# FINAL

# BEST DEMONSTRATED AVAILABLE TECHNOLOGY (BDAT)

# BACKGROUND DOCUMENT

FOR WASTES FROM THE PRODUCTION OF

# CHLORINATED ALIPHATIC HYDROCARBONS

F024

James R. Berlow, Chief Waste Treatment Branch

> Jerry Vorbach Project Manager

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U.S. Environmental Protection Agency Office of Solid Waste 401 M Street, S.W. Washington, D.C. 20460

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#### 1.0 INTRODUCTION

In accordance with the amendments to the Resource Conservation and Recovery Act (RCRA) of 1976, enacted in the Hazardous and Solid Waste Amendments (HSWA) of November 8, 1984, the Environmental Protection Agency (EPA) is establishing best demonstrated available technology (BDAT) treatment standards for the listed waste identified in Title 40, <u>Code of Federal Regulations</u>, Section 261.3! (40 CFR 261.31) as F024, wastes from the production of chlorinated aliphatic hydrocarbons. Compliance with these BDAT treatment standards is a prerequisite under 40 CFR Part 268 for placement of the waste in units designated as land disposal units. The BDAT treatment standards are effective as of June 8, 1989.

This background document provides the Agency's rationale and tecnnical support for selecting the constituents to be regulated in F024 and for developing treatment standards for these constituents. The document also provides waste characterization data that serve as a basis for determining whether a variance from a treatment standard may be warranted for a particular type of F024 that is more difficult to treat than the wastes that were analyzed in developing treatment standards for F024.

The Agency's legal authority and promulgated methodology for establishing treatment standards and the petition process necessary for requesting a variance from the treatment standards are summarized in EPA's <u>Methodology</u> for Developing BDAT Treatment <u>Standards</u> (Reference 1).

This background document presents waste-specific information on the numbers and locations of facilities that may be affected by the land disposal restrictions for F024; the waste-generating processes; the waste characterization data; the technologies used to treat the waste (or similar wastes, if any); and the treatment performance data on which the treatment standards are based (Sections 2.0-4.0). This document also explains how EPA determines BDAT, selects constituents to be regulated, and calculates treatment standards (Sections 5.0-7.0).

Under 40 CFR 261.31, wastes identified as F024 are listed as follows:

> F024 - Wastes, including but not limited to, distillation residues, heavy ends, tars, and reactor clean-out wastes from the production of chlorinated aliphatic hydrocarbons, having carbon content from one to five, utilizing free radical catalyzed processes. (This listing does not include light ends, spent filters and filter aids, spent desiccants, wastewater, wastewater treatment sludges, spent catalysts, and wastes listed in 40 CFR 261.32.)

The Agency estimates that there are 29 domestic facilities that may generate F024.

For both F024 nonwastewaters and wastewaters, the BDAT treatment standards for organic constituents are based on treatment performance data from rotary kiln incineration of F024. The BDAT treatment standards for metal constituents in F024 wastewaters are based on a transfer of treatment performance data from chemical precipitation followed by vacuum filtration treatment of K062 mixed with metal-bearing characteristic wastes.

The Agency is regulating nine organic constituents and five cloxins and furans in F024 nonwastewaters and nine organic constituents, two metal constituents, and five dioxins and furans in F024 wastewaters. The Agency is reserving treatment standards for metal constituents in F024 nonwastewaters. Due to the wide variation in the types of F024 generated by industry, a large number of constituents were selected for regulation to ensure that the different hazardous constituents that may be present in F024 are controlled. To determine the applicability of the treatment standards, wastewaters for F024 are defined as wastes containing less than 1% (weight basis) total suspended solids<sup>1</sup> and less than 1% (weight basis) total organic carbon (TOC). Wastes not meeting this definition are classified as nonwastewaters and must comply with the nonwastewater treatment standards.

The Agency has recently completed an analysis of TCLP extracts obtained from the stabilization of FO24 incinerator ash residues. The results of this analysis show substantial reduction of metals in TCLP extracts following stabilization. Therefore, the Agency has decided to reserve the final

<sup>&</sup>lt;sup>1</sup>The term "total suspended solids" (TSS) clarifies EPA's previously used terminology of "total solids" and "filterable solids." Specifically, total suspended solids is measured by Method 209C (total suspended solids dried at 103-105°C) in <u>Standard Methods for the Examination of Water and</u> <u>Wastewater Sixteenth Edition</u> (Reference 36).

treatment standards for metals in F024 nonwastewaters in the Second Thirds promulgated rule. The Agency will instead propose revised treatment standards for metals in F024 nonwastewaters based on the F024 stabilization results as part of the Third Third proposed rule so that sufficient time is provided for notice and comment on these revisions.

The BDAT treatment standards for dioxin and furan constituents in both FO24 nonwastewaters and wastewaters are set at the analytical detection limit that can be routinely achieved for these constituents, consistent with the dioxins rule promulgated by the Agency on November 8, 1986 (51 <u>Federal</u> Register 40572, 40638).

The tables following this section list the specific BDAT treatment standards for F024 nonwastewaters and wastewaters. The treatment standards reflect the total concentration of each organic constituent regulated in F024 nonwastewaters and the total concentration of each organic and metal constituent regulated in F024 wastewaters. Any future treatment standards for metal constituents in nonwastewaters will be based on analysis of leachate obtained by the Toxicity Characteristic Leaching Procedure (TCLP) described in Appendix I of 40 CFR Part 268. The units used for total constituent concentration are mg/kg (parts per million on a weight-by-weight basis) for nonwastewaters and mg/l (parts per million on a weight-by-volume basis) for wastewaters. The units used for leachate analysis of nonwastewater are mg/l (parts per million on a weight-by-volume basis). If the concentrations of the regulated

constituents in F024, as generated, are lower than or equal to the treatment standards, then treatment of F024 is not required prior to land disposal.

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# BDAT TREATMENT STANDARDS FOR F024 MCNWASTEWATERS

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BDAT	List Constituent	Total Composition (mg/kg)					
10.	2-Chloro-1,3-butadiene	0.28					
16.		0.28					
	1,1-Dichloroethane	0.014					
	1,2-Dichloroethane	0.014					
	1,2-Dichloropropane	0.014					
27.	• •	0.014					
28.		0.014					
70.	Bis(2-ethylhexyl)phthalate	1.8					
13.	• • •	1.8					
207.		0.001					
208.	•	0.001					
209.	Pentachlorodibenzo-p-dioxins	0.001					
	Pentachlorodibenzofurans	0.001					
212.	Tetrachlorodibenzofurans	0.001					
		TCLP Leachate <u>Concentration (mg</u> /					

Maximum for Any Single Grab Sample

159. Chromium (total) 163. Nickel

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#### BDAT TREATMENT STANDARDS FOR F024 WASTEWATERS

#### Total Composition (mg/l) BDAT List Constituent 0.28 10. 2-Chloro-1,3-butadiene 0.28 16. 3-Chloropropene 0.014 22. 1,1-Dichloroethane 0.014 23. 1,2-Dichloroethane 26. 1,2-Dichloropropane 0.014 27. trans-1,3-Dichloropropene 0.014 28. cis-1,3-Dichloropropene 70. Bis(2-ethylhexyl)phthalate 0.014 0.036 113. Hexachloroethane 0.036 207. Hexachlorodibenzo-p-dioxins 0.001 0.001 208. Hexachlorodibenzofurans 209. Pentachlorodibenzo-p-dioxins 0.001 210. Pentachlorodibenzofurans 0.001 212. Tetrachlorodibenzofurans 0.001 159. Chromium (total) 0.35 163. Nickel 0.47

# Maximum for Any Single Grab Sample

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#### 2.0 INDUSTRY AFFECTED AND WASTE CHARACTERIZATION

This section describes the industry affected by the land disposal restrictions for F024 and presents available characterization data for this waste.

Under 40 CFR 261.31 (hazardous wastes from non-specific sources), wastes identified as F024 are listed as follows:

> Wastes, including but not limited to, distillation residues, heavy ends, tars, and reactor clean-out wastes from the production of chlorinated aliphatic hydrocarbons, having carbon content from one to five, utilizing free radical catalyzed processes. (This listing does not include light ends, spent filters and filter aids, spent desiccants, wastewater, wastewater treatment sludges, spent catalysts, and wastes listed in 40 CFR 261.32.)

# 2.1 Industry Affected and Process Description

As defined in 40 CFR 261.31, F024 is waste specifically generated from the production of  $C_1$ - $C_5$  chlorinated aliphatic hydrocarbons by free radical catalyzed processes. The Agency estimates that there are 29 domestic facilities that may generate F024. Table 2-1 lists the number of facilities by state and EPA region. These facilities were identified using the <u>1987</u> <u>Stanford Research Institute Directory of Chemical Producers</u> (Reference 2) for major  $C_1$ - $C_5$  chlorinated aliphatic hydrocarbons as well as plant reports prepared for EPA's Characterization and Assessment Division (References 3-14).

# Table 2-1

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# FACILITIES THAT MAY GENERATE F024, BY STATE AND EPA REGION

State (EPA Region)	Number of Facilities
Alabama (IV) California (IX) Illinois (V) Kansas (VII) Kentucky (IV) Louisiana (VI) Maryland (III) Michigan (V) South Carolina (IV) Tennessee (IV) Texas (VI)	1 1 1 2 10 1 1 1 1 1
West Virginia (III)	2
	Total: 29

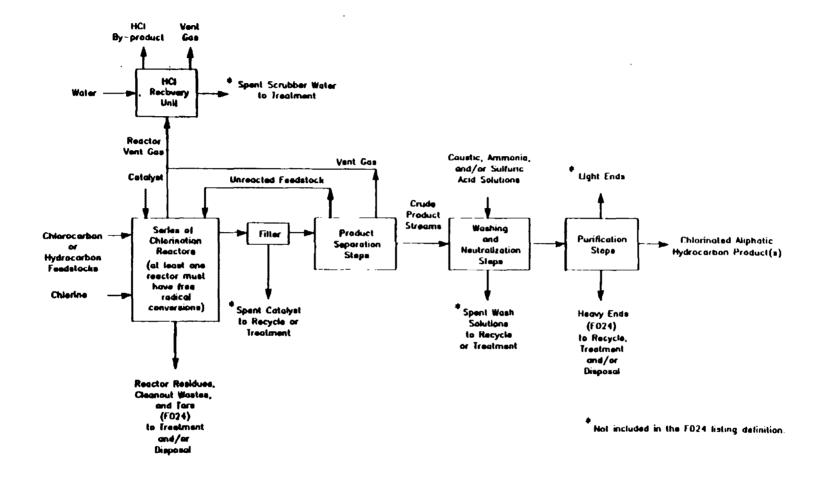
EPA Region	Number of Facilities
I	0
II	0
III	3
IV	5
V	2
VI	17
VII	1
VIII	0
IX	1
Х	<u> </u>
	Total: 29

The chemicals used in chlorinated aliphatic hydrocarbon manufacturing consist of chlorocarbon or hydrocarbon feedstocks and chlorine sources (Cl<sub>2</sub> or HCl). The majority of chlorinated aliphatic hydrocarbon manufacturing is based on five general chemical processes, as follows:

- Free radical initiated addition, substitution, and pyrolysis reactions;
- (2) Lewis acid catalyzed addition and substitution reactions;
- (3) Oxychlorination;
- (4) Base catalyzed dehydrochlorination; and
- (5) Zinc chloride catalyzed chlorination of alcohols.

Several of these processes may be integrated within a facility to convert the feedstock into a variety of desirable products. F024 may be generated when free radical catalyzed processes are either utilized solely or are combined with other reaction processes. A generic process diagram of the production of  $C_1-C_5$  chlorinated aliphatic hydrocarbons (F024) is presented in Figure 2-1.

As shown in Figure 2-1, an organic feedstock is fed, along with a chlorine source, into a series of chlorination reactors. Desired chemical conversions are catalyzed in the reactor by heat, by a combination of heat and ultraviolet radiation, or by catalysis chemicals (such as FeCl<sub>3</sub>, CuCl<sub>2</sub>, ZnCl<sub>2</sub>, or NaOH); at least one such conversion must be catalyzed by free radicals for a generated waste to be considered F024. These reactions may be conducted in either the gas or the liquid phase. F024 is generated by these processes as reactor residues, tars, and periodic clean-out wastes.



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Figure 2-1. Generic Flow Diagram for Processes Generating F024

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The reaction product is then sent to a filtration unit where spent catalysts, if any, are removed for recycle or treatment. Spent catalysts are not included in the F024 listing definition.

The remaining product undergoes a separation step in which it is quenched or cooled with water and/or distilled. This step separates the unreacted feedstock from the crude product and helps prevent product decomposition. The unreacted feedstock is then recycled back to the reactors. Hydrochloric acid is usually a major co-product from the reactor and is typically recovered from the vent gases generated by both the reactor and the quenching/cooling step.

The crude product streams are then generally put through a series of washing, neutralization, and drying steps, which aid in separating the organic phase (product) from the aqueous phase. The spent wash solutions generated in these steps, along with the wastewater generated by HCl recovery, are then recycled or treated, resulting in treated wastewater and various wastewater treatment sludges. This wastewater and the wastewater treatment sludges are not included in the F024 listing definition.

Finally, the product stream goes through a series of purification and separation steps, using fractional distillation and filtration techniques. The distillation residues or heavy ends resulting from these steps are included in the F024 listing definition. The light ends resulting from these steps are not included in the F024 listing definition.

#### 2.2 Waste Characterization

Table 2-2 presents a summary of the available characterization data for F024. Data are presented for all BDAT List constituents that were detected in one or more F024 samples. The summary in Table 2-2 was compiled from data submitted by industry and from data collected by EPA for a wide range of F024 waste matrices. These data include wastes generated from various  $C_1$ - $C_5$  manufacturing processes and include a wide range of physical forms (i.e., liquids, solids, and sludges). Specific data for 11 of the 16 facilities represented in this table have been claimed RCRA Confidential Business Information (CBI), and can be found in the confidential portion of the rulemaking record for F024.

As shown in Table 2-2, the variability among different types of F024 is quite extensive. The waste contains up to 95% organic constituents (BDAT List organics and non BDAT List organics), less than 1% BDAT List metals and up to 10% each of oil and grease, moisture, and ash. The wide variations in both the detection limits and the analytical results are caused by the differences in the F024 waste matrices.

## 2.3 Determination of Waste Treatability Group

EPA bases its treatability group decisions primarily on whether wastes are generated by the same or similar industries from similar processes. EPA believes that such groupings can be made because of the high probability

that the waste characteristics that affect treatment performance will be similar for the different types of F024 wastes and therefore, similar levels of treatment performance can be achieved. Based on the similarities among the industries and processes generating F024, the various types of F024 were combined into one waste treatability group.

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#### Table 2-2

# SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

					<u>Co</u>	ncuntratio	<u>n In FO</u>	24 (pp	n)					
<b>BOV</b>	List Constituent	(a)	<u>(b)</u>	<u>(c)</u>	(1)	<u>(e)</u>	(+)	<u>(a)</u>	<u>(h)</u>	<u>(i)</u>	<u>(j)</u>	<u>(</u> +)	())	Range (ppm)
222.	Acutone	NR	NR	NR	×200	< 20 <b>,000</b>	< 200	< 10	4.6	21,000	< ZO, 000	+0.05	· 500	<0.05-21.000
4.	Bunzene	33-1,900	NR	20	< 100	< 10,000	< 100	<5	< 1	<1,000	•10,000	•0.025	• 250	<0.025-1.900
5.	Bromodichioro- methane	1,260	NR	NR	< 100	< 10,000	< 100	<5	<1	<1,000	<10,000	· B. 025	<250	<0.025.7.200
7.	Carbon tetra- chlaride	ND-50.400	100-1,000	NR	< 100	< 10,000	< 100	<5	9.0	<1,000	< 10,000	+0.025	×250	<0,025-50,400
<b>9</b> .	Chlorobenzene	1.7-3,200	NR	NR	<100	<10,000	174	<5	< I	<1,000	<10,000	<0.025	×25D	<0.025 d. 200
10.	2-Chloro-1,3- butadiene	NR	NR	NR	5,462	< 200,00 <b>0</b>	<2,000	<100	< 20	139,721	× 200 , 000	50.5	<5,000	<0.5.139,721
14.	Chloroform	ND-136	100-1,000	NR	<100	<10,000	< 100	<5	• 1	<1,000	<10,000	•0.025	· 250	×0.025 1,000
16.	3-Chloropropene	ND	ND	ND	<2,000	<200,000	<2.000	<100	< 20	«20,000	285,486	·0 5	۰5, <b>0</b> 00	<0.5.285,486
20.	trans-1,4-Di- chioro-2-buiene	NR	NR	NR	4,691	<200,000	<2,000	<100	< 20	2,112	<200,000	×0.5	-5,000	<0.5 4,09)
22.	1,1-Dichloro- ethane	1.7-440,000	NR	NH	<100	<10,000	<100	<5	< 1	<1,000	<10,000	×0-025	·250	50-025-440, mm
23	1,2-Dichloro- ethane	ND-950,000	10,000-500,000	NR	<100	<10,000	2,708	<5	<1	<1,000	<10,000	KU, U25	11,000	<0.052-320.000
26	. 1,2-Dichioropro- pang	54-191	NR	NR	<100	177,024	< 100	<5	<1	<1,000	230,000	+0-025	×250	< <b>9 025 2</b> 30,000
27	trans-1,3-Di- chloropropene	540	NR	NR	<100	260,036	< 100	<5	۲ ا	<1,000	290,000	\$0.025	× 250	\$ <b>0.025.290</b> .000
ND - NH -	- Not detected; det - Not reported.	action limit	not avaliable.											

(a) CBI EPA Listing Reports (References 3-8, 10, 12-14)
(b) Response to 3007 Questionnaire from plant L (Reference 9)
(c) Listing Background Document for F024, p. 34 (Reference 11)
(d) Characterization data from plant A (Reference 27)
(e) Characterization data from plant B (Reference 28)
(f) Characterization data from plant C (Reference 29)
(g) Characterization data from plant D (Reference 30)
(n) Characterization data from plant E (Reference 31)
(i) Plant A (Reference 32)
(j) Plant B (Reference 32)
(k) Plant C (Reference 32)

(I) Plant D (Reference 32)

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## SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

						<u>C</u>	incentratio	n in fo	24 (pp	m)				_	
Ŭ	DAI	List Constituent	<u>(a)</u>	<u>(b)</u>	<u>(c)</u>	<u>(a)</u>	<u>(e)</u>	<u>(f)</u>	<u>(8)</u>	<u>(h)</u>	(1)	<u>(j)</u>	<u>(k)</u>	$\overline{\Omega}$ ).	- <u>Rauñe (Uber</u>
	28.	cis-1,3-Dichlo- ropropene	570	NŔ	NR	< 100	139, <b>760</b>	< 100	۰5	< I	<1,000 <sub>.</sub>	160,000	+0.025	· 25Ú	<0.025 160.000
2	26.	Ethyl benzene	1,1-230	NH	NR	< 100	< 10,000	<100	<b>&lt;</b> 5	<b>\$1</b>	<1,000	<10,000	-0.025	< 250	\$0,025,230
	34.	Nethyl sthyl ketone	ND	ND	ND	< 100	<10,000	< 100	<5	< 1	2,200	×20,000	×0-05	<500	\$0,05-7,200
	<b>38</b> .	Methylene chlor- Ide	5-1,900	NR	ND	< 100	< 10,000	< 100	<b>\$</b> 5	<1	<1,000	s (0,000	ND	< 250	· 1 - 1,900
	40.	l,l,l,2~letra- chlorosthane	50,000	NR	ND	< 100	< 10,000	< 100	<5	~1	<1,000	× 10 , BuO	+0.025	×250	×0-025-58,080
	41.	1,1,2,2-Tetra- chlurgethane	16,000	ND	NÜ	< 100	< 10,000	< 100	<5	< I	<1,000	< 10 , 000	<0.025	+ 250	<0.025-16.000
	42.	Tetrachioro- ethene	1.5-47,200	1,000-10, <b>00</b> 0	ND	< 100	<10,000	< 100	<5	41	<1,000	< 1 <b>0</b> ,000	0,330	× 250	<ul><li>47,200</li></ul>
	43.	Taluene	31~34,000	NR	ND	<100	<10,000	<100	< 5	- 4	<1,000	\$10,000	\$0.025	· 250	<ul><li>0.025_34_000</li></ul>
2-	45.	1,1,1-Tri- chioroethane	1.1-620	NA	ND	< 100	< 10 <b>,000</b>	< 100	<5	<1	«1,000	×10,000	×0.025	<250	<0.025.620
Ð	46.	1,1,2-Trichloro- ethene	260-92,000	NR	ND	< 100	<10,000	891	<b>&lt;5</b>	< 1	<1,000	< 10,000	×0.025	860	\$0.025.92,000
	47.	Trichlorouthene	ND-81,800	1.000-10.000	ND	< 100	< 10,000	<100	<b>&lt;</b> 5	1	<1,000	s10,000	×0,025	· 250	-0.025 B1,B00
	49.	1,2,3-Trichloro- propane	ND	ND	ND	< 100	<10,000	< 100	<5	4	<1,000	9,712	<0.025	· 250	•0.025.9,712
	50.	Vinyl chioride	ND	100-1,000	NR	< 200	< 20,000	<200	<10	< 2	<2,000	< 20 , UUU	×0.05	s500	<0.05 (j.000)
	69.	Benz(a)anthra- cena	ND	ND	ND	<200	<50	<24	<340	<900	<172	< 189	ព. អ្នង	NU	×24-0 ВВН

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ND - ND - Not detected; detection limit not available. NR - Not reported.

(a) CBI EPA Listing Reports (References 3-8, 10, 12-14)

(b) Response to 3007 Questionnaire from plant L (Reference 9)
 (c) Listing Background Document for F024, p. 34 (Reference 11)

(d) Characterization data from plant A (Reference 27)

(a) Characterization data from plant B (Reference 28)

(f) Characterization data from plant C (Reference 29)

- (9) Characterization data from plant D (Reference 30)
- [11] Characterization data from plant E (Reference 31)

(1) Plant A (Reference 32)

- (j) Plant B (Reference 32)
- (k) Plant ( (Reference 32)
- (1) Flant D (Reference 32)

#### SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

		Concentration in F024 (ppm)													
	BOAT	List Constituent	<u>(a)</u>	<u>[b]</u>	<u>(c)</u>	<u>(a)</u>	<u>(a)</u>	(f)	<u>(9)</u>	<u>(h)</u>	<u>()</u>	$\overline{\Omega}$	(k),	<u>(1)</u>	Range (ppm)
	62.	Benzu(a)pyrene	ND	ND	ND	× 200	<50	< 24	< 340 ·	<900	N172	< 189	0 600	+0.351	<0_151_0_600
	<b>63</b> .	Benza(b)fluor- anthene	MD	ND	ND	< 20D	<50	< 24	< 340 ·	<900	÷172	* 189	(1-716	NÜ	- 24 0 716
	64.	Banzo(ghi)pery- lena	DM	ND	ND	< 200	<50	<24	<340	<900	(1/2	× 18A	0.424	×0.351	<0.351/0.421
	65.	Benzo(k)fluor- anthene	HD	ND	ND	<200	< 50	<24	<340 ·	<900	<172	< 189	U.874	ND	×24 U.874
	6H .	Bis(2-chiora- uthyi)ether	ND 9,800	NR	NR	<200	<50	64	<340	<900	<172	<189	+0,354	32.4	sii 351 9,000
	<b>70</b> .	Bis(2-ethyl- hexyl)phthalate	7.9-480	HR	NR	<200	<50	<24	5.9	<900	<172	< 189	7,63	NU	· 24 4hu
	17.	2-Chloro- nephthelene	ND-260	NR	NR	< 20 <b>D</b>	s50	•24	<340	«90Q	×172	- 189	+ () - 35 <del>-</del>	NÛ	· 0 353 260
	<b>B</b> Q .	Chrysene	ND	ND	ND	< 200	<50	<24	<340	<90 <b>0</b>	<172	< 189	ι,υ <del>ο</del>	0.407	- 24 1 110
-1 -1		1,3-Dichioro- benzene	ND-1,300	NR	NH	< 200	<5D	<24	< 340	<900	<172	< 189	·0 (5)	- 0.351	•0.351 1, NO
Ö	87.	1,2 Dichioro- benzone	ND-24,000	NR	NŔ	«200	<50	<24	<340	<900	<1/2	< 189	•0.351	(1) 351	<ul><li>0.351 (24) 000</li></ul>
	<b>88</b> .	1,4-Dichioro- benzene	HD-8,000	NR	NR	<200	· <50	<24	<340	<90I)	<172	193	·0, (5)	2 06	<ul> <li>(1) 353 −8,0000</li> </ul>

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ND - Not detected; detection limit not available.

NR - Not reported.

(a) CBI EPA Listing Reports (References 3.8, 10, 12 14)
(b) Response to 3007 Questionnaire from plant L (Reference 9)
(c) Listing Background Document for F024, p. 34 (Reference 11)
(d) Engracterization data from plant A (Reference 27)
(e) Characterization data from plant B (Reference 28)
(f) Characterization data from plant C (Reference 29)
(g) Characterization data from plant D (Reference 30)
(h) Cheracterization data from plant E (Reference 30)
(h) Cheracterization data from plant E (Reference 30)
(h) Plant A (Reference 32)
(j) Plant B (Reference 32)
(k) Plant C (Reference 32)

(i) Plant D (Reference 32)

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#### SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

					Co	ncentratio	on in FC	)24 (pp	<b>m)</b>					
BDAT	List Constituent		<u>(b)</u>	<u>(c)</u>	<u>(ā)</u>	(•)	(1)	(9)	<u>(n)</u>	<u>_(1)</u>	(1)	( <u>h</u> )	$\overline{\mathbf{m}}$	<u>Bauña (hlm)</u>
92.	Diethyl phihelaic	1.2-120	NR	NR	<200	<50	<24	<340	<900	<172	< 189	NÐ	<0.351	<0.351-120
104.	Di-n-octyl phthelate	34	NR	NR	<200	<50	<24	5.5	<b>&lt;90</b> 0	<172	< 189	ND	<0.351	<0,351-34
110.	Hexachloro- benzene	ND-3,198	NR	NR	<200	< 50	<24	4.7	18,018	<172	< 189	2.06	0.628	<24-18,018
111.	Hexachloro- butadiene	ND-4,074	100-1,000	NR	<200	<5Q	<24	<340	16.470	<172	< 189	<172	<0.351	<0,351-16,4/U
112.	Hexachioro- cyclopentadiene	1.3	NR	NR	<200	<50	<24	<340	<900	<172	<189	<0.351	<0.351	<0.351 ) 1
113.	Hesechloroethane	ND-460,000	1,000-10,000	NR	< 200	<50	<24	<340	<900	<172	<189	0.44?	<0.351	<b>40.351-460</b> , 000
116.	Indeno(1,2,3~cd) pyrene	ND	ND	ND	<200	<50	<24	<340	< 900	<172	< 189	0.411	<0.351	<0.351-0-411
121.	Naphthalene	3.4-330	NR	NR	· < 200	<5Q	<24	<340	<900	<172	< 189	ND	ND	< 24 - 330
126.	Nitrobenzene	1.4	NR	NR	<200	<50	<24	<340	<900	<172	< 189	<0.351	<0. <b>35</b> 1	<0.351-j.4
136.	Pentachloro- benzene	500	NR	NR	ND	ND	ND	0.53	1,290	<860	<945	<1.76	«1.76	<1.76-1,290
137.	Pentachloroethane	ND-26,000	100-1,000	NR	ND	ND	ND	ND	ND	<172	<189	<0.351	ND. 351	<0.351_26,000
141.	Phenanthrene	ND	ND	ND	<200	<50	<24	<340	<900	<172	< 189	1.27	Q.892	<24 1.27
150.	1,2,4-Trichloro- benzene	160-1,40 <b>0</b>	HR	NR	<700	<50	<24	<340	<900	<172	< 189	+0.351	ND	<0.351 1,400
154.	Antimony	NR	NR	NR	<1.8	<1.8	<2.1	2.2	<2.1	<2.9	< 2 . 9	•2.9	×2.9	≤ 1 (0 + 2 − 2)
155.	Arsonic	NR	NR	NR	<0.85	<0.85	7.8	<1.0	<1.0	<1.0	<1.0	<1 D	2	«Q.06.7.N

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ND - Not detected; detection limit not available.

NR - Not reported.

2-11

(a) CB1 EPA Listing Reports (References 3-8, 10, 12-14)

- (b) Response to 3007 Questionnaire from plant L (Reference 9)
- (c) Listing Background Document for F024, p. 34 (Reference 11)
- (d) Characterization data from plant A (Reference 27)
- (e) Cheracterization data from plant B (Reference 28)
- (f) Characterization data from plant C (Reference 29)
- (y) Characterization data from plant D (Reference 30)
- (1) Characterization data from plant E (Reference 31)
- (i) Plant A (Reference 32)
- ()) Plant B (Reference 32)
- (6) Ofant C (Reference 32)

#### SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

						Con	centratio	on in FC	D24 (ppm	)					
	BOAT	List Constituent	<u>(a)</u>	<u>(b)</u>	(c)	(d)	(0)	(1)	<u>(g)</u>	<u>(n)</u>	<u>(i)</u>	<u>(i)</u>	(+)_	_(1)	Range (plum)
	156.	Barlum	NR	NR	NR	0.3	0.22	26	1.4	0.26	0.27	0.30	.14	6.8	0.22.34
	158.	Cadmium	NR	NR	NR	<0.26	<0. <b>26</b>	<0,3	<0.3	<0.3	<0,4	<0.4	, <b>)</b> L	<b>40 4</b>	<0.26 E I
	159.	Chromium (total)	'NR	NR	NR	40.43	<0.46	88	3.5	2.B	<0.4	×0.4	205	57	50 4 285
	160.	Capper	108-110	NR	NR	4.9	2.2	638	406	I	<0.4	×0.4	45	800	5 D . 4 - 800
	161.	Lesd	<50-5	NR	NR	1.36	•0.43	3.6	2.5	< 1 . O	<0.5	•0.5	9-0	3-8	•0,439 a
	162.	Mercury	NR	NR	NR	0.24	· 0 - 1	0.13	<0.I	<0.1	<0,I	<0.1	0 19	·0 I	• U , 1 • U , 2 4
	163.	Nickel	240	NR	NR	«2.2	<2. <b>2</b>	256	71	8	<0.9	<0.9	318	636	s0 9 630
	167.	Vanadium	NR	NR	NR	<0.17	<0.17	10	<0.2	<0.2	<0.J	<u.3< th=""><th>ı</th><th>د. ۱</th><th>50.17 (O</th></u.3<>	ı	د. ۱	50.17 (O
	168.	21nc	7.9-90	NR	NR	0.74	1.9	104	15	0.82	1.6	0.13	443	92	U.73-443
Ņ	169.	Cymnide	NR	NR	NR	<0.50	<0.61	4.57	<0.52	NR	<0.43	×Q.44	×0.49	2.92	su 4.1 4 57
-12	170.	Fluoride	MR	NR	NR	NR	NR	NR	NR	NR	<0.99	<1.00	E. 08	10.5	50,99 iu s
	171.	Sulfide	NR	NR	NR	153	349	<6.4	<5.1	NR	9.1	7.8	×4,8	4.6	×4 6 J49

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NR - Not reported.

- CBI EPA Listing Reports (References 3-8, 10, 12-14) (a)
- Respunse to 3007 Questionnairs from plant L (Reference 9) (6)
- (c) Listing Background Document for F024, p. 34 (Reference 11)
- (d) Characterization data from plant A (Reference 27)
- (a) Characterization data from plant 8 (Reference 28)
- (f) Characterization data from plant C (Reference 29)
- (g) Characterization data from plant D (Reference 30)
- (h) Characterization data from plant E (Reference 31)
- (1) Plant A (Reference 32)
- ()) Plant B (Reference 32)
- (k) Plant C (Reference 32)
- (1) Plant D (Reference 32)

#### SUMMARY OF AVAILABLE CHARACTERIZATION DATA FOR F024

				Conce	entratio	in FO	24 (pot						
BDAT LIST Constituent	(a)	<u>(b)</u>	(c)	(d)	(•)	(f)	(g)	(n)	(1)	<u>(j)</u>	(k)	$\overline{(1)}$	<u>Range (ppp)</u>
207. Hexachloro- dibenzo-p-dioxins	NR	NR	NR	NR	NR	NR	NR	NR	<0.0014	<0.0005	10.4	2.2	<0.005 10 4
208. Hexachlorodibenzo- furans	NR	NR	NR	NR	NR	NR	NR	NR	<0.0007	<0.0003	3.1	50.5	<0,0003-50-5
209. Pentachlorodibenzo~ p-dtoxins	NR ,	NR	NR	NR	NR	NR	NR	NR	<0.0014	<0.0005	2.3	0.31	<b>×0</b> ,0005÷2−3
210. Pentachlorodibenzo~ furans	NR	NR	NR	NR	NR	NR	NR	NR	<0.000 <b>5</b>	<0.0002	1.6	28.7	<b>NO.0002</b> 28,7
212. Tetrachlorodibenzo- furens	NR	NR	NR	NR	NR	NR	NR	NR	<d.0002< td=""><td>&lt;0.00D2</td><td>0.63</td><td>12</td><td>&lt;0.0002-12</td></d.0002<>	<0.00D2	0.63	12	<0.0002-12
Other Parameters													
Heating Value (BTU/1b)	NR	NR	NR	NR	NR	NR	NR	NR	9,565	7,876	8,214	9,283	7,876-9,565
Total Solids (%)	NR	NR	NR	NR	NR	NR	NR	NR	51.5	1.9	89	79.H	1.9-79.8
Ash Content (% of Tula; Solids)	NR	NR	NR	NR	NR	NR	NR	NR	ND	0.5	13.6	11.7	ND+13.8
Total Carbon (mg/kg)	NR	NR	NR	NR	NR	NR	NR	NR	395.000	350,000	388,000	453,000	350,000 453,000
Total Halogens (mg/kg)	NR	NR	NR	NR	NR	NR	NR	NR	458,000	26,300	31,900	64,300	26,300°458,000
Moisture (%)	NR	NR	NR	NR	NR	NR	NR	NR	0.03	0.04	3.53	8.48	0.03 8 48

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2-13

ND - Not detected; detection limit not available. NR - Not reported.

(a) CBI EPA Listing Reports (References 3-8, 10, 12-14) (b) Response to 3007 Questionnaire from plant L (Reference 9) (c) Listing Background Document for F024, p. 34 (Reference 11) (d) Characterization data from plant A (Reference 27) (e) Characterization data from plant B (Reference 28) (f) Characterization data from plant C (Reference 29) (g) Characterization data from plant D (Reference 30) (h) Characterization data from plant E (Reference 31) (i) Plant A (Reference 32) (j) Plant B (Reference 32) (H) Plant C (Reterence 32) (1) Plant D (Reference 32) F024 BGD - 2

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#### 3.0 APPLICABLE AND DEMONSTRATED TREATMENT TECHNOLOGIES

This section identifies the treatment technologies that are applicable to F024 and determines which, if any, of the applicable technologies can be considered demonstrated for the purpose of establishing BDAT.

To be applicable, a technology must theoretically be usable to treat the waste in question or to treat a waste that is similar in terms of parameters that affect treatment selection. (For detailed descriptions of the technologies applicable to these wastes, or for similar wastes, refer to EPA's <u>Treatment Technology Background Document</u> (Reference 15).)

To be demonstrated, a technology must be employed in full-scale operation for treatment of the waste in question or a similar waste. Technologies available only at pilot- and bench-scale operations are not considered in identifying demonstrated technologies.

#### 3.1 Applicable Treatment Technologies

Since F024 contains high concentrations of organic compounds (Section 2.0), applicable treatment technologies include those that destroy, reduce, or recover the total amount of various organic compounds in the waste. The Agency has identified the following treatment technologies as applicable for F024: (1) incineration (fluidized-bed, rotary kiln, and liquid injection)

followed by stabilization of incinerator ash (if necessary to control leacning of metals), and chemical precipitation followed by vacuum filtration of scrubber water (if dissolved metals are present at treatable levels); (2) solvent extraction followed by incineration or recycle of the extract, and stabilization for nonwastewater raffinate with treatable leachable metals levels and/or chemical precipitation followed by vacuum filtration for wastewater raffinate if dissolved metals are present at treatable levels; and (3) total recycle or reuse. These treatment technologies were identified based on current literature sources, field testing, and current waste treatment practices.

#### Incineration/Stabilization/Chemical Precipitation/Sludge Filtration

Incineration is a destruction technology in which energy, in the form of heat, is transferred to the waste to destabilize chemical bonds and eventually destroy hazardous constituents. In general, two residuals are generated by incineration processes: ash and scrubber water. Incinerator ash may require stabilization (which results in the formation of a chemically- or physically-stabilized treatment residual) to reduce the leachability of metals in the waste. Scrubber water may require treatment using chemical precipitation followed by vacuum filtration to remove dissolved metals from the wastewater. Chemical precipitation removes dissolved metals from wastewater by forming an insoluble metal precipitate sludge. Vacuum filtration separates the precipitated sludge from the wastewater.

#### Solvent Extraction

Solvent extraction is a separation technology in which organics are removed from the waste due to greater constituent solubility in the solvent phase than in the waste phase. This technology results in the formation of two treatment residuals: the treated waste residual and the extract. The treated waste residual may be further treated by stabilization for nonwastewater raffinate with treatable leachable metals levels and/or chemical precipitation followed by vacuum filtration for wastewater raffinate if dissolved metals are present at treatable levels. The extract may be recycled or treated further by incineration.

# Total Recycle or Reuse

Total recycle or reuse processes do not generate residuals. The applicability of these processes depends on the type of F024 generated. For example, heavy ends or distillation bottoms are recycled in other production processes at some facilities; however, reactor cleanout wastes are generally not suitable for recycle or reuse.

The Agency recognizes that wastewater forms of F024 may be generated from the treatment of F024, including scrubber water from incineration technologies and wastewater raffinate from solvent extraction processes. For wastewater forms of F024 that may contain hazardous organic constituents at treatable levels, applicable technologies include those that destroy or reduce

the total amount of various organic compounds in the waste. Therefore, the Agency has identified biological treatment and carbon adsorption as potentially applicable for treatment of F024 wastewaters with treatable organic constituent concentrations.

# 3.2 Demonstrated Treatment Technologies

The Agency has identified incineration (including rotary kiln, liquid injection, and fluidized-bed incineration), stabilization, and total recycle or reuse as the demonstrated treatment technologies for F024 and F024 nonwastewater residuals. The Agency is not aware of any facilities that treat, on a full-scale operational basis, F024 or similar wastes using solvent extraction; therefore, EPA believes that solvent extraction is not currently demonstrated for F024. The Agency has identified chemical precipitation followed by sludge filtration as demonstrated for treatment of metals in F024 wastewater residuals.

#### Incineration

Rotary kiln incineration is demonstrated on a full-scale operational basis for treatment of F024 at five facilities. Liquid injection incineration is demonstrated on a full-scale operational basis for treatment of F024 at four facilities. In addition, fluidized-bed incineration is demonstrated on a full-scale basis for treatment of F024 at one facility. The treatment process

the Agency tested was a rotary kiln incinerator with a secondary compustor and combustion gas scrubber system.

### Stabilization

The Agency evaluated the performance of a pilot-scale stabilization process on incinerator ash similar to rotary kiln incinerator ash from treatment of F024. (These data are presented in EPA's BDAT Background Document for K048, K049, K050, K051, and K052 (Reference 34).) In addition, the Agency has recently completed an evaluation of the performance of a pilot-scale stabilization process on F024 rotary kiln incinerator ash.

### Total Recycle or Reuse

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EPA is aware of four plants that recycle or reuse F024 in manufacturing processes on a full-scale basis. Specific information regarding the recycle or reuse of these wastes has been claimed RCRA Confidential Business Information (CBI) by the facilities.

#### Chemical Precipitation Followed by Vacuum Filtration

Although the Agency is not aware of any facilities that treat F024 wastewaters, chemical precipitation followed by vacuum filtration is demonstrated for wastewaters judged to be similar to scrubber water generated from the incineration of F024. Therefore, this technology is determined to be demonstrated for F024 wastewaters. The Agency tested chemical precipitation followed by vacuum filtration for a waste mixture similar to F024 at one facility (as shown in Section 4.0).

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#### 4.0 TREATMENT PERFORMANCE DATA BASE

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This section presents the data available to EPA on the performance of demonstrated treatment technologies for F024. These data are used elsewhere in this document to determine which technologies represent BDAT (Section 5.0), to select constituents for regulation (Section 6.0), and to develop treatment standards (Section 7.0). In addition to using full-scale treatment performance data, eligible data may include that developed at research facilities or obtained at less than full-scale operations, provided that the technology is demonstrated in full-scale operation for a similar waste or wastes (as described in Section 3.0).

Treatment performance data, to the extent that they are available to EPA, include the concentrations for a given constituent in the untreated and treated waste, values of operating parameters measured at the time the waste was being treated, values of relevant design parameters for the treatment technology, and data on waste characteristics that affect performance of the treatment technology.

Where data are not available on the treatment of the specific waste of concern, the Agency may elect to transfer performance data on the treatment of a similar waste or wastes, using a demonstrated technology. To transfer data from another waste treatability group, EPA must find that the waste of concern is no more difficult to treat (based on the waste characteristics that

affect performance of the demonstrated treatment technology) than the treated wastes from which treatment performance data are being transferred.

Treatment performance data were not available for BDAT List metals in FO24 wastewaters. Treatment performance data from lime and sulfide precipitation followed by vacuum filtration of KO62 mixed with metal-bearing characteristic wastes were transferred to FO24 wastewaters.

Table 4-1 presents the BDAT List constituents that were detected in the untreated waste streams during the rotary kiln incineration of F024 from plants A, B, C, and D, as well as a biological sludge and an organic liquid that were then burned with the F024. Table 4-2 presents the BDAT List constituents detected in the kiln ash residual, and Table 4-3 presents the BDAT List constituents detected in the scrubber water residual. Table 4-4 presents design and operating data for the rotary kiln and the secondary combustor. Table 4-5 presents treatment performance data for lime and sulfide precipitation followed by vacuum filtration of K062 mixed with other metalbearing characteristic wastes. Testing procedures used to analyze for the BDAT List constituents are identified in the analytical quality assurance/quality control discussion in Appendix A of this background document.

4-2

# Table 4-1

# WASTE CHARACTERIZATION DATA COLLECTED BY EPA FOR FO24 FROM PLANTS<sup>a</sup> A, B, C, AND D, AND OTHER WASTES TREATED BY ROTARY KILN INCINERATION

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	:	Wastes <sup>b</sup> Fed		tary Kiln	Concentration in Untreated Wastes <sup>b</sup> Fed to the Secondary Combustor					
	· ·	Plant C	Plant D	Sludge	Plant A	Plant B	Organic Liquid			
BDAT	List Constiluent	<u>(ppm)</u>	<u>(ppm)</u>	(ppm)	<u>(ppm)</u>	(ppm)	(ppm)			
222.	Acetone	<0.05	<500	<500	21,000	<20,000	<2,000			
10.	2-Chloro-1,3-butadiene	<0.5	<5,000	<5,000	139,721	<200,000	<20,000			
14.	Chloroform	<0.025	<250	<250	<1,000	<10,000	<1,000			
16.	3-Chloropropene	<0.5	<5,000	<5,000	<20,000	285,486	<20,000			
20.	trans-1,4-Dichloro-2-butene	<0.5	<5,000	<5,000	2,112	<200,000	<20,000			
23.	1,2-Dichloroethane	<0,25	11,000	<250	<1,000	<10,000	26,068			
26.	1,2-Dichloropropane	<0.025	<250	<250	<1,000	230,000	<1,000			
27.	trans-1,3-Dichloropropene	<0.025	<250	<250	<1,000	290,000	<1,000			
28.	cis-1, 3-Dichloropropene	<b>&lt;0.025</b>	<250	<250	<1,000	160,000	<1,000			
34.	Hethyl ethyl ketone	<0. <b>05</b>	<250	<500	2,200	<20,000	<2,000			
38.	Methylene chloride	<0.025	<250	<250	<1,000	<10,000	1,884			
42.	Tetrach) or oe thene	0.3	<250	< 500 ·	<1,000	<10,000	1,838			
43.	Toluene	<0.025	<250	<250	<1,000	<10,000	25,930			
45.	1, 1, 1-Trichloroethane	<0.025	<250	<250	<1,000	<10,000	1,214			
46.	1,1,2-Trichloroethane	<0.02 <b>5</b>	860	<250	<1,000	<10,000	<1,000			
49.	1,2,3-Trichloropropane	<0.025	<250	<2,500	<1,000	9,712	×1,000			
59.	Benz (a) anthracene	0.89	(0.351	<134	(172	< 189	× 196			
62.	Benzo(a)pyrene	0.6	<0.351	<134	<172	< 189	+ 196			
63.	Benzo(b)fluoranthene	0.72	(0.351	<134	<172	< 189	× 196			
64.	Benzo(ghi)perylene	0.42	<0.351	< 134	< 172	< 189	- 196			

Note: The variation in detection limits for the same constituent was caused by the variation in the waste matrices.

NA - Not analyzed. <sup>a</sup>Plant codes are listed in Appendix B. <sup>b</sup>F024 from Plants A and B were liquids; F024 from Plants C and D were sludges.

# WASTE CHARACTERIZATION DATA COLLECTED BY EPA FOR FO24 FROM PLANTS<sup>a</sup> A, B, C, AND D, AND OTHER WASTES THEATED BY ROTARY KILN INCINERATION

	:	Wastes <sup>D</sup> Fed	tion in Un to the Ro				reated Wastes <sup>b</sup>
BDAT	List Constituent	Plant C <u>(ppm)</u>	Plant D (ppm)	Sludge (ppm)	Plant A <u>(ppm)</u>	Plant B (ppm)	Organic Liquid (ppm)
65.	Benzo(k)fluoranthene	0.87	<0.351	<134	< 172	< 189	<196
68.	Bis(2-chloroethyl)ether	< 0.351	32.4	< 134	<172	< 189	< 196
70.	Bis(2-ethylhexy))phthalate	7.6	0.58	< 134	< 172	< 189	< 196
80.	Chrysene	1.1	0.41	< 134	<172	< 189	<196
87.	1,3-Dichlorobenzene	<0. <b>351</b>	<0.351	<134	<172	< 189	302
<b>8</b> 8.	1,4-Dichlorobenzene	<0.351	2.08	<134	< 1 <b>7</b> 2	< 189	< 196
89.	3,3'-Dichlorobenzidine	<1.76	<1.76	<266	< <b>86</b> 0	< 945	915
92.	Diethyl phthalate	0.05	<0.351	15.45	< 172	< 189	+ 196
110.	Hexachlorobenzene	2.1	0.63	<134	< 1 <b>7</b> 2	< 189	× 196
113.	Hexachloroethane	0.44	<0.351	<134	<172	<189	< <b>196</b>
116.	Indeno(1,2,3-cd)pyrene	0.41	<0.351	<134	<172	< 189	< 196
121.	Naphthalene	<0.351	<0.351	1,500	<172	< 189	< 196
141.	Phenanthrene	1.27	0.90	<134	<1 <b>7</b> 2	< 189	< 196
142.	Phenol	< 0.351	<0.351	<134	<1 <b>7</b> 2	< 189	1,842
150.	1,2,4-Trichlorobenzene	<0.351	<0.351	<134	< 172	< 189	279
155.	Arsenic	<1.0	2.0	<1.0	· <1.0	<1.0	NA
156.	Bartum	34	6.8	35	0.27	0.30	NA
158.	Cadmium	3.1	<0.4	3.1	<0,4	<0.4	NA
159.	Chromium (total)	285	57	294	<0,4	<0.4	NA
160.	Copper	45	800	46	<0.4	< <b>0.</b> 4	NA

NA - Not analyzed.

Note: The variation in detection limits for the same constituent was caused by the variation in the waste matrices.

<sup>A</sup>Plant codes are listed in Appendix B. <sup>b</sup>F024 from Plants A and B were liquids; F024 from Plants C and D were sludges.

# WASTE CHARACTERIZATION DATA COLLECTED BY EPA FOR FO24 FROM PLANTS<sup>a</sup> A, B, C, AND D, AND OTHER WASTES TREATED BY ROTARY KILN INCINERATION

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	:		ation in Un 1 to the Ro				reated Wastes <sup>b</sup> ry Combustor
		Plant C	Plant D	Sludge	Plant A	Plant B	Organic Liquid
BDAT	List Constituent	(ppm)	<u>(ppm)</u>	(ppm)	(ppm)	(ppm)	(ppm)
161.	Lead	9.0	3.8	10	<0.5	<0.5	NA
162.	Mercury	0.19	<0.1	0.28	<0.1	<0.1	NA
163.	Nickel	318	636	333	<0.9	<0.9	NA
165.	Silver	<0.4	<0.4	0.56	<0,4	<0.4	NA
167.	Vanadium	1,0	1.3	1.4	<0.3	<0.3	NA
168.	Zinc	443	92	455	1.6	0.73	NA
169.	Cyanide	<0,49	2.92	NA	<0.43	<0.44	NA
170.	Fluoride	1,08	10.5	NA	<0.99	<1.00	NA
171.	Sulfide	<4.8	<4.6	NA	9.1	7.8	NA
207.	Hexachlorodibenzo-p-dioxins	0.01	0.002	NA	$< 1.4 \times 10^{-6}$	<5.3x10 <sup>-7</sup>	NA
208.	Hexachlorodibenzofurans	0.003	0.05	NA	<7.1x10-7	<2.7x10 <sup>-7</sup>	NA
209.	Pentachlorodibenzo-p-dioxins	0.002	0.0003	NA	<1.4x10 <sup>-D</sup>	$(5.0 \times 10^{-7})$	NA
210.	Pentachlorodibenzofurans	0.002	0.03	NA	<5.1x10 <sup>-1</sup>	<2.0x10 <sup>-1</sup>	NA
212.	Tetrachlorod ibenzofurans	0.001	0.01	NA	<1.6x10 <sup>-7</sup>	<1,6x10 <sup>-1</sup>	NA

NA - Not analyzed.

Note: The variation in detection limits for the same constituent was caused by the variation in the waste matrices.

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<sup>A</sup>Plant codes are listed in Appendix B. <sup>b</sup>F024 from Plants A and B were liquids; F024 from Plants C and D were sludges.

## TREATMENT PERFORMANCE DATA COLLECTED BY EPA FROM ROTARY KILN INCINERATION OF FO24 FROM PLANTS A, B, C, AND D

#### KILN ASH RESIDUAL

	ï		Concentra	tion In Kilr	n Ash - Tota	1 Compositio	n (mg∕kg)	
BDAT	List	Sample Set	Sample Set	Sample Set		Sample Set	Sample Set	Range
Const	ituent			13		15	16	(mg/kg)
21,	Dichlorodi- fluoromethane <sup>8</sup>	0.71	1.40	1.50	1.60	1.70	1.40	0.71-1.70
<b>9</b> 2.	Diethyl phthalate <sup>b</sup>	0.67	0.67	0.67	0.60	0.45	1.14	0.45-1.14
170.	Fluoride	NA	NA	2.11	NA	NA	NA	2.11
X			Con	centration	in Kiln Ash	- TCLP (mg/1	)	Range (mg/1)
154.	Antimony	0.062	<0.029	<0.029	<0.029	0.042	0.030	<0.029-0.062
155.	Arsenic	0.079	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01-0.079
156.	Barium	1.64	1.67	1.94	1.78	1.64	1.56	1.56-1.94
157.	Beryllium	<0.001	0.002	0.001	0.002	0.002	0.001	<0.001-0.002
158.	Cadmium	<0.004	0.037	0.033	0,028	0.016	0.027	<0.004-0.037
159.	Chromium (total)	<0.004	0.28	0.29	0.31	0.32	0.13	<0.004-0.32
160.	Copper	0.026	6.48	5.37	4.57	4.54	4.80	0.026-6.48
	• -		1.2	29.3	22.0	21.6	1.25	0.032-29.3
162.	Hercury	0.0002	<0.0002	<0.0002	0.0003	<0.0002	<0.0005	0,0002-0,0003
167	Nickel	0.24	3.08	3.68	4.45	4.78	5.3.	0.94-5 X.2
				-			0.005	0.003-0.008
168.	Zine	0.060	1.84	1.90	1.61	1.56	2.78	0.060 2.78
163. 167.	Nickel Vanadium	0.24 0.005	<0,0002 3.08 0.004	<0.0002 3.68 <0.003	0.0003 4.45 0.008	<0.0002 4.78 <0.003	<0.0002 5.3. <sup>3</sup> 0.005	0.0002-0.00 0.24-5.3. 0.003-0.00

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NA - Not analyzed. <sup>a</sup>This constituent was also detected in the laboratory blank at 0.27 mg/kg. <sup>b</sup>This constituent was also detected in the laboratory blank at 0.51 mg/kg.

## TREATMENT PERFORMANCE DATA COLLECTED BY EPA FROM ROTARY KILN INCINERATION OF FO24 FROM PLANTS A, B, C, AND D

#### SCRUBBER WATER RESIDUAL

		Conc	entration in	Scrubber Wate	er - Total Co	mposition (mg	/1)
BDAT	List	Sample Set	Sample Set	Sample Set	Sample Set	Sample Set	Sample Set
Const	ituent	<b>#</b> 1		13			
6.	Bromomethane	< <b>0</b> .01	<0.01	<0.01	<0.01	<b>&lt;0.01</b>	0.032
12.	Chloroethane	<0.01	<0.01	<0.01	<0.01	<0.01	0.031
15.	Chloromethane	<0.01	<0.01	<0.01	<0.01	<0.01	0.020
21.	Dichlorodifluoromethane	<0.01	<0.01	0.29 <sup>a</sup>	0.36 <sup>a</sup>	0.40 <sup>a</sup>	0.44 <sup>a</sup>
50.	Vinyl chlorid <del>e</del>	<0.01	<0.01	<0.01	<0.01	<0.01	0.026
92.	Diethyl phthalate	<0.0116	<0.0121	<0.0108	<0.0104	0.057	<0.0107
154.	Antimony	11.3	12.3	9.27	10.8	7.94	8.72
155.	Arsenic	0.48	<0.2	0.58	0.51	<0.1	(0.5
156.	Barium	33.5	18.3	24.7	38.4	27.0	26.8
158.	Cadmium	6.62	6.47	6.95	6.63	2.90	4.81
159.	Chromium (total)	27.6	32.1	15.1	33.1	24.4	24.8
160.	Copper	175	190	107	209	144	158
161.	Lead	361	401	433	358	386	371
162.	Mercury	0.31	0.20	0.45	0.29	0.42	0.34
163.	Nickel	13.7	16.5	14.2	17.3	14.2	13.4
164.	Selenium	3.08	1.09	<2.5	<2.5	<5.0	<0.5
165.	Silver	10.9	10.7	10.1	9.45	7.12	5.38
167.	Vanadium	1.05	1.72	0.37	1.84	1.52	0.97
168.	Zinc	160	162	174	17 <del>9</del>	128	137
170.	Fluoride	NA	NA	153	NA	NA	NA
207.	Hexachlorodibenzo-p-dioxins	NA	NA	<7.2×10 <sup>-6</sup>	NA	NA	NA
208.	Hexachlorodibenzofurans	NA	NA	0.0003	NA	NA	NA
209.	Pentachlorodibenzo-p-dioxin		NA	<5.6×10 <sup>-6</sup>	NA	NA	NA
210.	Pentachlorodibenzofurans	NA	NA	0.0001	NA	NA	NA
212.	Tetrachlorodibenzofurans	NA	NA	9.6x10 <sup>-6</sup>	NA	NA	NA

NA - Not analyzed.  $^{\rm a}{\rm This}$  constituent was also detected in the laboratory blank at 0.23 mg/l.

				Operating V	alue		
		Sample Set	Sample Set	Sample Set	Sample Set	Sample Set	Sample Set 16
<u>Parameter (units)</u>	Design Value	(11:20 am) <sup>a</sup> (11:45 am) <sup>a</sup>	(12:15 pm) <sup>a</sup> (12:45 pm) <sup>a</sup>	(1:15 pm) <sup>a</sup> (1:45 pm) <sup>a</sup>	(2:15 pm) <sup>a</sup> (2:45 pm) <sup>a</sup>		(4:15 pm) <sup>a</sup> (4:45 pm) <sup>a</sup>
Kiln Outlet Temperature ( <sup>o</sup> F)	<b>1,000-1,</b> 700	1,391 1,255	1,393 1,405	1,360 1,604	1,425 1,413	1,452 1,450	1,207 1,551
Rotary Kiln Solid Waste <sup>b</sup> Feed Rate (lbs/minute)	NS	32 32	32 32	32 32	32 32	32 32	32 32
Kiln Rotational Speed <sup>C</sup> (RPM)	0.2-0.6	0.45 0.45	0.45 0.45	0.45 0.45	0.25 0.45	0.23 0.27	0.27 0.1
BTU Loading - Rotary Kiln (MMBTU/hour)	11	16 17	17 16	16 16	16 16	16 16	16 15
Secondary Combustor Temperature ( <sup>O</sup> F)	1,800-2,200	2,025 2,021	2,056 2,142	2,132 1,990	2,093 2,119	2,123 2,126	2,006 1,877
Secondary Combustor Liquid Waste <sup>d</sup> Feed Rate (1bs/minute)	NS	18.2 27.7	29.2 27.7	29.4 32.3	25.9 25.6	29.6 25.4	13.4 1,2
BTU Loading-Secondary Combustor (MMBTU/hour)	19	10 14	15 14	15 17	14 13	15 15	11 7

#### DESIGN AND OPERATING DATA FOR THE ROTARY KILN AND SECONDARY COMBUSTOR

Note: Kiln solids residence time is estimated to be 30-60 minutes. Secondary combustor residence time is estimated to be approximately 3 seconds. Both estimates are based on discussions with plant personnel.

NS - Not specified.

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a Time of data collection.

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b This solid represents F024 from plants C and D and the sludge that were treated by the rotary kill incineration system.

<sup>c</sup> The kiln rotational speed was decreased during the run to allow for a longer residence time of the kiln solids , and to lower the temperature of the waste heat boiler.

d This liquid represents F024 from plants A and B and the organic liquid that were treated by the rotary kills incineration system.

#### TREATMENT PERFORMANCE DATA COLLECTED BY EPA FOR K062 WASTE MIXTURE PLANT Z - LIME AND SULFIDE PRECIPITATION FOLLOWED BY VACUUM FILTRATION<sup>a</sup>

	•	Sample	<u>Set #8</u>	Sample	Set #11	Sample	Set #12
	Detected BDAT List Metal <u>Constituent</u>	Concentration in Untreated KD62 (ppm)	Concentration in KO62 Wastewater (ppm)	Concentration in Untreated KD62 (ppm)	Concentration In Treated K062 Wastewater (ppm)	Concentration in Untreated KO62 (ppm)	Concentration in Treated KU62 Wastewaler (ppm)
154.	Antimony	<10	< 1	< 10	< 1	<10	<1.00
155.	Arsentc	<1	<o.1< td=""><td>&lt; 1</td><td>&lt;0.1</td><td>&lt; 1</td><td>&lt;0.10</td></o.1<>	< 1	<0.1	< 1	<0.10
156.	Bartum	<10	<1	< 10	<1	12	<1.00
157.	Beryllium	< 2	<0.2	<2	<0.2	< 2	<0.20
158.	Cadintum	<5	<0.5	<5	<0.5	23	<u>د ج</u>
221.	Chromium (hexavalent)	D.13	<0.01	0.08	0.106	0.30	×0 01
159.	Chromium (total)	831	0.15	395	0.12	617	0.18
160.	Саррег	217	0.16	191	0.14	137	0.24
161.	Lead	212	<0.01	<10	<0.01	136	<b>\$0.01</b>
162.	Mercury	< 1	<0.1	<1	< <b>0</b> . I	×1	•0.10
163.	Nickel	669	Π.36	712	0.33	382	0,39
164.	Selenium	<10	< 1	< 10	< 1	< 10	< F. 00
165.	Silver	≪2	<0.2	<2	<0.2	×2	÷0,20
166.	Thallium	< 110	∢ ۱	< 10	< 1	<10	• E 00
168.	Zinc	151	0.130	5	0.070	135	U. 100 -

<sup>B</sup>Only 3 of the 12 data sets analyzed at Plant Z represented treatment by lime and sulfide precipitation followed by vacuum filtration. The remaining nine data sets included pretreatment technologies such as chromium reduction and treatment for cyanide. These data are not included since these pretreatment technologies are not applicable to F024.

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Source: Envirite Onsite Engineering Report (Reference 33).

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The technology that is found to perform best on a particular waste stream is then evaluated to determine whether it is "available." To be available, the technology must (1) be commercially available, and (2) provide "substantial" treatment of the waste, as determined through evaluation of treatment performance data that have been corrected for accuracy. In determining whether treatment is substantial, EPA may consider data on a treatment technology's performance on a waste similar to the waste in question, provided that the similar waste is at least as difficult to treat. If it is determined that the best technology is not available, then the next best technology is evaluated, and so on.

#### 5.1 Review of Treatment Performance Data

The available treatment performance data presented in Section 4.0 were reviewed and assessed to determine whether they represent operation of a well-designed and well-operated system, whether sufficient quality assurance/quality control measures were employed to ensure the accuracy of the data, and whether the appropriate measures of performance were used to assess the performance of the treatment technology.

The treatment performance data and the design and operating data collected during the test of rotary kiln incineration of F024 at plant X, stabilization of F024 incinerator ash at plant Y, and lime and sulfide precipitation followed by vacuum filtration of K062 mixed with other metal-bearing characteristic wastes at plant Z were reviewed for the points described above.

The appropriate measures of performance (total constituent concentration for incineration and lime and sulfide precipitation followed by vacuum filtration, and TCLP for stabilization) were used to assess the treatment systems. Design and operating data and quality assurance/quality control information for samples collected at plants X, Y, and Z are presented in References 32, 21, and 33, respectively. These data were considered to determine BDAT for F024.

F024 treatment performance data are not available for liquid injection and fluidized-bed incineration. Therefore, in the absence of treatment performance data for this waste, liquid injection and fluidized-bed incineration were not selected as BDAT for F024. However, the Agency believes that well-designed and well-operated liquid injection and fluidized-bed incineration systems can meet the BDAT treatment standards established for organic constituents in F024.

#### 5.2 Accuracy Correction of Treatment Performance Data

As part of the review of treatment performance data for rotary kiln incineration, the data were adjusted to take into account any analytical interferences associated with the chemical makeup of the samples. Generally, performance data were corrected for accuracy as follows: (1) a matrix spike recovery was determined, as explained below, for each BDAT List constituent detected in the untreated or treated waste; (2) an accuracy correction factor was determined for each of the above constituents by dividing 100 by the matrix spike recovery (expressed as a percentage) for that constituent; and

(3) the reported concentration of each BDAT List constituent detected in the untreated or treated waste was corrected by multiplying the concentration by the corresponding accuracy correction factor.

Matrix spike recoveries are developed by analyzing a sample of a treated waste for a constituent and then re-analyzing the sample after the addition of a known amount of the same constituent (i.e., spike) to the sample. The matrix spike recovery represents the total amount of constituent recovered after spiking minus the initial concentration of the constituent in the sample, and the result divided by the spike concentration of the constituent.

#### 5.2.1 Nonwastewaters

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Matrix spike recoveries used to adjust of the treatment performance data for the kiln ash residue are presented in Table A-8 of Appendix A of this background document. Duplicate matrix spikes were performed for some BDAT List constituents in kiln ash. If duplicate matrix spikes were performed for a constituent, the matrix spike recovery used for that constituent was the lower of the two values from the first matrix spike and the duplicate spike.

Where a matrix spike was not performed for a constituent, the matrix spike recovery for that constituent was derived from the average matrix spike recoveries of the appropriate group of constituents (e.g., volatile organics) for which recovery data were available. In these cases, the matrix spike

recoveries for all volatile organics, semivolatile organics, or dioxins/ furans from the first matrix spike were averaged. Similarly, an average matrix spike recovery was calculated for the duplicate matrix spike recoveries. The lower of the two average matrix spike recoveries of the volatile, semivolatile, or dioxin/furan group was used for any volatile, semivolatile, or dioxin/furan constituent for which no matrix spike was performed.

Using this method, the accuracy correction factors calculated for nonwastewater (rotary kiln ash) data are presented in Table A-10 of Appendix A of this document. The concentrations of each BDAT List constituent detected in either the untreated F024 or the rotary kiln ash were corrected for accuracy and are presented in Table 5-1. Constituent concentrations in the kiln ash residue were not adjusted to values below their detection limit. If accuracy correction resulted in a value less than the detection limit for a constituent, the accuracy-corrected concentration was set equal to the detection limit.

#### Table 5-1

## BDAT LIST CONSTITUENT CONCENTRATIONS IN KILN ASH RESIDUE, CORRECTED FOR ACCURACY<sup>4</sup>

		(	Corrected	Total C	oncentra	tion in	
			24 Rotary				
				Sample			
BDAT	List Constituent	1	_2	3	<u>– 1</u>	5	<u>ć</u>
Volat	iles						
222.	Acetone	0.010	0.010	0.010	0.010	0.010	0.010
10.	2-Chloro-1,3-butadiere	0.100	0.100	0.100	0.100	0.100	0.100
16.	3-Chloropropene	0.100	0.100	0.100	0.100	0.100	0.100
20.	trans-1,4-Dichloro-2- butene	0.100	0.100	0.100	0.100	0.100	0.100
21.	Dichlorodifluoromethane	0.710	1.400	1.500	1.600	1.700	1.400
23.	1,2-Dichloroethane	0.005	0.005	0.005	0.005	0.005	0.005
26.	1,2-Dichloropropane	0.005	0.005	0.005	0.005	0.005	0.005
27.	trans-1,3-Dichloropropene	0.005	0.005	0.005	0.005	0.005	0.005
28.	cis-1,3-Dichloropropene	0.005	0.005	0.005	0.005	0.005	0.005
34.	Methyl ethyl ketone	0.010	0.010	0.010	0.010	0.010	0.010
42.	Tetrachloroethene	0.005	0.005	0.005	0.005	0.005	0.005
46.	1,1,2-Trichloroethane	0.005	0.005	0.005	0.005	0.005	0.005
49.	1,2,3-Trichloropropane	0.005	0.005	0.005	0.005	0.005	0.005
<u>Semiv</u>	olatiles						
59.	Benz(a)anthracene	0.632	0.632	0.666	0.632	0.632	0.632
62.	Benzo(a)pyrene	0.632	0.632	0.666	0.632	0.632	0.632
63.	Benzo(b)fluoranthene	0.632	0.632	0.666	0.632	0.632	0.632
64.	Benzo(ghi)perylene	0.632	0.632	0.666	0.632	0.632	0.632
65.	Benzo(k)fluoranthene	0.632	0.632	0.666	0.632	0.632	0.632
68.	Bis(2-chloroethyl)ether	0.632	0.632	0.666	0.632	0.632	0.632
70.	Bis(2-ethylhexyl)phthal- ate	0.632	0.632	0.666	0.632	0.632	0.632
80.	Chrysene	0.632	0.632	0.666	0.632	0.632	0.632
88.	1,4-Dichlorobenzene	0.462	0.462	0.488	0.462	0.462	0.462
92.	Diethyl phthalate	1.264	1.264	1.280	1.145	0.854	2.164
110.	Hexachlorobenzene	0.632	0.632	0.666	0.632	0.632	0.632
113.	Hexachloroethane	0.632	0.632	0.666	0.632	0.632	0.632
-		-	-				-

NA - Not analyzed.

<sup>a</sup>This table presents data for the BDAT List constituents detected in either the untreated F024 from plants A, B, C, and D or the rotary kiln ash. These data were obtained by multiplying the concentration found in the incinerator ash (Section 4.0) by the accuracy correction factor for that constituent (Appendix A).

## BDAT LIST CONSTITUENT CONCENTRATIONS IN KILN ASH RESIDUE, CORRECTED FOR ACCURACY<sup>a</sup>

				i Total Co <u>/ Kiln Inc</u> Sample	inerato		mg∕kg)
BDAT	<u>List Constituent</u>	1	_2	3	<u> </u>		<u> </u>
Semiv	olatiles (Cont.)						
116. 141.	Indeno(1,2,3-cd)pyrene Phenanthrene	0.632 0.632	0.632 0.632	0.666 0.666	0.632 0.632	0.632 0.632	0.632 0.632
Inorg	anics						
170.	Fluoride	NA	NA	2.11	NA	NA	NA
<u>Dioxi</u>	ns and furans						
207.	Hexachlorodibenzo-p- dioxins	NA	NA	0.0001	NA	NA	NA
208.	Hexachlorodibenzofurans	NA	NA	0.00004	NA	NA	NA
209.	Pentachlorodibenzo-p- dioxins	NA	NA	0.0001	NA	NA	NA
210.	Pentachlorodibenzofurans	NA	NA	0.00003	NA	NA	NA
212.	Tetrachlorodibenzofurans	NA	NA	0.00003	NA	NA	NA

NA - Not analyzed.

<sup>a</sup>This table presents data for the BDAT List constituents detected in either the untreated F024 from plants A, B, C, and D or the rotary kiln ash. These data were obtained by multiplying the concentration found in the incinerator ash (Section 4.0) by the accuracy correction factor for that constituent (Appendix A).

#### 5.2.2 <u>Wastewaters</u>

Matrix spike recoveries used to calculate accuracy correction factors for adjustment of the treatment performance data for the combustion gas scrubber water are presented in Table A-9 of Appendix A. If duplicate matrix spikes were performed for a constituent, the matrix spike recovery used for that constituent was the lower of the two values from the first matrix spike and the duplicate spike.

Where a matrix spike was not performed for a constituent, the matrix spike recovery for that constituent was derived from the average matrix spike recoveries of the appropriate group of constituents (e.g., volatile organics) for which recovery data were available. In these cases, the matrix spike recoveries for all volatile organics, semivolatile organics, or dioxins/furans from the first matrix spike were averaged. Similarly, an average matrix spike recovery was calculated for the duplicate matrix spike recoveries. The lower of the two average matrix spike recoveries of the volatile, semivolatile, or dioxin/furan group was used for any volatile, semivolatile, or dioxin/furan constituent for which no matrix spike was performed.

Using this method, the accuracy correction factors calculated for wastewater (scrubber water) data are presented in Table A-10 of Appendix A of this document. The concentrations of each BDAT List constituent detected in either the untreated F024 or the scrubber water were corrected for accuracy and are presented in Table 5-2. Constituent concentrations in the scrubber

## Table 5-2

## BDAT LIST CONSTITUENT CONCENTRATIONS IN SCRUBBER WATER, CORRECTED FOR ACCURACY<sup>a</sup>

				Total Con			
		<u>F024</u>	Lombust	ion Gas S Sample		water (	<u>mg/1)</u>
BDAT	List Constituent	1	2		4	5	6
Volat	iles						
222.	Acetone	0.010	0.010	0.010	0.010	0.010	0.010
6.	Bromomethane	0.010	0.010	0.010	0.010	0.010	0.032
10.	2-Chloro-1,3-butadiene	0.101	0.101	0.101	0.101	0.101	0.101
12.	Chloroethane	0.010	0.010	0.010	0.010	0.010	0.031
15.	Chloromethane	0.010	0.010	0.010	0.010	0.010	0.020
16.	3-Chloropropene	0.101	0.101	0.101	0.101	0.101	0.101
20.	trans-1,4-Dichloro-2-	0.101	0.101	0.101	0.101	0.101	0.101
	butene				_		
21.	Dichlorodifluoromethane	0.010	0.010	0.292	0.363	0.403	0.444
23.	1,2-Dichloroethane	0.005	0.005	0.005	0.005	0.005	0.005
26.	1,2-Dichloropropane	0.005	0.005	0.005	0.005	0.005	0.005
27.	trans-1,3-Dichloropropene	0.005	0.005	0.0 <b>05</b>	0.005	0.005	0.005
28.	cis-1,3-Dichloropropene	0.005	0.005	0.005	0.005	0.005	0.005
34.	Methyl ethyl ketone	0.010	0.010	0.010	0.010	0.010	0.010
42.	Tetrachloroethene	0.005	0.005	0.005	0.005	0.005	0.005
46.	1,1,2-Trichloroethane	0.005	0.005	0.005	0.005	0.005	0.005
49.	1,2,3-Trichloropropane	0.005	0.005	0.005	0.005	0.005	0.005
Semiv	volatiles						
59.	Benz(a)anthracene	0.014	0.014	0.013	0.012	0.013	0.013
62.	Benzo(a)pyrene	0.014	0.014	0.013	0.012	0.013	0.013
63.	Benzo(b)fluoranthene	0.014	0.014	0.013	0.012	0.013	0.013
64.	Benzo(ghi)perylene	0.014	0.014	0.013	0.012	0.013	0.013
65.	Benzo(k)fluoranthene	0.014	0.014	0.013	0.012	0.013	0.013
68.	Bis(2-chloroethyl)ether	0.014	0.014	0.013	0.012	0.013	0.013
70.	Bis(2-ethylhexyl)phthal-	0.014	0.014	0.013	0.012	0.013	0.013
101	ate	0.017	0.017	9.01	9.9)£	0.013	U.U.J
80.	Chrysene	0.014	0.014	0.013	0.012	0.013	0.013

NA - Not analyzed.

<sup>a</sup>This table presents data for the BDAT List constituents detected in either the untreated F024 from plants A, B, C, and D or the scrubber water. These data were obtained by multiplying the concentration found in the incinerator ash (Section 4.0) by the accuracy correction factor for that constituent (Appendix A).

## BDAT LIST CONSTITUENT CONCENTRATIONS IN SCRUBBER WATER, CORRECTED FOR ACCURACY<sup>a</sup>

				Total Cond tion Gas <u>So</u>			
				Sample			
BDAT	List Constituent	1	2	3	4	5	6
Semiv	volatiles (Continued)						
88. 92. 110. 113. 116. 141.	1,4-Dichlorobenzene Diethyl phthalate Hexachlorobenzene Hexachloroethane Indeno(1,2,3-cd)pyrene Phenanthrene	0.021 0.014 0.014 0.014 0.014 0.014	0.021 0.014 0.014 0.014 0.014 0.014	0.019 0.013 0.013 0.013 0.013 0.013	0.017 0.012 0.012 0.012 0.012 0.012	0.019 0.066 0.013 0.013 0.013 0.013	0.013 0.013 0.013
Inorg	anics						
170.	Fluoride	NA	NA	153	NA	NA	NA
Dioxi	ins and furans						
207.	Hexachlorodibenzo-p- dioxins	NA	NA	0.00001	NA	NA	NA
208.	Hexachlorodibenzofurans	NA	NA	0.0 <b>003</b>	NA	NA	NA
209.	Pentachlorodibenzo-p- dioxins	NA	NA	0.00001	NA	NA	NA
210.	Pentachlorodibenzofurans	NA	NA	0.0001	NA	NA	NA
212.	Tetrachlorodibenzofurans	NA	NA	0.00001	NA	NA	NA
				ed Total Co			

		in '	the Trea	ted K062	Wastewater	(mg/1)	_
			Sample :	Set			-
Metal	<u>s</u>	1	_2	3			
	Chromium (total) Nickel		0.176 0.355				

NA - Not analyzed.

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<sup>a</sup>This table presents data for the BDAT List constituents detected in either the untreated FO24 from plants A, B, C, and D or the scrubber water. These data were obtained by multiplying the concentration found in the incinerator ash (Section 4.0) by the accuracy correction factor for that constituent (Appendix A). water residual were not adjusted to values below the detection limit for that constituent. If accuracy correction resulted in a value less than the detection limit, the accuracy-corrected value was set equal to the detection limit.

#### 5.3 Statistical Comparison of Treatment Performance Data

In cases where the Agency has treatment performance data from more than one technology, EPA uses the statistical method known as the analysis of variance (ANOVA) test (discussed in EPA's <u>Methodology for Developing BDAT</u> <u>Treatment Standards</u> (Reference 1)), to determine if one technology performs significantly better than the rest. For FO24, the Agency has treatment performance data for only one treatment system and, therefore, an ANOVA comparison is not appropriate.

#### 5.4 Available Treatment Technologies

The demonstrated technologies for treatment of F024 (rotary kiln incineration, stabilization, and chemical precipitation followed by vacuum filtration) are considered to be commercially available. Furthermore, the Agency has determined that these technologies will provide substantial treatment of F024. Therefore, these technologies are available for treatment of F024.

Methods of total recycle or reuse are not considered to be commercially available for F024 since they are proprietary or patented processes and

cannot be purchased or licensed from the proprietor. In addition, a process used successfully by one facility may not work for another because of the wide variations in the waste. Therefore, total recycle or reuse cannot be further considered BDAT, as it is not an available treatment technology.

## 5.5 BDAT for F024

As discussed above, rotary kiln incineration followed by: (1) stabilization of the incinerator ash, and (2) chemical precipitation followed by vacuum filtration of the scrubber water have been determined to be demonstrated and available. Because the Agency does not have treatment performance data for any other technologies treating F024 or similar wastes, this treatment train is the best. Therefore, the best demonstrated available technology (BDAT) for F024 has been determined to be rotary kiln incineration followed by: (1) stabilization of the incinerator ash, and (2) chemical precipitation followed by vacuum filtration of the scrubber water.

## 6.0 SELECTION OF REGULATED CONSTITUENTS

The Agency has developed a list of hazardous constituents (the BDAT Constituent List, presented in EPA's <u>Methodology for Developing BDAT Treatment</u> <u>Standards</u> (Reference 1)) from which constituents to be regulated are selected. EPA may revise this list as additional data and information become available. The list is divided into the following categories: volatile organics, semivolatile organics, metals, inorganics other than metals, organochlorine pesticides, phenoxyacetic acid herbicides, organophosphorus insecticides, PCBs, and dioxins and furans. This section presents the rationale for the selection of constituents to be regulated in wastewater and nonwastewater forms of F024.

Generally, constituents selected for regulation must satisfy the following criteria:

- <u>The constituent must be on the BDAT List of regulated constituents</u>. Presence on the BDAT List implies the existence of approved methods for analyzing the constituent in treated waste matrices.
- 2) The constituent must be present in, or be suspected of being present in, the untreated waste. For example, in some cases, analytical difficulties (such as masking) may prevent a constituent from being identified in the untreated waste, but its identification in a treatment residual may lead the Agency to conclude that it is present in the untreated waste.
- 3) Where treatment performance data are transferred, the constituents selected for regulation must be easier to treat than the constituent(s) from which performance data are transferred. Waste characteristics affecting performance (WCAPs) of treatment vary according to the technology of concern. For instance, for incineration, the WCAPs include bond dissociation energy, thermal conductivity, and boiling point.

From a group of constituents that are eligible for regulation because they meet the above criteria, EPA may select a subset of constituents that represent the broader group. For example, from a group of constituents that react similarly to treatment, the Agency may select for regulation only those constituents that are the most difficult to treat, to facilitate implementation of the compliance and enforcement program.

The Agency initially considered all constituents on the HDAT List for regulation. Available F024 characterization data for all BDAT List constituents are summarized in Table 6-1. (All tables are presented at the end of Section 6.0.) A range of detected concentrations is shown in the table for all constituents known to be present in the untreated F024. Constituents that were not detected in the untreated waste but were detected in the treated waste are identified by the symbol "\*." Constituents for which the Agency does not have analytical characterization data are identified by the notation "NA" (not analyzed).

The Agency is not regulating all of the BDAT List constituents considered for regulation. A BDAT List constituent was deleted from further consideration for regulation if (1) the constituent was not detected in the untreated and/or treated wastes, (2) the constituent was not analyzed for in the untreated waste, or (3) other reasons, as discussed in Section 6.1. Constituents that were selected for regulation are discussed in Section 6.2.

### 6.1 BDAT List Constituents Deleted from Consideration for Regulation

A BDAT List constituent that was detected in untreated F024 was deleted from consideration for regulation if (1) available treatment performance data for the constituent did not show effective treatment by BDAT, (2) the constituent was not present at treatable concentrations in the waste, or (3) other reasons, as described below. BDAT List constituents that remained following the deletions described in this subsection were further considered for regulation. These constituents are listed in Table 6-2 for nonwastewaters and in Table 6-3 for wastewaters. All tables are included at the end of this section.

Sulfide was not further considered for regulation in F024 wastewaters and nonwastewaters because the technology determined to be BDAT for F024 (rotary kiln incineration followed by stabilization of nonwastewater and chemical precipitation followed by vacuum filtration of wastewater) does not provide effective treatment for this BDAT List constituent. Moreover, the Agency is unaware of any demonstrated technology for treatment of sulfide in F024 or similar wastes.

Barium was deleted from further consideration for regulation in F024 wastewaters because the transferred BDAT treatment performance data obtained from chemical precipitation followed by vacuum filtration of K062 do not show effective treatment.

Arsenic was deleted from further consideration for regulation in nonwastewaters because it was not present in the untreated waste at treatable concentrations.

Similarly, benz(a)anthracene, benzo(a)pyrene, benzo(b)fluoranthene, benzo(ghi)perylene, benzo(k)fluoranthene, chrysene, and indeno(1,2,3-cd)pyrene were deleted from further consideration for regulation in F024 because they were not present in the untreated waste at treatable concentrations.

Copper, vanadium, and zinc were considered for regulation in F024 wastewaters but were not selected as regulated constituents. Although the metal compounds, copper cyanide, vanadium pentoxide, and zinc cyanide are listed in Appendix VIII of 40 CFR Part 261, the individual metals, copper, vanadium, and zinc, are not listed. In this Second-Thirds rulemaking, the Agency is only regulating copper, vanadium, and/or zinc in listed hazardous wastes when they cannot be controlled by regulation of other metal constituents. For F024, copper, vanadium, and zinc are believed to be controlled by the regulation of total chromium and nickel and are therefore not being regulated.

Dichlorodifluoromethane, diethyl phthalate, antimony, and beryllium were not detected in untreated F024 but were detected in the rotary kiln ash

residual. These constituents were not selected for regulation in F024 nonwastewaters. Dichlorodifluoromethane and diethyl phthalate are not being regulated because they were detected in the laboratory blank at 0.27 mg/kg and 0.51 mg/kg, respectively, and are believed to be laboratory contaminants. The Agency believes antimony and beryllium concentrations detected in the kiln ash are attributable to their presence in the sludge and organic liquid wastes incinerated with F024. These constituents were not typically found in untreated F024 (Table 2-3). Therefore, these constituents were not selected for regulation in F024 nonwastewaters.

Bromomethane, chloroethane, chloromethane, dichlorodifluoromethane, diethyl phthalate, antimony, and selenium were not detected in untreated F024 but were detected in the scrubber water residual from rotary kiln incineration. These constituents were not selected for regulation in F024 wastewaters. Dichlorodifluoromethane is not being regulated because it was detected in the laboratory blank at 0.23 mg/l and is believed to be a laboratory contaminant. Diethyl phthalate is not being regulated because it was detected in only one scrubber water sample at a low concentration (0.057 mg/l). In addition, diethyl phthalate, (BDE 3,145 kcal/mole) is controlled by regulation of bis (2-ethylhexyl) phthalate (BDE 6,465 kcal/mole). The Agency believes the concentrations of the five remaining constituents detected in the scrubber water (bromomethane, chloroethane, chloromethane, antimony, and selenium) are attributable to their presence in the sludge and organic liquid wastes incinerated with F024. These constituents were not typically found in F024 (Table

2-3). Therefore, these constituents were not selected for regulation in F024 wastewaters.

All other metals were deleted from further consideration for regulation in F024 nonwastewaters because the Agency has only recently completed an analysis of TCLP extracts obtained from the stabilization of F024 incinerator ash residues.

## 6.2 Selection of BDAT List Constituents for Regulation

The selection of constituents for regulation in nonwastewaters and wastewaters is discussed in Sections 6.2.1 and 6.2.2, respectively.

#### 6.2.1 Nonwastewaters

Constituents regulated in F024 nonwastewaters were selected from BDAT List constituents detected in the untreated waste, unless they were deleted from consideration as discussed in Section 6.1.

Table 6-4 presents each constituent selected for regulation in F024 nonwastewaters after consideration of (1) the constituent concentration in the untreated waste, (2) whether the constituent is adequately controlled by regulation of another constituent, and (3) the relative difficulty in achieving effective treatment of the constituent by BDAT for nonwastewaters (rotary kiln incineration followed by stabilization).

The Agency's determination of adequate control for organic constituents was based on an evaluation of the characteristics of the constituents that would affect the performance of rotary kiln incineration relative to the kiln ash residual, specifically, the boiling points of the constituents. In general, a constituent is believed to be controlled by regulation of another constituent that has a higher boiling point. Boiling points for the BDAT List organic constituents further considered for regulation, as shown in Table 6-2, are listed in Appendix C.

Based on the above discussion, the nine BDAT List organic constituents selected for regulation in F024 nonwastewaters are 2-chloro-1,3-butadiene; 3-chloropropene; 1,1-dichloroethane; 1,2-dichloroethane; 1,2-dichloropropane; trans-1,3-dichloropropene; cis-1,3-dichloropropene; bis(2-ethylhexyl)phthalate; and hexachloroethane.

The remaining 36 BDAT List organic constituents that were further considered for regulation in F024 nonwastewaters due to their presence in the untreated waste (Table 6-2) are not being regulated. These 36 constituents were found at lower concentrations in the untreated waste than the organic constituents selected for regulation. Deletion of these constituents from consideration for regulation in F024 nonwastewaters was also based on a comparison of their boiling points with the boiling points of the constituents selected for regulation. These 36 constituents were then deleted from consideration for regulation because they are believed to be adequately controlled by incineration of other organic constituents that have been selected for

regulation in F024 nonwastewaters. The control of constituents deleted from consideration for regulation by constituents selected for regulation in F024 nonwastewaters is presented in Table 6-5.

Cyanide was detected in untreated F024 but was not selected for regulation in F024 nonwastewaters because it was found at low concentrations in the untreated waste and is believed to be adequately controlled by treatment standards for the BDAT List organic constituents.

Two metals, total chromium and nickel, are being reserved for regulation in F024 nonwastewaters because the Agency has recently completed an analysis of *ICLP* extracts obtained from the stabilization of F024 incinerator ash residues. The results of this analysis show substantial reduction of metals in TCLP extracts following stabilization. Therefore, the Agency has decided to reserve the final treatment standards for metals in F024 nonwastewaters in the Second Thirds promulgated rule. The Agency will instead propose revised treatment standards for metals in F024 non the F024 stabilization results as part of the Third Thirds proposed rule so that sufficient time is provided for notice and comment on these revisions.

Five dioxins and furans, hexachlorodibenzo-p-dioxins, hexachlorodibenzofurans, pentachlorodibenzo-p-dioxins, pentachlorodibenzofurans, and tetrachlorodibenzofurans, were selected for regulation in F024 nonwastewaters based on the difficulty in treating these constituents, reflected by their high boiling points, as well as their inherent toxicity.

#### 6.2.2 Wastewaters

Constituents regulated in F024 wastewaters were selected using the same constituent selection method used for F024 nonwastewaters; that is, they were selected from the BDAT List constituents that were detected in the untreated waste, unless they were deleted from consideration as discussed in Section 6.1.

Table 6-6 presents each constituent selected for regulation in F024 wastewaters after consideration of (1) the constituent concentration in the untreated waste, (2) whether the constituent is adequately controlled by regulation of another constituent, and (3) the relative difficulty associated with achieving effective treatment of the constituent by the BDAT for wastewaters (rotary kiln incineration followed by chemical precipitation and vacuum filtration).

The Agency's determination of adequate control for organic constituents was based on an evaluation of the characteristics of the constituents that would affect performance of incineration relative to the scrubber water residual, specifically, their estimated bond dissociation energies. In general, a constituent is believed to be controlled by regulation of another constituent that has a higher bond dissociation energy. Estimated bond dissociation energies for the BDAT List organic constituents further considered for regulation, as shown in Table 6-3, are listed in Appendix C.

Eased on the above discussion, the nine BDAT List organic constituents selected for regulation in F024 wastewaters are 2-chloro-1,3-butadiene; 3-chloropropene; 1,1-dichloroethane; 1,2-dichloroethane; 1,2-dichloropropane; trans-1,3-dichloropropene; cis-1,3-dichloropropene; bis(2-ethylhexyl)phthalate; and hexachloroethane.

The remaining 39 BDAT List organic constituents that were further considered for regulation in F024 wastewaters due to their presence in the untreated waste (Table 6-3) are not being regulated. These 39 constituents were found at lower concentrations in the untreated waste than the organic constituents selected for regulation. Deletion of these constituents from consideration for regulation in F024 wastewaters was also based on a comparison of their bond dissociation energies (BDEs) with the BDEs of the constituents selected for regulation. These 39 constituents were then deleted from consideration for regulation because they are believed to be adequately controlled by incineration of other organic constituents that have been selected for regulation in F024 wastewaters. The control of constituents deleted from consideration for regulation by constituents selected for regulation in F024 wastewaters is presented in Table 6-7.

Cyamide was detected in untreated F024 but was not selected for regulation F024 wastewaters because it was found at low concentrations in the untreated waste and is believed to be adequately controlled by treatment standards for the BDAT List organic constituents.

Two metals, total chromium and nickel, were selected for regulation in F024 wastewaters. All other BDAT List metal constituents initially considered for regulation in F024 wastewaters were not selected because these constituents were found at low concentrations in the untreated waste and are believed to be adequately controlled by the treatment standards for total chromium and nickel. Although high lead levels were found in F024 scrubber water, these levels are attributed to high lead levels found in the background scrubber water sample. Control of metal constituents is provided by the use of chemical precipitation followed by vacuum filtration. By removing the metals present at the highest concentrations in the untreated waste, adequate treatment will be provided for other metals present at lower treatable concentrations.

Five dioxins and furans, hexachlorodibenzo-p-dioxins, hexachlorodibenzofurans, pentachlorodibenzo-p-dioxins, pentachlorodibenzofurans, and tetrachlorodibenzofurans, were selected for regulation in F024 wastewaters based on the detection of these BDAT List constituents in the scrubber water residual, as well as their inherent toxicity.

## Table 6-1

#### STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

Li	AT st ituent	Detection Status	Concentration in Untreated F024 (mg/kg)
Volat	liles		
222.	Acetone	x	<0.05-21,000
1.	Acetonitrile		<0.5
2.	Acrolein		<0.5
3. 4.	Acrylonitrile Benzene	X	0.5 <0.025-1,900
5.	Bromodichloromethane	x	<0.025-7,260
6.	Bromomethane		<0.05*
223.	n-Butyl alcohol		NA
7.	Carbon tetrachloride	Х	<0.025-50,400
8.	Carbon disulfide		<0.025
9.	Chlorobenzene	x	<0.025-3,200
10.	2-Chloro-1,3-butadiene	X	<0.5-139,721
11.	Chlorodibromomethane		<0.025
12.	Chloroethane		<0.05*
13.	2-Chloroethyl vinyl ether		NA
14.	Chloroform	X	0.025-1,000
15.	Chloromethane		0.05*
16.	3-Chloropropene	X	<0.5-285,486
17.	1,2-Dibromo-3-chloropropane		<0.05
18.	1,2-Dibromoethane		<0.025
19.	Dibromomethane		<0.025
20.	trans-1,4-Dichloro-2-butene	X	<0.5-4,691
21.	Dichlorodifluoromethane		<0.05*
22 <i>.</i>	1,1-Dichloroethane	X	<0.025-440,000
23.	1,2-Dichloroethane	X	<0.025-950,000
24.	1,1-Dichloroethylene		<0.025
25.	trans-1,2-Dichloroethene		NA
26.	1,2-Dichloropropane	X	<0.025-230,000
27.	trans-1,3-Dichloropropene	X	<0.025-290,000
28.	cis-1,3-Dichloropropene	X	<0.025-160,000

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

\* - Not detected in the untreated waste, but detected in the treatment residual.

# STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDAT List	Detection	Concentration in Untreated F024
Constituent	Status	(mg/kg)
Volatiles (Continued)		
29. 1,4-Dioxane		< 1
224. 2-Ethoxyethanol		NA
225. Ethyl acetate		NA
226. Ethyl benzene	Х	<0.025-230
30. Ethyl cyanide		<20
227. Ethyl ether		NA
31. Ethyl methacrylate		<0.5
214. Ethylene oxide		<2
32. Iodomethane		<0.25
33. Isobutyl alcohol		< 1
228. Methanol		NA
34. Methyl ethyl ketone	X	<0.05-2,200
229. Methyl isobutyl ketone		<0.05
35. Methyl methacrylate		<0.5
37. Methacrylonitrile		<0.5
38. Methylene chloride	х	<1-1,900
230. 2-Nitropropane		NA
39. Pyridine		<2
40. 1,1,1,2-Tetrachloroethane	X	<0.02 <b>5-58,</b> 000
1. 1,1,2,2-Tetrachloroethane	X	<0.025-16,000
12. Tetrachloroethene	x	< 1-47,200
43. Toluene	· X	<0.025-34,000
44. Tribromomethane		<0.025
45. 1,1,1-Trichloroethane	Х	<0.025-620
46. 1,1,2-Trichloroethane	X	<0.025-92,000
47. Trichloroethene	X	<0.025-81,800
48. Trichloromonofluoromethane		<1
49. 1,2,3-Trichloropropane	X	<0.025-9,712
231. 1,1,2-Trichloro-1,2,2-trifluoroethane		NA
50. Vinyl chloride	X	<0.05-1,000

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

• - Not detected in the untreated waste, but detected in the treatment residual.

#### STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDA Lis Consti	st	Detection Status	Concentration in Untreated F024 (mg/kg)
<u>Volati</u>	les (Continued)		
215. 216. 217.	1,2-Xylene 1,3-Xylene 1,4-Xylene		<0.025 <0.025 <0.025
Semivo	latiles		
	Acenaphthalene Acenaphthene Acetophenone 2-Acetylaminofluorene 4-Aminobiphenyl		<0.351 <0.351 <1.76 <0.702 <0.702
57. 58. 59.	Aniline Anthracene Aramite Benz(a)anthracene Benzal chloride	X	<0.351 <172 <1.76 <24-0.888 NA
63.	Benzenethiol Benzo(a)pyrene Benzo(b)fluoranthene Benzo(ghi)perylene Benzo(k)fluoranthene	X X X X	<0.702 <0.351-0.60 <24-0.716 <0.351-0.421 <24-0.874
67. 68. 69.	p-Benzoquinone Bis(2-chloroethoxy)methane Bis(2-chloroethyl)ether Bis(2-chloroisopropyl)ether Bis(2-ethylhexyl)phthalate	X X	<0.351 <0.351 <0.351-9,800 <0.351 <24-480
72. 73. 74.	4-Bromophenyl phenyl ether Butyl benzyl phthalate 2-sec-Butyl-4,6-dinitrophenol p-Chloroaniline Chlorobenzilate		<0.351 <0.351 <1.76 <0.351 <0.702

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

\* - Not detected in the untreated waste, but detected in the treatment residual.

## STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDAT		Concentration
List	Detection	in Untreated F024
Constituent	Status	(mg/kg)
emivolatiles (Continued)		
76. p-Chloro-m-cresol		<0.351
7. 2-Chloronaphthalene	X	<0.351-260
8. 2-Chlorophenol		< 0.351
9. 3-Chloropropionitrile		<0.702
0. Chrysene	X	<24-1.06
1. ortho-Cresol		<0.351
2. para-Cresol		<0.351
32. Cyclohexanone		NA
3. Dibenz(a,h)anthracene		<0.351
4. Dibenzo(a,e)pyrene		NA
5. Dibenzo(a,i)pyrene		NA
6. m-Dichlorobenzene	Х	<0.351-1,300
7. o-Dichlorobenzene	X	<0.351-24,000
<ol><li>p-Dichlorobenzene</li></ol>	Х	<0.351-8,000
9. 3,3'-Dichlorobenzidine		<1.76
0. 2,4-Dichlorophenol		<0.351
1. 2,6-Dichlorophenol		<0.702
2. Diethyl phthalate	X	<0.351-120
3. 3,3'-Dimethoxybenzidine		<0.702
4. p-Dimethylaminoazobenzene		<0.702
5. 3,3'-Dimethylbenzidine		<0.702
<ol> <li>2,4-Dimethylphenol</li> </ol>		<0.351
7. Dimethyl phthalate		<0.351
8. Di-n-butyl phthalate		<0.351
9. 1,4-Dinitrobenzene		<0.351
00. 4,6-Dinitro-o-cresol		<1.76
01. 2,4-Dinitrophenol		<1.76
02. 2,4-Dinitrotoluene		<0.351
03. 2,6-Dinitrotoluene		<0.351
04. Di-n-octyl phthalate	Х	<0.351-34

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

\* - Not detected in the untreated waste, but detected in the treatment residual.

## STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDAT	Detection	Concentration
List Constituent	DetectionStatus	in Untreated F024 (mg/kg)
		(mg/ kg/
Semivolatiles (Continued)		
105. Di-n-propylnitrosamine		<0.351
106. Diphenylamine		<0.702
219. Diphenylnitrosamine		NA
107. 1,2-Diphenylhydrazine		<1.76
108. Fluoranthene		<0.35 t
109. Fluorene		<0.351
10. Hexachlorobenzene	Х	<24-18,018
11. Hexachlorobutadiene	X	<0.351-16,470
112. Hexachlorocyclopentadi <del>ene</del>	X	<0.351-1.3
113. Hexachloroethane	X	<0.351-460,000
114. Hexachlorophene		NA
15. Hexachloropropene		<0.702
116. Indeno(1,2,3-cd)pyrene	X	<0.351-0.411
17. Isosafrole		<0.702
18. Methapyrilene		<0.702
119. 3-Methylcholanthrene		<0.702
120. 4,4'-Methylenebis(2-chloroani	line)	<0.702
6. Methyl methanesulfonate		NA
21. Naphthalene	X	<24-330
22. 1,4-Naphthoquinone		<0.702
23. 1-Naphthylamine		<1.76
24. 2-Naphthylamine		<1.76
25. p-Nitroaniline		<1.76
26. Nitrobenzene	X	<0.351-1.4
27. 4-Nitrophenol		<1.76
28. n-Nitrosodi-n-butylamine		<0.702
29. n-Nitrosodiethylamine		<0.351
30. n-Nitrosodimethylamine		<0.351
31. n-Nitrosomethylethylamine		<0.351
32. n-Nitrosomorpholine		<0.702

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

 Not detected in the untreated waste, but detected in the treatment residual.

## STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

Ĺ.	DAT ist tituent	Detection Status	Concentration in Untreated F024 (mg/kg)	
Semivolatiles (Continued)				
133.	n-Nitrosopiperidine		<0.351	
134.			<1.76	
35.			< 1.76	
136.	Pentachlorobenzene	Х	<1.76-1,290	
37.	Pentachloroethane	X	<0.351-26,000	
38.	Pentachloronitrobenzene		<3.51	
139.			<1.76	
140.			<0.702	
141.	Phenanthrene	Х	<24-1.27	
142.	Phenol		<0.351	
220.	Phthalic anhydride		<0.351	
143.	2-Picoline		NA	
144.	Pronamide		<0.351	
145.	Pyrene		<0.702	
146.	Resorcinol		<0.351	
147.			<0.351	
148.	1,2,4,5-Tetrachlorobenzene		<1.76	
149.	2,3,4,6-Tetrachlorophenol		<0.702	
150.	1,2,4-Trichlorobenzene	X	<0.351-1,400	
151.	2,4,5-Trichlorophenol		<1.76	
152.	2,4,6-Trichlorophenol		<0.351	
153.	Tris(2,3-dibromopropyl)phosphate		<1.76	
Meta.	13			
154.	Antimony	X	<1.8-2.2	
155.	Arsenic	X	<0.86-7.8	
156.	Barium -	X	0.22-34	
157.	5		<0.1 <del>*</del>	
158.	Cadmium	X	<0.26-3.1	

limit in one or more untreated F024 samples.
 - Not detected in the untreated waste, but detected in the treatment
 residual.

NA - Not analyzed.

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#### STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDAT List Constituent	Detection Status	Concentration in Untreated F024 (mg/kg)
Metals (Continued)		
159. Chromium (total) 221. Chromium (hexavalent)	x	<0.4-285 <10
160. Copper	Х	<0.4-800
161. Lead	X	<0.43-9.0
162. Mercury	X	<0.1-0.24
163. Nickel	X	<0.9-636
164. Selenium		<0.5 <b>*</b>
165. Silver		<0.4
166. Thallium	v	<10
167. Vanadium 168. Zinc	X X	<0.17-10 0.73-443
108. Zine	~	0.75-445
Inorganics		
169. Cyanide	x	<0.43-4.57
170. Fluoride	X	<0.99-10.5
171. Sulfide	X	<4.6-349
Organochlorine pesticides		NA
Phenoxyacetic acid herbicides		NA
Organophosphorous insecticides		NA
PCBs		
200. Aroclor 1016		< 10
201. Araclor 1221		<10
202. Aroclor 1232		<10
203. Aroclor 1242		< 10
204. Aroclor 1248		< 10
205. Aroclor 1254		< 10
206. Aroclor 1260		<10

NOTE: The BDAT List pesticides were not expected to be seen in the F024 samples or treatment residuals and were therefore not analyzed for.

X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated F024 samples.

\* - Not detected in the untreated waste, but detected in the treatment residual.

# Table 6-1 (Continued)

# STATUS OF BDAT LIST CONSTITUENTS IN UNTREATED F024

BDAT List Constitu <del>ent</del>	Detection Status	Concentration in Untreated F024 (mg/kg)
Dioxins and furans		(ug/kg)
207. Hexachlorodibenzo-p-dioxins	х	<0.0005-10
208. Hexachlorodibenzofurans	Х	<0.0007-50
209. Pentachlorodibenzo-p-dioxins	Х	<0.0005-2
210. Pentachlorodibenzofurans	X	<0.0005-30
211. Tetrachlorodibenzo-p-dioxins		<0.03
212. Tetrachlorodibenzofurans	Х	<0.0002-10
213. 2,3,7,8-Tetrachlorodibenzo-p-dioxin		<0.03

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X - Indicates that a constituent was quantified at or above its detection limit in one or more untreated FO24 samples.

# Table 6-2

BDAT LIST CONSTITUENTS FURTHER CONSIDERED FOR REGULATION<sup>a</sup> IN F024 NONWASTEWATERS AFTER INITIAL SCREENING

> 222. Acetone 4. Benzene 5. Bromodichloromethane 7. Carbon tetrachloride 9. Chlorobenzene 10. 2-Chloro-1,3-butadiene 14. Chloroform 16. 3-Chloropropene 20. trans-1.4-Dichloro-2-butene 21. Dichlorodifluoromethane 22. 1,1-Dichloroethane 23. 1,2-Dichloroethane 26. 1,2-Dichloropropane 27. trans-1,3-Dichloropropene 28. cis-1,3-Dichloropropene 226. Ethyl benzene 34. Methyl ethyl ketone 38. Methylene chloride 40. 1,1,1,2-Tetrachloroethane 41. 1,1,2,2-Tetrachloroethane 42. Tetrachloroethene 43. Toluene 45. 1.1.1-Trichloroethane 1,1,2-Trichloroethane 46. 47. Trichloroethene 49. 1,2,3-Trichloropropane 50. Vinyl chloride 68. Bis(2-chloroethyl)ether 70. Bis(2-ethylhexyl)phthalate 77. 2-Chloronaphthalene 86. 1,3-Dichlorobenzene 87. 1,2-Dichlorobenzene 88. 1,4-Dichlorobenzene 92. Diethyl phthalate

<sup>a</sup>All constituents on this list were detected in F024 and were either selected for regulation (as shown in Table 6-4) or are believed to be controlled by regulation of another consitituent.

# Table 6-2 (Continued)

## BDAT LIST CONSTITUENTS FURTHER CONSIDERED FOR REGULATION<sup>a</sup> IN FO24 NONWASTEWATERS AFTER INITIAL SCREENING

- 104. Di-n-octyl phthalate
- 110. Hexachlorobenzene
- 111. Hexachlorobutadiene
- 112. Hexachlorocyclopentadiene
- 113. Hexachloroethane
- 121. Naphthalene
- 126. Nitrobenzene
- 136. Pentachlorobenzene 137. Pentachloroethane
- 141. Phenanthrene
- 150. 1,2,4-Trichlorobenzene
- 169. Cyanide
- 207. Hexachlorodibenzo-p-dioxins
- 208. Hexachlorodibenzofurans
- 209. Pentachlorodibenzo-p-dioxins
- 210. Pentachlorodibenzofurans 212. Tetrachlorodibenzofurans

<sup>a</sup>All constituents on this list were detected in FO24 and were either selected for regulation (as shown in Table 6-4) or are believed to be controlled by regulation of another consitituent.

# Table 6-3

#### BDAT LIST CONSTITUENTS FURTHER CONSIDERED FOR REGULATION<sup>a</sup> IN FO24 WASTEWATERS AFTER INITIAL SCREENING

- 222. Acetone
- 4. Benzene
- 5. Bromodichloromethane
- 6. Bromomethane
- 7. Carbon tetrachloride
- 9. Chlorobenzene
- 10. 2-Chloro-1,3-butadiene
- 12. Chloroethane
- 14. Chloroform
- 15. Chloromethane
- 16. 3-Chloropropene
- 20. trans-1,4-Dichloro-2-butene
- 21. Dichlorodifluoromethane
- 22. 1,1-Dichloroethane
- 23. 1,2-Dichloroethane
- 26. 1,2-Dichloropropane
- 27. trans-1,3-Dichloropropene
- 28. cis-1,3-Dichloropropene
- 226. Ethyl benzene
- 34. Methyl ethyl ketone
- 38. Methylene chloride
- 40. 1,1,1,2-Tetrachloroethane
- 41. 1,1,2,2-Tetrachloroethane
- 42. Tetrachloroethene
- 43. Toluene
- 45. 1,1,1-Trichloroethane
- 46. 1.1.2-Trichloroethane
- 47. Trichloroethene
- 49. 1,2,3-Trichloropropane
- 50. Vinyl chloride
- 68. Bis(2-chloroethyl)ether
- 70. Bis(2-ethylhexyl)phthalate
- 77. 2-Chloronaphthalene
- <sup>a</sup>All constituents on this list were detected in F024 and were either selected for regulation (as shown in Table 6-5) or are believed to be controlled by regulation of another constituent.

# Table 6-3 (Continued)

#### BDAT LIST CONSTITUENTS FURTHER CONSIDERED FOR REGULATION<sup>a</sup> IN F024 WASTEWATERS AFTER INITIAL SCREENING

86. 1,3-Dichlorobenzene 87. 1.2-Dichlorobenzene 88. 1,4-Dichlorobenzene 92. Diethyl phthalate 104. Di-n-octyl phthalate 110. Hexachlorobenzene 111. Hexachlorobutadiene 112. Hexachlorocyclopentadiene 113. Hexachloroethane 121. Naphthalene 126. Nitrobenzene 136. Pentachlorobenzene 137. Pentachloroethane 141. Phenanthrene 150. 1,2,4-Trichlorobenzene 154. Antimony 155. Arsenic 158. Cadmium 159. Chromium (total) 161. Lead 162. Mercury 163. Nickel 164. Selenium 169. Cyanide 207. Hexachlorodibenzo-p-dioxins 208. Hexachlorodibenzofurans 209. Pentachlorodibenzo-p-dioxins 210. Pentachlorodibenzofurans 212. Tetrachlorodibenzofurans

<sup>a</sup>All constituents on this list were detected in F024 and were either selected for regulation (as shown in Table 6-5) or are believed to be controlled by regulation of another constituent.

# Table 6-4

## BDAT LIST CONSTITUENTS SELECTED FOR REGULATION IN FO24 NONWASTEWATERS

- 10. 2-Chloro-1,3-butadiene
- 3-Chloropropene 16.
- 1,1-Dichloroethane 22.
- 23. 1,2-Dichloroethane
- 26. 1,2-Dichloropropane
- 27. trans-1,3-Dichloropropene
- 28. cis-1,3-Dichloropropene
- 70. Bis(2-ethylhexyl)phthalate
- 113. Hexachloroethane

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- 207. Hexachlorodibenzo-p-dioxins
- 208. Hexachlorodibenzofurans 209. Pentachlorodibenzofurans 210. Pentachlorodibenzofurans
- 212. Tetrachlorodibenzofurans

# Table 6-5

## CONTROL OF CONSTITUENTS DELETED FROM FURTHER CONSIDERATION FOR REGULATION BY CONSTITUENTS SELECTED FOR REGULATION IN FO24 NONWASTEWATERS

	BDAT List Constituent Deleted from Further Consideration for			BOAT	List Co	nstitu	not Sa	lected	for R		100 (B	ilioa	Point	,	
	Regulation (Boiling Point)	<u>A</u>	8	<u>_C</u>	<u>_P_</u>	<u>_£</u> _	F	G	H			ĸ	<u>L</u>	<u> </u>	N
50.	Vinyl chloride (~13.37°C)	x	x	×	×	x	x	×	×	x	×	x	ж	×	×
38.	Methylene chloride (39.75°C)	X	X	×	×	×	×	X	x	x	×	x	X	×	×
222.	Acatone (56.5°C)		X	x	×	×	x	X	x	x	x	×	X	X	X
14.	Chlaraform (61-62 <sup>0</sup> C)				х	×	X	×	X	X	X	×	х	×	×
45.	1,1,1-Trichloroethane (74-74.1°C)				X	×	x	×	x	X	×	x	X	X	×
7.	Carbon tetrachloride (76.7-77°C)				×	×	×	X	x	×	x	x	x	X	X
34.	Methyl ethyl ketone (79.6°C)				×	×	X	ж.	x	x	×	×	×	X	X
4.	Benzene (80 <sup>0</sup> C)				x	×	x	×	x	X	x	×	ĸ	×	x
47.	Trichloroethene (86.7-87 <sup>0</sup> C)					х	x	X,	×	X	×	×	×	X	×
5.	Bromodichioromethane (90°C)					×	x	X	×	×	×	×	×	×	×
43.	Taluana (110.6°C)							×	X	X	x	×	x	х	×
46.	1,1,2-Trichioroethane (113-114°C)								X	×	X	×	X	x	X
42.	Tetrachloroethene (121°C)								×	x	X	×	×	X	×
9.	Chlorobenzene (131-132°C)								X	×	×	×	×	X	X
226.	Ethyl benzene (136.3°C)								X	×	x	×	×	×	×
40.	1,1,1,2-Tetrachloroethane (146.5°C)								X	×	×	×	x	×	x
41.	1,1,2,2-Tetrachloroethane (146.5- 147 <sup>0</sup> C)								x	×	x	×	×	×	×
20.	trans-1,4-Dichloro-2-butene (155.5°C)								X	×	X	×	×	X	x
49.	1,2,3-Trichloropropane (156.8 <sup>0</sup> C)								x	X	x	×	×	×	×
137.	Pentachloroethane (161-162°C)								X	×	X	×	X	×	ĸ
86.	1,3-Dichlorobenzene (173 <sup>0</sup> C)								×	×	×	×	х	×	x
88.	1.4-Dichlorobenzene (174-174.1°C)								×	×	X	х	×	×	×
68.	Bis(2-chloroethyl)ether (178°C)								X	×	×	X	×	X	x
87.	1,2-Dichlorobenzene (180.5-181 <sup>0</sup> C)								×	X	×	x	×	×	×
KE	1:														

H - 113. Hexachlorpethane (186.8 187<sup>0</sup>C) A - 16. 3-Chloropropene (44-45°C) I - 70. Bis(2-ethyinesyi)phthalate (385°C) 8 - 22. 1,1-Dichloroethane  $(57-57.3^{\circ}C)$ C - 10. 2-Chloro-1,3-butadiene  $(59.4^{\circ}C)$ J - 207 Hemachlorodibenzo-p-dioxins (400:500°C) D - 23. 1,2-Dichloroethane (83-84°C) K - 208 Hexachigrodibenzofurans (400-500°C) L = 209. Pentachlorodibenzo-p-dioxins (400-500°C) E - 26. 1,2-Dichloropropane (96.4°C) N - 210. Pentachjørodibenzofurans (400-500<sup>0</sup>C) F - 28. cis-1,3-Dichtoropropene (108°C)

G - 27, trans-1,3-Dichloropropene (112<sup>0</sup>C)

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N - 212. Tetrachlorodibenzufurans (400-500°C)

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X - Indicates EPA's belief that the constituent deleted from further consideration for regulation will be adequately controlled by regulation of the indicated constituent selected for regulation.

# Table 6-5 (Continued)

# CONTROL OF CONSTITUENTS DELETED FROM FURTHER CONSIDERATION FOR REGULATION BY CONSTITUENTS SELECTED FOR REGULATION IN FO24 NONWASTEWATERS

	BDAT List Constituent Deleted from Further Consideration for		BD	AT LIS	Consti	tuent_	Select	ed to:	Regul	ation (	Boilin	g Por	<u>nt)</u>		
	Regulation (Boiling Point)	A	B	<u> </u>	-P_	<u> </u>	<u> </u>	<u> </u>	H	1	1	<u>*</u>	L	M	N
126.	Nitrobunzene (210-211 <sup>D</sup> C)									×	x	×	x	x	×
111.	Herachlorobutadiene (210-220°C)									x	X	×	x	X	x
150.	1,2,4-Trichlorobenzene (213°C)									x	X	×	x	x	×
121.	Naphthalene (217,9-218°C)									X	x	x	х	х	х
112.	Hexachlorocyclopentadiene (234°C)									X	X	x	х	×	x
77.	2-Chloronaphthalene (256°C)									X	×	X	х	X	×
136.	Pentachlorobenzene (275-277°C)									X	ĸ	x	x	X	x
92.	Disthyl phthalate (298°C)									X	X	x	x	x	x
110.	Hexachlorobenzene (323-326°C)									×	X	×	x	×	X
141.	Phenanthrene (340°C)									×	x	х	ж	×	×
104.	Dj-n-octyl phthalate (385°C)									X	×	×	x	×	×

KEY:

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8 - 22. C - 10. D - 23. E - 26. F - 28.	), 3-Chloropropene (44-45 <sup>o</sup> C) 1,1-Dichloroethene (67-57.3 <sup>o</sup> C) 2-Chloro-1,3-butadiene (59.4 <sup>o</sup> C) 1,2-Dichloroethene (83-84 <sup>o</sup> C) 1,2-Dichloropropene (96.4 <sup>o</sup> C) 1,2-Dichloropropene (108 <sup>o</sup> C) 1,cis-1,3-Dichloropropene (108 <sup>o</sup> C) 1, trans-1,3-Dichloropropene (112 <sup>o</sup> C)	1 - 70. J - 207. K - 208. L - 209. M - 210.	Mexachlorpethane (186.8-187 <sup>o</sup> C) Bis(2-ethylhexyl)phthalate (385 <sup>o</sup> C) Mexachlorpdibenzo-p-dioxins (400-500 <sup>o</sup> C) Mexachlorpdibenzofurans (400-500 <sup>o</sup> C) Pentachlorodibenzofurans (400-500 <sup>o</sup> C) Pentachlorodibenzofurans (400-500 <sup>o</sup> C) Tetrachlorodibenzofurans (400-500 <sup>o</sup> C)	
8 - 22.	1, 1-Dichloroethene (§7-57.3°C)	1 - 70.	Bis(2-ethylhexyl)phthalate (385°C)	
C - 10.	2-Chloro-1,3-butadiene (59.4°C)	J - 207.	Hexachlorpdibenzo-p-dioxins (400-500°C)	
D - 23.	1, 2-Dichloroethene (83-84°C)	K - 208.	Hexachlorpdibenzofurans (400-500°C)	
E - 26.	5, 1,2-Dichloropropene (86.4°C)	L - 209.	Pentachlorodibenzo-p-dioxins (400-500°C)	
F - 28.	5, cis-1,3-Dichloropropene (108°C)	M - 210.	Pentachlorodibenzofurans (400-500°C)	

K - Indicates EPA's belief that the constituent deleted from further consideration for regulation will be adequately controlled by regulation of the indicated constituent selected for regulation.

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# Table 6-6

# BDAT LIST CONSTITUENTS SELECTED FOR REGULATION IN FO24 WASTEWATERS

- 10. 2-Chloro-1,3-butadiene
- 16. 3-Chloropropene
- 1,1-Dichloroethane 22.
- 1,2-Dichloroethane 23.
- 1,2-Dichloropropane 26.
- 27. trans-1,3-Dichloropropene
- cis-1,3-Dichloropropene 28.
- 70. Bis(2-ethylhexyl)phthalate
- 113. Hexachloroethane
- 159. Chromium (total)
- 163. Nickel

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- 207. Hexachlorodibenzo-p-dioxins
- 208. Hexachlorodibenzofurans
- 209. Pentachlorodibenzo-p-dioxins
  210. Pentachlorodibenzofurans
  212. Tetrachlorodibenzofurans

# Table 6-7

## CONTROL OF CONSTITUENTS DELETED FROM FURTHER CONSIDERATION FOR REGULATION BY CONSTITUENTS SELECTED FOR REGULATION IN F024 WASTEWATERS

	BDAT List Constituent Onjeted from Further Consideration for		А		st Const	ituent	Salar	lud for	Reow	lation	(Bood	Nieso		. Frac	
		<u>A</u>	8	<u> </u>	P	E	F	<u> </u>	H	1		<u></u>		M	<u>_N</u>
7.	Carbon tetrachioride (320 kcal/mole)	x	x	×	x	×	×	×	×	×	×	×	ĸ	×	x
5.		X	x	X	X	X	X	×	X	×	×	x	x	X	x
14.	Chloroform (340 kcml/molm)	X	X	X	X	х	x	X	X	x	X	x	x	×	x
38	Nethylene chlaride (360 kcsl/mole)	X	X	X	X	X	X	х	X	x	X	×	X	x	x
42.	Tetrachloroethene (465 kcel/mole)	X	x	X	x	x	x	X	×	x	х	×	X	X	×
47.	Trichlorosthene (485 kcal/mole)	X	X	х	×	X	X	X	х	×	х	×	x	×	x
50.	Vinyl chloride (525 kcal/mole)	X	x	X	X.	×	X	х	ж	x	X	×	X	x	X
137.	Pentachloroethane (585 kcal/mole)		×	X	×	х	×	X	X	X	x	x	x	x	×
40.	, , ,2-Tetrachioroethane (605 kcal/ mole)		×	×	X	×	×	X	×	X	×	×	×	×	×
41.	1,1,2,2-Tetrachloroethane (605 kcal/ mole)		X	×	x	×	X	X	×	×	×	×	×	×	×
45.			x	X	×	x	х	x	X	x	X	x	x	x	x
46.			X	X	ĸ	x	X	X	ж	х	X	ж	х	ж	×
111.								X	X	х	×	×	×	х	x
49.	1,2,3-Trichloropropane (910 kcel/mole)							X	×	x	×	X	X	x	×
222.	Acetone (945 kcal/mole)								×	x	ĸ	x	x	x	×
112.	Hexachlorocyclopentadiene (1,025 kcel/ mole)												×	×	ĸ
20.	trans-1,4-Dichloro-2-butana (1,075 kcal/mole)												x	×	×
34.													x	x	x
68.													×	×	×
86.	1,3-Dichiarabenzene (1,295 kcal/mole)												x	x	ĸ
KEV:								•							
A -	113. Hexachioroethane (585 kcal/mole)		н-		2-Chior						)				
8 -	22. 1,1-Dichloroethane (57-57.3°C)		l -	208.	Herachi										
с-	23. 1,2-Dichloroethane (645 kcal/mole)		J -	210.	Pentach										
D -	27. trans-1,3-Dichlaropropene (790 kcel/mol	• }	K -	212.	Tetrach										
ε-	28. cis-1.3-Dichloropropens (108°C)		L -	207.	Hexachi	0r0010	euso-b.	-diautr	15 (40)	0.500*(	.) 				

F - 16, 3-Chioropropene (810 kca)/mole) G - 26, 1,2-Dichioropropence (930 kca)/mole)

i.

M = 209. Pentachlorodibenzo-p-diaxins (400-500<sup>0</sup>C) N = 70. Bis(2-sthylhexyl)phthalate (385<sup>0</sup>C)

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X - Indicates EPA's belief that the constituent deleted from further consideration for regulation will be adequately controlled by regulation of the indicated constituent selected for regulation.

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# Table 6-7 (Continued)

# CONTROL OF CONSTITUENTS DELETED FROM FURTHER CONSIDERATION FOR REGULATION BY CONSTITUENTS SELECTED FOR REGULATION IN F024 WASTEWATERS

	BDAT list Constituent Dejeted from											_			
	Further Consideration for			BDAT LIS	Const	Ituent	Select	ted for	Regul	ation	(Band	Disso	ciation	Energ	iy)
A	egulation (Bond Dissociation Energy)	<u> </u>	8	<u>c</u>	P	E	F	<u> </u>	н	j	ī.	K	_ L	M	N
87.	1,2-Dichlarobenzene (1,295 kcml/mole)												×	×	×
88.	1,4-Dichlorobenzene (1,295 kcal/mole)												×	х	x
110.	Hexachlorobenzene (1,305 kcel/mole)												x	X	х
136.	Pentachlorobenzene (1,310 kcsl/mole)												×	X	x
150.	1,2,4-Trichlorobenzene (1,320 kcal/												×	×	X
	mole)														
9.	Chlorobenzene (1,330 kcel/mole)												×	X	x
4.	Benzene (1,340 kcal/mole)												×	X	×
126.	Nitrobenzene (1,435 kcal/mple)												×	X	x
43.	Toluene (1,620 kcal/mole)												×	×	×
226.	Ethyl benzene (1,905 kcel/mole)														X
77.	2-Chloronaphthalene (2,115 #c#1/mole)												×	×	x
121.	Naphthalene (2,120 kcal/mole)												x	×	x
141.	• • • • • • •														х
92.	Diethyl phthalate (3,145 kcsl/mole)														×
104.															хd

#### KEV;

A -	113.	Hexachloroethane (565 kcal/mole)		2-Chlorg-1,3-butadione (955 kcal/mole)
8 -	21.	1,1-Dichloroethane (645 kcal/mole)	1 - 208.	Hexachlorodibenzpfurans (960 kcal/mole)
Č -	25	1,2-Dichloroethane (645 kcal/mole)		Pantachlorodibangofurans (980 kcal/mole)
		trans-1,3-Dichloropropene (790 kcal/mole)	K - 212.	Tetrachlorodibenzofurans (1,000 kcal/mole)
		cis-1,3-Dichloropropane (790 kcal/mole)	1 - 207.	Hexachiorodibenzo-p-dioxins (2,470 kcal/mule)
		3-Chloropropene (810 kcsl/mole)	M - 209.	Pentachlorodibenzo-p-dioxins (2,490 kcal/mole)
		1_2-Dichlaropropane (930 kcal/mole)	N - 70.	Bis(2-ethylhexyl)phthalate (6,465 kcal/mole)

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- X Indicates EPA's belief that the constituent deleted from further consideration for regulation will be adequately controlled by regulation of the indicated constituent selected for regulation.
- The 100 kcal/mole difference between di-n-octyl phthalate and bis(2-athylhusyl)phthalate is believed to be within the accuracy of the BDE calculation; therefore, the Agency expects that di-n-octyl phthalate will be controlled by regulation of bis(2-athylhusyl)phthalate.

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# 7.0 CALCULATION OF BDAT TREATMENT STANDARDS

The Agency bases treatment standards for constituents on the performance of well-designed and well-operated BDAT treatment systems. These standards must account for analytical limitations in available treatment performance data, and the data must be adjusted for variability related to treatment, sampling, and analytical techniques and procedures.

BDAT treatment standards are determined for each constituent by multiplying the arithmetic mean of accuracy-adjusted constituent concentrations detected in treated waste by a "variability factor" specific to each constituent in a treatment performance data set. Accuracy adjustment of treatment performance data was discussed in Section 5.0 in relation to defining "substantial treatment." Variability factors correct for normal variations in the performance of a particular technology over time and are designed to reflect the 99th percentile level of performance that the technology achieves in commercial operation. (For more information on the principles of calculating variability factors, see EPA's Methodology for Developing BDAT Treatment Standards (Reference 1).)

In cases where EPA has identified BDAT for a particular waste, but cannot define specific concentration-based treatment standards for that waste because of data limitations or for some other compelling reason, the Agency can require the use of that treatment technology as the BDAT treatment standard. Similarly, where there are no known generators of a waste, or where EPA

believes that the waste can be totally recycled or reused as a raw material, the Agency may specify a "no land disposal" standard, which effectively means establishing a treatment standard of zero for all BDAT List constituents.

In Section 5.0, the best demonstrated available technology for treatment of F024 was selected based on available treatment performance data. In Section 6.0, the regulated constituents were selected to ensure effective treatment of F024. The purpose of Section 7.0 is to calculate treatment standards for each of these constituents using the available treatment performance data from the BDAT treatment technologies. A step-by-step discussion of the calculation of treatment standards for nonwastewater and wastewater forms of F024 is included in this section.

Rotary kiln incineration followed by stabilization of incinerator ash and chemical precipitation followed by vacuum filtration of scrubber water was determined to be BDAT for F024 (Section 5.0). Rotary kiln incineration generally results in the generation of two treatment residuals: ash (a nonwastewater form of F024) and combustion gas scrubber water (a wastewater form of F024). The best measure of performance for a destruction technology such as rotary kiln incineration is the total amount of constituent remaining after treatment. Therefore, BDAT treatment standards for organic constituents were calculated based on total constituent concentration data. BDAT treatment standards for metal constituents in F024 nonwastewater residuals are being reserved by the Agency in the Second-Thirds promulgated rule and will instead be proposed as part of the Third-Thirds proposed rule so that sufficient

time is provided for notice and comment on revisions based on results obtained from the stabilization of F024 incinerator ash. BDAT treatment standards for metal constituents in F024 wasteswater residuals were calculated based on treatment performance data from K062 mixed with other metal-bearing characteristic wastes. BDAT treatment standards for dioxin and furan constituents are set at the analytical detection limit that can be routinely achieved for these constituents, consistent with the dioxins rule promulgated by the Agency on November 8, 1986 (51 Federal Register, 40572, 40638).

# 7.1 Calculation of Treatment Standards for Nonwastewater Forms of F024

# 7.1.1 BDAT List Organics

The treatment standards for nonwastewater forms of F024 were calculated using treatment performance data from rotary kiln incineration of F024. Table 7-1 presents the concentrations of organic constituents in the treatment residual (ash) resulting from rotary kiln incineration of F024. Concentrations are presented for only those constituents used to develop treatment standards for constituents in F024 nonwastewaters, as discussed further in this subsection. The concentration data presented in Table 7-1 have been corrected for accuracy to account for analytical recovery, as described in Section 5.0.

The treatment standards for F024 nonwastewaters were calculated for organic constituents being regulated in F024, as shown in Table 7-2. The following three steps were used to calculate the treatment standards:

# Table 7-1

# CORRECTED CONCENTRATION DATA FOR ORGANIC CONSTITUENTS IN ROTARY KILN INCINERATOR ASH FROM TREATMENT OF F024

		Corrected Concentration <sup>a</sup> in the Treated Waste										
BDAT	List	Total Composition (mg/kg)										
Const	ituent			<u>Sample</u>	Set							
			2		<u> </u>	5	_6					
Volat	liles											
23. 46.	1,2-Dichloroethane 1,1,2-Trichloroethane	0.005 0.005	0.005 0.005	0.005 0.005	0.005 0.005	0.005 0.005	0.005 0.005					
Semiv	olatiles											
70.	Bis(2-ethylhexyl) phthalate	0.632	0.632	0.666	0.632	0.632	0.632					
113.	Hexachloroethane	0.632	0.632	0.666	0.632	0.632	0.632					

<sup>a</sup>Constituent concentrations have been adjusted for accuracy to account for analytical recoveries ("corrected"), as discussed in Section 5.0.

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# Table **7-**2

#### CALCULATION OF TREATMENT STANDARDS FOR F024 NONWASTEWATERS

Regu	i Lated Constituent	F024 Constituent From Which Treatment Performance Date Wore Trensforred	Arithmatic Average of Corrected Treatment Values (ppm)	Variability Factor (VF)	Treatment Standard <sup>a</sup> (Average x VF) (ppm)
Organ (Tota	<u>iics</u> <u>1 Cumposition)</u>				
10.	2-Chloro-1,3-butsdiene <sup>b</sup>	NA	0,101	2.8	0.28
16,	3-Chloropropens <sup>b</sup>	NA	0,101	2.8	D. 28
22.	1,1-Dichlorgethens	1,2-Dichloroethane	0,005	2.8	0,014
23.	1.2-Dichlorgethene	NA	0.005	2.8	0.014
26.	1,2-Dichloropropane <sup>b</sup>	1,1,2-Trichjøroethane	0.005	2.8	0.014
27.	trans-1,3-Dichloropropane <sup>D</sup>	1,1,2-Trichioroethane	0.005	2.6	0.014
28.	cis-1,3-Dichloropropene <sup>b</sup>	1,1,2-Trichtoroethane	0.005	2.8	0.014
70.	Bis(2-ethylhexyl)phthel- ate	NA	0.63	2.8	· 1.8
113.	Hexachloroethane	NA	0.63	2.8	1.8

Note: The treatment standard for BDAT List dioxins and furans in FD24 is 1 ppb. This represents the analytical detection limit that can be routinely achieved for these constituents by laboratories in the United States, consistent with the dioxing rule promulgated by the Agency on November 8, 1986 (51 <u>Federal Register</u>, 40572, 40638).

NA - Not applicable.

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<sup>B</sup>The values shown in this table for treatment standards have been rounded off to show algorificant figures only.

<sup>b</sup>This constituent was found in a liquid form of F024, which was incinerated in the secondary combustor, and therefore did not contribute to the kiln ash residual.

- The arithmetic average of the accuracy-corrected concentrations for each regulated constituent in the untreated waste was calculated using the data presented in Table 7-1.
- (2) Using the same data, a variability factor (discussed in EPA's <u>Methodology for Developing BDAT Treatment Standards</u>) was calculated that represents the variability inherent in the performance of the treatment system, the collection of treated samples, and the analysis of samples. Where concentrations in the treated waste were reported as less than or equal to the detection limit for all the data points in the data set, variability is still expected, since the actual concentrations could range from zero to the detection limit. In these cases, the Agency assumed a lognormal distribution of data points between the detection limit and a value one-tenth of the detection limit and then calculated a variability factor of 2.8.
- (3) The treatment standard for each constituent being regulated was calculated by multiplying the arithmetic average of the accuracy-corrected concentrations (step (1) above) by the variability factor (step (2) above).

As explained in Section 6.0, regulated constituents were selected based on all available F024 characterization data. However, treatment performance data were not available from treatment of F024 for some regulated organic constituents because not all of these regulated constituents were detected in the F024 that was treated by rotary kiln incineration and sampled by EPA. Where treatment performance data were not available for a regulated organic constituent, treatment performance data were transferred to the organic constituent from another organic constituent that was detected in the untreated F024 sampled by EPA, based on the boiling points of the constituents. (Boiling point (bp) is a waste characteristic that affects the performance of rotary kiln incineration, as discussed in EPA's <u>Treatment Technology</u> <u>Background Document</u> (Reference 1). Appendix C of this background document for constituent with the same or the next highest boiling point for which the Agency had treatment performance data from rotary kiln incineration of F024 was selected for transfer of treatment performance data to the constituent for which there was no treatment performance data. Cases where such a transfer of data occurred are summarized below and appear in Table 7-2, which shows the calculations of the treatment standards for F024 nonwastewaters.

Five of the organic constituents (2-chloro-1,3-butadiene; 3-chloropropene; 1,2-dichloropropane; trans-1,3-dichloropropene; and cis-1,3-dichloropropene) were detected in an F024 liquid. Liquid wastes were incinerated in a secondary combustor, and as such did not contribute to the kiln ash residual. Thus, treatment standards for these constituents were calculated based on treatment performance data transferred from constituents detected in a solid form of untreated F024, since solid wastes were incinerated in the rotary kiln and contributed to the kiln ash residual.

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## 2-Chloro-1, 3-butadiene

The treatment standard for 2-chloro-1,3-butadiene (bp 59.4°C) was determined based on the 0.1 ppm detection limit for this constituent. The treatment standard was not based on data transferred from treatment of 1,2-dichloroethane (bp 83-84°C), as may be expected from the methodology described in this section, because the detection limit for 2-chloro-1,3-butadiene is two orders of magnitude greater than the detection limit for 1,2-dichloroethane (0.005 ppm). Calculating a treatment standard based on 1,2-dichloroethane would yield a value of 0.014 ppm, which is below the analytical detection limit for 2-chloro-1,3-butadiene.

#### 3-Chloropropene

The treatment standard for 3-chloropropene (bp  $44-45^{\circ}$ C) was determined based on the 0.1 ppm detection limit for this constituent. The treatment standard was not based on data transferred from treatment of 1,2-dichloroethane (bp 83-84°C), as may be expected from the methodology described in this section, because the detection limit for 3-chloropropene is two orders of magnitude greater than the detection limit for 1,2-dichloroethane (0.005-ppm). Calculating a treatment standard based on 1,2-dichloroethane would yield a value of 0.014 ppm, which is below the analytical detection limit for 3-chloropropene.

#### 1,1-Dichloroethane

',1-Dichloroethane was not found in the F024 that was treated by rotary kiln incineration and sampled by EPA. The treatment standard for 1,1-dichloroethane (bp 57-57.3°C) is based on data transferred from treatment of 1,2-dichloroethane (bp 83-84°C). The Agency expects that 1,1-dichloroethane can be treated to concentrations as low as or lower than 1,2-dichloroethane.

# 1,2-Dichloropropane

The treatment standard for 1,2-dichloropropane (bp 96.4°C) is based on data transferred from treatment of 1,1,2-trichloroethane (bp 113-114°C). The Agency expects that 1,2-dichloropropane can be treated to concentrations as low as or lower than 1,1,2-trichloroethane.

#### trans-1,3-Dichloropropene

The treatment standard for trans-1,3-dichloropropene (bp  $112^{\circ}C$ ) is based on data transferred from treatment of 1,1,2-trichloroethane (bp  $113-114^{\circ}C$ ). The Agency expects that trans-1,3-dichloropropene can be treated to concentrations as low as or lower than 1,1,2-trichloroethane.

# cis-1,3-Dichloropropene

The treatment standard for cis-1,3-dichloropropene (bp  $108^{\circ}$ C) is based on data transferred from treatment of 1,1,2-trichloroethane (bp  $113-114^{\circ}$ C). The Agency expects that cis-1,3-dichloropropene can be treated to concentrations as low as or lower than 1,1,2-trichloroethane.

# 7.1.2 BDAT List Metals

The Agency has recently completed an analysis of TCLP extracts obtained from the stabilization of F024 incinerator ash residues. The results of this analysis show substantial reduction of metals in TCLP extracts following stabilization. Therefore, the Agency has decided to reserve the final treatment standards for metals in F024 nonwastewaters in the Second Thirds promulgated rule. The Agency will instead propose revised treatment standards for metals in F024 nonwastewaters based on the F024 stabilization results as part of the Third Third proposed rule so that sufficient time is provided for notice and comment on these revisions.

#### 7.1.3 BDAT List Dioxins and Furans

The treatment standard for dioxins and furans in F024 nonwastewaters is 1 ppb. This standard represents the analytical detection limit that can be routinely achieved by laboratories in the United States, consistent with the dioxins rule promulgated on November 8, 1986 (51 <u>Federal Register</u>, 40572, 40638).

## 7.2 Calculation of Treatment Standards for Wastewater Forms of F024

#### 7.2.1 BDAT List Organics

The treatment standards for wastewater forms of F024 were calculated using treatment performance data from rotary kiln incineration of F024. Table 7-3 presents the concentrations of organic constituents in the treatment residual (scrubber water) resulting from rotary kiln incineration of F024. Concentrations are presented for only those constituents used to develop treatment standards for constituents in F024 wastewaters, as discussed further in this subsection. The concentration data presented in Table 7-3 have been corrected for accuracy to account for analytical recovery, as described in Section 5.0.

The treatment standards for F024 wastewaters were calculated for organic constituents being regulated in F024, as shown in Table 7-4. The following three steps were used to calculate the treatment standards:

- The arithmetic average of the accuracy-corrected concentrations for each regulated constituent in the untreated waste was calculated using the data presented in Table 7-3.
- (2) Using the same data, a variability factor (discussed in Appendix A of this document) was calculated that represents the variability inherent in the performance of the treatment system, the collection of treated samples, and the analysis of samples. Where concentrations in the treated waste were reported as less than or equal to the detection limit for all the data points in the data set, variability is still expected, since the actual concentrations could range from zero to the detection limit. In these cases, the Agency assumed a lognormal distribution of data points between the detection limit and a value one-tenth of the detection limit and then calculated a variability factor of 2.8.

# Table 7-3

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# CORRECTED CONCENTRATION DATA FOR ORGANICS AND METALS IN COMBUSTION GAS SCRUBBER WATER FROM TREATMENT OF F024

		Corrected Concentration <sup>a</sup> in the Treated Waste									
BDAT	List Lituent		Total	Composit Sample		/1)					
001130		1	2	3	<u> </u>		6				
Volat	iles										
10. 16. 23. 26. 27. 28.	2-Chloro-1-3-butadiene 3-Chloropropene 1,2-Dichloroethane 1,2-Dichloropropane trans-1,3-Dichloropropene cis-1,3-Dichloropropene	0.101 0.101 0.005 0.005 0.005 0.005	0.101 0.101 0.005 0.005 0.005 0.005	0.101 0.101 0.005 0.005 0.005 0.005	0.101 0.005 0.005 0.005 0.005	0.101 0.101 0.005 0.005 0.005 0.005	0.101 0.101 0.005 0.005 0.005 0.005				
<u>Semiv</u>	volatiles										
70. 113.	Bis(2-ethylhexyl) phthalate Hexachloroethane	0.014 0.014	0.014 0.014	0.013 0.013	0.012 0.012	0.013 0.013	0.013 0.013				
Metal	<u>.s</u> b										
159. 163.	Chromium (total) Nickel	0.22 0.39	0.18 0.36	0.26 0.42							

<sup>a</sup>Constituent concentrations have been adjusted for accuracy to account for analytical recoveries ("corrected"), as discussed in Section 5.0.

<sup>b</sup>These data are from the lime and sulfide precipitation followed by vacuum filtration treatment of K062 mixed with other metal-bearing characteristic wastes.

# Table 7-4

#### CALCULATION OF TREATMENT STANDARDS FOR F024 WASTEWATERS

Repu	lated Constituent	F024 Constituent From Which Treatment Performance Data Were Transferred	Arithmetic Average of Corrected Treatment Values (ppm)	Variability Factor (VF)	Treatment Standard <sup>a</sup> (Average × VF) (ppm)
Organ					
(101)	al Composition)				
10.	2-Chloro-1,3-butadiene	NA	0.101	2.8	0.28
16,	3-Chloropropene	NA	0.101	2.8	0.28
22.	1,1-Dichloroethane	1,2-Dichloroethane	0.005	2.8	0.014
23.	1,2-Dichloroethane	NA	0.005	2.8	0.014
26,	1,2-Dichloropropane	MA	0.005	2.8	0.014
27.	trans-1,3-Dichloropropene	NA	0.005	2.8	0.014
28.	cis-1,3-Dichloropropene	ŊA	0.005	2.8	0.014
70.	Bis(2-athylhexyl)phthal- ate	AA	0.013	2.8	0.036
113.	Hexachloroethane	NA	0.013	2.8	0.036
		KD62 Constituent From Which Treatment Performance Data Were	Arithmetic Average of Corrected Treatment Values	Variability Factor	Treatment Standard <sup>a</sup> (Average × VF)
Regu	lated Constituent	Transferred	(ppm)	(VF)	(ppm)
Meta (Tota	ls al Composition)				
159.	Chromium (total)	Chromium (total)	0.221	1.58	0.35
163.	Nickel	Nickel	0.307	1.21	0.47

Note: The treatment standard for BDAT List dioxina and furans in FO24 is 1 ppb. This represents the analytical detection limit that can be routinely achieved for these constituents by laboratories in the United States, consistent with the dioxins rule promulgated by the Agency on November 8, 1986 (51 Federal Register, 40572, 40638).

NA - Not applicable.

<sup>a</sup>The values shown in this table for treatment standards have been rounded off to show significant figures only.

(3) The treatment standard for each constituent being regulated was calculated by multiplying the arithmetic average of the accuracy-corrected concentrations (step (1) above) by the variability factor (step (2) above).

As discussed in Section 6.0, regulated constituents were selected based on all available F024 characterization data. However, treatment performance data were not available from treatment of F024 for some regulated organic constituents because not all of the regulated constituents were detected in the F024 that was treated by rotary kiln incineration and sampled by EPA. Where treatment performance data were not available for a regulated organic constituent, treatment performance data were transferred to the organic constituent from another organic constituent that was detected in the untreated F024 sampled by EPA, based on the bond dissociation energies of the constituents. (Bond dissociation energy (BDE) is a waste characteristic that affects the performance of rotary kiln incineration, as discussed in EPA's Treatment Technology Background Document (Reference 1). Appendix C of this background document for F024 presents detailed information on this waste characteristic.) The constituent with the same or the closest bond dissociation energy for which the Agency had treatment data from rotary kiln incineration of F024 was selected for transfer of treatment performance data to the constituent for which there was no treatment performance data. The case in which such a transfer of data occurred is summarized below and appears in Table 7-4. which shows the calculations of the treatment standards for FO24 wastewaters.

## 1,1-Dichloroethane

1,1-Dichloroethane was not found in the F024 that was treated by rotary kiln incineration and sampled by EPA. The treatment standard for 1,1-dichloroethane (BDE 645 kcal/mole) is based on data transferred from treatment of 1,2-dichloroethane (BDE 645 kcal/mole). The Agency expects that 1,1-dichloroethane can be treated to concentrations as low as or lower than 1,2-dichloroethane.

## 7.2.2 BDAT List Metals

The Agency has no treatment performance data for metals in F024 wastewaters. The treatment standards for metal constituents in F024 wastewaters were based on treatment performance data transferred from the lime and sulfide precipitation followed by vacuum filtration of K062 mixed with other metal-bearing characteristic wastes. The Agency believes that wastewater residuals from K062 mixed with other metal-bearing characteristic wastes are sufficiently similar to F024 wastewater residuals such that treatment performance data can be transferred. Treatment performance data for each metal constituent being regulated in F024 wastewaters were transferred from K062 mixed with other metal-bearing characteristic wastes to F024.

The concentrations of metal constituents in the wastewater residual following the lime and sulfide precipitation and vacuum filtration of K062 mixed with other metal-bearing characteristic wastes are presented in Table

7-3. These concentration data have been corrected for accuracy to account for analytical recovery, as described in Section 5.0.

The F024 wastewater treatment standards were calculated for the regulated metal constituents, as shown in Table 7-4. These calculations are consistent with the methodology previously described in this section for BDAT List organic constituents.

# 7.2.3 BDAT List Dioxins and Furans

The treatment standard for dioxins and furans in F024 wastewaters is 1 ppb. This standard represents the analytical detection limit that can be routinely achieved by laboratories in the United States, consistent with the dioxins rule promulgated on November 8, 1986 (51 <u>Federal\_Register</u>, 40572, 40638).

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APPENDIX A

ANALYTICAL QUALITY ASSURANCE/QUALITY CONTROL

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# APPENDIX A

# ANALYTICAL QA/QC

The analytical methods used for analysis of the constituents being regulated (identified in Section 6.0) are presented in Table A-1. SW-846 Methods (EPA's <u>Test Methods for Evaluating Solid Waste: Physical/Chemical Methods, SW-846</u> (Reference 16)) were used in most cases for determining total constituent concentrations.

In some instances, it was necessary to deviate from the SW-846 Methods. Deviations from SW-846 Methods required to analyze the sample matrix are listed in Table A-2. SW-846 allows for the use of alternative or equivalent procedures or equipment; these alternatives are noted in Tables A-3 through A-7.

The accuracy determination for a constituent is based on the matrix spike recovery values. Tables A-8 and A-9 present the matrix spike recovery data for BDAT List constituents in the kiln-ash and scrubber water residuals, respectively.

The accuracy correction factors for BDAT List constituents detected in untreated F024 and in the kiln ash and scrubber water residuals are summarized in Table A-10. The accuracy correction factors were determined for each constituent by dividing 100 by the matrix spike recovery (expressed as a percentage) for that constituent.

A-2

# Table A-1

## ANALYTICAL METHODS FOR CONSTITUENTS SELECTED FOR REGULATION IN F024

		Cons	tituent Concentration in Kiin Ash		Constituent Concentration						
Reg	ulated Constituent	Preparation Method	Analytical Method	Reference	Preparation Method	Analytical Method	Reference				
VOLAT	ILES										
10. 16. 22. 23. 26. 27. 28.	2-Chloro-1,3-butadiene 3-Chloropropene 1,1-Dichloroethane 1,2-Dichloroethane 1,2-Dichloropropene trans-1,3-Dichloropro- pene cis-1,3-Dichloropropene	Purge and Trap (Method 5030)	Gas Chromatography/ Mass Spectrometry for Vplatile Organics (Method 8240)	١	Purge and Trap (Method 5030)	Gas Chromatography/ Mass Spectrometry for Volatile Organics (Method 8240)	ı				
SEMIN	OLATILES										
70. 113.	Bis(2-ethylhexyl)- phthalate Hexachloroethane		Column Technique (Method 8270)			Column Technique (Method 8270)					
METAL	<u>s</u>										
159. 163.	Chromium (total) Nickel	Acid Digestion of Sediments, Sludges, and Soils (Method '3050)	Inductively Coupled Plasma Atomic Emission Spectroscopy (Method 6010)	1	Acid Digestion of Liquids (Method 3010/3020)	Inductively Coupled Plasma Atomic Emission Spectroscopy (Method 6010)	١				

NA - Not applicable.

Source: Test Methods for Evaluating Solid Waste, SW-846 Third Edition (Reference 16)

A-3

#### Table A-1 (Continued)

#### ANALYTICAL METHODS FOR CONSTITUENTS REGULATED IN F024

		Constituent Concentration			Constituent Concentration in Scrubber Water			
Reg	ulated Constituent	Preparation Method	Analytical Method	Reference	Preparation Method	Analytical <u>Method</u>	Reference	
DIOX	NS/FURANS							
207.	Hexachlorodibenzo-p- dioxins	NA	Analysis of Poly- chlorinated Dibanzo-	1	NA	Analysis of Poly- chlorinated Dibenxo-	1	
208.	Hexachlorodibenzofurans		p-dioxins and Poly-			p-dioxins and Poly-		
209.	Pentachlorodibenzo- p-dioxins		chlorinated Dibenzo- furans (Method 8280)			chlorinated Dibenzo- furans (Method 8280)		
210.	Pentachlorodibenzofurans							
212,	Tetrachiorodibenzofurans							

NA - Not applicable.

Source: Test Methods for Evaluating Solid Waste, SW-846 Third Edition (Reference 16)

# DEVIATIONS FROM SW-846 IN ANALYSIS OF F024

Analysis Metho	SW-846 Specifications	Deviation from SW-846	Rationale for Deviation
Acid digestion for 3010 metals analysis 3020	Digest 100 ml of sample in a conical beaker.	Initial sample volume of 50 ml was digested in Griffin straight-side beakers. All acids and peroxides were halved.	Sample volume and reagents were reduced by half; therefore, time required to reduce sample to near dryness was reduced. However, this procedure produced no impact on the precision and accuracy of the data.

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#### SPECIFIC PROCEDURES USED IN EXTRACTION OF ORGANIC COMPOUNDS WHEN ALTERNATIVES TO SW-846 METHODS ARE ALLOWED BY APPROVAL OF EPA CHARACTERIZATION AND ASSESSMENT DIVISION

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Ana tys is	SW-846 method	Sample aliquot	SW-846 specification	Specific procedures allowed by approval of EPA-CAD
Continuous liquid- liquid extraction	3520	1 liter	<ul> <li>The internal standards are prepared by dissolution in carbon disulfide and then dilution to such volume that the final solvent is 20% carbon disulfide and 80% methylene chloride.</li> </ul>	<ul> <li>The preparation of the internal standards is changed to eliminate the use of carbon disulfide. The internal standards are prepared in methylene chloride only.</li> </ul>
Soxhlet extraction	3540	l gram	<ul> <li>The internal standards are prepared by dissolution in carbon disulfide and then dilution to such volume that the final solvent is 20% carbon disulfide and 80% methylene chloride.</li> </ul>	<ul> <li>The preparation of the internal standards is changed to eliminate the use of carbon disultide. The internal standards are prepared in methylene chloride only.</li> </ul>

# SPECIFIC PROCEDURES OR EQUIPMENT USED IN EXTRACTION OF ORGANIC COMPOUNDS WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

Ana lys is	SW-846 method	Sample aliquot	Alternatives or equivalents allowed by SW-846 methods	Specific procedures or equipment used
Purge-and-trap	5030	5 moulliliters of liquid, 1 gram of solid	<ul> <li>The purge-and-trap device to be used is specified in Figure 1 of the method. The desorber to be used is described in Figures 2 and 3, and the packing materials are described in Section 4.10.2 of SW-846. The method allows equivalents of this equipment or materials to be used.</li> </ul>	<ul> <li>The purge-and-trap equipment and the desorber used are as specified in SW-846. The purge-and-trap equipment is a Teckmar LSC-2 with standard purging chambers?(Supelco cat. 2-0293) The packing materials for the traps are 1/3 silica gel and 2/3 2,6-diphenylene.</li> </ul>
			<ul> <li>The method specifies that the trap must be at least 25 cm long and have an inside diameter of at least 0.105 cm.</li> </ul>	<ul> <li>The length of the trap is 30 cm and the diameter is 0.105 cm.</li> </ul>
			<ul> <li>The surrogates recommended are toluene-d8.4 bromofluorobenzene, and 1.2-dichloroethane-d4. The recommended concentration level is 50 μg/1.</li> </ul>	<ul> <li>The surrogates are added as specified in SW-846.</li> </ul>
Soxhiet extraction	3540	) gram of solid	<ul> <li>The recommended surrogates and their concentrations are the same as for Method 3520.</li> </ul>	<ul> <li>The surrogates used and their concentration levels are the same as for Method 3520</li> </ul>
			<ul> <li>Sample grinding may be required for sample not passing through a 1-mm standard sieve or a 1-mm opening</li> </ul>	<ul> <li>Sample granding is not required</li> </ul>

### Table A-4 (Continued)

Ana lys is	SW-846 method	Sample aliquot	Alternatives or equivalents allowed by SW-846 methods	Specific procedures or equipment used
Continuous liquid- liquid extraction	3520	1 liter of liquid	<ul> <li>Acid and base/neutral extracts are usually combined before analysis by GC/MS. Under some situations, however, they may be extracted and analyzed separately</li> </ul>	<ul> <li>Acid and base/neutral extracts are combined</li> <li>.</li> <li>.</li> </ul>
			<ul> <li>The base/neutral surrogates recommended are 2-fluorobiphenyl, nitrobenzene-d5, and terphenyl-d14. The acid surrogates recommended are 2-fluorophenol,</li> <li>2,4,6-tribromophenol, and phenol-d6. Additional compounds may be used for surrogates. The recommended concentrations for low-medium concentration level samples are 100 ppm for acid surrogates and 200 ppm for base/ neutral surrogates. Volume of surrogate may be adjusted.</li> </ul>	<ul> <li>Surrogates are the same as those recommended by SM-846, with the exception that phenol-d5 is substituted for phenol-d6. The concentrations used are the concentrations recommended in SM-84</li> </ul>

#### SPECIFIC PROCEDURES OR EQUIPMENT USED IN EXTRACTION OF ORGANIC COMPOUNDS WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

# SPECIFIC PROCEDURES OR EQUIPMENT USED FOR ANALYSIS OF ORGANIC COMPOUNDS WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

Ana lysis	SW 846 method	Sample preparation method	Alternatives of allowed in equipment or	SV 846 for	Specific equipment or procedures used			
Gas chromatography/ mass spectrometry	8240	5030	Recommended GC/NS operating	conditions:	Actual GC/MS operating condi	tions -		
for volatile			Electron energy:	70 ev (nominal)	Electron energy:	70 ev		
organics			Mass range:	35-260 amu	Mass range:	35-260 amu		
			Scan time:	To give 5 scans/peak but not to exceed 7 sec/scan	Scan time:	2.5 sec/scan		
Ą			Initial column temperature:	45°C	Initial column temperature.	38°C		
A-9			Initial column holding time:	3 min	Initial column holding time:	2 min		
			Column temperature program:	8°C/min	Column temperature program	10°C/min		
			Final column temperature:	200-C	Final column temperature:	225°C		
			Final column holding time.	15 min	Final column holding time	30 min or xylene elutes		
			Injector temperature:	200 225°C	Injector temperature:	225°C		
			Source temperature:	According to manufacturer's specification	Source temperature:	manufacturer's recommended value of 100°C		
			Transfer line temperature:	250-300°C	Transfer line temperature:	275*0		
			Carrier gas:	Hydrogen at 50 cm/sector helium at 30 cm/sec	Carrier gas:	Helium at 30 ml/min		
			The column should be 6 ft x packed with 1% SP-1000 on Ca an equivalent	•	The column used is an 8 ft x with 1% SP-1000 on Carbopack			
			Samples may be analyzed by p or by direct injection	burge-and-trap technique	The samples are analyzed usi technique.	ng the purge and trap		
					Additional information on ac	•		
					Equipment. Finneyan model 5	•		
					Data system — SHPERINCOS Aut Modu — Electrop import	ndnau		
					Mode flectron impact NBS libitary available			
					nds thraty available			

#### Table A-5 (Continued)

#### SPECIFIC PROCEDURES OR EQUIPMENT USED FOR ANALYSIS OF ORGANIC COMPOUNDS WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

Analysis	SW-846 method	Sample preparation method	Alternatives or allowed in S equipment or i	W-846 for	Specific equipment or procedures used		
			Recommended GC/HS operating co	unditions.	Actual GC/MS operating condition	ons :	
Gas chromatography/	8270	3520-11quids	Mass range:	35-500 amu	Mass range.	35-500 amu	
mass spectrometry		3540-solids	Scan time:	) sec/scan	Scan time.	l sec/scan	
for semivolatile			Initial column temperature	40°C	Initial column temperature.	30°C	
organics: capillary			Initial column holding time.	4 ตาก	Initial column holding time:	4 mid	
alumn technique			Column temperature program:	40 270°C at	Column temperature program:	8°C/min to 275°	
				10°C/min		and 10°C/min until	
			Final column temperature hold:	270°C (unti)		305°C	
				benzo[g,h,1,]perylene has	Final column temperature hold.	305°C	
				e luted)	Injector temperature.	240-260°C	
			Injector temperature:	250-300°C	Transfer line temperature	30 <b>0°C</b>	
			Transfer line temperature.	250-300°C	Source temperature.	Manufacturer's	
			Source temperature:	According to		recommendation	
				manufacturer's		(nonheated)	
				specification	Injector	Grob-type, splitless	
			Injector:	Grob-type, splitless	Sample volume:	<b>ΙμΙ of sample extract</b>	
			Sample volume:	1-2 µ1	Carrier gas	Helium at 40 cm/sec	
			Carrier gas.	Hydrogen at 50 cm/sec or			
				helium at 30 cm/sec			
			The column should be 30 m by 0	125 mm 10.,1μm 111m	The column used is a 30 m x 0 3	)2.mm ).D.	
			thickness silicon-coated fused (J&W Scientific DB-5 or equiva	Esilica capillary column	RT <sub>K</sub> -5 (5% phenyl methyl silico	one) FSCC.	
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Additional information on actual system used Equipment - Finnegan model 5100 GC/MS/DS system Software Package - SUPERINCOS Autoquan

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SPECIFIC PROCEDURES OR EQUIPMENT USED IN PREPARATION AND ANALYSIS OF METALS WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

Analysis	SW-846 method	_		Alternative or equivalent allowed by SW-846 methods		Specific procedures or equipment used
Inductively coupled plasma atomic emission spectroscopy	6010	Jarrell Ash 1140	•	Operate equipment following instructions provided by instrument's manufacturer.	•	Equipment is operated using procedures specified in the Jarrell Ash (JA) 1140 Operator's Manual.
			٠	For operation with organic solvents, auxiliary argon gas inlet is recommended	٠	Auxiliary argon gas is not required for sample matrix

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# SPECIFIC PROCEDURES OR EQUIPMENT USED FOR ANALYSIS OF CYANIDE AND SULFIDE WHEN ALTERNATIVES OR EQUIVALENTS ARE ALLOWED IN THE SW-846 METHODS

Ana iysıs	Su-d46 method	Samp'e aliquot	Alternatives or equivalent allowed by SW-846 methods	Specific procedures used
lotal and amenable cyarice	9012	500 m)	<ul> <li>Hydrogen sulfide treatment may be required.</li> </ul>	<ul> <li>Hydrogen sulfide treatment is not required.</li> </ul>
			<ul> <li>A Fisher-Mulligan absorber or equivalent should be used.</li> </ul>	<ul> <li>A Wheaton Distilling Apparatus absorber is used</li> </ul>
iu Ifide	9030	200 m]	<ul> <li>An aqueous starch solution or a soluble starch powder may be used.</li> </ul>	<ul> <li>An aqueous starch solution is used.</li> </ul>
			<ul> <li>The titrant used may be either sodium thiosulfate or phenylarsine oxide.</li> </ul>	<ul> <li>The titrant used is sodium thicsu<sup>*f</sup>ate.</li> </ul>
			<ul> <li>Sample pretreatment may be required.</li> </ul>	<ul> <li>For pretreatment of an aqueous sample, zinc acetate is added to precipitate the sulfice and the zinc sulfice precipitate is filtered and analyzed.</li> </ul>

#### MATRIX SPIKE RECOVERIES FOR KILN ASH RESIDUE

				Sample Result		Dupli	cate Sample	Result
	<u>Spike Constituent</u>	Origina) Amount Found <sup>a</sup> (ppm)	Amount Spiked (ppm)	Amount Recovered (ppm)	Percent Recoveryb (%)	Amount Spiked (ppm)	Ainounit Recovered (ppm)	Percent Recovery <sup>t</sup> (%)
VOLAT	ILES							
4.	Benzene	DL	0.050	0.063	126	0,050	0,065	130
9.	Chlorobenzene	DL	0.050	0.045	90	0.050	0.054	108
24.	1,1-Dichloroethene	DL	0.050	0.040	80	0.050	0.061	122
43.	Toluéne	DL	0.050	0.085	170	0.050	0.111	222
47.	Trichlargethene	DL	0.050	D.089	178	0.050	0.108	216
VERA	GE RECOVERY FOR VOLATILES				128.8			159.6
SEMIV	OLATILES (BASE/NEUTRAL FRACTION	)						
52.	Acenaphthene	DL	0.050	0.015	30	0.050	0.044	88
68.	1,4-Dichlorobenzene	DL	0.050	0.036	72	0.050	0.045	90
02.	2,4-Dinitrotoluene	DL	0.050	0.019	38	0,050	0.050	100
05.	N-Nitroso-di-n-propylamine	DL	0.050	0.055	110	0.050	0.059	118
45.	Pyrene	DL	0.050	0.003	6	0.050	0.041	82
150.	1,2,4-Trichlorobenzene	DL	0.050	0.030	60	0.050	0.045	90
	GE RECOVERY FOR SEMIVOLATILES /NEUTRAL FRACTION)				52.7			94.7
INORG	ANICS							
169.	Cyanide	DL	4.97	3.93	79	5.00	6.25	125
170.	Fluoride	2.11	4.91	7.04	100	4.99	1.44	107
171.	Sulfide	DL	25	27.4	110	24.6	26.1	106
AVERA	GE RECOVERY FOR INORGANICS				96.3			112.7

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DL - Detection limit. BFor constituents not detected at or above the detection limit (DL), the original amount found is considered as zero in calculating percent recovery. Percent recovery = 100 x  $(C_1 - C_0)/C_t$ , where  $C_1 = amount recovered$ ,  $C_0 = original amount found, and <math>C_t = amount spiked$ .

#### Table A-8 (Continued)

#### MATRIX SPIKE RECOVERIES FOR KILN ASH RESIDUE

			Sample Result		Dupl	Duplicate Sample Result			
Spike Constituent	Original Amount Found <sup>a</sup> (ppm)	Amount Spiked (ppm)	Amount Recovered (ppm)	Percent Recovery <sup>b</sup> (%)	Amount Spiked (ppm)	Amount Recovered (ppm)	Percent Recovery <sup>b</sup> (%)		
DIOXINS/FURANS									
1,2,3,4,7,8-Hexachiorodibenzo- p-diuxin	DL		0.010	101		0.010	104		
1,2,3,4,7,8-Hexachterodibenze- furan	DL		0.0096	97		0.0098	98		
1,2,3,7,8-Pentachlorodibenzo- p-dioxin	DL		0.010	102		0.010	102		
1,2,3,7,8-Pentachlorodibenzo- furan	DL		0.0086	87		0.0086	86		
2,3,7,8-Tetrachlorodibenzo- p-dioxin	OL		0.0099	100		0.0097	97		
2,3,7,B-Tetrachlorodibenzofuran	DL		0.011	107		0.011	108		
AVERAGE RECOVERY FOR DIOXINS/FURANS				99			99		

DL - Detection limit.

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<sup>a</sup>For constituents not detected at or above the detection limit (DL), the original amount found is considered as zero in calculating percent recovery. Percent recovery = 100 x  $(C_i \cdot C_o)/C_t$ , where  $C_i$  = amount recovered,  $C_o$  = original amount found, and  $C_t$  = amount spiked.

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# MATRIX SPIKE RECOVERIES FOR COMBUSTION GAS SCRUBBER WATER

				Sample Result		Ոսը Լ	icate Sample	Result
	Spike Constituent	Origina) Amount Found <sup>®</sup> (ppm)	Amount Spiked (ppm)	Amount Recovered (ppm)	Percent Recovery <sup>D</sup> (%)	Amount Spiked (ppm)	Anount Recovered <u>(ppm)</u>	Percent Recovery <sup>D</sup>
VOLAT	TLES							
4.	Benzana	DL	0.050	0.043	86	0.050	0.045	90
9.	Chlorobenzene	DI	0.050	0.055	110	0 050	0.056	112
24.	1,1-Dichloroethene	DL	0.050	0.040	80	0.050	0.040	60
43.	Toluene	DL	0.050	0.057	114	0 050	0.057	114
47.	Trichlaroethene	DL	0.050	0.053	106	0.050	0.054	108
AVERA	NGE RECOVERY FOR VOLATILES				99.2			100.8
SEMIN	DLATILES (BASE/NEUTRAL FRACTION	)						
52.	Acanaphthana	DL	0.050	0.039	78	0.050	0.032	64
68.	1,4-Dichlorobenzene	DL	0.050	0.032	64	0 050	0.029	58
102.	2.4-Dinitratoluene	DL	0.050	0.057	114	0.050	0.057	114
105	N-Nitroso di-n-propylamine	PL	0.050	0.066	132	0.050	0.058	116
145	Pyrana	DL	0.050	0.044	66	0.050	0.041	. 82
150.	1,2,4-Trichlorobenzene	ÐL	D.050	0.031	62	0.050	0.032	64
	NGE RFCOVFRY FOR SEMIVOLATILES E/NEUTRAL FRACTION)				89.7			85.7
METAL	<u>s</u> c							
159.	Chromium (total)	DL	0.050	0.035	70	0.050	0.034	68
161.	Lead	DL	0.025	0.022	88	0.025	0.019	76
168.		2.64	10	12.6	108	10	12.4	98
AVER	AGE RECOVERY FOR METALS				86			80.7
INOR	SANICS							
169.	Cyanide	DL	0.100	0.014	14	0.100	0.022	22
170.	fluoride	153	160	330	111	160	336	114
171.	Sulfide	ÐL	5.0	4.0	80	5.0	3.2	64
AVER	AGE RECOVERY FOR INORGANICS			68.3				66.7

DL - Detection limit.

NA - Nut analyzed.

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<sup>a</sup>For constituents not detected at or above the detection limit (DL), the original amount found was considered as zero in calculating

percent recovery. Percent recovery = 100 x  $(C_1-C_0)/C_1$ , where  $C_1 = amount recovered, C_0 = original amount found, and <math>C_1 = amount spiked.$ <sup>C</sup>Source: Waterways Onsite Engineering Report for F024 (Reference 21).

#### Table A-9 (Continued)

#### MATRIX SPIKE RECOVERIES FOR COMBUSTION GAS SCRUBBER WATER

		Sample Result		Duplicate Sample Result	
<u>Spike Constituent</u>	Original Amount Found <sup>a</sup> (ppm)	Amount Recovered (ppm)	Percent Recovery <sup>b</sup> (%)	Amount Recovered (ppm)	Percent Recovery <sup>11</sup> (%)
DIOXINS/FURANS					
1,2,3,4,7,8-Hexachiorodibenzo-p-dioxin	DL	0.0001	99	0.0001	101
1, 2, 3, 4, 7, B-Hesachlorudibenzofuran	0.0003	0.0010	101	0.0010	105
1,2,3,7,8-Pentechlorodibenzo-p-dioxin	<b>DL</b>	0.0010	104	0.0010	102
1,2,3,7,8-Pentachlorodibenzofuran	0.0001	0.0009	89	0.0009	87
2,3,7,8-Tetrachlorodibenzo-p-dioxin	DL	. 0.009	94	0.0009	95
2,3,7,8-Tetrachlorodibenzofuran	PL	0.0010	104	0.0011	106
AVERAGE RECOVERY FOR DIDXINS/FURANS			99		99

DL - Detection limit.

NA - Not analyzed.

"A For constituents not detected at or above the detection limit (DL), the original amount found was considered as zero in calculating percent recovery = 100 x  $(C_1-C_0)/C_1$ , where  $C_1$  = amount recovered,  $C_0$  - original amount found, and  $C_1$  = amount spiked.

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#### SUMMARY OF ACCURACY CORRECTION FACTORS

		Accuracy Correction Factor <sup>a</sup>		
		Kiln Ash	Scrubber Water	
BDAT	List Constituent	Total Composition	Total Composition	
222.	Acetone	1.000	1.008	
6.	Bromomethane	NA	1.008	
10.	2-Chloro-1,3-butadiene	1,000	1.008	
12.	Chloroethane	NA	1.008	
	Chloromethane	NA	1.008	
16.	3-Chloropropene	1.000	1.008	
20.	trans-1,4-Dichloro-2-butene	1.000	1.008	
21.	Dichlorodifluoromethane	1.000	1.008	
23.	1,2-Dichloroethane	1.000	1.008	
26.	1,2-Dichloropropane	1.000	1,008	
27.	trans-1,3-Dichloropropene	1.000	1.008	
28.	cis-1,3-Dichloropropene	1.000	1.008	
34.	Methyl ethyl ketone	1.000	1.008	
42.	Tetrachloroethene	1.000	1.008	
46.	1,1,2-Trichloroethane	1.000	1.008	
49.	1,2,3-Trichloropropane	1.000	1.008	
47. 59.	Benz(a)anthracene	1.899	1.167	
62.	Benzo(a)pyrene	1.899	1.167	
63.	Benzo(b)fluoranthene	1.899	1.167	
64.	Benzo(ghi)perylene	1.899	1.167	
65.	Benzo(k)fluoranthene	1.899	1.167	
68.		1.899	1.167	
70.	Bis(2-ethylhexyl)phthalate	1.899	1.167	
80.	Chrysene	1.899	1.167	
88.	1,4-Dichlorobenzene	1.389	1.724	
92.	Diethyl phthalate	1.899	1.167	
110.	Hexachlorobenzene	1.899	1.167	
	Hexachloroethane	1.899	1.167	
113. 116.		1.899	1.167	
141.	Indeno(1,2,3-cd)pyrene Phenanthrene	1.899	1.167	
		1.266		
	Cyanide Elucrido		7.143	
	Fluoride	1.000	1.000	
	Sulfide	1.000	1.562	
201.	Hexachlorodibenzo-p-dioxins	1.000	1.010	

NA - Not applicable; this constituent was not detected in either the untreated F024 or the kiln ash residual.

<sup>a</sup>The accuracy correction factor is equal to 1 divided by the percent recovery correction factor.

#### Table A-10 (Continued)

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#### SUMMARY OF ACCURACY CORRECTION FACTORS

		rection Factor <sup>a</sup>	
	·	Kiln Ash	Scrubber Water
BDAT List Constituent		<u>Total Composition</u>	Total Composition
208.	Hexachlorodibenzofurans	1.031	1.000
209.	Pentachlorodibenzo-p-dioxins	1.000	1.000
210.	Pentachlorodibenzofurans	1.163	1.149
212.	Tetrachlorodibenzofurans	1.000	1.000
			Scrubber Water Total Composition <sup>b</sup>
159. 163.	Chromium (total) Nickel		1.47 1.08

<sup>a</sup>The accuracy correction factor is equal to 1 divided by the percent recovery correction factor. <sup>b</sup>These data are from the Onsite Engineering Report for KO61 (Reference 35). Accuracy correction factors are listed only for those constituents regulated in F024 wastewater.

APPENDIX B

PLANT CODES

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# Table B-1

# PLANT CODES

Code	Plant	Location	References
А	DuPont Chemical Company	LaPlace, Louisiana	27, 32
В	Shell Chemical Company	Norco, Louisiana	28, 32
С	Vista Chemical Company	Lake Charles, Louisiana	29, 32
D	Vulcan Chemical Company	Wichita, Kansas	30, 32
	Velsicol Chemical Company	Memphis, Tennessee	31
E F	LCP Chemicals	Moundsville, West Virginia	3
G	Dow Chemical Company	Freeport, Texas	4
н	PPG Industries	Lake Charles, Louisiana	5
I	Formosa Plastics	Baton Rouge, Louisiana	6
J	Borden Chemical	Geismar, Louisiana	7
К	Vulcan Materials Company	Geismar, Louisiana	8
L	PPG Industries	Pittsburgh, Pennsylvania	9
М	Dow Chemical Company	Plaquemine, Louisiana	10
N	Shell Oil Company	Not Specified	11
0	FMC Corporation	Baltimore, Maryland	12
Р	Denka Chemical Corporation	Houston, Texas	13
х	ENSCO	El Dorado, Arkansas	32
Y	Waterways Experiment Station	Vicksburg, Mississippi	21
Z	Envirite Corporation	York, Pennsylvania	33

APPENDIX C

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WASTE CHARACTERISTICS AFFECTING PERFORMANCE