S for pooled sample sizes up to n = 20.

The Terry-Hoeffding test has several advantages over the classical t test:

- It is distribution-free; i.e., one need not be concerned that violations of distributional assumptions will affect the probability of a Type I error.
- Its large-sample power is better except when the normality assumption holds (then, it is equivalent to the t in large sample power)
- It is less sensitive to extreme observations (outliers) than the t. For example, the actual value of the largest observation in the pooled sample doesn't affect the Terry-Hoeffding test, but can have a great impact on the classical t test.

Rendall and Stuart (page 520) note that it is "difficult to make a case for the customary routine use of Student's test" in comparing two populations when the sample numbers are reasonably large.

To illustrate the application of the Terry-Hoeffding test, the hypothetical influent BOD data from Plastics and Not Plastics plants in Table B-3 will be used. The table gives original observations and normal scores for ten Plastics plants and fourteen Not Plastics plants (the normal scores were calculated using formula (8)). The null hypothesis is that Plastics and Not Plastics plants do not differ in median influent BOD. The sum of normal scores for plastics plants is

**B-32** 

RANK (r)	BOD	SOURCE*	NORMAL SCORE (E(r,n))
1	150	P	-1.751
2	170	P	-1.405
3	190	P	-1.175
4	210	P	-0.994
5	230	Р	-0.842
6	250	P	-0.706
7	270	P	-0.583
8	320	NP	-0.468
9	370	NP	-0.358
10	420	P	-0.253
11	470	P	-0.151
12	520	NP	-0.050
13	550	P	0.050
14	590	NP	0.151
15	610	NP	0.253
16 `	630	NP	0.358
17	680	NP	0.468
18	730	NP	0.583
19	780	NP	0.706
20	800	NP	0.842
21	810	NP	0.994
22	870	NP	1.175
23	930	NP	1.405
24	990	NP	1.751

TABLE B-3. EXAMPLE OF SAMPLE DATA FOR THE TERRY-HOEFFDING TEST

**\*P** = Plastics

NP = Not plastics

so the test statistic based on formula (10) is

$$T = -3.71.$$

S = -7.81,

Since this is less than -1.96, the critical value for the normal distribution with  $\alpha = 0.05$  for a two-tailed test, we reject the null hypothesis at this  $\alpha$ level. That is, we conclude that Plastics plants are different from Not Plastics plants in the example. Note in Table B-3 that observed BOD values (and normal scores) for Plastics tend to be lower than values for Not Plastics.

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................

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### APPENDIX C

## See Chapter 13 of the BAT Volume of this report for the glossary.

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#### APPENDIX D

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Rationale for Exclusion of Daily Data Base Plants From Variability Analysis

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#### APPENDIX D

### Rationale For Exclusion of Daily Data Base Plants From Variability Analysis

<u>Plant No. 1</u>: The plant upgraded its treatment system significantly during the first data period and since the second data collection period, and no data are available for the upgraded plant. The aerated lagoons were converted to activated sludge in 1980. In addition, the effluent daily data for both periods included unqualified cooling water dilution. Plant personnel expressed doubts as to available data's validity.

<u>Plant No. 3</u>: The treatment system for this plant was upgraded significantly during the data collection period. The aerated lagoons were replaced by activated sludge units after August 1976. In addition, the plant had operational problems from July 1978 to February 1980, including clarifier shutdowns, clarifier floodings and difficulties with the equalization basins, which make the data unrepresentative of a steadystate operation.

<u>Plant No. 292</u>: This plant is not considered representative of Organic Chemical Industry plants due to the inclusion of municipal effluent in its influent. The municipal flow may be as much as 25 percent of the total influent, and therefore is considered a significant, and atypical, influent stream.

<u>Plant No. 18</u>: During the data collection period, this plant underwent major expansion and modification. Screening, equalization, increased capacity, chlorination facilities and sludge handling facilities were implemented in 1976. Available data are, therefore, not considered to be from a steady-state system. Also, the effluent sample point is upstream of the final clarifiers, and therefore not representative of actual plant performance.

<u>Plant No. 293</u>: This plant is a poor performer (BOD removal 89%, effluent BOD 87 mg/l). Available information indicates poor solids control may be the cause of poor performance (TSS removal 66%).

<u>Plant No. 20</u>: No BOD, COD or TSS effluent data are available for this plant.

Plant No. 24: No BOD, COD, or TSS data are available for this plant.

<u>Plant No. 28</u>: The treatment system of this plant does not conform to accepted engineering practice. In particular, the use of chlorination before trickling filters is noted as being extremely unusual, and not in accordance with accepted wastewater treatment methodology.

<u>Plant No. 42</u>: The poor performance by this plant seems to be related to poor operational procedures. Flow is reported to be alternated between two equalization lagoons, with the switch between the two noted as a cause for high effluent BOD. Steady-state operation is also reported to be interrupted by periodic sludge wasting from the anaerobic lagoon. The treatment train used at this plant (anaerobic lagoon followed by activated sludge) is unorthodox. Plant was zero discharge until 1978. The data period is therefore shortly after startup and may not represent steady-state operations.

<u>Plant No. 53</u>: The treatment system for this plant has been upgraded and no data are available for the upgraded system. Plant had aerated lagoons, but they were inadequate for loadings received. The plant was converted to activated sludge in 1977 in order to improve solids control (effluent TSS 74 mg/l).

<u>Plant No. 60</u>: This plant achieved poor treatment levels (BOD removal 88%, effluent BOD 54 mg/l).

<u>Plant No. 61</u>: This plant has upgraded its treatment system due to poor performance, and no data are available for the upgraded system. Increased pretreatment, neutralization and dual media filters were added in 1977 in order to improve poor (BOD removal 79%, effluent BOD 81 mg/1) plant performance. (Available information indicates that insufficient air supply to aeration basins and poor solids control may have been responsible for the poor performance noted during the data period.)

Plant No. 72: There are no continuous data for this plant available at present.

<u>Plant No. 73</u>: This plant phased out a major production unit during collection period, which resulted in an estimated 70 percent reduction in influent BOD load from 12/75 through 12/76. Thus, the available data are not for a system operating under steady state conditions.

<u>Plant No. 74</u>: The data available for this plant are not representative of actual treatment system performance. The available effluent daily data include an unquantified stormwater stream. In addition, no BOD, TSS or COD data are available; only TOC and flow data were reported.

<u>Plant No. 75</u>: The data available for this plant are not representative of actual treatment system performance. The available effluent daily data include an unqualified waste stream which consists of untreated boiler blowdown, stormwater runoff and cooling water. No daily data are available for this stream.

<u>Plant No. 87</u>: For this plant, the effluent sampling site was downstream from a mixing point of biological and nonbiological effluents (i.e., effluent is diluted). Also, no BOD or TSS data are available for this plant.

<u>Plant No. 89</u>: The data available for this plant do not accurately represent actual treatment plant performance. The effluent daily data include an unquantified stream which consists of untreated process water and stormwater runoff. No daily data are available for this stream.

<u>Plant No. 90</u>: The data available for this plant are not representative of actual treatment plant performance. Available effluent daily data

include an unquantified stream consisting of untreated cooling water. No daily data are available for the cooling water stream.

Plant No. 103: No effluent BOD, COD or TSS data are available for this plant.

Plant No. 106: Effluent data for this plant contain cooling water dilution. This data is therefore not representative of actual treatment plant performance.

<u>Plant No. 109</u>: Data available for this plant are not representative of actual treatment plant performance. Effluent data contain dilution by an unqualified wastestream for which only <u>estimated</u> average BOD is reported.

<u>Plant No. 120</u>: This plant treats petroleum refinery wastewater as well as organic chemical wastewater, and is therefore not representative of treatment plants in the OCPS data base.

Plant No. 123: Data available for this plant are not representative of actual treatment plant performance. Available effluent data include stormwater dilution.

<u>Plant No. 124</u>: The data avilable for this plant are not representative of actual treatment plant performance. The available effluent data include an unquantified stream consisting of untreated stormwater. No daily data are available for the stormwater stream.

Plant No. 138: This plant achieved poor removals during the period for which data are available. (Effluent BOD 251 mg/l.) This appears to be caused by inadequate treatment system design. In particular, the lack of final clarification is noted.

Plant No. 146: Data available for this plant are not representative of actual treatment plant performance. Available effluent data include stormwater dilution.

<u>Plant No. 176</u>: There are only three months of data available for this plant.

Plant No. 245: This plant has been rejected due to the use of a nonbiological (air stripping) treatment system.

Plant No. 268: This plant was found to have high effluent BOD (248 mg/l) and BOD removal below 95% (93% BOD removal).

<u>Plant No. 269</u>: The treatment system of the plant has been upgraded and no data are available for the upgraded system. Additions to the plant in 10/77 included primary settling, equalization, increased secondary clarification capacity and sludge handling modifications. Prior to these modifications plant performance was poor (BOD removal 83%, effluent BOD 66 mg/1). Plant No. 274: A significant portion of flow treated by this plant is sanitary wastewater.

Plant No. 294: Plant performance (effluent TSS 95 mg/l) is poor considering the level of treatment technology employed (activated sludge followed by sand filtration).

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## APPENDIX E

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## CAPDET Methodology

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#### APPENDIX E

#### METHODOLOGY

The basic calculation tool used to develop alternative engineering costs (with the exception of RBC costs) is the computer program, CAPDET (Computer Assisted Procedure For The Design and Evaluation of Wastewater Treatment Facilities).[E-1] The CAPDET computer model was developed jointly by the Corps of Engineers Waterway Experiment Station, Vicksburg, Mississipi and the EPA Office of Water and Waste Management.

The major purpose of the CAPDET model is to provide for the rapid design, cost estimating and ranking by cost of wastewater treatment plant alternatives. The model can be applied to industrial wastewater treatment system design, as well as Publicly Owned Treatment Works (POTWs), with modifications of selected computer program default values.

The CAPDET model provides flexibility and accuracy while maintaining ease of operation. The algorithm contains default values applicable to all wastewater parameters. These default values enable complete wastewater treatment design and costing by stipulating only average flow and a list of the treatment schemes to be considered. Any value in the CAPDET model (chemical data, cost data, unit operations parameters, etc.) can be varied, however, to reflect more current information, more localized cost data or specific waste treatment parameters applicable to industrial wastewater treatment.

The CAPDET model is not designed to select and sequence unit operations into a treatment system. The user must input a series of up to 20 "blocks" or waste treatment process functions into the model. Thus, the user inputs the basic treatment scheme to be costed into the model.

In each treatment block, up to ten different treatment alternatives can be considered. For example, a block containing complete mix activated sludge could also contain conventional activated sludge, extended aeration activated sludge, pure oxygen activated sludge and six other alternatives. The model will then design, estimate costs, and rank from cheapest to most expensive all possible combinations. Further, the model can also consider four separate treatment trains per run.

Another characteristic of the CAPDET model is its ability to consider up to three modifications of each treatment process. The model contains the process treatment parameters necessary to solve the unit process equations, namely chemical default data for certain wastewaters and such typical physical criteria as removal efficiency, overflow rate and average temperature. The CAPDET model allows the user direct access to any of these parameters. In fact, the user can run three different versions of any treatment process as alternatives in the same block or in different blocks. An example would be checking in one run the impact on effluent quality and plant costs of changing the overflow rate of a clarifier. CAPDET, then, is able to consider many different alternative treatment schemes, including multiple variants on one unit process.

After the treatment process alternatives to be considered have been selected and input into the model, each possible alternative treatment train is designed. The CAPDET model contains in its treatment catalog 64 large facility (>.5 MGD) and 23 small facility (<.5 MGD) unit processes, including sludge handling processes. Certain common industrial processes (API oil separation, steam stripping etc.) were not included in CAPDET because of CAPDET's original sanitary waste orientation. The writers of CAPDET, however, have included as part of the model a process known as the "dummy" process. To use this procedure, one calculates the percentage reduction for any parameters affected by the applicable unit process, plus any increase or decrease in effluent flow rate and sludges due to the process. These numbers are entered in the model and the dummy process is treated as if it were a process in the CAPDET treatment catalog. Like any other unit process, different modifications of the dummy processes can be considered as treatment alternatives. This flexibility enables the CAPDET model to simulate unit processes not in its treatment process file.

As the CAPDET model designs all possible alternative treatment systems, it determines the cost of each system. As with the design data, a complete set of default unit construction, operation and maintenance cost data are contained in the model. Thus, the cost of the treatment system can be calculated. For each treatment process, secondary design calculations are performed by the computer to determine the specific amounts of materials required: foundations, tanks, basins, walls and many other items which are individually designed by the model to arrive at values for such construction items as concrete, sand and gravel, size and length of pipe, chemical use, number of valves and pounds of steel plate. These items are then totaled for the specific treatment train being considered and the construction costs are computed. Equipment costs are calculated parametrically, with an attempt to optimize accu-The equipment cost data were provided by vendors of treatment racy. equipment. The parameter was chosen which best represented the cost of that equipment as a straight line on a log-log plot (clarifier diameter being a more appropriate parameter than total clarifier flow rate for costing a clarifier mechanism). The cotal capital cost of the treatment system is, therefore, a hybrid of construction design calculations and parametric cost estimates.

Default values for all cost data are contained in the CAPDET model. As with all other data in the model, however, the user has the option of altering or updating the costs for nearly anything in the model. These changes can be made in the following ways:

- 1. Equipment and construction default data are based on first quarter 1977 costs. The model updates these cost data to the present.
- 2. Equipment costs, operating costs, operating manhours per year and construction material costs are all user selectable. Thus, if a user knows that standard blower costs are lower or highter than the default value in a particular area, then correct costs can be input. If parametric cost curves are to be updated, the user inputs the local cost for the standard sized unit (i.e., 90-foot clarifier mechanis<sup>r</sup>) used in the model. The cost curve then will be adjusted

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up or down, with the slope remaining unchanged. If local or current data are input, they will not be updated again by one of the cost indices.

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- 3. Some costs are given as percentages of the total facility costs (e.g., administrative costs and inspection costs). These percentages can be changed depending on the judgment of the user or on local conditions.
- 4. Other costs are also accessible to the user. CAPDET calculates labor rates based on ratios with an Operator II rate. While these ratios are not user selectable, the default Operation II rate is alterable. Thus, the CAPDET model can account for variations in labor costs.

The CAPDET model also has some optional cost items which can be included as appropriate (i.e., special foundations, mobilization, site electrical costs, lab and administration buildings, etc.).

The CAPDET model compares treatment alternatives rather than adding unit operations until the effluent is below the desired efflent level. The model calculates the expected effluent from each alternative train and compares it with the desired effluent level. The model rejects those alternatives that do not meet the desired effluent quality and ranks (from the least expensive to the most expensive) the least expensive hundred alternatives that meet the criteria.

The general design bases for the CAPDET model were:

- 1. Costs expressed as first quarter 1979 dollars.
- Capital costs representing total equipment costs (installed), contractor's overhead/profit, administration/legal expenses, architectural/engineering design fees, inspection fees, contingencies, technical costs and land costs.
- 3. Operating costs equal to the sum of costs for operating and maintenance labor, power, materials, chemicals, and administration and laboratory expenses.
- 4. Equivalent annual costs representing the amortization of the capital costs (capital recovery plus interest) at 10 percent and twenty years plus the annual operating costs.[E-2],[E-3]
- 5. Design constants based on a statistical analysis of the municipal wastewater treatment industry as well as on literature publications.[E-4]
- 6. Cost equations taken from correlations developed from data based on municipally owned treatment plants.[E-5]

Unit price input data and details of the direct nonconstruction costs are shown in Table E-1. Default waste characteristics present in the

# TABLE E-1-COST INPUT PARAMETERS (10 PERCENT, 20 YEARS)

COST INDEXES	UNIT PRICES		
Building	48.00	\$/sg ft	
Excavation	1.20	\$/cu yd	
Wall concrete (reinforced, in-place)	207.00	\$/cu yd	
Slab concrete (reinforced, in-place)	91.00	\$/cu yd	
Marshall and Swift Index (equipment cost)	577.0 <b>0</b>	-	
Crane rental	67.0 <b>0</b>	\$/hr	
EPA Construction Cost Index	132.00		
Canopy roof	15.75	\$/sg_ft	
Labor rate (equipment installation)	13.40	\$/h <b>r</b>	
Operator Class II	7.50	\$/hr	
Electricity	0.04	\$/kwhr	
Chemical Cost			
Lime	0.03	\$/lb	
Alum	0.04	\$/lb	
Iron salts	0.06	\$/lb	
Polymer	1.62	\$/1Ь	
Engineering News Record Cost Index	2886 <b>.00</b>		
Handrail, (in-place)	25.20	\$/ft	
Pipe Cost Index	295.20		
Pipe installation labor rate	14.70	\$/hr	
Eight-inch pipe	9.08	\$/ft	
Eight-inch pipe bend	86.82	\$/unit	
Eight-inch pipe tee	128.49	\$/unit	
Eight-inch pipe valve	1346.16	\$/unit	

## INDIRECT NON-CONSTRUCTION COSTS

Miscellaneous non-construction cost Admin/legal Inspection Contingencies Profit and overhead (contractor's) Technical cost

## (AS % OF CONSTRUCTION COST)

5.0% 2.0% 2.0% 11.5% 22.0% 2.0% CAPDET model are summarized in Table E-2. Table E-3 lists removal efficiencies of pollutants built into the program.

An evaluation of CAPDET suggests the following limitations:

1. CAPDET was originally designed for municipal wastewater plants and therefore requires certain adjustments for industrial applications. The characteristics of municipal raw wastewaters are normally more consistent among different locations than those of industrial wastewater, and the parameters relating to the treatability are more applicable among the different locations than are the treatment parameters of industrial wastewaters.

This permits the use of CAPDET for the POTW cost estimating without changes in the default values. The use of CAPDET for evaluations based on industrial wastewaters requires changes in the default values, particularly in the reaction rate constant, influent BOD concentrations, influent TSS, nutrient balances and other factors which differentiate the composition of industrial wastewaters from municipal wastewaters.

- 2. The costs generated by CAPDET reflected grass roots installations. This version of CAPDET is not capable of directly generating upgrade costs for modifications or replacements of existing equipment. Many OCPS plants may require only minor adjustments to current facilities or management practices to reach proposed effluent targets. Although these engineering options may be available to specific plants to achieve the target effluent concentrations, CAPDET, as used in this study, is limited to estimating the designs and costs for entire wastewater treatment unit process additions (second stage treatment).
- 3. Land requirements and administrative and laboratory costs calculated by CAPDET are related to plant capacity rather than to discrete unit operations. The relationship of costs to flow rate limits CAPDET in the estimation of discrete (singular) units as opposed to entire plant facilities. As a result, CAPDET predicts similar land costs for clarification and an entire activated sludge unit for the same flow rate. As a percentage of the total costs, land, administrative and laboratory costs represent a small portion of an entire facility estimate. For a single unit, these costs have a much greater effect.

## TABLEB-2

## CAPDET DEFAULT INFLUENT WASTE CHARACTERISTICS

TEMPERATURE	18 °Ç	
SUSPENDED SOLIDS	200	MG/L
VOLATILE SOLIDS	60	% OF SS
SETTLEABLE SOLIDS	15	MG/L
BOD5	25 <b>0</b>	MG/L
SBOD	75	MG/L
COD	500	MG/L
SCOD	400	MG/L
Ъң	7, 6	
CATIONS	160	MG/L
ANIONS	160	MG/L
PO4	18	MG/L
TKN	45	MG/L
NH <sub>3</sub>	25	MG/L
NO2	0	MG/L
NO <sub>3</sub>	0	MG/L
OIL AND GREASE	80	MG/L

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	WASTE CHARACTERISTIC REMOVALS FOR											
PROCESS	BOD5	TSS	COD	OIL & GREASE	דונא	PIIOS	NH3	Settle SOLII				
Dissolved Air Flotation	30%	80%	30%	-	10%	-	-	-				
Clarifica- tion	32%	58%	40%	-	5%	5%	-	-				
Activated Sludge	USER INPUT INFLUENT AND EFFLUENT	USER INPUT TO SECODARY CLARIFIER	1.5 × BOD <sub>EFF</sub>	0	30%	30%	SET EQUAL TO TKN	0				
Aerated Lagoon	USER INPUT INFLUEN'C AND EFFLUEN'C	USER INPUT TO SECONDARY CLARIFIER	ASSUME	SAME AS ASL								
Multimedia Filtration	SET EFFLUENT EQUAL TO BOD SOLUBLE	60%	SET EFFLUEN EQUAL TO COD <sub>SOLUBLE</sub>	т 0 :	PASS ON THROUGH	PASS ON THROUGH	PASS THROUGH	0				
	INFLUENT		INFLUENT									

#### WASTE CHARACTERISTIC REMOVAL DEFAULT VALUES FOR CAPDET PROCESSES

TABLE E-3

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E-7

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### APPENDIX <sup>P</sup> - DISCRETE UNIT COST CURVES

## 1 GENERAL

This appendix supplements the engineering cost section of this report (Section 8) with a detailed presentation of the discrete unit cost curves discussed in Paragraph 8.2.3.

The curves presented contain costs for the treatment technology alternatives considered in the plant-by-plant analysis for various influent and effluent concentrations (mg/l) and flow rates (MGD). The terms frequently listed on the curves are defined as follows:

(1) <u>Capital Costs</u> are the costs representing the total installed equipment costs, contractor's overhead/profit, administration/legal expenses, A/E design fees, inspection fees, contingencies, technical costs, and land costs.

(2) <u>Operating Costs</u> are the costs equal to the sum of costs for operating and maintenance labor, power, materials, chemicals, and administration and laboratory expenses.

(3) <u>Equivalent Annual Costs (Annual Costs)</u> are those costs representing the amortization of the capital costs (capital recovery plus interest) at 10 percent and twenty years, plus the annual operating costs.

(4) <u>M\$, M\$/yr</u> are costs expressed as first quarter 1979 million dollars (per year for operating and equivalent annual costs).

The procedure for obtaining the costs used for the plant-by-plant analysis, is as follows:

(1) Locate the set of curves for the technology being considered.

(2) Look for the influent concentration (reported effluent) for BOD in the upper right hand corner of the page among the various graphs. Note that solids removal technologies are directly related to flow and that concentration levels are only limiting conditions of the treatment alternative. The influent concentrations are not listed on the graphs for solids removals alternatives.

(3) Select the capital cost (left axis) corresponding to the flow (bottom axis) and effluent concentration (curve) from the capital, operating, and annual cost pages. One cost should be obtained from each page for a given flow, influent BOD and effluent BOD. The costs for solid removal are flow related and all three costs can be obtained from one page for each technology alternative.

(4) Repeat this procedure for each target and continue for any plant being considered.

(5) For influent concentrations, assume a 15 percent accuracy range. Occasionally interpretation between two sets of costs may be required. This estimation technique is within the accuracy of the costs.

The cost curves are presented for a variety of effluent BOD concentrations for each biological alternative. Given a relative accuracy of 15 percent for the BOD reported value, the cost curves are basically equivalent for BOD targets with the same range. For example, the costs for a target of 50 mg/l BOD are approximately equivalent for targets ranging from 40 to 60 mg/l. The range for 30 mg/l is 25 to 35 mg/l. For estimating purposes, the curves for 10 mg/l represent the 20 mg/l target. In many cases, there is a lack of sensitivity in the costs for changes in effluent targets at lower influent concentrations and flows. A discussions of the sensitivities of the costs are provided in Section 8. The design bases and cost model limitations are also discussed in Section 8.

From the information and bases presented in this appendix, additional analysis of the plants may be pursued.





# FIGURE 1-DISSOLVED AIR FLOTATION ANNUAL, CAPITAL AND OPERATING COSTS



FLOW (MILLION GAL/DAY)

# FIGURE 2-SEDIMENTATION ANNUAL, CAPITAL AND OPERATING COSTS





# FIGURES 3-ACTIVATED SLUDGE ANNUAL (85mg/1)



### INFLUENT = 85 mg/L BOD

FIGURE 4—ACTIVATED SLUDGE CAPITAL COSTS (85 mg/l)



INFLUENT = 85 mg/L BOD

FIGURE 5-ACTIVATED SLUDGE OPERATING COSTS (85 mg/l)



INFLUENT = 100 mg/L BOD

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FIGURE 6-ACTIVATED SLUDGE ANNUAL COSTS (100 mg/l)



INFLUENT = 100 mg/L BOD

FIGURE 7-ACTIVATED SLUDGE CAPITAL COSTS (100 mg/l)



INFLUENT = 100 mg/L BOD

FIGURE 8-ACTIVATED SLUDGE OPERATING COSTS (100 mg/l)



# INFLUENT = 150 mg/L BOD

FIGURE 9-ACTIVATED SLUDGE ANNUAL COSTS (150 mg/l)

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INFLUENT = 150 mg/L BOD

FIGURE 10-ACTIVATED SLUDGE CAPITAL COSTS (150 mg/l)


FIGURE 11-ACTIVATED SLUDGE OPERATING COSTS (150 mg/l)



FIGURE 12-ACTIVATED SLUDGE ANNUAL COSTS (250 mg/l)



# FIGURE 13-ACTIVATED SLUDGE CAPITAL COSTS (250 mg/l)



## FIGURE 14—ACTIVATED SLUDGE OPERATING COSTS (250 mg/l)



FIGURE 15

## ACTIVATED SLUDGE ANNUAL COSTS (400 mg/l)



FIGURE 16-ACTIVATED SLUDGE CAPITAL COSTS (400 mg/l)



FIGURE 17-ACTIVATED SLUDGE OPERATING COSTS (400 mg/l)



FIGURE 18-ACTIVATED SLUDGE ANNUAL COSTS (500 mg/l)

F-20

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FIGURE 19-ACTIVATED SLUDGE CAPITAL COSTS (500 mg/l)



FIGURE 20-ACTIVATED SLUDGE OPERATING COSTS (500 mg/l)



BOD INFLUENT = 1,000 mg/L

FIGURE 21-ACTIVATED SLUDGE ANNUAL COSTS (1000 mg/l)



FIGURE 22\_ACTIVATED SLUDGE CAPITAL COSTS (1000 mg/l)



FIGURE 23-ACTIVATED SLUDGE OPERATING COSTS (1000 mg/l)



FIGURE 24-ACTIVATED SLUDGE ANNUAL COSTS (1500 mg/l)



FIGURE 25-ACTIVATED SLUDGE CAPITAL COSTS (1500 mg/l)



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INFLUENT = 1500 mg/L BOD

FIGURE 26-ACTIVATED SLUDGE OPERATING COSTS (1500 mg/l)

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## INFLUENT = 2000 mg/L 80D

FIGURE 27-ACTIVATED SLUDGE ANNUAL COSTS (2000 mg/l)



FIGURE 28—ACTIVATED SLUDGE CAPITAL COSTS (2000 mg/l)







FIGURE 30-AERATED LAGOONS ANNUAL COSTS (100 mg/l)

INFLUENT = 100 mg/L BOD



FIGURE 31—AERATED LAGOONS CAPITAL COSTS (100 mg/l)



FIGURE 32—AERATED LAGOONS OPERATING COSTS (100 mg/l)



FIGURE 33

## AERATED LAGOONS ANNUAL COSTS (200 mg/l)



FIGURE 34-AERATED LAGOONS CAPITAL COSTS (200 mg/l)



FIGURE 35-AERATED LAGOONS OPERATING COSTS (200 mg/l)



FIGURE 36-AERATED LAGOONS ANNUAL COSTS (500 mg/l)



FIGURE 37-AERATED LAGOONS CAPITAL COSTS (500 mg/l)



FIGURE 38-AERATED LAGOONS OPERATING COSTS (500 mg/l)



# FIGURE 39—AERATED LAGOONS ANNUAL COSTS (1000 mg/l)



FIGURE 40-AERATED LAGOONS CAPITAL COSTS (1000 mg/l)



FIGURE 41-AERATED LAGOONS OPERATING COSTS (1000 mg/l)



FIGURE 42-ROTATING BIOLOGICAL CONTACTORS ANNUAL COSTS











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FIGURE 45

# MULTI-MEDIA FILTRATION ANNUAL, CAPITAL AND OPERATING COSTS

APPENDIX G

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#### Wastestream Data Listing

G-1
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}***********		**********			SUBCAT=PL	ASTICS ONLY				*	
PLANT NUMBER *	n PP	FLOWINF (MgD)	FLOWEFF (MGD)	BODINF (PPM)	HODEFF (PPM)	CODINF (PPM)	CODEFF (PPH)	T8SINF (PPM)	T8SEFF (PPH)	TOCINF (PPM)	TOCEFF (PPN)
			# <b>47000</b>								
2	2	1.83	1.83	•	3.00	•	10.00		27.00	•	•
3	2	0,29	0,30	1394,00	45,00	2000,00	113,00	22,00	32,00	464,00	42,00
4	3	0.07	•	•	•	_ •	•	237.00	÷	31,00	•
7	2	0.04	•	12542.00	٠	18549,00	•	1055.00	•	•	•
9	2	0.07	0.07	118,00	8,00	•	•••••	28.00	31,00	•	74,00
10	3	0.71	0,76	•	7.00	•	30.00	٠	14.00	•	•
11	1	."		•	•	•	•	•	٠	•	•
11	:	0,50	9,20	•	•	•	٠	•	•	•	•
12	1		•	٠	•	•	•	٠	٠	•	٠
13	<u> </u>	2.00	•	•	•	•	•	•	•	•	• •
17	i	0.00	0.08	1200 00	52.00	2874 00	188 00	•	30 00	•	
17	ž	0.03	4.43	1500.00	36.00	2310.00	100.00	•	30.00	•	•
19	Á	1.40	<b>1</b> .80	237 00	7.00	470 00	•	•	15.00	•	•
0 25	4	0-17	0 14	9.00		42.00	42.00	31.00	12.00	9.00	•
27	ŝ	10.60	10.70	131.00	23.00	258.00	120.00	38.00	25.00		•
29	ž	8.70	8.70	114.00	7.00	210.00	115.00	86.00	114.00	•	17.00
30	3	0.04					•				
33	ī	5.24	5,22								
34	1	0.08	0.06	666.00	35.00	1669.00	194.00		23.00		
35	1	0.70	•	388,00		416.00	•	0.30	•		
39	4	0.39	0,39		6.00	•		•	9,00	•	•
44	1	0.67	0,67	754,00	9,00	1234.00	83,00	2898.00	22,00	•	•
45	2	2.22	2,22	390.00	3.00	639,00	57.00	285,00	24,00	•	•
46	2	•	•	•	•	•	•	•	•	•	٠
47	5	٠	•	•	•	•	•	•	•	•	•
47	•	•	•	•	•	•	•	•	•	٠	•
48	1			•	••••	•		•		•	•
56	e .	0.00	0.86	104.00	8,00		67.00		13.00	•	٠
34		0.20	V, CZ	424.00	30.00	05/.00	110.00	44.00	43.00		•
50	1	0.01	٠	1991-00	•	0002,0V	•	200,00	•	1430.00	•
50	•	0.00	٠	. •	•	•	•	•	•	•	٠
65	i	8.70	8.70	349.00	18.00	710.00	110.00	153.00	52.00	•	•
67	1	0.74	~	344800	10.00	110.00	110.00	133400	JE . VV		•
68	ż	•	•	•	•	•	•	•	•	•	•
70	2		•		•	•	•		•		•
73	3	0.25	0.25	14.00	3,00	473,00	38,00	1359,00	11,00	•	•

\*Note: Each line entry represents a separate waste stream. Plant code numbers are repeated for plants with multiple waste streams.

### WASTE STREAM Data Listing - Subcategorization file

						SUBCATEPLA	STICS ONLY					
	PLANT NUMBER	# Pp	FLOWINF (MgD)	FLOH <b>eff</b> (Mgd)	BODIN <b>F</b> (PPM)	BODEFF (PPH)	CODIN <b>F</b> (PPM)	CODEFF (PPM)	TS81NF (PPM)	T88EFF (PPH)	TOCINF (PPH)	TUCEFF (PPM)
									44444			
	75	2	0.42	0.42	170.00	24.00	876.00	176.00				
	77	Ā	0 94	0,94	2.00	2,00	27.00	27,00	10,00	10,00	•	•
	77	•	0,01	•	895,00	•	48.00	•	16.00	•	•	•
	77	•	•	•	•	•	•	•	•	•	•	•
	78	1	•	•	•	•	•	•	•	•	•	•
	79	2	0_03	•	•	•	•	•	•	•	•	•
	85	2	•	•	•	•	•	•	•	•	•	•
	82	:	<b>~•</b> •	A <sup>•</sup> ••	•	•	•	•	•	•	•	•
	84	-	0 14		•	•	•	•	•	•	. •	•
	80		1 71	1 71	979 00	AA 00	1908 00	A20 00	101 00	76.00	•	• •
	99	2			51 00	40,00	97 00	15.00	20.00	15.00	•	•
	91	2	2 06	2.04	174 00	10.00	329.00	69.00	79.00	84.00	•	•
	91	-	0_47								•	•
G	93	i	0.12	0.12	80.00	15.00	504.00	146.00	205.00	78.00		
1	96	i	3.00	3.00	128.00	3.00			24.00	19.00		
10	100	1	1.00	1.00	424,00	15,00	607.00	98,00	106.00	37.00	•	
	100	Ĩ	0.85	0.85		10.00	•	•	•	4,00	•	•
	104	7	1,54	1,54	1082.00	11,00	1557,00	115,00	62.00	60,00	434.00	45,00
	104	•	0.10	•	•	•	•	•	•	•	•	•
	104	•	•	•	•	•	•	•	•	•	•	•
	104	•	•_	•	•	•	· · · •			• • • •	•	•
	105	1	0.30	0.30	1847.00	6.00	2689.00	131,00	<b>61</b> ,00	32.00	•	•
	106	6	5,95	5,95	94.00	11.00	179.00	53.00			•	•
	107	Į	2,16	2,55	215.00	47.00	405.00	1/5.00	142,00	127.00	300 00	B4 <sup>1</sup> 00
	109	3	1,57	1,57	427.00	12.00	373 40	17.00	121-00		240,00	46.00
	111	1	0.93	0.43	42.00	8.00	73 00	58 00		26 00	25 00	19 00
	121	2	1,03	1,05	12.00	10.00	807 00	96 00	A1.00	18 00		
	124	2	0 36	0 16	548.00	17.00	874 00	61.00	A3.00	22.00	•	-
	125	5	0.52	0.52	1764.00	45.00	3173.00	205.00	187.00	30.00	•	
	126	3	0.35	0.35	1021.00	8.00	2292.00	102.00	561.00	28.00		
	129	3	••••			••••						
	131	ĩ	0.00	-								-
	132	Ī	0.28	0.28		21.00		130,00	-	13,00	-	•
	140	2	• • -		-	•	•	•	•	•	•	•
	140	•	-	-	-	•	•	•	•	•	•	•
	146	5	0,35	0.35	187_00	6.00	1681.00	45.00	2547.00	23.00	•	•

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### WASTE STREAM Data listing - Subcategorization file

PLANT NUMBER 147 148 150 152	# Pp 	FLOHINF (MGD) 0.30 0.58 1.96	FLOHEFF (MGD) 0,29	80DINF (PPH) 1000,00	80DEFF (PPH) 	CODINF (PPH)	CODEFF (PPH)	T85INF (PPM)	T83EFF (PPH)	TOCINF (PPH)	TOCEFF (PPH)
NUMBER 147 148 150 152	# PP  3 1 6 1 1 3	(MGD) 0.30 0.58 1.96	(MGD) 0,29 0,63	(PPH)  1000.00	(PPH)  9.00	(PPH) 	(PPH) 	(PPM)	(PP+) 	(PPH)	(PPH)
147 148 150 152	3 1 6 1 1 3	0,30 0,58 1,96	0,29	1000.00	9.00						
147 148 150 152	3 1 6 1 1 3	0.30 0.58 1.96	0,29	1000.00	9.00					3	
148 150 152	1 6 1 3	0,58	0.63	_	• ·	2152,00	80,00	868.00	29.00	•	•
150 152 153	6 1 1 3	0.58	0.63		•	•	•		•	٠	•
152	1 - 1 - 3	1.96		447.00	43.00	•	124,00	•	11.00	•	•
163	13		1.96	•	7.00	•	58,00	•	21,00	•	•
1 26	3	0.00	0.06	•	4.00	. •	49.00	•	16.00	•	•
154		0.07	•	•	•	•	•	•	•	•	•
155	6	• •	•	•	•	•	•	•	•	•	•
155	•	10.70	•	•	•	•	•	•	•	•	•
155	•	0.50	•	•	•	•	•	•	•	•	•
156	2	1.06	0.00	•	• • • • •	•		14.00	••••	•	٠
157	1	1.72	1.31	•	18.00	•	37.00	•	52.00	•	•
161	2		٠	•	٠	•	•	•	•	•	•
167	1	0.00	•	•	•	•	•	٠	٠	•	•
108	1		•		•	•	•	•	•	••••	•
173	2	0.00		8500.00	••••		( <b>*</b> ***	•	••••	800.00	•
174	4	0.04	0.04	372.00	9.00	1178.00	62.00	*	14,00	•	•
1/9	2	0,48	0,48	578.00	10.00	1195.00	140.00	78.00	37.00	•	•
184		1.02	1.02	•	16,00	•	02.00	•	26.00	•	•
105	1		.*	•	• • • • •	•	•••	•		•	•
104	÷.	0.32	0.34	••••	80,00		334,00	•	38.00		•
172	2	0.13	0,11	003.00	21,00	1000.00	100.00		¥0.00	1218*00	•
194	č	0.34	0.34	7.00	7.00	<b>•••</b> •••	· • • • • •	10,00	10.00	•	٠
170	-	3.70	3.70	20.00	24.00	218,00	12/.00	51.00	17.00	•	•
197	3	•	•	•	•	•	•	•	•	•	•
177	-	•	•	•	•	•	•	٠	•	•	•
198	1	•	. •	•	•	•	•	•	•	•	٠
199	-	•	٠	•	•	•	•	•	•		•
300			٠	•	•	•	•	•	• •	•	•
200		0.00	•	•	•	•	•	•	•	•	•
200	:			•	<b>!</b>	•		•		•	•
202	1	0.13	0.15	•	<b>K</b> 0.00	•	82.00	٠	10,00	•	٠
200	<b>E</b>		•	•	•	•	•	•	•	•	0-25-17 🛛 🕚
214	1	0.00		. •	<b>^*</b> ^^	•	14	•		•	· •
210	0	1.21	1.21	•	4.00	•	20.00	•	6.00	•	
210	;	٠	•	•	•	•	•	•	•	•	٠
213	1	<b>^</b> • <b>^</b>	•	•	•	•	•	•	•	• • •	• • •
217	-	0.00		1520 00			A1 AC	<b>.</b>	-		

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	PLANT	# PP	FLOWINF (Mgd)	FLOW <b>eff</b> (Mgd)	BODINF (PPM)	BODEFF (PPM)	CODINF (PPM)	CODEFF (PPM)	TSSINF (PPM)	T8SEFF (PPM)	TOCINF (PPM)	TOCEFF (PPH)
										******	*****	
	221	1	•	•	•	•	•	•	•	•	•	٠
	221	•	0,00	•	•	•	•	•	•	• • • •	•	٠
	223	1	0,74	0,74	•	15,00	•	•	• • • •	16,00	•	٠
	224	5	0.07	0.07	400,00	32.00	520,00	64.00	2078.00	64,00	٠	•
	225	1	0,17	0.17	•	•	•	•	•	•	•	•
	227	1	0,09		478,00	•	1422,00	•	•	•	•	•
	229	•	1,17	1,17	204,00	6.00	405.00	77.00	74.00	18.00	•	•
	232	Ś		•		•	•	•	•	•	•	<b>.</b>
	233	2	0.60	0.60	61.00	36.00	109,00	75,00	•	•	100.00	71.00
	237	1	0.70	0.70	•	•	•	•	•	•	•	•
	240	3	0.00		•	•	•	•	•	•	•	•
	241	1	0,10	0.10	•	•	•	•	•	•	•	•
	242	1		•	•	•	•	•	•	•	•	•
G	243	1	0.00		•	•	•	•	•	•	٠	•
4	244	1	•	•		•	•	•	•	•	•	•
-	244					•	•	•	•	•	•	•
	246	Ĵ	0.48	0.48	572.00	99.00	1219.00	455,00	97.00	90,00	•	•
	249	1	0,03	0.04	-	4,00	•	18,00	•	8,00	•	•
	250	1	0,03	0,03	•			•	5.00	5.00	•	•
	251	1	•		•	•	•	•	•	•	•	•
	253	1	-			•	•	•	•	٠	•	•
	254	4	88_0	0.88	119,00	14,00	531,00	85,00	643.00	19,00	•	•
	262	3	0.31	0.31	808.00	58.00	3080,00	323.00	2713.00	179.00	•	•
	265	2	0.00	•		•	•	•	•	•	•	٠
	273	3	3.98	3.98	470,00	34,00	900,00	353.00	69,00	97.00	•	•
	277	3	2.47	2.30	291.00	16.00	1893,00	94,00	407,00	45.00	•	•
	280	ž	0.00			•		•	•	•	•	•
	282	3					•	•	•	•	•	•
	287	ī	1.50	1.50		7.00		•	•	6.00	•	•
	288	ž	0.50			•		•	•	•	•	•
	288	ī		•		-	-	•	•	•	•	•
	289	ä	•	•	•							•

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### WASTE STREAM Data LIBTING = BUBCATEGORIZATION FILE

				SUBCATENOT	P, TYPE I	C. MATERUSE	>,165 GAL/L	.8		********	
PLANT Number 	# PP	FLOHINF (MgD)	FLOH <b>eff</b> (MGD)	BODINF (PPH)	BODEFF (PPM)	CODINF (PPH)	CODEFF (PPM)	<b>T88INF</b> (PPH)	TSSEFF (PPH)	TOCINF (PPN)	10CEFF (PPH)
20	3	0,72	0,72	2113,00	190.00	5370.00	725.00	•	•	•	•
20	.:				***	7400 00			••••	•	•
31 36	11	1.10	1.10	2527 00	22.00 391 00	2404.00 5629 00	244,00	86.UU 1946 00	148.00	•	•
41	3			E JE I I V	1,1,00						•
41	•										
43	3	•	•	•	•	•	•	•	•	•	•
43	.:		•	•	•		•		•	••••	•
49	13	0.44	٠	•	•	27632.00	•	459.00	٠	11006.00	•
55	11	0.90	•	6022.00	•	10037.00	•	•	•	3613 00	•
57	5	3.11	3.10	1385.00	38.00	1488.00	201.00		118.00		•
60	4	5.03	5,07	501.00	41.00	•	•		88.00	469.00	159,00
61	11	2.43	2,43	382.00	61.00	862,00	189.00	•	•	599.00	135,00
4° 62	5	2.13	1,35	199,00	68,00	•	1 <b>•</b>	27,00	20,00	• • • • • •	
01 74	14	2.34	2,34	•	•	•	•	•	٠	473.00	23,00
74	•	•	•	•	•	•	•	•	٠	•	•
76	51	18.36	15.80	582.00	15.00	1327.00	162.00	23.00	36.00	332.00	64.00
80	25	29.70	29.00	153.00	10.00				9.00		
84	4	0.55	0,55	•=-••	•				•		
84	4	0.27	0,27	•	•		•	•	•	•	•
84	2	• . •	•		••••	•	•	•	•	•	•
84	13	6.67	0.66	244.00	54,00	•	•	•	٠	٠	•
84	26	0.21	U.C/ 1 11	1407 00	A.ª 0.0	2454 44	•	•	143 00	1345 44	
95	1	1		1407.00	44,00	5434100	•	•	146,00	1503.00	110.00
98	21	2.11	2,00	793.00	53,00	562.00	351.00		50.00		•
98	•	•	•			•	•	•	•		
98	•	•	•	•	•	•	•	•	•	•	•
99	2		."	····		•	•	•		··-·	•
102	7	2.73	2.75	850,00	15.00	<b>51</b>	28 00	•	74.00	505.00	34.00
108	11	2.80	16.60	314400	14.00	33,00	E9.00	•	٠	150,00	. <b></b>
110	· 3	2.04	2.04	1506.00	6.00	2883.00	55.00	57.00	13.00	•	•
112	5	0.82	58,0	1177.00	53,00			72.00	26,00	685.00	69.00
113	41	17.46	16,70	1719,00	28,00	•	•	36,00	21,00	793.00	61.00
114	45	3.92	4,80	823.00	13,00		135.00	73.00	62.00	513.00	61.00

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		WASTE STREAM	
DATA	LISTING	- SUBCATEGORIZATION	FILE

					SUBCATENOT P	, TYPE ISC	, WATERUSE >	.165 GAL/LE				***********
	PLANT NUMBER	# PP	FLOWINF (MgD)	FLOWEFF (MGD)	BODINF (PPH)	BODEFF (PPH)	CODINF (PPH)	CODEFF (PPM)	T88INF (PPM)	TSSEFF (PPM)	TOCINF (PPM)	TOCEFF (PPH)
	127	17	1,13	1.13	684.00	127.00	653.00	535,00	43.00	46.00	• .	198.00
	134	3	•	•	•	•	•	•	•	•	•	•
	134	•	•	•	•	•	•	•	•	•	•	•
	135	3		•	•	•	•	•	•	•	•	•
	135	•	•	•	•	•	•	•	•	•	•	•
	136	5	•	•	•	•	•	•	•	•	•	•
	136	•	•	•	•	•	•	•	•	•	•	•
	137	3	•	•	•	•	•	•	•	•	•	•
	137	•	•	•	•	•	•	•	•	•	•	•
	139	5	•	•	•	•	•	•	•	٠	•	•
	139		•	•	•	•	•	•	•	•	•	•
	141	3	•	•	•	•	•	•	•	•	•	•
	141	<u>.</u>	•	•	٠	•	•	•	•	•	•	٠
	142	3	•	•	•	•	•	•	٠	٠	•	•
G	142	<b>±</b> .	•	٠	•	•	•	•	•	•	•	•
Ĩ	143	5	•	•	•	•	•	•	•	•	•	•
C	n 143	•	••••		•		<b>**</b> ***	500 00	•		•	•
	158		0,40	0.37	<b>··</b> *•••	61.00	4324.00	340.00	7	60,00	•	•
	160	2	0.55	0.49	5710.00	85.00	6/5/.00	220.00	<b>KO</b> <sup>0</sup> 00	104-00	•	•
	175		0.04	0.84	443.00	40.00	442.00	322,00	A 4 0 0 0 0		•	•
	1/0	4	1.0/	1.0/	2507,00	40.00	37744 44	•	000.00	19400		•
	100	'	5.00	•	14409 00	•	23/44.00	•	•	•	0000.00	•
	180	:	• • • •	•				E 31 00	•	170 00	•	•
	10/	۲	1.00	1.00	2024.UU	12/.00	4010-00	353.00	a•^^	1/0.00	33 00	21 00
	101	.:	0.20	0.20	E17 <sup>0</sup> 00	E17 00	1577	1577 00	9 00	9 00	EJOV	23,00
	143	13	0.30	0.50	53/.00	3/ 00	1211-00	13// 00	98 00	A2 00	116.00	100 00
	143	10	0,70	0430	224.00	24.00 A1 00	•	334 00	10.00	53.00	330.00	100,00
	215				•	41.00	•	334800	•		•	•
	216		^*^o	^ ^	2400 00	10 00	•	•	•	29.00	•	•
	216	•	v.v.	••••	2400.00		•	•	•		•	•
	220	į	^*^9	^ <b>^</b> ^	1741 00	ຮດ້າວ	6178 00	368.00	344.00	95.00	•	•
	>>>	i	0 48	0.11	803 00	368.00	2976.00	654.00		16.00	•	•
	228	29	3 75	3 75	003.00	43 00		364.00	•	78.00	3112.00	•
	214	11	7 95	7.91	546.00	12.00	1354.00	199.00	•			•
	235	Ā	5 10	5.10	3-0.00	89.00	2605.00	367.00	428.00	235.00	719.00	•
	236	18	4 08	4.08	•	37.00		193.00		65.00	334.00	53.00
	236	•		•			•		•	•	•	•

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 *********				UBCATENOT P.	TYPE ISC,	WATERUSE 3	.165 GAL/L	8			**********
PLANT NUMBER	# Pp •••••	FLOWINF (MGD)	FLOHEFF (MGD)	BODINF (PPH)	BODEFF (PPH)	CODINF (PPH)	CODEFF (PPH)	TSSINF (PPM)	183EFF (PPH)	TOCINF (PPH)	TOCEFF (PPH)
238 248	3	2.01	1.60	4490.00	72.00	8334.00	217.00	86.00	74.00	3202.00	73.00
249 257	•	0.68	0.62	1211.00	12.00	5141.00 •	150.00	•	14.00	•	•
272 274	15	3,99	4,25	970.00	22.00	2189.00	147.00	103.00	86.00	695.00	74.00
276	6	0.95	0,95	-30.00	•	•	256.00	•	132,00	EGE.00	•

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					SUBCATENOT P	TYPE ILC	, WATERUSE<	.165 GAL/LB				**********
F •	PLANT NUMBER	# PP	FLOWINF (MgD)	FLOWEFF (MGD)	BODINF (PPM)	BODEFF (PPH)	CODINF (PPM)	CODEFF (PPH)	TSSINF (PPM)	TSSEFF (PPH)	TOCINF (PPH)	TOCEFF (PPM)
	37		0.02			•	•	•	•	•		•
	38	8	1.30		670.00		960.00		61.00			
	40	Ś	0.01	0.01					•			
	42	š	0.05	0.05	5961.00	380.00	3683.00	541.00	•			
	49	•	1.01	1.01		7.00				34,00	20.00	15.00
	50	21	0.38	0.25	3913.00	33.00		278,00		74,00	•	•
	50		0.12					- • •				
	51	10	0.68	•	•	•						
	51		0.07	•	•				•		•	
	63		0.01	•								
	63	10	0.56	0.56	891.00	468.00	5496.00	4079.00		194_00	1079.00	685.00
	63	••	6.74	6.74	43.00	43.00	271.00	271.00	12.00	12.00	•	
	63	•	0 27						•			
0	A1	;	0 12	0.11	•	104.00	•				1199.00	417.00
Ϋ́	Å1	ī	0.26	0.30	•	288.00	· ·			140.00	• • •	360.00
ω	A1	ī	0 00		•		•					
	84		0 12	0.12	•	•	•	•	•	•	•	•
	114	10	A 40	0.49	•	251.00	3091.00	583.00		250.00		•
	150	10	0.05	v v	52550 00		67095.00		•			•
	141	<b>.</b>	0 03	0.01	1977 00	154.00	8237.00	638.00	•	•	•	•
	103	71	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	2 22	288 00	16 00			•	•	•	•
	177	21	6 - CE		200.00	10.00	20000 00	•	•	•	•	•
	177	•	1 27	•	•	•		•	•	•	•	•
	1//		1,63		107.00	34 00	9229 00	240 00	4110 00	45 00	•	•
	100	24	1.00	1.00	1010.00	30.00	A2181 AA	240.00			•	•
	143	•	0.01	A*A3	E14E3.00	111 00		1178 00	AA5 00	395 00	•	•
	210	Ę	0.02	0.02	7510 00	30.00	1550.00	545 00		105 00	•	•
	219	2	0.05	0.05	5214 00	30.00	3314400		•	37 00	•	•
	234	2	0.01	0.01	<b>*</b> • • • • • •	4.00	•	00.00	•	31.00	•	•
	201		0,35	<b>,</b> *,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	2010.00	30	8040 00		•	•	•	•
	200	20	<b>6,</b> 30	2,30	4047.00	240,00		1007.00	•	150,00	1475 00	A5 00
	271	10	1,72	1.12	2012.00	24 00	2046.00	«««•v0	•	130.00	1413400	03000
	276	:	0,11	••••	•	•	31178	3/10 0-	385 00	111 00	٠	•
	279	3	0.01	0.01	•	•	≪11\o*00	£47.VQ	202.00	111*00	•	•

### HASTE STREAM Data listing - Subcategorization file

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					UBCATENOT	P, TYPE IEND	)T C				
PLANT	s PP	FLOHINF (MgD)	FLOWEFF (HGD)	BODINF (PPM)	BODEFF (PPH)	CODINF (PPH)	CODEFF (PPM)	TSSINF (PPH)	TSSEFF (PPH)	TOCINF (PPH)	TOCEFF (PPM)
1	9	0.88	0.66	1137.00	47.00	2073.00	354.00	494.00	62.00	•	
15	9	2.44	2,44	809.00	23.00	1822.00	380,00	•			
16	6	4.60	4.60		632.00	•	523,00		804,00		276.00
16	2	0.84						93,00		•	•
22	11	2.49	2.49					-	18,00	175.00	36.00
22		0.22	-•	781.00		1810.00			•	364,00	
26	6	4.03								•	•
28	4	3,57	3,57	279,00	31,00	210.00	63,00	30,00	13,00	•	
32	9	0.02	0,03	647.00	80,00	5546,00		2486,00	36,00	2056,00	
35	3	0.08	0,08	9,00	9,00	28,00	28,00	1.00	1,00		•
53	3	2.20	2,28	1201,00	28,00		117,00	•	74,00		
64	3	0.86	0,86	655,00	13,00	854,00	30.00	991.00	13,00		•
. 69	3	•	•		•	•	•	•	•	•	
63	17	•	•	•	•	•	•	•	•	•	
85	4	0.31	0,30	•	30,00	•	818,00	•	67,00	•	•
୍ନ <b>87</b>	8	0.84	0,84	•	•	•	•	•	•	431,00	42,00
ú 87 ·	•	0.05	•	•	•	•	•	•	•	•	141.00
87	•	•	•	` •	•	•	•		•	•	
117	11	3,60	3,60		11.00	120,00	40.00	•	13,00	•	14,00
118	16	7.10	7,16	467,00	13,00	•	•	•	•	•	•
128	. 5	3.38	3,38	200.00	26,00	300.00	105.00	58,00	38,00	75.00	26,00
128	•	•	<b>.</b>	•	•	•	•	•	•	•	•
130	8	32.05	32,10	62,00	17.00	•	•	_ •	8,00	•	•
149	15	0.30	•	3044,00	•	11966.00	•	33.00	. • .	•	•
159	1	0.64	0,64	_• .	16.00	•	•	•	22,00	•	•
162	3	0.08	. •	42,00	••••	14.00	• • • •		•	•	•
164	12	0.36	0,36	2436.00	82,0Q	3449.00	335,00	177,00	106,00	•	98.00
165	15	0.00			<b>.</b> . <b>.</b>	•	•	•	•	•	•
170	22	5.40	2,40	344.00	28,00	•	•	•		••••	•
171	10	1.44	1,44	••••	26,00	•	<b>.!</b>	•	48,00	272,00	60,00
170		2,00	2,76	1327.00	168.00	3067.00	713.00	٠	78.00	971.00	501.00
183	11	3.10	3*35	78,00	9,00	•	•		19.00	46,00	19,00
102	ć	0.20		•		•	•	32,00	•	130,00	•
184	11	0.22	25,0	•	12,00	•	•	•	46,00	•	• • •
141	D	0.00		<b>.</b>	•••		• • • • •	••••		•	•
201	7	0.02	0.02	2000,00	/80.00	32476,00	11000.00	780,00	051,00	•	•
203	•	0.11	U.11	£725.00	181.00	7427.00	1647,00	164.00	147.00		•
204	27	0.10	U.10	•	25,00	•	44.00	•			35.00

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#### WASTE STREAM Data listing - subcategorization file

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	PLANT NUMBER	# PP	FLÖWINF (MGD) +++++	FLOWEFF (MGD)	BODINF (PPM)	BODEFF (PPH)	CODINF (PPH)	CODEFF (PPM)	T83INF (PPH)	T85EFF (PPH)	TOCINF (PPN)	TOCEFF (PPH)	
	213	3	0.00	•	•	•	•	•	•	•	•	•	
	230	3	0.42	0,32	129.00	115,00	423.00	385,00	170.00	13,00	143.00	118,00	
	256	4	0.17	0,17	204.00	20,00	• • •	100.00	£6/.UV •	38,00	112,00	•	
	259	12	0.03	0.03	•	13.00	•	75.00	•	•	+ +	24.00	
	264	6 19	1.80	1,80	266,00	17,00	333.00	40,00	133.00	27.00	67,00	7.00	
	269	20	4.31	4.31	830.00	66.00	•	•	472.00	71.00	867.00	176.00	
	281	12	0.14 1.00	1.00	116.00 66.00	116,00	•	•	80,00 •	<b>6</b> 0,00	•	•	
G-1	284	•	0.33	0.33	•	•	•	2960.00	•	18.00	•	•	
0	286	3	0.00	•	•	•	•	•	•	•	•	•	٠

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### HASTE STREAM Data Listing - Subcategorization file

PLANT NUMBER	g PP =====	FLOWINF (MgD)	FLOWEFF (MGD)	BODINF (PPM) 	BODEFF (PPH)	CODINF (PPM)	CODEFF (PPM)	TSSINF (PPM)	133EFF (PPM)	TOCINF (PPM)	TOC <b>EFF</b> (PPM)
5	5	0.37	0,37	•	•		43,00	•	7,00	•	17.00
6	1	0.01	0.01	•	279,00	•	1045.00	•	<b></b> •	•	•
8	2	0.08	0,08	•	104,00	•	241.00	•	37,00	•	•
8	:	. •	• • • •	•	••••	•	<b></b> *	•		•	•
	1	0.14	0,14	*	17.00	<b>•••</b> •••	22,00		19.00	•	•
10	5	4.39	4.34	330,00	10,00	240.00	65,00	1/,00	24.00	•	•
21	2	4.04	4.04	•	12,00	•	40.00	•	<b>K</b> .00	•	٠
24	3	0.54	A* AT	•	•	•	•	•	•	<b>F</b> ^3 <sup>1</sup> ^A	
24		0.43	0.43	<b>#</b> > <b>^</b> *	<b>`</b> ^	5088 00		•	30.00	205.00	133,00
37	2	1.40	1,40	250.00	24.00	2400.00	110.00	•	24.00	•	<b>.</b>
71	<b>E</b>	1.33	1.40	•	15.00	•	/5.00	٠	442.00	•	<b>₹1,00</b>
72			1 1 1	•	•	218 00		•	•	•	•
	Ň		1 28	•	<b>E a a a</b>	£10.00	111400	•	13.00	•	. •
02	Š,	1.20	4 14	٠	4 00	•	10 00	•	10.00	•	<b>e</b> *
92		0.00	- <u>-</u> 30	•		•	34.00	•		•	<b>4</b> ,00
94	22	An 00	40 <sup>°</sup> 00	100.00	<b>*</b> ***	•	•	111 00	101 00	194 00	•
97	5	0.86	0 86	429 00	14.00	A94.00	278.00	121.00	78.00		•
101	10	7.50	7.50							•	•
101	10	0.13		•	•	•	•	•	•	•	•
115	13	0.74	•	•	•	•	•	•	•	•	•
116	14	13.90	13.90	•	•	•	•	138.00	15.00	35.00	16.00
120	14	19.16	19.90	95.00	8.00	266.00	89.00	34.00	32.00	68.00	34.00
121	4	0.09	0.09		8.00	286.00	68.00	19.00	18.00	92.00	
121		•						•			•
122	Ĵ	2.00	2.00		27.00		8.00	•	51.00		6.00
129	•	•	•								
131		0.00									
133	6	•									
144	5	0.95	0.95		27.00	744.00	413,00	1266.00	34,00		
145	6	1.57	1,57		42.00		252,00	•	37.00		48.00
151 -	2	3,00	3,00	•	22,00	•	102,00	•	18,00	•	41.00
151	4	•	•	•	•	•	•	•	•	•	•
153	3	0.33	0,33	•	122,00	•	162.00	. •	42,00	•	62,00
166	1	0.01	•	•		•	•	•	•	-	•
169	5	0.01	•	•	•	•	•	•	•	•	•
172	6	2.60	•	261,00	•	1384,00	•	108.00	•	•	-
181	6	0.01		1439.00		17186.00		7554.00	-	9592.00	-

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	PLANT Number	8 PP	FLOWINF (HGD)	FLOWEFF (MGD)	BODINF (PPH)	BODEFF (PPH)	CODINF (PPH)	CODEFF (PPH)	T88INF (PPH)	TSSEFF (PPM)	TOCINF (PPM)	TOCEFF (PPH)
	181 182 186 205	1 1 5 3	0.53 0.32 0.00	0.55 0.33 0.01	238.00 162.00	36.00 5.00 14.00	5224.00	146.00 159.00	85.00 670.00	26.00 20.00 154.00	492,00	61.00
	208 214 226 231	11 7 10 2	0,54 1,36 0,15	0,45 1,20 0,16	318,00 481,00 1743,00	27,00 23,00 772,00	456,00 1979,00 4556,00	139,00 250,00 1326,00	1225.00	41.00 55.00 21.00	175,00	315,00
	245 247 252 252	1 <b>5</b> 10 10	3,10 0,11 0,24	0.28 3.10	•	12,00 3,00	260,000 2956,00	69.00	1198.00	18,00 26,00	• • •	15.00
G-]	252 255 258 261	1 8 5 2	0.67 0.40 0.01 0.10	0.40 0.01 0.10	• • •	51.00	27872,00	31.00 243.00	33178,00	16.00 37.00	• • •	70.00
2	270 278 285 290 291	4 8 9 3	0.97 1.80 0.10	0,95	143.00 193.00	17,00	345.00 546.00	51.00	204.00 B1.00	12,00	161.00	•
		•	•	•	٠	•	•	•	•	•	•	•

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---- SUBCATEPLASTICS ONLY ------

PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	HTF	DISCHARGE
2	0	2	0	8 3 K	DIH
3	0	2	0	ASL	DIR
4	Ö	2	1	DPH	ZERO
7	0	2	0	LAP	ZERO
9	0	2	0	ASL	DIR
10	0	3	0	ALA	DIR
11	0	1	0	BRN	ZERO
11	0	0	0	CLR	DIR
12	0	1	0	BRN	ZERO
13	0	2	0	RTE	ZERO
19	0	3	0	CON	ZERO
17	0	3	0	RBC	DIR
17	0	0	0	CON	ZERO
19	0	5	3	ASL	DIR
25	0	1	0	CLR	DIR
27	0	3	2	ASL	DIR
29	0	1	1	ASL	DIR
30	0	2	1	CON	ZERO
33	0	1	0	PCF	DIR
34	0	1	0	ASL	DIR
35	0	1	0	BRN	ZERO
39	0	4	0	ASL	DIR
44	0	1	0	ASL	DIR
45	0	2	0	ASL	DIR
46	0	2	0	RTE	ZERO
47	0	5	0	BRN	ZERO
47	0	0	0	CON	ZERO
48	0	1	0	RTE	ZERO
52	0	2	0	ASL	DIR
54	0	1	0	RBC	DIR
56	0	1	0	IMP	ZERO
58	0	1	0	RTE	ZERO
62	0	1	0	IND	ZERO
65	0	2	1	ASL	DIR
67	0	1	0	DRY	ZERO
68	0	2	0	DRY	ZERD
70	0	2	0	DRY	ZERO
73	0	2	1	ASL	DIR

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	PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	WTF 	DISCHARGE	
	75	0	2	0	TRF	DIR	
	11	ŏ	Ā	Ó	NOT	DIR	
	77	ŏ	Ó	0	INP	ZERO	
	77	Ŏ	ō	Ó	ŬL8	DIR	
	78	ŏ	1	0	EVP	ZERO	
	79	Ō	Ž	0	0Pm	ZERO	
,	82	Ō	3	Ó	BRN	ZERO	
	82	Ō	Š	Ó	DRY	ZERO	
	84	ŏ	Ĩ	0	CLR	DIR	
	84	ŏ	ĩ	0	OL S	DIR	
	89	ŏ	4	0	ASL	DIR	
	90	Ŏ	2	0	RBC	DIR	
	91	Ŏ	ž	0	ASL	DIR	
	91	ŏ	ō	Ō	IMP	ZĒRO	
	93	Ō	i	0	ABL	DIR	
Б С	96	ŏ	ī	Ó	ASL	DIR	
ī,	100	ō	ż	Ŏ	ASL	DIR	
14	100	ŏ	ō	Ō	ARL	DIH	
	104	ŏ	7	Ō	ABL	DIR	
	104	ŏ	à	ŏ	DPH	ZERO	
	104	ŏ	ŏ	ŏ	IMP	ZERO	
	104	ŏ	ŏ	ŏ	NEU	DIR	
	105	ŏ	i	ŏ	ASL	DIR	
	106	ů.	Ś	i	ALA	DIR	
	107	ŏ	ī	ō	ASL	DIR	
	109	0	j	ŏ	ASL	DIR	
	111	Ŏ	ī	ŏ	ASL	DIR	
	119	ŏ	i	ŏ	OL S	DIR	
	123	ŏ	i	i	ASL	DIR	
	124	Ŏ	ż	ō	ASL	DIR	
	125	ŏ	2	ŏ	451	DIR	
	124	ŏ	ĩ	õ	ASL	DIR	
	129	0	í	ŏ	CON	7680	
	111	v 0	1	ŏ	CON	7580	
	171	v A	;	•		DTR	
	176	U A	2	Ň	087	7580	
	140	v	<u>د</u>	Ň	ote	7500	
	140	v	V 2			ALNU A10	
	146	0	4	1	AGL	D1K	

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### WASTE STREAM DATA LISTING - SUBCATEGORIZATION FILE SUBCATEPLASTICS ONLY

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PLANT Number	ORGANIC	PLASTIC PP	OTHER	WTF	DISCHARGE
49887					
	_	_			
147	0	2	1	ASL	DIR
148	0	1	0	DRY	ZERO
150	0	3	3	ALA	DIR
152	0	1	. 0	CLR	DIR
152	0	1	0	CLR	DIR
154	0	2	1	CON	ZERO
155	0	4	2	DRY	ZERO
155	0	0	0	RTE	ZERO
155	0	0	0	CON	ZERO
156	0	5	0	88K	DIR
157	0	1	0	ASL	DIH
141	0	2	0	DRY	ZERO
167	0	1	0	CON	ZERO
168	0	1	0	UNK	UNK
173	0	4	1	CON	ZEHO
174	Ó	4	3	ASL	DIR
179	ŏ	2	1	ASL	DIR
184	ŏ	4	õ	88K	DIR
185	ŏ	1	Ō	CON	ZERO
189	ò	i	ŏ	ALA	DIR
192	ŏ	ż	i	ASL	DIR
194	ŏ	ī	i	NOT	DIR
196	Ň		ò	CLR	DIR
197	0	i	ŏ	RTF	7580
197		Ĩ		APN	7680
198	Ň	;		084	2580
199	Ň	i		OTE	7590
100	0			975	7590
200	Ň	:	Ň		7500
200	v			TMD	7590
202	0			101	
202	U A	i.		ARL	7500
207	0	E .		0.01	2ERU
204	0	1	0	BHN	ZERU
210	0	0	U	AJL	UIN UIN
210	0	0	Ū	BKN	<b>TEND</b>
211	0	1	U	LAP	ZERU
212	0	4	0	LAP	ZENU
217	0	2	0	ASL	DIR

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		PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	WTF	DISCHARGE
		221	0	1	0	DRY	ZEHO
		221	Ō	Ó	0	CON	ZERO
		223	ŏ	1	0	ALA	DIR
		224	õ	ž	0	ABL	DIR
		225	Ŏ	ĩ	Ó	ALA	DIR
		227		i	ŏ	OF 8	ZERO
		229		ŝ	i	ASL	DIR
		212	0	Š	ŏ	NOT	DIR
		211	Ň	5	ŏ	RBC	DIR
		217	Ň	1	ŏ	ASL	DIR
		240	Ň	· ;	i	EVP	ZERO
		241	Ň	ī	ō	OL 8	DIR
		342	Ň	i	ŏ	CON	ZERO
		241	<b>.</b> .	i	ŏ	CON	ZERO
	G	244	0	;	ŏ	ORY	ZERO
	<u> </u>	244	Ň	à	õ	IMP	ZERO
	6	344	Ň	i	ò	CLR	DIR
		240	Ň		ů.	01.8	DIR
		350		;	ő	NOT	DIR
		230	Ň	•	ŏ	DRY	ZERO
		251	Ň	:	ŏ	ARN	ZERO
		273	, v	÷			DIR
		234		7	ŏ	ASL	DIR
		202	v	2	ŏ	CON	TERO
		202	Ű	6	0	4.91	DIR
		2/3	0	,	0		018
		2//	Q	3	v v	DIE	7580
		200	0	E S			LINK
		202	0	ć			018
		267	0		0		7580
		288	0	<b>C</b>			7680
		288	. 0 .	.0	1		7580
		289	0 .	2	1	UKT	4CRU

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		SUBCATENOT P	, TYPE ISC.	NATERUSE	>,165 GA	L/LB	
	PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	NTF	DISCHARGE	
				•			
	20	1	e e	U	ASL	DIW	
	20	0	0	0	BRN	ZENU	
	21		0	2	ASL	DIM	
	30	1	2	U A	ASL	DIR	
	41	1	<b>4</b>	0		ZENU JEDO	
	41	0	0	U	TWM	ZERU	
	43	1	ć	U A	DRY	ZERU	
	43	0	0	0	INH	ZERU	
	44	11	0	2	DPW	ZERU	
	51	0	0	0	DHY	ZERO	
	22	11	0	0	TWH	ZERD	
	57	3	2	0	ASL	DIR	
	00	3	1	0	ABL	DIM	
	•1	4	2	0	ABL	DIR	
	50	1	3	<b>U</b>	ALA	DIR	
	74	12	1	1	ASL	DIR	
G.	74	0	0	0	BRN	ZERO	
<b>_</b>	74	0	0	0	DPH	ZERO	
7	76	44	7	1	ALA	DIR	·
	80	15	5	5	ASL	DIR	
	84	4	0	0	OLS	DIR	
	84	5	0	0	NOT	DIR	
	84	2	0	0	UNK	UNK	
	84	11	1	0	ASL	DIR	
	84	4	0	0	NOT	DIR	
	88	23	0	3	ASL	DIR	
	95	1	0	0	BRN	ZERO	
	98	11	0	8	ASL	DIR	
	98	1	1	0	DPW	ZERO	
	98	0	0	0	BRN	ZERO	
	99	1	0	1	8RN	ZERD	
	102	5	2	1	ASL	DIR	
	103	8	3	3	ANL	DIR	
	108	5	2	4	DPw	ZERO	
	110	2	1	0	ASL	DIR	
	112	2	2	1	ASL	DIR	
	113	29	12	0	ASL	DIR	
	114	39	5	1	ALA	DIR	

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***************************************	SUBCATENOT	P, TYPE ILC,	WATERUSE	: >,165 G	AL/LB
PLA NUM 	NT ORGANIC BER PP	PLASTIC PP 	OTHER	NTF	DISCHARGE
32	7 14	4.	0	ABL	DIH
13	4 1	2	0	DRY	ZERO
13	4 Ō	0	0	RTE	ZERO
13	5 <u>i</u>	2	0	DRY	ZERO
13	5 0	0	0	RTE	ZERO
13	6 2	2	1	DRY	ZERO
13	6 0	ō	Ū i	RTE	ZERO
13	7 1	2	0	OPY	ZERO
13	7 Ō	0	0	RTE	ZERO
13	9 2	2	1	DRY	ZERO
13	9 0	0	0	RTE	ZERO
14	1 Í	2	0	DRY	ZERO
14	i õ	0	0	RTE	ZERO
14	2 i	2	0	DRY	ZERO
ရှိ ၂ ရ	2 0	Ō	0	RTE	ZERO
	3 2	2	1	DRY	ZERO
<sup>00</sup> 14	<b>3</b> 0	Ō	Ō	RTE	ZERO
15	8 4	0	2	ASL	DIR
16	0 4	Í.	0	ASL	DIR
17	5 2	6	1	ASL	DIR
17	6 6	0	1	ASL	DIR
16	0 3	2	2	DPW	ZERO
18	o	Ō	Ō	OLS	DIN
18	7 1	0	1	AŠL	DIR
19	0 2	Ó	0	NOT	DIR
19	3 3	3	0	NOT	DIR
19	Š 16	Ō	0	ASL	DIR
20	6 8	Ĩ	0	ALA	DIR
21	<b>5</b> 1	ŏ	Ö	RTE	ZERO
21	<b>6</b> 1	Ś	Ó	ASL	DIR
21	6 0		0	BRN	ZERD
22	0 1	ž	Ō	ASL	DIR
22	żi	ä	ō	ACR	DIR
22	8 21	1	7	ASL	DIR
23	4 31	i	1	ASL	DIR
23	5 6	Ō	Ō	ASL	DIR
23	6 15	Ŏ	3	ASL	DIR
23	6 0	ō	ŏ	DPN	ZERO

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	- SUBCATENOT	P, TYPE ISC.	WATERUSE	>.165	GAL/LB	\$= <b>*</b> ** <b>************</b>
PLANT Numbe	ORGANIC R PP	PLABTIC PP	OTHER	NTF 	DISCHARGE	
238	t	2	0	RTE	ZENO	
248	ź	Ō	0	ASL	DIR	
249	Š	Ó	0	ASL	DIR	
257	4	2	. 0	ASL	DIR	
260	3	Ó	2	OFS	ZERO	
272	12	3	ō	ANL	DIR	•
274	36	3	Ó	OF 8	ZERO	
276	5	ī	Ŏ	ALA	DIR	

### WABTE BTREAM Data libting - Subcategorization file

•••••••••	3	UBCATENOT P	, TYPE ILC,	NATERUSE	<8,165 GA	L/LB	
	PLANT Number	ORGANIC PP	PLASTIC PP	OTHER	#TF	DISCHARGE	
	37	2	2	0	RTE	ZERO	
	38	6	0	2	DPH	ZERO	
	40	1	3	1	MMF	DIR	
	42	1	2	0	ABL	DIR	
	49	Ō	0	0	ALA	DIR	
	50	18	3	0	ASL	DIR	
	50	0	0	0	IMP	ZERO	
	51	10	0	0	DPW	ZERO	
	51	0	0	0	BRN	ZERO	
	63	0	0	0	8RN	ZERO	
	63	6	0	4	ASL	DIR	
	63	0	0	0	NOT	DIR	
	63	0	0	0	DPW	ZERQ	
	81	2	0	0	ASL	DIR	
	81	3	0	. 0	STR	DIR	
ត្	61	1	0	0	BRN	ZERO	
N N	84	5	0	0	NOT	DIR	
0	138	6	1	3	ALA	DIR	
	160	0	0	0	BRN	ZERO	
	163	5	0	0	ASL	DIR	
	177	29	2	0	ASL	DIR	
	177	0	0	0	DPH	ZERO	
	177	0	0	0	DPm	ZERO	
	188	15	2	3	ASL	DIR	
	193	1	2	0	BRN	ZERO	
	218	1	1	0	ALA	DIR	
	219	ī	2	0	ABL	DIR	
	239	ī	2	0	ABL	DIR	
	267	ā	0	1	IND	ZERO	
	268	26	0	0	ABL	DIR	
	271	15	3	0	ASL	DIR	
	276	0	0	0	DPW	ZERO	
· · ·	279	Í.	2	0	ALA	DIR	

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## WASTE STREAM Data Listing - Subcategorization File

		34	BCATENOT P	TYPE ILN	OT C	
	PLANT Number	ORGANIC PP	PLASTIC PP	OTHER	WTF	DISCHARGE
	1	2	6	1	ALA	DIR
	15	1	8	0	ASL	DIR
	16	3	1	2	PCF	DIR
	16	1	0	1	DPW	ZERO
	22	9	5	0	ASL	DIR
	22	0	0	0	DPW	ZERO
	26	6	0	0	RTE	ZERO
	28	1	2	1	TRF	DIR
	32	9	0	0	ALA	DIR
	35	2	1	0	NOT	DIR
	53	1	2	0	ASL	DIR
·	64	1	1	1	ALA	DIR
	69	2	1	0	DRY	ZERO
	83	11	1	5	RTE	ZERO
	85	2	2	0	NEU	DIR
	87	4	1	1	ASL	DIR
ې م	87	0	0	0	DPW	ZERO
N	87	2	0	0	ACR	DIR
<b>_</b>	117	8	0	2	ALA	DIR
	118	14	1	1	OXY	DIR
	128	2	2	1	OXY	DIR
	128	0	0	0	DPW	ZERO
	130	4	3	1	PCF	DIR
•	149	13	Ū.	Ž	INP	ZERO
	159	1	Ó	0	NEU	DIR
	162	3	0	0	IMP	ZERO
	164	11	0	1	ASL	DIR
	165	7	Ś	3	CON	ZERO
	170	14	3	5	ASL	DIR
	171	8	2	6	ASL	DIR
	178	2	5	0	ASL	DIR
	183	7	2	2	ARL	DIR
	183	1	ī	1	UPW	ZERO
	184	ī	Ŏ	Ō	ASL	DIR
	191	5	1	0	BRN	ZERO
	201	3	Ō	5	ASL	DIR
	203	4	0	Ó	ASL	DIR
·	204	22	Ō	2	ACR	DIR

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	PLANT NUMBER	ORGANIC PP	PLABTIC PP	OTHER	WTF	DIBCHARGE		
					*****			
	213	1	2	0	CON	ZERO		
	213	ŏ	3	0	DRY	ZERO		
	230	š	ō	Ō	CLR	DIR		
	230	7	i	Ō	OPW	ZERO		
	256	i	j	Ō	BSK	DIA	•	
	256	ò	ō	ŏ	THP	ZERO	·	
	259	12	ŏ	ŏ	ACR	DIR		
	263	17	i	ō	4.81	DIR		
	264		i	i	451	019		
	266	<b>E</b>	i		140	7580		
•	240			1	4.91			
	176	12	E A	<b>,</b>	NOL	010		
	613				NUT	DIM		
	201	11	0	1	ASL	DIM		
	593	3	1	0	DRY	ZERO		
~	264	8	0	1	NEU	DIR		
, i	286	5	4	0	CON	ZERO		
2	286	Ō	2	1	DRY	ZERO		

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	PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	WTF 	DISCHARGE	
	5	1	1	3	CLR	DIR	
	6	1	0	0	ALA	DIR	
	8	2	0	0	OLS	DIR	
	6	0	0	0	OFS	ZERO	
	8	1	0	0	NEU	DIR	
	18	1	1	1	ASL	DIR	
	21	3	2	0	ALA	DIR	
	23	3	0	0	RTE	ZERO	
	24	1	3	0	ALA	DIR	
	59	4	3	0	ASL	DIR	
	66	2	0	0	NEU	DIR	
	71	3	0	1	DRY	ZERO	
	72	4	3	1	ASL	DIR	
	86	5	0	- 3	CLR	DIR	
â	92	5	0	0	NEU	DIR	
L 7 I	92	Ó	0	0	DPW	ZERO	
N	94	19	1	2	ASL	DIR	
8	97	1	1	0	ASL	DIR	
	101	6	3	1	NEU	DIR	
	101	Ō	Ō	Ö	DPH	ZERO	
	115	9	2	2	DPH	ZERO	
	116	9	Ž	3	ALA	DIR	
	120	8	3	3	ALA	DIR	
	121	4	õ	ō	ACR	DIR	
	121	ů.	Ó	0 .	DRY	ZERO	
	122	Ĵ	Ō	Ō	OLS	DIR	
	129	õ	ŏ	Ō	BRN	ZERO	
	131	Ō	ŏ	Ō	INP	ZERO	
	133	i	Š	ŏ	CON	ZERO	
	144	i	2	õ	ASL	DIR	
	145	ű.	1	i	ARL	DIR	
	151	2	ò	ō	ASL	DIR	
	151	3	ĭ	ŏ	ALA	DIR	
	153	3	ō	ŏ	DAF	DIR	
	166	ī	ŏ	ŏ	CON	ZERO	
	169	ä	ŏ	ĭ	RTE	ZERO	
	172	6	ŏ	ŏ	CON	ZERO	
	181	ĩ	š	ĩ	IMP	ZERO	
		-	_	-			

	***********		UBCAT=NOT P	. NOT TYPE	E I		***********
	PLANT NUMBER	ORGANIC PP	PLASTIC PP	OTHER	WTF 	DISCHARGE	
	181	0	1	0	DRY	ZEHO	
	182	1	Ō	ō	STR	DIR	
	186	i	4	ŏ	ASL	DIR	
	205	ż	Ó	1	NEU	DIR	
	208	10	õ	ī	ASI	OIR	
•	214	1	Š	i i	DRY	ZERD	
	226	i	Ā	ĩ '	TRF	DIR	
	231	i	ī	0	ACR	DIR	
	245	14	Ŏ	ĩ	STR	DIR	
	247	ŝ	Å	i	ASL	DIR	
	252	2	Õ	ż	OFB	ZERD	
	252	0	i	Ō	OFS	ZERO	
	252	i	ō	ŏ	0Pm	TERO	
	255		ò	2	NEU	DTR	
	254	š	ŏ	0	ACR	DTR	
	261	í	i	ŏ	NOT	DIR	
0	270	i	i	ŏ	011	018	
1	278	;	ξ.	i	ORY	7580	
24	286	Š	ő	Å		7580	
<del>~ -</del>	200	2	1	õ	0 P m	7580	
	201	1		ň	APN	7580	
	671	•	~	•	write .	FPUA .	

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	PLANT NUMBER	OLGINF (PPH)	08GEFF (PPH)	PHENOLINF (PPH)	PHENOLEFF (PPH)	NH3NINF (PPH)	NH3NEFF (PPH)	CHROMIUMINF (PPH)	CHROMIUMEFF (PPH)
	2	•	4.00	•	•	•	•	•	0.50
	3	41.00	13,00	0,14	0.03	7.60	34,00	0.11	0.03
	4	•	•	•	•	•	•	•	•
	7	•	•	•	•	•	•	٠	•
	10	•	•	•	•	•	•	•	•
	11	•	•	•	•	•	•	•	•
	ii	•	•	•	•	•	•		
	12								
	13	•	•	•	•	•	•	•	
	14	•	•	•	•	•	•	•	•
	17	•	•	•	•	•	29.00	•	•
	17	•	•	•	•	•	•	•	•
	19	•	•	•	0.00	•	0.60	••••	
൭	(2)	•	•	•	•	•	•	1.40	0.01
Ĩ.	20	•	•	•	•	•	<b>^</b> •••	•	•
ι Υ	27	•	•	•	•	•	V. 6V	•	•
	11	•	•	•	•	•	•	•	•
	34		•	•	•	8.00	6.00	•	•
	35					0.00	•		
	39	•	•			•	•	•	
	44	•	•	•	•	•	35,00	•	•
	45	•	•	•	0.13	8.00	6.00	•	0.02
	46	•	•	•	•	•	•	•	•
	47	•	•	•	•	٠	•	•	•
	47	•	•	•	•	•	•	•	•
	52	•	•	•	•		•	•	
	54	•	•	•	•	•	•	0.09	0.02
	56	•	•	. •	•	•	•		
	58			•	•	•			•
	62	•	•	-		•	•	•	•
	65	•	•	•	•	•	•	•	•
	67	•	•	•	•	•	•		•
	68	•	•	•	•	•	•	•	•
	70	•	٠	•	•	•	•	•	•
	73	•				•	•		•

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### WABTE STREAM Data listing - Subcategorization File

	PLANT ORGINE ORGEFE PHENOLINE PHENOLEFE NHININE NHINEFE CHROMIUMINE CHROMIUMEFE										
	PLANT	OLGINF (PPH)	OLGEFF (PPm)	PHENOLINF (PPH)	PHENOLEFF (PPH)	NH3NINF (PPH)	NH3NEFF (PPN)	CHROMIUMINF (PPm)	CHROMIUMEFF (PPN)		
			••••						**********		
	75		•		•	•	•	•	•		
	77	•	•	•	•	•	2.00	•	•		
	77	• •	•	•	•	•	•	•	•		
	77	•	•	•	•	٠	•	•	•		
	78	•	•	•	•	•	•	•	•		
	79	•	•	•	•	•	•	•	•		
	82	٠	٠	•	•	•	•	•	•		
	82	٠	•	•	•	•	•	•	•		
	84	•	٠	•	•	•	•	•	٠		
	04	•	•	•	•	<b>^^</b> ^		•	•		
	04	•	•	•	•	47.00	03.00	•	•		
	90	•	•	•	•	٠	•	•	•		
	91	•	•	•	•	•	•	•	•		
	91	•	•	•	•	•	0.46	•	•		
ດ	96	•	•	•	•	2.00					
<b>.</b> .	100	•	•	0.10	0.01	6.20	4.40				
ת	100	•				••=•					
	104		1.20	0.09	0.01	4.50	0 47	•			
	104		•		•		•	•	•		
	104	•	•	•	•	•	•	•	•		
	104	•	•	•	•	•	•	•	•		
	105	•	•	•	0,48	•	0.36	•	•		
	106	•	•	•	•	•	•	•	•		
	107	•	•	•	•	•	•	•	•		
	109	•	•	•	•	•	•	•	•		
	111	•	•	•	•	•	•	•	٠		
	119	23.00	2.00	•	•	•	•	•	•		
	123	• • • •	. •	•	•			<b>• • •</b>	<b>^</b> *••		
	124	\$3,00	6.00	-*••	.*	20.00	20,00	V.2V	0 <b>*</b> 20		
	125	•	•	2.00	0.14	٠	•	3.00	0 <sup>*</sup> 05		
	120	•	•	•	•	•	•	<b>E</b> . VV	V . V 3		
	124	•	•	•	•	٠	•	•	•		
	171	٠	•	•	•	•	•	•	•		
	136	•	•	•	•	•	•	•	•		
	140	•	•	•	•	•	•	•	•		
	144	•	•	•	•	•	•	•	•		
	140	•	•	•	•	•	•	T	•		

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	PLANT NUMBER	OLGINF (PPH)	08GEFF (PPH)	PHENOLIN <b>F</b> (PPH) 	PHENOLEFF (PPM) +	NH3NINF (PPH)	NH3NEFF (PPM)	CHROMIUMINF (PPH)	CHROMIUMEFF (PPN)
	147	•	•	•	•	•	3.10	0.09	0.01
	148	•	•	•	•	•	•	•	•
	150	•	14.00	•	0.13	•	•	•	•
	152	•	•	•	•	•	•	•	•
	152	•	٠	•	•	٠	٠	•	٠
	154	•	•	•	•	•	٠	•	•
	155	•	•	•	•	•	•	•	•
	122	•	•	•	•	•	•	•	•
	177	•	٠	•	•	•	٠	•	•
	170	٠	1	٠	•	•	•	•	<b>^*</b> 30
	137	•	3.00	•	•	•	•	•	Veev
	147	•	•	•	•	•	•	•	•
	148	•	•	•	•	•	. •	•	•
	173	•		26.00	•	•	•		•
с,	174		8.00			•	109.00		
Ň	179					11.00	11.00		
7	184	•	•	•			•		0.67
	185	•	•	•	•		•	•	•
	189	•	1.10	•	•	•	•	•	0.38
	192	•	•	426.00	0.03	•	•	•	•
	194	•	•	•	•	•	•	•	•
	196	•	•	•	•	•	•	•	•
	197	•	•	•	•	•	•	•	•
	197	•	٠	٠	•	•	•	•	•
	148	•	•	•	•	•	•	•	•
	144	•	•	•	•	•	•	•	•
	144	•	•	•	•	•	•	•	•
	200	•	•	٠	•	•	•	•	•
	202	•	•	•	•	•	A 10	•	A A2
	207	•	•	•	•	•	V.3V	•	VOVE
	209	•	•	•	•	٠	•	•	•
	210	•	•	•	•	•	•	•	•
	210			•	•	•			•
	211	-	-			•	-		
	212	•	•	•			•	•	•
	217			-		•	-	-	•

### WASTE STREAM Data listing - Subcategorization file

	*******			**********	SUBCAT=PLAST	ICS ONLY				
	PLANT NUMBER 	QLGINF (PPM)	04GEFF (PPM)	PHENOLINF (PPH)	PHENOLEFF (PPH)	NH3NINF (PPH) TTTT	NH3NEFF (PPH) =====	CHROMIUMINF (PPH)	CHROMIUMEFF (PPM)	
	551	•	•	•	•	•	•	•	•	
	155	•	•	•	•	•	•	•	•	
	252	•	•	•	•	•	•	•	•	
	224	•	•	•	•	•	•	•	•	
	223	•	•	•	•	•	•	•	•	
	227	<b>~~~~~</b>	•••••	•	•	•	•	•	•	
	227	242.00	eu.cu	•	•	•	•	•	•	
	232	•	3 10 .	•	•	•	•	•	<b>^*</b> 1 <i>a</i>	
	233	•	2,30	•	•	•	•	•	0.14	
	240	•	•	•	•	•	•	•	•	
	241	•	•	•	₽	•	•	•	•	
	242	•	•	•	•	•	•	•	•	
	243	•	•	•	•	•	•	•	•	
	244	•		•	•	•	•	•	•	
G	244			•	•	•	•			
	246				•	•				
8	249			-		•				
	250		•	-						
	251	•	•		-		•			
	253		•				•		•	
	254	•	•	•	•		•	•	•	
	262	•	•	0,13	0,05	39,00	50,00	•	•	
	265	•	•	•	•	•	•	•	•	
	273	•	9.00	•	•	•	•	٠	•	
	277	•	4.10	•	•	•	•	•	•	
	280	•	•	•	•	•	. <b>.</b>	•	• •	
	202	•	•	•	•	•	•	•	•	
	207	•	8.00	•	0.04	•	•	. •	•	
	200	•	•	•	•	•	•	•	•	
	200	٠	•	•	•	•	•	•	•	
	504	•	•	•	•	•	•	•	•	
						1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -				

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	PLANT NUMBER	OLGINF (PPH)	016EFF (PPm)	PHENOLINF (PPn) 	PHENOLEFF (PPM)	NH3NINF (PPM) 	NH3NE <b>FF</b> (PPM) 	CHROMIUMINF (PPh)	CHROMIUHEFF (PPM)
	20	•	•	•	•	•	•	•	•
	31	16.00	•	•	•	•	•	•	0.05
	36	•	•	420.00	4,00	•	•	•	•
	41	•	•	•	•	•	•	•	•
	43	•	•	•	•	•	•	•	•
	43	•	•	•	•	•	•	•	•
	49	•	•	1,56	•	•	•	•	•
	\$5	•	•	•	•	•	•	•	•
	57	•	•	•	0.01	•	4.60	0.18	0.30
	60 61	17.40	15.20	•	•	2.40	2.60	1.00	0.50
G	62		•		•	•	•	•	•
2	74	•	•	•	•	•	•	•	•
9	74	•	•	•	•	•	•	•	• •
	76	•	•	•	•	12,00	6.00	•	•
	80	•	0,58	•	0.04	6,50	4.30	•	٠
	84	•	30.00	•	•	•	•	•	•
	84	•	•	•	•	•	•	•	•
	84 84	•	•	•	•	•	•	•	•
	66	•	•	•	•	•	•	•	•
	95	•	•	•	•	•	••••	•	•
	98	•	•	•	•	•	63,00	•	0.42
	98	•	•	•	•	•	•	•	•
	99	•	•	•	•	•	•	•	•
	102	•	•	•	•	•	•	•	•
	108	•	•	•	•	•	•	•	•
	110	•	•	•	•	·	<b>F*</b> 00	•	•
	112	•	•	1.60	0.09	30.00	16.00	0.14	0.03
	114	•	•	•	•	•	•	•	1.30

	PLANT NUMBER	OLGINF (PPM)	OLGEFF (PPM)	PHENOLINF (PPM)	PHENOLEFF (PPM)	NH3NINF (PPH)	NH3NEFF (PPH)	CHROMIUMINF (PPH)	CHROMIUMEFF (PPM)
	127	•	•	18.00	3.60 .	303.00	80,00	•	•
	134	•	•	•	•	•	•	•	•
	134	•	•	•	•	•	•	•	•
	135	•	•	•	•	•	•	•	•
	132	•	•	•	•	•	٠	•	•
	130	•	•	•	•	•	٠	•	•
	137	•	•	•	•	•	•	•	•
	137	•	•	•	•	•	•	•	•
	139	•	•	•		•			•
	139	•	•	•	-	•	•	•	•
	141	•	•	•	•	•	•	•	•
	141	•	•	•	•	•	•	•	•
	142	•	•	•	•	•	•	•	•
	142	•	•	•	•	•	•	•	•
•	143	•	•	•	•	•	•	•	•
	154	•	•	•	•	•	20.00	•	•
5	160	•	•	•	•	•	7.80	•	•
	175	•		•	•	•			•
	176					•			
	180	•	•						
	180	•	•	•	•	•	•	•	•
	187	•	•	•	•	390,00	274.00	•	•
	190	•	•	••••	-•	••••	••••	0.10	0,10
	193	•	•	7.80	7.80	29.00	29.00	•	.•
	142	•	•	•	0.01	•	•	•	0.04
	215	•	•	•	•	•	•	•	v.ua
	216	•	•	•	•	•	•	•	•
	216	•	•	•	•	•	•		•
	220	•	•	23.00	0.04	•		•	-
	222	•	•	173,00	0,60		•	•	•
	228	•	•	•	•	•	55,00	•	•
	234	•	•	•	•	•	•	•	•
	235	•	•	•	•	•	•	•	•
	236	•	•	٠	•	•	•	•	•
	539	•	•	•	•	•	•	•	•

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********	*********	**********	********	SUBCATENOT P	, TYPE ILC,	ATERUSE >.1	65 GAL/LB		**************	
	PLANT Number	OLGINF (PPH)	016277 (PPM)	PHENOLINF (PPM)	PHENOLEFF (PPM)	NH3NINF (PPH) 	NH3NEFF (PPH)	CHROMIUHINF (PPH)	CHROMIUHE <b>FF</b> (PPM)	
	238	•	•	•	•	•	•	•	•	
	248	•		•			42.00	•		
	249	•	7.00	•		•	1.80	•	•	
	257	•		•	•		•	•	•	
	260		•	•	•		•	•	•	
	272	•	•	•				-		
	274		•		•					
	276	•	9.00	•	•		•	•	0,30	

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	PLANT Number	DEGINF (PPN)	OLGEFF (PPM)	PHENÖLINF' (PPH)	PHEND <b>LEFF</b> (PPH)	NH3NINF (PPM)	NH3NEFF (PPH)	CHROMIUMINF (PPH)	CHROMJUHEFF (PPM)
			*****						
	37	•	٠	• .	•	<b></b>	•	. •	•
	38	539.00	٠	•	•	545°00	•	1,05	•
	40	•	•	7	TA 00	•	•	•	•
	42 49	•	•	147.00	0.02	•	1.12	•	0.23
	50	•	•	•		•			
	50						•	•	•
	51	•	•	•	•	•	•	•	•
	51	•	•	•	٠	•	•	0.27	•
	63	٠		•	٠	51.00	A3 00		<b>^*1</b> 0
	<b>5</b> 3	0 <sup>1</sup> 75	0,75	•	•	1 20	1 20	0.14	0.14
	61	V.73	V.13	•	•			••••	
	81	•		146.00	16.00	•			
G _	41			•	•		•	•	•
ώ	81	•	•	•	•	•	•	•	•
	84	•	•	• • • • •	. •	•	•		
	138	•	•	445.00	1,40	•	•	0.34	0.10
	163	•	٠	705 00	n <sup>*</sup> 21	380.00	146.00	•	•
	177	•	•	143444	****				
	177					:	•	•	•
	177	•	•	•	•	•	•	•	•
	188	•	•	•	•	6,40	5.60	•	•
	193	•	•	•	•	•	175 00	•	•
	216	•	•	•	•	•	153.00	•	•
	219	•	•	•	•	•	•		0.02
	267	•		•	•	:	•	•	•
	268	•	-	•	•	-	•	•	•
	271	-		95.00	0.57	-	2.20	•	•

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	PLANT NUMBER	O&GINF (PPM)	08GE <b>FF</b> (PPH)	PHENOLINF (PPH) 	PHENOLEFF (PPM)	NH3N]NF (PPM) ===4=	NH3NEFF (PPH) ++++	CHROMIUNINF (PPM)	CHROMIUMEFF (PPm)
	1	•	•	•	•	•	•	•	•
	15	•	•	•	٠	•	782 00	•	0.43
	16	•	•	•	•	•	102.00		v.83
	22	•	•		•	•			
	22				•		•	•	•
	26	•	•	•	•	•	•	•	•
	28		••••	•	••••	3.00	1.00	•	•
	32	570.00	112,00	2.50	0.05	1,80	1.40	•	•
	)) 61	•	٠	•	•	•	•	•	•
	64	•	•	•	•	•	•	•	•
	69	•	•	•		•			
	83							•	
	85	•	•	•	•	•	•	•	•
6	67	•	•	•	•	32,00	2.00	•	•
Ĩ	87	•	٠	•	•	•	•	•	•
<b>ω</b>	67	•	1 00	•	1.0 0.0	•	•	•	^ <b>*</b> ^ <b>7</b>
	118	•	1.00	•	14.00	•	•	•	0.07
	128		•	2.00	0.02	•	•		•
	126			•	•			•	
	130 -	•	•	•	0,08	•	•	•	•
	149	• •	٠	٠	•	294.00	•	0.08	•
	159	•	•	•	٠	•	•	•	•
	102	•	•	•	•	•	•	•	•
	165	•	•	•	•	•	•	•	•
	170	17.00	11.00	18.00	0.17	•	•		•
	171	•	15,00	•	1.00		29,00	•	•
	178	•	•	•	•	•	•	•	•
	183	•	•	•	•	•	•	0.09	0.04
	183	•	•	•	•	•	•	•	
	104	•	•	٠	•	•	•	•	0.38
	201	•	•	•	•	•	•	•	•
	203	•		10.70	0.79	•	•	•	
	204	-	•	•	•	-	•	•	

F	PLANT NUMBER	OLGINF (PPM)	O\$GEFF (PPN)	PHENOLINF (PPN)	PHENOLEFF (PPM)	NH3NINF (PPM)	NH3NEFF (PPh)	CHROMIUMINF (PPH)	CHROMIUHE <b>FF</b> (PPH)
•					*****	*****			*******
	213		•	•	•	-	•	•	•
	213			•					-
	310	•	•	•	•	•	•	0 50	0 30
	230	•	•	•	•	•	•	0.30	4.30
	230	• • • • •	•	•	•	•	•	•	•
	256	418.00	\$3.00	•	•	•	•	•	•
	256		•	•	•			•	•
	259		1.00			-			
	263	•		•	•		1.80		-
	344	•	•	•	•	283 00	80.00	•	•
	204	•	•	•	•	633.00	00.00	•	•
	200	•	•	•	•	•	•	•	•
	269	•	•	•	•		•	•	•
	275	•		0_15	0.15	22.70	22.70	0.01	0.01
	281					-	•	•	•
	283	•	•	•	•	•	•	-	•
	504	•	•	•	•	•	•	•	•
	204	•	•	•	•	•	•	•	•
•	200	•	•	•	•	•	٠	•	•
	286	•	•	•	-	•	•	•	•

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	PLANT NUMBER	OLGINF (PPM)	0162FF (PPH)	PHENOLINF (PPH)	PHENOLEFF (PPM)	NH3NINF (PPH)	NH3NEFF (PPH)	CHROMIUMINF (PPM) P++++++++	CHROMIUHEFF (PPH)
	Ę		A #2						0 03
	6	•	46.00	2105.00	10.00	•	240.00	•	4146
	ě				15.00	•	4.50		•
	6				•	•	•	•	
	6	•	•	•	•	•	•	•	•
	18	•	•	•	•	•	•	•	•
	. 21	•	•	•	•	•	•	•	0,10
	23	•	•	٠	•	•	•	•	•
	59	•	3.00	•	0.12	•	٠	0.00	0.00
	66	•		•	0.10	•	•	••••	Q.10
	71					•		•	•
	72	•	•		•		•	•	•
	86	•	1.60	•	•	•	•	•	•
	92	•	•	•	٠	•	•	•	0.05
0	42	•	•		<b>^'</b> 30	45 40		A 31	<b>^</b> *•>
Y.	97	٠	•	4,00	0.∉0	03.04	04,1V	0.61	0,04
ω 5	101	•	•	•	•	•	•	•	•
	101			•	•	•	•		•
	115		•		•				
	116	•	•	•	•	•	•	•	•
	120	•	•	•		8,80	3,70	•	•
	121	•	•	120,00	0,20	•	•	•	•
	121	430.00	1 00	٠	•	•	•	•	A <sup>*</sup> A1
	129	420,00		•	•	•	•	•	0.03
	131			•	•	•			•
	133		•			:			
	144	•	•	•	•	•	•	•	•
	145	•	•	•	0.65	•	•	•	0.29
	151	•	•	•	0.05	•	2.20	•	0,23
	171	•	•	•	A <sup>*</sup> 10	•	•	•	2.00
	153	•	•	. •	V.3V	•	•	•	<b>«</b> •00
	169	•		•	•		•	•	•
	172	286.00	•	23.00	•		•	0.74	•
	161	•	-		•	80,00			-

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	PLANT NUMBER	OLGINF (PPH)	OLGEFF (PPN)	PHENOLINF (Pph)	PHENOLEFF (PPM)	NH3NINF (PPH)	NH3NEFF (PPM)	CHROMIUMINF (PPM)	CHROMIUME <b>FF</b> (PPM)	
	181	•	•	. •	•	•	•	•	•	
	182	•	•	•	•	•	•	•	•	
	186	•	•	•	0,08	•	•	•	•	
	205	•	•	•	•	•	1.70	•	•	
	208	50,00	0.67	0.32	0.01	0.80	0,90	0.70	10.00	
	214	•	•	•	•	•	•	•	•	
	220	•	•	•	•	•	•	•	•	
	231	•	•	88,00	5.50	•	٠	•	•	
	245	•	•	•	•	•	•	•	•	
	247	•	0.77	٠	•	•	•	•	0.08	
	252	•		•	•	189,00	•	•	•	
	252	•	•	•	•	•	•	•	•	
	252	•	•	•	•	5331.00	•	•	•	
	255	•	5,00	•	0.04	•	•	•	0 44	
	258	•	1.46	•	•	•	•	•	0.05	
ភ្	261	•	•	•	•	•	•	•	•	
ப் ப	270	•	,	8.30	0.44	•	•	•	•	
o	278	•	,	•	•	•	•	•	•	
	285	•	•	•	•	•	•	•	•	
	290		•	•	•	•	•	•	•	
	291	•			•	•	•	•	-	
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